

Gas-Phase Ion and Neutral Thermochemistry

Journal of

Physical and Chemical Reference Data

David R. Lide, Jr., Editor

The *Journal of Physical and Chemical Reference Data* (ISSN 0047-2689) is published quarterly by the American Chemical Society (1155 16th St., N. W., Washington, DC 20036-9976) and the American Institute of Physics (335 E. 45th St., New York, NY 10017-3483) for the National Bureau of Standards. Second-class postage paid at Washington, DC and additional mailing offices POSTMASTER: Send address changes to *Journal of Physical and Chemical Reference Data*, Membership and Subscription Services, P. O. Box 3337, Columbus, Ohio 43210

The objective of the *Journal* is to provide critically evaluated physical and chemical property data, fully documented as to the original sources and the criteria used for evaluation. Critical reviews of measurement techniques, whose aim is to assess the accuracy of available data in a given technical area, are also included. The *Journal* is not intended as a publication outlet for original experimental measurements such as those that are normally reported in the primary research literature, nor for review articles of a descriptive or primarily theoretical nature.

Supplements to the *Journal* are published at irregular intervals and are not included in subscriptions to the *Journal*. They contain compilations which are too lengthy for a journal format.

The Editor welcomes appropriate manuscripts for consideration by the Editorial Board. Potential contributors who are interested in preparing a compilation are invited to submit an outline of the nature and scope of the proposed compilation, with criteria for evaluation of the data and other pertinent factors, to:

David R. Lide, Jr., Editor
J. Phys. Chem. Ref. Data
National Bureau of Standards
Gaithersburg, MD 20899

One source of contributions to the *Journal* is The National Standard Reference Data System (NSRDS), which was established in 1963 as a means of coordinating on a national scale the production and dissemination of critically evaluated reference data in the physical sciences. Under the Standard Reference Data Act (Public Law 90-396) the National Bureau of Standards of the U.S. Department of Commerce has the primary responsibility in the Federal Government for providing reliable scientific and technical reference data. The Office of Standard Reference Data of NBS coordinates a complex of data evaluation centers, located in university, industrial, and other Government laboratories as well as within the National Bureau of Standards, which are engaged in the compilation and critical evaluation of numerical data on physical and chemical properties retrieved from the world scientific literature. The participants in this NBS-sponsored program, together with similar groups under private or other Government support which are pursuing the same ends, comprise the National Standard Reference Data System.

The primary focus of the NSRDS is on well-defined physical and chemical properties of well-characterized materials or systems. An effort is made to assess the accuracy of data reported in the primary research literature and to prepare compilations of critically evaluated data which will serve as reliable and convenient reference sources for the scientific and technical community.

Information for Contributors

Manuscripts submitted for publication must be prepared in accordance with *Instructions for Preparation of Manuscripts for the Journal of Physical and Chemical Reference Data*, available on request from the Editor.

Editorial Board

Term ending 31 December 1988

William A. Goddard III, Ronald A. Phaneuf, Alfons Weber

Term ending 31 December 1989

Mostafa A. El-Sayed, Glen A. Slack, Barry N. Taylor

Term ending 31 December 1990

Sidney C. Abrahams, Carlos M. Bowman, Malcolm W. Chase, Jr.

Management Board

David R. Lide, Jr., Charles R. Bertsch, John T. Scott

Editorial Staff at NBS: Julian M. Ives, Joan Sauerwein

Editorial Staff at AIP: Kathleen Strum, Managing Editor;

Susan A. Walsh, Chief Copy Editor; Thomas J. Buckley, Copy Editor

New and renewal subscriptions should be sent with payment to the Office of the Controller at the American Chemical Society, 1155 Sixteenth Street, N.W., Washington, DC 20036-9976. Address changes, with at least six weeks advance notice, should be sent to *Journal of Physical and Chemical Reference Data*, Membership and Subscription Services, American Chemical Society, P.O. Box 3337, Columbus, OH 43210. Changes of address must include both old and new addresses and ZIP codes and, if possible, the address label from the mailing wrapper of a recent issue. Claims for missing numbers will not be allowed: if loss was due to failure of the change-of-address notice to be received in the time specified; if claim is dated (a) North America, more than 90 days beyond issue date, (b) all other foreign, more than one year beyond issue date.

Members of AIP member and affiliate societies requesting member subscription rates should direct subscriptions, renewals, and address changes to American Institute of Physics, Dept S/F, 335 E 45th St., NY 10017-3483.

Subscription Prices (1988) (not including supplements)

| | U.S.A. | Foreign (surface mail) | Optional air freight Europe Mideast N. Africa | Asia and Oceania |
|---|----------|------------------------------|--|---------------------|
| Members (of ACS, AIP, or affiliated society) | \$ 60.00 | \$ 70.00 | \$ 80.00 | \$ 80.00 |
| Regular rate | \$265.00 | \$275.00 | \$285.00 | \$285.00 |

Rates above do not apply to nonmember subscribers in Japan, who must enter subscription orders with Maruzen Company Ltd., 3-10 Nihonbashi 2-chome, Chuo-ku, Tokyo 103, Japan. Tel: (03) 272-7211.

Back numbers are available at a cost of \$75 per single copy and \$295 per volume.

Orders for reprints, supplements, and back numbers should be addressed to the American Chemical Society, 1155 Sixteenth Street, N. W., Washington, DC 20036-9976. Prices for reprints and supplements are listed at the end of this issue.

Copying Fees: The code that appears on the first page of articles in this journal gives the fee for each copy of the article made beyond the free copying permitted by AIP. (See statement under "Copyright" elsewhere in this journal.) If no code appears, no fee applies. The fee for pre-1978 articles is \$0.25 per copy. With the exception of copying for advertising and promotional purposes, the express permission of AIP is not required provided the fee is paid through the *Copyright Clearance Center, Inc. (CCC), 21 Congress Street, Salem, MA 01970*. Contact the CCC for information on how to report copying and remit payment.

Microfilm subscriptions of the *Journal of Physical and Chemical Reference Data* are available on 16 mm and 35 mm. This journal also appears in *Sec. I of Current Physics Microform (CPM)* along with 26 other journals published by the American Institute of Physics and its member societies. A *Microfilm Catalog* is available on request.

Copyright 1988 by the U.S. Secretary of Commerce; copyright assigned to the American Institute of Physics (AIP) and the American Chemical Society (ACS). Individual teachers, students, researchers, and libraries acting for them are permitted to make copies of articles in this journal for their own use in research or teaching, including multiple copies for classroom or library reserve use, provided such copies are not sold. Copying for sale is subject to payment of copying fees. (See "Copying Fees" paragraph elsewhere in this journal.) Permission is granted to quote from this journal with the customary acknowledgment of the source. To reprint a figure, table, or other excerpt requires in addition the consent of one of the original authors and notification to AIP. Reproduction for advertising or promotional purposes, or republication in any form, is permitted only under license from AIP, which will normally require that the permission of one of the authors also be obtained. Direct inquiries to Office of Rights and Permissions, American Institute of Physics, 335 East 45th Street, New York, NY 10017-3483. The right of the U.S. Government to unrestricted copying for its own use of copyrighted material originating in its laboratories or under its contracts is specifically recognized.

Journal of
**Physical and
Chemical
Reference Data**

Volume 17, 1988
Supplement No. 1

Gas-Phase Ion and Neutral Thermochemistry

Sharon G. Lias

*Center for Chemical Physics, National Bureau of Standards,
Gaithersburg, Maryland 20899*

John E. Bartmess

*Department of Chemistry, The University of Tennessee,
Knoxville, Tennessee 37996-1600*

Joel F. Liebman

*Department of Chemistry, University of Maryland,
Baltimore County Campus, Baltimore, Maryland 21228*

John L. Holmes

*Department of Chemistry, University of Ottawa,
Ottawa, Ontario K1N 9B4, Canada*

Rhoda D. Levin

*Center for Chemical Physics, National Bureau of Standards,
Gaithersburg, Maryland 20899*

and

W. Gary Mallard

*Center for Chemical Physics, National Bureau of Standards,
Gaithersburg, Maryland 20899*



Published by the **American Chemical Society**
and the **American Institute of Physics** for
the **National Bureau of Standards**

845, 1111

Copyright © 1988 by the U.S. Secretary of Commerce on behalf of the United States. This copyright will be assigned to the American Institute of Physics and the American Chemical Society, to whom all requests regarding reproduction should be addressed.

Library of Congress Catalog Card Number 88-70606

International Standard Book Number
0-88318-562-8

American Institute of Physics, Inc.
335 East 45th Street
New York, New York 10017-3483

Printed in the United States of America

Foreword

The *Journal of Physical and Chemical Reference Data* is published jointly by the American Institute of Physics and the American Chemical Society for the National Bureau of Standards. Its objective is to provide critically evaluated physical and chemical property data, fully documented as to the original sources and the criteria used for evaluation. One of the principal sources of material for the journal is the National Standard Reference Data System (NSRDS), a program coordinated by NBS for the purpose of promoting the compilation and critical evaluation of property data.

The regular issues of the *Journal of Physical and Chemical Reference Data* are published quarterly and contain compilations and critical data reviews of moderate length. Longer monographs, volumes of collected tables, and other material unsuited to a periodical format are published separately as *Supplements to the Journal*. This tabulation, "Gas-Phase Ion and Neutral Thermochemistry", by Sharon G. Lias, John E. Bartmess, Joel F. Liebman, John L. Holmes, Rhoda D. Levin, and W. Gary Mallard, is presented as Supplement No. 1 to Volume 17 of the *Journal of Physical and Chemical Reference Data*.

David R. Lide, Jr., Editor
Journal of Physical and Chemical Reference Data

Gas-Phase Ion and Neutral Thermochemistry

Sharon G. Lias

Center for Chemical Physics, National Bureau of Standards, Gaithersburg, Maryland 20899

John E. Bartmess

Department of Chemistry, The University of Tennessee, Knoxville, Tennessee 37996-1600

Joel F. Liebman

Department of Chemistry, University of Maryland, Baltimore County Campus, Baltimore, Maryland 21228

John L. Holmes

Department of Chemistry, University of Ottawa, Ottawa, Ontario K1N 9B4, Canada

Rhoda D. Levin

Center for Chemical Physics, National Bureau of Standards, Gaithersburg, Maryland 20899

and

W. Gary Mallard

Center for Chemical Physics, National Bureau of Standards, Gaithersburg, Maryland 20899

Critically evaluated data on heats of formation of positive and negative ions in the gas phase are compiled and presented in these tables (GIANT tables), along with auxiliary information on ionization energies, proton affinities, electron affinities and acidities, as well as relevant thermochemistry of related neutral species. The literature coverage is through the middle of 1986. The criteria used in carrying out evaluations of data are described, and a short discussion is presented of special concerns for the thermochemistry of charged species.

Key words: acidity; anion; basicity; cation; Franck-Condon principle; electron affinity; heats of formation; ion/molecule equilibrium; ionization energy; negative ion; proton affinity.

Contents

| | |
|--|----|
| 1. Introduction | 5 |
| 1.1. History | 5 |
| 1.2. Definitions | 5 |
| 1.3. Scope, Limitations to Coverage, and Organization..... | 6 |
| 1.4. Literature References | 7 |
| 1.5. Units | 7 |
| 1.6. Ion Thermochemistry at Finite Temperatures..... | 8 |
| 1.6.1. Thermochemical Conventions for the Electron..... | 8 |
| 1.6.2. Thermochemistry of Positive Ions at Finite Temperatures | 10 |
| 1.6.3. Thermochemistry of Negative Ions at Finite Temperatures | 12 |
| 2. Positive Ions..... | 12 |
| 2.1. The Evaluation of Experimentally-Determined Ionization Energies: The Franck-Condon Principle | 12 |
| 2.2. Interpretation of Appearance Energies..... | 13 |
| 2.3. Experimental Techniques..... | 14 |
| 2.3.1. Optical Spectroscopy | 14 |
| 2.3.2. Beam Studies Involving Laser Photoionization | 14 |
| 2.3.3. Determination of Ionization/Appearance Energies by Threshold Techniques..... | 14 |
| 2.3.4. Photoelectron Spectroscopy | 15 |
| 2.3.5. Ion/Molecule Equilibrium Constant Determinations..... | 15 |
| 2.3.6. Ion/Molecule Bracketing Experiments | 16 |
| 2.3.7. Onsets of Endothermic Reactions..... | 16 |
| 2.3.8. Other Techniques..... | 16 |
| 2.4. Reliability of Ionization Energy Data and Criteria for Evaluation..... | 16 |
| 2.4.1. Comparisons between Results of Different Techniques ... | 16 |
| 2.4.2. Reliability of Data; Error Limits..... | 17 |
| 2.5. Trends in the Data | 18 |
| 2.5.1. Estimation Schemes for Heats of Formation of Cations... | 18 |
| 2.5.2. Correlations of Ionization Energies with Proton Affinities, Substituent Constants | 19 |
| 3. Negative Ions..... | 25 |
| 3.1. Aims | 25 |
| 3.2. Experimental Techniques..... | 26 |
| 3.2.1. Laser Photoelectron Spectroscopy [LPES]..... | 26 |
| 3.2.2. Laser Photodetachment [LPD] | 26 |
| 3.2.3. Photodetachment [PD] | 26 |
| 3.2.4. Photodissociation [PDis]..... | 26 |
| 3.2.5. Ion/Molecule Reaction Equilibrium Constant Determinations [IMRE, Kine, TDEq, TDAs] | 26 |
| 3.2.6. Ion/Molecule Reaction Bracketing [IMRB] | 27 |
| 3.2.7. Electron Impact Appearance Potentials [EIAP]..... | 27 |
| 3.2.8. Neutral Beam Ionization/Appearance Potentials [NBIP/NBAP] | 28 |
| 3.2.9. Photoionization [PI]..... | 28 |
| 3.2.10. Endothermic Reaction Energy, Including Charge Transfer [Endo, EnCT, CIDT] | 28 |
| 3.2.11. Surface Ionization (Magetron) [SI] | 28 |
| 3.2.12. Electron Swarm [ES] | 28 |
| 3.2.13. Lattice Energy [Latt] | 28 |
| 3.2.14. Kinetic Branching Methods [Bran, CIDC] | 29 |
| 3.2.15. Electron Transmission Spectroscopy [ETS] | 29 |

| | |
|---|----|
| 3.2.16. Electron Capture Detector [ECD] | 29 |
| 3.2.17. Mobility of Ions in a Gas [Mobl] | 29 |
| 3.2.18. Laser Optogalvanic Photodetachment Spectroscopy [LOG] | 29 |
| 3.3. Thermochemical Cycles | 29 |
| 3.4. Priority of Data | 30 |
| 4. Thermochemistry of Neutral Species | 30 |
| 4.1. Literature Sources | 31 |
| 4.1.1. Organic Compounds | 31 |
| 4.1.2. Inorganic Compounds | 31 |
| 4.1.3. Specialized Classes of Compounds and Radicals | 31 |
| 4.1.4. Other Literature Sources | 31 |
| 4.2. Conventions Used and Assumptions Made | 32 |
| 4.3. Use of Condensed Phase Heats of Formation | 32 |
| 4.4. Estimated Heats of Formation | 32 |
| 4.4.1. Estimates from Data on Isomeric Species | 32 |
| 4.4.2. Summing of Increments | 33 |
| 5. Summary of Conventions Used in Tables | 34 |
| 5.1. Positive Ion Table (Table 1) | 34 |
| 5.2. Negative Ion Table (Table 2) | 34 |
| 5.3. References to Tables 1 and 2 | 36 |
| 6. Acknowledgements | 36 |
| 7. References to Introduction | 37 |

List of Tables

Tables in Introductory Text

| | |
|---|----|
| 1.6.1.1. Summary of assumptions about electron thermochemistry in data compilations | 9 |
| 2.5.1.1. Estimation scheme of Holmes, Fingas and Lossing: $\Delta_r H(M^+) \text{ kcal/mol} = A - Bn + C/n$ | 19 |
| 2.5.1.2. Estimation scheme of Bachiri, Mouvier, Carlier, and DuBois: $\log_{10} \{ [IP(R_1XR_2) - IP_\infty] / (IP_0 - IP_\infty) \} = 0.106 [I(R_1) + I(R_2)]$ | 20 |
| 2.5.1.3. Comparison of ionization energies/heats of formation with estimated values predicted from estimation schemes | 20 |
| 2.5.2.1. The relationship between proton affinity, ionization energy, and hydrogen affinity for homologous series | 23 |
| 5.2. Acronyms, abbreviations and symbols used in negative ion table | 35 |

Tables of Data

| | |
|--|-----|
| 1. Ionization energies and related heats of formation of positive ions | 40 |
| 2. Acidities, electron affinities, and heats of formation of anions | 647 |
| References to Tables 1 and 2 | 797 |

1. Introduction

1.1. History

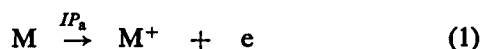
This publication is the direct linear descendant of two earlier compilations of evaluated heats of formation of ions derived from ionization potential and appearance potential data, both carried out under the auspices of the National Bureau of Standards. The first such volume, "Ionization Potentials, Appearance Potentials, and Heats of Formation of Gaseous Positive Ions" by J. L. Franklin, J. G. Dillard, H. M. Rosenstock, J. T. Herron, K. Draxl, and F. H. Field¹ appeared in 1969, and included all data on threshold energies for formation of positive ions which had appeared in the literature through mid-1966. That book, although woefully out-of-date now, is still occasionally referred to in the mass spectrometric literature, and is one of the most widely cited publications in the history of mass spectrometry. In 1977, H. M. Rosenstock, K. Draxl, B. W. Steiner, and J. T. Herron published an update, "Energetics of Gaseous Ions," which covered the literature through mid-1971²; the scope of the work was also extended to cover data on anions. In both of these books, the data for the threshold energies for formation of ions (ionization potentials and appearance potentials) were evaluated where possible, and where thermochemical data for relevant neutral species were available, values for heats of formation of the corresponding ions were derived.

In 1982, two of the present authors published an extensive compilation of *unevaluated* ionization potential and appearance potential data ("Ionization Potential and Appearance Potential Measurements, 1971-1981")³ which covered the literature from the 1971 cut-off date of the 1977 book through mid-1981.

Since the mid-1970's, much information about ion thermochemistry has been derived from determinations of the equilibrium constants of ion/molecule reactions, a type of data which was not covered in the earlier compilations. Much of the work on equilibria of positive ions involves proton transfer reactions. These data have been compiled and evaluated by some of the present authors⁴.

1.2. Definitions

The *heat of formation of a positive ion in the gas phase* is obtained by taking the heat of formation of the corresponding neutral species and adding the energy required to remove an electron, the so-called adiabatic ionization potential or, more correctly, the *adiabatic ionization energy*, IP_a (sometimes designated IE, or, in the older literature, I):



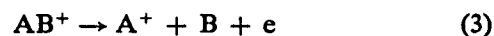
$$\Delta_f H(M^+) = \Delta_f H^\circ(M) + IP_a - \Delta_f H(e) \quad (2)$$

As discussed in Sec. 1.6.2., Eq. (2) is rigorously correct only at absolute zero. According to the convention adopted in this work for dealing with the thermochem-

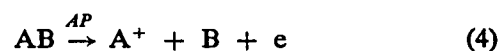
istry of the electron (the "ion convention", sometimes called the "stationary electron convention", see Sec. 1.6.1.) "298 K heats of formation" of positive ions are often derived by simply adding the 0 K value for the ionization energy to the 298 K heat of formation of the molecule. The assumptions inherent in this treatment are discussed in Secs. 1.6.1. and 1.6.2. The user of these tables is cautioned that there is an alternate convention for dealing with the thermochemistry of the electron, which results in numerically different values for heats of formation for ions than those given here; details are discussed in Sec. 1.6.1.

The *vertical ionization energy* is the energy change corresponding to formation of the ion in a configuration which is effectively the same as that of the equilibrium geometry of the ground state neutral molecule. See Sec. 2.1. for a more complete discussion.

Accepting the simplifications described and justified in Sec. 1.6., determination of the heat of formation of a *molecular ion* is, in principle, straightforward, requiring only a value for the heat of formation of the corresponding neutral molecule and a reliable value for the adiabatic ionization energy. Many positive ions of interest, however, do not have stable neutral molecular counterparts. These include many of the ions which originate by fragmentation of a molecular ion, *fragment ions*:



Heats of formation of fragment ions, A^+ , are usually based on mass spectrometric determinations of the energy required to generate the ion from the neutral precursor molecule. This energy is called the "appearance potential" or, more correctly, the "*appearance energy*", AP :

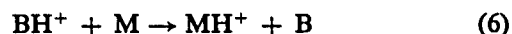


In the case that there is no potential barrier in the reaction coordinate, and little or no kinetic shift (see Sec. 2.2.), the onset energy for formation of A^+ corresponds approximately to the enthalpy change of reaction 4. Under these conditions, the heat of formation of A^+ is usually assumed to be given by:

$$\Delta_f H(A^+) = \Delta_f H^\circ(AB) - \Delta_f H^\circ(B) + AP \quad (5)$$

where, according to the ion convention, the term accounting for the electron has been taken to be zero. For a discussion of a more exact treatment of the energetics of ionic fragmentation processes, see Sec. 1.6.2.

Stable cations formed in the gas phase also include ions formed by protonating a neutral molecule:



In practice, heats of formation of most *protonated molecules* are derived from experiments in which the

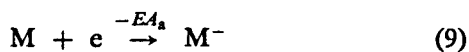
equilibrium constant of a proton transfer reaction such as 6 is determined (given that a heat of formation of a reference BH^+ ion is available from appearance potential determinations). Formally, the relationship between the heat of formation of MH^+ and its neutral counterpart, M , is defined in terms of a quantity called the *proton affinity*, PA . The proton affinity is the negative of the enthalpy change of the hypothetical protonation reaction:



$$\Delta_f H(MH^+) = \Delta_f H^\circ(M) + \Delta_f H(H^+) - PA \quad (8)$$

The term proton affinity, as universally used, is a quantity defined at 298 K (and therefore not strictly analogous to the adiabatic ionization energy, which is the 0 K enthalpy change of reaction 1). (The Gibbs energy change associated with reaction 7 is called the *gas basicity*, GB , of molecule M .) At 298 K, the heat of formation of the proton, using the ion ("stationary electron") convention, is 365.7 kcal/mol, 1530.0 kJ/mol.

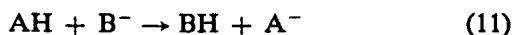
The *electron affinity* (EA) of a molecule is, for *negative ions* or *anions*, the quantity which is analogous to the ionization energy for positive ions. That is, the electron affinity is equal to the energy difference between the heat of formation of a neutral species and the *heat of formation of the negative ion* of the same structure. The electron affinity is defined as the negative of the 0 K enthalpy change for the electron attachment reaction:



The *gas phase acidity* (or merely, acidity) of a molecule AH , $\Delta_{acid} G(AH)$, is the Gibbs energy change of the reaction:



usually defined at 298 K. The enthalpy change of reaction 10, $\Delta_{acid} H$, is, of course, the proton affinity of the anion. The Gibbs energy change of the reaction:



is called the *relative acidity* of species AH and BH .

1.3. Scope, Limitations to Coverage, and Organization

The intent of the present effort is to give (a) the "best" available experimentally-determined values for ionization potentials, electron affinities, acidities or proton affinities of molecules or molecular fragments, and (b) the heats of formation of the corresponding positive and negative ions. Also included are values for the heats of formation of the relevant neutral species which were

used to obtain the heats of formation of the ions. Appearance energies are not specifically listed here, although heats of formation of ions derived from such data are given where the accuracy is sufficiently great to warrant inclusion.

In evaluating heats of formation of ions for the present work, all data presented in the previous compilations^{1,2,3,4} have been considered, along with data from the more recent literature, 1981–1986. In addition, thermochemical information about ions derived from ion/molecule equilibrium constant determinations has been fully utilized, both in evaluations of ionization potential/appearance potential, proton affinity, acidity, and electron affinity data, and in deriving values for heats of formation of ions for which no other information is available.

Because the values for the heats of formation of ions are, of course, dependent on the larger corpus of thermochemical data on uncharged species, the values for heats of formation of relevant neutral species which were utilized are included as an intrinsic part of the tables.

The user familiar with the previous compilations in this series will note that the format of the present work is considerably different from that of its predecessors. In all three previous volumes, *all* ionization energy or appearance energy data pertaining to a particular ionic species were displayed, so that the books served as complete summaries and guides to the literature. Because of the increasing volume of such an archive with time, such a display is no longer practical for the positive ion data. Furthermore, because the general quality of mass spectrometric measurements has increased greatly over the last decade, display of some of the now out-of-date early data is no longer even desirable for ionization potentials/appearance potentials.

On the other hand, there has been a tremendous increase in the number of anions for which some thermochemical information is known, and this publication presents the first extensive evaluated compilation of those data. The table of anion thermochemistry (Table 2), therefore, includes both as *complete* a collection as possible of the literature data, and an assignment where possible of the "best" value for the thermochemistry.

Because earlier volumes in this series^{1,2} were devoted to deriving values for *heats of formation* of ions, this work has been defined in the same way. It should be emphasized that in Table 1 (the positive ion table) molecules for which heats of formation are not known or have not been estimated are not included, even if the corresponding ionization energies or proton affinities are known. On the other hand, Table 2 (the negative ion table) presents a complete archive of data on electron affinities and gas phase acidities, whether or not the thermochemistry of relevant neutral species is available; the evaluation of the scale of gas phase acidities will, however, be the subject of a separate publication⁵. The total archive of ionization energy and appearance energy data will be published separately⁶, as will the updated scales of gas phase basicities/proton affinities⁷.

Another consequence of defining this work in terms of thermochemical data is that the abundant data on excited states of ions from photoelectron spectroscopy are not included here. The combined bibliographies of this work and its predecessors, however, do include the entire corpus of literature of photoelectron spectroscopy, since values for the *lowest* ionization energy derived from photoelectron experiments are included. Also not included are data on multiply charged ions.

Thermochemical information about ion/molecule clusters has been published in a recent compilation⁸, and is not specifically included here, although some information derived from the enthalpy changes associated with the association of the first solvent molecule have been used in evaluating certain heats of formation.

At this writing, publications are beginning to appear in increasing numbers giving quantum mechanical calculations of very high accuracy on the thermochemical properties of ions, especially small ions^{9,10}. The present work includes *only* data derived from experimental determinations. However, conclusions derived from some high level calculations have been taken into account in the evaluation of data for particular species.

The solution phase reduction potentials of a variety of species have been correlated with gas phase electron affinities (EAs), and values for a large number of EAs have been extrapolated from such correlations. More recent determinations of accurate gas phase data have shown that such relationships hold only for limited classes of compounds, so that the solution phase data can be taken only as an approximate guide to predicting electron affinities. Thus, any electron affinity values derived from reduction potentials have been omitted from this compilation. Such values were included in a recent compilation of anion data¹¹.

This compilation also does not attempt to cover negative electron affinities — cases where the electron in the highest occupied molecular orbital is unbound (resonance states), and therefore the lifetime of the anion with respect to autodetachment is on the order of microseconds, at most. Electron transmission spectrometry¹² is used to determine thermochemical data for such species. Brief mention is made for certain small molecules and elements for which the anion is known to be unbound, to differentiate from cases for which there is just no data available.

The data on positive ions and on negative ions are not interdependent, and have been evaluated separately. Data on the positive ions were collected and evaluated at the National Bureau of Standards (ionization energies, equilibrium constant data) and the University of Ottawa (appearance energies), while information concerning the negative ions was handled at the University of Tennessee. The data on cations and anions are presented in two separate tables.

Since heats of formation of ions are derived using data on heats of formation of neutral molecules and radicals, data on the thermochemistry of uncharged species are an integral part of this work. Although only experimen-

tally-determined values for heats of formation of neutral species were utilized in the 1977 evaluation, estimation schemes for arriving at thermochemical information are now widely accepted and used. Estimated heats of formation are included for many species for which no experimental data are available. These estimations, and a literature search for thermochemical data not available in compilations, were performed primarily at the University of Maryland, Baltimore County Campus.

1.4. Literature References

With respect to ionization energies, appearance energies, or proton affinities, the present publication gives specific citations only to publications which were not included in the previous compilations^{1,2,3,4}. The bibliography includes *all* references which have appeared since the previous publications^{1,2,3,4} even if the data from a particular paper are not given here because of a lack of information about the thermochemistry of relevant neutral molecules. When no literature reference is given for these kinds of data in the positive ion table, it should be assumed that the primary reference can be obtained from the secondary sources, references 1, 2, 3, or 4. When the source of the data on ion thermochemistry is a recent paper which was not included in any of these previous compilations, the reference is specifically cited in a footnote. The literature citations for which a specific column is provided in Table 1 refer only to the source of the data on the thermochemistry of the neutral species.

In Table 2, specific citations are given for the data on both the ion thermochemistry and the relevant neutral thermochemistry.

1.5. Units

Information is displayed in the tables using different units, dictated by the current practices for reporting data of a particular kind. For example, ionization energy and electron affinity values are usually reported in electron volts, and that is the unit used here for these data. Heats of formation of positive ions are given here in both kcal/mol and kJ/mol. The reason for this duplication is simply that both units are extensively used in the literature, and users of these tables will be about equally divided between those who prefer kilocalories and those who prefer kilojoules. Furthermore, because of the duplication in units, the data can always be displayed as they appeared in the original paper, a practice which helps in elimination of transcribing errors. While the same statements certainly apply to data on negative ions, the amount of information which needs to be displayed in Table 2 is sufficiently great that including the same information twice, in two sets of units, would crowd the page too much; therefore, the negative ion heats of formation and acidities are given only in the SI unit, kJ/mol.

The conversion factors which were used in this work are: 1 electron volt (eV) = 23.06036 kilocalories/mole = 96.4845 kilojoules/mole; 1 kilocalorie/mole = 4.184 kilojoules/mole.

1.6. Ion Thermochemistry at Finite Temperatures

The auxiliary thermochemical information required for citation of ion heats of formation—heats of formation of relevant neutral species—is available mostly for species at 298 K. These thermochemical data are correct for use in deriving ion heats of formation from equilibrium constant determinations, i.e., for treatment of data derived from processes occurring at temperatures other than 0 K. However, strictly speaking, the ionization energy and the electron affinity of a molecule are quantities which correspond to processes occurring at 0 K. As mentioned above in Sec. 1.2., a rigorously correct treatment of heats of formation of ions requires explicit treatment of the differences in thermochemical values at 0 K and at higher temperatures. This section describes the principles involved in such a correct treatment, considers the simplifications which are often made in the literature, and specifies how data have been treated in this work.

1.6.1. Thermochemical Conventions for the Electron

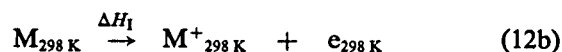
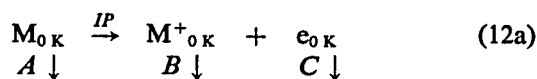
We are concerned with the way in which the enthalpies of formation of the chemical species, M^+ and M^- , are *defined*, particularly at temperatures other than 0 K. The enthalpy of formation of any chemical species is always taken as the difference between the enthalpy of the compound and the sum of the enthalpies of the elements of which it is composed. However, in the case of an ion, M^+ or M^- , a special problem arises—one must explicitly take into account the enthalpy of the electron in some way.

There are two conventions for dealing with the thermochemistry of the electron, one used predominantly by thermodynamicists^{13,14,15} and one adopted by scientists studying ion physics/chemistry^{16,17}. The thermodynamicists' convention, commonly called the "thermal electron convention" or merely the "electron convention", defines the electron as a standard *chemical element* and treats its thermochemistry accordingly. The mass spectrometrists' convention, known as the "stationary electron convention" or the "ion convention", defines the electron as a *sub-atomic particle*. Because of differences in the treatment of the thermochemistry under these two definitions, except at absolute zero values cited for the enthalpies of formation of ions in certain thermochemical compilations such as the JANAF tables¹³ or the NBS Tables of Chemical Thermodynamic Properties¹⁴ differ from those cited here, or in most mass spectrometric literature, by 1.481 kcal/mol, 6.197 kJ/mol. Our values are lower for positive ions and higher for negative ions. Problems arise when users unknowingly mix inconsistent values for heats of formation in the same equation.

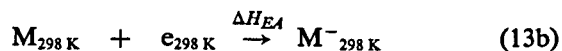
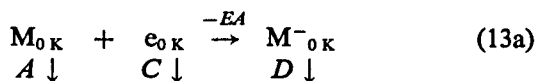
There is considerable confusion and misunderstanding of the basic assumptions and treatment of the thermochemistry of the electron in the two approaches. Many scientists who regularly use one or the other convention in their work can not clearly explain the differences. In-

deed, some hold that the two ways of dealing with the thermochemistry of the electron are not merely two conventions, but two scientifically different concepts, one of which must be incorrect. The discussion which follows is an attempt to present the question of how the electron is treated in a thermochemical equation in as simple and straightforward a manner as possible, in the hope that some of the confusion will be dispelled and the identity of the two treatments as *conventions* will become clear. This discussion is also intended to justify the choice of the usual mass spectrometrists' convention for use in these tables.

The relationships between the various quantities which must be considered are shown in the thermochemical cycles:



and



where A , B , C , and D are the integrated heat capacities for the various indicated species, e.g., A is the energy required to raise M from 0 K to 298 K, and ΔH_I and ΔH_{EA} are the 298 K enthalpies of reaction. This discussion will be concerned with the standard temperature, 298 K, but the arguments can obviously be extended to any other temperature.

At 0 K, the heat of formation of the electron is zero and the heats of formation of the ions are exactly equal to the 0 K heat of formation of the molecule M plus the energy difference between M and the corresponding ion:

$$\Delta_f H(M^+)_{0\text{ K}} = \Delta_f H^\circ(M)_{0\text{ K}} + IP_a \quad (14)$$

$$\Delta_f H(M^-)_{0\text{ K}} = \Delta_f H^\circ(M)_{0\text{ K}} - EA \quad (15)$$

At absolute zero, there is no difference between the two conventions.

When the temperature is raised to 298 K, the heats of formation of M^+ and M^- will be related to the heat of formation of M at 298 K through the enthalpy changes of reactions 12b and 13b:

$$\Delta_f H(M^+)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} - \Delta_f H(e)_{298\text{ K}} + \Delta H_I \quad (16)$$

$$\Delta_f H(M^-)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} + \Delta_f H(e)_{298\text{ K}} + \Delta H_{EA} \quad (17)$$

The enthalpy changes of reaction at 298 K are related to the 0 K ionization energy and electron affinity through the relationships:

$$\Delta H_I = IP_a + (C + B - A) \quad (18)$$

$$\Delta H_{EA} = -EA - (C + A - D) \quad (19)$$

If the electron is defined to be a chemical element (the "electron convention"), its heat of formation by definition is zero at all temperatures in its standard state. Thermodynamicists start from this assumption and then make a second one, that an electron gas can be treated as an ideal gas following Boltzmann statistics; this second assumption is used to calculate the integrated heat capacity of the electron, C . In many thermodynamics data compilations, the integrated heat capacity terms for M and the corresponding ion, M^+ or M^- , are taken to be approximately equal for many ions, i.e. $A = B = D$. (See Sec. 1.6.2. for a discussion of this assumption.) Under this set of assumptions, Eqs. (16) and (17) can be written:

$$\Delta_f H(M^+)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} + [IP_a + C] \quad (20)$$

$$\Delta_f H(M^-)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} - [EA - C] \quad (21)$$

(where the term $\Delta_f H(e)_{298\text{ K}}$ has been taken to be equal to zero and the quantity in brackets is the assumed enthalpy change of reaction at 298 K). What most often causes confusion for non-thermodynamicists is the de facto assignment of the integrated heat capacity of the electron, C , to the ion M^+ or M^- , rather than to the electron in going from 0 K to 298 K. This is required if the heat of formation of the electron is constrained to be zero at all temperatures. It is questionable whether an ion is any more "ideal" than an electron, due to the Coulombic forces between the particles, but this assignment is a necessity if the original assumptions are carried through the argument.

In contrast, the standard treatment of ion heats of formation followed in almost the entire corpus of literature on ion physics/chemistry essentially assumes that:

$$\Delta_f H(M^+)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} + [IP_a + B - A] \quad (22)$$

$$\Delta_f H(M^-)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} + [-EA - A + D] \quad (23)$$

(where the expressions in brackets are assumed to be equal to the enthalpy change of reaction at 298 K, and the quantities A , B , and D are often, but not always, taken to be equal). Since this is equivalent to taking a value of zero for the integrated heat capacity of the electron (the term C in Eqs. (18) and (19)), this way of treating the thermochemistry of the electron has come to be known as the "stationary electron" convention. The use of this term has unfortunately led to the widespread conception that this convention defines the ionization process as producing an electron which has no thermal energy at 298 K. Since this is not the case, it is preferable

to choose another designation for the convention. In this publication we will adopt the term originally suggested by Syverud¹⁸ for the mass spectrometrists' convention, "ion convention".

At 298 K, the integrated heat capacity of an ideal Boltzmann gas is 1.481 kcal/mol, 6.197 kJ/mol. The relationship between 298 K heats of formation of ions in the ion convention (IC) and the thermodynamicists' convention (TC) is:

$$\Delta_f H(M^+)_{298\text{ K}}(\text{IC}) = \Delta_f H(M^+)_{298\text{ K}}(\text{TC}) - 6.197\text{ kJ/mol} \quad (24)$$

$$\Delta_f H(M^-)_{298\text{ K}}(\text{IC}) = \Delta_f H(M^-)_{298\text{ K}}(\text{TC}) + 6.197\text{ kJ/mol} \quad (25)$$

Table 1.6.1.1. summarizes the assumptions made in the two conventions and the data compilations where they are used.

TABLE 1.6.1.1. Summary of assumptions about electron thermochemistry in data compilations.

| Convention | Compilation | Convention Includes $H_T - H_0$ for Species: | | | Value of C , kJ/mol |
|-----------------------------|-----------------------------|--|----------------|-----|-----------------------|
| | | M | M^+ or M^- | e | |
| Thermal electron | JANAF Tables ¹³ | Yes* | Yes* | Yes | 6.197 |
| Thermal electron | Gurvich et al ¹⁵ | Yes* | Yes* | Yes | 6.197 |
| Thermal electron (Modified) | TN270 ¹⁴ | No | No | Yes | 6.197 |
| Ion convention | This work, Refs. 1-4 | Yes* | Yes* | No | 0 |
| Ion convention | Some papers | No | No | No | 0 |

*When sufficient information is available. See discussion in Sec. 1.6.2.

The objection has been made that the mass spectrometrists' convention is scientifically incorrect because the electron actually does have thermal energy at 298 K. Note, however, that the values derived in the mass spectrometrists' convention for the heats of formation of the ions are numerically identical to those one would obtain if one assigned the thermal energy of the electron to the electron rather than to the enthalpy of formation of the accompanying ion (as is done in the thermodynamicists' convention). That is, in Eqs. (16) and (17) if one assigns a value of C to $\Delta_f H(e)_{298\text{ K}}$ and takes the value for the enthalpy change of reaction from Eqs. (18) and (19), one obtains:

$$\Delta_f H(M^+)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} - C + IP_a + (C + B - A) \quad (26)$$

$$\Delta_f H(M^-)_{298\text{ K}} = \Delta_f H^\circ(M)_{298\text{ K}} + C - EA - (C + A - D) \quad (27)$$

which are identical to expressions 22 and 23. Although this is a nonstandard treatment, it is possible to justify using a special convention for the thermochemical properties of the electron, since this species is not normally considered to be a chemical element; a stronger justification is found by considering that the use of the standard treatment for an element in this case results in heats of formation for a large body of molecular species — ions — which reflect an arbitrary temperature dependence which can not be experimentally measured or verified at the present time.

In fact, the mass spectrometrists' convention for treating the electron was not derived from a conscious treatment of the electron as having a non-zero heat of formation at 298 K. Indeed, earlier discussions of this convention^{16,17,18} have centered mainly on the reluctance to assign a purely arbitrary temperature dependence to ionization or electron attachment events and a recognition that absolute values of the various parameters, ΔH_i , ΔH_{E_e} , B , D , and especially C were not available. The enthalpy changes of reactions 12b and 13b are not directly measured by any currently-available experimental techniques, and can not be said to be known within ± 6.197 kJ/mol. As will be discussed below, accurate values for the integrated heat capacities of ions M^+ and M^- are not available except for a few small species, and the assumption that $(B - A)$ and $(A - D)$ are exactly equal to zero is often not warranted (see Sec. 1.6.2.). Most important, however, the value chosen for the integrated heat capacity of the electron, C , is completely arbitrary. To quote from the 1985 edition of the JANAF thermochemical tables^{13c}:

"As shown by Sommerfeld¹⁹, the electron gas is a degenerate Fermi-Dirac gas and its properties will differ from the classical (Boltzmann) gas. These deviations will increase as the temperature decreases or as the density increases. Due to the low mass of the electron, these departures from classical behavior will persist to higher temperatures and lower densities than for atomic systems. Under conditions of 1 atm pressure, Gordon²⁰ showed that the deviation of the Fermi-Dirac gas from the Boltzmann gas is negligible above 1250 K. Below this temperature the deviation between classical and quantum statistics will be significant.

Despite these known deviations we have chosen to present the classical (Boltzmann) values here since the primary purpose of this table is to serve as a reference state for the calculation of tables of thermodynamic properties for atomic and molecular ions..... Therefore, although this ideal-gas table has the formalism of 1 bar as the standard reference state, it should not be applied to real systems where the electron partial pressure exceeds 10^{-6} bar."

That is, the authors of the JANAF tables¹³ recognize that the standard thermodynamicists' convention for dealing with the electron does involve a completely arbitrary assumption about the value assigned to the enthalpy of the electron (as does the mass spectrometrists'

convention when expressed by Eqs. (22) and (23)—but not in the assumptions built into the equivalent Eqs. (26) and (27)). Syverud¹⁸, in an unpublished discussion of conventions for treating the thermochemical properties of the electron, cites a value of approximately 3.3 kJ/mol, 0.8 kcal/mol for the value of C derived from a quantum chemical calculation (source not quoted). Furthermore, while the rationale for the thermodynamicists' convention is that the values "correspond to a meaningful thermal process"²¹, the use of that convention is excluded for a substantial set of possible thermal conditions.

The mass spectrometrists' approach to the problem recognizes that the specific inclusion of the term for the enthalpy of the electron in deriving ion heats of formation is not physically meaningful if it is based on the assumption that an electron gas can be treated like an ideal gas. In fact, at this time neither the enthalpy changes of reactions 12b and 13b nor the enthalpy of the electron are established; a solution is to adopt a convention (the "ion convention") which sidesteps the problem, that is, in which the enthalpy change of reaction and the enthalpy of the electron need not be known or assumed. If, in the future, information about the integrated heat capacities of the electron and the ions does become available, the values for heats of formation of ions can be fine-tuned; however, in the meantime, there is no real problem with using data in the present form as long as internal consistency is maintained.

It will be noted that in the tables, the symbol $\Delta_f H$ rather than $\Delta_f H^\circ$ is used to denote the standard heats of formation of the ions. This convention has been adopted here to emphasize that the heats of formation are referred to the ion convention rather than the electron convention used by thermodynamicists.

1.6.2. Thermochemistry of Positive Ions at Finite Temperatures

Molecular ions. Using the ion convention (also known as the stationary electron convention, see Sec. 1.6.1.) the heat of formation of molecular ion M^+ at temperature T can be defined in terms of the heat of formation of the corresponding neutral species, M , at temperature T , and a quantity labelled ΔH_i , the gas phase enthalpy change of ionization, which represents the energy required to bring about ionization at temperature T :

$$\Delta_f H(M^+)_T = \Delta_f H^\circ(M)_T + \Delta H_i \quad (28)$$

In applying Eq. (28), the value for ΔH_i is usually taken to be exactly equal to the adiabatic ionization potential. Although the use of the ion convention obviates the necessity of assigning an exact value to the increase in the ionization energy at temperature T due to energy imparted to the electron, the assumption that ΔH_i is the same as IP_a is still not correct. The adiabatic ionization energy of a molecule is the energy difference between the lowest rotational and vibrational levels of the ground

state of the molecule and the lowest rotational and vibrational levels of the electronic ground state of the ion, i.e., the difference between the heats of formation of the molecule and the corresponding ion at absolute zero. The adiabatic ionization energy—the quantity obtained from analysis of a Rydberg series (Sec. 2.3.1.) or from determinations of an ionization onset energy (Sec. 2.3.3.) — is a measure of the $0 \rightarrow 0$ transition, and does not depend on the temperature at which the determination is made.

However, it is a common practice to derive "298 K heats of formation" of positive ions by simply adding the 0 K value for the ionization energy to the 298 K heat of formation of the molecule. This practice probably gains impetus from the fact that much of the available thermochemical data for chemical compounds (particularly for organic and other large polyatomic compounds) correspond to values for heats of formation at 298 K.

The relationship between the enthalpy change associated with ionization at temperature T , ΔH_T , and the adiabatic ionization energy is shown in thermochemical cycle 12, and given explicitly in Eq. 18. When using the ion convention for dealing with thermochemistry of the electron (Sec. 1.6.1.), the integrated heat capacity of the electron (the quantity C in the cycle) can be ignored, and the relationship between the adiabatic ionization energy and the enthalpy change of ionization at temperature T is given by:

$$\Delta H_T = IP_a + B - A \quad (29)$$

That is, IP_a and ΔH_T are the same only when the integrated heat capacities of the neutral molecule, M , and the ion, M^+ , are identical over this temperature range. An analysis²² of the differences between integrated heat capacities of M and M^+ for various molecules demonstrated that (a) there will be no discernable differences between the translational and rotational heat capacities of M and M^+ , (b) that differences arising from a splitting of degenerate energy levels in multiplet ground states of M or M^+ will never be larger than 0.009 eV at temperatures in the 300–400 K range, and (c) when the frequency of a particular vibration changes upon ionization, there will be a difference between the integrated heat capacities of M and M^+ . However, even this contribution will usually be sufficiently small that a significant error will not be introduced if it is ignored. For example, the lowest ionization energy of ethylene corresponds to removal of an electron from the C–C pi bond, which leads to a lowering of the frequency of the symmetric C–C stretch from 1623 to 1230 cm^{-1} and a reduction in the frequency of the twisting around the C–C bond from 1027 to 430 cm^{-1} . Although these differences in vibrational frequencies are significant, the predicted effect on the 298 K enthalpy of ionization is to raise it above the value for the adiabatic ionization potential by only 0.0069 eV, i.e. only the most accurate experimental measurements would detect an increment of this size. Thus for most species, the simplifying assumption that the adi-

abatic ionization energy and the 298 K enthalpy of ionization, ΔH_T , are approximately the same:

$$IP_a \sim \Delta H_T \quad (30)$$

will not introduce significant errors in the 298 K heats of formation of molecular radical cations.

In this compilation, most values of heats of formation of molecular ions correspond to 298 K. Most of these were obtained by simply adding the value for the adiabatic ionization energy to the 298 K heat of formation of the neutral species, that is, the assumption stated in Eq. (30) was usually made. Of course, a rigorously correct treatment would require calculating exact values for integrated heat capacities A and B from complete sets of vibrational frequencies for the molecule and the ion. This complete procedure has been applied to only a few of the species listed in this compilation. Vibrational frequencies for most of the ions are not available, and the correction would simply cancel out if one made the often-used assumption that the vibrational frequencies of the ion and its neutral counterpart are the same. Whenever the original authors carried out such a complete analysis (a routine procedure only for photoelectron-photoion coincidence studies), the results of that analysis are included here, and both 0 K and 298 K values for the ion heat of formation are given. In addition, for those diatomic and triatomic and other small molecules for which values for the 0 K heats of formation as well as the vibrational frequencies of the molecule²³ and the ion²⁴ were readily available, the heats of formation of the ion at absolute zero and at 298 K were derived by the more correct procedure. In the course of this work, we did not, however, carry out a comprehensive literature search for sets of vibrational frequencies, but only made use of readily available compilations^{23,24}.

Fragment ions. Analogous arguments can be applied to the use of appearance energies for the derivation of heats of formation of fragment ions, A^+ , at temperature T in Eq. (5). If there are no complicating factors (see Sec. 2.2.), the appearance energy, AP , corresponds to the enthalpy change for the fragmentation reaction 4, and can be used to derive a value for the heat of formation of the fragment ion, A^+ . Correctly, a 0 K heat of formation of A^+ must be obtained using 0 K heats of formation of AB and B in the calculation, and this heat of formation can then be corrected to some other temperature, T , taking into account the vibrational frequencies of the ion and appropriate thermodynamic functions of the elements.

For the most common experimental techniques (energy selected electron impact, photoionization mass spectroscopy, etc.) for measuring the appearance energy of a fragment ion starting from a molecule or radical at temperature, T , the major problem is to identify the internal energies of the reaction products. This matter has been discussed at length by Traeger and McLaughlin²⁵. At *onset* the products of the unimolecular decomposition will be formed with zero translational energy with respect to the center of mass (provided that the fragmenta-

tion does not involve a reverse energy barrier) and a center of mass translational energy the same as that of the precursor molecule. The products thus are at a translational quasi-temperature, T^* . In principle, if the observational time scale of the experiment and the sensitivity of the ion detector are great enough, then the observed appearance energy approaches that for products having 0 K internal energy (i.e., all internal energy modes have contributed to reaching the transition state). Traeger and McLaughlin²⁵ showed that for the molecule AB:

$$AP_T(\text{exp}) = \Delta_f H[A^+ + B + e]_T - \Delta_f H^*[AB]_T + 5/2RT - \int C_p[A^+ + B + e]dT \quad (31)$$

In effect, this equation corrects the observed threshold energy for the fragmentation process to an effective 0 K value by adding the thermal rotational and vibrational energy contained in AB to the onset.

Most heats of formation of fragment ions are derived making the simplifying assumption that the last two terms of Eq. (31) will cancel one another. That is, values for heats of formation of fragment ions at 298 K derived from appearance potential data are more often obtained by simply using an observed onset energy and 298 K heats of formation of relevant neutral species in Eq. (5). When such a value for a heat of formation has been reported in the literature, the value is given here as it appeared in the original paper, with only the imposed requirement that the thermochemistry of the relevant neutral species employed must be internally consistent with the values of those species used in this publication. Where the original authors have used a more sophisticated analysis, such as that represented by Eq. (31), or that routinely used in the interpretation of photoelectron-photoion data, both 0 K and 298 K values of the ion are cited. The user should be cautioned that the 298 K value assigned to a heat of formation of a fragment ion may differ by as much as 3 or 4 kcal/mol, 12–18 kJ/mol, depending on which of these treatments has been used. For example, Baer and Brand²⁶, and Lossing²⁷ determined the appearance energies for formation of $C_4H_7^+$ ions in C_3H_{10} isomers. Although the appearance energies reported in the two studies were almost identical, the 298 K values for heats of formation of the $C_4H_7^+$ ions derived by Baer and Brand²⁶, using a complete treatment of the temperature dependence of the heat of formation, are higher than the values derived by Lossing²⁷ by 4.3 kcal/mol, 18 kJ/mol.

1.6.3. Thermochemistry of Negative Ions at Finite Temperatures

The electron affinity is a quantity which is analogous to the ionization energy. That is, the electron affinity is a 0 K quantity which corresponds to the transition from the ground state of the neutral species to the ground state of the anion. Thus, the heat of formation of an anion at 298 K can not rigorously be taken as the heat of formation of the corresponding neutral species (298 K) minus the (positive) electron affinity (0 K) without some

estimate of the temperature dependence of the electron affinity. Although the use of the ion convention ("stationary electron" convention) allows one to ignore the integrated heat capacity of the electron, a term for correcting for the integrated heat capacity of the anion from 0 K to 298 K is required. Statistical mechanics permits a calculation of this quantity if the structure and vibrational frequencies of the anion are known. However, at present the necessary data are not readily available for most anions, and therefore this correction is generally ignored in this work.

Under the assumption that the temperature dependence of the electron affinity and that of the ionization energy of the H atom are equal, one can relate the (298 K) gas phase acidity, Eq. (10), to the (0 K) electron affinity:

$$\Delta_{\text{acid}}H(\text{AH}) = D(\text{A-H}) - EA_{0\text{K}}(\text{A}) + IP_{0\text{K}}(\text{H}) \quad (32)$$

There is not extensive data on the validity of this assumption, although it appears to hold¹³ to ± 2 kJ/mol for Cl^- and OH^- .

2. Positive Ions

In the discussion which follows, a brief description of the Franck-Condon principle along with a discussion of the implications for an analysis of data obtained from experimental determinations of ionization energies will be given in Sec. 2.1. In Sec. 2.2., special problems in the interpretation of appearance potential data will be summarized, followed in Sec. 2.3. by short descriptions of the various experimental techniques used in obtaining the data given here, with attention to intrinsic experimental problems which may affect the reliability of data. Section 2.4. will give a discussion of the rationale used in evaluating ionization energy and appearance energy data from the various approaches, and a description of the conventions and symbols used in the tables. Finally, Sec. 2.5. summarizes a few of the regular trends observed in the data, and describes schemes for estimating data on heats of formation of positive ions.

Detailed discussions of the ionization process and of the experimental techniques used in studying ion chemistry, as well as of thermodynamics, are available in many books and reviews. Therefore, no attempt will be made to present a comprehensive discussion or review of these subjects. Rather, attention will be given only to those aspects which have a bearing on the evaluation of data on ionization energies, appearance energies, or ion/molecule equilibrium constants.

2.1. The Evaluation of Experimentally-Determined Ionization Energies:

The Franck-Condon Principle

Ionization of a molecule by photoionization or by electron impact is governed by the Franck-Condon prin-

ciple, which states that the most probable ionizing transition will be that in which the positions and momenta of the nuclei are unchanged^{28,29}. Thus, when the equilibrium geometries of an ion and its corresponding neutral species are closely similar, the energy dependence of the onset of ionization will be a sharp step function leading to the ion vibrational ground state. However, when the equilibrium geometry of the ion involves a significant change in one or more bond lengths/angles from that of the neutral species, the transition to the lowest vibrational level of the ion is no longer the most intense, and the maximum transition probability (the vertical ionization energy) will favor population of a higher vibrational level of the ion; if the geometry change is great, it is possible that the transition to the lowest vibrational level of the ion will not even be observed. These situations are illustrated for hypothetical diatomic species in Fig. 1.

In evaluating ionization energy data, the shapes of photoelectron bands are useful indicators as to which of the situations pictured in Fig. 1 prevails for the particular molecule. A sharp onset indicates that the equilibrium geometries of ion and neutral are quite similar, and that photoionization or electron impact determinations of the ionization threshold are likely to be free of complications. When an ionization process proceeds according to the second situation pictured in the figure, the *onset* of the photoelectron band is observed approximately at the adiabatic ionization energy; adiabatic ionization energies derived from observation of the onsets of photoelectron bands are usually in excellent agreement with adiabatic ionization energies obtained from analyses of Rydberg series or from the most reliable threshold determinations.

When the equilibrium geometry of the ion is very different from that of the corresponding neutral molecule and the lowest vibrational level is not populated in ionization by photon absorption or electron impact, it has been shown that values for the adiabatic ionization energies can be obtained by determining the equilibrium constant for charge transfer to another molecule of known ionization energy:



The enthalpy change for this reaction, which (Sec. 2.3.5.) is obtained from the equilibrium constant determination, is just the difference between the enthalpies of ionization, ΔH_i , of species A and B. As shown above (Sec. 1.6.) this difference is likely to be quite close to the difference in the adiabatic ionization energies:

$$\Delta H(33) = [\Delta H_i(B) - \Delta H_i(A)] \sim [IP_a(B) - IP_a(A)] \quad (34)$$

In such determinations, the ions are at thermal equilibrium with their surroundings, and one measures the thermochemical properties of the ions in their equilibrium geometries.

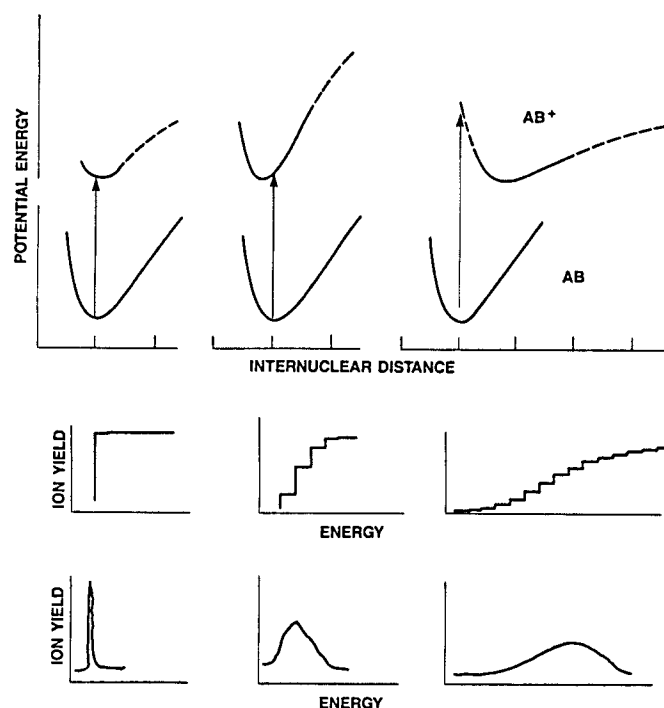


Fig. 1. Potential energy curves for hypothetical diatomic molecule AB, and the corresponding positive ion, AB^+ for the cases in which the equilibrium internuclear distance is (a) the same, (b) slightly different, or (c) greatly different. Below the potential energy curves are hypothetical probabilities for ionization as a function of energy for cases (a), (b), and (c), and, at bottom, shapes of observed photoelectron bands for the three corresponding cases.

2.2. Interpretation of Appearance Energies

In the discussion above, the appearance energy for formation of a fragment ion (reaction 4) was defined, and Eqs. (5) and (31), for obtaining values for the heat of formation of the fragment ion, were derived, with the proviso that the equations were valid only when there is no potential barrier in the reaction coordinate, and no significant "kinetic shift" associated with the determination.

The "kinetic shift"^{2,30,31} is the term applied to describe the experimental observation of ionization onsets which are higher than the thermodynamic onset energy due to the fact that the apparatus samples the (fragmenting) ions at a certain time (usually around 10^{-5} s) after ionization has occurred, when ions undergoing a slow fragmentation process have not yet had time to dissociate. One approach for getting around this problem is an analysis based on the determination of the so-called rate-energy curve for a given fragmentation, in which the rate constant of the dissociating ion is derived as a function of energy. This kind of information is derived by analysis of the data from an elegant technique which is, moreover, capable of delivering very accurate thermochemical information for fragmentation processes, photoelectron-photoion coincidence spectroscopy (PEPICO)³². Another approach to detecting a barrier in the reaction

coordinate is the determination of the kinetic energy carried off by the fragment ion. Studies of metastable peaks, for example, permit such an evaluation³³.

2.3. Experimental Techniques

The 1977 evaluated compilation included an extensive review of the experimental techniques which provide ionization energy and appearance energy data, along with a detailed description of how the data derived from each type of experiment are interpreted to give ionization energies^{2,31}. Although technological advances have been made in mass spectrometric instrumentation since that review was written, the detailed presentation given there is still recommended reading for anyone interested in an in-depth description of the basic principles of the various approaches. For the present purposes, it will suffice to summarize briefly the different types of experiments from which the data presented here originate, and to give some general indications of the strengths and limitations of the different techniques, and how these influence the evaluator in arriving at a recommended value for an ionization energy.

2.3.1. Optical Spectroscopy

The identification of a Rydberg series in an atomic or molecular spectrum leads to a value for the ionization energy; in cases where the analysis of the spectrum is straightforward, the spectroscopic ionization energy values are highly accurate. The determination of atomic ionization energies through optical spectroscopy is a highly developed field which has been extensively reviewed. A large fraction of atomic ionization energies listed here are from expert evaluations of atomic spectra³⁴. In the evaluation of ionization energies of atoms and diatomic molecules, spectroscopic ionization energies have been chosen where they are available. For polyatomic species, a value derived from an analysis of the optical spectrum has been given great weight, unless several determinations from other highly reliable techniques are in conflict with the spectroscopic value. As pointed out by Rosenstock^{2,31}, the evaluation of molecular Rydberg series is not always straightforward, and reported spectroscopic ionization energies of polyatomic species may disagree with values derived from ionization onset determinations or the onsets of photoelectron bands due to complications in the analysis of vibrational and rotational structure.

2.3.2. Beam Studies Involving Laser Photoionization

In the years since the cut-off date of the literature search for the previous volume of this series², several highly accurate ionization energy values have been reported based on multi-photon ionization of vibrationally-cooled species in a molecular beam³⁵. In these studies, a vibrationally and rotationally cooled beam of molecules

is raised to a specific excited state by irradiation with a tunable laser; while this excitation energy is held constant, a second independently tunable laser is used to ionize the beam of excited molecules, with the photon energy being tuned through the ionization onset. The excitation laser is then tuned to a different transition, and the ionization scan is repeated. In this way, the entire Franck-Condon accessible region of the intermediate electronic state is mapped out, insuring that the molecular geometry corresponding to the adiabatic ionization energy is accessed. Since every intermediate vibronic state leads to an independent value of the ionization threshold, the experiment contains an internal consistency check.

2.3.3. Determination of Ionization/Appearance Energies by Threshold Techniques

In the several techniques which fall under this heading, the onset of ionization or of the appearance of a particular fragment ion is detected as a function of the energy of the ionizing agent, either photons or an electron beam. The most obvious problem which must be considered with regard to this technique is the accurate characterization of the energy of the ionizing medium, photons or electrons. When ionization is brought about by photon absorption, this is usually not a problem; monochromators capable of delivering photons with a high energy resolution are available. The most sophisticated photoionization experiments involve detection of energy-selected electrons; in the so-called "threshold photoelectron spectroscopy" technique, only those photoelectrons which correspond to essentially zero energy of ejection are detected.

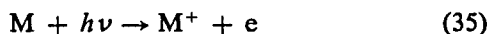
In the past, many experimental determinations of ionization onsets were carried out in instruments in which ionization was effected by bombarding the sample of interest with an electron beam in which the electrons had a known energy. This technique, called "electron ionization" or in the older literature, "electron impact", resulted in many determinations which were unreliable because of the energy spread of the electrons in a conventional beam. Several approaches have been utilized to overcome this problem; the most successful has been the use of a so-called "electron monochromator", in which the energy of the electron beam is narrowly defined by passing the beam through electron energy selectors of various designs^{31,36,37,38,39}. Results obtained using electron beams with well-defined energies are in excellent agreement with analogous results derived from determinations of photoionization thresholds. At this writing, reliable data on ion thermochemistry are being obtained from experiments of this sort. Although studies are still being published which report ionization energy and appearance energy data from less accurate electron ionization techniques, the intent of the authors of those studies is rarely to examine the thermochemistry of the ionization process.

In the powerful threshold technique known as photoelectron-photoion coincidence (PEPICO)³², the thermochemistry and detailed mechanism of an ionic fragmentation process can be mapped out very accurately. Ejected electrons which originated with "zero" kinetic energy are matched with their corresponding positive ions. At energies where parent ions, M^+ , are undergoing dissociation to form one or more fragment ions, one obtains the relative probabilities for the formation of the daughter ions from parent ions of known energy (i.e. the breakdown curve). The ions can be detected at differing times after the ionization event for the determination of the time dependence of the dissociation process. The complete interpretation of such data requires a modeling of the dissociation using statistical theories of unimolecular decomposition (i.e. quasi-equilibrium/RRKM theory)^{40,41}. As pointed out by Danacher in a recent review³², in spite of its great strengths, this technique has not been widely utilized, possibly because of the intricate instrumentation required, the complexity of the data analysis, and the fact that each determination requires the investment of a great amount of time on the part of the experimentalist.

To summarize, intrinsic problems associated with threshold determinations of ionization energies are: a) the difficulty of detecting the onset when there is a large change of molecular geometry in the ionization process, as discussed in Sec. 2.1.; and, b) the observation of ionization at energies below the adiabatic ionization energy when there is a significant population of vibrationally excited molecules in the system ("hot bands").

2.3.4. Photoelectron Spectroscopy

It is also possible to determine the energy change associated with ionization process 1 by effecting ionization with a photon of well-defined energy and measuring the energy of the ejected electrons:



where

$$KE(e) = h\nu - I - E^*(\text{vib,rot}) \quad (36)$$

(where $E^*(\text{vib,rot})$ is the internal energy of M^+ and I is the binding energy of the electron).

The most widely-used technique of this type is conventional photoelectron spectroscopy⁴² in which the photon sources are usually the helium resonance lines of 58.4331 nm (21.218 eV) or 30.3781 nm (40.813 eV); some work is done with neon resonance lines (73.589 nm and 74.370 nm, 16.848 and 16.671 eV) or other intense monochromatic sources. In such an experiment, the ejected electrons will have differing energies depending on the distribution of energy levels in the M^+ ions formed; a map of the abundances of the electron as a function of energy is called the photoelectron spectrum. As described in Sec. 2.1., the shapes of the photoelectron bands will reflect not only the energy differences in the different states of M^+ but the $M \rightarrow M^+$ transition proba-

bilities as governed by the Franck-Condon principle. In cases where the equilibrium geometry of the ion and the corresponding neutral are the same or are similar, it is found that the observed onset of the first photoelectron band is usually a reliable indicator of the adiabatic ionization potential (see Fig. 1).

2.3.5. Ion/Molecule Equilibrium Constant Determinations

This evaluation takes into account (although previous works in the series did not) all information on ion thermochemistry generated by ion/molecule equilibrium constant determinations.

An ion/molecule equilibrium:



is established in a high pressure mass spectrometer⁴³, flow tube⁴⁴, or ion cyclotron resonance spectrometer⁴⁵, and the equilibrium constant is determined by observing the relative abundances of the two ions, A^+ and C^+ , after a large number of collisions:

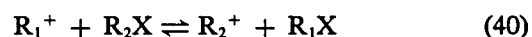
$$K_{\text{eq}} = \frac{[C^+][D]}{[A^+][B]} \quad (38)$$

The neutral reactants, B and D, are present in great abundance compared to the ionic reactants, and therefore, the ratio $[D]/[B]$ does not change as equilibrium is established. A single measurement leads to a value for the Gibbs energy change of reaction 37 at the temperature of the measurement, while a series of measurements at different temperatures permits an experimental evaluation of the entropy and enthalpy changes associated with the reaction:

$$-RT \ln K_{\text{eq}} = \Delta G = \Delta H - T\Delta S \quad (39)$$

In practice, many studies have been published in which measurements were made at a single temperature, the (usually small) entropy change for the reaction was estimated from statistical mechanical considerations (usually just from consideration of changes in symmetry numbers), and the corresponding enthalpy change was derived from these two pieces of information.

Published ion/molecule equilibrium studies involving cations provide data on charge transfer (reaction 33), proton transfer (reaction 6), and hydride or halide transfer equilibria:



(where X is H, F, Cl, Br, or I). Studies of hydride transfer and halide transfer equilibria have led to quantitative information about the relative heats of formation of alkyl carbocations. These data were used to supplement information from appearance potential determinations in evaluating heats of formation of alkyl carbocations.

Most ion/molecule equilibrium studies involving positive ions have been devoted to the derivation of an extensive (more than 100 kcal/mol in length) scale of relative proton affinities (see Eqs. (7) and (8)). The results were mainly derived from interlocking ladders of enthalpy changes for reaction 6. These data have recently been evaluated to establish internal consistency⁴. Most of the values for heats of formation of protonated molecules given in this evaluation are taken from that publication. When this is the case, no specific literature reference is given, it being understood that the source is the evaluated compilation⁴.

As noted above (reaction 33), in determinations of charge transfer equilibrium constants, the difference in the ionization energies of two reacting molecules is obtained. A thermochemical ladder of relative ionization energies determined in this way²² closely reproduces the equivalent scale of spectroscopic ionization energies, thus demonstrating the reliability of the approach for deriving information on relative ionization energies. The most useful application of this approach for ionization energy data has proved to be the determination of ionization energies for species which undergo a large change of geometry upon ionization (case 3 in Fig. 1), and which therefore exhibit very slow onsets of ionization as a function of energy. For example, the only reliable data on the adiabatic ionization energies of *n*-alkanes⁴⁶ and of alkyl hydrazines^{47,48} come from thermochemical ladders established through equilibrium constant determinations.

The main uncertainty associated with this technique, aside from the necessity of relating the thermochemical ladder to a reliable comparison standard, is the temperature of the reacting system. However, the reproduction of relative spectroscopic ionization energies through equilibrium measurements²² demonstrates that this is not a serious problem.

2.3.6. Ion/Molecule Bracketing Experiments

There are some ion/molecule systems for which an equilibrium can not be established in an ion source, either because one of the relevant neutral species is unstable (e.g. a radical or unstable molecule) or because of competing reactions in the system. In such cases, it is sometimes possible to obtain an experimental estimate of the enthalpy change of a particular reaction (charge transfer, proton transfer, hydride transfer, etc.) by use of a technique known as "bracketing" in which the ion of interest is reacted with a series of molecules chosen for variations in the relevant thermochemical parameter (proton affinity, ionization energy, etc.). The occurrence, and sometimes the rate constant, of reaction is monitored as a function of the parameter of interest; the approximate onset energy is usually assumed to lie on the energy scale at a point where the rate of reaction becomes very slow. Few data in this work are derived from such measurements, but in cases where heats of

formation are derived from this kind of experiment, a specific comment describes the experiment.

2.3.7. Onsets of Endothermic Reactions

Several pieces of data given here have been derived from an analysis of the enthalpy changes of endothermic ion/molecule reactions. Although some such information has been obtained from straightforward kinetic treatments (Arrhenius plots) of the temperature dependences of the rate constants of endothermic ion/molecule reactions^{49,50} recent quantitative studies^{51,52,53} cover a much broader energy range by generating a beam of energy- and mass-selected ions which is focussed into a collision chamber containing the reactant gas; product ions are detected as a function of the energy of the ions in the beam.

2.3.8. Other Techniques

Essentially all of the ionization potentials and heats of formation of positive ions included in this evaluation have been derived from results obtained using the experimental approaches listed above. Several additional techniques (Auger electron spectroscopy, Penning ionization, Born-Haber cycle calculations, and analyses of so-called charge transfer spectra) were described in the Introduction to the 1977 compilation^{2,31} but are not widely used for the quantitative determination of data of interest to this compilation. Such data, when available, have been taken into account in the evaluation, except for ionization energies derived from charge transfer spectra. The latter technique is mainly used for obtaining values for ionization potentials of compounds of low vapor pressure. Since the cut-off date for inclusion of literature in the 1977 volume, numerous quantitative determinations of ionization energies for such species, mainly by photoelectron spectroscopy or by ion/molecule equilibrium constant determinations, have appeared in the literature. These have made the charge transfer spectra data obsolete for many species. Since it is generally seen that the gas phase ionization potentials derived from charge transfer spectra may be very inaccurate, all these data have been ignored in the present volume.

2.4. Reliability of Ionization Energy Data and Criteria for Evaluation

2.4.1. Comparisons between Results of Different Techniques

The data on ionization energies summarized here are derived from the different types of measurements described above, and are consequently of widely varying quality, not only because the accuracies of the measurement techniques differ, but also because of differences in the focusses of the research in which the measurements were made. For example, many of the ionization energies reported for inorganic species were never intended by the original authors to be quantitative ionization

energy measurements, but are simply qualitative indicators of whether or not a given ion observed in the vapor over a heated Knudsen cell has been formed by electron impact ionization of the corresponding neutral species (in which case it exhibits an onset at a relatively low energy) or through fragmentation of a molecular ion (which would correspond to a higher onset energy). In these experiments, error limits of 0.5 to 1 eV are commonly cited by the original authors. Similarly, most photoelectron spectroscopic studies are carried out for the purpose of examining molecular orbital energy levels; thermochemistry is not a concern, and often, although the accuracy of the measurements is very high, only vertical ionization energies, which are not necessarily related to thermochemical onsets, are reported.

Because many of the values for ionization energies given here are derived from evaluations of several different determinations carried out using different techniques, there is no specific indication in Table 1 of an experimental method associated with a particular value. In carrying out the evaluation, an attempt was made to integrate the entire corpus of information about any given ion, giving weight to various determinations depending on the nature of the ionization onset, the measurement techniques used, the attention to detail by the original authors, and so forth. Usually (but not always) a spectroscopically-determined ionization energy was considered more reliable than a contradictory value obtained by observation of an ionization threshold. A value obtained from an observed ionization onset using photoionization or an electron monochromator was considered more reliable than an onset obtained using less accurate techniques. In all of these cases, an observed onset of a photoelectron band was given great weight in carrying out the analysis, with values from any of the above three techniques being downgraded if they did not match the photoelectron onset (unless, of course, the differences could be rationalized in terms of the principles outlined above).

As mentioned above, many photoelectron spectroscopy studies do not cite values for adiabatic ionization potentials. In these cases, where the authors have provided a figure showing the photoelectron spectrum, it is usually possible to estimate from the figure the value for the adiabatic onset; where adiabatic ionization energies have been obtained in this way, a specific comment to that effect is made.

Data derived from ion/molecule equilibrium constant determinations have been utilized as an aid in evaluating information obtained from other sources. For example, where scales of relative ionization energies were available from equilibrium constant determinations, internal consistency with these scales was required in the assigned ionization energy or heat of formation values. Where this was not possible, a specific comment spells out the discrepancy. As described in Sec. 2.3.5., ionization energy values derived from equilibrium constant determinations provide the only values for ionization energies of species which undergo large changes of ge-

ometry upon ionization such as normal alkanes with six or more C-atoms⁴⁶, or hydrazines^{47,48}. When an ionization energy has been obtained solely from this approach, the source of the data is indicated in a comment, and the identity of the reference compound is given.

Heats of formation of protonated molecules derived from the evaluated proton affinity scale⁴ are taken from that publication. More recent data are included, with the internal consistency requirement rigorously maintained. The value for the corresponding proton affinity of the molecule is given in a comment. Note that to locate a value for a *proton affinity*, one must look under the empirical formula of the corresponding *protonated molecule*, i.e. the proton affinity of methane is located by looking under CH₅. When data from recent publications are given, the literature source is specifically cited.

2.4.2. Reliability of Data; Error Limits

Ionization Energies. The experimentally-determined ionization energies collected here display widely varying uncertainties, ranging from ± 0.0001 eV or smaller for some spectroscopic or multiphoton-laser determinations to ± 1 eV for measurements carried out on the vapor above a heated Knudsen cell. The error limits associated with a particular ionization energy are specifically listed when the original work(s) gave an estimate of this quantity. In other cases, the error limits are indicated by the number of significant figures displayed; in these cases, it can be assumed that the error limits are five times the last significant figure displayed.

Some of the ionization energy values are shown enclosed in parentheses. Data enclosed in parentheses are considered not to be firmly established for one of three reasons:

(1) The measurement itself must be considered unreliable (as in, for example, threshold determinations in which the energy spread of the electrons was not well defined).

(2) The relevant ionization energy has been determined more than once but with poor agreement between the different results, and there is no auxiliary information available which allows a choice between the divergent values. In such cases, the evaluation gives either (a) the value determined by the most reliable technique, or (b) an average of two or more values determined by the same technique, with error limits indicating the scatter in the data. In a very few cases where the scatter in the reported values is very great or where the value obtained by the "most reliable" technique appears to be specious, no evaluated ionization energy is cited, but a note is included which lists the various determined values.

(3) Parentheses are also used to indicate data which are unevaluated. That is, when a particular molecule has been studied only once, and additional information which would permit one to judge the reliability of the data is unavailable, the ionization energy is given exactly as it appears in the original reference but is enclosed in

parentheses. Many of these untested determinations are undoubtedly reliable; the cited error limits and the number of significant figures shown in the table will give an indication of the probable reliability of the technique by which such a value was obtained.

As described above, some ionization energy values were obtained by reading onsets of photoelectron bands in figures reproduced in papers, where the original authors did not assign a numerical value to the band onset. In every such case, a specific comment is made indicating that the value has been derived from a figure. The accuracy with which such onsets can be read should be assumed to be not better than 0.1–0.2 eV, except where the authors have given an enlarged view of the band onset, in which case, an additional significant figure is cited. When a figure was not given, the lowest vertical ionization potential from the original paper is cited as the upper limit to the adiabatic ionization energy.

Heats of Formation. The cited heats of formation of ions necessarily reflect both the uncertainties in the ionization (or appearance) energy values and the uncertainties in the heats of formation of the relevant neutral species. Values of ionic heats of formation which are not firmly established - either because of a poorly established ionization/appearance energy or because of large uncertainties in the heat of formation of the neutral species - are shown enclosed in parentheses.

Although the values which were used for heats of formation of neutral species will be discussed separately in Sec. 4, it should be emphasized here that many of these data are based on estimates. Some of the estimation schemes for particular classes of compounds are sophisticated and well-documented, and can be considered to lead to values for heats of formation which are as reliable as most experimental data. Other estimations have been carried out by various authors with varying degrees of attention to complexities, or in some cases, with little or no documentation about how the estimate was accomplished. A large fraction of the estimates used were made specifically for this publication, and even among this fraction, there is a broad spectrum of quality depending on the size of the network of related information which was available. Rather than try to sort out and make judgments about the quality of each estimate of the heat of formation of a neutral molecule, the policy has been followed of enclosing in parentheses each ion heat of formation based on an estimated value for the heat of formation of relevant neutral molecules or radicals; this practice is not meant to disparage the quality of the estimated data, but simply to alert the reader to the fact that it is being used. As a first approximation, the user can assume that the reliability of an estimate varies inversely with the complexity of the molecule.

2.5. Trends in the Data

2.5.1. Estimation Schemes for Heats of Formation of Cations

Within the past few years, a sufficient amount of reliable information on ionization energies and heats of for-

mation of many classes of positive ions has become available so that regular trends as a function of molecular size and structure can be discerned. These can be used to develop empirical schemes for estimating ionization energies and/or heats of formation of cations. Since ionization energies for a homologous series do not have a linear dependence on molecular size, values for heats of formation of ions can not be reproduced satisfactorily by simple additivity systems like those in widespread use for the prediction of thermochemical data for neutral molecules. The predictive schemes put forward to date utilize equations which are empirical.

One series of several papers^{54,55,56,57} presents a scheme which is designed to predict values for the heat of formation of positive ions at 298 K from equations of the form:

$$\Delta_f H(M^+) = A - Bn + C/n \quad (41)$$

where A , B , and C are constants derived from the data for any particular series, and n is the total number of atoms in the molecule. The parameters derived in the paper of Holmes, Fingas, and Lossing⁵⁴ for predicting heats of formation of the parent ions of several common classes of compounds are listed in Table 2.5.1.1.

This method works because to an excellent approximation, the ionization energies of a homologous series vary linearly as n^{-1} , as expressed in the term C/n in Eq. (41). The other two terms, A and Bn , reflect the additive nature of heats of formation of neutral molecules. Also, for molecules in which there is multiple substitution by characteristic groups on charge-bearing atoms or at the position of charge delocalized pi-electron systems, good straight-line relationships exist between ionic heats of formation and the logarithm of the number of atoms (i.e. ion size). Such correlations permit reasonably accurate estimates of ion enthalpies of formation^{54,55,56,57}.

Bachiri, Mouvier, Carlier, and DuBois⁵⁸ have advanced a scheme for the estimation of ionization energies of alkenes, alkynes, aldehydes, ketones, alcohols, ethers, mercaptans, and thioethers. Their empirical equation takes the form:

$$\log_{10} \frac{IP(R_1XR_2) - IP_\infty}{IP_0 - IP_\infty} = 0.106[I(R_1) + I(R_2)] \quad (42)$$

where X is a functional group (i.e. $-\text{CH}=\text{CH}-$ or $>\text{C}=\text{CH}_2$ for alkenes, $-\text{O}-$ for alcohols and ethers, $>\text{C}=\text{O}$ for aldehydes and ketones, etc.), R_1 and R_2 are the attached alkyl groups, IP_0 is the ionization potential of the reference compound for which $R_1 = R_2 = \text{H}$. IP_∞ in Eq. (42) is a constant for each compound type. (A modification of this scheme which does away with the need for the parameter IP_∞ has also been put forward recently⁵⁹). Table 2.5.1.2. lists the constants for the alkyl substituent groups and the different compound types (modified slightly from the values given in the original publication to predict adiabatic rather than vertical ionization energies).

TABLE 2.5.1.1. Estimation scheme^c of Holmes, Fingas, and Lossing⁵⁴:
 $\Delta_f H(M^+) \text{ kcal/mol} = A - Bn + C/n$

| Compound Type | kcal/mol ^a | | | Correction Terms |
|-------------------|-----------------------|------|-----|---|
| | A | B | C | |
| Alkanes | 224 | 2.2 | 298 | For each branch: -3 |
| 1-Alkenes | 231.6 | 1.61 | 110 | For each branch on C-2: -13 For each remote branch: -2.5 |
| x-Alkenes | 219.6 | 1.61 | 110 | For each branch on =C: -13 For each branch elsewhere: -2.5 One cis correction: +1 Two cis corrections at one double bond: +3 If one group is t-butyl: +4 If both groups are t-butyl: +10 |
| 1-Alkynes | 278 | 1.57 | 110 | For each branch: -4 |
| 2-Alkynes | 260 | 1.58 | 110 | For each branch: -4 |
| 3-Alkynes | 257 | 1.57 | 110 | For each branch: -4 |
| 4-Alkynes | 257 | 1.57 | 110 | For each branch: -4 |
| 5-Alkynes | 256 | 1.57 | 110 | For each branch: -4 |
| Alkanols | 175 | 1.59 | 216 | For each branch adjacent to -OH: -6 For each branch elsewhere: -2 |
| Aliphatic ethers | 157 | 1.41 | 368 | For each branch adjacent to -O-: -6 For each branch elsewhere: -3 ^b Asymmetry correction per carbon: +1 |
| Aliphatic | 188 | 1.65 | 135 | For each branch adjacent to C=O: -5 For each branch elsewhere: -3 |
| Aliphatic ketones | 166 | 1.78 | 252 | For each branch adjacent to C=OD: -3.5 ^b Asymmetry correction per carbon: +1.5 |
| Alkanoic acids | 142 | 1.90 | 112 | For each branch adjacent to C=O: -3.5 For each branch elsewhere: -1.5 |
| Chloroalkanes | 236 | 1.98 | 57 | For each branch adjacent to halogen: -5 For each branch elsewhere: -3 |
| Bromoalkanes | 219 | 1.40 | 115 | For each branch adjacent to halogen: -5 For each branch elsewhere: -3 |
| Iodoalkanes | 222 | 1.69 | 44 | For each branch adjacent to halogen: -5 |

^aConstants are given here in the units used in the original paper⁵⁴.

^bAsymmetry correction for ethers and ketones having different numbers of C-atoms on either side of the functional group is based on the smallest numbers of C-atoms which must be transferred to give the most symmetrical species, e.g. for methyl pentyl ketone, +3 kcal/mol.

^c n is the total number of atoms in the molecule.

A comparison of ionization energy values⁵⁸ or heats of formation of cations⁵⁴ predicted from expressions 41 or 42 with the corresponding evaluated experimental values is given in Table 2.5.1.3. For both predictive schemes, the agreement between estimated values and experiment is generally quite good — good enough to inspire confidence in the use of the equations for filling in blanks in the data series.

As pointed out in one of the papers advancing these empirical estimation schemes⁵⁴ the equations are “not only useful for predicting new $\Delta_f H$ values, but also for revealing misfits which could indicate incorrect values for $\Delta_f H^\circ$ (Neutral) or the ionization energy, or, more interestingly, an ion structure having special stabilizing or destabilizing properties.” In fact, the trends described by these equations were routinely examined in evaluating the data for just these reasons.

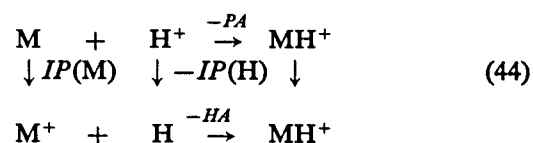
2.5.2. Correlations of Ionization Energies with Proton Affinities or Substituent Constants

The proton affinity of molecule M, defined by Eq. 7, is equal to the M-H⁺ bond energy of the MH⁺ ion. The

M⁺-H bond energy is called the *hydrogen affinity* (HA) of M⁺:



Consider the thermodynamic cycle, constructed from reactions 7 and 43:



From cycle 44 we write:

$$\begin{aligned}
 PA(M) &= HA(M^+) + IP(H) - IP(M) \\
 &= HA(M^+) + 13.6 \text{ eV} - IP(M) \quad (45)
 \end{aligned}$$

If the hydrogen affinity were a constant for a given compound type, the proton affinity values would vary linearly with the ionization potentials for a homologous

TABLE 2.5 1.2 Estimation scheme of Bachiri, Mouvier, Carlier, and DuBois⁵⁸:

$$\log_{10} \frac{IP(R_1XR_2) - IP_0}{IP_0 - IP_0} = 0.106[I(R_1) + I(R_2)]$$

| X | | IP ₀ (eV) | IP ₀ (eV) |
|--------------------|--|----------------------|----------------------|
| -C≡C- | (Alkynes) | 11.400 | 6.577 |
| -HC=CH- | (Alkenes) | 10.507 | 6.849 |
| >C=CH ₂ | (Alkenes, gem) | 10.737 | 6.814 |
| -(C=O)-H | (Aldehydes except CH ₂ O) | 12.063 | 3.575 |
| >C=O | (Ketones) | 13.334 | 3.936 |
| -OH | (Alcohols) | 12.607 | (3.7) |
| -O- | (Ethers) | 12.612 | 5.483 |
| -S- | (H ₂ S, Thiols, Thioethers) | 10.473 | 5.725 |

| R | I |
|--|----------------|
| H | 0 (Convention) |
| Methyl | 1 (Convention) |
| Ethyl | 1.166 |
| n-Propyl | 1.271 |
| i-Propyl | 1.291 |
| n-Butyl | 1.330 |
| s-Butyl | 1.400 |
| t-Butyl | 1.358 |
| n-Pentyl | 1.394 |
| i-Pentyl | 1.340 |
| neo-Pentyl | 1.389 |
| t-Pentyl | 1.369 |
| s-Pentyl [-CH(C ₂ H ₅) ₂] | 1.479 |
| n-Hexyl | 1.462 |
| t-Hexyl [-C(CH ₃) ₂ (n-C ₃ H ₇)] | 1.355 |
| t-Hexyl [-C(CH ₃) ₂ (i-C ₃ H ₇)] | 1.524 |
| neo-Hexyl [-CH ₂ CH ₂ C(CH ₃) ₃] | 1.570 |
| | 1.360 |

TABLE 2.5 1.3. Comparison of ionization energies/heats of formation with estimated values predicted from estimation schemes^a

| Compound | IP (eV) | Δ _f H/(Ion) (kJ/mol) | Holmes <i>et al.</i> ⁵⁴ | | Bachiri <i>et al.</i> ⁵⁸ | |
|---|------------|------------------------------------|------------------------------------|-----------------------------------|-------------------------------------|-----------------------------------|
| | | | IP (eV) | Δ _f H(Ion) (kJ/mol) | IP (eV) | Δ _f H(Ion) (kJ/mol) |
| Alkynes | | | | | | |
| CH ₃ C≡CH | 10.36 | 1186 | [10.34] | 1184 | 10.36 | [1184] |
| C ₂ H ₃ C≡CH | 10.178 | 1147 | [10.13] | 1142 | 10.21 | [1151] |
| n-C ₃ H ₇ C≡CH | 10.05 | 1113 | [10.04] | 1113 | 10.11 | [1121] |
| n-C ₄ H ₉ C≡CH | (9.95) | (1079) | [10.02] | 1088 | 10.06 | [1092] |
| n-C ₃ H ₁₁ C≡CH | (10.04) | (1071) | [9.93] | 1063 | 10.06 | [1075] |
| n-C ₄ H ₁₃ C≡CH | (9.95) | (1038) | [9.93] | 1038 | 10.04 | [1046] |
| i-C ₃ H ₇ C≡CH | 9.97 | 1096 | [9.95] | 1096 | 10.10 | [1109] |
| i-C ₃ H ₇ C≡CCH ₃ | 9.31 | 996 | [9.32] | 996 | 9.33 | [996] |
| t-C ₄ H ₉ C≡CH | (9.80) | (1050) | [9.80] | 1050 | 10.01 | [1071] |
| CH ₃ C≡CCH ₃ | 9.562 | 1068 | [9.55] | 1067 | 9.54 | [1067] |
| C ₂ H ₅ C≡CCH ₃ | 9.44 | 1038 | [9.43] | 1038 | 9.42 | [1038] |
| n-C ₃ H ₇ C≡CCH ₃ | 9.366 | 1013 | [9.37] | 1013 | 9.35 | [1013] |
| n-C ₄ H ₉ C≡CCH ₃ | (9.33) | (983) | [9.37] | 987 | 9.31 | [983] |
| n-C ₃ H ₁₁ C≡CCH ₃ | 9.31 | 962 | [9.32] | 962 | 9.30 | [958] |
| n-C ₄ H ₁₃ C≡CCH ₃ | (9.30) | (941) | [9.28] | 941 | 9.29 | [941] |
| C ₂ H ₅ C≡CC ₂ H ₅ | 9.323 | 1004 | [9.28] | 1000 | 9.31 | [1004] |
| n-C ₃ H ₇ C≡C ₂ H ₅ | (9.26) | 975) | [9.24] | 975 | 9.24 | [975] |
| n-C ₄ H ₉ C≡CC ₂ H ₅ | 9.22 | 954 | [9.19] | 950 | 9.20 | [950] |
| n-C ₃ H ₁₁ C≡CC ₂ H ₅ | 9.20 | 929 | [9.19] | 929 | 9.19 | [929] |
| n-C ₄ H ₁₃ C≡CC ₂ H ₅ | 9.19 | 908 | [9.19] | 908 | 9.18 | [908] |

TABLE 2.5.1.3. Comparison of ionization energies/heats of formation with estimated values^a — Continued

| Compound | <i>IP</i> (eV) | $\Delta_f H^{\circ}(\text{Ion})$ (kJ/mol) | Holmes <i>et al.</i> ⁵⁴ | | Bachiri <i>et al.</i> ⁵⁸ | |
|--|-------------------|--|------------------------------------|--|-------------------------------------|--|
| | | | <i>IP</i> (eV) | $\Delta_f H^{\circ}(\text{Ion})$ (kJ/mol) | <i>IP</i> (eV) | $\Delta_f H^{\circ}(\text{Ion})$ (kJ/mol) |
| Alkenes | | | | | | |
| CH ₃ CH=CH ₂ | 9.73 | 958 | [9.73] | 958 | 9.71 | [958] |
| C ₂ H ₃ CH=CH ₂ | 9.58 | 925 | [9.59] | 925 | 9.60 | [925] |
| n-C ₃ H ₇ CH=CH ₂ | 9.52 | 895 | [9.54] | 900 | 9.53 | [895] |
| n-C ₄ H ₉ CH=CH ₂ | 9.44 | 870 | [9.48] | 874 | 9.49 | [874] |
| n-C ₅ H ₁₁ CH=CH ₂ | 9.44 | 849 | [9.45] | 849 | 9.49 | [853] |
| n-C ₆ H ₁₃ CH=CH ₂ | 9.43 | 828 | [9.41] | 828 | 9.48 | [833] |
| i-C ₃ H ₇ CH=CH ₂ | 8.96 | 812 | [8.94] | 812 | 8.92 | [808] |
| t-C ₄ H ₉ CH=CH ₂ | 9.45 | 849 | [9.43] | 853 | 9.45 | [853] |
| (C ₂ H ₅) ₂ C=CH ₂ | 9.06 | 820 | [9.06] | 820 | 9.03 | [816] |
| E-C ₂ H ₃ CH=CHCH ₃ | 9.100 | 866 | [9.12] | 866 | 9.09 | [866] |
| E-C ₂ H ₃ CH=CHCH ₃ | 9.036 | 840 | [9.05] | 841 | 9.00 | [837] |
| E-n-C ₃ H ₇ CH=CHCH ₃ | (8.97) | (812) | [8.99] | 816 | 8.95 | [812] |
| i-C ₃ H ₇ CH=CHCH ₃ | 8.97 | 803 | [8.96] | 803 | 8.94 | [803] |
| E/Z-n-C ₄ H ₉ CH=CHCH ₃ | (8.84) | (782) | [8.92] | 791 | 8.92 | [791] |
| E-n-C ₅ H ₁₁ CH=CHCH ₃ | 8.85 | 757 | [8.97] | 766 | 8.91 | [761] |
| C ₂ H ₃ CH=CHC ₂ H ₅ | 8.96 | 812 | [8.94] | 812 | 8.92 | [808] |
| (CH ₃) ₂ C=C(CH ₃) ₂ | 8.27 | 728 | [8.04] | 707 | 8.23 | [724] |
| Alcohols | | | | | | |
| CH ₃ OH | 10.85 | 845 | [10.82] | 845 | 10.68 | [828] |
| C ₂ H ₅ OH | 10.47 | 774 | [10.45] | 774 | 10.40 | [770] |
| n-C ₃ H ₇ OH | 10.22 | 732 | [10.20] | 728 | 10.23 | [732] |
| n-C ₄ H ₉ OH | 10.06 | 695 | [10.03] | 695 | 10.14 | [703] |
| n-C ₅ H ₁₁ OH | 10.00 | 669 | [9.96] | 661 | 10.12 | [678] |
| n-C ₆ H ₁₃ OH | (9.89) | (640) | [9.86] | 636 | 10.10 | [661] |
| i-C ₃ H ₇ OH | 10.12 | 703 | [10.10] | 703 | 10.20 | [711] |
| s-C ₄ H ₉ OH | 9.88 | 661 | [9.96] | 669 | 10.03 | [674] |
| i-C ₄ H ₉ OH | 10.09 | 690 | [10.03] | 686 | 10.09 | [690] |
| t-C ₄ H ₉ OH | 9.97 | 648 | [9.90] | 644 | 10.09 | [661] |
| Ethers | | | | | | |
| CH ₃ OCH ₃ | 10.025 | 782 | [9.94] | 774 | 9.86 | [766] |
| C ₂ H ₅ OCH ₃ | 9.72 | 720 | [9.77] | 715 | 9.69 | [717] |
| n-C ₃ H ₇ OCH ₃ | — | — | 9.58 | — | — | — |
| i-C ₃ H ₇ OCH ₃ | (9.42) | 657 | [9.41] | 657 | 9.56 | [669] |
| n-C ₄ H ₉ OCH ₃ | (9.54) | (661) | [9.32] | 640 | 9.52 | [661] |
| n-C ₃ H ₇ OC ₂ H ₅ | (9.45) | (640) | [9.46] | 640 | 9.42 | [636] |
| C ₂ H ₅ OC ₂ H ₅ | 9.51 | 665 | [9.56] | 669 | 9.52 | [665] |
| n-C ₄ H ₉ OC ₂ H ₅ | 9.36 | 611 | [9.36] | 611 | 9.36 | [611] |
| t-C ₄ H ₉ OCH ₃ | 9.41 | 619 | [9.26] | 602 | 9.46 | [623] |
| Aldehydes | | | | | | |
| CH ₃ CHO | 10.229 | 821 | [10.21] | 820 | 10.23 | [820] |
| C ₂ H ₃ CHO | 9.953 | 773 | [9.97] | 774 | 9.96 | [774] |
| n-C ₃ H ₇ CHO | 9.84 | 741 | [9.82] | 741 | 9.80 | [736] |
| n-C ₄ H ₉ CHO | 9.74 | 711 | [9.76] | 711 | 9.71 | [707] |
| n-C ₅ H ₁₁ CHO | 9.67 | 686 | [9.66] | 685 | 9.69 | [688] |
| i-C ₃ H ₇ CHO | 9.705 | 721 | [9.69] | 719 | 9.77 | [727] |
| i-C ₄ H ₉ CHO | 9.70 | 699 | [9.71] | 700 | 9.67 | [695] |
| s-C ₄ H ₉ CHO | (9.59) | (690) | [9.58] | 690 | 9.61 | [692] |
| t-C ₄ H ₉ CHO | 9.50 | 674 | [9.45] | 669 | 9.62 | [686] |
| neo-C ₅ H ₁₁ CHO | (9.61) | (661) | [9.61] | 661 | 9.65 | [665] |
| Ketones | | | | | | |
| CH ₃ COCH ₃ | 9.705 | 719 | [9.77] | 724 | 9.704 | [761] |
| C ₂ H ₅ COCH ₃ | 9.51 | 678 | [9.53] | 680 | 9.48 | [675] |
| n-C ₃ H ₇ COCH ₃ | 9.38 | 644 | [9.40] | 646 | 9.33 | [639] |
| n-C ₄ H ₉ COCH ₃ | 9.35 | 628 | [9.26] | 619 | 9.26 | [619] |
| C ₂ H ₅ COC ₂ H ₅ | 9.31 | 640 | [9.32] | 642 | 9.26 | [636] |

TABLE 2.5.1.3. Comparison of ionization energies/heats of formation with estimated values^a — Continued

| Compound | IP (eV) | $\Delta_f H^\circ(\text{Ion})$ (kJ/mol) | Holmes <i>et al.</i> ⁵⁴ | | Bachiri <i>et al.</i> ⁵⁸ | |
|---|------------|--|------------------------------------|--|-------------------------------------|--|
| | | | IP (eV) | $\Delta_f H^\circ(\text{Ion})$ (kJ/mol) | IP (eV) | $\Delta_f H^\circ(\text{Ion})$ (kJ/mol) |
| Ketones - Continued | | | | | | |
| n-C ₃ H ₇ COCH ₃ | 9.12 | 598 | [9.25] | 611 | 9.12 | [598] |
| n-C ₄ H ₉ COCH ₃ | (9.02) | (573) | [9.14] | 586 | 9.05 | [577] |
| i-C ₃ H ₇ COCH ₃ | 9.30 | 636 | [9.21] | 628 | 9.31 | [636] |
| i-C ₃ H ₇ COC ₂ H ₅ | (9.10) | (594) | [9.15] | 598 | 9.10 | [590] |
| (i-C ₃ H ₇) ₂ CO | 8.95 | 552 | [8.92] | 548 | 8.94 | [552] |
| s-C ₄ H ₉ COCH ₃ | 9.21 | 598 | [9.24] | 602 | 9.17 | [594] |
| t-C ₄ H ₉ COCH ₃ | 9.30 | 607 | [9.43] | 619 | 9.22 | [602] |
| t-C ₄ H ₉ COCH ₃ | 9.11 | 590 | [9.14] | 590 | 9.17 | [594] |
| neo-C ₅ H ₁₁ COCH ₃ | (9.23) | (573) | [9.29] | 577 | 9.21 | [569] |

^aIn these lists, values obtained through the use of the estimation scheme of Holmes *et al.*⁵⁴ are *heats of formation* of ions at 298 K. The scheme of Bachiri *et al.*⁵⁸ predicts *ionization energies*. For purposes of comparison, both quantities are given here, the conversion being made using standard heats of formation of corresponding neutral molecules from Table 1. The derived quantity is enclosed in brackets. Parentheses indicate a quantity which is not well established (see conventions for Table 1).

series, the slope of the plot would be -1 and the intercept would be $[HA(M^+) + 13.6 \text{ eV}]$.

It has been observed^{45a,60,61} that the value which can be assigned to the M^+-H bond strength (i.e. the HA) is indeed often approximately constant for a homologous series, at least over a limited range. For instance, it was reported⁶² that linear plots of PA versus IP for primary, secondary, and tertiary amines display the same slope, but have different intercepts (i.e. different values of HA). However, a detailed statistical analysis⁶³ of the relationships between proton affinities and ionization potentials for many different compound types (alcohols, ethers, primary-, secondary- and tertiary- amines, nitriles, mercaptans, sulfides, aldehydes, ketones, carboxylic acids, esters, amides, and atoms) demonstrated that only the parent radical cations of sulfides and mercaptans displayed a characteristic (constant) value of the hydrogen affinity. For other compound types, it was concluded that the hydrogen affinity itself varies linearly with the ionization energy:

$$HA(M^+) = c + \Delta IP(M) \quad (46)$$

Several series of compounds for which reliable evaluated ionization energy and proton affinity data are both available are summarized in Table 2.5.2.1. along with values for the hydrogen affinities. The published analysis utilized vertical ionization energies corresponding to the orbital of the site of protonation, and a proton affinity scale which, although internally consistent, was constricted in length (due to the incorrect assumption in early equilibrium studies using ICR that the operating temperature was 300 K rather than 320 K) and related to an absolute standard whose proton affinity value has now been revised downward by 3 kcal/mol. Repeating that statistical analysis, but using instead the thermochemically more meaningful adiabatic ionization energies which relate to the M^+-H bond strengths, and the

evaluated scale of proton affinities⁴, it is seen that Eq. (46) does hold for alcohols, aldehydes, ketones, primary amines, cyclic ethers and esters. In the series of aliphatic ethers, thioethers, and secondary and tertiary amines, values of the hydrogen affinity appear to decrease slightly with decreasing ionization energy, but the differences are too small to be meaningful (i.e. the slope of a plot of Eq. (46) is -0.7 or greater), and the assumption that the hydrogen affinity is constant will be approximately valid. The hydrogen affinities of mercaptans and of aromatic amines are indeed constant. Substituting Eq. (46) into Eq. (45), we derive an expression which permits the estimation of an unknown proton affinity/ionization energy when one of these two parameters is known:

$$PA_1 - PA_2 = (d-1)(IP_1 - IP_2) = K(IP_1 - IP_2) \quad (47)$$

where $K = (d-1)$ is the slope of a plot of PA versus IP for a compound series:

$$PA_x = C + KIP_x \quad (48)$$

Values for C and K derived from the statistical analysis of the data are given in Table 2.5.2.1. for those compound types for which sufficient information was available to make a meaningful analysis.

Attention has also been given to relating ionization energies and proton affinities of various series of compounds to the appropriate Taft substituent constants^{64,65,66,67}. It has been shown that the adiabatic ionization energies of compounds RX (where R is an alkyl group) correlate linearly with $\sigma^*(R)$ and $\sigma_1(R)$ (measures of the polarizability and electron-releasing and donating ability of R) for constant electron-withdrawing group X . This is easily understood in terms of a lowering of the energy required to remove an electron with increasing electron-donating ability of the groups, R .

TABLE 2.5.2.1 The relationship between proton affinity, ionization energy, and hydrogen affinity for homologous series

| | kJ/mol | | |
|---|-----------|-----------|-----------|
| | <i>IP</i> | <i>PA</i> | <i>HA</i> |
| $PA(M) = C + K \cdot IP(M)$ $HA(M^+) = c + d \cdot IP(M)$ ($C - c = 1312$ kJ/mol, $d - K = 1.00$) | | | |
| Alcohols: $K = -0.54$, $C = 1335$ kJ/mol | | | |
| CH ₃ OH | 1047 | 761 | 494 |
| C ₂ H ₅ OH | 1010 | 788 | 485 |
| n-C ₃ H ₇ OH | 986 | 798 | 472 |
| n-C ₄ H ₉ OH | 971 | 800 | 456 |
| i-C ₄ H ₉ OH | 974 | 805 | 464 |
| i-C ₃ H ₇ OH | 976 | 800 | 464 |
| s-C ₄ H ₉ OH | 953 | 799 | 439 |
| t-C ₄ H ₉ OH | 958 | 810 | 460 |
| Acyclic Ethers: ($K = -0.77$, $C = 1548$ kJ/mol) | | | |
| CH ₃ OCH ₃ | 967 | 804 | 460 |
| C ₂ H ₅ OCH ₃ | 938 | 822 | 448 |
| C ₂ H ₅ OC ₂ H ₅ | 918 | 838 | 444 |
| (n-C ₃ H ₇) ₂ O | 895 | 846 | 427 |
| (n-C ₄ H ₉) ₂ O | 910 | 852 | 448 |
| (s-C ₄ H ₉) ₂ O | 879 | 874 | 439 |
| t-C ₄ H ₉ OCH ₃ | <908 | 846 | <444 |
| Cyclic Ethers: ($K = -0.40$, $C = 1192$ kJ/mol) | | | |
| c-C ₂ H ₄ O | 1020 | 786 | 494 |
| c-C ₃ H ₆ O | 933 | 824 | 448 |
| c-C ₄ H ₈ O | 908 | 832 | 427 |
| c-C ₅ H ₁₀ O | 892 | 836 | 414 |
| Aldehydes: ($K = -0.50$, $C = 1276$ kJ/mol) | | | |
| CH ₃ CHO | 987 | 781 | 456 |
| C ₂ H ₅ CHO | 960 | 793 | 444 |
| n-C ₃ H ₇ CHO | 949 | 801 | 439 |
| i-C ₃ H ₇ CHO | 936 | 806 | 431 |
| i-C ₄ H ₉ CHO | 936 | 806 | 431 |
| Ketones: ($K = -0.39$, $C = 1188$ kJ/mol) | | | |
| CH ₃ COCH ₃ | 936 | 823 | 448 |
| C ₂ H ₅ COCH ₃ | 917 | 836 | 444 |
| C ₂ H ₅ COC ₂ H ₅ | 898 | 843 | 427 |
| i-C ₃ H ₇ COCH ₃ | 897 | 851 | 435 |
| (i-C ₃ H ₇) ₂ CO | 864 | 857 | 410 |
| t-C ₄ H ₉ COCH ₃ | 879 | 846 | 414 |
| (t-C ₄ H ₉) ₂ CO | 836 | 864 | 389 |
| Primary Amines: ($K = -0.59$, $C = 1406$ kJ/mol) | | | |
| CH ₃ NH ₂ | 866 | 896 | 448 |
| C ₂ H ₅ NH ₂ | 855 | 908 | 452 |
| n-C ₃ H ₇ NH ₂ | 847 | 912 | 448 |
| i-C ₃ H ₇ NH ₂ | 841 | 915 | 444 |
| n-C ₄ H ₉ NH ₂ | 841 | 914 | 444 |
| s-C ₄ H ₉ NH ₂ | 839 | 923 | 448 |
| i-C ₄ H ₉ NH ₂ | 839 | 915 | 444 |
| t-C ₄ H ₉ NH ₂ | 833 | 924 | 444 |
| n-C ₅ H ₁₁ NH ₂ | 836 | 916 | 439 |

TABLE 2 5 2 1 The relationship between proton affinity, ionization energy, and hydrogen affinity for homologous series—Continued

$$PA(M) = C + C \cdot IP(M)$$

$$HA(M^+) = c + d \cdot IP(M)$$

$$(C - c = 1312 \text{ kJ/mol}, d - K = 1.00)$$

| | kJ/mol | | |
|---|--------|-----|-----|
| | IP | PA | HA |
| Secondary Amines: ($K = -0.72$, $C = 1502$ kJ/mol, $HA = 397$ kJ/mol) | | | |
| (CH ₃) ₂ NH | 794 | 923 | 406 |
| (CH ₃)(C ₂ H ₅)NH | 786 | 932 | 406 |
| (C ₂ H ₅) ₂ NH | 773 | 945 | 406 |
| (n-C ₃ H ₇) ₂ NH | 756 | 952 | 397 |
| (i-C ₃ H ₇) ₂ NH | 746 | 963 | 397 |
| (n-C ₄ H ₉) ₂ NH | 742 | 956 | 385 |
| (s-C ₄ H ₉) ₂ NH | 736 | 966 | 389 |
| (i-C ₄ H ₉) ₂ NH | 754 | 956 | 397 |
| Tertiary Amines: ($K = -0.83$, $C = 1573$ kJ/mol, $HA = 385$ kJ/mol) | | | |
| (CH ₃) ₃ N | 754 | 942 | 385 |
| (CH ₃) ₂ (C ₂ H ₅)N | 747 | 952 | 385 |
| (CH ₃)(C ₂ H ₅) ₂ N | 723 | 962 | 372 |
| (C ₂ H ₅) ₃ N | 723 | 972 | 385 |
| (n-C ₃ H ₇) ₃ N | 715 | 979 | 381 |
| Aromatic Amines: ($K = -1.0$, $C = 1636$ kJ/mol, $HA = 305$ kJ/mol) | | | |
| C ₆ H ₅ NH ₂ | 741 | 877 | 305 |
| C ₆ H ₅ N(CH ₃) ₂ | 687 | 935 | 310 |
| 3-(CH ₃)C ₆ H ₄ N(CH ₃) ₂ | 677 | 939 | 305 |
| 4-(CH ₃)C ₆ H ₄ N(CH ₃) ₂ | 669 | 944 | 301 |
| 3,5-(CH ₃) ₂ C ₆ H ₃ N(CH ₃) ₂ | 671 | 950 | 301 |
| C ₆ H ₅ N(C ₂ H ₅) ₂ | 674 | 952 | 314 |
| Mercaptans: ($K = -0.98$, $C = 1678$ kJ/mol, $HA = 381$ kJ/mol) | | | |
| CH ₃ SH | 911 | 784 | 381 |
| C ₂ H ₅ SH | 896 | 798 | 381 |
| n-C ₃ H ₇ SH | 887 | 802 | 377 |
| i-C ₃ H ₇ SH | 882 | 812 | 381 |
| t-C ₄ H ₉ SH | 871 | 824 | 381 |
| Thioethers: ($K = -0.83$, $C = 1531$ kJ/mol, $HA = 360$ kJ/mol) | | | |
| CH ₃ SCH ₃ | 838 | 839 | 364 |
| C ₂ H ₅ SCH ₃ | 824 | 851 | 364 |
| (C ₂ H ₅) ₂ S | 813 | 858 | 360 |
| (n-C ₃ H ₇) ₂ S | 801 | 864 | 351 |
| (i-C ₃ H ₇) ₂ S | 796 | 877 | 360 |
| (n-C ₄ H ₉) ₂ S | 793 | 873 | 356 |
| (t-C ₄ H ₉) ₂ S | 779 | 890 | 356 |
| Nitriles: | | | |
| CH ₃ CN | 1177 | 788 | 653 |
| C ₂ H ₅ CN | 1142 | 806 | 636 |
| n-C ₃ H ₇ CN | 1129 | 810 | 628 |
| i-C ₃ H ₇ CN | 1133 | 813 | 632 |
| Esters: ($K = -0.58$, $C = 1401$ kJ/mol) | | | |
| HCOOCH ₃ | 1043 | 790 | 523 |
| HCOOC ₂ H ₅ | 1024 | 808 | 519 |
| HCOO(n-C ₃ H ₇) | 1015 | 813 | 515 |
| HCOO(i-C ₃ H ₇) | 1008 | 820 | 515 |
| HCOO(n-C ₄ H ₉) | 1013 | 815 | 515 |
| CH ₃ COOCH ₃ | 991 | 828 | 506 |
| CH ₃ COOC ₂ H ₅ | 966 | 840 | 494 |
| CH ₃ COO(n-C ₃ H ₇) | 969 | 839 | 494 |

TABLE 2.5.2.1. The relationship between proton affinity, ionization energy, and hydrogen affinity for homologous series—Continued

$$PA(M) = C + K \cdot IP(M)$$

$$HA(M^+) = c + d \cdot IP(M)$$

$$(C - c = 1312 \text{ kJ/mol}, d - K = 1.00)$$

| | kJ/mol | | |
|---|-----------|-----------|-----------|
| | <i>IP</i> | <i>PA</i> | <i>HA</i> |
| Esters: (<i>K</i> = -0.58, <i>C</i> = 1401 kJ/mol) | | | |
| C ₂ H ₅ COOCH ₃ | 979 | 838 | 506 |
| n-C ₃ H ₇ COOCH ₃ | 971 | 837 | 498 |
| i-C ₃ H ₇ COOCH ₃ | 951 | 843 | 481 |
| t-C ₄ H ₉ COOCH ₃ | 955 | 849 | 490 |
| Acids: | | | |
| CH ₃ COOH | 1028 | 796 | 510 |
| C ₂ H ₅ COOH | 1015 | 802 | 506 |

3. Negative Ions

The previous publication of evaluated heats of formation of ions, "Energetics of Gaseous Ions"², contains ionization/appearance potential data for over 4000 species leading to evaluated heats of formation for more than 600 positive ions. By contrast, that volume contains thermochemical data for only 117 anions, including only 12 organic (C, H containing) anions. These statistics reflect the relative importance of studies on cation versus anion thermochemistry at the time of the cut-off for the literature search for that volume, 1971.

The large discrepancy in the numbers of early studies on anions as compared to cations is easy to rationalize. Most neutral species display a much lower cross section for production of anions than for cation production, with the necessary consequence that conventional electron impact mass spectrometry is much more adaptable to studies of positive ions. Similarly, the presence of excess energy in a cation can cause fragmentation, with the identities of the fragment cations providing useful structural information. In contrast, loss of an electron from a bound anion to form the corresponding neutral species is often energetically preferred to a dissociation process producing a fragment anion. Thus, the "cation bias" of much of gaseous ion thermochemistry until the last decade is understandable.

The renaissance in gas phase anion chemistry and thermochemistry came about with the development of chemical ionization mass spectrometry as a commonly-used technique. Anions are often more useful than cations for analytical work in that they can originate with less internal energy. In a proton transfer reaction leading to an anion product, the new bond which is formed—with its share of the excess energy of reaction—is in the departing neutral species (reaction 11) while in the analogous reaction involving cations (reaction 6), the new bond is

in the ion. Likewise, thermal electron attachment to those species which form stable radical anions is considerably faster than particle transfer, so that the chemical ionization step can be much more sensitive.

3.1. Aims

The compilation of anion thermochemistry in this work has slightly different aims than the companion cation compilation, in that the latter presents only the "best" available values for the ionization energy/heat of formation of a given structure. A complete archive of the literature having to do with cations has not been given, because the previous compilations^{1,2,3,4} have summarized the literature exhaustively. Although there have been a number of compilations concerned with the thermochemistry of anions in the last few years^{11,68-75} these have not presented data which are critically evaluated, i.e., the best values are not assigned, save for atomic ions⁷⁰. The advances in the last decade in ion/molecule chemistry and in such techniques as photoelectron and photodetachment spectrometry have resulted in a tremendous increase in the number of chemical structures for which some anionic thermochemistry is known. This publication therefore includes a collection of the literature data which is as complete as possible, and an assignment of the "best" value for the thermochemistry where sufficient information is available.

The thermochemical parameters of critical interest in this compilation are the heat of formation of the anion and the electron affinity of the radical or neutral molecule corresponding to the anion. In order to properly evaluate these, however, data on the energetics of chemical processes involving the anions (Brønsted basicity of the anion, parameters for solvation by neutral species, etc.) are also included. The extensive thermochemical ladders of relative acidities, electron affinities,

solvation thermochemistry, and similar data derived from chemical equilibria have provided a powerful tool for evaluating the thermochemistry of anions: chemical intuition. The use of structure-reactivity relationships allows the examination of the structure of an acid and a prediction about what its acidity, and therefore anion heat of formation, should be. The extra thermodynamic techniques such as linear free energy and enthalpy relationships often allow prediction of expected values accurate to better than a kJ/mol. Although results derived from such relationships can not always be trusted in cases of unusual structures, they nevertheless provide a reasonable rationale for assigning "best" values in many cases.

A problem that has become increasingly important recently is the question of the thermochemistry of the allied neutral species. As indicated below in the section on thermochemical relationships, the limiting factor in deriving anion thermochemical data is often the reliability of the data on the related thermochemistry of the neutral species (heats of formation, bond strengths). The information generated by the field of ion chemistry has outpaced the availability of neutral thermochemical data in recent years. In many cases, the best values for certain bond strengths are derived from data on the thermochemical properties of ions, rather than the other way around.

3.2. Experimental Techniques

Detailed descriptions of the various techniques used to obtain anion thermochemical data will not be presented here, since these are well documented in the literature by their practitioners. Brief descriptions of each technique follow, with comments about accuracy and limitations. The phrase in square brackets following the name is the acronym used in the database to refer to the method.

3.2.1. Laser Photoelectron Spectroscopy [LPES]

A fixed frequency laser (commonly 2.54 eV photons) is used to irradiate a beam of anions, and the energies of the detached electrons are analyzed⁷⁰. The method often provides information on the vibrational states of the neutral and ionic species as well. However, the assignment of the (0-0) threshold can be complicated by these states. The precision is commonly better than 0.2 kJ/mol, and can be much better.

3.2.2. Laser Photodetachment [LPD]

In this technique, which may be considered the converse of photoelectron spectroscopy, the laser wavelength is varied to determine the threshold for detachment of a (presumably) thermal electron⁷¹ from an anion. This experiment has usually been carried out in an ICR ion trap, with the decrease in the ICR signal of the ion as the detected quantity; the lower power of variable

wavelength lasers often requires a longer irradiation period than with the ion beam in photoelectron spectroscopy. Precision is ca. 1-4 kJ/mol. The detection of the true threshold is often complicated by a gradual onset, although the general theory of the onset has been worked out^{68,69}. This method actually yields the vertical detachment energy, which is equated with the electron affinity. This assumption is usually valid, but fails for molecules for which the geometries of the anion and neutral are considerably different (i.e., for which there is poor Franck-Condon overlap). A notable case is CF_3^- , where the photodetachment value is larger than the adiabatic value by 0.8 eV⁷⁷.

A recent determination of the spectrum of the hydroxide anion is at a resolution of ca. 2 J/mol⁷⁰ while coaxial LPD for O^- furnishes a resolution of 0.006 cm^{-1} , or 0.07 J/mol⁷⁶.

3.2.3. Photodetachment [PD]

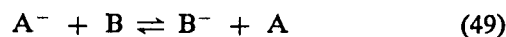
Early photodetachment experiments were carried out using an arc lamp and a monochromator to irradiate the ICR cell⁷¹. Precision was lower than with the laser experiment.

3.2.4. Photodissociation [PDIs]

Irradiation of anions does not always yield electron detachment as the first threshold process since bond cleavage may also be an allowed process. The wavelength threshold for such a process can provide information on the heat of formation of the anion, if the heats of formation of the products are known.

3.2.5. Ion/Molecule Equilibrium Constant Determinations [IMRE, K_{ine}, TDEq, TDAs]

As discussed in Sec. 2.3.5., this evaluation takes into account (although previous works in the series did not) all information on ion thermochemistry generated by ion/molecule equilibrium constant determinations. In the case of anions, ion/molecule equilibrium studies on electron transfer reactions:



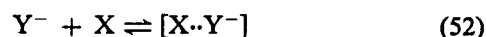
lead to scales of relative electron affinities.

$$K_{\text{eq}} = \frac{[\text{B}^-][\text{A}]}{[\text{A}^-][\text{B}]} \quad (50)$$

while equilibrium constants for hydride or halide transfer reactions:



(where Y is H, F, Cl, Br, or I) lead to thermochemical ladders of relative acidities or halide affinities. Other scales of anionic thermochemistry are derived from equilibrium constants for solvation equilibria:



Further, if the forward and reverse rate constants for a reaction are known, then the equilibrium constant, and thus ΔG , can be calculated from kinetic data [Kine].

The bulk of the available data on anion thermochemistry in the past decade has been derived from ion/molecule equilibrium constant determinations and photoelectron/photodetachment spectroscopy. Extensive scales, spanning an energy range of nearly 400 kJ/mol, have been determined for proton transfer (leading to relative Brønsted acidities of molecules), electron transfer, and halide transfer reactions.

In general, the free energy changes associated with such equilibria are measured to a precision of ca. 0.5 kJ/mol. The absolute uncertainty of anion heats of formation derived from such results is usually on the order of 5–10 kJ/mol, and depends on the accuracy of the method of "anchoring" the resulting scales of relative thermochemical values.

There remain at present several points of uncertainty regarding these data. There is an active debate⁷⁸ about the actual temperature of the ions in an ICR cell. Although some early results indicated that effective ion temperatures could be as much as several hundred degrees above ambient⁷⁸, the accuracy of the kinetic rate constant "thermometer" used as the basis of that judgement was not established. Comparisons of equilibrium constant data obtained in ICR cells with data derived from other sources indicate that the effective ion temperatures in ICR cells are not more than 10 degrees higher than measured gas temperatures in the cells^{22,78}.

The measured equilibrium constant data lead directly to values of free energies, which require some knowledge of the entropy changes of the processes under consideration in order to derive the desired enthalpy changes. Entropy changes have been obtained either through statistical mechanical calculations^{79,80}, or by measuring equilibrium constants as a function of temperature [TDEq = temperature dependent equilibrium constant], leading through a van't Hoff treatment of the results to experimental values for the entropy and enthalpy changes. Finally, the dynamic range (i.e. ion trapping time) of all the mass spectrometric techniques now in use for ion/molecule equilibrium constant determinations is such that the maximum free energy change which can be determined for particle transfer reactions is no greater than ca. 30–40 kJ/mol at most, and often only 10 kJ/mol at room temperature. The dynamic range for determinations of thermochemical parameters of association reactions is much greater [TDAs = temperature dependent association].

There have been questions raised regarding the accuracy of this method, since alcohol bond strengths derived in this way were consistently 9 kJ/mol smaller than accepted values. It was originally thought that this discrepancy was due to the temperature problem alluded to above⁷⁸, because the acidity scale measured in the ICR spectrometer⁸⁰ was compressed relative to that determined by pulsed high pressure mass spectrometry⁷⁹. If the equilibria established in the reaction cell of the ICR

spectrometer were actually at a higher temperature than the value which was used to convert K_{eq} to $\Delta(\Delta_{acid}G)$, then the calculated free energy scale from ICR experiments would be compressed. However, the gas phase basicity scales measured by ICR and by high pressure mass spectrometric methods agree quite well⁴. In addition, Taft⁸¹ has recently redetermined many of the relative acidities that make up the thermochemical ladder, and finds that the region of the acidity scale from trifluoroethanol to acetone has a larger range than the original work indicated. These results have been confirmed in the laboratory of one of the present authors⁸². The region of the acidity scale from trifluoroethanol up to methanol has therefore been adjusted to include these new data. The data affected by this revision are still referred to by the original literature reference, e.g. 79BAR/SCO, and the original values are still displayed, but the method is denoted IMRE°. The revised values are *preferred* in the evaluation.

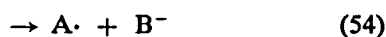
3.2.6. Ion/Molecule Reaction Bracketing [IMRB]

For most of the techniques currently used for studying thermal ion/molecule reaction equilibria and kinetics, ions can only be examined for, at most, several thousand collisions with the reactive neutral gas. Thus, any reaction more endothermic than a few kcal/mol can not be observed on the time scale of the presently used techniques. In the observation of a series of reactions for which the functional groups present at the reactive site of the molecule are always the same, and the energy of the reaction is being varied by changing some distant substituent, then if the rate constant falls to less than the observable rate over some small energy range, it is a fair assumption that the reaction pathway has become endothermic at that point. From this, an estimate of the thermoneutral (equilibrium) point may be made. This technique must be applied with caution, because the mechanism of the observed reaction may not be the same for the entire series of molecules, so that apparent variations in reactivity may not actually reflect the thermochemistry of the assumed reaction.

3.2.7. Electron Impact Appearance Potentials [EIAP]

Since a bound anion must be thermochemically more stable than the combined energies of the free electron plus the neutral species, simple attachment of electrons, even thermal ones, in general results in rapid autodetachment. In certain cases, however, the excited anion state can fragment to yield either an anion plus a neutral species (dissociative attachment), or an anion plus a cation (ion pair production). The latter process has not been well studied save for relatively small species, and is not at present a source of much thermochemical data. On the other hand, a considerable amount of thermochemical data has been derived from experiments in which the onset energy for dissociative attachment is measured. A

complication in the interpretation of such onsets involves the unknown internal energy of both the anionic and neutral fragments. A particularly useful case is where two onsets are observed, with the fragments differing only in the identity of the species associated with the electron:



If the electron affinity of one of the product species is known, that of the other can be inferred from the known electron affinity and the difference in the onset energies for the two channels.

Most workers have not used monoenergetic electron beams, so the precision in the energy onsets is generally larger than 0.1 eV (10 kJ/mol). The resulting anion heats of formation include that uncertainty plus the uncertainties in the heats of formation of the associated reactant and neutral species. A few retarding potential difference measurements have been carried out, to improve the accuracy of such results.

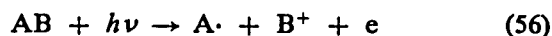
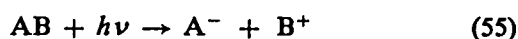
3.2.8. Neutral Beam Ionization/Appearance Potentials [NBIP/NBAP]

Collision of a neutral species with an energetic particle of low ionization potential, such as an alkali atom, can result in electron transfer, giving an alkali cation and an anion⁸³. The electron affinity of the neutral species is equal to the translational energy of the alkali atom less its ionization potential. Determinations of electron affinities by this method have the advantage that one obtains values for the true electron affinity: electron attachment to a neutral species, rather than detachment from an anion. Certain anions can be produced by this technique which are not accessible via electron impact due to low energy exit channels, e.g. CCl_4^- . Due to the limited energy resolution of the neutral alkali beam, the precision of this technique is not high, typically 20 kJ/mol. The onset energies of fragment ions can also provide useful thermochemical information, if the thermochemistry of the co-produced neutral species is known.

Normally this technique results in a determination of the adiabatic electron affinity, but for a sufficiently fast beam of neutral species, the onset corresponds to the vertical attachment energy of the electron, which, in contrast to detachment methods, is smaller than the adiabatic value.

3.2.9. Photolionization [PI]

This technique involves production of cation-anion pairs by vacuum ultraviolet photons. It has been used primarily for small molecules (O_2 , F_2 , etc.). The difference in onset for dissociative ion pair production and dissociative ionization



corresponds to the electron affinity of A.

3.2.10. Endothermic Reaction Energy, Including Charge Transfer [Endo,EnCT,CIDT]

If an ion/molecule reaction is appreciably endothermic at thermal (room temperature) energies, it is not observable by present techniques. For some processes it is possible to increase the rate by increasing the translational energy of the reactants so that products can be observed. Assuming that all the translational energy is available to bring about the reaction through the intermediacy of a long-lived complex in which energy is statistically distributed, the onset energy for observation of a given reaction can be taken as the threshold for the process, and thermochemistry assigned accordingly. Here the acronym "Endo" describes the use of such onset energies for deriving thermochemical data, "EnCT" the use of such onsets in charge transfer processes.

A variant is the case of collision of a *non-reactive* species, which serves only to provide the energy necessary for the negative ion to fragment or detach the electron. This is termed the "Collision Induced Dissociation Threshold" method [CIDT].

3.2.11. Surface Ionization (Magnetron) [SI]

The production of ions on a surface can yield thermochemical data if a number of parameters are known, including the work function of the surface. A common version of this experiment, the Magnetron technique⁸⁴ [Surface Ionization, SI], lacks mass analysis, and therefore many of the values for thermochemical parameters resulting from this method correspond to anions of uncertain identity. Precision is thought to be several tenths of a volt (>20 kJ/mol).

3.2.12. Electron Swarm [ES]

In this technique⁸⁵, the electron affinity of a neutral species (usually a closed shell molecule) is calculated by a statistical method, using the rate of electron attachment, the autodetachment lifetime, and the vibrational frequencies of the species. The attachment rate is measured in a drift tube — electron swarm experiment, and extrapolated to thermal energy. The autodetachment lifetime is taken from results of beam experiments. The precision is probably a few tenths of an eV (30-40 kJ/mol) at best.

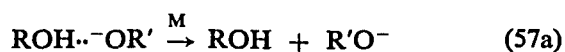
3.2.13. Lattice energy [Latt]

The heat of formation of an anion can be derived from a Born-Haber cycle using the lattice energy and heat of formation of a crystal and the thermochemistry of the appropriate gas phase cation. This method is not espe-

cially accurate relative to more recent techniques, but for some singly charged inorganic anions it provides the only data available.

3.2.14. Kinetic Branching Methods [Bran, CIDC]

If certain ion/molecule complexes are subjected to collision induced dissociation (CID), the weakest bond between the two species in the complex is the most likely one to break. If the functional groups forming the bond are identical, with the acids differing only in distant substitution, then either species has a chance to acquire the proton on breakup of the complex. The branching ratio in the reaction:



has been shown to reflect the relative acidities of the two species⁸⁷. Once the sensitivity of this branching ratio for compounds of known acidity has been established, then CID of clusters with one compound of known acidity and one unknown can lead to an estimate for the acidity of the unknown species. This appears to be reliable to 1–2 kJ/mol in determining relative acidities. This approach has some limitations. First, values for the gas phase acidities for several members of the series must first be known from other sources for proper calibration. Further, the temperature of the reacting system is not defined, and so problems may arise in interpretation for systems with significant entropy changes. The general method has also been applied to the estimation of relative electron affinities⁸⁷ for complexes of aromatic radical anions with aromatic molecules.

The excited intermediate complex can be prepared other ways than by collision. If an ion/molecule reaction is sufficiently exothermic, and has more than one available reaction channel, then the branching ratio of products formed on breakup of the complex can reflect product stabilities. This assumption has been used to estimate the acidities of the simple alkanes⁸⁸ since many of the localized carbanions from those compounds do not appear to be bound with respect to electron loss. Such ions exist only in ion/molecule complexes, where the cluster energy may serve to prevent electron detachment before reaction.

3.2.15. Electron Transmission Spectroscopy [ETS]

In this technique, the scattering angles of a monoenergetic electron beam impacting on a gas at less than the ionization threshold are determined. The presence of resonances in the spectrum implies electron capture to produce a temporary state, followed by autodetachment. This is the principal technique for measurement of negative electron affinities. Occasionally, a series of resonances can be extrapolated to below zero electron

energy to give an estimate of a positive electron affinity¹¹.

3.2.16. Electron Capture Detector [ECD]

An electron capture detector for a gas chromatograph, when operated in a variable temperature pulse sampling mode, can provide data on electron capture/detachment ratios. These can be converted into electron affinities. Use of the method is limited to the determination of electron affinities in the 0.2–0.8 eV (20–80 kJ/mol) range. The precision of such measurements is commonly quoted as less than 1 kJ/mol⁸⁹.

3.2.17. Mobility of Ions in a Gas [Mobl]

If the mobility of an ion in a gas can be measured in response to a weak electric field, the potential well depth, corresponding to $\Delta_{\text{aff}}H$, for the ion associating with the neutral gas can be determined.

3.2.18. Laser Optogalvanic Photodetachment Spectroscopy [LOG]

The gas of interest is subjected to an electrical discharge, and the discharge region is probed by a laser. The LOG⁹⁰ spectrum is recorded by scanning the wavelength of the laser, and monitoring laser-induced changes in the discharge impedance. The spectrum produced will be similar to the laser absorption spectrum but relative intensities of spectral features may be very different. The method is particularly suitable for detecting unstable (radical) species.

3.3. Thermochemical Cycles

The relationships between the different quantities measured in the above experimental techniques can be exploited to derive additional thermochemical information. In Table 2, such derivations have been made wherever possible. In the table, the quantities which have been *derived* from the experimentally-determined value are indicated by superscripted letters, which correspond to the various types of derivation described here, while the quantity actually determined in the reported experiment is given without any superscripted letter. A list of the various approaches to derivation and their corresponding superscript letters is given in the Table in Sec. 5.2.

The *heat of formation of an anion* can be derived from the heat of formation of the acid, its gas phase acidity, the heat of formation of the proton:

$$\Delta_f H(\text{A}^-) = \Delta_{\text{acid}} H(\text{AH}) - \Delta_f H(\text{H}^+) + \Delta_f H^\circ(\text{AH}) \quad (58)$$

The quantity $\Delta_f H^\circ(\text{AH})$ is lacking in many cases where acidities are now available; various group additivity estimation schemes (see below, and Sec. 4.) have been employed to fill in this information.

As discussed in Sec. 1.6.3., the calculation of the anion heat of formation as the heat of formation of the neutral species less the electron affinity:

$$\Delta_f H(A^-) = \Delta_f H^\circ(A) - EA(A) \quad (59)$$

is not, strictly speaking, correct, since for most of the species given here the heat of formation of the neutral species is a 298 K value, while the electron affinity is a threshold 0 K value. The preference is for anion heats of formation calculated by Eq. 58.

In an inversion of the bond strength/electron affinity Eq. 32 for calculating acidities, a known acidity and bond strength can yield an *electron affinity*.

$$EA(A) = BDE(A-H) + IP(H\cdot) - \Delta_{\text{acid}}H(AH) \quad (60)$$

Based on the temperature cancellation effect, this should correspond to the 0 K value. This is also an adiabatic value, which can be less than the vertical electron affinity obtained from the optical techniques if the geometries of the neutral and anion differ appreciably.

The difference between the anion and neutral heats of formation (at 298 K) give a 298 K electron affinity:

$$EA(A) = \Delta_f H^\circ(A) - \Delta_f H(A^-) \quad (61)$$

If the geometry change is small, this should be a reasonable approximation to the 0 K value.

Gas phase acidities, taken as the enthalpy of acidity, can be calculated from the homolytic bond strength of the acidity site, the electron affinity of the resulting radical, and the ionization energy of the hydrogen atom:

$$\Delta_{\text{acid}}H(AH) = BDE(A-H) - EA(A) + IP(H) \quad (62)$$

The last is common to all acids, and is very accurately known (1311.98 kJ/mol), and does not present a limitation in determining the values. A more valid concern is the temperature of definition for these terms. The acidity and bond strength are commonly taken as 298 K values, while the electron affinity and ionization potential are threshold values defined at 0 K. The cancellation necessary for this equation to be considered valid is discussed in Sec. 1.6.3.

Sometimes a heat of formation of an anion or an electron affinity value may be known without a value for the bond strength being available. The *acidity of the conjugate acid* can be derived in those cases from the acid heat of formation:

$$\Delta_{\text{acid}}H(AH) = \Delta_f H(A^-) + \Delta_f H(H^+) - \Delta_f H^\circ(AH) \quad (63)$$

While the primary goal of this work is not to obtain values for *homolytic bond strengths*, such values can be derived from gas phase acidities and electron affinities in cases where they are not known from more conventional sources.

$$BDE(A-H) = \Delta_{\text{acid}}H(AH) + EA(A) - IP(H\cdot) \quad (64)$$

$$BDE(A-H) = \Delta_f H^\circ(AH) - \Delta_f H^\circ(A) - \Delta_f H^\circ(H) \quad (65)$$

3.4. Priority of Data

At the present time, the heat of formation of an anion in the gas phase is not directly measurable, since gas phase plasma calorimetry is not a known technique. Likewise, direct measurement of an electron affinity, in the sense of exothermic electron attachment to a neutral, is not feasible in a calorimetric sense, although the combination of attachment and detachment rate constants can be used. The electron affinity and anionic heat of formation are available from either thermochemical cycles, based on other known and measurable quantities, or by reasonable assumptions about the reversibility of processes such as electron detachment from anions.

For electron affinities, we adopt the following order of priority for the evaluation of "best" values. There are exceptions in many cases to this order, where a given method is known not to be suitable. The user should be aware of the difference between adiabatic and vertical values that these techniques yield.

Laser photoelectron spectroscopy

Laser photodetachment

Photodetachment

From bond strengths and gas phase acidities

Neutral beam ionization/appearance potentials

Electron impact appearance potentials

Ion/molecule bracketing reactions

Electron swarm

For gas phase acidities, the following priorities are assigned to data sources:

Direct gas phase equilibrium constant determinations

Kinetic methods for gas phase acidities: branching ratios in collisional dissociation and ion/molecule complex breakup.

From bond strengths and electron affinities

Ion/molecule bracketing reactions (using either the heat of formation of the anion or of the acid as the unknown quantity).

4. Thermochemistry of Neutral Species

Tables 1 and 2 display values for heats of formation of the neutral gas phase molecules which are "related" to the archived ions. In Table 1, which is concerned with cation thermochemistry, the "related" neutral species is either (a) the neutral molecule which corresponds to the ion plus an electron (for ionization potential data) or (b) the molecule which has one less proton than the ion of interest (for proton affinity data). In Table 2, concerned with anion thermochemistry, the term "related" means that the neutral is formed from the ion either by loss of an electron (electron affinity) or gaining of a proton (gas phase acidity). For every case, the identity of the neutral

molecule corresponding to the displayed heat of formation is made unambiguous.

4.1. Literature Sources

Values for the heats of formation of neutral molecules were taken from the experimental literature whenever possible. If a value for a particular compound was available from an evaluated data compilation, this value was generally selected for inclusion here. The primary compilations which were used were as follows.

4.1.1. Organic Compounds

J. B. Pedley and J. Rylance, "Sussex-N. P. L. Computer Analysed Thermochemical Data: Organic and Organometallic Compounds," University of Sussex (1977). The numerous data from this evaluated compilation⁹¹ of 298 K heats of formation of gas phase organic compounds are identified by the squib 77PED/RYL. A second edition of this work (86PED/NAY) has appeared⁹², but regrettably, was available to the authors of the current compilation too late to obviate an extensive literature search for heats of formation from the primary literature to cover the period 1976-mid-1986. Since the updated compilation of Pedley, Naylor and Kirby⁹² (which is complete only through 1982) became available only as this work was nearing completion, references to 77PED/RYL or to recent primary literature have been retained even in cases where the data are given in 86PED/NAY.

4.1.2. Inorganic Compounds

(1) D. D. Wagman, W. H. Evans, V. B. Parker, R. H. Schumm, I. Halow, S. M. Bailey, K. L. Churney, and R. L. Nuttall, "The NBS Tables of Chemical Thermodynamic Properties: Selected Values for Inorganic and C₁ and C₂ Organic Substances in SI Units," *J. Phys. Chem. Ref. Data*, Vol. 11, Suppl. 2 (1982), hereafter referred to as 82TN270, from the original publication of this compilation¹⁴ as a series of NBS Technical Notes called the 270-series. (It should be noted that when this source is used in Table 2, both the value and an associated error limit are given, while only the former is given in Table 1.)

(2) (a) D. R. Stull and H. Prophet, "JANAF Thermochemical Tables," NSRDS-NBS 37 (1971); (b) M. W. Chase, J. C. Curnutt, H. Prophet, R. A. McDonald, and A. N. Syverud, "JANAF Thermochemical Tables," 1975 Supplement, *J. Phys. Chem. Ref. Data* 2, 1 (1975); (c) M. W. Chase, Jr., J. L. Curnutt, J. R. Downey, Jr., R. A. McDonald, A. N. Syverud, and E. A. Valenzuela, *J. Phys. Chem. Ref. Data*, 11, 695 (1982). Information from these sources¹⁵ is referenced as 71JANAF, 75JANAF, or 82JANAF. An updated composite edition^{13c} of this com-

pilation was in press at the time this work was being prepared, but was not actually available until these tables were near completion. A few values for heats of formation from the updated version have been inserted here where warranted by changes in recommended values, but an exhaustive check of the new publication was not made.

(3) L. V. Gurvich, I. V. Veits, V. A. Medvedev, G. A. Khachkuruzov, V. S. Yungman, G. A. Bergman, et al, "Termodinamicheskie Svoistva Individual'nykh Veshchestv" (Thermodynamic Properties of Individual Substances); V. P. Glushko, Gen. Ed., Vols. 1 through 4 (in 8 parts), (1978-1982), Izdatel'stvo "Nauka" Moscow. These volumes¹⁵ are collectively cited as 82TPIS.

4.1.3. Specialized Classes of Compounds and Radicals

In addition, various literature compilations which cover well-defined, but restricted, classes of compounds such as nitriles, organometallic compounds, free radicals, or strained hydrocarbons were utilized. The preferential use of data compilations as sources of experimental data recognizes that these data have been evaluated for internal consistency insofar as possible.

Many values for heats of formation of free radicals were taken from the review of McMillen and Golden⁹³. With respect to the alkyl radicals, however, a recent re-evaluation of data from the literature has led to the suggestion that C-H bond energies in alkanes should be revised upwards⁹⁴. The heats of formation of these radicals are still a matter of controversy at this writing. Values cited in the tables are based on the following revised⁹⁴ C-H bond strengths: Primary C-H bond (101 kcal/mol, 422 kJ/mol); Secondary C-H bond (99 kcal/mol, 415 kJ/mol); Tertiary C-H bond (95 kcal/mol, 398 kJ/mol).

The corresponding values for the heats of formation of the alkyl radicals are in some cases (e.g. t-butyl radical) entirely consistent with the differences between well-established heats of formation of the corresponding alkyl cations and experimental ionization energy values, but there are also cases for which the relevant values show inconsistencies. These are pointed out in Table 1 by a specific comment.

4.1.4. Other Literature Sources

Experimental information about heats of formation of species not included in evaluated compilations was generally obtained from primary literature sources. When more than one value for a heat of formation was available from such unevaluated primary sources, and no supplementary information was available which would allow an educated choice, the most recent value was usually arbitrarily chosen in the possibly naive expectation that "improved instrumentation" as well as a greater (historical) awareness of the problems inherent in an analysis of the thermochemistry of the particular species would lead to a more reliable value.

Care was taken in utilizing these data from unrelated literature sources to be aware of ambiguities in thermochemical reference states. Values for heats of formation derived from heats of reaction (e.g. hydrolysis, bromination) were used in preference to directly-determined heats of combustion because of the inherent problems associated with numbers derived from relatively small differences between two large numbers. In many cases, heats of formation of neutral species were derived using well-established ionization energies or proton affinity values in combination with well-established heats of formation of relevant ions; these values are identified by an explanatory sentence in the comment field.

4.2. Conventions Used and Assumptions Made

Implicitly, in assigning gas phase heats of formation to the neutral species, the compounds are assumed to be ideal gases at S. T. P. Whenever sufficient information is readily available, values for heats of formation at both 0 and 298 K are given. In some cases, the 0 K value has been obtained by combining the 298 K values from a preferred literature source (i.e., an evaluated compilation) with the 298—0 K difference taken from another reference. In other cases, this difference was calculated from experimental or estimated extrathermodynamic quantities such as vibrational frequencies and the appropriate elemental thermochemical functions. Another approach was to use isoelectronic or isostructural analogies (e.g., data for O_3 may be compared with that on NO_2^- and neutral NO_2).

4.3. Use of Condensed Phase Heats of Formation

Numerous methods exist for measuring and interpreting experimental heats of sublimation and heats of vaporization. Where available, such measurements were used to translate condensed phase data into gas phase values for heats of formation. In such cases, the cited reference is the source of the condensed phase heat of formation data, although the bibliography includes the references from which the information about heats of sublimation or vaporization were obtained.

In most cases, data on heats of sublimation (and the associated methodologies for translating condensed phase heats of formation to gas phase values at 298 K) are from the recent publication⁹⁵, 87CHI, for organic compounds.

Regrettably, while work on this publication was in progress no such single literature source for heats of vaporization was available, although such a compilation has since appeared⁹⁶. In fact, however, experimental data on heats of vaporization do not exist for numerous species of interest here, either because of experimental difficulties associated with such determinations (i.e. lack of adequate volatility, purity, or thermal stability) or per-

haps because of a lack interest in, or availability of, the compound.

Estimation methods for heats of vaporization and sublimation have been described in the literature^{95,97}. Some require auxiliary experimental data (e.g. critical constants). Other such estimation methods can be applied only to well-defined classes of compounds. For heats of vaporization, these estimates are usually reliable to approximately 4 kJ/mol (1 kcal/mol). In presenting data on heats of formation incorporating the use of such estimated heats of vaporization, a choice had to be made of whether to cite the source of the experimental heat of formation of the liquid or the paper from which the method for estimating the heat of vaporization was obtained; the experimental work is given as the primary citation. For heats of sublimation, no generally accurate estimation approaches exist. Thermochemical values obtained using estimated heats of sublimation are clearly labelled as estimates in the tables.

4.4. Estimated Heats of Formation

Estimates were also made for heats of formation of neutral molecules and radicals for which no experimental data were available. Several estimation approaches were utilized and are now briefly described. The relationships between the various estimation approaches have been described in a recent review⁹⁸.

4.4.1. Estimates from Data on Isomeric Species

One approach utilizes experimental information about isoenergetic processes for the formation of two or more isomeric species in a particular reaction. That is, given a pair of isomers for which information about the heat of formation is available for only one of the pair, an estimate of the thermochemistry of the second compound can be based on the casual and generally rather reliable assumption that if two isomers are formed in comparable yield in a particular process then their Gibbs energies and enthalpies of formation are generally comparable. Likewise, though less reliable, one may assert if one isomer is formed in higher yield, then this is the more stable product. Estimates made in this way often include the assumption that heats of vaporization and of solution are also comparable for the relevant pair of isomers; this will be a valid approximation except when there are great differences in the extent of hydrogen bonding (such as might exist for isomeric alcohols and ethers), and even in these cases, approximate corrections (such as assuming constant H-bond strengths) can be made.

In estimating thermochemical data from known information about an isomeric species, a common assumption made is that $\Delta(\Delta_r H) = \Delta(\Delta_r G)$ for the pair of isomers (i.e. $\Delta(\Delta_r S)$ is negligible). A related approach examines experimentally-determined reaction rates and/or kinetic activation energies. The thermochemical estimate is

based on the assumption that the structural effects on rates and equilibria will vary in a parallel manner, and that thermodynamic and kinetic control of arbitrary reactions result in the same products. While not in fact absolutely true, experience has shown this to be a useful assumption for predicting substituent effects for numerous homologous series.

Another approach to estimating heats of formation is based on the assumption that $\Delta(\Delta_f H)$ can be equated with ΔE_{tot} for two isomers, where ΔE_{tot} is the difference in total energies of the two species calculated by quantum mechanics. For this assumption, as well as all other estimation approaches in this study employing results from quantum mechanics, *ab initio* calculations were given preference over results from any of the plethora of semi-empirical methods in the literature. (The reader should note that heats of formation from MNDO and from molecular mechanical calculations were occasionally used, however). Care was explicitly taken to contrast only species studied with the same basis set and degree of geometry optimization. Implicit, however, are the requirements that both the zero-point energy and 0 K - 298 K corrections are essentially identical for a pair of isomers. These last assumptions are surprisingly valid where sufficient experimental data are available to test them.

All values for heats of formation based on these approaches are labelled as estimates (EST) in the Tables.

4.4.2. Summing of Increments

There remain three related approaches which were employed to estimate heats of formation of molecules here. The best characterized is Benson's "group increment" approach⁹ in which the molecule of interest is defined as a collection of groups, and a "group" is then defined as a polyvalent atom (ligancy > 2) with all of its associated ligands in the molecule. The heat of formation of the molecule is obtained by summing the contributions of the heats of formation of the various groups, correcting for various higher order interactions and "correction" terms. These corrections include the presence of gauche configurations in substituted alkanes, gem-substitution of large and/or polar groups, and the presence of rings that are strained because of heteroatoms and/or are not six-membered. These group energies and the various corrections have been obtained using both statistical analysis and by chemical intuition, and for "reasonable" molecules generally give reasonable results. Estimates using this approach are better defined, though not necessarily of better quality, than the others.

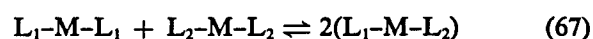
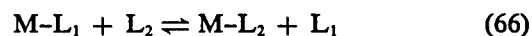
Because this approach is now very common in the chemical literature, many of the papers included here which are primarily concerned with aspects of ion chemistry (e.g. ionization energy, gas phase basicity or acidity determinations) include estimates of heats of formation of relevant neutral species based on this approach; rather

than cite the work of the authors who made the (rather standard) estimate, these values, when they were used, have been labelled as estimates. In Table 2 estimates utilizing this approach are labelled "Est", but are accompanied by error bars.

A related approach takes advantage of regularities in trends in heats of formation of different homologous series of compounds. As an example of how this approach works, if it is observed that the heats of formation of several RXR compounds differ from those of RYR compounds by some approximately constant increment, then this difference is defined as a "correction term" for deriving heats of formation of any -Y- compound from the heat of formation of the corresponding -X- compound (or vice versa)¹⁰⁰. The unknown heat of formation is taken to be the sum of the known heat of formation and the suitable correction term associated with the exchange of the substituent and parent components. One may also derive correction terms from suitable bond energies, e.g., assume that the O-H bond energy in all carboxylic acids is the same. This approach is more commonly used in anion chemistry and is designated as Est2 in Table 2 (but as EST without a special designator in Table 1).

The final estimation approach used here is commonly called "macroincrementation"^{101,102}, and, as the name implies, involves building up the molecule of interest by adding increments (as in the Benson approach), but with the difference that the incremental heats of formation are specifically derived from thermochemical data for molecules or large ("macro") molecular fragments which incorporate factors which need to be considered, such as resonance, strain energy, steric effects, etc. This approach assumes that "if for each of two sets of molecules the total number of bonds, atoms and structural types is the same, then the total heat of formation of each set of molecules is the same. Then, if all but one of the heats of formation are available, the remaining one can be estimated by simple arithmetic." Macroincrementation maximizes the direct use of chemical intuition with regard to electronic and/or steric effects, as well as the direct use of available experimental data. The majority of estimates for organic compounds in Table 1 were made using this approach.

The heats of formation of only a few inorganic or organometallic compounds were estimated. Where estimates were made, it generally was assumed that the heat of ligand exchange was negligible, i.e. the heats of reaction of the following generic reactions for suitably similar ligands (L_1 and L_2) could be taken to be zero:



Likewise, simple additive assumptions were made as to heats of vaporization and sublimation and the 0 K - 298 K energy differences.

5. Summary of Conventions Used in Tables

In an attempt to present as much information as possible in Tables 1 and 2, while keeping the pages uncluttered, it was sometimes necessary to resort to the use of bold face, italic typefaces, asterisks, etc. to convey additional information. The various conventions are summarized below. The user is particularly cautioned that these conventions are different for Table 1 and Table 2. In particular, italicized numbers have different meanings in Table 1 (zero Kelvin heats of formation) and Table 2 (a hydride or halide affinity, and information relating to thermochemistry of neutral species).

5.1. Positive Ion Table (Table 1)

Value underscored: A well-established value of an ionization energy or ion heat of formation.

Value enclosed in parentheses: A value of an ionization energy or heat of formation which is not well established, or not evaluated, for one of the following reasons: (1) Only one determination of the ionization energy has been reported, and there are no auxiliary data which would permit one to judge its accuracy; (2) The heat of formation of relevant neutral species is/are not well established; (3) Two or more contradictory values for the ionization energy or appearance energy have been reported, and while one value has been selected, there is sufficient doubt that one can not regard the selected value as well-established. (For data falling under category (3), an explanatory comment is always included.)

Value given in italics: Thermochemical data corresponding to a temperature of absolute zero.

Literature citations: In Table 1, there is no column giving references to the source of ionization energy/appearance energy data. Such data are always taken from the earlier compilations^{1,2,3,4} unless specifically noted in a comment. When data are from the 1981–1986 primary literature, the reference is always specifically mentioned in the comment, and is specifically given in the bibliography. Heats of formation derived from proton affinity data (and the proton affinity data themselves) are taken from the evaluated compilation⁴ or from more recent literature, which will always be specifically cited in the comment and listed in the bibliography.

Sort scheme: Data are sorted by empirical formula ordered according to the so-called Hill scheme, which is the same sort scheme used by Chemical Abstracts. Formulas are written as $C_nH_mX_xY_y\dots$, where the primary sort is ordered by n , the number of carbon atoms, and the first sub-sort is ordered according to m , the number of H atoms. All other atoms in the molecule (X, Y, etc) are ordered alphabetically, and the various sub-sorts follow accordingly. Any molecules which do not contain carbon appear according to a strictly alphabetical sort.

Proton affinity data: To locate the proton affinity of a molecule, look under the empirical formula of the protonated molecule, i.e. the proton affinity of CH_4 appears under CH_5 .

Estimated heats of formation of neutral molecules: The literature citation column contains the acronym EST for estimated values.

5.2 Negative Ion Table (Table 2)

Chemical species: Each entry is headed by an empirical formula of the relevant anion, with the atoms ordered according to the Hill formulation. Below this there appears a structural representation of the anion where this can be conveniently represented on one line; the last-listed atom is usually the atom judged to carry the negative charge (insofar as this can be ascertained). These formulas may contain simplifying abbreviations in common use by organic chemists, for example "Me" for CH_3 , "Et" for C_2H_5 , "Pr" for C_3H_7 , "COT" for cyclooctatetraene, or "Ph" for phenyl. For chemical species which have structures which are too complex to be represented by a semi-structural formulation, a name is given. The names chosen for inclusion are easily recognizable by most chemists, or at least can be readily located in standard texts.

Units: In Table 2, all data are presented in kJ/mol, except the values for electron affinities, which (as specifically indicated) are given in electron volts.

Presentation of Data: Each line presents data from a different reference, which is cited at the end of the line. The value (or values) which results (or result) from a primary experimental measurement will appear without an affiliated superscript alphabetic letter. These letters point out data which have been derived from the experimental result; the derivations are described in Sec. 3.3., and summarized (along with their alphabetic identifiers) in Table 5.2. The data in the Table are divided into columns as follows:

| | | | | | |
|-----|-----------------|-------------------------------|-------------------------------|-----------------------|----------------------------|
| Ion | $\Delta_f H(A)$ | EA(A) | $\Delta_{acid} H(AH)$ | $\Delta_{acid} G(AH)$ | Method, Comment; Reference |
| | or | eV | or | or | |
| | $[X \cdot Y^-]$ | $\Delta_{aff} H(X \cdot Y^-)$ | $\Delta_{aff} G(X \cdot Y^-)$ | | |

Ion: The chemical formula of the anion of interest.

$\Delta_f H(A^-)$ or $[X \cdot Y^-]$: The second column presents the heat of formation of the listed anion in kJ/mol. The column heading specifies that the data correspond to anion A^- which may also be represented as $[X \cdot Y^-]$. The second designation is included for the cases where the heat of formation of the anion has been derived from data on the clustering of anion Y^- to neutral molecule X (see reaction 52). For example, data on the heat of formation of AlF_4^- ($X \cdot Y^-$) is derived from information on the fluoride affinity of AlF_3 (that is, AlF_3 is X and F^- is Y^-).

EA(A): The electron affinity of neutral species A is listed in this column in electron volts.

$\Delta_{\text{acid}}H(AH)$ or $\Delta_{\text{aff}}H(X\cdots Y^-)$ and $\Delta_{\text{acid}}G(AH)$ or $\Delta_{\text{aff}}G(X\cdots Y^-)$: The fourth and fifth columns serve double purposes, with normal typefaced data representing the enthalpy change (fourth column) or Gibbs energy-change (fifth column), respectively, of reaction 10 for the species AH leading to a value for the heat of formation of anion A^- . Data given in italics represent enthalpy changes for reaction 52, that is the *affinity* of molecule X for anion Y^- . These have been derived either from direct determinations of equilibrium constants for reaction 52, or from equilibrium constants for Y^- transfer reaction 51 which yield scales of relative Y^- -affinities.

Method: This column gives an acronym to indicate the experimental technique used in determining a particular piece of data. These are discussed in detail in Sec. 3.2. For quick reference, an alphabetized summary of the acronyms with their definitions, and the locations of the relevant discussions, is given in Table 5.2. This table also includes other acronyms, abbreviations, and symbols used in Table 2 for ready reference.

Comment: Where necessary for clarity, details of a particular experiment are given as a comment. In this column, there also appears information about auxiliary thermochemistry concerning neutral species. All data pertaining to neutral species appear in a different italicized typeface.

Reference: The squib given in this column refers to the article in which the primary datum reported on a particular line was reported. The complete reference can be found in the bibliography for Tables 1 and 2.

Thermochemistry of neutral species: The relevant heats of formation of neutral species and accompanying references are given in the top line of the "Comment" column. All data and the references pertaining to neutral molecules are presented in a different italicized typeface, so that they will not be mistaken for data concerning the anion.

Sort scheme: Data are sorted by empirical formula using the same sort scheme as that used for the Positive Ion Table. This is the Hill (or Chemical Abstracts) scheme.

Acidity data: Data on the acidity of a given neutral species is given under the empirical formula of the conjugate base, i.e. the acidity of CH_3OH is found under CH_3O^- .

Asterisk in left hand margin: Due to the comprehensive nature of the Negative Ion compilation, there can be numerous entries in Table 2 for a given quantity associated with a particular negative ion, unlike the convention adopted for the cation table, where only one value of an ionization energy/heat of formation is given. A special indication must be given, therefore, to denote the preferred value. Any line with an asterisk in the left margin contains the selected "best" value for a given piece of data pertaining to that ion. There may be more than one line thus marked for a given anion, since the best values for an acidity value and an electron affinity value may be from different sources.

(The quantity without a superscripted letter is the primary piece of information for any given line.)

Absence of asterisk in margin: If no line is marked as preferred in the data collected for an ion, then no definitive evaluation could be made. Some preference should be given to the first reference cited in such cases, but this is a qualitative judgement on the part of the compiler, and should not be given undue weight.

Superscript "o" after method acronym [IMRE]: Original data which were re-evaluated to take into account new results which expanded a portion of the acidity^{81,82} scale; corrected values are shown above "original" data, with original reference cited for both values.

Primary data originating from cited experimental reference: In Table 2, on any given line (which presents information derived from a single paper) items which were *derived* from the primary experimental data using the relationships listed in Sec. 3.3. and summarized in Table 5.2., have a superscripted letter indicating the relationship used to derive the value (see Table 5.2.). The primary data do not display a superscript.

TABLE 5.2. Acronyms, abbreviations and symbols used in Negative Ion Table

| | |
|------------------|---|
| BDE(A-H): | Bond dissociation energy of A-H bond |
| Bran: | Branching ratio in an exothermic reaction (see Sec. 3.2.14.) |
| Calc: | Calculation |
| CIDC: | Collision-induced dissociation of cluster ion-branching ratio (see Sec. 3.2.7.) |
| CIDT: | Collision induced dissociation threshold (see Sec. 3.2.10.) |
| Def: | Defined |
| EA(A): | Electron affinity of A. |
| ECD: | Electron capture detector (see Sec. 3.2.16.) |
| EIAP: | Electron impact appearance potentials (see Sec. 3.2.7.) |
| EnCT: | Endothermic charge transfer threshold (see Sec. 3.2.10.) |
| Endo: | Endothermic reaction threshold energy (see Sec. 3.2.10.) |
| ES: | Electron swarm (see Sec. 3.2.12.) |
| Est: | Estimate, based on addition of increments |
| Est2: | Estimate, based on thermochemistry of analogous compounds |
| ETS: | Electron transmission spectroscopy (see Sec. 3.2.15.) |
| IMRB: | Ion/molecule reaction—bracketing (see Sec. 3.2.6.) |
| IMRE: | Ion/molecule reaction equilibrium constant determination (see Sec. 3.2.5.) |
| Kine: | Attachment/detachment rate ratio (see Sec. 3.2.5.) |
| Latt: | Lattice energy calculation (see Sec. 3.2.13.) |
| LOG: | Laser optogalvanic spectroscopy |
| LPD: | Laser photodetachment (see Sec. 3.2.2.) |
| LPES: | Laser photoelectron spectroscopy (see Sec. 3.2.1.) |
| Mobl: | Mobility of ion in gas (see Sec. 3.2.17.) |
| NBAP: | Neutral beam appearance potential (see Sec. 3.2.8.) |
| NBIP: | Neutral beam ionization potential (see Sec. 3.2.8.) |
| PD: | Photodetachment (see Sec. 3.2.3.) |
| PDIs: | Photodissociation (see Sec. 3.2.4.) |

TABLE 5.2. Acronyms, abbreviations and symbols used in Negative Ion Table — Continued

| | |
|---------------------------------|--|
| PI: | Photoionization (see Sec. 3.2.9.) |
| PLA: | Plasma absorption |
| SI: | Surface ionization (Magnetron) (see Sec. 3.2.11.) |
| TDA: | Temperature dependent association equilibrium constant determination (see Sec. 3.2.5.) |
| TDEq: | Temperature dependent equilibrium constant determination (see Sec. 3.2.5.) |
| $\Delta_f H(A^-)$: | Heat of formation of A^- |
| $\Delta_{aff} H(X \cdot Y^-)$: | Enthalpy of association of neutral X to anion Y^- , the affinity of X for Y^- |
| $\Delta_{aff} G(X \cdot Y^-)$: | Gibbs energy of association of neutral X to anion Y^- |
| $\Delta_{acid} H(AH)$: | Acidity of molecule AH; see definition below under f |

Single letter codes which define chemical reaction types (superscripts)

- a: $\Delta_f H(A^-) = \Delta_{acid} H(AH) - \Delta_f H^*(AH) + \Delta_f H(H^+)$
 b: $\Delta_f H(A^-) = \Delta_f H^*(A) - EA(A)$
 c: $\Delta_f H(X \cdot Y^-) = -\Delta_{aff} H(X \cdot Y^-) + \Delta_f H^*(X) + \Delta_f H(Y^-)$
 d: $EA(A) = \Delta_{acid} H(AH) - IP(H^+) - BDE(A-H)$
 e: $BDE(A-H) = \Delta_{acid} H(AH) - IP(H^+) + EA(A)$
 f: $\Delta_{acid} H(AH) = \Delta_f H(A^-) + \Delta_f H(H^+) - \Delta_f H^*(AH)$
 g: $\Delta_{rxn} H = \Delta_{rxn} G + T \Delta_{rxn} S$
 h: $\Delta_{rxn} G = \Delta_{rxn} H - T \Delta_{rxn} S$
 i: $EA(A) = \Delta_f H^*(A) - \Delta_f H(A^-)$
 j: $\Delta_{aff} H(X \cdot Y^-) = \Delta_f H^*(X) + \Delta_f H(Y^-) - \Delta_f H(X \cdot Y^-)$

5.3. References to Tables 1 and 2

The bibliography given at the back of the volume includes (a) references to the sources data having a bearing on the thermochemistry of the positive ions given in Table 1 (including ionization potentials, appearance potentials, proton affinities, and other related information) *except when those references appeared in the bibliographies of references 1 through 4*; (b) references to the sources of all data on the thermochemistry of negative ions from Table 2; and (c) references to the sources of the data on the thermochemistry of neutral molecules.

The references are identified in the tables, and in the bibliography, by a squib, made up of the year of the publication, the first three letters of the surname of the first author, followed by a slash and the first three letters of the surname of the second author. Example: A publication by J. B. Pedley and J. Rylance which appeared in 1977 would be designated by 77PED/RYL.

The references given in the bibliography are sorted according to these squibs, that is, first according to year, and then alphabetically according to the first three letters of the names of the first two authors. Example: Within the papers which appeared during a given year, reference to a paper by "Beauchamp and Armentrout" (BEA/ARM) would precede a reference to a paper by "Beach and Jackson" (BEA/JAC), which in turn would appear above a reference to "Beauchamp and Schwarz" (BEA/SCH). Note that papers of a given first author do not necessarily follow one another in the listing.

6. Acknowledgements

This work was initiated by Henry M. Rosenstock in the late 1970's. Until his death in 1982, Dr. Rosenstock was the guru of the project, assembling the current list of collaborators, participating vigorously in planning sessions, and, on occasion, engaging in scientific discussions which sometimes had the distinct character of teaching seminars, with Henry Rosenstock making certain that the rest of us clearly understood the science associated with our data evaluation project. There is no question that the quality of the evaluations reported here has been substantially improved by the interactions of the authors with this distinguished scientist.

The authors also acknowledge the contributions of the original group, D. Sims, S. S. Shroyer, and W. J. Webb, who started work on this project under Dr. Rosenstock's direction in 1978. Much of the literature search for the years 1971-1977 was completed by this team, as was the abstracting of data for the years 1971-72.

John Holmes especially wishes to record the contributions of Fred Lossing to this formidable task; his unquenchable enthusiasm and countless stimulating discussions over the past 15 or more years are most gratefully acknowledged.

At the National Bureau of Standards, Kathy Maugh, Jose Portal, Carol Martin, Clairemarie Lanthier, Kathy Whalen, and Elizabeth Rogers have helped in abstracting data from the literature, retrieving Chemical Abstracts Registry Numbers, proofreading, and numerous other tasks necessary to the completion of the project. Kathy Whalen and Elizabeth Rogers were responsible for affixing the drawings of molecules to the camera-ready pages. The careful, detailed work of all of these collaborators is acknowledged with thanks. We should also mention the help received from Julian Ives and Connie Seymour on preparation of the camera-ready copy.

Dr. Stephen E. Stein has collaborated with one of the present authors (W. G. M.) to develop a searchable PC version of this work, which can be purchased through the Office of Standard Reference Data.

Mahnaz Motevalli-Aliabadi of the University of Maryland, Baltimore County Campus, participated in the task of estimating heats of formation of neutral molecules during the early part of the project. Clairemarie Lanthier also provided estimations.

The molecular representations for the camera-ready copy were drawn by Andrea Fladager Buckley and Frances Baldwin.

The list of professional colleagues who have contributed data in preprint form or who have critically examined these tables and contributed advice, or pointed out errors is so long that a complete accounting is impossible. However, special mention should go to Dr. Charles DePuy of the University of Colorado for consultations on the table of anion thermochemistry. Dr. Malcolm Chase (NBS), editor of the JANAF tables, consulted with us about the discussion given here on the thermochemical conventions for the electron. Drs.

Eugene Domalski, Arthur Greenberg, Wing Tsang, and Deborah Van Vechten were frequently consulted about all aspects of that part of the project involving compilation of thermochemistry of neutral species. Dr. Pierre Ausloos critically reviewed the work several times during the five year period, and also contributed expert evaluations of thermochemical data from time to time. Also, we thank the three reviewers who provided thoughtful, informative, helpful comments on the manuscript; one of these, Dr. David A. Dixon, took the trouble to discuss the publication with us in some depth.

The project was carried out under the joint sponsorship of the Office of Standard Reference Data and the Center for Chemical Physics, National Bureau of Standards. We would like to acknowledge the support of Dr. David R. Lide, Director of OSRD, and Dr. Pierre Ausloos, Director of CCP, as well as Dr. Howard White, program manager in the Office of Standard Reference Data.

6. References to Introduction

- ¹J. L. Franklin, J. G. Dillard, H. M. Rosenstock, J. T. Herron, K. Draxl, and F. H. Field, *Ionization Potentials, Appearance Potentials, and Heats of Formation of Gaseous Positive Ions*, Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U. S.) 26, (1969).
- ²H. M. Rosenstock, K. Draxl, B. W. Steiner, and J. T. Herron, "Energetics of Gaseous Ions," J. Phys. Chem. Ref. Data, Vol. 6, Suppl. 1 (1977).
- ³R. D. Levin and S. G. Lias, *Ionization Potential and Appearance Potential Measurements, 1971-1981*, Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U. S.) 71, (1982).
- ⁴S. G. Lias, J. F. Liebman, and R. D. Levin, "Evaluated Gas Phase Basicities and Proton Affinities of Molecules; Heats of Formation of Protonated Molecules," J. Phys. Chem. Ref. Data, Vol. 13, 695 (1984).
- ⁵J. E. Bartmess, to be published.
- ⁶S. G. Lias and R. D. Levin, to be published.
- ⁷S. G. Lias, J. F. Liebman, and R. D. Levin, to be published.
- ⁸R. G. Keese and A. W. Castleman, Jr., "Thermochemical Data on Gas-Phase Ion-Molecule Association and Clustering Reactions," J. Phys. Chem. Ref. Data 15, 1011 (1986).
- ⁹D. A. Dixon and S. G. Lias, "Absolute Values of Gas Phase Proton Affinities and Basicities of Molecules: A Comparison between Theory and Experiment," Chapter 7 in *Molecular Structure and Energetics, Vol. 2* (J. F. Liebman and A. Greenberg, Editors), VCH Publishers, Inc.: Deerfield Beach, Fla., 1987, p. 269.
- ¹⁰J. E. Del Bene, "Quantum Chemical Reaction Enthalpies," Chapter 9 in *Molecular Structure and Energetics, Vol. 1* (J. F. Liebman and A. Greenberg, editors), VCH Publishers, Inc.: Deerfield Beach, Fla., 1986, p. 319.
- ¹¹A.A. Christodoulides; D.L. McCorkle, L.G. Christophorou, "Electron Affinities of Atoms, Molecules, and Radicals", Ch. 13 in *Electron-Molecule Interactions and Their Applications, Vol. 2*, Academic Press: New York, 1984, pp. 423-641.
- ¹²K. D. Jordan and P. D. Burrow, "Studies of the Temporary Anion States of Unsaturated Hydrocarbons by Electron Transmission Spectroscopy," *Acc. Chem. Res.* 11, 341 (1978).
- ¹³(a) D. R. Stull and H. Prophet, "JANAF Thermochemical Tables," Nat. Stand. Ref. Data Ser. (U. S.) 37, (1971); (b) M. W. Chase, J. L. Curnutt, A. T. Hu, H. Prophet, A. N. Syverud, and L. C. Walker, "JANAF Thermochemical Tables, 1974 Supplement," J. Phys. Chem. Ref. Data 3, 311 (1974); (c) M. W. Chase, Jr., C. A. Davies, J. R. Downey, Jr., D. J. Frurip, R. A. McDonald, and A. N. Syverud, "JANAF Thermochemical Tables, Third Edition, Parts I and II," J. Phys. Chem. Ref. Data 14, Suppl. 1 (1985).
- ¹⁴D. D. Wagman, W. H. Evans, V. B. Parker, R. H. Schumm, I. Halow, S. M. Bailey, K. L. Churney, and R. L. Nuttall, "The NBS Tables of Chemical Thermodynamic Properties: Selected Values for Inorganic and C₁ and C₂ Organic Substances in SI Units," J. Phys. Chem. Ref. Data 11, Suppl. 2 (1982).
- ¹⁵L. V. Gurvich, I. V. Veits, V. A. Medvedev, G. A. Khachkuruzov, V. S. Yungman, G. A. Bergman, et al., *Termodinamicheskie Svoistva Individual'nykh Veshchestv* (Thermodynamic Properties of Individual Substances); V. P. Glushko, Gen. Ed., Vols. 1 through 4 (in 8 parts), (1978-1982), Izdatel'stvo "Nauka" Moscow.
- ¹⁶S. G. Lias, "Thermochemistry of Polyatomic Cations," in *Kinetics of Ion-Molecule Reactions*, (P. Ausloos, Editor), New York: Plenum, 1979, p. 223.
- ¹⁷H. M. Rosenstock, "Standard States in Gas Phase Ion Thermochemistry," in *Kinetics of Ion-Molecule Reactions*, (P. Ausloos, Editor), Plenum: New York, 1979, p. 246.
- ¹⁸A. N. Syverud, personal communication (1978).
- ¹⁹A. Sommerfeld, *Z. Physik* 47, 1 (1928).
- ²⁰A. R. Gordon, *J. Chem. Phys.* 4, 678 (1936).
- ²¹W. H. Evans, personal communication to H. M. Rosenstock (1976).
- ²²S. G. Lias and P. Ausloos, "Ionization Energies of Organic Compounds by Equilibrium Measurements," *J. Am. Chem. Soc.* 100, 6027 (1978).
- ²³(a) T. Shimanouchi, "Tables of Molecular Vibrational Frequencies. Consolidated Volume I," Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U. S.) 39, (1972); (b) T. Shimanouchi, "Tables of Molecular Vibrational Frequencies. Consolidated Volume II," J. Phys. Chem. Ref. Data 6, 993 (1977); (c) T. Shimanouchi, "Tables of Molecular Vibrational Frequencies, Part 9," J. Phys. Chem. Ref. Data 7, 1323 (1978); (d) T. Shimanouchi, "Tables of Molecular Vibrational Frequencies, Part 10," J. Phys. Chem. Ref. Data 9, 1149 (1980).
- ²⁴M. E. Jacox, "Ground State Vibrational Energy Levels of Polyatomic Transient Molecules," J. Phys. Chem. Ref. Data 13, 945 (1984).
- ²⁵J. C. Traeger and R. G. McLaughlin, "Absolute Heats of Formation for Gas Phase Cations," *J. Am. Chem. Soc.* 103, 3647 (1981).
- ²⁶W. A. Brand and T. Baer, *J. Am. Chem. Soc.* 106, 3154 (1984).
- ²⁷F. P. Lossing, *Can. J. Chem.* 50, 3973 (1972).
- ²⁸J. Franck, "Elementary Processes of Photochemical Reactions," *Trans. Faraday Soc.* 21, 536 (1926).
- ²⁹(a) E. U. Condon, "Nuclear Motions Associated with Electron Transitions in Diatomic Molecules," *Phys. Rev.* 32, 858 (1928); (b) E. U. Condon, "The Franck-Condon Principle and Related Topics," *Am. J. Phys.* 15, 365 (1947).
- ³⁰W. A. Chupka, "Effect of Unimolecular Decay Kinetics on the Interpretation of Appearance Potentials," *J. Chem. Phys.* 30, 191 (1959).
- ³¹H. M. Rosenstock, "The Measurement of Ionization and Appearance Potentials," *Int. J. Mass Spectrom. Ion Phys.* 20, 139 (1976).
- ³²J. Dannacher, "The Study of Ionic Fragmentation by Photoelectron-Photoion Coincidence Spectroscopy," *Org. Mass Spectrom.* 19, 253 (1984).
- ³³J. L. Holmes, "Assigning Structures to Ions in the Gas Phase," *Org. Mass Spectrom.* 20, 169 (1985).
- ³⁴C. E. Moore, "Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra," Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U. S.) 34, (1970).
- ³⁵S. Leutwyler, M. Hofmann, H.-P. Harri, and E. Schumacher, "The Ionization Potentials of the Alkali Dimers Na₂, NaK, and K₂," *Chem. Phys. Lett.* 77, 257 (1981) and references cited therein.
- ³⁶K. Maeda, G. P. Semeluk, and F. P. Lossing, "A Two-Stage Double Hemispherical Electron Energy Selector," *Int. J. Mass Spectrom. Ion Phys.* 1, 395 (1968).
- ³⁷E. M. Clarke, "Ionization Probability Curves Using an Electron Selector," *Can. J. Phys.* 32, 764 (1954).
- ³⁸P. Marmet, and L. Kerwin, "An Improved Electrostatic Electron Selector," *Can. J. Phys.* 38, 787 (1960).
- ³⁹P. Marmet and J. D. Morrison, "Secondary Reactions in the Ion Chamber of a Mass Spectrometer," *J. Chem. Phys.* 36, 1238 (1962).
- ⁴⁰H. M. Rosenstock and M. Krauss, *Mass Spectrometry of Organic Ions*, Chapter 1, New York: Academic Press (1963).

- ⁴¹H. M. Rosenstock, M. B. Wallenstein, A. L. Wahrhaftig and H. Eyring, *Proc. Natl. Acad. Sci. U. S.*, **38**, 667 (1952).
- ⁴²(a) C. R. Brundle and A. D. Baker, *Electron Spectroscopy: Theory, Techniques and Applications*, New York: Academic Press, 1977; (b) J. W. Rabalais, *Principles of Ultraviolet Photoelectron Spectroscopy*, New York: Wiley-Interscience, 1977; (c) J. H. D. Eland, *Photoelectron Spectroscopy: An Introduction to Ultraviolet Photoelectron Spectroscopy in the Gas Phase*, New York: John Wiley & Sons, 1974; (d) D. W. Turner, C. Baker, A. D. Baker, and C. R. Brundle, *Molecular Photoelectron Spectroscopy*, New York: Wiley-Interscience, 1970.
- ⁴³(a) P. Kebarle, "Higher-Order Reaction-Ion Clusters and Ion Solvation," Chapter 7 in *Ion-Molecule Reactions* (J. L. Franklin, Editor), New York: Plenum Press, 1972; (b) P. Kebarle, "Ion Thermochemistry and Solvation from Gas Phase Ion Equilibria," *Ann. Rev. Phys. Chem.* **28**, 445 (1977).
- ⁴⁴(a) E. E. Ferguson, "Flowing Afterglow Studies," Chapter 8 in *Ion-Molecule Reactions* (J. L. Franklin, Editor), New York: Plenum Press, 1972; (b) D. K. Bohme, "The Kinetics and Energetics of Proton Transfer," in *Interactions between Ions and Molecules*, (P. Ausloos, Editor), New York: Plenum Press, 1975, p. 504.
- ⁴⁵(a) J. L. Beauchamp, "Ion Cyclotron Resonance Spectroscopy," *Ann. Rev. Phys. Chem.* **22**, 527 (1971); (b) R. W. Taft, "Gas Phase Proton Transfer Equilibria," in *Proton Transfer Reactions*, (E. F. Caldin and V. Gold, Editors), New York: John Wiley & Sons, 1975, p. 31; (c) D. H. Aue and M. T. Bowers, "Stabilities of Positive Ions from Equilibrium Gas Phase Basicity Measurements," in *Gas Phase Ion Chemistry*, (M. T. Bowers, Editor), New York: Academic Press, 1979, p. 1; (d) C. R. Moylan and J. I. Brauman, "Gas Phase Acid-Base Chemistry," *Ann. Rev. Phys. Chem.* **34**, 187 (1983).
- ⁴⁶(a) S. G. Lias, P. Ausloos, and Z. Horvath, "Charge Transfer Reactions in Alkane and Cycloalkane Systems. Estimated Ionization Potentials," *Int. J. Chem. Kinetics* **8**, 725 (1976); (b) M. Meot-Ner (Mautner), L. W. Sieck, and P. Ausloos, "Ionization of Normal Alkanes: Enthalpy, Entropy, Structural, and Isotope Effects," *J. Am. Chem. Soc.* **103**, 5342 (1981).
- ⁴⁷M. Meot-Ner (Mautner), S. F. Nelsen, M. R. Willi, and T. B. Frigo, "Special Effects of an Unusually Large Neutral to Radical Cation Geometry Change. Adiabatic Ionization Energies and Proton Affinities of Alkylhydrazines," *J. Am. Chem. Soc.* **106**, 7384 (1984).
- ⁴⁸S. F. Nelsen, "Hydrazine-Hydrazine Cation Electron Transfer Reactions," in *Molecular Structure and Energetics, Volume 3*, (J. F. Liebman and A. Greenberg, Editors), VCH Publishers, Inc.: Deerfield Beach, Fla., 1986, p. 1.
- ⁴⁹K. Hiroaka, "Endothermic Ion-Molecule Reactions: The Reactions of H_3O^+ and H_3O^+ with Isobutane," *Int. J. Mass Spectrom. Ion Phys.* **27**, 139 (1978).
- ⁵⁰S. G. Lias, "Thermochemical Information from Ion-Molecule Rate Constants," in *Ion Cyclotron Resonance Spectrometry II*, Berlin: Springer-Verlag (1982), p. 409.
- ⁵¹(a) P. B. Armentrout and J. L. Beauchamp, "Ion Beam Studies of the Reactions of Atomic Cobalt Ions with Alkanes: Determination of Metal-Hydrogen and Metal-Carbon Bond Energies and an Examination of the Mechanism by which Transition Metals Cleave Carbon-Carbon Bonds," *J. Am. Chem. Soc.* **103**, 784 (1981); (b) P. B. Armentrout, L. F. Halle, and J. L. Beauchamp, "Reaction of Cr^+ , Mn^+ , Fe^+ , Co^+ , and Ni^+ with O_2 and N_2O . Examination of the Translational Energy Dependence of the Cross Sections of Endothermic Reactions," *J. Chem. Phys.* **76**, 2449 (1982); and references cited therein.
- ⁵²E. Murad, "Abstraction Reactions of Ca^+ and Sr^+ Ions," *J. Chem. Phys.* **78**, 6611 (1983) and references cited therein.
- ⁵³W. S. Koski, "Reactions of Electronically Excited Positive Ions," in *Interactions between Ions and Molecules*, (P. Ausloos, ed.) New York and London: Plenum (1975), p. 215, and references cited therein.
- ⁵⁴J. L. Holmes, M. Fingas, and F. P. Lossing, "Towards a General Scheme for Estimating the Heats of Formation of Organic Ions in the Gas Phase. Part I. Odd-Electron Cations," *Can. J. Chem.* **59**, 80 (1981).
- ⁵⁵J. L. Holmes and F. P. Lossing, "Towards a General Scheme for Estimating the Heats of Formation of Organic Ions in the Gas Phase. Part II. The Effect of Substitution at Charge-Bearing Sites," *Can. J. Chem.* **60**, 2365 (1982).
- ⁵⁶J. L. Holmes and F. P. Lossing, "The Need for Adequate Thermochemical Data for the Interpretation of Fragmentation Mechanisms and Ion Structure Assignments," *Int. J. Mass Spectrom. Ion Phys.* **47**, 133 (1983).
- ⁵⁷F. P. Lossing and J. L. Holmes, "Stabilization Energy and Ion Size in Carbocations in the Gas Phase," *J. Am. Chem. Soc.* **106**, 6917 (1984).
- ⁵⁸M. Bachiri, G. Mouvier, P. Carlier, and J. E. Dubois, "Evaluation Quantitative des Effets de Substituants sur les Premiers Potentiels d'Ionisation de Composés Monofonctionnels Aliphatiques," *J. Chim. Phys.* **77**, 899 (1980).
- ⁵⁹J. F. Liebman, unpublished results, derived from his earlier equations discussed in *Structural Thermochemistry and Reactivity of Ions*, P. Ausloos and S. G. Lias, Editors, Amsterdam, D. Reidel, 371 (1987).
- ⁶⁰S. G. Lias and P. Ausloos in "Ion-Molecule Reactions: Their Role in Radiation Chemistry," Washington: Am. Chem. Soc. (1975), pp. 91-95.
- ⁶¹S. G. Lias, J.-A. A. Jackson, H. Argentar, and J. F. Liebman, "Substituted N,N-Dialkylanilines: Relative Ionization Energies and Proton Affinities through Determination of Ion-Molecule Reaction Equilibrium Constants," *J. Org. Chem.* **50**, 333 (1985).
- ⁶²D. H. Aue, H. M. Webb, and M. T. Bowers, "Quantitative Relative Gas Phase Basicities of Alkylamines: Correlations with Solution Basicity," *J. Am. Chem. Soc.* **94**, 4726 (1972).
- ⁶³J. Koppel, U. Molder, and R. Pikver, "On Relationship Between Ionization Potentials and Proton Affinities in Gas Phase," *Org. Reactivity (Engl. Ed.)* **17**, 457 (1980).
- ⁶⁴L. S. Levitt and H. F. Widing, "The Alkyl Inductive Effect. Calculation of Inductive Substituent Parameters," *Prog. Phys. Org. Chem.* **12**, 119 (1977).
- ⁶⁵J. A. Koppel and U. H. Molder, "The Dependence of Ionization Potentials and Proton Affinities on Structure. II. Proton Affinities. Correlations with Substituent Constants and Polarizability," *Org. Reactivity* **20**, 3 (1983).
- ⁶⁶J. A. Koppel and U. H. Molder, "Internal Substituent Parameters and Correlation of Proton Affinities," *Org. Reactivity (Engl. Ed.)* **21**, 213 (1984).
- ⁶⁷J. A. Koppel, U. H. Molder, and R. I. Pikver, "The Dependence of Ionization Potentials and Proton Affinities on Structure. I. Ionization Potentials. Correlations with Substituent Constants and Polarizability," *Org. Reactivity (Engl. Ed.)* **18**, 380 (1981).
- ⁶⁸R. D. Mead, A. E. Stevens, W. C. Lineberger, *Gas Phase Ion Chemistry*, Vol. 3, M.T. Bowers, Ed. Academic Press, Orlando, FL, 1984; Ch. 22 "Photodetachment in Negative Ion Beams".
- ⁶⁹P. S. Drzaic, J. Marks, J. I. Brauman *Gas Phase Ion Chemistry*, Vol. 3, M.T. Bowers, Ed. Academic Press, Orlando, FL, (1984); Ch. 21 "Electron Photodetachment from Gas Phase Molecular Anions".
- ⁷⁰H. Hotop and W. C. Lineberger, "Binding Energies in Atomic Negative Ions: II," *J. Phys. Chem. Ref. Data* **14**, 731 (1985).
- ⁷¹T. M. Miller, *Adv. Electron. Electron Phys.* 1981 **55**, 119.
- ⁷²R. R. Corderman and W.C. Lineberger *Ann. Rev. Phys. Chem.* **30**, 347 (1979).
- ⁷³R. C. Dunbar *Gas Phase Ion Chemistry*, Vol. 2, M.T. Bowers, Ed. Academic Press, Orlando, FL, (1979).
- ⁷⁴R. S. Berry and S. Leach, *Adv. Electron. Electron Phys.* **57**, 1 (1982).
- ⁷⁵P. Kebarle and S. Chowdhury, "Electron Affinities and Electron Transfer Reactions," *Chem. Rev.* **87**, 513 (1987).
- ⁷⁶D. M. Neumark, K. R. Lykke, T. Andersen and W. C. Lineberger, "Laser Photodetachment Measurement of the Electron Affinity of Atomic Oxygen," *Phys. Rev. A* **32**, 1890 (1985).
- ⁷⁷J. H. Richardson, L. M. Stephenson and J. I. Brauman, "Photodetachment of Electrons from Trifluoromethyl and Trifluorosilyl Ions; the Electron Affinities of CF_3 and SiF_3 ," *Chem. Phys. Lett.* **30**, 17 (1975).
- ⁷⁸J. E. Bartmess in "Structure/Thermochemistry and Reactivity of Ions" (P. Ausloos and S. G. Lias, Editors), Amsterdam: D. Reidel (1987).
- ⁷⁹J. B. Cumming and P. Kebarle, "Summary of Gas Phase Measurements Involving Acids, AH. Entropy Changes in Proton Transfer Reactions Involving Negative Ions. Bond Dissociation Energies D(A-H) and Electron Affinities, EA(A)," *Can. J. Chem.* **56**, 1 (1978).

- ⁸⁰J. E. Bartmess, J. A. Scott, and R. T. McIver, Jr., "The Gas Phase Acidity Scale from Methanol to Phenol," *J. Am. Chem. Soc.* **101**, 6047 (1979).
- ⁸¹R. W. Taft, personal communication.
- ⁸²J. E. Bartmess, to be published.
- ⁸³K. Lacmann and D. R. Herschbach, "Collisional Excitation and Ionization of K Atoms by Diatomic Molecules: Role of Ion-pair States," *Chem. Phys. Lett.* **6**, 106 (1970).
- ⁸⁴F. M. Page and G. C. Goode, "Negative Ions and the Magnetron," New York: Wiley (1969).
- ⁸⁵L. G. Christophorou, P. M. Collins and J. C. Carter, "Electron Attachment in the Field of the Ground and Excited States of the Azulene Molecule," *J. Chem. Phys.* **52**, 4413 (1970).
- ⁸⁶G. S. A. McLuckey, D. Cameron and R. G. Cooks, "Proton Affinities from the Dissociation of Proton Bound Dimers," *J. Am. Chem. Soc.* **103**, 1313 (1981); (b) Boand, R. Houriet, and T. Gaumann, "The Gas Phase Acidity of Aliphatic Alcohols," *J. Am. Chem. Soc.* **105**, 2203 (1983).
- ⁸⁷D. J. Burinsky, E. K. Fukuda and J. E. Campana, "Electron Affinities from Dissociations of Mixed Negative Ion Dimers," *J. Am. Chem. Soc.* **106**, 2770 (1984).
- ⁸⁸C. H. DePuy, V. M. Bierbaum and R. Damrauer, "Relative Gas Phase Acidities of the Alkanes," *J. Am. Chem. Soc.* **106**, 4051 (1984).
- ⁸⁹E. C. M. Chen and W. E. Wentworth, *J. Phys. Chem.* **87**, 45 (1983).
- ⁹⁰C. R. Webster, I. S. McDermid, and C. T. Rettner, "Laser Optogalvanic Photodetachment Spectroscopy: A New Technique for Studying Photodetachment Thresholds with Application to I^- ," *J. Chem. Phys.* **78**, 646 (1983).
- ⁹¹J. B. Pedley and J. Rylance, *Sussex-N. P. L. Computer Analysed Thermochemical Data: Organic and Organometallic Compounds*, University of Sussex (1977).
- ⁹²J. B. Pedley, R. D. Naylor, and S. P. Kirby, *Thermochemical Data of Organic Compounds*, Second Edition, London/New York: Chapman and Hall (1986).
- ⁹³D. F. McMillen and D. M. Golden, *The Annual Review of Physical Chemistry*, Vol. 33, Annual Reviews, Palo Alto, CA (1982).
- ⁹⁴W. Tsang, "The Stability of Alkyl Radicals," *J. Am. Chem. Soc.* **107**, 2872 (1985).
- ⁹⁵J. S. Chickos, "Heats of Sublimation" in *Molecular Structure and Energetics*, Vol. 2 (J. F. Liebman and A. Greenberg, Editors), VCH Publishers, Deerfield Beach 67 (1987).
- ⁹⁶V. Majer and V. Svoboda, *Enthalpies of Vaporization of Organic Compounds: A Critical Review and Data Compilation*, Oxford: Blackwell Scientific Publications, 1985.
- ⁹⁷J. S. Chickos, A. S. Hyman, L. H. Ladon, and J. F. Liebman, *J. Org. Chem.* **46**, 4294 (1981).
- ⁹⁸J. F. Liebman and D. Van Vechten, "Universality: The Differences and Equivalences of Heats of Formation, Strain Energy, and Resonance Energy," in *Molecular Structure and Energetics*, Vol. 2, 315 (J. F. Liebman and A. Greenberg, Editors), VCH Publishers, Deerfield Beach (1987).
- ⁹⁹(a) S. W. Benson and J. H. Buss, "Additivity Rules for the Estimation of Molecular Properties. Thermodynamic Properties," *J. Chem. Phys.* **29**, 546 (1958); (b) S. W. Benson, *Thermochemical Kinetics: Methods for the Estimation of Thermochemical Data and Rate Parameters*, 2nd Ed. John Wiley & Sons, New York (1976).
- ¹⁰⁰For a recent example, see: K. Bystrom, "Enthalpies of Combustion, Vaporization and Formation for Di-n-propyldiazene N-oxide and Di-t-butyldiazene N-oxide," *J. Chem. Thermodyn.* **13**, 139 (1981).
- ¹⁰¹H. M. Rosenstock, J. Dannacher, and J. F. Liebman, "The Role of Excited Electronic States in Ion Fragmentation: $C_6H_6^+$," *Radiat. Phys. Chem.* **20**, 7 (1982).
- ¹⁰²J. F. Liebman, "Macroincrementation Reactions: A Holistic Estimation Approach for the Properties of Organic Compounds," in *Molecular Structure and Energetics*, Vol. 3 (J. F. Liebman and A. Greenberg, Editors), VCH Publishers, Deerfield Beach 262 (1986).

Table 1. Positive Ion Table

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------------|-------------------|---|--------------------------|--------------------|------------------------------|------------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Ac ⁺ | Ac | 5.17±0.12 | 216 | 905 | 97 | 406 | 82TN270 | 7440-34-8 |
| Ag ⁺ | Ag | 7.576 | 242.7 242.6 | 1015.6 1015.1 | 68.0 67.9 | 284.6 284.1 | 82TN270 | 7440-22-4 |
| | | See also: 80KRA. | | | | | | |
| AgAl ⁺ | AgAl | (7.8±0.5) 0 K values. | (287) | (1200) | 107 | 448 | 79HUB/HER | 12379-67-8 |
| AgBr ⁺ | AgBr | ≤9.59 0 K values. | ≤246 | ≤1028 | 25 | 103 | 79HUB/HER | 7785-23-1 |
| AgCl ⁺ | AgCl | (≤10.08) 0 K values. | (≤255) | (≤1065) | 22 | 93 | 79HUB/HER | 7783-90-6 |
| AgF ⁺ | AgF | (11.0±0.3) 0 K values. | (256) | (1071) | 2 | 10 | 79HUB/HER | 7775-41-9 |
| AgH ⁺ | AgH | (9.2) $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM). 0 K value. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | (280) | (1170) | 67 | 282 | 79HUB/HER | |
| Ag ₂ ⁺ | Ag ₂ | (7.35) | (267) (268) | (1119) (1120) | 98.0 98 | 410.0 411 | 82TN270 | 12187-06-3 |
| Al ⁺ | Al | 5.986 | 216.3 216.8 | 904.9 907.3 | 78.2 78.8 | 327.3 329.7 | 85JANAF | 7429-90-5 |
| AlAu ⁺ | AuAl | (7.6±0.3) 0 K values. | (263) | (1101) | 88 | 368 | 79HUB/HER | 12250-38-3 |
| AlBO ₂ ⁺ | AlBO ₂ | (9.5±0.5) | (90) | (376) | -129±4 | -541±17 | 71JANAF | |
| AlBr ⁺ | AlBr | (9.3) | (218.3) (220.0) | (913.2) (920.4) | 3.8±3.0 5.5±3.0 | 15.9±12.6 23.1±12.5 | 85JANAF | 22359-97-3 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| AlBr_3^+ AlBr ₃ | (10.4) | (142) (147) | (593) (616) | -98.1 -92.5 | -410.4 -387.2 | 85JANAF | 7727-15-3 |
| IP is onset of photoelectron band. | | | | | | | |
| AlCl^+ AlCl | 9.4 | 204 204 | 855 855 | -12.3 -12.3 | -51.5 -51.7 | 85JANAF | 13595-81-8 |
| AlCIF^+ AlCIF | (7.9±1.0) | (66) (65) | (276) (271) | -117 -117 | -489 -488 | 85JANAF | |
| AlCl_3^+ AlCl ₃ | (12.01) | (137) (138) | (574) (576) | -140 -139 | -585 -583 | 85JANAF | 7446-70-0 |
| AlF^+ AlF | 9.73±0.01 | 160.9 160.9 | 673.1 673.9 | -63.5±0.8 -63.5±0.8 | -265.7±3.4 -265.6±3.4 | 79HUB/HER | 13595-82-9 |
| IP from 84DYK/KIR. | | | | | | | |
| AlF_2^+ AlF ₂ | (8.1) | (8) | (33) | -179 | -749 | 81WOO | 13569-23-8 |
| IP from 85JANAF. | | | | | | | |
| AlF_3^+ AlF ₃ | ≤15.45 | ≤67 ≤68 | ≤282 ≤285 | -289 -288 | -1209 -1206 | 85JANAF | 7784-18-1 |
| IP from 84DYK/KIR. | | | | | | | |
| AlI^+ AlI | (9.3±0.3) | (230.7) (231.0) | (965.3) (966.6) | 16.3±1 16.6±1 | 68.0±4.2 69.3±4.2 | 85JANAF | 29977-41-1 |
| AlI_3^+ AlI ₃ | (9.1) | (160) | (670) | -49 | -208 | 82TN270 | 7784-23-8 |
| IP is onset of photoelectron band. | | | | | | | |
| AlO^+ AlO | 9.46±0.06 | 234.1 234.2 | 979.6 979.7 | 16.0±2 16.0±2 | 66.9±8 67.0±8 | 85JANAF | 14457-64-8 |
| IP from 82ARM/HAL. See also: 80MUR/HIL, 81KAP/STA. | | | | | | | |
| AlO_2^+ AlO ₂ | (10.0±1.0) | (200) | (835) | -31 | -130 | 82KAS/CHE | 11092-32-3 |
| AlP^+ AlP | (8.4±0.4) | (295) | (1232) | 101 | 422 | 79HUB/HER | 20859-73-8 |
| 0 K values. | | | | | | | |

Table 1. Positive Ion Table - Continued


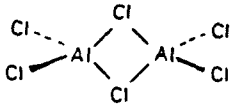
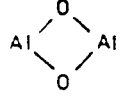
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | ● CAS registry number |
|--|--|---|----------------------------------|------------------------------|------------------------|----------------------|--------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| AlSe^+ AlSe | (8.3±0.5) 0 K values. | (243) | (1016) | 52 | 215 | 79HUB/HER | 23330-87-2 |
| Al_2^+ Al ₂ | (5.4±1.0) | (240.9) (240.8) | (1008.0) (1007.3) | 116.4±0.8 116.2±0.8 | 487.0±3.5 486.3±3.5 | 85JANAF | 32752-94-6 |
| Al_2Br_6^+  | (10.97) | (21) | (87) | -232 | -971 | 82TN270 | 18898-34-5 |
| Al_2Cl_6^+  | (12.18) | (-28) | (-116) | -309 | -1291 | 82TN270 | 13845-12-0 |
| Al_2O^+ Al ₂ O | (7.7±0.2) | (144) (145) | (603) (605) | -33±5 -33 | -140±22 -138 | 82KAS/CHE | 12004-36-3 |
| Al_2O_2^+  | (9.9±0.5) | (131) | (551) | -97±12 | -404±48 | 82KAS/CHE | 12252-63-0 |
| Am^+ Am | 5.99 See also: 81CHE/GAB. | 206 | 862 | 68 | 284 | 85KLE/WAR | 7440-35-9 |
| Ar^+ Ar | 15.75973±0.00001 See also: 81KIM/KAT. | <u>363.42</u> <u>363.42</u> | <u>1520.57</u> <u>1520.57</u> | 0 | 0 | *DEF | 7440-37-1 |
| ArH^+ ArH | | 277 | 1159 | | | | |
| | | From proton affinity of Ar (RN 7440-37-1). PA = 88.6 kcal/mol, 371. kJ/mol. | | | | | |
| ArHe^+ ArHe | 15.735 $\Delta_f H(\text{Ion})$ from 81DAB/HER. 0 K values. | 362.8 | 1518.0 | -0.055 | -0.23 | 79HUB/HER | 12254-69-2 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|------------------|------------------------------|----------------|----------------------|---|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| ArHg^+ ArHg | (10.217±0.012) IP from 85LIN/BRO. | (250.6) | (1048.6) | 15.0 | 62.8 | 84BOU/BRA | 87193-95-1 |
| ArKr^+ ArKr | 13.484±0.015 IP from 82DEH/PRA. 0 K values. | 310.6 | 1299.7 | -0.3 | -1.3 | 79HUB/HER | 51184-77-1 |
| ArNe^+ ArNe | 15.685±0.004 IP from 82PRA/DEH2. 0 K values. | 361.6 | 1512.9 | -0.10 | -0.43 | 76BOB/BAR | 12301-65-4 |
| ArXe^+ ArXe | 11.968±0.012 IP from 82DEH/PRA. See also: 85PRA/DEH, 85PRA/DEH2. 0 K values. | 275.7 | 1153.4 | -0.32 | -1.35 | 76BOB/BAR | 58206-67-0 |
| Ar_2^+ Ar ₂ | 14.501±0.025 IP from 82DEH/PRA2. See also: 81DEH/POL, 82LEV/LIA. 0 K values. | 334.2 | 1398.1 | -0.24 | -1.01 | 79HUB/HER | 12595-59-4 |
| As^+ As | 9.7883±0.0002 | 298.0 297.7 | 1246.9 1245.8 | 72.3 72.0 | 302.5 301.4 | 82TN270 | 7440-38-2 |
| AsBr_3^+ AsBr ₃ | (10.0) IP is onset of photoelectron band. | (200) (205) | (835) (858) | -31 -25.5 | -130 -106.9 | 82TN270 | 7784-33-0 |
| AsClO^+ AsOCl | (11.1) IP from 83BIN. | (249) | (1040) | -7 | -31 | 83BIN | 14525-25-8 |
| AsCl_3^+ AsCl ₃ | (10.55±0.025) See also: 83OZG. | (181) (181) | (756) (758) | -63 -62 | -262 -260 | 82TN270 | 7784-34-1 |
| AsF_3^+ AsF ₃ | (12.84±0.05) | (108) (109) | (453) (457) | -188 -187 | -786 -782 | 82TN270 | 7784-35-2 |
| AsF_3H^+ F ₃ AsH | | 23 | 96 | | | | From proton affinity of AsF ₃ (RN 7784-35-2). PA = 155 kcal/mol, 648 kJ/mol. |
| AsH_3^+ AsH ₃ | 9.89 See also: 82ELB/DIE. | 244 246 | 1020 1028 | 16 18 | 66 74 | 82TN270 | 7784-42-1 |

Table 1. Positive Ion Table - Continued


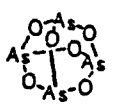
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------------|---|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| AsH_4^+ AsH ₄ | | 202 | 846 | | | | |
| | | From proton affinity of AsH ₃ (RN 7784-42-1). PA = 179.2 kcal/mol, 750 kJ/mol. | | | | | |
| As_2^+ As ₂ | (10.1±0.2) | (278) | (1165) | 45.5±0.7 | 190.4±2.9 | 73BEN/MAR | 23878-46-8 |
| | | (278) | (1164) | 45.5±0.7 | 190.4±2.9 | | |
| | See also: 8SHIR/STR. | | | | | | |
| As_4^+  | (9.07±0.07) | (244) | (1019) | 34 | 144 | 82TN270 | 12187-08-5 |
| | See also: 8SHIR/STR. | | | | | | |
| As_4O_6^+  | (9.6) | (-68) | (-283) | -289 | -1209 | 82TN270 | 12505-67-8 |
| | IP is onset of photoelectron band. | | | | | | |
| Au^+ Au | 9.225 | 300 | 1256 | 87 | 366 | 82TN270 | 7440-57-5 |
| | | 300.2 | 1256.0 | 87.5 | 365.9 | | |
| AuB^+ AuB | (8.7±0.5) | (337) | (1411) | 137 | 572 | 79HUB/HER | 12408-81-0 |
| | 0 K values. | | | | | | |
| AuCe^+ AuCe | (6.0±0.3) | (248) | (1036) | 109 | 457 | 82TN270 | 12408-82-1 |
| | | (248) | (1039) | 110 | 460 | | |
| AuHo^+ AuHo | (6.2±0.5) | (242) | (1013) | 99.1 | 414.5 | 82TN270 | 12044-80-3 |
| | | (243) | (1016) | 100 | 418 | | |
| AuLa^+ AuLa | (5.9±0.5) | (247) | (1033) | 111 | 464 | 82TN270 | 12429-32-2 |
| | | (247) | (1035) | 111 | 466 | | |
| AuNd^+ AuNd | (5.8±0.8) | (228) | (955) | 94 | 395 | 82TN270 | 12429-33-3 |
| | | (229) | (957) | 95 | 397 | | |
| AuPr^+ AuPr | (5.4±0.8) | (224) | (937) | 99 | 416 | 82TN270 | 12429-34-4 |
| | | (224) | (939) | 100 | 418 | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--------------------|------------------------------------|------------------------------|--------------------------------|------------------------------|----------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| AuSi^+ | AuSi | (9.5±0.5) 0 K values. | (340) | (1422) | (121) | (505) | 79HUB/HER | 12256-53-0 |
| Au_2^+ | Au ₂ | (9.5±0.3) | (343) (341) | (1433) (1427) | 123.5 122.0 | 516.7 510.4 | 79HUB/HER | 12187-09-6 |
| B^+ | B | 8.29808±0.00002 | <u>325.9</u> <u>324.7</u> | <u>1363.3</u> <u>1358.2</u> | 134.5 133.3 | 562.7 557.6 | 82TN270 | 7440-42-8 |
| BBr_2H^+ | BHBr ₂ | (10.92±0.02) IP from 81FRO/KIR. | (227) (231) | (949) (966) | -25±5 -21±5 | -105±21 -88±21 | 71JANAF | 13709-65-4 |
| BBr_3^+ | BBr ₃ | (10.51±0.02) | (194) | (810) | -49±0.2 | -204±1 | 71JANAF | 10294-33-4 |
| BCl^+ | BCl | (10.2) | (269.0) (268.3) | (1125.5) (1122.5) | 33.8 33.1 | 141.4 138.4 | 85JANAF | 20583-55-5 |
| $\Delta_f H(\text{Ion})$ from appearance potential (18.37±0.02 eV) in BCl ₃ . Cited ionization potential is difference between heats of formation of ion and neutral. | | | | | | | | |
| BClF_2^+ | BClF ₂ | (13.06±0.11) | (88) | (370) | -213 | -890 | 82TN270 | 14720-30-0 |
| BCl_2^+ | BCl ₂ | (7.8) | (159) | (664) | -20±15 | -83±63 | 71JANAF | 13842-52-9 |
| $\Delta_f H(\text{Ion})$ from appearance potential (12.30±0.02 eV) in BCl ₃ . Cited ionization potential is difference between heats of formation of ion and neutral. | | | | | | | | |
| BCl_2F^+ | BCl ₂ F | (12.18±0.10) | (130) | (544) | -151 | -631 | 82TN270 | 14720-31-1 |
| BCl_2H^+ | HBCl ₂ | (11.91±0.02) IP from 81FRO/KIR. | (215) (216) | (901) (904) | -59.3±1 -58.6±1 | -248.1±4 -245.2±4 | 71JANAF | 10325-39-0 |
| BCl_3^+ | BCl ₃ | 11.60±0.02 | 171 171 | 715 716 | -96 -96 | -404 -403 | 82TN270 | 10294-34-5 |
| BF^+ | BF | 11.12±0.01 IP from 83DYK/KIR. | 229 228 | 957 954 | -27.7 -28.4 | -115.8 -118.8 | 79HUB/HER | 13768-60-0 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| BFS⁺ FBS | (10.90±0.01) IP from 84COO/KRO. | (165) | (690) | -86 | -362 | *EST | 83995-89-5 |
| BF₂⁺ BF ₂ | (9.4) From appearance potential of 15.81±0.04 eV in BF ₃ . IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | (75) | (314) | -141.0±3 | -589.9±12 | 71JANAF | |
| BF₂H⁺ HBF ₂ | (13.60±0.05) IP from 81CHO/KIR. | (138) (139) | (578) (582) | -175.4±0.8 -174.5±0.8 | -733.8±3.3 -730.1±3.3 | 71JANAF | 13709-83-6 |
| BF₃⁺ BF ₃ | 15.56±0.03 See also: 84FAR/SRI, 84DEH/PAR, 81ASB/SVE, 81KIM/KAT. | 87.1 87.8 | 364.3 367.3 | -271.7 -271.0 | -1137.0 -1134.0 | 82TN270 | 7637-07-2 |
| BH⁺ BH | 9.77±0.05 | 331.1 330.3 | 1385.4 1382.2 | 105.8±2.0 105.0±2.0 | 442.7±8.4 439.5±8.4 | 85JANAF | 13766-26-2 |
| BHO₂⁺ BHO ₂ | (12.6±0.2) | (156) | (654) | -134 | -562 | 82TN270 | 13460-50-9 |
| BHS⁺ HBS | 11.11±0.03 | (268) | (1122) | 12±10 | 50±42 | 78JANAF | 14457-85-3 |
| BH₂⁺ BH ₂ | (9.8±0.2) | (274) | (1146) | 48±15 | 201±63 | 71JANAF | 14452-64-3 |
| BH₃⁺ BH ₃ | 12.3±0.1 | (308) | (1287) | 24 | 100 | 82TN270 | 13283-31-3 |
| BI₃⁺ BI ₃ | (9.25±0.03) | (230) (231) | (964) (967) | 17.0 18 | 71.1 75 | 82TN270 | 13517-10-7 |
| BKO₂⁺ KBO ₂ | (8.62±0.14) See also: 85FAR/SRI. | (38) | (160) | -161±2 | -672±10 | 85FAR/SRI | |
| BLiO⁺ LiBO | 7.7±0.5 IP from 85NEU. | (136) | (568) | -42 | -175 | *EST | 77965-53-8 |
| BLiO₂⁺ LiBO ₂ | (9.8±0.5) IP from 85NEU. | (66) | (274) | -160 | -671 | 71JANAF | |

Table 1. Positive Ion Table - Continued

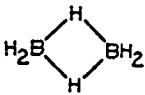
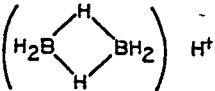
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---------------------------|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| BNaO_2^+ | NaBO_2 | (9.18±0.10) | (58) | (242) | -154 | -644 | 82TN270 | |
| | | | (59) | (244) | -153 | -642 | | |
| BO^+ | BO | 13.0±0.3 | (300) | (1254) | 0 | 0 | 83PED/MAR | 12505-77-0 |
| | | | (299) | (1251) | -1 | -3 | | |
| | | See also: 79BAG/NIK, 85NEU. | | | | | | |
| BO_2^+ | BO_2 | (13.5±0.3) | (240) | (1003) | -72 | -300 | 82TN270 | 13840-88-5 |
| | | IP from 79BAG/NIK. | | | | | | |
| BSe^+ | BSe | (10.3) | (315) | (1320) | 78 | 326 | 79HUB/HER | 29750-36-5 |
| | | 0 K values. | | | | | | |
| B_2Cl_2^+ | B_2Cl_2 | | (223) | (934) | | | | |
| | | From appearance potential (17.24±0.03 eV) in B_2Cl_4 assumed to give $\text{B}_2\text{Cl}_2^+ + 2\text{Cl}$. | | | | | | |
| B_2Cl_3^+ | B_2Cl_3 | | 120 | 502 | | | | |
| | | From appearance potential of 11.52±0.02 eV in B_2Cl_4 . 0 K values. | | | | | | |
| B_2Cl_4^+ | B_2Cl_4 | 10.32±0.02 | 121 | 506 | -117 | -490 | 82TN270 | 13701-67-2 |
| | | | 121 | 506 | -117 | -490 | | |
| B_2F_3^+ | B_2F_3 | | -7 | -28 | | | | |
| | | From appearance potential of 15.40±0.01 eV in B_2F_4 . 0 K values. | | | | | | |
| B_2F_4^+ | B_2F_4 | 12.07±0.01 | -66 | -275 | -344 | -1440 | 82TN270 | 13965-73-6 |
| | | | -65 | -272 | -343 | -1437 | | |
| B_2H_6^+ |  | 11.38±0.03 | 271 | 1134 | 8.5 | 35.6 | 82TN270 | 19287-45-7 |
| | | | 275 | 1149 | 12.3 | 51.4 | | |
| | | See also: 81ASB/SVE, 81KIM/KAT. | | | | | | |
| B_2H_7^+ |  | | 228 | 955 | | | | |
| | | From proton affinity of Diborane(6). (RN 19287-45-7). PA = -146 kcal/mol, -611 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

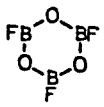
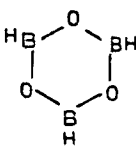
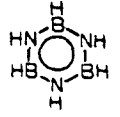
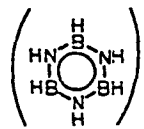
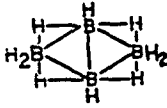
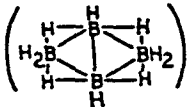
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------|---|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| B_2O_2^+ O=BB=O | 13.58 | 204 | 854 | -109±2 | -456±8 | 71JANAF | 13766-28-4 |
| IP from 84RUS/CUR. See also: 79BAG/NIK. | | | | | | | |
| B_2O_3^+ B_2O_3 | 13.5±0.15 | 110 110 | 460 462 | -201.3 -201 | -842.1 -841 | 79BAG/NIK | 1303-86-2 |
| IP from 79BAG/NIK. | | | | | | | |
| $\text{B}_3\text{F}_3\text{O}_3^+$  | (13.9±0.1) | (-244) | (-1024) | -565±1 | -2365±4 | 71JANAF | 13703-95-2 |
| $\text{B}_3\text{H}_3\text{O}_3^+$  | (13.5±0.5) | (20) | (85) | -291 | -1218 | 71JANAF | 289-56-5 |
| $\text{B}_3\text{H}_6\text{N}_3^+$  | 9.88±0.02 | 106 112 | 441 467 | -122.3 -116.2 | -511.8 -486.2 | 82TN270 | 6569-51-3 |
| $\text{B}_3\text{H}_7\text{N}_3^+$  | | 49 | 206 | From proton affinity of borazine (RN 6569-51-3). PA = 194.1 kcal/mol, 812 kJ/mol. | | | |
| $\text{B}_4\text{H}_{10}^+$  | 10.76±0.04 | 264 | 1104 | 16 | 66 | 82TN270 | 18283-93-7 |
| $\text{B}_4\text{H}_{11}^+$  | | 237 | 993 | From proton affinity of B_4H_{10} (RN 18283-93-7). PA = -144 kcal/mol, -602 kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

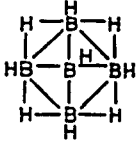
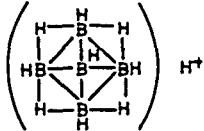
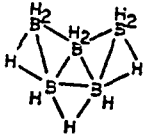
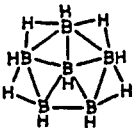
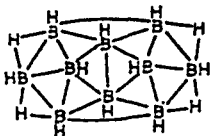
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| B_5H_9^+ | 9.90±0.04 | 246 | 1028 | 17.5 | 73.2 | 82TN270 | 19624-22-7 |
|  | | 253 | 1057 | 24.4 | 102.1 | | |
| $\text{B}_5\text{H}_{10}^+$ | | 214 | 896 | | | | |
|  | | From proton affinity of B_5H_9 (RN 19624-22-7). PA = 169 kcal/mol, 707 kJ/mol. | | | | | |
| $\text{B}_5\text{H}_{11}^+$ | (10.1) | (257) | (1078) | 24.6 | 103.3 | 82TN270 | 18433-84-6 |
|  | IP is onset of photoelectron band. | | | | | | |
| $\text{B}_6\text{H}_{10}^+$ | (9.0) | (230) | (963) | 23 | 95 | 82TN270 | 2377-80-2 |
|  | IP is onset of photoelectron band. | | | | | | |
| $\text{B}_{10}\text{H}_{14}^+$ | 9.88±0.03 | 235 | 985 | 7.6 | 31.6 | 82TN270 | 17702-41-9 |
|  | | 247 | 1031 | 18.7 | 78.1 | | |
| Ba^+ | | | | | | | |
| Ba | 5.212 | 163 | 683 | 43 | 180 | 82TN270 | 7440-39-3 |
| | | 163 | 684 | 43 | 181 | | |
| BaBr^+ | | | | | | | |
| BaBr | (5.0) | (88.9) | (371.8) | -26.4±10.0 | -111±41 | 85JANAF | 14832-97-4 |
| | | (91.0) | (380.8) | -24.3±10.0 | -102±41 | | |
| BaBr_2^+ | | | | | | | |
| BaBr ₂ | (8.5) | (90) | (377) | -106 | -443 | 82EMO/KIE | 10553-31-8 |
| | | (93) | (391) | -102 | -429 | | |
| | IP is onset of photoelectron band (79LEE/POT). See also: 82EMO/KIE, 79LEE/POT2. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| BaCl⁺ | | | | | | | |
| BaCl | 5.01±0.010 | 81 | 340 | -34 | -143 | 85JANAF | 14832-99-6 |
| | | 82 | 341 | -34 | -142 | | |
| BaCl₂⁺ | | | | | | | |
| BaCl ₂ | (9.4) | (97) | (405) | -120 | -502 | 82EMO/KIE | 10361-37-2 |
| | IP is onset of photoelectron band (79LEE/POT, 79LEE/POT2). See also: 82EMO/KIE. 0 K values. | | | | | | |
| BaF⁺ | | | | | | | |
| BaF | (4.8±0.3) | (34) | (144) | -77 | -324 | 82TN270 | 13966-70-6 |
| | | (34) | (142) | -77 | -326 | | |
| BaHO⁺ | | | | | | | |
| BaOH | 4.35±0.3 | 45 | 189 | -55±4 | -230±17 | 81MUR | 12009-08-4 |
| | IP from 81MUR. See also: 81FAR/SRI. | | | | | | |
| BaH₂O₂⁺ | | | | | | | |
| Ba(OH) ₂ | (8) | (44) | (186) | -140 | -586 | 82TN270 | 17194-00-2 |
| | IP from 81FAR/SRI. | | | | | | |
| BaI⁺ | | | | | | | |
| BaI | (5.0±0.3) | (105.2) | (440.0) | -10.1±20.1 | -42.4±84 | 85JANAF | 12524-20-8 |
| | | (105.9) | (443.1) | -9.4±20.1 | -39.3±84 | | |
| BaI₂⁺ | | | | | | | |
| BaI ₂ | (8.24) | (116) | (487) | -74 | -308 | 82EMO/KIE | 13718-50-8 |
| | | (117) | (490) | -73 | -305 | | |
| | IP is onset of photoelectron band (79LEE/POT, 79LEE/POT2). See also: 82EMO/KIE. | | | | | | |
| BaO⁺ | | | | | | | |
| BaO | 6.91±0.06 | 129.8 | 542.9 | -29.6±2 | -123.8±8 | 85JANAF | 1304-28-5 |
| | | 130.3 | 545.1 | -29.1±2 | -121.6±8 | | |
| | See also: 81MUR. | | | | | | |
| BaO₄W⁺ | | | | | | | |
| BaWO ₄ | (9.8±0.5) | (-181) | (-757) | -407 | -1703 | 76DEL/HAL | |
| Be⁺ | | | | | | | |
| Be | 9.322 | <u>292.5</u> | <u>1223.7</u> | 77.5 | 324.3 | 82TN270 | 7440-41-7 |
| | | <u>291.5</u> | <u>1219.4</u> | 76.5 | 320.0 | | |
| BeCl₂⁺ | | | | | | | |
| BeCl ₂ | (11.15) | (171) | (717) | -86 | -359 | 82TN270 | 7787-47-5 |
| | IP is onset of photoelectron band (79LEE/POT2). | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ kcal/mol kJ/mol | $\Delta_f H(\text{Neutral})$ kcal/mol kJ/mol | Neutral reference | CAS registry number | |
|--|----------------------------|---|---|----------------------|------------------------|------------|
| BeF⁺ BeF | | | | | | |
| | | | -40.6±2 -41.2±2 | -169.9±8 -172.2±8 | 85JANAF | 13597-96-1 |
| A value of 168 kcal/mol, 703 kJ/mol is obtained for the enthalpy of formation of BeF ⁺ based on experimental ionization potential values of 9.1±0.5 or 9.3±1.0 eV; the enthalpy of formation based on an appearance potential of 15.4±0.4 eV in BeF ₂ is 147 kcal/mol, 615 kJ/mol. | | | | | | |
| BeF₂⁺ BeF ₂ | (14.6±0.5) | (147) (615) | -190 | -794 | 82TN270 | 7787-49-7 |
| BeH⁺ BeH | 8.21±0.04 | 272 1136 271 1132 | 82 81.2 | 344 339.8 | 79HUB/HER | 13597-97-2 |
| BeO⁺ BeO | (10.1±0.4) | (265.5) (1110.9) (264.9) (1108.5) | 32.6±3 32.0±3 | 136.4±13 134.0±13 | 85JANAF | 1304-56-9 |
| Be₂O⁺ Be ₂ O | (10.5±0.5) | (227) (950) | -15±10 | -63±42 | 71JANAF | 12009-99-3 |
| Be₂O₂⁺ (BeO) ₂ | (10.8±0.7) | (151) (632) | -98±12 | -410±50 | 71JANAF | 70478-90-9 |
| Be₃O₃⁺ (BeO) ₃ | (10.9±0.6) | (-1) (-2) | -252±9 | -1054±38 | 71JANAF | 61279-73-0 |
| Be₄O₄⁺ (BeO) ₄ | (11.0) | (-126) (-529) | -380±12 | -1590±50 | 71JANAF | 61279-74-1 |
| Be₅O₅⁺ (BeO) ₅ | (-11) | (-251) (-1052) | -505±23 | -2113±95 | 71JANAF | 61279-75-2 |
| Be₆O₆⁺ (BeO) ₆ | (-11) | (-239) (-1000) | -492±22 | -2061±92 | 71JANAF | 61279-76-3 |
| Bi⁺ Bi | 7.289 | 218 910 217.7 910.7 | 49 49.6 | 207 207.4 | 82TN270 | 7440-69-9 |
| BiCl₃⁺ BiCl ₃ | (10.4) | (176) (738) (177) (739) | -64 -63 | -266 -264 | 82TN270 | 7787-60-2 |
| IP is onset of photoelectron band (83NOV/POT). See also: 83OZG. | | | | | | |
| BiH₃⁺ BiH ₃ | (10.1) | (288) (1204) | 55 | 230 | 64GUN | 18288-22-7 |

Table 1. Positive Ion Table - Continued

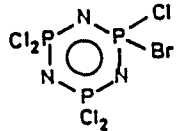
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---------------------------------------|---|--------------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| BiO^+ | BiO | (9.0±0.5) | (236) | (989) | 29±3 | 121±13 | 83PED/MAR | 1332-64-5 |
| | | | (236) | (988) | 29 | 120 | | |
| BiS^+ | BiS | (8.7±0.5) 0 K values. | (243) | (1017) | 42 | 178 | 79HUB/HER | 12048-34-9 |
| BiTe^+ | BiTe | (8.4±0.5) 0 K values. | (235) | (983) | 41 | 173 | 79HUB/HER | 12010-57-0 |
| Bi_2^+ | Bi ₂ | (7.3±0.5) | (221) | (924) | 53 | 220 | 82TN270 | 12187-12-1 |
| | | | (221) | (926) | 53.1 | 222.2 | | |
| Bk^+ | Bk | 6.30±0.09 | 219 | 918 | 74 | 310 | 85KLE/WAR | 7440-40-6 |
| Br^+ | Br | 11.814 | 299.2 | 1251.7 | 26.7 | 111.9 | 82BAU/COX | 10097-32-2 |
| | | | 300.6 | 1257.8 | 28.2 | 117.9 | | |
| | | See also: 81KIM/KAT. | | | | | | |
| BrCa^+ | CaBr | 5.54 | 123 | 513 | -5 | -21 | 79HUB/HER | 10024-43-8 |
| | | IP from 84MEY/SCH. 0 K values. | | | | | | |
| BrCl^+ | BrCl | 11.01 | 257 | 1077 | 4 | 15 | 82TN270 | 13863-41-7 |
| | | | 259 | 1084 | 5 | 22 | | |
| | | IP from 84DYK/JOS. | | | | | | |
| $\text{BrCl}_5\text{N}_3\text{P}_3^+$ |  | (9.83±0.1) | (52) | (218) | -174 | -730 | *EST | 14740-93-3 |
| BrCs^+ | CsBr | 7.72±0.05 | 130 | 545 | -48 | -200 | 84PAR/WEX | 7787-69-1 |
| | | | 133 | 554 | -45.5±1.8 | -190.4±7.5 | | |
| BrF^+ | BrF | 11.77±0.01 | 257.4 | 1077.1 | -14.0±0.4 | -58.5±1.7 | 85JANAF | 13863-59-7 |
| | | | 259.3 | 1084.8 | -12.1±0.4 | -50.8±1.7 | | |
| | | See also: 84DYK/JOS. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------------|------------------------------|--------------------|----------------------|--------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| BrF_3^+ BrF ₃ | (12.15±0.04) | (219) (222) | (917) (928) | -61.1 -58.4 | -255.6 -244.4 | 82TN270 | 7787-71-5 |
| BrF_5^+ BrF ₅ | (13.17±0.01) | (201) (205) | (842) (858) | -103 -99 | -429 -413 | 82TN270 | 7789-30-2 |
| BrH^+ HBr | 11.66±0.03 | 260 262 | 1089 1096 | -9 -7 | -36 -29 | 82TN270 | 10035-10-6 |
| IP from 79HUB/HER, 77ROS/DRA, 82LEV/LIA. See also: 81KIM/KAT. | | | | | | | |
| BrH_2^+ H ₂ Br | | 218 | 911 | | | | |
| From proton affinity of HBr (RN 10035-10-6) (84POL/MUN, 85MCM/KEB). PA = 139 kcal/mol, 582 kJ/mol. | | | | | | | |
| BrH_3Si^+ SiH ₃ Br | 10.6 | (226) | (945) | -19±4 | -78±17 | 82JANAF | 13465-73-1 |
| IP is onset of photoelectron band. | | | | | | | |
| BrI^+ IBr | 9.790±0.004 | 235.5 237.7 | 985.4 994.4 | 9.8 11.9 | 40.8 49.8 | 82TN270 | 7789-33-5 |
| See also: 84DYK/JOS, 71POT/PRI. | | | | | | | |
| BrIn^+ InBr | (9.09) | (202) (204) | (845) (854) | -8 -6 | -32 -23 | 79HUB/HER | 14280-53-6 |
| BrK^+ KBr | 7.85±0.1 | 138 140 | 577 586 | -43 -41 | -180 -171 | 82TN270 | 7758-02-3 |
| BrLi^+ LiBr | (8.7) | (164) (166) | (685) (693) | -37±3 -35±3 | -154±13 -146±13 | 71JANAF | 7550-35-8 |
| IP is onset of photoelectron band. | | | | | | | |
| BrNO^+ NOBr | 10.17±0.03 | 254 | 1063 | 20 | 82 | 82BAU/COX | 13444-87-6 |
| BrNa^+ NaBr | 8.31±0.1 | 157 160 | 659 668 | -34 -32 | -143 -134 | 82TN270 | 7647-15-6 |
| BrO^+ BrO | (10.2) | (265) (267) | (1110) (1118) | 30.1 31.9 | 125.8 133.5 | 82TN270 | 15656-19-6 14380-62-2 |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

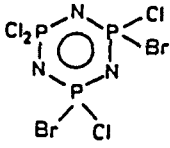
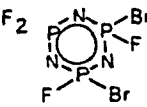
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|--------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| BrRb⁺ RbBr | 7.94±0.03 | 139 142 | 583 592 | -43.7 -41.4 | -182.8 -173.4 | 82TN270 | 7789-39-1 |
| BrSi⁺ SiBr | (7.3) | (224.6) (225.9) | (939.6) (945.1) | 56.3±11.0 57.6±11.0 | 235.3±46.0 240.8±46.0 | 85JANAF | 14791-57-2 |
| BrSr⁺ SrBr | (5.5) | (106) (108) | (442) (450) | -21±10 -19.2±10 | -89±42 -80.4±42 | 85JANAF | 14519-13-2 |
| BrTl⁺ TlBr | 9.14±0.02 | 202 204 | 844 853 | -9 -7 | -38 -29 | 82TN270 | 7789-40-4 |
| | | See also: 83BAN/BRI. | | | | | |
| Br₂⁺ Br ₂ | 10.515±0.005 | 250 253 | 1046 1061 | 7.4 11 | 31.0 46 | 82BAU/COX | 7726-95-6 |
| | | Cited IP leads to Br ₂ ⁺ (² Π _{3/2} g); formation of Br ₂ ⁺ (² Π _{1/2} g) requires 10.865±0.005 eV. IP from 84VAN/DEL2, 84DYK/JOS, 77ROS/DRA. See also: 81KIM/KAT. | | | | | |
| Br₂Ca⁺ CaBr ₂ | ≤9.68 | ≤130 ≤134 | ≤545 ≤560 | -93 -89±2 | -389 -374±9 | 82TPIS | 7789-41-5 |
| | | IP is onset of photoelectron band (79LEE/POT2). | | | | | |
| Br₂Cl₄N₃P₃⁺  | (9.80±0.1) | (63) | (265) | -163 | -681 | *EST | 15964-99-5 |
| Br₂F₄N₃P₃⁺  | (10.63±0.03) | (-135) | (-563) | -380 | -1589 | *EST | 29871-63-4 |
| | | IP from 81CLA/SOW. | | | | | |
| Br₂Fe⁺ FcBr ₂ | (10.7±0.5) | (237) | (991) | -10±0.5 | -41±2 | 71JANAF | 7789-46-0 |
| Br₂Ge⁺ GeBr ₂ | (9.60±0.05) | (206) | (863) | -15 | -63 | 82TN270 | 24415-00-7 |
| | | IP from 82JON/VAN. | | | | | |

Table 1. Positive Ion Table - Continued


| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|----------------------------|--------------------------|----------------|------------------------------|-----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{Br}_2\text{H}_2\text{Si}^+$ | SiH_2Br_2 | (10.7) | (201) | (842) | -45±4 | -190±17 | 82JANAF | 13768-94-0 |
| IP is onset of photoelectron band. | | | | | | | | |
| Br_2Hg^+ | HgBr_2 | 10.560±0.003 | 223 | 934 | -20±2 | -85±8 | 71JANAF | 7789-47-1 |
| Cited ionization potential (83LIN/TZE) refers to formation of $\text{HgBr}_2^+ (^2\Pi_{3/2} \text{g})$. IP for formation of $\text{HgBr}_2^+ (^2\Pi_{1/2} \text{g})$ is 10.8846±0.0012 eV. See also: 81LEE/POT. | | | | | | | | |
| Br_2Li_2^+ |  | (≤10.05±0.08) | (≤112) | (≤469) | -120 | -501 | 81LIN/BES | 12380-84-6 |
| Br_2Mg^+ | MgBr_2 | 10.47 | 169 173 | 708 723 | -72 -69±4 | -302 -287±15 | 82TPIS | 7789-48-2 |
| IP is onset of photoelectron band (79LEE/POT2). | | | | | | | | |
| Br_2OS^+ | SOBr_2 | (10.1) | (204) (209) | (851) (872) | -29 -24 | -123 -102 | 82TN270 | 507-16-4 |
| IP is onset of photoelectron band. | | | | | | | | |
| Br_2Pb^+ | PbBr_2 | 9.6 | (196) | (822) | -25±1 | -104±6 | 75JANAF | 10031-22-8 |
| IP is onset of photoelectron band (84NOV/POT2, 82LEV/LIA). | | | | | | | | |
| Br_2S_2^+ | S_2Br_2 | (9.23±0.03) | (221) | (923) | 8 | 33 | 82TN270 | 13172-31-1 |
| IP from 81KAU/VAH. | | | | | | | | |
| Br_2Se^+ | SeBr_2 | 9.07 | 204 | 854 | -5 | -21 | 82TN270 | 22987-45-7 |
| Br_2Sn^+ | SnBr_2 | 9.0 | 201 | 839 | -7 | -29 | 82TPIS | 10031-24-0 |
| IP is onset of photoelectron band (84NOV/POT2, 82LEV/LIA). | | | | | | | | |
| Br_2Sr^+ | SrBr_2 | (9.11) | (114) (118) | (477) (492) | -96 -92±3 | -402 -387±11 | 82TPIS | 10476-81-0 |
| IP is onset of photoelectron band (79LEE/POT2). See also: 82EMO/KIE. | | | | | | | | |

Table 1. Positive Ion Table - Continued

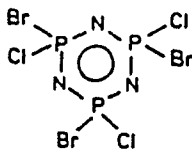
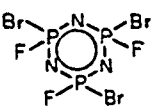
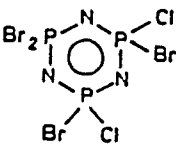
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{Br}_3\text{Cl}_3\text{N}_3\text{P}_3^+$  | (9.72±0.1) | (73) | (306) | -151 | -632 | *EST | 16032-52-3 |
| $\text{Br}_3\text{F}_3\text{N}_3\text{P}_3^+$  | (10.37±0.03) IP from 81CLA/SOW. | (-74) | (-311) | -314 | -1312 | *EST | 67336-18-9 |
| Br_3Ga^+ GaBr ₃ | 10.40 | 170 | 710 | -70 | -293 | 82TN270 | 13450-88-9 |
| Br_3In^+ InBr ₃ | (10.0) IP is onset of photoelectron band. | (163) | (683) | -67 | -282 | 82TN270 | 13465-09-3 |
| Br_3La^+ LaBr ₃ | (9.85) IP is onset of photoelectron band (83RUS/GOO). | (87) | (364) | -140±2 | -586±7 | 78TPIS | 13536-79-3 |
| Br_3OP^+ POBr ₃ | 10.75±0.02 | 151 161 | 632 673 | -97 -87 | -405 -364 | 71JANAF | 7789-59-5 |
| Br_3P^+ PBr ₃ | 9.7 IP is onset of photoelectron band. | (190) (196) | (797) (821) | -33 -27 | -139 -115 | 82TN270 | 7789-60-8 |
| $\text{Br}_4\text{Cl}_2\text{N}_3\text{P}_3^+$  | (9.60±0.1) | (82) | (343) | -139 | -583 | *EST | 15965-00-1 |
| Br_4Hf^+ HfBr ₄ | (10.9) IP is onset of photoelectron band. | (87) | (365) | -164 | -687 | 81SPE | 13777-22-5 |
| Br_4Sn^+ SnBr ₄ | 10.6 IP is onset of photoelectron band. | 169 177 | 708 739 | -75 -68 | -315 -284 | 82TN270 | 7789-67-5 |

Table 1. Positive Ion Table - Continued

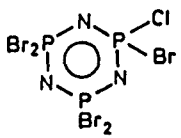
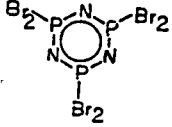
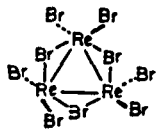
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Br_4Ti^+ TiBr_4 | 10.3 IP is onset of photoelectron band. | (90) | (376) | -148±1 | -618±5 | 71JANAF | 7789-68-6 |
| Br_4Zr^+ ZrBr_4 | (10.7) IP is onset of photoelectron band. | (93) | (387) | -154±2 | -645±8 | 78JANAF | 13777-25-8 |
| $\text{Br}_5\text{ClN}_3\text{P}_3^+$  | (9.47±0.1) | (91) | (380) | -128 | -534 | *EST | 15608-37-4 |
| Br_5W^+ WBr_5 | (8.3±0.2) | (144) (153) | (602) (638) | -48±5 -39 | -199±21 -163 | 71JANAF | 13470-11-6 |
| $\text{Br}_6\text{N}_3\text{P}_3^+$  | 9.62±0.03 IP from 81CLA/SOW. | (82) | (343) | -140 | -585 | *EST | 13701-85-4 |
| Br_9Re_3^+  | (8.4) IP is onset of photoelectron band. | (125) | (521) | -69 | -289 | 82TN270 | 33517-16-7 |
| C^+ C | 11.260 | 431.0 429.7 | 1803.2 1797.6 | 171.3 170.0 | 716.7 711.2 | 82TN270 | 7440-44-0 |
| CBr^+ CBr | (10.43±0.02) | (362.5) (363.4) | (1516.7) (1520.6) | 122.0±15 122.9±15 | 510.4±63 514.3±63 | 85JANAF | |
| CBrClF_2^+ CF_2BrCl | (≤11.83) | (≤168) | (≤703) | -105±2 | -438±8 | 78KUD/KUD | 353-59-3 |
| CBrCl_3^+ CCl_3Br | (10.6) IP is onset of photoelectron band (81NOV/CVI3). | (234) | (980) | -10.2±0.6 | -42.7±2.4 | 77PED/RYL | 75-62-7 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|------------------|------------------------------|-------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CBrF_3^+ CF_3Br | 11.4 | 108 111 | 450 463 | -155 -152 | -650 -637 | 78KUD/KUD | 75-63-8 |
| IP is onset of photoelectron band. See also: 82BOC/WIT. | | | | | | | |
| CBrN^+ BrCN | 11.84±0.01 | 316 | 1323 | 43±1 | 181±4 | 77PED/RYL | 506-68-3 |
| $\text{CBr}_2\text{Cl}_2^+$ CCl_2Br_2 | (10.4) | (242) | (1012) | 2±2 | 9±8 | 78KUD/KUD | 594-18-3 |
| IP is onset of photoelectron band. | | | | | | | |
| CBr_2F_2^+ CF_2Br_2 | 11.07±0.03 | 165 | 689 | -91±2 | -379±8 | 78KUD/KUD | 75-61-6 |
| CBr_2O^+ COBr_2 | (10.8) | (222) | (929) | -27±0.5 | -113±2 | 77PED/RYL | 593-95-3 |
| IP is onset of photoelectron band. | | | | | | | |
| CBr_3^+ CBr_3 | (8.2) | (239) | (1000) | 49.6 | 207.5 | *EST | |
| From appearance potential (10.47±0.02 eV) in CBr_4 ; IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| CBr_3F^+ CFBr_3 | 10.67±0.01 | 190 | 793 | -56±2 | -236±8 | 78KUD/KUD | 353-54-8 |
| CBr_4^+ CBr_4 | (10.31±0.02) | (258) (265) | (1079) (1109) | 20.1±0.8 27.2 | 83.9±3.4 113.8 | 84BIC/MIN | 558-13-4 |
| CCe^+ CCe | (7.5±1.0) | (336) | (1406) | 163 | 682 | 82TN270 | 12011-58-4 |
| CCl^+ CCl | (8.9±0.2) | (297) | (1243) | (92) | (384) | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination. $\Delta_f H(\text{Neutral})$ is $\Delta_f H(\text{Ion}) - \text{IP}$. IP from 82HEP/TRE. | | | | | | | |
| CClF^+ CClF | (10.7) | (243) | (1017) | -5±7 | -20±29 | 85LIA/KAR | 1691-88-9 |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See: 85LIA/KAR (re-evaluated here). | | | | | | | |
| CClF_2^+ CClF_2 | (8.3) | 126 | 528 | -66 | -275 | *EST | 1691-89-0 |
| Cited heat of formation based on observation of near-thermoneutral reaction: ($\text{C}_2\text{H}_5^+ + \text{CF}_2\text{Cl}_2 \rightarrow \text{CF}_2\text{Cl}^+ + \text{C}_2\text{H}_5\text{Cl}$). Value based on appearance potential of ion (11.99 eV) in CF_2Cl_2 is 133 kcal/mol, 556 kJ/mol. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|----------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CClF_3^+ CF_3Cl | 12.39 | 116 117 | 485 491 | -169.7±0.6 -168 | -710.0±2.3 -704 | 77PED/RYL | 75-72-9 |
| | | See also: 85KIS/MOR. | | | | | |
| CCIN^+ ClCN | 12.34±0.01 | 318 317 | 1329 1328 | 33.0 32.8 | 138.0 137.3 | 77PED/RYL | 506-77-4 |
| | | Cited ionization potential corresponds to the formation of $\text{CNCl}^+ (^2\Pi_{3/2})$. Formation of $\text{CNCl}^+ (^2\Pi_{1/2})$ requires 12.37 eV. | | | | | |
| CCINO^+ ClNCO | (10.72±0.01) | (253) | (1057) | 5.5 | 23.0 | 83DEW/RZE | 13858-09-8 |
| CCl_2^+ CCl_2 | 10.36 | 278 | 1163 | 39 | 163 | 85LIA/KAR | 1605-72-7 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations. Cited IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See 85LIA/KAR. | | | | | |
| CCl_2F^+ CCl_2F | (8.0) | (168) | (703) | -17.5 | -73 | *EST | 1691-90-3 |
| | | From observation of near-thermoneutral reaction: $(\text{C}_2\text{H}_5^+ + \text{CF}_2\text{Cl}_2 \rightarrow \text{CCl}_2\text{F}^+ + \text{C}_2\text{H}_5\text{F})$ (77LIA/AUS). Appearance potential determinations lead to values of 175 kcal/mol, 732 kJ/mol, for the heat of formation of this ion. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | |
| CCl_2F_2^+ CF_2Cl_2 | 11.75±0.04 | 157 158 | 656 661 | -114.1±1.3 -113 | -477.5±5.6 -473 | 77PED/RYL | 75-71-8 |
| | | See also: 85KIS/MOR. | | | | | |
| CCl_2O^+ COCl_2 | (11.4) | (210) (211) | (880) (882) | -53 -52 | -220 -218 | 82BAU/COX | 75-44-5 |
| | | IP is onset of photoelectron band. | | | | | |
| CCl_2S^+ CSCl_2 | 9.61±0.02 | 215 | 900 | -6 | -27 | 79JOS | 463-71-8 |
| CCl_3^+ CCl_3 | (7.8) | (199) | (831) | 19 | 79 | 82MCM/GOL | 3170-80-7 |
| | | $\Delta_f H(\text{Ion})$ is based on the observation of the reaction $(\text{H}_3\text{O}^+ + \text{CFCl}_3 \rightarrow \text{CCl}_3^+ + \text{HF} + \text{H}_2\text{O})$ and lack of occurrence of $(\text{sec-C}_3\text{H}_7^+ + \text{CFCl}_3 \rightarrow \text{CCl}_3^+ + \text{C}_3\text{H}_7\text{F})$ which brackets the heat of formation between 197 and 200 kcal/mol, 824 and 837 kJ/mol (77LIA/AUS). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. Experimental value: 8.28 eV. | | | | | |
| CCl_3F^+ CFCl_3 | 11.77±0.02 | 207 208 | 868 871 | -64±2 -63 | -268±8 -265 | 77PED/RYL | 75-69-4 |
| | | See also: 85KIS/MOR. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CCl_4^+ CCl_4 | 11.47±0.01 | 241 | 1010 | -23.2±0.7 | -97.1±3 | 77PED/RYL | 56-23-5 |
| | | 242 | 1012 | -22.7 | -95.0 | | |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |
| CCo^+ CCo | | (364) | (1524) | | | | |
| | | $\Delta_f H(\text{Ion})$ from photodissociation onset to give Co^+ (86HET/FRE). | | | | | |
| CF^+ CF | 9.11±0.01 | 271.1 | 1134.2 | 61.0±2 | 255.2±8 | 85JANAF | 3889-75-6 |
| | | 270.2 | 1130.6 | 60.1±2 | 231.6±8 | | |
| IP from 84DYK/LEW. See also: 82HEP/TRE. | | | | | | | |
| FCN^+ FCN | 13.32±0.01 | 316 | 1321 | 9±4 | 36±17 | 71JANAF | 1495-50-7 |
| FCO^+ FCO | 8.76±0.32 | (160) | (669) | -42±4 | -175±16 | 81DYK/JON2 | |
| IP from 81DYK/JON2. | | | | | | | |
| CF_2^+ CF_2 | 11.42±0.01 | 214 | 897 | -49±3 | -205±12 | 85LIA/KAR | 2154-59-8 |
| CF_2O^+ COF_2 | 13.03 | 147 | 617 | -153 | -640 | 77PED/RYL | 353-50-4 |
| | | 148 | 620 | -152 | -637 | | |
| CF_2S^+ CSF_2 | (10.45±0.01) | (157) | (658) | -84 | -350 | 79JOS | 420-32-6 |
| See also: 85BIN/GRO. | | | | | | | |
| CF_2Se^+ CSeF_2 | (9.6±0.2) | (154) | (646) | -67 | -280 | *EST | 54393-39-4 |
| IP from 85BIN/GRO, 84BOC/AYG. | | | | | | | |
| CF_3^+ CF_3 | (≤8.9) | (95.4) | (399.0) | -110 | -460 | 86TSA | 2264-21-3 |
| | | (96.1) | (402.0) | -109 | -457 | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations (82BOM/DAN). See also: 81BER/BEA, 83WAN/LER. IP estimated in 81LOG/TAK. | | | | | | | |
| CF_3I^+ CF_3I | 10.23 | 95 | 397 | -141±5 | -590±21 | 78KUD/KUD | 2314-97-8 |
| See also: 81BER/BEA, 84BAN/YAT. | | | | | | | |
| CF_3NO^+ CF_3NO | (10.5±0.1) | (116) | (484) | -126 | -529 | *EST | 334-99-6 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--------------------------|----------------------------|--------------------------|--------|------------------------------|------------|------------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CF_4^+ | CF_4 | | | | -223.4±0.1 | -934.5±0.4 | 77PED/RYL | 75-73-0 |
| | | | | | -221.6 | -927 | | |
| The stable region of the CF_4^+ ground state is not accessible by a vertical transition from the CF_4 molecule; no CF_4^+ ions have been experimentally observed. The onset of the photoelectron spectrum is at -15.3 eV (81BIE/ASB, 84CAR/PAH, 85NOV/POT). A value of < 14.7 eV was suggested in 77ROS/DRA. See also: 75LLO/ROB, 85KIS/MOR. | | | | | | | | |
| CF_4O^+ | CF_3OF | (13.0) | (112) | (469) | -188 | -785 | 69STU/WES | 373-91-1 |
| IP is onset of photoelectron band. | | | | | | | | |
| CF_5N^+ | CF_3NF_2 | (11.9) | (105) | (440) | -169±0.5 | -708±2 | 77PED/RYL | 335-01-3 |
| IP from 82BUR/PAW. | | | | | | | | |
| CFe^+ | FeC | | (358) | (1499) | | | | |
| $\Delta_f H(\text{Ion})$ from photodissociation onset to give Fe^+ (86HET/FRE). | | | | | | | | |
| CGe^+ | GeC | (10.3±0.3) | (388) | (1622) | 150 | 628 | 79HUB/HER | 12334-26-8 |
| 0 K values. | | | | | | | | |
| CH^+ | CH | 10.64±0.01 | 387.8 | 1622.4 | 142.4 | 595.8 | 79HUB/HER | 3315-37-5 |
| | | | 387.0 | 1619.1 | 141.6 | 592.5 | | |
| See also: 83PLE/MAR. | | | | | | | | |
| CHBrCl_2^+ | CHBrCl_2 | 10.6 | 233 | 974 | -12 | -49 | 78KUD/KUD | 75-27-4 |
| IP is onset of photoelectron band (81NOV/CVI3). | | | | | | | | |
| CHBrF_3^+ | CF_3BrH | | 73 | 305 | | | | |
| From proton affinity of CBrF_3 (RN 75-63-8) (85MCM/KEB) re-evaluated relative to CO standard (84LIA/LIE). PA = 137.5 kcal/mol, 575.3 kJ/mol. | | | | | | | | |
| CHBrN^+ | BrCNH | | 231 | 965 | | | | |
| From proton affinity of BrCN (RN 506-68-3). PA = 178.3 kcal/mol, 746 kJ/mol. | | | | | | | | |
| CHBr_2^+ | CHBr_2 | (7.4) | (224) | (936) | 54 | 227 | 82MCM/GOL [✓] | 14362-13-1 |
| Ion heat of formation from appearance potential (10.70±0.02 eV) in CHBr_3 . Cited ionization potential is difference between this heat of formation and that of neutral. Experimental determinations of this ionization potential gave values of 8.13±0.16 eV (77ROS/DRA), 8.41±0.03 eV (V) (84AND/DYK3). | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CHBr ₂ Cl ⁺ CHClBr ₂ | 10.59±0.01 | 246 | 1031 | 2±2 | 9±8 | 78KUD/KUD | 124-48-1 |
| IP (77ROS/DRA) in good agreement with onset of photoelectron band (81NOV/CVI3). | | | | | | | |
| CHBr ₃ ⁺ CHBr ₃ | 10.48±0.02 | 247.4 | 1035.0 | 5.7±1.1 | 23.8±4.5 | 84BIC/MIN | 75-25-2 |
| See also: 82VON/ASB. | | | | | | | |
| CHCl ⁺ CHCl | 9.84 | 298 | 1247 | 71 | 297 | 85LIA/KAR | 2108-20-5 |
| $\Delta_f H(\text{Ion})$ derived from hydrogen affinity considerations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| CHClF ⁺ CClFH | (8.81±0.02) | (178) | (743) | (-25) | (-105) | | 33272-71-8 |
| $\Delta_f H(\text{Ion})$ from observation of: (CF ₂ Cl ⁺ + CHFCl ₂ → CHFCl ⁺ + CF ₂ Cl ₂) and non-observation of: (C ₂ H ₅ ⁺ + CHFCl ₂ → CHFCl ⁺ + C ₂ H ₅ Cl)(77LIA/AUS). Appearance potential determinations lead to a value of 205 kcal/mol, 858 kJ/mol. IP from 84AND/DYK. | | | | | | | |
| CHClF ₂ ⁺ CHF ₂ Cl | (12.2) | (166) | (694) | -115.6±0.5 | -483.5±2.2 | 77PED/RYL | 75-45-6 |
| See also: 81NOV/CVI3. | | | | | | | |
| CHClF ₃ ⁺ CF ₃ ClH | | 60 | 251 | | | | |
| From proton affinity of CF ₃ Cl(RN 75-72-9) (85MCM/KEB) re-evaluated relative to CO standard(84LIA/LIE). PA = 136 kcal/mol, 569 kJ/mol. | | | | | | | |
| CHClN ⁺ ClCNH | | 224 | 937 | | | | |
| From proton affinity of ClCN (RN 506-77-4). PA = 174.8 kcal/mol, 731 kJ/mol (86MAR/TOP). | | | | | | | |
| CHCl ₂ ⁺ CHCl ₂ | (8.1) | (212) | (887) | 26±1 | 108±4 | 83WEI/BEN | 3474-12-2 |
| $\Delta_f H(\text{Ion})$ from appearance potential (11.49±0.02 eV) in CCl ₃ H. Cited IP is difference between heats of formation of ion and neutral. An experimental determination of the IP gave a value of 8.32 eV(84AND/DYK) which would correspond to a $\Delta_f H(\text{Radical})$ of 20 kcal/mol, 84 kJ/mol. | | | | | | | |
| CHCl ₂ F ⁺ CHFCl ₂ | (11.5) | (198) | (829) | -67±2 | -281±8 | 78KUD/KUD | 75-43-4 |
| IP is onset of photoelectron band (82LEV/LIA, 81NOV/CVI3). | | | | | | | |
| CHCl ₃ ⁺ CHCl ₃ | 11.37±0.02 | 237 | 992 | -25.0±0.5 | -104.8±2 | 77PED/RYL | 67-66-3 |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------------|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CHCo ⁺ | CHCo | | (325) | (1361) | | | | |
| $\Delta_f H(\text{Ion})$ from photodissociation onset to give Co ⁺ (86HET/FRE). | | | | | | | | |
| CHF ⁺ | CHF | (10.49) | (268) | (1121) | 26±3 | 109±12 | 85LIA/KAR | 13453-52-6 |
| $\Delta_f H(\text{Ion})$ from hydrogen affinity considerations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$ (85LIA/KAR). | | | | | | | | |
| CHFN ⁺ | FCNH | | (224) | (934) | | | | |
| $\Delta_f H(\text{Ion})$ from core binding energies of isoelectronic neutral HNCO (84BEA/EYE). | | | | | | | | |
| CHFO ⁺ | HFCO | (12.37±0.02) | (195) | (817) | -90 | -377 | 71JANAF | 1493-02-3 |
| CHF ₂ ⁺ | CHF ₂ | (8.78) | (146) | (611) | -57±1 | -237±5 | 83PIC/ROD | 2670-13-5 |
| Heat of formation of ion derived from observed ion-molecule reactions (74BLI/MCM, 77LIA/AUS); cited ionization potential is the difference between the heats of formation of the ion and the radical. | | | | | | | | |
| CHF ₂ O ⁺ | F ₂ COH | | 52 | 219 | | | | |
| From proton affinity of CF ₂ O (RN 353-50-4). PA = 160.5 kcal/mol, 671.5 kJ/mol. | | | | | | | | |
| CHF ₃ ⁺ | CHF ₃ | 13.86 | 154 | 642 | -166±2 | -695±8 | 78KUD/KUD | 75-46-7 |
| | | | 156 | 649 | -164 | -688 | | |
| See also: 81BIE/ASB, 85NOV/POT, 82BOC/WIT. | | | | | | | | |
| CHF ₃ I ⁺ | CF ₃ IH | | (78) | (326) | | | | |
| From proton affinity of CF ₃ I (RN 2314-97-8) (85MCM/KEB) re-evaluated relative to CO standard (84LIA/LIE). PA = 146.7 kcal/mol, 614 kJ/mol. | | | | | | | | |
| CHF ₃ NO ⁺ | CF ₃ NHO | | (70) | (294) | | | | |
| From the proton affinity of CF ₃ NO (RN 334-99-6). PA = 70. kcal/mol, 294. kJ/mol. | | | | | | | | |
| CHF ₄ ⁺ | F ₃ CFH | | 17 | 70 | | | | |
| From proton affinity of CF ₄ (RN 75-73-0). PA = -126 kcal/mol, -527 kJ/mol. | | | | | | | | |
| CHF ₄ N ⁺ | CHF ₂ NF ₂ | (11.5) | (156) | (655) | -109 | -455 | *EST | 24708-53-0 |
| IP from 82BUR/PAW. | | | | | | | | |
| CHFe ⁺ | CHFe | | (322) | (1349) | | | | |
| $\Delta_f H(\text{Ion})$ from photodissociation onset to give Fe ⁺ (86HET/FRE). | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---------------------------------|--------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CHI₃⁺ CHI ₃ | 9.25±0.02 | 241 244 | 1010 1019 | 28±5 30 | 118±21 127 | 78KUD/KUD | 75-47-8 |
| CHN⁺ HCN | 13.60±0.01 | 346 346 | 1447 1448 | 32.3 32.4 | 135.1 135.5 | 82TN270 | 74-90-8 |
| | | See also: 82KRE/SCH, 81KIM/KAT. | | | | | |
| HNC | (12.5±0.1) | (336) | (1407) | 48±2 | 201±8 | 82PAU/HEH | 6914-07-4 |
| | IP by charge exchange bracketing of HNC ⁺ ions generated in CH ₃ NC(78BIE/JON). See also: 80MCL/MCG. | | | | | | |
| CHNO⁺ HNCO | 11.61±0.03 | 243 | 1015 | -25±3 | -105±13 | 86SPI/PER | 75-13-8 |
| HCNO | (10.83) | (302) | (1263) | 52 | 218 | *EST | 506-85-4 |
| CHNS⁺ HNCS | 9.94±0.02 | 260 | 1087 | 31 | 128 | 82TN270 | 3129-90-6 |
| CHO⁺ HCO | 8.10±0.05 | 197.3 | 825.6 | 10.7 | 44.8 | 77BEC/LIP | 17030-74-9 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (85TRA2). See also: 76GUY/CHU, 84WAN/CAP, 80DYK/JON. | | | | | | |
| COH | | (230) | (963) | | | | |
| | $\Delta_f H(\text{Ion})$ from correlation with oxygen 1s binding energy (85MCM/KEB2). See also: 85WAG/KEM, 83BUR/MOM. | | | | | | |
| CHOS⁺ COSH | | 181 | 757 | | | | |
| | From proton affinity of COS (RN 463-58-1) (85MCM/KEB, 85MCM/KEB2) re-evaluated relative to CO standard (84LIA/LIE). PA = 150.7 kcal/mol, 631 kJ/mol. | | | | | | |
| CHOSe⁺ COSeH | | 230 | 962 | | | | |
| | From proton affinity of COSe (RN 1603-84-5) (85KAR). PA = 152. kcal/mol, 637. kJ/mol. | | | | | | |
| CHO₂⁺ COOH | | 141 | 589 | | | | 2564-86-5 |
| | $\Delta_f H(\text{Ion})$ from appearance potential in HCOOH. | | | | | | |
| CHP⁺ HCP | (10.79±0.01) | (289) | (1208) | 40±15 | 167±63 | 71JANAF | 6829-52-3 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------------------------------|----------------------------|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CHS⁺ | HCS | >(7.3) | 243 | 1018 | ≤73 | ≤305 | 83BUT/BAE | |
| | | | 243 | 1018 | ≤74 | ≤310 | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations in thiirane (RN 420-12-2) (82BUT/BAE) in good agreement with value derived from proton affinity of CS (RN 2944-05-0) (85SMI/ADA). PA = 188.2 kcal/mol, 787 kJ/mol. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See also: 82KUT/EDW, 82KUT/EDW. | | | | | | | | |
| CHS₂⁺ | HSCS | | 229 | 959 | | | | |
| From proton affinity of CS ₂ (RN 75-15-0), re-evaluated. PA = 164.4 kcal/mol, 688. kJ/mol. See also: 85MCM/KBB, 85WEI/PLA. | | | | | | | | |
| CHTi⁺ | TiCH | | (289) | (1209) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86BLK/ARM). 0 K value. | | | | | | | | |
| CHV⁺ | VCH | | (307) | (1283) | | | | |
| $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM, 86ARI/ARM). 0 K value. | | | | | | | | |
| CH₂⁺ | CH ₂ | 10.396±.003 | 331 | 1386 | 93 | 390 | 82TN270 | 60528-76-9 |
| | | | 331 | 1386 | 93 | 390 | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination (83PLE/MAR). | | | | | | | | |
| CH₂Br⁺ | CH ₂ Br | (7.9) | (224) | (937) | 42 | 174 | 82MCM/GOL | 16519-97-4 |
| Heat of formation of ion from appearance potential (11.35±0.02) in CH ₂ Br ₂ . Cited ionization potential is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. An experimental value of 8.61±0.01 eV has been reported for the ionization potential (84AND/DYK3). | | | | | | | | |
| CH₂BrCl⁺ | CH ₂ ClBr | 10.77±0.01 | 259 | 1084 | 11±2 | 45±8 | 78KUD/KUD | 74-97-5 |
| IP from 77ROS/DRA, 81NOV/CVI3. | | | | | | | | |
| CH₂Br₂⁺ | CH ₂ Br ₂ | 10.50±0.02 | 242 | 1013 | 0±1 | 0±4 | EST | 74-95-3 |
| See also: 82VON/ASB. | | | | | | | | |
| CH₂Cl⁺ | CH ₂ Cl | (8.6) | (229.2) | (959.0) | 31 | 130 | 83WEI/BEN | 6806-86-6 |
| | | | (229.9) | (962.1) | | | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations. Cited ionization potential is difference in heats of formation of ion and radical; an experimental determination of the ionization potential gives 8.75±0.01 eV (84AND/DYK) which would correspond to a radical heat of formation of 27 kcal/mol, 115 kJ/mol. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------------|---|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_2ClF^+ CH_2FCl | 11.71±0.01 IP from 84AND/DYK. | 208 | 869 | -62±2 | -261±8 | 78KUD/KUD2 | 593-70-4 |
| CH_2Cl_2^+ CH_2Cl_2 | 11.32±0.01 See also: 82VON/ASB, 81KIM/KAT. | 238 240 | 997 1003 | -22.9±0.2 -21.2 | -95.7±0.8 -88.8 | 77PED/RYL | 75-09-2 |
| $\text{CH}_2\text{Cl}_4\text{Si}^+$ $\text{Cl}_3\text{SiCH}_2\text{Cl}$ | (10.7) IP is onset of photoelectron band (81ZYK/KHV). | (116) | (486) | -130 | -546 | *EST | 1558-25-4 |
| CH_2Co^+ $\text{CH}_2 = \text{Co}$ | | (290) | (1213) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction and photodissociation (81ARM/HAL, 81ARM/BEA2, 86HET/FRE). 0 K values. | | | |
| CH_2Cr^+ $\text{CH}_2 = \text{Cr}$ | | (292) | (1223) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARI). See also: 81ARM/HAL, 81HAL/ARM. 0 K values. | | | |
| CH_2F^+ CH_2F | 9.05±0.01 IP from 84AND/DYK. $\Delta_f H(\text{Ion})$ evaluated from observed ion-molecule reactions (77LIA/AUS). | 199 | 833 | -8±2 | -33±8 | 82MCM/GOL | 3744-29-4 |
| CH_2F_2^+ CH_2F_2 | 12.71 See also: 81BIE/ASB. | 185 | 773 | -108±2 | -453±8 | 78KUD/KUD | 75-10-5 |
| CH_2F_3^+ F_2CHFH | | 53 | 220 | From proton affinity of CF_3H (RN 75-46-7). PA = 147 kcal/mol, 615 kJ/mol. | | | |
| $\text{CH}_2\text{F}_3\text{O}_3\text{S}^+$ $\text{CF}_3\text{SO}_3\text{H}_2$ | | (-85) | (-356) | From proton affinity of $\text{CF}_3\text{SO}_3\text{H}$ (RN 1493-13-6). PA = (169) kcal/mol, (707) kJ/mol. | | | |
| CH_2Fe^+ $\text{CH}_2 = \text{Fe}$ | | (292) | (1222) | $\Delta_f H(\text{Ion})$ from photodissociation onset to give Fe^+ (86HET/FRE). See also: 81ARM/HAL, 84JAC/JAC. | | | |
| CH_2I_2^+ CH_2I_2 | 9.46±0.02 | 246 249 | 1031 1040 | 28±5 30 | 118±21 127 | 78KUD/KUD | 75-11-6 |
| CH_2Mn^+ $\text{CH}_2 = \text{Mn}$ | | (237) | (992) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (81ARM/HAL). 0 K values. | | | 65127-77-7 |

Table 1. Positive Ion Table - Continued


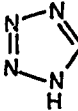
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_2N^+ | | | | | | | |
| HCNH | | 226 | 947 | | | | |
| | | From proton affinity of HCN (RN 74-90-8) (PA = 171 kcal/mol, 717 kJ/mol) and HNC (RN 6914-07-4) (PA = 190 kcal/mol, 796 kJ/mol). | | | | | |
| CNH ₂ | | (265) | (1109) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (84BUR/HOL). | | | | | |
| CH_2NO^+ | | | | | | | |
| H ₂ NCO | | 167 | 700 | | | | |
| | | From proton affinity of HNCO (RN 75-13-8) (PA = 173 kcal/mol, 725 kJ/mol). | | | | | |
| CH_2N_2^+ | | | | | | | |
| CH ₂ N ₂ | 8.999±0.001 | 263 | 1098 | 55±4 | 230±17 | 78VOG/WIL | 334-88-3 |
| H ₂ NCN | (10.4) | (272) | (1137) | 32 | 134 | 77PED/RYL | 420-04-2 |
| | | IP is onset of photoelectron band. | | | | | |
|  | (10.3) | (301) | (1259) | 63.3±2.7 | 264.8±11 | 72LAU/OKA | 157-22-2 |
| CH_2N_4^+ | | | | | | | |
|  | (10.95) | (333) | (1392) | 80±1 | 335±4 | 77PED/RYL | 288-94-8 |
| | | IP is onset of photoelectron band (82LEV/LIA, 81PAL/SIM). | | | | | |
| CH_2Ni^+ | | | | | | | |
| CH ₂ =Ni | | (285) | (1193) | | | | 60187-22-6 |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (81ARM/HAL). 0 K values. | | | | | |
| CH_2O^+ | | | | | | | |
| CH ₂ O | 10.874±0.002 | <u>224.8</u> | <u>940.5</u> | -26.0±0.2 | -108.7±0.7 | 77PED/RYL | 50-00-0 |
| | | <u>225.8</u> | <u>944.5</u> | -25.0 | -104.7 | | |
| | | See also: 81BOM/DAN, 76GUY/CHU, 80VON/BIE, 84WAN/CAP, 81KIM/KAT. | | | | | |
| HCOH | | 230 | 962 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential measurement (83BUR/MOM). | | | | | |
| CH_2O_2^+ | | | | | | | |
| HCOOH | 11.33±0.01 | 170.7 | 714.3 | -90.5±0.1 | -378.8±0.5 | 78CHA/ZWO | 64-18-6 |
| | | See also: 80VON/BIE, 81KIM/KAT. | | | | | |
| C(OH) ₂ | | 175 | 732 | | | | 71946-83-3 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (82BUR/HOL, 83BUR/MOM). | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|-------------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_2S^+ CH_2S | 9.34±0.01 | 240 | 1006 | 25 | 105 | 82ROY/MCM | 865-36-1 |
| | See also: 83BERM/AKO, 82KUT/EDW. | | | | | | |
| HCSH | | (270) | (1130) | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential determination (82KUT/EDW). 0 K values. | | | | | | |
| CH_2Se^+ CH_2Se | (8.95) | (245) | (1024) | 38 | 160 | *EST | 6596-50-5 |
| | IP from 84BOC/AYG. | | | | | | |
| CH_2Ti^+ $\text{CH}_2=\text{Ti}$ | | (277) | (1158) | | | | |
| | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM). 0 K value. | | | | | | |
| CH_2V^+ $\text{CH}_2=\text{V}$ | | (295) | (1234) | | | | |
| | $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM, 86ARI/ARM). 0 K value. | | | | | | |
| CH_3^+ CH_3 | 9.84±0.01 | 261.3±0.4 | 1093.3±1.7 | 34.8±0.3 | 145.8±1 | 81HEN/KNO | 2229-07-4 |
| | | <u>262</u> | <u>1098</u> | 35.6 | 149.0 | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations (81TRA/MCL). See also: 83PLE/MAR. | | | | | | |
| $\text{CH}_3\text{BBr}_2^+$ CH_3BBr_2 | 10.60 | 197 | 824 | -48 | -199 | 82HOL/SMI | 17933-16-3 |
| $\text{CH}_3\text{BCl}_2^+$ CH_3BCl_2 | (11.51) | (185) | (774) | -81 | -337 | 82HOL/SMI | 7318-78-7 |
| CH_3BF_2^+ CH_3BF_2 | (12.54±0.03) | (90) | (377) | -199 | -833 | 82HOL/SMI | 373-64-8 |
| CH_3BO^+ BH_3CO | 11.14±0.02 | 230 | 964 | -27 | -111 | 82TN270 | 13205-44-2 |
| | | 232 | 970 | -25.0 | -104.8 | | |
| CH_3Br^+ CH_3Br | 10.541±0.003 | 234 | 979 | -9.1±0.3 | -38.1±1.3 | 84BIC/MIN | 74-83-9 |
| | | 238 | 994 | -5.5 | -23.0 | | |
| | Cited IP leads to $\text{CH}_3\text{Br}^+ (^2E_{3/2})$; formation of $\text{CH}_3\text{Br}^+ (^2E_{1/2})$ requires 10.857 eV. IPs from 82BAI/CON, 82LEV/LIA, 77ROS/DRA, 82VON/ASB, 81HOL/FIN, 84AND/DYK3, 81KIM/KAT, 77KAR/JAD. | | | | | | |
| CH_2BrH | | (237) | (990) | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential determination (83HOL/LOS2). | | | | | | |
| CH_3BrHg^+ CH_3HgBr | (9.9) | (224) | (937) | -4±0.7 | -18±3 | 77PED/RYL | 506-83-2 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|-----------------------------------|----------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_3Cd^+ | CH_3Cd | | (213) | (891) | | | | |
| From appearance potential (9.69 eV) in $(\text{CH}_3)_2\text{Cd}$. | | | | | | | | |
| CH_3Cl^+ | CH_3Cl | 11.22±0.01 | 239 | 1000 | -19.6±0.1 | -82.0±0.5 | 79KUD/KUD | 74-87-3 |
| | | | 241 | 1009 | -17.5 | -73.4 | | |
| See also: 81KIM/KAT, 77KAR/JAD. | | | | | | | | |
| CH_2CH^+ | | | (246) | (1029) | | | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination (83HOL/LOS2). | | | | | | | | |
| CH_3ClHg^+ | CH_3HgCl | (10.5) | (230) | (962) | -12±0.7 | -51±3 | 77PED/RYL | 115-09-3 |
| IP is onset of photoelectron band (77ROS/DRA, 81BAI/CHI2). | | | | | | | | |
| CH_3ClO^+ | CH_3OCl | (10.39±0.02) | (226) | (944) | -14 | -58 | *EST | 593-78-2 |
| IP from 81COL/FRO. | | | | | | | | |
| $\text{CH}_3\text{ClO}_2\text{S}^+$ | $\text{CH}_3\text{SO}_2\text{Cl}$ | 11.3 | (173) | (722) | -88 | -368 | *EST | 124-63-0 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{CH}_3\text{Cl}_2\text{N}^+$ | CH_3NCl_2 | 9.52 | (264) | (1104) | 44 | 185 | *EST | 7651-91-4 |
| $\text{CH}_3\text{Cl}_2\text{OP}^+$ | CH_3POCl_2 | 10.91 | 119 | 497 | -133±6 | -556±25 | 77PED/RYL | 676-97-1 |
| IP from 80ZVE/VIL, 82LEV/LIA. | | | | | | | | |
| $\text{CH}_3\text{Cl}_2\text{P}^+$ | CH_3PCl_2 | (9.5) | (168) | (703) | -51 | -214 | *EST | 676-83-5 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{CH}_3\text{Cl}_3\text{Si}^+$ | CH_3SiCl_3 | (11.36±0.03) | (131) | (547) | -131 | -549 | 81BEL/PER | 75-79-6 |
| CH_3Co^+ | CH_3Co | (7.0±0.3) | (257) | (1075) | (96) | (400) | 81ARM/BEA | 76826-90-9 |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (81ARM/HAL). IP from 81ARM/BEA. 0 K values. | | | | | | | | |
| CH_3Cr^+ | CH_3Cr | (7.2) | (257) | (1074) | 90 | 375 | 86ELK/ARI | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARI). See also: 81ARM/HAL. 0 K values. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | | |
| CH_3F^+ | CH_3F | 12.47±0.02 | 228 | 956 | -59 | -247 | 85LIA/KAR | 593-53-3 |
| See also: 81BIE/ASB, 81KIM/KAT, 77KAR/JAD. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------------------|----------------------------|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_3F^+ | CH_2FH | | 217 | 908 | | | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination (83HOL/LOS2). | | | | | | | | |
| CH_3F_2^+ | FCH_2FH | | 110 | 462 | | | | |
| From proton affinity of CH_2F_2 (RN 75-10-5). PA = 147 kcal/mol, 615 kJ/mol. | | | | | | | | |
| $\text{CH}_3\text{F}_2\text{P}^+$ | CH_3PF_2 | (9.8) | (68) | (285) | -158 | -661 | *EST | 753-59-3 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{CH}_3\text{F}_2\text{Si}^+$ | CH_3SiF_2 | | 23 | 95 | | | | |
| From appearance potential (11.70±0.03) of ion in $(\text{CH}_3)_2\text{SiF}_2$. | | | | | | | | |
| $\text{CH}_3\text{F}_3\text{Si}^+$ | CH_3SiF_3 | 12.48±0.04 | -8 | -33 | -296 | -1237 | 71JANAF | 373-74-0 |
| CH_3Fe^+ | CH_3Fe | (8.1) | (257) | (1075) | 71 | 298 | 86ELK/ARI | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARI). See also: 81ARM/HAL, 84JAC/JAC. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | | | |
| CH_3Hg^+ | CH_3Hg | | 221 | 926 | | | | |
| | | | 225 | 942 | | | | |
| From appearance potential (10.10±0.02 eV) in $(\text{CH}_3)_2\text{Hg}$. | | | | | | | | |
| CH_3HgI^+ | CH_3HgI | (9.0) | (213) | (891) | 5.3±0.4 | 22.4±1.9 | 77PED/RYL | 143-36-2 |
| IP is onset of photoelectron band. | | | | | | | | |
| CH_3I^+ | CH_3I | 9.538 | 223.6 | 935.7 | 3.7±0.2 | 15.4±0.9 | 77PED/RYL | 74-88-4 |
| | | | 226 | 945 | 6 | 25 | | |
| See: 78LIA/AUS, 83POW, 81KIM/KAT, 77KAR/JAD. | | | | | | | | |
| CH_3Mn^+ | CH_3Mn | | (223) | (934) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ARM). See also: 81ARM/HAL. 0 K values. | | | | | | | | |
| CH_3Mn_2^+ | CH_3Mn_2 | | (261) | (1090) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ARM). 0 K values. | | | | | | | | |
| CH_3N^+ | $\text{CH}_2=\text{NH}$ | (9.9) | (260) | (1090) | 32 | 135 | 78DEF/HEH | 2053-29-4 |
| IP is onset of photoelectron band (82SCH/SCH, 86WER). | | | | | | | | |
| | HCNH_2 | | 258 | 1079 | | | | 35430-17-2 |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations(84BUR/HOL). | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|----------------------------|----------------------------------|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_3NO^+ | | | | | | | | |
| | HCONH ₂ | 10.16±0.06 | 190 | 794 | -44 | -186 | 69BEN/CRU | 75-12-7 |
| | | See also: 81KIM/KAT, 81ASB/SVE, 81HEN/ISA. | | | | | | |
| | CH ₂ =NOH | 10.11 | (240) | (1004) | 7 | 29 | *EST | 75-17-2 |
| | | IP is onset of photoelectron band (82FRO/LAU, 84DOG/POU). | | | | | | |
| | CH ₃ NO | 9.3 | 231 | 967 | 17±0.7 | 70±3 | 73BAT/MIL | 865-40-7 |
| | | IP is onset of photoelectron band (82CHO/FRO, 82FRO/LAU). | | | | | | |
| CH_3NO_2^+ | | | | | | | | |
| | CH ₃ NO ₂ | 11.02±0.04 | 236 | 987 | -17.9±0.2 | -74.8±1.0 | 77PED/RYL | 75-52-5 |
| | | See also: 83GIL/HSI, 83OGD/SHA, 81ALL/MIG, 81ASB/SVE, 81KIM/KAT. | | | | | | |
| | CH ₃ ONO | 10.38±0.03 | 223 | 935 | -15.9±0.2 | -66.5±0.9 | 74BAT/CHR | 624-91-9 |
| | | IP from 83GIL/HSI, 83GIL/HSI2, 80MEI/HSI, 83OGD/SHA. | | | | | | |
| CH_3NO_3^+ | | | | | | | | |
| | CH ₃ ONO ₂ | (11.53±0.01) | (237) | (990) | -29±1 | -122±4 | 77PED/RYL | 598-58-3 |
| CH_3NS^+ | | | | | | | | |
| | HCSNH ₂ | 8.69 | (210) | (877) | 9 | 39 | *EST | 115-08-2 |
| | | See also: 81HEN/ISA. | | | | | | |
| CH_3N_2^+ | | | | | | | | |
| | CH ₃ N ₂ | | 216 | 902 | | | | |
| | | From appearance energy and from proton affinity of CH ₂ N ₂ (RN 334-88-3)(PA = 205 kcal/mol, 858 kJ/mol). | | | | | | |
| | H ₂ NCNH | | (234) | (978) | | | | |
| | | From core binding energy of isoelectronic CH ₃ CN (84BEA/EYE). PA of H ₂ NCN = (164) kcal/mol, (686) kJ/mol. | | | | | | |
| CH_3N_3^+ | | | | | | | | |
| | CH ₃ N ₃ | 9.81±0.02 | 293 | 1227 | 67 | 280 | 69BEN/CRU | 624-90-8 |
| | | See also: 81BOC/DAM. | | | | | | |
| CH_3Ni^+ | | | | | | | | |
| | CH ₃ Ni | | (265) | (1109) | | | | 63583-16-4 |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (81ARM/HAL, 86ELK/ARI). 0 K values. | | | | | | |
| CH_3O^+ | | | | | | | | |
| | CH ₂ OH | 7.56±0.01 | 168 | 703 | -6.2±1.5 | -25.9±6 | 82MCM/GOL | 17691-31-5 |
| | | $\Delta_f H(\text{Ion})$ from proton affinity of formaldehyde. PA = 171.7 kcal/mol, 718 kJ/mol. $\Delta_f H(\text{Ion})$ from appearance potential measurements is 169 kcal/mol, 709 kJ/mol. (82MAC, 83HOL/LOS2, 84LOS/HOL). IP from 84DYK/ELL2. | | | | | | |
| | CH ₃ O | (8.6) | (201) | (842) | 3.7±0.7 | 15.5±2.9 | 74BAT/CHR | 2143-68-2 |
| | | The reaction: HCO ⁺ + H ₂ → CH ₃ O ⁺ is 3.9 kcal/mol, 16.3 kJ/mol, exothermic (77HIR/KEB). A value of 247 kcal/mol, 1034 kJ/mol, has been reported for ³ CH ₃ O ⁺ (84BUR/HOL2) in agreement with 87FER/RON. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_3O_2^+ $\text{HC}(\text{OH})_2$ | | 96 | 403 | | | | |
| | | From proton affinity of HCOOH (RN 64-18-6) and appearance potential determinations (84HOL/LOS). PA = 178.8 kcal/mol, 748 kJ/mol. | | | | | |
| CH_2OOH | | (185) | (774) | | | | |
| | | $\Delta_f H(\text{Ion})$ from 87FER/RON. | | | | | |
| CH_3O_3^+ $\text{C}(\text{OH})_3$ | | 37 | 155 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations(82HOL/LOS2). | | | | | |
| CH_3S^+ CH_2SH | | 206 | 862 | | | | 20879-50-9 |
| | | 208 | 870 | | | | |
| | | Heat of formation of ion from appearance potential determinations (83BUT/BAE, 82LEV/LIA). See also: 83ERM/AKO, 83HOL/LOS2. | | | | | |
| CH_3S | (8.06±0.1) | (215) | (901) | 29.4±2.1 | 123.0±8.8 | 82MCM/GOL | 7175-75-9 |
| | | Collisional activation results (79DIL/MCL) indicate that this structure is a stable triplet; ab initio calculations predict its heat of formation to be -10 kcal/mol above that of CH_2SH^+ , in agreement with the experimentally obtained value given here. | | | | | |
| CH_3S_2^+ CH_3SS | (8.0) | 200 | 835 | 16 | 69 | 86HAW/GRI | |
| | | 201 | 839 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (83BUT/BAE). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | |
| CH_3Sc^+ CH_3Sc | (5.1) | (212) | (887) | 93 | 391 | 86SUN/ARI | |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). See also: 87SUN/ARI. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | |
| CH_3Se^+ CH_2SeH | | 219 | 916 | | | | |
| | | From proton affinity of CH_2Se (RN 6596-50-5)(85KAR). PA = 185 kcal/mol, 774 kJ/mol. | | | | | |
| CH_3Ti^+ CH_3Ti | (6.3) | (248) | (1039) | (102) | (426) | 86ELK/ARI | |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | |
| CH_3V^+ CH_3V | (6.6) | (263) | (1102) | 111 | 463 | 86ARI/ARM | |
| | | $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM, 86ARI/ARM). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_3Xe^+ CH_3Xe | | (210) | (877) | | | | |
| | | $\Delta_f H(\text{Ion})$ derived from results of 86HOV/MCM. | | | | | |
| CH_3Zn^+ CH_3Zn | (7.2) | (213) | (890) | (47) | (197) | | |
| | From appearance potential (10.22 ± 0.02 eV) in $(\text{CH}_3)_2\text{Zn}$. Value from onset of endothermic reaction (86GEO/ARM) is in agreement. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | |
| CH_4^+ CH_4 | 12.51 | 271 | 1132 | -17.8±0.1 | -74.5±0.4 | 77PED/RYL | 74-82-8 |
| | | 272 | 1140 | -16.0 | -66.8 | | |
| | See also: 83PLE/MAR, 81KIM/KAT, 84CHA/HIL. | | | | | | |
| CH_4Br^+ CH_3BrH | | 191 | 800 | | | | |
| | From proton affinity of CH_3Br (RN 74-83-9). PA = 165.7 kcal/mol, 693 kJ/mol. | | | | | | |
| CH_4Cl^+ CH_3ClH | | 183 | 767 | | | | |
| | From proton affinity of CH_3Cl (RN 74-87-3). PA = -163 kcal/mol, -682 kJ/mol. | | | | | | |
| CH_4CIN^+ CH_3NHCl | (9.19±0.02) | (230) | (964) | 18 | 77 | *EST | 6154-14-9 |
| $\text{CH}_4\text{Cl}_2\text{Si}^+$ $\text{CH}_3\text{SiHCl}_2$ | (11.47) | (168) | (705) | -96±2 | -402±8 | 81BEL/PER | 20156-50-7 |
| CH_4F^+ CH_3FH | | (162) | (678) | | | | |
| | From proton affinity of CH_3F (RN 593-53-3). PA = 145 kcal/mol, 605 kJ/mol (86MCM/KEB, 85MCM/KEB3). | | | | | | |
| CH_4I^+ CH_3IH | | (198) | (830) | | | | |
| | From proton affinity of CH_3I (RN 74-88-4). PA = -171 kcal/mol, -715 kJ/mol. | | | | | | |
| CH_4N^+ CH_2NH_2 | 6.1 | (178) | (745) | 38±2 | 159±8 | 81GRI/LOS | 54088-53-8 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations (81LOS/LAM). See also: 81GRI/LOS, 84LOS/HOL, 82MAC, 83BUR/CAS. | | | | | | |
| CH_3NH | (6.7) | (199) | (833) | 43.6±3.0 | 182.4±12.5 | 78SEN/FRA | 49784-84-1 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations (84LOS/HOL). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |
| CH_4NO^+ HC(OH)NH_2 | | 123 | 514 | | | | |
| | From proton affinity of HCONH_2 (RN 75-12-7). PA = 198.4 kcal/mol, 830 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------------|--------------------------------------|----------------------------|--|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_4NO_2^+ | CH_3NOOH | | 169 | 705 | | | | |
| | | | From proton affinity of CH_3NO_2 (RN 75-52-5). PA = 179.2 kcal/mol, 750 kJ/mol. | | | | | |
| | CH_3ONHO | | 157 | 658 | | | | |
| | | | From proton affinity of CH_3ONO (RN 624-91-9). PA = 192.5 kcal/mol, 805 kJ/mol. | | | | | |
| CH_4N_2^+ | (E)- $\text{CH}_3\text{N}=\text{NH}$ | 8.8±0.1 | (248) | (1037) | 45±2 | 188±8 | *EST | 26981-93-1 |
| $\text{CH}_4\text{N}_2\text{O}^+$ | $(\text{NH}_2)_2\text{CO}$ | 9.7 | 165 | 690 | -58.8±0.5 | -245.9±2.1 | 77PED/RYL | 57-13-6 |
| | | | See also: 82BIE/ASB. | | | | | |
| $\text{CH}_4\text{N}_2\text{S}^+$ | $(\text{NH}_2)_2\text{CS}$ | 7.9 | 188 | 785 | 5±0.5 | 23±2 | 82TOR/SAB | 62-56-6 |
| CH_4O^+ | CH_3OH | 10.85±0.01 | 202.0 | 845.3 | -48.2±0.1 | -201.6±0.2 | 77PED/RYL | 67-56-1 |
| | | | 204.6 | 856.2 | -45.6 | -190.7 | | |
| | | | See also: 82MIS/POK, 80VON/BIE, 82ALL/MIG, 84BOW/MAC, 81KIM/KAT, 80BAC/MOU, 77KAR/JAD. | | | | | |
| | CH_2OH_2 | | 195±2 | 815±8 | | | | 25765-84-8 |
| | | | $\Delta_f H(\text{Ion})$ from appearance potential measurements (82HOL/LOS). | | | | | |
| CH_4S^+ | CH_3SH | 9.44±0.005 | 212.3 | 888.2 | -5.5±0.1 | -22.9±0.6 | 77PED/RYL | 74-93-1 |
| | | | 214.8 | 899.0 | -2.9 | -12.1 | | |
| | | | IP from 83BUT/BAE, 81KIM/KAT, 82KUT/EDW. | | | | | |
| | CH_2SH_2 | | 219 | 916 | | | | 63933-47-1 |
| | | | 221 | 925 | | | | |
| | | | $\Delta_f H(\text{Ion})$ from appearance potential determination (83HOL/LOS2). | | | | | |
| CH_4S_2^+ | $\text{CH}_2(\text{SH})_2$ | (9.42) | (225) | (942) | 8±2 | 33±8 | 78BEN | 6725-64-0 |
| CH_4Sc^+ | CH_3ScH | | (214) | (895) | | | | |
| | | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). See also: 86ELK/ARI. | | | | | |
| CH_5^+ | CH_5 | | 216 | 905 | | | | |
| | | | From proton affinity of CH_4 . (RN 74-82-8) See also: 85MCM/KEB. PA = 131.6 kcal/mol, 551. kJ/mol. | | | | | |
| CH_5As^+ | CH_3AsH_2 | (8.5) | (207) | (868) | 11 | 48 | *EST | 593-52-2 |
| | | | IP is onset of photoelectron band (82ELB/DIE). | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|----------------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_5N^+ CH_2NH_3 | | (≤ 201) | (≤ 841) | | | | |
| | | The reaction $\text{c-C}_3\text{H}_6^+ + \text{NH}_3 \rightarrow \text{CH}_2\text{NH}_3^+ + \text{C}_2\text{H}_4$ is at least 15 kcal/mol exothermic (84LIA/BUC). See also: 83HOL/LOS2, 72GRO. | | | | | |
| CH_3NH_2 | 8.97±0.02 | 201 | 842 | -5.5±0.1 | -23.0±0.4 | 77PED/RYL | 74-89-5 |
| | | See also: 81KIM/KAT, 82BIE/ASB, 82ELB/DIE. | | | | | |
| CH_5NO^+ CH_3ONH_2 | 9.55 | (214) | (895) | -6±2 | -26±8 | 69BEN/CRU | 67-62-9 |
| | | IP from 83MOL/PIK. See also: 81KIM/KAT. | | | | | |
| CH_3NHOH | (9.0) | (196) | (818) | -12±2 | -50±8 | 69BEN/CRU | 593-77-1 |
| | | IP is onset of photoelectron band. | | | | | |
| CH_5N_3^+ $(\text{NH}_2)_2\text{C}=\text{NH}$ | (9.10±0.05) | (218) | (910) | 8 | 32 | 82JOS | 113-00-8 |
| CH_5O^+ CH_3OH_2 | | 136 | 567 | | | | |
| | | From proton affinity of CH_3OH (RN 67-56-1). PA = 181.9 kcal/mol, 761 kJ/mol. | | | | | |
| CH_5P^+ CH_3PH_2 | 9.12±0.07 | (206) | (862) | -4 | -18 | *EST | 593-54-4 |
| | | See also: 82COW/KEM, 82ELB/DIE. | | | | | |
| CH_5S^+ CH_3SH_2 | | 173 | 723 | | | | |
| | | From proton affinity of CH_3SH (RN 74-93-1). PA = 187.4 kcal/mol, 784 kJ/mol. | | | | | |
| CH_6N^+ CH_3NH_3 | (4.3±0.1) | (146) | (611) | | | | |
| | | $\Delta_f H(\text{Ion})$ from proton affinity of CH_3NH_2 (RN 74-89-5). PA = 214.1 kcal/mol, 896 kJ/mol. IP estimated from neutralized ion-beam spectroscopy data (85JEO/RAK). | | | | | |
| CH_6N_2^+ CH_3NHNH_2 | 7.67±0.02 | 199 | 835 | 22.6±0.1 | 94.6±0.6 | 77PED/RYL | 60-34-4 |
| | | IP from charge transfer equilibrium constant determinations (84MAU/NEL) is in agreement. See also: 81KIM/KAT. | | | | | |
| CH_6P^+ CH_3PH_3 | | 158 | 658 | | | | |
| | | From proton affinity of CH_3PH_2 (RN 593-54-4). PA = 204.1 kcal/mol, 854 kJ/mol. | | | | | |
| CH_6Si^+ CH_3SiH_3 | 10.7 | 240 | 1003 | -7±1 | -29±4 | 86DON/WAL | 992-94-9 |
| CH_7N_2^+ $\text{CH}_3\text{NH}_2\text{NH}_2$ | | (174) | (729) | | | | |
| | | From proton affinity of CH_3NHNH_2 (RN 60-34-4). PA = (214.1) kcal/mol, (896) kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------------|----------------------------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CH_3BN^+ $\text{CH}_3\text{NH}_2\text{BH}_3$ | (9.66±0.01) | (210) | (878) | -13±1 | -54±4 | 80TEL/RAB | 1722-33-4 |
| CIN^+ ICN | 10.87±0.02 | 305 305 | 1274 1275 | 53.9 54.0 | 225.5 226.1 | 82TN270 | 506-78-5 |
| Cl_4^+ Cl_4 | 8.95 IP is onset of photoelectron band (82JON/DEL). | 142 | 596 | -64 | -268 | 78KUD/KUD | 507-25-5 |
| CIr^+ IrC | (9.5±1) 0 K values. | (400) | (1670) | 180 | 753 | 79HUB/HER | 12385-37-4 |
| CKN^+ KCN | (9.3±0.3) | (236) (236) | (988) (987) | 22 21 | 91 90 | 82TN270 | 151-50-8 |
| CN^+ CN | (14.09) | (428.9) (429.3) | (1794.6) (1796.3) | 104.0±2 104.4±2 | 435.1±10 436.8±10 | 85JANAF | 57-12-5 |
| $\Delta_f H(\text{Ion})$ from appearance potential measurements. IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| CNO^+ NCO | (11.76±0.01) | (308) | (1289) | 37±3 | 154±14 | 70OKA | |
| IP from 83DYK/JON. | | | | | | | |
| CN_2O^+ ONCN | 10.93 | 300.9 | 1259.0 | 48.85±0.03 | 204.4±0.1 | 84NAD/REI | 4343-68-4 |
| IP from 81JON/MOO. See also: 81KIM/KAT. | | | | | | | |
| CN_4^+ $\text{N}=\text{CN}_3$ | (≤10.98±0.02) | (≤361) | (≤1512) | 108±5 | 453±20 | 69OKA/MEL | 764-05-6 |
| CO^+ CO | 14.0139 | <u>296.74</u> <u>295.97</u> | <u>1241.59</u> <u>1238.32</u> | -26.42 -27.20 | -110.53 -113.80 | 82TN270 | 630-08-0 |
| See also: 81KIM/KAT. | | | | | | | |
| COS^+ COS | 11.1736±0.0015 | 224 224 | 936 936 | -34 -34 | -142 -142 | 77PED/RYL | 463-58-1 |
| Cited ionization potential corresponds to formation of $\text{COS}^+(^2\Pi_{3/2})$. Formation of $\text{COS}^+(^2\Pi_{1/2})$ requires 11.2204±0.0015 eV. IP from 81ONO/OSU,80DEL/HUB. | | | | | | | |
| COSe^+ COSe | 10.36±0.01 | (222) | (928) | -17 | -72 | *EST | 1603-84-5 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CO_2^+ CO_2 | 13.773±0.002 | 223.6 223.7 | 935.4 935.7 | -94.05 -93.96 | -393.51 -393.14 | 82TN270 | 124-38-9 |
| See also: 81KIM/KAT. | | | | | | | |
| CP^+ CP | (10.5±0.5) | (365) | (1529) | 123 | 516 | 79HUB/HER | 12326-85-1 |
| CRh^+ CRh | (8.9±0.5) | (370) | (1550) | 165±1 | 692±4 | 84SHI/GIN | 12127-42-3 |
| 0 K values. See also: 81HAQ/GIN. | | | | | | | |
| CS^+ CS | 11.33±0.01 | 327 324 | 1368 1356 | 64 63 | 267 262 | | 2944-05-0 |
| Heat of formation of ion from appearance potentials in CS_2 of 13.64±0.02 eV (to give $\text{CS}^+ + \text{S}^-$) and 15.75±0.02 eV (to give $\text{CS}^+ + \text{S}$). $\Delta_f H(\text{Neutral}) = \Delta_f H(\text{Ion}) - \text{IP}$, in good agreement with 79HUB/HER. | | | | | | | |
| CS_2^+ CS_2 | 10.0685±0.0020 | 260 260 | 1088 1088 | 28±0.2 28 | 117±1 117 | 77PED/RYL | 75-15-0 |
| See also: 81KIM/KAT. | | | | | | | |
| CSe_2^+ CSe_2 | 9.258±0.0002 | 275 | 1149 | 61±5 | 256±20 | 82PIL/SKI | 506-80-9 |
| CSi_2^+ Si_2C | (9.2±0.4) | (344) (343) | (1440) (1437) | 132 131 | 552 549 | 82TN270 | 12070-04-1 |
| CV^+ CV | | (360) | (1506) | | | | |
| $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM). 0 K value. | | | | | | | |
| C_2^+ C_2 | 12.11 | 478 476 | 1998 1992 | 198.8 196.8 | 831.9 823.4 | 79HUB/HER | 12070-15-4 |
| IP from 79HUB/HER. | | | | | | | |
| C_2BrI^+ BrC≡CI | (9.34) | (276.56) | (1157.15) | 61.18 | 255.98 | 84DEW/HEA | 26395-29-9 |
| C_2Br_2^+ BrC≡CBr | 9.67 | 285 | 1192 | 61.8 | 258.6 | 83DEW/HEA | 624-61-3 |
| $\text{C}_2\text{Br}_2\text{F}_4^+$ (CF_2Br) ₂ | (11.1) | (67) | (282) | -189±1 | -789±4 | 83KOL/PAP | 124-73-2 |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-------------------------------------|---|---|--------------------------|-----------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{Br}_2\text{O}_2^+$ | BrCOCOBBr | (10.49±0.1) | (180) | (752) | -62 | -260 | *EST | 15219-34-8 |
| C_2Ce^+ | C_2Ce | (5.6±0.5) | (265) | (1110) | 136 | 570 | 82TN270 | 12012-32-7 |
| | | | (265) | (1109) | 136 | 569 | | |
| C_2ClF_3^+ | $\text{C}_2\text{F}_3\text{Cl}$ | 9.81±0.03 | (89) | (374) | -137±2 | -573±8 | 77PED/RYL | 79-38-9 |
| | | | (90) | (377) | -136 | -570 | | |
| C_2ClF_5^+ | $\text{CF}_3\text{CF}_2\text{Cl}$ | (12.6) | (23) | (98) | -267±1 | -1118±4 | 81BUC/FOR | 76-15-3 |
| | | IP is onset of photoelectron band. | | | | | | |
| C_2ClH^+ | ClC=Cl | (9.44) | (271.94) | (1137.79) | 54.25 | 226.98 | 84DEW/HEA | 25604-71-1 |
| C_2Cl_2^+ | ClC=CCl | 10.09 | 283 | 1183 | 50±10 | 209±42 | 71JANAF | 7572-29-4 |
| | | | 282 | 1180 | 49±10 | 205±42 | | |
| | | See: 81BOC/RIE, 82MAI/THO, 83KLA/MAI. | | | | | | |
| $\text{C}_2\text{Cl}_2\text{F}_2^+$ | $\text{CF}_2=\text{CCl}_2$ | 9.65±0.03 | 142 | 593 | -81±3 | -338±11 | 83KOL/PAP | 79-35-6 |
| | $\text{CFCl}=\text{CFCl}$ | (10.2±0.1) | (157) | (657) | -78 | -327 | 82TN270 | 598-88-9 |
| $\text{C}_2\text{Cl}_2\text{F}_4^+$ | $(\text{CF}_2\text{Cl})_2$ | 12.2 | 60 | 252 | -221±1 | -925±4 | 83KOL/PAP | 76-14-2 |
| $\text{C}_2\text{Cl}_2\text{O}^+$ | $\text{Cl}_2\text{C}=\text{C}=\text{O}$ | 9.0 | (191) | (799) | -16 | -69 | *EST | 4591-28-0 |
| | | IP is onset of photoelectron band (81BOC/HIR, 82LEV/LIA). | | | | | | |
| $\text{C}_2\text{Cl}_2\text{O}_2^+$ | $(\text{COCl})_2$ | 10.91±0.05 | 173 | 724 | -79±1 | -329±5 | 77PED/RYL | 79-37-8 |
| | | See also: 81KIM/KAT. | | | | | | |
| $\text{C}_2\text{Cl}_3\text{F}_3^+$ | CF_3CCl_3 | 11.5 | 92 | 385 | -173±2 | -725±10 | 83KOL/PAP | 354-58-5 |
| | | IP is onset of photoelectron band (81DUM/DUP). See also: 77ROS/DRA. | | | | | | |
| | $\text{CFCl}_2\text{CF}_2\text{Cl}$ | 11.99±0.02 | 103 | 430 | -174±0.7 | -727±3 | 83KOL/PAP | 76-13-1 |
| $\text{C}_2\text{Cl}_3\text{N}^+$ | CCl_3CN | 11.89 | (294) | (1229) | 20 | 82 | *EST | 545-06-2 |
| | | IP from 83MOL/PIK2. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_2Cl_4^+ C_2Cl_4 | 9.32 | 212 212 | 888 889 | -3±0.5 -2 | -11±2 -10 | 83KOL/PAP | 127-18-4 |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |
| $\text{C}_2\text{Cl}_4\text{F}_2^+$ $\text{CFCl}_2\text{CFCl}_2$ | 11.3 | 135 | 563 | -126±2 | -527±10 | 83KOL/PAP | 76-12-0 |
| IP is onset of photoelectron band (81DUM/DUP). | | | | | | | |
| $\text{C}_2\text{Cl}_4\text{O}^+$ CCl_3COCl | (11.0) | (198) | (828) | -56±2 | -236±9 | 77PED/RYL | 76-02-8 |
| IP is onset of photoelectron band (81KIM/KAT). | | | | | | | |
| C_2Cl_6^+ CCl_3CCl_3 | 11.1 | 220 | 921 | -36±1 | -150±5 | 83KOL/PAP | 67-72-1 |
| IP is onset of photoelectron band (81KIM/KAT). See also: 82LEV/LIA. | | | | | | | |
| C_2F_2^+ $\text{FC}\equiv\text{CF}$ | 11.18 | (263) | (1100) | 5±5 | 21±21 | 71JANAF | 689-99-6 |
| See also: 81BIE/ASB. | | | | | | | |
| $\text{C}_2\text{F}_2\text{O}_2^+$ FCOCOF | (12.20±0.02) | (107) | (449) | -174 | -728 | *EST | 359-40-0 |
| C_2F_3^+ C_2F_3 | (10.2) | (189) | (791) | -45.9±2.0 | -192.0±8.4 | 83SPY/SAU | |
| From appearance potentials of 15.84±0.02 eV in C_2F_4 and 15.4±0.1 eV in $\text{C}_2\text{F}_3\text{Cl}$. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| $\text{C}_2\text{F}_3\text{N}^+$ CF_3CN | 13.86 | 200 | 837 | -119.4±0.3 | -499.8±1.2 | 77PED/RYL | 353-85-5 |
| IP from 81ASB/SVE. See also: 83MOL/PIK2. | | | | | | | |
| C_2F_4^+ C_2F_4 | 10.12±0.02 | 75 76 | 316 319 | -158±0.7 -157 | -659±3 -657 | 83KOL/PAP | 116-14-3 |
| See also: 81BIE/VON, 81BIE/ASB. | | | | | | | |
| C_2F_5^+ C_2F_5 | | (0) | (0) | -213±1 | -893±4 | 82MCM/GOL | 3369-48-0 |
| Appearance potentials of this ion in C_2F_6 (15.46 eV), $\text{C}_2\text{F}_5\text{I}$ (11.71 eV); C_3F_8 (13.32 eV), and n- C_4F_{10} (13.05 eV) lead to estimated values for the heat of formation of 15 kcal/mol, 5 kcal/mol, -5 kcal/mol and -14 kcal/mol, respectively. See: 80ING/HAN. | | | | | | | |
| $\text{C}_2\text{F}_5\text{I}^+$ $\text{C}_2\text{F}_5\text{I}$ | (10.66±0.1) | (6) | (25) | -240±1 | -1004±4 | 81BUC/FOR | 354-64-3 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|----------------------------|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_2F_6^+ | C_2F_6 | (13.4) | (-12) | (-50) | -321 | -1343 | 75CHE/ROD | 76-16-4 |
| | | | (-10) | (-41) | -319 | -1334 | | |
| IP is onset of photoelectron band. (80ING/HAN). | | | | | | | | |
| $\text{C}_2\text{F}_7\text{N}^+$ | $(\text{CF}_3)_2\text{NF}$ | (11.6) | (-10) | (-44) | -278 | -1163 | *EST | 359-62-6 |
| IP from 82BUR/PAW. | | | | | | | | |
| C_2H^+ | C_2H | (11.7) | (405) | (1693) | 135±1 | 565±4 | 82MCM/GOL | 2122-48-7 |
| | | | (404) | (1689) | 134 | 560 | | |
| Heat of formation of ion from appearance potential measurement; IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | | |
| C_2HBr^+ | $\text{HC}=\text{CBr}$ | 10.31±0.02 | 297.0 | 1242.4 | 59.2 | 247.7 | 75OKA | 593-61-3 |
| IP from 77ALL/KLO. See also: 82LEV/LIA. | | | | | | | | |
| $\text{C}_2\text{HBrClF}_3^+$ | CF_3CHClBr | 11.0 | 86 | 361 | -167±1 | -700±4 | 83KOL/PAP | 151-67-7 |
| IP is onset of photoelectron band (81DUM/DUP). | | | | | | | | |
| C_2HBrO^+ | $\text{CHBr}=\text{C}=\text{O}$ | (≤9.10) | (≤207) | (≤868) | -2 | -10 | *EST | 78957-22-9 |
| IP from 81BOC/HIR. | | | | | | | | |
| C_2HCl^+ | $\text{HC}=\text{CCl}$ | 10.58±0.02 | 305 | 1276 | 61 | 255 | 70KLO/PAS | 593-63-5 |
| IP from 77ALL/KLO. See also: 84MAI/THO. | | | | | | | | |
| $\text{C}_2\text{HClF}_2^+$ | $\text{CF}_2=\text{CHCl}$ | 9.80±0.04 | 150 | 629 | -76 | -316 | 82TN270 | 359-10-4 |
| $\text{C}_2\text{HClF}_3\text{O}^+$ | $\text{CF}_3\text{C}(\text{OH})\text{Cl}$ | | 4 | 14 | | | | |
| From proton affinity of CF_3COCl (RN 354-32-5)(85MCM/KEB, 85MCM/KEB2). PA = 161.2 kcal/mol, 674 kJ/mol. | | | | | | | | |
| C_2HClO^+ | $\text{CHCl}=\text{C}=\text{O}$ | (≤9.3) | (≤201) | (≤840) | -14 | -57 | *EST | 29804-89-5 |
| See also: 81BOC/HIR. | | | | | | | | |
| $\text{C}_2\text{HCl}_2\text{F}_3^+$ | CF_3CHCl_2 | 11.5 | 88 | 370 | -177±2 | -740±10 | 83KOL/PAP | 306-83-2 |
| IP is onset of photoelectron band (81DUM/DUP). | | | | | | | | |
| | CF_2ClCHCl | ≤12.00 | ≤104 | ≤434 | -173±2 | -724±10 | 83KOL/PAP | 354-23-4 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_2HCl_3^+ C_2HCl_3 | 9.47±0.01 | 214 215 | 895 898 | -4.5±0.7 -4 | -19±3 -16 | 85PAP/KOL | 79-01-6 |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |
| $\text{C}_2\text{HCl}_3\text{N}^+$ CCl_3CNH | | 209 | 876 | | | | |
| From proton affinity of CCl_3CN (RN 545-06-2). PA = 175.8 kcal/mol, 735.5 kJ/mol. | | | | | | | |
| $\text{C}_2\text{HCl}_3\text{O}^+$ CCl_3CHO | (10.5) | (195) | (816) | -47 | -197 | 82TN270 | 75-87-6 |
| IP is onset of photoelectron band (81KIM/KAT). See also: 85GUI/PFI2. | | | | | | | |
| CHCl_2COCl | (11.0) | (196) | (820) | -58±2 | -241±9 | 77PED/RYL | 79-36-7 |
| IP is onset of photoelectron band. | | | | | | | |
| C_2HCl_5^+ $\text{CHCl}_2\text{CCl}_3$ | (11.0) | (220) | (919) | -34±2 | -143±7 | 78GUN/HEA | 76-01-7 |
| IP is onset of photoelectron band (81KIM/KAT). | | | | | | | |
| C_2HF^+ $\text{HC}\equiv\text{CF}$ | 11.26 | 285 | 1193 | 26 | 107 | 80STA/VOG | 2713-09-9 |
| See also: 81BIE/ASB. | | | | | | | |
| C_2HF_3^+ C_2HF_3 | 10.14 | 117 | 487 | -117±2 | -491±8 | 77PED/RYL | 359-11-5 |
| See also: 81BIE/VON, 81BIE/ASB. | | | | | | | |
| $\text{C}_2\text{HF}_3\text{N}^+$ CF_3CNH | | 82 | 343 | | | | |
| From proton affinity of CF_3CN (RN 353-85-5) (85MCM/KEB, 85MCM/KEB2). PA = 164.3 kcal/mol, 687. kJ/mol. | | | | | | | |
| $\text{C}_2\text{HF}_3\text{O}_2^+$ CF_3COOH | 11.46 | 18 | 75 | -246.3±0.3 | -1030.7±1 | 77PED/RYL | 76-05-1 |
| See also: 81ASB/SVE. | | | | | | | |
| $\text{C}_2\text{HF}_4\text{O}^+$ $\text{CF}_3\text{C}(\text{OH})\text{F}$ | | -44 | -182 | | | | |
| From proton affinity of CF_3COF (RN 354-34-7). PA = 160.2 kcal/mol, 670 kJ/mol. | | | | | | | |
| C_2HN^+ HCCN | | (366) | (1531) | | | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations. See also: 85HAR/MCI. 0 K values. | | | | | | | |
| C_2HN_2^+ NCCNH | | 277 | 1161 | | | | |
| From proton affinity of NCCN (RN 460-19-5) (87DEA/MAU). PA = 162 kcal/mol, 678 kJ/mol. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|------------------------------|--------------------------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_2HO^+ HCCO | (9.5) | (262) | (1096) | 42.4±2.1 | 177.4±8.8 | *EST | 51095-15-9 |
| Heat of formation from appearance potential determination (84LOS/HOL). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| C_2HV^+ VC_2H | | (303) | (1268) | | | | |
| $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM). 0 K value. | | | | | | | |
| C_2H_2^+ C_2H_2 | 11.400±0.002 | <u>317.4</u> <u>317.5</u> | <u>1327.9</u> <u>1328.5</u> | 54.5±0.25 54.7 | 228.0±1 228.6 | 77PED/RYL | 74-86-2 |
| See also: 81KIM/KAT, 82HAY/TWA. | | | | | | | |
| $\text{C}_2\text{H}_2\text{Br}_2^+$ (E)-CHBr=CHBr | 9.51±0.04 | (245) | (1024) | 25 | 106 | *EST | 590-12-5 |
| An IP of 9.30±0.02 has also been reported (72CHA/FRO). | | | | | | | |
| (Z)-BrCH=CHBr | 9.63±0.01 | 247 | 1035 | 25 | 106 | *EST | 590-11-4 |
| An IP of 9.32±0.02 eV has also been reported (72CHA/FRO). | | | | | | | |
| $\text{CBr}_2 = \text{CH}_2$ | 9.78±0.01 | (247) | (1034) | 21 | 90 | *EST | 593-92-0 |
| See also: 82VON/ASB. | | | | | | | |
| $\text{C}_2\text{H}_2\text{Br}_2\text{F}_2^+$ $\text{CF}_2\text{BrCH}_2\text{Br}$ | 10.83±0.01 | 147 | 614 | -103±5 | -431±20 | 83KOL/PAP | 75-82-1 |
| $\text{C}_2\text{H}_2\text{CIN}^+$ CH_2CICN | 11.95±0.01 | (296) | (1239) | 21 | 86 | *EST | 107-14-2 |
| $\text{C}_2\text{H}_2\text{Cl}_2^+$ $\text{CH}_2 = \text{CCl}_2$ | 9.79±0.04 | 226 228 | 947 953 | 0.5±0.2 2.0 | 2.3±0.7 8.4 | 77PED/RYL | 75-35-4 |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |
| (Z)-CHCl=CHCl | 9.66±0.01 | 224 225 | 936 942 | 1±0.2 2 | 4±1 10 | 83KOL/PAP | 156-59-2 |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |
| (E)-CHCl=CHCl | 9.65±0.02 | 224 225 | 937 942 | 1±0.2 3 | 6±1 11 | 83KOL/PAP | 156-60-5 |
| See also: 82VON/ASB, 81KIM/KAT. | | | | | | | |
| $\text{C}_2\text{H}_2\text{Cl}_2\text{F}_2^+$ $\text{CF}_2\text{ClCH}_2\text{Cl}$ | ≤11.8 | ≤142 | ≤596 | -130±2 | -543±10 | 83KOL/PAP | 1649-08-7 |
| IP from 81DUM/DUP. | | | | | | | |
| $\text{C}_2\text{H}_2\text{Cl}_2\text{O}^+$ CHCl_2CHO | 10.5 | (199) | (833) | -43±5 | -180±20 | *EST | 79-02-7 |
| IP is onset of photoelectron band (81KIM/KAT). | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_2\text{Cl}_2\text{O}^+$ CH_2ClCOCl | (11.0) IP is onset of photoelectron band. | (195) | (815) | -59±2 | -246±9 | 77PED/RYL | 79-04-9 |
| $\text{C}_2\text{H}_2\text{Cl}_3\text{O}_2^+$ $\text{CCl}_3\text{C}(\text{OH})_2$ | | 76 | 318 | | | | |
| | | From proton affinity of CCl_3COOH (RN 76-03-9) (PA = 183.5 kcal/mol, 768 kJ/mol). | | | | | |
| $\text{C}_2\text{H}_2\text{Cl}_4^+$ $\text{CH}_2\text{ClCCl}_3$ | (11.1) IP is onset of photoelectron band (81KIM/KAT). | (220) | (919) | -36±0.2 | -152±1 | 83KOL/PAP | 630-20-6 |
| $(\text{CHCl}_2)_2$ | (≤11.62) IP from 81KIM/KAT. | (≤232) | (≤971) | -36±1 | -150±5 | 77PED/RYL | 79-34-5 |
| $\text{C}_2\text{H}_2\text{F}^+$ CH_2CF | | 227 | 951 | | | | |
| | | From appearance potential of 13.56 eV in $\text{C}_2\text{H}_3\text{F}$ in agreement with value from proton affinity of HCCF (PA = 165 kcal/mol, 689 kJ/mol). See also: 8SHEI/BAR, 84BEA/EYE. | | | | | |
| $\text{C}_2\text{H}_2\text{F}_2^+$ $\text{CH}_2=\text{CF}_2$ | 10.29±0.01 | 155 | 648 | -82±2 | -345±10 | 76WIL/LEB | 75-38-7 |
| | | 157 | 655 | -81 | -338 | | |
| | | See also: 81BIE/VON, 81BIE/ASB. | | | | | |
| (Z)-CHF=CHF | 10.23 | 165 | 690 | -71 | -297 | 80STA/VOG | 1630-77-9 |
| | | See also: 81BIE/VON, 81BIE/ASB, 79JOC/LOH, 81MAI/THO2. | | | | | |
| (E)-CHF=CHF | 10.21 | 165 | 692 | -70 | -293 | 80STA/VOG | 1630-78-0 |
| | | See also: 81BIE/VON, 81BIE/ASB, 79JOC/LOH. | | | | | |
| $\text{C}_2\text{H}_2\text{F}_3^+$ CHF_2CHF | | (79) | (332) | | | | |
| | | From proton affinity of $\text{CF}_2=\text{CHF}$ (RN 359-11-5). PA = -169 kcal/mol, -707 kJ/mol. | | | | | |
| CF_3CH_2 | (10.6±0.1) | (120) | (506) | -124±2 | -517±8 | 82MCM/GOL | 3248-58-6 |
| $\text{C}_2\text{H}_2\text{F}_3\text{I}^+$ $\text{CF}_3\text{CH}_2\text{I}$ | 9.998 | 75 | 316 | -155±1 | -649±4 | 83KOL/PAP | 353-83-3 |
| $\text{C}_2\text{H}_2\text{F}_3\text{NO}^+$ CF_3CONH_2 | (10.8) IP from 81ASB/SVE. | (49) | (206) | -200 | -836 | *EST | 354-38-1 |
| $\text{C}_2\text{H}_2\text{F}_3\text{O}^+$ CF_3CHOH | | 12 | 49 | | | | |
| | | From proton affinity of CF_3CHO (RN 75-90-1). PA = 165.1 kcal/mol, 691 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_2\text{F}_3\text{O}_2^+$ $\text{CF}_3\text{C}(\text{OH})_2$ | | -50 | -208 | | | | |
| | | From proton affinity of CF_3COOH (RN 76-05-1). PA = 169.0 kcal/mol, 707 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued



| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|--|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_2\text{I}_2^+$ (Z)-CHI=CHI | (8.6) | (248) | (1037) | 49.5±0.3 | 207.2±1.1 | 77PED/RYL | 590-26-1 |
| | IP is onset of photoelectron band. | | | | | | |
| (E)-CHI=CHI | (8.6) | (248) | (1037) | 49.5±0.3 | 207.2±1.1 | 77PED/RYL | 590-27-2 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_2\text{H}_2\text{N}^+$ CH ₂ CN | (10.0) | (290) | (1214) | 59±2 | 245±10 | 82MCM/GOL | 2932-82-3 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (77ROS/DRA, 85HAR/MCI) IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See also: 82ALL/MIG. | | | | | | |
| $\text{C}_2\text{H}_2\text{N}_2\text{Se}^+$  | (8.9) | (290) | (1212) | 84 | 353 | *EST | 26223-16-5 |
| | IP from 80BOC/AYG, 82LEV/LIA. | | | | | | |
| $\text{C}_2\text{H}_2\text{N}_4^+$  | (9.14) | (322) | (1346) | 111 | 464 | 82JOS | 290-96-0 |
| $\text{C}_2\text{H}_2\text{O}^+$ HC=COH | | 247 | 1033 | $\Delta_f H(\text{Ion})$ from appearance potential determination (86BAA/WEI). | | | |
| CH ₂ CO | 9.61±0.02 | 210.2 | 879.6 | -11.4±0.6 | -47.7±2.5 | 71NUT/LAU | 463-51-4 |
| | | 210.9 | 882.7 | -10.7 | -44.6 | | |
| | See also: 81BOC/HIR. | | | | | | |
| $\text{C}_2\text{H}_2\text{O}_2^+$ (CHO) ₂ | 10.1 | 182 | 763 | -50.6±0.2 | -211.9±0.8 | 77PED/RYL | 107-22-2 |
| | IP is onset of photoelectron band (80VON/BIE, 81KIM/KAT). | | | | | | |
| $\text{C}_2\text{H}_2\text{O}_4^+$ HOOCOOH | (10.8) | (74) | (310) | -175±0.7 | -732±3 | 77PED/RYL | 144-62-7 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_2\text{H}_2\text{S}^+$ CH ₂ =C=S | (8.77) | (242) | (1011) | 39 | 165 | *EST | 18282-77-4 |
| | | (234) | (979) | Cited IP is onset of photoelectron band (77ROS/SOL). Heat of formation of ion from appearance potential in CH ₃ SSCH ₃ (83BUT/BAE). $\Delta_f H(\text{Neutral})$ is ($\Delta_f H(\text{Ion}) - \text{IP}$). | | | |

Table 1. Positive Ion Table - Continued


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|------------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_2\text{S}_2^+$  | (8.5) IP is onset of photoelectron band (83SCH/SCH). | (258) | (1080) | 62 | 260 | *EST | 7092-01-5 |
| $\text{C}_2\text{H}_2\text{Se}^+$ $\text{CH}_2 = \text{C} = \text{Se}$ | 8.7 IP is onset of photoelectron band (80BOC/AYG). | (256) | (1071) | 55 | 232 | *EST | 61134-37-0 |
| C_2H_3^+ C_2H_3 | 8.9 Heat of formation of ion from appearance potential measurement; IP from J.L. Beauchamp, personal communication. | 265.9 267.9 | 1112 1120.9 | 63.4±1 62.7 | 265.3±4 262.2 | 85KIE/WEI | 2669-89-8 |
| $\text{C}_2\text{H}_3\text{Br}^+$ $\text{C}_2\text{H}_3\text{Br}$ | 9.80±0.02 See also: 82VON/ASB, 83CAM/CIU, 84MIL/BAE. | 244.9 248.5 | 1024.8 1039.7 | 18.9±0.5 22.5 | 79.3±1.9 94.2 | 77PED/RYL | 593-60-2 |
| $\text{C}_2\text{H}_3\text{BrHg}^+$ $\text{CH}_2 = \text{CHHgBr}$ | (9.8) IP is onset of photoelectron band (81BAI/CHI). | (256) | (1072) | 30 | 126 | *EST | 16188-37-7 |
| $\text{C}_2\text{H}_3\text{BrO}^+$ CH_3COBr | 10.4±0.1 IP is onset of photoelectron band (82LEV/LIA, 81KIM/KAT). | 194 | 813 | -45.5±0.1 | -190.4±0.5 | 77PED/RYL | 506-96-7 |
| $\text{C}_2\text{H}_3\text{BrO}_2^+$ CH_2BrCOOH | (10.4) IP is onset of photoelectron band. | (145) | (608) | -94.4±1.5 | -395±6 | *EST | 79-08-3 |
| $\text{C}_2\text{H}_3\text{Cl}^+$ $\text{C}_2\text{H}_3\text{Cl}$ | 9.99±0.02 See also: 83CAM/CIU, 82VON/ASB, 81KIM/KAT. | 236 238 | 987 995 | 5±0.5 7 | 23±2 31 | 83KOL/PAP | 75-01-4 |
| $\text{C}_2\text{H}_3\text{ClF}_2^+$ $\text{CH}_3\text{CF}_2\text{Cl}$ | 11.98±0.01 | 149.7 | 626.2 | -126.6±1.2 | -529.7±5.0 | 78PAP/KOL | 75-68-3 |
| $\text{C}_2\text{H}_3\text{ClN}^+$ ClCH_2CNH | From proton affinity of ClCH_2CN (RN 107-14-2). PA = 179.5 kcal/mol, 751 kJ/mol. | 207 | 865 | | | | |
| $\text{C}_2\text{H}_3\text{ClO}^+$ CH_3COCl | 10.85±0.05 See: 81KIM/KAT | 192 194 | 804 813 | -58±0.2 -56 | -243±1 -234 | 77PED/RYL | 75-36-5 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_3\text{ClO}^+$ CH_2ClCHO | 10.48±0.03 See: 81KIM/KAT. | (195) | (816) | -47±4 | -195±15 | *EST | 107-20-0 |
| $\text{C}_2\text{H}_3\text{ClO}_2^+$ CH_2ClCOOH | (10.7) IP is onset of photoelectron band. | (143) | (597) | -104±2 | -435±9 | 77PED/RYL | 79-11-8 |
| $\text{C}_2\text{H}_3\text{Cl}_3^+$ $\text{CHCl}_2\text{CH}_2\text{Cl}$ | 11.0 IP is onset of photoelectron band (81KIM/KAT). | 218 | 912 | -36±0.5 | -149±2 | 77PED/RYL | 79-00-5 |
| CH_3CCl_3 | (11.0) IP is onset of photoelectron band (81KIM/KAT). | (219) | (916) | -34.6±0.1 | -144.9±0.6 | 83KOL/PAP | 71-55-6 |
| $\text{C}_2\text{H}_3\text{Cl}_3\text{O}^+$ $\text{CCl}_3\text{CH}_2\text{OH}$ | (10.94) IP from 83KOP/MOL. | (182) | (763) | (-70) | (-293) | *EST | 115-20-8 |
| $\text{C}_2\text{H}_3\text{Cl}_3\text{Si}^+$ $\text{CH}_2=\text{CHSiCl}_3$ | (≤11.0) IP from 81KHV/ZYK. | (≤144) | (≤603) | -109 | -458 | *EST | 75-94-5 |
| $\text{C}_2\text{H}_3\text{F}^+$ $\text{C}_2\text{H}_3\text{F}$ | 10.363±0.015 See also: 81BIE/VON, 81BIE/ASB. | 205.8 | 861.1 | -33.2±0.4 | -138.8±1.7 | 76WIL/LEB | 75-02-5 |
| $\text{C}_2\text{H}_3\text{FO}^+$ CH_3COF | 11.51±0.02 See: 81KIM/KAT. | 159 | 667 | -106±0.7 | -444±3 | 77PED/RYL | 557-99-3 |
| $\text{C}_2\text{H}_3\text{F}_2^+$ CH_2FCHF | | 130 | 543 | | | | |
| | From proton affinity of (E)-CHF=CHF (RN 1630-78-0). PA = 166 kcal/mol, 694 kJ/mol. | | | | | | |
| CH_3CF_2 | (7.92) | (109) | (458) | -72±2 | -303±8 | 82MCM/GOL | 40640-67-3 |
| | Value of $\Delta_f H(\text{Ion})$ from appearance potential determination (84HEI/BAR, 85HEI/BAR); value from proton affinity of $\text{CH}_2=\text{CF}_2$ (RN 75-38-7) = 108 kcal/mol, 451 kJ/mol. PA = 176 kcal/mol, 736 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_3\text{F}_3^+$ CH_3CF_3 | 12.9±0.1 | 118 122 | 496 509 | -179±0.7 -176 | -749±3 -736 | 83KOL/PAP | 420-46-2 |
| | IP from 73GOL/KOR. | | | | | | |
| $\text{C}_2\text{H}_3\text{F}_3\text{O}^+$ $\text{CF}_3\text{CH}_2\text{OH}$ | 11.49 IP from 83KOP/MOL. | 53 | 221 | -212±1 | -888±5 | 77PED/RYL | 75-89-8 |
| $\text{C}_2\text{H}_3\text{I}^+$ $\text{CH}_2=\text{CHI}$ | 9.30 | (246) | (1027) | 31 | 130 | *EST | 593-66-8 |

Table 1. Positive Ion Table - Continued

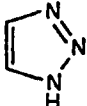
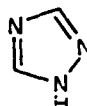

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|------------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_3\text{N}^+$ | | | | | | | |
| CH_3CN | 12.194±0.005 | 299 300 | 1251 1258 | 18±0.2 19 | 74±1 81 | 83AN/MAN | 75-05-8 |
| | | IP from 81RID/RAY. See also: 82CHE/LAP, 84OHN/MAT, 82ALL/MIG, 81KIM/KAT, 85HAR/MCI. | | | | | |
| CH_2CNH | | (240) (242) | (1004) (1011) | | | | |
| | | From appearance potential determinations. | | | | | |
| CH_3NC | 11.24 | 300 302 | 1257 1262 | 41±0.2 43 | 173±1 178 | 83AN/MAN | 593-75-9 |
| | | See also: 82CHE/LAP, 81BEV/SAN, 85HAR/MCI. | | | | | |
| $\text{C}_2\text{H}_3\text{NO}^+$ | | | | | | | |
| CH_3NCO | (10.67±0.02) | (215) | (899) | -31 | -130 | 75COM/DES | 624-83-9 |
| $\text{C}_2\text{H}_3\text{NS}^+$ | | | | | | | |
| CH_3SCN | (9.96±0.05) | (268) | (1121) | 38 | 160 | 82TN270 | 556-64-9 |
| CH_3NCS | (9.25±0.03) | (245) (247) | (1023) (1032) | 31 33 | 131 140 | 82TN270 | 556-61-6 |
| $\text{C}_2\text{H}_3\text{N}_3^+$ | | | | | | | |
|  | 10.06 | 291 | 1218 | 59 | 247 | 82IOS | 288-36-8 |
| | | See: 81PAL/SIM. | | | | | |
|  | (9.8) | (272) | (1140) | 46±0.5 | 194±2 | 85FAO/AKA | 288-88-0 |
| | | IP is onset of photoelectron band (81PAL/SIM). | | | | | |
| $\text{C}_2\text{H}_3\text{O}^+$ | | | | | | | |
| CH_3CO | 7.0 | 156 | 653 | -6±0.5 | -24±2 | 82MCM/GOL | 15762-07-9 |
| | | $\Delta_f H(\text{Ion})$ at 298 K from 82TRA/MCL, 84LIA/LIE, and 81LIF/TZI. See also: 84LOS/HOL, 83LIF/BER. Value derived from proton affinity of ketene is 157 kcal/mol, 657 kJ/mol. PA = 198.0 kcal/mol, 828 kJ/mol. IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. Experimentally determined IP of this radical is 8.05±0.17 eV. See also: 82BUR/HOL2. | | | | | |
| $\text{CH}_2 = \text{COH}$ | | (192) | (803) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (82HOL/LOS, 82HOL/LOS2, 83BUR/HOL2). | | | | | |
|  | | (201) | (841) | | | | 31586-84-2 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential measurements. See also: 83BUR/HOL2. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|------------------------------|----------------------------|------------------------------|-------------------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_3\text{O}_2\text{I}^+$ CH_2ICOOH | (9.6) | (327) | (1367) | 105 | 441 | *EST | 64-69-7 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_2\text{H}_3\text{S}^+$ CH_3CS | | 204 | 853 | | | | |
| From proton affinity of $\text{CH}_2=\text{C}=\text{S}$ (RN 18282-77-4) (83CAS/KIM). PA = 201.2 kcal/mol, 842 kJ/mol. Original authors recommend value of 210 kcal/mol, 879 kJ/mol, using $\Delta_f H(\text{CH}_2=\text{C}=\text{S}) = 46$ kcal/mol, 196 kJ/mol, from MNDO calculation. | | | | | | | |
| $\text{C}_2\text{H}_3\text{V}^+$ VC_2H_3 | | (266) | (1115) | | | | |
| $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM). 0 K value. | | | | | | | |
| C_2H_4^+ C_2H_4 | 10.507±0.004 | <u>254.8</u> <u>256.8</u> | <u>1066</u> <u>1074</u> | 12.5±0.2 14.5 | 52.2±1 60.7 | 77PED/RYL | 74-85-1 |
| See also: 81KIM/KAT, 84POL/TRE. | | | | | | | |
| $\text{C}_2\text{H}_4\text{BrCl}^+$ $\text{CH}_2\text{BrCH}_2\text{Cl}$ | 10.67±0.03 | 225 | 942 | -21±1 | -87±5 | 83KOL/PAP | 107-04-0 |
| See: 81KIM/KAT. | | | | | | | |
| $\text{C}_2\text{H}_4\text{BrCl}^+$ CH_3CHClBr | 10.37 | 219±1 | 918±5 | -20±1 | -83±5 | 83KOL/PAP | 593-96-4 |
| $\text{C}_2\text{H}_4\text{BrF}^+$ $\text{CH}_2\text{FCH}_2\text{Br}$ | ≤10.57 | (≤184) | (≤769) | -60±5 | -251±20 | 83KOL/PAP | 762-49-2 |
| $\text{C}_2\text{H}_4\text{Br}_2^+$ $\text{CH}_2\text{BrCH}_2\text{Br}$ | 10.37 | 230 | 962 | -9±0.2 | -39±1 -38.3 ± 1.2 | 83KOL/PAP | 106-93-4 |
| See: 78GAN/PBE, 81KIM/KAT, 77STA/WIE. | | | | | | | |
| $\text{C}_2\text{H}_4\text{Br}_2^+$ CH_3CHBr_2 | 10.17 | 226 | 944 | -9±1 | -37±6 -26.7 ± 1.9 | 83KOL/PAP | 557-91-5 |
| $\text{C}_2\text{H}_4\text{Cl}^+$ CH_3CHCl | | (199) | (832) | | | | |
| From appearance potential (11.20 eV) in CH_3CHCl_2 . | | | | | | | |
| $\text{C}_2\text{H}_4\text{Cl}^+$ CH_2ClCH_2 | | (204) | (855) | | | | |
| From appearance potential (11.47 eV) in $\text{CH}_2\text{ClCH}_2\text{Cl}$. | | | | | | | |
| $\text{C}_2\text{H}_4\text{ClO}_2^+$ $\text{CH}_2\text{ClC}(\text{OH})_2$ | | 79 | 332 | | | | |
| From proton affinity of CH_2ClCOOH (RN 79-11-8). PA = 182.4 kcal/mol, 763 kJ/mol. | | | | | | | |
| $\text{C}_2\text{H}_4\text{Cl}_2^+$ CH_3CHCl_2 | 11.06 | 224 229 | 936 959 | -31±0.7 -26 | -131±3 -108 | 83KOL/PAP | 75-34-3 |
| See also: 81KIM/KAT. | | | | | | | |

Table 1. Positive Ion Table - Continued

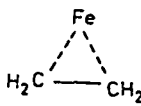
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_4\text{Cl}_2^+$ $\text{CH}_2\text{ClCH}_2\text{Cl}$ | 11.04 | 222 225 | 931 942 | -32 ± 0.2 -29 | -134 ± 1 -123 | 83KOL/PAP | 107-06-2 |
| See also: 81KIM/KAT. | | | | | | | |
| $\text{C}_2\text{H}_4\text{Cl}_2\text{O}^+$ $\text{CH}_3\text{OCHCl}_2$ | (10.6) | (191) | (800) | -53 | -222 | *EST | 4885-02-3 |
| IP is onset of photoelectron band (80VER/SAL). | | | | | | | |
| $\text{C}_2\text{H}_4\text{Cl}_3\text{O}^+$ $\text{CCl}_3\text{CH}_2\text{OH}_2$ | | 118 | 495 | | | | |
| From proton affinity of $\text{CCl}_3\text{CH}_2\text{OH}$ (RN 115-20-8). PA = 177.4 kcal/mol, 742 kJ/mol. | | | | | | | |
| $\text{C}_2\text{H}_4\text{F}^+$ CH_3CHF | 7.93 | 157 | 659 | -26 | -106 | ? | |
| $\Delta_f H(\text{Ion})$ from proton affinity of $\text{C}_2\text{H}_3\text{F}$ (RN 75-02-5). PA = 175 kcal/mol, 732 kJ/mol. $\Delta_f H(\text{Neutral}) = \text{IP} - \Delta_f H(\text{Ion})$. | | | | | | | |
| $\text{C}_2\text{H}_4\text{FO}_2^+$ $\text{CH}_2\text{FC}(\text{OH})_2$ | | 42 | 176 | | | | |
| From proton affinity of CH_2FCOOH (RN 144-49-0). PA = 183.5 kcal/mol, 768. kJ/mol. | | | | | | | |
| $\text{C}_2\text{H}_4\text{F}_2^+$ CH_3CHF_2 | 11.87 ± 0.03 | 154 | 644 | -120 ± 1 | -501 ± 6 | 75CHE/ROD | 75-37-6 |
| IP from 84HEI/BAR, 85HEI/BAR. | | | | | | | |
| $\text{C}_2\text{H}_4\text{F}_3\text{N}^+$ $\text{CF}_3\text{CH}_2\text{NH}_2$ | (9.8 ± 0.1) | (58) | (244) | (-167) | (-701) | *EST | 753-90-2 |
| IP is average of values from 83MOL/PIK3, 79AUE/BOW. | | | | | | | |
| $\text{C}_2\text{H}_4\text{F}_3\text{O}^+$ $\text{CF}_3\text{CH}_2\text{OH}_2$ | | -16 | -65 | | | | |
| From proton affinity of $\text{CF}_3\text{CH}_2\text{OH}$ (RN 75-89-8). See also: 85MCM/KEB. PA = 169.0 kcal/mol, 707 kJ/mol. | | | | | | | |
| $\text{C}_2\text{H}_4\text{Fe}^+$ | | (256) | (1071) | | | | |
|  | | $\Delta_f H(\text{Ion})$ from 84JAC/JAC. | | | | | |
| $\text{C}_2\text{H}_4\text{I}_2^+$ $\text{CH}_2\text{ICH}_2\text{I}$ | (9.4) | (233) | (973) | 15.8 ± 0.3 | 66.3 ± 1.4 | 77PED/RYL | 624-73-7 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_2\text{H}_4\text{N}^+$ CH_3CNH | | 195 | 817 | | | | |
| From proton affinity of CH_3CN (RN 75-05-8). PA = 188.2 kcal/mol, 787 kJ/mol. | | | | | | | |

Table 1. Positive Ion Table - Continued

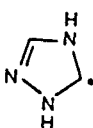
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|------------------------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_4\text{N}^+$ CH_3NCH | | 205 | 860 | | | | |
| | | From proton affinity of CH_3NC (RN 593-75-9) (86KNI/FRE, 86MAU/KAR). PA = 201.4 kcal/mol, 843 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_4\text{NO}^+$ CH_3NHCO | | 150 | 628 | | | | |
| | | From proton affinity of CH_3NCO (RN 624-83-9) (85KAR/STE). PA = 184.5 kcal/mol, 772. kJ/mol. | | | | | |
| $\text{C}_2\text{H}_4\text{NS}^+$ CH_3SCNH | | 212 | 886 | | | | |
| | | From proton affinity of CH_3SCN (RN 556-64-9) (85KAR/STE). PA = 192. kcal/mol, 804. kJ/mol. | | | | | |
| CH_3NCSH | | 204 | 853 | | | | |
| | | From proton affinity of CH_3NCS (RN 556-61-6) (85KAR/STE). PA = 193.0 kcal/mol, 807.5 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_4\text{N}_2^+$ $\text{CH}_2=\text{NN}=\text{CH}_2$ | (8.95) | (264) | (1104) | 58 | 241 | 82JOS | 503-27-5 |
| | See also: 84KIR/POP. | | | | | | |
| $\text{C}_2\text{H}_4\text{N}_2\text{O}_2^+$ $\text{NH}_2\text{COCONH}_2$ | (9.41) | (121) | (505) | -96±1 | -403±5 | 77PED/RYL | 471-46-5 |
| $\text{C}_2\text{H}_4\text{N}_3^+$  | | 199 | 835 | | | | |
| | | From proton affinity of 1H-1,2,4-Triazole (RN 288-88-0) (86MAU/LIE). PA = 212.4 kcal/mol, 889. kJ/mol. | | | | | |
| $\text{C}_2\text{H}_4\text{N}_4^+$ $\text{NCN}=\text{C}(\text{NH}_2)_2$ | (8.4) | (230) | (963) | 36 | 153 | 77PED/RYL | 10191-60-3 |
| | IP is onset of photoelectron band (80KLA/BUT). | | | | | | |
| $\text{C}_2\text{H}_4\text{O}^+$ CH_3CHO | 10.229±0.0007 | <u>196.3</u> <u>198.9</u> | <u>821.1</u> <u>831.9</u> | -39.6±0.1 -37.0 | -165.8±0.4 -155.0 | 77PED/RYL | 75-07-0 |
| | See also: 82JOH/POW, 72POT/SOR, 81ELS/ALL, 81KIM/KAT, 77STA/WIE. | | | | | | |
| $\text{CH}_2=\text{CHOH}$ | 9.14 | 181 | 757 | -30 | -125 | 82HOL/LOS3 | 557-75-5 |
| | From 82HOL/LOS3, 84ALB/ALL. | | | | | | |
| CH_3COH | | (207) | (865) | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations (83TER/WEZ). | | | | | | |

Table 1. Positive Ion Table - Continued



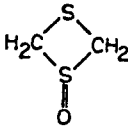
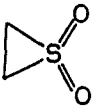
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_4\text{O}^+$ | | 10.566±0.01 | 231.0 | 966.8 | -12.6±0.1 | -52.6±0.6 | 77PED/RYL | 75-21-8 |
|  | | | 234.1 | 979.4 | -9.6 | -40.1 | | |
| | | See also: 82JOH/POW, 81KIM/KAT, 82BIE/ASB. | | | | | | |
| $\text{C}_2\text{H}_4\text{OS}^+$ | CH_3COSH | 10.00±0.02 | 189 | 790 | -42±2 | -175±8 | 77PED/RYL | 507-09-5 |
|  | | 9.2 | (205) | (858) | -7 | -30 | *EST | 7117-41-1 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_2\text{H}_4\text{OS}_2^+$ | | (8.8) | (199) | (831) | -4 | -18 | *EST | 58816-63-0 |
|  | | IP is onset of photoelectron band (82BLO/COR). | | | | | | |
| $\text{C}_2\text{H}_4\text{O}_2^+$ | HCOOCH_3 | 10.815±0.005 | 164.4 | 688.0 | -85.0±0.2 | -355.5±0.7 | 77PED/RYL | 107-31-3 |
| | | See also: 81KIM/KAT, 8SCAN/HAM. | | | | | | |
| | CH_3COOH | 10.66±0.02 | 142.5 | 596.4 | -103.3±0.1 | -432.1±0.4 | 78CHA/ZWO | 64-19-7 |
| | | | 145.9 | 610.4 | -99.9±0.1 | -418.1±0.4 | | |
| | | See also: 81HOL/FIN, 80VON/BIE, 81KIM/KAT. | | | | | | |
| | $\text{CH}_2\text{C}(\text{OH})_2$ | | 120 | 503 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations. | | | | | | |
| | $\text{HOCH}=\text{CHOH}$ | (9.62±0.10) | (146) | (612) | -76 | -316 | *EST | |
| | | IP from 86TUR/HAV3. | | | | | | |
| | CH_3OCOH | | 158 | 661 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential of metastable ion (83TER/WEZ). | | | | | | |
| | $\text{CH}_2\text{CO}(\text{H}_2\text{O})$ | | (138) | (579) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (86POS/RUT). | | | | | | |
| $\text{C}_2\text{H}_4\text{O}_2\text{S}^+$ | | (10.3) | (177) | (741) | -60 | -253 | *EST | 1782-89-4 |
|  | | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

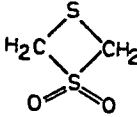
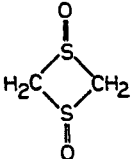

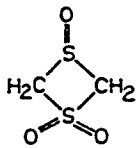
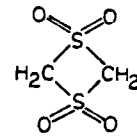


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_4\text{O}_2\text{S}_2^+$ | | | | | | | |
|  | (9.4) | (159) | (666) | -58 | -241 | *EST | 60743-07-9 |
| | IP is onset of photoelectron band (82BLO/COR). | | | | | | |
|  | (8.6) | (167) | (700) | -31 | -130 | *EST | 60743-08-0 |
| | IP is onset of photoelectron band (82BLO/COR). | | | | | | |
| $\text{C}_2\text{H}_4\text{O}_3^+$ | | | | | | | |
|  | (10.1) | (183) | (765) | -50 | -209 | 82CRE | 289-14-5 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_2\text{H}_4\text{O}_3\text{S}_2^+$ | | | | | | | |
|  | (9.6) | (137) | (573) | -84 | -353 | *EST | 60743-10-4 |
| | IP is onset of photoelectron band (82BLO/COR). | | | | | | |
| $\text{C}_2\text{H}_4\text{O}_4\text{S}_2^+$ | | | | | | | |
|  | (10.6) | (107) | (447) | -138 | -576 | *EST | 21511-46-6 |
| | IP is onset of photoelectron band (82BLO/COR). | | | | | | |
| $\text{C}_2\text{H}_4\text{S}^+$ | | | | | | | |
| CH_3CHS | 8.98±0.02 | 218 | 910 | 11 | 44 | 79JOS | 6851-93-0 |
| | See also: 83BUT/BAE2. | | | | | | |
|  | 9.051±0.006 | 228 | 955 | 19.6±0.3 | 82.1±1.2 | 77PED/RYL | 420-12-2 |
| | | 231 | 967 | 22.4 | 93.7 | | |
| | See also: 82BUT/BAE, 83BUT/BAE2. | | | | | | |
| $\text{C}_2\text{H}_4\text{S}_2^+$ | | | | | | | |
|  | (8.5) | (218) | (914) | 22 | 94 | *EST | 287-53-6 |
| | IP is onset of photoelectron band (82BLO/COR). | | | | | | |

Table 1. Positive Ion Table - Continued


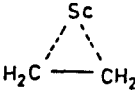
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-------------------------------------|---|----------------------------|--|----------------|------------------------------|------------------|----------------------|------------------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_4\text{S}_3^+$ |  | (≤ 8.72) | (≤ 196) | (≤ 818) | -5 | -23 | *EST | 289-16-7 |
| $\text{C}_2\text{H}_4\text{Sc}^+$ |  | | (215) | (899) | | | | |
| | | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). See also: 86ELK/ARI. | | | | | |
| $\text{C}_2\text{H}_4\text{Se}^+$ | $\text{CH}_3\text{CH}=\text{Se}$ | (8.3) | (219) | (915) | 27 | 114 | *EST | 67281-48-5 |
| | | | IP is onset of photoelectron band (84BOC/AYG). | | | | | |
| C_2H_5^+ | C_2H_5 | 8.13 | 215.6 ± 1.0 | 902 ± 4 | 28 | 118 | 84CAO/BAC | 14936-94-8 2025-Sc-1 |
| | | | 218.5 ± 1.0 | 914 ± 4 | 31 | 130 | | |
| | | | Heat of formation of ion from appearance potential measurements (See: 81TRA/MCL, 80BAE, 82DYK/JON2, 82ROS/BUF). IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 100.5$ kcal/mol. Experimental IP of radical $\leq 8.26 \pm 0.02$ eV. (84DYK/ELL). | | | | | |
| $\text{C}_2\text{H}_5\text{Br}^+$ | $\text{C}_2\text{H}_5\text{Br}$ | 10.28 | 222.2 | 929.6 | -14.9 ± 0.2 | -62.3 ± 1.0 | 77PED/RYL | 74-96-4 |
| | | | 227.4 | 951.5 | -9.6 ± 0.2 | -40.4 ± 1.0 | | |
| | | | See also: 81KIM/KAT, 85OHN/IMA. | | | | | |
| $\text{C}_2\text{H}_5\text{BrO}^+$ | $\text{CH}_2\text{BrCH}_2\text{OH}(\text{gauche})$ | (≤ 10.75) | (≤ 196) | (≤ 820) | (-52) | (-217) | *EST | 540-51-2 |
| | | | See also: 84KOB, 81KIM/KAT, 85OHN/IMA. | | | | | |
| | $\text{CH}_2\text{BrCH}_2\text{OH}(\text{trans})$ | (≤ 10.65) | (≤ 194) | (≤ 811) | (-52) | (-217) | *EST | 540-51-2 |
| | | | See also: 84KOB, 81KIM/KAT, 85OHN/IMA. | | | | | |
| $\text{C}_2\text{H}_5\text{Cl}^+$ | $\text{C}_2\text{H}_5\text{Cl}$ | 10.97 ± 0.02 | 226 | 946 | -26.8 ± 0.1 | -112.1 ± 0.5 | 77PED/RYL | 75-00-3 |
| | | | 230 | 961 | -23.3 | -97.6 | | |
| | | | See also: 83OHN/IMA, 81KIM/KAT. | | | | | |
| | CH_3CHClH | | 227 | 951 | | | | |
| | | | $\Delta_f H(\text{Ion})$ from appearance potential determination (83HOL/BUR). | | | | | |
| $\text{C}_2\text{H}_5\text{ClHg}^+$ | $\text{C}_2\text{H}_5\text{HgCl}$ | 9.9 | 212 | 888 | -16 ± 1 | -67 ± 4 | 80TEL/RAB | 107-27-7 |
| | | | IP is onset of photoelectron band (81BAI/CHI2). | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_5\text{ClO}^+$ $\text{C}_2\text{H}_5\text{OCl}$ | (10.13±0.02) | (212) | (886) | -22 | -91 | *EST | 624-85-1 |
| | IP from 81COL/FRO. | | | | | | |
| $\text{CH}_2\text{ClCH}_2\text{OH}$ | (10.52) | (181) | (756) | -62 | -259 | *EST | 107-07-3 |
| | IP is onset of photoelectron band (81KIM/KAT, 85OHN/IMA). See also: 84KOB. | | | | | | |
| $\text{CH}_3\text{OCH}_2\text{Cl}$ | (10.2) | (184) | (769) | -51 | -215 | *EST | 107-30-2 |
| | IP is onset of photoelectron band (80VER/SAL). | | | | | | |
| $\text{C}_2\text{H}_5\text{ClS}^+$ $\text{CH}_3\text{SCH}_2\text{Cl}$ | (≤7.74) | (≤157) | (≤657) | -22 | -90 | *EST | 2373-51-5 |
| $\text{C}_2\text{H}_5\text{Cl}_2\text{P}^+$ $\text{C}_2\text{H}_5\text{PCl}_2$ | 9.3 | 153 | 638 | -62±4 | -259±18 | 80TEL/RAB | 1498-40-4 |
| | IP is onset of photoelectron band (83ZVE/BAZ, 82LEV/LIA). | | | | | | |
| $\text{C}_2\text{H}_5\text{Cl}_3\text{Si}^+$ $\text{C}_2\text{H}_5\text{SiCl}_3$ | (10.74±0.04) | (122) | (509) | -126±6 | -527±25 | 80TEL/RAB | 115-21-9 |
| $\text{CH}_3\text{SiCl}_2(\text{CH}_2\text{Cl})$ | (10.4) | (129) | (538) | -111 | -465 | *EST | 1558-33-4 |
| | IP is onset of photoelectron band (81ZYK/KHV). | | | | | | |
| $\text{C}_2\text{H}_5\text{F}^+$ $\text{C}_2\text{H}_5\text{F}$ | (11.6) | (205) | (856) | -63±0.5 | -263±2 | 75CHE/ROD | 353-36-6 |
| | IP is onset of photoelectron band (81BIE/ASB). See also: 81KIM/KAT. | | | | | | |
| $\text{C}_2\text{H}_5\text{FO}^+$ $\text{CH}_2\text{FCH}_2\text{OH}$ | (10.66) | (146) | (612) | -100 | -417 | *EST | 371-62-0 |
| | IP from 83KOP/MOL. | | | | | | |
| $\text{C}_2\text{H}_5\text{F}_2\text{N}^+$ $\text{CF}_2\text{HCH}_2\text{NH}_2$ | (9.4) | (326) | (1366) | 110 | 462 | *EST | 430-67-1 |
| | IP from 79AUE/BOW. | | | | | | |
| $\text{C}_2\text{H}_5\text{F}_2\text{O}^+$ $\text{CF}_2\text{HCH}_2\text{OH}_2$ | | 34 | 144 | | | | |
| | From proton affinity of $\text{CF}_2\text{HCH}_2\text{OH}$ (RN 359-13-7). PA = 176.2 kcal/mol, 737 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_5\text{F}_3\text{N}^+$ $\text{CF}_3\text{CH}_2\text{NH}_3$ | | -4 | -18 | | | | |
| | From proton affinity of $\text{CF}_3\text{CH}_2\text{NH}_2$ (RN 753-90-2). PA = 202.5 kcal/mol, 847 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_5\text{I}^+$ $\text{C}_2\text{H}_5\text{I}$ | 9.346 | 213.3 | 891.9 | -2.2±0.2 | -9.0±0.9 | 77PED/RYL | 75-03-6 |
| | | 216.9 | 907.8 | 1.6 | 6.9 | | |
| | See: 78LIA/AUS, 83OHN/IMA, 81KIM/KAT. | | | | | | |

Table 1. Positive Ion Table - Continued


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|--|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_5\text{IO}^+$ gauche- $\text{ICH}_2\text{CH}_2\text{OH}$ | 9.73 | (186) | (778) | -38 | -161 | *EST | 624-76-0 |
| trans- $\text{ICH}_2\text{CH}_2\text{OH}$ | 9.60 | (183) | (765) | -38 | -161 | *EST | 624-76-0 |
| $\text{C}_2\text{H}_5\text{N}^+$ $\text{CH}_2=\text{NCH}_3$ | (9.4) IP is onset of photoelectron band. See also: 86WER. | (234) | (979) | 17 | 72 | 69BEN/CRU | 1761-67-7 |
| $\text{CH}_3\text{CH}=\text{NH}$ | (9.6) IP is onset of photoelectron band (86LAF/GON). | (222) | (930) | 2±4 | 8±17 | 79ELL/EAD | 20729-41-3 |
| $\text{CH}_2=\text{CHNH}_2$ | (8.20) IP from 84ALB/ALL2. | (196) | (820) | 7 | 29 | 81ELL/DIX | 593-67-9 |
|  | 9.2±0.1 See also: 82BIE/ASB. | 242 | 1014 | 30.2±0.2 | 126.5±0.9 | 77PED/RYL | 151-56-4 |
| $\text{C}_2\text{H}_5\text{NO}^+$ CH_3CONH_2 | 9.65±0.03 See also: 81ASB/SVE. | 165 | 693 | -57.0±0.2 | -238.3±0.8 | 77PED/RYL | 60-35-5 |
| (E)- $\text{CH}_3\text{CH}=\text{NOH}$ | (10.0) IP is onset of photoelectron band. | (226) | (945) | -4.7±2 | -20±8 | 69BEN/CRU | 107-29-9 |
| HCONHCH_3 | 9.79 See also: 81KIM/KAT. | (181) | (758) | -45±0.7 | -187±3 | *EST | 123-39-7 |
| $\text{C}_2\text{H}_5\text{NO}_2^+$ $\text{NH}_2\text{CH}_2\text{COOH}$ | 8.8 See also: 83CAN/HAM. | 109 | 458 | -93±1 | -391±5 | 77NGA/SAB | 56-40-6 |
| $\text{C}_2\text{H}_5\text{NO}_2$ | 10.88±0.05 See also: 81KIM/KAT. | 226.5 | 947.5 | -24.4±0.1 | -102.2±0.6 | 77PED/RYL | 79-24-3 |
| $\text{C}_2\text{H}_5\text{ONO}$ | (10.53±0.01) | (218) | (913) | -25 | -103 | 74BAT/CHR | 109-95-5 |
| $\text{C}_2\text{H}_5\text{NO}_3^+$ $\text{C}_2\text{H}_5\text{ONO}_2$ | (11.22) | (222) | (928) | -36.8±0.2 | -154.1±1.0 | 77PED/RYL | 625-58-1 |
| $\text{C}_2\text{H}_5\text{NS}^+$ CH_3CSNH_2 | 8.33 | 194 | 814 | 2±0.2 | 10±1 | 82TOR/SAB2 | 62-55-5 |
| $\text{C}_2\text{H}_5\text{N}_2^+$ NCCH_2NH_3 | | 194 | 812 | From proton affinity of NCCH_2NH_2 (RN 540-61-4). PA = 197.4 kcal/mol, 826 kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

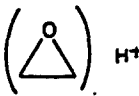

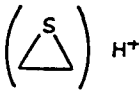
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_5\text{O}^+$ CH_3CHOH | 6.7 | 139 | 583 | -16±1 | -66±4 | 82MCM/GOL | 17104-36-8 |
| | | $\Delta_f H(\text{Ion})$ from proton affinity of acetaldehyde (RN 75-07-0). PA = 186.6 kcal/mol, 781 kJ/mol. The IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See also: 82MAC, 84LOS/HOL. | | | | | |
| CH_3OCH_2 | 6.94 | (157) (165) | (657) (690) | -3±1 | -13±4 | 82MCM/GOL | 16520-04-0 |
| | | $\Delta_f H(\text{Ion})$ at 0 K from appearance potential determination (82MAC, 84BUT/HOL). See also: 84BOW/MAC. | | | | | |
| $\text{CH}_2 = \text{CHOH}_2$ | | 148 | 619 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (82BUR/TER2). | | | | | |
|  | | 165 | 691 | | | | |
| | | From proton affinity of oxirane (RN 75-21-8). PA = 187.9 kcal/mol, 786 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_5\text{O}_2^+$ $\text{CH}_3\text{C}(\text{OH})_2$ | | 72 | 302 | | | | |
| | | From proton affinity of CH_3COOH (RN 64-19-7). See also: 84HOL/LOS. 85AUD/MIL. PA = 190.2 kcal/mol, 796. kJ/mol. | | | | | |
| $\text{HC}(\text{OH})\text{OCH}_3$ | | 92 | 386 | | | | |
| | | From proton affinity of HCOOCH_3 (RN 107-31-3). PA = 188.4 kcal/mol, 788. kJ/mol. (86KNI/FRE, 84LIA/LIE). | | | | | |
| $\text{C}_2\text{H}_5\text{P}^+$  | (9.4±0.1) | (200) | (838) | -16±0.5 | -69±2 | *EST | 6569-82-0 |
| $\text{C}_2\text{H}_5\text{S}^+$ CH_3CHSH | | 197 200 | 823 836 | | | | 58794-14-2 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (83BUT/BAE). | | | | | |
| CH_3SCH_2 | | (194) | (812) | | | | 31533-72-9 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations. See also: 83ERM/AKO. 0 K values. | | | | | |
|  | | 191 | 798 | | | | |
| | | From proton affinity of thiirane (RN 420-12-2). PA = 194.6 kcal/mol, 814 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

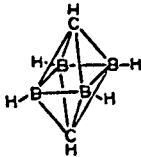
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_2H_6^+ | C_2H_6 | 11.52±0.01 | 245.6 | 1028 | -20.1±0.05 | -84.0±0.2 | 77PED/RYL | 74-84-0 |
| | | | 249.3 | 1043 | -16.4 | -68.4 | | |
| See also: 81KIM/KAT, 84CHA/HIL. | | | | | | | | |
| $\text{C}_2\text{H}_6\text{BBr}^+$ | $(\text{CH}_3)_2\text{BBr}$ | 10.25 | 192 | 804 | -44 | -185 | 82HOL/SMI | 5158-50-9 |
| $\text{C}_2\text{H}_6\text{BCl}^+$ | $(\text{CH}_3)_2\text{BCl}$ | (10.2) | (173) | (725) | -62 | -259 | 82HOL/SMI | 1803-36-7 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_2\text{H}_6\text{BCl}_2\text{N}^+$ | $(\text{CH}_3)_2\text{NBCl}_2$ | 9.56 | 125 | 521 | -96±1 | -401±4 | 77PED/RYL | 1113-31-1 |
| $\text{C}_2\text{H}_6\text{B}_4^+$ |  | (9.77) | (236.3) | (988.6) | 11.0±2.9 | 45.9±12.1 | 85GAL/TAM | 20693-67-8 |
| $\text{C}_2\text{H}_6\text{Br}^+$ | $\text{C}_2\text{H}_5\text{BrH}$ | | (180) | (753) | | | | |
| From proton affinity of $\text{C}_2\text{H}_5\text{Br}$ (RN 74-96-4). PA = -171 kcal/mol, -715 kJ/mol. | | | | | | | | |
| $\text{C}_2\text{H}_6\text{BrSi}^+$ | $(\text{CH}_3)_2\text{SiBr}$ | | 146 | 612 | | | | |
| | | | 151 | 633 | | | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination (84SZE/BAE). | | | | | | | | |
| $\text{C}_2\text{H}_6\text{Cd}^+$ | $(\text{CH}_3)_2\text{Cd}$ | (8.56±0.02) | (223) | (932) | 25.3±0.3 | 105.8±1.3 | 77PED/RYL | 506-82-1 |
| $\text{C}_2\text{H}_6\text{Cl}^+$ | $\text{C}_2\text{H}_5\text{ClH}$ | | 170 | 711 | | | | |
| From proton affinity of $\text{C}_2\text{H}_5\text{Cl}$ (RN 75-00-3). PA = 169 kcal/mol, 707 kJ/mol. | | | | | | | | |
| | CH_3ClCH_3 | | (177) | (743) | | | | 24400-15-5 |
| Derived (85SHA/HOJ). | | | | | | | | |
| $\text{C}_2\text{H}_6\text{ClN}^+$ | $(\text{CH}_3)_2\text{NCl}$ | 8.75 | (221) | (925) | 19 | 81 | *EST | 1585-74-6 |
| $\text{C}_2\text{H}_6\text{ClP}^+$ | $(\text{CH}_3)_2\text{PCl}$ | (8.7) | (163) | (681) | -38 | -158 | *EST | 811-62-1 |
| IP is onset of photoelectron band (82LEV/LIA, 86BOC/BAN). | | | | | | | | |
| $\text{C}_2\text{H}_6\text{Cl}_2\text{NOP}^+$ | $(\text{CH}_3)_2\text{NPOCl}_2$ | (9.5) | (86) | (361) | -133 | -556 | *EST | 677-43-0 |
| IP is onset of photoelectron band. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|----------------|------------------------------|-------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_6\text{Cl}_2\text{NP}^+$ (CH_3) ₂ NP ₂ Cl ₂ | (8.9) IP is onset of photoelectron band. | (171) | (716) | -34 | -143 | *EST | 683-85-2 |
| $\text{C}_2\text{H}_6\text{Cl}_2\text{Si}^+$ (CH_3) ₂ SiCl ₂ | (10.7) IP is onset of photoelectron band. | (137) | (574) | -109 | -458 | 81BEL/PER | 75-78-5 |
| $\text{C}_2\text{H}_6\text{Cl}_2\text{Sn}^+$ (CH_3) ₂ SnCl ₂ | (10.43) | (174) | (727) | -67 | -279 | *EST | 753-73-1 |
| $\text{C}_2\text{H}_6\text{F}^+$ CH ₃ FCH ₃ | | (147) | (614) | | | | |
| | | $\Delta_f H(\text{Ion})$ derived from results of 86HOV/MCM. | | | | | |
| $\text{C}_2\text{H}_5\text{FH}$ | | 138 | 577 | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{F}$ (RN 75-02-5). PA = 165 kcal/mol, 690 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_6\text{FN}^+$ CH ₂ FCH ₂ NH ₂ | (9.1) IP from 79AUE/BOW. | (155) | (650) | -55 | -229 | *EST | 406-34-8 |
| $\text{C}_2\text{H}_6\text{FP}^+$ (CH_3) ₂ PF | (8.8) IP is onset of photoelectron band. | (112) | (468) | -91 | -381 | *EST | 507-15-3 |
| $\text{C}_2\text{H}_6\text{FSi}^+$ (CH_3) ₂ SiF | | 86 | 359 | | | | |
| | | From appearance potential (10.70±0.04 eV) of ion in (CH_3) ₃ SiF. | | | | | |
| $\text{C}_2\text{H}_6\text{F}_2\text{N}^+$ CF ₂ HCH ₂ NH ₃ | | 269 | 1124 | | | | |
| | | From proton affinity of CF ₂ HCH ₂ NH ₂ (RN 430-67-1). PA = 207.5 kcal/mol, 868 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_6\text{F}_2\text{Si}^+$ (CH_3) ₂ SiF ₂ | 11.03±0.03 | 42 | 177 | -212 | -887 | 77MUR/BEA | 353-66-2 |
| $\text{C}_2\text{H}_6\text{Hg}^+$ (CH_3) ₂ Hg | (9.10±0.05) | (232) (237) | (972) (991) | 22.5±0.2 27.0 | 94.0±1.0 113.3 | 77PED/RYL | 593-74-8 |
| $\text{C}_2\text{H}_6\text{I}^+$ C ₂ H ₅ IH | | (188) | (785) | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{I}$ (RN 75-03-6). PA = -176 kcal/mol, -736 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_6\text{N}^+$ CH ₂ NHCH ₃ | 5.9 | 166 | 695 | 30 | 126 | 83BUR/CAS | 31277-24-4 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination(81LOS/LAM); IP derived (81GRI/LOS, 83BUR/CAS). | | | | | |
| CH ₃ CHNH ₂ | 5.7 | 157 | 657 | 26 | 109 | 83BUR/CAS | 30208-36-7 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination(81LOS/LAM); IP derived(83BUR/CAS). | | | | | |

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_6\text{N}^+$ $(\text{CH}_3)_2\text{N}$ | (5.17) | (154) | (644) | 35±2 | 145±8 | 82MCM/GOL | 15337-44-7 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurement. IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |
|  H^+ | | 180 | 755 | | | | |
| | From proton affinity of aziridine (RN 151-56-4). PA = 215.7 kcal/mol, 902 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_6\text{NO}^+$ $\text{CH}_3\text{C}(\text{OH})\text{NH}_2$ | | 103 | 429 | | | | |
| | From proton affinity of CH_3CONH_2 (RN 60-35-5). PA = 206.2 kcal/mol, 863 kJ/mol. | | | | | | |
| $\text{HC}(\text{OH})\text{NHCH}_3$ | | 115 | 481 | | | | |
| | From proton affinity of HCONHCH_3 (RN 123-39-7). PA = 205.8 kcal/mol, 861 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_6\text{NO}_2^+$ $\text{NH}_3\text{CH}_2\text{COOH}$ | | 61 | 254 | | | | |
| | From proton affinity of $\text{NH}_2\text{CH}_2\text{COOH}$ (RN 56-40-6). PA = 211.6 kcal/mol, 885 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_5\text{ONHO}$ | | 144 | 602 | | | | |
| | From proton affinity of $\text{C}_2\text{H}_5\text{ONO}$ (RN 109-95-5). PA = 197.3 kcal/mol, 825.5 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_5\text{NOOH}$ | | 157 | 655 | | | | |
| | From proton affinity of $\text{C}_2\text{H}_5\text{NO}_2$ (RN 79-24-3). PA = 184.8 kcal/mol, 773 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_6\text{N}_2^+$ $(\text{E})\text{-CH}_3\text{N}=\text{NCH}_3$ | 8.45±0.05 | 231 | 964 | 36 | 149 | 82PAM/ROG | 4143-41-3 |
| $\text{C}_2\text{H}_6\text{N}_2\text{O}^+$ $(\text{E})\text{-CH}_3\text{NN}(\text{O})\text{CH}_3$ | (9.7) | (238) | (997) | 15 | 61 | *EST | 54168-20-6 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{CH}_3\text{NHCONH}_2$ | (≤9.66) | (≤164) | (≤688) | -58 | -244 | *EST | 598-50-5 |
| $(\text{CH}_3)_2\text{NNO}$ | 8.69 | 200 | 835 | -0.7±2 | -3±8 | 67KOR/PEP | 62-75-9 |
| $\text{C}_2\text{H}_6\text{N}_2\text{O}_2^+$ $(\text{CH}_3)_2\text{NNO}_2$ | (9.53) | (219) | (914) | -1±0.8 | -5±3 | 77PED/RYL | 4164-28-7 |
| $(\text{E})\text{-(CH}_3\text{NO)}_2$ | (≤8.68) | (≤217) | (≤908) | 17±0.2 | 71±1 | 73BAT/MIL | 37765-15-4 |
| $\text{C}_2\text{H}_6\text{O}^+$ $\text{C}_2\text{H}_5\text{OH}$ | 10.47±0.02 | 185.3 | 775.4 | -56.1±0.1 | -234.8±0.2 | 77PED/RYL | 64-17-5 |
| | | 189.5 | 793.1 | -51.9 | -217.1 | | |
| | See also: 82MIS/POK, 72POT/SOR, 80VON/BIE, 84BOW/MAC, 83OHN/IMA, 81KIM/KAT, 80BAC/MOU, 74BET/BAK. | | | | | | |

Table 1. Positive Ion Table - Continued


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|----------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_6\text{O}^+$ (CH_3) ₂ O | 10.025±0.025 | 187.2 191.5 | 783.3 801.0 | -44.0±0.1 -39.7 | -184.0±0.5 -166.3 | 77PED/RYL | 115-10-6 |
| | IP from 84BUT/HOL. See also: 84BOW/MAC, 81KIM/KAT, 80BAC/MOU, 82BIE/ASB. | | | | | | |
| $\text{C}_2\text{H}_4\text{OH}_2$ | | 175 | 732 | | | | 60786-90-5 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (85BUR/HOL). See also: 82HOL/LOS, 82BUR/HOL). The authors propose the structure $\text{C}_2\text{H}_3\text{..H}^+ \text{..OH}_2$ for the ion. See also: 81TER/HEE. | | | | | | |
| $\text{C}_2\text{H}_6\text{OS}^+$ (CH_3) ₂ SO | (9.01) | (172) (176) | (718) (738) | -36.2±0.2 -31.4 | -151.3±0.8 -131.5 | 77PED/RYL | 67-68-5 |
| | See: 81KIM/KAT. | | | | | | |
| $\text{C}_2\text{H}_6\text{O}_2^+$ HOCH ₂ CH ₂ OH | 10.16 | 142 | 593 | -92.6±0.4 | -387.6±1.7 | 77PED/RYL | 107-21-1 |
| | IP from 82HOL/LOS2. See also: 80VON/BIE, 81KIM/KAT. | | | | | | |
| (CH_3O) ₂ | 9.1 | 180 | 752 | -30.0±0.3 | -125.7±1.3 | 77PED/RYL | 690-02-8 |
| | IP is onset of photoelectron band (81KIM/KAT, 82LEV/LIA). | | | | | | |
| $\text{C}_2\text{H}_6\text{O}_2\text{S}^+$ (CH_3) ₂ SO ₂ | (10.3) | (148) (154) | (621) (644) | -89±0.7 -84 | -373±3 -350 | 77PED/RYL | 67-71-0 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_2\text{H}_6\text{O}_3\text{S}^+$ (CH_3O) ₂ SO | (9.9) | (113) | (472) | -115±0.5 | -483±2 | 77PED/RYL | 616-42-2 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_2\text{H}_6\text{P}^+$ | | 158 | 660 | | | | |
|  | From proton affinity of phosphirane (RN 6569-82-0). PA = 191.4 kcal/mol, 801 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_6\text{S}^+$ $\text{C}_2\text{H}_5\text{SH}$ | 9.285±0.005 | 203 207 | 850 867 | -11.1±0.1 -7.0 | -46.3±0.6 -29.5 | 77PED/RYL | 75-08-1 |
| | See also: 83OHN/IMA, 81KIM/KAT. | | | | | | |
| (CH_3) ₂ S | 8.69±0.01 | 191 195 | 801 817 | -9.0±0.1 -5.1 | -37.5±0.5 -21.3 | 77PED/RYL | 75-18-3 |
| | See also: 81KIM/KAT. | | | | | | |
| $\text{C}_2\text{H}_6\text{SSi}^+$ (CH_3) ₂ Si=S | | (203) | (848) | | | | 1111-83-7 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determination (81GUS/VOL). | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|-------------------|------------------------------|-------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_6\text{S}_2^+$ (CH_3S) ₂ | (7.4±0.3) | (165±4) (169) | (690±15) (707) | -5.8±0.2 -1.6 | -24.2±1.0 -6.8 | 77PED/RYL | 624-92-0 |
| Adiabatic ionization potential determined from consideration of dissociation rates; experimentally observed onset of ionization, 8.33 eV, is much higher because of change in the CSSC bond angle upon ionization from 90° to 180°. (83BUT/BAE). See also: 81KIM/KAT. | | | | | | | |
| $\text{C}_2\text{H}_6\text{S}_3^+$ $\text{CH}_3\text{SSSCH}_3$ | (8.73±0.03) | (199) | (831) | -3 | -11 | *EST | 3658-80-8 |
| $\text{C}_2\text{H}_6\text{Sc}^+$ $\text{C}_2\text{H}_5\text{ScH}$ | | (205) | (858) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | | | | | |
| (CH_3) ₂ Sc | | 189 | 791 | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). See also: 86ELK/ARI. | | | | | | | |
| $\text{C}_2\text{H}_4\text{ScH}_2$ | | (218) | (912) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | | | | | |
| $\text{C}_2\text{H}_6\text{Se}^+$ (CH_3) ₂ Se | 8.40±0.01 | (198) | (827) | 4 | 17 | *EST | 593-79-3 |
| IP from 84BOC/AYG, 82LEV/LIA. | | | | | | | |
| $\text{C}_2\text{H}_6\text{Se}_2^+$ (CH_3Se) ₂ | (8.1) | (197) | (826) | 11 | 44 | *EST | 7101-31-7 |
| IP is onset of photoelectron band (84BOC/AYG). | | | | | | | |
| $\text{C}_2\text{H}_6\text{Si}^+$ $\text{CH}_2 = \text{CHSiH}_3$ | 10.1 | 234 | 978 | 1±3 | 4±13 | 80TEL/RAB | 7291-09-0 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_2\text{H}_6\text{Zn}^+$ (CH_3) ₂ Zn | (9.00±0.02) | (220) | (919) | 12.1±0.3 | 50.6±1.3 | 77PED/RYL | 544-97-8 |
| C_2H_7^+ C_2H_7 | | 202 | 845 | | | | |
| From proton affinity of C_2H_6 (RN 74-84-0). See also: 85MCM/KEB. PA = 143.6 kcal/mol, 601 kJ/mol. | | | | | | | |
| $\text{C}_2\text{H}_7\text{As}^+$ (CH_3) ₂ AsH | (8.1) | (194) | (813) | 7 | 31 | *EST | 593-57-7 |
| IP is onset of photoelectron band (82ELB/DIE). | | | | | | | |
| $\text{C}_2\text{H}_7\text{BO}_2^+$ (CH_3O) ₂ BH | (9.7±1.0) | (85) | (355) | -138.8±0.4 | -580.7±1.7 | 77PED/RYL | 4542-61-4 |

Table 1. Positive Ion Table - Continued

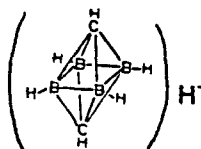
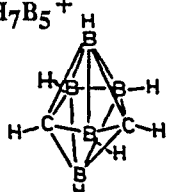
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_7\text{B}_4^+$  | | (170) | (710) | | | | |
| | | From proton affinity of 1,6-dicarbahexaborane(6) (RN 20693-67-8). PA = 207. kcal/mol, 866. kJ/mol. | | | | | |
| $\text{C}_2\text{H}_7\text{B}_5^+$  | 10.54 | 240.5 | 1006.3 | -2.5±2.6 | -10.6±10.9 | 85GAL/TAM | 20693-69-0 |
| $\text{C}_2\text{H}_7\text{ClO}^+$ (CH_3) ₂ OHC1 | (10.4) | (167) | (698) | -73 | -305 | 82TN270 | 24521-77-5 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_2\text{H}_7\text{FN}^+$ $\text{CH}_2\text{FCH}_2\text{NH}_3$ | | 99 | 413 | | | | |
| | | From proton affinity of $\text{CH}_2\text{FCH}_2\text{NH}_2$ (RN 406-34-8). PA = 212.3 kcal/mol, 888 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_7\text{Hg}^+$ (CH_3) ₂ HgH | | (202) | (846) | | | | |
| | | From proton affinity of CH_3HgCH_3 (RN 593-74-8). PA = -186 kcal/mol, -778 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_7\text{N}^+$ $\text{C}_2\text{H}_5\text{NH}_2$ | 8.86±0.02 | 193 | 807 | -11.3±0.2 | -47.5±0.7 | 77PED/RYL | 75-04-7 |
| | | See also: 83OHN/IMA, 81KIM/KAT. | | | | | |
| (CH_3) ₂ NH | 8.23±0.08 | 185 | 776 | -4.4±0.1 | -18.5±0.4 | 77PED/RYL | 124-40-3 |
| | | See also: 81KIM/KAT. | | | | | |
| $\text{C}_2\text{H}_7\text{NO}^+$ $\text{NH}_2\text{CH}_2\text{CH}_2\text{OH}$ | 8.96 | 158 | 662 | -48 | -202 | 77REI/PRA | 141-43-5 |
| | | IP from 83KOP/MOL, 83MOL/PIK3, in agreement with onset of photoelectron band (81KIM/KAT). | | | | | |
| $\text{CH}_3\text{NHOCH}_3$ | 8.92 | (197) | (824) | -9 | -37 | *EST | 1117-97-1 |
| | | IP from 83MOL/PIK. | | | | | |
| $\text{C}_2\text{H}_7\text{N}_2^+$ $\text{CH}_3\text{NNHCH}_3$ | | 194 | 813 | | | | |
| | | From proton affinity of (E)- $\text{CH}_3\text{N}=\text{NCH}_3$ (RN 4143-41-3). PA = 206.9 kcal/mol, 866 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_7\text{O}^+$ $\text{C}_2\text{H}_5\text{OH}_2$ | | 121 | 507 | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{OH}$ (RN 64-17-5). PA = 188.3 kcal/mol, 788 kJ/mol. | | | | | |
| (CH_3) ₂ OH | | 130 | 542 | | | | |
| | | From proton affinity of (CH_3) ₂ O (RN 115-10-6). PA = 192.1 kcal/mol, 804 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

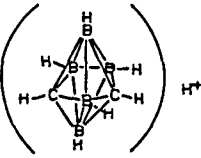
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_7\text{OS}^+$ (CH_3) ₂ SOH | | 118 | 495 | | | | |
| | | From proton affinity of (CH_3) ₂ SO (RN 67-68-5). PA = 211.3 kcal/mol, 834 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_7\text{O}_3\text{P}^+$ (CH_3O) ₂ PHO | (10.53) | (43) | (179) | (-200) | (-837) | *EST | 868-85-9 |
| | IP from 80ZVE/VIL. | | | | | | |
| $\text{C}_2\text{H}_7\text{P}^+$ (CH_3) ₂ PH | 8.47±0.07 | (181) | (757) | -14 | -60 | *EST | 676-59-5 |
| | See also: 82COW/KEM. | | | | | | |
| $\text{C}_2\text{H}_7\text{S}^+$ $\text{C}_2\text{H}_5\text{SH}_2$ | | 164 | 686 | | | | |
| | From proton affinity of $\text{C}_2\text{H}_5\text{SH}$ (RN 75-08-1). PA = 190.8 kcal/mol, 798 kJ/mol. | | | | | | |
| (CH_3) ₂ SH | | 156 | 653 | | | | |
| | From proton affinity of (CH_3) ₂ S (RN 75-18-3). PA = 200.6 kcal/mol, 839 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_7\text{S}_2^+$ $\text{CH}_3\text{SSHCH}_3$ | | (164) | (686) | | | | |
| | From proton affinity of CH_3SSCH_3 (RN 624-92-0). PA = -196 kcal/mol, -820 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_8\text{B}_5^+$ | | (195) | (816) | | | | |
|  | | From proton affinity of 2,4-dicarbaheptaborane(7) (RN 20693-69-0). PA = 168. kcal/mol, 703. kJ/mol. | | | | | |
| $\text{C}_2\text{H}_8\text{N}^+$ $\text{C}_2\text{H}_5\text{NH}_3$ | | 137 | 574 | | | | |
| | From proton affinity of $\text{C}_2\text{H}_5\text{NH}_2$ (RN 75-04-7). PA = 217.0 kcal/mol, 908. kJ/mol. | | | | | | |
| (CH_3) ₂ NH ₂ | | 141 | 588 | | | | |
| | From proton affinity of (CH_3) ₂ NH (RN 124-40-3). PA = 220.6 kcal/mol, 923. kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_8\text{NO}^+$ $\text{H}_3\text{N}(\text{CH}_2)_2\text{OH}$ | | 96 | 402 | | | | |
| | From proton affinity of $\text{NH}_2(\text{CH}_2)_2\text{OH}$ (RN 141-43-5). PA = 221.3 kcal/mol, 926. kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_8\text{N}_2^+$ $\text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$ | (8.6) | (194) | (812) | -4.3±0.5 | -17.8±2.1 | 77PED/RYL | 107-15-3 |
| | IP is onset of photoelectron band (81KIM/KAT). | | | | | | |
| (CH_3) ₂ NNH ₂ | 7.28±0.04 | 188 | 786 | 20±0.5 | 84±2 | 77PED/RYL | 57-14-7 |
| | IP from charge transfer equilibrium constant determination. Reference standard: IP ($\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2$) = 7.12 eV (84MAU/NEL). See also: 81KIM/KAT. | | | | | | |

Table 1. Positive Ion Table - Continued

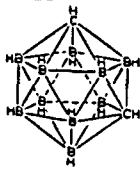
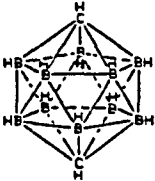
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|-----------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_2\text{H}_8\text{N}_2^+$ $\text{C}_2\text{H}_5\text{NHNH}_2$ | | | | 16±0.2 | 69±1 | *EST | 624-80-6 |
| | | A value of 8.12 eV has been reported for the adiabatic IP of this compound. Values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. | | | | | |
| $(\text{CH}_3\text{NH})_2$ | | | | 22±1 | 92±4 | 77PED/RYL | 540-73-8 |
| | | Values of 7.75 and 8.22 eV have been reported for the adiabatic IP of this compound. Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 81KIM/KAT. | | | | | |
| $\text{C}_2\text{H}_8\text{P}^+$ $(\text{CH}_3)_2\text{PH}_2$ | | 134 | 559 | | | | |
| | | From proton affinity of $(\text{CH}_3)_2\text{PH}$ (RN 676-59-5). PA = 216.3 kcal/mol, 905 kJ/mol. | | | | | |
| $\text{C}_2\text{H}_8\text{Si}^+$ $\text{C}_2\text{H}_5\text{SiH}_3$ | (10.18±0.05) | (262) | (1095) | 27±3 | 113±13 | 80TEL/RAB | 2814-79-1 |
| $(\text{CH}_3)_2\text{SiH}_2$ | 10.3 | 215 | 899 | -23±1 | -95±4 | 86DON/WAL | 1111-74-6 |
| $\text{C}_2\text{H}_9\text{N}_2^+$ $\text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_3$ | | 135 | 567 | | | | |
| | | From proton affinity of $\text{H}_2\text{NCH}_2\text{CH}_2\text{NH}_2$ (RN 107-15-3). PA = 225.9 kcal/mol, 945 kJ/mol. | | | | | |
| $(\text{CH}_3)_2\text{NHNH}_2$ | | 166 | 694 | | | | |
| | | From proton affinity of $(\text{CH}_3)_2\text{NNH}_2$ (RN 57-14-7). PA = 219.9 kcal/mol, 920 kJ/mol (84MAU/NEL). | | | | | |
| $\text{C}_2\text{H}_{10}\text{BN}^+$ $((\text{CH}_3)_2\text{NH})(\text{BH}_3)$ | (9.39±0.01) | (202) | (847) | -14±1 | -59±4 | 80TEL/RAB | 74-94-2 |
| $\text{C}_2\text{H}_{12}\text{B}_{10}^+$ | | | | | | | |
|  | (10.19) | (191) | (800) | -44±2 | -183±8 | 82PIL/SKI | 16986-24-6 |
|  | (10.2) | (175) | (733) | -60±2 | -251±8 | 82PIL/SKI | 20644-12-6 |
| | | IP is onset of photoelectron band. | | | | | |
| C_2I_2^+ $\text{IC}\equiv\text{CI}$ | (9.03) | (269.57) | (1127.90) | 61.34 | 256.64 | 84DEW/HEA | 624-74-8 |

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|------------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_2La^+ LaC ₂ | (5.4±0.3) | (266) (266) | (1113) (1112) | 141±2 141 | 592±6 591 | 81GIN/PEL | 12071-15-7 |
| C_2N^+ CCN | 12.0 | (410) | (1715) | 133 | 556 | 85JANAF | 12327-12-7 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (83SMI, 85HAR/MCI). IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |
| CNC | | (387) | (1620) | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (85HAR/MCI). | | | | | | |
| C_2N_2^+ NCCN | 13.37±0.01 | 381.6 381.1 | 1596.7 1594.8 | 73.3±0.2 72.8 | 306.7±0.7 304.8 | 77PED/RYL | 460-19-5 |
| | See also: 83SMI. | | | | | | |
| $\text{C}_2\text{N}_2\text{O}^+$ NCNCO | (11.49±0.02) | (296) | (1238) | 31 | 129 | *EST | 22430-66-6 |
| $\text{C}_2\text{N}_2\text{S}_2^+$ (SCN) ₂ | (10.5) | (326) | (1363) | 84±1 | 350±6 | 77PED/RYL | 505-14-6 |
| | IP is onset of photoelectron band. | | | | | | |
| C_2Sc^+ C ₂ Sc | 7.7±0.2 | 325 324 | 1360 1357 | 147±3 147 | 617±12 614 | 81HAQ/GIN | 12175-91-6 |
| | See also: 81HAQ/GIN. | | | | | | |
| C_2Si^+  | (10.2±0.5) | (382) (381) | (1599) (1594) | 147 146 | 615 610 | 82TN270 | 12071-27-1 |
| C_2Th^+ C ₂ Th | (6.4±0.5) | (321) | (1341) | 173 | 724 | 82TN270 | 12071-31-7 |
| C_2V^+ C ₂ V | | (335) | (1401) | | | | |
| | $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM). 0 K value. | | | | | | |
| C_2Y^+ C ₂ Y | 6.7±0.3 | 297 296 | 1243 1240 | 143 142 | 597 594 | 82TN270 | 12071-35-1 |
| C_3^+ C ₃ | (12.1±0.3) | (479) | (2004) | 200±4 | 837±17 | 83RAK/BOH | 12075-35-3 |

Table 1. Positive Ion Table - Continued

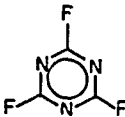
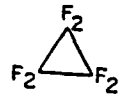
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|------------------------------------|---|---|--|--------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_3BrN^+ | $\text{BrC}\equiv\text{CCN}$ | (10.71±0.02) See also: 84KUH/MAI. | (350) | (1466) | 103±5 | 433±20 | 79BUC/VOG | 3114-46-3 |
| C_3CIN^+ | $\text{ClC}\equiv\text{CCN}$ | 10.95±0.02 | 334 | 1396 | 81±5 | 339±20 | 79BUC/VOG | 2003-31-8 |
| C_3FN^+ | $\text{CF}\equiv\text{CCN}$ | (11.51±0.02) | (305) | (1278) | 40±12 | 167±50 | 79BUC/VOG | 32038-83-8 |
| $\text{C}_3\text{F}_3\text{N}^+$ | $\text{CF}_2=\text{CFCN}$ | (10.6±0.1) | (139) | (584) | -105±0.7 | -439±3 | 71JANAF | 433-43-2 |
| $\text{C}_3\text{F}_3\text{N}_3^+$ |  | (11.3) IP is onset of photoelectron band (81ASB/SVE). | (131) | (548) | -129 | -542 | *EST | 675-14-9 |
| C_3F_4^+ | $\text{CF}_2=\text{C}=\text{CF}_2$ | (10.88) | (109) | (456) | -142 | -594 | 86SMA | 461-68-7 |
| C_3F_6^+ | $\text{CF}_3\text{CF}=\text{CF}_2$ | 10.60±0.03 IP from 81BER/BOM. | -24 | -102 | -269 | -1125 | 75CHE/ROD | 116-15-4 |
| |  | 11.18±0.03 IP from 81BER/BOM. | 24 | 101 | -234 | -978 | 81BOM/BER | 931-91-9 |
| $\text{C}_3\text{F}_6\text{O}^+$ | $(\text{CF}_3)_2\text{CO}$ | (11.44) | (-70) | (-293) | -334 | -1397 | 72GOR | 684-16-2 |
| C_3F_8^+ | C_3F_8 | 13.38 | -118 | -492 | -426±2 | -1783±7 | 77PED/RYL | 76-19-7 |
| $\text{C}_3\text{F}_9\text{N}^+$ | $(\text{CF}_3)_3\text{N}$ | 11.7 IP is onset of photoelectron band (82ELB/DIE, 82BUR/PAW). | (-168) | (-703) | -438 | -1832 | *EST | 432-03-1 |
| C_3H^+ | HCCC | | (381) | (1593) | | | | |
| | | | From proton affinity of C_3 (RN 12075-35-3). PA = -185 kcal/mol, -774 kJ/mol. | | | | | |
| C_3HF_3^+ | $\text{CF}_3\text{C}=\text{CH}$ | (11.96±0.02) See also: 81BIE/ASB. | (177) | (741) | -99 | -413 | 86SMA | 661-54-1 |

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|-----------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{HF}_5\text{N}^+$ $\text{C}_2\text{F}_5\text{CNH}$ | | -21 | -86 | | | | |
| | | From proton affinity of $\text{C}_2\text{F}_5\text{CN}$ (RN 422-04-8). PA = 167.1 kcal/mol, 699 kJ/mol. | | | | | |
| $\text{C}_3\text{HF}_6\text{O}^+$ $(\text{CF}_3)_2\text{COH}$ | | -118 | -495 | | | | |
| | | From proton affinity of $(\text{CF}_3)_2\text{CO}$ (RN 684-16-2) (85MCM/KEB, 85MCM/KEB2) re-evaluated relative to CO standard (84LIA/LIE). PA = 150.0 kcal/mol, 628. kJ/mol. | | | | | |
| $\text{C}_3\text{HF}_8\text{N}^+$ $(\text{CF}_3)_2\text{NCHF}_2$ | (11.7) | (-110) | (-461) | -380 | -1590 | *EST | 73563-15-2 |
| | IP from 82BUR/PAW. | | | | | | |
| C_3HN^+ $\text{HC}\equiv\text{CCN}$ | 11.64±0.01 | 352 | 1474 | 84 | 351 | 85HAR | 1070-71-9 |
| C_3HNO^+ $\text{NCCH}=\text{C}=\text{O}$ | (≤10.07) | (≤256) | (≤1073) | 24 | 101 | *EST | |
| | IP from 81BOC/HIR. | | | | | | |
| C_3HO^+ $\text{HC}\equiv\text{C}-\text{C}=\text{O}$ | | 232 | 971 | | | | |
| | From appearance potential determinations (83TER/HOL). | | | | | | |
| C_3H_2^+ $\text{HC}\equiv\text{CCH}$ | | (330±3) | (1381±12) | | | | 2008-19-7 |
| | $\text{C}_3\text{H}^+ + \text{H}_2 \rightarrow \text{C}_3\text{H}_2^+ + \text{H}$ is -1 kcal/mol endothermic. (84SMI/ADA). | | | | | | |
|  | | 281±3 | 1176±12 | | | | 75123-91-0 |
| | From appearance potentials in CH_3CCX compounds. (84HOL/SZU). | | | | | | |
| $\text{C}_3\text{H}_2\text{F}_2^+$ $\text{CF}_2=\text{C}=\text{CH}_2$ | (9.79±0.03) | (178) | (743) | -48 | -202 | 86SMA | 430-64-8 |
| $\text{C}_3\text{H}_2\text{F}_4\text{O}^+$ $(\text{CHF}_2)_2\text{CO}$ | (10.7) | (15) | (61) | -232±4 | -971±16 | *EST | 360-52-1 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_3\text{H}_2\text{F}_6\text{O}^+$ $\text{CF}_3\text{CH}(\text{OH})\text{CF}_3$ | 11.94 | (-92) | (-384) | -367±2 | -1536±8 | *EST | 920-66-1 |
| | IP from 83KOP/MOL. | | | | | | |
| $\text{C}_3\text{H}_2\text{F}_7\text{N}^+$ $(\text{CF}_2\text{H})_2\text{NCF}_3$ | (11.4) | (-60) | (-250) | -323 | -1350 | *EST | 73551-02-7 |
| | IP from 82BUR/PAW. | | | | | | |
| $\text{C}_3\text{H}_2\text{N}^+$ HCCCNH | | 269 | 1127.5 | | | | |
| | From proton affinity of $\text{HC}\equiv\text{CCN}$ (RN 1070-71-9) (87DEA/MAU, 85KNI/FRE). PA = 180. kcal/mol, 753.5 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

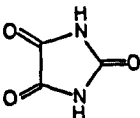

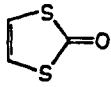
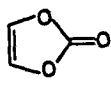
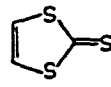
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------|-----------|--------------------------|--------|------------------------------|-----------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_2\text{N}_2^+$ <chem>CH2(CN)2</chem> | | (12.70) | (356) | (1491) | 63.5±0.4 | 265.5±1.5 | 77PED/RYL | 109-77-3 |
| IP from 83MOL/PIK2. | | | | | | | | |
| $\text{C}_3\text{H}_2\text{N}_2\text{O}_3^+$ | | (10.67) | (134) | (559) | -112 | -470 | *EST | 120-89-8 |
|  | | | | | | | | |
| $\text{C}_3\text{H}_2\text{O}^+$ <chem>HC=CCHO</chem> | | (10.8) | (276) | (1157) | 27 | 115 | *EST | 624-67-9 |
| IP from 80VON/BIE. See also: 79CAR/MOU. | | | | | | | | |
| <chem>CH2=C=C=O</chem> | | 9.12±0.05 | (233) | (975) | 23 | 95 | *EST | 61244-93-7 |
| IP from 83TER/HOL. See also: 85MCN/SUF. | | | | | | | | |
|  | | (9.47) | (251) | (1052) | 33±2 | 138±8 | *EST | 2961-80-0 |
| $\text{C}_3\text{H}_2\text{OS}_2^+$ | | (8.6) | (195) | (815) | -3.6±1.2 | -15.0±5.1 | 77PED/RYL | 2314-40-1 |
|  | | | | | | | | |
| IP is onset of photoelectron band (83SCH/SCH). | | | | | | | | |
| $\text{C}_3\text{H}_2\text{O}_2^+$ <chem>HC=CCOOH</chem> | | (10.45) | (213) | (891) | -28 | -117 | *EST | 471-25-0 |
| IP is onset of photoelectron band (80VON/BIE). | | | | | | | | |
| $\text{C}_3\text{H}_2\text{O}_3^+$ | | (9.8) | (126) | (527) | -100±5 | -419±21 | 77PED/RYL | 872-36-6 |
|  | | | | | | | | |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_3\text{H}_2\text{S}_3^+$ | | 8.26 | 251 | 1050 | 60.5±2 | 253±7 | 77PED/RYL | 930-35-8 |
|  | | | | | | | | |

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|---------|----------------------|-----------------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_3H_3^+ $\text{CH}_2\text{C}\equiv\text{CH}$ | 8.68 | 282 | 1179 | 82 | 343 | | 2932-78-7 |
| | HCCCH ⁺ + H ₂ → CH ₂ C≡CH ⁺ + H is 4 kcal/mol endothermic (84SMI/ADA); value derived from appearance potential measurements is 281±3 kcal/mol; 1176 kJ/mol. $\Delta_f H(\text{Neutral}) = \Delta_f H(\text{Ion}) - \text{IP}$. | | | | | | |
|  | 6.6 | 257 | 1075 | 105±4 | 440±17 | 82MCM/GOL | 28933-84-8 |
| | Heat of formation of ion from appearance potential measurements; IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |
| $\text{C}_3\text{H}_3\text{Cl}^+$ $\text{CH}_2=\text{C}=\text{CHCl}$ | (9.57) | (263) | (1102) | 43 | 179 | *EST | 3223-70-9 |
| $\text{CH}_3\text{C}\equiv\text{CCl}$ | 9.82 | (276) | (1153) | 49±4 | 206±15 | *EST | 7747-84-4 |
| $\text{CH}_2\text{ClC}\equiv\text{CH}$ | 10.68 | (285) | (1192) | 39 | 162 | *EST | 624-65-7 |
| | See also: 81ZVE/ERM, 82BIE/ASB. | | | | | | |
| $\text{C}_3\text{H}_3\text{F}_3^+$ CH_2GHCF_3 | (10.9) | (104) | (438) | -147±2 | -614±7 | 77PED/RYL | 677-21-4 32718-30-2 |
| | $\text{C}_3\text{H}_3\text{F}_3^+$ (put in double bond) | | | | | | |
| $\text{C}_3\text{H}_3\text{F}_3\text{O}^+$ CH_3COCF_3 | 10.67 | (52) | (217) | -194 | -812 | *EST | 421-50-1 |
| $\text{C}_3\text{H}_3\text{F}_3\text{O}_2^+$ $\text{HCOOCH}_2\text{CF}_3$ | (11.31) | (5) | (18) | -256 | -1073 | *EST | 32042-38-9 |
| $\text{C}_3\text{H}_3\text{F}_4\text{O}^+$ $(\text{CF}_2\text{H})_2\text{COH}$ | | -32 | -134 | | | | |
| | From proton affinity of $\text{CF}_2\text{HCOCF}_2\text{H}$ (RN 360-52-1). PA = 170 kcal/mol, 711 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_3\text{F}_5\text{O}^+$ $\text{C}_2\text{F}_5\text{CH}_2\text{OH}$ | (11.2) | (-55) | (-229) | -313±0.7 | -1310±3 | 77PED/RYL | 422-05-9 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_3\text{H}_3\text{F}_6\text{N}^+$ $(\text{CF}_2\text{H})_3\text{N}$ | (11.2) | (-7) | (-29) | -265 | -1110 | *EST | 73551-03-8 |
| | IP from 82BUR/PAW. | | | | | | |
| $\text{C}_3\text{H}_3\text{F}_6\text{O}^+$ $(\text{CF}_3)_2\text{CHOH}_2$ | | -180 | -755 | | | | |
| | From proton affinity of $(\text{CF}_3)_2\text{CHOH}$ (RN 920-66-1). PA = 165.0 kcal/mol, 690 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_3\text{N}^+$ CH_2CHCN | 10.91±0.01 | 296 | 1237 | 44 | 184 | 82CHU/NGU | 107-13-1 |
| | See also: 84OHN/MAT, 81KIM/KAT. | | | | | | |

Table 1. Positive Ion Table - Continued







| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_3\text{NO}^+$ $\text{CH}_2=\text{CHNCO}$ | (9.3) | (208) | (872) | -6 | -25 | *EST | 3555-94-0 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{HC}\equiv\text{CCONH}_2$ | (9.85) | (244) | (1023) | 17 | 73 | *EST | 7341-96-0 |
| | IP is onset of photoelectron band (81ASB/SVE). | | | | | | |
|  | 9.93±0.05 | 248 | 1037 | 19 | 79 | 78MCC/HAM | 288-14-2 |
| | IP from 81BOU/HOP. | | | | | | |
|  | (9.6) | (217) | (910) | -4±0.2 | -16±1 | 78MCC/HAM | 288-42-6 |
| $\text{C}_3\text{H}_3\text{NS}^+$ | | | | (1029) | | (1029) | |
|  | (9.55) | (261) | (1090) | 40 | 169 | *EST | 288-16-4 |
|  | (≤9.50) | (≤256) | (≤1070) | 37±2 | 153±10 | *EST | 288-47-1 |
| $\text{C}_3\text{H}_3\text{N}_2^+$ NCCH_2CNH | | 254 | 1061 | | | | |
| | From proton affinity of $\text{CH}_2(\text{CN})_2$ (RN 109-77-3). PA = 175.6 kcal/mol, 735 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_3\text{N}_3^+$ | | | | | | | |
|  | (9.3) | (314) | (1313) | 99 | 416 | *EST | 289-96-3 |
| | IP is onset of photoelectron band (83GLE/SPA). | | | | | | |
|  | (9.2) | (292) | (1222) | 80 | 334 | *EST | 290-38-0 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued


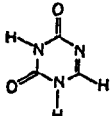
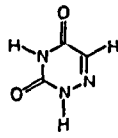

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_3\text{N}_3^+$  | 10.03±0.05 See also: 84SHA/URA. | 285 | 1194 | 54±0.2 | 226±1 | 82BYS | 290-87-9 |
| $\text{C}_3\text{H}_3\text{N}_3\text{O}_2^+$  | (10.59) IP from 81AJO/CAS2. | (181) | (756) | -64 | -266 | *EST | |
|  | 10.18 IP from 81AJO/CAS2, 77ROS/DRA. | (181) | (756) | -54 | -226 | *EST | |
| $\text{C}_3\text{H}_3\text{O}^+$ $\text{CH}_2 = \text{CHCO}$ | (7.0) IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. $\Delta_f H(\text{Ion})$ from appearance potential determination. | (179) | (751) | 17 | 72 | 82MCM/GOL | 72241-20-4 |
| $\text{HC} = \text{CCH}_2\text{O}$ | | 227 | 950 | | | | 92056-62-7 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (84LOS/HOL). | | | | | |
| C_3H_4^+ $\text{CH}_2 = \text{C} = \text{CH}_2$ | 9.69±0.01 See also: 81KIM/KAT, 84MOM/BUR. | 269 | 1126 | 45.6±.2 | 190.6±1 | 77PED/RYL | 463-49-0 |
| | | 271 | 1134 | 47.7 | 199.5 | | |
| $\text{CH}_3\text{C} = \text{CH}$ | 10.36±0.01 See also: 81KIM/KAT, 84MOM/BUR. | 283.5 | 1186.2 | 44.6±.5 | 186.6±2 | 77PED/RYL | 74-99-7 |
| | | 285.5 | 1194.5 | 46.6 | 195.1 | | |
|  | 9.67±0.01 | 289 | 1210 | 66±0.7 | 277±3 | 77PED/RYL | 2781-85-3 |
| $\text{C}_3\text{H}_4\text{F}_3\text{O}^+$ $\text{CH}_3\text{C}(\text{OH})\text{CF}_3$ | | -3 | -11 | | | | |
| | From proton affinity of CH_3COCF_3 (RN 421-50-1). PA = 174.2 kcal/mol, 729 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_4\text{F}_3\text{O}_2^+$ $\text{HC}(\text{OH})\text{CH}_2\text{CF}_3$ | | -70 | -294 | | | | |
| | From proton affinity of $\text{HCOOCH}_2\text{CF}_3$ (RN 32042-38-9). PA = 179.4 kcal/mol, 751 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

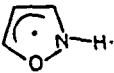




| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_4\text{F}_3\text{O}_2^+$ $\text{CF}_3\text{C}(\text{OH})\text{CH}_3$ | | | -55 | -231 | | | | |
| From proton affinity of $\text{CF}_3\text{COOCH}_3$ (RN 431-47-0). PA = 178.8 kcal/mol, 748 kJ/mol. | | | | | | | | |
| $\text{C}_3\text{H}_4\text{N}^+$ CH_2CHCNH | | (7.37) | 220 | 920 | 50±2 | 209±10 | 82MCM/GOL | 74738-52-6 |
| From proton affinity of $\text{CH}_2=\text{CHCN}$ (RN 107-13-1). PA = 189.7 kcal/mol, 794. kJ/mol. IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | | |
| $\text{C}_3\text{H}_4\text{NO}^+$ CH_3COCNH | | | 181 | 759 | | | | |
| From proton affinity of CH_3COCN (RN 631-57-2). PA = 179.5 kcal/mol, 751. kJ/mol (86MAR/TOP). | | | | | | | | |
|  | | | 182 | 761 | | | | |
| From proton affinity of isooxazole (RN 288-14-2). PA = 202.7 kcal/mol, 848 kJ/mol. | | | | | | | | |
|  | | | 154 | 643 | | | | |
| From proton affinity of oxazole (RN 288-42-6). PA = 208.2 kcal/mol, 871 kJ/mol. | | | | | | | | |
| $\text{C}_3\text{H}_4\text{NO}_2^+$ CH_3COOCNH | | | 138 | 576 | | | | |
| From proton affinity of CH_3OCCN (86MAR/TOP). PA = 179.5 kcal/mol, 751. kJ/mol. | | | | | | | | |
| $\text{C}_3\text{H}_4\text{NS}^+$  | | | 189 | 791 | | | | |
| From proton affinity of thiazole (RN 288-47-1). PA = 213.2 kcal/mol, 892 kJ/mol. | | | | | | | | |
| $\text{C}_3\text{H}_4\text{N}_2^+$  | | 9.25±0.01 | 258 | 1077 | 44±0.5 | 185±2 | 80SAB | 288-13-1 |
| IP from 86MAI/OLE. | | | | | | | | |
|  | | 8.81±0.01 | 238 | 995 | 35±0.5 | 145±2 | 80SAB | 288-32-4 |
| IP from 86MAI/OLE. | | | | | | | | |

Table 1. Positive Ion Table - Continued



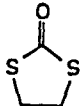

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_4\text{N}_3^+$ | | 219 | 915 | | | | |
|  | | | | | | | |
| | | From proton affinity of 1,3,5-triazine (RN 290-87-9). PA = 201.1 kcal/mol, 841 kJ/mol. | | | | | |
| $\text{C}_3\text{H}_4\text{O}^+$ | | | | | | | |
| $\text{CH}_3\text{CH}=\text{C}=\text{O}$ | 8.95 | 181 | 759 | -25 | -105 | 80DEM/WUL | 6004-44-0 |
| | | IP from 81BOC/HIR. | | | | | |
| $\text{CH}_2=\text{CHCHO}$ | 10.103±0.006 | 215 | 898 | -18 | -77 | 79VAJ/HAR | 107-02-8 |
| | | See also: 80VON/BIE, 81KIM/KAT, 78VAN/OSK. | | | | | |
| $\text{HC}\equiv\text{CCH}_2\text{OH}$ | 10.51 | (253) | (1060) | 11 | 46 | *EST | 107-19-7 |
| | | IP from 83KOP/MOL, 80VON/BIE. | | | | | |
| $\text{HC}\equiv\text{COCH}_3$ | 9.48 | (236) | (989) | 18 | 74 | *EST | 6443-91-0 |
| | | IP from 86HOL/LOS. | | | | | |
|  | (9.1±0.1) | (214) | (894) | 4 | 16 | 76ROD/CHA | 5009-27-8 |
| $\text{C}_3\text{H}_4\text{OS}_2^+$ | | | | | | | |
|  | | | | | | | |
| | (9.2) | (182) | (762) | -30±1 | -126±5 | 77PED/RYL | 2080-58-2 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_3\text{H}_4\text{O}_2^+$ | | | | | | | |
| $\text{CH}_2=\text{CHCOOH}$ | 10.60 | 167 | 699 | -77 | -324 | 80VIL/PER | 79-10-7 |
| | | See also: 78VAN/OSK. | | | | | |
| CH_3COCHO | 9.60±0.06 | 156 | 655 | -65±1 | -271±5 | 77PED/RYL | 78-98-8 |
| | | See also: 81KIM/KAT. | | | | | |
|  | (9.70±0.01) | (156) | (653) | -67.6±0.2 | -282.9±0.8 | 77PED/RYL | 57-57-8 |

Table 1. Positive Ion Table - Continued


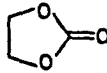
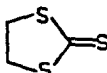
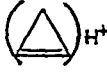
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|------------------------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_4\text{O}_2\text{S}^+$  | (8.6) IP is onset of photoelectron band. | (129) | (538) | -70 | -292 | *EST | 20628-59-5 |
| $\text{C}_3\text{H}_4\text{O}_3^+$  | (10.4) | (117) | (491) | -122±1 | -512±4 | 83CAL | 96-49-1 |
| $\text{CH}_3\text{COCO}^+\text{OH}$ IP is onset of photoelectron band. | 9.9 | 97 | 407 | -131 | -548 | 83TER/WEZ | 127-17-3 |
| $\text{C}_3\text{H}_4\text{S}^+$ $\text{CH}_2=\text{CHCH}=\text{S}$ IP from 82BOC/MOH. | (8.3) | (223) | (934) | 32 | 133 | *EST | 53439-64-8 |
| $\text{CH}_3\text{SC}\equiv\text{CH}$ IP is onset of photoelectron band (81BOC/RIE). | (8.3) | (247) | (1036) | 56 | 235 | *EST | 10152-75-7 |
| $\text{C}_3\text{H}_4\text{S}_3^+$  | (8.40) | (216) | (904) | 22.4±0.5 | 93.8±2.2 | 77PED/RYL | 822-38-8 |
| C_3H_5^+ $\text{CH}_2\text{CH}=\text{CH}_2$ | 8.13 | <u>226.0</u> <u>228.9</u> | <u>945.6</u> <u>957.7</u> | 39 41 | 161 173 | 84HOL/LOS | 1981-80-2 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential measurements. (See also: 84TRA, 82MAC). For IP determination, see also 83KAG/UJS. For $\Delta_f H(\text{Neutral})$, 81TSA recommends 43 kcal/mol, 179 kJ/mol and 82MCM/GOL recommends 40 kcal/mol, 167 kJ/mol. | | | | | |
| CH_3CCH_2 | | 231 | 969 | | | | ? |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (83BUR/HOL). | | | | | |
|  H^+ | 8.18±0.03 IP from 85DYK/ELL. | 255 | 1069 | 66.9 | 279.9 | 82MCM/GOL | 2417-82-5 |
| $\text{C}_3\text{H}_5\text{Br}^+$ $\text{CH}_3\text{CH}=\text{CHBr}$ | (9.30±0.05) | (224) | (938) | 10±1 | 41±4 | 77PED/RYL | 41407-21-0 |
| $\text{CH}_2=\text{CHCH}_2\text{Br}$ See also: 82BIE/ASB. | 10.06 | 243 | 1018 | 11.4±0.6 | 47.7±2.4 | 84TRA | 106-95-6 |

Table 1. Positive Ion Table - Continued

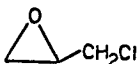
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------------------------|--------|------------------------------|----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_5\text{BrO}^+$ | $\text{CH}_3\text{COCH}_2\text{Br}$ | (9.73) | (181) | (758) | -43 ± 2 | -181 ± 8 | 77PED/RYL | 598-31-2 |
| | | IP from 84OLI/GUE. | | | | | | |
| $\text{C}_3\text{H}_5\text{Cl}^+$ | $\text{CH}_2 = \text{CHCH}_2\text{Cl}$ | 9.9 | 227 | 949 | -1.3 ± 0.6 | -5.6 ± 2.4 | 84TRA | 107-05-1 |
| | | IP is onset of photoelectron band (82BIE/ASB). See also: 82LEV/LIA, 81ZVE/ERM. | | | | | | |
| $\text{C}_3\text{H}_5\text{CIN}^+$ | $\text{ClCH}_2\text{CH}_2\text{CNH}$ | | 188 | 787 | | | | |
| | | From proton affinity of $\text{ClCH}_2\text{CH}_2\text{CN}$ (RN 542-76-7). PA = 187.5 kcal/mol, 784.5 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_5\text{ClO}^+$ | $\text{CH}_3\text{COCH}_2\text{Cl}$ | 9.91 ± 0.03 | (175) | (731) | -54 | -225 | *EST | 78-95-5 |
| | | See also: 84OLI/GUE. | | | | | | |
|  | | (10.2) | (209) | (876) | -26 ± 1 | -108 ± 4 | 77PED/RYL | 106-89-8 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_3\text{H}_5\text{ClO}_2^+$ | $\text{ClH}_2\text{CCOOCH}_3$ | (10.3) | (138) | (577) | -100 | -417 | *EST | 96-34-4 |
| | | IP is onset of photoelectron band (85CAN/HAM). | | | | | | |
| $\text{C}_3\text{H}_5\text{F}^+$ | $\text{CH}_2 = \text{CHCH}_2\text{F}$ | 10.11 | 196 | 819 | -37 | -156 | 82DOL/MED | 818-92-8 |
| $\text{C}_3\text{H}_5\text{FO}^+$ | $\text{CH}_3\text{COCH}_2\text{F}$ | (9.9) | (136) | (572) | -92 | -383 | *EST | 430-51-3 |
| | | See also: 84OLI/GUE. | | | | | | |
| $\text{C}_3\text{H}_5\text{F}_2\text{O}^+$ | $(\text{CFH}_2)_2\text{COH}$ | | 52 | 219 | | | | |
| | | From proton affinity of $\text{CFH}_2\text{COCFH}_2$ (RN 453-14-5). PA = 187 kcal/mol, 782 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_5\text{F}_3\text{O}^+$ | $\text{CF}_3\text{CH}_2\text{OCH}_3$ | 10.53 | (35) | (147) | -208 | -869 | *EST | 460-43-5 |
| | | IP from 83MOL/PIK. | | | | | | |
| $\text{C}_3\text{H}_5\text{I}^+$ | $\text{CH}_2\text{CHCH}_2\text{I}$ | 9.298 | 238.2 | 996.6 | 23.8 | 99.5 | 84TRA | 556-56-9 |
| $\text{C}_3\text{H}_5\text{IO}^+$ | $\text{CH}_3\text{COCH}_2\text{I}$ | (9.3) | (183) | (767) | -31 ± 1 | -130 ± 5 | 77PED/RYL | 3019-04-3 |
| | | IP is onset of photoelectron band (84OLI/GUE). | | | | | | |

Table 1. Positive Ion Table - Continued

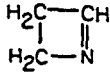

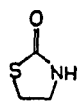
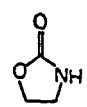
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_5\text{N}^+$ | | | | | | | |
| $\text{C}_2\text{H}_5\text{CN}$ | 11.84±0.02 | 285 | 1194 | 12.3±0.1 | 51.5±0.5 | 82CHU/NGU | 107-12-0 |
| | See also: 82CHE/LAP, 81KIM/KAT. | | | | | | |
| $\text{C}_2\text{H}_5\text{NC}$ | 11.2±0.1 | 292 | 1222 | 33.8±1 | 141.4±4.2 | 77BAG/COL | 624-79-3 |
| | IP from 82CHE/LAP, 77ROS/DRA. | | | | | | |
| (E)- $\text{CH}_2 = \text{CHCH} = \text{NH}$ | (9.65) | (249) | (1043) | 27 | 112 | *EST | 73311-40-7 |
| | IP is onset of photoelectron band (82SCH/SCH). | | | | | | |
|  | (9.30) | (265) | (1108) | 50 | 211 | *EST | 6788-85-8 |
| | IP from 83DAM/BOC. | | | | | | |
|  | (≤9.76±0.22) | (≤300) | (≤1256) | 75 | 314 | *EST | 19540-05-7 |
| $\text{C}_3\text{H}_5\text{NO}^+$ | | | | | | | |
| $\text{C}_2\text{H}_5\text{NCO}$ | (10.1) | (196) | (819) | -37 | -155 | *EST | 109-90-0 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{NCCH}_2\text{OCH}_3$ | 10.75 | (240) | (1002) | -8 | -35 | *EST | 1738-36-9 |
| | IP from 83MOL/PIK. | | | | | | |
| $\text{CH}_2\text{CHCONH}_2$ | 9.5 | (172) | (722) | -47 | -195 | *EST | 79-06-1 |
| | IP is onset of photoelectron band (78VAN/OSK). | | | | | | |
| $\text{C}_3\text{H}_5\text{NOS}^+$ | | | | | | | |
|  | (9.2) | (177) | (743) | -35 | -145 | *EST | 2682-49-7 |
| | IP is onset of photoelectron band (80AND/DEV). | | | | | | |
| $\text{C}_3\text{H}_5\text{NO}_2^+$ | | | | | | | |
|  | (9.6) | (139) | (582) | -82 | -344 | *EST | 497-25-6 |
| | IP is onset of photoelectron band (80AND/DEV). | | | | | | |

Table 1. Positive Ion Table - Continued






| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|----------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_5\text{NSSe}^+$  | 7.3 IP is onset of photoelectron band (80AND/DEV). | (155) | (650) | -13 | -54 | *EST | 63369-86-8 |
| $\text{C}_3\text{H}_5\text{NS}_2^+$  | ≤ 8.25 IP from 80AND/DEV, 82LEV/LIA. | (≤ 161) | (≤ 672) | -30 | -124 | *EST | 96-53-7 |
| $\text{C}_3\text{H}_5\text{N}_2^+$  | | 177 | 740 | | | | |
| | | From proton affinity of imidazole (RN 288-32-4) (86MAU/LIE, 84FLA/MAQ, 86TAF/ANV). PA = 223.4 kcal/mol, 935. kJ/mol. | | | | | |
|  | | 197 | 825 | | | | |
| | | From proton affinity of pyrazole (RN 288-13-1) (86MAU/LIE, 84FLA/MAQ). PA = 212.8 kcal/mol, 890. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_5\text{O}^+$ $\text{C}_2\text{H}_5\text{CO}$ | (5.7) | 141 | 591 | 10 ± 1 | 43 ± 4 | 82MCM/GOL | 15843-24-0 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential measurements (85TRA) and from proton affinity of $\text{CH}_3\text{CH}=\text{CO}$ (RN 6004-44-0). PA = 199.4 kcal/mol, 834 kJ/mol. IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | |
| CH_2CHCHOH | | 153 | 642 | | | | |
| | | From proton affinity of $\text{CH}_2=\text{CHCHO}$ (RN 107-02-8). PA = 193.9 kcal/mol, 811 kJ/mol. | | | | | |
| C_3H_6^+ $\text{CH}_3\text{CH}=\text{CH}_2$ | 9.73 ± 0.02 See also: 81KIM/KAT. | 229 | 959 | 4.8 ± 0.2 | 20.2 ± 0.4 | 77PED/RYL | 115-07-1 |
|  | 9.86 | 240 244 | 1004 1022 | 12.7 ± 2 16.9 | 53.3 ± 0.5 70.9 | 77PED/RYL | 75-19-4 |
| | | IP from 84LIA/BUC. See also: 81KIM/KAT. | | | | | |
| $\text{C}_3\text{H}_6\text{Br}_2^+$ $\text{CH}_2\text{BrCHBrCH}_3$ | 10.1 IP is onset of photoelectron band (81KIM/KAT). | 216 | 903 | -17 ± 0.2 | -71 ± 1 | 77PED/RYL | 78-75-1 |
| $\text{CH}_2\text{BrCH}_2\text{CH}_2\text{Br}$ | ≤ 10.26 | (≤ 220) | (≤ 919) | -17 | -71 | *EST | 109-64-8 |

Table 1. Positive Ion Table - Continued


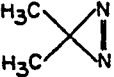
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|---------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_6\text{Cl}_2^+$ $\text{CH}_3\text{CHClCH}_2\text{Cl}$ | (10.87±0.05) | (212) | (886) | -38.9±0.3 | -162.6±0.3 | 77PED/RYL | 78-87-5 |
| $\text{CH}_2\text{ClCH}_2\text{CH}_2\text{Cl}$ | 10.85±0.05 | 212 | 887 | -38±2 | -160±8 | 77PED/RYL | 142-28-9 |
| $\text{C}_3\text{H}_6\text{FO}^+$ $\text{CH}_3\text{C}(\text{OH})\text{CH}_2\text{F}$ | | 82 | 344 | | | | |
| | From proton affinity of $\text{CH}_3\text{COCH}_2\text{F}$ (RN 430-51-3). PA = 192.0 kcal/mol, 803. kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_6\text{F}_2^+$ $(\text{CH}_3)_2\text{CF}_2$ | (11.42±0.02) | (138) | (578) | -125±3 | -524±13 | 82DOL/MED | 420-45-1 |
| $\text{C}_3\text{H}_6\text{F}_3\text{N}^+$ $\text{CF}_3\text{CH}_2\text{CH}_2\text{NH}_2$ | (9.3) IP from 79AUE/BOW | (40) | (166) | -175 | -731 | *EST | 460-39-9 |
| $\text{CF}_3\text{N}(\text{CH}_3)_2$ | (9.2) IP from 79AUE/BOW. | (25) | (104) | -187 | -784 | *EST | 677-41-8 |
| $\text{C}_3\text{H}_6\text{N}^+$ $\text{C}_2\text{H}_5\text{CNH}$ | | 185 | 775 | | | | |
| | From proton affinity of $\text{C}_2\text{H}_5\text{CN}$ (RN 107-12-0). PA = 192.6 kcal/mol, 806 kJ/mol. | | | | | | |
| $\text{C}_2\text{H}_5\text{NCH}$ | | 196 | 819 | | | | |
| | From proton affinity of $\text{C}_2\text{H}_5\text{NC}$ (RN 624-79-3) (86MAU/KAR). PA = 203.7 kcal/mol, 852. kJ/mol. | | | | | | |
| $\text{HCCCH}_2\text{NH}_3$ | | 208 | 870 | | | | |
| | From proton affinity of $\text{HC}\equiv\text{CCH}_2\text{NH}_2$ (RN 2450-71-7). PA = 210.8 kcal/mol, 882 kJ/mol. | | | | | | |
|  H^+ | | (229) | (957) | | | | |
| | From proton affinity of 1-azabicyclo[1.1.0]butane (RN 19540-05-7). PA = (212) kcal/mol, (887) kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_6\text{N}_2^+$ $(\text{CH}_3)_2\text{NC}\equiv\text{N}$ | (9.0) | (241) | (1007) | 33 | 139 | *EST | 1467-79-4 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (≤9.76) | (≤267) | (≤1118) | 42 | 176 | *EST | 5161-49-9 |

Table 1. Positive Ion Table - Continued

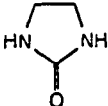



| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|------------|----------------------|--------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_6\text{N}_2\text{O}^+$  | (8.9) IP is onset of photoelectron band (80AND/DEV). | (163) | (683) | -42 | -176 | *EST | 120-93-4 |
| $\text{C}_3\text{H}_6\text{N}_2\text{S}^+$  | 8.15 | 210 | 880 | 22 | 94 | *EST | 96-45-7 |
| $\text{C}_3\text{H}_6\text{N}_2\text{Se}^+$  | (7.0) IP is onset of photoelectron band (80AND/DEV). | (192) | (803) | 31 | 128 | *EST | 33251-51-3 |
| $\text{C}_3\text{H}_6\text{O}^+$ | | | | | | | |
| $\text{C}_2\text{H}_5\text{CHO}$ | 9.953±0.005 See also: 81ELS/ALL, 85TRA, 81KIM/KAT, 77STA/WIE. | 184.7 | 772.9 | -44.8±0.4 | -187.4±1.5 | 77PED/RYL | 123-38-6 |
| $(\text{CH}_3)_2\text{CO}$ | 9.705 See also: 72POT/SOR, 81KIM/KAT, 77STA/WIE. | 171.9 | 719.2 | -51.9±0.1 | -217.2±0.4 | 76CHA/ZWO | 67-64-1 |
| $\text{CH}_2 = \text{CHCH}_2\text{OH}$ | 9.67±0.05 See also: 83BOM/DAN, 82HOL/BUR. | 193 | 809 | -30±0.5 | -124±2 | 77PED/RYL | 107-18-6 |
| (E)- $\text{CH}_3\text{CH} = \text{CHOH}$ | 8.64±0.02 $\Delta_f H(\text{Ion})$ from appearance potential determinations. (82HOL/LOS3, 82HOL/BUR). IP from 84TUR2. $\Delta_f H(\text{Neutral})$ is $\Delta_f H(\text{Ion}) - \text{IP}$. (See 84TUR2). | 159 | 665 | -40 | -169 | 84TUR2 | 57642-95-2 |
| (Z)- $\text{CH}_3\text{CH} = \text{CHOH}$ | 8.70±0.03 $\Delta_f H(\text{Ion})$ from appearance potential determinations. (82HOL/LOS3, 82HOL/BUR). IP from 84TUR2. $\Delta_f H(\text{Neutral})$ is $\Delta_f H(\text{Ion}) - \text{IP}$. (See 84TUR2). | 159 | 665 | -42 | -174 | 84TUR2 | 57642-96-3 |
| $\text{CH}_2 = \text{C}(\text{OH})\text{CH}_3$ | 8.67±0.05 $\Delta_f H(\text{Ion})$ from appearance potential determinations 82HOL/LOS3. (See also: 82LIF2). IP from 84TUR/HAN. | 158 | 661 | -42 | -176 | 84TUR/HAN | 29456-04-0 94324-85-9 |
| $\text{CH}_2 = \text{CHOCH}_3$ | (8.93±0.02) | (182) | (762) | -24±2 | -100±7 | *EST | 107-25-5 |
|  | 9.668±0.005 See also: 79AUE/BOW. | 203.7 | 852.3 | -19.2±0.1 | -80.5±0.6 | 77PED/RYL | 503-30-0 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_6\text{O}^+$ | 10.22±0.02 See also: 81KIM/KAT, 79AUE/BOW. | 213 | 891 | -22.6±0.1 | -94.7±0.6 | 77PED/RYL | 75-56-9 |
| | (9.10) IP from 83BOM/DAN. See also: 82HOL/BUR. | (188) | (785) | -22 | -93 | *EST | 16545-68-9 |
| $\text{C}_3\text{H}_6\text{OS}^+$ $\text{CH}_3\text{C}(=\text{O})\text{SCH}_3$ | (9.5) IP is onset of photoelectron band. | (182) | (761) | -37 | -156 | *EST | 1534-08-3 |
| | (8.1) IP is onset of photoelectron band (83JOR/CAR). | (231) | (966) | 44 | 184 | *EST | 5684-29-7 |
| | (8.5) IP is onset of photoelectron band (83JOR/CAR). | (184) | (769) | -12 | -51 | *EST | 13153-11-2 |
| | (8.3) IP is onset of photoelectron band (83JOR/CAR). | (137) | (572) | -55 | -229 | *EST | 50879-06-6 |
| $\text{C}_3\text{H}_6\text{O}_2^+$ $\text{C}_2\text{H}_5\text{COOH}$ | 10.525±0.003 See also: 81HOL/FIN, 81KIM/KAT. | 136 | 567 | -107±0.5 | -448±2 | 77PED/RYL | 79-09-4 |
| HCOOC_2H_5 | 10.61±0.01 | (153) | (637) | -92 | -387 | *EST | 109-94-4 |
| | 10.27±0.02 See also: 85CAN/HAM. | 139 | 581 | -98.0±0.2 | -410.0±0.8 | 77PED/RYL | 79-20-9 |
| $\text{CH}_2=\text{C}(\text{OH})\text{OCH}_3$ | | 114 | 477 | | | | 4453-91-2 |
| | From appearance potential determination. | | | | | | |
| $\text{CH}_3\text{CH}=\text{C}(\text{OH})_2$ | | 104 | 437 | | | | |
| | From appearance potential determinations. | | | | | | |
| | | (127) | (533) | | | | |
| | Estimated in 86BUR/HOL. | | | | | | |

Table 1. Positive Ion Table - Continued






| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_6\text{O}_2^+$ | | | | | | | |
|  | (≤ 9.86) | (≤ 203) | (≤ 847) | -25 | -104 | *EST | 4362-13-4 |
|  | (9.9) IP is onset of photoelectron band. | (157) | (658) | -71.1 \pm 0.1 | -297.5 \pm 0.6 | 77PED/RYL | 646-06-0 |
| $\text{C}_3\text{H}_6\text{O}_2\text{S}^+$ (CH_3O) ₂ CS | (8.7) IP is onset of photoelectron band. | (121) | (504) | -80 | -335 | *EST | 1115-13-5 |
| $\text{C}_3\text{H}_6\text{O}_3^+$ $\text{CH}_3\text{OCOOCOCH}_3$ | (10.5) IP is onset of photoelectron band. | (103) | (432) | -139 | -581 | *EST | 616-38-6 |
|  | (10.3) IP is onset of photoelectron band. | (126) | (528) | -111.4 \pm 0.1 | -465.9 \pm 0.3 | 77PED/RYL | 110-88-3 |
| $\text{C}_3\text{H}_6\text{O}_3\text{P}^+$ | | | | | | | |
|  | | 25 | 105 | | | | |
| | | From proton affinity of 2,6,7-trioxa-1-phosphabicyclo[2.2.1]heptane (RN 279-53-8). PA = 194.0 kcal/mol, 812. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_6\text{S}^+$ (CH_3) ₂ CS | $\leq 8.60 \pm 0.05$ | ≤ 196 | ≤ 821 | -2 | -9 | 79JOS | 4756-05-2 |
| $\text{CH}_2 = \text{CHCH}_2\text{SH}$ | 9.25 | (228) | (956) | 15 \pm 2 | 64 \pm 9 | *EST | 870-23-5 |
| $\text{CH}_2 = \text{CHSCH}_3$ | 8.2 | (207) | (865) | 18 \pm 0.2 | 74 \pm 1 | *EST | 1822-74-8 |
| | | IP is onset of photoelectron band. | | | | | |
|  | 8.69 | 214.9 | 899.1 | 14.5 | 60.7 | 77PED/RYL | 287-27-4 |
| | | 219.5 | 918.4 | 19.1 | 79.9 | | |
| | Results from 83BUT/BAE2. See also: 79AUE/BOW. | | | | | | |

Table 1. Positive Ion Table - Continued

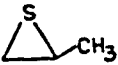




| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_6\text{S}^+$ | | | | | | | |
|  | 8.7 | 212 | 885 | 11±0.5 | 46±2 | 77PED/RYL | 1072-43-1 |
| | IP is onset of photoelectron band. See also: 79AUE/BOW. | | | | | | |
| $\text{C}_3\text{H}_6\text{S}_2^+$ | | | | | | | |
| $\text{CH}_3\text{CSSCH}_3$ | (8.1) | (211) | (882) | 24±3 | 100±13 | *EST | 2168-84-5 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (7.6) | (170) | (712) | -5 | -21 | *EST | 557-22-2 |
| | IP is onset of photoelectron band (80BOC/STE). | | | | | | |
|  | 8.6 | (201) | (840) | 2 | 10 | *EST | 4829-04-3 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_3\text{H}_6\text{S}_3^+$ | | | | | | | |
| $(\text{CH}_3)_2\text{CS}$ | (7.9) | (203) | (851) | 21 | 89 | *EST | 2314-48-9 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (7.7) | (190) | (797) | 13 | 54 | *EST | 291-21-4 |
| | IP is onset of photoelectron band. (81BOC/SCH). | | | | | | |
| C_3H_7^+ | | | | | | | |
| n- C_3H_7 | 8.09±0.01 | 211 | 881 | 24.0±0.5 | 100.5±2.1 | 85TSA | 2143-61-5 |
| | | 214 | 896 | 27 | 115 | | |
| | IP from 85DYK/ELL. See also: 84SCH/HOU. $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 100.5$ kcal/mol. | | | | | | |
| iso- C_3H_7 | 7.36±0.02 | 190.9 | 798.9 | 22.3±0.6 | 93.3±2.5 | 85TSA | 19252-53-0 |
| | | 195.3 | 817.1 | 25.6 | 107.0 | | |
| | Heat of formation of ion from appearance potential measurements (80BAE, 82ROS/BUF, 81TRA/MCL). IP from 85DYK/ELL. See also: 83BRA/BAE2. $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 99$ kcal/mol. PA ($\text{CH}_3\text{CH}=\text{CH}_2$) = 179.5 kcal/mol, 751. kJ/mol. | | | | | | |
|  | | 198.5 | 831 | | | | |
| | From proton affinity of c- C_3H_6 . (RN 75-19-4). PA = 179.8 kcal/mol, 752 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₃H₇Br⁺ | | | | | | | |
| n-C ₃ H ₇ Br | 10.18±0.01 | 214 | 898 | -20.2±0.1 | -84.5±0.5 | 77PED/RYL | 106-94-5 |
| | | 221 | 926 | -13 | -56 | | |
| | See also: 81KIM/KAT. | | | | | | |
| iso-C ₃ H ₇ Br | 10.07±0.01 | 209 | 873 | -23.4±0.2 | -98.3±0.9 | 80TRA | 75-26-3 |
| | | 215 | 901 | -17 | -70 | | |
| | See also: 81KIM/KAT. | | | | | | |
| C₃H₇Cl⁺ | | | | | | | |
| n-C ₃ H ₇ Cl | 10.82±0.03 | 218 | 911 | -31.6±0.1 | -132.4±0.6 | 77PED/RYL | 540-54-5 |
| | See also: 81KIM/KAT. | | | | | | |
| iso-C ₃ H ₇ Cl | 10.78±0.02 | 214 | 895 | -34.6±0.1 | -145.0±0.6 | 80TRA | 75-29-6 |
| | See also: 81KIM/KAT. | | | | | | |
| C₃H₇ClHg⁺ | | | | | | | |
| n-C ₃ H ₇ HgCl | ≤10.15 | ≤213 | ≤891 | -21±2 | -88±8 | 80TEL/RAB | 2440-40-6 |
| | IP from 81BAI/CHI2. | | | | | | |
| iso-C ₃ H ₇ HgCl | ≤9.80 | ≤206 | ≤863 | -20±2 | -83±8 | 80TEL/RAB | 30615-19-1 |
| | IP from 81BAI/CHI2. | | | | | | |
| C₃H₇ClO⁺ | | | | | | | |
| ClCH ₂ OC ₂ H ₅ | 10.30 | (184) | (771) | -53 | -223 | *EST | 3188-13-4 |
| | IP from 83MOL/PIK. | | | | | | |
| C₃H₇F⁺ | | | | | | | |
| n-C ₃ H ₇ F | (11.3) | (192) | (804) | -68±0.5 | -286±2 | 77PED/RYL | 460-13-9 |
| | IP is onset of photoelectron band. | | | | | | |
| iso-C ₃ H ₇ F | (11.08±0.02) | (185) | (776) | -70±0.5 | -293±2 | 77PED/RYL | 420-26-8 |
| C₃H₇F₃N⁺ | | | | | | | |
| CF ₃ NH(CH ₃) ₂ | | -15 | -65 | | | | |
| | From proton affinity of CF ₃ N(CH ₃) ₂ (RN 677-41-8). PA = 193.8 kcal/mol, 811 kJ/mol. | | | | | | |
| CF ₃ CH ₂ CH ₂ NH ₃ | | -20 | -82 | | | | |
| | From proton affinity of CF ₃ CH ₂ CH ₂ NH ₂ (RN 460-39-9). PA = 210.6 kcal/mol, 881 kJ/mol. | | | | | | |
| CF ₃ CH ₂ NH ₂ CH ₃ | | -11 | -47 | | | | |
| | From proton affinity of CF ₃ CH ₂ NHCH ₃ (RN 2730-67-8). PA = 209.8 kcal/mol, 878 kJ/mol. | | | | | | |
| C₃H₇I⁺ | | | | | | | |
| n-C ₃ H ₇ I | 9.269 | 206 | 862 | -7.8±0.4 | -32.5±1.7 | 77PED/RYL | 107-08-4 |
| | | 211 | 884 | -2.4±0.5 | -10.2±2 | | |
| | See: 82ROS/BUF, 83BRA/BAE2, 81KIM/KAT. | | | | | | |

Table 1. Positive Ion Table - Continued



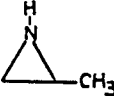
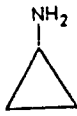
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_7\text{I}^+$ iso- $\text{C}_3\text{H}_7\text{I}$ | 9.175 | 202 207 | 844 865 | -9.9±0.4 -4.8±0.5 | -41.6±1.7 -20.1±2 | 77PED/RYL | 75-30-9 |
| See: 82ROS/BUF, 83BRA/BAE2, 81KIM/KAT. | | | | | | | |
| $\text{C}_3\text{H}_7\text{N}^+$ $\text{CH}_2=\text{CHCH}_2\text{NH}_2$ | 8.76 See also: 79AUE/BOW. | (213) | (893) | 11 | 48 | *EST | 107-11-9 |
|  | (8.3) IP from 79AUE/BOW. | (215) | (898) | 24±1 | 99±4 | *EST | 503-29-7 |
|  | (8.7) IP from 79AUE/BOW. See also: 86CAU/DIV. | (230) | (964) | 30±0.5 | 127±2 | *EST | 1072-44-2 |
|  | (9.0) IP from 79AUE/BOW. | (230) | (961) | 22±1 | 91±6 | *EST | 75-55-8 |
|  | (8.7) IP is onset of photoelectron band (81KIM/KAT). See also: 79AUE/BOW. | (219) | (916) | 18.4±0.1 | 77.0±0.6 | 77PED/RYL | 765-30-0 |
| $\text{C}_3\text{H}_7\text{NO}^+$ $\text{HCON}(\text{CH}_3)_2$ | 9.13±0.02 See also: 82BIE/ASB, 81HEN/ISA. | 165 | 689 | -45.8±0.4 | -191.7±1.7 | 77PED/RYL | 68-12-2 |
| $(\text{CH}_3)_2\text{C}=\text{NOH}$ | (9.1) IP is onset of photoelectron band. | (195) | (815) | -15±3 | -63±12 | *EST | 127-06-0 |
| $\text{CH}_3\text{CONHCH}_3$ | 9.3 IP is onset of photoelectron band. | 158 | 661 | -56 | -236 | *EST | 79-16-3 |
| $\text{C}_3\text{H}_7\text{NO}_2^+$ n- $\text{C}_3\text{H}_7\text{NO}_2$ | 10.81±0.03 See also: 81KIM/KAT. | 220 | 919 | -29.7±0.1 | -124.0±0.6 | 77PED/RYL | 108-03-2 |
| i- $\text{C}_3\text{H}_7\text{NO}_2$ | 10.71±0.05 See also: 81KIM/KAT. | 214 | 894 | -33.2±0.2 | -139.0±0.9 | 77PED/RYL | 79-46-9 |
| n- $\text{C}_3\text{H}_7\text{ONO}$ | (10.34±0.01) | (210) | (879) | -28±1 | -119±4 | 74BAT/CHR | 543-67-9 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|---|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₃H₇NO₂⁺ | | | | | | | |
| i-C ₃ H ₇ ONO | (10.23±0.01) | (204) | (854) | -32±1 | -133±4 | 74BAT/CHR | 541-42-4 |
| H ₂ NCH ₂ CH ₂ COOH | (8.8) | (101) | (425) | -101±0.5 | -424±2 | 83SKO/SAB | 28854-76-4 |
| | | | | | | | IP is onset of photoelectron band (83CAN/HAM). |
| CH ₃ NHCH ₂ COOH | (8.4) | (106) | (443) | -88±0.2 | -367±1 | 78SAB/LAF | 107-97-1 |
| | | | | | | | IP is onset of photoelectron band (83CAN/HAM). |
| L-CH ₃ CH(NH ₂)COOH | 8.88 | 106 | 442 | -99±1 | -415±4 | 77NGA/SAB | 56-41-7 |
| | | | | | | | See also: 83CAN/HAM. |
| NH ₂ COOC ₂ H ₅ | (10.15) | (127) | (533) | -107 | -446 | 75BER/BOU | 51-79-6 |
| | | | | | | | IP is onset of photoelectron band. |
| NH ₂ CH ₂ COOCH ₃ | (9.1) | (121) | (505) | -89 | -373 | *EST | 616-34-2 |
| | | | | | | | IP is onset of photoelectron band (83CAN/HAM). |
| C₃H₇NO₂S⁺ | | | | | | | |
| L-HSCH ₂ CH(NH ₂)COOH | (9.5) | (128) | (534) | -92 | -383 | *EST | 3374-22-9 |
| | | | | | | | IP from 83CAN/HAM. |
| C₃H₇NO₃⁺ | | | | | | | |
| n-C ₃ H ₇ ONO ₂ | (11.07±0.02) | (214) | (894) | -41.6±0.3 | -173.9±1.3 | 77PED/RYL | 627-13-4 |
| L-HOCH ₂ CH(NH ₂)COOH | (8.7) | (67) | (278) | -134 | -561 | *EST | 302-84-1 |
| | | | | | | | IP is onset of photoelectron band(83CAN/HAM). |
| C₃H₇NS⁺ | | | | | | | |
| HCSN(CH ₃) ₂ | (≤8.2) | (≤201) | (≤840) | 12 | 49 | *EST | 758-16-7 |
| | | | | | | | IP from 81HEN/ISA. |
| C₃H₇N₂⁺ | | | | | | | |
| H ₃ N(CH ₂) ₂ CN | | 180 | 755 | | | | |
| | | | | | | | From proton affinity of H ₂ N(CH ₂) ₂ CN (RN 151-18-8). PA = 207.0 kcal/mol, 866. kJ/mol. |
| CH ₃ NH ₂ CH ₂ CN | | 185 | 775 | | | | |
| | | | | | | | From proton affinity of CH ₃ NHCH ₂ CN (RN 5616-32-0). PA = 206.0 kcal/mol, 862. kJ/mol. |
| (CH ₃) ₂ NCNH | | 194 | 811 | | | | |
| | | | | | | | From proton affinity of (CH ₃) ₂ NCN (RN 1467-79-4) (86MAR/TOP). PA = 205.0 kcal/mol, 858 kJ/mol. |
| C₃H₇O⁺ | | | | | | | |
| n-C ₃ H ₇ O | (9.20±0.05) | (202) | (847) | -10 | -41 | 82MCM/GOL | 16499-18-6 |
| i-C ₃ H ₇ O | (9.20±0.05) | (197) | (825) | -15 | -63 | 82MCM/GOL | 3958-66-5 |

Table 1. Positive Ion Table - Continued

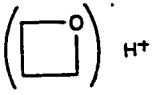
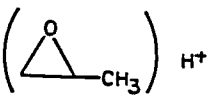
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_7\text{O}^+$ | | | | | | | |
| $\text{C}_2\text{H}_5\text{CHOH}$ | | 131 | 550 | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{CHO}$ (RN 123-38-6). PA = 189.6 kcal/mol, 793. kJ/mol. $\text{C}_3\text{H}_7\text{O}^+$ formed in $n\text{-C}_3\text{H}_7\text{OH}$ with appearance potential of 10.71 eV is probably the same species. See also: 82MAC. | | | | | |
| $(\text{CH}_3)_2\text{COH}$ | | 117 | 490 | | | | |
| | | From proton affinity of $(\text{CH}_3)_2\text{CO}$ (RN 67-64-1). PA = 196.7 kcal/mol, 823 kJ/mol. See also: 84LOS/HOL. | | | | | |
| $\text{C}_2\text{H}_5\text{OCH}_2$ | | (142) | (593) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination. See also: 82MAC. | | | | | |
| $\text{CH}_3\text{CHOCH}_3$ | | 134 | 562 | | | | |
| | | From proton affinity of $\text{CH}_2=\text{CHOCH}_3$ (RN 107-25-5). PA = 207.4 kcal/mol, 868 kJ/mol. See also: 82MAC. | | | | | |
|  H^+ | | 149 | 625 | | | | |
| | | From proton affinity of oxetane (RN 503-30-0). PA = 196.9 kcal/mol, 824 kJ/mol. | | | | | |
|  H^+ | | 148 | 620 | | | | |
| | | From proton affinity of methyloxirane (RN 75-56-9). PA = 194.7 kcal/mol, 815 kJ/mol. | | | | | |
| $\text{C}_3\text{H}_7\text{OS}^+$ | | | | | | | |
| $\text{CH}_3\text{C}(\text{SH})\text{OCH}_3$ | | 125 | 522 | | | | |
| | | From proton affinity of $\text{CH}_3\text{C}(=\text{S})\text{OCH}_3$ (RN 21119-13-1) (83CAS/KIM). PA = 203.7 kcal/mol, 852. kJ/mol. | | | | | |
| $\text{CH}_3\text{C}(\text{OH})\text{SCH}_3$ | | 106 | 443 | | | | |
| | | From proton affinity of $\text{CH}_3\text{C}(=\text{O})\text{SCH}_3$ (RN 1534-08-3) (83CAS/KIM). PA = 199.7 kcal/mol, 836. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_7\text{O}_2^+$ | | | | | | | |
| $\text{C}_2\text{H}_5\text{C}(\text{OH})_2$ | | 67 | 280 | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{COOH}$ (RN 79-09-4). PA = 191.8 kcal/mol, 802 kJ/mol. | | | | | |
| $\text{HC}(\text{OH})\text{OC}_2\text{H}_5$ | | 80 | 335 | | | | |
| | | From proton affinity of HCOOC_2H_5 (RN 109-94-4). PA = 193.1 kcal/mol, 808 kJ/mol. | | | | | |
| $\text{CH}_3\text{C}(\text{OH})\text{OCH}_3$ | | 69 | 288 | | | | |
| | | From proton affinity of $\text{CH}_3\text{COOCH}_3$ (RN 79-20-9). PA = 197.8 kcal/mol, 828. kJ/mol. | | | | | |
| $\text{CH}(\text{OCH}_3)_2$ | | 97 | 406 | | | | 4483-45-8 |
| | | From appearance potential determination (82HOL/LOS2). | | | | | |

Table 1. Positive Ion Table - Continued

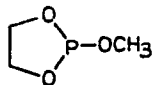
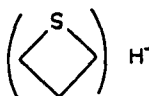
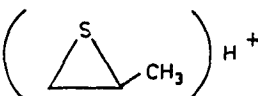
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_7\text{O}_3^+$ <chem>CH3OC(OH)OCH3</chem> | | 27 | 111 | | | | |
| | | From proton affinity of <chem>CH3OCOCH3</chem> (RN 616-38-6). PA = 200.2 kcal/mol, 838. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_7\text{O}_3\text{P}^+$ | | | | | | | |
|  | (9.06±0.1) | (45) | (186) | -164 | -688 | *EST | 3741-36-4 |
| | See also: 82WOR/HAR. | | | | | | |
| $\text{C}_3\text{H}_7\text{S}^+$ | | | | | | | |
|  | | (179) | (749) | | | | |
| | From proton affinity of thietane (RN 287-27-4). PA = (201.3) kcal/mol, (842) kJ/mol. | | | | | | |
|  | | (176) | (737) | | | | |
| | From proton affinity of 2-methylthiirane (RN 1072-43-1). PA = (200.6) kcal/mol, (839) kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_7\text{S}_2^+$ <chem>CH3C(SH)SCH3</chem> | | 182 | 763 | | | | |
| | From proton affinity of <chem>CH3C(=S)SCH3</chem> (RN 2168-84-5). PA = 207.3 kcal/mol, 867. kJ/mol. | | | | | | |
| C_3H_8^+ <chem>C3H8</chem> | 10.95±0.05 | 227.5 | 951.5 | -25.0±0.1 | -104.5±0.3 | 77PED/RYL | 74-98-6 |
| | See also: 81KIM/KAT. | | | | | | |
| $\text{C}_3\text{H}_8\text{Cl}^+$ <chem>CH3ClC2H5</chem> | | (164) | (688) | | | | |
| | $\Delta_f H(\text{Ion})$ from equilibrium constant determination (85SHA/HOJ). | | | | | | |
| $\text{C}_3\text{H}_8\text{Cl}_2\text{Si}^+$ <chem>(CH3)2SiCl(CH2Cl)</chem> | (9.2) | (126) | (527) | -86 | -361 | *EST | 1719-57-9 |
| | IP is onset of photoelectron band (81ZYK/KHV). | | | | | | |
| $\text{C}_3\text{H}_8\text{N}^+$ <chem>CH3CH2CHNH2</chem> | | 152 | 636 | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (81LOS/LAM). | | | | | | |
| <chem>CH3CHCH2NH2</chem> | | (161) | (673) | | | | |
| | From proton affinity of <chem>CH2=CHCH2NH2</chem> (RN 107-11-9). PA = 215.6 kcal/mol, 903 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

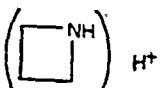
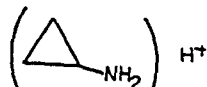
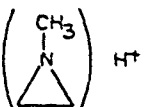
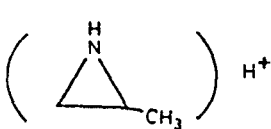
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_8\text{N}^+$ $(\text{CH}_3)_2\text{CNH}_2$ | (5.4) | (141) | (590) | (17) | (69) | 81LOS/LAM | 26374-12-9 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (81LOS/LAM). IP derived (83BUR/CAS). Value derived from proton affinity of $\text{CH}_2=\text{C}(\text{CH}_3)\text{NH}_2$ (PA = 226.3 kcal/mol, 947 kJ/mol) and $(\text{CH}_3)_2\text{C}=\text{NH}_2$ (PA = 221 kcal/mol, 925 kJ/mol) is 147 kcal/mol, 615 kJ/mol. (84LIA/LIE). | | | | | | |
| $\text{CH}_3\text{CH}_2\text{NHCH}_2$ | | 156 | 653 | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (81LOS/LAM). | | | | | | |
| $\text{CH}_3\text{CHNHCH}_3$ | | (147) | (615) | | | | |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (81LOS/LAM). | | | | | | |
| $\text{CH}_2\text{N}(\text{CH}_3)_2$ | 5.7 | (158) | (661) | (26) | (109) | 81GRI/LOS | 30208-47-0 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (81LOS/LAM), IP derived (81GRI/LOS, 83BUR/CAS, 81LOG/TAK). | | | | | | |
|  | | (167) | (698) | | | | |
| | From proton affinity of azetidine. (RN 503-29-7). PA = 222.8 kcal/mol, 932. kJ/mol. | | | | | | |
|  | | 169 | 707 | | | | |
| | From proton affinity of $c\text{-C}_3\text{H}_5\text{NH}_2$ (RN 765-30-0). PA = 215.2 kcal/mol, 900 kJ/mol. | | | | | | |
|  | | 174 | 730 | | | | |
| | From proton affinity of N-methylaziridine (RN 1072-44-2). PA = 221.6 kcal/mol, 927. kJ/mol. | | | | | | |
|  | | (168) | (704) | | | | |
| | From proton affinity of 2-methylaziridine (RN 75-55-8). PA = (219.2) kcal/mol, (917) kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_8\text{NO}^+$ $\text{HC}(\text{OH})\text{N}(\text{CH}_3)_2$ | | 108 | 454 | | | | |
| | From proton affinity of $\text{HCON}(\text{CH}_3)_2$ (RN 68-12-2). PA = 211.4 kcal/mol, 884 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_8\text{NO}_2^+$ $i\text{-C}_3\text{H}_7\text{ONHO}$ | | 132 | 552 | | | | |
| | From proton affinity of $i\text{-C}_3\text{H}_7\text{ONO}$ (RN 541-42-4). PA = 201.9 kcal/mol, 845 kJ/mol. | | | | | | |
| $\text{CH}_3\text{CH}(\text{NH}_3)\text{COOH}$ | | 52 | 216 | | | | |
| | From proton affinity of L-alanine (RN 56-41-7). PA = 214.8 kcal/mol, 899. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

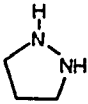
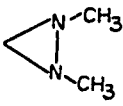
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|----------------|------------------------------|----------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_8\text{NO}_2^+$ <chem>CH3NH2CH2COOH</chem> | | 59 | 248 | | | | |
| | | From proton affinity of sarcosine (RN 107-97-1). PA = 218.7 kcal/mol, 915. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_8\text{NO}_3^+$ <chem>HOCH2CH(NH3)COOH</chem> | | 15 | 62 | | | | |
| | | From proton affinity of L-serine (RN 302-84-1). PA = 216.8 kcal/mol, 907. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_8\text{N}_2^+$  | (≤ 7.90) | (≤ 216) | (≤ 903) | 34 | 141 | *EST | 504-70-1 |
|  | (8.7) | (259) | (1082) | 58 | 243 | *EST | 6794-95-2 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_3\text{H}_8\text{N}_2\text{O}^+$ <chem>(CH3NH)2CO</chem> | (≤ 9.23) | (≤ 155) | (≤ 649) | -58 | -242 | *EST | 96-31-1 |
| <chem>(CH3)2NCONH2</chem> | (≤ 8.96) | (≤ 149) | (≤ 622) | -58 | -242 | *EST | 598-94-7 |
| $\text{C}_3\text{H}_8\text{N}_2\text{S}^+$ <chem>(CH3NH)2CS</chem> | ($\leq 8.08 \pm 0.03$) | (≤ 194) | (≤ 814) | 8 | 34 | *EST | 534-13-4 |
| $\text{C}_3\text{H}_8\text{O}^+$ n- <chem>C3H7OH</chem> | 10.22 \pm 0.03 | 175 181 | 731 756 | -60.9 \pm 0.2 -55.1 | -254.8 \pm 1. -230.4 | 77PED/RYL | 71-23-8 |
| | | See also: 84BOW/MAC, 81KIM/KAT, 80BAC/MOU. | | | | | |
| iso- <chem>C3H7OH</chem> | 10.12 \pm 0.08 | 168 174 | 704 729 | -65.1 \pm 0.1 -59.2 | -272.5 \pm 0.4 -247.7 | 77PED/RYL | 67-63-0 |
| | | See also: 72POT/SOR, 81KIM/KAT, 80BAC/MOU, 84BOW/MAC. | | | | | |
| <chem>C2H5OCH3</chem> | 9.72 | 172 | 721 | -51.7 \pm 0.1 | -216.4 \pm 0.6 | 77PED/RYL | 540-67-0 |
| | | IP from 81HOL/FIN, 84BOW/MAC, 81KIM/KAT, 80BAC/MOU, 82LEV/LIA, 79AUE/BOW. | | | | | |
| <chem>CH2CH2CH2OH2</chem> | | 171 | 714 | | | | |
| | | From appearance potential determinations (84HOL/MOM). | | | | | |
| <chem>CH2CHCH2HOH2</chem> | | 172 | 721 | | | | |
| | | From appearance potential determinations (84HOL/MOM). Authors propose that ion is proton-bound dimer of water and allyl radical. | | | | | |

Table 1. Positive Ion Table - Continued

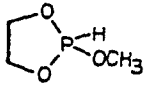
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_8\text{O}_2^+$ $(\text{CH}_3\text{O})_2\text{CH}_2$ | 9.5 | 136 | 568 | -83.2±0.2 | -348.2±0.7 | 77PED/RYL | 109-87-5 |
| | IP from 82HOL/LOS2. See also: 81JOR, 82ZVE/VIL, 81KIM/KAT. | | | | | | |
| $\text{HOCH}_2\text{CH}_2\text{OCH}_3$ | 9.6 | 134 | 562 | -87 | -364 | *EST | 109-86-4 |
| | IP is onset of photoelectron band (83BIE/MOR, 81KIM/KAT). | | | | | | |
| $\text{C}_3\text{H}_8\text{O}_3\text{P}^+$ | | -11 | -48 | | | | |
|  | | From proton affinity of 2-methoxy-1,3,2-dioxaphospholane (RN 3741-36-4). PA = 212.7 kcal/mol, 890 kJ/mol. | | | | | |
| $\text{C}_3\text{H}_8\text{S}^+$ n- $\text{C}_3\text{H}_7\text{SH}$ | 9.195±0.005 | 195.8 | 819.2 | -16.2±0.1 | -67.9±0.6 | 77PED/RYL | 107-03-9 |
| | See also: 81KIM/KAT. | | | | | | |
| iso- $\text{C}_3\text{H}_7\text{SH}$ | 9.14 | 193 | 806 | -18.2±0.1 | -76.2±0.6 | 77PED/RYL | 75-33-2 |
| | See: 81KIM/KAT. | | | | | | |
| $\text{C}_2\text{H}_5\text{SCH}_3$ | 8.54±0.1 | 183 | 764 | -14.2±0.3 | -59.6±1.1 | 77PED/RYL | 624-89-5 |
| | See also: 79AUE/BOW. | | | | | | |
| $\text{C}_3\text{H}_8\text{S}_2^+$ $\text{CH}_3\text{SCH}_2\text{SCH}_3$ | (8.4) | (195) | (815) | 1±2 | 5±8 | *EST | 1618-26-4 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_3\text{H}_8\text{Sc}^+$ $\text{C}_2\text{H}_4\text{ScH}(\text{CH}_3)$ | | (197) | (824) | | | | |
| | $\Delta_f H(\text{Ion})$ from 84TOL/BEA. | | | | | | |
| $\text{C}_3\text{H}_8\text{Si}^+$ $(\text{CH}_3)_2\text{Si}=\text{CH}_2$ | 7.71±0.03 | 183 | 765 | 5 | 21 | 86WAL | 4112-23-6 |
| | IP from 82DYK/JOS. See also: 81KOE/MCK. | | | | | | |
| C_3H_9^+ C_3H_9 | | 191 | 797 | | | | |
| | From proton affinity of C_3H_8 . (RN 74-98-6). PA = 150 kcal/mol, 628 kJ/mol. | | | | | | |
| $\text{C}_3\text{H}_9\text{Al}^+$ $(\text{CH}_3)_3\text{Al}$ | (≤9.76) | (≤206) | (≤861) | -19±3 | -81±11 | 77PED/RYL | 75-24-1 |
| $\text{C}_3\text{H}_9\text{As}^+$ $(\text{CH}_3)_3\text{As}$ | (8.2) | (192) | (804) | 3±2 | 13±10 | 77PED/RYL | 593-88-4 |
| | IP is onset of photoelectron band (82ELB/DIE). | | | | | | |
| $\text{C}_3\text{H}_9\text{AsO}_3^+$ $\text{As}(\text{OCH}_3)_3$ | (7.93) | (51) | (215) | -131±0.5 | -550±2 | 77PED/RYL | 6596-95-8 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_9\text{B}^+$ (CH_3) ₃ B | (9.5) | (190) (196) | (794) (820) | -29±2 -23 | -123±10 -97 | 77PED/RYL | 593-90-8 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_3\text{H}_9\text{BO}_3^+$ B(OCH ₃) ₃ | (10.0) | (15) | (65) | -215±0.5 | -900±2 | 77PED/RYL | 121-43-7 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_3\text{H}_9\text{BS}_3^+$ B(SCH ₃) ₃ | (8.74) | (164) | (687) | -37±0.7 | -156±3 | 77PED/RYL | 997-49-9 |
| $\text{C}_3\text{H}_9\text{BrPb}^+$ (CH ₃) ₃ PbBr | (≤9.30) | (≤229) | (≤956) | 14 | 59 | 85DEW/HOL | 6148-48-7 |
| $\text{C}_3\text{H}_9\text{BrSi}^+$ (CH ₃) ₃ SiBr | 10.0 | (161) (169) | (672) (707) | -70±1 -61±0.8 | -293±4 -258±4 | 77PED/RYL | 2857-97-8 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_3\text{H}_9\text{BrSn}^+$ (CH ₃) ₃ SnBr | (9.4) | (184) | (769) | -33±1 | -138±6 | 77PED/RYL | 1066-44-0 |
| $\text{C}_3\text{H}_9\text{ClGe}^+$ (CH ₃) ₃ GeCl | (9.2) | (148) | (620) | -64±3 | -268±13 | 80TEL/RAB | 1529-47-1 |
| IP is onset of photoelectron band. See also: 79DRA/GLA2. | | | | | | | |
| $\text{C}_3\text{H}_9\text{ClSi}^+$ (CH ₃) ₃ SiCl | (10.15) | (149) (156) | (625) (654) | -85 -78 | -354 -325 | 81BEL/PER | 75-77-4 |
| IP is onset of photoelectron band. See also: 84SZE/BAE, 81ZYK/KHV. | | | | | | | |
| $\text{C}_3\text{H}_9\text{ClSn}^+$ (CH ₃) ₃ SnCl | (9.90) | (185) | (773) | -43 | -182 | *EST | 1066-45-1 |
| IP from 82LEV/LIA. | | | | | | | |
| $\text{C}_3\text{H}_9\text{FN}^+$ CH ₂ FCH ₂ CH ₂ NH ₃ | | 87 | 365 | | | | |
| From proton affinity of CH ₂ FCH ₂ CH ₂ NH ₂ (RN 462-41-9). PA = 217.8 kcal/mol, 911. kJ/mol. | | | | | | | |
| $\text{C}_3\text{H}_9\text{FSi}^+$ (CH ₃) ₃ SiF | 10.31±0.04 | 112 | 468 | -126 | -527 | 77MUR/BEA | 420-56-4 |
| $\text{C}_3\text{H}_9\text{Ga}^+$ (CH ₃) ₃ Ga | (8.9) | (195) | (817) | -10±1 | -42±6 | 77PED/RYL | 1445-79-0 |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_9\text{Ge}^+$ (CH_3) ₃ Ge | | 180 | 754 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations. Proton affinity of (CH_3) ₂ Ge = CH_2 (RN 82064-99-1) = 204.9 kcal, 857. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_9\text{N}^+$ n- $\text{C}_3\text{H}_7\text{NH}_2$ | 8.78±0.02 See also: 81KIM/KAT. | 186 | 777 | -16.8±0.1 | -70.2±0.4 | 77PED/RYL | 107-10-8 |
| iso- $\text{C}_3\text{H}_7\text{NH}_2$ | 8.72±0.03 See also: 81KIM/KAT. | 181 | 758 | -20.0±0.1 | -83.8±0.5 | 77PED/RYL | 75-31-0 |
| (CH_3)(C_2H_5)NH | (8.15) IP from 79AUE/BOW. | (177) | (740) | -11±0.5 | -46±2 | *EST | 624-78-2 |
| (CH_3) ₃ N | 7.82±0.06 See also: 81KIM/KAT, 82ELB/DIE. | 175 | 731 | -5.7±0.1 | -23.7±0.6 | 77PED/RYL | 75-50-3 |
| $\text{C}_3\text{H}_9\text{NO}^+$ $\text{NH}_2(\text{CH}_2)_3\text{OH}$ | (9.0) IP is onset of photoelectron band. | (156) | (650) | -52 | -218 | *EST | 156-87-6 |
| $\text{CH}_3\text{OCH}_2\text{CH}_2\text{NH}_2$ | (8.9) IP is onset of photoelectron band. | (161) | (675) | -44±0.7 | -184±3 | *EST | 109-85-3 |
| $\text{CH}_3\text{ON}(\text{CH}_3)_2$ | ≤8.78 IP from 83MOL/PIK. See also: 82LEV/LIA. | (≤194) | (≤810) | -9 | -37 | *EST | 5669-39-6 |
| $\text{C}_3\text{H}_9\text{N}_3\text{Si}^+$ (CH_3) ₃ SiN ₃ | (≤9.7±0.1) | (≤241) | (≤1007) | 17±2 | 71±8 | 80TEL/RAB | 4648-54-8 |
| $\text{C}_3\text{H}_9\text{O}^+$ n- $\text{C}_3\text{H}_7\text{OH}_2$ | | 114 | 476 | | | | |
| | | From proton affinity of n- $\text{C}_3\text{H}_7\text{OH}$ (RN 71-23-8). PA = 190.8 kcal/mol, 798. kJ/mol. | | | | | |
| i- $\text{C}_3\text{H}_7\text{OH}_2$ | | 109 | 457 | | | | |
| | | From proton affinity of i- $\text{C}_3\text{H}_7\text{OH}$ (RN 67-63-0). PA = 191.2 kcal/mol, 800. kJ/mol. | | | | | |
| $\text{C}_2\text{H}_5\text{OHCH}_3$ | | 118 | 492 | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{OCH}_3$ (RN 540-67-0). PA = 196.4 kcal/mol, 822. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_9\text{OP}^+$ (CH_3) ₃ PO | (9.5) IP is onset of photoelectron band. | (115) | (482) | -104±2 | -434±8 | 77PED/RYL | 676-96-0 |
| $\text{C}_3\text{H}_9\text{O}_2^+$ $\text{HOCH}_2\text{CH}_2\text{OHCH}_3$ | | 96 | 402 | | | | |
| | | From proton affinity of $\text{HOCH}_2\text{CH}_2\text{OCH}_3$ (RN 109-86-4) (78TAF/TAA). PA = 182.6 kcal/mol, 764. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------|---|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_9\text{O}_3\text{P}^+$ $\text{P}(\text{OCH}_3)_3$ | (8.50) See also: 81CHA/FIN, 82WOR/HAR, 77COW/GOO. | (29) | (123) | -167±5 | -697±20 | 77PED/RYL | 121-45-9 |
| $\text{C}_3\text{H}_9\text{O}_3\text{PS}^+$ $(\text{CH}_3\text{O})_3\text{PS}$ | (≤9.16) | (≤28) | (≤117) | -183 | -767 | *EST | 152-18-1 |
| $\text{C}_3\text{H}_9\text{O}_4\text{P}^+$ $(\text{CH}_3\text{O})_3\text{PO}$ | 9.99 See also: 81CHA/FIN. | (-34) | (-143) | -265 | -1107 | *EST | 512-56-1 |
| $\text{C}_3\text{H}_9\text{P}^+$ $(\text{CH}_3)_3\text{P}$ | 8.06±0.05 See also: 82IKU/KEB, 82COW/KEM, 82ELB/DIE, 82BAN/CHA2. | 162 | 677 | -24±1 | -101±5 | 77PED/RYL | 594-09-2 |
| $\text{C}_3\text{H}_9\text{Pb}^+$ $(\text{CH}_3)_3\text{Pb}$ | | 200 | 840 | $\Delta_f H(\text{Ion})$ from appearance potential determinations. Proton affinity of $(\text{CH}_3)_2\text{Pb} = \text{CH}_2$ (RN 82065-01-8) = 223.9 kcal/mol, 937. kJ/mol. | | | |
| $\text{C}_3\text{H}_9\text{S}^+$ n- $\text{C}_3\text{H}_7\text{SH}_2$ | | 158 | 660 | From proton affinity of n- $\text{C}_3\text{H}_7\text{SH}$ (RN 107-03-9). PA = 191.6 kcal/mol, 802 kJ/mol. | | | |
| i- $\text{C}_3\text{H}_7\text{SH}_2$ | | 153 | 642 | From proton affinity of i- $\text{C}_3\text{H}_7\text{SH}$ (RN 75-33-2). PA = 194.1 kcal/mol, 812 kJ/mol. | | | |
| $\text{CH}_3\text{SHC}_2\text{H}_5$ | | 148 | 619 | From proton affinity of $\text{CH}_3\text{SC}_2\text{H}_5$ (RN 624-89-5). PA = 203.5 kcal/mol, 851 kJ/mol. | | | |
| $\text{C}_3\text{H}_9\text{Sb}^+$ $(\text{CH}_3)_3\text{Sb}$ | (7.7) | (185) | (775) | 8±6 | 32±25 | 77PED/RYL | 594-10-5 |
| | | IP is onset of photoelectron band(82ELB/DIE). | | | | | |
| $\text{C}_3\text{H}_9\text{Si}^+$ $(\text{CH}_3)_3\text{Si}$ | (6.5) | (150) | (630) | -0.8±2 | -3±8 | 86DON/WAL | 16571-41-8 |
| | | (157) | (656) | 6 | 26 | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (84SZE/BAE, 84SZE/BAE2). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | |
| $\text{C}_3\text{H}_9\text{Sn}^+$ $(\text{CH}_3)_3\text{Sn}$ | (7.10±0.05) | (181) | (759) | $\Delta_f H(\text{Ion})$ from appearance potential determinations. Proton affinity of $(\text{CH}_3)_2\text{Sn} = \text{CH}_2$ (RN 82065-00-7) = 215.8 kcal/mol, 903. kJ/mol. | | | |
| $\text{C}_3\text{H}_{10}\text{As}^+$ $(\text{CH}_3)_3\text{AsH}$ | | 155 | 650 | From proton affinity of $(\text{CH}_3)_3\text{As}$ (RN 593-88-4). PA = 213.4 kcal/mol, 893 kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_{10}\text{N}^+$ n- $\text{C}_3\text{H}_7\text{NH}_3$ | | 131 | 548 | | | | |
| | | From proton affinity of n- $\text{C}_3\text{H}_7\text{NH}_2$ (RN 107-10-8). PA = 217.9 kcal/mol, 912. kJ/mol. | | | | | |
| i- $\text{C}_3\text{H}_7\text{NH}_3$ | | 127 | 531 | | | | |
| | | From proton affinity of i- $\text{C}_3\text{H}_7\text{NH}_2$ (RN 75-31-0). PA = 218.6 kcal/mol, 915. kJ/mol. | | | | | |
| $(\text{CH}_3)(\text{C}_2\text{H}_5)\text{NH}_2$ | | 132 | 552 | | | | |
| | | From proton affinity of $(\text{CH}_3)(\text{C}_2\text{H}_5)\text{NH}$ (RN 624-78-2). PA = 222.8 kcal/mol, 932. kJ/mol. | | | | | |
| $(\text{CH}_3)_3\text{NH}$ | | 135 | 564 | | | | |
| | | From proton affinity of $(\text{CH}_3)_3\text{N}$ (RN 75-50-3). PA = 225.1 kcal/mol, 942. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{10}\text{NO}^+$ $\text{NH}_3(\text{CH}_2)_3\text{OH}$ | | 85 | 356 | | | | |
| | | From proton affinity of $\text{NH}_2(\text{CH}_2)_3\text{OH}$ (RN 156-87-6). PA = 228.6 kcal/mol, 956.5 kJ/mol. | | | | | |
| $\text{CH}_3\text{OCH}_2\text{CH}_2\text{NH}_3$ | | 98 | 412 | | | | |
| | | From proton affinity of $\text{CH}_3\text{OCH}_2\text{CH}_2\text{NH}_2$ (RN 109-85-3). PA = 223.3 kcal/mol, 934 kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{10}\text{N}_2^+$ $(\text{CH}_3)_2\text{NNH}(\text{CH}_3)$ | | | | 21 | 87 | 69BEN/CRU | 1741-01-1 |
| | | A value of 7.93 eV has been reported for the adiabatic IP of this compound. Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. | | | | | |
| $\text{C}_3\text{H}_{10}\text{OP}^+$ $(\text{CH}_3)_3\text{POH}$ | | 124 | 518 | | | | |
| | | From proton affinity of $(\text{CH}_3)_3\text{PO}$ (RN 676-96-0) (84BOL/HOU). PA = 217.1 kcal/mol, 908. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{10}\text{O}_3\text{P}^+$ $\text{HP}(\text{OCH}_3)_3$ | | -22 | -92 | | | | |
| | | From proton affinity of $\text{P}(\text{OCH}_3)_3$ (RN 121-45-9). PA = 220.6 kcal/mol, 923. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{10}\text{O}_3\text{PS}^+$ $(\text{CH}_3\text{O})_3\text{PSH}$ | | -32 | -134 | | | | |
| | | From proton affinity of $(\text{CH}_3\text{O})_3\text{PS}$ (RN 152-18-1). PA = 214.5 kcal/mol, 897. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{10}\text{O}_4\text{P}^+$ $(\text{CH}_3\text{O})_3\text{POH}$ | | -111 | -464 | | | | |
| | | From proton affinity of $(\text{CH}_3\text{O})_3\text{PO}$ (RN 512-56-1). PA = 212.0 kcal/mol, 887. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{10}\text{P}^+$ $(\text{CH}_3)_3\text{PH}$ | | 114 | 479 | | | | |
| | | From proton affinity of $(\text{CH}_3)_3\text{P}$ (RN 594-09-2). PA = 227.1 kcal/mol, 950. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

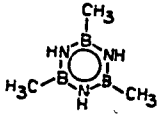
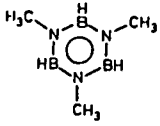
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|----------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_3\text{H}_{10}\text{Si}^+$ (CH_3) ₃ SiH | 9.9 IP from 81HOT. | 189 | 792 | -39±1 | -163±4 | 86DON/WAL | 993-07-7 |
| $\text{C}_3\text{H}_{10}\text{Sn}^+$ (CH_3) ₃ SnH | (≤9.9) | (≤228) | (≤955) | 0±2 | 0±8 | 80TEL/RAB | 1631-73-8 |
| $\text{C}_3\text{H}_{11}\text{N}_2^+$ NH ₂ (CH ₂) ₃ NH ₃ | | 124 | 518 | | | | |
| | | From proton affinity of NH ₂ (CH ₂) ₃ NH ₂ (RN 109-76-2). PA = 234.1 kcal/mol, 979. kJ/mol. | | | | | |
| $\text{C}_3\text{H}_{12}\text{BN}^+$ (CH_3) ₃ NBH ₃ | (9.28±0.2) | (194) (203) | (810) (848) | -20 -11 | -85 -47 | 82TN270 | 75-22-9 |
| $\text{C}_3\text{H}_{12}\text{B}_3\text{N}_3^+$  | (9.1±0.15) | (-13) | (-55) | -223 | -933 | 70FIN/GAR | 1004-35-9 |
|  | (9.07) | (-8) | (-33) | -217±1 | -908±4 | 80TEL/RAB | 5314-85-2 |
| C_3IN^+ IC≡CCN | (10.18±0.02) See also: 84KUH/MAI. | (347) | (1451) | 112±10 | 469±40 | 79BUC/VOG | 2003-32-9 |
| C_3La^+ LaC ₃ | (6.8±0.5) | (336) | (1404) | 179±1 | 748±1 | 81GIN/PEL | 12602-63-0 |
| $\text{C}_3\text{N}_2\text{O}^+$ (CN) ₂ CO | (≤12.56) | (≤349) | (≤1459) | 59±1 | 247±6 | 77PED/RYL | 1115-12-4 |
| C_3O_2^+ C ₃ O ₂ | 10.60 | 222 | 929 | -22±0.5 | -94±2 | 71JANAF | 504-64-3 |
| C_4^+ C ₄ | (12.6) | (522) | (2187) | 232±8 | 971±33 | 71JANAF | 12184-80-4 |
| $\text{C}_4\text{Cl}_2\text{Hg}^+$ (ClC≡C) ₂ Hg | 9.58±0.02 IP is onset of photoelectron band (81FUR/PIA). | (373) | (1559) | 152 | 635 | *EST | 64771-59-1 |

Table 1. Positive Ion Table - Continued

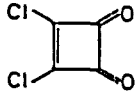
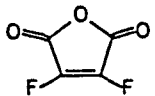
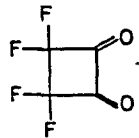
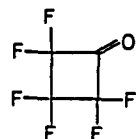
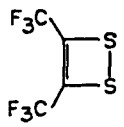
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{Cl}_2\text{O}_2^+$ | | | | | | | |
|  | (9.5) | (204) | (856) | -14 | -61 | *EST | 2892-63-9 |
| | IP is onset of photoelectron band (81BOC/RIE). | | | | | | |
| $\text{C}_4\text{F}_2\text{O}_3^+$ | | | | | | | |
|  | (11.45) | (79) | (330) | -185 | -775 | *EST | 669-78-3 |
| | IP from 81ASB/SVE. | | | | | | |
| $\text{C}_4\text{F}_4\text{O}_2^+$ | | | | | | | |
|  | 10.05±0.1 | (47) | (199) | -184 | -771 | *EST | 663-45-6 |
| | IP is onset of photoelectron band (85GLE/SCH, 85ALB/HEL). | | | | | | |
| C_4F_6^+ | | | | | | | |
| $\text{CF}_2 = \text{CFCF} = \text{CF}_2$ | (9.5) | (-5) | (-21) | -224 | -938 | 68LAC/SKI | 685-63-2 |
| $\text{C}_4\text{F}_6\text{O}^+$ | | | | | | | |
| $(\text{CF}_3)_2\text{C} = \text{C} = \text{O}$ | (10.67) | (-95) | (-398) | -341 | -1427 | *EST | |
| | IP is onset of photoelectron band (83GLE/SAA). | | | | | | |
|  | (10.7) | (-30) | (-124) | -276 | -1157 | *EST | 699-35-4 |
| | IP is onset of photoelectron band (85GLE/SCH). | | | | | | |
| $\text{C}_4\text{F}_6\text{S}^+$ | | | | | | | |
| $(\text{CF}_3)_2\text{C} = \text{C} = \text{S}$ | (9.4) | (-71) | (-295) | -287 | -1202 | *EST | 7445-60-5 |
| | IP is onset of photoelectron band (83GLE/SAA). | | | | | | |
| $\text{C}_4\text{F}_6\text{S}_2^+$ | | | | | | | |
|  | 9.6 | (37) | (154) | -185 | -772 | *EST | 360-91-8 |
| | IP is onset of photoelectron band (83SCH/SCH, 83JIA/MOH). | | | | | | |
| C_4F_8^+ | | | | | | | |
| (Z)-2- C_4F_8 | (11.1) | (-126) | (-526) | -382 | -1597 | 70BEN/O'N | 1516-65-0 |
| | IP is onset of photoelectron band. | | | | | | |
| (E)-2- C_4F_8 | (11.0) | (-129) | (-540) | -383 | -1601 | 70BEN/O'N | 1516-64-9 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

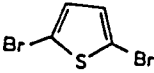

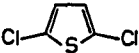
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------|----------------------------|--------------------------|---------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{FeI}_2\text{O}_4^+$ $\text{Fe}(\text{CO})_4\text{I}_2$ | | (8.4) | (42) | (174) | -152±2 | -636±9 | 82PIL/SKI | 14911-55-8 |
| IP is onset of photoelectron band. | | | | | | | | |
| C_4HCl^+ $\text{CH}=\text{CC}=\text{CCl}$ | | (9.72±0.02) | (345) | (1443) | 121±0.5 | 505±2 | *EST | 6089-44-7 |
| $\text{C}_4\text{HC}_6\text{O}_4^+$ $\text{HCo}(\text{CO})_4$ | | (8.2) | (53) | (222) | -136±0.5 | -569±2 | 77PED/RYL | 16842-03-8 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_4\text{HF}_7\text{N}^+$ $n\text{-C}_3\text{F}_7\text{CNH}$ | | | -110 | -460 | | | | |
| From proton affinity of $n\text{-C}_3\text{F}_7\text{CN}$ (RN 375-00-8). PA = 167.4 kcal/mol, 700. kJ/mol. | | | | | | | | |
| $\text{C}_4\text{HF}_9\text{O}^+$ $(\text{CF}_3)_3\text{COH}$ | | 12.25 | (-266) | (-1115) | (-549) | (-2297) | *EST | 2378-02-1 |
| IP from 83KOP/MOL. | | | | | | | | |
| $\text{C}_4\text{HNiO}_4^+$ $\text{HNi}(\text{CO})_4$ | | | (43) | (179) | | | | |
| From proton affinity of $\text{Ni}(\text{CO})_4$ (RN 13463-39-3). PA = (180) kcal/mol, (753) kJ/mol. | | | | | | | | |
| C_4H_2^+ $\text{HC}=\text{CC}=\text{CH}$ | | 10.180±0.003 | 340 | 1422 | 105 | 440 | 85STE/FAH | 460-12-8 |
| See also: 80MAI/THO. | | | | | | | | |
| $\text{C}_4\text{H}_2\text{Br}_2\text{S}^+$ | | | | | | | | |
|  | | (≤8.49) | (≤233) | (≤976) | 38 | 157 | *EST | 3141-27-3 |
|  | | (≤8.94) | (≤246) | (≤1028) | 39 | 165 | *EST | 3141-26-2 |
| $\text{C}_4\text{H}_2\text{Cl}_2\text{S}^+$ | | | | | | | | |
|  | | (8.60±0.05) | (213) | (890) | 14 | 60 | *EST | 3172-52-9 |
| $\text{C}_4\text{H}_2\text{F}_4^+$ $\text{CF}_2\text{CHCHCF}_2$ | | (10.6±0.1) | (82) | (343) | -163 | -680 | *EST | 407-70-5 |

Table 1. Positive Ion Table - Continued






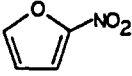
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₄H₃BrS⁺ | | | | | | | |
|  | 8.6 IP is onset of photoelectron band. | (231) | (966) | 33 | 136 | *EST | 1003-09-4 |
|  | 8.812±0.005 | (236) | (986) | 33 | 136 | *EST | 872-31-1 |
| C₄H₃ClS⁺ | | | | | | | |
|  | 8.89±0.05 | (225) | (941) | 20 | 83 | *EST | 96-43-5 |
| C₄H₃F₉N⁺ (CF ₃) ₃ CNH ₃ | | | | | | | |
| | | (-329) | (-1375) | | | | |
| | From proton affinity of (CF ₃) ₃ CNH ₂ (RN 2809-92-9). PA = (191.5) kcal/mol, (801.) kJ/mol. | | | | | | |
| C₄H₃IS⁺ | | | | | | | |
|  | ≤8.46 | (≤242) | (≤1010) | 46 | 194 | *EST | 3437-95-4 |
|  | (≤8.46) | (≤241) | (≤1010) | 46 | 194 | *EST | 10486-61-0 |
| C₄H₃N⁺ | | | | | | | |
| CH ₂ =C=CHCN | (10.1) IP is onset of photoelectron band. | (259) | (1084) | 26 | 110 | *EST | 1001-56-5 |
| CH ₃ C≡CCN | 10.78±0.02 | (329) | (1378) | 81±0.7 | 338±3 | *EST | 13752-78-8 |
| C₄H₃NO₃⁺ | | | | | | | |
|  | (≤9.75±0.05) | (≤218) | (≤910) | -7 | -31 | *EST | 609-39-2 |

Table 1. Positive Ion Table - Continued

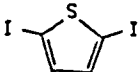

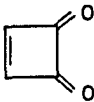
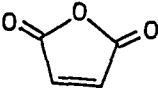
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|-----------------|------------------------------|-----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_2\text{F}_9\text{N}^+$ | $(\text{CF}_3)_3\text{CNH}_2$ | (10.4) | (-263) | (-1100) | -503 | -2104 | *EST | 2809-92-9 |
| | | IP from 79AUE/BOW. | | | | | | |
| $\text{C}_4\text{H}_2\text{F}_9\text{O}^+$ | $(\text{CF}_3)_3\text{COH}_2$ | | -346 | -1449 | | | | |
| | | From proton affinity of $(\text{CF}_3)_3\text{COH}$ (RN 2378-02-1). PA = 163.1 kcal/mol, 682. kJ/mol. | | | | | | |
| $\text{C}_4\text{H}_2\text{I}_2\text{S}^+$ |  | ≤ 8.28 | (≤ 256) | (≤ 1072) | 65 | 273 | *EST | 625-88-7 |
| |  | (≤ 8.45) | (≤ 263) | (≤ 1099) | 68 | 284 | *EST | 19259-08-6 |
| $\text{C}_4\text{H}_2\text{N}_2^+$ | (Z)-CH(CN)CH(CN) | (11.15) | (338) | (1416) | 81.3 ± 0.5 | 340.2 ± 1.9 | 77PED/RYL | 928-53-0 |
| | (E)-CH(CN)CH(CN) | 11.16 ± 0.03 | 338 | 1417 | 81 | 340 | 82CHU/NGU | 764-42-1 |
| $\text{C}_4\text{H}_2\text{O}_2^+$ |  | (≤ 9.79) | (≤ 239) | (≤ 1002) | 14 | 57 | *EST | 32936-74-6 |
| $\text{C}_4\text{H}_2\text{O}_3^+$ |  | (10.8) | (154) | (644) | -95 ± 1 | -398 ± 5 | 77PED/RYL | 108-31-6 |
| | | IP is onset of photoelectron band (81KIM/KAT). | | | | | | |
| C_4H_3^+ | HCCCCH ₂ | | (291) | (1217) | | | | |
| | | From proton affinity of HC≡CC≡CH (RN 460-12-8) (87DEA/MAU). PA = 180 kcal/mol, 753 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued








| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_4H_4^+ | | | | | | | |
| $\text{CH}_2=\text{C}=\text{C}=\text{CH}_2$ | (9.15) | (294) | (1232) | 83 | 349 | 82ROS/DAN | 2873-50-9 |
| | See also: 85DEW/TIE. | | | | | | |
| $\text{CH}_2=\text{CHC}\equiv\text{CH}$ | 9.58±0.02 | (294) | (1229) | 73 | 305 | 69STU/WES | 689-97-4 |
|  | 8.15 | (289) | (1209) | 101 | 423 | 87STA/NOR | 4095-06-1 |
| | Heat of formation of ion from appearance potential measurements (82ROS/DAN); $\Delta_f H(\text{Neutral})$ is $\Delta_f H(\text{Ion}) - \text{IP}$ (87STA/NOR). | | | | | | |
| $\text{C}_4\text{H}_4\text{N}_2^+$ | | | | | | | |
| $\text{NCCH}_2\text{CH}_2\text{CN}$ | 12.1±0.25 | 329 | 1377 | 50.1±0.1 | 209.7±0.6 | 77PED/RYL | 110-61-2 |
| | IP from 82CHE/LAP. | | | | | | |
|  | (8.64) | (266) | (1112) | 66.5±0.2 | 278.3±1 | 77PED/RYL | 289-80-5 |
|  | 9.23 | 260 | 1087 | 47.0±0.2 | 196.6±0.9 | 77PED/RYL | 289-95-2 |
| | See also: 83PIA/KEL. | | | | | | |
|  | 9.29±0.01 | 261 | 1092 | 46.8±0.3 | 196.0±1.3 | 77PED/RYL | 290-37-9 |
| | See also: 83PIA/KEL. | | | | | | |
| $\text{C}_4\text{H}_4\text{N}_2\text{O}^+$ | | | | | | | |
|  | (8.89±0.02) | (252) | (1056) | 47 | 198 | *EST | 1457-42-7 |
|  | (8.80±0.02) | (231) | (966) | 28 | 117 | *EST | 17043-94-6 |
|  | (10.06±0.05) | (221) | (924) | -11 | -47 | *EST | 557-01-7 |

Table 1. Positive Ion Table - Continued


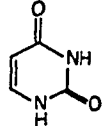
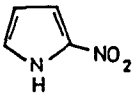
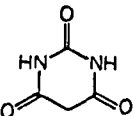
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|-----------|--|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_4\text{N}_2\text{O}^+$  | (9.0) IP is onset of photoelectron band. | (235) | (984) | 28 | 116 | *EST | 2423-65-6 |
| $\text{C}_4\text{H}_4\text{N}_2\text{O}_2^+$  | (9.2) IP is onset of photoelectron band. See also: 81YU/ODO. | (140) | (585) | -72 ± 0.5 | -303 ± 2 | 77NAB/SAB | 66-22-8 |
|  | (9.30±0.05) | (237) | (990) | 23 | 93 | *EST | 5919-26-6 |
| $\text{C}_4\text{H}_4\text{N}_2\text{O}_3^+$  | (10.20) | (103) | (430) | -132 | -554 | 72DOM | 67-52-7 |
| $\text{C}_4\text{H}_4\text{O}^+$ $\text{CH}_3\text{CH}=\text{C}=\text{C}=\text{O}$ | 8.68±0.05 IP from 83TER/HOL. See also: 79HOL/TER, 81MOH/HIR. | (215) | (900) | 15 | 63 | *EST | 78957-08-1 |
| $(\text{CH}_2)_2\text{C}=\text{C}=\text{O}$ | | 198 | 828 | From appearance potential determination (82BUR/HOL). | | | |
| $\text{CH}_2=\text{CHCH}=\text{C}=\text{O}$ | 8.29±0.05 IP from 79TER/BUR. See also: 82BUR/HOL, 81MOH/HIR, 79HOL/TER, 85MCN/SUF, 81BOC/HIR. | (195) | (817) | 4 | 17 | *EST | 50888-73-8 |
| $\text{CH}_2=\text{C}=\text{CHCHO}$ | (9.5) $\Delta_f H(\text{Ion})$ from 82BUR/HOL. IP from 79HOL/TER. | (236) | (987) | 18 | 75 | *EST | 53268-92-1 |
| $\text{HC}=\text{CCH}_2\text{CHO}$ | (9.85) IP estimated in 82BUR/HOL. | (247) | (1034) | 20 | 84 | *EST | 52844-23-2 |
| $\text{CH}_3\text{C}=\text{CCHO}$ | 10.20±0.02 IP from 79CAR/MOU. See also: 79TER/BUR, 82BUR/HOL, 79HOL/TER. | (253) | (1057) | 17 | 73 | *EST | 1119-19-3 |
| $\text{CH}_3\text{COC}=\text{CH}$ | 10.17±0.02 IP from 79CAR/MOU. See also: 79TER/BUR, 82BUR/HOL, 79HOL/TER. | (250.18) | (1046.75) | 15.6 ± 2 | 65.5 ± 1 | 85FUC | 1423-60-5 |
| $\text{CH}_2=\text{C}=\text{C}=\text{CHOH}$ | | 222 | 931 | From appearance potential determination(82BUR/HOL). | | | |

Table 1. Positive Ion Table - Continued


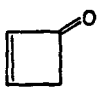
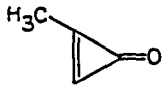
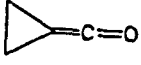


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_4\text{O}^+$ HC=CCH=CHOH | | 220 | 922 | | | | 59095-55-5 |
| | | From appearance potential determination(82BUR/HOL). | | | | | |
| $\text{CH}_2=\text{C}(\text{OH})\text{C}=\text{CH}$ | (8.92) | (226) | (944) | 20±2 | 83±7 | 86TUR/HAV2 | |
| | $\Delta_f H(\text{Ion})$ from appearance potential determination (86TUR/HAV2). See also: 82BUR/HOL, 79HOL/TER. | | | | | | |
| $\text{HC}=\text{COCH}=\text{CH}_2$ | 9.40 | (273) | (1142) | 56 | 235 | *EST | |
| | IP from 82BUR/HOL. See also: 79HOL/TER. | | | | | | |
|  | 8.883±0.003 | 196.5 | 822.3 | -8.3±0.1 | -34.8±0.4 | 77PED/RYL | 110-00-9 |
| | See also: 82BUR/HOL, 82KOB/KUB, 79HOL/TER, 81GAL/KLA, 81KIM/KAT, 82BIE/ASB, 80TED/VID, 83BOC/ROT, 83ZYK/ERC, 82KLA/SAB. | | | | | | |
|  | (9.3) | (223) | (933) | 8 | 33 | *EST | 32264-87-2 |
| | From appearance potential determination; kinetic energy release = 0.19 eV (82BUR/HOL). IP from 79HOL/TER. | | | | | | |
|  | 9.15±0.05 | (240) | (1004) | 29 | 121 | *EST | 4883-96-9 |
| | See also: 79TER/BUR, 82BUR/HOL, 79HOL/TER. | | | | | | |
|  | (8.78) | (222) | (931) | 20 | 84 | *EST | |
| | IP from 81BOC/HIR. | | | | | | |
|  | (9.6) | (235) | (983) | 14 | 58 | *EST | 36998-21-7 |
| | IP from 79HOL/TER. | | | | | | |
| $\text{C}_4\text{H}_4\text{O}_2^+$ HC=CCOOCH ₃ | (10.3) | (214) | (894) | -24 | -100 | *EST | 922-67-8 |
| | IP is onset of photoelectron band (82BIE/ASB). | | | | | | |
|  | (7.75±0.02) | (152) | (633) | -27±1 | -115±5 | *EST | 290-67-5 |

Table 1. Positive Ion Table - Continued


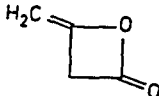
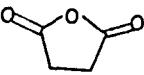


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|----------------------|---|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_4\text{O}_2^+$  | (9.4) IP is onset of photoelectron band. | (178) | (745) | -39 | -163 | *EST | 33689-28-0 |
| $\text{C}_4\text{H}_4\text{O}_2^+$  | (9.6±0.02) IP from 84OLI/FLE. | (176) | (736) | -45.5±0.1 | -190.3±0.4 | 77PED/RYL | 674-82-8 |
| $\text{C}_4\text{H}_4\text{O}_3^+$  | (10.6) IP is onset of photoelectron band (81KIM/KAT). | (119) | (498) | -125 | -525 | 77PED/RYL | 108-30-5 |
| $\text{C}_4\text{H}_4\text{O}_4^+$ (E)-HO ₂ CCH=CHCO ₂ H | (10.7) IP is onset of photoelectron band. | (85) | (352) | -162±0.6 | -680±3 | 77PED/RYL | 110-17-8 |
| $\text{C}_4\text{H}_4\text{S}^+$  | 8.87±0.04 See also: 80TED/VID, 83BOC/ROT, 81GAL/KLA, 82KLA/SAB. | 232 | 971 | 27.5±0.1 | 115.0±0.4 | 81KUD/KUD3 | 110-02-1 |
| $\text{C}_4\text{H}_4\text{S}_2^+$  | (7.7) IP is onset of photoelectron band. | (233) | (976) | 56±3 | 233±13 | *EST | 290-79-9 |
| C_4H_5^+ CH ₂ =CCH=CH ₂ | | (246) | (1029) | | | | 62698-26-4 |
| | From appearance potential measurements (84LOS/HOL). | | | | | | |
| CH=CCHCH ₃ | 7.97 | 257 | 1074 | 73 | 305 | 82MCM/GOL | 3315-42-2 |
| | From appearance potential measurements (84LOS/HOL). | | | | | | |
| CH ₃ C=CCH ₂ | 7.95 | 252 | 1056 | 69 | 289 | 82MCM/GOL | 64235-83-2 ⁸²²⁵²⁻³⁸⁻⁸ |
| | From appearance potential measurements (84LOS/HOL). | | | | | | |

Table 1. Positive Ion Table - Continued


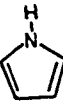

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_4H_5^+  | | (237) | (992) | | | | 60824-24-0 |
| | | From appearance potential measurements (84LOS/HOL). | | | | | |
| $\text{C}_4\text{H}_5\text{ClO}^+$ (E)- $\text{CH}_3\text{CH}=\text{CHCOCl}^+$ | (9.4) | (216) | (906) | -0.2 | -1 | *EST | 625-35-4 |
| | IP is onset of photoelectron band (81MOH/HIR). | | | | | | |
| $\text{C}_4\text{H}_5\text{F}_3\text{O}_2^+$ $\text{CF}_3\text{COOC}_2\text{H}_5$ | (11.0) | (5) | (19) | -249 | -1042 | *EST | 383-63-1 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_4\text{H}_5\text{F}_4\text{O}_2^+$ $\text{CF}_3\text{C}(\text{OH})\text{OCH}_2\text{CH}_2\text{F}$ | | -105 | -441 | | | | |
| | From proton affinity of $\text{CF}_3\text{COOCH}_2\text{CH}_2\text{F}$ (RN 1683-88-1). PA = 178.6 kcal/mol, 747. kJ/mol. | | | | | | |
| $\text{C}_4\text{H}_5\text{F}_6\text{O}^+$ $(\text{CF}_3)_2\text{C}(\text{CH}_3)\text{OH}_2$ | | -192 | -805 | | | | |
| | From proton affinity of $(\text{CF}_3)_2\text{C}(\text{CH}_3)\text{OH}$ (RN 1515-14-6). PA = 167.0 kcal/mol, 699. kJ/mol. | | | | | | |
| $\text{C}_4\text{H}_5\text{N}^+$ $\text{CH}_2=\text{CHCH}_2\text{CN}$ | 10.20±0.05 | 273 | 1140 | 37±0.5 | 156±2 | 77PED/RYL | 109-75-1 |
| | See also: 84OHN/MAT. | | | | | | |
| $\text{CH}_2\text{C}(\text{CH}_3)\text{CN}$ | 10.34 | 269 | 1128 | 31 | 130 | 80WIL/BAE | 126-98-7 |
| (E)- $\text{CH}_3\text{CH}=\text{CHCN}$ | (≤10.23±0.05) | (≤272) | (≤1137) | 36 | 150 | 82CHU/NGU | 627-26-9 |
|  | 8.208±0.005 | 215.2 | 900.2 | 25.9±0.1 | 108.3±0.4 | 80WIL/BAE | 109-97-7 |
| | See also: 81GAL/KLA, 82BIE/ASB, 80TED/VID, 82KLA/SAB. | | | | | | |
|  | 10.25 | 280 | 1172 | 44±0.2 | 183±1 | 82FUC/HAL | 5500-21-0 |

Table 1. Positive Ion Table - Continued




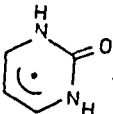
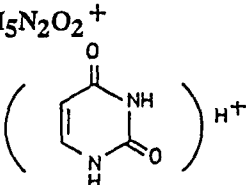
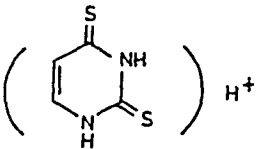
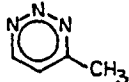

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_5\text{N}_2^+$ | | | | | | | |
|  | | 216 | 906 | | | | |
| | | From proton affinity of pyridazine (RN 289-80-5). PA = 215.7 kcal/mol, 902 kJ/mol. | | | | | |
|  | | 202 | 846 | | | | |
| | | From proton affinity of pyrimidine (RN 289-95-2). PA = 210.8 kcal/mol, 882. kJ/mol. | | | | | |
|  | | 203 | 852 | | | | |
| | | From proton affinity of pyrazine (RN 290-37-9). PA = 209.0 kcal/mol, 874. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_5\text{N}_2\text{O}^+$ | | | | | | | |
|  | | 146 | 613 | | | | |
| | | From proton affinity of 2(1H)-pyrimidinone (RN 557-01-7). PA = -208 kcal/mol, -870 kJ/mol. | | | | | |
| $\text{C}_4\text{H}_5\text{N}_2\text{O}_2^+$ | | | | | | | |
|  | | (85) | (357) | | | | |
| | | From proton affinity of uracil (RN 66-22-8). PA = -208 kcal/mol, -870 kJ/mol. | | | | | |
| $\text{C}_4\text{H}_5\text{N}_2\text{S}_2^+$ | | | | | | | |
|  | | (200) | (836) | | | | |
| | | From proton affinity of dithiouracil (RN 2001-93-6). PA = -217 kcal/mol, -907 kJ/mol. | | | | | |
| $\text{C}_4\text{H}_5\text{N}_3^+$ | | | | | | | |
|  | (≤ 9.7) IP from 83GLE/SPA. | (≤ 313) | (≤ 1311) | 90 | 375 | *EST | 77202-08-5 |
|  | (9.1) IP is onset of photoelectron band (83GLE/SPA). | (301) | (1258) | 91 | 380 | *EST | 86402-30-4 |

Table 1. Positive Ion Table - Continued

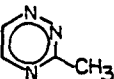
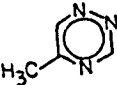
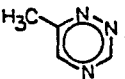
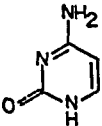
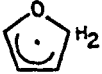
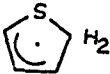
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|-----------------|---|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_5\text{N}_3^+$  | (8.6) IP is onset of photoelectron band. | (268) | (1123) | 70 | 293 | *EST | 24108-33-6 |
|  | (≤ 9.31) | (≤ 285) | (≤ 1191) | 70 | 293 | *EST | 21134-95-2 |
|  | (≤ 9.35) | (≤ 286) | (≤ 1195) | 70 | 293 | *EST | 21134-96-3 |
| $\text{C}_4\text{H}_5\text{N}_3\text{O}^+$  | (8.45) | (181) | (756) | -14 ± 2 | -59 ± 10 | 80SAB2 | 71-30-7 |
| $\text{C}_4\text{H}_5\text{O}^+$  | | 165 | 691 | From proton affinity of furan (RN 110-00-9). PA = 192.2 kcal/mol, 804. kJ/mol. | | | |
| $\text{C}_4\text{H}_5\text{O}_2^+$ $\text{CH}_3\text{C}(\text{OH})=\text{CHCO}$ | | 110 | 461 | From appearance potential of 10.24 eV in $\text{CH}_3\text{COCH}_2\text{COCH}_3$. | | | |
| $\text{C}_4\text{H}_5\text{S}^+$  | | 197 | 826 | From proton affinity of thiophene (RN 110-02-1) (86MAU, 84LIA/LIE). PA = 195.8 kcal/mol, 819. kJ/mol. | | | |
| C_4H_6^+ $\text{CH}_2=\text{C}=\text{CHCH}_3$ | (9.03) | (247) | (1033) | 38.8 ± 0.1 | 162.3 ± 0.5 | 77PED/RYL | 590-19-2 |

Table 1. Positive Ion Table - Continued


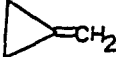


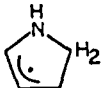

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|---|-------------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_4H_6^+ | | | | | | | | |
| | $\text{CH}_3\text{C}\equiv\text{CCH}_3$ | 9.562±0.005 | <u>255.2</u> | <u>1068</u> | 34.7±0.2 | 145.4±0.8 | 77PED/RYL | 503-17-3 |
| |  | 9.43 | 255 | 1067 | 37.5±0.4 | 156.7±1.5 | 77PED/RYL | 822-35-5 |
| |  | (9.57) See also: 81KIM/KAT. | (269) | (1124) | 48±0.5 | 201±2 | 77PED/RYL | 6142-73-0 |
| |  | 8.700±0.005 IP from 83BOM/DAN3. | 253 | 1057 | 51.9±0.2 | 217.2±0.8 | 77PED/RYL | 157-33-5 |
| $\text{C}_4\text{H}_6\text{Cl}_2\text{Si}^+$ | | | | | | | | |
| | $\text{CH}_2=\text{CHSiCl}_2\text{CH}=\text{CH}_2$ | (≤10.8) IP from 81KHV/ZYK. | (≤192) | (≤802) | -57 | -240 | *EST | 1745-72-8 |
| |  | ≤9.65 See also: 81KHV/ZYK. | (≤143) | (≤598) | -80 | -333 | *EST | 872-46-8 |
| $\text{C}_4\text{H}_6\text{F}_3\text{O}_2^+$ | | | | | | | | |
| | $\text{CF}_3\text{C}(\text{OH})\text{OC}_2\text{H}_5$ | | -68 | -284 | | | | |
| | | | From proton affinity of $\text{CF}_3\text{COOC}_2\text{H}_5$ (RN 383-63-1). PA = 184.6 kcal/mol, 772. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_6\text{N}^+$ | | | | | | | | |
| | $(\text{CH}_3)_2\text{CCN}$ | (8.2) | (229) | (960) | 40.3±2.2 | 168.6±9.2 | 82MCM/GOL | 3225-31-8 |
| | | IP is onset of photoelectron band. | | | | | | |
| |  | | 184 | 769 | | | | |
| | | From proton affinity of pyrrole (RN 109-97-7). PA = 207.6 kcal/mol, 868 kJ/mol. | | | | | | |
| |  | | 214 | 895 | | | | |
| | H^+ | From proton affinity of cyclopropylcarbonitrile (RN 5500-21-0). PA = 195.4 kcal/mol, 817.5 kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

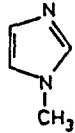
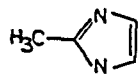
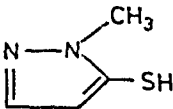
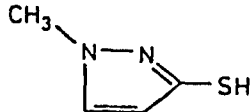
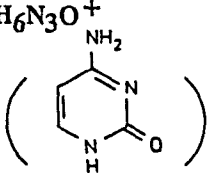
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|-----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_6\text{NO}_2^+$ (NCCOOC ₂ H ₅)H | | 134 | 562 | | | | |
| | | From proton affinity of NCCOOC ₂ H ₅ (RN 623-49-4). PA = 179.5 kcal/mol, 751. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_6\text{N}_2^+$ | | | | | | | |
|  | (≤ 8.66) | ≤ 236 | ≤ 986 | 36 | 150 | *EST | 616-47-7 |
|  | (≤ 8.50) | ≤ 225 | ≤ 942 | 29 | 122 | *EST | 693-98-1 |
| $\text{C}_4\text{H}_6\text{N}_2\text{S}^+$ | | | | | | | |
|  | ≤ 9.1 IP from 83GUI/PFI. | (≤ 265) | (≤ 1107) | 55 | 229 | *EST | |
|  | (≤ 8.6) IP from 83GUI/PFI. | (≤ 251) | (≤ 1049) | 52 | 219 | *EST | 79208-64-3 |
| $\text{C}_4\text{H}_6\text{N}_3\text{O}^+$ | | | | | | | |
|  | | 128 | 535 | | | | |
| | | From proton affinity of cytosine (RN 71-30-7). PA = 223.8 kcal/mol, 936. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_6\text{O}^+$ | | | | | | | |
| $\text{C}_2\text{H}_5\text{CH}=\text{C}=\text{O}$ | 8.80 IP from 81BOC/HIR. | (171) | (714) | -32 | -135 | *EST | 20334-52-5 |
| $(\text{CH}_3)_2\text{C}=\text{C}=\text{O}$ | (8.45) IP from 81BOC/HIR. | (163) | (681) | -32 ± 1 | -134 ± 4 | 80DEM/WUL | 598-26-5 |
| (E)- $\text{CH}_3\text{CH}=\text{CHCHO}$ | 9.73 ± 0.01 See also: 78VAN/OSK. | 200 | 835 | -24.8 ± 0.4 | -103.6 ± 1.5 | 79VAJ/HAR | 4170-30-3 |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{CHO}$ | (9.86) IP from 86HOL/LOS. | (199) | (834) | -28 | -117 | 79VAJ/HAR | 78-85-3 |

Table 1. Positive Ion Table - Continued



| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_6\text{O}^+$ | | | | | | | |
| $\text{CH}_2 = \text{CHCOCH}_3$ | 9.64 See also: 80TER/HEE, 82MOR/MER. | (189) | (792) | -33 | -138 | 79VAJ/HAR | 78-94-4 |
| $\text{CH}_3\text{C}=\text{COCH}_3$ | (8.79) IP from 86HOL/LOS. | (206) | (860) | 2.9 | 12.1 | *EST | 13169-01-2 |
| $\text{CH}_2 = \text{CHCH} = \text{CHOH}(\text{E})$ | (8.51±0.03) IP from 86TUR/HAV, 86TUR/HAV3. See also: 80TER/HEE. | (175) | (733) | -21±1 | -88±5 | 86TUR/HAV | 70411-98-2 |
| $\text{CH}_2 = \text{CHCH} = \text{CHOH}(\text{Z})$ | (8.47±0.03) IP from 86TUR/HAV, 86TUR/HAV3. See also: 80TER/HEE. | (174) | (728) | -21±2 | -89±9 | 86TUR/HEE | 70415-58-6 |
| $\text{CH}_2 = \text{C} = \text{CHCH}_2\text{OH}$ | (8.74) IP from 80TER/HEE. | (206) | (861) | 4.3 | 18.0 | *EST | 18913-31-0 |
| $\text{HC}=\text{CCH}_2\text{CH}_2\text{OH}$ | (9.66) IP from 86HOL/LOS. | (226) | (945) | 3.2 | 13.4 | *EST | 927-74-2 |
| $\text{CH}_3\text{C}=\text{CCH}_2\text{OH}$ | (9.78) IP from 86HOL/LOS. | (227) | (948) | 1.1 | 4.6 | *EST | 764-01-2 |
| $\text{HC}=\text{CCH}(\text{CH}_3)\text{OH}$ | (10.15) | (236) | (987) | 2 | 8 | *EST | 2028-63-9 |
| $\text{CH}_2=\text{CHC}(\text{OH})=\text{CH}_2$ | 8.68±0.03 $\Delta_f H(\text{Ion})$ from appearance potential determination(80TER/HEE). IP from 84TUR. | 182 | 761 | -18 | -76 | 84TUR | 59120-04-6 |
| $\text{CH}_2 = \text{C} = \text{CHOCH}_3$ | (8.64) IP from 86HOL/LOS, onset of photoelectron band (86KAM/BOS). | (207) | (866) | 7.7 | 32.2 | *EST | 13169-00-1 |
| $\text{HC}=\text{CCH}_2\text{OCH}_3$ | (9.78) IP from 86HOL/LOS. | (240) | (1005) | 14.7 | 61.5 | *EST | 627-41-8 |
| $\text{CH}_2 = \text{CHOCH} = \text{CH}_2$ | (8.7) $\Delta_f H(\text{Ion})$ from appearance potential determination (81HOL/BUR). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | (197) | (824) | -3 | -13 | *EST | 109-93-3 |
|  | 9.14±0.02 | 195 | 816 | -16±1 | -66±3 | 81ALL/GLA | 1708-29-8 |
|  | 9.354 | (194) | (814) | -21 | -89 | *EST | 1191-95-3 |

Table 1. Positive Ion Table - Continued

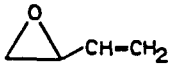
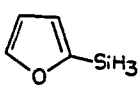


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₄H₆O⁺  | 9.52 IP from 86HOL/LOS. | (222) | (928) | 2 | 10 | *EST | 930-22-3 |
| C₄H₆OSi⁺  | (<8.0) IP is onset of photoelectron band (83ZYK/ERC). | (<183) | (<765) | -2 | -7 | *EST | 73726-79-1 |
| C₄H₆O₂⁺ (Z)-CH ₃ CH=CHCOOH | (10.08) | (150) | (626) | -83 | -346 | *EST | 503-64-0 |
| (E)-CH ₃ CH=CHCOOH | (9.9) IP is onset of photoelectron band (78VAN/OSK, 81MOH/HIR). | (145) | (605) | -84 | -350 | *EST | 107-93-7 |
| CH ₂ =CHCH ₂ COOH | (9.75) IP is onset of photoelectron band (81MOH/HIR). | (141) | (589) | -84 | -352 | *EST | 625-38-7 |
| CH ₂ =C(CH ₃)COOH | (10.15) | (146) | (610) | -88 | -369 | 84BOU/HOP | 79-41-4 3724-65-0 |
| CH ₃ CO ₂ CH=CH ₂ | 9.19 "Doubtful" IP value reported in K. Watanabe, T. Nakayama, and J. R. Mottl, J. Quant. Spectrosc. Radiat. Transfer 2, 369 (1962) is in good agreement with onset of photoelectron band (78VAN/OSK). See also: 82LEV/LIA. | 137 | 572 | -75.3±0.1 | -314.9±0.5 | 77PED/RYL | 108-05-4 |
| CH ₂ =CHCOOCH ₃ | (9.9) IP is onset of photoelectron band (78VAN/OSK). See also: 82LEV/LIA. | (154) | (643) | -75 | -312 | 80VIL/PER | 96-33-3 |
| (CH ₃ CO) ₂ | 9.24±0.04 See also: 80VON/BIE, 81KIM/KAT. | 135 | 564 | -78.2±0.3 | -327.1±1.1 | 77PED/RYL | 431-03-8 |
|  | (9.5) IP is onset of photoelectron band (81KIM/KAT). | (224) | (938) | 5 | 21 | *EST | 18715-02-1 |
|  | (8.07±0.02) | (131) | (549) | -55 | -230 | *EST | 543-75-9 |

Table 1. Positive Ion Table - Continued


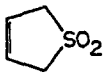
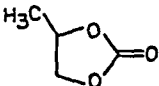

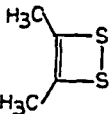
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_6\text{O}_2^+$ COOH  | 10.64 | (167) | (699) | -78 | -328 | *EST | 1759-53-1 |
| $\text{C}_4\text{H}_6\text{O}_2\text{S}^+$ $(\text{CH}_2=\text{CH})_2\text{SO}_2$ | 10.59±0.03 | 208 | 871 | -36±0.9 | -151±4 | 77PED/RYL | 77-77-0 |
|  | (10.0) | (169) | (709) | -61±0.7 | -256±3 | 77PED/RYL | 77-79-2 |
| IP is onset of photoelectron band (82LEV/LIA, 84AIT/GOS). | | | | | | | |
| $\text{C}_4\text{H}_6\text{O}_3^+$ $(\text{CH}_3\text{CO})_2\text{O}$ | (10.0) | (95) | (398) | -135.6±0.3 | -567.3±1.3 | 77PED/RYL | 108-24-7 |
| IP is onset of photoelectron band (81BOC/HIR). | | | | | | | |
|  | (10.52) | (103) | (432) | -139±0.5 | -583±2 | 77PED/RYL | 108-32-7 |
| $\text{C}_4\text{H}_6\text{O}_4^+$ $\text{CH}_3\text{OCOCOOCH}_3$ | (10.0) | (69) | (289) | -162 | -676 | 76ANT/CAR | 553-90-2 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_4\text{H}_6\text{S}^+$ $(\text{CH}_2=\text{CH})_2\text{S}$ | (8.25±0.01) | (232) | (970) | 42±2 | 174±9 | *EST | 627-51-0 |
|  | (8.4) | (215) | (897) | 20.8±0.3 | 87.0±1.1 | 81KUD/KUD3 | 1708-32-3 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_4\text{H}_6\text{S}_2^+$ $\text{CH}_3\text{SC}\equiv\text{CSCH}_3$ | (7.8) | (238) | (995) | 58 | 242 | *EST | 59507-56-1 |
| IP is onset of photoelectron band (81BOC/RIE). | | | | | | | |
|  | (8.0) | (231) | (966) | 46 | 194 | *EST | 74378-81-7 |
| IP is onset of photoelectron band (83SCH/SCH). | | | | | | | |

Table 1. Positive Ion Table - Continued

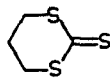
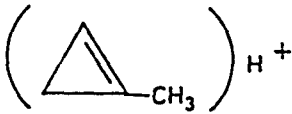

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|--|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_6\text{S}_3^+$ |  | (8.2) | (208) | (869) | 19±0.7 | 78±3 | 77PED/RYL | 1748-15-8 |
| | | | | | | | | IP is onset of photoelectron band. |
| C_4H_7^+ | | | | | | | | |
| | $\text{CH}_3\text{CHCH}=\text{CH}_2$ | (7.49±0.02) | 202 | 845 | 31.7 | 132.6 | 87LIA/AUS | 65338-31-0 |
| | | | 206 | 863 | 35.7 | 149.4 | | |
| | | | | | | | | IP from 84SCH/HOU2. Value of $\Delta_f H(\text{Ion})$ from proton affinity of 1,3-butadiene (RN 106-99-0). PA = 190 kcal/mol, 795 kJ/mol (87LIA/AUS). See also: 86TRA. |
| | $\text{CH}_2\text{C}(\text{CH}_3)=\text{CH}_2$ | 7.90±0.02 | (211) | (883) | 29 | 121 | 87LIA/AUS | 15157-95-6 |
| | | | | | | | | IP from 84SCH/HOU2. |
| |  | | (218) | (912) | | | | 65338-31-0 |
| | | | | | | | | From proton affinity of 1-methylcyclopropene (RN 3100-04-7). PA = (206) kcal/mol, (862) kJ/mol. |
| | $\text{CH}_3\text{CCHCH}_3$ | | (213) | (893) | | | | |
| | | | (217) | (908) | | | | |
| | | | | | | | | From proton affinity of 2-butyne. (RN 503-17-3). PA = 188 kcal/mol, 787 kJ/mol (87LIA/AUS). |
| | $\text{CH}_2=\text{CHCH}_2\text{CH}_2$ | 8.04 | (231) | (968) | 46 | 191 | 84SCH/HOU2 | 2154-62-3 |
| | | | | | | | | IP from 84SCH/HOU2. |
| |  | 7.54±0.02 | (225) | (941) | 51.2 | 214.2 | 82MCM/GOL | 4548-06-5 |
| | | | | | | | | IP from 84SCH/HOU2. Value of $\Delta_f H(\text{Ion})$ formed by protonation of cyclobutene (RN 822-35-5) = 212 kcal/mol, 888 kJ/mol. |
| $\text{C}_4\text{H}_7\text{F}_3\text{O}^+$ | $\text{CF}_3\text{CH}_2\text{OC}_2\text{H}_5$ | 10.27 | (21) | (86) | -216 | -905 | *EST | 461-24-5 |
| | | | | | | | | IP from 83MOL/PIK. |
| $\text{C}_4\text{H}_7\text{IO}_2^+$ | $\text{CH}_3\text{CHICOOCH}_3$ | (9.1) | (122) | (510) | -88 | -368 | *EST | 56905-18-1 |
| | | | | | | | | IP from 83BUR/HOL3. |
| $\text{C}_4\text{H}_7\text{N}^+$ | | | | | | | | |
| | n- $\text{C}_3\text{H}_7\text{CN}$ | (11.2) | (266) | (1112) | 7 | 31 | 82CHU/NGU | 109-74-0 |
| | | | | | | | | IP is onset of photoelectron band (84OHN/MAT, 81KIM/KAT). See also: 82CHE/LAP. |
| | n- $\text{C}_3\text{H}_7\text{NC}$ | (11.8) | (302) | (1262) | 29.5 | 123.4 | *EST | 627-36-1 |
| | | | | | | | | IP from 82CHE/LAP. |

Table 1. Positive Ion Table - Continued

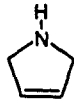

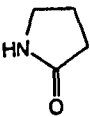
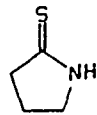

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-------------------------------------|---|----------------------------|--------------------------|--------|------------------------------|----------------|----------------------|---|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_7\text{N}^+$ | iso- $\text{C}_3\text{H}_7\text{CN}$ | (11.3) | (266) | (1115) | 5.8 ± 0.2 | 24.5 ± 0.7 | 77PED/RYL | 78-82-0 |
| | | | | | | | | IP is onset of photoelectron band (84OHN/MAT, 81KIM/KAT). |
| |  | (8.0) | (210) | (882) | 26 ± 0.7 | 110 ± 3 | *EST | 109-96-6 |
| | | | | | | | | IP is onset of photoelectron band. |
| |  | (8.2) | (251) | (1048) | 61 | 257 | *EST | |
| | | | | | | | | IP from 81MUL/PRE. |
| $\text{C}_4\text{H}_7\text{NO}^+$ |  | (9.2) | (161) | (675) | -51 | -213 | 77PED/RYL | 616-45-5 |
| | | | | | | | | IP is onset of photoelectron band (80AND/DEV). |
| $\text{C}_4\text{H}_7\text{NO}_2^+$ | $\text{CH}_2 = \text{CHCH}_2\text{CH}_2\text{ONO}$ | (9.7) | (224) | (939) | 0.7 | 3 | *EST | 67428-02-8 |
| | | | | | | | | IP is onset of photoelectron band. |
| $\text{C}_4\text{H}_7\text{NO}_3^+$ | $\text{CH}_3\text{CONHCH}_2\text{COOH}$ | (9.4) | (72) | (303) | -144 | -604 | *EST | 543-24-8 |
| | | | | | | | | IP is onset of photoelectron band (83CAN/HAM). |
| $\text{C}_4\text{H}_7\text{NS}^+$ |  | (8.14) | (192) | (801) | 4 | 16 | *EST | 2295-35-4 |
| | | | | | | | | IP is onset of photoelectron band (80AND/DEV). |
| $\text{C}_4\text{H}_7\text{NSe}^+$ |  | 7.6 | (196) | (819) | 21 | 86 | *EST | 23164-74-1 |
| | | | | | | | | IP is onset of photoelectron band (80AND/DEV). |

Table 1. Positive Ion Table - Continued

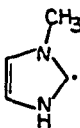
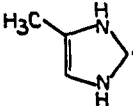
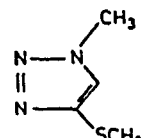
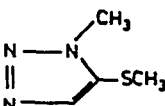
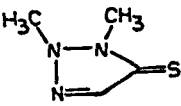

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|---|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_7\text{N}_2^+$  | | (173) | (723) | | | | |
| | | From proton affinity of 1-methylimidazole (RN 616-47-7). PA = 228.9 kcal/mol, 958. kJ/mol. | | | | | |
|  | | (170) | (713) | | | | |
| | | From proton affinity of 4-methylimidazole (RN 822-36-6). PA = 224.4 kcal/mol, 939. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_7\text{N}_3\text{S}^+$ | | | | | | | |
|  | ≤ 8.33 IP from 83GUI/PFI. | (≤ 250) | (≤ 1047) | 58 | 243 | *EST | 36811-14-0 |
|  | ≤ 8.65 IP from 83GUI/PFI. | (≤ 258) | (≤ 1077) | 58 | 243 | *EST | 35262-23-8 |
|  | (7.4) IP from 83GUI/PFI. | (259) | (1084) | 88 | 370 | *EST | 64808-28-2 |
| $\text{C}_4\text{H}_7\text{O}^+$ (E)- $\text{CH}_3\text{CHCHCHOH}$ | | 141 | 591 | | | | |
| | | From proton affinity of (E)- $\text{CH}_3\text{CH}=\text{CHCHO}$ (RN 4170-30-3). PA = 199.7 kcal/mol, 835.5 kJ/mol. | | | | | |
| $\text{CH}_2\text{C}(\text{CH}_3)\text{CHOH}$ | | 142 | 596 | | | | |
| | | From proton affinity of $\text{CH}_2=\text{C}(\text{CH}_3)\text{CHO}$ (RN 78-85-3). PA = 195.2 kcal/mol, 817. kJ/mol. | | | | | |
| $\text{CH}_2\text{CHC}(\text{OH})\text{CH}_3$ | | 133 | 554 | | | | |
| | | From proton affinity of $\text{CH}_2=\text{CHCOCH}_3$ (RN 78-94-4). PA = 200.2 kcal/mol, 838. kJ/mol. | | | | | |
|  | | 141 | 589 | | | | |
| | | From proton affinity of 2,3-dihydrofuran (RN 1191-99-7) (86BOU/DJA). PA = 206.8 kcal/mol, 865 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued


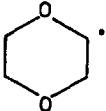
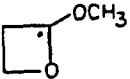

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_7\text{O}^+$  | | 152 | 634 | | | | |
| | | From proton affinity of 2,5-dihydrofuran (RN 1708-29-8) (86BOU/DJA). PA = 198.4 kcal/mol, 830. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_7\text{O}_2^+$ (Z)- $\text{CH}_3\text{CH}=\text{CHC}(\text{OH})_2$ | | 78 | 327 | | | | |
| | | From proton affinity of (Z)- $\text{CH}_3\text{CH}=\text{CHCOOH}$ (RN 503-64-0)(84BOU/HOP). PA = 199.7 kcal/mol, 836. kJ/mol. | | | | | |
| $\text{CH}_3\text{C}(\text{CH}_2)\text{C}(\text{OH})_2$ | | 81 | 338 | | | | |
| | | From proton affinity of $\text{CH}_3\text{C}(\text{=CH}_2)\text{COOH}$ (RN 3724-65-0)(84BOU/HOP). PA = 196.8 kcal/mol, 823. kJ/mol. | | | | | |
| $\text{CH}_2=\text{CHC}(\text{OH})\text{OCH}_3$ | | 92 | 386 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (83BUR/HOL3). | | | | | |
| $\text{CH}_3\text{CO}_2\text{CHCH}_3$ | | 94 | 392 | | | | |
| | | From proton affinity of $\text{CH}_3\text{CO}_2\text{CH}=\text{CH}_2$ (RN 108-05-4) (86MAU). PA = 196.7 kcal/mol, 823. kJ/mol. | | | | | |
| $\text{CH}_3\text{CHCOOCH}_3$ | | 115 | 480 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (83BUR/HOL3). | | | | | |
| $\text{CH}_3\text{COC}(\text{OH})\text{CH}_3$ | | 93 | 388 | | | | |
| | | From proton affinity of $(\text{CH}_3\text{CO})_2$ (RN 431-03-8). PA = 194.8 kcal/mol, 815. kJ/mol. | | | | | |
|  | | 126 | 529 | | | | |
| | | From proton affinity of dihydro-1,4-dioxin (RN 543-75-9) (86BOU/HAN). PA = 198.4 kcal/mol, 830. kJ/mol. | | | | | |
|  | | 108 | 450 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (83BUR/HOL3). | | | | | |
| $\text{C}_4\text{H}_7\text{O}_3\text{P}^+$  | (9.42±0.1) | (89) | (371) | -129 | -538 | *EST | 280-45-5 |

Table 1. Positive Ion Table - Continued



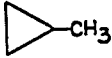
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_7\text{S}^+$ | | | | | | | |
|  | | 237 | 992 | | | | 39925-70-7 |
| | | 214 | 895 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential in tetrahydrothiophene (83BUT/BAE2). | | | | | |
| C_4H_8^+ | | | | | | | |
| 1- C_4H_8 | 9.58±0.02 | 221 | 924 | -0.1±0.1 | -0.4±0.5 | 77PED/RYL | 106-98-9 |
| | See also: 83HOL/LOS, 86TRA. | | | | | | |
| (Z)-2- C_4H_8 | 9.108±0.008 | 208 | 871 | -1.9±0.1 | -7.8±0.5 | 77PED/RYL | 590-18-1 |
| | IP from 78LIA/AUS. See also: 81KIM/KAT, 86TRA. | | | | | | |
| (E)-2- C_4H_8 | 9.100±0.008 | 207 | 866 | -2.9±0.2 | -12.2±0.5 | 77PED/RYL | 624-64-6 |
| | IP from 78LIA/AUS. See also: 81KIM/KAT, 86TRA. | | | | | | |
| iso- C_4H_8 | 9.239±0.003 | 209 | 874 | -4.0±0.1 | -16.9±0.6 | 77PED/RYL | 115-11-7 |
| | See also: 83HOL/LOS, 81KIM/KAT, 86TRA. | | | | | | |
|  | (9.92±0.05) | (235) | (985) | 6.8±0.2 | 28.4±0.5 | 77PED/RYL | 287-23-0 |
|  | (9.46) | (224) | (938) | 5.5 | 23 | 77PED/RYL | 594-11-6 |
| $\text{C}_4\text{H}_8\text{Br}_2^+$ | | | | | | | |
| $\text{CH}_3\text{CHBrCHBrCH}_3$ -(R,R(±)) | (≤10.12) | (≤206) | (≤860) | -28 | -116 | *EST | 598-71-0 |
| $\text{CH}_3\text{CHBrCHBrCH}_3$ -(R,S) | (≤10.16) | (≤207) | (≤864) | -28 | -116 | *EST | 5780-13-2 |
| $\text{BrCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{Br}$ | (10.15) | (210) | (880) | -24 | -99 | 77PED/RYL | 110-52-1 |
| | IP from 77STA/WIE. | | | | | | |
| $\text{C}_4\text{H}_8\text{F}_3\text{N}^+$ | | | | | | | |
| $\text{CF}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2$ | (9.1) | (29) | (123) | -180 | -755 | *EST | 819-46-5 |
| | IP from 79AUE/BOW. | | | | | | |
| $(\text{CH}_3)_2\text{NCH}_2\text{CF}_3$ | (8.42) | (27) | (112) | -167 | -700 | 81LOG/TAK | 819-06-7 |
| | IP from 81LOG/TAK. See also: 79AUE/BOW. | | | | | | |

Table 1. Positive Ion Table - Continued

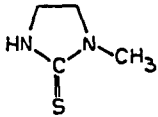
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_8\text{F}_3\text{O}^+$ | $\text{C}_2\text{H}_5\text{OHCH}_2\text{CF}_3$ | | -37 | -154 | | | | |
| | | | From proton affinity of $\text{C}_2\text{H}_5\text{OCH}_2\text{CF}_3$ (RN 461-24-5). PA = 186.4 kcal/mol, 780. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_8\text{N}^+$ | n- $\text{C}_3\text{H}_7\text{CNH}$ | | 179 | 751 | | | | |
| | | | From proton affinity of n- $\text{C}_3\text{H}_7\text{CN}$ (RN 109-74-0). PA = 193.7 kcal/mol, 810. kJ/mol. | | | | | |
| | i- $\text{C}_3\text{H}_7\text{CNH}$ | | 177 | 740 | | | | |
| | | | From proton affinity of i- $\text{C}_3\text{H}_7\text{CN}$ (RN 78-82-0). PA = 194.3 kcal/mol, 813. kJ/mol. | | | | | |
| | i- $\text{C}_3\text{H}_7\text{NCH}$ | | 186 | 778 | | | | |
| | | | From proton affinity of i- $\text{C}_3\text{H}_7\text{NC}$ (RN 598-45-8) (86MAU/KAR). PA = 206. kcal/mol, 862. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_8\text{NO}_4^+$ | $\text{HOOCCH}_2\text{CH}(\text{NH}_3)\text{COOH}$ | | -44 | -184 | | | | |
| | | | From proton affinity of L-aspartic acid (RN 617-45-8). PA = 216.7 kcal/mol, 907. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_8\text{N}_2^+$ | $(\text{CH}_3)_2\text{NCH}_2\text{CN}$ | (8.72±0.05) | (228) | (953) | 27 | 112 | *EST | 926-64-7 |
| | | | See also: 83MOL/PIK2. | | | | | |
| $\text{C}_4\text{H}_8\text{N}_2\text{OS}^+$ | $(\text{CH}_3)_2\text{NCSOCNH}_2$ | (≤8.21) | (≤171) | (≤714) | -19 | -78 | *EST | 41168-96-1 |
| | | | IP from 81HEN/ISA. | | | | | |
| | $(\text{CH}_3)_2\text{NCOCSNH}_2$ | ≤8.37 | (≤168) | (≤704) | -25 | -104 | *EST | 18138-14-2 |
| | | | IP from 81HEN/ISA. | | | | | |
| $\text{C}_4\text{H}_8\text{N}_2\text{O}_2^+$ | $\text{CH}_3\text{NHCOCONHCH}_3$ | (9.33) | (121) | (504) | -95 | -396 | *EST | 615-35-0 |
| $\text{C}_4\text{H}_8\text{N}_2\text{S}^+$ |  | (7.7) | (201) | (842) | 24 | 99 | *EST | 13431-10-2 |
| | | | IP is onset of photoelectron band (80AND/DEV). | | | | | |
| $\text{C}_4\text{H}_8\text{N}_2\text{S}_2^+$ | $\text{CH}_3\text{NHCSCSNHCH}_3$ | ≤8.23 | (≤163) | (≤684) | -26 | -110 | *EST | 120-79-6 |
| | | | IP from 81HEN/ISA. | | | | | |
| $\text{C}_4\text{H}_8\text{N}_4^+$ | $\text{NCN}=\text{C}(\text{NHCH}_3)_2$ | (8.5) | (234) | (977) | 38 | 157 | *EST | 31857-31-5 |
| | | | IP is onset of photoelectron band (80KLA/BUT). | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|-------------------|---------------------|
| | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_8\text{O}^+$ | | | | | | | |
| n- $\text{C}_3\text{H}_7\text{CHO}$ | 9.84±0.02 | 177 | 742 | -49.6±0.4 | -207.5±1.5 | 77PED/RYL | 123-72-8 |
| | See also: 81ELS/ALL, 83MCA/HUD, 81KIM/KAT, 86TRA/MCA. | | | | | | |
| iso- $\text{C}_3\text{H}_7\text{CHO}$ | 9.705±0.005 | 172 | 721 | -51.5±0.1 | -215.6±0.6 | 77PED/RYL | 78-84-2 |
| | See also: 83MCA/HUD, 86TRA/MCA. | | | | | | |
| $\text{C}_2\text{H}_5\text{COCH}_3$ | 9.51±0.04 | 162 | 677 | -57.5±0.1 | -240.8±0.6 | 77PED/RYL | 78-93-3 |
| | See also: 72POT/SOR, 85TRA, 81KIM/KAT. | | | | | | |
| $\text{CH}_3\text{CH}_2\text{CH}=\text{CHOH}$ | (8.34±0.05) | (150) | (628) | -42 | -177 | *EST | 56640-69-8 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$ (83HOL/LOS). See also: 83MCA/HUD. | | | | | | |
| (E)- $\text{CH}_3\text{CH}=\text{CHCH}_2\text{OH}$ | (9.13±0.02) | (173) | (726) | -37 | -155 | *EST | |
| | IP from 86TRA/MCA. See also: 83MCA/HUD. | | | | | | |
| $\text{CH}_2=\text{CHCH}_2\text{CH}_2\text{OH}$ | (9.56±0.05) | (184) | (770) | -36 | -152 | *EST | 627-27-0 |
| | IP from 83HOL/LOS. | | | | | | |
| $\text{CH}_3\text{CHCH}_2\text{CHOH}$ | | (165) | (690) | | | | |
| | Based on appearance energy measurements of metastable processes (83MCA/HUD). | | | | | | |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{OH}$ | (9.26±0.02) | (176) | (734) | -38 | -159 | *EST | 513-42-8 |
| | IP is average of values from 83HOL/LOS and 86TRA/MCA. | | | | | | |
| $(\text{CH}_3)_2\text{C}=\text{CHOH}$ | (8.27±0.05) | (145) | (607) | -46 | -192 | *EST | 56640-70-1 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$ (83HOL/LOS). | | | | | | |
| $\text{CH}_2\text{CH}(\text{CH}_3)\text{CHOH}$ | | (154) | (644) | | | | |
| | Based on appearance energy measurements of metastable processes (83MCA/HUD). | | | | | | |
| $\text{CH}_3\text{CH}_2\text{C}(\text{OH})=\text{CH}_2$ | (8.36±0.05) | (150) | (628) | -43 | -179 | *EST | 61923-55-5 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$ (83HOL/LOS). See also: 83MCA/HUD. | | | | | | |
| $\text{CH}_3\text{C}(\text{OH})=\text{CHCH}_3$ | | 139 | 581 | | | | 21411-38-1 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations. See also: 83MCA/HUD. | | | | | | |
| $\text{CH}_2=\text{CHCH}(\text{OH})\text{CH}_3$ | 9.50±0.05 | (180) | (756) | -38 | -161 | *EST | 598-32-3 |
| | IP from 83MCA/HUD, 83HOL/LOS, 86TRA/MCA. | | | | | | |
| $\text{CH}_3\text{C}(\text{OH})\text{CH}_2\text{CH}_2$ | | (147) | (613) | | | | |
| | Based on appearance energy measurements of metastable processes (83MCA/HUD). | | | | | | |
| $\text{CH}_2=\text{CHCH}_2\text{OCH}_3$ | (9.56) | (195) | (817) | -25 | -105 | *EST | 627-40-7 |
| | IP from 86HOL/LOS. | | | | | | |

Table 1. Positive Ion Table - Continued


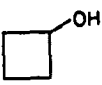
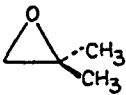

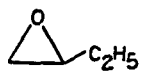

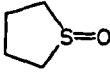
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₄H₈O⁺ | | | | | | | |
| CH ₂ =CHOC ₂ H ₅ | (8.8) IP from 86HOL/LOS. See also: 82MOR/MER. | (169) | (708) | -34 | -141 | 77PED/RYL | 109-92-2 |
| CH ₂ =C(CH ₃)OCH ₃ | (8.64) IP from 82HOL/LOS2. | (164) | (688) | -35 | -146 | *EST | 116-11-0 |
|  | 9.41±0.02 See also: 81KIM/KAT. | 173 | 724 | -44.0±0.2 | -184.2±0.7 | 77PED/RYL | 109-99-9 |
|  | 9.25 IP from 83MCA/HUD, 86TRA/MCA. | (181) | (756) | -32 | -136 | *EST | 2919-23-5 |
|  | (10.00) | (198) | (830) | -32 | -135 | *EST | 558-30-5 |
|  | (9.98) | (199) | (832) | -31 | -131 | *EST | 21490-63-1 |
|  | (10.15) | (206) | (864) | -28 | -115 | *EST | 106-88-7 |
| C₄H₈OS⁺ | | | | | | | |
| CH ₃ COSC ₂ H ₅ | (9.2) IP is onset of photoelectron band. | (158) | (660) | -54±0.2 | -228±1 | 66WAD | 625-60-5 |
|  | (8.67) | (164) | (688) | -36 | -149 | *EST | 15980-15-1 |
|  | 8.5 IP is onset of photoelectron band. | (161) | (674) | -35 | -146 | *EST | 1600-44-8 |

Table 1. Positive Ion Table - Continued



| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_8\text{O}_2^+$ | | | | | | | |
| n-C ₃ H ₇ COOH | 10.17±0.05 | 121 127 | 507 533 | -113±1 -107 | -473±4 -447 | 82BUT/FRA | 107-92-6 |
| | | See also: 82BUT/FRA, 81HOL/FIN. | | | | | |
| iso-C ₃ H ₇ COOH | 10.33±0.03 | (123) | (517) | -115 | -480 | *EST | 79-31-2 |
| HCOOCH ₂ CH ₂ CH ₃ | 10.52±0.02 | 132 | 553 | -110 | -462 | 77PED/RYL | 110-74-7 |
| HCOOCH(CH ₃) ₂ | 10.44±0.05 | (144) | (602) | -97 | -405 | *EST | 625-55-8 |
| CH ₃ COOC ₂ H ₅ | 10.01±0.05 | 125 131 | 523 548 | -106.1±0.1 -99.9±0.1 | -443.9±0.4 -418.0±0.4 | 77PED/RYL | 141-78-6 |
| | | IP from 82FRA/FRA2. | | | | | |
| C ₂ H ₅ COOCH ₃ | 10.15±0.03 | (131) | (547) | -103 | -432 | *EST | 554-12-1 |
| CH ₃ CH ₂ CH=C(OH) ₂ | | 97 | 405 | | | | 12542-32-4 |
| | | From appearance potential of 10.14 eV in (C ₂ H ₅) ₂ CHCOOH (RN 88-09-5). | | | | | |
| (CH ₃) ₂ C=C(OH) ₂ | | 92 | 387 | | | | |
| | | From appearance potentials of 10.02 eV in C ₂ H ₅ C(CH ₃) ₂ COOH (RN 595-37-9) and 9.96 eV in n-C ₃ H ₇ C(CH ₃) ₂ COOH (RN 1185-39-3). | | | | | |
| CH ₂ =C(OH)OC ₂ H ₅ | | 104 | 433 | | | | |
| | | From appearance potential of 10.06 eV in n-C ₃ H ₇ COOC ₂ H ₅ (RN 105-54-4) and 9.96 eV in n-C ₅ H ₁₁ COOC ₂ H ₅ (RN 123-66-0). | | | | | |
| CH ₃ CH=C(OH)OCH ₃ | | 99 | 413 | | | | |
| | | From appearance potential of 9.81 eV in sec-C ₄ H ₉ COOCH ₃ (RN 868-57-5). | | | | | |
| CH ₃ COCH ₂ OCH ₃ | ≤9.66 | (≤143) | (≤598) | -80 | -334 | *EST | 5878-19-3 |
| | | IP from 84OLI/GUE. | | | | | |
|  | (≤10.0) | (≤195) | (≤816) | -36 | -149 | *EST | 5703-46-8 |
|  | 9.8 | 145 | 608 | -81±0.2 | -338±1 | 82BYS/MAN | 505-22-6 |
| | | See also: 84ASF/ZYK. | | | | | |

Table 1. Positive Ion Table - Continued


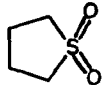
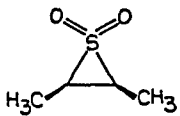

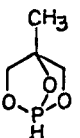
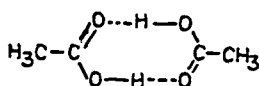
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_8\text{O}_2^+$ | | | | | | | |
|  | 9.19±0.01 | 136 144 | 571 602 | -75.5±0.2 -68.2±0.2 | -316.0±0.7 -285.3±0.8 | 82BYS/MAN | 123-91-1 |
| | | IP from 82FRA/FRA. See also: 81KIM/KAT, 73GOL/KOR, 82BIE/ASB. | | | | | |
| $\text{C}_4\text{H}_8\text{O}_2\text{S}^+$ | | | | | | | |
|  | (9.8) | (138) | (577) | -88 | -369 | *EST | 126-33-0 |
| | IP is onset of photoelectron band. See also: 84AIT/GOS. | | | | | | |
|  | (9.5) | (142) | (593) | -77 | -324 | *EST | 54697-52-8 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_4\text{H}_8\text{O}_3^+$ | | | | | | | |
| $(\text{CH}_3)_2\text{COHCOOH}$ | ≤10.9 | (≤96) | (≤404) | -155 | -648 | *EST | 594-61-6 |
| | IP from 73GOL/KOR. | | | | | | |
| $\text{C}_4\text{H}_8\text{O}_3\text{P}^+$ | | | | | | | |
|  | | 30 | 126 | | | | |
| | From proton affinity of 2,6,7-trioxa-1-phosphabicyclo[2.2.2]octane (RN 280-45-5). PA = 207.1 kcal/mol, 866.5 kJ/mol. | | | | | | |
|  | | 13 | 55 | | | | |
| | From proton affinity of 4-methyl-2,6,7-trioxa-1-phosphabicyclo[2.2.1]heptane (RN 61580-09-4). PA = 198.1 kcal/mol, 829. kJ/mol. | | | | | | |
| $\text{C}_4\text{H}_8\text{O}_4^+$ | | | | | | | |
|  | (≤10.6) | (≤22) | (≤94) | -222 | -929 | *EST | 6993-75-5 |
| $\text{C}_4\text{H}_8\text{S}^+$ | | | | | | | |
| $\text{CH}_3\text{SCH}_2\text{CH}=\text{CH}_2$ | 8.6 | (210) | (880) | 12±2 | 50±9 | *EST | 10152-76-8 |
| $\text{CH}_2=\text{CHSC}_2\text{H}_5$ | (8.21±0.01) | (201) | (840) | 11±1 | 48±6 | *EST | 627-50-9 |

Table 1. Positive Ion Table - Continued






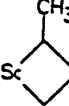
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|----------------------------|--------------------------|----------------|------------------------------|-------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_8\text{S}^+$ | | | | | | | | |
|  | | 8.47 | 187.1 193.8 | 782.8 810.8 | -8.2±0.2 -1.5 | -34.1±0.9 -6.2 | 81KUD/KUD3 | 110-01-0 |
| Results from 83BUT/BAE2. | | | | | | | | |
| $\text{C}_4\text{H}_8\text{S}_2^+$ | | | | | | | | |
| (Z)- $\text{CH}_3\text{SCH}=\text{CHSCH}_3$ | | (≤7.80) | (≤203) | (≤849) | 23 | 96 | *EST | 764-44-3 |
| (E)- $\text{CH}_3\text{SCH}=\text{CHSCH}_3$ | | (≤7.85) | (≤204) | (≤853) | 23 | 96 | *EST | 764-45-4 |
| $\text{CH}_2=\text{C}(\text{SCH}_3)_2$ | | (≤8.2) | (≤212) | (≤887) | 23 | 96 | *EST | 51102-74-0 |
|  | | 8.1 | (178) | (746) | -9 | -36 | *EST | 505-20-4 |
| IP is onset of photoelectron band. | | | | | | | | |
|  | | 8.2 | (188) | (786) | -1 | -5 | *EST | 505-23-7 |
| IP is onset of photoelectron band. | | | | | | | | |
|  | | (8.4) | (193) | (805) | -1 | -5 | *EST | 505-29-3 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_4\text{H}_8\text{S}_4^+$ | | | | | | | | |
|  | | (7.8) | (197) | (825) | 17 | 72 | *EST | 2373-00-4 |
| IP is onset of photoelectron band (81BOC/SCH). | | | | | | | | |
| $\text{C}_4\text{H}_8\text{Sc}^+$ | | | | | | | | |
| (E)- $\text{CH}_3\text{CH}=\text{CHCH}_3$ Sc | | | (191) | (799) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | | | | | | |
|  | | | (179) | (749) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|---|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₄H₉⁺ | | | | | | | |
| n-C ₄ H ₉ | 8.02 | (203) | (849) | 18 | 74 | *EST | 2492-36-6 |
| | IP from 84SCH/HOU. Error limits +0.04-0.1. $\Delta_f H(\text{Neutral})$ based on D[C-H] = 100.5 kcal/mol. | | | | | | |
| sec-C ₄ H ₉ | 7.25±0.02 | 183 | 766 | 17.0±0.4 | 71.0±1.6 | 85TSA | 2548-55-2 4630-45-9 |
| | IP from 84SCH/HOU. Heat of formation of ion at 298 K from appearance potential measurements (81TRA). $\Delta_f H(\text{Neutral})$ based on D[C-H] = 99 kcal/mol. $\Delta_f H(\text{Ion})$ - IP leads to $\Delta_f H(\text{Neutral})$ = 16 kcal/mol, 66 kJ/mol. | | | | | | |
| iso-C ₄ H ₉ | 7.93 | (199) | (832) | 16 | 70 | 81TSA | 1630-45-9 65114-21-8 |
| | IP from 84SCH/HOU. Error limits +0.03-0.1. $\Delta_f H(\text{Neutral})$ based on D[C-H] = 100.5 kcal/mol. | | | | | | |
| tert-C ₄ H ₉ | 6.70±0.03 | 165.8 | 693.7 | 11.0±0.6 | 46.2±2.5 | 85TSA | 1605-73-8 |
| | Same value is obtained for heat of formation of ion from appearance potential measurements (81TRA/MCL). Neutral $\Delta_f H$ based on D[C-H] = 95.5 kcal/mol. $\Delta_f H(\text{Ion})$ - IP leads to $\Delta_f H(\text{Neutral})$ = 11.3 kcal/mol, 47.2 kJ/mol. | | | | | | |
| C₄H₉Br⁺ | | | | | | | |
| n-C ₄ H ₉ Br | 10.13 | 208 | 870 | -25.6±0.3 | -107.1±1.3 | 77PED/RYL | 109-65-9 |
| | See: 81KIM/KAT. | | | | | | |
| sec-C ₄ H ₉ Br | 9.98±0.01 | 201 | 842 | -28.9±0.1 | -120.9±0.4 | 77PED/RYL | 78-76-2 |
| | See also: 81TRA, 81KIM/KAT. | | | | | | |
| iso-C ₄ H ₉ Br | 10.09±0.02 | (205) | (858) | -27 | -115 | *EST | 78-77-3 |
| | See: 81KIM/KAT. | | | | | | |
| tert-C ₄ H ₉ Br | 9.92±0.03 | 197 | 824 | -32 | -133 | 79WIB/SQU | 507-19-7 |
| | See: 81KIM/KAT. | | | | | | |
| C₄H₉Cl⁺ | | | | | | | |
| n-C ₄ H ₉ Cl | 10.67±0.03 | 209 | 874 | -36.9±0.2 | -154.5±1 | 78SEL/STR | 109-69-3 |
| | See also: 81KIM/KAT. | | | | | | |
| sec-C ₄ H ₉ Cl | 10.53 | 204 | 855 | -38±2 | -161±8 | 77PED/RYL | 78-86-4 |
| | See also: 81KIM/KAT. | | | | | | |
| iso-C ₄ H ₉ Cl | 10.66±0.03 | 208 | 869 | -38±2 | -159±8 | 77PED/RYL | 513-36-0 |
| | See also: 81KIM/KAT. | | | | | | |
| tert-C ₄ H ₉ Cl | 10.61±0.03 | 201 | 842 | -43.5±0.3 | -182.1±1.2 | 77PED/RYL | 507-20-0 |
| | See also: 81KIM/KAT. | | | | | | |
| C₄H₉ClHg⁺ | | | | | | | |
| n-C ₄ H ₉ HgCl | ≤10.08 | (≤206) | (≤864) | -26 | -109 | *EST | 543-63-5 |
| | IP from 81BAI/CHI2. | | | | | | |
| sec-C ₄ H ₉ HgCl | 9.5 | (194) | (814) | -25 | -103 | *EST | 38455-12-8 |
| | IP is onset of photoelectron band (81BAI/CHI2). | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------------------------|----------------|---|-----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_9\text{ClHg}^+$ | iso- $\text{C}_4\text{H}_9\text{HgCl}$ | ≤ 10.04 IP from 81BAI/CHI2. | (≤ 204) | (≤ 852) | -28 | -117 | *EST | 27151-74-2 |
| | tert- $\text{C}_4\text{H}_9\text{HgCl}$ | ≤ 9.52 IP from 81BAI/CHI2. | (≤ 198) | (≤ 830) | -21 | -89 | *EST | 38442-51-2 |
| $\text{C}_4\text{H}_9\text{ClO}^+$ | tert- $\text{C}_4\text{H}_9\text{OCl}$ | ≤ 9.91 IP from 81COL/FRO. | ≤ 188 | ≤ 788 | -40 | -168 | 68WAL/PAP | 507-40-4 |
| $\text{C}_4\text{H}_9\text{Cl}_2\text{P}^+$ | tert- $\text{C}_4\text{H}_9\text{PCl}_2$ | (9.0) IP is onset of photoelectron band. | (136) | (570) | -71 | -298 | *EST | 25979-07-1 |
| $\text{C}_4\text{H}_9\text{F}_2\text{P}^+$ | tert- $\text{C}_4\text{H}_9\text{PF}_2$ | (9.2) IP is onset of photoelectron band. | (34) | (143) | -178 | -745 | *EST | 29149-32-4 |
| $\text{C}_4\text{H}_9\text{F}_3\text{N}^+$ | $\text{CF}_3\text{CH}_2\text{NH}(\text{CH}_3)_2$ | | -17 | -69 | From proton affinity of $\text{CF}_3\text{CH}_2\text{N}(\text{CH}_3)_2$ (RN 819-06-7). PA = 215.0 kcal/mol, 900. kJ/mol. | | | |
| | $\text{CF}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_3$ | | -29 | -122 | From proton affinity of $\text{CF}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{NH}_2$ (RN 819-46-5). PA = 214.3 kcal/mol, 897 kJ/mol. | | | |
| $\text{C}_4\text{H}_9\text{I}^+$ | n- $\text{C}_4\text{H}_9\text{I}$ | 9.229 Sec: 81KIM/KAT. | (200) | (838) | -12 | -52 | *EST | 542-69-8 |
| | sec- $\text{C}_4\text{H}_9\text{I}$ | 9.09 \pm 0.02 See also: 81TRA, 81KIM/KAT. | (195) | (815) | -15 | -62 | *EST | 513-48-4 |
| | iso- $\text{C}_4\text{H}_9\text{I}$ | 9.202 See also: 81KIM/KAT. | (197) | (826) | -15 | -62 | *EST | 513-38-2 |
| | tert- $\text{C}_4\text{H}_9\text{I}$ | 9.02 \pm 0.03 See also: 81KIM/KAT. | 191 | 798 | -17.2 \pm 0.5 | -72.0 \pm 2.2 | 77PED/RYL | 558-17-8 |
| $\text{C}_4\text{H}_9\text{N}^+$ | $\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH}_2\text{NH}_2$ | (8.8) IP from 79AUE/BOW. | (207) | (866) | 5 | 21 | *EST | 2878-14-0 |
| | (E)- $\text{CH}_3\text{CH} = \text{NC}_2\text{H}_5$ | (9.29) See also: 79AUE/BOW. | (218) | (914) | 4 \pm 0.2 | 18 \pm 1 | *EST | 1190-79-0 |

Table 1. Positive Ion Table - Continued

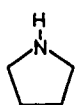
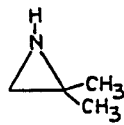
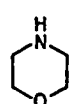
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₄H₉N⁺  | (8.0) | (184) | (771) | -0.8±0.1 | -3.4±0.6 | 77PED/RYL | 123-75-1 |
| | IP from 79AUE/BOW. Predicted $\Delta_f H(\text{Ion})$ based on hydrogen affinities of homologous series is 186 kcal/mol, 778 kJ/mol corresponding to IP of 8.1 eV. | | | | | | |
|  | (8.94) | (222) | (929) | 16±0.5 | 66±2 | *EST | 2658-24-4 |
| C₄H₉NO⁺ | | | | | | | |
| tert-C ₄ H ₉ NO | (7.5) | (163) | (681) | -10±1 | -43±6 | 74CHO/MEN | 917-95-3 |
| | IP is onset of photoelectron band. | | | | | | |
| CH ₃ CON(CH ₃) ₂ | 8.81 | 147 | 617 | -56 | -233 | 78BEA/LEE | 127-19-5 |
| (E)-CH ₃ (CH ₂) ₂ CH=NOH | (9.5) | (203) | (849) | -16 | -68 | *EST | 110-69-0 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (8.2) | (201) | (842) | 12 | 51 | *EST | 110-91-8 |
| C₄H₉NOS⁺ | | | | | | | |
| (CH ₃) ₃ CNSO | (10.0) | (166) | (695) | -65 | -270 | *EST | 38662-39-4 |
| | IP is onset of photoelectron band. | | | | | | |
| C₄H₉NO₂⁺ | | | | | | | |
| H ₂ NCH ₂ CH ₂ CH ₂ COOH | (8.7) | (95) | (398) | -105±0.5 | -441±2 | 83SKO/SAB | 56-12-2 |
| | IP is onset of photoelectron band (83CAN/HAM). | | | | | | |
| C ₂ H ₅ CH(NH ₂)COOH | (8.70) | (97) | (402) | -104±2 | -437±10 | *EST | 80-60-4 |
| H ₂ NCH ₂ COOC ₂ H ₅ | (8.8) | (107) | (447) | -96 | -402 | *EST | 459-73-4 |
| n-C ₄ H ₉ NO ₂ | (10.71±0.01) | (213) | (889) | -34.4±0.3 | -143.9±1.4 | 77PED/RYL | 627-05-4 |
| sec-C ₄ H ₉ NO ₂ | (10.71±0.01) | (208) | (870) | -39.1±0.4 | -163.6±1.6 | 77PED/RYL | 600-24-8 |
| C₄H₉NO₂S⁺ | | | | | | | |
| L-CH ₃ SCH ₂ CH(NH ₂)COOH | (8.4) | (99) | (412) | -95 | -398 | *EST | 1187-84-9 |
| | IP is onset of photoelectron band (83CAN/HAM). | | | | | | |

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------------------------------|---|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_9\text{NO}_3^+$ L- $\text{CH}_3\text{CH}(\text{OH})\text{CH}(\text{NH}_2)\text{COOH}$ | (≤ 10.2) IP from 83CAN/HAM. | (≤ 94) | (≤ 392) | -141 | -592 | *EST | 72-19-5 |
| $\text{C}_4\text{H}_9\text{N}_2^+$ NCCH ₂ NH(CH ₃) ₂ | | (188) | (788) | | | | |
| | | From proton affinity of NCCH ₂ N(CH ₃) ₂ (RN 926-64-7). PA = 211.1 kcal/mol, 883. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_9\text{N}_2\text{O}_3^+$ L-H ₂ NCOCH ₂ CH(NH ₃)COOH | | 5 | 19 | | | | |
| | | From proton affinity of L-asparagine (RN 3130-87-8). PA = 219.8 kcal/mol, 920. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_9\text{O}^+$ n-C ₄ H ₉ O | (9.22) | (196) | (820) | -17 | -69 | 82MCM/GOL | 21576-64-7 |
| n-C ₃ H ₇ CHOH | | 124 | 521 | | | | |
| | | From proton affinity of n-C ₃ H ₇ CHO (RN 123-72-8). PA = 191.5 kcal/mol, 801. kJ/mol. | | | | | |
| i-C ₃ H ₇ CHOH | | 121 | 508 | | | | |
| | | From proton affinity of i-C ₃ H ₇ CHO (RN 78-84-2). PA = 192.6 kcal/mol, 806. kJ/mol. | | | | | |
| (CH ₃)(C ₂ H ₅)COH | | 109 | 455 | | | | |
| | | From proton affinity of CH ₃ COC ₂ H ₅ (RN 78-93-3). PA = 199.8 kcal/mol, 836. kJ/mol. See 82MAC for appearance potential determination. | | | | | |
| C ₂ H ₅ OCHCH ₃ | | 125 | 521 | | | | |
| | | From proton affinity of C ₂ H ₅ OCH=CH ₂ (RN 109-92-2). PA = 207.4 kcal/mol, 868. kJ/mol (86BOU/DJA). See also: 82MAC. | | | | | |
| (CH ₃) ₂ COCH ₃ | | (114) | (477) | | | | |
| | | From appearance potential determination (82MAC). | | | | | |
|  | | 123 | 514 | | | | |
| | | From proton affinity of tetrahydrofuran (RN 109-99-9). PA = 198.8 kcal/mol, 831. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_9\text{O}_2^+$ 1HOH(O-n-C ₃ H ₇) | | 61 | 256 | | | | |
| | | From proton affinity of HCOO(n-C ₃ H ₇) (RN 110-74-7). PA = 194.2 kcal/mol, 812.5 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued



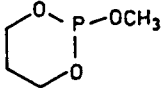
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_9\text{O}_2^+$ HCOH(O-i-C ₃ H ₇) | | 73 | 305 | | | | |
| | | From proton affinity of HCOOCH(CH ₃) ₂ (RN 625-55-8). PA = 196.0 kcal/mol, 820. kJ/mol. | | | | | |
| CH ₃ COH(O-C ₂ H ₅) | | 59 | 247 | | | | |
| | | From proton affinity of CH ₃ COOC ₂ H ₅ (RN 141-78-6). PA = 200.7 kcal/mol, 840. kJ/mol. | | | | | |
| C ₂ H ₅ COH(O-CH ₃) | | 62 | 260 | | | | |
| | | From proton affinity of C ₂ H ₅ COOCH ₃ (RN 554-12-1). PA = 200.2 kcal/mol, 838. kJ/mol. | | | | | |
|  H ⁺ | | 86 | 360 | | | | |
| | | From proton affinity of 1,3-dioxane (RN 505-22-6). PA = 198.8 kcal/mol, 832. kJ/mol. | | | | | |
|  H ⁺ | | 96 | 403 | | | | |
| | | From proton affinity of 1,4-dioxane (RN 123-91-1). PA = 193.8 kcal/mol, 811. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_9\text{O}_2\text{S}^+$ C ₂ H ₅ S(OCH ₃)COH | | 64 | 269 | | | | |
| | | From proton affinity of C ₂ H ₅ S(OCH ₃)CO (RN 38103-96-7). PA = 201.0 kcal/mol, 841. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_9\text{O}_3^+$ C(OCH ₃) ₃ | | 53 | 223 | | | | |
| | | From appearance potential of 10.22 eV in CH(OCH ₃) ₃ , and appearance potential of 9.86 eV in CH ₃ C(OCH ₃) ₃ (82HOL/LOS2). | | | | | |
| C ₂ H ₅ OC(OH)OCH ₃ | | 22 | 90 | | | | |
| | | From proton affinity of C ₂ H ₅ OCOOCH ₃ (RN 623-53-0). PA = 202.7 kcal/mol, 848 kJ/mol. | | | | | |
| $\text{C}_4\text{H}_9\text{O}_3\text{P}^+$  | (8.74±0.1) | (26) | (110) | -175 | -733 | *EST | 31121-06-9 |

Table 1. Positive Ion Table - Continued

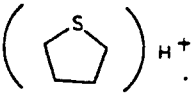
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_9\text{S}^+$  | | 153 | 640 | | | | |
| | | From proton affinity of tetrahydrothiophene (RN 110-01-0) (83CAS/KIM). PA = 204.6 kcal/mol, 856. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{10}^+$ n- C_4H_{10} | 10.53±0.10 | 213 | 889 | -30.2±0.1 | -126.5±0.4 | 77PED/RYL | 106-97-8 |
| | | IP based on charge transfer equilibrium constant in cyclopentane: n-butane system. IP (c- C_5H_{10}) = 10.51 eV (81MAU/SIE). Threshold value = 10.55 eV. See also: 81KIM/KAT. | | | | | |
| iso- C_4H_{10} | 10.57 | (212) | (885) | -32.1±0.1 | -134.5±0.5 | 77PED/RYL | 75-28-5 |
| | | See also: 81KIM/KAT. | | | | | |
| $\text{C}_4\text{H}_{10}\text{Cd}^+$ (C_2H_5) $_2\text{Cd}$ | (8.0) | (210) | (877) | 25±0.7 | 105±3 | 77PED/RYL | 592-02-9 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_4\text{H}_{10}\text{Cl}^+$ (CH_3) $_2\text{CHClCH}_3$ | | (150) | (628) | | | | |
| | | From equilibrium constant determination (85SHA/HOJ). | | | | | |
| $\text{C}_4\text{H}_{10}\text{Cl}_2\text{Si}^+$ (CH_3) $_3\text{SiCHCl}_2$ | (9.7) | (163) | (683) | -60 | -253 | *EST | 5926-38-5 |
| | | IP is onset of photoelectron band (81ZYK/KHV). | | | | | |
| (CH_3) $_2\text{Si}(\text{CH}_2\text{Cl})_2$ | (9.7) | (165) | (689) | -59 | -247 | *EST | 2917-46-6 |
| | | IP is onset of photoelectron band (81ZYK/KHV). | | | | | |
| $\text{C}_4\text{H}_{10}\text{Hg}^+$ (C_2H_5) $_2\text{Hg}$ | ≤8.45 | ≤212 | ≤887 | 17.3±0.2 | 72.3±0.8 | 77PED/RYL | 627-44-1 |
| $\text{C}_4\text{H}_{10}\text{N}^+$ $\text{CH}_2\text{C}(\text{CH}_3)\text{CH}_2\text{NH}_3$ | | (152) | (638) | | | | |
| | | From proton affinity of $\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH}_2\text{NH}_2$ (RN 2878-14-0). PA = (218.2) kcal/mol, (913.) kJ/mol. | | | | | |
| $\text{CH}_3\text{CHN}(\text{CH}_3)_2$ | | 153 | 639 | | | | |
| | | From proton affinity of (CH_3) $_2\text{NCH} = \text{CH}_2$ (RN 5763-87-1). PA = 227.8 kcal/mol, 953. kJ/mol. | | | | | |
| $\text{CH}_3\text{CHNHC}_2\text{H}_5$ | | 147 | 616 | | | | |
| | | From proton affinity of $\text{CH}_3\text{CH} = \text{NC}_2\text{H}_5$ (RN 1190-79-0). PA = 222.7 kcal/mol, 932. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued



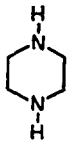
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{10}\text{N}^+$  H^+ | | 140 | 585 | | | | |
| | | From proton affinity of pyrrolidine (RN 123-75-1). PA = 225.2 kcal/mol, 942. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{10}\text{NO}^+$ $\text{CH}_3\text{C}(\text{OH})\text{N}(\text{CH}_3)_2$ | | 94 | 392 | | | | |
| | | From proton affinity of $\text{CH}_3\text{CON}(\text{CH}_3)_2$ (RN 127-19-5) (86TAF/GAL). PA = 216.2 kcal/mol, 905. kJ/mol. | | | | | |
| $n\text{-C}_3\text{H}_7\text{NHCHOH}$ | | 95 | 395 | | | | |
| | | From proton affinity of $n\text{-C}_3\text{H}_7\text{NHCHO}$ (RN 6281-94-3). PA = (210.0) kcal/mol, (879.) kJ/mol. | | | | | |
|  H^+ | | 158 | 663 | | | | |
| | | From proton affinity of morpholine (RN 110-91-8). PA = 219.4 kcal/mol, 918. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{10}\text{NO}_2^+$ $t\text{-C}_4\text{H}_9\text{ONHO}$ | | 119 | 497 | | | | |
| | | From proton affinity of $t\text{-C}_4\text{H}_9\text{ONO}$ (RN 540-80-7). PA = 205.7 kcal/mol, 861. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{10}\text{NO}_3^+$ $\text{CH}_3\text{CH}(\text{OH})\text{CH}(\text{NH}_3)\text{COOH}$ | | (6) | (23) | | | | |
| | | From proton affinity of L-threonine (RN 72-19-5). PA = 218.6 kcal/mol, 915. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{10}\text{N}_2^+$ $(\text{CH}_3)_2\text{NN}=\text{CHCH}_3$ | (7.54) | (176) | (736) | 2 | 9 | 80LEB/MAS | 7422-90-4 |
|  | (≤ 8.72) | (≤ 207) | (≤ 866) | 6 ± 0.2 | 25 ± 1 | *EST | 110-85-0 |
| $\text{C}_4\text{H}_{10}\text{N}_2\text{O}^+$ $(\text{CH}_3)_2\text{NCONHCH}_3$ | (≤ 8.80) | (≤ 146) | (≤ 609) | -57 | -240 | *EST | 632-14-4 |
| $\text{C}_4\text{H}_{10}\text{O}^+$ $n\text{-C}_4\text{H}_9\text{OH}$ | 10.06 ± 0.03 | 166 | 696 | -65.7 ± 0.1 | -275.0 ± 0.4 | 77PED/RYL | 71-36-3 |
| | | See also: 81KIM/KAT, 80BAC/MOU, 84BOW/MAC. | | | | | |
| $\text{sec-C}_4\text{H}_9\text{OH}$ | 9.88 | 158 | 660 | -70.5 ± 0.1 | -295.0 ± 0.4 | 77PED/RYL | 78-92-2 |
| | | IP from 81HOL/FIN, 84BOW/MAC. See also: 80BAC/MOU. | | | | | |

Table 1. Positive Ion Table - Continued

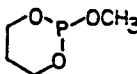
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{10}\text{O}^+$ | | | | | | | |
| iso- $\text{C}_4\text{H}_9\text{OH}$ | 10.12±0.04 IP from 81HOL/FIN, 84BOW/MAC, 77ROS/DRA. | 166 | 692 | -67.8±0.1 | -283.6±0.4 | 77PED/RYL | 78-83-1 |
| tert- $\text{C}_4\text{H}_9\text{OH}$ | 9.97±0.02 See also: 84BOW/MAC. | 155 | 650 | -74.7±0.7 | -312.5±2.9 | 77PED/RYL | 75-65-0 |
| $(\text{C}_2\text{H}_5)_2\text{O}$ | 9.51±0.03 See also: 81KIM/KAT, 80BAC/MOU, 84BOW/MAC. | 159 | 666 | -60.1±0.1 | -251.7±0.3 | 77PED/RYL | 60-29-7 |
| n- $\text{C}_3\text{H}_7\text{OCH}_3$ | (9.42) IP from 84BOW/MAC. See also: 80BAC/MOU. | (160) | (671) | -56.8±0.1 | -237.9±0.5 | 77PED/RYL | 557-17-5 |
| i- $\text{C}_3\text{H}_7\text{OCH}_3$ | 9.42 IP from 81HOL/FIN, 84BOW/MAC. | 157 | 657 | -60.2±0.2 | -252.0±0.9 | 77PED/RYL | 598-53-8 |
| $\text{C}_4\text{H}_{10}\text{OS}^+$ | | | | | | | |
| $(\text{CH}_3\text{CH}_2)_2\text{SO}$ | ≤8.76 | ≤153 | ≤640 | -49.1±0.4 | -205.6±1.5 | 77PED/RYL | 70-29-1 |
| $\text{C}_4\text{H}_{10}\text{O}_2^+$ | | | | | | | |
| n- $\text{C}_4\text{H}_9\text{OOH}$ | (9.36±0.03) IP from 77ASH/BUR. | (166) | (696) | -49 | -207 | *EST | 4813-50-7 |
| tert- $\text{C}_4\text{H}_9\text{OOH}$ | (≤10.24) | (≤178) | (≤744) | -58±1 | -244±6 | 77PED/RYL | 75-91-2 |
| $\text{HOCH}_2\text{CH}_2\text{CH}_2\text{OCH}_3$ | (9.3) IP is onset of photoelectron band (83BIE/MOR). | (122) | (509) | -93 | -388 | *EST | 1320-67-8 |
| $\text{HOCH}_2\text{CH}_2\text{OC}_2\text{H}_5$ | (9.6) IP is onset of photoelectron band (81KIM/KAT). | (126) | (528) | -95 | -398 | *EST | 110-80-5 |
| $\text{CH}_3\text{OCH}_2\text{CH}_2\text{OCH}_3$ | (9.3) IP is onset of photoelectron band (83BAK/ARM, 81KIM/KAT). | (133) | (557) | -81 | -340 | 67LOU/LAI | 110-71-4 |
| $\text{CH}_3\text{CH}(\text{OCH}_3)_2$ | (9.65±0.03) | (129) | (541) | -93.1±0.2 | -389.7±0.8 | 77PED/RYL | 534-15-6 |
| $\text{C}_4\text{H}_{10}\text{O}_2\text{S}^+$ | | | | | | | |
| $(\text{C}_2\text{H}_5)_2\text{SO}_2$ | (9.96±0.03) | (127) | (532) | -103±0.7 | -429±3 | 77PED/RYL | 597-35-3 |
| $\text{C}_4\text{H}_{10}\text{O}_3^+$ | | | | | | | |
| $\text{CH}(\text{OCH}_3)_3$ | (9.5) IP from 82HOL/LOS2. | (89) | (372) | -130±0.2 | -545±1 | 77PED/RYL | 149-73-5 |
| $\text{C}_4\text{H}_{10}\text{O}_3\text{P}^+$ | | | | | | | |
|  | (-29) (-121) From proton affinity of 2-methoxy-1,3,2-dioxaphosphorinane (RN 31121-06-9). PA = 219.4 kcal/mol, 918. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

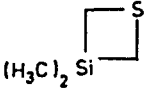
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|------------|---|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{10}\text{O}_3\text{S}^+$ ($\text{C}_2\text{H}_5\text{O}$) ₂ SO | (9.68) | (91) | (382) | -132±0.5 | -552±2 | 77PED/RYL | 623-81-4 |
| $\text{C}_4\text{H}_{10}\text{S}^+$ | | | | | | | |
| n-C ₄ H ₉ SH | 9.14±0.02 | 190 | 794 | -21.1±0.3 | -88.1±1.2 | 77PED/RYL | 109-79-5 |
| sec-C ₄ H ₉ SH | (9.10) | (187) | (781) | -23.2±0.2 | -96.9±0.8 | 77PED/RYL | 513-53-1 |
| iso-C ₄ H ₉ SH | (9.12) | (187) | (783) | -23.3±0.2 | -97.3±0.8 | 77PED/RYL | 513-44-0 |
| tert-C ₄ H ₉ SH | (9.03) | (182) | (762) | -26.2±0.2 | -109.6±0.8 | 77PED/RYL | 75-66-1 |
| n-C ₃ H ₇ SCH ₃ | (8.8±0.2) | (183) | (767) | -19.6±0.2 | -82.2±0.9 | 77PED/RYL | 3877-15-4 |
| iso-C ₃ H ₇ SCH ₃ | (8.7±0.2) | (179) | (748) | -21.6±0.2 | -90.5±0.7 | 77PED/RYL | |
| (C ₂ H ₅) ₂ S | 8.43±0.01 | 174 181 | 729 757 | -20±0.2 -13 | -84±1 -56 | 77PED/RYL | 352-93-2 |
| $\text{C}_4\text{H}_{10}\text{SSi}^+$ (C ₂ H ₅) ₂ Si=S | | (188) | (787) | $\Delta_f H(\text{Ion})$ from appearance potential determination (81GUS/VOL). | | | |
|  | (8.25±0.03) | (193) | (806) | (2) | (10) | 81GUS/VOL | 77205-52-8 |
| | | IP from 81GUS/VOL | | | | | |
| $\text{C}_4\text{H}_{10}\text{S}_2^+$ (C ₂ H ₅ S) ₂ | ≤8.27±0.03 | ≤173 | ≤723 | -17.8±0.3 | -74.7±1.1 | 77PED/RYL | 110-81-6 |
| | Dialkyl disulfides undergo a change in the dihedral CSSC angle from 90° to 180° upon ionization; adiabatic ionization potentials are probably well below the the experimentally observed ionization onset. | | | | | | |
| CH ₃ SCH ₂ CH ₂ SCH ₃ | (≤8.64) | (≤190) | (≤797) | -9 | -37 | *EST | 6628-18-8 |
| $\text{C}_4\text{H}_{10}\text{Sc}^+$ | | | | | | | |
| HScCH(CH ₃)C ₂ H ₅ | | (195) | (816) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | |
| CH ₃ ScC ₃ H ₇ | | (174) | (728) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | |
| C ₂ H ₄ Sc(CH ₃) ₂ | | (175) | (732) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | |
| (CH ₃ CH=CH ₂)ScH(CH ₃) | | (184) | (770) | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | |

Table 1. Positive Ion Table - Continued

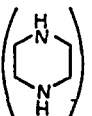
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{10}\text{Sc}^+$ ($\text{CH}_3\text{CH}=\text{CHCH}_3$)ScH ₂ | | (195) | (816) | | | | |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). | | | | | |
| $\text{C}_4\text{H}_{10}\text{Se}^+$ (C_2H_5) ₂ Se | (8.3±0.3) | (178) | (743) | -14±1 | -58±5 | 77PED/RYL | 627-53-2 |
| $\text{C}_4\text{H}_{10}\text{Zn}^+$ (C_2H_5) ₂ Zn | (≤8.6) | (≤212) | (≤888) | 14±0.7 | 58±3 | 77PED/RYL | 557-20-0 |
| $\text{C}_4\text{H}_{11}^+$ ((CH_3) ₃ CH)H | | 170 | 712 | | | | |
| | | From proton affinity of iso- C_4H_{10} . (RN 75-28-5). PA = 163.3 kcal/mol, 683. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{11}\text{ClO}_3\text{Si}^+$ (CH_3O) ₃ SiCH ₂ Cl | (10.0) | (4) | (17) | -226 | -948 | *EST | 5926-26-1 |
| | | IP is onset of photoelectron band (81ZYK/KHV). | | | | | |
| $\text{C}_4\text{H}_{11}\text{ClSi}^+$ (CH_3) ₃ SiCH ₂ Cl | (9.4) | (159) | (667) | -57 | -240 | *EST | 2344-80-1 |
| | | IP is onset of photoelectron band (81ZYK/KHV, 82LEV/LIA). | | | | | |
| $\text{C}_4\text{H}_{11}\text{N}^+$ n- $\text{C}_4\text{H}_9\text{NH}_2$ | 8.71±0.03 | 179 | 748 | -22±0.2 | -92±1 | 77PED/RYL | 109-73-9 |
| | | See also: 81KIM/KAT, 79AUE/BOW. | | | | | |
| sec- $\text{C}_4\text{H}_9\text{NH}_2$ | (8.70) | (176) | (734) | -25.0±0.2 | -104.8±0.9 | 77PED/RYL | 13952-84-6 |
| iso- $\text{C}_4\text{H}_9\text{NH}_2$ | (8.70) | (177) | (741) | -23.6±0.1 | -98.8±0.4 | 77PED/RYL | 78-81-9 |
| tert- $\text{C}_4\text{H}_9\text{NH}_2$ | (8.64) | (170) | (713) | -28.9±0.1 | -120.9±0.4 | 77PED/RYL | 75-64-9 |
| (C_2H_5) ₂ NH | 8.01±0.01 | 167 | 700 | -17.4±0.5 | -72.6±2 | 77PED/RYL | 109-89-7 |
| $\text{C}_2\text{H}_5\text{N}(\text{CH}_3)_2$ | (7.74±0.05) | (167) | (701) | -11 | -48 | *EST | 598-56-1 |
| | | IP is onset of photoelectron band. See also: 81LOG/TAK, 79AUE/BOW. | | | | | |
| $\text{C}_4\text{H}_{11}\text{NO}^+$ (CH_3) ₂ NCH ₂ CH ₂ OH | (8.2) | (140) | (587) | -49 | -204 | 81LOS/LAM | 108-01-0 |
| | | IP is onset of photoelectron band (82LEV/LIA, 86VOR/BRO). | | | | | |
| $\text{C}_4\text{H}_{11}\text{N}_2^+$  H ⁺ | | 147 | 617 | | | | |
| | | From proton affinity of piperazine (RN 110-85-0). PA = 224.2 kcal/mol, 938. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

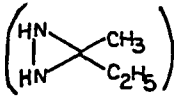
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|---|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{11}\text{N}_2^+$ | | (182) | (764) | | | | |
|  H^+ | | | | | | | |
| | | | | | | | From proton affinity of 3-ethyl-3-methyldiaziridine (RN 4901-75-1). PA = (214.9) kcal/mol, (899) kJ/mol. |
| $\text{C}_4\text{H}_{11}\text{O}^+$ | | | | | | | |
| n- $\text{C}_4\text{H}_9\text{OH}_2$ | | 109 | 456 | | | | From proton affinity of n- $\text{C}_4\text{H}_9\text{OH}$ (RN 71-36-3). PA = 191.1 kcal/mol, 799.5 kJ/mol. |
| sec- $\text{C}_4\text{H}_9\text{OH}_2$ | | (101) | (421) | | | | From proton affinity of sec- $\text{C}_4\text{H}_9\text{OH}$ (RN 78-92-2) (78PAU/KIM). PA = (195) kcal/mol, (816) kJ/mol. |
| iso- $\text{C}_4\text{H}_9\text{OH}_2$ | | 105 | 441 | | | | From proton affinity of iso- $\text{C}_4\text{H}_9\text{OH}$ (RN 78-83-1) (78TAF/TAA). PA = 192.4 kcal/mol, 805. kJ/mol. |
| tert- $\text{C}_4\text{H}_9\text{OH}_2$ | | 97 | 408 | | | | From proton affinity of tert- $\text{C}_4\text{H}_9\text{OH}$ (RN 75-65-0). PA = 193.7 kcal/mol, 810. kJ/mol. |
| $(\text{C}_2\text{H}_5)_2\text{OH}$ | | 105 | 440 | | | | From proton affinity of $(\text{C}_2\text{H}_5)_2\text{O}$ (RN 60-29-7) (86KNI/FRE, 86MAU/LIE). PA = 200.2 kcal/mol, 838 kJ/mol. |
| $\text{C}_4\text{H}_{11}\text{O}_2^+$ | | | | | | | |
| $\text{HO}(\text{CH}_2)_4\text{OH}_2$ | | 52 | 216 | | | | From proton affinity of $\text{HO}(\text{CH}_2)_4\text{OH}$ (RN 110-63-4). PA = (212) kcal/mol, (887) kJ/mol. |
| $\text{CH}_3\text{OCH}_2\text{CH}_2\text{OHCH}_3$ | | 80 | 333 | | | | From proton affinity of $\text{CH}_3\text{OCH}_2\text{CH}_2\text{OCH}_3$ (RN 110-71-4). PA = 204.9 kcal/mol, 857. kJ/mol. |
| $\text{C}_4\text{H}_{11}\text{O}_3\text{P}^+$ | | | | | | | |
| $\text{OPH}(\text{OC}_2\text{H}_5)_2$ | (10.31) | (19) | (79) | -219 | -916 | *EST | 762-04-9 |
| | | | | | | | See also: 80ZVE/VIL. |
| $\text{C}_4\text{H}_{11}\text{P}^+$ | | | | | | | |
| tert- $\text{C}_4\text{H}_9\text{PH}_2$ | (8.9) | (181) | (757) | -24 | -102 | *EST | 2501-94-2 |
| | | | | | | | IP is onset of photoelectron band. |
| $(\text{C}_2\text{H}_5)_2\text{PH}$ | (8.69) | (176) | (736) | -24 | -102 | *EST | 627-49-6 |
| $\text{C}_4\text{H}_{11}\text{S}^+$ | | | | | | | |
| $(\text{C}_2\text{H}_5)_2\text{SH}$ | | 141 | 588 | | | | From proton affinity of $(\text{C}_2\text{H}_5)_2\text{S}$ (RN 352-93-2). PA = 205.0 kcal/mol, 858. kJ/mol. |

Table 1. Positive Ion Table - Continued

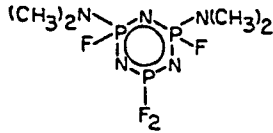
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{11}\text{S}^+$ tert- $\text{C}_4\text{H}_9\text{SH}_2$ | | 143 | 596 | | | | |
| | | From proton affinity of t- $\text{C}_4\text{H}_9\text{SH}$ (RN 75-66-1). PA = 196.9 kcal/mol, 824. kJ/mol. | | | | | |
| sec- $\text{C}_4\text{H}_9\text{SH}_2$ | | (148) | (621) | | | | |
| | | From proton affinity of sec- $\text{C}_4\text{H}_9\text{SH}$ (RN 513-53-1) (78PAU/KIM). PA = (194.0) kcal/mol, (812.) kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{11}\text{SSi}^+$ $(\text{C}_2\text{H}_5)_2\text{SiSH}$ | | (157) | (657) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (81GUS/VOL). | | | | | |
| $\text{C}_4\text{H}_{12}\text{BCIN}_2^+$ $\text{B}(\text{N}(\text{CH}_3)_2)_2\text{Cl}$ | 8.08 | 106 | 445 | -80±1 | -335±5 | 77PED/RYL | 6562-41-0 |
| $\text{C}_4\text{H}_{12}\text{CIN}_2\text{OP}^+$ $((\text{CH}_3)_2\text{N})_2\text{POCl}$ | (8.61) | (75) | (316) | -123 | -515 | *EST | 1605-65-8 |
| $\text{C}_4\text{H}_{12}\text{CIN}_2\text{P}^+$ $((\text{CH}_3)_2\text{N})_2\text{PCl}$ | (7.6) | (127) | (531) | -48 | -202 | *EST | 3348-44-5 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_4\text{H}_{12}\text{F}_4\text{N}_5\text{P}_3^+$  | (8.96) | (-169) | (-706) | -375.5 | -1571 | *EST | 30004-14-9 |
| | | IP from 81CLA/SOW. | | | | | |
| $\text{C}_4\text{H}_{12}\text{Ge}^+$ $(\text{CH}_3)_4\text{Ge}$ | 9.33±0.05 | 198 | 828 | -17±2 | -72±9 | 77PED/RYL | 865-52-1 |
| $\text{C}_4\text{H}_{12}\text{N}^+$ n- $\text{C}_4\text{H}_9\text{NH}_3$ | | 122 | 524 | | | | |
| | | From proton affinity of n- $\text{C}_4\text{H}_9\text{NH}_2$ (RN 109-73-9). PA = 218.4 kcal/mol, 914. kJ/mol. | | | | | |
| sec- $\text{C}_4\text{H}_9\text{NH}_3$ | | 120 | 502 | | | | |
| | | From proton affinity of sec- $\text{C}_4\text{H}_9\text{NH}_2$ (RN 13952-84-6). PA = 220.5 kcal/mol, 922. kJ/mol. | | | | | |
| iso- $\text{C}_4\text{H}_9\text{NH}_3$ | | 123 | 515 | | | | |
| | | From proton affinity of iso- $\text{C}_4\text{H}_9\text{NH}_2$ (RN 78-81-9). PA = 218.8 kcal/mol, 915. kJ/mol. | | | | | |
| tert- $\text{C}_4\text{H}_9\text{NH}_3$ | | 116 | 485 | | | | |
| | | From proton affinity of tert- $\text{C}_4\text{H}_9\text{NH}_2$ (RN 75-64-9). PA = 220.8 kcal/mol, 924. kJ/mol. | | | | | |
| $(\text{C}_2\text{H}_5)_2\text{NH}_2$ | | 125 | 512 | | | | |
| | | From proton affinity of $(\text{C}_2\text{H}_5)_2\text{NH}$ (RN 109-89-7). PA = 225.9 kcal/mol, 945. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{12}\text{N}^+$ (CH_3) ₂ (C_2H_5)NH | | 127 | 531 | | | | |
| | | From proton affinity of (CH_3) ₂ (C_2H_5)N (RN 598-56-1). PA = 227.5 kcal/mol, 952. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{12}\text{NO}^+$ $\text{NH}_3(\text{CH}_2)_4\text{OH}$ | | 75 | 312 | | | | |
| | | From proton affinity of $\text{NH}_2(\text{CH}_2)_4\text{OH}$ (RN 13325-10-5). PA = 233.8 kcal/mol, 978. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{12}\text{NO}_2\text{P}^+$ (CH_3O) ₂ PN(CH_3) ₂ | (8.1) | (71) | (296) | -116 | -486 | *EST | 597-07-9 |
| | IP is onset of photoelectron band (82WOR/HAR). | | | | | | |
| $\text{C}_4\text{H}_{12}\text{N}_2^+$ (CH_3) ₂ NN(CH_3) ₂ | (6.87) | (175) | (732) | 16 | 69 | 61GOW/JON | 6415-12-9 |
| | IP from charge transfer equilibrium constant determination (86RUM). Reference standard: IP($\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2$) = 7.12 eV. See also: 84MAU/NEL. | | | | | | |
| $\text{C}_4\text{H}_{12}\text{N}_2\text{S}_2^+$ (CH_3) ₂ NSSN(CH_3) ₂ | (7.2) | (163) | (683) | -3 | -12 | *EST | 928-05-2 |
| | IP is onset of photoelectron band (81BOC/SCH). | | | | | | |
| $\text{C}_4\text{H}_{12}\text{N}_4^+$ (E)-(CH_3) ₂ NN=NN(CH_3) ₂ | (7.0) | (227) | (948) | 65±0.7 | 273±3 | 77PED/RYL | 6130-87-6 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_4\text{H}_{12}\text{OSi}^+$ (CH_3) ₃ SiOCH ₃ | 9.61 | (110) | (459) | -112±2 | -468±8 | *EST | 1825-61-2 |
| | IP from 83MOL/PIK. | | | | | | |
| $\text{C}_4\text{H}_{12}\text{Pb}^+$ (CH_3) ₄ Pb | (8.50) | (229) | (956) | 33±1 | 136±4 | 82PIL/SKI | 75-74-1 |
| $\text{C}_4\text{H}_{12}\text{Si}^+$ (CH_3) ₄ Si | 9.80±0.04 | 170 | 711 | -55.7±0.7 | -233.0±2.9 | 83STE2 | 75-76-3 |
| | | 178 | 743 | -48 | -202 | | |
| (C_2H_5) ₂ SiH ₂ | (9.8) | (182) | (763) | -44±1 | -183±6 | 77PED/RYL | 542-91-6 |
| $\text{C}_4\text{H}_{12}\text{SiS}^+$ (CH_3) ₃ SiSCH ₃ | (8.4) | (128) | (534) | -66 | -276 | *EST | 3908-55-2 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_4\text{H}_{12}\text{Sn}^+$ (CH_3) ₄ Sn | 8.89±0.05 | 200 | 838 | -5±0.5 | -20±2 | 77PED/RYL | 594-27-4 |

Table 1. Positive Ion Table - Continued

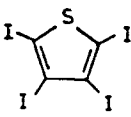
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|---|---------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{H}_{13}\text{N}_2^+$ | $\text{NH}_2(\text{CH}_2)_4\text{NH}_3$ | | 115 | 483 | | | | |
| | | | From proton affinity of $\text{NH}_2(\text{CH}_2)_4\text{NH}_2$ (RN 110-60-1). PA = 237.6 kcal/mol, 994 kJ/mol. | | | | | |
| | $(\text{CH}_3)_2\text{NNH}(\text{CH}_3)_2$ | | 157 | 658 | | | | |
| | | | From proton affinity of $(\text{CH}_3)_2\text{NN}(\text{CH}_3)_2$ (RN 6415-12-9) (84MAU/NEL). PA = 224.8 kcal/mol, 941. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{13}\text{OSi}^+$ | $(\text{CH}_3)_3\text{Si}(\text{OH})\text{CH}_3$ | | (51) | (213) | | | | |
| | | | From proton affinity of $(\text{CH}_3)_3\text{SiOCH}_3$ (RN 1825-61-2). PA = -203 kcal/mol, -849 kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{14}\text{N}_3\text{OP}^+$ | $((\text{CH}_3)_2\text{N})_2(\text{NH}_2)\text{PO}$ | (8.60±0.05) | (83) | (348) | -115 | -482 | *EST | 3732-86-3 |
| $\text{C}_4\text{H}_{15}\text{N}_3\text{OP}^+$ | $((\text{CH}_3)_2\text{N})_2(\text{NH}_2)\text{POH}$ | | 26 | 109 | | | | |
| | | | From proton affinity of $((\text{CH}_3)_2\text{N})_2(\text{NH}_2)\text{PO}$ (RN 3732-86-3) (85BOL/HOU). PA = 224.4 kcal/mol, 939. kJ/mol. | | | | | |
| $\text{C}_4\text{H}_{15}\text{OSi}_2^+$ | $((\text{CH}_3)_2\text{SiH})_2\text{OH}$ | | (6) | (26) | | | | |
| | | | From proton affinity of $((\text{CH}_3)_2\text{SiH})_2\text{O}$ (RN 3277-26-7). PA = -203 kcal/mol, -849 kJ/mol. | | | | | |
| $\text{C}_4\text{I}_4\text{S}^+$ |  | (≤8.27) | (≤302) | (≤1262) | 111 | 464 | *EST | 19259-11-1 |
| C_4La^+ | LaC_4 | (4.7±0.5) | (288) | (1207) | 180±2 | 754±8 | 81GIN/PEL | 12603-31-5 |
| C_4N_2^+ | $\text{NCC}=\text{CCN}$ | 11.81±0.01 | 400 | 1673 | 128 | 534 | 82CHU/NGU | 1071-98-3 |
| | | | See also: 82MAI/MIS. | | | | | |
| $\text{C}_4\text{N}_2\text{O}^+$ | $(\text{NC})_2\text{C}=\text{C}=\text{O}$ | (10.56) | (300) | (1255) | 56.5 | 236 | *EST | 4361-47-1 |
| | | | IP is onset of photoelectron band (80HOT/NEI). | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|----------------------------|--------------------------|--------|------------------------------|---------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_4\text{N}_2\text{S}^+$ | $(\text{NC})_2\text{C}=\text{C}=\text{S}$ | (9.94) | (339) | (1419) | 110 | 460 | 80SCH/SCH2 | 54856-36-9 |
| IP is onset of photoelectron band (80SCH/SCH2). | | | | | | | | |
| C_4NiO_4^+ | $\text{Ni}(\text{CO})_4$ | 8.27 ± 0.04 | 48 | 200 | -143 ± 1 | -598 ± 4 | 77PED/RYL | 13463-39-3 |
| See also: 86REU/WAN. | | | | | | | | |
| C_4Sc^+ | ScC_4 | (6.7±1.0) | (339) | (1418) | 184 ± 4 | 772 ± 18 | 81HAQ/GIN | 12547-95-4 |
| IP from 81HAQ/GIN. | | | | | | | | |
| $\text{C}_5\text{BrMnO}_5^+$ | $\text{Mn}(\text{CO})_5\text{Br}$ | 8.4 | (-16) | (-65) | -209 ± 1 | -876 ± 5 | 82CON/ZAF | 14516-54-2 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_5\text{BrO}_5\text{Re}^+$ | $\text{Re}(\text{CO})_5\text{Br}$ | 8.5 | (-9) | (-38) | -205 ± 1 | -858 ± 5 | 83ALT/CON | 14220-21-4 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_5\text{ClMnO}_5^+$ | $\text{Mn}(\text{CO})_5\text{Cl}$ | 8.6 | (-21) | (-88) | -219 ± 2 | -918 ± 10 | 82CON/ZAF | 14100-30-2 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_5\text{ClO}_5\text{Re}^+$ | $\text{Re}(\text{CO})_5\text{Cl}$ | 8.55 | (-12) | (-52) | -210 ± 4 | -877 ± 18 | 83ALT/CON | 14099-01-5 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_5\text{FeO}_4\text{S}^+$ | $\text{Fe}(\text{CO})_4\text{CS}$ | (7.8) | (65) | (273) | -115 | -480 | *EST | 66517-47-3 |
| IP is onset of photoelectron band (82BOH/GLE). | | | | | | | | |
| C_5FeO_5^+ | $\text{Fe}(\text{CO})_5$ | 7.96 ± 0.01 | 10 | 43 | -173 ± 2 | -725 ± 7 | 82PIL/SKI | 13463-40-6 |
| See also: 83HAR/OHN. | | | | | | | | |
| $\text{C}_5\text{HFeO}_5^+$ | $\text{HFe}(\text{CO})_5$ | | (-10) | (-40) | | | | |
| From proton affinity of $\text{Fe}(\text{CO})_5$ (RN 13463-40-6). PA = -202 kcal/mol, -845 kJ/mol. | | | | | | | | |
| $\text{C}_5\text{HMnO}_5^+$ | $\text{Mn}(\text{CO})_5\text{H}$ | 8.5 ± 0.1 | 19 | 80 | -177 ± 2 | -740 ± 10 | 82CON/ZAF | 16972-33-1 |
| C_5HN_3^+ | $\text{C}(\text{CN})_2=\text{CHCN}$ | (-11.55) | (390) | (1632) | 124 | 518 | 82CHU/NGU | 997-76-2 |
| $\text{C}_5\text{H}_2\text{MnO}_5^+$ | $\text{H}_2\text{Mn}(\text{CO})_5$ | | (-12) | (-51) | | | | |
| From proton affinity of $\text{HMn}(\text{CO})_5$ (RN 16972-33-1). PA = (201) kcal/mol, (841) kJ/mol. | | | | | | | | |

Table 1. Positive Ion Table - Continued

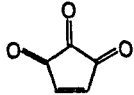
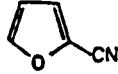
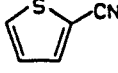
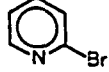
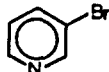
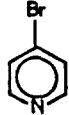
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|------------------------------------|---|--|--------------------------|---------------|------------------------------|-------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_2\text{O}_3^+$ |  | (9.3) | (144) | (603) | -70 | -294 | *EST | 15548-56-8 |
| | | IP is onset of photoelectron band (82GLE/DOB). | | | | | | |
| $\text{C}_5\text{H}_3\text{Cl}^+$ | $\text{CH}_3\text{C}=\text{CC}=\text{CCl}$ | 9.19 ± 0.01 | (321) | (1342) | 110 ± 0.2 | 459 ± 1 | *EST | |
| | | IP from 84KLA/KUH. | | | | | | |
| $\text{C}_5\text{H}_3\text{NO}^+$ |  | $(\leq 9.47 \pm 0.05)$ | (≤ 243) | (≤ 1018) | 25 | 104 | *EST | 617-90-3 |
| $\text{C}_5\text{H}_3\text{NS}^+$ |  | (9.83 ± 0.05) | (293) | (1226) | 66 | 278 | *EST | 1003-31-2 |
| C_5H_4^+ | $\text{CH}_2=\text{C}=\text{C}=\text{C}=\text{CH}_2$ | (8.67) | (315) | (1318) | 115 | 481 | *EST | 21986-03-8 |
| | $\text{CH}=\text{CCH}_2\text{C}\equiv\text{CH}$ | 10.1 | (338) | (1413) | 105 | 439 | *EST | 24442-69-1 |
| | | IP from 83HOL. | | | | | | |
| | $\text{CH}_3\text{C}=\text{CC}=\text{CH}$ | 9.4 | (318) | (1332) | 101 | 425 | *EST | 4911-55-1 |
| | | IP from 81FOR/MAI. See also: 81MAI. | | | | | | |
| $\text{C}_5\text{H}_4\text{BrN}^+$ |  | 9.65 ± 0.05 | (261) | (1092) | 38 | 161 | *EST | 109-04-6 |
| |  | (9.75 ± 0.1) | (263) | (1102) | 38 | 161 | *EST | 626-55-1 |
| |  | 9.94 ± 0.05 | (268) | (1120) | 38 | 161 | *EST | 1120-87-2 |

Table 1. Positive Ion Table - Continued

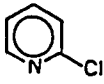
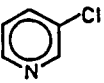
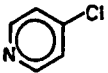
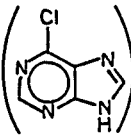
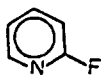
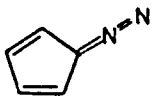
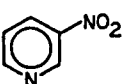
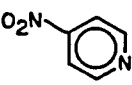
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_5H_4ClN^+$ | | | | | | | |
|  | 9.0 | (232) | (971) | 25 | 103 | *EST | 109-09-1 |
| IP is onset of photoelectron band (81MOD/DIS2). | | | | | | | |
|  | 9.1 | (236) | (986) | 26 | 108 | *EST | 626-60-8 |
| IP is onset of photoelectron band (81MOD/DIS2). | | | | | | | |
|  | 9.5 | (245) | (1025) | 26 | 108 | *EST | 626-61-9 |
| IP is onset of photoelectron band (81MOD/DIS2). | | | | | | | |
| $C_5H_4ClN_4^+$ | | | | | | | |
|  | | (200) | (839) | | | | |
| From proton affinity of 6-chloropurine (RN 87-42-3). PA = -208 kcal/mol, -870 kJ/mol. | | | | | | | |
| $C_5H_4FN^+$ | | | | | | | |
|  | (9.4) | (201) | (839) | -16 | -68 | *EST | 372-48-5 |
| IP is onset of photoelectron band (83PIA/KEL). | | | | | | | |
| $C_5H_4N_2^+$ | | | | | | | |
|  | (8.09±0.01) | (277) | (1161) | 91±4 | 380±16 | *EST | 1192-27-4 |
| $C_5H_4N_2O_2^+$ | | | | | | | |
|  | (10.3±0.1) | (270) | (1130) | 33 | 136 | *EST | 2530-26-9 |
|  | (10.4) | (273) | (1140) | 33 | 137 | *EST | 1122-61-8 |

Table 1. Positive Ion Table - Continued

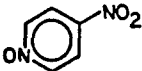
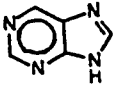
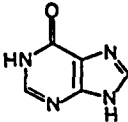
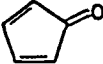
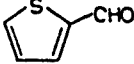

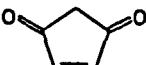
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_4\text{N}_2\text{O}_3^+$ | | | | | | | |
|  | (9.03±0.02) | (222) | (930) | 14 | 59 | *EST | 1124-82-9 |
| $\text{C}_5\text{H}_4\text{N}_4^+$ | | | | | | | |
|  | (≤9.52±0.03) | (≤275) | (≤1149) | 55 | 230 | *EST | 120-73-0 |
| $\text{C}_5\text{H}_4\text{N}_4\text{O}^+$ | | | | | | | |
|  | (≤8.55±0.03) | (≤209) | (≤875) | 12 | 50 | 77PED/RYL | 68-94-0 |
| $\text{C}_5\text{H}_4\text{O}^+$ | | | | | | | |
|  | (9.49) | (211) | (881) | -8±8 | -35±35 | *EST | 13177-38-3 |
| $\text{C}_5\text{H}_4\text{OS}^+$ | | | | | | | |
|  | (≤9.37±0.05) | (≤222) | (≤928) | 6 | 24 | *EST | 98-03-3 |
| $\text{C}_5\text{H}_4\text{O}_2^+$ | | | | | | | |
|  | 9.35±0.05 | (176) | (733) | -40 | -169 | *EST | 108-97-4 |
|  | (9.6) | (168) | (704) | -53 | -222 | *EST | 930-60-9 |
| | IP is onset of photoelectron band (82GLE/DOB). | | | | | | |

Table 1. Positive Ion Table - Continued

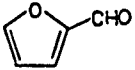
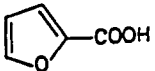
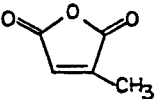
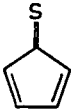

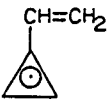
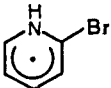
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|---|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_4\text{O}_2^+$  | 9.21±0.01 | 176 | 738 | -36±1 | -151±5 | 77PED/RYL | 98-01-1 |
| $\text{C}_5\text{H}_4\text{O}_3^+$  | (≤9.16±0.05) | (≤118) | (≤493) | -93±0.7 | -391±3 | 77PED/RYL | 488-93-7 |
|  | (10.7) IP is onset of photoelectron band (81KIM/KAT). | (140) | (585) | -106.9±0.6 | -447.2±2.5 | 77PED/RYL | 616-02-4 |
| $\text{C}_5\text{H}_4\text{S}^+$  | (8.4) IP is onset of photoelectron band (81SCH/SCH). | (239) | (1000) | 45 | 190 | *EST | 77825-99-1 |
| C_5H_5^+ HC≡CCHCH=CH ₂ IP from 84LOS/HOL. | 7.88 | 271 | 1132 | 89 | 372 | 82MCM/GOL | 50706-18-8 |
|  | 8.41 | (252) | (1052) | 58±1 | 241±6 | 82MCM/GOL | 62744-94-9 |
|  | | (242) | (1012) | From appearance energy from $\text{C}_6\text{H}_5\text{CH}_2^+$ precursor, 3.55 eV (78MCC/FRE). | | | |
| $\text{C}_5\text{H}_5\text{BrN}^+$  | | 189 | 793 | From proton affinity of 2-bromopyridine (RN 109-04-6). | | | |

Table 1. Positive Ion Table - Continued

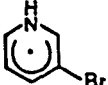
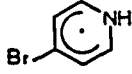
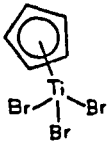
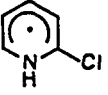
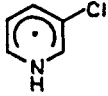

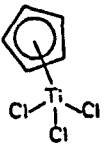
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_5\text{BrN}^+$ | | | | | | | |
|  | | 189 | 791 | | | | |
| | | From proton affinity of 3-bromopyridine (RN 626-55-1). | | | | | |
|  | | 186 | 779 | | | | |
| | | From proton affinity of 4-bromopyridine (RN 1120-87-2). | | | | | |
| $\text{C}_5\text{H}_5\text{Br}_3\text{Ti}^+$ | | | | | | | |
|  | (9.1) | (102) | (428) | -108 | -450 | *EST | 12240-42-5 |
| | | IP is onset of photoelectron band (84TER/LOU). | | | | | |
| $\text{C}_5\text{H}_5\text{ClN}^+$ | | | | | | | |
|  | | 176 | 736 | | | | |
| | | From proton affinity of 2-chloropyridine (RN 109-09-1). PA = 214.4 kcal/mol, 897. kJ/mol. | | | | | |
|  | | 177 | 739 | | | | |
| | | From proton affinity of 3-chloropyridine (RN 626-60-8). PA = 214.8 kcal/mol, 899. kJ/mol. | | | | | |
|  | | 174 | 727 | | | | |
| | | From proton affinity of 4-chloropyridine (RN 626-61-9). PA = 217.8 kcal/mol, 911 kJ/mol. | | | | | |
| $\text{C}_5\text{H}_5\text{Cl}_3\text{Ti}^+$ | | | | | | | |
|  | (9.1) | (76) | (319) | -133±3 | -559±12 | 77PED/RYL | 1270-98-0 |
| | | IP is onset of photoelectron band (84TER/LOU). | | | | | |

Table 1. Positive Ion Table - Continued

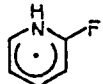
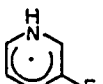
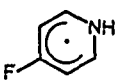


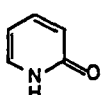
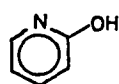
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_5\text{FN}^+$ | | | | | | | |
|  | | 139 | 581 | | | | |
| | | From proton affinity of 2-fluoropyridine (RN 372-48-5). PA = 210.6 kcal/mol, 881 kJ/mol. | | | | | |
|  | | 138 | 577 | | | | |
| | | From proton affinity of 3-fluoropyridine (RN 372-47-4). PA = 214.3 kcal/mol, 897 kJ/mol. | | | | | |
|  | | 135 | 567 | | | | |
| | | From proton affinity of 4-fluoropyridine (RN 694-52-0). PA = 216.6 kcal/mol, 906 kJ/mol. | | | | | |
| $\text{C}_5\text{H}_5\text{F}_3\text{O}_2^+$ $\text{CF}_3\text{COH} = \text{CHCOCH}_3$ | (9.5) | (-20.7) | (-86.7) | -239.8 | -1003.3 | 84ERA/KOL | 367-57-7 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_5\text{H}_5\text{N}^+$ | | | | | | | |
|  | 9.25 | 247 | 1032 | 33±0.2 | 140±1 | 79KUD/KUD3 | 110-86-1 |
| | See also: 83PIA/KEL, 82LIF, 81KIM/KAT. | | | | | | |
| $\text{C}_5\text{H}_5\text{NO}^+$ | | | | | | | |
|  | 8.38±0.02 | (207) | (869) | 14 | 61 | *EST | 694-59-7 |
|  | (8.4) | (176) | (733) | -18±0.5 | -77±2 | 82SUR/ELS | 142-08-5 |
| | IP is onset of photoelectron band. | | | | | | |
|  | 8.6 | (179) | (750) | -19±0.5 | -80±2 | 82SUR/ELS | 109-10-4 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

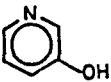
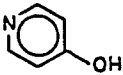
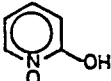
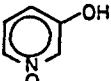
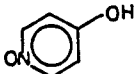

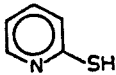
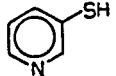
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₅NO⁺ | | | | | | | |
|  | ($\leq 9.15 \pm 0.03$) | (≤ 200) | (≤ 839) | -11 \pm 0.5 | -44 \pm 2 | 82SUR/ELS | 109-00-2 |
|  | 9.75 \pm 0.11 | (215) | (900) | -10 \pm 0.5 | -41 \pm 2 | 82SUR/ELS | 626-64-2 |
| C₅H₅NO₂⁺ | | | | | | | |
|  | (8.90 \pm 0.05) | (167) | (699) | -38 | -160 | *EST | |
|  | (8.60 \pm 0.05) | (168) | (706) | -30 | -124 | *EST | 6602-28-4 |
|  | (8.18 \pm 0.05) | (160) | (668) | -29 | -121 | *EST | 6890-62-6 |
| C₅H₅NS⁺ | | | | | | | |
|  | (7.7) IP is onset of photoelectron band. | (220) | (921) | 43 | 178 | *EST | 2637-34-5 |
|  | ≤ 8.7 IP from 81DRE/BEC, 82LEV/LIA. | (≤ 230) | (≤ 963) | 30 | 124 | *EST | 73018-10-7 |
|  | ($\leq 8.89 \pm 0.03$) | (≤ 239) | (≤ 999) | 34 | 141 | *EST | 16133-26-9 |

Table 1. Positive Ion Table - Continued


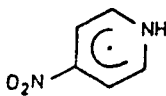
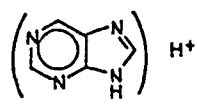
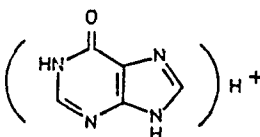
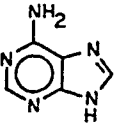
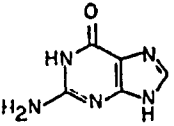
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|-----------------|------------------------------|-------------|---|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_5\text{NS}^+$  | $\leq 9.25 \pm 0.03$ | (≤ 247) | (≤ 1033) | 34 | 141 | *EST | 4556-23-4 |
| $\text{C}_5\text{H}_5\text{N}_2\text{O}_2^+$  | | 190 | 795 | | | From proton affinity of 4-nitropyridine (RN 1122-61-8). PA = 208.5 kcal/mol, 872. kJ/mol. | |
| $\text{C}_5\text{H}_5\text{N}_4^+$  | | 201 | 843 | | | From proton affinity of 9H-purine (RN 120-73-0). PA = 219.3 kcal/mol, 917.5 kJ/mol. | |
| $\text{C}_5\text{H}_5\text{N}_4\text{O}^+$  | | (161) | (673) | | | From proton affinity of hypoxanthine (RN 68-94-0). PA = -217 kcal/mol, -907 kJ/mol. | |
| $\text{C}_5\text{H}_5\text{N}_5^+$  | (7.8) | (229) | (960) | 49 ± 2 | 207 ± 8 | 83KIR/DOM | 73-24-5 |
| $\text{C}_5\text{H}_5\text{N}_5\text{O}^+$  | (7.85) | (181) | (759) | 0.5 | 2 | 77PED/RYL | 73-40-5 |
| C_5H_6^+ | | | | | | | |
| $\text{CH}_2 = \text{C} = \text{CHCH} = \text{CH}_2$ | (8.88) | (265) | (1108) | 60 | 251 | *EST | 10563-01-6 |
| (Z)- $\text{CH}_3\text{CH} = \text{CHC}\equiv\text{CH}$ | 9.14 ± 0.04 | 272 | 1138 | 61 ± 1 | 256 ± 6 | 78SHA | 1574-40-9 |
| (E)- $\text{CH}_3\text{CH} = \text{CHC}\equiv\text{CH}$ | (9.05) | (270) | (1130) | 61 ± 0.7 | 257 ± 3 | 78SHA | 2004-69-5 |
| $\text{CH}_2 = \text{CHC}\equiv\text{CCH}_3$ | 9.00 ± 0.01 | (267) | (1118) | (60) | (250) | *EST | 646-05-9 |

Table 1. Positive Ion Table - Continued


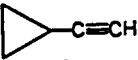



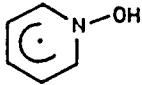
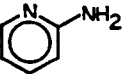
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|------------------------------------|---|---|---|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_5H_6^+ | $\text{CH}_2=\text{C}(\text{CH}_3)\text{C}\equiv\text{CH}$ | 9.23 ± 0.01 | 275 | 1148 | 62 | 258 | 77LEB/RYA | 78-80-8 |
| |  | 8.56 ± 0.01 | 229 | 957 | 31 ± 1 | 131 ± 4 | 77PED/RYL | 542-92-7 |
| |  | (8.7) IP is onset of photoelectron band. | (275) | (1152) | 75 | 313 | *EST | 6746-94-7 |
| |  | (8.0) | (264) | (1103) | 79 | 331 | *EST | 5164-35-2 |
| |  | 9.74 IP from 85HON/HUB. | 308 | 1291 | 84 ± 1 | 351 ± 4 | 85WIB/DAI | 35634-10-7 |
| $\text{C}_5\text{H}_6\text{N}^+$ |  | | 178 | 746 | | | | |
| | | | From proton affinity of pyridine (RN 110-86-1). PA = 220.8 kcal/mol, 924 kJ/mol. | | | | | |
| $\text{C}_5\text{H}_6\text{NO}^+$ |  | | 160 | 669 | | | | |
| | | | From proton affinity of pyridine-N-oxide (RN 694-59-7). PA = 220.3 kcal/mol, 922. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_6\text{N}_2^+$ |  | (8.0) | (213) | (890) | 28 ± 0.2 | 118 ± 1 | 84BIC/PIL | 504-29-0 |
| | | | IP is onset of photoelectron band (82LEV/LIA, 82GUI/KHA). | | | | | |

Table 1. Positive Ion Table - Continued

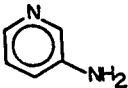
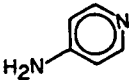
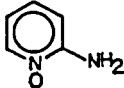
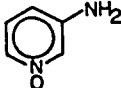
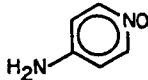
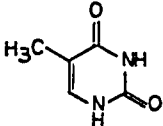
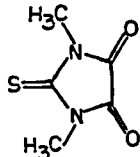
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_6\text{N}_2^+$  | (8.1) | (221) | (926) | 34±0.5 | 144±2 | 84BIC/PIL | 462-08-8 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (8.4) | (225) | (940) | 31±0.2 | 130±1 | 84BIC/PIL | 504-24-5 |
| | IP is onset of photoelectron band. Value of $\Delta_f H(\text{Ion})$ predicted from hydrogen affinity considerations: 214 kcal/mol, 895 kJ/mol, corresponding to IP of 7.9 eV. | | | | | | |
| $\text{C}_5\text{H}_6\text{N}_2\text{O}^+$  | (8.04±0.05) | (197) | (825) | 12 | 49 | *EST | 14150-95-9 |
|  | (8.21±0.05) | (204) | (853) | 15 | 61 | *EST | 1657-32-5 |
|  | (7.67±0.05) | (191) | (797) | 14 | 57 | *EST | 3535-75-9 |
| $\text{C}_5\text{H}_6\text{N}_2\text{O}_2^+$  | (8.8) | (124) | (520) | -79±1 | -329±4 | 77NAB/SAB | 65-71-4 |
| | IP from onset of photoelectron band. | | | | | | |
| $\text{C}_5\text{H}_6\text{N}_2\text{O}_2\text{S}^+$  | 8.6 | (152) | (635) | -47 | -195 | *EST | 21035-65-4 |

Table 1. Positive Ion Table - Continued

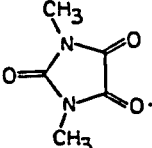
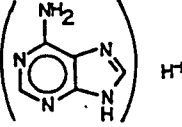
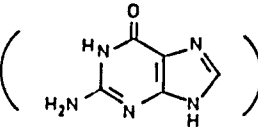

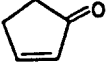
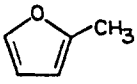
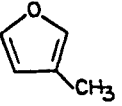
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_6\text{N}_2\text{O}_3^+$  | 10.19 IP from 85ROT/BOC. | (≤ 124) | (≤ 519) | -111 | -464 | *EST | 5176-82-9 |
| $\text{C}_5\text{H}_6\text{N}_5^+$  | | 191 | 802 | | | | |
| | From proton affinity of adenine (RN 73-24-5). PA = 223.5 kcal/mol, 935. kJ/mol. | | | | | | |
| $\text{C}_5\text{H}_6\text{N}_5\text{O}^+$  | | (143) | (599) | | | | |
| | From proton affinity of guanine (RN 73-40-5). PA = -223 kcal/mol, -933 kJ/mol. | | | | | | |
| $\text{C}_5\text{H}_6\text{O}^+$  | 8.4 IP from 86SPI/GRU. | (192) | (803) | -2 \pm 1 | -7 \pm 5 | *EST | 289-65-6 |
|  | $\leq 9.34 \pm 0.02$ | (≤ 196) | (≤ 823) | -19 | -78 | *EST | 930-30-3 |
|  | 8.39 \pm 0.01 IP from 78LIA/AUS, 77ROS/DRA. See also: 83ZYK/ERC, 86SPI/GRU. | (174) | (730) | -19 | -80 | *EST | 534-22-5 |
|  | (8.64) IP from 86SPI/GRU. | (182) | (763) | -17 | -71 | *EST | 930-27-8 |

Table 1. Positive Ion Table - Continued

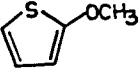
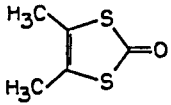
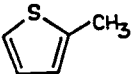
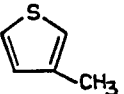


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₆OS⁺ | | | | | | | |
|  | (7.8) IP is onset of photoelectron band (85BAJ/HUM). | (131) | (547) | -49 | -206 | *EST | 16839-97-7 |
| C₅H₆OS₂⁺ | | | | | | | |
|  | (≤ 8.5) IP from 83SCH/SCH. | (≤ 177) | (≤ 740) | -19 | -80 | *EST | 49675-88-9 |
| C₅H₆O₃⁺ | | | | | | | |
| CH ₃ (CO) ₃ CH ₃ | (≤ 9.52) | (≤ 115) | (≤ 482) | -104 | -437 | *EST | 921-11-9 |
| C₅H₆S⁺ | | | | | | | |
|  | 8.61 \pm 0.02 | 218 | 914 | 20.0 \pm 0.2 | 83.5 \pm 0.8 | 77PED/RYL | 554-14-3 |
|  | (8.40) | (213) | (893) | 19.7 \pm 0.2 | 82.6 \pm 0.8 | 77PED/RYL | 616-44-4 |
|  | (7.9) IP is onset of photoelectron band. | (224) | (940) | 42 \pm 2 | 178 \pm 8 | *EST | 289-70-3 |
| C₅H₆Si⁺ | | | | | | | |
|  | (7.8) IP is onset of photoelectron band (84BOC/ROS). | (197) | (824) | 17 | 71 | 83GOR/BOU | 289-77-0 |
| C₅H₇⁺ | | | | | | | |
| CH ₂ =CHCHCH=CH ₂ | Fw sup _g RN goes with (7.25) See also: 80WOL/HOL. | (220) | (922) | 53 | 222 | 69GOL/BEN | 14362-08-4 |
| HC \equiv CC(CH ₃) ₂ | (7.44) See also: 80WOL/HOL. | (234) | (981) | 63 | 263 | 76LOS/TRA | 56897-57-5 |

Table 1. Positive Ion Table - Continued


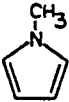
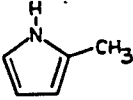
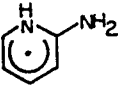
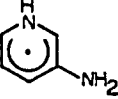
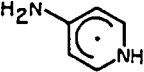
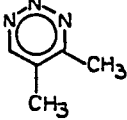
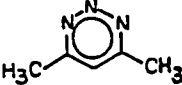
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_5H_7^+  | 7.00 | 199 | 833 | 38 | 159 | 70FUR/GOL | 54846-63-8 |
| | Proton affinity of cyclopentadiene (RN 542-92-7) (PA = 199.6 kcal/mol, 835. kJ/mol) leads to $\Delta_f H(\text{Ion}) = 197$ kcal/mol, 826 kJ/mol. IP from 80WOL/HOL. | | | | | | |
| $\text{C}_5\text{H}_7\text{N}^+$  | 7.94±0.02 | 207.6 | 869.2 | 24.6±0.1 | 103.1±0.5 | 77PED/RYL | 96-54-8 |
|  | (7.78±0.01) | (197) | (825) | 18±0.2 | 74±1 | *EST | 636-41-9 |
| $\text{C}_5\text{H}_7\text{N}_2^+$  | | 170 | 711 | | | | |
| | From proton affinity of 2-pyridinamine (RN 504-29-0). PA = 223.8 kcal/mol, 936. kJ/mol. | | | | | | |
|  | | 179 | 747 | | | | |
| | From proton affinity of 3-pyridinamine (RN 462-08-8). PA = 221.0 kcal/mol, 925. kJ/mol. | | | | | | |
|  | | (169) | (706) | | | | |
| | From proton affinity of 4-pyridinamine (RN 504-24-5). PA = (230) kcal/mol, (962) kJ/mol. | | | | | | |
| $\text{C}_5\text{H}_7\text{N}_3^+$  | (≤9.5) | (≤302) | (≤1263) | 83 | 346 | *EST | 86402-31-5 |
| | IP from 83GLE/SPA. | | | | | | |
|  | (≤9.5) | (≤300) | (≤1257) | 81 | 340 | *EST | 77202-09-6 |
| | IP from 83GLE/SPA. | | | | | | |

Table 1. Positive Ion Table - Continued

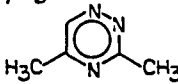
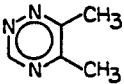

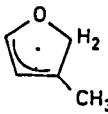
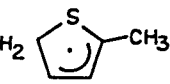
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------------|--|-----------------|------------------------------|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₇N₃⁺ | | | | | | | |
|  | (≤ 9.02) | (≤ 268) | (≤ 1123) | 60 | 253 | *EST | 24108-34-7 |
|  | (≤ 9.15) | (≤ 274) | (≤ 1145) | 63 | 262 | *EST | 21134-90-7 |
| C₅H₇O⁺ | | | | | | | |
| (CH ₃) ₂ C=CHCO | | (138) | (577) | | | | 44391-34-6 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (85ALA/ATT). | | | | | |
|  | | 140 | 587 | | | | |
| | | From proton affinity of 2-methylfuran (RN 534-22-5) (85HOU/ROL, 86MAU/LIE, 86SAN/BAL). PA = 206.2 kcal/mol, 863. kJ/mol. | | | | | |
|  | | 145 | 606 | | | | |
| | | From proton affinity of 3-methylfuran (RN 930-27-8) (85HOU/ROL). PA = 204.0 kcal/mol, 853. kJ/mol. | | | | | |
| C₅H₇S⁺ | | | | | | | |
|  | | 180 | 754 | | | | |
| | | From proton affinity of 2-methylthiophene (RN 554-14-3) (86MAU). PA = 205.4 kcal/mol, 859. kJ/mol. | | | | | |
| C₅H₈⁺ | | | | | | | |
| CH ₂ =C=CHCH ₂ CH ₃ | 9.22 | 246 | 1030 | 33.6 \pm 0.2 | 140.7 \pm 0.6 | 77PED/RYL | 591-95-7 |
| (Z)-CH ₂ =CHCH=CHCH ₃ | 8.63 \pm 0.03 | 218 | 914 | 19.4 \pm 0.2 | 81.1 \pm 1.0 | 77PED/RYL | 1574-41-0 |
| | IP from 81MAS/MOU. | | | | | | |
| (E)-CH ₂ =CHCH=CHCH ₃ | 8.59 \pm 0.02 | 216 | 905 | 18.2 \pm 0.1 | 76.3 \pm 0.6 | 77PED/RYL | 2004-70-8 |
| | IP from 81MAS/MOU. | | | | | | |
| CH ₂ =CHCH ₂ CH=CH ₂ | (9.62 \pm 0.02) | (247) | (1034) | 25.3 \pm 0.2 | 105.7 \pm 0.6 | 77PED/RYL | 591-93-5 |
| CH ₃ CH=C=CHCH ₃ | (8.7) | (232) | (972) | 31.8 \pm 0.2 | 133.1 \pm 0.7 | 77PED/RYL | 591-96-8 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued


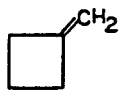
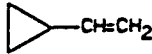
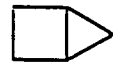


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_5H_8^+ <chem>CH2=C(CH3)CH=CH2</chem> | 8.84±0.01 See also: 81MAS/MOU. | 221.8 | 927.9 | 17.9±.2 | 75±1 | 77PED/RYL | 78-79-5 |
| <chem>C3H7C=CH</chem> | 10.05 IP from 81HOL/FIN. | 266 | 1114 | 34.4±1 | 144±4 | 79ROG/DAG | 627-19-0 |
| <chem>C2H5C=CCH3</chem> | 9.44±0.01 | 248 | 1039 | 30.6±1 | 128±4 | 79ROG/DAG | 627-21-4 |
| <chem>(CH3)2CHC=CH</chem> | 9.97 IP from 81HOL/FIN. | 262 | 1098 | 32.5 | 136 | 69BEN/CRU | 598-23-2 |
|  | 9.01±0.02 See also: 81KIM/KAT. | 216 | 905 | 8.6 | 36 | 82ALL/DOD | 142-29-0 |
|  | 9.16±0.02 | 241 | 1008 | 29.6±.2 | 124±1 | 78LEB/TSV | 1120-56-5 |
|  | (8.7) | (236) | (988) | 35.6±.2 | 149±1 | 77PED/RYL | 693-86-7 |
|  | (8.7±0.1) | (238) | (997) | 37.8 | 158 | 82WIB/WEN | 185-94-4 |
|  | (9.65) | (272) | (1139) | 49.7 | 208 | 82WIB/WEN | 311-75-1 |
|  | 9.26 See also: 86GLE/KRE. | 258 | 1078 | 44.2±0.2 | 185.1±0.7 | 77PED/RYL | 157-40-4 |

Table 1. Positive Ion Table - Continued

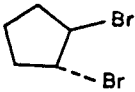
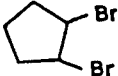
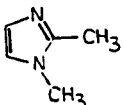


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_8\text{Br}_2^+$ | | | | | | | |
|  | 10.06 | (218) | (913) | -14 | -58 | *EST | 10230-26-9 |
|  | (10.02±0.02) | (217) | (909) | -14 | -58 | *EST | 33547-17-0 |
| $\text{C}_5\text{H}_8\text{F}_3\text{O}_2^+$ $\text{CF}_3\text{C}(\text{OH})\text{O}(\text{n-C}_3\text{H}_7)$ | | | | | | | |
| | | -74 | -311 | | | | |
| | | From proton affinity of $\text{CF}_3\text{COO}(\text{n-C}_3\text{H}_7)$ (RN 383-66-4). | | | | | |
| $\text{C}_5\text{H}_8\text{N}_2^+$ | | | | | | | |
|  | (≤8.38) | (≤224) | (≤936) | 30 | 127 | *EST | 1739-84-0 |
|  | 8.45±0.04 | 244 | 1022 | 49±0.7 | 207±3 | 80ENG | 2721-32-6 |
| $\text{C}_5\text{H}_8\text{N}_2\text{O}^+$ | | | | | | | |
|  | (9.2) IP is onset of photoelectron band. | (243) | (1015) | 30.55±0.3 | 127.8±1.4 | 83BYS | 22509-00-8 |
| $\text{C}_5\text{H}_8\text{O}^+$ | | | | | | | |
| (E)- $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCHO}$ | (9.70) | (194) | (810) | -30 | -126 | 83HOL | 764-39-6 |
| $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{CHO}$ | (9.60) | (188) | (787) | -33 | -139 | 83HOL | 497-03-0 |
| $\text{C}_2\text{H}_5\text{COCH}=\text{CH}_2$ | (9.50) | (186) | (781) | -33 | -136 | 83HOL | 1629-58-9 |
| (E)- $\text{CH}_3\text{CH}=\text{CHC}(=\text{O})\text{CH}_3$ | (9.39) | (175) | (732) | -42 | -174 | 84BOU/HOP | 625-33-2 |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{C}(=\text{O})\text{CH}_3$ | (9.50) | (177) | (741) | -42 | -176 | 84BOU/HOP | 814-78-8 |

Table 1. Positive Ion Table - Continued


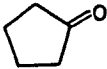
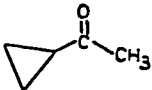

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_8\text{O}^+$ (E)- $\text{CH}_3\text{OCH}=\text{CHCH}=\text{CH}_2$ | (8.03) | (222) | (931) | 37 | 156 | *EST | 10034-09-0 |
|  | 8.34±0.01 | 162 | 679 | -29.9±0.4 | -125.2±1.5 | 77PED/RYL | 110-87-2 |
|  | 9.25±0.01 See also: 82BIE/ASB. | 167 | 698 | -46±0.5 | -194±2 | 77PED/RYL | 120-92-3 |
|  | 9.46 | 190 | 794 | -28±0.2 | -119±1 | 83FUC/SMI | 765-43-5 |
| $\text{C}_5\text{H}_8\text{OS}^+$ (Z)- $\text{CH}_3\text{C}(\text{S})\text{CH}=\text{C}(\text{OH})\text{CH}_3$ | (8.4) IP is onset of photoelectron band (81JOR/CAR). | (160) | (670) | -33 | -140 | *EST | 73059-87-7 |
| (Z)- $\text{CH}_3\text{C}(\text{SH})\text{CHC}(\text{O})\text{CH}_3$ | (≤8.73) IP from 81JOR/CAR. | (≤168) | (≤702) | -33 | -140 | *EST | 65581-04-6 |
|  | (8.90±0.05) | (168) | (704) | -37±0.7 | -155±3 | 77PED/RYL | 1072-72-6 |
| $\text{C}_5\text{H}_8\text{O}_2^+$ $\text{C}_2\text{H}_5\text{CH}=\text{CHCOOH}$ | (10.14) | (144) | (601) | -90±2 | -377±8 | *EST | 626-98-2 |
| $(\text{CH}_3)_2\text{C}=\text{CHCOOH}$ | (9.63) | (124) | (519) | -98 | -410 | *EST | 541-47-9 |
| $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{COOH}$ | (9.50) | (121) | (507) | -98 | -410 | *EST | 13201-46-2 |
| $\text{CH}_2=\text{C}(\text{C}_2\text{H}_5)\text{COOH}$ | (10.06) | (139) | (582) | -93 | -389 | *EST | 3586-58-1 |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{CH}_2\text{COOH}$ | (9.52) | (128) | (536) | -92 | -383 | *EST | 53774-20-2 |

Table 1. Positive Ion Table - Continued


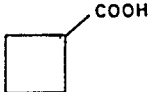

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|---------|------------------------------|--------|----------------------|---------------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_8\text{O}_2^+$ $\text{CH}_3\text{CH}=\text{CHCH}_2\text{COOH}$ | (9.41) | (126) | (527) | -91 | -381 | *EST | 1617-32-9 |
| $\text{CH}_2=\text{CHCOOC}_2\text{H}_5$ | (> 10.3) IP from 82MOR/MER. | (> 147) | (> 617) | -90 | -377 | *EST | 140-88-5 |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{COOCH}_3$ | (9.7) IP is onset of photoelectron band(78VAN/OSK). | (141) | (588) | -83 | -348 | 80VIL/PER | 80-62-6 |
| $\text{CH}_3\text{COCH}_2\text{COCH}_3$ | 8.85±0.02 Enol form, $\text{CH}_3\text{COCH}=\text{C}(\text{OH})\text{CH}_3$, is preferred. | 112 | 470 | -92±0.2 | -384±1 | 79HAC/PIL | 123-54-6 |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{OC}(=\text{O})\text{CH}_3$ | 9.1 IP is onset of photoelectron band (78VAN/OSK). See also: 82LEV/LIA. | 126 | 529 | -83 | -349 | 77PED/RYL | 108-22-5 591-87-7 |
|  | ≤9.54 IP from 82ZVE/VIL. | (≤163) | (≤682) | -57 | -238 | *EST | 5417-32-3 |
|  | (10.35) | (154) | (645) | -85 | -354 | *EST | 3721-95-7 |
| $\text{C}_5\text{H}_8\text{Si}^+$ | | | | | | | |
|  | (9.1) IP is onset of photoelectron band (84BOC/ROS). | (238) | (997) | 28 | 119 | *EST | 81200-77-3 |
| C_5H_9^+ | | | | | | | |
| $\text{CH}_2=\text{CHCHCH}_2\text{CH}_3$ | (7.30) | (193) | (810) | 25 | 106 | 76LOS/TRA | 17829-37-7 |
| $\text{CH}_3\text{CHCH}=\text{CHCH}_3$ | (7.07) Heat of formation of ion from proton affinity of (E)-1,3-pentadiene (RN 2004-70-8). PA = (201.8) kcal/mol, (844.) kJ/mol. | (182) | (763) | (22) | (92) | 76LOS/TRA | 51685-67-7 ^a |
| $\text{CH}_3\text{CH}=\text{CC}_2\text{H}_5$ | | 200 | 838 | | | | |
| | | From proton affinity of 2-pentyne (RN 627-21-4). PA = (196) kcal/mol, (820) kJ/mol. | | | | | |
| $(\text{CH}_3)_2\text{CCH}=\text{CH}_2$ | (7.13) Heat of formation of ion from proton affinity of 2-methyl-1,3-butadiene (RN 78-79-5). PA = (200.4) kcal/mol, (838.) kJ/mol. See also: 85LAD/HAR. | (183) | (767) | 19 | 81 | 76LOS/TRA | 29791-12-6 |

Table 1. Positive Ion Table - Continued


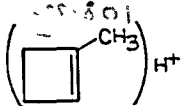
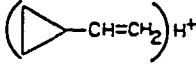
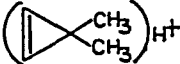
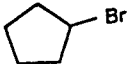
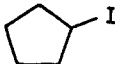
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_5H_9^+ $(\text{CH}_3)_2\text{CHC}=\text{CH}_2$ | | (200) | (838) | | | | |
| | | From proton affinity of 3-methyl-1-butyne (RN 598-23-2). PA = (198) kcal/mol, (828) kJ/mol. | | | | | |
| $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{CH}_2$ | | 190 | 797 | | | | 60288-51-9 |
| | | Heat of formation of ion from appearance potential measurements (84LOS/HOL). | | | | | |
|  | 7.21 | 191.4 | 800.8 | 24±1 | 102±4 | 82MCM/GOL | 3889-74-5 |
| | | Value of $\Delta_f H(\text{Ion})$ from hydride and chloride transfer equilibrium constant determinations (76SOL/FIE, 85SHA/SHA), and from proton affinity of cyclopentene (84LIA/LIE). PA = 183.4 kcal/mol, 767.5 kJ/mol. IP from 79HOU. $\Delta_f H(\text{Ion})$ -IP leads to $\Delta_f H(\text{Neutral}) = 25$ kcal/mol, 105 kJ/mol. | | | | | |
|  | | (193) | (807) | | | | 53249-17-5 |
| | | From proton affinity of 1-methylcyclobutene (RN 1489-60-7). PA = 201 kcal/mol, 841 kJ/mol. | | | | | |
|  | | 204 | 852 | | | | |
| | | From proton affinity of vinylcyclopropane (RN 693-86-7). PA = 197.6 kcal/mol, 827. kJ/mol. | | | | | |
|  | | (213) | (890) | | | | 63974-90-3 |
| | | From proton affinity of 3,3-dimethylcyclopropene (RN 3907-06-0). PA = 203 kcal/mol, 849 kJ/mol. | | | | | |
| $\text{C}_5\text{H}_9\text{Br}^+$ | | | | | | | |
|  | (9.94±0.02) | (213) | (891) | -16 | -68 | *EST | 137-43-9 |
| $\text{C}_5\text{H}_9\text{BrO}^+$ | | | | | | | |
| $(\text{CH}_3)_2\text{CBrCOCH}_3$ | (9.35) | (154) | (646) | -61 | -256 | *EST | 2647-11-7 |
| | | IP from 84BOU/DAG. | | | | | |
| $\text{C}_5\text{H}_9\text{I}^+$ | | | | | | | |
|  | 9.07 | (206) | (861) | -3 | -14 | *EST | 1556-18-9 |

Table 1. Positive Ion Table - Continued



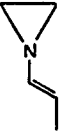
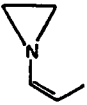
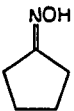
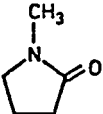
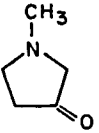
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------------|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_9\text{N}^+$ | | | | | | | | |
| | n-C ₄ H ₉ NC | (11.1) | (280) | (1173) | 24±0.5 | 102±2 | *EST | 2769-64-4 |
| | (CH ₃) ₂ NCH ₂ C≡CH | (8.17) | (242) | (1013) | 54±1 | 225±5 | *EST | 7223-38-3 |
| | | See also: 81LOG/TAK. | | | | | | |
| |  | (8.0) | (201) | (845) | 17 | 73 | 74PIH/TAS | 694-05-3 |
| | | IP is onset of photoelectron band. | | | | | | |
| |  | (≤8.21±0.05) | (≤216) | (≤907) | 27±0.5 | 115±2 | *EST | 554-15-4 |
| |  | (7.9) | (224) | (939) | 42 | 177 | *EST | |
| | | IP is onset of photoelectron band (81MUL/PRE). | | | | | | |
| |  | (8.0) | (227) | (949) | 42 | 177 | *EST | |
| | | IP from 81MUL/PRE. | | | | | | |
| $\text{C}_5\text{H}_9\text{NO}^+$ | | | | | | | | |
| | n-C ₄ H ₉ NCO | (10.14±0.05) | (186) | (776) | -48 | -202 | *EST | 111-36-4 |
| | tert-C ₄ H ₉ CNO | ≤9.55±0.005 | (≤223) | (≤931) | 2 | 10 | *EST | 27143-81-3 |
| |  | (8.92±0.03) | (193) | (809) | -12 | -52 | *EST | 1192-28-5 |
| | | IP from 79GOL/KUL. | | | | | | |
| |  | ≤9.17 | ≤161 | ≤674 | -50 | -211 | 77PED/RYL | 872-50-4 |
| | | IP from 85TRE/RAD. | | | | | | |
| |  | (8.3) | (165) | (691) | -26 | -110 | *EST | 68165-06-0 |
| | | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

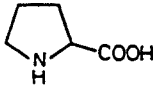
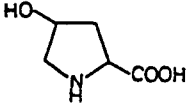
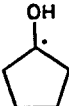

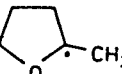
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|--|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_9\text{NO}_2^+$ | | (8.3) | (104) | (435) | -87±1 | -366±4 | 78SAB/LAF | 609-36-9 |
|  | | | | | | | | IP is onset of photoelectron band(83CAN/HAM). |
| $\text{C}_5\text{H}_9\text{NO}_3^+$ | | (9.2) | (62) | (260) | -150 | -628 | *EST | 97-69-8 |
| $\text{CH}_3\text{CONHCH}(\text{CH}_3)\text{COOH}$ | | | | | | | | IP is onset of photoelectron band (83CAN/HAM). |
|  | | (≤9.1) | (≤87) | (≤362) | -123 | -516 | 77PED/RYL | 51-35-4 |
| $\text{C}_5\text{H}_9\text{O}^+$ | | | | | | | | |
| $\text{CH}_3\text{C}(\text{OH})\text{C}(\text{CH}_2)\text{CH}_3$ | | | 121 | 507 | | | | From proton affinity of $\text{CH}_3\text{C}(=\text{O})\text{C}(\text{=O})\text{CH}_3$ (RN 814-78-8) (84BOU/HOP). PA = 202.4 kcal/mol, 847. kJ/mol. |
| (E)- $\text{CH}_3\text{CHCHC}(\text{OH})\text{CH}_3$ | | | 117 | 491 | | | | From proton affinity of (E)- $\text{CH}_3\text{CH}=\text{CHC}(\text{=O})\text{CH}_3$ (RN 625-33-2) (84BOU/HOP). PA = 206.7 kcal/mol, 865. kJ/mol. |
|  | | | 121 | 506 | | | | From proton affinity of cyclopentanone (RN 120-92-3). PA = 198.8 kcal/mol, 832. kJ/mol. |
|  | | | 129 | 539 | | | | From proton affinity of 3,4-dihydro-4H-pyran (RN 110-87-2) (86BOU/HAN). PA = 206.9 kcal/mol, 866. kJ/mol. |
|  | | | 121 | 507 | | | | From proton affinity of 2-methyl-4,5-dihydrofuran (RN 1487-15-6) (86BOU/DJA). PA = 215.6 kcal/mol, 902. kJ/mol. |

Table 1. Positive Ion Table - Continued

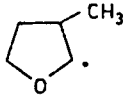
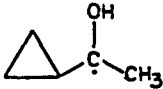
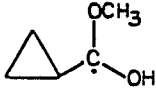
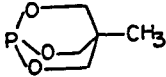
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₉O⁺ | | | | | | | |
|  | | 132 | 552 | | | | |
| | | From proton affinity of 3-methyl-4,5-dihydrofuran (RN 557-31-3) (86BOU/DJA). PA = 207.0 kcal/mol, 866. kJ/mol. | | | | | |
|  | | 133 | 555 | | | | |
| | | From proton affinity of 1-cyclopropylethanol (RN 765-43-5). PA = 205.1 kcal/mol, 858. kJ/mol. | | | | | |
| C₅H₉O₂⁺ | | | | | | | |
| CH ₃ C(OH)CHC(OH)CH ₃ | | 66 | 277 | | | | |
| | | From proton affinity of CH ₃ COCH = C(OH)CH ₃ (RN 123-54-6). PA = 207.8 kcal/mol, 869. kJ/mol. | | | | | |
|  | | 89 | 373 | | | | |
| | | From proton affinity of methylcyclopropane carboxylate (RN 2868-37-3). PA = 202.9 kcal/mol, 849. kJ/mol. | | | | | |
| C₅H₉O₃P⁺ | | | | | | | |
|  | (9.2) | (76) | (317) | -136 | -571 | *EST | 1449-91-8 |
| | IP is onset of photoelectron band. (77COW/GOO). | | | | | | |
| C₅H₁₀⁺ | | | | | | | |
| 1-C ₅ H ₁₀ | 9.52±0.02 | 214 | 897 | -5.1±0.1 | -21.4±0.4 | 84WIB/WAS | 109-67-1 |
| | | 221.5 | 926.9 | 2.0 | 8.4 | | |
| | See also: 83HOL/LOS, 86TRA, 84BRA/BAE. | | | | | | |
| 2-(Z)-C ₅ H ₁₀ | 9.036±0.005 | 202.0 | 845.3 | -6.3±0.1 | -26.5±0.4 | 84WIB/WAS | 627-20-3 |
| | See also: 86TRA. | | | | | | |
| 2-(E)-C ₅ H ₁₀ | 9.036±0.005 | 200.8 | 840.3 | -7.5±0.1 | -31.5±0.4 | 84WIB/WAS | 646-04-8 |
| | | 208.0 | 870.1 | -0.4 | -1.7 | | |
| | See also: 84BRA/BAE. | | | | | | |
| (CH ₃) ₂ CHCH = CH ₂ | 9.52±0.02 | 213 | 891 | -6.5±0.2 | -27.4±0.6 | 77PED/RYL | 563-45-1 |
| | | 220.3 | 921.6 | 0.7 | 3.1 | | |
| | See also: 84BRA/BAE. | | | | | | |
| C ₂ H ₅ C(CH ₃) = CH ₂ | 9.13±0.03 | 202 | 845 | -8.5±0.2 | -35.6±0.7 | 77PED/RYL | 563-46-2 |
| | | 209.3 | 875.8 | -1.2 | -5.1 | | |
| | See also: 86TRA, 84BRA/BAE. | | | | | | |

Table 1. Positive Ion Table - Continued


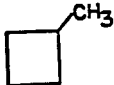

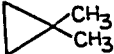


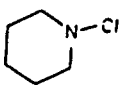
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{10}^+$ $(\text{CH}_3)_2\text{C}=\text{CHCH}_3$ | 8.68±0.01 | 190 197.4 | 795 825.8 | -10.1±0.1 -2.8 | -42.1±0.6 -11.6 | 77PED/RYL | 513-35-9 |
| | | See also: 86TRA, 84BRA/BAE. | | | | | |
|  | 10.51±0.05 | 224 231.8 | 936 969.8 | -18.7±0.2 -10.6 | -78.4±0.8 -44.2 | 77PED/RYL | 287-92-3 |
| | | See also: 81MAU/SIE, 81KIM/KAT, 86TRA, 84BRA/BAE. | | | | | |
|  | (9.60) | (221) | (923) | -0.7 | -3 | *EST | 598-61-8 |
|  | (9.50) | (218) | (912) | -1 | -5 | 77PED/RYL | 1191-96-4 |
|  | (9.08) | (207) | (868) | -2 | -8 | 77PED/RYL | 1630-94-0 |
| | | See also: 81PLE/VIL. | | | | | |
|  | (9.76±0.02) | (225) | (942) | 0 | 0 | 77PED/RYL | 930-18-7 |
|  | (9.73±0.02) | (223) | (934) | -1 | -5 | 77PED/RYL | 2402-06-4 |
| $\text{C}_5\text{H}_{10}\text{Br}_2^+$ $\text{Br}(\text{CH}_2)_5\text{Br}$ | (≤10.23) | (≤207) | (≤868) | -28 | -119 | *EST | 111-24-0 |
| $\text{C}_5\text{H}_{10}\text{ClN}^+$ | | | | | | | |
|  | (8.5) | (208) | (871) | 12 | 51 | *EST | 2156-71-0 |
| | | IP is onset of photoelectron band. | | | | | |

Table 1. Positive Ion Table - Continued

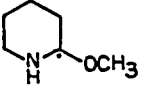
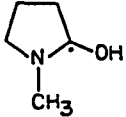
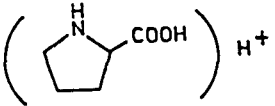
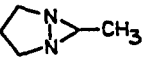
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₁₀N⁺ | | | | | | | |
| n-C ₄ H ₉ CNH | | 174 | 728 | | | | |
| | | From proton affinity of n-C ₄ H ₉ CN (RN 110-59-8). PA = 194.0 kcal/mol, 812. kJ/mol. | | | | | |
| t-C ₄ H ₉ CNH | | 169 | 709 | | | | |
| | | From proton affinity of t-C ₄ H ₉ CN (86MAR/TOP, 86MAU/KAR). PA = 195.7 kcal/mol, 819. kJ/mol. | | | | | |
| t-C ₄ H ₉ NCH | | 178 | 744 | | | | |
| | | From proton affinity of t-C ₄ H ₉ CN (RN 7188-38-7) (86MAU/KAR). PA = 207.5 kcal/mol, 868 kJ/mol. | | | | | |
| C₅H₁₀NO⁺ | | | | | | | |
|  | | 103 | 433 | | | | |
| | | From proton affinity of 2-methoxy-1-pyrroline (RN 5264-35-7). PA = 225.9 kcal/mol, 945. kJ/mol. | | | | | |
|  | | 98 | 412 | | | | |
| | | From proton affinity of 1-methyl-2-pyrrolidinone (RN 872-50-4). PA = 216.8 kcal/mol, 907. kJ/mol. | | | | | |
| C₅H₁₀NO₂⁺ | | | | | | | |
|  | | 58 | 243 | | | | |
| | | From proton affinity of L-proline (RN 609-36-9). PA = 220.2 kcal/mol, 921. kJ/mol. | | | | | |
| C₅H₁₀NO₃⁺ | | | | | | | |
| CH ₃ C(OH)NHCH ₂ COOCH ₃ | | 8 | 34 | | | | |
| | | From proton affinity of CH ₃ CONHCH ₂ COOCH ₃ . PA = 217.7 kcal/mol, 911. kJ/mol. | | | | | |
| C₅H₁₀NO₄⁺ | | | | | | | |
| L-HOOC(CH ₂) ₂ CH(NH ₃)COOH | | 29 | 121 | | | | |
| | | From proton affinity of L-HOOC(CH ₂) ₂ CH(NH ₂)COOH (RN 617-65-2). PA = 216.5 kcal/mol, 906. kJ/mol. | | | | | |
| C₅H₁₀N₂⁺ | | | | | | | |
|  | (≤8.78) | (≤264) | (≤1108) | 62 | 261 | *EST | 6794-96-3 |

Table 1. Positive Ion Table - Continued

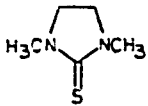
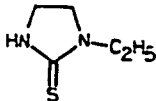
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₁₀N₂S⁺ | | | | | | | |
|  | ≤ 7.95 See also: 80AND/DEV. | (≤ 208) | (≤ 869) | 24 | 102 | *EST | 13461-16-0 |
|  | (7.7) IP is onset of photoelectron band (80AND/DEV). | (194) | (813) | 17 | 70 | *EST | 29704-02-7 |
| C₅H₁₀N₄⁺ NCN = C(N(CH ₃) ₂)(NHCH ₃) | (8.2) IP is onset of photoelectron band (80KLA/BUT). | (227) | (950) | 38 | 159 | *EST | 17686-53-2 |
| C₅H₁₀O⁺ | | | | | | | |
| n-C ₄ H ₉ CHO | 9.74±0.04 See also: 81HOL/FIN. | 169 | 709 | -55.1±0.5 | -230.5±2 | 77PED/RYL | 110-62-3 |
| sec-C ₄ H ₉ CHO | (9.59±0.01) | (165) | (689) | -56 | -236 | *EST | 96-17-3 |
| iso-C ₄ H ₉ CHO | 9.70±0.02 See also: 81HOL/FIN. | (167) | (699) | -57 | -237 | *EST | 590-86-3 |
| tert-C ₄ H ₉ CHO | 9.50 | (161) | (673) | -58 | -244 | *EST | 630-19-3 |
| n-C ₃ H ₇ COCH ₃ | 9.38±0.01 See also: 84OLI/GUE. | 154.4 | 645.9 | -61.9±0.2 | -259.1±0.8 | 77PED/RYL | 107-87-9 |
| (C ₂ H ₅) ₂ CO | 9.31±0.01 See also: 81HOL/FIN. | 153.0 | 639.9 | -61.7±0.2 | -258.4±0.7 | 77PED/RYL | 96-22-0 |
| iso-C ₃ H ₇ COCH ₃ | 9.30±0.01 | 151.8 | 634.9 | -62.7±0.2 | -262.4±0.8 | 77PED/RYL | 563-80-4 |
| CH ₂ = CHCH ₂ CH ₂ CH ₂ OH | (9.42±0.05) IP from 83HOL/LOS. | (176) | (737) | -41 | -172 | *EST | 821-09-0 |
| CH ₂ = CHC(CH ₃) ₂ OH | (≤ 9.90) | (≤ 198) | (≤ 830) | -30 | -125 | 84GUB/GER | 115-18-4 |
| CH ₂ = CHCH ₂ CH(OH)CH ₃ | (9.38±0.05) IP from 83HOL/LOS. | (171) | (717) | -45 | -188 | *EST | 625-31-0 |
| CH ₂ = CHCH(OH)CH ₂ CH ₃ | 9.40±0.05 IP from 83HOL/LOS. See also: 84ZWI/HAR. | (173) | (725) | -43 | -182 | *EST | 616-25-1 |

Table 1. Positive Ion Table - Continued


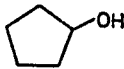

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------------------------|----------------|------------------------------|------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{10}\text{O}^+$ | $\text{CH}_2 = \text{CHOCH}(\text{CH}_3)_2$ | (≤ 8.90) | (≤ 164) | (≤ 685) | -42 \pm 1 | -174 \pm 5 | 81TRO/NED | 926-65-8 |
|  | | 9.25 \pm 0.01 See also: 81KIM/KAT. | 160 | 669 | -53.3 \pm 0.2 | -223.0 \pm 0.7 | 77PED/RYL | 142-68-7 |
|  | | 9.72 IP from 85TRA. | 166 | 695 | -58.0 \pm 0.3 | -242.6 \pm 1.2 | 77PED/RYL | 96-41-3 |
| $\text{C}_5\text{H}_{10}\text{OS}^+$ | $\text{CH}_3\text{COCH}_2\text{SCH}_2\text{CH}_3$ | (≤ 8.72) IP from 84OLI/GUE. | (≤ 153) | (≤ 638) | -49 | -203 | *EST | 20996-62-7 |
| $\text{C}_5\text{H}_{10}\text{O}_2^+$ | n-C ₄ H ₉ COOH | (≤ 10.53) | (≤ 126) | (≤ 526) | -117 \pm 0.5 | -490 \pm 2 | 77PED/RYL | 109-52-4 |
| | iso-C ₄ H ₉ COOH | (≤ 10.51) | (≤ 119) | (≤ 499) | -123 \pm 1 | -515 \pm 6 | 77PED/RYL | 503-74-2 |
| | tert-C ₄ H ₉ COOH | (10.08) IP from 81HOL/FIN. | (110) | (460) | -122 | -512 | *EST | 75-98-9 |
| | HCOO(CH ₂) ₃ CH ₃ | 10.50 \pm 0.02 | (139) | (583) | -103 | -430 | *EST | 592-84-7 |
| | CH ₃ COOCH ₂ CH ₂ CH ₃ | 10.04 \pm 0.03 | (123) | (515) | -109 | -454 | *EST | 109-60-4 |
| | CH ₃ COOCH(CH ₃) ₂ | 9.99 \pm 0.03 | 115 | 482 | -115.1 \pm 0.1 | -481.5 \pm 0.6 | 77PED/RYL | 108-21-4 |
| | C ₂ H ₅ COOC ₂ H ₅ | (10.00 \pm 0.02) | (120) | (501) | -111 \pm 0.5 | -464 \pm 2 | 77PED/RYL | 105-37-3 |
| | n-C ₃ H ₇ COOCH ₃ | 10.07 \pm 0.03 | (124) | (520) | -108 | -452 | *EST | 623-42-7 |
| | iso-C ₃ H ₇ COOCH ₃ | 9.86 IP from 83BUR/HOL3. | 118 | 495 | -109 \pm 0.2 | -456 \pm 1 | 83FUC/SMI | 547-63-1 |
|  | | (≤ 9.75) | (≤ 190) | (≤ 797) | -34 | -144 | *EST | 505-63-5 |

Table 1. Positive Ion Table - Continued


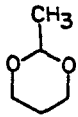
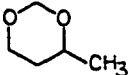
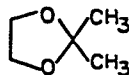

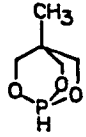

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|---|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_5H_{10}O_2^+$ | | | | | | | |
|  | (9.45) | (135) | (565) | -83 ± 0.5 | -347 ± 2 | 77PED/RYL | 505-65-7 |
|  | (≤ 10.03) IP from 84ASF/ZYK | (≤ 136) | (≤ 570) | -95.1 ± 0.07 | -397.8 ± 2.9 | 77PED/RYL | 626-68-6 |
|  | (≤ 10.04) IP from 84ASF/ZYK. | (≤ 141) | (≤ 592) | -90.1 ± 0.7 | -376.9 ± 3.1 | 77PED/RYL | 1120-97-4 |
|  | (9.2) IP is onset of photoelectron band. | (120) | (502) | -92 ± 0.2 | -386 ± 1 | *EST | 2916-31-6 |
| $C_5H_{10}O_2S^+$ | | | | | | | |
|  | 8.67 ± 0.05 IP from 72CON/COL. | 128 | 537 | -72 | -300 | 72CON/COL | 2094-92-0 |
| $C_5H_{10}O_3P^+$ | | | | | | | |
|  | | 19 | 80 | From proton affinity of 4-methyl-2,6,7-trioxa-1-phosphabicyclo[2.2.2]octane (RN 1449-91-8). PA = 210.0 kcal/mol, 879. kJ/mol. | | | |
| $C_5H_{10}S^+$ | | | | | | | |
| $CH_2 = CHCH_2SC_2H_5$ | (8.51 ± 0.01) | (200) | (839) | 4 ± 0.7 | 18 ± 3 | 77PED/RYL | 5296-62-8 |
|  | (8.2) IP is onset of photoelectron band (80SAR/WOR, 82LEV/LIA). | (174) | (730) | -15.2 ± 0.2 | -63.5 ± 0.7 | 77PED/RYL | 1613-51-0 |

Table 1. Positive Ion Table - Continued

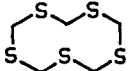
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{10}\text{S}_5^+$ |  | (7.6) | (197) | (823) | 22 | 90 | *EST | 2372-99-8 |
| IP is onset of photoelectron band (81BOC/SCH). | | | | | | | | |
| $\text{C}_5\text{H}_{11}^+$ | | | | | | | | |
| $1\text{-C}_5\text{H}_{11}$ | | (7.85) | (194) | (812) | 13 | 56 | *EST | 2672-01-7 |
| $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 100.5$ kcal/mol. IP estimated by J.L. Holmes, personal communication. | | | | | | | | |
| $\text{CH}_3\text{CH}_2\text{CH}_2\text{CHCH}_3$ | | (7.1) | (175) | (732) | 12 | 50 | *EST | 2492-34-4 |
| Cited ionization potential is difference between heats of formation of ion and neutral. $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 99$ kcal/mol. Experimental value = 7.41 eV. | | | | | | | | |
| $(\text{CH}_3)_2\text{CCH}_2\text{CH}_3$ | | 6.6 | 158±1 | 661±4 | 6.5 | 27 | *EST | 4348-35-0 |
| $\Delta_f H(\text{Ion})$ from hydride transfer equilibrium constant determinations (75SOL/FIE, 76GOR/MUN). $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 95.5$ kcal/mol. IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. Experimental IP of radical = 6.65 eV (86KRU/BEA). | | | | | | | | |
| $(\text{CH}_3)_3\text{CCH}_2$ | | 7.88±0.05 | (190) | (795) | 8 | 33 | *EST | 3744-21-6 |
| IP from 84SCH/HOU. $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 100.5$ kcal/mol. | | | | | | | | |
| $\text{C}_5\text{H}_{11}\text{Br}^+$ | | | | | | | | |
| $n\text{-C}_5\text{H}_{11}\text{Br}$ | | 10.09±0.02 | 202 | 844 | -30.8±0.3 | -129.1±1.4 | 77PED/RYL | 110-53-2 |
| $(\text{CH}_3)_3\text{CCH}_2\text{Br}$ | | 10.04 | 196 | 822 | -35 | -147 | 81HOL/FIN | 630-17-1 |
| $\text{C}_5\text{H}_{11}\text{ClHg}^+$ | | | | | | | | |
| $n\text{-C}_5\text{H}_{11}\text{HgCl}$ | | ≤9.99 | (≤200) | (≤835) | -31 | -129 | *EST | 544-15-0 |
| IP from 81BAI/CHI2. | | | | | | | | |
| $\text{iso-C}_5\text{H}_{11}\text{HgCl}$ | | ≤9.95 | (≤197) | (≤823) | -33 | -137 | *EST | 17774-08-2 |
| IP from 81BAI/CHI2. | | | | | | | | |
| $\text{C}_5\text{H}_{11}\text{I}^+$ | | | | | | | | |
| $n\text{-C}_5\text{H}_{11}\text{I}$ | | 9.201 | (195) | (816) | -17 | -72 | *EST | 628-17-1 |
| $(\text{CH}_3)_2\text{C}(\text{C}_2\text{H}_5)\text{I}$ | | (8.93) | (184) | (769) | -22 | -93 | *EST | 594-38-7 |
| $\text{CH}_2\text{ICH}_2\text{CH}(\text{CH}_3)_2$ | | 9.192 | (193) | (807) | -19 | -80 | *EST | 541-28-6 |
| $\text{C}_5\text{H}_{11}\text{N}^+$ | | | | | | | | |
| $\text{C}_2\text{H}_5\text{CH}=\text{NC}_2\text{H}_5$ | | (8.7) | (201) | (839) | 0 | 0 | 69BEN/CRU | 18328-91-1 |
| IP is onset of photoelectron band. | | | | | | | | |
| $(\text{CH}_3)_2\text{C}=\text{NC}_2\text{H}_5$ | | (8.83) | (195) | (816) | -9±2 | -36±9 | *EST | 15673-04-8 |
| See also: 79AUE/BOW. | | | | | | | | |

Table 1. Positive Ion Table - Continued

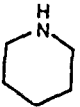
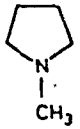
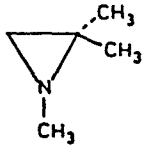
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{11}\text{N}^+$ $(\text{CH}_3)_2\text{NCH}_2\text{CH}=\text{CH}_2$ | 7.84 See also: 81LOG/TAK. | 195 | 813 | 14 | 57 | 70BEN/O'N | 2155-94-4 |
|  | 8.05±0.05 See also: 82ROZ/HOU. | 174 | 728 | -11.7±0.4 | -48.9±1.5 | 77PED/RYL | 110-89-4 |
|  | (≤8.41±0.02) $\Delta_f H(\text{Ion})$ from hydrogen affinities of homologues, 178kcal/mol; 746 kJ/mol; corresponding IP, 7.8 eV. | (≤193) | (≤809) | -0.5±0.5 | -2±2 | *EST | 120-94-5 |
|  | (≤8.68±0.02) | (≤201) | (≤842) | 1 | 5 | *EST | 23132-47-0 |
| $\text{C}_5\text{H}_{11}\text{NO}^+$ $(\text{CH}_3)_2\text{NCH}_2\text{COCH}_3$ | (7.71) IP from 81LOG/TAK. See also: 84OLI/GUE. | (135) | (567) | -42 | -177 | 81LOG/TAK | 15364-56-4 |
| $\text{C}_5\text{H}_{11}\text{NO}_2^+$ $\text{H}_2\text{N}(\text{CH}_2)_4\text{COOH}$ | (≤9.4) | (≤107) | (≤447) | -110±.7 | -460±3 | 83SKO/SAB | 660-88-8 |
| $n\text{-C}_3\text{H}_7\text{CH}(\text{NH}_2)\text{COOH}$ | (8.53) | (87) | (364) | -110±2 | -459±10 | *EST | 6600-40-4 |
| $L\text{-iso-C}_3\text{H}_7\text{CH}(\text{NH}_2)\text{COOH}$ | (8.71) | (92) | (385) | -108.8±0.2 | -455.1±1.0 | 77PED/RYL | 72-18-4 |
| $(\text{CH}_3)_2\text{NCH}_2\text{COOCH}_3$ | (7.96) IP from 81LOG/TAK. | (98) | (411) | -85 | -357 | 81LOG/TAK | 7148-06-3 |
| $\text{C}_5\text{H}_{11}\text{NO}_2\text{S}^+$ $L\text{-CH}_3\text{SCH}_2\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$ | (8.3) IP is onset of photoelectron band (83CAN/HAM). | (92) | (387) | -99±1 | -414±4 | 81SAB/MIN | 59-51-8 |
| $\text{C}_5\text{H}_{11}\text{N}_2\text{O}_3^+$ $L\text{-H}_2\text{NCO}(\text{CH}_2)_2\text{CH}(\text{NH}_3)\text{COOH}$ | | 73 | 304 | | | | |
| | | From proton affinity of $L\text{-H}_2\text{NCO}(\text{CH}_2)_2\text{CH}(\text{NH}_2)\text{COOH}$ (RN 585-21-7). PA = 218.4 kcal/mol, 914. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

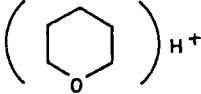
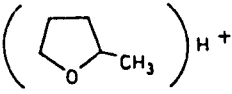
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---------------------------------------|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{11}\text{O}^+$ | | | | | | | | |
| | $i\text{-C}_3\text{H}_7\text{C}(\text{OH})\text{CH}_3$ | | 102 | 427 | | | | |
| | | | From proton affinity of $i\text{-C}_3\text{H}_7\text{COCH}_3$ (RN 563-80-4). PA = 201.1 kcal/mol, 841. kJ/mol. | | | | | |
| | $n\text{-C}_4\text{H}_9\text{CHOH}$ | | 118 | 493 | | | | |
| | | | From proton affinity of $n\text{-C}_4\text{H}_9\text{CHO}$ (RN 110-62-3). PA = 192.6 kcal/mol, 806. kJ/mol. | | | | | |
| | $(\text{CH}_3)_2\text{COC}_2\text{H}_5$ | | (104) | (435) | | | | |
| | | | From appearance potential determination (82MAC). | | | | | |
| | $(\text{C}_2\text{H}_5)_2\text{COH}$ | | 102 | 429 | | | | |
| | | | From proton affinity of $(\text{C}_2\text{H}_5)_2\text{CO}$ (RN 96-22-0). PA = 201.4 kcal/mol, 843. kJ/mol. | | | | | |
| | $\text{C}_2\text{H}_5\text{OCHCH}_2\text{CH}_3$ | | (114) | (476) | | | | |
| | | | From proton affinity of $\text{C}_2\text{H}_5\text{OCH}=\text{CHCH}_3$ (86BOU/DJA). PA = (210.) kcal/mol, (880.) kJ/mol. \leftarrow needs RN! check with khoda | | | | | |
| | $\text{C}_2\text{H}_5\text{O}(\text{H})\text{CH}_2\text{CHCH}_2$ | | (132) | (552) | | | | |
| | | | From proton affinity of $\text{C}_2\text{H}_5\text{OCH}_2\text{CH}=\text{CH}_2$. (PA = ?) (RN = ?) \leftarrow <u>RHODA</u> | | | | | |
| |  | | 113 | 472 | | | | |
| | | | From proton affinity of tetrahydropyran (RN 142-68-7). PA = 199.7 kcal/mol, 835.5 kJ/mol. | | | | | |
| |  | | 110 | 461 | | | | |
| | | | From proton affinity of 2-methyltetrahydrofuran (RN 96-47-9). PA = 203.6 kcal/mol, 852. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{11}\text{O}_2^+$ | | | | | | | | |
| | $\text{HC}(\text{OH})(\text{O}-n\text{-C}_4\text{H}_9)$ | | 68 | 285 | | | | |
| | | | From proton affinity of $\text{HCOO}(n\text{-C}_4\text{H}_9)$ (RN 592-84-7). PA = 194.8 kcal/mol, 815. kJ/mol. | | | | | |
| | $n\text{-C}_3\text{H}_7\text{C}(\text{OH})(\text{OCH}_3)$ | | 57 | 241 | | | | |
| | | | From proton affinity of $\text{C}_3\text{H}_7\text{COOCH}_3$ (RN 623-42-7). PA = 200.1 kcal/mol, 837. kJ/mol. | | | | | |
| | $\text{CH}_3\text{C}(\text{OH})(\text{O}-\text{C}_3\text{H}_7)$ | | 57 | 237 | | | | |
| | | | From proton affinity of $\text{CH}_3\text{COOC}_3\text{H}_7$ (RN 109-60-4). PA = 200.6 kcal/mol, 839. kJ/mol. | | | | | |
| | $i\text{-C}_3\text{H}_7\text{C}(\text{OH})\text{OCH}_3$ | | 55 | 231 | | | | |
| | | | From proton affinity of $i\text{-C}_3\text{H}_7\text{COOCH}_3$ (RN 547-63-7). PA = 201.6 kcal/mol, 843. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

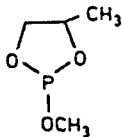
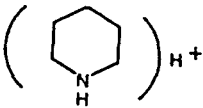
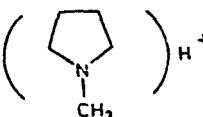
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{11}\text{O}_3\text{P}^+$  | (8.7) | (22) | (93) | -178 | -746 | *EST | 33892-95-4 |
| | | IP is onset of photoelectron band (81ARS/ZVE). | | | | | |
| $\text{C}_5\text{H}_{12}^+$ n- C_5H_{12} | 10.35±0.01 | 204 | 852 | -35.0±0.1 | -146.5±0.4 | 77PED/RYL | 109-66-0 |
| | | 211 | 884 | -27.3±0.1 | -114.2±0.4 | | |
| | See also: 81MAU/SIE, 82LIA, 81KIM/KAT. | | | | | | |
| iso- C_5H_{12} | ≤10.22 | ≤199 | ≤832 | -36.7±0.1 | -153.8±0.5 | 77PED/RYL | 78-78-4 |
| | | ≤207 | ≤867 | -28.4 | -118.8 | | |
| | IP from 81TRA. See also: 81KIM/KAT. | | | | | | |
| neo- C_5H_{12} | ≤10.21±0.04 | ≤195 | ≤818 | -40.0±0.1 | -167.4±0.7 | 77PED/RYL | 463-82-1 |
| | | ≤203 | ≤850 | -32.4 | -135.6 | | |
| | See also: 81KIM/KAT. | | | | | | |
| $\text{C}_5\text{H}_{12}\text{Cl}^+$ (CH_3) ₃ CClCH ₃ | | (137) | (572) | | | | |
| | From equilibrium constant determination (8SSHA/HOJ). | | | | | | |
| $\text{C}_5\text{H}_{12}\text{N}^+$ (CH_3) ₂ CNHC ₂ H ₅ | | (128) | (534) | | | | |
| | From proton affinity of (CH_3) ₂ C=NC ₂ H ₅ (RN 15673-04-8). PA = (229.5) kcal/mol, (960.) kJ/mol. | | | | | | |
| CH ₃ CH ₂ CHN(CH ₃) ₂ | | 142 | 596 | | | | |
| | From proton affinity of CH ₃ CH=CHN(CH ₃) ₂ (RN 6163-56-0). PA = 229.4 kcal/mol, 960. kJ/mol. | | | | | | |
|  | | 128 | 535 | | | | |
| | From proton affinity of piperidine (RN 110-89-4). PA = 226.4 kcal/mol, 947. kJ/mol. | | | | | | |
|  | | 136 | 571 | | | | |
| | From proton affinity of N-methylpyrrolidine (RN 120-94-5). PA = 228.7 kcal/mol, 957. kJ/mol. | | | | | | |
| $\text{C}_5\text{H}_{12}\text{NO}_2^+$ (CH_3) ₂ NC(OH)OC ₂ H ₅ | | 43 | 180 | | | | |
| | From proton affinity of (CH_3) ₂ NCOOC ₂ H ₅ (RN 687-48-9). PA = 213.7 kcal/mol, 894. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

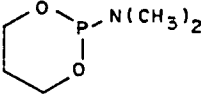
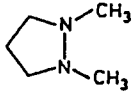
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|---|---|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{12}\text{NO}_2^+$ | L-i-C ₃ H ₇ CH(NH ₃)COOH | | 40 | 167 | | | | |
| | | | From proton affinity of L-i-C ₃ H ₇ CH(NH ₂)COOH (RN 72-18-4). PA = 217.0 kcal/mol, 908. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{12}\text{NO}_2\text{P}^+$ |  | 7.8 | (50) | (208) | -130 | -545 | *EST | 17454-25-0 |
| | | IP from 81ARS/ZVE. | | | | | | |
| $\text{C}_5\text{H}_{12}\text{NO}_2\text{S}^+$ | L-CH ₃ SCH ₂ CH ₂ CH(NH ₃)COOH | | 45 | 190 | | | | |
| | | From proton affinity of L-CH ₃ SCH ₂ CH ₂ CH(NH ₂)COOH (RN 59-51-8). PA = 221.4 kcal/mol, 926. kJ/mol. | | | | | | |
| $\text{C}_5\text{H}_{12}\text{N}_2^+$ |  | 6.66 | (180) | (754) | 27 | 111 | *EST | 38704-89-1 |
| | | IP from charge transfer equilibrium constant determination (86RUM). Reference standard: IP (C ₆ H ₅ N(CH ₃) ₂) = 7.12 eV. See also: 84NEL. | | | | | | |
| $\text{C}_5\text{H}_{12}\text{N}_2\text{O}^+$ | ((CH ₃) ₂ N) ₂ CO | ≤8.64 | (≤142) | (≤595) | -57 | -238 | *EST | 632-22-4 |
| $\text{C}_5\text{H}_{12}\text{N}_2\text{S}^+$ | ((CH ₃) ₂ N) ₂ CS | (7.5) | (184) | (769) | 11±0.5 | 45±2 | 82INA/MUR2 | 2782-91-4 |
| | | IP is onset of photoelectron band. See also: 85ROT/BOC. | | | | | | |
| $\text{C}_5\text{H}_{12}\text{O}^+$ | n-C ₅ H ₁₁ OH | 10.00±0.03 | 160 | 668 | -70.9±0.4 | -296.7±1.6 | 77PED/RYL | 71-41-0 |
| | | IP from 77ASH/BUR. See also: 80BAC/MOU. | | | | | | |
| | CH ₃ CH ₂ CH(CH ₃)CH ₂ OH | (9.86) | (155) | (649) | -72.2±0.3 | -302.0±1.4 | 77PED/RYL | 137-32-6 |
| | | IP from 81HOL/FIN. | | | | | | |
| | n-C ₃ H ₇ CH(OH)CH ₃ | (9.78±0.03) | (151) | (630) | -75.0±0.2 | -313.8±0.8 | 77PED/RYL | 6032-29-7 |
| | | IP from 77ASH/BUR, 84BOW/MAC. | | | | | | |
| | (C ₂ H ₅) ₂ CHOH | 9.78 | 150 | 628 | -75.4±0.2 | -315.5±0.9 | 77PED/RYL | 584-02-1 |
| | | IP from 81HOL/FIN, 84BOW/MAC. See also: 77ASH/BUR. | | | | | | |
| | (CH ₃) ₂ CHCH(OH)CH ₃ | (10.01) | (155) | (650) | -75.4±0.3 | -315.7±1.1 | 77PED/RYL | 598-75-4 |
| | | IP from 84BOW/MAC. | | | | | | |

Table 1. Positive Ion Table - Continued

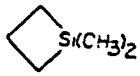
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₅H₁₂O⁺ | | | | | | | |
| C ₂ H ₅ C(CH ₃) ₂ OH | 9.80 IP from 84BOW/MAC, 82LEV/LIA. | 147 | 615 | -79.1±0.3 | -330.8±1.3 | 77PED/RYL | 75-85-4 |
| n-C ₄ H ₉ OCH ₃ | (9.54) IP from 81HOL/FIN. | (158) | (662) | -61.7±0.3 | -258.1±1.1 | 77PED/RYL | 628-28-4 |
| n-C ₃ H ₇ OC ₂ H ₅ | (9.45±0.1) IP cited in 81HOL/FIN. | (153) | (640) | -65.0±0.2 | -272.2±1 | 77PED/RYL | 628-32-0 |
| tert-C ₄ H ₉ OCH ₃ | (9.24) IP from 84BOW/MAC. See also: 80BAC/MOU. | (145) | (608) | -67.8±0.2 | -283.6±1 | 77PED/RYL | 1634-04-4 |
| C₅H₁₂O₂⁺ | | | | | | | |
| n-C ₃ H ₇ CH(CH ₃)OOH | (9.35±0.03) IP from 77ASH/BUR. | (159) | (666) | -56 | -236 | *EST | 14018-58-7 |
| CH ₃ O(CH ₂) ₃ OCH ₃ | (9.3) IP is onset of photoelectron band (83BIE/MOR). | (126) | (526) | -89 | -371 | *EST | 17081-21-9 |
| n-C ₅ H ₁₁ OOH | (9.50±0.03) IP from 77ASH/BUR. | (167) | (698) | -52 | -219 | *EST | 74-80-6 |
| C₅H₁₂O₃⁺ | | | | | | | |
| CH ₃ C(OCH ₃) ₃ | (9.65) IP from 82HOL/LOS2. | (82) | (343) | -140±0.5 | -588±2 | 77PED/RYL | 1445-45-0 |
| C₅H₁₂S⁺ | | | | | | | |
| n-C ₃ H ₇ SC ₂ H ₅ | (8.50±0.05) | (171) | (715) | -25.0±0.2 | -104.7±0.7 | 77PED/RYL | 4110-50-3 |
| (CH ₃) ₃ CSCH ₃ | (8.38±0.05) | (164) | (687) | -29.0±0.2 | -121.3±0.7 | 77PED/RYL | 6163-64-0 |
| C ₂ H ₅ S(iso-C ₃ H ₇) | (8.35±0.01) | (165) | (689) | -28±0.6 | -117±2 | 77PED/RYL | 5145-99-3 |
| C₅H₁₂S₂⁺ | | | | | | | |
| C ₂ H ₅ SCH ₂ SC ₂ H ₅ | (8.22±0.02) | (179) | (750) | -10 | -43 | *EST | 4396-19-4 |
| C₅H₁₂Si⁺ | | | | | | | |
| CH ₂ =CHSi(CH ₃) ₃ | (9.5) IP is onset of photoelectron band (81KHV/ZYK, 82LEV/LIA). | (190) | (794) | -29 | -123 | *EST | 754-05-2 |
|  S(CH ₃) ₂ | 8.83±0.07 See also: 82DYK/JOS, 81KOE/MCK, 81GUS/VOL2. | 184 | 769 | -19.8 | -82.8 | 81GUS/VOL2 | 2295-12-7 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|-----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{12}\text{Sn}^+$ $\text{CH}_2 = \text{CHSn}(\text{CH}_3)_3$ | (≤ 9.7) | (≤ 246) | (≤ 1028) | 22 \pm 3 | 92 \pm 13 | 77PED/RYL | 754-06-3 |
| $\text{C}_5\text{H}_{13}\text{N}^+$ n- $\text{C}_5\text{H}_{11}\text{NH}_2$ | (8.67) IP from 79AUE/BOW. | (174) | (726) | -26 | -110 | *EST | 110-58-7 |
| tert- $\text{C}_5\text{H}_{11}\text{NH}_2$ | (8.46 \pm 0.1) | (165) | (689) | -30 | -127 | *EST | 594-39-8 |
| neo- $\text{C}_5\text{H}_{11}\text{NH}_2$ | (8.54 \pm 0.1) | (166) | (692) | -31 | -132 | *EST | 5813-64-9 |
| $(\text{C}_2\text{H}_5)_2(\text{CH}_3)\text{N}$ | (7.50 \pm 0.1) IP from 79AUE/BOW. | (156) | (654) | -17 | -70 | *EST | 616-39-7 |
| $(\text{CH}_3)_2(i\text{-C}_3\text{H}_7)\text{N}$ | (7.3) $\Delta_f H(\text{Ion})$ from hydrogen affinities of homologous series. IP cited is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See also: 84NEL. | (150) | (628) | -18 | -76 | *EST | 996-35-0 |
| $\text{C}_5\text{H}_{13}\text{N}_2\text{O}^+$ [(CH_3) $_2\text{N}$] $_2\text{COH}$ | | 88 | 367 | | | | |
| | | From proton affinity of [(CH_3) $_2\text{N}$] $_2\text{C}=\text{O}$ (RN 632-22-4) (86TAF/GAL). PA = 221.1 kcal/mol, 925. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{13}\text{O}^+$ neo- $\text{C}_5\text{H}_{11}\text{OH}_2$ | | 97 | 406 | | | | |
| | | From proton affinity of neo- $\text{C}_5\text{H}_{11}\text{OH}$ (RN 75-84-3) (78TAF/TAA). PA = 193.6 kcal/mol, 810. kJ/mol. | | | | | |
| $\text{C}_2\text{H}_5\text{OH}(i\text{-C}_3\text{H}_7)$ | | 94 | 393 | | | | |
| | | From proton affinity of $\text{C}_2\text{H}_5\text{O}(i\text{-C}_3\text{H}_7)$ (RN 625-54-7). PA = 203.5 kcal/mol, 851. kJ/mol. | | | | | |
| t- $\text{C}_4\text{H}_9\text{OHCH}_3$ | | 96 | 400 | | | | |
| | | From proton affinity of t- $\text{C}_4\text{H}_9\text{OCH}_3$ (RN 1634-04-4). PA = 202.2 kcal/mol, 846. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{14}\text{N}^+$ n- $\text{C}_5\text{H}_{11}\text{NH}_3$ | | 121 | 504 | | | | |
| | | From proton affinity of n- $\text{C}_5\text{H}_{11}\text{NH}_2$ (RN 110-58-7). PA = 218.9 kcal/mol, 916. kJ/mol. | | | | | |
| tert- $\text{C}_5\text{H}_{11}\text{NH}_3$ | | 112 | 468 | | | | |
| | | From proton affinity of tert- $\text{C}_5\text{H}_{11}\text{NH}_2$ (RN 594-39-8). PA = 222.3 kcal/mol, 930. kJ/mol. | | | | | |
| neo- $\text{C}_5\text{H}_{11}\text{NH}_3$ | | 115 | 481 | | | | |
| | | From proton affinity of neo- $\text{C}_5\text{H}_{11}\text{NH}_2$ (RN 5813-64-9). PA = 219.3 kcal/mol, 917.5 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

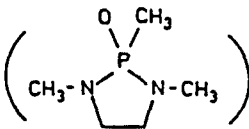
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{14}\text{N}^+$ $(\text{C}_2\text{H}_5)(i\text{-C}_3\text{H}_7)\text{NH}_2$ | | 113 | 474 | | | | |
| | | From proton affinity of $(\text{C}_2\text{H}_5)(i\text{-C}_3\text{H}_7)\text{NH}$ (RN 19961-27-4). PA = 227.4 kcal/mol, 951. kJ/mol. | | | | | |
| $(\text{CH}_3)(\text{C}_2\text{H}_5)_2\text{NH}$ | | 119 | 498 | | | | |
| | | From proton affinity of $(\text{CH}_3)(\text{C}_2\text{H}_5)_2\text{N}$ (RN 616-39-7). PA = 230.0 kcal/mol, 962. kJ/mol. | | | | | |
| $(\text{CH}_3)_2(i\text{-C}_3\text{H}_7)\text{NH}$ | | 118 | 493 | | | | |
| | | From proton affinity of $(\text{CH}_3)_2(i\text{-C}_3\text{H}_7)\text{N}$ (RN 996-35-0). PA = 229.8 kcal/mol, 961. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{14}\text{N}_2^+$ $(\text{C}_2\text{H}_5)(\text{CH}_3)\text{NN}(\text{CH}_3)_2$ | | (8.18) | (201) | (839) | 12 | 50 | *EST 50599-41-2 |
| | | Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL. | | | | | |
| $((\text{CH}_3)_2\text{N})_2\text{CH}_2$ | (7.74±0.05) | (174) | (729) | -4.2±0.3 | -17.6±1.4 | 77PED/RYL | 51-80-9 |
| | | See also: 81LOG/TAK. | | | | | |
| $\text{C}_5\text{H}_{14}\text{N}_2\text{OP}^+$  H^+ | | 42 | 176 | | | | |
| | | From proton affinity of 2,5-dimethyl-1,3,2-diazaphospholidine-2-oxide (RN 16606-18-1) (84MAU/NEL). PA = 224.8 kcal/mol, 941. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{14}\text{N}_3^+$ $[(\text{CH}_3)_2\text{N}]_2\text{CNH}_2$ | | 192 | 803 | | | | |
| | | From proton affinity of $[(\text{CH}_3)_2\text{N}]_2\text{C}=\text{NH}$ (RN 31081-16-0) (86TAF/GAL). PA = 241.0 kcal/mol, 1008. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{14}\text{Si}^+$ $(\text{CH}_3)_3\text{SiC}_2\text{H}_5$ | (9.6) | (164) | (685) | -58 | -241 | *EST | 3439-38-1 |
| $\text{C}_5\text{H}_{14}\text{Sn}^+$ $(\text{CH}_3)_3\text{SnC}_2\text{H}_5$ | (8.6) | (191) | (800) | -7±0.7 | -30±3 | 77PED/RYL | 3531-44-0 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_5\text{H}_{15}\text{N}_2^+$ $\text{NH}_2(\text{CH}_2)_5\text{NH}_3$ | | 110 | 461 | | | | |
| | | From proton affinity of $\text{NH}_2(\text{CH}_2)_5\text{NH}_2$ (RN 462-94-2). PA = 238.1 kcal/mol, 996. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

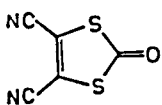
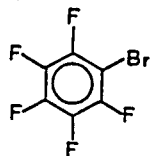
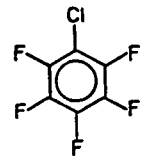
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_5\text{H}_{15}\text{N}_2^+$ (CH_3) ₂ NH(CH ₂) ₃ NH ₂ | | 118 | 494 | | | | |
| | | From proton affinity of (CH_3) ₂ N(CH ₂) ₃ NH ₂ (RN 109-55-7). Data re-evaluated. PA = 241. kcal/mol, 1006. kJ/mol. | | | | | |
| $\text{C}_5\text{H}_{15}\text{Si}_2^+$ (CH_3) ₃ SiSi(CH ₃) ₂ | | 117 | 489 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (84SZE/BAE, 84SZE/BAE2). 0 K values. | | | | | |
| $\text{C}_5\text{H}_{15}\text{Ta}^+$ Ta(CH ₃) ₅ | 8.25 | 241 | 1007 | 51±6 | 212±26 | 82PIL/SKI | 53378-72-6 |
| | IP is onset of photoelectron band (75GAL/WIL, 82LEV/LIA). | | | | | | |
| $\text{C}_5\text{H}_{16}\text{NSi}^+$ (CH_3) ₃ SiNH(CH ₃) ₂ | | (81) | (336) | | | | |
| | | From proton affinity of (CH_3) ₃ SiN(CH ₃) ₂ (RN 18135-05-2). PA = (226) kcal/mol, (946) kJ/mol. | | | | | |
| $\text{C}_5\text{IMnO}_5^+$ Mn(CO) ₅ I | (8.1) | (-13) | (-52) | -199±1 | -834±5 | 82CON/ZAF | 14879-42-6 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_5\text{N}_2\text{OS}_2^+$ | | | | | | | |
|  | ≤9.94 | (≤294) | (≤1229) | 65 | 270 | *EST | 934-31-6 |
| | IP from 83SCH/SCH. | | | | | | |
| C_5N_4^+ C(CN) ₄ | (13.94) | (482) | (2018) | 161±2 | 673±9 | 82CHU/NGU | 24331-09-7 |
| C_6BrF_5^+ | | | | | | | |
|  | 9.57±0.02 | 51 | 211 | -170±4 | -712±17 | 77KRE/PRI | 344-04-7 |
| C_6ClF_5^+ | | | | | | | |
|  | (9.72±0.02) | (30) | (128) | -194±3 | -810±11 | 77PED/RYL | 344-07-0 |

Table 1. Positive Ion Table - Continued

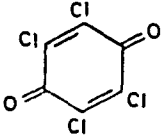
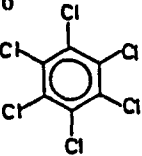
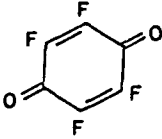
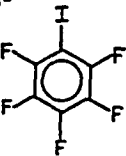
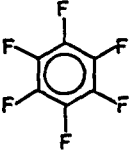
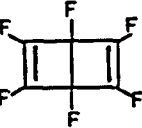
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{Cl}_4\text{O}_2^+$  | 9.74 IP from 81SAT/SEK. | 180 | 754 | -44.4±2.8 | -185.7±11 | 77PED/RYL | 118-75-2 |
| C_6Cl_6^+  | 8.98 IP from 81SAT/SEK. See also: 81RUS/KLA, 81KIM/KAT. | 196.4 | 821.7 | -10.7 | -44.7 | 83PLA/SIM | 118-74-1 |
| C_6CrO_6^+ $\text{Cr}(\text{CO})_6$ | 8.142±0.017 See also: 82HUB/LIC, 85DAS/NIS. | -29 | -122 | -217.0±0.3 | -908±1.2 | 77PED/RYL | 13007-92-6 |
| $\text{C}_6\text{F}_3\text{MnO}_5^+$ $\text{CF}_3\text{Mn}(\text{CO})_5$ | 8.8 IP is onset of photoelectron band. | -128 | -537 | -331±1 | -1386±4 | 82CON/ZAF | 13601-14-4 |
| $\text{C}_6\text{F}_4\text{O}_2^+$  | (10.7) IP is onset of photoelectron band. | (52) | (216) | -195.0±9.9 | -816±41 | *EST | 527-21-9 |
| $\text{C}_6\text{F}_5\text{I}^+$  | 9.54 See also: 81BIE/ASB. | 87 | 362 | -133±3 | -558±13 | 77PED/RYL | 827-15-6 |
| C_6F_6^+ $\text{CF}_3\text{C}\equiv\text{CC}\equiv\text{CCF}_3$ | (10.99±0.01) | (78) | (326) | -175 | -734 | 77PRA/HUB | 10524-09-1 |
|  | 9.906 IP from 81BIE/ASB. A value of 9.91 eV is assigned to the ionization potential at 300 K based on determinations of charge transfer equilibrium constants (reference standard, ionization potential of $\text{C}_6\text{H}_5\text{CF}_3 = 9.685$ eV). See also: 81MAI/THO. | 2 | 10 | -226±2 | -946±8 | 79PRI/SAP | 392-56-3 |
|  | 10.08±0.05 | 62 | 260 | -170 | -713 | 77PRA/HUB | 6733-01-3 |

Table 1. Positive Ion Table - Continued

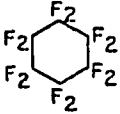
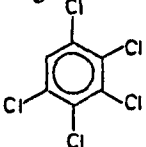
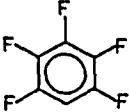
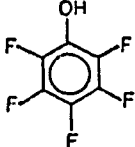
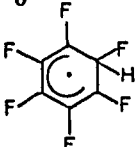
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{F}_{12}^+$  | (13.2) | (-262) | (-1095) | -566±2 | -2369±8 | 79PRI/SAP | 355-68-0 |
| C_6HCl_5^+  | (8.9) IP is onset of photoelectron band (81RUS/KLA3). | (195.7) | (818.7) | -9.6±2.1 | -40.0±8.7 | 85PLA/SIM2 | 608-93-5 |
| $\text{C}_6\text{HCrO}_6^+$ $\text{HCr}(\text{CO})_6$ | | (-31) | (-131) | | | | |
| | | From proton affinity of $\text{Cr}(\text{CO})_6$ (RN 13007-92-6). PA = (180) kcal/mol, (753) kJ/mol. | | | | | |
| C_6HF_5^+  | 9.63 | 29 | 123 | -193±2 | -806±7 | 77PED/RYL | 363-72-4 |
| | Ionization potential from charge transfer equilibrium constant determinations (standard: $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV)(78LIA/AUS). Value of 9.64 eV reported from photoelectron spectroscopy measurement. See also: 81BIE/ASB. | | | | | | |
| $\text{C}_6\text{HF}_5\text{O}^+$  | 9.20±0.02 | -17 | -69 | -229±0.5 | -957±2 | 77PED/RYL | 771-61-9 |
| C_6HF_6^+  | | -38 | -159 | | | | |
| | From proton affinity of C_6F_6 (RN 392-56-3). PA = 177.7 kcal/mol, 743. kJ/mol. | | | | | | |
| $\text{C}_6\text{HMoO}_6^+$ $\text{HMo}(\text{CO})_6$ | | -38 | -160 | | | | |
| | From proton affinity of $\text{Mo}(\text{CO})_6$ (RN 13939-06-5). PA = (185) kcal/mol, (774) kJ/mol. | | | | | | |
| $\text{C}_6\text{HO}_6\text{V}^+$ $\text{HV}(\text{CO})_6$ | | (-33) | (-138) | | | | |
| | From proton affinity of $\text{V}(\text{CO})_6$ (RN 20644-87-5). PA = (194.5) kcal/mol, (814) kJ/mol. | | | | | | |
| $\text{C}_6\text{HO}_6\text{W}^+$ $\text{HW}(\text{CO})_6$ | | -30 | -127 | | | | |
| | From proton affinity of $\text{W}(\text{CO})_6$ (RN 14040-11-0). PA = (184) kcal/mol, (770) kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

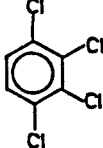
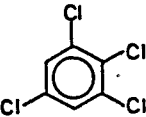
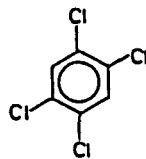
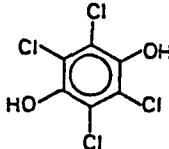
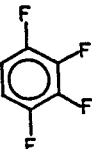
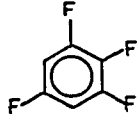
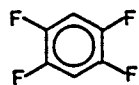
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6H_2^+ <chem>HC#CC#CC#CH</chem> | (9.50) | (375) | (1569) | 155 | 652 | *EST | 3161-99-7 |
| $\text{C}_6\text{H}_2\text{Cl}_4^+$  | (8.9) IP is onset of photoelectron band (81RUS/KLA3). | (199.1) | (833.3) | -6.1 | -25.4 | 85PLA/SIM | 634-66-2 |
|  | (9.0) IP is onset of photoelectron band (81RUS/KLA3). | (199) | (833) | -8.3 | -34.9 | 85PLA/SIM | 634-90-2 |
|  | 8.9 IP is onset of photoelectron band (81RUS/KLA3, 81KIM/KAT). | 197.4 | 826.1 | -7.8 | -32.6 | 83PLA/SIM | 95-94-3 |
| $\text{C}_6\text{H}_2\text{Cl}_4\text{O}_2^+$  | (8.30±0.05) | (104) | (437) | -87 | -364 | 77PED/RYL | 87-87-6 |
| $\text{C}_6\text{H}_2\text{F}_4^+$  | 9.53±0.01 Ionization potential from charge transfer equilibrium constant determinations (standard: IP of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV)(78LIA/AUS). Value of ionization potential from photoelectron spectroscopy, 9.56 eV. See also: 81BIE/ASB. | (68) | (281) | -152±0.2 | -638±1 | *EST | 551-62-2 |
|  | 9.53±0.01 Ionization potential from charge transfer equilibrium constant determinations (standard: ionization potential of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV)(78LIA/AUS). Value of ionization potential from photoionization, 9.55 eV; from photoelectron spectroscopy, 9.56 eV. See also: 81BIE/ASB. | (63) | (262) | -157±0.2 | -657±1 | *EST | 2367-82-0 |
|  | 9.35±0.01 Ionization potential from charge transfer equilibrium constant determinations (standard: ionization potential of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV)(78LIA/AUS). Value of IP from photoelectron spectroscopy, 9.36 eV (82LEV/LIA, 81BIE/ASB). | 61 | 255 | -155±1 | -647±3 | 78HAR/HEA | 327-54-8 |

Table 1. Positive Ion Table - Continued

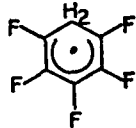
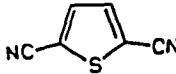
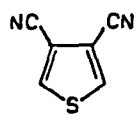
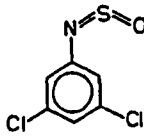
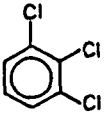
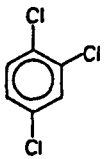
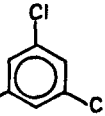
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_2\text{F}_5^+$  | | | -7 | -29 | | | |
| | | From proton affinity of C_6HF_5 (RN 363-72-4). PA = 179.9 kcal/mol, 753. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_2\text{N}_2\text{S}^+$  | ≤ 9.76 IP from 83BOC/ROT. | (≤ 319) | (≤ 1337) | 94 | 395 | *EST | 18853-40-2 |
|  | (≤ 10.20) IP from 83BOC/ROT. | (≤ 331) | (≤ 1384) | 96 | 400 | *EST | 18853-32-2 |
| $\text{C}_6\text{H}_3\text{Cl}_2\text{NOS}^+$  | (≤ 9.46) IP from 82LOU/VAN. | (≤ 197) | (≤ 826) | -21 | -87 | *EST | |
| $\text{C}_6\text{H}_3\text{Cl}_3^+$  | 9.18 IP from 81RUS/KLA3. | 209.8 | 877.6 | -1.9 | -8.1 | 85PLA/SIM | 87-61-6 |
|  | 9.04 IP from 81RUS/KLA3 | 210 | 880 | 1.9 | 8.1 | 85PLA/SIM | 120-82-1 |
|  | 9.32±0.02 IP from 81RUS/KLA(3), onset of photoelectron band (81KIM/KAT). See also: 82MAI/THO2. | 215 | 899 | 0 | 0 | 82SHA | 108-70-3 |

Table 1. Positive Ion Table - Continued

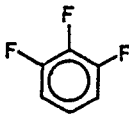
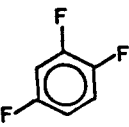
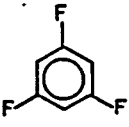
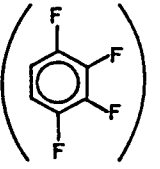
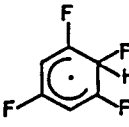
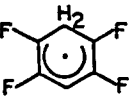
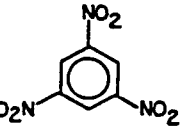
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_3\text{F}_3^+$  | (9.7) IP from 81BIE/ASB. | (107) | (448) | -117 | -488 | *EST | 1489-53-8 |
|  | 9.30±0.05 IP from charge transfer equilibrium constant determinations (standard: IP of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV) (78LIA/AUS) and from photoelectron spectroscopy (81BIE/ASB, 77ROS/DRA). | (96) | (401) | -119 | -496 | *EST | 367-23-7 |
|  | 9.64 See also: 81BIE/ASB. | (100) | (418) | -122±0.7 | -512±3 | *EST | 372-38-3 |
| $\text{C}_6\text{H}_3\text{F}_4^+$  H^+ | | 32 | 134 | From proton affinity of 1,2,3,4- $\text{C}_6\text{F}_4\text{H}_2$ (RN 551-62-2). PA = 181.1 kcal/mol, 758. kJ/mol. | | | |
|  | | 28 | 117 | From proton affinity of 1,2,3,5- $\text{C}_6\text{F}_4\text{H}_2$ (RN 2367-82-0). PA = 180.6 kcal/mol, 756. kJ/mol. | | | |
|  | | 31 | 131 | From proton affinity of 1,2,4,5- $\text{C}_6\text{F}_4\text{H}_2$ (RN 327-54-8). PA = 179.7 kcal/mol, 752. kJ/mol. | | | |
| $\text{C}_6\text{H}_3\text{MnO}_5^+$ $\text{CH}_3(\text{CO})_5\text{Mn}$ | (8.4) IP is onset of photoelectron band. | (14) | (57) | -180±1 | -753±4 | 82CON/ZAF | 13601-24-6 |
| $\text{C}_6\text{H}_3\text{N}_3\text{O}_6^+$  | (10.96±0.02) | (268) | (1119) | 15±0.5 | 62±2 | 77PED/RYL | 99-35-4 |

Table 1. Positive Ion Table - Continued


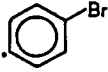
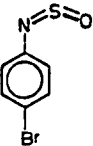
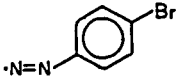
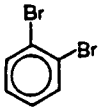
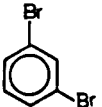
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_3\text{O}_5\text{Re}^+$ (CO) ₅ CH ₃ Re | 8.5 IP is onset of photoelectron band. | 14 | 60 | -182±1 | -760±6 | 82PIL/SKI | 14524-92-6 |
| C_6H_4^+ (Z)-HC≡CCH = CHC≡CH | (9.10±0.02) | (333) | (1394) | 123 | 516 | *EST | 16668-67-0 |
| (E)-HC≡CCH = CHC≡CH | (9.07±0.02) | (334) | (1400) | 125 | 525 | *EST | 16668-68-1 |
|  | 8.6 | 313 316 | 1311 1321 | 115 118 | 481 492 | 80POL/HEH | 462-80-6 |
| $\Delta_f H(\text{Ion})$ from 80ROS/STO2. Cited IP = $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$; a value of 8.95 eV has been estimated (80ROS/STO2). See also: 82ROS/DAN, 85DEW/TIE. | | | | | | | |
| $\text{C}_6\text{H}_4\text{Br}^+$  | 9.04 $\Delta_f H(\text{Ion})$ from 77NUY/MES. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | 298 | 1247 | 89.6 | 374.9 | 77NUY/MES | 2973-43-5 |
| $\text{C}_6\text{H}_4\text{BrNOS}^+$  | (≤8.91) IP from 82LOU/VAN. | (≤203) | (≤851) | -2 | -9 | *EST | 26516-62-1 |
| $\text{C}_6\text{H}_4\text{BrN}_2^+$  | (8.18) $\Delta_f H(\text{Ion})$ from 77NUY/MES. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | (276) | (1155) | 87 | 366 | *EST | |
| $\text{C}_6\text{H}_4\text{Br}_2^+$  | 8.8 IP is onset of photoelectron band. | (234) | (981) | 31.5 | 132 | *EST | 583-53-9 |
|  | 8.85 IP is onset of photoelectron band. | 235 | 985 | 31 | 131 | 83DEW/HEA | 108-36-1 |

Table 1. Positive Ion Table - Continued

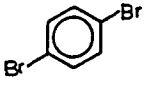
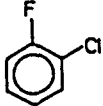
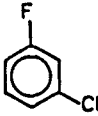
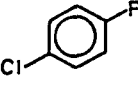
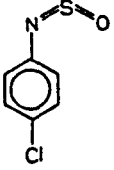
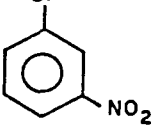
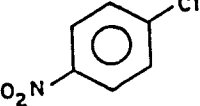
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_4\text{Br}_2^+$  | 8.7 IP is onset of photoelectron band. | (232) | (970) | 31 | 131 | *EST | 106-37-6 |
| $\text{C}_6\text{H}_4\text{ClF}^+$  | 9.18±0.01 See also: 78LIA/AUS. | (181) | (756) | -31 | -130 | *EST | 348-51-6 |
|  | 9.21±0.01 | (179) | (749) | -33 | -140 | *EST | 625-98-9 |
|  | 9.01±0.01 IP from 78LIA/AUS. | (174) | (728) | -34 | -141 | *EST | 352-33-0 |
| $\text{C}_6\text{H}_4\text{ClNOS}^+$  | (8.8) IP is onset of photoelectron band (82LOU/VAN). | (188) | (787) | (-15) | (-62) | ? | |
| $\text{C}_6\text{H}_4\text{ClNO}_2^+$  | (9.92±0.1) | (238) | (995) | 9.1±2.0 | 38.1±8.4 | *EST | 121-73-3 |
|  | 9.96±0.1 | (239) | (999) | 9.1±2.0 | 38.1±8.4 | *EST | 100-00-5 |

Table 1. Positive Ion Table - Continued

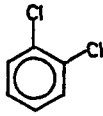
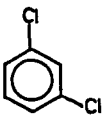
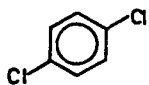
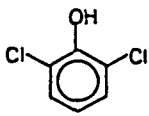
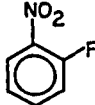
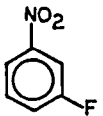
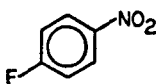
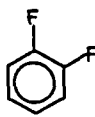
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_4\text{Cl}_2^+$  | 9.08±0.01 | 217.3 | 909.1 | 7.9 | 33.0 | 84PLA/SIM | 95-50-1 |
| | IP from charge transfer equilibrium constant determination (78LIA/AUS). See also: 81RUS/KLA2, 81KIM/KAT. | | | | | | |
|  | 9.11±0.01 | 216.8 | 907.1 | 6.7 | 28.1 | 84PLA/SIM | 541-73-1 |
| | IP from charge transfer equilibrium constant determination(78LIA/AUS). See also: 82LEV/LIA, 81RUS/KLA2, 81KIM/KAT. | | | | | | |
|  | 8.89±0.01 | 210.9 | 882.3 | 5.9 | 24.6 | 84PLA/SIM | 106-46-7 |
| | IP from 81RUS/KLA2 (onset of photoelectron band), and 78LIA/AUS (charge transfer equilibrium constant determination). See also: 81KIM/KAT. | | | | | | |
| $\text{C}_6\text{H}_4\text{Cl}_2\text{O}^+$  | (8.65±0.02) | (174) | (729) | -25 | -106 | 82SHA | 87-65-0 |
| $\text{C}_6\text{H}_4\text{FNO}_2^+$  | (≤9.86) | (≤199) | (≤833) | -28 | -118 | *EST | 1493-27-2 |
|  | 9.88 | (198) | (827) | -30 | -126 | *EST | 402-67-5 |
|  | 9.90 | (197) | (824) | -31 | -131 | *EST | 350-46-9 |
| $\text{C}_6\text{H}_4\text{F}_2^+$  | (9.28±0.01) | (144) | (602) | -70.2±0.2 | -293.8±1.0 | 77PED/RYL | 367-11-3 |
| | Ionization potential from charge transfer equilibrium constant determinations (standard: ionization potential of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV). Value of ionization potential from Rydberg series and from photoelectron spectroscopy (81BIE/ASB) = 9.30 eV. | | | | | | |

Table 1. Positive Ion Table - Continued

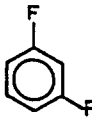
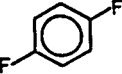
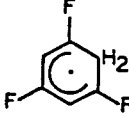
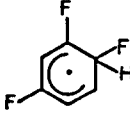
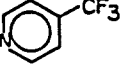

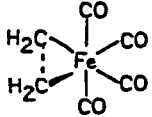
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_4\text{F}_2^+$  | 9.33±0.01 | 141 | 591 | -73.9±0.3 | -309.2±1.1 | 77PED/RYL | 372-18-9 |
| | Ionization potential from charge transfer equilibrium constant determinations (standard: ionization potential of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV). Value of ionization potential from Rydberg series, 9.35 eV, from photoelectron spectroscopy, 9.32 eV. See also: 81BIE/ASB. | | | | | | |
|  | 9.14±0.01 | 137 | 575 | -73.3±0.3 | -306.6±1.1 | 77PED/RYL | 540-36-3 |
| | Ionization potential from charge transfer equilibrium constant determinations (standard: ionization potential of $\text{C}_6\text{H}_5\text{CF}_3$, 9.685 eV). Value of ionization potential from Rydberg series, 9.18 eV, from photoionization and photoelectron spectroscopy, 9.14 eV (81BIE/ASB). | | | | | | |
| $\text{C}_6\text{H}_4\text{F}_3^+$  | | 62 | 261 | | | | |
| | From proton affinity of 1,3,5- $\text{C}_6\text{F}_3\text{H}_3$ (RN 372-38-3). PA = 181. kcal/mol, 757. kJ/mol. | | | | | | |
|  | | 69 | 289 | | | | |
| | From proton affinity of 1,2,4- $\text{C}_6\text{F}_3\text{H}_3$ (RN 367-23-7). PA = 181.4 kcal/mol, 759. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_4\text{F}_3\text{N}^+$  | (≤10.1) | (≤105) | (≤438) | -128 | -536 | *EST | 3796-24-5 |
| $\text{C}_6\text{H}_4\text{F}_3\text{NO}^+$  | (≤8.90) | (≤58) | (≤243) | -147 | -616 | *EST | |
| $\text{C}_6\text{H}_4\text{FeO}_4^+$  | (7.6) | (46) | (192) | -129±2 | -541±10 | 82PIL/SKI | 32799-25-0 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

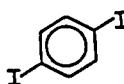
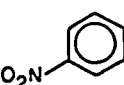
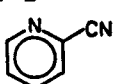
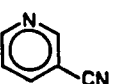
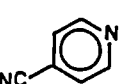
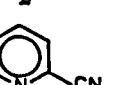
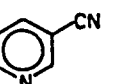
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_4\text{I}_2^+$  | 8.45 | (252) | (1056) | 58 | 241 | *EST | 624-38-4 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_6\text{H}_4\text{MnO}_5^+$ $\text{CH}_3\text{MnH}(\text{CO})_5$ | | 3 | 11 | | | | |
| From proton affinity of $\text{CH}_3\text{Mn}(\text{CO})_5$ (RN 13601-24-6). PA = 183 kcal/mol, 766 kJ/mol. | | | | | | | |
| $\text{C}_6\text{H}_4\text{NO}_2^+$  | (9.06) | (283) | (1183) | 74 | 309 | *EST | 2395-99-5 |
| IP from 77NUY/MES. | | | | | | | |
| $\text{C}_6\text{H}_4\text{N}_2^+$  | 10.12 | 301 | 1260 | 67 ± 0.5 | 281 ± 2 | 84BIC/PIL | 100-70-9 |
|  | (10.0) | (297) | (1243) | 66 ± 0.5 | 278 ± 2 | 84BIC/PIL | 100-54-9 |
| IP is onset of photoelectron band. | | | | | | | |
|  | (9.9) | (296) | (1239) | 68 ± 0.2 | 284 ± 1 | 84BIC/PIL | 100-48-1 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_6\text{H}_4\text{N}_2\text{O}^+$  | (8.96 ± 0.02) | (256) | (1069) | 49 | 204 | *EST | 2402-98-4 |
|  | $(\leq 8.93 \pm 0.02)$ | (≤ 254) | (≤ 1064) | 48 | 202 | *EST | 14906-64-0 |

Table 1. Positive Ion Table - Continued

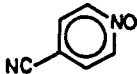
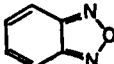
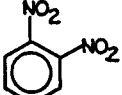
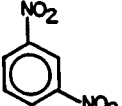
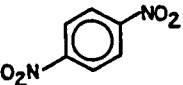
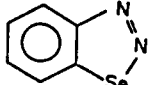
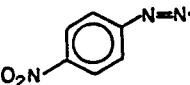
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_4\text{N}_2\text{O}^+$  | 8.95±0.02 | (255) | (1068) | 49 | 204 | *EST | 14906-59-3 |
|  | (9.37) | (288) | (1205) | 72±0.5 | 301±2 | 80ARS | 273-09-6 |
| $\text{C}_6\text{H}_4\text{N}_2\text{O}_4^+$  | (≤10.71) | (≤267) | (≤1119) | 21±0.5 | 86±2 | 76FER/PIA | 528-29-0 |
|  | 10.43±0.02 | 255 | 1065 | 14±0.2 | 59±1 | 76FER/PIA | 99-65-0 |
|  | 10.3±0.1 | 251 | 1051 | 14±0.7 | 57±3 | 76FER/PIA | 100-25-4 |
| $\text{C}_6\text{H}_4\text{N}_2\text{Se}^+$  | (8.5) IP is onset of photoelectron band. | (292) | (1223) | 96 | 403 | *EST | 273-92-7 |
| $\text{C}_6\text{H}_4\text{N}_3\text{O}_2^+$  | (7.89) IP from 77NUY/MES. | (258) | (1079) | 76 | 318 | *EST | |

Table 1. Positive Ion Table - Continued

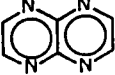
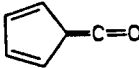
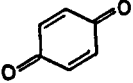

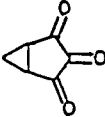

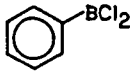
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|------------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_4\text{N}_4^+$  | (8.9) IP is onset of photoelectron band (84GLE/SPA2). | (295) | (1233) | 89 | 374 | *EST | 255-53-8 |
| $\text{C}_6\text{H}_4\text{O}^+$  | 8.2 IP is onset of photoelectron band (79SCH/SCH, 81BOC/HIR). | (214) | (895) | 25±1 | 104±6 | *EST | 4727-22-4 |
| $\text{C}_6\text{H}_4\text{O}_2^+$  | 10.04±0.18 See also: 83BOC/MOH. | 202 | 846 | -29±1 | -123±4 | 77PED/RYL | 106-51-4 |
|  | (9.3) IP is onset of photoelectron band. | (189) | (791) | -25±1 | -106±4 | *EST | 583-63-1 |
| $\text{C}_6\text{H}_4\text{O}_3^+$  | (9.0) IP is onset of photoelectron band (81BEC/HOF). | (119) | (498) | -88 | -370 | *EST | 81640-31-5 |
| C_6H_5^+  | 8.25 $\Delta_f H(\text{Ion})$ from appearance potential measurements(86MAL/LIF). See also: 83DAN/ROS, 85MAL/ARA, 76BAE/TSA, 84LIF/MAL, 80ROS/STO, 81PRA/CHU, 84GEF/LIF, 85DUN, 74BEA, 84PAN/BAE, 85PAN/BAE, 85PAN/BAE2, 86NIS/DAS and 84BUR/HOL. IP given is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. Experimental IP of radical = 8.1-8.2 eV. | 269.3 272.8 | 1126.9 1141.3 | 79±1 | 329±4 | 82MCM/GOL | 2396-01-2 |
| $\text{C}_6\text{H}_5\text{BCl}_2^+$  | (9.3) IP is onset of photoelectron band. | (151) | (631) | -64±0.5 | -266±2 | 77PED/RYL | 873-51-8 |

Table 1. Positive Ion Table - Continued

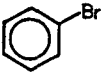
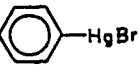
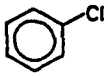
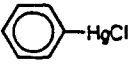
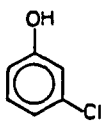
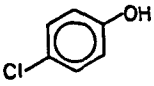
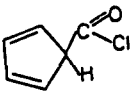
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_5\text{Br}^+$  | 8.98±0.02 | 232 237 | 971 993 | 24.9±0.7 30.1±0.7 | 104.3±3.1 126.1±3.1 | 77PED/RYL | 108-86-1 |
| | | IP from 78LIA/AUS, 82LEV/LIA, onset of photoelectron band in 81KIM/KAT. See also: 82VON/ASB, 83KLA/KOV, 86FUJ/OHN. | | | | | |
| $\text{C}_6\text{H}_5\text{BrHg}^+$  | (9.1) | (248) | (1037) | 38 | 159 | *EST | 1192-89-8 |
| | | IP is onset of photoelectron band (81BAI/CHI). | | | | | |
| $\text{C}_6\text{H}_5\text{Cl}^+$  | 9.06±0.02 | 222 | 929 | 13.0±0.2 | 54.4±0.9 | 85PLA/SIM | 108-90-7 |
| | | See also: 78LIA/AUS, 81RUS/KLA2, 82VON/ASB, 83KLA/KOV, 81KIM/KAT, 86FUJ/OHN. | | | | | |
| $\text{C}_6\text{H}_5\text{ClHg}^+$  | 9.14±0.04 | 240.5 | 1006.2 | 29.7 | 124.3 | 85DEW/GRA | 100-56-1 |
| | | See also: 81BAI/CHI. | | | | | |
| $\text{C}_6\text{H}_5\text{ClO}^+$  | (8.65) | (163) | (682) | -37±2 | -153±9 | 77PED/RYL | 108-43-0 |
| | | IP from 85OIK/ABE. | | | | | |
|  | (≤8.69) | (≤165) | (692) | -35±2 | -146±9 | 77PED/RYL | 106-48-9 |
|  | (8.9) | (194) | (813) | -11 | -46 | *EST | 78957-21-8 |
| | | IP is onset of photoelectron band (81BOC/HIR). | | | | | |

Table 1. Positive Ion Table - Continued

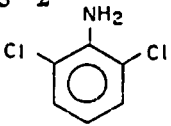
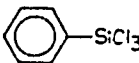

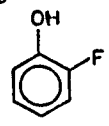
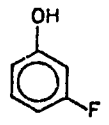
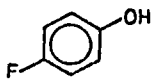
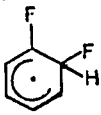
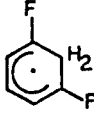
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|--|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_5\text{Cl}_2\text{N}^+$  | (7.60±0.02) | (182) | (763) | 7 | 30 | *EST | 608-31-1 |
| $\text{C}_6\text{H}_5\text{Cl}_3\text{Si}^+$  | (9.10) IP from 84VES/HAR. | (115) | (481) | -95 | -397 | *EST | 98-13-5 |
| $\text{C}_6\text{H}_5\text{F}^+$  | 9.200±0.005 See also: 81BIE/ASB, 81KIM/KAT, 86FUJ/OHN. | 184.4 | 771.6 | -27.7±0.3 | -116.0±1.4 | 77PED/RYL | 462-06-6 |
| $\text{C}_6\text{H}_5\text{FO}^+$  | 8.68±0.02 IP from 85OIK/ABE. | (131) | (548) | -69 | -289 | *EST | 367-12-4 |
|  | 8.73±0.02 IP from 85OIK/ABE. | (131) | (547) | -71 | -295 | *EST | 372-20-3 |
|  | (8.5) IP is onset of photoelectron band. | (126) | (529) | -70 | -291 | *EST | 371-41-5 |
| $\text{C}_6\text{H}_5\text{F}_2^+$  | | 114 | 475 | From proton affinity of 1,2-difluorobenzene (RN 367-11-3). PA = 181.8 kcal/mol, 761. kJ/mol. | | | |
|  | | 110 | 460 | From proton affinity of 1,3-difluorobenzene (RN 372-18-9) (82MAS/BOH). PA = 181.9 kcal/mol, 761. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

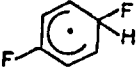
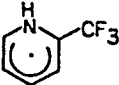
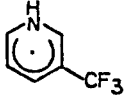
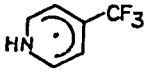
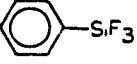
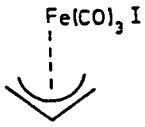
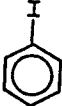
| ION Neutral | Ionization potential eV | | $\Delta_f H(\text{Ion})$ kcal/mol kJ/mol | | $\Delta_f H(\text{Neutral})$ kcal/mol kJ/mol | | Neutral reference | CAS registry number |
|--|---|--|---|--------|---|-----------|----------------------|------------------------|
| | | | | | | | | |
| $\text{C}_6\text{H}_5\text{F}_2^+$  | | | 111 | 465 | | | | |
| | | | From proton affinity of 1,4-difluorobenzene (RN 540-36-3). PA = 181.2 kcal/mol, 758. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_5\text{F}_3\text{N}^+$  | | | 27 | 113 | | | | |
| | | | From proton affinity of 2-trifluoromethylpyridine (RN 368-48-9). PA = 211.5 kcal/mol, 885. kJ/mol. | | | | | |
|  | | | 25 | 104 | | | | |
| | | | From proton affinity of 3-trifluoromethylpyridine (RN 3796-23-4). PA = 212.6 kcal/mol, 889. kJ/mol. | | | | | |
|  | | | 25 | 104 | | | | |
| | | | From proton affinity of 4-trifluoromethylpyridine (RN 3796-24-5). PA = 212.8 kcal/mol, 890. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_5\text{F}_3\text{Si}^+$  | (9.18) | | (-50) | (-207) | -261 | -1093 | *EST | 368-47-8 |
| | IP from 84VES/HAR. | | | | | | | |
| $\text{C}_6\text{H}_5\text{FeIO}_3^+$  | 8.17 | | 106 | 444 | -82±3 | -344±11 | 82PIL/SKI | 12189-10-5 |
| | IP is onset of photoelectron band (82LOU/HAR). | | | | | | | |
| $\text{C}_6\text{H}_5\text{I}^+$  | 8.685 | | 240 | 1003 | 39.4±1.4 | 164.9±5.9 | 77PED/RYL | 591-50-4 |
| | | | 243 | 1019 | 43.2±1.4 | 180.9±5.9 | | |
| | See also: 83KLA/KOV, 83DAN/ROS, 81KIM/KAT, 86FUJ/OHN. | | | | | | | |

Table 1. Positive Ion Table - Continued

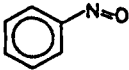
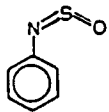
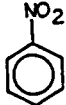
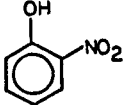
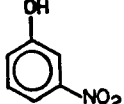
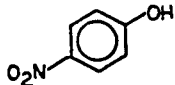
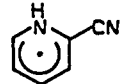
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------------|--|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_5\text{NO}^+$  | (8.09) | (235) | (982) | 48±1 | 201±4 | 75CHO/GOL | 586-96-9 |
| $\text{C}_6\text{H}_5\text{NOS}^+$  | (8.8) IP is onset of photoelectron band (82LOU/VAN). | (196) | (819) | -7 | -30 | *EST | 1122-83-4 |
| $\text{C}_6\text{H}_5\text{NO}_2^+$  | 9.86±0.02 | 243 250 | 1019 1045 | 16.1±0.2 22 | 67.6±1 92 | 77PED/RYL | 98-95-3 |
| | | See also: 83KLA/KOV, 81KIM/KAT, 81ALL/MIG, 82ALL/MIG, 73GOL/KOR. 0 K values from 84PAN/BAE, 85PAN/BAE2. | | | | | |
| $\text{C}_6\text{H}_5\text{NO}_3^+$  | (9.1) IP is onset of photoelectron band. | (187) | (780) | -23 | -98 | *EST | 88-75-5 |
|  | (9.0) IP is onset of photoelectron band. | (181) | (757) | -27 | -111 | *EST | 554-84-7 |
|  | (9.1) IP is onset of photoelectron band. | (182) | (762) | -28 | -116 | *EST | 100-02-7 |
| $\text{C}_6\text{H}_5\text{N}_2^+$  | | 225 | 943 | From proton affinity of 2-pyridinecarbonitrile. (RN 100-70-9). 208.1 kcal/mol, 871. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

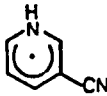
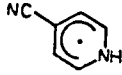
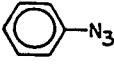
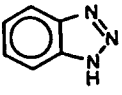
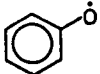
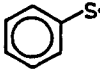
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|------------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_5\text{N}_2^+$  | | 222 | 932 | | | | |
| | | From proton affinity of 3-pyridinecarbonitrile (RN 100-54-9). PA = 209.3 kcal/mol, 876. kJ/mol. | | | | | |
|  | | 223 | 934 | | | | |
| | | From proton affinity of 4-pyridinecarbonitrile (RN 100-48-1). PA = 210.3 kcal/mol, 880 kJ/mol. | | | | | |
| $\text{C}_6\text{H}_5\text{N}_3^+$  | (8.4) | (286) | (1195) | 92 | 385 | 29ROT/MUE | 622-37-7 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (9.20±0.05) | (295) | (1236) | 83 | 348 | 61ZIM/GEI | 95-14-7 |
| $\text{C}_6\text{H}_5\text{O}^+$  | (8.56) | (208) (212) | (870) (888) | 11.4 | 48 | 82MCM/GOL | 2122-46-5 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations (84LOS/HOL). IP from 80DEW/DAV. See also: 84PAN/BAE, 85PAN/BAE, 85PAN/BAE2, 86DAS/GIL. | | | | | | |
| $\text{C}_6\text{H}_5\text{S}^+$  | (8.63±0.10) | (254) | (1063) | 55±2 | 230±8 | 82MCM/GOL | 4985-62-0 |
| C_6H_6^+ $\text{CH}_2 = \text{C} = \text{CHCH} = \text{C} = \text{CH}_2$ | (8.53) | (295) (299) | (1234) (1251) | 98 102 | 411 428 | 82ROS/DAN | 29776-96-3 |
| $\text{HC} = \text{CCH}_2\text{CH} = \text{C} = \text{CH}_2$ | (9.40) | (316) (320) | (1321) (1339) | 99 103 | 414 432 | 82ROS/DAN | 33142-15-3 |
| | IP from 82ROS/DAN (onset of photoelectron band). | | | | | | |

Table 1. Positive Ion Table - Continued


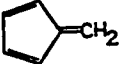
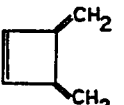
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6H_6^+ HC≡CCH=CHCH=CH ₂ | (9.20) | (299) | (1253) | 87 | 365 | 82ROS/DAN | 10420-90-3 |
| | | (303) | (1270) | 91 | 382 | | |
| IP from 82ROS/DAN. | | | | | | | |
| CH ₂ =CHC≡CCH=CH ₂ | (8.50±0.02) | (280) | (1172) | 84 | 352 | 82ROS/DAN | 821-08-9 |
| | | (284) | (1189) | 88 | 369 | | |
| See also: 85DEW/TIE. | | | | | | | |
| HC≡CC≡CC ₂ H ₅ | (9.41) | (312) | (1306) | 95 | 398 | 82ROS/DAN | 4447-21-6 |
| | | (316) | (1323) | 99 | 415 | | |
| IP from 82ROS/DAN. | | | | | | | |
| HC≡CCH ₂ C≡CCH ₃ | (9.50) | (317) | (1328) | 98 | 411 | 82ROS/DAN | 10420-91-4 |
| | | (321) | (1345) | 102 | 428 | | |
| IP from 82ROS/DAN (onset of photoelectron band). | | | | | | | |
| HC≡CCH ₂ CH ₂ C≡CH | 9.90 | 327 | 1369 | 99 | 414 | 82ROS/DAN | 628-16-0 |
| | | 331 | 1387 | 103 | 432 | | |
| See also: 82ROS/DAN. | | | | | | | |
| CH ₃ C≡CC≡CCH ₃ | 8.92±0.05 | 296 | 1238 | 90 | 377 | 82ROS/DAN | 2809-69-0 |
| | | 299 | 1255 | 94 | 394 | | |
| See also: 82ROS/DAN. | | | | | | | |
|  | 9.2459±0.0002 | 233.2 | 975.8 | 19.8±0.1 | 82.9±0.3 | 77PED/RYL | 71-43-2 |
| | | 237.2 | 992.6 | 24.0±0.2 | 100.4±1 | | |
| IP from 84GRU/WHE. IP at 298 K = 9.225±0.005 (78LIA/AUS). See also: 81KIM/KAT, 81KIM/KAT, 84HOW/GON. <i>duplicate</i> | | | | | | | |
|  | (8.36) | (246) | (1030) | 53.5 | 223.8 | 84ROT | 497-20-1 |
| | | (251) | (1048) | 57.8 | 241.9 | | |
|  | (8.80) | (283) | (1184) | 80.4 | 335.5 | 86ROT/LEN | 5291-90-7 |
| | | (286) | (1198) | 83 | 349 | | |

Table 1. Positive Ion Table - Continued

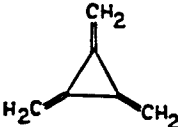

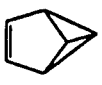
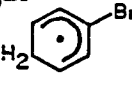
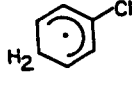
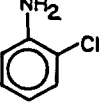
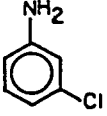
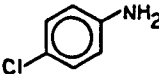
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|------------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6H_6^+  | (8.80) | (298) (302) | (1245) (1263) | 95 99 | 396 414 | 82ROS/DAN | 3227-90-5 |
| | | IP from 82ROS/DAN (onset of photoelectron band). | | | | | |
|  | (9.0) | (294) (298) | (1232) (1250) | 87 91 | 364 382 | 82ROS/DAN | 5649-95-6 |
| | | IP from 82ROS/DAN (onset of photoelectron band). | | | | | |
|  | 8.1 | (274) (278) | (1144) (1163) | 87 91 | 363 381 | 82ROS/DAN | 659-85-8 |
| | | IP from 82ROS/DAN (onset of photoelectron band). | | | | | |
| $\text{C}_6\text{H}_6\text{Br}^+$  | | 208 | 871 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{Br}$ (RN 108-86-1). PA = 182.4 kcal/mol, 763. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_6\text{Cl}^+$  | | 196 | 821 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{Cl}$ (RN 108-90-7). PA = 181.7 kcal/mol, 760. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_6\text{ClN}^+$  | (8.50) | (211) | (881) | 15 | 61 | *EST | 95-51-2 |
|  | (8.09±0.1) | (200) | (836) | 13 | 55 | *EST | 108-42-9 |
|  | (≤8.18) | (≤202) | (≤844) | 13 | 55 | *EST | 106-47-8 |

Table 1. Positive Ion Table - Continued

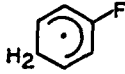
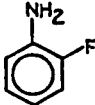
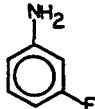
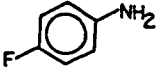
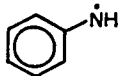
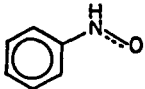
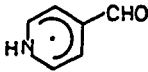
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|---|------------------------------|--------|----------------------|--|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_6\text{F}^+$ | | | 155 | 650 | | | |
|  | | | | | | | |
| | | | From proton affinity of $\text{C}_6\text{H}_5\text{F}$ (RN 462-06-6). PA = 182.6 kcal/mol, 764. kJ/mol. | | | | |
| $\text{C}_6\text{H}_6\text{FN}^+$ | | | | | | | |
|  | (≤ 8.18) | (≤ 164) | (≤ 683) | -25 | -106 | *EST | 348-54-9 |
|  | (≤ 8.32) | (≤ 165) | (≤ 691) | -27 | -112 | *EST | 372-19-0 |
|  | (≤ 8.18) | (≤ 163) | (≤ 680) | -26 | -109 | *EST | 371-40-4 |
| $\text{C}_6\text{H}_6\text{Hg}^+$ ($\text{CH}_3\text{C}\equiv\text{C}$) ₂ Hg | 8.98±0.07 | (323) | (1351) | 116 | 485 | *EST | 64705-15-3 |
| | | | | | | | IP is onset of photoelectron band (81FUR/PIA). |
| $\text{C}_6\text{H}_6\text{N}^+$ | | | | | | | |
|  | (8.26±0.1) | (247) | (1034) | 57±2 | 237±8 | 82MCM/GOL | 2835-77-0 |
| $\text{C}_6\text{H}_6\text{NO}^+$ | | | | | | | |
|  | | 209 | 874 | | | | |
| | | From proton affinity of nitrosobenzene (RN 586-96-9). PA = 204.8 kcal/mol, 857. kJ/mol. | | | | | |
|  | | (156) | (654) | | | | |
| | | From proton affinity of 4-pyridinecarboxaldehyde (RN 872-85-5). PA = (215.2) kcal/mol, (900.) kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

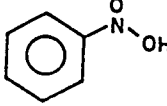
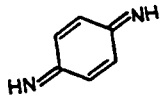
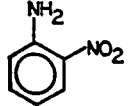
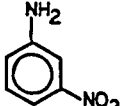
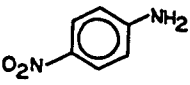
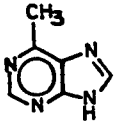
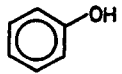
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|------------|--|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_6\text{NO}_2^+$  | | 189 | 789 | From proton affinity of nitrobenzene (RN 98-95-3). PA = 193.4 kcal/mol, 809. kJ/mol. | | | |
| $\text{C}_6\text{H}_6\text{N}_2^+$  | (9.36±0.03) | (294) | (1229) | 78±1 | 326±5 | *EST | 4377-73-5 |
| $\text{C}_6\text{H}_6\text{N}_2\text{O}_2^+$  | 8.27±0.01 | 206 | 862 | 15±1 | 64±4 | 77PED/RYL | 88-74-4 |
|  | 8.31±0.02 | 207 | 864 | 15±0.5 | 62±2 | 83NIS/SAK | 99-09-2 |
|  | 8.34±0.01 | 205 | 860 | 13±0.5 | 55±2 | 83NIS/SAK | 100-01-6 |
| $\text{C}_6\text{H}_6\text{N}_4^+$  | (8.9) IP is onset of photoelectron band. | (248) | (1037) | 43 | 178 | *EST | 2004-03-7 |
| $\text{C}_6\text{H}_6\text{O}^+$  | 8.47 | 173 175 | 722 732 | -23.0±0.2 -20.4 | -96.3±0.8 -85.2 | 78KUD/KUD | 108-95-2 |
| IP from 84FRA/FRA. See also: 84FUK/YOS, 83KLA/KOV, 81KIM/KAT. | | | | | | | |

Table 1. Positive Ion Table - Continued

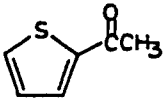
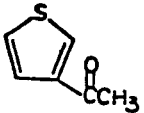
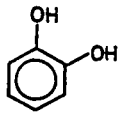
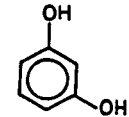
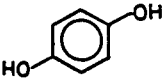
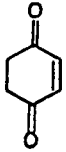
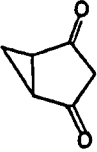
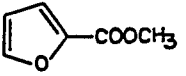
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₆H₆OS⁺ | | | | | | | | |
|  | | (9.20±0.05) | (206) | (862) | -6 | -26 | *EST | 88-15-3 |
|  | | (9.32±0.05) | (209) | (875) | -6 | -24 | *EST | 1468-83-3 |
| C₆H₆O₂⁺ | | | | | | | | |
|  | | (8.15) | (123) | (514) | -65±1 | -272±5 | 79KUD/KUD | 120-80-9 |
| | | IP is onset of photoelectron band. | | | | | | |
|  | | (8.2) | (123) | (514) | -65.6±0.5 | -274.7±2.1 | 79KUD/KUD | 108-46-3 |
| | | IP is onset of photoelectron band. | | | | | | |
|  | | 7.95±0.03 | 121 | 505 | -63±0.5 | -262±2 | 79KUD/KUD | 123-31-9 |
| | | IP from 85OIK/ABE. | | | | | | |
|  | | (9.77) | (205) | (859) | -20 | -84 | *EST | 4505-38-8 |
| | | IP is onset of photoelectron band (85GLE/JAH). | | | | | | |
|  | | (9.4) | (170) | (713) | -46 | -194 | *EST | 29798-87-6 |
| | | IP is onset of photoelectron band (81BEC/HOF). | | | | | | |
| C₆H₆O₃⁺ | | | | | | | | |
|  | | (≤9.00±0.05) | (≤110) | (≤463) | -97 | -405 | 80BAL/LEB | 611-13-2 |

Table 1. Positive Ion Table - Continued

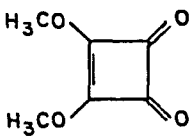
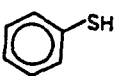
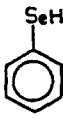

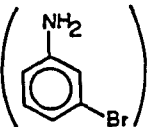
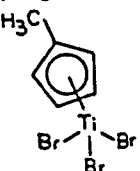
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_6\text{O}_4^+$  | ≤ 9.20 | (≤ 107) | (≤ 447) | -105 | -441 | *EST | 5222-73-1 |
| $\text{C}_6\text{H}_6\text{S}^+$  | 8.30 ± 0.02 See also: 82CAR/KIB, 81KIM/KAT. | 218 | 913 | 26.9 ± 0.2 | 112.4 ± 0.8 | 77PED/RYL | 108-98-5 |
| $\text{C}_6\text{H}_6\text{Se}^+$  | (≤ 7.7) IP is onset of photoelectron band (81BAK/ARM). | (≤ 217) | (≤ 906) | 39 | 163 | *EST | 645-96-5 |
| C_6H_7^+  | | 204 | 854 | | | | |
| ($\text{HC}=\text{CCH}_2\text{CH}_2\text{C}=\text{CH}$)H | | 269 | 1124 | | | | |
| | From proton affinity of $\text{HC}=\text{C}(\text{CH}_2)_2\text{C}=\text{CH}$ (RN 628-16-0) (85LIA/AUS). PA = 196 kcal/mol, 819 kJ/mol. | | | | | | |
| ($\text{CH}_3\text{C}=\text{CC}=\text{CCH}_3$)H | | 260 | 1087 | | | | |
| | From proton affinity of $\text{CH}_3\text{C}=\text{CC}=\text{CCH}_3$ (RN 2809-69-0) (85LIA/AUS). PA = 196 kcal/mol, 819 kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_7\text{BrN}^+$  | H ⁺ | 183 | 767 | | | | |
| | From proton affinity of 3-BrC ₆ H ₄ NH ₂ (RN 591-19-5). | | | | | | |
| $\text{C}_6\text{H}_7\text{Br}_3\text{Ti}^+$  | (8.6) IP is onset of photoelectron band (84TER/LOU). | (80) | (337) | -118 | -493 | *EST | 1277-45-8 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_7\text{ClN}^+$ | | | | | | | |
| | | 172 | 718 | | | | |
| | | From proton affinity of 3-ClC ₆ H ₄ NH ₂ (RN 108-42-9). PA = 207.2 kcal/mol, 867. kJ/mol. | | | | | |
| | | 170 | 712 | | | | |
| | | From proton affinity of 4-ClC ₆ H ₄ NH ₂ (RN 106-47-8). PA = 208.6 kcal/mol, 873. kJ/mol. | | | | | |
| | | (163) | (681) | | | | |
| | | From proton affinity of 2-chloro-4-methylpyridine (RN 3678-62-4). (218.6) kcal/mol, (915.) kJ/mol. | | | | | |
| | | (161) | (675) | | | | |
| | | From proton affinity of 2-chloro-6-methylpyridine (RN 18368-63-3). PA = (219) kcal/mol, (916) kJ/mol. | | | | | |
| $\text{C}_6\text{H}_7\text{ClNO}^+$ | | | | | | | |
| | | 127 | 531 | | | | |
| | | From proton affinity of 6-chloro-1-methyl-2(1H)pyridinone (RN 17228-63-6). PA = 217.8 kcal/mol, 911. kJ/mol. | | | | | |
| | | 129 | 538 | | | | |
| | | From proton affinity of 2-chloro-6-methoxypyridine (RN 17228-64-7). PA = 215.9 kcal/mol, 903. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_7\text{Cl}_3\text{Ti}^+$ | | | | | | | |
| | | (9.1) | (66) | (276) | -144 | -602 | *EST 1282-31-1 |
| | | IP is onset of photoelectron band (84TER/LOU). | | | | | |
| $\text{C}_6\text{H}_7\text{FN}^+$ | | | | | | | |
| | | 132 | 552 | | | | |
| | | From proton affinity of 3-fluorobenzencamine (RN 372-19-0). PA = 207.0 kcal/mol, 866. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

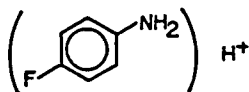
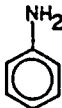
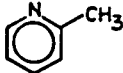
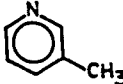

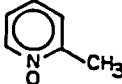
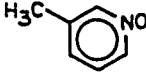
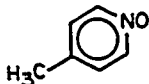
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|---|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_7\text{FN}^+$  | | 132 | 550 | From proton affinity of 4-fluorobenzeneamine (RN 371-40-4). PA = 208.1 kcal/mol, 871. kJ/mol. | | | |
| $\text{C}_6\text{H}_7\text{N}^+$  | 7.720±0.002 | 198 | 829 | 20.8±0.2 | 87.1±0.8 | 77PED/RYL | 62-53-3 |
| | IP from 84SMI/HAG. See also: 83KLA/KOV, 81KIM/KAT, 85MBE/SEK, 85HAG/SMI. | | | | | | |
|  | 9.02±0.03 | 232 | 969 | 23.7±0.2 | 99.2±0.7 | 77PED/RYL | 109-06-8 |
| | See also: 81KIM/KAT. | | | | | | |
|  | 9.04±0.03 | 234 | 979 | 25.4±0.1 | 106.4±0.5 | 77PED/RYL | 108-99-6 |
| | See also: 81MOD/DIS2, 81KIM/KAT. | | | | | | |
|  | 9.04±0.03 | 233 | 976 | 24.8±0.3 | 103.8±1.2 | 77PED/RYL | 108-89-4 |
| $\text{C}_6\text{H}_7\text{NO}^+$  | ≤8.21±0.02 | (≤194) | (≤811) | 5 | 19 | *EST | 931-19-1 |
|  | (≤8.20±0.02) | (≤195) | (≤817) | 6 | 26 | *EST | 1003-73-2 |
|  | 8.12±0.02 | (193) | (807) | 6 | 24 | *EST | 1003-67-4 |

Table 1. Positive Ion Table - Continued

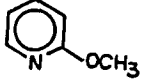
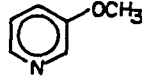
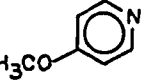
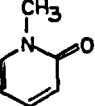
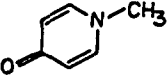
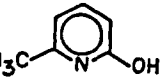
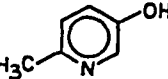
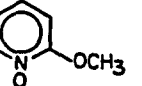
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₆H₇NO⁺ | | | | | | | |
|  | (8.7) IP is onset of photoelectron band. | (189) | (787) | -12 | -52 | *EST | 1628-89-3 |
|  | (9.34±0.02) | (211) | (885) | -4 | -16 | *EST | 7295-76-3 |
|  | (9.58±0.02) | (218) | (911) | -3 | -13 | *EST | 620-08-6 |
|  | (8.2) IP is onset of photoelectron band. See also: 81DRE/BEC. | (169) | (706) | -20±2 | -85±10 | *EST | 694-85-9 |
|  | (≤8.20±0.03) | (≤186) | (≤778) | -3±2 | -13±8 | *EST | 695-19-2 |
|  | (8.33) | (163) | (684) | -29±0.7 | -120±3 | 82SUR/ELS | 73229-70-6 |
|  | (9.15±0.05) | (194) | (813) | -17±0.7 | -70±3 | 82SUR/ELS | 1121-78-4 |
| C₆H₇NO₂⁺ | | | | | | | |
|  | (7.5) IP is onset of photoelectron band. | (151) | (631) | -22 | -93 | *EST | 1122-96-9 |

Table 1. Positive Ion Table - Continued

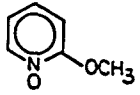
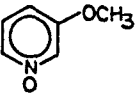
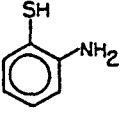
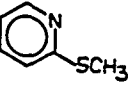
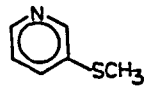
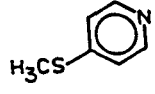
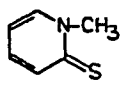
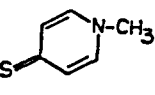
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_7\text{NO}_2^+$ | | | | | | | |
|  | (8.21±0.05) | (158) | (660) | -32 | -132 | *EST | 20773-98-2 |
|  | (8.40±0.05) | (171) | (714) | -23 | -96 | *EST | 14906-61-7 |
| $\text{C}_6\text{H}_7\text{NS}^+$ | | | | | | | |
|  | (7.6) IP is onset of photoelectron band (82ZVE/ASH). | (203) | (849) | 28 | 116 | *EST | 137-07-5 |
|  | (8.24±0.03) See also: 81DRE/BEC. | (223) | (933) | 33 | 138 | *EST | 18438-38-5 |
|  | (≤8.41±0.03) | (≤231) | (≤966) | 37 | 155 | *EST | 18794-33-7 |
|  | (≤8.73±0.03) | (≤238) | (≤997) | 37 | 155 | *EST | 22581-72-2 |
|  | (7.69±0.03) See also: 81DRE/BEC. | (218) | (912) | 41 | 170 | *EST | 2044-27-1 |
|  | 7.54±0.02 | (239) | (999) | 65 | 272 | *EST | 6887-59-8 |

Table 1. Positive Ion Table - Continued

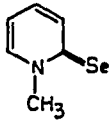
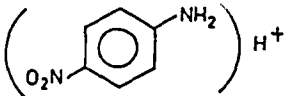
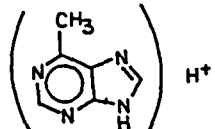
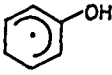
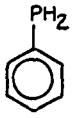
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|---|----------------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_7\text{NSe}^+$ |  | ≤ 7.22 IP from 81DRE/BEC. | (≤ 224) | (≤ 937) | 57 | 240 | *EST | 2240-85-9 |
| $\text{C}_6\text{H}_7\text{N}_2\text{O}_2^+$ |  | | 172 | 719 | | | | |
| | | | From proton affinity of 4- $\text{NO}_2\text{C}_6\text{H}_4\text{NH}_2$ (RN 100-01-6) (84ROL/HOU). PA = 207.0 kcal/mol, 866. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_7\text{N}_4^+$ |  | | 185 | 775 | | | | |
| | | | From proton affinity of 6-methylpurine (RN 2004-03-7). PA = (223) kcal/mol, (933) kJ/mol. | | | | | |
| $\text{C}_6\text{H}_7\text{O}^+$ |  | | 146 | 613 | | | | |
| | | | From proton affinity of $\text{C}_6\text{H}_5\text{OH}$ (RN 108-95-2). PA = 196.3 kcal/mol, 821. kJ/mol. | | | | | |
| | $(\text{HC}=\text{CCH}_2)_2\text{OH}$ | | (246) | (1031) | | | | |
| | | | From proton affinity of $(\text{HC}=\text{CCH}_2)_2\text{O}$ (RN 6921-27-3). PA = 190.8 kcal/mol, 798. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_7\text{P}^+$ |  | (8.47 ± 0.01) See also: 81CAB/COW2. | (226) | (945) | 31 | 128 | *EST | 638-21-1 |
| C_6H_8^+ | $(E)\text{-CH}_2=\text{C}=\text{CHCH}=\text{CHCH}_3$ (8.32) | | (244) | (1020) | 52 | 217 | *EST | 20130-95-4 |
| | $(Z)\text{-CH}_2=\text{CHCH}=\text{CHCH}=\text{CH}_2$ 8.31 \pm 0.01 | | 233 | 973 | 41 | 171 | 70BEN/O'N | 2612-46-6 |
| | $(E)\text{-CH}_2=\text{CHCH}=\text{CHCH}=\text{CH}_2$ 8.28 \pm 0.02 | | (231) (237) | (965) (991) | 40 46 | 166 192 | *EST | 821-07-8 |
| | $\text{CH}_3\text{CH}=\text{C}=\text{CHCH}=\text{CH}_2$ (8.56) | | (250) | (1048) | 53 | 222 | *EST | 33755-64-5 |

Table 1. Positive Ion Table - Continued



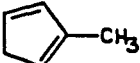
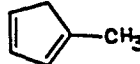
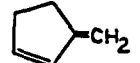
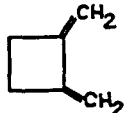
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------------|------------------------------|------------------------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6H_8^+ | | | | | | | |
| $\text{CH}_2 = \text{C} = \text{C}(\text{CH}_3)\text{CH} = \text{CH}_2$ | (8.54) | (249) | (1040) | 52 | 216 | *EST | 25054-29-9 |
| $\text{CH}_2 = \text{C} = \text{CHC}(\text{CH}_3) = \text{CH}_2$ | (8.54) | (249) | (1040) | 52 | 216 | *EST | 14763-81-6 |
| $\text{C}_2\text{H}_5\text{C} \equiv \text{CCH} = \text{CH}_2$ | (8.91±0.01) | (260) | (1090) | 55 | 230 | *EST | 13721-54-5 |
| $\text{CH}_3\text{C} \equiv \text{CC}(\text{CH}_3) = \text{CH}_2$ | (8.72±0.01) | (253) | (1058) | 52 | 217 | *EST | 926-55-6 |
|  | 8.25±0.02 | <u>215.6</u> <u>221.3</u> | <u>902.3</u> <u>926.1</u> | 25.4±0.1 31.1±0.1 | 106.3±0.5 130.1±0.5 | 77PED/RYL | 592-57-4 |
| | See also: 81KIM/KAT. | | | | | | |
|  | 8.82±0.02 | 229 235 | 959 985 | 25.8±0.5 32.0±0.5 | 107.9±2 133.9±2 | 77SHA/GOL | 628-41-1 |
| | See also: 81KIM/KAT. | | | | | | |
|  | 8.40±0.05 | (217) | (907) | 23 | 97 | *EST | 96-39-9 |
|  | 8.45±0.05 | (218) | (911) | 23 | 96 | *EST | 3727-31-9 |
|  | (8.40) | (223) | (931) | 29 | 121 | *EST | 930-26-7 |
|  | (8.4) | (242) | (1011) | 48 | 201 | 80GAJ | 14296-80-1 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

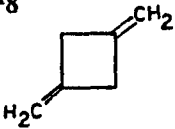
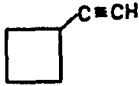
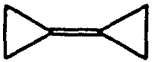


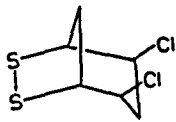
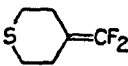
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6H_8^+  | (8.7) IP is onset of photoelectron band. | (254) | (1063) | 54 | 224 | 80GAJ | 2045-78-5 |
|  | (9.6) IP is onset of photoelectron band. | (284) | (1188) | 63 | 262 | *EST | 50786-62-4 |
|  | (8.5) IP is onset of photoelectron band. | (280) | (1170) | 84 | 350 | *EST | 27567-82-4 |
|  | (9.1) IP is onset of photoelectron band. | (271) | (1135) | 61 | 257 | 80ROT/KLA | 3097-63-0 |
|  | (≤ 9.43) | (≤ 272) | (≤ 1138) | 54.5 | 228.0 | 85SVY/IOF | 287-12-7 |
| $\text{C}_6\text{H}_8\text{Cl}_2\text{S}_2^+$  | (7.8) IP is onset of photoelectron band (83JOR/MCC). | (158) | (660) | -22 | -93 | *EST | 74796-12-6 |
| $\text{C}_6\text{H}_8\text{F}_2\text{S}^+$  | (9.34) IP from 80SAR/WOR. | (129) | (538) | -87 | -363 | *EST | 77471-71-7 |
| $\text{C}_6\text{H}_8\text{N}^+$ ($\text{HC}=\text{CCH}_2$) ₂ NH ₂ | | 262 | 1098 | | | | |
| | From proton affinity of ($\text{HC}=\text{CCH}_2$) ₂ NH (RN 6921-28-4). PA = 216.1 kcal/mol, 904. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

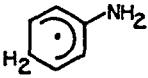
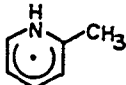
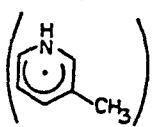
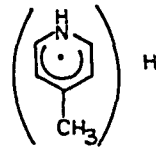
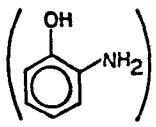
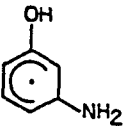
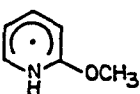
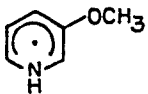
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₆H₈N⁺ | | | | | | | |
|  | | 177 | 740 | From proton affinity of C ₆ H ₅ NH ₂ (RN 62-53-3). PA = 209.5 kcal/mol, 876. kJ/mol. | | | |
|  | | 164 | 688 | From proton affinity of 2-methylpyridine (RN 109-06-8). PA = 225.0 kcal/mol, 942. kJ/mol. | | | |
|  | H ⁺ | 167 | 698 | From proton affinity of 3-methylpyridine (RN 108-99-6). PA = 224.1 kcal/mol, 938. kJ/mol. | | | |
|  | H ⁺ | 165 | 692 | From proton affinity of 4-methylpyridine (RN 108-89-4). PA = 225.2 kcal/mol, 942. kJ/mol. | | | |
| C₆H₈NO⁺ | | | | | | | |
|  | H ⁺ | 131 | 547 | From proton affinity of 2-HOC ₆ H ₄ NH ₂ . PA = 214.2 kcal/mol, 896. kJ/mol. | | | |
|  | | 130 | 545 | From proton affinity of 3-(OH)C ₆ H ₄ NH ₂ (RN 591-27-5). PA = 214.2 kcal/mol, 896. kJ/mol. | | | |
|  | | 131 | 550 | From proton affinity of 2-methoxypyridine (RN 1628-89-3). PA = 221.9 kcal/mol, 928. kJ/mol. | | | |
|  | | 138 | 579 | From proton affinity of 3-methoxypyridine (RN 7295-76-3). PA = 223.6 kcal/mol, 935. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

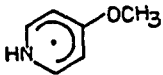
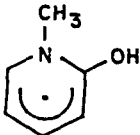
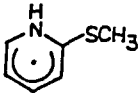
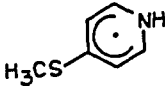
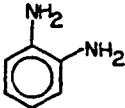
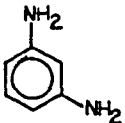
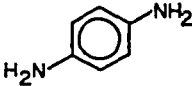
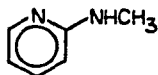
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₆H₈NO⁺ | | | | | | | |
|  | | 135 | 565 | | | | |
| | | From proton affinity of 4-methoxy-pyridine (RN 620-08-6). PA = 227.6 kcal/mol, 952. kJ/mol. | | | | | |
|  | | 125 | 524 | | | | |
| | | From proton affinity of 1-methyl-2-pyridinone (RN 694-85-9). PA = 220.2 kcal/mol, 921. kJ/mol. | | | | | |
| C₆H₈NS⁺ | | | | | | | |
|  | | 177 | 739 | | | | |
| | | From proton affinity of 2-(methylthio)pyridine (RN 18438-38-5). PA = 222.0 kcal/mol, 929. kJ/mol. | | | | | |
|  | | (177) | (742) | | | | |
| | | From proton affinity of 4-(methylthio)pyridine (RN 22581-72-2). PA = (225.5) kcal/mol, (943.) kJ/mol. | | | | | |
| C₆H₈N₂⁺ | | | | | | | |
|  | 7.2 | (188) | (787) | 22±1 | 92±5 | *EST | 95-54-5 |
| | See also: 81NEL/GRE. | | | | | | |
|  | 7.14 | (186) | (777) | 21 | 88 | *EST | 108-45-2 |
|  | 6.87±0.05 | (181) | (760) | 23 | 97 | *EST | 106-50-3 |
| | See also: 81CAB/COW2. | | | | | | |
|  | (8.26±0.05) | (220) | (924) | 30 | 127 | *EST | 4597-87-9 |

Table 1. Positive Ion Table - Continued

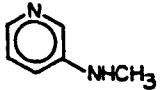
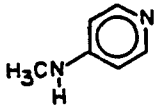
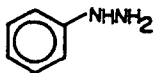
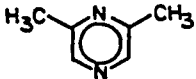
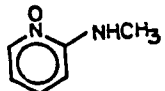
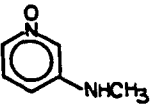
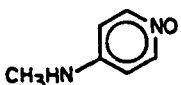
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-------------------------------------|--------------------------|--------|---|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_8\text{N}_2^+$ | | | | | | | |
|  | (8.53±0.05) | (231) | (965) | 34 | 142 | *EST | 18364-47-1 |
|  | (8.75±0.05) | (233) | (972) | 31 | 128 | *EST | 1121-58-0 |
|  | | | | 48.6±0.2 | 203.5±0.8 | 77PED/RYL | 100-63-0 |
| | | | | Values of 7.64 and 7.74 have been reported for the adiabatic IP of this compound. Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. | | | |
|  | (8.80) | (270) | (1128) | 67±0.7 | 279±3 | *EST | 108-50-9 |
| $\text{C}_6\text{H}_8\text{N}_2\text{O}^+$ | | | | | | | |
|  | (7.67±0.05) | (188) | (787) | 11 | 47 | *EST | 54818-70-1 |
|  | (7.97±0.05) | (198) | (829) | 14 | 60 | *EST | 54818-71-2 |
|  | (7.45±0.05) | (185) | (775) | 13 | 56 | *EST | 1122-92-5 |
| $\text{C}_6\text{H}_8\text{O}^+$ | | | | | | | |
| HC=CCOCH ₂ CH ₂ CH ₃ | (10.00±0.04) IP from 86TUR/HAV2. | (233) | (975) | 2.5 | 10.5 | *EST | |

Table 1. Positive Ion Table - Continued

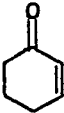
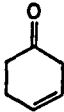
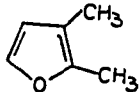
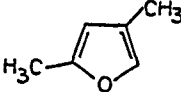
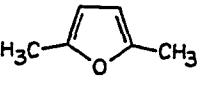
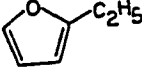

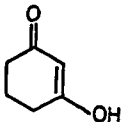
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_8\text{O}^+$ | | | | | | | |
|  | 9.23±0.05 | (185) | (775) | -28±0.7 | -116±3 | *EST | 930-68-7 |
|  | (≤9.42) | (≤185) | (≤773) | -33±0.7 | -136±3 | *EST | 4096-34-8 |
|  | (8.25±0.10) IP from 85GRU/SPI. | (166) | (694) | -24 | -102 | *EST | |
|  | (8.39±0.10) IP from 85GRU/SPI. | (166) | (694) | -28 | -116 | *EST | 3710-43-8 |
|  | (8.25±0.10) IP from 85GRU/SPI. | (165) | (690) | -25 | -106 | *EST | 625-86-5 |
|  | (8.45±0.05) | (171) | (715) | -24 | -100 | *EST | 3208-16-0 |
|  | (≤9.44±0.02) | (≤207) | (≤867) | -11 | -44 | *EST | 6705-50-6 |
| $\text{C}_6\text{H}_8\text{O}_2^+$ | | | | | | | |
|  | 9.52±0.05 | (141) | (589) | -79 | -330 | *EST | 504-02-9 |

Table 1. Positive Ion Table - Continued


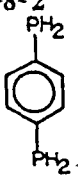
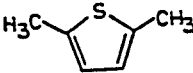
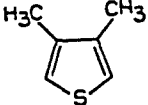
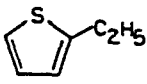
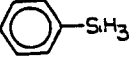

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_8\text{O}_2\text{S}^+$  | <9.6 IP is onset of photoelectron band (84AIT/GOS). | (<187) | (<784) | -34 | -142 | *EST | 84451-42-3 |
| $\text{C}_6\text{H}_8\text{P}_2^+$  | ≤ 8.78 IP from 81CAB/COW2. | (≤ 237) | (≤ 990) | 34 | 143 | *EST | 78550-67-1 |
| $\text{C}_6\text{H}_8\text{S}^+$  | (8.10) See also: 83BOC/ROT. | (199) | (832) | 12 | 50 | *EST | 638-02-8 |
|  | (≤ 8.55) IP from 83BOC/ROT. | (≤ 209) | (≤ 875) | 12 | 50 | *EST | 632-15-5 |
|  | (8.67 ± 0.05) | (215) | (898) | 15 | 61 | *EST | 872-55-9 |
| $\text{C}_6\text{H}_8\text{Si}^+$  | (9.09) | (236) | (988) | 27 | 111 | *EST | 694-53-1 |
| C_6H_9^+ $\text{CH}_3\text{C}=\text{CC}(\text{CH}_3)_2$  | | 216 | 904 | | | | 77920-98-0 |
| | From appearance potential measurements (84LOS/HOL). | | | | | | |
| | | 191 | 800 | | | | |
| | From proton affinities of 1,3-c- C_6H_8 (RN 592-57-4) PA = (200) kcal/mol, (837) kJ/mol and 1,4-c- C_6H_8 (RN 628-41-1) (83GAU/HOU) PA = (200) kcal/mol, (837) kJ/mol. Value derived from appearance potential measurements (84LOS/HOL) is the same. | | | | | | |

Table 1. Positive Ion Table - Continued

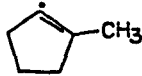

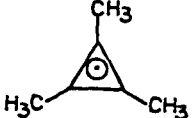

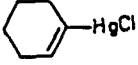

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6H_9^+  | | 190 | 795 | | | | 72026-92-7 |
| | | From appearance potential measurements (84LOS/HOL). | | | | | |
|  | | (202) | (845) | | | | |
| | | From proton affinity of 1-methyl-3-methylenecyclobutene. (RN 15082-13-0). PA = (212) kcal/mol, (887) kJ/mol. | | | | | |
|  | | 199 | 833 | | | | 26827-04-3 |
| | | From appearance potential measurements (84LOS/HOL). | | | | | |
| $\text{C}_6\text{H}_9\text{Br}^+$  | (9.5) | (235) | (983) | 16 | 66 | *EST | 77379-00-1 |
| | | IP is onset of photoelectron band (84DEL/ABE). | | | | | |
| $\text{C}_6\text{H}_9\text{ClHg}^+$  | (8.8) | (212) | (887) | 9 | 38 | *EST | 10080-39-4 |
| | | IP is onset of photoelectron band (81BAI/CHI). | | | | | |
| $\text{C}_6\text{H}_9\text{Cl}_2\text{P}^+$ (CH_3) ₃ CC=CPCl ₂ | (≤ 9.58) | (≤ 211) | (≤ 883) | -10 | -41 | *EST | 77376-08-0 |
| | | IP from 81CAB/COW. | | | | | |
| $\text{C}_6\text{H}_9\text{I}^+$  | (8.8) | (233) | (976) | 30 | 127 | *EST | 74725-75-0 |
| | | IP is onset of photoelectron band (84DEL/ABE). | | | | | |
| $\text{C}_6\text{H}_9\text{N}^+$ (E)-(CH ₃) ₂ NCH=CHC=CH | (7.7) | (260) | (1087) | 82±1 | 344±6 | *EST | 2206-24-8 |

Table 1. Positive Ion Table - Continued

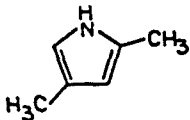
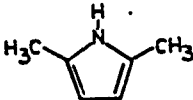
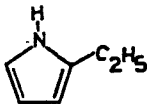
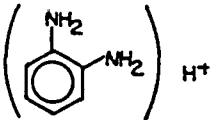
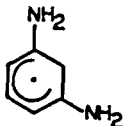
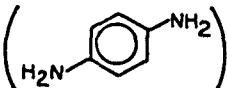
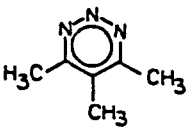
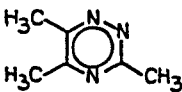
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------|---|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₆H₉N⁺ | | | | | | | |
|  | (7.54±0.02) | (184) | (767) | 10 | 40 | *EST | 625-82-1 |
|  | (≤7.69) | (≤187) | (≤782) | 9.5±0.2 | 39.8±0.8 | 77PED/RYL | 625-84-3 |
|  | (7.97±0.05) | (197) | (823) | 13 | 54 | *EST | 1551-06-0 |
| C₆H₉N₂⁺ | | | | | | | |
|  | | 175 | 732 | From proton affinity of 1,2-C ₆ H ₄ (NH ₂) ₂ (RN 95-54-5). PA = 212.8 kcal/mol, 890. kJ/mol. | | | |
|  | | 164 | 688 | From proton affinity of 1,3-C ₆ H ₄ (NH ₂) ₂ (RN 108-45-2). PA = 222.4 kcal/mol, 930.5 kJ/mol. | | | |
|  | | 173 | 723 | From proton affinity of 1,4-C ₆ H ₄ (NH ₂) ₂ (RN 106-50-3). PA = 215.9 kcal/mol, 903. kJ/mol. | | | |
| C₆H₉N₃⁺ | | | | | | | |
|  | ≤9.4 IP from 83GLE/SPA. | (≤267) | (≤1118) | 50 | 211 | *EST | 33209-85-7 |
|  | (≤8.84) | (≤274) | (≤1146) | 70 | 293 | *EST | 24108-36-9 |

Table 1. Positive Ion Table - Continued

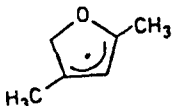
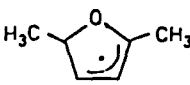
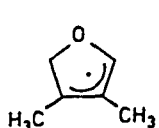

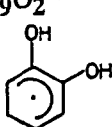
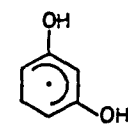
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_9\text{O}^+$ | | | | | | | |
|  | | 125 | 523 | | | | |
| | | From proton affinity of 2,4-dimethylfuran (RN 3710-43-8) (85HOU/ROL). PA = 213.0 kcal/mol, 819. kJ/mol. | | | | | |
|  | | 131 | 550 | | | | |
| | | From proton affinity of 2,5-dimethylfuran (RN 625-86-5) (85HOU/ROL, 86MAU). PA = 209.0 kcal/mol, 874. kJ/mol. | | | | | |
|  | | 133 | 556 | | | | |
| | | From proton affinity of 3,4-dimethylfuran (RN 20843-07-6) (85HOU/ROL). PA = 207.1 kcal/mol, 867. kJ/mol. | | | | | |
|  | | 155 | 649 | | | | |
| | | From proton affinity of bicyclo[2.2.1]hept-2-ene, 7-oxa- (RN 6705-50-6) (86HOU/SCH). PA = 200.0 kcal/mol, 837. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_9\text{O}_2^+$ | | | | | | | |
|  | | 92 | 384 | | | | |
| | | From proton affinity of 1,2-cyclohexanedione (RN 765-87-7) (83MAU). PA = 203.9 kcal/mol, 853. kJ/mol. | | | | | |
|  | | 76 | 318 | | | | |
| | | From proton affinity of 1,3-cyclohexanedione (RN 504-02-9). PA = 210.8 kcal/mol, 882. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{10}^+$ | | | | | | | |
| $\text{CH}_2 = \text{C} = \text{CHCH}_2\text{C}_2\text{H}_5$ | (9.00±0.05) | (237) | (990) | 29 | 122 | *EST | 592-44-9 |
| (E)- $\text{CH}_2 = \text{CHCH} = \text{CHC}_2\text{H}_5$ | 8.51 IP from 81MAS/MOU. | (210) | (878) | 14 | 57 | *EST | 20237-34-7 |
| (Z)- $\text{CH}_2 = \text{CHCH}_2\text{CH} = \text{CHCH}_3$ | (9.04±0.05) | (227) | (950) | 19 | 80 | *EST | 7318-67-4 |
| (E)- $\text{CH}_2 = \text{CHCH}_2\text{CH} = \text{CHCH}_3$ | (8.98±0.05) | (225) | (940) | 18 | 74 | *EST | 7319-00-8 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------------------------------|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}^+$ | | | | | | | |
| $\text{CH}_2 = \text{CHCH}_2\text{CH}_2\text{CH} = \text{CH}_2$ | 9.29±0.05 | 234 | 980 | 20.1±0.1 | 84.1±0.6 | 77PED/RYL | 592-42-7 |
| $\text{CH}_3\text{CH} = \text{C} = \text{CHC}_2\text{H}_5$ | (8.76±0.05) | (228) | (955) | 26 | 110 | *EST | 592-49-4 |
| (Z),(Z)- $\text{CH}_3\text{CH} = \text{CHCH} = \text{CHCH}_3$ | (8.27) See also: 81MAS/MOU. | (203) | (850) | 12 | 52 | *EST | 6108-61-8 |
| (E),(Z)- $\text{CH}_3\text{CH} = \text{CHCH} = \text{CHCH}_3$ | 8.24±0.02 See also: 81MAS/MOU. | (202) | (844) | 12 | 49 | *EST | 5194-50-3 |
| (E),(E)- $\text{CH}_3\text{CH} = \text{CHCH} = \text{CHCH}_3$ | 8.18±0.06 See also: 81MAS/MOU. | (199) | (832) | 11 | 43 | *EST | 5194-51-4 |
| $\text{CH}_2 = \text{C} = \text{CHCH}(\text{CH}_3)_2$ | (9.06±0.05) | (236) | (987) | 27 | 113 | *EST | 13643-05-5 |
| $\text{CH}_2 = \text{C} = \text{C}(\text{CH}_3)\text{C}_2\text{H}_5$ | (8.74±0.05) | (227) | (951) | 26 | 108 | *EST | 7417-48-3 |
| $(\text{CH}_3)_2\text{C} = \text{CHCH} = \text{CH}_2$ | 8.25 IP from 81MAS/MOU, 82LEV/LIA. | (201) | (839) | 10 | 43 | 70BEN/O'N | 926-56-7 |
| (Z)- $\text{CH}_2 = \text{CHC}(\text{CH}_3) = \text{CHCH}_3$ | 8.42 IP from 81MAS/MOU. | (205) | (859) | 11 | 47 | *EST | 2787-43-1 |
| (E)- $\text{CH}_2 = \text{CHC}(\text{CH}_3) = \text{CHCH}_3$ | (8.38) IP from 81MAS/MOU. | (204) | (852) | 10 | 43 | *EST | 2787-45-3 |
| (E)- $\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH} = \text{CHCH}_3$ | 8.43 IP from 81MAS/MOU. | (205) | (856) | 10 | 43 | *EST | 926-54-5 |
| $\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH}_2\text{CH} = \text{CH}_2$ | (9.16±0.05) | (228) | (956) | 17 | 72 | *EST | 763-30-4 |
| $\text{CH}_2 = \text{CHCH}(\text{CH}_3)\text{CH} = \text{CH}_2$ | (9.40±0.05) | (235) | (985) | 19 | 78 | *EST | 1115-08-8 |
| $(\text{CH}_3)_2\text{C} = \text{C} = \text{CHCH}_3$ | 8.64±0.05 | (222) | (930) | 23 | 96 | *EST | 3043-33-2 |
| $\text{CH}_2 = \text{C}(\text{CH}_3)\text{C}(\text{CH}_3) = \text{CH}_2$ | 8.71 See also: 81MAS/MOU. | 211 | 884 | 10±0.2 | 44±1 | 77PED/RYL | 513-81-5 |

Table 1. Positive Ion Table - Continued


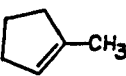
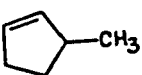
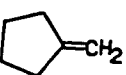
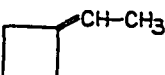
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}^+$ $\text{C}_2\text{H}_5\text{C}(\text{=CH}_2)\text{CH}=\text{CH}_2$ | (8.79±0.02) See also: 81MAS/MOU. | (216) | (904) | 13 | 56 | *EST | 3404-63-5 |
| $\text{C}_4\text{H}_9\text{C}\equiv\text{CH}$ | (9.95±0.05) See 81HOL/FIN. | (258) | (1082) | 29±0.2 | 122±1 | 79ROG/DAG | 693-02-7 |
| $\text{C}_3\text{H}_7\text{C}\equiv\text{CCH}_3$ | 9.366±0.005 | 242 | 1012 | 26±0.5 | 108±2 | 79ROG/DAG | 764-35-2 |
| $\text{C}_2\text{H}_5\text{C}\equiv\text{CC}_2\text{H}_5$ | 9.323±0.005 | 240 | 1005 | 25±0.5 | 106±2 | 79ROG/DAG | 928-49-4 |
| $(\text{CH}_3)_2\text{CHCH}_2\text{C}\equiv\text{CH}$ | (9.83±0.05) | (254) | (1064) | 28 | 116 | *EST | 7154-75-8 |
| $\text{CH}_3\text{CH}_2\text{CH}(\text{CH}_3)\text{C}\equiv\text{CH}$ | 9.79±0.05 | 253 | 1058 | 27±0.2 | 113±1 | 79ROG/DAG | 922-59-8 |
| $(\text{CH}_3)_3\text{CC}\equiv\text{CH}$ | (9.80±0.05) See also: 81CAB/COW, 85ORL/BOG. | (251) | (1051) | 25±0.7 | 106±3 | 77KUP/SHI | 917-92-0 |
| $(\text{CH}_3)_2\text{CHC}\equiv\text{CCH}_3$ | 9.31±0.05 | (238) | (995) | 23 | 97 | *EST | 21020-27-9 |
|  | 8.945±0.01 See also: 81KIM/KAT. | 205.2 | 858.4 | -1.1±0.1 | -4.6±0.5 | 77PED/RYL | 110-83-8 |
|  | 8.55±0.05 | 196 | 821 | -1±0.2 | -4±1 | 82ALL/DOD | 693-89-0 |
|  | 8.95±0.01 | 208 | 871 | 2±0.5 | 7±2 | 79FUC/PEA | 1120-62-3 |
|  | 8.55±0.01 | 200 | 837 | 3±0.5 | 12±2 | 82ALL/DOD | 1528-30-9 |
|  | (8.70±0.05) | (221) | (925) | 21 | 86 | *EST | 1528-21-8 |

Table 1. Positive Ion Table - Continued

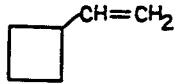
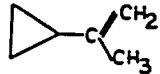
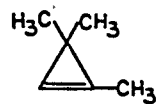



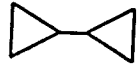

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|---|--------------------------|-----------------|------------------------------|----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}^+$ | | | | | | | | |
|  | | (≤ 9.44) | (≤ 242) | (≤ 1010) | 24 | 99 | *EST | 2597-49-1 |
|  | | (8.66 ± 0.05) | (222) | (930) | 22 | 94 | 82KOZ/MAS | 4663-22-3 |
|  | | (8.58 ± 0.05) See also: 81PLE/VIL. | (239) | (1001) | 41 | 173 | *EST | 3664-56-0 |
|  | | (9.16 ± 0.02) | (220.4) | (922.1) | 9.2 ± 0.1 | 38.3 ± 0.4 | 77PED/RYL | 285-58-5 |
|  | | (9.0) IP is onset of photoelectron band. | (237) | (993) | 30 | 125 | 82WIB/WEN | 186-04-9 |
|  | | (9.7) IP is onset of photoelectron band (84DEL/PIG). | (239) | (1000) | 15.3 | 64.0 | 82WIB/WEN | 285-86-9 |
|  | | (8.9) IP is onset of photoelectron band (82SPA/GLE). | (236) | (988) | 31 ± 1 | 129 ± 4 | 77PED/RYL | 5685-46-1 |
|  | | (9.1) IP is onset of photoelectron band. | (250) | (1045) | 40 | 167 | *EST | 157-45-9 |

Table 1. Positive Ion Table - Continued


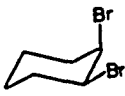
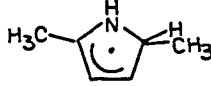

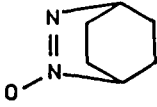
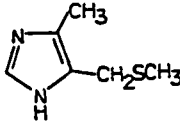
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}\text{Br}_2^+$ | | | | | | | |
|  | 10.02±0.02 | (206) | (863) | -25 | -104 | *EST | 7429-37-0 |
|  | (9.94±0.02) | (204) | (855) | -25 | -104 | *EST | 19246-38-9 |
| $\text{C}_6\text{H}_{10}\text{F}_3\text{O}_2^+$ $\text{CF}_3\text{C}(\text{OH})\text{O}(\text{n-C}_4\text{H}_9)$ | | | | | | | |
| | | -79 | -332 | | | | |
| | | From proton affinity of $\text{CF}_3\text{COO}(\text{n-C}_4\text{H}_9)$ (RN 367-64-6). PA = 185.8 kcal/mol, 777. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{10}\text{N}^+$ | | | | | | | |
|  | | 157 | 656 | | | | |
| | | From proton affinity of 2,5-dimethylpyrrole (RN 625-84-3) (86MAU/LIE). PA = 218.2 kcal/mol, 913. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{10}\text{N}_2^+$ | | | | | | | |
|  | (7.79±0.04) | (218) | (913) | 38 | 161 | 77OTH/OLS | 3310-62-1 |
| $\text{C}_6\text{H}_{10}\text{N}_2\text{O}^+$ | | | | | | | |
|  | (≤9.30±0.03) | (≤237) | (≤990) | 22.08±.44 | 92.38±1.84 | 83BYS | 25926-96-9 |
| $\text{C}_6\text{H}_{10}\text{N}_2\text{S}^+$ | | | | | | | |
|  | (7.9) | (208) | (869) | 26 | 107 | *EST | 75899-43-3 |
| | | IP is onset of photoelectron band (80KLA/BUT). | | | | | |

Table 1. Positive Ion Table - Continued

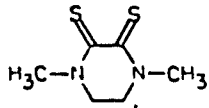
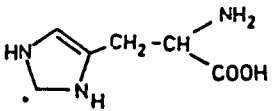

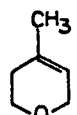
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|-----------------------------------|--------------------------|----------------|------------------------------|--------|----------------------|--|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}\text{N}_2\text{S}_2^+$  | ≤ 7.82 IP from 81HEN/ISA. | (≤ 236) | (≤ 986) | 55 | 232 | *EST | 78134-03-9 |
| $\text{C}_6\text{H}_{10}\text{N}_3\text{O}_2^+$  | | 103 | 431 | - | - | | From proton affinity of L-histidine. PA = 231.9 kcal/mol, 970. kJ/mol. |
| $\text{C}_6\text{H}_{10}\text{O}^+$ (E)-n-C ₃ H ₇ CH=CHCHO | (9.65) | (187) | (782) | -36 | -149 | *EST | 505-57-7 |
| CH ₃ CH ₂ CH=C(CH ₃)CHO | (9.54) | (181) | (758) | -39 | -162 | *EST | 623-36-9 |
| CH ₃ CH=C(C ₂ H ₅)CHO | (9.53) | (181) | (757) | -39 | -162 | *EST | 19780-25-7 |
| iso-C ₃ H ₇ COCH=CH ₂ | (9.39) | (177) | (741) | -39 | -165 | *EST | 1606-47-9 |
| (E)-CH ₃ CH=CHC(=O)C ₂ H ₅ | (9.32) | (175) | (730) | -40 | -169 | *EST | 2497-21-4 |
| CH ₃ CH=C(CH ₃)C(=O)CH ₃ | (9.35) | (172) | (719) | -44 | -183 | *EST | 565-62-8 |
| (CH ₃) ₂ C=CHC(=O)CH ₃ | 9.08±0.03 | (165) | (693) | -44 | -183 | *EST | 141-79-7 |
|  | 9.14±0.01 See also: 86SPA/RAD. | 157 | 656 | -54±0.5 | -226±2 | 77PED/RYL | 108-94-1 |
|  | (8.88) IP from 84ALA/RYE. | (173) | (724) | -32 | -133 | *EST | 2270-61-3 |

Table 1. Positive Ion Table - Continued


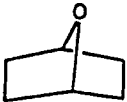
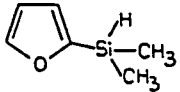
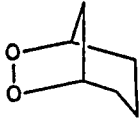

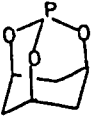
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}\text{O}^+$ | | | | | | | |
|  | (9.82) IP from 84ALA/RYE. | (197) | (822) | -30 | -125 | *EST | 286-20-4 |
|  | ($\leq 9.57 \pm 0.02$) | (≤ 177) | (≤ 740) | -44 | -183 | 74PIH/TAS | 279-49-2 |
| $\text{C}_6\text{H}_{10}\text{OS}^+$ | | | | | | | |
| $\text{CH}_3\text{SC}(\text{CH}_3)=\text{CHC}(=\text{O})\text{CH}_3$ | (8.15) IP is onset of photoelectron band (81JOR/CAR). | (152) | (636) | -36 | -150 | *EST | 60887-86-7 |
| $\text{C}_6\text{H}_{10}\text{OSi}^+$ | | | | | | | |
|  | ≤ 8.62 IP from 83ZYK/ERC. | (≤ 165) | (≤ 689) | -34 | -143 | *EST | 13271-68-6 |
| $\text{C}_6\text{H}_{10}\text{O}_2^+$ | | | | | | | |
| (E)- $\text{CH}_3\text{CH}=\text{CHCOOC}_2\text{H}_5$ | (≤ 10.11) | (≤ 143) | (≤ 599) | -90 ± 0.5 | -376 ± 2 | 77PED/RYL | 623-70-1 |
|  | (8.6) IP is onset of photoelectron band (84GLE/DOB). | (163) | (683) | -35 | -147 | *EST | 51272-66-3 |
|  | 8.4 IP is onset of photoelectron band. | (159) | (663) | -35 | -147 | *EST | 280-53-5 |
| $\text{C}_6\text{H}_{10}\text{O}_3\text{P}^+$ | | | | | | | |
|  | | -14 | -58 | | | | |
| | From proton affinity of 2,8,9-trioxa-1-phosphaadamantane (RN 281-33-4). PA = 213.8 kcal/mol, 894. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

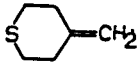

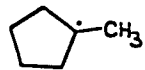
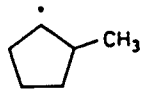
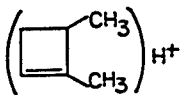
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{10}\text{O}_4^+$ <chem>C2H5OCOCOCOC2H5</chem> | (9.8) | (49) | (206) | -177±2 | -740±9 | 77PED/RYL | 95-92-1 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_6\text{H}_{10}\text{S}^+$  | 9.22 | (221) | (923) | 8 | 34 | *EST | 50550-56-6 |
| | IP from 80SAR/WOR. | | | | | | |
| $\text{C}_6\text{H}_{11}^+$ <chem>CH3CH=CHC(CH3)2</chem> | | (169) | (706) | | | | |
| | From proton affinity of <chem>CH3CH=CHC(CH3)2</chem> (RN 1118-58-7). PA = (207.9) kcal/mol, (870.) kJ/mol. | | | | | | |
| <chem>C2H5C(CH3)CH=CH2</chem> | | (170) | (712) | | | | |
| | From proton affinity of <chem>CH3CH=C(CH3)CH=CH2</chem> (RN 4549-74-0). PA = (205.7) kcal/mol, (860.6) kJ/mol. | | | | | | |
| <chem>(CH3)2CC(CH3)=CH2</chem> | | (174) | (728) | | | | |
| | From proton affinity of <chem>CH2=C(CH3)C(CH3)=CH2</chem> (RN 513-81-5). PA = (202.1) kcal/mol, (846.) kJ/mol. | | | | | | |
|  | | (175) | (733) | 18 | 77 | 81TSA | 3170-58-9 |
| | From proton affinity of cyclohexene (RN 110-83-8). PA = (189) kcal/mol, (792) kJ/mol. | | | | | | |
|  | | 167 | 698 | | | | |
| | From proton affinities of methylenecyclopentane (RN 1528-30-9) PA = 200.8 kcal/mol, 840. kJ/mol and 1-methylcyclopentene (RN 693-89-0), PA = 196.9 kcal/mol, 824. kJ/mol, and from hydride and chloride transfer equilibrium constant determinations (76SOL/FIE, 76GOR/MUN, 85SHA/SHA). | | | | | | |
|  | | (179) | (747) | | | | |
| | From appearance potential measurements (81HER/SIC). | | | | | | |
|  | | (182) | (762) | | | | |
| | From proton affinity of 1,2-dimethylcyclobutene. (RN 1501-58-2). PA = (201) kcal/mol, (841) kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

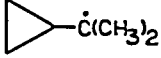
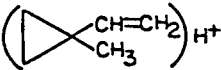
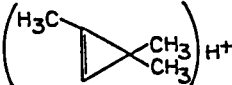
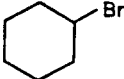
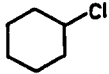
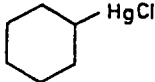
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{11}^+$ | | | | | | | |
|  | | 179 | 750 | | | | |
| | | From proton affinity of 2-cyclopropylpropene (RN 4663-22-3). PA = 209.0 kcal/mol, 874. kJ/mol. | | | | | |
|  | | (181) | (756) | | | | |
| | | From proton affinity of 1-methyl-1-vinylcyclopropane (RN 16906-27-7). PA = (206) kcal/mol, (862) kJ/mol. | | | | | |
|  | | (193) | (808) | | | | |
| | | From proton affinity of 1,3,3-trimethylcyclopropane. (RN 3664-56-0). PA = (214) kcal/mol, (895) kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{11}\text{Br}^+$ | | | | | | | |
|  | (9.85±0.01) | (200) | (835) | -27 | -115 | *EST | 108-85-0 |
| $\text{C}_6\text{H}_{11}\text{Cl}^+$ | | | | | | | |
|  | (10.10±0.01) | (194) | (810) | -39±1 | -164±4 | 77PED/RYL | 542-18-7 |
| $\text{C}_6\text{H}_{11}\text{ClHg}^+$ | | | | | | | |
|  | 9.2 | (188) | (787) | -24 | -101 | *EST | 24371-94-6 |
| | | IP is onset of photoelectron band (81BAI/CHI2). | | | | | |
| $\text{C}_6\text{H}_{11}\text{F}_3\text{NO}^+$ | | | | | | | |
| $\text{CF}_3\text{C}(\text{OH})\text{NH}(\text{n-C}_4\text{H}_9)$ | | -54 | -226 | | | | |
| | | From proton affinity of $\text{CF}_3\text{CONH}(\text{n-C}_4\text{H}_9)$ (RN 400-59-9). PA = 203.6 kcal/mol, 852. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

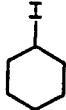
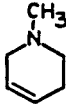
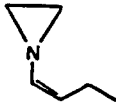
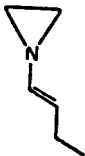
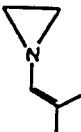
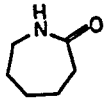
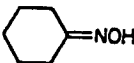
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{11}\text{I}^+$ |  | 9.003 | 195 | 818 | -12±1 | -51±4 | 77PED/RYL | 626-62-0 |
| $\text{C}_6\text{H}_{11}\text{N}^+$ | | | | | | | | |
| (E)- $\text{CH}_3\text{CH}=\text{CHCH}=\text{NC}_2\text{H}_5$ | | (8.9) IP is onset of photoelectron band. | (225) | (941) | 20±1 | 82±6 | *EST | 3653-19-8 |
| $(\text{CH}_2=\text{CHCH}_2)_2\text{NH}$ | | (8.2) IP is onset of photoelectron band. | (224) | (937) | 35±1 | 146±6 | *EST | 124-02-7 |
|  | | (≤8.67±0.05) | (≤219) | (≤914) | 19±2 | 78±10 | *EST | 694-55-3 |
|  | | (7.9) IP from onset of photoelectron band (81MUL/PRE). | (220) | (919) | 38 | 157 | *EST | |
|  | | (7.7) IP is onset of photoelectron band (81MUL/PRE, 81MUL/PRE2). | (215) | (900) | 38 | 157 | *EST | |
|  | | (7.6) IP is onset of photoelectron band (81MUL/PRE, 81MUL/PRE2). | (220) | (919) | 44 | 186 | *EST | |
| $\text{C}_6\text{H}_{11}\text{NO}^+$ | | | | | | | | |
|  | | (9.07±0.02) | (150) | (629) | -58.8±0.3 | -246.2±1.2 | 77PED/RYL | 105-60-2 |
|  | | (8.97±0.03) IP from 79GOL/KUL. | (186) | (779) | -21 | -86 | *EST | 100-64-1 |

Table 1. Positive Ion Table - Continued

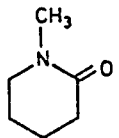
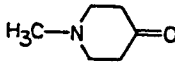

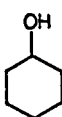
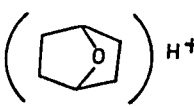
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|-----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{11}\text{NO}^+$ | | | | | | | |
|  | ≤ 8.92 IP from 85TRE/RAD. | ≤ 149 | ≤ 624 | -57 ± 0.7 | -237 ± 3 | 77PED/RYL | 931-20-4 |
|  | (8.3) IP from 80SAR/WOR. See also: 86SPA/RAD. | (155) | (648) | -37 | -153 | *EST | 1445-73-4 |
| $\text{C}_6\text{H}_{11}\text{NOS}^+$ | | | | | | | |
|  | (≤ 10.0) | (≤ 169) | (≤ 707) | -62 | -258 | *EST | 30980-11-1 |
| $\text{C}_6\text{H}_{11}\text{O}^+$ | | | | | | | |
| $(\text{CH}_3)_2\text{CCHC}(\text{OH})\text{CH}_3$ | | (112) | (470) | | | | |
| | From proton affinity of $(\text{CH}_3)_2\text{C}=\text{CHC}(\text{=O})\text{CH}_3$ (RN 141-79-7). PA = (210) kcal/mol, (877) kJ/mol. | | | | | | |
| $(\text{CH}_2\text{CHCH}_2)_2\text{OH}$ | | 158 | 661 | | | | |
| | From proton affinity of $(\text{CH}_2=\text{CHCH}_2)_2\text{O}$ (RN 557-40-4). PA = 200.4 kcal/mol, 838. kJ/mol. | | | | | | |
|  | | 111 | 466 | | | | |
| | From proton affinity of cyclohexanone (RN 108-94-1) (86SAN/BAL). PA = 201.4 kcal/mol, 843. kJ/mol. | | | | | | |
|  | | 119 | 498 | | | | |
| | From proton affinity of bicyclo[2.2.1]heptane, 7-oxa- (RN 279-49-2). PA = 203 kcal/mol, 849 kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{11}\text{O}_2^+$ | | | | | | | |
| $\text{CH}_3\text{C}(\text{OH})\text{CH}_2\text{CH}_2\text{COCH}_3$ | | 64 | 269 | | | 83MAU | |
| | From proton affinity of $\text{CH}_3\text{COCH}_2\text{CH}_2\text{COCH}_3$ (RN 110-13-4). PA = 213.2 kcal/mol, 892. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{11}\text{P}^+$ | | | | | | | |
| $(\text{CH}_3)_3\text{CC}=\text{CPH}_2$ | ≤ 9.05 IP from 81CAB/COW. | (≤ 246) | (≤ 1028) | 37 | 155 | *EST | 77376-07-9 |

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{12}^+$ | | | | | | | |
| 1- C_6H_{12} | 9.44±0.04 | 207.7 | 869.0 | -10.0±0.2 | -41.8±1 | 81WIB/WAS | 592-41-6 |
| (Z)-2- C_6H_{12} | (8.97±0.01) | (195.5) | (817.8) | -11.4±0.2 | -47.7±1 | 81WIB/WAS | 7688-21-3 |
| (E)-2- C_6H_{12} | (8.97±0.01) | (194.5) | (814.0) | -12.3±0.2 | -51.5±1 | 81WIB/WAS | 4050-45-7 |
| (Z)-3- C_6H_{12} | (8.95±0.01) | (195.2) | (816.7) | -11.2±0.2 | -46.8±0.8 | 81WIB/WAS | 7642-09-3 |
| (E)-3- C_6H_{12} | 8.96±0.02 | 194.5 | 813.9 | -12.1±0.2 | -50.6±1 | 81WIB/WAS | 13269-52-8 |
| $\text{C}_2\text{H}_5\text{CH}_2\text{C}(\text{CH}_3)=\text{CH}_2$ | (9.08±0.01) | (195) | (817) | -14.2±0.3 | -59.4±1 | 77PED/RYL | 763-29-1 |
| $\text{C}_2\text{H}_5\text{CH}(\text{CH}_3)\text{CH}=\text{CH}_2$ | (9.44) IP from 81HOL/FIN. | (206) | (861) | -11.8±0.4 | -49.5±1.5 | 77PED/RYL | 760-20-3 29564-68-9 |
| $(\text{CH}_3)_2\text{CHCH}_2\text{CH}=\text{CH}_2$ | (9.45±0.01) | (206) | (861) | -12±0.5 | -51±2 | 77PED/RYL | 691-37-2 |
| $(\text{C}_2\text{H}_5)_2\text{C}=\text{CH}_2$ | (9.06±0.02) | (196) | (818) | -13.4±0.3 | -56.0±1 | 77PED/RYL | 760-21-4 |
| $(\text{CH}_3)_2\text{CHC}(\text{CH}_3)=\text{CH}_2$ | (9.07±0.01) | (194) | (812) | -15.1±0.2 | -63.3±0.8 | 77PED/RYL | 563-78-0 |
| $(\text{CH}_3)_3\text{CCH}=\text{CH}_2$ | 9.45±0.01 | 203 | 851 | -14.5±0.2 | -60.7±0.9 | 77PED/RYL | 558-37-2 |
| (Z)- $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{C}_2\text{H}_5$ | (8.58) IP from 81HOL/FIN. | (183) | (766) | -14.9±0.4 | -62.3±1 | 77PED/RYL | 922-61-2 |
| (Z)- $(\text{CH}_3)_2\text{CHCH}=\text{CHCH}_3$ | (8.98±0.01) | (193) | (809) | -13.7±0.2 | -57.5±1 | 77PED/RYL | 691-38-3 |
| (E)- $(\text{CH}_3)_2\text{CHCH}=\text{CHCH}_3$ | (8.97±0.01) | (192) | (804) | -14.7±0.3 | -61.5±1 | 77PED/RYL | 674-76-0 |
| $(\text{CH}_3)_2\text{C}=\text{CHC}_2\text{H}_5$ | (8.58) IP from 81HOL/FIN. | (182) | (761) | -16.0±0.3 | -66.8±1 | 77PED/RYL | 625-27-4 |
| $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$ | 8.27±0.01 | 174 | 729 | -16.6±0.2 | -69.3±0.8 | 77PED/RYL | 563-79-1 |
|  | 9.86±0.03 | 198 | 828 | -29.5±0.1 | -123.3±0.3 | 77PED/RYL | 110-82-7 |
| | From charge transfer equilibria relative to fluorobenzenes; data re-interpreted. (82SIE/MAU; 82LIA). Threshold measurement leads to IP = 9.88 eV. See also: 81KIM/KAT. | | | | | | |

Table 1. Positive Ion Table - Continued

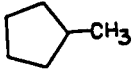

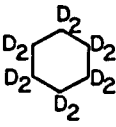
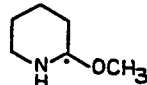
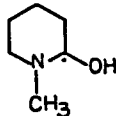
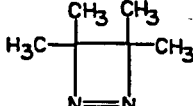
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{12}^+$ | | | | | | | |
|  | 9.85±0.03 | 202 | 844 | -25.3±0.1 | -105.9±0.4 | 77PED/RYL | 96-37-7 |
| | From charge transfer equilibrium constant relative to cyclohexane (76LIA/AUS). | | | | | | |
|  | (8.90) | (197) | (825) | -8.0 | -33.5 | *EST | |
| | IP from 85LAD/HAR. | | | | | | |
| $\text{C}_6\text{D}_{12}^+$ | | | | | | | |
|  | 9.89 | | | | | | 1735-17-7 |
| | From charge transfer equilibria relative to fluorobenzenes; data re-interpreted. (82SIE/MAU; 82LIA). Threshold measurement leads to IP = 9.88 eV. | | | | | | |
| $\text{C}_6\text{H}_{12}\text{N}^+$ | | | | | | | |
| $(\text{CH}_2\text{CHCH}_2)_2\text{NH}_2$ | | 175 | 735 | | | | |
| | From proton affinity of $(\text{CH}_2=\text{CHCH}_2)_2\text{NH}$ (RN 124-02-7). PA = 224.7 kcal/mol, 940. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{12}\text{NO}^+$ | | | | | | | |
|  | | 96 | 400 | | | | |
| | From proton affinity of 2,3,4,5-tetrahydro-6-methoxypyridine (RN 53687-79-9). PA = 228.1 kcal/mol, 954. kJ/mol. | | | | | | |
|  | | 90 | 376 | | | | |
| | From proton affinity of 1-methylpiperidine-2-one (RN 931-20-4). PA = 219.3 kcal/mol, 917.5 kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{12}\text{NO}_3^+$ | | | | | | | |
| $\text{CH}_3\text{C}(\text{OH})\text{NHCH}(\text{CH}_3)\text{COOCH}_3$ | | -4 | -18 | | | | |
| | From proton affinity of $\text{CH}_3\text{CONHCH}(\text{CH}_3)\text{COOCH}_3$. (RN 3619-02-1). PA = 224.5 kcal/mol, 939. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{12}\text{N}_2^+$ | | | | | | | |
|  | (8.2) | (225) | (941) | 36±0.7 | 150±3 | 80ENG | 54166-22-2 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

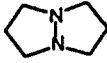
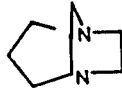

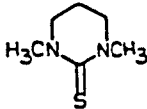
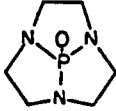
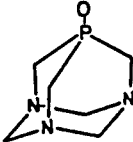
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{12}\text{N}_2^+$ | | | | | | | |
|  | (7.87) IP from 82LEV/LIA. See also: 84NEL. | (219) | (915) | 37 | 156 | *EST | 5397-67-1 |
|  | ≤ 8.24 | (≤ 210) | (≤ 878) | 20 | 83 | *EST | 280-28-4 |
|  | 7.197 ± 0.001 IP from 84SMI/HAG2. | 187 | 784 | 21 ± 2 | 89 ± 7 | 71RAP/WES | 280-57-9 |
| $\text{C}_6\text{H}_{12}\text{N}_2\text{O}_2^+$ (CH_3) ₂ NCOCON(CH_3) ₂ | 9.02 IP from 82LEV/LIA, 85ROT/BOC. | (132) | (554) | -76 | -316 | *EST | 1608-14-6 |
| $\text{C}_6\text{H}_{12}\text{N}_2\text{S}^+$ | | | | | | | |
|  | (7.3) IP is onset of photoelectron band. | (192) | (802) | 23 | 98 | *EST | 16597-35-6 |
| $\text{C}_6\text{H}_{12}\text{N}_2\text{S}_2^+$ (CH_3) ₂ NC(=S)C(=S)N(CH_3) ₂ | ≤ 7.75 IP from 81HEN/ISA. | (≤ 222) | (≤ 930) | 43 | 182 | *EST | 35840-78-9 |
| $\text{C}_6\text{H}_{12}\text{N}_3\text{OP}^+$ | | | | | | | |
|  | ≤ 8.89 IP from 82COW/LAT. | (≤ 190) | (≤ 794) | -15 | -64 | *EST | 71771-37-4 |
|  | $\leq 8.19 \pm 0.10$ IP from 82COW/LAT. | (≤ 151) | (≤ 631) | -38 | -159 | *EST | 53597-70-9 |

Table 1. Positive Ion Table - Continued

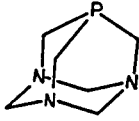
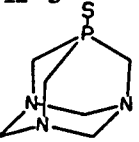
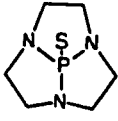
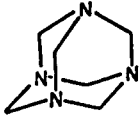
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|-----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{12}\text{N}_3\text{P}^+$  | $\leq 8.05 \pm 0.10$ IP from 82COW/LAT. | (≤ 227) | (≤ 952) | 42 | 175 | *EST | 53597-69-6 |
| $\text{C}_6\text{H}_{12}\text{N}_3\text{PS}^+$  | $\leq 8.02 \pm 0.10$ IP from 82COW/LAT. | (≤ 205) | (≤ 857) | 20 | 83 | *EST | 56796-56-6 |
|  | $\leq 8.43 \pm 0.10$ IP from 82COW/LAT. | (≤ 237) | (≤ 991) | 43 | 178 | *EST | |
| $\text{C}_6\text{H}_{12}\text{N}_4^+$  | (≤ 8.53) See also: 82COW/LAT. | (≤ 244) | (≤ 1022) | 47 ± 0.7 | 199 ± 3 | 77PED/RYL | 100-97-0 |
| $\text{C}_6\text{H}_{12}\text{O}^+$ | | | | | | | |
| n-C ₅ H ₁₁ CHO | 9.67 ± 0.05 | 164 | 686 | -59 | -247 | 78TRC | 66-25-1 |
| n-C ₃ H ₇ CH(CH ₃)CHO | (9.70) | (163) | (679) | -61 | -257 | *EST | 123-15-9 |
| (C ₂ H ₅) ₂ CHCHO | (9.54) IP from 81HOL/FIN. | (158) | (663) | -61 | -257 | *EST | 97-96-1 |
| C ₂ H ₅ CH(CH ₃)CH ₂ CHO | (9.68) IP from 81HOL/FIN. | (161) | (676) | -62 | -258 | *EST | 15877-57-3 |
| neo-C ₅ H ₁₁ CHO | 9.61 ± 0.01 | (158) | (658) | -64 | -269 | *EST | 2987-16-8 |
| n-C ₄ H ₉ COCH ₃ | 9.35 ± 0.02 | 150 | 624 | -66 ± 0.2 | -278 ± 1 | 77PED/RYL | 591-78-6 |
| n-C ₃ H ₇ COC ₂ H ₅ | 9.12 ± 0.02 See also: 81HOL/FIN. | 143 | 601 | -67 ± 0.2 | -279 ± 1 | 77PED/RYL | 589-38-8 |
| sec-C ₄ H ₉ COCH ₃ | 9.21 ± 0.01 IP from 81HOL/FIN, 82LEV/LIA, 84BOU/FLA. | (144) | (602) | -69 | -287 | *EST | 565-61-7 |
| iso-C ₄ H ₉ COCH ₃ | 9.30 ± 0.01 | (145) | (610) | -69 | -287 | *EST | 108-10-1 |

Table 1. Positive Ion Table - Continued

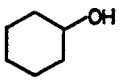

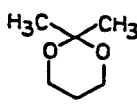
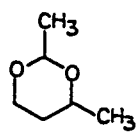
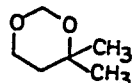
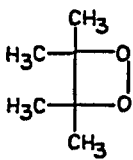
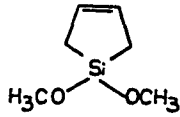
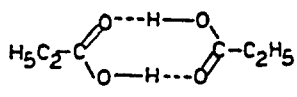
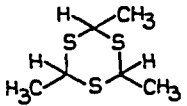
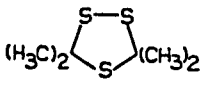
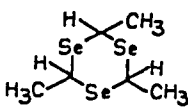
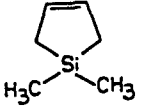
| ION | Ionization potential | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-------------------------------|--------------------------|---------|------------------------------|------------|-------------------|---------------------|
| | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{12}\text{O}^+$ | | | | | | | |
| tert- $\text{C}_4\text{H}_9\text{COCH}_3$ | 9.11±0.02 | 141 | 589 | -69.3±0.2 | -289.8±0.9 | 77PED/RYL | 75-97-8 |
| iso- $\text{C}_3\text{H}_7\text{COC}_2\text{H}_5$ | 9.10±0.01 | 141 | 592 | -68.3±0.2 | -286.1±0.9 | 77PED/RYL | 565-69-5 |
|  | (9.75) IP from 83RAB/SEL. | (155.5) | (650.7) | -69.3±0.2 | -290.0±0.9 | 85WIB/WAS | 108-93-0 |
| $\text{C}_6\text{H}_{12}\text{O}_2^+$ | | | | | | | |
| $\text{CH}_3(\text{CH}_2)_4\text{COOH}$ | ≤10.12 IP from 81HOL/FIN. | ≤111 | ≤463 | -122.8±0.4 | -513.6±1.6 | 77PED/RYL | 142-62-1 |
| $\text{CH}_3\text{COO}(\text{CH}_2)_3\text{CH}_3$ | 10.0 | 114 | 479 | -116.1±0.1 | -485.6±0.5 | 77PED/RYL | 123-86-4 |
| $\text{CH}_3\text{COOCH}(\text{CH}_3)\text{C}_2\text{H}_5$ | 9.90 IP from 82GRE/MCC. | 109 | 454 | -120 | -501 | 82GRE/MCC | 105-46-4 |
| $\text{CH}_3(\text{CH}_2)_3\text{COOCH}_3$ | (10.4±0.2) | (127) | (532) | -112.7±0.3 | -471.5±1.4 | 77PED/RYL | 624-24-8 |
| tert- $\text{C}_4\text{H}_9\text{COOCH}_3$ | (9.90±0.04) | (111) | (464) | -117±0.2 | -491±1 | 77PED/RYL | 598-98-1 |
|  | (≤9.29) | (≤178) | (≤746) | -36 | -150 | *EST | 6572-89-0 |
|  | ≤9.84 IP from 84ASF/ZYK. | (≤124) | (≤519) | -103 | -430 | 77PED/RYL | 695-30-7 |
|  | (≤9.90) IP from 84ASF/ZYK. | (≤127) | (≤530) | -102±1 | -425±4 | 77PED/RYL | 766-20-1 |
|  | ≤9.80 IP from 84ASF/ZYK. | (≤124) | (≤521) | -102 | -425 | *EST | 766-15-4 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{12}\text{O}_2^+$  | (8.53) | (156) | (653) | -41 | -170 | 78GRE/LIE | 35856-82-7 |
| $\text{C}_6\text{H}_{12}\text{O}_2\text{Si}^+$  | ≤ 9.59 IP from 81KHV/ZYK. | (≤ 39) | (≤ 163) | -182 | -762 | *EST | 61667-33-2 |
| $\text{C}_6\text{H}_{12}\text{O}_4^+$  | (≤ 10.4) | (≤ 10) | (≤ 40) | -230 | -963 | *EST | |
| $\text{C}_6\text{H}_{12}\text{S}_3^+$  | (8.0) IP is onset of photoelectron band. | (178) | (746) | -6 | -26 | *EST | 2765-04-0 |
|  | 8.0 IP is onset of photoelectron band. | (151) | (633) | -33 | -139 | *EST | 38348-31-1 |
| $\text{C}_6\text{H}_{12}\text{Se}_3^+$  | (7.7) IP is onset of photoelectron band (84BOC/AYG). | (211) | (882) | 33 | 139 | *EST | 15732-69-1 |
| $\text{C}_6\text{H}_{12}\text{Si}^+$  | ≤ 9.0 | (≤ 182) | (≤ 760) | -26 | -108 | *EST | 16054-12-9 |
| $\text{C}_6\text{H}_{13}^+$ $1\text{-C}_6\text{H}_{13}$ | 7.92 ± 0.06 | (191) | (800) | 8 | 33 | *EST | 2679-29-0 |

$\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 100.5$ kcal/mol.

Table 1. Positive Ion Table - Continued


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{13}^+$ 2- C_6H_{13} | 7.0 | (168) | (704) | 7 | 29 | *EST | 2493-44-9 |
| | $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 99$ kcal/mol. | | | | | | |
| n- $\text{C}_3\text{H}_7\text{C}(\text{CH}_3)_2$ | | 152 | 636 | | | | 21058-26-4 |
| | From hydride transfer equilibrium constant (75SOL/FIE and 76GOR/MUN); Heat of formation relative to $\Delta_f H(\text{tert-C}_4\text{H}_9^+)$. | | | | | | |
| $(\text{CH}_3)_2\text{CHC}(\text{CH}_3)_2$ | | 150 | 628 | | | | 24436-98-4 |
| | From hydride transfer equilibrium constant (75SOL/FIE and 76GOR/MUN); Heat of formation relative to $\Delta_f H(\text{tert-C}_4\text{H}_9^+)$. | | | | | | |
| $(\text{C}_2\text{H}_5)_2(\text{CH}_3)\text{C}$ | | 152 | 638 | | | | 23088-03-1 |
| | From proton affinity of $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{C}_2\text{H}_5$. (RN 922-61-2). PA = 198.2 kcal/mol, 829. kJ/mol. | | | | | | |
|  H^+ | | (167) | (700) | | | | |
| | From proton affinity of cyclohexane. (RN 110-82-7). PA = (169) kcal/mol, (707) kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{13}\text{ClHg}^+$ n- $\text{C}_6\text{H}_{13}\text{HgCl}$ | ≤ 9.96 | (≤ 194) | (≤ 811) | -36 | -150 | *EST | 17774-09-3 |
| | IP from 81BAI/CHI2. | | | | | | |
| $\text{C}_6\text{H}_{13}\text{I}^+$ n- $\text{C}_6\text{H}_{13}\text{I}$ | 9.179 | 190 | 794 | -22 | -92 | 81HOL/FIN | 638-45-9 |
| $\text{C}_6\text{H}_{13}\text{N}^+$ n- $\text{C}_3\text{H}_7\text{CH}=\text{NC}_2\text{H}_5$ | (9.00) | (203) | (847) | -5 | -21 | *EST | 1611-12-7 |
| | See also: 79AUE/BOW. | | | | | | |
| (iso- C_3H_7) $\text{CH}=\text{NC}_2\text{H}_5$ | (8.7) | (192) | (805) | -8 | -34 | *EST | 1743-56-2 |
| | IP is onset of photoelectron band. | | | | | | |
| n- $\text{C}_3\text{H}_7\text{N}=\text{CHCH}_2\text{CH}_3$ | (8.55 \pm 0.2) | (192) | (802) | -5 | -23 | *EST | 7707-70-2 |
| n- $\text{C}_3\text{H}_7\text{N}=\text{C}(\text{CH}_3)_2$ | (8.31 \pm 0.2) | (178) | (742) | -14 \pm 2 | -60 \pm 8 | *EST | 22023-64-9 |
| iso- $\text{C}_3\text{H}_7\text{N}=\text{CHCH}_2\text{CH}_3$ | (8.50 \pm 0.2) | (186) | (780) | -10 | -40 | 69BEN/CRU | 28916-23-6 |
| $(\text{CH}_3)_2\text{NCH}=\text{CHC}_2\text{H}_5$ | ≤ 7.57 | (≤ 174) | (≤ 730) | 0 | 0 | *EST | 14548-12-0 |
| | IP from 81MUL/PRE2. | | | | | | |

Table 1. Positive Ion Table - Continued

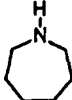
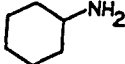
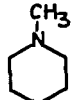
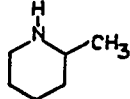
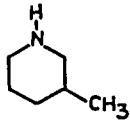
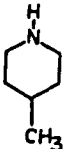

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------------------|---|---|--------------------------|--------------|------------------------------|-----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{13}\text{N}^+$ | $(\text{CH}_3)_2\text{NCH}=\text{C}(\text{CH}_3)_2$ | ≤ 8.15 IP from 81MUL/PRE2. | (≤ 189) | (≤ 791) | 1 | 5 | *EST | 6906-32-7 |
| |  | $(\leq 8.41 \pm 0.02)$ | (≤ 183) | (≤ 767) | -10 | -44 | *EST | 111-49-9 |
| |  | (8.62 ± 0.24) See also: 79AUE/BOW. | (174) | (727) | -25 ± 0.2 | -105 ± 1 | 79STE | 108-91-8 |
| |  | 7.74 See also: 82ROZ/HOU, 80SAR/WOR, 86SPA/RAD, 86CAU/DIV. | (166) | (697) | -12 ± 1 | -50 ± 4 | *EST | 626-67-5 |
| |  | 7.76 ± 0.05 See also: 82ROZ/HOU. | 159 | 664 | -20.2 ± 0.2 | -84.4 ± 1.0 | 77PED/RYL | 109-05-7 |
| |  | 7.94 ± 0.05 See also: 82ROZ/HOU. | (164) | (685) | -19 ± 0.4 | -81 ± 2 | *EST | 626-56-2 |
| |  | 8.01 ± 0.05 See also: 82ROZ/HOU. | (166) | (692) | -19 ± 0.4 | -81 ± 2 | *EST | 626-58-4 |
| $\text{C}_6\text{H}_{13}\text{NO}^+$ |  | (≤ 9.49) | (≤ 186) | (≤ 777) | -33 | -139 | *EST | 6982-39-4 |
| | $\text{CH}_3\text{CON}(\text{C}_2\text{H}_5)_2$ | (8.60 ± 0.02) | (130) | (543) | -69 | -287 | *EST | 685-91-6 |

Table 1. Positive Ion Table - Continued

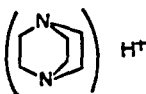
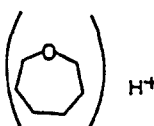
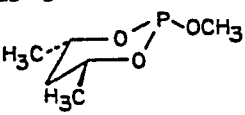
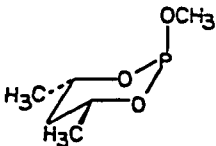
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{13}\text{NO}_2^+$ n- $\text{C}_4\text{H}_9\text{CH}(\text{NH}_2)\text{COOH}$ | (8.52) | (82) | (343) | -114±2 | -479±10 | *EST | 327-57-1 |
| sec- $\text{C}_4\text{H}_9\text{CH}(\text{NH}_2)\text{COOH}$ | (8.66) | (83) | (349) | -116±2 | -487±10 | *EST | 73-32-5 |
| iso- $\text{C}_4\text{H}_9\text{CH}(\text{NH}_2)\text{COOH}$ | (8.51) | (80) | (333) | -117±0.7 | -488±3 | 77PED/RYL | 61-90-5 |
| $\text{C}_6\text{H}_{13}\text{N}_2^+$  H^+ | | 158 | 661 | | | | |
| | | From proton affinity of 1,4-diazabicyclo[2.2.2]octane (RN 280-57-9). PA = 229.0 kcal/mol, 958. kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{13}\text{O}^+$ t- $\text{C}_4\text{H}_9\text{C}(\text{OH})\text{CH}_3$ | | 94 | 394 | | | | |
| | | From proton affinity of t- $\text{C}_4\text{H}_9\text{COCH}_3$ (RN 75-97-8). PA = 202.3 kcal/mol, 846. kJ/mol. | | | | | |
|  H^+ | | (161) | (674) | | | | |
| | | From proton affinity of oxepane (RN 592-90-5). PA = (202) kcal/mol, (845) kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{13}\text{O}_2^+$ t- $\text{C}_4\text{H}_9\text{C}(\text{OH})\text{OCH}_3$ | | 46 | 191 | | | | |
| | | From proton affinity of t- $\text{C}_4\text{H}_9\text{COOCH}_3$ (RN 598-98-1). PA = 202.8 kcal/mol, 848.5 kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{13}\text{O}_3\text{P}^+$  | (8.34±0.1) | (11) | (45) | -182 | -760 | *EST | 7735-82-2 |
|  | (8.69±0.1) | (19) | (78) | -182 | -760 | *EST | 41821-91-4 |

Table 1. Positive Ion Table - Continued

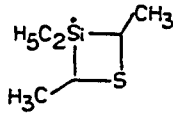
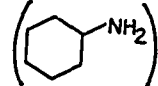
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|--------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{13}\text{SSi}^+$ | | (162) | (679) | | | | 79126-87-7 |
|  | | $\Delta_f H(\text{Ion})$ from appearance potential determination (81GUS/VOL). | | | | | |
| $\text{C}_6\text{H}_{14}^+$ | | | | | | | |
| n- C_6H_{14} | 10.13 | 194 202 | 810 847 | -39.9±0.1 -31.1±0.1 | -167.1±0.4 -130.1±0.4 | 74SCO | 110-54-3 |
| | | From charge transfer equilibrium constant determinations relative to fluorobenzenes; data re-interpreted (81MAU/SIE; 82LLA). Threshold measurement leads to IP = 10.2 eV. | | | | | |
| $(\text{CH}_3)_2\text{CH}(\text{CH}_2)_2\text{CH}_3$ | (10.12) | (191) (201±0.2) | (802) (842±0.9) | -41.6±0.2 -32.2±0.2 | -173.8±0.9 -134.6±0.9 | 74SCO | 107-83-5 |
| $(\text{C}_2\text{H}_5)_2\text{CHCH}_3$ | (10.08) | (191) (201) | (801) (841) | -40.9±0.2 -31.5±0.2 | -171.3±0.9 -131.9±0.9 | 74SCO | 96-14-0 |
| $(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)_2$ | (10.02) | (189) (199) | (791) (832) | -42.1±0.2 -32.3±0.2 | -176.2±0.9 -135.1±0.9 | 74SCO | 79-29-8 |
| $(\text{CH}_3)_3\text{CCH}_2\text{CH}_3$ | (10.06) | (188) (198) | (787) (827) | -43.9±0.2 -34.3±0.2 | -183.9±0.9 -143.5±0.9 | 74SCO | 75-83-2 |
| $\text{C}_6\text{H}_{14}\text{Hg}^+$ | | | | | | | |
| (n- C_3H_7) $_2\text{Hg}$ | (≤8.29) | (≤200) | (≤836) | 9±2 | 36±6 | 77PED/RYL | 628-85-3 |
| (iso- C_3H_7) $_2\text{Hg}$ | (≤8.03) | (≤195) | (≤815) | 10±1 | 40±6 | 77PED/RYL | 1071-39-2 |
| $\text{C}_6\text{H}_{14}\text{N}^+$ | | | | | | | |
| n- $\text{C}_3\text{H}_7\text{CHNHC}_2\text{H}_5$ | | (135) | (566) | | | | |
| | | From proton affinity of n- $\text{C}_3\text{H}_7\text{CH}=\text{NC}_2\text{H}_5$ (RN 1611-12-7). PA = (225.3) kcal/mol, (943) kJ/mol. | | | | | |
| $(\text{CH}_3)_2\text{NC}(\text{CH}_3)\text{CH}_2\text{CH}_3$ | | 129 | 539 | | | | |
| | | From proton affinity of $(\text{CH}_3)_2\text{NC}(\text{CH}_3)=\text{CHCH}_3$ (RN 52113-79-8). PA = 237 kcal/mol, 992 kJ/mol. | | | | | |
|  H^+ | | 120 | 500 | | | | |
| | | From proton affinity of cyclohexanamine (RN 108-91-8). PA = 221.2 kcal/mol, 925.5 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

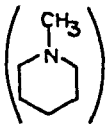
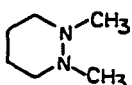
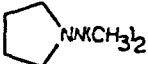
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{14}\text{N}^+$  H^+ | | 124 | 519 | | | | |
| | | From proton affinity of 1-methylpiperidine (RN 626-67-5). PA = 229.7 kcal/mol, 961 kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{14}\text{NO}_2^+$ L- $\text{C}_2\text{H}_5\text{CH}(\text{CH}_3)\text{CH}(\text{NH}_3)\text{COOH}$ | | 30 | 127 | | | | |
| | | From proton affinity of L- $\text{C}_2\text{H}_5\text{CH}(\text{CH}_3)\text{CH}(\text{NH}_2)\text{COOH}$ (RN 73-32-5). PA = 218.9 kcal/mol, 916. kJ/mol. | | | | | |
| L- $(\text{CH}_3)_2\text{CHCH}_2\text{CH}(\text{NH}_3)\text{COOH}$ | | 31 | 130 | | | | |
| | | From proton affinity of L- $(\text{CH}_3)_2\text{CHCH}_2\text{CH}(\text{NH}_2)\text{COOH}$ (RN 61-90-5). PA = 218.1 kcal/mol, 912.5 kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{14}\text{N}_2^+$ (E)- $(\text{C}_3\text{H}_7)_2\text{NN}$ | (8.1) | (199) | (833) | 12±1 | 51±4 | 80ENG | 55204-42-7 |
| | | IP is onset of photoelectron band. | | | | | |
| (Z)-iso- $(\text{C}_3\text{H}_7)_2\text{NN}$ | (≤8.24) | (≤210) | (≤879) | 20 | 84 | *EST | 23201-84-5 |
| (E)-(iso- $\text{C}_3\text{H}_7)_2\text{NN}$ | (8.0) | (193) | (808) | 9±0.5 | 36±2 | 80ENG | 15464-00-3 |
| | | IP is onset of photoelectron band. | | | | | |
|  | 6.54 | (167) | (699) | 16 | 68 | *EST | 26163-37-1 |
| | | IP from charge transfer equilibrium constant determinations (84MAU/NEL). Reference standard: IP $(\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2 = 7.12 \text{ eV.}$ See also: 82LEV/LIA, 84NEL, 80SCH/THO. | | | | | |
|  | (≤7.97) | (≤207) | (≤865) | 23 | 96 | *EST | 53779-90-1 |
| | | Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 82LEV/LIA, 84NEL. | | | | | |
| $\text{C}_6\text{H}_{14}\text{N}_2\text{O}_2^+$ L- $\text{H}_2\text{N}(\text{CH}_2)_4\text{CH}(\text{NH}_2)\text{COOH}$ | (8.6) | (74) | (308) | -125 | -522 | *EST | 56-87-1 |
| | | IP is onset of photoelectron band (83CAN/HAM). | | | | | |
| $\text{C}_6\text{H}_{14}\text{O}^+$ n- $\text{C}_6\text{H}_{13}\text{OH}$ | (9.89±0.03) | (153) | (639) | -75.3±0.3 | -315.1±1.4 | 77PED/RYL | 111-27-3 |
| | | IP from 77ASH/BUR. | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{14}\text{O}^+$ | | | | | | | |
| $n\text{-C}_4\text{H}_9\text{CH}(\text{OH})\text{CH}_3$ | (9.80±0.03) IP from 77ASH/BUR. | (146) | (612) | -80 | -334 | 84WIB/WAS | 626-93-7 |
| $\text{C}_2\text{H}_5\text{CH}(\text{OH})\text{C}_3\text{H}_7$ | (9.63±0.03) IP from 77ASH/BUR. | (143) | (597) | -79 | -332 | *EST | 623-37-0 |
| $n\text{-C}_5\text{H}_{11}\text{OCH}_3$ | (≤9.67) IP from 80BAC/MOU. | (≤157) | (≤656) | -66 | -277 | *EST | 628-80-8 |
| $(\text{CH}_3)_2\text{CHCH}_2\text{CH}_2\text{OCH}_3$ | (≤9.65) IP from 80BAC/MOU. | (≤154) | (≤646) | -68 | -285 | *EST | 626-91-5 |
| $(\text{CH}_3)_3\text{CCH}_2\text{OCH}_3$ | (≤9.41) IP from 80BAC/MOU. | (≤146) | (≤611) | -71 | -297 | *EST | 1118-00-9 |
| $n\text{-C}_4\text{H}_9\text{OC}_2\text{H}_5$ | 9.36 IP from 81HOL/FIN. See also: 82AUD/BOU, 80BAC/MOU. | 146 | 609 | -70 | -294 | 81HOL/FIN | 628-81-9 |
| $\text{sec-C}_4\text{H}_9\text{OC}_2\text{H}_5$ | (9.32) IP from 81HOL/FIN. See also: 82AUD/BOU. | (140) | (587) | -75 | -312 | 81HOL/FIN | 2679-87-0 |
| $(\text{CH}_3)_2\text{CHCH}_2\text{OC}_2\text{H}_5$ | (9.30) IP from 82AUD/BOU. | (140) | (585) | -75 | -312 | *EST | 627-02-1 |
| $\text{tert-C}_4\text{H}_9\text{OC}_2\text{H}_5$ | (≤9.39±0.015) | (≤139) | (≤582) | (-77) | (-324) | *EST | 637-92-3 |
| $(n\text{-C}_3\text{H}_7)_2\text{O}$ | 9.27±0.05 See also: 80BAC/MOU. | 144 | 601 | -70±0.5 | -293±2 | 77PED/RYL | 111-43-3 |
| $(\text{iso-C}_3\text{H}_7)_2\text{O}$ | 9.20±0.05 See also: 80BAC/MOU. | 136 | 569 | -76.2±0.4 | -318.8±1.8 | 77PED/RYL | 108-20-3 |
| $\text{C}_6\text{H}_{14}\text{OS}^+$ | | | | | | | |
| $(n\text{-C}_3\text{H}_7)_2\text{SO}$ | (≤8.60) | (≤137) | (≤575) | -60.9±0.4 | -254.9±1.5 | 77PED/RYL | 4253-91-2 |
| $[(\text{CH}_3)_2\text{CH}]_2\text{SO}$ | (≤8.46) | (≤134) | (≤562) | -61 | -254 | *EST | 2211-89-4 |
| $\text{C}_6\text{H}_{14}\text{O}_2^+$ | | | | | | | |
| $n\text{-C}_4\text{H}_9\text{CH}(\text{CH}_3)\text{OOH}$ | 9.25±0.03 IP from 77ASH/BUR. | (152) | (636) | -61 | -256 | *EST | 24254-55-5 |
| $n\text{-C}_6\text{H}_{13}\text{OOH}$ | (9.47±0.03) IP from 77ASH/BUR. | (162) | (677) | -57 | -237 | *EST | 4312-76-9 |
| $(\text{iso-C}_3\text{H}_7\text{O})_2$ | (≤9.16) | (≤147) | (≤614) | -65 | -270 | 74BAT/CHR | 16642-57-2 |

Table 1. Positive Ion Table - Continued

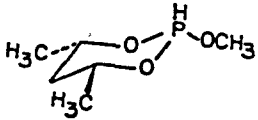
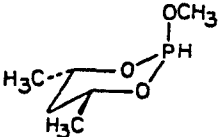
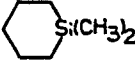
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|-----------------------------------|--|--------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{14}\text{O}_2^+$ $\text{CH}_3\text{CH}(\text{OC}_2\text{H}_5)_2$ | ≤ 9.78 IP from 82ZVE/VIL. | ≤ 117 | ≤ 490 | -108.4 ± 0.6 | -453.5 ± 2.4 | 77PED/RYL | 105-57-7 |
| $\text{C}_6\text{H}_{14}\text{O}_3^+$ $\text{CH}_3\text{OCH}_2\text{CH}_2\text{OCH}_2\text{CH}_2\text{OCH}_3$ | ≤ 9.8 IP from 83BAK/ARM. | (≤ 107) | (≤ 448) | -119 | -498 | *EST | 111-96-6 |
| $\text{C}_6\text{H}_{14}\text{O}_3\text{P}^+$ | | | | | | | |
|  | | -42 | -176 | | | | |
| | | From proton affinity of 2-ax-methoxy-4,6-dimethyl-1,3,2-dioxaphosphorinane (RN 7735-82-2). PA = 226.2 kcal/mol, 946 kJ/mol. | | | | | |
|  | | -41 | -171 | | | | |
| | | From proton affinity of 2-ax-methoxy-4,6-dimethyl-1,3,2-dioxaphosphorinane (RN 41821-91-4). PA = 225 kcal/mol, 941 kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{14}\text{S}^+$ (<i>n</i> - C_3H_7) ₂ S | 8.30 ± 0.02 | 161 | 676 | -29.9 ± 0.2 | -125.3 ± 0.8 | 77PED/RYL | 111-47-7 |
| (<i>i</i> - C_3H_7) ₂ S | 8.0 | (150) | (630) | -33.9 ± 0.2 | -141.9 ± 0.9 | 77PED/RYL | 625-80-9 |
| | | IP is onset of photoelectron spectrum. See also: 82HIR/MOH. | | | | | |
| $\text{C}_6\text{H}_{14}\text{S}_2^+$ (<i>n</i> - $\text{C}_3\text{H}_7\text{S}$) ₂ | (≤ 8.62) | (≤ 171) | (≤ 714) | -28.0 ± 0.3 | -117.3 ± 1.1 | 77PED/RYL | 629-19-6 |
| | | Dialkyl disulfides undergo a change in the CSSC bond angle from 90° to 180° upon ionization; adiabatic ionization potentials are probably well below the experimentally observed ionization onset. | | | | | |
| (<i>i</i> - $\text{C}_3\text{H}_7\text{S}$) ₂ | ≤ 8.51 | (≤ 164) | (≤ 688) | -32 | -133 | *EST | 4253-89-8 |
| | | Dialkyl disulfides undergo a change in the CSSC bond angle from 90° to 180° upon ionization; adiabatic ionization potentials are probably well below the experimentally observed ionization onset. | | | | | |
| $\text{C}_6\text{H}_{14}\text{Si}^+$ $(\text{C}_2\text{H}_5)_2\text{Si} = \text{CHCH}_3$ | | (201) | (839) | | | | 2372-29-4 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (81GUS/VOL). | | | | | |
|  | (9.0) | (164) | (686) | -43 ± 3 | -182 ± 12 | 77PED/RYL | 1072-54-4 |
| | | IP is onset of photoelectron band. See also: 81GUS/VOL2. | | | | | |

Table 1. Positive Ion Table - Continued

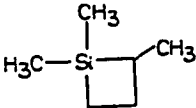
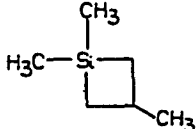
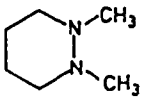
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{14}\text{Si}^+$  | 8.59±0.03 IP from 81GUS/VOL2. | (167) | (699) | -31 | -130 | 81GUS/VOL2 | 30681-90-4 |
|  | (8.67±0.03) IP from 81GUS/VOL2. | (170) | (709) | -30 | -127 | 81GUS/VOL2 | 2295-13-8 |
| $\text{C}_6\text{H}_{15}\text{B}^+$ (C_2H_5) ₃ B | 9.6 | (186) | (777) | -36±1 | -149±6 | 77PED/RYL | 97-94-9 |
| $\text{C}_6\text{H}_{15}\text{BO}_3^+$ B(OC ₂ H ₅) ₃ | (10.13) | (-6) | (-25) | -239±0.5 | -1002±2 | 77PED/RYL | 150-46-9 |
| $\text{C}_6\text{H}_{15}\text{N}^+$ n-C ₆ H ₁₃ NH ₂ | (8.63±0.05) See also: 79AUE/BOW. | (167) | (700) | -32±0.7 | -133±3 | *EST | 111-26-2 |
| (n-C ₃ H ₇) ₂ NH | 7.84±0.02 | 153 | 640 | -27.7±0.1 | -116.0±1.4 | 77PED/RYL | 142-84-7 |
| (iso-C ₃ H ₇) ₂ NH | (7.73±0.03) | (144) | (602) | -34.4±0.1 | -144.0±0.4 | 77PED/RYL | 108-18-9 |
| n-C ₄ H ₉ N(CH ₃) ₂ | ≤8.35 IP from 84NEL. | (≤172) | (≤722) | -20 | -84 | *EST | 927-62-8 |
| i-C ₄ H ₉ N(CH ₃) ₂ | ≤8.31 IP from 84NEL. | (≤170) | (≤711) | -22 | -91 | *EST | ? same as above. |
| t-C ₄ H ₉ N(CH ₃) ₂ | ≤8.08 IP from 84NEL. | (≤166) | (≤694) | -21 | -86 | *EST | 918-02-5 |
| (C ₂ H ₅) ₃ N | 7.50 IP values of 7.11 and 7.20 eV have also been reported; selected value gives hydrogen affinity value consistent with other tertiary amine ions. | 151 | 631 | -22.1±0.1 | -92.8±0.6 | 77PED/RYL | 121-44-8 |
| $\text{C}_6\text{H}_{15}\text{NO}_3^+$ N(CH ₂ CH ₂ OH) ₃ | (7.9) IP is onset of photoelectron band. | (49) | (205) | -133±0.7 | -558±3 | 82MIN/SAB | 102-71-6 |
| $\text{C}_6\text{H}_{15}\text{N}_2^+$  | | 152 | 637 | | | | |
| | | From proton affinity of hexahydro-1,2-dimethylpyridazine (RN 26163-37-1) (84MAU/NEL). PA = 229.8 kcal/mol, 961. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

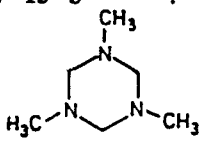
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{15}\text{N}_2\text{O}_2^+$ L-H ₃ N(CH ₂) ₄ CH(NH ₂)COOH | | 11 | 45 | | | | |
| | | From proton affinity of L-H ₂ N(CH ₂) ₄ CH(NH ₂)COOH (RN 56-87-1). PA = 230.3 kcal/mol, 963.5 kJ/mol. | | | | | |
| $\text{C}_6\text{H}_{15}\text{N}_3^+$  | 7.6 | 185 | 772 | 9.4 | 39 | 69BEN/CRU | 108-74-7 |
| | IP is onset of photoelectron band (86BEC/HUN). | | | | | | |
| $\text{C}_6\text{H}_{15}\text{O}^+$ (n-C ₃ H ₇) ₂ OH | | 93 | 391 | | | | |
| | From proton affinity of (n-C ₃ H ₇) ₂ O (RN 111-43-3). PA = 202.3 kcal/mol, 846. kJ/mol. | | | | | | |
| (i-C ₃ H ₇) ₂ OH | | 84 | 350 | | | | |
| | From proton affinity of (i-C ₃ H ₇) ₂ O (RN 108-20-3). PA = 206.0 kcal/mol, 862. kJ/mol. | | | | | | |
| C ₂ H ₅ OH(t-C ₄ H ₉) | | 83 | 347 | | | | |
| | From proton affinity of C ₂ H ₅ O(t-C ₄ H ₉) (RN 637-92-3). PA = 205.3 kcal/mol, 859. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{15}\text{OSi}^+$ (CH ₃) ₂ COSi(CH ₃) ₃ | | 40 | 168 | | | | |
| | From proton affinity of CH ₂ =C(CH ₃)OSi(CH ₃) ₃ (RN 1833-53-0). PA = 221. kcal/mol, 925. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{15}\text{O}_2^+$ CH ₃ OH(CH ₂) ₄ OCH ₃ | | 46 | 194 | | | | |
| | From proton affinity of CH ₃ O(CH ₂) ₄ OCH ₃ (RN 13179-96-9). PA = 221.8 kcal/mol, 928. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{15}\text{O}_3^+$ (CH ₃ OCH ₂ CH ₂) ₂ OH | | 27 | 114 | | | | |
| | From proton affinity of CH ₃ (OCH ₂ CH ₂) ₂ OCH ₃ (RN 111-96-6). PA = 219.4 kcal/mol, 918. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{15}\text{O}_3\text{P}^+$ (C ₂ H ₅ O) ₃ P | (8.4) | (0.6) | (2.5) | -193±1 | -808±5 | 80TEL/RAB | 122-52-1 |
| | IP is onset of photoelectron band (81ARS/ZVE, 81CHA/FIN, 82LEV/LIA). | | | | | | |
| $\text{C}_6\text{H}_{15}\text{O}_3\text{PS}^+$ (C ₂ H ₅ O) ₃ PS | (8.49±0.02) | (-35) | (-148) | -231 | -967 | *EST | 126-68-1 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|--|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{15}\text{O}_3\text{PSe}^+$ ($\text{C}_2\text{H}_5\text{O}$) ₃ PSe | (<7.9) IP from 81ZVE/VIL. | (<-27) | (<-113) | -209 | -875 | *EST | 2651-89-0 |
| $\text{C}_6\text{H}_{15}\text{O}_4\text{P}^+$ ($\text{C}_2\text{H}_5\text{O}$) ₃ PO | (9.79) See also: 81CHA/FIN. | (-58) | (-242) | -284±1 | -1187±6 | 77PED/RYL | 78-40-0 |
| $\text{C}_6\text{H}_{15}\text{P}^+$ (C_2H_5) ₃ P | 8.15±0.11 See also: 77COW/GOO, 69BOG/GRI, 79AUE/BOW. | (134) | (561) | -54 | -225 | *EST | 554-70-1 |
| $\text{C}_6\text{H}_{15}\text{S}^+$ (n- C_3H_7) ₂ SH | | 129 | 541 | From proton affinity of (n- C_3H_7) ₂ S (RN 111-47-7). PA = 206.5 kcal/mol, 864. kJ/mol. | | | |
| (i- C_3H_7) ₂ SH | | 122 | 511 | From proton affinity of (i- C_3H_7) ₂ S (RN 625-80-9). PA = 209.6 kcal/mol, 877. kJ/mol. | | | |
| $\text{C}_6\text{H}_{15}\text{Sb}^+$ (C_2H_5) ₃ Sb | (9.2±0.3) | (224) | (937) | 12±3 | 49±11 | 82TN270 | 617-85-6 |
| $\text{C}_6\text{H}_{16}\text{N}^+$ n- $\text{C}_6\text{H}_{13}\text{NH}_3$ | | 116 | 484 | From proton affinity of n- $\text{C}_6\text{H}_{13}\text{NH}_2$ (RN 111-26-2). PA = 218.9 kcal/mol, 916. kJ/mol. | | | |
| (n- C_3H_7) ₂ NH ₂ | | 110 | 462 | From proton affinity of (n- C_3H_7) ₂ NH (RN 142-84-7). PA = 227.5 kcal/mol, 952. kJ/mol. | | | |
| (i- C_3H_7) ₂ NH ₂ | | 101 | 423 | From proton affinity of (i- C_3H_7) ₂ NH (RN 108-18-9). PA = 230.2 kcal/mol, 963. kJ/mol. | | | |
| (CH_3) ₂ (tert- C_4H_9)NH | | 109 | 457 | From proton affinity of (CH_3) ₂ (tert- C_4H_9)N (RN 918-02-5). PA = 232.0 kcal/mol, 971. kJ/mol. | | | |
| (C_2H_5) ₃ NH | | 111 | 465 | From proton affinity of (C_2H_5) ₃ N (RN 121-44-8). PA = 232.3 kcal/mol, 972. kJ/mol. | | | |
| $\text{C}_6\text{H}_{16}\text{NO}^+$ $\text{NH}_3(\text{CH}_2)_6\text{OH}$ | | (68) | (285) | From proton affinity of $\text{NH}_2(\text{CH}_2)_6\text{OH}$ (RN 4048-33-3). PA = (231.0) kcal/mol, (966.5) kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

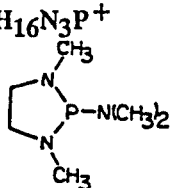
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{16}\text{N}_2^+$ $(\text{CH}_3)_2\text{NCH}_2\text{CH}_2\text{N}(\text{CH}_3)_2$ | 7.59±0.3 | 170 | 713 | -4.7 | -19.7 | 81LOG/TAK | 110-18-9 |
| | IP from 81LOG/TAK, 82LEV/LIA. | | | | | | |
| $(\text{C}_2\text{H}_5)_2\text{NN}(\text{CH}_3)_2$ | ≤8.10 | (≤196) | (≤819) | 9 | 37 | *EST | 21849-74-1 |
| | Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 82LEV/LIA, 84NEL. | | | | | | |
| $(n\text{-C}_3\text{H}_7)(\text{CH}_3)\text{NN}(\text{CH}_3)_2$ | (6.63) | (160) | (671) | 7 | 31 | *EST | 60678-65-1 |
| | IP from charge transfer equilibrium determinations (84MAU/NEL). Reference standard: IP $(\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2 = 7.12 \text{ eV.})$ See also: 84NEL. | | | | | | |
| $(\text{C}_2\text{H}_5)(\text{CH}_3)\text{NN}(\text{CH}_3)(\text{C}_2\text{H}_5)$ | 6.75 | (164) | (686) | 8 | 35 | *EST | 23337-93-1 |
| | IP from charge transfer equilibrium constant determinations (86RUM). See also: 82LEV/LIA, 84NEL. | | | | | | |
| $\text{C}_6\text{H}_{16}\text{N}_3\text{P}^+$  | (7.1) | (159) | (666) | -4 | -19 | *EST | 6069-38-1 |
| | IP is onset of photoelectron band (82WOR/HAR). | | | | | | |
| $\text{C}_6\text{H}_{16}\text{OP}^+$ $(\text{C}_2\text{H}_5)_3\text{POH}$ | | 70 | 292 | | | | |
| | From proton affinity of $(\text{C}_2\text{H}_5)_3\text{PO}$ (RN 597-50-2) (85BOL/HOU). PA = 222.6 kcal/mol, 931. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{16}\text{O}_4\text{P}^+$ $\text{HOP}(\text{OC}_2\text{H}_5)_3$ | | -135 | -565 | | | | |
| | From proton affinity of $\text{OP}(\text{OC}_2\text{H}_5)_3$ (RN 78-40-0). PA = (217) kcal/mol, (910) kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{16}\text{P}^+$ $(\text{C}_2\text{H}_5)_3\text{PH}$ | | 80 | 336 | | | | |
| | From proton affinity of $(\text{C}_2\text{H}_5)_3\text{P}$ (RN 554-70-1). PA = (231.7) kcal/mol, (969.) kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{16}\text{Si}^+$ $(\text{C}_2\text{H}_5)_3\text{SiH}$ | 9.5 | 171 | 716 | -48±4 | -201±15 | 77PED/RYL | 617-86-7 |
| | See also: 81HOT. | | | | | | |

Table 1. Positive Ion Table - Continued

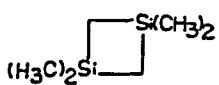
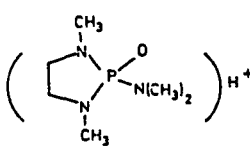
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{16}\text{Si}_2^+$ | | | | | | | |
|  | (8.56±0.07) See also: 81KHV/ZYK. | (125) | (525) | -72±3 | -301±14 | 77PED/RYL | 1627-98-1 |
| $\text{C}_6\text{H}_{16}\text{Sn}^+$ | | | | | | | |
| $(\text{C}_2\text{H}_5)_3\text{SnH}$ | (≤9.1) | (≤210) | (≤878) | 0±2 | 0±8 | 80TEL/RAB | 997-50-2 |
| $i\text{-C}_3\text{H}_7\text{Sn}(\text{CH}_3)_3$ | 8.2 IP is onset of photoelectron band. | (178) | (744) | -11±1 | -47±5 | 77PED/RYL | 3531-46-2 |
| $\text{C}_6\text{H}_{17}\text{NSi}^+$ | | | | | | | |
| $(\text{CH}_3)_2\text{NCH}_2\text{Si}(\text{CH}_3)_3$ | 7.61 See also: 81LOG/TAK. | (126) | (527) | -49 | -207 | *EST | 18182-40-6 |
| $\text{C}_6\text{H}_{17}\text{N}_2^+$ | | | | | | | |
| $\text{NH}_3(\text{CH}_2)_6\text{NH}_2$ | | 106 | 442 | | | | |
| | From proton affinity of $\text{NH}_2(\text{CH}_2)_6\text{NH}_2$ (RN 124-09-4). PA = 237.7 kcal/mol, 994.4 kJ/mol. | | | | | | |
| $(n\text{-C}_3\text{H}_7)(\text{CH}_3)\text{HNN}(\text{CH}_3)_2$ | | 145 | 605 | | | | |
| | From proton affinity of $(n\text{-C}_3\text{H}_7)(\text{CH}_3)\text{NN}(\text{CH}_3)_2$ (RN 60678-65-1) (84MAU/NEL). PA = 229.1 kcal/mol, 959. kJ/mol. | | | | | | |
| $(\text{CH}_3)_2\text{NH}(\text{CH}_2)_2\text{N}(\text{CH}_3)_2$ | | 121 | 507 | | | | |
| | From proton affinity of $(\text{CH}_3)_2\text{N}(\text{CH}_2)_2\text{N}(\text{CH}_3)_2$ (RN 110-18-9) re-evaluated. PA = 240 kcal/mol, 1003 kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{17}\text{N}_3\text{OP}^+$ | | | | | | | |
|  | | 37 | 154 | | | | |
| | From proton affinity of 1,3,2-Diazaphospholidine-2-amine, N,N',1,3-tetramethyl-2-oxide- (RN 7778-06-5) (85BOL/HOU). PA = 226.9 kcal/mol, 949. kJ/mol. | | | | | | |
| $\text{C}_6\text{H}_{18}\text{BN}_3^+$ | | | | | | | |
| $\text{B}(\text{N}(\text{CH}_3)_2)_3$ | 7.60 | 116 | 487 | -59 | -246 | 82HOL/SMI | 4375-83-1 |
| $\text{C}_6\text{H}_{18}\text{NSi}^+$ | | | | | | | |
| $(\text{CH}_3)_3\text{SiCH}_2\text{NH}(\text{CH}_3)_2$ | | 85 | 354 | | | | |
| | From proton affinity of $(\text{CH}_3)_3\text{SiCH}_2\text{N}(\text{CH}_3)_2$ (RN 18182-40-6). PA = 231.5 kcal/mol, 968. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

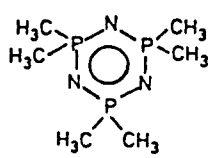
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------------|------------------------------|------------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_6\text{H}_{18}\text{N}_3\text{OP}^+$ $((\text{CH}_3)_2\text{N})_3\text{PO}$ | 7.82 | 66 | 277 | -114 | -477 | 69BEN/CRU | 630-31-9 |
| IP is onset of photoelectron band. (82LEV/LIA, 82WOR/HAR). See also: 82COW/LAT. | | | | | | | |
| $\text{C}_6\text{H}_{18}\text{N}_3\text{P}^+$ $((\text{CH}_3)_2\text{N})_3\text{P}$ | 6.75 | (124) | (517) | -32 | -134 | 69BEN/CRU | 1608-26-0 |
| IP is onset of photoelectron band (82LEV/LIA, 82WOR/HAR, 77COW/GOO). See also: 82COW/LAT. | | | | | | | |
| $\text{C}_6\text{H}_{18}\text{N}_3\text{PS}^+$ $\text{SP}[\text{N}(\text{CH}_3)_2]_3$ | $\leq 8.63 \pm 0.10$ | (≤ 162) | (≤ 677) | -37 | -156 | *EST | 3732-82-9 |
| IP from 82COW/LAT. | | | | | | | |
| $\text{C}_6\text{H}_{18}\text{N}_3\text{P}_3^+$  | (8.35 ± 0.05) | (64) | (268) | -129 ± 5 | -538 ± 23 | 77PED/RYL | 6607-30-3 |
| $\text{C}_6\text{H}_{18}\text{OSi}_2^+$ $((\text{CH}_3)_3\text{Si})_2\text{O}$ | 9.64 ± 0.01 | 36 (48) | 153 (202) | -186 ± 1 -174 | -777 ± 6 -728 | 77PED/RYL | 107-46-0 |
| IP from 83MOL/PIK, 85SEE/MOL. | | | | | | | |
| $\text{C}_6\text{H}_{18}\text{Si}_2^+$ $(\text{CH}_3)_6\text{Si}_2$ | 8.27 ± 0.05 | 110 122 | 459 513 | -81 ± 2 -68 ± 2 | -339 ± 8 -285 ± 8 | 81WAL | 1450-14-2 |
| IP from 84SZE/BAE, 81SZE/KOR. See also: 81KHV/ZYK, 85MOC/WOR. | | | | | | | |
| $\text{C}_6\text{H}_{18}\text{Sn}_2^+$ $((\text{CH}_3)_3\text{Sn})_2$ | (7.8) | (173) | (726) | -6 ± 2 | -27 ± 8 | 77PED/RYL | 661-69-8 |
| IP is onset of photoelectron band (85GRA/BER, 81SZE/KOR). See also: 85MOC/WOR. | | | | | | | |
| $\text{C}_6\text{H}_{18}\text{W}^+$ $(\text{CH}_3)_6\text{W}$ | (8.3) | (376) | (1572) | 185 ± 8 | 772 ± 35 | 82PIL/SKI | 36133-73-0 |
| IP is onset of photoelectron band (82LEV/LIA, 75GAL/WIL). | | | | | | | |
| $\text{C}_6\text{H}_{19}\text{NSi}_2^+$ $((\text{CH}_3)_3\text{Si})_2\text{NH}$ | ≤ 8.55 | ≤ 83 | ≤ 348 | -114 ± 1 | -477 ± 6 | 77PED/RYL | 999-97-3 |
| IP from 83MOL/PIK3. | | | | | | | |
| $\text{C}_6\text{H}_{19}\text{N}_3\text{P}^+$ $\text{HP}(\text{N}(\text{CH}_3)_2)_3$ | | 113 | 472 | | | | |
| From proton affinity of $\text{P}(\text{N}(\text{CH}_3)_2)_3$ (RN 1608-26-0). PA = 220.9 kcal/mol, 924. kJ/mol. | | | | | | | |
| $\text{C}_6\text{H}_{19}\text{OSi}_2^+$ $((\text{CH}_3)_3\text{Si})_2\text{OH}$ | | (-23) | (-96) | | | | |
| From proton affinity of $((\text{CH}_3)_3\text{Si})_2\text{O}$ (RN 107-46-0). PA = (203) kcal/mol, (849) kJ/mol. | | | | | | | |

Table 1. Positive Ion Table - Continued

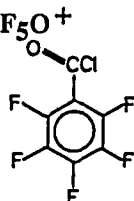
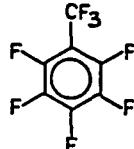
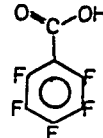
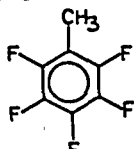
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_6MoO_6^+ $\text{Mo}(\text{CO})_6$ | 8.227±0.011 | -28 -29 | -118 -121 | -218 -219 | -912 -915 | 77ROS/DRA | 13939-06-5 |
| See also: 82HUB/LIC. | | | | | | | |
| C_6N_4^+ $(\text{NC})_2\text{CC}(\text{CN})_2$ | 11.77±0.01 | 440 | 1842 | 169±1 | 706±6 | 77PED/RYL | 670-54-2 |
| $\text{C}_6\text{O}_6\text{V}^+$ $\text{V}(\text{CO})_6$ | 7.52 | -31 | -128 | -204±7 | -854±29 | 67BID/MCI | 20644-87-5 |
| $\text{C}_6\text{O}_6\text{W}^+$ $\text{W}(\text{CO})_6$ | 8.20 | -23 | -96 | -212±1 | -887±4 | 84ALT/CON2 | 14040-11-0 |
| IP from 82HUB/LIC, 77ROS/DRA. | | | | | | | |
| $\text{C}_7\text{ClF}_5\text{O}^+$  | (9.8) | (-3) | (-13) | -229 | -959 | *EST | 2251-50-5 |
| IP is onset of photoelectron band (81MEE/WAH). | | | | | | | |
| $\text{C}_7\text{F}_3\text{MnO}_6^+$ $\text{CF}_3\text{COMn}(\text{CO})_5$ | (8.5) | (-164) | (-688) | -360±1 | -1508±6 | 82CON/ZAF | 14099-62-8 |
| IP is onset of photoelectron band. | | | | | | | |
| C_7F_8^+  | (9.9) | (-56) | (-232) | -284±2 | -1187±8 | 77PED/RYL | 434-64-0 |
| $\text{C}_7\text{HF}_5\text{O}_2^+$  | (9.2) | (-62) | (-260) | -274±1 | -1148±4 | 77PED/RYL | 602-94-8 |
| IP is onset of photoelectron band (81MEE/WAH). | | | | | | | |
| $\text{C}_7\text{H}_3\text{F}_5^+$  | (9.4) | (15) | (64) | -201.5±0.4 | -842.9±1.8 | 77PED/RYL | 771-56-2 |
| Value of IP from charge transfer equilibrium constant determinations is 9.63 eV. | | | | | | | |

Table 1. Positive Ion Table - Continued

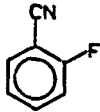
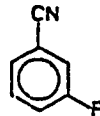
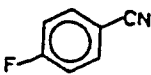

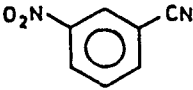
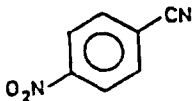
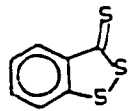
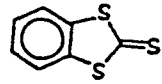
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_7H_4FN^+$ | | | | | | | |
|  | (9.78) | (231) | (965) | 5 | 21 | *EST | 394-47-8 |
|  | (9.79) | (231) | (966) | 5 | 21 | *EST | 403-54-3 |
|  | (9.74) | (229) | (957) | 4 | 17 | *EST | 1194-02-1 |
| $C_7H_4F_4^+$ | | | | | | | |
|  | 9.98 IP from 82CAB/COW. | (41) | (171) | -189±0.3 | -792±1 | *EST | 402-44-8 |
| $C_7H_4N_2O_2^+$ | | | | | | | |
|  | (10.29±0.1) | (286) | (1197) | 49 | 204 | *EST | 619-24-9 |
|  | (10.23±0.1) | (284) | (1189) | 48 | 202 | *EST | 619-72-7 |
| $C_7H_4S_3^+$ | | | | | | | |
|  | (7.9) IP is onset of photoelectron band. | (242) | (1013) | 60±1 | 251±5 | 72GEI/RAU | 3354-42-5 |
|  | (8.14) | (246) | (1027) | 57.8±0.4 | 242.0±1.7 | 77PED/RYL | 934-36-1 |

Table 1. Positive Ion Table - Continued


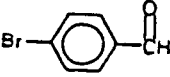
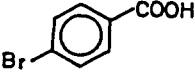
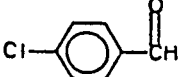
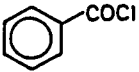
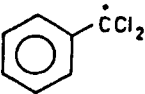
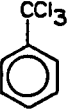
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_5\text{BrO}^+$  | (9.65) IP from 79MCL/TRA. See also: 84GAN/LIV. | (211) | (882) | -12 | -49 | 79MCL/TRA | 618-32-6 |
|  | ≤ 9.22 IP from 85GAL/GER. | (≤ 209) | (≤ 874) | -4 | -16 | *EST | 1122-91-4 |
| $\text{C}_7\text{H}_5\text{BrO}_2^+$  | (9.72 ± 0.2) | (155) | (648) | -69 ± 1 | -290 ± 5 | 77PED/RYL | 586-76-5 |
| $\text{C}_7\text{H}_5\text{ClO}^+$  | 9.59 ± 0.02 IP from 85GAL/GER, 77ROS/DRA. | (205) | (856) | -16 | -69 | *EST | 104-88-1 |
|  | 9.54 IP is onset of photoelectron band (84GAN/LIV, 81MBE/WAH). See also: 80GOF/YAR, 79MCL/TRA. | 195 | 817 | -25 ± 1 | -103 ± 4 | 75MOS/PRI | 98-88-4 |
| $\text{C}_7\text{H}_5\text{Cl}_2^+$  | (197) (824) $\Delta_f H(\text{Ion})$ from chloride transfer equilibrium constant determinations (85SHA/SHA). | | | | | | |
| $\text{C}_7\text{H}_5\text{Cl}_3^+$  | ≤ 9.60 IP from 81ZVE/ERM. | (≤ 219) | (≤ 915) | -3 | -11 | *EST | 98-07-7 |

Table 1. Positive Ion Table - Continued

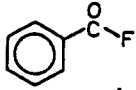
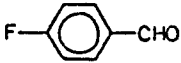
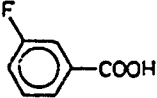
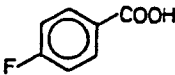
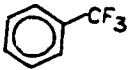
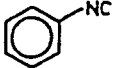
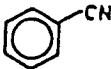
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₅FO⁺ | | | | | | | |
|  | 9.78 IP from 79MCL/TRA, 84GAN/LIV. | (153) | (639) | -73 | -305 | *EST | 455-32-3 |
|  | ≤9.60 IP from 85GAL/GER. | (≤167) | (≤700) | -54 | -226 | *EST | 459-57-4 |
| C₇H₅FO₂⁺ | | | | | | | |
|  | (9.91±0.2) | (111) | (466) | -117 | -490 | *EST | 455-38-9 |
|  | (9.91±0.2) | (111) | (461) | -118±1 | -495±3 | 77PED/RYL | 456-22-4 |
| C₇H₅F₃⁺ | | | | | | | |
|  | 9.685±0.004 See also: 81BER/BOM. | 80.1 | 335.4 | -143.2±0.2 | -599.0±0.9 | 77PED/RYL | 98-08-8 |
| C₇H₅N⁺ | | | | | | | |
|  | (9.4) IP is onset of photoelectron band. | (289) | (1208) | 72±2 | 301±7 | *EST | 931-54-4 |
|  | 9.62 See also: 83KLA/KOV, 81KIM/KAT. | 274 | 1147 | 52 | 219 | 82CHU/NGU | 100-47-0 |

Table 1. Positive Ion Table - Continued

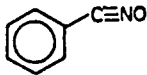
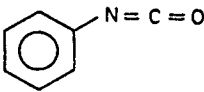

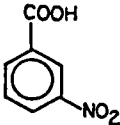
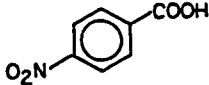
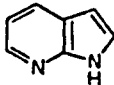
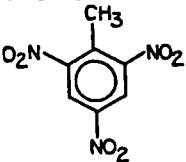
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|------------------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_5\text{NO}^+$ |  | (8.96±0.02) | (275) | (1148) | 68 | 283 | *EST | 873-67-6 |
| |  | (8.8) | (206) | (862) | 3 | 13 | *EST | 103-71-9 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_7\text{H}_5\text{NO}_3^+$ |  | 10.27±0.01 See also: 85GAL/GER. | (249) | (1043) | 12 | 52 | *EST | 555-16-8 |
| $\text{C}_7\text{H}_5\text{NO}_4^+$ |  | (10.31±0.2) | (143) | (600) | -94.3±0.3 | -394.7±1.3 | 77PED/RYL | 121-92-6 |
| |  | 10.18±0.2 | 141 | 589 | -93.7±0.4 | -392.2±1.5 | 77PED/RYL | 62-23-7 |
| $\text{C}_7\text{H}_5\text{N}_2^+$ |  | 8.11±0.01 IP from 84FUK/YOS. | (238) | (995) | 51 | 213 | *EST | |
| $\text{C}_7\text{H}_5\text{N}_3\text{O}_6^+$ |  | (10.59±0.04) | (252) | (1054) | 8±0.5 | 32±2 | 77PEL | 118-96-7 |

Table 1. Positive Ion Table - Continued

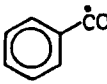
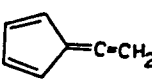

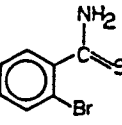
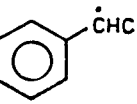
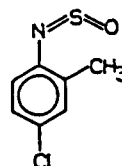
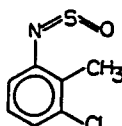
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_5\text{O}^+$  | | 168±1 | 705±6 | | | | 2652-65-5 |
| From appearance potential measurements (79MCL/TRA, 82BUR/HOL2. See also: 85TAJ/TOB. | | | | | | | |
| C_7H_6^+  | (8.29) | (275) | (1150) | 84±2 | 350±10 | *EST | 27041-32-3 |
|  | (≤8.82) | (≤292) | (≤1223) | 89±1 | 372±4 | 73BIL/CHO | 4646-69-9 |
| $\text{C}_7\text{H}_6\text{BrNS}^+$  | (8.5) IP from 81GRU. | (232) | (972) | 36 | 152 | *EST | 30216-44-5 |
| $\text{C}_7\text{H}_6\text{Cl}^+$  | | (209) | (873) | | | | |
| $\Delta_f H(\text{Ion})$ from chloride transfer equilibrium constant determinations (85SHA/SHA); $\Delta_f H(\text{C}_6\text{H}_5\text{CCl}_2\text{H})$ estimated as 3 kcal/mol, 13 kJ/mol. | | | | | | | |
| $\text{C}_7\text{H}_6\text{ClNOS}^+$  | (8.5) IP is onset of photoelectron band (82LOU/VAN). | (166) | (694) | -30 | -126 | *EST | |
|  | (≤9.23) IP from 82LOU/VAN. | (≤184) | (≤769) | -29 | -122 | *EST | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-------------------------------------|---------|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_6\text{CINS}^+$ | | 8.8 IP from 81GRU. | (226) | (948) | 24 | 99 | *EST | 15717-17-6 |
| $\text{C}_7\text{H}_6\text{ClO}^+$ | | | 149 | 623 | | | | |
| | | | From proton affinity of 4-ClC ₆ H ₄ CHO (RN 104-88-1). PA = 200.2 kcal/mol, 838. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_6\text{F}^+$ | | | (170) | (710) | | | | 40880-01-1 |
| | | | $\Delta_f H(\text{Ion})$ from chloride-transfer equilibrium constants (85SHA/SHA); $\Delta_f H(\text{o-C}_6\text{H}_4\text{FCH}_2\text{Cl})$ estimated as -43.5 kcal/mol, -182 kJ/mol. | | | | | |
| | | | (173) | (725) | | | | 2599-73-7 |
| | | | $\Delta_f H(\text{Ion})$ from chloride-transfer equilibrium constants (85SHA/SHA); $\Delta_f H(\text{m-C}_6\text{H}_4\text{FCH}_2\text{Cl})$ estimated as -43.5 kcal/mol, -182 kJ/mol. | | | | | |
| | | | (166) | (696) | | | | 2194-09-4 |
| | | | $\Delta_f H(\text{Ion})$ from chloride-transfer equilibrium constants (85SHA/SHA); $\Delta_f H(\text{o-C}_6\text{H}_4\text{FCH}_2\text{Cl})$ estimated as -43.5 kcal/mol, -182 kJ/mol. | | | | | |
| $\text{C}_7\text{H}_6\text{FO}^+$ | | | 113 | 472 | | | | |
| | | | From proton affinity of 3-FC ₆ H ₄ CHO (RN 456-48-4). PA = 196.4 kcal/mol, 822. kJ/mol. | | | | | |
| | | | 110 | 462 | | | | |
| | | | From proton affinity of 4-FC ₆ H ₄ CHO (RN 459-57-4). PA = 199.2 kcal/mol, 833. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

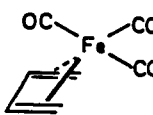
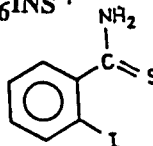
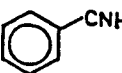
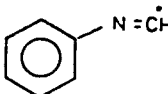
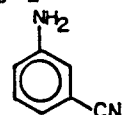

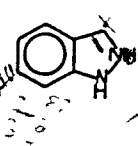
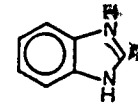
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|--|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_6\text{FeO}_3^+$  | 8.04 IP is onset of photoelectron band. See also: 82GRE/KEL. | 100 | 417 | -86±2 | -359±9 | 82PIL/SKI | 12078-32-9 |
| $\text{C}_7\text{H}_6\text{INS}^+$  | 8.5 IP from 81GRU. | (246) | (1030) | 50 | 210 | *EST | 81568-85-6 |
| $\text{C}_7\text{H}_6\text{N}^+$  | | 222 | 929 | From proton affinity of $\text{C}_6\text{H}_5\text{CN}$ (RN 100-47-0). PA = 195.9 kcal/mol, 820. kJ/mol. | | | |
|  | | 231 | 965 | From proton affinity of $\text{C}_6\text{H}_5\text{NC}$ (RN 931-54-4). (86MAU/KAR). PA = 207 kcal/mol, 866 kJ/mol. | | | |
| $\text{C}_7\text{H}_6\text{N}_2^+$  | (8.61±0.05) | (252) | (1053) | 53 | 222 | *EST | 2237-30-1 |
|  | (8.17) IP is onset of photoelectron band (81MOD/DIS). | (240) | (1004) | 52 | 216 | *EST | 873-74-5 |
|  | (8.35) | (253) | (1060) | 60.8±1.1 | 254.2±4.6 | 85FAO/AKA | 271-44-3 |
|  | (8.0) IP is onset of photoelectron band. | (228) | (957) | 44±2 | 185±10 | *EST | 51-17-2 |

Table 1. Positive Ion Table - Continued

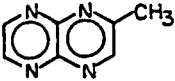
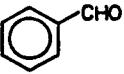
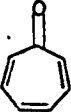
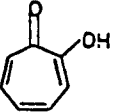
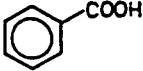
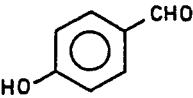
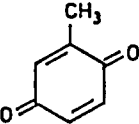
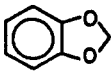
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_6\text{N}_4^+$ | | | | | | | |
|  | (8.8) | (283) | (1182) | 80 | 333 | *EST | 6499-38-3 |
| | IP is onset of photoelectron band (84GLE/SPA2). | | | | | | |
| $\text{C}_7\text{H}_6\text{O}^+$ | | | | | | | |
|  | 9.49±0.02 | 210 | 879 | -9±0.5 | -37±2 | 77PED/RYL | 100-52-7 |
| | IP from 79MCL/TRA. See also: 83KLA/KOV, 85GAL/GER. | | | | | | |
|  | 8.90±0.02 | 215 | 903 | 10±0.7 | 44±3 | 77PED/RYL | 539-80-0 |
| $\text{C}_7\text{H}_6\text{O}_2^+$ | | | | | | | |
|  | (9.86±0.02) | (191) | (797) | -36.8±0.2 | -154.0±0.9 | 77PED/RYL | 533-75-5 |
|  | (9.47) | (148) | (620) | -70.3±0.4 | -294.1±1.6 | 77PED/RYL | 65-85-0 |
| | IP from onset of photoelectron band (83KLA/KOV). See also: 81MEE/WAH. | | | | | | |
|  | (9.32±0.02) | (159) | (666) | -56±2 | -233±8 | *EST | |
|  | 9.78±0.02 | 188 | 789 | -37.1±2.0 | -155±9 | *EST | 553-97-9 |
|  | (8.0) | (150) | (629) | -34±0.7 | -143±3 | 77PED/RYL | 274-09-9 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

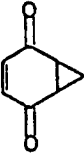
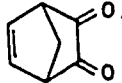

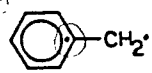
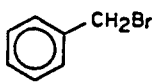
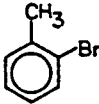
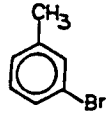
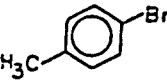
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_6\text{O}_2^+$ | | | | | | | |
|  | (9.64) IP from 85GLE/JAH. | (232) | (971) | 10 | 41 | *EST | 53735-22-1 |
|  | (8.4) IP is onset of photoelectron band. | (175) | (731) | -19 | -79 | *EST | 17994-26-2 |
| C_7H_7^+ | | | | | | | |
|  | 6.24±0.01 | 203 208 | 849 872 | 59 | 247 | 82MCM/GOL | 3551-27-7 |
| | $\Delta_f H(\text{Ion})$ from appearance potential measurements (83BOM/DAN, 83BOM/DAN2); Heat of formation of radical derived from $\Delta_f H(\text{C}_7\text{H}_7^+)$ -IP; 82MCM/GOL cite 65±2 kcal/mol, 271±8 kJ/mol. | | | | | | |
|  | 7.20±0.02 | 215 219 | 899 917 | 49 53 | 204 223 | 81TSA | 2154-56-5 |
| | $\Delta_f H(\text{Ion})$ from chloride transfer equilibrium constants (81SEN/KEB) is in agreement. | | | | | | |
| $\text{C}_7\text{H}_7\text{Br}^+$ | | | | | | | |
|  | 9.0 IP is onset of photoelectron band. | (224) | (935) | 16±0.5 | 67±2 | 76ASH | 100-39-0 |
|  | 8.58±0.1 See also: 85BAI/MIS. | (213) | (890) | 15 | 62 | *EST | 95-46-5 |
|  | 8.79±0.02 | (217) | (909) | 15 | 61 | *EST | 591-17-3 |
|  | 8.67±0.01 IP from 82LEV/LIA, 78LIA/AUS, 77ROS/DRA. See also: 85BAI/MIS. | (217) | (908) | 17 | 71 | *EST | 106-38-7 |

Table 1. Positive Ion Table - Continued

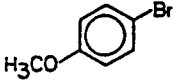
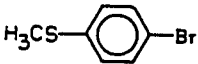
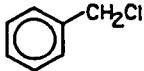
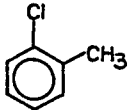
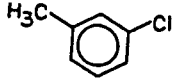
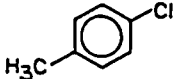


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_7H_7BrO^+$ | | | | | | | |
|  | (8.11) | (177) | (739) | -10 | -43 | *EST | 104-92-7 |
| $C_7H_7BrS^+$ | | | | | | | |
|  | (7.5) IP is onset of photoelectron band (81BAK/ARM). | (201) | (843) | 28 | 119 | *EST | 104-95-0 |
| $C_7H_7Cl^+$ | | | | | | | |
|  | 9.14±0.01 See also: 81ZVE/ERM, 81KIM/KAT. | 215 | 899 | 4±0.7 | 17±3 | 76ASH | 25168-05-2 |
|  | (8.83±0.02) See also: 85BAI/MIS. | (208) | (871) | 4 | 18 | *EST | 95-49-8 |
|  | (8.83±0.02) | (208) | (870) | 4 | 18 | *EST | 108-41-8 |
|  | 8.69±0.02 See also: 85BAI/MIS. | (205) | (856) | 4 | 18 | *EST | 106-43-4 |
|  | ≤8.77 IP from 83HOU/RON. | (≤256) | (≤1071) | 54 | 225 | *EST | 2294-41-9 |
| $C_7H_7ClHg^+$ | | | | | | | |
|  | (8.7) IP is onset of photoelectron band (81FUR/PIA). | (223) | (931) | 22 | 92 | *EST | 539-43-5 |

Table 1. Positive Ion Table - Continued

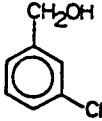
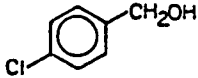
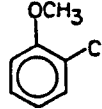
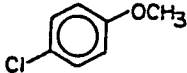
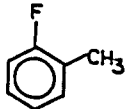
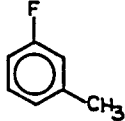
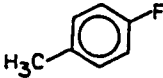
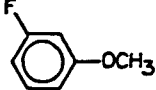
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₇ClO⁺ | | | | | | | |
|  | (8.51) IP from 83RUS/FRE. | (166) | (697) | -30 | -124 | 83RUS/FRE | 873-63-2 |
|  | (8.58) IP from 83RUS/FRE. | (167) | (698) | -31 | -130 | 83RUS/FRE | 873-76-7 |
|  | (8.42) IP from 83RUS/FRE. | (169) | (707) | -25 | -105 | 83RUS/FRE | 766-51-8 |
|  | (7.79) IP from 83RUS/FRE. | (153) | (641) | -26 | -111 | 83RUS/FRE | 623-12-1 |
| C₇H₇F⁺ | | | | | | | |
|  | 8.91±0.01 See also: 78LIA/AUS. | (170) | (711) | -36 | -149 | *EST | 95-52-3 |
|  | 8.91±0.01 | (170) | (710) | -36 | -150 | *EST | 352-70-5 |
|  | 8.79±0.01 See also: 78LIA/AUS. | 167 | 700 | -35.2±0.3 | -147.5±1.2 | 77PED/RYL | 352-32-9 |
| C₇H₇FO⁺ | | | | | | | |
|  | 8.41 IP from 85OIK/ABE. | (130) | (544) | -64 | -267 | *EST | 456-49-5 |

Table 1. Positive Ion Table - Continued

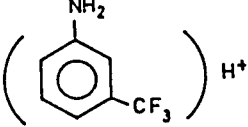
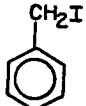
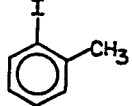
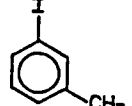
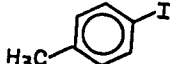
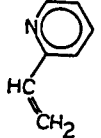
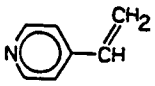

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_7\text{F}_3\text{N}^+$  | | 19 | 81 | | | | |
| | | From proton affinity of 3-CF ₃ C ₆ H ₄ NH ₂ (RN 98-16-8). PA = 204.2 kcal/mol, 854. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_7\text{I}^+$ | | | | | | | |
|  | (8.6) | (223) | (933) | 25±1 | 103±4 | 76ASH | 620-05-3 |
|  | (8.62±0.01) See also: 8SBAI/MIS. | (231) | (965) | 32±1 | 133±6 | 77PED/RYL | 615-37-2 |
|  | (8.61±0.03) | (231) | (965) | 32±1 | 134±6 | 77PED/RYL | 625-95-6 |
|  | (8.50±0.01) See also: 8SBAI/MIS. | (225) | (942) | 29±1 | 122±6 | 77ROS/DRA | 624-31-7 |
| $\text{C}_7\text{H}_7\text{N}^+$ | | | | | | | |
|  | (8.6) IP is onset of photoelectron band (81MOD/DIS2). | (246) | (1030) | 48 | 200 | *EST | 100-69-6 |
|  | (8.9) IP is onset of photoelectron band (81MOD/DIS2). | (254) | (1061) | 48 | 202 | *EST | 100-43-6 |
|  | (≤9.11) IP from 79AUE/BOW. | (≤270) | (≤1129) | 60 | 250 | *EST | 56911-25-2 |

Table 1. Positive Ion Table - Continued

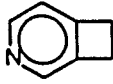
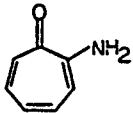
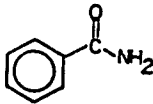
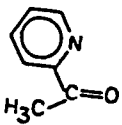
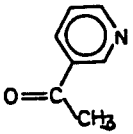
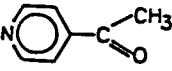
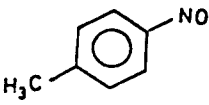
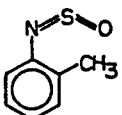
| ION Neutral | Ionization potential eV | | $\Delta_f H(\text{Ion})$ kcal/mol kJ/mol | | $\Delta_f H(\text{Neutral})$ kcal/mol kJ/mol | | Neutral reference | CAS registry number |
|---|----------------------------|----------------|---|---------|---|------------|----------------------|--|
| | | | | | | | | |
| C₇H₇N⁺ | | | | | | | | |
|  | (≤ 9.37) | (≤ 276) | (≤ 1156) | 60 | 252 | *EST | 56911-27-4 | IP from 79AUE BOW. $\Delta_f H(\text{Ion})$ estimated from hydrogen affinities of pyridine ions = 268 kcal/mol, 1121 kJ/mol. Corresponding IP = 9.02 eV. |
| C₇H₇NO⁺ | | | | | | | | |
|  | (9.43±0.02) | (227) | (949) | 9.4±0.6 | 39.5±2.5 | 77PED/RYL | 6264-93-3 | |
|  | 9.45 | 194 | 811 | -24±0.2 | -101±1 | 82TOR/SAB2 | 55-21-0 | |
|  | (8.9) | (200) | (838) | -5 | -21 | *EST | 1122-62-9 | IP is onset of photoelectron band (81MOD/DIS2). |
|  | (9.1) | (204) | (852) | -6 | -26 | *EST | | IP is onset of photoelectron band (81MOD/DIS2). |
|  | (9.3) | (208) | (871) | -6 | -26 | *EST | 1122-54-9 | IP is onset of photoelectron band (81MOD/DIS2). |
|  | (8.79±0.1) | (216) | (903) | 13.2±1 | 55±4 | *EST | 623-11-0 | |
| C₇H₇NOS⁺ | | | | | | | | |
|  | (8.75) | (187) | (782) | -15 | -62 | *EST | | IP is onset of photoelectron band (82LOU/VAN). |

Table 1. Positive Ion Table - Continued

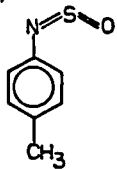
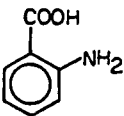
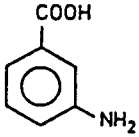
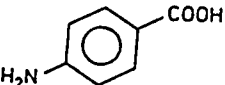
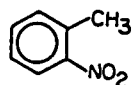
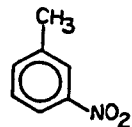
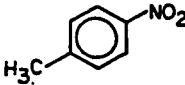
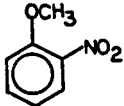
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_7H_7NOS^+$  | (≤ 8.84) IP from 82LOU/VAN. | (≤ 190) | (≤ 795) | -14 | -58 | *EST | |
| $C_7H_7NO_2^+$  | (7.6) IP is onset of photoelectron band (81MEE/WAH). | (104) | (435) | -71 ± 0.5 | -298 ± 2 | 77NAB/SAB | 118-92-3 |
|  | (7.8) IP is onset of photoelectron band (81MEE/WAH). | (111) | (463) | -69 ± 1 | -289 ± 4 | 77NAB/SAB | 99-05-8 |
|  | (7.8) IP is onset of photoelectron band (81MEE/WAH). See also: 84TOB/TAJ. | (110) | (458) | -70 ± 1 | -294 ± 4 | 77NAB/SAB | 150-13-0 |
|  | 9.45 ± 0.04 IP from 82LEV/LIA, 82BAL/CAR. See also: 73GOL/KOR. | 231 | 965 | 13 | 53 | 77PED/RYL | 88-72-2 |
|  | (9.48 ± 0.02) | (226) | (946) | 7 | 31 | 77PED/RYL | 99-08-1 |
|  | (9.4) IP is onset of photoelectron band. | (224) | (938) | 7 ± 1 | 31 ± 4 | 77PED/RYL | 99-99-0 |
| $C_7H_7NO_3^+$  | (8.8) IP is onset of photoelectron band. | (186) | (779) | -17 | -70 | *EST | 91-23-6 |

Table 1. Positive Ion Table - Continued

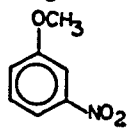
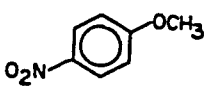
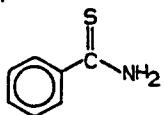
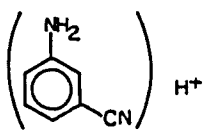
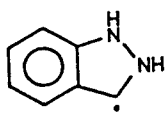
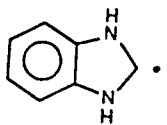
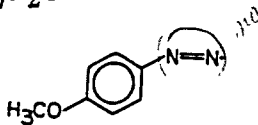
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|---|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_7\text{NO}_3^+$  | (8.7) IP is onset of photoelectron band. | (179) | (749) | -22 | -90 | *EST | 555-03-3 |
|  | (8.79) | (182) | (760) | -21 | -88 | *EST | 100-17-4 |
| $\text{C}_7\text{H}_7\text{NS}^+$  | (8.8) IP from 81GRU. | (234) | (980) | 31.3±0.3 | 131.0±1.3 | 82TUR/SAB2 | 2227-79-4 |
| $\text{C}_7\text{H}_7\text{N}_2^+$  | | 218 | 912 | From proton affinity of 3-cyanobenzamide (RN 2237-30-1). PA = 200.7 kcal/mol, 840. kJ/mol. | | | |
|  | | 208 | 870 | From proton affinity of indazole (RN 271-44-3) (84FLA/MAQ). PA = 218 kcal/mol, 914 kJ/mol. | | | |
|  | | 181 | 757 | From proton affinity of benzimidazole (RN 51-17-2) (83CAT/ELG, 84FLA/MAQ). PA = 227 kcal/mol, 958 kJ/mol. | | | |
| $\text{C}_7\text{H}_7\text{N}_2\text{O}^+$  | (7.28) IP from 77NUY/MES. | (178) | (746) | 11 | 44 | *EST | 17333-79-8 |

Table 1. Positive Ion Table - Continued

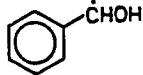
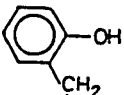
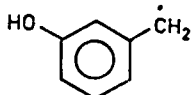
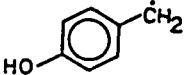
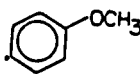
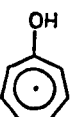
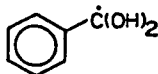
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_7\text{O}^+$ | | | | | | | |
|  | | 157 | 655 | | | | |
| | | From proton affinity of benzaldehyde (RN 100-52-7). PA = 200.2 kcal/mol, 838. kJ/mol. | | | | | |
|  | | 176 | 735 | | | | 65108-16-9 |
| | | From appearance potential determinations (83RUS/FRE). | | | | | |
|  | | 174 | 728 | | | | 65108-08-9 |
| | | From appearance potential determinations (83RUS/FRE). | | | | | |
|  | | 175 | 731 | | | | 29180-18-5 |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determinations (83RUS/FRE). Value derived from proton affinity of 4-methylene-2,5-cyclohexadiene-1-one is (153) kcal/mol, (641) kJ/mol, a serious discrepancy. PA = (222) kcal/mol, (929) kJ/mol. | | | | | |
|  | (8.32) | (202) | (845) | 10 | 42 | *EST | 2396-03-4 |
| | IP from 77NUY/MES. | | | | | | |
|  | | 157 | 656 | | | | |
| | | From proton affinity of 2,4,6-cycloheptatriene-1-one. PA = 219 kcal/mol, 918 kJ/mol (RN 539-80-0) and appearance potential determinations (83RUS/FRE). | | | | | |
| $\text{C}_7\text{H}_7\text{O}_2^+$ | | | | | | | |
|  | | 97 | 407 | | | | |
| | | From proton affinity of benzoic acid (RN 65-85-0). PA = 198.2 kcal/mol, 829. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued


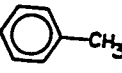





| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------|---|---|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_7H_8^+ | | | | | | | | |
| | <chem>HC#C(CH2)3C#CH</chem> | (9.85) | (322) | (1346) | 94.7 | 396 | 58BEN/BUS | 2396-63-6 |
| | | IP from 78TRA/MCL. | | | | | | |
| |  | 8.29 | 235 | 982 | 43.7±0.2 | 182.8±1 | 77PED/RYL | 544-25-2 |
| | | | 240 | 1004 | 48.7 | 203.8 | | |
| | | IP from 78TRA/MCL. | | | | | | |
| |  | 8.82±0.01 | 215 | 901 | 12.0±0.1 | 50.1±0.3 | 77PED/RYL | 108-88-3 |
| | | | 221 | 924 | 17.5 | 73.3 | | |
| | | See also: 82SEL/HEL, 81KIM/KAT, 78LIA/AUS, 84HOW/GON. | | | | | | |
| |  | 7.9 | 223 | 934 | 41 | 172 | 85BAL/HAS | 20679-59-8 |
| | | IP from 85BAL/HAS. See also: 82BUR/TER, 82BAR. | | | | | | |
| |  | (8.6) | (233) | (975) | 35±3 | 146±13 | 82BAR | 3217-87-6 |
| | | IP from 82BAR. | | | | | | |
| |  | 8.35 | 250 | 1046 | 57±1 | 240±4 | 80ROG/CHO | 121-46-0 |
| | | IP from 78TRA/MCL. See also: 83HOU/RON, 82BIE/ASB, 85OHN/ISH. | | | | | | |
| |  | 8.8 | 288 | 1206 | 85 | 357 | 85BAL/HAS | 67254-49-3 |
| | | IP from 85BAL/HAS. | | | | | | |
| |  | (7.8) | (260) | (1086) | 80±1 | 333±4 | 80ROG/CHO | 278-06-8 |
| | | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

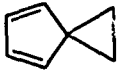
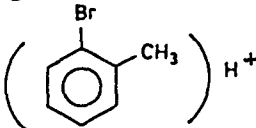
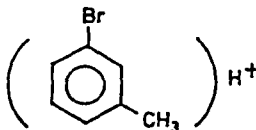
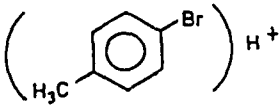
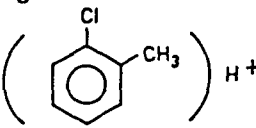
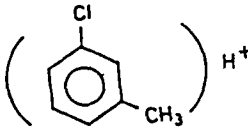
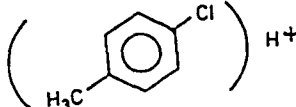
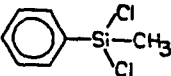
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_7H_8^+ |  | (8.14) | (250) | (1045) | 62 | 260 | *EST | 765-46-8 |
| $\text{C}_7\text{H}_8\text{Br}^+$ | | | | | | | | |
|  | | | 193 | 809 | | | | |
| | | | From proton affinity of 1,2- $\text{C}_6\text{H}_4(\text{CH}_3)\text{Br}$ (RN 95-46-5) (82MAS/BOH). PA = 187.2 kcal/mol, 783. kJ/mol. | | | | | |
|  | | | 191 | 801 | | | | |
| | | | From proton affinity of 1,3- $\text{C}_6\text{H}_4(\text{CH}_3)\text{Br}$ (RN 591-17-3) (82MAS/BOH). PA = 188.8 kcal/mol, 790. kJ/mol. | | | | | |
|  | | | 196 | 818 | | | | |
| | | | From proton affinity of 1,4- $\text{C}_6\text{H}_4(\text{CH}_3)\text{Br}$ (RN 106-38-7) (82MAS/BOH). PA = 187.2 kcal/mol, 783. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_8\text{Cl}^+$ | | | | | | | | |
|  | | | 186 | 777 | | | | |
| | | | From proton affinity of 1,2- $\text{C}_6\text{H}_4(\text{CH}_3)\text{Cl}$ (RN 95-49-8) (82MAS/BOH). PA = 184.3 kcal/mol, 771. kJ/mol. | | | | | |
|  | | | 181 | 758 | | | | |
| | | | From proton affinity of 1,3- $\text{C}_6\text{H}_4(\text{CH}_3)\text{Cl}$ (RN 108-41-8) (82MAS/BOH). PA = 188.9 kcal/mol, 790. kJ/mol. | | | | | |
|  | | | 189 | 792 | | | | |
| | | | From proton affinity of 1,4- $\text{C}_6\text{H}_4(\text{CH}_3)\text{Cl}$ (RN 106-43-4) (82MAS/BOH). PA = 180.6 kcal/mol, 756. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_8\text{Cl}_2\text{Si}^+$ | | | | | | | | |
|  | | (8.97) | (132) | (551) | -75 | -314 | *EST | 149-74-6 |
| | | | IP from 84VES/HAR. | | | | | |

Table 1. Positive Ion Table - Continued

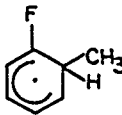
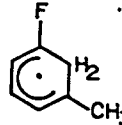
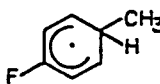
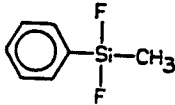
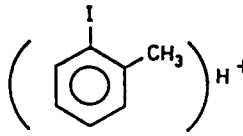
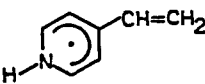
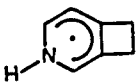
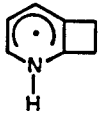
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_8\text{F}^+$ | | | | | | | |
|  | | 143 | 599 | | | | |
| | | From proton affinity of 2- $\text{FC}_6\text{H}_4\text{CH}_3$ (RN 95-52-3) (82MAS/BOH). PA = 187.0 kcal/mol, 782. kJ/mol. | | | | | |
|  | | 140 | 587 | | | | |
| | | From proton affinity of 3- $\text{FC}_6\text{H}_4\text{CH}_3$ (RN 352-70-5) (82MAS/BOH). PA = 189.5 kcal/mol, 793. kJ/mol. | | | | | |
|  | | 149 | 625 | | | | |
| | | From proton affinity of 4- $\text{FC}_6\text{H}_4\text{CH}_3$ (RN 352-32-9) (82MAS/BOH). PA = 180.9 kcal/mol, 757. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_8\text{F}_2\text{Si}^+$ | | | | | | | |
|  | (8.97) | (29) | (122) | -178 | -743 | *EST | 328-57-4 |
| | | IP from 84VES/HAR. | | | | | |
| $\text{C}_7\text{H}_8\text{I}^+$ | | | | | | | |
|  | | 209 | 875 | | | | |
| | | From proton affinity of 2- $\text{IC}_6\text{H}_4\text{CH}_3$ (RN 615-37-2) (82MAS/BOH). PA = 188.4 kcal/mol, 788. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_8\text{N}^+$ | | | | | | | |
|  | | (191) | (798) | | | | |
| | | From proton affinity of 4-vinylpyridine (RN 100-43-6). PA = (223.2) kcal/mol, (934) kJ/mol. | | | | | |
|  | | (202) | (846) | | | | |
| | | From proton affinity of 3,4-cyclobutenopyridine (RN 56911-27-4). PA = (225.9) kcal/mol, (945) kJ/mol. | | | | | |
|  | | (200) | (838) | | | | |
| | | From proton affinity of 2,3-cyclobutenopyridine. PA = (223.3) kcal/mol, (934) kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

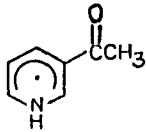
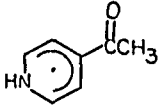
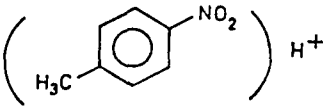
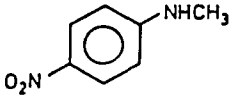
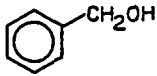
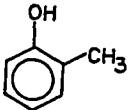
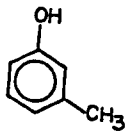
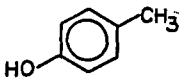
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|--|---|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₈NO⁺ | | | | | | | | |
|  | | | 142 | 595 | | | | |
| | | | From proton affinity of 1-(3-pyridinyl)-ethanone (RN 350-03-8). PA = 217.2 kcal/mol, 909 kJ/mol. | | | | | |
|  | | | 142 | 594 | | | | |
| | | | From proton affinity of 1-(4-pyridinyl)-ethanone (RN 1122-54-9). PA = 217.4 kcal/mol, 910. kJ/mol. | | | | | |
| C₇H₈NO₂⁺ | | | | | | | | |
|  | | | 176 | 738 | | | | |
| | | | From proton affinity of 4-nitrotoluene (RN 99-99-0) (84ROL/HOU). PA = 196.8 kcal/mol, 823. kJ/mol. | | | | | |
| C₇H₈N₂O₂⁺ | | | | | | | | |
|  | | (8.1) | (201) | (843) | 15 | 61 | *EST | 100-15-2 |
| | | IP is onset of photoelectron band. | | | | | | |
| C₇H₈O⁺ | | | | | | | | |
|  | | (8.5) | (172) | (720) | -24.0 | -100.4 | 77PED/RYL | 100-51-6 |
| | | See also: 82DES/DUT, 83RUS/FRE. IP is onset of photoelectron band (86BAL/JON). | | | | | | |
|  | | 8.14 | 158 | 661 | -30 | -124 | 79KUD/KUD | 95-48-7 |
| | | IP from 83RUS/FRE. | | | | | | |
|  | | 8.29 | 160 | 668 | -31.6±0.3 | -132.3±1.2 | 79KUD/KUD | 108-39-4 |
| | | IP from 85OIK/ABE, 83RUS/FRE. | | | | | | |
|  | | 8.13 | 157 | 659 | -29.9 | -125.1 | 79KUD/KUD | 106-44-5 |
| | | IP from 83RUS/FRE. | | | | | | |

Table 1. Positive Ion Table - Continued

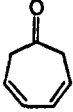
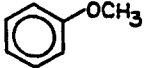
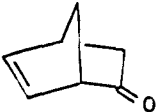
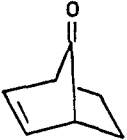
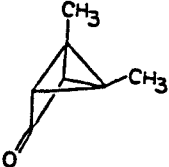
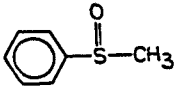
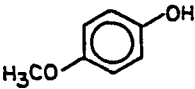
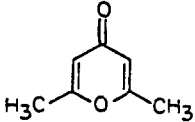
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₈O⁺ | | | | | | | |
|  | (8.24) IP from 83RUS/FRE. | (188) | (788) | -1.7 | -7.1 | 83RUS/FRE | 1121-65-9 |
|  | 8.21±0.02 See also: 83KLA/KOV. | 173 | 724 | -16.2±0.3 | -68.0±1.1 | 77PED/RYL | 100-66-3 |
|  | (≤8.86) | (≤197) | (≤826) | -7 | -29 | *EST | 694-98-4 |
|  | (≤9.25) | (≤210) | (≤877) | -4 | -15 | *EST | 694-71-3 |
|  | (8.9) IP is onset of photoelectron band (84GLE/HAI). | (251) | (1049) | 45 | 190 | *EST | 3350-02-5 |
| C₇H₈OS⁺ | | | | | | | |
|  | (8.5) IP is onset of photoelectron band. | (191) | (800) | -5 | -20 | *EST | 1193-82-4 |
| C₇H₈O₂⁺ | | | | | | | |
|  | (7.50) | (115) | (482) | -58 | -242 | *EST | 150-76-5 |
|  | (9.03) IP from 85GRU/SPI | (152) | (636) | -56 | -235 | *EST | 1004-36-0 |

Table 1. Positive Ion Table - Continued

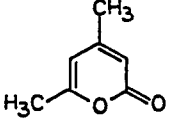
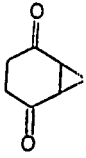
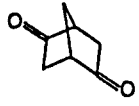
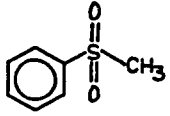
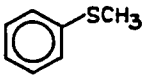
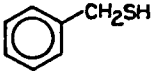
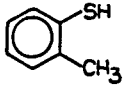
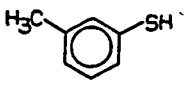
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_8\text{O}_2^+$ | | | | | | | |
|  | (8.51) IP from 85GRU/SPI | (135) | (564) | -61 | -257 | *EST | 675-09-2 |
|  | (9.4) IP is onset of photoelectron band (85GLE/JAH). | (168) | (702) | -49 | -205 | *EST | 60582-65-2 |
|  | (9.26) IP is onset of photoelectron band (80FRO/WES). | (161) | (674) | -52 | -219 | *EST | 27943-47-1 |
| $\text{C}_7\text{H}_8\text{O}_2\text{S}^+$ | | | | | | | |
|  | (9.5) IP is onset of photoelectron band (81MOH/JIA). | (159) | (663) | -60.6±0.7 | -253.4±3.0 | 77PED/RYL | 3112-85-4 |
| $\text{C}_7\text{H}_8\text{S}^+$ | | | | | | | |
|  | 7.94±0.02 | 206 | 864 | 23.4±0.3 | 97.8±1.2 | 77PED/RYL | 100-68-5 |
|  | (8.5) IP is onset of photoelectron band. | (218) | (914) | 22±0.7 | 94±3 | 77PED/RYL | 100-53-8 |
|  | (≤8.31) | (≤211) | (≤881) | 19 | 79 | *EST | 137-06-4 |
|  | (≤8.44) | (≤214) | (≤893) | 19 | 79 | *EST | 108-40-7 |

Table 1. Positive Ion Table - Continued

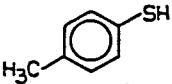
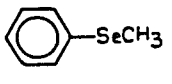
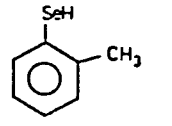
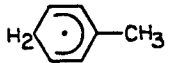
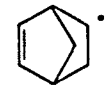
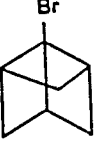
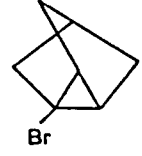
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------------|---|--|--------------------------|----------------|--|--------|----------------------|--|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_8\text{S}^+$ |  | (8.0) IP is onset of photoelectron band. | (203) | (851) | 19 | 79 | *EST | 106-45-6 |
| $\text{C}_7\text{H}_8\text{Se}^+$ |  | (7.4) IP is onset of photoelectron band (81BAK/ARM). | (207) | (867) | 36 | 153 | *EST | 4346-64-9 |
| |  | ≤ 8.4 IP from 81BAK/ARM. | (≤ 225) | (≤ 940) | 31 | 130 | *EST | 37773-21-0 |
| C_7H_9^+ |  | | 188 | 786 | From proton affinity of $\text{C}_6\text{H}_5\text{CH}_3$ (RN 108-88-3). PA = 189.8 kcal/mol, 794. kJ/mol. | | | |
| |  | | 220 | 919 | From proton affinity of 2,5-norbornadiene (RN 121-46-0) (86HOU/SCH). PA = 203.4 kcal/mol, 851. kJ/mol. | | | |
| $\text{C}_7\text{H}_9\text{Br}^+$ |  | (8.7) IP is onset of photoelectron band (84ABE/DEL). | (261) | (1091) | 60 | 252 | *EST | 78995-60-5 59346-69-9 <i>duplicate by C8H7Br p338</i> |
| |  | (8.55) IP is onset of photoelectron band (85DEL/PIG). | (218) | (912) | 21 | 87 | *EST | 31991-53-4 |

Table 1. Positive Ion Table - Continued

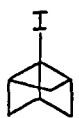
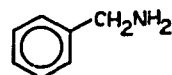
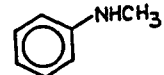
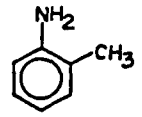
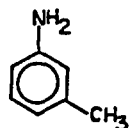
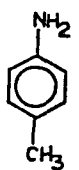
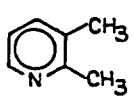
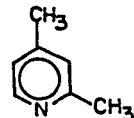
| ION Neutral | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------|-------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | eV | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₉I⁺  | (8.6) | (273) | (1143) | 75 | 313 | *EST | 74725-76-1 | |
| IP is onset of photoelectron band (84ABE/DEL). | | | | | | | | |
| C₇H₉N⁺ | | | | | | | | |
|  | 8.64±0.05 | 219 | 918 | 20±0.7 | 84±3 | 77CAR/LAY | 100-46-9 | |
| See also: 79AUE/BOW. | | | | | | | | |
|  | 7.33±0.02 | 189 | 792 | 20 | 85 | 78COL/BEN | 100-61-8 | |
| See also: 84MAU/NEL, 83KLA/KOV. | | | | | | | | |
|  | 7.44±0.02 | (185) | (773) | 13±0.2 | 55±1 | *EST | 95-53-4 | |
|  | 7.50±0.02 | (186) | (778) | 13±0.4 | 54±2 | *EST | 108-44-1 | |
|  | (7.24±0.02) | (180) | (753) | 13 | 54 | *EST | 106-49-0 | |
|  | (8.85±0.02) | (220) | (922) | 16.3 | 68.3 | 77PED/RYL | 583-61-9 | |
| $\Delta_f H(\text{Ion})$ predicted from hydrogen affinities of pyridines: 224 kcal/mol, 937 kJ/mol. Corresponding IP = 9.01 eV. See also: 79AUE/BOW. | | | | | | | | |
|  | (8.85±0.03) | (219) | (918) | 15.3 | 63.9 | 77PED/RYL | 108-47-4 | |

Table 1. Positive Ion Table - Continued

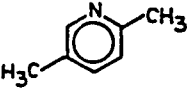
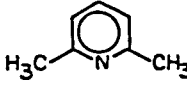
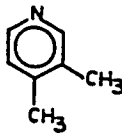
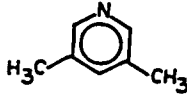
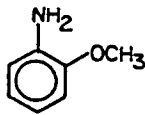
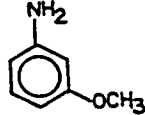
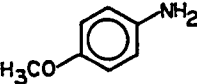
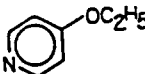
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|---|--------------------------|----------------|------------------------------|----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₉N⁺ | | | | | | | | |
|  | | ($\leq 8.80 \pm 0.05$) | (≤ 219) | (≤ 916) | 15.9 \pm 0.2 | 66.5 \pm 1.0 | 77PEI/RYL | 589-93-5 |
| | | $\Delta_f H(\text{Ion})$ predicted from hydrogen affinities of pyridines: 223 kcal/mol, 933 kJ/mol. Corresponding IP = 8.98 eV. See also: 79AUE/BOW. | | | | | | |
|  | | 8.86 \pm 0.03 | 218 | 913 | 14.0 \pm 0.4 | 58.7 \pm 1.6 | 77PED/RYL | 108-48-5 |
| | | See also: 81KIM/KAT. | | | | | | |
|  | | (≤ 9.15) | (≤ 228) | (≤ 953) | 16.7 \pm 0.2 | 70.1 \pm 1.1 | 77PED/RYL | 583-58-4 |
| | | $\Delta_f H(\text{Ion})$ predicted from hydrogen affinities of pyridines: 224 kcal/mol, 937 kJ/mol. Corresponding IP = 8.98 eV. See also: 79AUE/BOW. | | | | | | |
|  | | (≤ 9.25) | (≤ 231) | (≤ 965) | 17.4 \pm 0.2 | 72.8 \pm 0.9 | 77PED/RYL | 591-22-0 |
| | | $\Delta_f H(\text{Ion})$ predicted from hydrogen affinities of pyridines: 226 kcal/mol, 946 kJ/mol. Corresponding IP = 9.05 eV. | | | | | | |
| C₇H₉NO⁺ | | | | | | | | |
|  | | (7.46 \pm 0.1) | (158) | (663) | -14 | -57 | *EST | 90-04-0 |
|  | | (7.76 \pm 0.1) | (163) | (682) | -16 | -67 | *EST | 536-90-3 |
|  | | (7.44) | (158) | (660) | -14 | -58 | *EST | 104-94-9 |
|  | | ($\leq 9.25 \pm 0.03$) | (≤ 202) | (≤ 845) | -11 | -47 | *EST | 33399-46-1 |

Table 1. Positive Ion Table - Continued

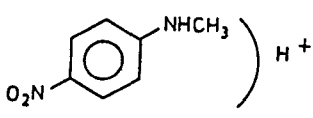
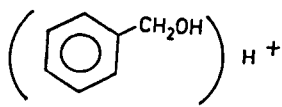
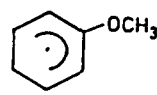
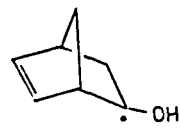
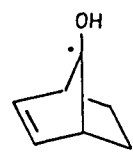
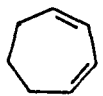
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--------------------------------------|--|-----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_9\text{N}_2\text{O}_2^+$  | | 167 | 700 | | | | |
| | | From proton affinity of N-methyl-4-nitroaniline (RN 100-23-2) (84ROL/HOU). PA = 212.9 kcal/mol, 891. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_9\text{O}^+$  | | 153 | 641 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{CH}_2\text{OH}$ (RN 100-51-6) (78TAF/TAA). PA = 188.5 kcal/mol, 789. kJ/mol. | | | | | |
|  | | 149 | 624 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{OCH}_3$ (RN 100-66-3). PA = 200.3 kcal/mol, 838. kJ/mol. | | | | | |
|  | | 157 | 655 | | | | |
| | | From proton affinity of bicyclo[2.2.1]hept-2-ene-5-ol (RN 694-98-4). (86HOU/SCH). PA = 202.1 kcal/mol, 846. kJ/mol. | | | | | |
|  | | 164 | 686 | | | | |
| | | From proton affinity of bicyclo[2.2.1]hept-2-ene-7-ol (RN 694-71-3). (86HOU/SCH). PA = 198.1 kcal/mol, 829. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{10}^+$ | | | | | | | |
| $(\text{CH}_2 = \text{CH})_3\text{CH}$ | (≤ 9.5) IP from 83GLE/HAI. | (≤ 265) | (≤ 1108) | 46 | 191 | *EST | 26456-63-3 |
| (E,E)- $\text{CH}_2 = \text{CHCH} = \text{CHCH} = \text{CHCH}_3$ | 7.96 ± 0.02 | (215) | (901) | 32 ± 1 | 133 ± 4 | *EST | 17679-93-5 |
| $\text{C}_2\text{H}_5\text{C} \equiv \text{CC}(\text{CH}_3) = \text{CH}_2$ | (8.66 ± 0.01) | (247) | (1033) | 47 | 197 | *EST | 23056-94-2 |
| (E)- $\text{HC} \equiv \text{CC}(\text{C}_2\text{H}_5) = \text{CHCH}_3$ | (8.70 ± 0.01) | (247) | (1031) | 46 | 192 | *EST | 14272-82-3 |
|  | $\leq 8.31 \pm 0.03$ | ≤ 214 | ≤ 896 | 22.5 ± 0.2 | 94.2 ± 0.9 | 77PED/RYL | 4054-38-0 |

Table 1. Positive Ion Table - Continued


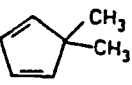
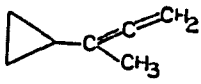
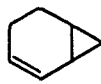
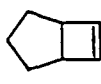

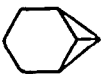

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{10}^+$ | | | | | | | |
|  | (8.85±0.03) | (232) | (970) | 28 | 116 | 76JEN | 7161-35-5 |
|  | 8.2 IP is onset of photoelectron band (85GUI/PFI3). | (206) | (860) | 16 | 69 | *EST | 4125-18-2 |
|  | (8.83) | (257) | (1075) | 53 | 223 | 82KOZ/MAS | 51549-86-1 |
|  | (≤8.69) | (≤227) | (≤951) | 27 | 113 | *EST | 2566-57-6 |
|  | (9.37) | (250) | (1046) | 34 | 142 | *EST | 4927-03-1 |
|  | 8.82±0.03 See also: 83HOU/RON. | 225 | 941 | 21±1 | 90±4 | 80ROG/CHO | 498-66-8 |
|  | 8.72 IP is onset of photoelectron band (85DEL/PIG). | 245.5 | 1027.1 | 44.4 | 185.8 | 85SVY/IOF | 287-13-8 |
|  | (8.7) IP is onset of photoelectron band (84ABE/DEL). | (260) | (1089) | 60 | 250 | *EST | 51273-50-8 |

Table 1. Positive Ion Table - Continued



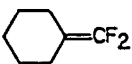
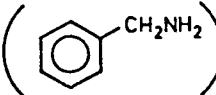
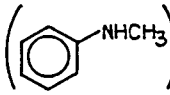
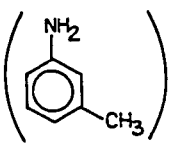
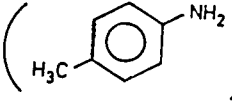
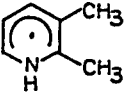
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{10}^+$ | | | | | | | |
|  | (8.72) | (221) | (926) | 20±1 | 85±4 | 80ROG/CHO | 279-19-6 |
| | IP is onset of photoelectron band (85DEL/PIG). | | | | | | |
|  | (≤8.48) | (≤236) | (≤986) | 40 | 168 | *EST | 52708-23-3 |
| $\text{C}_7\text{H}_{10}\text{F}_2^+$ | | | | | | | |
|  | 8.84 | (103) | (433) | -100 | -420 | *EST | 696-32-2 |
| | IP from 80SAR/WOR. | | | | | | |
| $\text{C}_7\text{H}_{10}\text{N}^+$ | | | | | | | |
|  H^+ | | 169 | 707 | From proton affinity of $\text{C}_6\text{H}_5\text{CH}_2\text{NH}_2$ (RN 100-46-9). PA = 216.8 kcal/mol, 907. kJ/mol. | | | |
|  H^+ | | 168 | 703 | From proton affinity of $\text{C}_6\text{H}_5\text{NHCH}_3$ (RN 100-61-8). PA = 218.1 kcal/mol, 912.5 kJ/mol. | | | |
|  H^+ | | 165 | 690 | From proton affinity of 3- $\text{CH}_3\text{C}_6\text{H}_4\text{NH}_2$ (RN 108-44-1). PA = 213.4 kcal/mol, 893. kJ/mol. | | | |
|  H^+ | | 165 | 690 | From proton affinity of 4- $\text{CH}_3\text{C}_6\text{H}_4\text{NH}_2$ (RN 106-49-0). PA = 213.7 kcal/mol, 894. kJ/mol. | | | |
|  | | 156 | 652 | From proton affinity of 2,3-dimethylpyridine (RN 583-61-9). PA = 226.2 kcal/mol, 946. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

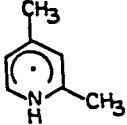
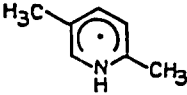
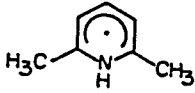
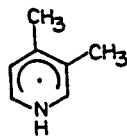
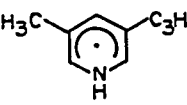
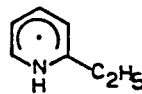
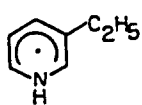
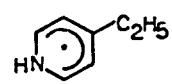
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₁₀N⁺ | | | | | | | |
|  | | 153 | 643 | From proton affinity of 2,4-dimethylpyridine (RN 108-47-4). PA = 227.3 kcal/mol, 951. kJ/mol. | | | |
|  | | 156 | 651 | From proton affinity of 2,5-dimethylpyridine (RN 589-93-5). PA = 226.2 kcal/mol, 946. kJ/mol. | | | |
|  | | 152 | 634 | From proton affinity of 2,6-dimethylpyridine (RN 108-48-5). PA = 228.2 kcal/mol, 955. kJ/mol. | | | |
|  | | 157 | 654 | From proton affinity of 3,4-dimethylpyridine (RN 583-58-4). PA = 226.0 kcal/mol, 946. kJ/mol. | | | |
|  | | 158 | 661 | From proton affinity of 3,5-dimethylpyridine (RN 591-22-0). PA = 225.5 kcal/mol, 943. kJ/mol. | | | |
|  | | 159 | 665 | From proton affinity of 2-ethylpyridine (RN 100-71-0). PA = 226.2 kcal/mol, 946. kJ/mol. | | | |
|  | | 162 | 679 | From proton affinity of 3-ethylpyridine (RN 536-78-7). PA = 223.9 kcal/mol, 937. kJ/mol. | | | |
|  | | (161) | (672) | From proton affinity of 4-ethylpyridine (RN 536-75-4). PA = (224.6) kcal/mol, (940) kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

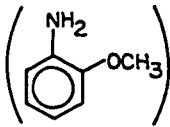
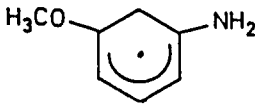
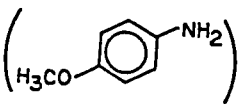
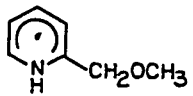
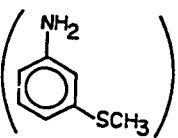
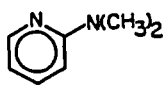
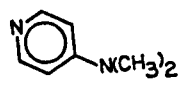
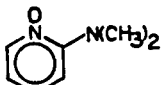
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₁₀NO⁺ | | | | | | | |
|  | | | 137 | 575 | | | |
| | | | From proton affinity of 2-CH ₃ OC ₆ H ₄ NH ₂ (RN 90-04-0). PA = 214.7 kcal/mol, 898. kJ/mol. | | | | |
|  | | | 132 | 553 | | | |
| | | | From proton affinity of 3-CH ₃ OC ₆ H ₄ NH ₂ (RN 536-90-3). PA = 217.6 kcal/mol, 910. kJ/mol. | | | | |
|  | | | 137 | 575 | | | |
| | | | From proton affinity of 4-CH ₃ OC ₆ H ₄ NH ₂ (RN 104-94-9). PA = 214.3 kcal/mol, 897. kJ/mol. | | | | |
|  | | | (134) | (563) | | | |
| | | | From proton affinity of pyridine,2-methoxymethyl (RN 23579-92-2). PA = (226.0) kcal/mol, (945.) kJ/mol. | | | | |
| C₇H₁₀NS⁺ | | | | | | | |
|  | | | 176 | 735 | | | |
| | | | From proton affinity of 3-methylthiobenzenamine (RN 1783-81-9). PA = 214.5 kcal/mol, 897. kJ/mol. | | | | |
| C₇H₁₀N₂⁺ | | | | | | | |
|  | 7.75±0.15 | (211) | (880) | 32 | 132 | *EST | 5683-33-0 |
|  | (≤7.82) | (≤214) | (≤898) | 34 | 144 | 84BIC/PIL | 1122-58-3 |
| C₇H₁₀N₂O⁺ | | | | | | | |
|  | (7.62±0.05) | (191) | (798) | 15 | 63 | *EST | 3618-79-9 |

Table 1. Positive Ion Table - Continued

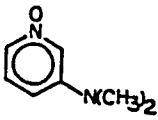
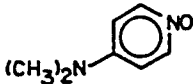
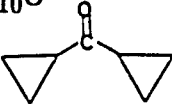

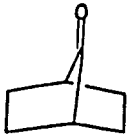
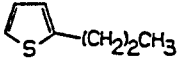

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{10}\text{N}_2\text{O}^+$ | | | | | | | |
|  | (7.85±0.05) | (197) | (824) | 16 | 67 | *EST | 36100-40-0 |
|  | (7.0) IP is onset of photoelectron band. | (178) | (746) | 17 | 71 | *EST | 1005-31-8 |
| $\text{C}_7\text{H}_{10}\text{O}^+$ | | | | | | | |
|  | (9.1) | (249) | (1041) | 39 | 163 | *EST | 1121-37-5 |
|  | 8.94±0.02 See also: 80FRO/WES. | 166 | 695 | -40±0.7 | -168±3 | 78STE2 | 497-38-1 |
|  | ≤9.01±0.02 | ≤176 | ≤735 | -32±0.7 | -134±3 | 78STE2 | 10218-02-7 |
| $\text{C}_7\text{H}_{10}\text{S}^+$ | | | | | | | |
|  | (≤8.6±0.2) | (≤208) | (≤870) | 10 | 40 | *EST | 1551-27-5 |
| $\text{C}_7\text{H}_{11}^+$ | | | | | | | |
|  | 6.84 | 186 | 777 | 28 | 117 | DERIVED | 30967-37-4 |
| | IP from 79HOU. $\Delta_f H$ of $\text{C}_7\text{H}_{11}^+$ from PA of norbornene, 187 kcal/mol, 784 kJ/mol; PA = 199.9 kcal/mol, 836. kJ/mol. From hydride transfer equilibria relative to $\Delta_f H(\text{t-C}_4\text{H}_9^+)$, 185 kcal/mol, 773 kJ/mol (76SOL/FIE, 85SHA/SHA); from chloride transfer equilibria 185.8 kcal/mol, 777.4 kJ/mol (85SHA/SHA). Cited $\Delta_f H$ of radical = $\Delta_f H(\text{C}_7\text{H}_{11}^+) - \text{IP}$. | | | | | | |

Table 1. Positive Ion Table - Continued

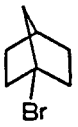
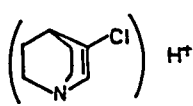




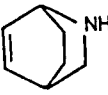
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{11}\text{Br}^+$ | | | | | | | |
|  | 9.55 | (209) | (874) | -11 | -47 | *EST | 13474-70-9 |
| | IP is onset of photoelectron band (84DEL/ABE, 85HON/HEI2). | | | | | | |
| $\text{C}_7\text{H}_{11}\text{Cl}^+$ | | | | | | | |
| $(\text{CH}_3)_3\text{CCH}=\text{C}=\text{CHCl}$ | 9.05 | (228) | (954) | 19 | 81 | *EST | 65388-53-6 |
| | IP is onset of photoelectron band (85ELS/VER). | | | | | | |
| $\text{C}_7\text{H}_{11}\text{ClN}^+$ | | | | | | | |
|  | | (167) | (697) | | | | |
| | From proton affinity of 3-chloro-1-azabicyclo[2.2.2]oct-2-ene PA = (224.0) kcal/mol, (937.) kJ/mol. | | | | | | |
| $\text{C}_7\text{H}_{11}\text{I}^+$ | | | | | | | |
|  | (8.8) | (206) | (863) | 3 | 14 | *EST | 930-80-3 |
| | IP is onset of photoelectron band (84DEL/ABE). | | | | | | |
|  | (9.00) | (210) | (878) | 2 | 10 | *EST | 57173-48-5 |
| | IP from 84HON/HEI. | | | | | | |
|  | 9.00 | (211) | (882) | 3 | 14 | *EST | 30983-85-8 |
| | IP from 84HON/HEI. | | | | | | |
| $\text{C}_7\text{H}_{11}\text{N}^+$ | | | | | | | |
|  | (8.02) | (222) | (930) | 37 | 156 | *EST | 13929-94-7 |
|  | ($\leq 8.35 \pm 0.05$) | (≤ 218) | (≤ 913) | 26 | 107 | *EST | 3693-58-1 |

Table 1. Positive Ion Table - Continued

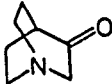
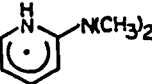
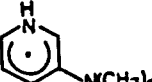
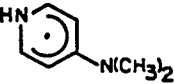
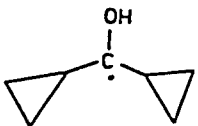

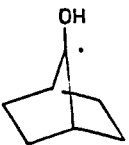
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{11}\text{NO}^+$ | | | | | | | |
|  | (≤ 8.2) IP from 79AUE/BOW. | (≤ 161) | (≤ 675) | -28 | -116 | *EST | 3731-38-2 |
| $\text{C}_7\text{H}_{11}\text{N}_2^+$ | | | | | | | |
|  | | 168 | 703 | | | | |
| | From proton affinity of N,N-dimethyl-2-pyridinamine (RN 5683-33-0). PA = 229.2 kcal/mol, 959. kJ/mol. | | | | | | |
|  | | (174) | (726) | | | | |
| | From proton affinity of N,N-dimethyl-3-pyridinamine (RN 18437-57-5). PA = (229.9) kcal/mol, (962) kJ/mol. | | | | | | |
|  | | 163 | 684 | | | | |
| | From proton affinity of N,N-dimethyl-4-pyridinamine (RN 1122-58-3). (86TAF/GAL, 77ARN/CHA). PA = 236.6 kcal/mol, 990. kJ/mol. | | | | | | |
| $\text{C}_7\text{H}_{11}\text{O}^+$ | | | | | | | |
|  | | 194 | 812 | | | | |
| | From proton affinity of dicyclopropylmethanone (RN 1121-37-5). PA = 210.7 kcal/mol, 881.5 kJ/mol. | | | | | | |
|  | | 123 | 514 | | | | |
| | From proton affinity of bicyclo[2.2.1]heptan-2-ol (RN 497-38-1) (86HOU/SCH). PA = 202.6 kcal/mol, 848. kJ/mol. | | | | | | |
|  | | 134 | 561 | | | | |
| | From proton affinity of bicyclo[2.2.1]heptan-7-ol (RN 10218-02-7) (86HOU/SCH). PA = 199.5 kcal/mol, 835. kJ/mol. | | | | | | |
| $\text{C}_7\text{H}_{12}^+$ | | | | | | | |
| (E)- $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}=\text{CHCH}=\text{CH}_2$ | (8.47) IP from 81MAS/MOU. | (204) | (852) | 8 | 35 | *EST | 2384-92-1 |

Table 1. Positive Ion Table - Continued

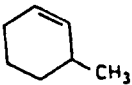

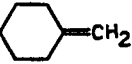
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{12}^+$ | | | | | | | |
| (E),(E)- $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}=\text{CHCH}_3$ | ≤ 8.17 IP from 81MAS/MOU. | (≤ 194) | (≤ 812) | 6 | 24 | *EST | 2384-94-3 |
|  | 8.89 ± 0.01 See also: 83BRO/BUS. | (196) | (822) | -9 | -36 | *EST | 591-48-0 |
| (E)- $(\text{CH}_3)_2\text{CHCH}=\text{CHCH}=\text{CH}_2$ | ≤ 8.47 IP from 81MAS/MOU. | (≤ 202) | (≤ 847) | 7 | 30 | *EST | 32763-70-5 |
| (E)- $\text{C}_2\text{H}_5\text{C}(\text{CH}_3)=\text{CHCH}=\text{CH}_2$ | (8.19) IP from 81MAS/MOU. | (195) | (814) | 6 | 24 | *EST | 4842-93-7 |
| (E)- $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{C}(\text{CH}_3)=\text{CH}_2$ | (8.28) IP from 81MAS/MOU. | (194) | (813) | 3 | 14 | *EST | 1625-49-6 |
| $\text{CH}_2=\text{C}(\text{CH}_3)\text{C}(\text{C}_2\text{H}_5)=\text{CH}_2$ | (8.65) IP from 81MAS/MOU. | (205) | (860) | 6 | 25 | *EST | 14145-44-9 |
| $n\text{-C}_5\text{H}_{11}\text{C}\equiv\text{CH}$ | (10.04) From plot of trends in IP's of 1-alkynes, an IP of -9.95 eV would be predicted. | (256) | (1073) | 25 ± 0.7 | 104 ± 3 | 79ROG/DAG | 628-71-7 |
| $n\text{-C}_4\text{H}_9\text{C}\equiv\text{CCH}_3$ | (9.33 ± 0.01) | (235) | (985) | 20 ± 0.5 | 85 ± 2 | 79ROG/DAG | 1119-65-9 |
| $n\text{-C}_3\text{H}_7\text{C}\equiv\text{CC}_2\text{H}_5$ | (9.26 ± 0.01) | (233) | (976) | 20 | 83 ± 2 | 79ROG/DAG | 2586-89-2 |
| (tert- C_4H_9) $\text{C}\equiv\text{CCH}_3$ | (9.276 ± 0.10) See also: 85ORL/BOG. | (235) | (984) | 21 | 89 | *EST | 999-78-0 |
|  | (8.91 ± 0.04) | (203) | (850) | -2.2 ± 0.2 | -9.4 ± 0.9 | 77PED/RYL | 628-92-2 |
|  | 8.93 ± 0.01 IP from 80SAR/WOR. See also: 86SPA/RAD. | 200 | 837 | -6 ± 1 | -25 ± 4 | 79FUC/PEA | 1192-37-6 |

Table 1. Positive Ion Table - Continued

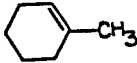
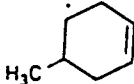
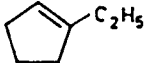
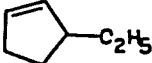
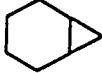

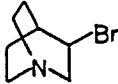
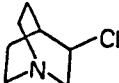
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₁₂⁺ | | | | | | | |
|  | 8.67±0.02 | 189.6 | 793.3 | -10.3±0.2 | -43.2±0.7 | 77PED/RYL | 591-49-1 |
|  | (8.91±0.01) | (197) | (824) | -9 | -36 | *EST | 591-47-9 |
|  | (8.53±0.01) | (192) | (801) | -5±0.7 | -22±3 | 82ALL/DOD | 2146-38-5 |
|  | 8.88±0.01 | 202 | 843 | -3 | -14 | 82ALL/DOD | 694-35-9 |
|  | (9.03±0.02) | (209) | (873) | 0.5±0.5 | 2±2 | 77PED/RYL | 286-08-8 |
|  | 9.77±0.03 | 213 | 894 | -12±1 | -49±4 | 80ROG/CHO | 279-23-2 |
| C₇H₁₂BrN⁺ | | | | | | | |
|  | (≤8.5) IP from 79AUE/BOW. | (≤197) | (≤823) | 1 | 3 | *EST | |
| C₇H₁₂ClN⁺ | | | | | | | |
|  | (≤8.8) IP from 79AUE/BOW. | (≤192) | (≤805) | -11 | -44 | *EST | 42332-45-6 |

Table 1. Positive Ion Table - Continued

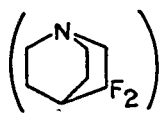
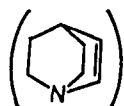
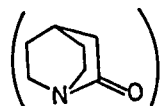
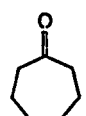

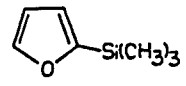
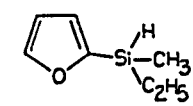
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{12}\text{F}_2\text{N}^+$  H^+ | | (43) | (179) | | | | |
| | | From proton affinity of 3,3-difluoro-1-azabicyclo[2.2.2]octane. PA = (221.8) kcal/mol, (928.) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{12}\text{N}^+$  H^+ | | (175) | (730) | | | | |
| | | From proton affinity of 1-azabicyclo[2.2.2]oct-2-ene (RN 13929-94-7). PA = (228.5) kcal/mol, (956.) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{12}\text{NO}^+$  H^+ | | (116) | (486) | | | | |
| | | From proton affinity of 1-azabicyclo[2.2.2]octan-3-one (RN 3731-38-2). PA = (221.9) kcal/mol, (928) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{12}\text{O}^+$  | ≤ 9.14 | ≤ 152 | ≤ 634 | -59.1 ± 0.4 | -247.5 ± 1.8 | 77PED/RYL | 502-42-1 |
|  CHO | (9.6±0.1) | (165) | (691) | -56.2 | -235.1 | 82SPL/CAL | 2043-61-0 |
| | | IP from 82SPL/CAL. | | | | | |
| $\text{C}_7\text{H}_{12}\text{OSi}^+$  | (8.1) | (134) | (563) | -52 | -219 | *EST | 1578-33-2 |
| | | IP is onset of photoelectron band (83ZYK/ERC). | | | | | |
|  | ≤ 8.53 | (≤ 151) | (≤ 630) | -46 | -193 | *EST | 13271-69-7 |
| | | IP from 83ZYK/ERC. | | | | | |

Table 1. Positive Ion Table - Continued

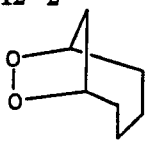
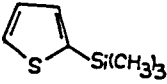
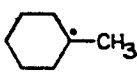
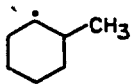
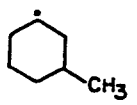
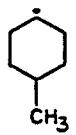
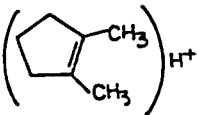
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{12}\text{O}_2^+$  | (8.5) IP is onset of photoelectron band (84GLE/DOB). | (161) | (673) | -35 | -147 | *EST | 68525-35-9 |
| $\text{C}_7\text{H}_{12}\text{SSi}^+$  | (8.1) IP is onset of photoelectron band (83VES/HAR). | (173) | (726) | -13 | -56 | *EST | 18245-28-8 |
| $\text{C}_7\text{H}_{13}^+$ $(\text{CH}_3)_2\text{CCHC}(\text{CH}_3)_2$ | | (157) | (655) | | | | 60602-30-4 |
| | From proton affinity of $(\text{CH}_3)_2\text{C}=\text{CHC}(\text{CH}_3)=\text{CH}_2$. (RN 1000-86-8). PA = (213.1) kcal/mol, (892.) kJ/mol. | | | | | | |
|  | | 157 | 655 | | | | 16998-65-5 |
| | From proton affinity of 1-methylcyclohexene. (RN 591-49-1). PA = 198.8 kcal/mol, 832. kJ/mol. | | | | | | |
|  | | (169) | (708) | | | | 41771-02-2 |
| | From appearance potential measurements (81HER/SIC). | | | | | | |
|  | | (173) | (722) | | | | 61838-22-0 |
| | From appearance potential measurements (81HER/SIC). | | | | | | |
|  | | (172) | (720) | | | | 21029-96-9 |
| | From appearance potential measurements (81HER/SIC). | | | | | | |
|  | | 158 | 660 | | | | 21029-96-9 |
| | From proton affinity of 1,2-dimethylcyclopentene. (RN 765-47-9). PA = 198.1 kcal/mol, 829. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

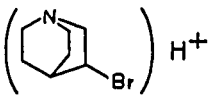
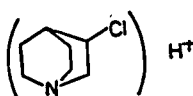
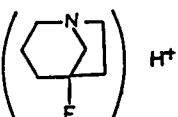
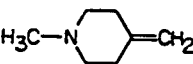
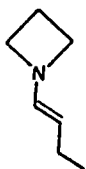
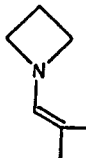

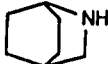
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{13}\text{BrN}^+$ | | | | | | | |
|  | | 139 | 583 | | | | |
| | | From proton affinity of 3-bromo-1-azabicyclo[2.2.2]octane. PA = (227.1) kcal/mol, (950.) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{13}\text{ClN}^+$ | | | | | | | |
|  | | (129) | (541) | | | | |
| | | From proton affinity of 3-chloro-1-azabicyclo[2.2.2]octane (RN 42332-45-6). PA = (225.8) kcal/mol, (945.) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{13}\text{FN}^+$ | | | | | | | |
|  | | (160) | (670) | | | | |
| | | From proton affinity of 3-fluoro-1-azabicyclo[3.2.1]octane. PA = (228.1) kcal/mol, (954.) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{13}\text{N}^+$ | | | | | | | |
|  | (≤ 8.36) IP from 80SAR/WOR. | (≤ 204) | (≤ 855) | 11 | 48 | *EST | 13669-28-8 |
|  | (7.1) IP is onset of photoelectron band (81MUL/PRE2). | (197) | (823) | 33 | 138 | *EST | 81156-87-8 |
|  | (6.9) IP is onset of photoelectron band (81MUL/PRE2). | (192) | (803) | 33 | 137 | *EST | 81156-88-9 |
|  | (7.4) IP is onset of photoelectron band. | (170) | (710) | -1.0 \pm 0.3 | -4.2 \pm 1.2 | 77PED/RYL | 100-76-5 |
|  | ($\leq 8.22 \pm 0.05$) | (≤ 187) | (≤ 782) | -3 | -11 | *EST | 280-38-6 |

Table 1. Positive Ion Table - Continued

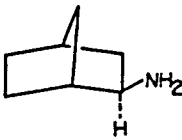
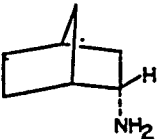
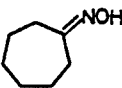
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|---------|------------------------------|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{13}\text{N}^+$ | | | | | | | |
|  | (8.33) IP from 79AUE/BOW. | (185) | (776) | -7 ± 0.2 | -28 ± 1 | *EST | 31002-73-0 |
|  | (8.41) IP from 79AUE/BOW. | (186) | (779) | -8 ± 0.2 | -32 ± 1 | *EST | 7242-92-4 |
| $\text{C}_7\text{H}_{13}\text{NO}^+$ | | | | | | | |
|  | (8.88±0.03) IP from 79GOL/KUL. | (179) | (749) | -26 | -108 | *EST | 2158-31-8 |
| $\text{C}_7\text{H}_{14}^+$ | | | | | | | |
| 1- C_7H_{14} | (9.44) | (202.8) | (848.9) | -14.8 | -61.9 | 84WIB/WAS | 592-76-7 |
| 2- C_7H_{14} | (8.84±0.02) IP from 77ASH/BUR. | (187) | (782) | -17 | -71 | 84WIB/WAS | 592-77-8 |
| 3- C_7H_{14} | (8.92) IP from 81HOL/FIN. | (189) | (790) | -17 | -71 | 84WIB/WAS | 14686-14-7 |
| $(\text{CH}_3)_3\text{CCH}_2\text{CH}=\text{CH}_2$ | 9.40±0.01 | 197 | 823 | -20.0 ± 0.2 | -83.8 ± 0.8 | 77PED/RYL | 762-62-9 |
| n- $\text{C}_4\text{H}_9\text{C}(\text{CH}_3)=\text{CH}_2$ | (9.04±0.01) | (190) | (796) | -18 | -76 | *EST | 6094-02-6 |
| $(\text{CH}_3)_2\text{CHCH}_2\text{C}(\text{CH}_3)=\text{CH}_2$ | (9.03±0.01) | (188) | (787) | -20.0 ± 0.3 | -83.8 ± 1.4 | 77PED/RYL | 2213-32-3 |
| $(\text{CH}_3)_3\text{CC}(\text{CH}_3)=\text{CH}_2$ | (9.02±0.01) | (187.5) | (784.4) | -20.4 ± 0.3 | -85.5 ± 1.4 | 77PED/RYL | 594-56-9 |
| (Z)- $(\text{CH}_3)_2\text{CHCH}_2\text{CH}=\text{CHCH}_3$ | (8.92±0.01) | (187) | (782) | -19 | -78 | *EST | 13151-17-2 |
| (E)- $(\text{CH}_3)_2\text{CHCH}_2\text{CH}=\text{CHCH}_3$ | (8.92±0.01) | (186) | (779) | -20 | -82 | *EST | 7385-82-2 |
| (E)- $\text{C}_2\text{H}_5\text{CH}(\text{CH}_3)\text{CH}=\text{CHCH}_3$ | (8.91±0.01) | (186) | (778) | -20 | -82 | *EST | 3683-22-5 |
| $\text{C}_3\text{H}_7\text{CH}=\text{C}(\text{CH}_3)_2$ | (8.62) IP from 81HOL/FIN. | (179) | (748) | -20 | -84 | *EST | 2738-19-4 |

Table 1. Positive Ion Table - Continued


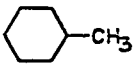
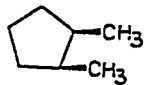
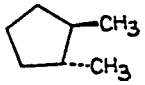
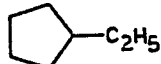
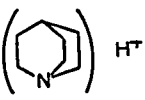
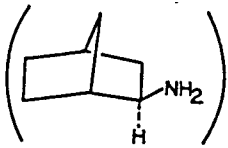
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{14}^+$ $\text{C}_2\text{H}_5\text{C}(\text{CH}_3)=\text{C}(\text{CH}_3)_2$ | (8.21±0.01) | (168) | (702) | -21 | -90 | *EST | 10574-37-5 |
| (Z)- $(\text{CH}_3)_3\text{CCH}=\text{CHCH}_3$ | (8.92±0.01) | (188) | (788) | -17.4±0.3 | -72.6±1.4 | 77PED/RYL | 762-63-0 |
| (E)- $(\text{CH}_3)_3\text{CCH}=\text{CHCH}_3$ | (8.91±0.01) | (184) | (771) | -21.3±0.3 | -88.8±1.1 | 77PED/RYL | 690-08-4 |
|  | 9.97 | 202 | 844 | -28.3±0.1 | -118.2±0.6 | 77PED/RYL | 291-64-5 |
| | 300 K ionization energy from charge transfer equilibrium constants relative to cyclohexane and cyclohexane-d ₁₂ = 9.99 eV. (82SIE/MAU). | | | | | | |
|  | 9.64 | 185 | 775 | -37.0±0.2 | -154.7±1.0 | 77PED/RYL | 108-87-2 |
| | IP from charge transfer equilibrium constant determinations (82SIE/MAU; 82LIA). Reference IP's, fluorobenzenes. Threshold determination gives IP = 9.76±0.03 eV. | | | | | | |
|  | (9.92±0.05) | (198) | (828) | -30.9±0.3 | -129.5±1.3 | 77PED/RYL | 1192-18-3 |
| | IP from 81HER/SIC. | | | | | | |
|  | (9.95±0.05) | (197) | (823) | -32.7±0.3 | -136.7±1.1 | 77PED/RYL | 822-50-4 |
| | IP from 81HER/SIC. | | | | | | |
|  | (10.12±0.02) | (203) | (850) | -30.3±0.2 | -126.7±0.9 | 77PED/RYL | 1640-89-7 |
| <hr/> | | | | | | | |
| $\text{C}_7\text{H}_{14}\text{N}^+$ | | | | | | | |
|  | | 132 | 551 | | | | |
| | From proton affinity of 1-azabicyclo[2.2.2]octane (RN 100-76-5). (86TAP/GAL). PA = 233.1 kcal/mol, 975. kJ/mol. | | | | | | |
|  | | 137 | 574 | | | | |
| | From proton affinity of bicyclo[2.2.1]heptan-2-amine, endo (RN 31002-73-0). PA = (221.7) kcal/mol, (927.) kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

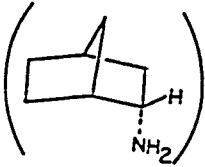
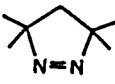
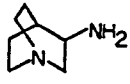
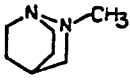
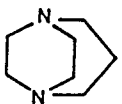
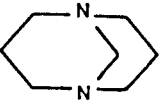
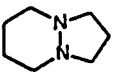
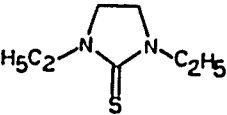
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|---|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{14}\text{N}^+$  | | | | | | | |
| | | 136 | 570 | | | | |
| | | From proton affinity of bicyclo[2.2.1]heptan-2-amine, exo (RN 7242-92-4). PA = (221.7) kcal/mol, (927.) kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{14}\text{N}_2^+$ | | | | | | | |
|  | (≤ 8.63) | (≤ 208) | (≤ 872) | 9.4 ± 0.8 | 39.3 ± 3.6 | 80ENG | 2721-31-5 |
|  | (≤ 8.0) | (≤ 188) | (≤ 785) | 4 | 17 | *EST | 6238-14-8 |
| | IP from 79AUE/BOW. | | | | | | |
|  | ≤ 8.02 | (≤ 215) | (≤ 898) | 30 | 124 | *EST | 6523-29-1 |
| | IP from 82LEV/LIA. See also: 84NEL. | | | | | | |
|  | ≤ 7.43 | (≤ 192) | (≤ 802) | 20 | 85 | *EST | 283-47-6 |
|  | ≤ 7.75 | (≤ 191) | (≤ 800) | 12 | 52 | *EST | 281-17-4 |
|  | (7.63) | (204) | (853) | 28 | 117 | *EST | 5721-43-7 |
| | IP from 82LEV/LIA. See also: 84NEL. | | | | | | |
| $\text{C}_7\text{H}_{14}\text{N}_2\text{S}^+$ | | | | | | | |
|  | (7.5) | (177) | (742) | 4 | 18 | *EST | 30826-80-3 |
| | IP is onset of photoelectron band (80AND/DEV). | | | | | | |

Table 1. Positive Ion Table - Continued

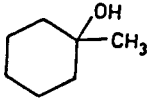
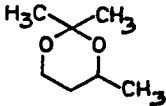
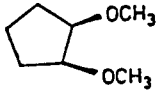
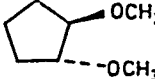
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₇H₁₄O⁺ | | | | | | | |
| n-C ₆ H ₁₃ CHO | (9.65±0.02) | (159) | (667) | -63±1 | -264±4 | 77PED/RYL | 111-71-7 |
| (CH ₃) ₂ CHCH(C ₂ H ₅)CHO | (9.44) IP from 81HOL/FIN. | (149) | (624) | -69 | -287 | *EST | 26254-92-2 |
| n-C ₅ H ₁₁ COCH ₃ | 9.30±0.01 | 142 | 596 | -72 | -301 | 75TRC | 110-43-0 |
| n-C ₄ H ₉ COC ₂ H ₅ | (9.22±0.04) IP is average of values from 81HOL/FIN, 86TRA/MCA. | (141) | (590) | -71 | -299 | 75TRC | 106-35-4 |
| (n-C ₃ H ₇) ₂ CO | 9.10±0.04 | 138 | 578 | -72 | -300 | 75TRC | 123-19-3 |
| (CH ₃) ₂ CHCH ₂ CH ₂ COCH ₃ | (9.28±0.01) | (140) | (587) | -74 | -308 | *EST | 110-12-3 |
| CH ₃ (CH ₂) ₂ CH(CH ₃)COCH ₃ | (9.20±0.02) IP is average of values from 81HOL/FIN, 86TRA/MCA. | (139) | (581) | -73 | -307 | *EST | 2550-21-2 |
| neo-C ₅ H ₁₁ COCH ₃ | (9.23±0.01) | (137) | (571) | -76 | -319 | *EST | 590-50-1 |
| C ₂ H ₅ C(CH ₃) ₂ COCH ₃ | (9.02±0.01) | (133) | (555) | -75 | -315 | *EST | 20669-04-9 |
| (iso-C ₃ H ₇) ₂ CO | 8.95±0.01 | 132 | 552 | -74.4±0.3 | -311.3±1.1 | 77PED/RYL | 565-80-0 |
|  | (9.8±0.2) | (140) | (588) | -86 | -358 | 85WIB/WAS | 590-67-0 |
| C₇H₁₄O₂⁺ | | | | | | | |
|  | ≤9.63 IP from 84ASF/ZYK. | ≤111 | ≤463 | -111 | -466 | 77PED/RYL | 696-79-7 |
|  | (8.6) IP is onset of photoelectron band. | (113) | (472) | -86 | -358 | *EST | 61011-51-6 |
|  | (8.7) IP is onset of photoelectron band. | (115) | (481) | -86 | -358 | *EST | 29887-56-7 |

Table 1. Positive Ion Table - Continued

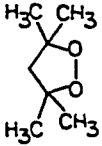
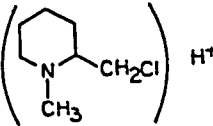
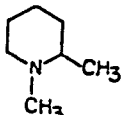
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{14}\text{O}_2^+$ |  | 8.9 | (139) | (583) | -66 | -276 | *EST | 22431-90-9 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_7\text{H}_{15}^+$ | | | (183) | (766) | 4 | 15 | *EST | |
| 1- C_7H_{15} | | From appearance potential measurements (82MAC). $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 100.5$ kcal/mol. | | | | | | |
| 2- C_7H_{15} | | (6.95) | (162) | (678) | 2 | 8 | *EST | |
| | | From appearance potential measurements (82MAC). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. $\Delta_f H(\text{Neutral})$ based on $D[\text{C-H}] = 99$ kcal/mol. | | | | | | |
| $(\text{CH}_3)_2\text{CCH}_2\text{CH}_2\text{CH}_2\text{CH}_3$ | | | 147 | 615 | | | | 40626-78-6 |
| | | From appearance potential measurement (84LOS/HOL). | | | | | | |
| $(\text{CH}_3)_2\text{CCH}_2\text{CH}(\text{CH}_3)_2$ | | | 148.4 | 620.9 | | | | 35443-14-2 |
| | | From hydride transfer equilibrium constant determinations relative to $\Delta_f H(\text{tert-C}_4\text{H}_9^+)$ (76MAU/SOL). | | | | | | |
| $(\text{C}_2\text{H}_5)_3\text{C}$ | | | 150.6 | 630.1 | | | | 28013-53-8 |
| | | From hydride transfer equilibrium constant determinations relative to $\Delta_f H(\text{tert-C}_4\text{H}_9^+)$ (76MAU/SOL). | | | | | | |
| $(\text{CH}_3)_3\text{CC}(\text{CH}_3)_2$ | | | 144.5 | 604.5 | | | | 24436-96-2 |
| | | From hydride transfer equilibrium constant determinations relative to $\Delta_f H(\text{tert-C}_4\text{H}_9^+)$ (76MAU/SOL). | | | | | | |
| $\text{C}_7\text{H}_{15}\text{ClN}^+$ |  | | (115) | (481) | | | | |
| | | From proton affinity of Pyrrolidine, 2-chloromethyl-1-methyl- (RN 49665-74-9). PA = (227.6) kcal/mol, (952.) kJ/mol. | | | | | | |
| $\text{C}_7\text{H}_{15}\text{N}^+$ | | | | | | | | |
| (E)- $\text{C}_2\text{H}_5\text{C}(\text{N}(\text{CH}_3)_2)=\text{CHCH}_3$ | | (≤ 7.61) | (≤ 173) | (≤ 724) | -2 | -10 | *EST | 32317-47-8 |
| | | IP from 81MUL/PRE2. | | | | | | |
|  | | (7.63) | (157) | (658) | -19 | -78 | *EST | 671-36-3 |
| | | IP from 82ROZ/HOU. | | | | | | |

Table 1. Positive Ion Table - Continued

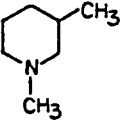
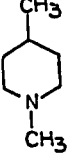
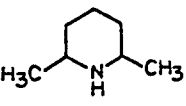
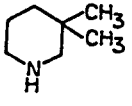
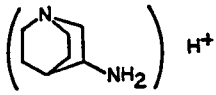
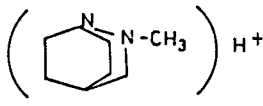
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{15}\text{N}^+$ | | | | | | | |
|  | (7.76) IP from 82ROZ/HOU. | (160) | (669) | -19 | -80 | *EST | 695-35-2 |
|  | (7.79) IP from 82ROZ/HOU. | (161) | (672) | -19 | -80 | *EST | 695-15-8 |
|  | (7.93) IP from 82ROZ/HOU. | (155) | (648) | -28 | -117 | *EST | 766-17-6 |
|  | (8.05) IP from 82ROZ/HOU. | (160) | (670) | -26 | -107 | *EST | 1193-12-0 |
| $\text{C}_7\text{H}_{15}\text{N}_2^+$ | | | | | | | |
|  | | 138 | 577 | | | | |
| | | From proton affinity of 3-amino-1-azabicyclo[2.2.2]octane (RN 6238-14-8). PA = (231.8) kcal/mol, (970) kJ/mol. | | | | | |
|  | | 165 | 690 | | | | |
| | | From proton affinity of 2-methyl-1,2-diazabicyclo[2.2.2]octane (RN 6523-29-1). (84MAU/NEL). PA = 230.4 kcal/mol, 964. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{15}\text{O}^+$ ($i\text{-C}_3\text{H}_7$) ₂ COH | | 87 | 363 | | | | |
| | | From proton affinity of ($i\text{-C}_3\text{H}_7$) ₂ CO (RN 565-80-0). PA = 204.9 kcal/mol, 857. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{16}^+$ $n\text{-C}_7\text{H}_{16}$ | 9.92±0.05 | 184 | 770 | -44.8±0.1 | -187.5±0.5 | 74SCO | 142-82-5 |
| | | 194 | 811 | -34.8±0.1 | -145.7±0.5 | | |
| | This value of IP from charge transfer equilibrium constant determinations (76LIA/AUS, 82LIA). | | | | | | |

Table 1. Positive Ion Table - Continued


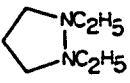
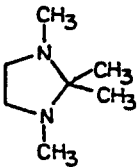
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{16}\text{N}^+$ $(\text{CH}_3)_2\text{NC}(\text{C}_2\text{H}_5)_2$ | | 127 | 531 | | | | |
| | | From proton affinity of (E)- $(\text{CH}_3)_2\text{NC}(\text{C}_2\text{H}_5)=\text{CHCH}_3$ (RN 78733-73-0). PA = 236.4 kcal/mol, 989. kJ/mol. | | | | | |
| $\text{C}_7\text{H}_{16}\text{N}_2^+$ | | | | | | | |
|  | (6.83) IP from 86RUM. See also: 84NEL. | (170) | (710) | 12 | 51 | *EST | 49840-60-0 |
|  | (≤ 8.06) See also: 84NEL. | (≤ 198) | (≤ 828) | 12 | 50 | *EST | 22825-58-7 |
|  | (7.2) IP is onset of photoelectron band (82WOR/HAR). | (196) | (822) | 30 | 127 | *EST | 33709-65-8 |
| $\text{C}_7\text{H}_{16}\text{O}^+$ n- $\text{C}_7\text{H}_{15}\text{OH}$ | (9.84 \pm 0.03) IP from 77ASH/BUR. | (147) | (614) | -80.2 \pm 0.4 | -335.5 \pm 1.5 | 77PED/RYL | 111-70-6 |
| $\text{CH}_3(\text{CH}_2)_4\text{CHOHCH}_3$ | (9.70 \pm 0.03) IP from 77ASH/BUR. | (139) | (582) | -85 | -354 | 84WIB/WAS | 543-49-7 |
| $\text{CH}_3(\text{CH}_2)_3\text{CHOHCH}_2\text{CH}_3$ | 9.68 \pm 0.03 IP from 77ASH/BUR. | 139 | 580 | -85 | -354 | 84WIB/WAS | 589-82-2 |
| $\text{CH}_3(\text{CH}_2)_2\text{CHOH}(\text{CH}_2)_2\text{CH}_3$ | (9.61 \pm 0.03) IP from 77ASH/BUR. | (137) | (573) | -85 | -354 | 84WIB/WAS | 589-55-9 |
| n- $\text{C}_5\text{H}_{11}\text{OC}_2\text{H}_5$ | (≤ 9.49) IP from 80BAC/MOU. | (≤ 144) | (≤ 602) | -75 | -314 | *EST | 17952-11-3 |
| (i- C_3H_7)O(t- C_4H_9) | (≤ 9.20) IP from 79AUE/BOW. Authors suggest that adiabatic IP is lower than cited vertical value by 25-29 kJ/mol, 6-7 kcal/mol. | (≤ 131) | (≤ 548) | -81 | -339 | *EST | 17348-59-3 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{16}\text{O}_2^+$ n- $\text{C}_7\text{H}_{15}\text{OOH}$ | (9.48±0.03) | (150) | (626) | -69 | -289 | *EST | 764-81-8 |
| | IP from 77ASH/BUR. | | | | | | |
| n- $\text{C}_5\text{H}_{11}\text{CH}(\text{CH}_3)\text{OOH}$ | (9.30±0.03) | (148) | (621) | -66 | -276 | *EST | 762-46-9 |
| IP from 77ASH/BUR. | | | | | | | |
| $\text{C}_7\text{H}_{17}\text{N}^+$ (C_2H_5) ₂ (n- C_3H_7)N | (7.67) | (150) | (626) | -27 | -114 | *EST | 4458-31-5 |
| | IP from 79AUE/BOW. Ion heat of formation predicted from hydrogen affinities of tertiary amine ions: 146 kcal/mol, 611 kJ/mol, corresponding to IP of 7.50 eV. | | | | | | |
| $\text{C}_7\text{H}_{17}\text{O}^+$ (i- C_3H_7)OH(t- C_4H_9) | | (76) | (317) | | | | |
| | From proton affinity of (i- C_3H_7)O(t- C_4H_9) (RN 17348-59-3). PA = (208.8) kcal/mol, (874.) kJ/mol. | | | | | | |
| $\text{C}_7\text{H}_{17}\text{O}_2^+$ $\text{CH}_3\text{OH}(\text{CH}_2)_5\text{OCH}_3$ | | 40 | 167 | | | | |
| | From proton affinity of $\text{CH}_3\text{O}(\text{CH}_2)_5\text{OCH}_3$ (RN 111-89-7). PA = 221.8 kcal/mol, 928. kJ/mol. | | | | | | |
| $\text{C}_7\text{H}_{18}\text{N}^+$ n- $\text{C}_7\text{H}_{15}\text{NH}_3$ | | 111 | 463 | | | | |
| | From proton affinity of n- $\text{C}_7\text{H}_{15}\text{NH}_2$ (RN 111-68-2). PA = 219.0 kcal/mol, 916. kJ/mol. | | | | | | |
| (CH_3) ₂ (neo- C_5H_{11})NH | | 108 | 450 | | | | |
| From proton affinity of (CH_3) ₂ (neo- C_5H_{11})N (RN 10076-31-0). PA = 229.9 kcal/mol, 962. kJ/mol. | | | | | | | |
| (C_2H_5) ₂ (n- C_3H_7)NH | | (106) | (445) | | | | |
| From proton affinity of (C_2H_5) ₂ (n- C_3H_7)N (RN 4458-31-5). PA = (232.0) kcal/mol, (971.) kJ/mol. | | | | | | | |
| $\text{C}_7\text{H}_{18}\text{N}_2^+$ (CH_3) ₂ N(CH ₂) ₃ N(CH ₃) ₂ | (7.6) | (168) | (704) | -7 | -29 | *EST | 110-95-2 |
| | IP is onset of photoelectron spectrum (81LIV/ROB). | | | | | | |
| (C_2H_5) ₂ NN(CH ₃)(C_2H_5) | (8.02) | (189) | (791) | 4 | 17 | *EST | 50599-43-4 |
| Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL, 79NEL/KES. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{18}\text{N}_2^+$ (n-C ₄ H ₉)(CH ₃)NN(CH ₃) ₂ | (6.63) | (156) | (652) | 3 | 12 | *EST | 52598-10-4 |
| | IP from charge transfer equilibrium constant determination (84MAU/NEL). See also: 80NEL/KES, 84NEL. | | | | | | |
| (t-C ₄ H ₉)(CH ₃)NN(CH ₃) ₂ | (6.80) | (159) | (667) | 3 | 11 | *EST | 60678-73-1 |
| | IP from charge transfer equilibrium constant determination (84MAU/NEL). | | | | | | |
| $\text{C}_7\text{H}_{18}\text{Si}^+$ (CH ₃) ₃ CSi(CH ₃) ₃ | (9.34±0.06) | (170) | (713) | -45 | -188 | 72TRI/ALL | 5037-65-0 |
| $\text{C}_7\text{H}_{18}\text{Si}_2^+$ CH ₂ =CHSi(CH ₃) ₂ Si(CH ₃) ₃ | (≤8.56) | (≤138) | (≤577) | -60 | -249 | *EST | 1112-06-7 |
| | IP from 81KHV/ZYK. | | | | | | |
| $\text{C}_7\text{H}_{18}\text{Sn}^+$ (C ₂ H ₅) ₃ (CH ₃)Sn | (≤8.95) | (≤152) | (≤638) | -54±1 | -226±4 | 80TEL/RAB | 2097-60-1 |
| (CH ₃) ₃ (tert-C ₄ H ₉)Sn | (8.0) | (168) | (705) | -16±1 | -67±6 | 77PED/RYL | 3531-47-3 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_7\text{H}_{19}\text{NSi}^+$ (CH ₃) ₃ SiN(C ₂ H ₅) ₂ | (7.68) | (97) | (406) | -80±2 | -335±8 | 80TEL/RAB | 996-50-9 |
| | IP from 83MOL/PIK3. | | | | | | |
| $\text{C}_7\text{H}_{19}\text{N}_2^+$ H ₃ N(CH ₂) ₇ NH ₂ | | 100 | 419 | | | | |
| | From proton affinity of H ₂ N(CH ₂) ₇ NH ₂ (RN 646-19-5). PA = 238. kcal/mol, 996. kJ/mol. | | | | | | |
| (CH ₃) ₂ NH(CH ₂) ₃ N(CH ₃) ₂ | | 116 | 484 | | | | |
| | From proton affinity of (CH ₃) ₂ N(CH ₂) ₃ N(CH ₃) ₂ (RN 110-95-2). PA = 243. kcal/mol, 1017. kJ/mol. | | | | | | |
| (n-C ₄ H ₉)(CH ₃)NHN(CH ₃) ₂ | | 139 | 580 | | | | |
| | From proton affinity of (n-C ₄ H ₉)(CH ₃)NN(CH ₃) ₂ (RN 52598-10-4). (84MAU/NEL). PA = 230.0 kcal/mol, 962. kJ/mol. | | | | | | |
| (t-C ₄ H ₉ (CH ₃)NHN(CH ₃) ₂ | | 139 | 582 | | | | |
| | From proton affinity of (t-C ₄ H ₉)(CH ₃)NN(CH ₃) ₂ (RN 60678-73-1). (84MAU/NEL). PA = 229.3 kcal/mol, 959. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

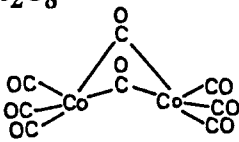
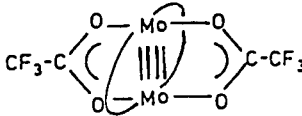
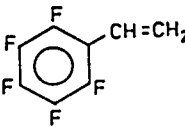
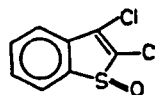
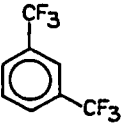
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------------|------------------------------|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_7\text{H}_{20}\text{NSi}^+$ $(\text{CH}_3)_3\text{Si}(\text{CH}_2)_2\text{NH}(\text{CH}_3)_2$ | | 79 | 332 | | | | |
| | | From proton affinity of $(\text{CH}_3)_3\text{Si}(\text{CH}_2)_2\text{N}(\text{CH}_3)_2$ (RN 23138-94-5). (84MAU/NEL). PA = 231.8 kcal/mol, 970. kJ/mol. | | | | | |
| $\text{C}_8\text{Co}_2\text{O}_8^+$  | (8.12 ± 0.22) | (-96) | (-402) | -283 ± 2 | -1185 ± 8 | 82PIL/SKI | 10210-68-1 |
| $\text{C}_8\text{F}_{12}\text{Mo}_2\text{O}_8^+$  | 8.07 | (-818) | (-3421) | -1004 | -4200 | *EST | 36608-07-8 |
| | IP is onset of photoelectron band (82BAN/PEL). | | | | | | |
| $\text{C}_8\text{F}_{18}\text{O}^+$ $(n\text{-C}_4\text{F}_9)_2\text{O}$ | 12.68 | -658.5 | -2755.2 | -950.9 ± 1 | -3978.6 ± 3 | 77PED/RYL | 308-48-5 |
| | IP from 83MOL/PIK. | | | | | | |
| C_8H_2^+ $\text{CH}=\text{CC}=\text{CC}=\text{CC}=\text{CH}$ | (9.09 ± 0.02) | (416) | (1741) | 207 | 864 | *EST | 6165-96-4 |
| $\text{C}_8\text{H}_3\text{F}_5^+$  | (9.18 ± 0.02) | (35) | (145) | -177 | -741 | *EST | 653-34-9 |
| $\text{C}_8\text{H}_4\text{Cl}_2\text{OS}^+$  | ≤ 9.00 | (≤ 205) | (≤ 857) | -3 | -11 | *EST | 30834-33-4 |
| | IP from 82BEN/DUR. | | | | | | |
| $\text{C}_8\text{H}_4\text{F}_6^+$  | 10.57 | (-61) | (-255) | -305 | -1275 | *EST | 402-31-3 |
| | IP from 82CAB/COW. | | | | | | |

Table 1. Positive Ion Table - Continued

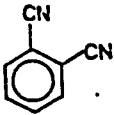
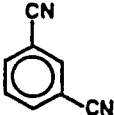
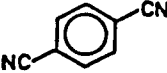
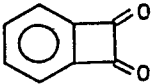
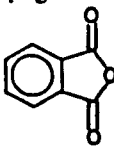
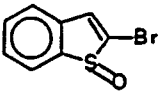
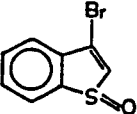
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|------------------------------------|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_4\text{N}_2^+$ | | | | | | | | |
|  | | 9.9 | (316) | (1323) | 88 ± 0.5 | 368 ± 2 | 80SAT/SAK | 91-15-6 |
| | | IP is onset of photoelectron band. | | | | | | |
|  | | 10.2 | (322) | (1347) | 87 ± 0.5 | 363 ± 2 | 80SAT/SAK | 626-17-5 |
| | | IP is onset of photoelectron band. | | | | | | |
|  | | 10.10 | 318 | 1331 | 85 ± 0.5 | 357 ± 2 | 80SAT/SAK | 623-26-7 |
| $\text{C}_8\text{H}_4\text{O}_2^+$ | | | | | | | | |
|  | | ≤ 9.23 | (≤ 237) | (≤ 992) | 24 | 101 | *EST | 6383-11-5 |
| $\text{C}_8\text{H}_4\text{O}_3^+$ | | | | | | | | |
|  | | (10.0) | (142) | (594) | -89 ± 0.5 | -371 ± 2 | 77PED/RYL | 85-44-9 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_8\text{H}_5\text{BrOS}^+$ | | | | | | | | |
|  | | ≤ 9.10 | (≤ 225) | (≤ 941) | 15 | 63 | *EST | 57147-27-0 |
| | | IP from 82BEN/DUR. | | | | | | |
|  | | ≤ 8.95 | (≤ 221) | (≤ 927) | 15 | 63 | *EST | 57147-26-9 |
| | | IP from 82BEN/DUR. | | | | | | |

Table 1. Positive Ion Table - Continued

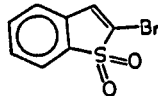
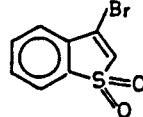
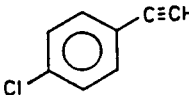
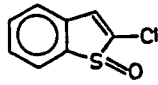
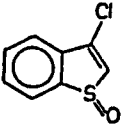
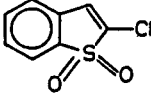
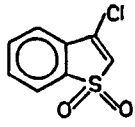
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_5\text{BrO}_2\text{S}^+$ | | | | | | | |
|  | ≤ 9.10 IP from 82BEN/DUR. | (≤ 171) | (≤ 715) | -39 | -163 | *EST | 5350-05-0 |
|  | ≤ 9.40 IP from 82BEN/DUR. | (≤ 178) | (≤ 744) | -39 | -163 | *EST | 16957-97-4 |
| $\text{C}_8\text{H}_5\text{Cl}^+$ | | | | | | | |
|  | (8.6) IP is onset of photoelectron band. | (264) | (1104) | 65 | 274 | *EST | 873-73-4 |
| $\text{C}_8\text{H}_5\text{ClO}_2\text{S}^+$ | | | | | | | |
|  | ≤ 9.10 IP from 82BEN/DUR. | (≤ 212) | (≤ 888) | 2 | 10 | *EST | 57147-28-1 |
|  | ≤ 8.95 IP from 82BEN/DUR. | (≤ 209) | (≤ 874) | 2 | 10 | *EST | 63724-95-8 |
| $\text{C}_8\text{H}_5\text{ClO}_2\text{S}^+$ | | | | | | | |
|  | ≤ 9.25 IP from 82BEN/DUR. | (≤ 162) | (≤ 676) | -52 | -216 | *EST | 10133-41-2 |
|  | ≤ 9.45 IP from 82BEN/DUR. | (≤ 166) | (≤ 696) | -52 | -216 | *EST | 21211-29-0 |

Table 1. Positive Ion Table - Continued

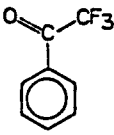
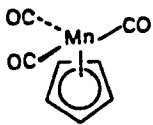
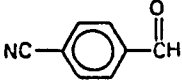
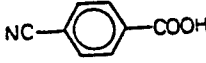
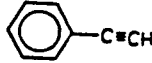
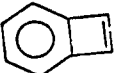
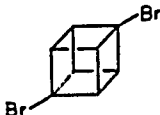
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_5\text{F}_3\text{O}^+$  | (9.72) IP from 79MCL/TRA. | (61) | (256) | -163 | -682 | *EST | 434-45-7 |
| $\text{C}_8\text{H}_5\text{MnO}_3^{+}$  | (7.6) IP is onset of photoelectron band (81CAL/HUB). See also: 86LIC/KEL. | (61) | (257) | -114±2 | -476±8 | 77PED/RYL | 12079-65-1 |
| $\text{C}_8\text{H}_5\text{NO}^+$  | ≤10.10 IP from 85GAL/GER. | (≤258) | (≤1081) | 25 | 107 | *EST | 105-07-7 |
| $\text{C}_8\text{H}_5\text{NO}_2^+$  | (10.0) IP from 84TOB/TAJ. | (193) | (807) | -38 | -158 | *EST | 619-65-8 |
| C_8H_6^+ (E),(E)-HC=CCH=CHCH=CHC≡CH (7.8) IP from 74KOP/SCH. | (7.8) IP from 74KOP/SCH. | (278) | (1161) | 98 | 409 | *EST | 53477-04-6 |
|  | 8.81±0.04 See also: 80BOC/AYG, 74KOP/SCH, 81ELB/LIE. | 276 | 1156 | 73±0.5 | 306±2 | 85DAV/ALL | 536-74-3 |
|  | (≤7.5) IP is onset of photoelectron band. | (≤291) | (≤1218) | 118 | 494 | 85DEW/MER | 4026-23-7 |
| $\text{C}_8\text{H}_6\text{Br}_2^+$  | (9.0) IP is onset of photoelectron band (85HON/HEI). | (358) | (1496) | 150 | 628 | *EST | 59346-70-2 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₆Cl⁺ | | | | | | | | |
| | | | 235 | 984 | | | | |
| | | | From proton affinity of 3-ClC ₆ H ₄ C=CH (RN 766-83-6). (85MAR/MOD). PA = 196.1 kcal/mol, 820. kJ/mol. | | | | | |
| | | | 232 | 970 | | | | |
| | | | From proton affinity of 4-ClC ₆ H ₄ C=CH (RN 873-73-4). (85MAR/MOD). PA = 199.4 kcal/mol, 834. kJ/mol. | | | | | |
| C₈H₆ClN⁺ | | | | | | | | |
| | | (9.48±0.05) | (256) | (1071) | 37 | 156 | *EST | |
| | | (9.43±0.05) | (255) | (1066) | 37 | 156 | *EST | |
| C₈H₆Cl₂⁺ | | | | | | | | |
| | | (9.15) | (337) | (1411) | 126 | 528 | *EST | |
| | | | IP is onset of photoelectron band (85HON/HEI). | | | | | |
| C₈H₆F⁺ | | | | | | | | |
| | | | 196 | 819 | | | | |
| | | | From proton affinity of 3-FC ₆ H ₄ C=CH (RN 2561-17-3). (85MAR/MOD). PA = 195.4 kcal/mol, 818. kJ/mol. | | | | | |
| | | | 190 | 797 | | | | |
| | | | From proton affinity of 4-FC ₆ H ₄ C=CH (RN 766-98-3). (85MAR/MOD). PA = 200.8 kcal/mol, 840. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

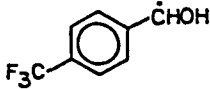
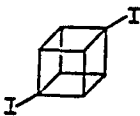
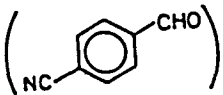
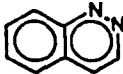
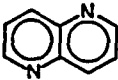

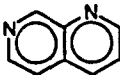

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_6\text{F}_3\text{O}^+$  | | 2 | 10 | | | | |
| | | From proton affinity of 4-CF ₃ C ₆ H ₄ CHO (RN 455-19-6). PA = 191.9 kcal/mol, 799. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_6\text{I}_2^+$  | (8.7) | (379) | (1587) | 179 | 748 | *EST | |
| | IP is onset of photoelectron band (85HON/HEI). | | | | | | |
| $\text{C}_8\text{H}_6\text{NO}^+$  | | 204 | 855 | | | | |
| | From proton affinity of 4-(CN)C ₆ H ₄ CHO (RN 105-07-7). PA = 187.0 kcal/mol, 782. kJ/mol. | | | | | | |
| $\text{C}_8\text{H}_6\text{N}_2^+$  | (8.2) | (270) | (1129) | 81±2 | 338±10 | *EST | 253-66-7 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (8.8) | (267) | (1116) | 64 | 267 | *EST | 254-79-5 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (9.0) | (271) | (1135) | 64 | 267 | *EST | 253-72-5 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (8.99) | (271) | (1134) | 64 | 267 | *EST | 253-69-0 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (8.8) | (267) | (1116) | 64 | 267 | *EST | 254-60-4 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued






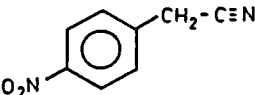
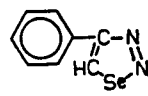
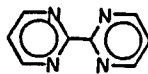
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|---------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_6\text{N}_2^+$ | | | | | | | |
|  | (8.4) IP is onset of photoelectron band. | (274) | (1148) | 81 | 338 | *EST | 253-52-1 |
|  | (≤ 8.8) IP is onset of photoelectron band. | (267) | (1116) | 64 | 267 | *EST | 253-50-9 |
|  | (8.8) IP is onset of photoelectron band. | (267) | (1116) | 64 | 267 | *EST | 253-45-2 |
|  | 9.00 ± 0.02 | (269) | (1125) | 61 | 257 | *EST | 253-82-7 |
|  | 9.01 ± 0.02 | 271 | 1131 | 63 ± 1 | 262 ± 4 | 81STE/BAR | 91-19-0 |
| $\text{C}_8\text{H}_6\text{N}_2\text{O}_2^+$ | | | | | | | |
|  | (10.11 ± 0.04) | (274) | (1146) | 41 ± 1 | 171 ± 4 | *EST | 555-21-5 |
| $\text{C}_8\text{H}_6\text{N}_2\text{Se}^+$ | | | | | | | |
|  | (8.1) IP is onset of photoelectron band (80BOC/AYG). | (295) | (1234) | 108.1 ± 2 | 452.3 ± 8 | 73ARS/SHA | 25660-64-4 |
| $\text{C}_8\text{H}_6\text{N}_4^+$ | | | | | | | |
|  | (8.3) IP is onset of photoelectron band (82BAR/CAU). | (293) | (1227) | 102 | 426 | *EST | 34671-83-5 |

Table 1. Positive Ion Table - Continued

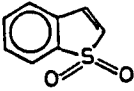
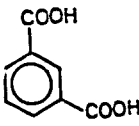
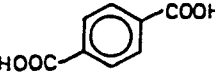
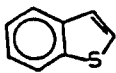
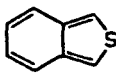

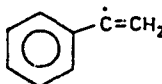
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_6\text{O}_2\text{S}^+$ | | | | | | | |
|  | (9.1) | (166) | (694) | -44 | -184 | *EST | 825-44-5 |
| | IP is onset of photoelectron band (82BEN/DUR). | | | | | | |
| $\text{C}_8\text{H}_6\text{O}_4^+$ | | | | | | | |
|  | (9.98±0.2) | (64) | (267) | -166±0.5 | -696±2 | 77PED/RYL | 121-91-5 |
|  | (9.86±0.2) | (55) | (233) | -172±0.7 | -718±3 | 77PED/RYL | 100-21-0 |
| $\text{C}_8\text{H}_6\text{S}^+$ | | | | | | | |
|  | 8.13±0.015 | 227 | 950 | 40±0.2 | 166±1 | 79SAB | 95-15-8 |
|  | (7.75) | (228) | (954) | 49 | 206 | *EST | 270-82-6 |
| $\text{C}_8\text{H}_6\text{S}_2^+$ | | | | | | | |
|  | (7.99) | (243) | (1017) | 59 | 246 | *EST | 3172-56-3 |
| C_8H_7^+ | | | | | | | |
|  | | 239 | 998 | From proton affinity of $\text{C}_6\text{H}_5\text{C}=\text{CH}$ (RN 536-74-3). (85MAR/MOD). PA = 200.2 kcal/mol, 838. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

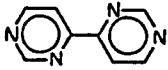
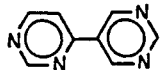
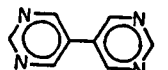
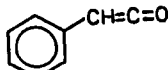
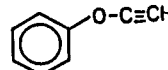
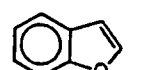
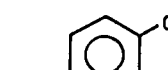

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₆N₄⁺ | | | | | | | |
|  | (9.0) IP is onset of photoelectron band (82BAR/CAU). | (309) | (1294) | 102 | 426 | *EST | 2426-94-0 |
|  | (9.0) IP is onset of photoelectron band (82BAR/CAU). | (306) | (1278) | 98 | 410 | *EST | 28648-89-7 |
|  | (9.0) IP is onset of photoelectron band (82BAR/CAU). | (306) | (1278) | 98 | 410 | *EST | 56598-46-0 |
| C₈H₆O⁺ | | | | | | | |
|  | (≤ 8.17) | (≤ 194) | (≤ 813) | 6 | 25 | 80DEM/WUL | 3496-32-0 |
|  | (8.7) | (266) | (1113) | 65 | 274 | *EST | 4279-76-9 |
|  | 8.37 \pm 0.015 | 199 | 833 | 6 \pm 2 | 26 \pm 10 ⁻ | 77PED/RYL | 271-89-6 |
| C₈H₆O₂⁺ | | | | | | | |
|  | (10.13 \pm 0.01) | (196) | (820) | -37.6 \pm 2 | -157 \pm 8 | *EST | 623-27-8 |
|  | (9.64) IP is onset of photoelectron band (85GLE/JAH). | (222) | (927) | -1 | -3 | *EST | 77627-49-7 |

Table 1. Positive Ion Table - Continued

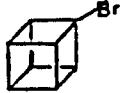
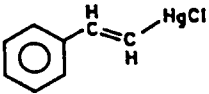
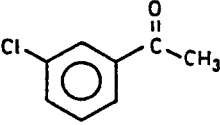
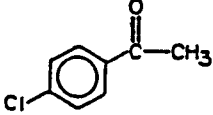
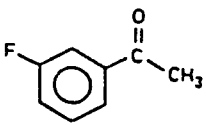
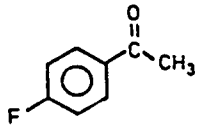
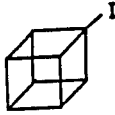
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|--|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_7\text{Br}^+$ | (8.76) | (351) | (1470) | 149 | 625 | *EST | 59346-69-9 <i>Duplicated see p 324</i> |
|  | IP is onset of photoelectron band (85HON/HEI, 84ABE/DEL). | | | | | | |
| $\text{C}_8\text{H}_7\text{ClHg}^+$ | 8.3 | (232) | (971) | 41 | 170 | *EST | 36525-03-8 |
|  | IP is onset of photoelectron band (81BAI/CHI). | | | | | | |
| $\text{C}_8\text{H}_7\text{ClO}^+$ | (9.51±0.1) | (191) | (801) | -28±2 | -117±8 | *EST | 99-02-5 |
|  | | | | | | | |
|  | (8.9) | (177) | (742) | -28±2 | -117±8 | *EST | 99-91-2 IP is onset of photoelectron band. See also: 85GAL/GER, 82PFI/GER, 77ROS/DRA. |
| $\text{C}_8\text{H}_7\text{FO}^+$ | (9.76±0.1) | (158) | (662) | -67±2 | -280±8 | *EST | 455-36-7 |
|  | | | | | | | |
|  | (9.57±0.2) | (154) | (643) | -67±2 | -280±8 | *EST | 403-42-9 |
| $\text{C}_8\text{H}_7\text{I}^+$ | (8.6) | (362) | (1515) | 164 | 685 | *EST | 74725-77-2 |
|  | IP is onset of photoelectron band (84ABE/DEL). | | | | | | |

Table 1. Positive Ion Table - Continued

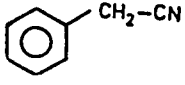
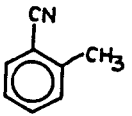
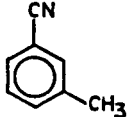
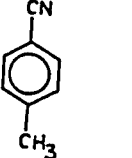
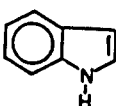
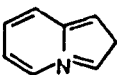
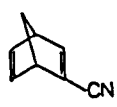
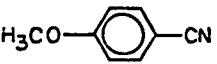
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₇N⁺ | | | | | | | |
|  | (9.34) | (260) | (1087) | 44.5 | 186 | *EST | 140-29-4 |
|  | 9.38 | (259) | (1083) | 43 | 178 | *EST | 529-19-1 |
|  | 9.34 | (259) | (1084) | 44 | 183 | *EST | 620-22-4 |
|  | 9.32 | (258) | (1081) | 44 | 182 | *EST | 104-85-8 |
|  | 7.761±0.001 IP from 85HAG/IVA. See also: 79COR. | 216 | 906 | 38±1 | 157±5 | 77PED/RYL | 120-72-9 |
|  | 7.26 | 243 | 1015 | 75.2 | 314.6 | 79COR | 274-40-8 |
|  | (≤9.26) IP from 83HOU/RON. | (≤302) | (≤1265) | 89 | 372 | *EST | 39863-20-2 |
| C₈H₇NO⁺ | | | | | | | |
|  | (8.6) IP is onset of photoelectron band (81MOD/DIS). | (213) | (892) | 15 | 62 | *EST | 874-90-8 |

Table 1. Positive Ion Table - Continued

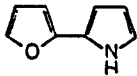
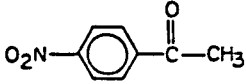
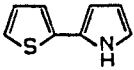
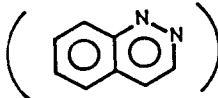
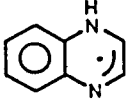



| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|----------------|---|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_7\text{NO}^+$ | | | | | | | |
|  | (6.95) IP is onset of photoelectron band (81GAL/KLA). | (182) | (760) | 21 | 89 | *EST | 63122-43-0 |
| $\text{C}_8\text{H}_7\text{NO}_3^+$ | | | | | | | |
|  | ≤ 9.98 IP from 85GAL/GER. | (≤ 206) | (≤ 861) | -24 | -102 | *EST | 100-19-6 |
| $\text{C}_8\text{H}_7\text{NS}^+$ | | | | | | | |
|  | (7.1) IP is onset of photoelectron band (81GAL/KLA). | (221) | (924) | 57 | 239 | *EST | 52707-46-7 |
| $\text{C}_8\text{H}_7\text{N}_2^+$ | | | | | | | |
|  H^+ | | (223) | (934) | From proton affinity of cinnoline (RN 253-66-7). PA = 223.2 kcal/mol, 934. kJ/mol. | | | |
|  | | (214) | (895) | From proton affinity of quinoxaline (RN 91-19-0). PA = 214.4 kcal/mol, 897. kJ/mol. | | | |
| C_8H_8^+ | | | | | | | |
|  | 8.01 ± 0.04 See also: 78FU/DUN. | 256 | 1070 | 71.1 ± 0.3 | 297.6 ± 1.3 | 77PED/RYL | 629-20-9 |
|  | (8.5) IP is onset of photoelectron band (85MEI/KON). | (287) | (1201) | 91 | 381 | 85KOL/MEI | |
|  | (8.2) IP is onset of photoelectron band (85MEI/KON). | (281) | (1176) | 92 | 385 | 85KOL/MEI | 68344-46-7 |

Table 1. Positive Ion Table - Continued


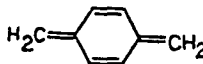
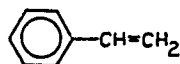
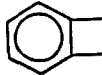

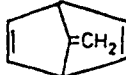

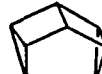
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|--|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_8H_8^+ | | | | | | | | |
|  | | (8.9) | (321) | (1346) | 116 | 487 | 78LEU/WIR | 49852-40-6 |
|  | | (7.5) | (221) | (927) | 48±4 | 203±17 | 81POL/RAI | 502-86-3 |
| | | IP is onset of photoelectron band. See also: 82DEW. | | | | | | |
|  | | 8.43±0.06 | 230 | 961 | 35.3±0.2 | 147.7±0.7 | 77PED/RYL | 100-42-5 |
| | | IP from 78FU/DUN, 81KIM/KAT. | | | | | | |
|  | | (≤8.66±0.03) | (≤248) | (≤1037) | 48±1 | 201±4 | 81ROT/SCH | 694-87-1 |
|  | | 8.23 | 263 | 1100 | 73 | 306 | 76ALL | 500-24-3 |
| | | See also: 82HAS/NEU, 83GLE/BOH. | | | | | | |
|  | | (8.5) | (274) | (1145) | 78 | 325 | *EST | 37846-63-2 |
| | | IP is onset of photoelectron band. See also: 85MAR/MAY. | | | | | | |
|  | | (8.5) | (345) | (1444) | 149±1 | 622±4 | 77PED/RYL | 277-10-1 |
| | | Values for this IP of 8.74, 8.64 (77ROS/DRA), 8.46 (82LEV/LIA) and 8.56 eV (83LIF/EAT) have been reported. | | | | | | |
|  | | (8.18) | (285) | (1189) | 96 | 400 | 81GOD/SCH | 20656-23-9 |

Table 1. Positive Ion Table - Continued

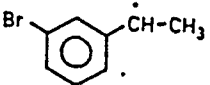
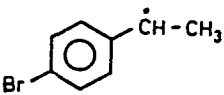
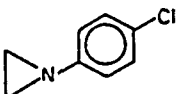
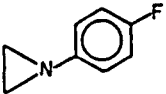
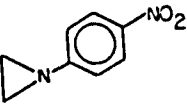
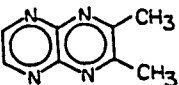
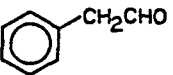
| ION Neutral | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------|--------|---|--------|------------------------------|--------|----------------------|------------------------|
| | eV | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₈Br⁺  | | | 199 | 831 | | | | |
| | | | From proton affinity of 3-BrC ₆ H ₄ CH = CH ₂ (RN 2039-86-3) (84HAR/HOU). PA = 197.4 kcal/mol, 826. kJ/mol. | | | | | |
|  | | | 195 | 815 | | | | |
| | | | From proton affinity of 4-BrC ₆ H ₄ CH = CH ₂ (RN 2039-82-9) (84HAR/HOU). PA = 201.3 kcal/mol, 842. kJ/mol. | | | | | |
| C₈H₈ClN⁺  | ≤8.3 | (≤255) | (≤1067) | 64 | 266 | | *EST | 28192-05-4 |
| | IP from 82CRI/LIC. | | | | | | | |
| C₈H₈FN⁺  | (≤8.2) | (≤213) | (≤890) | 24 | 99 | | *EST | 698-53-3 |
| | IP from 82CRI/LIC. | | | | | | | |
| C₈H₈N₂O₂⁺  | (≤8.9) | (≤273) | (≤1142) | 68 | 283 | | *EST | 30855-79-9 |
| | IP from 82CRI/LIC. | | | | | | | |
| C₈H₈N₄⁺  | (≤8.9) | (≤275) | (≤1151) | 70 | 292 | | *EST | 6499-39-4 |
| | IP from 84GLE/SPA2. | | | | | | | |
| C₈H₈O⁺  | (8.80) | (190) | (796) | -13 | -53 | | *EST | 122-78-1 |
| | See also: 81DAL/NIB. | | | | | | | |

Table 1. Positive Ion Table - Continued

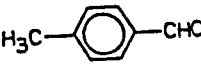
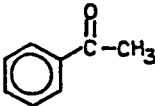
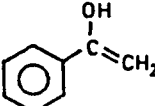
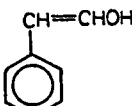
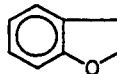
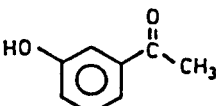
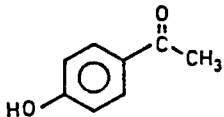
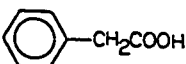
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₈O⁺ | | | | | | | |
|  | 9.33±0.05 See also: 85GAL/GER. | (197) | (825) | -18 | -75 | *EST | 104-87-0 |
|  | 9.29±0.03 See also: 81DAL/NIB, 79MCL/TRA, 85GAL/GER, 78CEN/FRA, 82PFI/GER. | 194 | 810 | -20.7±0.4 | -86.6±1.5 | 77PED/RYL | 98-86-2 |
|  | (175) (731) $\Delta_f H(\text{Ion})$ from appearance potential determination (81DAL/NIB). | | | | | | 4383-15-7 |
|  | (8.71±0.1) IP from 81DAL/NIB. | (194) | (812) | -7 | -28 | *EST | 4365-04-2 |
|  | (7.65) IP is onset of photoelectron band (81BAK/ARM). See also: 82LEV/LIA. | (163) | (683) | -13 | -55 | *EST | 496-16-2 |
| C₈H₈O₂⁺ | | | | | | | |
|  | (8.67±0.05) | (137) | (573) | -63±2 | -264±8 | *EST | |
|  | (8.70±0.03) | (138) | (575) | -63±2 | -264±8 | *EST | |
|  | (8.26) IP is onset of photoelectron band (83KLA/KOV). See also: 81MEE/WAH. | (114) | (478) | -76 | -319 | *EST | 103-82-2 |

Table 1. Positive Ion Table - Continued

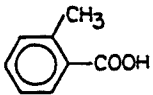
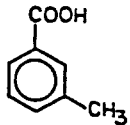
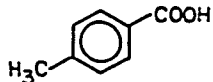
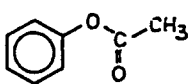
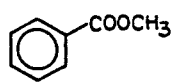
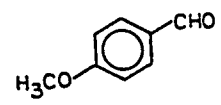
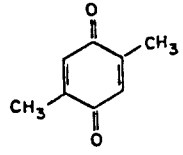
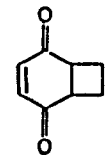
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_8\text{O}_2^+$ | | | | | | | |
|  | (9.1) IP from 81MEE/WAH. | (133) | (558) | -76.5 ± 0.2 | -320 ± 1 | 76COL/JIM | 118-90-1 |
|  | (9.43 \pm 0.2) See also: 81MEE/WAH. | (139) | (581) | -79 ± 0.2 | -329 ± 1 | 76COL/JIM | 99-04-7 |
|  | (9.23 \pm 0.2) See also: 81MEE/WAH. | (134) | (558) | -79 ± 0.2 | -332 ± 1 | 76COL/JIM | 99-94-5 |
|  | (8.6 \pm 0.05) | (131) | (550) | -66.8 ± 0.3 | -279.7 ± 1.1 | 77PED/RYL | 122-79-2 |
|  | 9.32 \pm 0.03 IP from 79MCL/TRA. See also: 81MEE/WAH, 82CAB/COW. | 146 | 611 | -69 ± 2 | -288 ± 7 | 77PED/RYL | 93-58-3 |
|  | (8.43) See also: 85GAL/GER. | (145) | (610) | -49 ± 1 | -203 ± 5 | 77PED/RYL | 123-11-5 |
|  | 9.58 | (176) | (737) | -45 | -187 | *EST | 137-18-8 |
|  | (9.3) IP is onset of photoelectron band 85GLE/JAH. | (218) | (910) | 3 | 13 | *EST | 77627-56-6 |

Table 1. Positive Ion Table - Continued

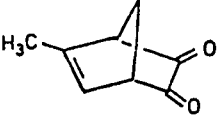
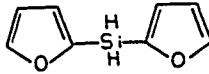
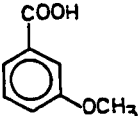
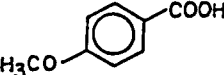
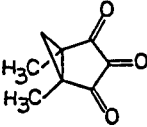
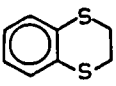
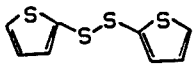
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_8\text{O}_2^+$  | (8.1) IP is onset of photoelectron band. | (160) | (669) | -27 | -113 | *EST | 60526-48-9 |
| $\text{C}_8\text{H}_8\text{O}_2\text{Si}^+$  | (8.2) IP is onset of photoelectron band (83ZYK/ERC). | (178) | (743) | -11 | -48 | *EST | 87027-12-1 |
| $\text{C}_8\text{H}_8\text{O}_3^+$  | (9.06±0.2) | (102) | (428) | -107±0.2 | -446±1 | 78COL/JIM | 586-38-9 |
|  | (9.04±0.2) | (100) | (420) | -108±0.2 | -452±1 | 78COL/JIM | 100-09-4 |
|  | (8.4) IP is onset of photoelectron band (81BEC/HOF). | (80) | (334) | -114 | -476 | *EST | 81640-32-6 |
| $\text{C}_8\text{H}_8\text{S}_2^+$  | (≤7.91) IP from 82BRE/SCH. | (≤219) | (≤916) | 37 | 153 | *EST | 6247-55-8 |
| $\text{C}_8\text{H}_8\text{S}_4^+$  | (7.5) IP is onset of photoelectron band (83BOC/ROT). | (246) | (1032) | 74 | 308 | *EST | |

Table 1. Positive Ion Table - Continued

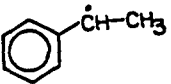
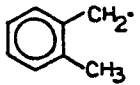
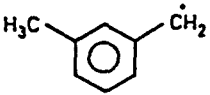
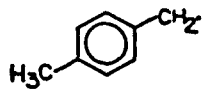
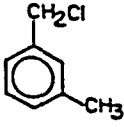
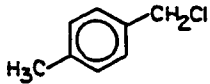
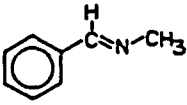
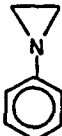
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₉⁺ | | | | | | | |
|  | (6.9) | (199) | (831) | 39.6±1.5 | 165.7±6.3 | 81ROB/STE | 2348-51-8 |
| | $\Delta_f H(\text{Ion})$ from proton affinity of C ₆ H ₅ CH=CH ₂ PA = 202.0 kcal/mol, 845. kJ/mol, and from hydride transfer equilibrium constant determinations (85SHA/SHA). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |
|  | 7.07 | (203) | (849) | 40 | 167 | 86HAY/KRU | 2348-48-3 |
| | IP from 86HAY/KRU. $\Delta_f H(\text{Ion})$ from chloride transfer equilibrium constants (85SHA/SHA). $\Delta_f H(2\text{-CH}_3\text{C}_6\text{H}_4\text{CH}_2\text{Cl})$ estimated as -3.6 kcal/mol, -15.0 KJ/mol. | | | | | | |
|  | 7.12 | (204) | (855) | 40 | 167 | 86HAY/KRU | 2348-47-2 |
| | IP from 86HAY/KRU. $\Delta_f H(\text{Ion})$ from chloride transfer equilibrium constants (85SHA/SHA); $\Delta_f H(3\text{-CH}_3\text{C}_6\text{H}_4\text{CH}_2\text{Cl})$ estimated as -3.8 kcal/mol, -15.9 kJ/mol. | | | | | | |
|  | 6.96 | (200) | (837) | 40 | 167 | 86HAY/KRU | 2348-52-9 |
| | IP from 86HAY/KRU. $\Delta_f H(\text{Ion})$ from chloride transfer equilibrium constants (85SHA/SHA); $\Delta_f H(4\text{-CH}_3\text{C}_6\text{H}_4\text{CH}_2\text{Cl})$ estimated as -3.8 kcal/mol, -15.0 KJ/mol. | | | | | | |
| C₈H₉Cl⁺ | | | | | | | |
|  | (8.82±0.03) | (200) | (835) | -4 | -16 | *EST | 620-19-9 |
|  | (8.79±0.03) | (199) | (832) | -4 | -16 | *EST | 104-82-5 |
| C₈H₉N⁺ | | | | | | | |
|  | 8.77 | (246) | (1031) | 44±2 | 185±10 | *EST | 622-29-7 |
|  | (8.0) | (256) | (1070) | 71 | 298 | *EST | 696-18-4 |
| | IP from 82ROZ/HOU2, 82CRI/LIC. | | | | | | |

Table 1. Positive Ion Table - Continued

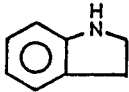
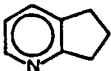
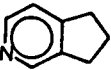
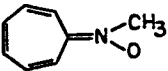
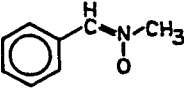
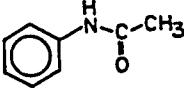
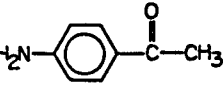
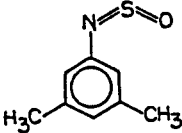
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₉N⁺ | | | | | | | |
|  | (7.15±0.02) | (212) | (888) | 47 | 198 | *EST | 496-15-1 |
|  | ≤9.15 | (≤238) | (≤994) | 27 | 111 | *EST | |
| | IP from 79AUE/BOW. Ion heat of formation predicted from hydrogen affinities of pyridines: 234 kcal/mol, 979 kJ/mol, corresponding to IP of 9.0 eV. | | | | | | |
|  | ≤9.19 | (≤239) | (≤1000) | 27 | 113 | *EST | |
| | IP from (79AUE/BOW). Ion heat of formation predicted from hydrogen affinities of pyridines: 234 kcal/mol, 979 kJ/mol, corresponding to IP of 9.0 eV. | | | | | | |
| C₈H₉NO⁺ | | | | | | | |
|  | (7.28) | (212) | (888) | 44 | 186 | *EST | 65194-06-1 |
|  | 7.89 | (207) | (866) | 25 | 105 | *EST | 3376-23-6 |
|  | (8.30) | (161) | (672) | -31±0.2 | -129±1 | 77PED/RYL | 103-84-4 |
| | Values reported for this ionization potential range from 8.18 eV to 8.60 eV. | | | | | | |
|  | (7.8±0.1) | (159) | (666) | -21 | -87 | *EST | 99-92-3 |
| | See also: 85GAL/GER. | | | | | | |
| C₈H₉NOS⁺ | | | | | | | |
|  | (8.2) | (166) | (695) | -23 | -96 | *EST | |
| | IP is onset of photoelectron band (82LOU/VAN). | | | | | | |

Table 1. Positive Ion Table - Continued

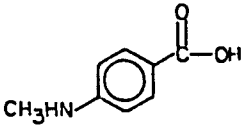
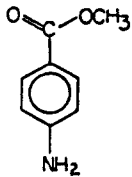
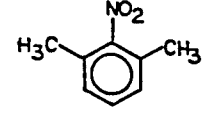
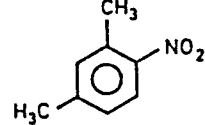
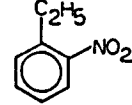
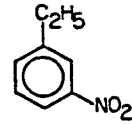
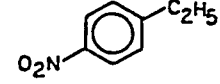
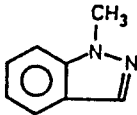
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_8H_9NO_2^+$ | | | | | | | |
|  | (7.3) IP is onset of photoelectron band (81MEE/WAH). | (99) | (412) | -70 | -292 | *EST | 10541-83-0 |
|  | (7.7) IP is onset of photoelectron band (81MEE/WAH). | (109) | (455) | -69 | -288 | *EST | 619-45-4 |
|  | 9.17±0.015 | (221) | (925) | 10 | 40 | *EST | 81-20-9 |
|  | (9.1) IP is onset of photoelectron band. | (215) | (898) | 5 | 20 | *EST | 89-87-2 |
|  | (9.39) IP from 82BAL/CAR. | (219) | (917) | 3±2 | 11±7 | 77PED/RYL | 612-22-6 |
|  | (9.64) IP from 82BAL/CAR. | (224) | (937) | 2 | 7 | *EST | 7369-50-8 |
|  | (9.71) IP from 82BAL/CAR. | (225) | (943) | 2±2 | 7±7 | 77PED/RYL | 100-12-9 |
| $C_8H_9N_2^+$ | | | | | | | |
|  | | 204 | 852 | From proton affinity of 1-methyl-1H-indazole (RN 13436-48-1) (84FLA/MAQ). PA = (221) kcal/mol, (925) kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

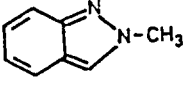
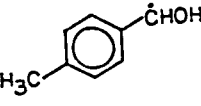
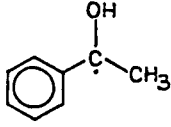
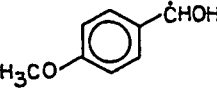
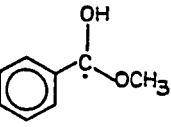


| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_9\text{N}_2^+$ | | | | | | | |
|  | | 218 | 913 | | | | |
| | | From proton affinity of 2-methyl-2H-indazole (RN 4838-00-0) (84FLA/MAQ). PA = (224) kcal/mol, (939) kJ/mol. | | | | | |
| $\text{C}_8\text{H}_9\text{O}^+$ | | | | | | | |
|  | | 144 | 603 | | | | |
| | | From proton affinity of 4-(CH_3) $\text{C}_6\text{H}_4\text{CHO}$ (RN 104-87-0). PA = 203.7 kcal/mol, 852. kJ/mol. | | | | | |
|  | | 140 | 584 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{COCH}_3$ (RN 98-86-2). PA = 205.4 kcal/mol, 859. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_9\text{O}_2^+$ | | | | | | | |
|  | | 104 | 434 | | | | |
| | | From proton affinity of 4-(CH_3O) $\text{C}_6\text{H}_4\text{CHO}$ (RN 123-11-5). PA = 213.5 kcal/mol, 893. kJ/mol. | | | | | |
|  | | 94 | 395 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{COOCH}_3$ (RN 93-58-3). PA = 203.7 kcal/mol, 852. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{10}^+$ | | | | | | | |
| (E)- $\text{CH}_2 = \text{CHCH} = \text{CHCH} = \text{CHCH} = \text{CH}_2$ | | | | | | | |
| | 7.79±0.02 | (235) | (981) | 55 | 229 | *EST | 3725-31-3 |
| | IP from 84HOL, 77ROS/DRA. | | | | | | |
| $\text{CH}_2 = \text{C}(\text{CH}_3)\text{C}=\text{CC}(\text{CH}_3)=\text{CH}_2$ | | | | | | | |
| | (8.95±0.1) | (324) | (1357) | 118 | 494 | 77LEB/R YA | 3725-05-1 |
|  | | (7.9) | (226) | (945) | 44 | 183 | 69BEN/CRU 1871-52-9 |
|  | | (8.5) | (243) | (1017) | 47 | 197 | *EST 3725-30-2 |

Table 1. Positive Ion Table - Continued


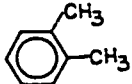
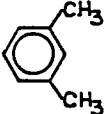
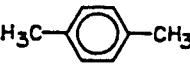
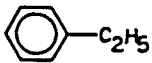
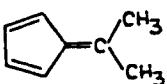
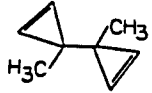
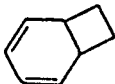
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{10}^+$ | | | | | | | | |
|  | | (8.90) | (276) | (1158) | 71 | 299 | 78LEU/WIR | 68177-00-4 |
|  | | 8.56±0.01 | 201.7 | 843.9 | 4.3±0.1 | 18.0±0.5 | 77PED/RYL | 95-47-6 |
| | | Value derived from charge transfer equilibrium constant determinations (78LIA/AUS) is in agreement. See: 84HOW/GON. | | | | | | |
|  | | 8.56±0.01 | 202 | 843 | 4.1±0.1 | 17.3±0.6 | 77PED/RYL | 108-38-3 |
| | | Value derived from charge transfer equilibrium constant determinations (78LIA/AUS) is in agreement. See: 84HOW/GON. | | | | | | |
|  | | 8.44±0.01 | 199 | 832 | 4.3±0.2 | 18.0±0.9 | 77PED/RYL | 106-42-3 |
| | | IP at 298 K from charge transfer equilibrium constant determinations (78LIA/AUS) is 8.52 eV. See: 84HOW/GON. | | | | | | |
|  | | 8.77±0.01 | 209 | 875 | 7.0±0.1 | 29.2±0.5 | 77PED/RYL | 100-41-4 |
| | | Value derived from charge transfer equilibrium constant determinations (78LIA/AUS) is in agreement. See also: 83KLA/KOV, 82SEL/HEL, 84HOW/GON. | | | | | | |
|  | | (≤8.03) | (≤217) | (≤909) | 32.1±1.3 | 134.4±5.4 | 77PED/RYL | 2175-91-9 |
|  | | (8.4) | (320) | (1340) | 127 | 530 | *EST | |
| | | IP is onset of photoelectron band (82SPA/KOR). | | | | | | |
|  | | (7.6) | (224) | (938) | 49 | 205 | *EST | 3725-28-8 |
| | | IP is onset of photoelectron band (81GLE/GUB2). | | | | | | |

Table 1. Positive Ion Table - Continued


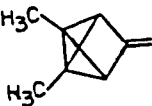

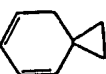

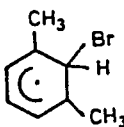
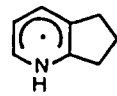
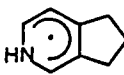
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_8H_{10}^+$ | | | | | | | |
|  | (8.5) | (230) | (964) | 34 | 144 | 76ALL | 657-23-8 |
| IP is onset of photoelectron band. See also: 82HAS/NEU. | | | | | | | |
|  | (8.0) | (273) | (1141) | 88 | 369 | *EST | 63001-13-8 |
| IP is onset of photoelectron band (84GLE/HAI). | | | | | | | |
|  | (8.4) | (242) | (1012) | 48.1 | 201.3 | 81GOD/SCH | 765-72-0 |
| IP is onset of photoelectron band. | | | | | | | |
|  | (≤ 7.89) | (≤ 242) | (≤ 1012) | 60 | 251 | *EST | 53143-64-9 |
|  | (8.20) | (246) | (1029) | 57 | 238 | *EST | 15439-15-3 |
| $C_8H_{10}Br^+$ | | | | | | | |
|  | | 178 | 743 | | | | |
| From proton affinity of 1,3,2- $C_6H_3(CH_3)_2Br$ (RN 576-22-7). PA = (199) kcal/mol, (832) kJ/mol. | | | | | | | |
| $C_8H_{10}N^+$ | | | | | | | |
|  | | 166 | 695 | | | | |
| From proton affinity of 2,3-cyclopentenopyridine. PA = (225.8) kcal/mol, (945.) kJ/mol. | | | | | | | |
|  | | 166 | 696 | | | | |
| From proton affinity of 3,4-cyclopentenopyridine. PA = (226.8) kcal/mol, (949.) kJ/mol. | | | | | | | |

Table 1. Positive Ion Table - Continued

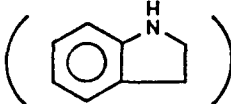
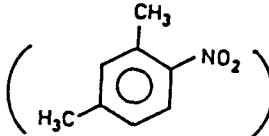
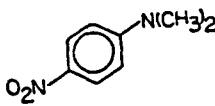
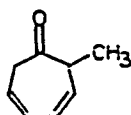
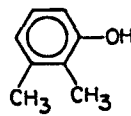
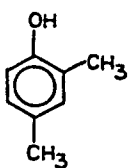
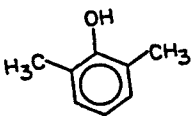
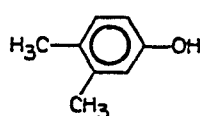
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{10}\text{N}^+$  | | 186 | 779 | | | | |
| | | From proton affinity of 2,3-dihydroindole (RN 496-15-1) (85BOL/HOU). PA = 226.7 kcal/mol, 949. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{10}\text{NO}_2^+$  | | 153 | 641 | | | | |
| | | From proton affinity of 1,3,4- $\text{C}_6\text{H}_3(\text{CH}_3)_2\text{NO}_2$ (RN 89-87-2) (84ROL/HOU). PA = 199.8 kcal/mol, 836. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{10}\text{N}_2\text{O}_2^+$  | (7.6±0.1) | (191) | (801) | 16.1±0.4 | 67.3±1.8 | 84FUR/MUR | 100-23-2 |
| $\text{C}_8\text{H}_{10}\text{O}^+$  | 8.23 IP from 83RUS/FRE. | (146) | (611) | -44 | -183 | *EST | 42104-03-0 |
|  | (8.26) IP from 83RUS/FRE. | (153) | (640) | -37.6±0.3 | -157.2±1.4 | 77PED/RYL | 526-75-0 |
|  | (8.0) IP is onset of photoelectron band. | (146) | (609) | -38.9±0.2 | -162.9±0.9 | 77PED/RYL | 105-67-9 |
|  | 8.05±0.02 | 147 | 615 | -38.7±0.2 | -161.8±1.0 | 77PED/RYL | 576-26-1 |
|  | (8.09) IP from 83RUS/FRE. | (149) | (624) | -37.4±0.3 | -156.6±1.1 | 77PED/RYL | 95-65-8 |

Table 1. Positive Ion Table - Continued

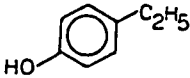
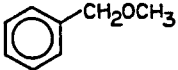
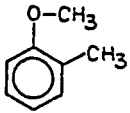
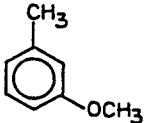
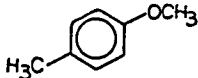
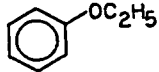
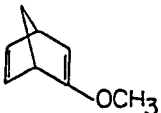
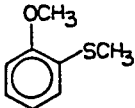
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_8H_{10}O^+$ | | | | | | | |
|  | (7.84) IP from 83RUS/FRE. | (146) | (612) | -34.4±0.2 | -144.1±1.0 | 77PED/RYL | 123-07-9 |
|  | 8.85±0.03 | 186 | 780 | -18 | -74 | 73BIL/CHO | 538-86-3 |
|  | 7.90 | (157) | (657) | -25 | -105 | *EST | 578-58-5 |
|  | (8.0) IP is onset of photoelectron band. | (160) | (668) | -25±1 | -104±5 | 77PED/RYL | 100-84-5 |
|  | 7.9 IP is onset of photoelectron band. | (158) | (662) | -24 | -100 | *EST | 104-93-8 |
|  | 8.13±0.02 | 163 | 683 | -24.3±0.1 | -101.7±0.5 | 77PED/RYL | 103-73-1 |
|  | (≤8.05) IP from 83HOU/RON. | (≤207) | (≤865) | 21 | 88 | *EST | 74437-38-0 |
| $C_8H_{10}OS^+$ | | | | | | | |
|  | (≤8.05) | (≤172) | (≤720) | -14 | -57 | *EST | 2388-73-0 |

Table 1. Positive Ion Table - Continued

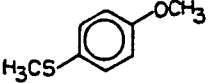
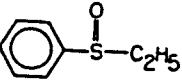
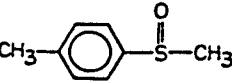
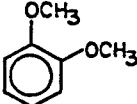
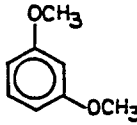
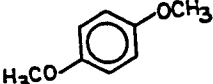
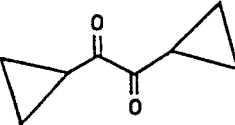
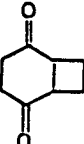
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₁₀OS⁺ | | | | | | | |
|  | ≤7.80 | (≤168) | (≤703) | -12 | -50 | *EST | 1879-16-9 |
|  | (≤8.75) IP from 81MOH/JIA. | (≤193) | (≤809) | -8 | -35 | *EST | 4170-80-3 |
|  | (≤8.70) IP from 81MOH/JIA. | (≤193) | (≤808) | -7.6 | -31.8 | *EST | 934-72-5 |
| C₈H₁₀O₂⁺ | | | | | | | |
|  | (7.8) IP is onset of photoelectron band. | (127) | (530) | -53±0.7 | -223±3 | 77PED/RYL | 91-16-7 |
|  | (7.8) IP is onset of photoelectron band. | (122) | (511) | -58 | -242 | *EST | 151-10-0 |
|  | 7.53 IP from 85OIK/ABE, 82LEV/LIA. | (118) | (493) | -56 | -234 | *EST | 150-78-7 |
|  | (8.8) IP is onset of photoelectron band. | (173) | (724) | -30 | -125 | *EST | 15940-88-2 |
|  | (9.1) IP is onset of photoelectron band (85GLE/JAH). | (154) | (645) | -56 | -233 | *EST | 54338-82-8 |

Table 1. Positive Ion Table - Continued

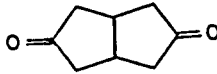
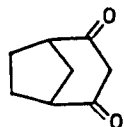
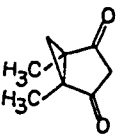
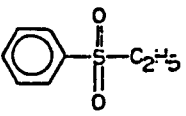
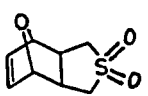
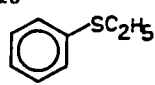
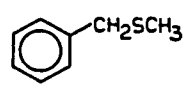
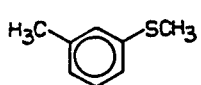
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{10}\text{O}_2^+$ | | | | | | | |
|  | (≤ 9.33) IP is onset of photoelectron band. | (≤ 138) | (≤ 576) | -77 | -324 | *EST | |
|  | (8.7) IP is onset of photoelectron band. | (127) | (532) | -73 | -307 | *EST | 74896-14-3 |
|  | (9.14) IP is onset of photoelectron band (81BEC/HOF). | (139) | (582) | -72 | -300 | *EST | 29978-55-0 |
| $\text{C}_8\text{H}_{10}\text{O}_2\text{S}^+$ | | | | | | | |
|  | (9.4) IP from 81MOH/JIA. | (150) | (628) | -67 | -279 | *EST | 599-70-2 |
| $\text{C}_8\text{H}_{10}\text{O}_3\text{S}^+$ | | | | | | | |
|  | (9.5) IP is onset of photoelectron band (84AIT/GOS). | (160) | (668) | -60 | -249 | *EST | |
| $\text{C}_8\text{H}_{10}\text{S}^+$ | | | | | | | |
|  | 7.88 \pm 0.02 | 200 | 837 | 18.4 \pm 0.6 | 77.0 \pm 2.6 | 77PED/RYL | 622-38-8 |
|  | (8.42) | (213) | (892) | 19.0 \pm 0.7 | 79.5 \pm 2.9 | 77PED/RYL | 766-92-7 |
|  | (≤ 8.00) | (≤ 200) | (≤ 838) | 16 | 66 | *EST | 4886-77-5 |

Table 1. Positive Ion Table - Continued

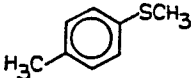
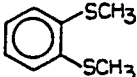
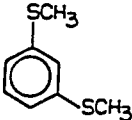
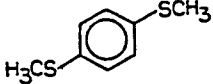
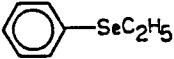
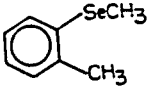
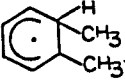
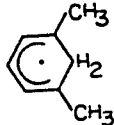
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{10}\text{S}^+$ | | | | | | | |
|  | 7.5 IP is onset of photoelectron band. | (189) | (790) | 16 | 66 | *EST | 623-13-2 |
| $\text{C}_8\text{H}_{10}\text{S}_2^+$ | | | | | | | |
|  | 7.7 IP is onset of photoelectron band (81TRA/RED, 82LEV/LIA). | (206) | (864) | 29 | 121 | *EST | 2388-68-3 |
|  | (≤ 8.0) | (≤ 211) | (≤ 885) | 27 | 113 | *EST | 2388-69-4 |
|  | (7.3) IP is onset of photoelectron band. | (195) | (817) | 27 | 113 | *EST | 699-20-7 |
| $\text{C}_8\text{H}_{10}\text{Se}^+$ | | | | | | | |
|  | (7.6) IP is onset of photoelectron band (81BAK/ARM). | (207) | (865) | 31 | 132 | *EST | 17774-38-8 |
|  | (7.5) IP is onset of photoelectron band (81BAK/ARM). | (200) | (837) | 27 | 113 | *EST | 1528-88-7 |
| $\text{C}_8\text{H}_{11}^+$ | | | | | | | |
|  | | 177 | 739 | From proton affinity of 1,2- $\text{C}_6\text{H}_4(\text{CH}_3)_2$ (RN 95-47-6). PA = 193.3 kcal/mol, 809. kJ/mol. | | | |
|  | | 174 | 727 | From proton affinity of 1,3- $\text{C}_6\text{H}_4(\text{CH}_3)_2$ (RN 108-38-3). PA = 195.9 kcal/mol, 820. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued


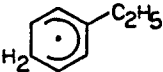
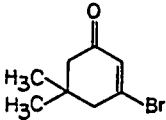
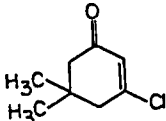
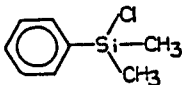
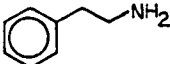
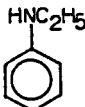
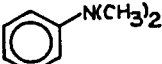
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|--|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_8H_{11}^+$ | | | | | | | |
|  | | 178 | 745 | From proton affinity of 1,4- $C_6H_4(CH_3)_2$ (RN 106-42-3). PA = 192.0 kcal/mol, 803. kJ/mol. | | | |
|  | | 181 | 757 | From proton affinity of $C_6H_5C_2H_5$ (RN 100-41-4). PA = 191.6 kcal/mol, 802. kJ/mol. | | | |
| $C_8H_{11}BrO^+$ | | | | | | | |
|  | (≤ 9.35) IP from 82PFI/GER. | (≤ 180) | (≤ 755) | -35 | -147 | *EST | 13271-49-3 |
| $C_8H_{11}ClO^+$ | | | | | | | |
|  | (9.35) IP from 82PFI/GER. | (170) | (713) | -45 | -189 | *EST | 17530-69-7 |
| $C_8H_{11}ClSi^+$ | | | | | | | |
|  | (8.93) IP from 84VES/HAR. | (156) | (652) | -50 | -210 | *EST | 768-33-2 |
| $C_8H_{11}N^+$ | | | | | | | |
|  | (8.5) IP is onset of photoelectron band. | (212) | (885) | 16 | 65 | *EST | 64-04-0 |
|  | (≤ 7.67) IP from 82ROZ/HOU2. | (≤ 190) | (≤ 796) | 13 \pm 1 | 56 \pm 6 | 77PED/RYL | 103-69-5 |
|  | 7.12 \pm 0.02 | 188 | 788 | 24 \pm 0.7 | 101 \pm 3 | 82FUR/SAK | 121-69-7 |

Table 1. Positive Ion Table - Continued

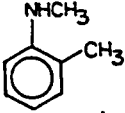
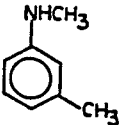
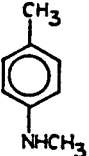
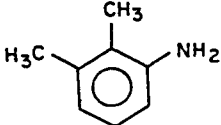
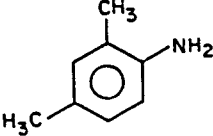
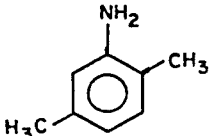
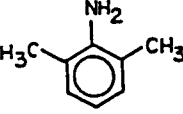
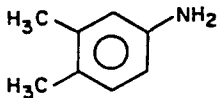
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{11}\text{N}^+$  | (7.27) | (182) | (759) | 14 | 58 | *EST | 611-21-2 |
|  | (7.26) | (180) | (753) | 13 | 53 | *EST | 696-44-6 |
|  | (7.13) | (177) | (741) | 13 | 53 | *EST | 623-08-5 |
|  | $\leq 7.77 \pm 0.05$ | (≤ 186) | (≤ 777) | 6 | 27 | *EST | 87-59-2 |
|  | ($\leq 7.65 \pm 0.05$) | (≤ 182) | (≤ 761) | 5 | 23 | *EST | 95-68-1 |
|  | 7.2 IP is onset of photoelectron band. | (172) | (718) | 5 | 23 | *EST | 95-78-3 |
|  | 7.33 ± 0.05 | (175) | (734) | 6 ± 0.2 | 27 ± 1 | *EST | 87-62-7 |
|  | ($\leq 7.68 \pm 0.05$) | (≤ 183) | (≤ 764) | 5 | 23 | *EST | 95-64-7 |

Table 1. Positive Ion Table - Continued

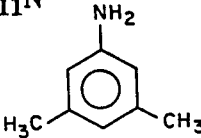
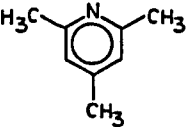
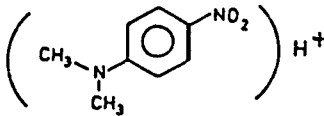
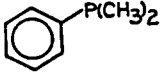

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|------------------------------------|--------------------------|----------------|---|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{11}\text{N}^+$  | 7.2 | (171) | (716) | 5 | 21 | *EST | 108-69-0 |
| | IP is onset of photoelectron band. | | | | | | |
|  | ($\leq 8.9 \pm 0.1$) | (≤ 210) | (≤ 879) | 5 | 20 | *EST | 108-75-8 |
| $\text{C}_8\text{H}_{11}\text{N}_2\text{O}_2^+$  | | 167 | 699 | From proton affinity of N,N-dimethyl-4-nitroaniline (RN 100-23-2) (84ROL/HOU). PA = 214.6 kcal/mol, 898. kJ/mol. | | | |
| $\text{C}_8\text{H}_{11}\text{P}^+$  | 7.58 ± 0.05 | (184) | (771) | 10 | 40 | *EST | 672-66-2 |
| $\text{C}_8\text{H}_{12}^+$ (E),(E)- $\text{CH}_3\text{CH}=\text{C}(\text{CH}_3)\text{CH}=\text{CHCH}=\text{CH}_2$ | (≤ 8.01) | (≤ 208) | (≤ 872) | 24 | 99 | *EST | 58434-77-8 |
| $\text{CH}_2=\text{CHCH}=\text{CHCH}(\text{CH}_3)\text{CH}=\text{CH}_2$ | (8.4 ± 0.1) | (226) | (945) | 32.2 | 134.7 | *EST | 925-52-0 |
| | IP from 84GRO/GRO. | | | | | | |
| n- $\text{C}_4\text{H}_9\text{C}\equiv\text{CCH}=\text{CH}_2$ | (8.83 ± 0.01) | (248) | (1038) | 44 ± 2 | 186 ± 7 | 78SHA | 17679-92-4 |
| (E)-n- $\text{C}_4\text{H}_9\text{CH}=\text{CHC}\equiv\text{CH}$ | (8.87 ± 0.01) | (248) | (1040) | 44 | 184 | *EST | 42104-42-7 |
| n- $\text{C}_3\text{H}_7\text{C}\equiv\text{CC}(\text{CH}_3)=\text{CH}_2$ | (8.62 ± 0.01) | (241) | (1008) | 42 | 176 | *EST | 17669-40-8 |
| $(\text{C}_2\text{H}_5)_2\text{C}=\text{CHC}\equiv\text{CH}$ | (8.54 ± 0.01) | (240) | (1004) | 43 | 180 | *EST | 2750-71-2 |
|  | (8.4) | (213) | (891) | 19 | 81 | 82KOZ/MAS | 1700-10-3 |

Table 1. Positive Ion Table - Continued



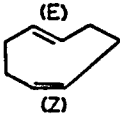

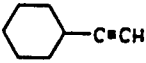
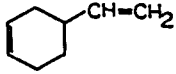
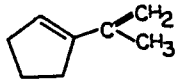
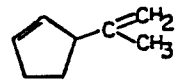
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-------------------------------------|--------------------------|---------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{12}^+$  | (8.5) | (221) | (925) | 25±2 | 105±8 | *EST | 1073-07-0 |
|  | (8.9) | (219) | (917) | 14.0±0.3 | 58.6±1.2 | 77PED/RYL | 111-78-4 |
|  IP is onset of photoelectron band. | (8.2) | (218) | (912) | 29 | 121 | *EST | 5259-71-2 |
|  | (8.9) | (248) | (1041) | 43±1 | 182±3 | 75ALL/MEY | 1781-78-8 |
|  | (≤9.92) | (≤257) | (≤1076) | 28±1 | 119±3 | 75ALL/MEY | 931-48-6 |
|  | (8.93±0.02) See also: 84GRO/GRO. | (221) | (927) | 15.6±0.3 | 65.1±1.2 | 77PED/RYL | 100-40-3 |
|  | (8.60±0.01) | (219) | (917) | 21 | 87 | *EST | 37689-19-3 |
|  | (8.89±0.02) | (227) | (950) | 22 | 92 | *EST | 14564-97-7 |

Table 1. Positive Ion Table - Continued

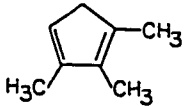
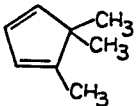
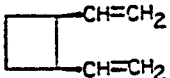
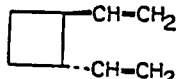

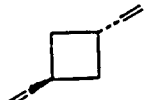
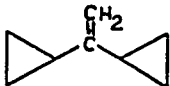

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{12}^+$ | | | | | | | |
|  | (7.96±0.05) | (192) | (804) | 9 | 36 | *EST | 3853-27-8 |
|  | (8.0±0.1) | (195) | (818) | 11 | 46 | *EST | 4249-09-6 |
|  | (≤9.22) IP from 81BIS/GLE. | (≤252) | (≤1056) | 39.8±0.8 | 166.5±3.5 | 77PED/RYL | 16177-46-1 |
|  | (≤9.20) IP from 81BIS/GLE. | (≤246) | (≤1031) | 34.3±0.8 | 143.5±3.4 | 77PED/RYL | 6553-48-6 |
|  | (8.9) IP is onset of photoelectron band (81BIS/GLE). | (247) | (1035) | 42 | 176 | *EST | 77614-53-0 |
|  | (8.9) IP is onset of photoelectron band (81BIS/GLE). | (247) | (1035) | 42 | 176 | *EST | 77614-67-6 |
|  | 8.08 | (237) | (993) | 51 | 213 | *EST | 822-93-5 |
|  | (8.92) See also: 82HAS/NEU. | (211) | (881) | 4.9±0.2 | 20.5±0.8 | 77PED/RYL | 931-64-6 |

Table 1. Positive Ion Table - Continued






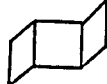


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{12}^+$  | ≤ 9.02 | ≤ 220 | ≤ 920 | 12 | 50 | 79AUE/BOW | 497-35-8 |
|  | (≤ 9.40) See also: 85MAR/MAY. | (≤ 231.2) | (≤ 967.2) | 14 ± 1 | 60 ± 3 | 77KOZ/BYC | 31463-35-1 |
|  | (≤ 8.95) | (≤ 234) | (≤ 980) | 28 | 116 | *EST | 50695-42-6 |
|  | (≤ 9.39) | (≤ 244) | (≤ 1022) | 28 | 116 | *EST | 50895-58-4 |
|  | (≤ 9.18) | (≤ 264) | (≤ 1102) | 52 ± 2 | 216 ± 8 | 73ENG/AND2 | 28636-10-4 |
|  | (≤ 9.23) | (≤ 259) | (≤ 1084) | 46 ± 2 | 193 ± 7 | 73ENG/AND2 | 13027-75-3 |
|  | (9.4) IP is onset of photoelectron band. | (238) | (997) | 21 ± 3 | 90 ± 14 | 81GOD/SCH2 | 250-21-5 |
|  | (≤ 8.44) | (≤ 228) | (≤ 954) | 33 | 140 | *EST | 7647-57-6 |

Table 1. Positive Ion Table - Continued




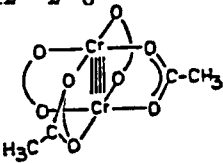
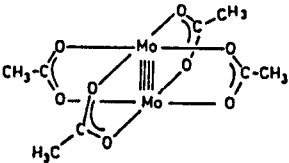
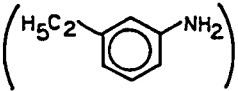
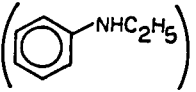
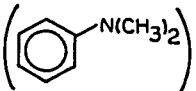
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|--|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_8H_{12}^+$ | | | | | | | |
|  | (8.65) | (233) | (975) | 33 | 140 | *EST | 14783-50-7 |
|  | (8.4) IP is onset of photoelectron band. | (266) | (1112) | 72 | 302 | *EST | 21426-37-9 |
|  | (8.8) IP is onset of photoelectron band. | (275) | (1151) | 72 | 302 | *EST | 25399-32-0 |
| $C_8H_{12}Cr_2O_8^+$ | | | | | | | |
|  | (8.0) IP is onset of photoelectron band. | (-290) | (-1212) | -474±7 | -1984±28 | 82PIL/SKI | 15020-15-2 |
| $C_8H_{12}Mo_2O_8^+$ | | | | | | | |
|  | 6.54 IP from 84LIC/BLE. | (-280) | (-1175) | -432±2 | -1806±10 | 81CAV/CON | 14221-06-8 |
| $C_8H_{12}N^+$ | | | | | | | |
|  H ⁺ | | 158 | 662 | From proton affinity of 3-C ₂ H ₅ C ₆ H ₄ NH ₂ (RN 587-02-0). PA = 214.0 kcal/mol, 895. kJ/mol. | | | |
|  H ⁺ | | 157 | 658 | From proton affinity of C ₆ H ₅ NHC ₂ H ₅ (RN 103-69-5). PA = 221.8 kcal/mol, 928. kJ/mol. | | | |
|  H ⁺ | | 166 | 696 | From proton affinity of C ₆ H ₅ N(CH ₃) ₂ (RN 121-69-7). PA = 223.4 kcal/mol, 935. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

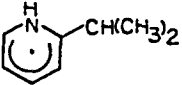
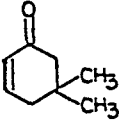
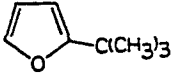
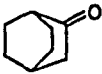
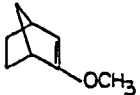
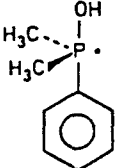
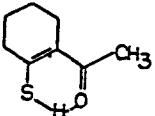
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{12}\text{N}^+$  | | 152 | 635 | | | | |
| | | From proton affinity of 2-isopropylpyridine (RN 75981-47-4). PA = 227.2 kcal/mol, 951. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{12}\text{N}_4^+$ (E)-(NCC(CH ₃) ₂) ₂ N ₂ | (9.2) | (271) | (1134) | 59±0.4 | 246±1.8 | 84LEB/GUT | 34241-39-9 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_8\text{H}_{12}\text{O}^+$  | (≤9.24) | (≤171) | (≤718) | -42 | -174 | *EST | 4694-17-1 |
| | IP from 82PFI/GER. | | | | | | |
|  | (8.38) | (157) | (656) | -37 | -153 | *EST | 7040-43-9 |
| | IP from 83ZYK/ERC. | | | | | | |
|  | (8.8) | (148) | (619) | -55±1 | -230±5 | 77PED/RYL | 2716-23-6 |
| | IP is onset of photoelectron band (81CAR/GAN). | | | | | | |
|  | (≤8.15) | (≤173) | (≤724) | -15 | -62 | *EST | 17190-90-8 |
| | IP from 83HOU/RON. | | | | | | |
| $\text{C}_8\text{H}_{12}\text{OP}^+$  | | 82 | 341 | | | | |
| | From proton affinity of (CH ₃) ₂ (C ₆ H ₅)PO (RN 10311-08-7) (86TRA/MUN). PA = 216 kcal/mol, 904 kJ/mol. | | | | | | |
| $\text{C}_8\text{H}_{12}\text{OS}^+$  | (8.0) | (125) | (523) | -60 | -249 | *EST | 76698-82-3 |
| | IP is onset of photoelectron band (81JOR/CAR). | | | | | | |

Table 1. Positive Ion Table - Continued

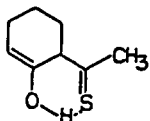
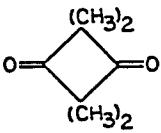
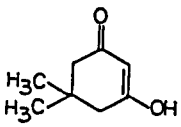
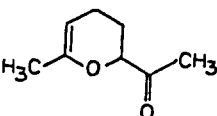
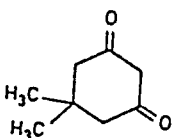
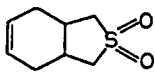
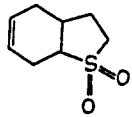
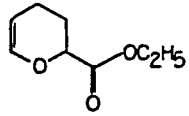
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------|--|--------------------------|--------|------------------------------|---------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₁₂OS⁺ | | | | | | | | |
|  | | (7.8) | (120) | (504) | -60 | -249 | *EST | |
| | | IP is onset of photoelectron band (81JOR/CAR). | | | | | | |
| C₈H₁₂O₂⁺ | | | | | | | | |
|  | | ≤8.80 | ≤129 | ≤541 | -74±0.5 | -308±2 | 77PED/RYL | 933-52-8 |
| | | See also: 84OLI/FLE. | | | | | | |
|  | | (≤9.45) | (≤149) | (≤625) | -68±3 | -287±13 | *EST | 3471-13-4 |
| | | IP from 82PFI/GER. | | | | | | |
|  | | (8.62) | (120) | (502) | -79 | -330 | 82MOR/MER | |
| | | IP from 82MOR/MER. | | | | | | |
|  | | (9.28±0.05) | (145) | (608) | -68.5±3 | -287±12 | *EST | 126-81-8 |
| C₈H₁₂O₂S⁺ | | | | | | | | |
|  | | (≤9.2) | (≤139) | (≤584) | -73 | -304 | *EST | |
| | | IP from 84AIT/GOS. | | | | | | |
|  | | (9.05) | (136) | (569) | -73 | -304 | *EST | |
| | | IP is onset of photoelectron band (84AIT/GOS). | | | | | | |
| C₈H₁₂O₃⁺ | | | | | | | | |
|  | | (8.6) | (69) | (288) | -129 | -542 | *EST | |
| | | IP from 82MOR/MER. | | | | | | |

Table 1. Positive Ion Table - Continued

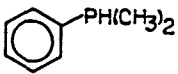
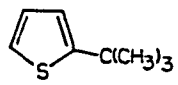
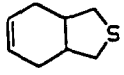
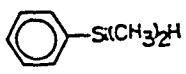
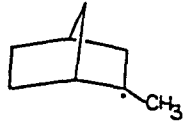
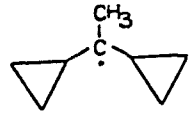
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{12}\text{P}^+$  | | 156 | 651 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{P}(\text{CH}_3)_2$ (RN 672-66-2). PA = 229.6 kcal/mol, 961. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{12}\text{S}^+$  | (8.32) | (194) | (812) | 2 | 9 | *EST | 1689-78-7 |
| | IP is onset of photoelectron band (83VES/HAR). | | | | | | |
|  | (8.0) | (192) | (804) | 8 | 32 | *EST | |
| | IP is onset of photoelectron band (84AIT/GOS). | | | | | | |
| $\text{C}_8\text{H}_{12}\text{Si}^+$ $(\text{CH}_2 = \text{CH})_4\text{Si}$ | (9.3) | (229) | (958) | 15 | 61 | 85GAD/GUB | 1112-55-6 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (8.92±0.15) | (203) | (848) | -3 | -13 | *EST | 766-77-8 |
| $\text{C}_8\text{H}_{12}\text{Sn}^+$ $\text{Sn}(\text{CH} = \text{CH}_2)_4$ | (8.4) | (277) | (1162) | 84 | 352 | *EST | 1112-56-7 |
| | IP is onset of photoelectron band (81NOV/CVI). | | | | | | |
| $\text{C}_8\text{H}_{13}^+$  | | 171 | 717 | | | | 3197-78-2 |
| | From proton affinities of 2-methylenebicyclo[2.2.1]heptane PA = (207) kcal/mol, (866) kJ/mol, (RN 497-35-8), 2-methylbicyclo[2.2.1]hept-2-ene PA = 206 kcal/mol, 862 kJ/mol, (RN 694-92-8) and hydride and chloride transfer equilibrium constants. (76SOL/FIE, 85SHA/SHA). | | | | | | |
|  | | 200 | 837 | | | | 50555-45-8 |
| | From proton affinity of 1,1-dicyclopropylethylene (RN 822-93-5). PA = 216.5 kcal/mol, 906. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued



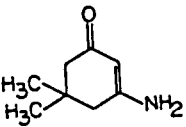
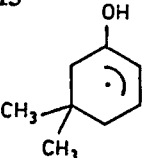
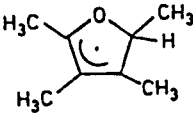
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------|---|--------------------------|--------|------------------------------|--------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₁₃Br⁺ | | | | | | | | |
|  | | (9.4±0.1) | (194) | (812) | -23 | -95 | *EST | 7697-09-8 |
| | | IP is onset of photoelectron band (84DEL/ABE). | | | | | | |
| C₈H₁₃I⁺ | | | | | | | | |
|  | | (8.7) | (192) | (805) | -8 | -34 | *EST | 931-98-6 |
| | | IP is onset of photoelectron band (84DEL/ABE). | | | | | | |
| C₈H₁₃NO⁺ | | | | | | | | |
|  | | (≤8.55) | (≤150) | (≤628) | -47 | -197 | *EST | 873-95-0 |
| | | IP from 82PFI/GER. | | | | | | |
| C₈H₁₃O⁺ | | | | | | | | |
|  | | | 116 | 487 | | | | |
| | | From proton affinity of 5,5-dimethylcyclohex-2-ene-1-one PA = 207.6 kcal/mol, 869. kJ/mol (86TAF/GAL). | | | | | | |
|  | | | 106 | 444 | | | | |
| | | From proton affinity of 2,3,4,5-tetramethylfuran (RN 10599-58-3) (85HOU/ROL). PA = 217.6 kcal/mol, 910. kJ/mol. | | | | | | |
| C₈H₁₄⁺ | | | | | | | | |
| (E)-CH ₃ CH ₂ CH ₂ CH ₂ CH=CHCH=CH ₂ | | (8.45) | (198) | (830) | 4 | 15 | *EST | 39491-65-1 |
| | | IP from 81MAS/MOU. | | | | | | |
| (E)-CH ₂ =CHCH ₂ CH=CH(CH ₂) ₂ CH ₃ | | (8.96) | (215) | (897) | 8 | 32 | *EST | 53793-31-0 |
| | | IP from 84HOL. | | | | | | |
| (E),(E)-CH ₃ CH ₂ CH ₂ CH=CHCH=CHCH ₃ | | (8.13) | (188) | (786) | 0.5 | 2 | *EST | 60919-80-4 |
| | | IP from 81MAS/MOU. | | | | | | |
| (E)-CH ₃ CH ₂ CH ₂ C(CH ₃)=CHCH=CH ₂ | | (8.02) | (185) | (776) | 0.5 | 2 | *EST | 40095-05-4 |
| | | IP from 81MAS/MOU. | | | | | | |

Table 1. Positive Ion Table - Continued


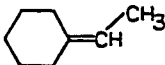
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{14}^+$ | | | | | | | |
| (Z)-(CH ₃) ₃ CCH=CHCH=CH ₂ | (8.46) | (199) | (833) | 4 | 17 | *EST | 59697-92-6 |
| | IP from 81MAS/MOU. | | | | | | |
| (E)-(CH ₃) ₃ CCH=CHCH=CH ₂ | (8.43) | (197) | (823) | 2 | 10 | *EST | 36320-14-6 |
| | IP from 81MAS/MOU. | | | | | | |
| (CH ₃) ₂ C=CHCH=C(CH ₃) ₂ | (7.67) | (171) | (716) | -6 | -24 | *EST | 764-13-6 |
| | IP from 81MAS/MOU. | | | | | | |
| (Z),(Z)-(CH ₃ CH=C(CH ₃)) ₂ | (8.1) | (182) | (761) | -5 | -20 | *EST | 21293-01-6 |
| | IP is onset of photoelectron band (84HON/ZHO). | | | | | | |
| (E),(E)-(CH ₃ CH=C(CH ₃)) ₂ | (7.8) | (177) | (740) | -3 | -12 | *EST | 18265-39-9 |
| | IP is onset of photoelectron band (84HON/ZHO). | | | | | | |
| (E),(Z)-(CH ₃ CH=C(CH ₃)) ₂ | (8.0) | (181) | (756) | -4 | -16 | *EST | 2417-88-1 |
| | IP is onset of photoelectron band (84HON/ZHO). | | | | | | |
| $\text{C}_2\text{H}_5\text{C}(\text{=CH}_2)\text{C}(\text{=CH}_2)\text{C}_2\text{H}_5$ | (8.58) | (199) | (834) | 1 | 6 | *EST | 16356-05-1 |
| | IP from 81MAS/MOU. | | | | | | |
| 1-C ₈ H ₁₄ | (9.95±0.02) | (248) | (1041) | 19±1 | 81±4 | 79ROG/DAG | 629-05-0 |
| n-C ₅ H ₁₁ C≡CCH ₃ | 9.31±0.01 | 230 | 962 | 15±0.2 | 64±2 | 79ROG/DAG | 2809-67-8 |
| C ₄ H ₉ C≡CC ₂ H ₅ | 9.22±0.01 | 228 | 953 | 15±0.5 | 63±2 | 79ROG/DAG | 15232-76-5 |
| n-C ₃ H ₇ C≡CC ₃ H ₇ | 9.20±0.01 | 226 | 948 | 14±0.5 | 60±2 | 79ROG/DAG | 1942-45-6 |
|  | 8.82 | 196.9 | 824.0 | -6.5±0.3 | -27.0±1.1 | 77PED/RYL | 931-88-4 |
|  | 8.44±0.05 | 174 | 726 | -21 | -88 | 76JEN | 1003-64-1 |

Table 1. Positive Ion Table - Continued

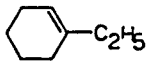
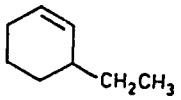
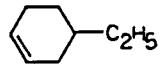
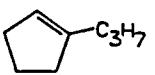
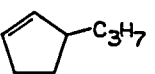
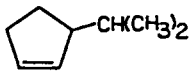
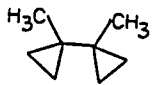

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{14}^+$ | | | | | | | |
|  | (8.48±0.01) | (180) | (755) | -15.2±0.2 | -63.4±1 | 77PED/RYL | 1453-24-3 |
|  | (8.83±0.01) | (191) | (799) | -13 | -53 | *EST | 2808-71-1 |
|  | (8.88±0.01) | (192) | (804) | -13 | -53 | *EST | 3742-42-5 |
|  | (8.48±0.01) | (186) | (779) | -9 | -39 | *EST | 3074-61-1 |
|  | (8.84±0.02) | (196) | (819) | -8 | -34 | *EST | 34067-75-9 |
|  | 8.81 IP from 84HOL. | (193) | (807) | -10 | -43 | *EST | 4276-45-3 |
|  | (8.8) IP is onset of photoelectron band (82SPA/GLE). | (222) | (931) | 20 | 82 | *EST | 59020-33-6 |
|  | (9.6) IP is onset of photoelectron band. | (220) | (921) | -1.2 | -5 | 81MAI/SCH | 7078-34-4 |

Table 1. Positive Ion Table - Continued




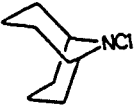
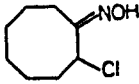
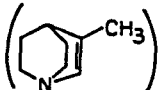
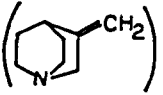
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{14}^+$  | 9.43±0.02 | 194 | 813 | -23±1 | -97±4 | 81GOD/SCH | 280-33-1 |
|  | (≤9.46) | (≤221) | (≤925) | 3 | 12 | *EST | 185-65-9 |
|  | (9.45) | (225) | (941) | 7 | 29 | *EST | 175-56-4 |
| $\text{C}_8\text{H}_{14}\text{ClN}^+$  | (≤8.55) IP from 82NEL/GAN. | (≤190) | (≤795) | -7 | -30 | *EST | |
| $\text{C}_8\text{H}_{14}\text{ClNO}^+$  | (9.19±0.03) IP from 79GOL/KUL. | (178) | (747) | -33 | -140 | *EST | 10499-33-9 |
| $\text{C}_8\text{H}_{14}\text{N}^+$  H^+ | | (164) | (687) | From proton affinity of 1-azabicyclo[2.2.2]oct-2-ene,3-methyl-. PA = (231.0) kcal/mol, (966.5) kJ/mol. | | | |
|  H^+ | | (156) | (651) | From proton affinity of 1-azabicyclo[2.2.2]octane, 3-methylene-. PA = (230.1) kcal/mol, (963.) kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

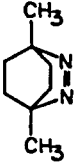
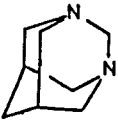
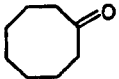
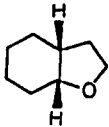
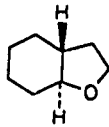
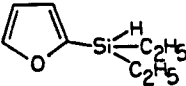
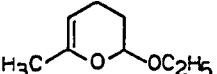
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{14}\text{N}_2^+$ | | | | | | | |
|  | (7.8) IP is onset of photoelectron band. | (202) | (846) | 22±1 | 93±5 | 80ENG | 49570-30-1 |
|  | ≤7.75 | (≤190) | (≤795) | 11 | 47 | *EST | 281-29-8 |
| $\text{C}_8\text{H}_{14}\text{O}^+$ | | | | | | | |
| $n\text{-C}_3\text{H}_7\text{CH}=\text{C}(\text{CH}_3)\text{C}(=\text{O})\text{CH}_3$ | (9.22) | (159) | (666) | -54 | -224 | *EST | 39899-08-6 |
|  | 9.08 See also: 86SPA/RAD. | 144 | 604 | -65±1 | -272±5 | 77PED/RYL | 502-49-8 |
|  | (9.0) IP is onset of photoelectron band (83TUR/HAN). | (149) | (624) | -58 | -244 | *EST | |
|  | (9.0) IP is onset of photoelectron band (83TUR/HAN). | (148) | (620) | -59 | -248 | *EST | |
| $\text{C}_8\text{H}_{14}\text{OSi}^+$ | | | | | | | |
|  | (8.1) IP is onset of photoelectron band (83ZYK/ERC). | (129) | (540) | -58 | -241 | *EST | 13271-67-5 |
| $\text{C}_8\text{H}_{14}\text{O}_2^+$ | | | | | | | |
|  | (≤8.6) IP from 82MOR/MER. | (≤86) | (≤358) | -113 | -471 | 82MOR/MER | |

Table 1. Positive Ion Table - Continued

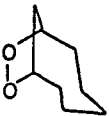
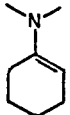
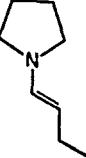
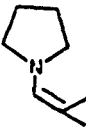
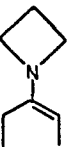
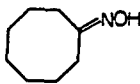
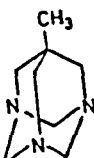
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---------------------------------------|---|--|--|----------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{14}\text{O}_2^+$ |  | (≤ 9.2) IP from 84GLE/DOB. | (≤ 177) | (≤ 741) | -35 | -147 | *EST | 69492-24-6 |
| $\text{C}_8\text{H}_{15}^+$ | $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)\text{C}(\text{CH}_3)_2$ | | (152) | (636) | | | | |
| | | | From proton affinity of $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)\text{C}(\text{CH}_3)=\text{CH}_2$ PA = (210.6) kcal/mol, (881.) kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{15}\text{N}^+$ |  | ≤ 7.50 IP from 81MUL/PRE2. | (≤ 170) | (≤ 711) | -3 | -12 | *EST | 13815-46-8 |
| |  | (6.7) IP is onset of photoelectron band (81MUL/PRE2). | (153) | (642) | -1 | -4 | *EST | 13937-89-8 |
| |  | (6.8) IP is onset of photoelectron band (81MUL/PRE2). | (160) | (671) | 4 | 15 | *EST | 2403-57-8 |
| |  | (≤ 7.48) IP from 81MUL/PRE2. | (≤ 193) | (≤ 809) | 21 | 87 | *EST | |
| $\text{C}_8\text{H}_{15}\text{NO}^+$ |  | (8.80 ± 0.03) IP from 79GOL/KUL. | (171) | (717) | -32 | -132 | *EST | 1074-51-7 |
| $\text{C}_8\text{H}_{15}\text{N}_3^+$ |  | (≤ 8.08) | (≤ 212) | (≤ 889) | 26 | 109 | *EST | 38705-10-1 |

Table 1. Positive Ion Table - Continued


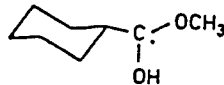
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{15}\text{O}^+$ | | | | | | | |
|  | | 98 | 410 | | | | |
| | | From proton affinity of cyclohexylethanone (RN 823-76-7). PA = 202.4 kcal/mol, 847. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{15}\text{O}_2^+$ | | | | | | | |
|  | | 51 | 212 | | | | |
| | | From proton affinity of methylcyclohexane carboxylate (RN 4630-82-4). PA = 203.7 kcal/mol, 852. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{16}^+$ | | | | | | | |
| 1- C_8H_{16} | 9.43±0.01 | 198 | 829 | -19.4±0.2 | -81.2±1 | 77PED/RYL | 111-66-0 |
| (Z)-2- C_8H_{16} | 8.91±0.01 | (184) | (767) | -22 | -91 | *EST | 7642-04-8 |
| (E)-2- C_8H_{16} | 8.91±0.01 | (183) | (765) | -23 | -95 | *EST | 13389-42-9 |
| (Z)-3- C_8H_{16} | 8.85±0.01 | (183) | (764) | -21 | -90 | *EST | 14850-22-7 |
| (E)-3- C_8H_{16} | 8.85±0.01 | (181) | (759) | -23 | -95 | *EST | 14919-01-8 |
| (Z)-4- C_8H_{16} | 8.84±0.01 | (182) | (763) | -21 | -90 | *EST | 7642-15-1 |
| (E)-4- C_8H_{16} | 8.83±0.01 | (181) | (758) | -22 | -94 | *EST | 14850-23-8 |
| $(\text{C}_2\text{H}_5)_2\text{C}=\text{CHC}_2\text{H}_5$ | (8.48±0.01) | (171) | (715) | -25 | -103 | *EST | 16789-51-8 |
| $\text{C}_2\text{H}_5\text{CH}_2\text{C}(\text{CH}_3)=\text{C}(\text{CH}_3)_2$ | (8.19±0.01) | (162) | (680) | -26 | -110 | *EST | 7145-20-2 |
| (Z)- $(\text{CH}_3)_2\text{CHCH}=\text{CHCH}(\text{CH}_3)_2$ | (8.85±0.01) | (179) | (749) | -25 | -105 | *EST | 10557-44-5 |
| (E)- $(\text{CH}_3)_2\text{CHCH}=\text{CHCH}(\text{CH}_3)_2$ | (8.84±0.01) | (178) | (743) | -26 | -110 | *EST | 692-70-6 |
| (Z)- $\text{C}_2\text{H}_5\text{C}(\text{CH}_3)=\text{C}(\text{CH}_3)\text{C}_2\text{H}_5$ | (8.17±0.01) | (162) | (678) | -26 | -110 | *EST | 19550-87-9 |
| (E)- $\text{C}_2\text{H}_5\text{C}(\text{CH}_3)=\text{C}(\text{CH}_3)\text{C}_2\text{H}_5$ | (8.16±0.01) | (162) | (677) | -26 | -110 | *EST | 19550-88-0 |
| (tert- C_4H_9) $\text{CH}_2\text{C}(\text{CH}_3)=\text{CH}_2$ | (8.91±0.01) | (179) | (749) | -26.4±0.2 | -110.4±1 | 77PED/RYL | 107-39-1 |
| $(\text{C}_2\text{H}_5)_2\text{C}=\text{C}(\text{CH}_3)_2$ | 8.17±0.01 | (162) | (678) | -26 | -110 | *EST | 19780-67-7 |

Table 1. Positive Ion Table - Continued


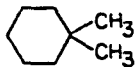
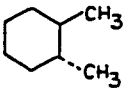
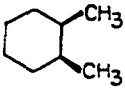
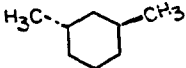
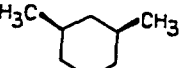
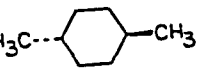
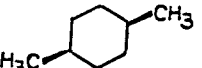
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------|---|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{16}^+$ | | | | | | | | |
| | $(\text{CH}_3)_2\text{CHC}(\text{CH}_3)=\text{C}(\text{CH}_3)_2$ | (8.17±0.01) | (160) | (670) | -28 | -118 | *EST | 565-77-5 |
| |  | 9.76 | 195 | 817 | -29.7±0.2 | -124.4±0.9 | 77PED/RYL | 292-64-8 |
| | | IP from charge transfer equilibrium constant determinations (82SIE/MAU, 82LIA). Reference IP's, fluorobenzenes. Photoelectron spectroscopy IP = 9.74±0.05 (79GOL/KUL). | | | | | | |
| |  | 9.42 | 174 | 728 | -43.2±0.5 | -180.9±1.9 | 77PED/RYL | 590-66-9 |
| | | IP from charge transfer equilibrium constant determinations (82SIE/MAU, 82LIA). Reference IP's, fluorobenzenes. | | | | | | |
| |  | 9.41 | 174 | 728 | -43.0±0.4 | -179.9±1.8 | 77PED/RYL | 6876-23-9 |
| | | IP from charge transfer equilibrium constant determinations (82SIE/MAU, 82LIA). Reference IP's, fluorobenzenes. Electron impact IP = 9.89 eV (81HER/SIC). | | | | | | |
| |  | (<9.78) | (<184) | (<771) | -41.1±0.4 | -172.3±1.8 | 77PED/RYL | 2207-01-4 |
| | | IP from 81HER/SIC. | | | | | | |
| |  | 9.53 | 178 | 743 | -42.2±0.4 | -176.5±1.7 | 77PED/RYL | 2207-03-6 |
| | | IP from charge transfer equilibrium constant determinations (82SIE/MAU, 82LIA). Reference IP's, fluorobenzenes. Electron impact IP = 9.89 eV (81HER/SIC). | | | | | | |
| |  | (<9.98) | (<186) | (<778) | -44.1±0.4 | -184.6±1.7 | 77PED/RYL | 638-04-0 |
| | | IP from 81HER/SIC. | | | | | | |
| |  | 9.56 | 176 | 738 | -44.1±0.4 | -184.5±1.7 | 77PED/RYL | 2207-04-7 |
| | | IP from charge transfer equilibrium constant determinations (82SIE/MAU, 82LIA). Reference IP's, fluorobenzenes. Threshold photoionization value of IP = 9.67 eV. | | | | | | |
| |  | (<9.93) | (<187) | (<781) | -42.2±0.4 | -176.6±1.7 | 77PED/RYL | 624-29-3 |
| | | IP from 81HER/SIC. | | | | | | |

Table 1. Positive Ion Table - Continued

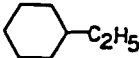
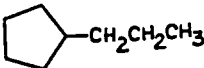
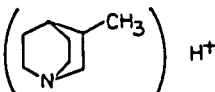
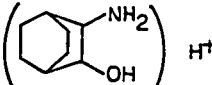
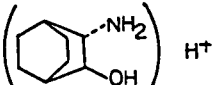
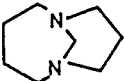
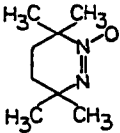
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₈H₁₆⁺ | | | | | | | |
|  | 9.54 | 178.8 | 748.1 | -41.2±0.1 | -172.4±0.6 | 77PED/RYL | 1678-91-7 |
| IP from charge transfer equilibrium constant determinations (82SIE/MAU, 82LIA). Reference IP's, fluorobenzenes. | | | | | | | |
|  | (10.00±0.04) | (195) | (817) | -35.3±0.2 | -147.8±0.6 | 77PED/RYL | 2040-96-2 |
| C₈H₁₆N⁺ | | | | | | | |
|  | | (126) | (528) | | | | |
| From proton affinity of 3-methyl-1-azabicyclo[2.2.2]octane (RN 695-88-5). PA = (231.7) kcal/mol, (969.) kJ/mol. | | | | | | | |
| C₈H₁₆NO⁺ | | | | | | | |
|  | | 84 | 353 | | | | |
| From proton affinity of cis-3-aminobicyclo[2.2.2]octan-2-ol (RN 17997-65-8). PA = 223.9 kcal/mol, 937. kJ/mol. | | | | | | | |
|  | | 86 | 359 | | | | |
| From proton affinity of trans-3-aminobicyclo[2.2.2]octan-2-ol (RN 40335-14-6). PA = 220.6 kcal/mol, 923. kJ/mol. | | | | | | | |
| C₈H₁₆N₂⁺ | | | | | | | |
|  | 7.0 | (174) | (729) | 13 | 54 | *EST | |
| IP is onset of photoelectron band (85HON/YAN). | | | | | | | |
| C₈H₁₆N₂O⁺ | | | | | | | |
|  | (≤9.13±0.03) | (≤204) | (≤854) | -6.33±0.55 | -26.48±0.3 | 83BYS | 54143-34-9 |
| C₈H₁₆O⁺ | | | | | | | |
| n-C ₆ H ₁₃ COCH ₃ | 9.40±0.03 | 140 | 586 | -77 | -321 | 75TRC | 111-13-7 |
| n-C ₄ H ₉ COCH ₂ CH ₂ CH ₃ | (9.10±0.05) | (133) | (558) | -76 | -320 | 75TRC | 589-63-9 |

Table 1. Positive Ion Table - Continued

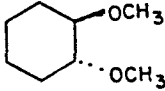
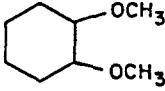
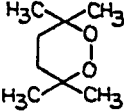
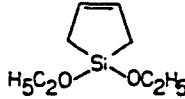
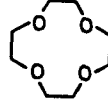
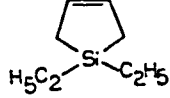
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{16}\text{O}^+$ tert-C ₄ H ₉ CO(iso-C ₃ H ₇) | (8.80±0.01) | (122) | (510) | -80.8±0.3 | -338.3±1.2 | 77PED/RYL | 5857-36-3 |
| $\text{C}_8\text{H}_{16}\text{O}_2^+$  | (8.7) IP is onset of photoelectron band. | (104) | (435) | -97 | -404 | *EST | 29887-60-3 |
|  | (8.6) IP is onset of photoelectron band. | (102) | (426) | -97 | -404 | *EST | 30363-80-5 |
|  | 9.2 IP is onset of photoelectron band. | (135) | (567) | -77 | -321 | *EST | 22431-89-6 |
| $\text{C}_8\text{H}_{16}\text{O}_2\text{Si}^+$  | ≤9.44 IP from 81KHV/ZYK. | (≤52) | (≤216) | -166 | -695 | *EST | 67059-49-8 |
| $\text{C}_8\text{H}_{16}\text{O}_4^+$  | (8.8) IP is onset of photoelectron band (83BAK/ARM, 82LEV/LIA). | (52) | (218) | -151±0.5 | -631±2 | 82BYS/MAN | 294-93-9 |
| $\text{C}_8\text{H}_{16}\text{Si}^+$  | (≤8.89) IP from 81KHV/ZYK. | (≤175) | (≤734) | -30 | -124 | *EST | 69657-20-1 |
| $\text{C}_8\text{H}_{17}^+$ (CH ₃) ₂ CCH ₂ CH ₂ CH ₂ CH ₂ CH ₃ | | 139 | 582 | | | | 40626-79-7 |
| | | From appearance potential measurement (84LOS/HOL). | | | | | |

Table 1. Positive Ion Table - Continued

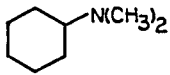
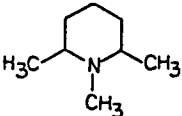
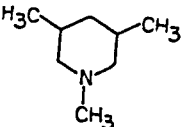
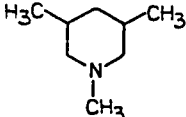
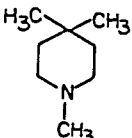
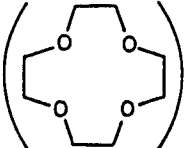
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_8H_{17}N^+$ | | | | | | | |
|  | (7.5) | (148) | (618) | -25 | -106 | *EST | 98-94-2 |
| IP is onset of photoelectron band. | | | | | | | |
|  | (7.77) | (148) | (620) | -31 | -130 | *EST | 2439-13-6 |
| IP from 82ROZ/HOU. | | | | | | | |
|  | (7.66) | (152) | (637) | -24 | -102 | *EST | 16544-52-8 |
| IP from 82ROZ/HOU. | | | | | | | |
|  | (7.63) | (149) | (625) | -27 | -111 | *EST | 14446-76-5 |
| IP from 82ROZ/HOU. | | | | | | | |
|  | (7.77) | (148) | (621) | -31 | -129 | *EST | 1003-84-5 |
| IP from 82ROZ/HOU. | | | | | | | |
| $C_8H_{17}O_4^+$ | | | | | | | |
|  | | -7 | -29 | | | | |
| H^+ | | From proton affinity of 1,4,7,10-tetraoxacyclododecane (12-Crown-4) (RN 294-93-9). PA = 221.6 kcal/mol, 927. kJ/mol. | | | | | |
| $C_8H_{18}^+$ | | | | | | | |
| n- C_8H_{18} | (9.82) | (177) | (739) | -49.8 | -208.5 | 74SCO | 111-65-9 |
| | | (188) | (786) | -38.6 | -161.4 | | |
| IP from charge transfer equilibrium constant determinations (81MAU/SIE, 82LIA). Reference IP's, fluorobenzenes. | | | | | | | |
| $(CH_3)_2CH(CH_2)_4CH_3$ | 9.84 | 176 | 734 | -51.4±0.3 | -215.1±1.4 | 74SCO | 592-27-8 |
| | | 187 | 784 | -39.6±0.3 | -165.9±1.4 | | |
| IP from charge transfer equilibrium constant determinations (81MAU/SIE, 82LIA). Reference IP's, fluorobenzenes. | | | | | | | |
| $(CH_3)_3CC(CH_3)_3$ | 9.8 | (172) | (720) | -53.9±0.3 | -225.7±1.1 | 77PED/RYL | 594-82-1 |
| IP is onset of photoelectron band (81SZE/KOR, 81KIM/KAT). | | | | | | | |

Table 1. Positive Ion Table - Continued

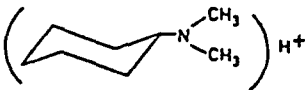
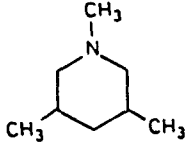
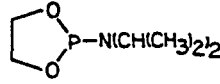
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|---|----------------|------------------------------|---------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{18}^+$ | $(\text{CH}_3)_2\text{CHCH}_2\text{C}(\text{CH}_3)_3$ | 9.86 | (171) | (714) | -57 | -238 | *EST | 540-84-1 |
| $\text{C}_8\text{H}_{18}\text{ClP}^+$ | $(\text{tert-C}_4\text{H}_9)_2\text{PCl}$ | (8.0) IP is onset of photoelectron band. | (112) | (469) | -72 | -303 | *EST | 13716-10-4 |
| $\text{C}_8\text{H}_{18}\text{FP}^+$ | $(\text{tert-C}_4\text{H}_9)_2\text{PF}$ | (8.2) IP is onset of photoelectron band. | (63) | (265) | -126 | -526 | *EST | 29146-24-5 |
| $\text{C}_8\text{H}_{18}\text{Hg}^+$ | $(n\text{-C}_4\text{H}_9)_2\text{Hg}$ | (≤ 8.35) | (≤ 185) | (≤ 774) | -8 ± 2 | -32 ± 8 | 77PED/RYL | 629-35-6 |
| | $(\text{iso-C}_4\text{H}_9)_2\text{Hg}$ | (≤ 8.30) | (≤ 182) | (≤ 763) | -9 ± 2 | -38 ± 8 | 77PED/RYL | 24470-76-6 |
| $\text{C}_8\text{H}_{18}\text{N}^+$ |  | | 108 | 450 | | | | |
| | | | From proton affinity of N,N-dimethylcyclohexanamine (RN 98-94-2) (86TAF/GAL). PA = 232.7 kcal/mol, 974. kJ/mol. | | | | | |
| |  | | (109) | (457) | | | | |
| | | | From proton affinity of N,3,5-trimethylpiperidine PA = (230) kcal/mol, (962) kJ/mol, (RN 14446-76-5) (84HOP/JAH). | | | | | |
| $\text{C}_8\text{H}_{18}\text{NO}^+$ | $(\text{tert-C}_4\text{H}_9)_2\text{NO}$ | (6.77) | (126) | (527) | -30 ± 3 | -126 ± 13 | *EST | 2406-25-9 |
| $\text{C}_8\text{H}_{18}\text{NO}_2\text{P}^+$ |  | (≤ 8.52) | (≤ 71) | (≤ 295) | -126 | -527 | *EST | |
| | | | IP from 82WOR/HAR. | | | | | |
| $\text{C}_8\text{H}_{18}\text{N}_2^+$ | $(E)\text{-}(\text{tert-C}_4\text{H}_9\text{N})_2$ | (7.7) | (169) | (707) | -9 ± 0.7 | -36 ± 3 | 80ENG | 927-83-3 |
| | | | IP is onset of photoelectron band. | | | | | |

Table 1. Positive Ion Table - Continued

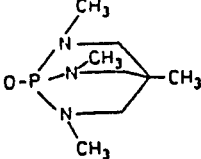
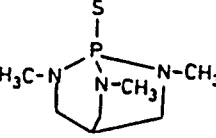
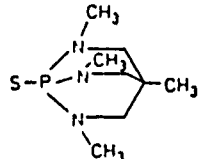
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{18}\text{N}_3\text{OP}^+$  | $(\leq 8.14 \pm 0.10)$ IP from 82COW/LAT. | (≤ 88) | (≤ 367) | -100 | -418 | *EST | 15199-21-0 |
| $\text{C}_8\text{H}_{18}\text{N}_3\text{P}^+$  | $(\leq 7.71 \pm 0.10)$ IP from 82COW/LAT. | (≤ 177) | (≤ 739) | -1 | -5 | *EST | 14418-26-9 |
| $\text{C}_8\text{H}_{18}\text{N}_3\text{PS}^+$  | $(\leq 8.14 \pm 0.10)$ IP from 82COW/LAT. | (≤ 164) | (≤ 688) | -23 | -97 | *EST | 15199-22-1 |
| $\text{C}_8\text{H}_{18}\text{O}^+$ $(n\text{-C}_4\text{H}_9)_2\text{O}$ | ≤ 9.43 | ≤ 138 | ≤ 577 | -80 | -333 | 77PED/RYL | 142-96-1 |
| | IP from 80BAC/MOU. Value derived from hydrogen affinity considerations: 9.37 eV. | | | | | | |
| $(\text{sec-C}_4\text{H}_9)_2\text{O}$ | (9.11) | (122) | (509) | -88 ± 0.5 | -370 ± 2 | 77PED/RYL | 6863-58-7 |
| | IP from 81HOL/FIN. | | | | | | |
| $(\text{tert-C}_4\text{H}_9)_2\text{O}$ | 8.81 | 117 | 488 | -87 ± 0.2 | -362 ± 1 | 77PED/RYL | 6163-66-2 |
| | See also: 80BAC/MOU. | | | | | | |
| $\text{C}_8\text{H}_{18}\text{OS}^+$ $[(\text{CH}_3)_3\text{C}]_2\text{SO}$ | 8.0 | (113) | (471) | -72 | -301 | *EST | 2211-92-9 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_8\text{H}_{18}\text{O}_2^+$ $(\text{tert-C}_4\text{H}_9\text{O})_2$ | (8.4) | (111) | (461) | -83 ± 0.7 | -349 ± 3 | 77PED/RYL | 110-05-4 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_8\text{H}_{18}\text{O}_2\text{S}^+$ $(\text{iso-C}_4\text{H}_9)_2\text{SO}_2$ | (9.54 ± 0.05) | (92) | (384) | -128 ± 0.7 | -536 ± 3 | 77PED/RYL | 10495-45-1 |
| $\text{C}_8\text{H}_{18}\text{O}_4^+$ $(\text{CH}_3\text{O}(\text{CH}_2)_2\text{OCH}_2)_2$ | (≤ 9.8) | (≤ 69) | (≤ 289) | -157 | -656 | *EST | 112-49-2 |
| | IP from 83BAK/ARM. | | | | | | |
| $\text{C}_8\text{H}_{18}\text{S}^+$ $(n\text{-C}_4\text{H}_9)_2\text{S}$ | (8.2) | (149) | (624) | -40.0 ± 0.3 | -167.3 ± 1.1 | 77PED/RYL | 544-40-1 |
| | IP is onset of photoelectron band. | | | | | | |
| $(\text{iso-C}_4\text{H}_9)_2\text{S}$ | 8.36 ± 0.05 | 150 | 628 | -43 ± 0.5 | -179 ± 2 | 77PED/RYL | 592-65-4 |

Table 1. Positive Ion Table - Continued

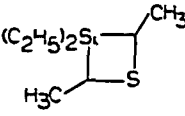
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{18}\text{S}^+$ (tert-C ₄ H ₉) ₂ S | (8.0) | (139) | (582) | -45.1±0.2 | -188.9±0.7 | 77PED/RYL | 107-47-1 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_8\text{H}_{18}\text{SSi}^+$  | (7.92±0.03) | (165) | (691) | (-17) | (-73) | 81GUS/VOL | |
| IP from 81GUS/VOL. | | | | | | | |
| $\text{C}_8\text{H}_{18}\text{S}_2^+$ (n-C ₄ H ₉ S) ₂ | (≤8.51) | (≤158) | (≤663) | -38±0.7 | -158±3 | 77PED/RYL | 629-45-8 |
| Dialkyl disulfides undergo a change in the CSSC bond angle from 90° to 180° upon ionization; adiabatic ionization potentials are probably well below the the experimentally observed ionization onset. | | | | | | | |
| (tert-C ₄ H ₉ S) ₂ | (7.7) | (130) | (542) | -48±0.7 | -200±3 | 77PED/RYL | 110-06-5 |
| IP is onset of photoelectron band. Dialkyl disulfides undergo a change in the CSSC bond angle from 90° to 180° upon ionization; adiabatic ionization potentials are probably well below the experimentally observed ionization onset. | | | | | | | |
| $\text{C}_8\text{H}_{18}\text{Si}_2^+$ CH ₂ =CH[Si(CH ₃) ₂] ₂ CH=CH ₂ | (≤8.63) | (≤166) | (≤694) | -33 | -139 | *EST | |
| IP from 81KHV/ZYK. | | | | | | | |
| $\text{C}_8\text{H}_{19}\text{ClNP}^+$ (CH ₃) ₃ CP(Cl)NHC(CH ₃) ₃ | (≤8.75) | (≤145) | (≤606) | -57 | -238 | *EST | |
| IP from 85ELB/ELL. | | | | | | | |
| $\text{C}_8\text{H}_{19}\text{N}^+$ n-C ₈ H ₁₇ NH ₂ | (8.5) | (155) | (648) | -41 | -172 | *EST | 111-86-4 |
| IP from 79AUE/BOW. | | | | | | | |
| (n-C ₄ H ₉) ₂ NH | (7.69±0.03) | (140) | (585) | -37.4±0.3 | -156.6±1.3 | 77PED/RYL | 111-92-2 |
| Ion heat of formation predicted from hydrogen affinities of secondary amines: 143 kcal/mol, 598 kJ/mol, corresponding to IP of 7.8 eV. | | | | | | | |
| (sec-C ₄ H ₉) ₂ NH | (7.63) | (138) | (579) | -38 | -157 | *EST | 626-23-3 |
| IP from 79AUE/BOW. | | | | | | | |
| (i-C ₄ H ₉) ₂ NH | (7.81) | (137) | (574) | -43±2 | -179±8 | 73PEP/GAF | 110-96-3 |
| IP from 79AUE/BOW. | | | | | | | |

Table 1. Positive Ion Table - Continued

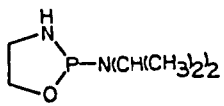
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{19}\text{N}_2\text{OP}^+$ | | | | | | | |
|  | (≤ 7.74) IP from 82WOR/HAR. | (≤ 79) | (≤ 330) | -100 | -417 | *EST | |
| $\text{C}_8\text{H}_{19}\text{O}^+$ | | | | | | | |
| (n-C ₄ H ₉) ₂ OH | | 82 | 345 | | | | |
| | From proton affinity of (n-C ₄ H ₉) ₂ O (RN 142-96-1). PA = 203.7 kcal/mol, 852. kJ/mol. | | | | | | |
| (sec-C ₄ H ₉) ₂ OH | | 68 | 286 | | | | |
| | From proton affinity of (sec-C ₄ H ₉) ₂ O (RN 6863-58-7). PA = 209.0 kcal/mol, 874. kJ/mol. | | | | | | |
| $\text{C}_8\text{H}_{19}\text{O}_4^+$ | | | | | | | |
| CH ₃ (OCH ₂ CH ₂) ₂ O(H)CH ₂ CH ₂ OCH ₃ | | -15 | -64 | | | | |
| | From proton affinity of CH ₃ (OCH ₂ CH ₂) ₃ OCH ₃ (RN 112-49-2). PA = 224.1 kcal/mol, 938. kJ/mol. | | | | | | |
| $\text{C}_8\text{H}_{19}\text{P}^+$ | | | | | | | |
| (tert-C ₄ H ₉) ₂ PH | (7.9) | (132) | (551) | -50 | -211 | *EST | 819-19-2 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_8\text{H}_{19}\text{S}^+$ | | | | | | | |
| (n-C ₄ H ₉) ₂ SH | | 117 | 490 | | | | |
| | From proton affinity of (n-C ₄ H ₉) ₂ S (RN 544-40-1). PA = 208.7 kcal/mol, 873. kJ/mol. | | | | | | |
| (t-C ₄ H ₉) ₂ SH | | 108 | 451 | | | | |
| | From proton affinity of (t-C ₄ H ₉) ₂ S (RN 107-47-1). PA = 212.8 kcal/mol, 890. kJ/mol. | | | | | | |
| $\text{C}_8\text{H}_{20}\text{Ge}^+$ | | | | | | | |
| (C ₂ H ₅) ₄ Ge | 8.9 | (167) | (698) | -38±2 | -161±8 | 77PED/RYL | 597-63-7 |
| | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_8\text{H}_{20}\text{N}^+$ | | | | | | | |
| n-C ₈ H ₁₇ NH ₃ | | (104) | (436) | | | | |
| | From proton affinity of n-C ₈ H ₁₇ NH ₂ (RN 111-86-4). PA = 220.4 kcal/mol, 922. kJ/mol. | | | | | | |
| (n-C ₄ H ₉) ₂ NH ₂ | | 100 | 417 | | | | |
| | From proton affinity of (n-C ₄ H ₉) ₂ NH (RN 111-92-2). PA = 228.4 kcal/mol, 956. kJ/mol. | | | | | | |
| (sec-C ₄ H ₉) ₂ NH ₂ | | (97) | (407) | | | | |
| | From proton affinity of (sec-C ₄ H ₉) ₂ NH (RN 626-23-3), re-evaluated (84HOP/JAH). PA = (230.9) kcal/mol, (966) kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

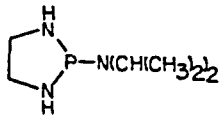
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|---------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{20}\text{N}^+$ (iso-C ₄ H ₉) ₂ NH ₂ | | 94 | 395 | | | | |
| | | From proton affinity of (iso-C ₄ H ₉) ₂ NH (RN 110-96-3). PA = 228.6 kcal/mol, 956. kJ/mol. | | | | | |
| (tert-C ₄ H ₉) ₂ NH ₂ | | 91 | 382 | | | | |
| | | From proton affinity of (tert-C ₄ H ₉) ₂ NH (RN 21981-37-3). PA = 233.2 kcal/mol, 976. kJ/mol. | | | | | |
| (i-C ₃ H ₇) ₂ (C ₂ H ₅)NH | | (97) | (406) | | | | |
| | | From proton affinity of (i-C ₃ H ₇) ₂ (C ₂ H ₅)N (RN 7087-68-5). PA = 235.3 kcal/mol, 984. kJ/mol. | | | | | |
| (CH ₃) ₃ C(CH ₂) ₂ NH(CH ₃) ₂ | | (100) | (417) | | | | |
| | | From proton affinity of (CH ₃) ₃ C(CH ₂) ₂ N(CH ₃) ₂ (RN 15673-04-8). PA = 230.4 kcal/mol, 964. kJ/mol. | | | | | |
| $\text{C}_8\text{H}_{20}\text{N}_2^+$ (C ₂ H ₅) ₂ NN(C ₂ H ₅) ₂ | (6.50) | (149) | (625) | -0.5 | -2 | *EST | 4267-00-9 |
| | IP from charge transfer equilibrium constant determination (84MAU/NEL). Reference standard: IP (C ₆ H ₅ N(CH ₃) ₂) = 7.12 eV. See also: 84NEL. | | | | | | |
| (i-C ₃ H ₇) ₂ NN(CH ₃) ₂ | (6.53) | (153) | (639) | 2 | 9 | *EST | 60678-72-0 |
| | IP from charge transfer equilibrium constant determination (86RUM). | | | | | | |
| (i-C ₃ H ₇)(CH ₃)NN(CH ₃)(i-C ₃ H ₇) | (6.58) | (154) | (645) | 2 | 10 | *EST | 60678-71-9 |
| | IP from charge transfer equilibrium constant determination (86RUM). See also: 84NEL. | | | | | | |
| $\text{C}_8\text{H}_{20}\text{N}_3\text{P}^+$  | (≤7.40) | (≤136) | (≤568) | -35 | -146 | *EST | |
| | IP from 82WOR/HAR. | | | | | | |
| $\text{C}_8\text{H}_{20}\text{N}_4^+$ (N ₂ (C ₂ H ₅) ₂) ₂ | (≤7.1) | (≤213) | (≤890) | 49 | 205 | 70BEN/O'N | 13304-29-5 |
| $\text{C}_8\text{H}_{20}\text{O}_4\text{Si}^+$ (C ₂ H ₅ O) ₄ Si | (≤9.77) | (≤-93) | (≤-388) | -318±5 | -1331±21 | 80TEL/RAB | 78-10-4 |
| $\text{C}_8\text{H}_{20}\text{Pb}^+$ (C ₂ H ₅) ₄ Pb | (11.1) | (282) | (1180) | 26±1 | 109±5 | 77PED/RYL | 78-00-2 |

Table 1. Positive Ion Table - Continued

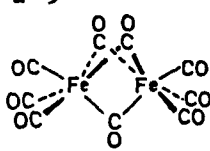
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_8\text{H}_{20}\text{Si}^+$ (C_2H_5) ₄ Si | (8.9) | (142) | (594) | -63±4 | -265±15 | 77PED/RYL | 631-36-7 |
| $\text{C}_8\text{H}_{20}\text{Sn}^+$ (C_2H_5) ₄ Sn | (8.1) | (176) | (737) | -11±0.7 | -45±3 | 77PED/RYL | 597-64-8 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_8\text{H}_{21}\text{N}_2^+$ (CH_3) ₂ NH(CH ₂) ₄ N(CH ₃) ₂ | | 108 | 450 | | | | |
| From proton affinity of (CH_3) ₂ N(CH ₂) ₄ N(CH ₃) ₂ (RN 111-51-3). PA = 246. kcal/mol, 1029. kJ/mol. | | | | | | | |
| (C_2H_5) ₂ NHN(C_2H_5) ₂ | | 135 | 564 | | | | |
| From proton affinity of (C_2H_5) ₂ NN(C_2H_5) ₂ (RN 4267-00-9) (84MAU/NEL). PA = 230.4 kcal/mol, 964. kJ/mol. | | | | | | | |
| $\text{C}_8\text{H}_{22}\text{NSi}^+$ (CH_3) ₃ Si(CH ₂) ₃ NH(CH ₃) ₂ | | 75 | 312 | | | | |
| From proton affinity of (CH_3) ₃ Si(CH ₂) ₃ N(CH ₃) ₂ (RN 28247-29-2). 231.8 kcal/mol, 970. kJ/mol. | | | | | | | |
| (CH_3) ₂ (tert-C ₄ H ₉)SiNH(CH ₃) ₂ | | 68 | 283 | | | | |
| From proton affinity of (CH_3) ₂ (tert-C ₄ H ₉)SiN(CH ₃) ₂ (RN 66365-05-7). PA = 229.7 kcal/mol, 961. kJ/mol. | | | | | | | |
| $\text{C}_8\text{H}_{24}\text{N}_4\text{Mo}^+$ ((CH_3) ₂ N) ₄ Mo | (≤5.30) | (≤153) | (≤642) | 31±2 | 131±8 | 81CAV/CON | |
| $\text{C}_8\text{H}_{24}\text{O}_2\text{Si}_3^+$ [(CH_3) ₃ SiO] ₂ Si(CH ₃) ₂ | (≤10.04) | (≤-99) | (≤-412) | -330±3 | -1381±12 | 77PED/RYL | 107-51-7 |
| IP from 82ERM/KIR. | | | | | | | |
| $\text{C}_8\text{H}_{24}\text{Si}_3^+$ Si ₃ (CH ₃) ₈ | (7.7) | (65) | (273) | -112±4 | -470±17 | 77PED/RYL | 3704-44-7 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_9\text{Fe}_2\text{O}_9^+$ | (7.91±0.01) | (-136) | (-571) | -319±6 | -1334±23 | 77PED/RYL | 15321-51-4 |
|  | | | | | | | |

Table 1. Positive Ion Table - Continued

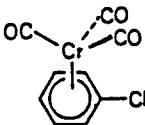
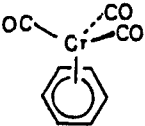
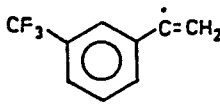
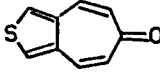
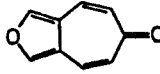
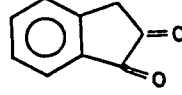
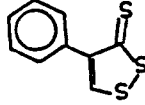
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_5\text{ClCrO}_3^+$  | (7.00±0.1) | (74) | (311) | -87±1 | -364±6 | 77PED/RYL | 12082-03-0 |
| $\text{C}_9\text{H}_6\text{CrO}_3^+$  | 7.0 IP is onset of photoelectron band. See also: 82GUI/PFI. | (78) | (325) | -84±2 | -350±9 | 77PED/RYL | 12082-08-5 |
| $\text{C}_9\text{H}_6\text{F}_3^+$  | 83 347 From proton affinity of 3-CF ₃ C ₆ H ₄ C=CH (RN 705-28-2) (85MAR/MOD). PA = 192.9 kcal/mol, 807. kJ/mol. | | | | | | |
| $\text{C}_9\text{H}_6\text{OS}^+$  | (8.5) IP is onset of photoelectron band (84GLE/BIS). | (222) | (930) | 26 | 110 | *EST | 10095-83-7 |
| $\text{C}_9\text{H}_6\text{O}_2^+$  | (8.65) IP is onset of photoelectron band (84GLE/BIS). | (190) | (795) | -10 | -40 | *EST | 18895-06-2 |
|  | (8.8) IP is onset of photoelectron band. | (167) | (699) | -36 | -150 | *EST | 16214-27-0 |
| $\text{C}_9\text{H}_6\text{S}_3^+$  | (7.8) IP is onset of photoelectron band. | (253) | (1060) | 74±2 | 311±10 | 72GEI/RAU | 3445-76-9 |

Table 1. Positive Ion Table - Continued

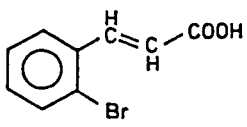
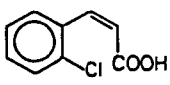
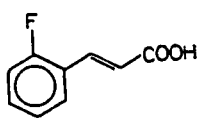
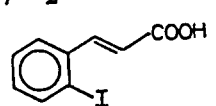
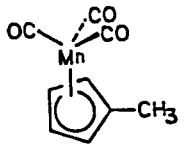
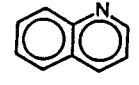
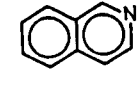
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_7\text{BrO}_2^+$  | (8.80) IP from 84SCH. | (152) | (638) | -50 | -211 | *EST | |
| $\text{C}_9\text{H}_7\text{ClO}_2^+$  | (8.85) IP from 84SCH. | (147) | (615) | -57 | -239 | *EST | 4513-41-1 |
| $\text{C}_9\text{H}_7\text{FO}_2^+$  | (9.00) IP from 84SCH. | (103) | (430) | -105 | -438 | *EST | 451-69-4 |
| $\text{C}_9\text{H}_7\text{IO}_2^+$  | (8.55) IP from 84SCH. | (160) | (668) | (-37) | (-156) | *EST | 90276-19-0 |
| $\text{C}_9\text{H}_7\text{MnO}_3^+$  | (7.4) IP is onset of photoelectron band (81CAL/HUB, 81CAL/LIC). | (47) | (196) | -124 | -518 | *EST | 12108-13-3 |
| $\text{C}_9\text{H}_7\text{N}^+$  | 8.62±0.01 | 249 | 1043 | 50±0.2 | 211±1 | 79VIS | 91-22-5 |
|  | 8.53±0.03 | 247 | 1031 | 50±0.2 | 208±1 | 79VIS/WIL | 119-65-3 |

Table 1. Positive Ion Table - Continued

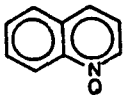
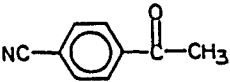

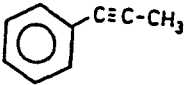
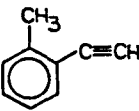
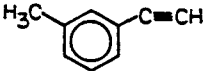
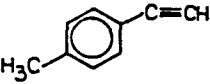
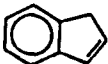
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_7\text{NO}^+$  | 8.00±0.02 | (215) | (903) | 31 | 131 | *EST | 1613-37-2 |
|  | (≤9.82) IP from 85GAL/GER. | (≤237) | (≤991) | 11 | 44 | *EST | 1443-80-7 |
|  | (7.9) IP is onset of photoelectron band. | (213) | (890) | 31 | 128 | *EST | 1532-72-5 |
| C_9H_8^+ | | | | | | | |
|  | 8.41 See also: 81ELB/LIE. | (258) | (1079) | 64 | 268 | 85DAV/ALL | 673-32-5 |
|  | (≤8.61±0.02) | (≤264) | (≤1105) | 65 | 274 | *EST | 766-47-2 |
|  | (≤8.63±0.02) | (≤264) | (≤1106) | 65 | 273 | *EST | 766-82-5 |
|  | 8.3 IP is onset of photoelectron band. | (257) | (1075) | 65 | 274 | *EST | 766-97-2 |
|  | 8.14±0.01 | 227 | 948 | 39±0.2 | 163±1 | 80KUD/KUD | 95-13-6 |

Table 1. Positive Ion Table - Continued

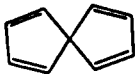
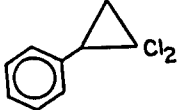
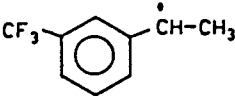
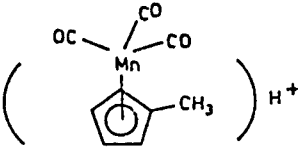
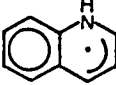
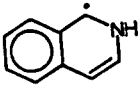
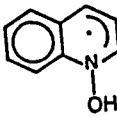
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C_9H_8^+  | (7.99) | (271) | (1134) | 87 | 363 | *EST | 14867-83-5 |
| $\text{C}_9\text{H}_8\text{Cl}_2^+$  | (8.7) IP is onset of photoelectron band. | (234) | (979) | 33 | 140 | *EST | 2415-80-7 |
| $\text{C}_9\text{H}_8\text{F}_3^+$  | | 43 | 182 | | | | |
| | | From proton affinity of 3-CF ₃ C ₆ H ₄ CH = CH ₂ (RN 402-24-4) (84HAR/HOU). PA = 194.6 kcal/mol, 814. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_8\text{MnO}_3^+$  | | (44) | (183) | | | | |
| | | From proton affinity of η^5 -methylcyclopentadienyl-(RN 12108-13-3). PA = (200.6) kcal/mol, (839.) kJ/mol. | | | | | |
| $\text{C}_9\text{H}_8\text{N}^+$  | | 190 | 793 | | | | |
| | | From proton affinity of quinoline (RN 91-22-5). PA = (226.5) kcal/mol, 948. kJ/mol. | | | | | |
|  | | 190 | 793 | | | | |
| | | From proton affinity of isoquinoline (RN 119-65-3). PA = 225.9 kcal/mol, 945. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_8\text{NO}^+$  | | 172 | 721 | | | | |
| | | From proton affinity of quinoline-1-oxide (RN 1613-37-2). PA = 224.6 kcal/mol, 940. kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued


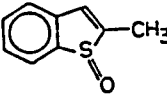
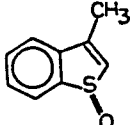
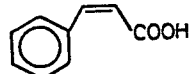
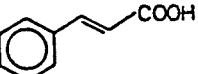
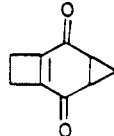
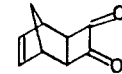
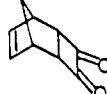
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_8\text{N}_2^+$  | (≤ 8.5) IP from 82CRI/LIC. | (≤ 299) | (≤ 1251) | 103 | 431 | *EST | 30855-80-2 |
| $\text{C}_9\text{H}_8\text{OS}^+$  | (≤ 8.75) IP from 82BEN/DUR. | (≤ 203) | (≤ 848) | 1 | 4 | *EST | 33945-86-7 |
|  | (8.2) IP is onset of photoelectron band (82BEN/DUR). | (190) | (795) | 1 | 4 | *EST | 51500-43-7 |
| $\text{C}_9\text{H}_8\text{O}_2^+$  | (8.90 ± 0.05) IP from 84SCH. | (155) | (649) | -50 | -210 | *EST | 102-94-3 |
|  | (9.00 ± 0.05) IP from 84SCH. | (153) | (641) | -54 | -227 | 77PED/RYL | 140-10-3 |
|  | (≤ 9.38) IP from 85GLE/JAH. | (≤ 249) | (≤ 1043) | 33 | 138 | *EST | 94499-50-0 |
|  | (≤ 8.65) IP from 78MAR/SCH. | (≤ 193) | (≤ 809) | -6 | -26 | *EST | |
|  | (≤ 8.90) IP from 78MAR/SCH. | (≤ 204) | (≤ 854) | -1 | -5 | *EST | |

Table 1. Positive Ion Table - Continued

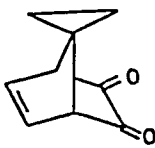
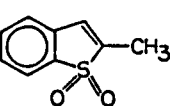
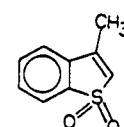
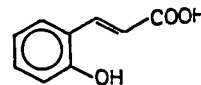
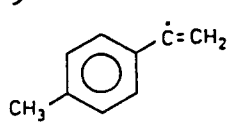
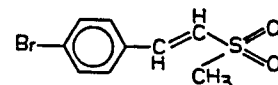
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_8\text{O}_2^+$ | | | | | | | |
|  | (8.1) IP is onset of photoelectron band. | (199) | (834) | 12 | 52 | *EST | 60526-40-1 |
| $\text{C}_9\text{H}_8\text{O}_2\text{S}^+$ | | | | | | | |
|  | (≤ 9.10) IP from 82BEN/DUR. | (≤ 158) | (≤ 662) | -52 | -216 | *EST | 6224-55-1 |
|  | (≤ 9.20) IP from 82BEN/DUR. | (≤ 160) | (≤ 672) | -52 | -216 | *EST | 6406-91-3 |
| $\text{C}_9\text{H}_8\text{O}_3^+$ | | | | | | | |
|  | (8.50 ± 0.05) IP from 84SCH. | (93) | (389) | -103 | -431 | 84SCH | 614-60-8 |
| C_9H_9^+ | | | | | | | |
|  | | 227 | 951 | | | | |
| | | From proton affinity of 4- $\text{CH}_3\text{C}_6\text{H}_4\text{C}=\text{CH}_2$ (RN 766-97-2) (85MAR/MOD). PA = 203.8 kcal/mol, 853. kJ/mol. | | | | | |
| | | (217) | (907) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination in $\text{C}_6\text{H}_5\text{C}(\text{CH}_3)=\text{CH}_2$ (85HON/SEG); structure may be indanyl or vinyltropylium. | | | | | |
| $\text{C}_9\text{H}_9\text{BrO}_2\text{S}^+$ | | | | | | | |
|  | ≤ 8.92 IP from 84CAU/FUR. | (≤ 174) | (≤ 727) | -32 | -134 | *EST | |

Table 1. Positive Ion Table - Continued

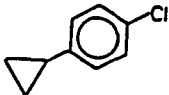
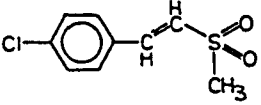
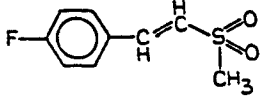
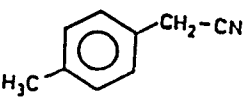
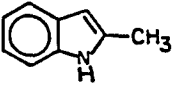
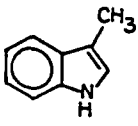
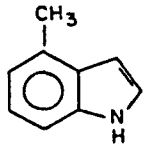
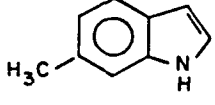
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------------------------------|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₉Cl⁺  | (≤ 8.64) | (≤ 228) | (≤ 953) | 28 | 119 | *EST | 1798-84-1 |
| C₉H₉ClO₂S⁺  | ≤ 8.94 IP from 84CAU/FUR. | (≤ 162) | (≤ 679) | -44 | -184 | *EST | |
| C₉H₉FO₂S⁺  | (≤ 9.05) IP from 84CAU/FUR. | (≤ 145) | (≤ 607) | -64 | -266 | *EST | |
| C₉H₉N⁺  | (9.16 ± 0.06) | (248) | (1037) | 36.6 | 153 | *EST | |
|  | (7.44 ± 0.015) | (203) | (850) | 32 | 132 | *EST | 95-20-5 |
|  | (7.54 ± 0.01) | (205) | (859) | 32 | 132 | *EST | 83-34-1 |
|  | (7.60 ± 0.015) | (207) | (865) | 32 | 132 | *EST | 16096-32-5 |
|  | (7.54 ± 0.015) | (205) | (859) | 32 | 132 | *EST | 3420-02-8 |

Table 1. Positive Ion Table - Continued

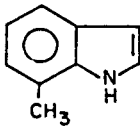
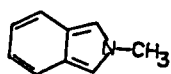
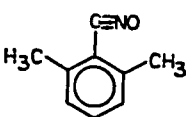
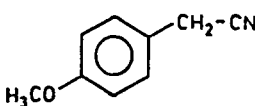
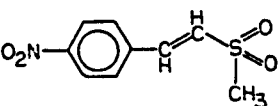
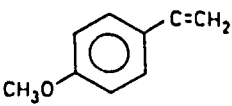
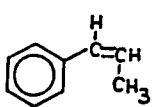
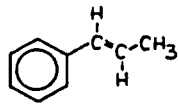
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₉N⁺ | | | | | | | |
|  | (7.53±0.015) | (205) | (859) | 32 | 132 | *EST | 933-67-5 |
|  | (≤7.12) | (≤214) | (≤895) | 50 | 208 | *EST | 33804-84-1 |
| C₉H₉NO⁺ | | | | | | | |
|  | (8.4) IP is onset of photoelectron band. | (249) | (1042) | 55 | 232 | *EST | 19111-74-1 |
|  | (8.77±0.05) | (209) | (876) | 7.1 | 30 | *EST | 104-47-2 |
| C₉H₉NO₄S⁺ | | | | | | | |
|  | (≤9.62) IP from 84CAU/FUR. | (≤181) | (≤758) | -41 | -170 | *EST | |
| C₉H₉O⁺ | | | | | | | |
|  | | 193 | 806 | | | | |
| | | From proton affinity of 4-CH ₃ OC ₆ H ₄ C=CH (RN 768-60-5) (85MAR/MOD). PA = 210.1 kcal/mol, 879. kJ/mol. | | | | | |
| C₉H₁₀⁺ | | | | | | | |
|  | 8.15 | 217 | 907 | 29 | 121 | 69BEN/CRU | 766-90-5 |
| | IP from 78FU/DUN and onset of photoelectron band (81KOB/ARA). | | | | | | |
|  | (8.08) | (214) | (897) | 28 | 117 | 69BEN/CRU | 873-66-5 |
| | IP is onset of photoelectron band (81KOB/ARA). | | | | | | |

Table 1. Positive Ion Table - Continued

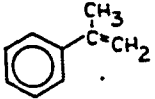
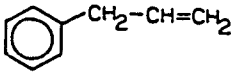
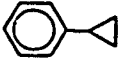
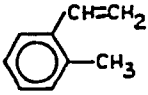
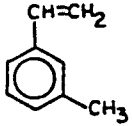
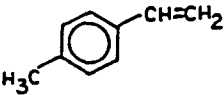
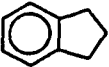

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{10}^+$  | 8.19±0.02 | 216 | 903 | 27 | 113 | 69BEN/CRU | 98-83-9 |
|  | 8.60 See also: 78FU/DUN. | (236) | (986) | 37±2 | 156±8 | 81CHI/HYM | 300-57-2 |
|  | 8.35 IP from 78FU/DUN. | 229 | 957 | 36±0.2 | 151±1 | 82FUC/HAL | 873-49-4 |
|  | 8.20±0.02 | 217 | 909 | 28 | 118 | 69BEN/CRU | 611-15-4 |
|  | 8.15±0.02 | 215 | 901 | 27 | 115 | 69BEN/CRU | 100-80-1 |
|  | 8.1±0.1 IP is onset of photoelectron band. | (214) | (896) | 27 | 115 | 69BEN/CRU | 622-97-9 |
|  | (8.3) IP is onset of photoelectron band. | (206) | (862) | 15±0.2 | 61±1 | 80KUD/KUD | 496-11-7 |
|  | (8.47) | (293) | (1225) | 97±2 | 408±8 | 73ENG/AND2 | 452-61-9 |

Table 1. Positive Ion Table - Continued




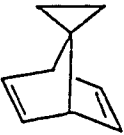
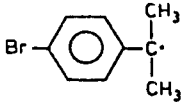
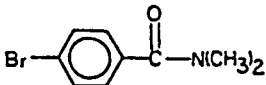
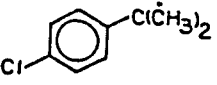
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₀⁺ | | | | | | | |
|  | (8.2) IP is onset of photoelectron band. | (246) | (1030) | 57 | 239 | *EST | 766-30-3 |
|  | (8.2) IP is onset of photoelectron band. | (246) | (1030) | 57 | 239 | *EST | 24430-29-3 |
|  | (8.3) IP is onset of photoelectron band (82SPA/KOR). | (344) | (1441) | 153 | 640 | *EST | 55980-70-6 |
|  | (7.9) IP is onset of photoelectron band. | (271) | (1133) | 89 | 371 | *EST | 7092-57-1 |
| C₉H₁₀Br⁺ | | | | | | | |
|  | | 186 | 777 | Value from appearance energy determinations (86ORL/MIS). | | | |
| C₉H₁₀BrNO⁺ | | | | | | | |
|  | (≤ 9.09) IP from 85GAL/GER. | (≤ 192) | (≤ 803) | -18 | -74 | *EST | 18469-37-9 |
| C₉H₁₀Cl⁺ | | | | | | | |
|  | | 180 | 753 | From proton affinity of 4-ClC ₆ H ₄ C(CH ₃) ₂ =CH ₂ (RN 1712-70-5). PA = 205.0 kcal/mol, 858. kJ/mol. Value from appearance energy determination (86ORL/MIS) = 171 kcal/mol, 716 kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

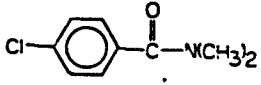
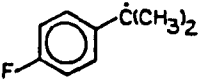
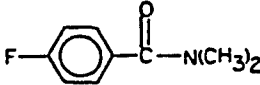
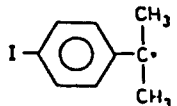
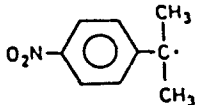
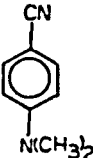
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{10}\text{ClNO}^+$ | (≤ 9.16) IP from 8SGAL/GER. | (≤ 182) | (≤ 760) | -30 | -124 | *EST | 14062-80-7 |
|  | | | | | | | |
| $\text{C}_9\text{H}_{10}\text{F}^+$ | | 138 | 579 | | | | |
|  | | | | | | | |
| | | From proton affinity of 4-FC ₆ H ₄ C(CH ₃)=CH ₂ (RN 350-40-3). PA = 206.7 kcal/mol, 865. kJ/mol. Value from appearance energy determinations (86ORL/MIS). | | | | | |
| $\text{C}_9\text{H}_{10}\text{FNO}^+$ | (≤ 9.13) IP from 8SGAL/GER. | (≤ 140) | (≤ 587) | -70 | -294 | *EST | 24167-56-4 |
|  | | | | | | | |
| $\text{C}_9\text{H}_{10}\text{I}^+$ | | 199 | 833 | | | | |
|  | | | | | | | |
| | | Value from appearance energy determinations (86ORL/MIS). | | | | | |
| $\text{C}_9\text{H}_{10}\text{N}^+$ (HC=CCH ₂) ₃ NH | | (319) | (1336) | | | | |
| | | From proton affinity of (HC=CCH ₂) ₃ N (RN 6921-29-5). PA = 220.2 kcal/mol, 921. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{10}\text{NO}_2^+$ | | 189 | 789 | | | | |
|  | | | | | | | |
| | | Value from appearance energy determinations (86ORL/MIS). | | | | | |
| $\text{C}_9\text{H}_{10}\text{N}_2^+$ | | | | | | | |
|  | (7.60) | (230) | (963) | 55 | 230 | *EST | 1197-19-9 |
| | IP is onset of photoelectron band (81MOD/DIS). | | | | | | |

Table 1. Positive Ion Table - Continued

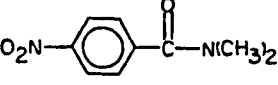
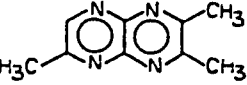
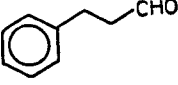
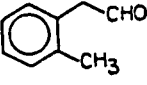
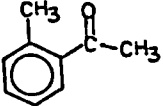
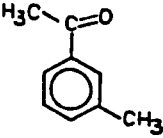
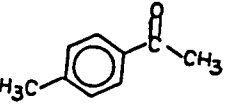
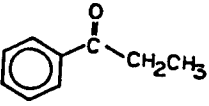
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|-----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{10}\text{N}_2\text{O}_3^+$  | (≤ 9.46) IP from 85GAL/GER. | (≤ 192) | (≤ 803) | -26 | -110 | *EST | 7291-01-2 |
| $\text{C}_9\text{H}_{10}\text{N}_4^+$  | (≤ 8.7) IP from 84GLE/SPA2 | (≤ 261) | (≤ 1090) | 60 | 251 | *EST | 6479-02-3 |
| $\text{C}_9\text{H}_{10}\text{O}^+$  | (8.7 ± 0.2) IP from 84DEN/AUD. | (182) | (763) | -17 | -73 | 84DEN/AUD | 1335-10-0 |
|  | (8.5) IP from 83AUD/MIL. | (176) | (735) | -20 | -85 | 83AUD/MIL | 69380-02-5 |
|  | (8.92) IP is onset of photoelectron band. See also: 81RAB/HEL. | (180) | (754) | -26 | -107 | *EST | 577-16-2 |
|  | (8.85) IP is onset of photoelectron band. See also: 81RAB/HEL. | (175) | (734) | -29 | -120 | *EST | 585-74-0 |
|  | (8.85) IP is onset of photoelectron band. See also: 81RAB/HEL, 85GAL/GER. | (176) | (735) | -28 | -119 | *EST | 122-00-9 |
|  | (9.16) IP from 79MCA/TRA. | (185) | (775) | -26 ± 0.5 | -109 ± 2 | 77PED/RYL | 93-55-0 |

Table 1. Positive Ion Table - Continued

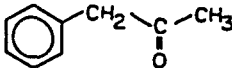
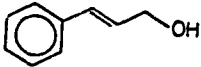
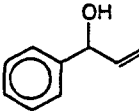
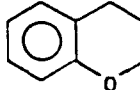
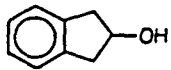
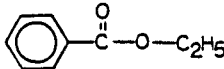
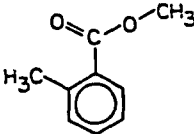
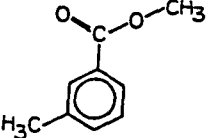
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₀O⁺ | | | | | | | |
|  | (8.7) IP is onset of photoelectron band (78CEN/FRA). | (177) | (741) | -23.6±0.3 | -98.6±1.4 | 77PED/RYL | 103-79-7 |
|  | (8.4±0.2) IP from 84DEN/AUD. | (186) | (779) | -7 | -29 | 84DEN/AUD | 104-54-1 |
|  | (8.6±0.2) IP from 84DEN/AUD. | (192) | (802) | -6 | -25 | 84DEN/AUD | |
|  | (7.93) | (161) | (673) | -22±1 | -92±5 | 77SHA/GOL | 493-08-3 |
|  | (8.6) IP from 83AUD/MIL. | (167) | (697) | -32 | -133 | 83AUD/MIL | 4254-29-9 |
| C₉H₁₀O₂⁺ | | | | | | | |
|  | (8.9) IP is onset of photoelectron band (81MEE/WAH). | (128) | (537) | -77 | -322 | *EST | 93-89-0 |
|  | (8.6) IP is onset of photoelectron band (81MEE/WAH). | (123) | (514) | -75 | -316 | *EST | 89-71-4 |
|  | (8.5) IP is onset of photoelectron band (81MEE/WAH). | (119) | (499) | -77 | -321 | *EST | 99-36-5 |

Table 1. Positive Ion Table - Continued

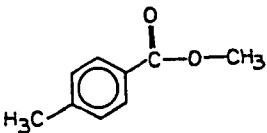
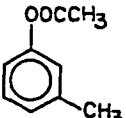
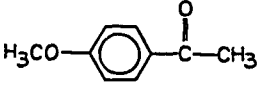
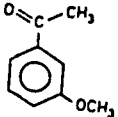
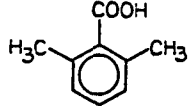
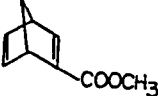
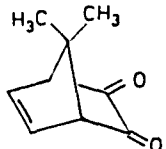
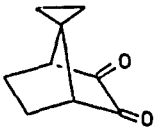
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{10}\text{O}_2^+$  | (8.4) | (117) | (489) | -77 | -321 | *EST | 99-75-2 |
| | IP is onset of photoelectron band (81MEE/WAH). | | | | | | |
|  | (8.98±0.2) | (132) | (553) | -75±0.5 | -313±2 | 77PED/RYL | 122-46-3 |
|  | 8.2±0.1 | (132) | (552) | -57 | -239 | *EST | 100-06-1 |
| | See also: 85GAL/GER. | | | | | | |
|  | (8.53±0.05) | (137) | (573) | -59.8±1 | -250±4 | *EST | 586-37-8 |
|  | (8.9) | (124) | (517) | -81.6±0.4 | -341.6±1.7 | 84COL/JIM | 632-46-2 |
| | IP from 81MEE/WAH. | | | | | | |
|  | (≤8.92) | (≤176) | (≤737) | -30 | -124 | *EST | 3604-36-2 |
| | IP from 83HOU/RON. | | | | | | |
|  | (8.2) | (155) | (649) | -34 | -142 | *EST | 60526-42-3 |
| | IP is onset of photoelectron band. | | | | | | |
|  | (≤9.3) | (≤191) | (≤799) | -23 | -98 | *EST | 94499-48-6 |
| | IP from 85GLE/JAH. | | | | | | |

Table 1. Positive Ion Table - Continued

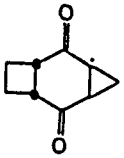
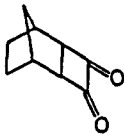
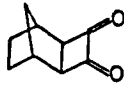
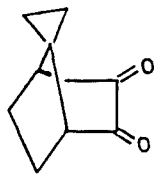
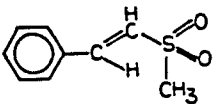
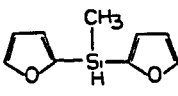
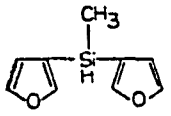
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{10}\text{O}_2^+$ | | | | | | | |
|  | (9.14) IP is onset of photoelectron band (85GLE/JAH). | (187) | (784) | -23 | -98 | *EST | 94595-48-9 |
|  | (≤ 8.85) IP from 78MAR/SCH. | (≤ 167) | (≤ 699) | -37 | -155 | *EST | 67843-62-3 |
|  | (≤ 8.85) IP from 78MAR/SCH. | (≤ 160) | (≤ 668) | -44 | -186 | *EST | 67843-61-2 |
|  | (8.4) IP is onset of photoelectron band. | (243) | (1017) | 49 | 207 | *EST | 70705-73-6 |
| $\text{C}_9\text{H}_{10}\text{O}_2\text{S}^+$ | | | | | | | |
|  | (8.7) IP from 84CAU/FUR. | (164) | (684) | -37 | -155 | *EST | |
| $\text{C}_9\text{H}_{10}\text{O}_2\text{Si}^+$ | | | | | | | |
|  | (8.0) IP is onset of photoelectron band (83ZYK/ERC). | (156) | (651) | -29 | -121 | *EST | 1911-24-6 |
|  | (8.1) IP is onset of photoelectron band (83ZYK/ERC). | (157) | (661) | -29 | -121 | *EST | 73357-16-1 |

Table 1. Positive Ion Table - Continued

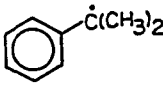
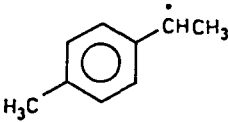
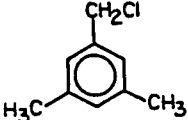
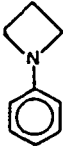
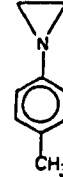
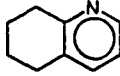
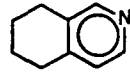
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|----------------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{11}^+$ | | | | | | | |
|  | (6.6) | (186) (196) | (777) (821) | 32.4±1.5 | 135.5±6.3 | 81ROB/STE | 16804-70-9 |
| | | Value at 298 K from proton affinity of $\text{C}_6\text{H}_5\text{C}(\text{CH}_3)=\text{CH}_2$ (RN 98-83-9). PA = 207.0 kcal/mol, 866. kJ/mol Value at 0 K from appearance potential measurements (83BRA/BAE, 85BUT/LER). See also: 85DOM/LAK. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | |
|  | | 186 | 780 | | | | |
| | | From proton affinity of 4- $\text{CH}_3\text{C}_6\text{H}_4\text{CH}=\text{CH}_2$ (RN 622-97-9) (84HAR/HOU). PA = 206.8 kcal/mol, 865. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{11}\text{Cl}^+$ | | | | | | | |
|  | (8.63±0.03) | (187) | (784) | -12 | -49 | *EST | 2745-54-2 |
| $\text{C}_9\text{H}_{11}\text{N}^+$ | | | | | | | |
|  | (7.1) | (222) | (929) | 58 | 244 | *EST | 3334-89-2 |
| | | IP is onset of photoelectron band (82ROZ/HOU2). | | | | | |
|  | (≤8.0) | (≤248) | (≤1037) | 63 | 265 | *EST | 38201-24-0 |
| | | IP from 82CRI/LIC. | | | | | |
|  | (≤9.15) | (≤229) | (≤957) | 18 | 74 | *EST | 10500-57-9 |
| | | IP from 79AUE/BOW. | | | | | |
|  | (≤9.19) | (≤230) | (≤963) | 18 | 76 | *EST | 36556-06-6 |
| | | IP from 79AUE/BOW. | | | | | |

Table 1. Positive Ion Table - Continued

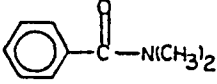
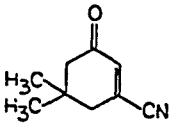
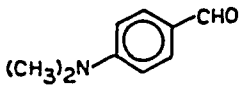
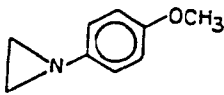
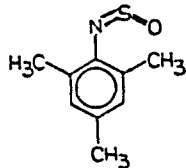
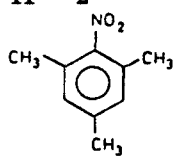
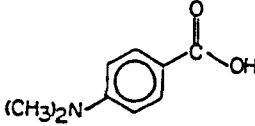
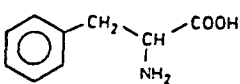
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₁NO⁺ | | | | | | | |
|  | (≤ 9.04) IP from 85GAL/GER. | (≤ 186) | (≤ 777) | -23 | -95 | *EST | 611-74-5 |
|  | (≤ 9.72) IP from 82PFI/GER. | (≤ 214) | (≤ 896) | -10 | -42 | *EST | 65115-71-1 |
|  | 7.36 \pm 0.02 See also: 85GAL/GER. | (160) | (670) | -10 | -40 | *EST | 100-10-7 |
|  | (≤ 7.6) IP from 82CRI/LIC. | (≤ 210) | (≤ 880) | 35 | 147 | *EST | 27347-09-7 |
| C₉H₁₁NOS⁺ | | | | | | | |
|  | (8.2) IP is onset of photoelectron band (82LOU/VAN). | (160) | (670) | -29 | -121 | *EST | |
| C₉H₁₁NO₂⁺ | | | | | | | |
|  | (8.8) IP is onset of photoelectron band. | (200) | (836) | -3 | -13 | *EST | 603-71-4 |
|  | (7.1) IP is onset of photoelectron band (81MEE/WAH). | (97) | (405) | -67 | -280 | *EST | 619-84-1 |
|  | 8.4 See also: 83CAN/HAM. | 119 | 497 | -74.8 \pm 0.3 | -312.9 \pm 1.2 | 77PED/RYL | 150-30-1 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------|--|--------------------------|----------------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{11}\text{NO}_3^+$ | | (8.0) | (68) | (286) | -116 | -486 | *EST | 556-03-6 |
| | | IP is onset of photoelectron band(83CAN/HAM). | | | | | | |
| $\text{C}_9\text{H}_{11}\text{NS}^+$ | | (≤ 7.70) | (≤ 212) | (≤ 885) | 34 | 142 | *EST | 15482-60-7 |
| | | IP from 82BER/HEN. | | | | | | |
| $\text{C}_9\text{H}_{11}\text{NSe}^+$ | | (≤ 7.33) | (≤ 220) | (≤ 919) | 51 | 212 | *EST | 13120-03-1 |
| | | IP from 82BER/HEN. | | | | | | |
| $\text{C}_9\text{H}_{11}\text{O}^+$ | | | 128 | 535 | | | | |
| | | From proton affinity of (4- CH_3) $\text{C}_6\text{H}_4\text{COCH}_3$ (RN 122-00-9). PA = 208.7 kcal/mol, 873. kJ/mol. | | | | | | |
| | | | 150 | 628 | | | | |
| | | From proton affinity of (4- CH_3O) $\text{C}_6\text{H}_4\text{CH}=\text{CH}_2$ (RN 637-69-4) (84HAR/HOU). PA = 214.4 kcal/mol, 897. kJ/mol. | | | | | | |
| $\text{C}_9\text{H}_{12}^+$ | | 8.43±0.02 | 228 | 954 | 34 | 141 | 76JEN | 696-86-6 |
| | | 8.72±0.01 | 203 | 849 | 1.9±0.2 | 7.9±0.7 | 77PED/RYL | 103-65-1 |
| | | Value from charge transfer equilibrium constant determinations (78LIA/AUS) is in agreement. See also: 82SEL/HEL. | | | | | | |
| | | 8.73±0.01 | 202 | 846 | 1.0±0.2 | 4.0±1.0 | 77PED/RYL | 98-82-8 |
| | | Value from charge transfer equilibrium constant determinations (78LIA/AUS) is in agreement. See also: 84HOW/GON. | | | | | | |

Table 1. Positive Ion Table - Continued

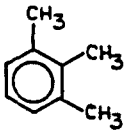
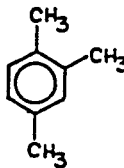
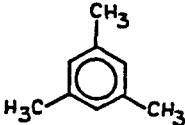
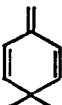
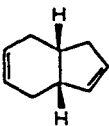
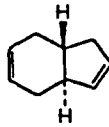
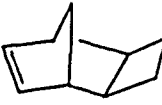
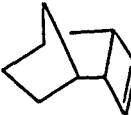
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{12}^+$  | 8.42±0.02 | 192 | 803 | -2.3±0.2 | -9.5±1.1 | 77PED/RYL | 526-73-8 |
| From charge transfer equilibrium constant determinations (78LIA/AUS). See: 84HOW/GON. | | | | | | | |
|  | 8.27±0.01 | 187 | 784 | -3.3±0.2 | -13.8±1.0 | 77PED/RYL | 95-63-6 |
| IP from 77ROS/DRA, 84HOW/GON. | | | | | | | |
|  | 8.41±0.01 | 190 | 795 | -3.8±0.3 | -15.9±1.3 | 77PED/RYL | 108-67-8 |
| Value from charge transfer equilibrium constant determinations (78LIA/AUS) is in agreement. See also: 83CET/LAP, 84HOW/GON. | | | | | | | |
|  | (8.07) | (207) | (868) | 21 | 89 | *EST | 29304-70-9 |
| IP from 83BAL/NEU. | | | | | | | |
|  | 8.81±0.03 | 229 | 959 | 26.1±0.3 | 109±1 | 72KOZ/TIM | |
| IP from 85TUR/PAN. | | | | | | | |
|  | (8.89) | (236) | (988) | 31 | 130 | *EST | |
| IP from 85TUR/PAN. | | | | | | | |
|  | (8.3) | (231) | (965) | 39 | 164 | *EST | 16529-82-1 |
| IP is onset of photoelectron band. | | | | | | | |
|  | (≤8.92±0.03) | (≤252) | (≤1056) | 47 | 195 | *EST | 16529-83-2 |

Table 1. Positive Ion Table - Continued


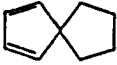
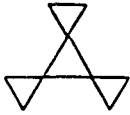
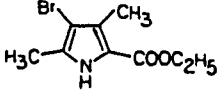
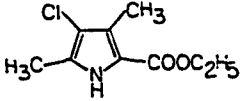
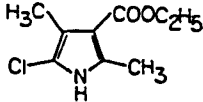
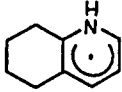
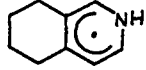
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₂⁺ | | | | | | | |
|  | (8.2) IP is onset of photoelectron band. | (283) | (1185) | 94.1 | 393.7 | 81GOD/SCH | 3105-29-1 |
|  | ≤8.10 IP is onset of photoelectron band. | (≤217) | (≤910) | 31 | 128 | *EST | 766-29-0 |
|  | (≤9.12) | (≤317) | (≤1325) | 106 | 445 | *EST | 31561-59-8 |
| C₉H₁₂BrO₂⁺ | | | | | | | |
|  | (7.75) IP is onset of photoelectron band (81CAU/GIA). | (88) | (369) | -91 | -379 | *EST | 5408-07-1 |
| C₉H₁₂ClNO₂⁺ | | | | | | | |
|  | (≤8.03) IP from 81CAU/GIA. | (≤83) | (≤346) | -103 | -429 | *EST | 58921-31-6 |
|  | (≤7.94) IP from 81CAU/GIA. | (≤81) | (≤337) | -103 | -429 | *EST | 56453-93-1 |
| C₉H₁₂N⁺ | | | | | | | |
|  | (156) (651) From proton affinity of 2,3-cyclohexenopyridine (RN 10500-57-9). PA = (227.7) kcal/mol, (953.) kJ/mol. | | | | | | |
|  | (156) (653) From proton affinity of 3,4-cyclohexenopyridine (RN 36566-06-6). PA = (227.7) kcal/mol, (953.) kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

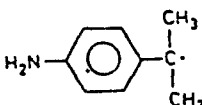
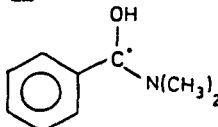
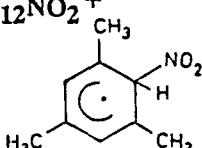
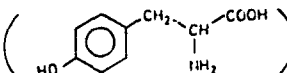
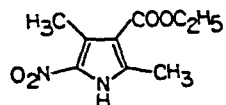
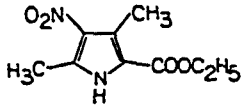
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{12}\text{N}^+$ | | | | | | | |
|  | | 179 | 750 | | | | |
| | | From appearance energy determination (86ORL/MIS). | | | | | |
| $\text{C}_9\text{H}_{12}\text{NO}^+$ | | | | | | | |
|  | | (122) | (510) | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{CON}(\text{CH}_3)_2$ (RN 611-74-5) (86TAF). PA = (221) kcal/mol, (925) kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{12}\text{NO}_2^+$ | | | | | | | |
|  | | 164 | 687 | | | | |
| | | From proton affinity of 2,4,6-(CH_3) $_3\text{C}_6\text{H}_2\text{NO}_2$ (RN 603-71-4) (84ROL/HOU). PA = 198.4 kcal/mol, 830. kJ/mol. | | | | | |
| $\left(\text{C}_6\text{H}_5\text{CH}_2\text{CH}(\text{COOH})\text{NH}_2 \right)^+$ | | 74 | 311 | | | | |
| | | From proton affinity of L- $\text{C}_6\text{H}_5\text{CH}_2\text{CH}(\text{NH}_2)\text{COOH}$ (RN 150-30-1). PA = 216.5 kcal/mol, 906. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{12}\text{NO}_3^+$ | | | | | | | |
|  | | 27 | 114 | | | | |
| | | From proton affinity of L-tyrosine. PA = 222.3 kcal/mol, 930. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{12}\text{N}_2\text{O}_4^+$ | | | | | | | |
|  | (≤ 8.78) | (≤ 103) | (≤ 432) | -99 | -415 | *EST | |
| | IP from 81CAU/GIA. | | | | | | |
|  | (≤ 8.76) | (≤ 103) | (≤ 430) | -99 | -415 | *EST | |
| | IP from 81CAU/GIA. | | | | | | |

Table 1. Positive Ion Table - Continued

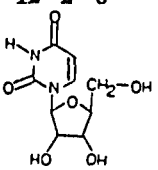
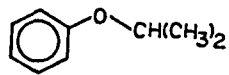
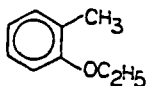
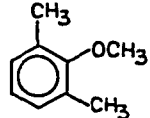
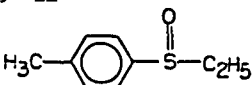
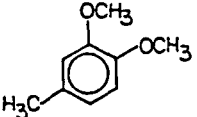
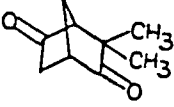
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{12}\text{N}_2\text{O}_6^+$  | (9.0) | (142) | (593) | -66 | -275 | *EST | 58-96-8 |
| $\text{C}_9\text{H}_{12}\text{O}^+$ | | | | | | | |
|  | (≤ 8.32) | (≤ 159) | (≤ 665) | -33 | -138 | *EST | 2741-16-4 |
|  | (8.0) IP from 81BAK/ARM. | (152) | (637) | -32 | -135 | *EST | 614-71-1 |
|  | 8.10 \pm 0.02 | (161) | (674) | -26 | -108 | *EST | 1004-66-6 |
| $\text{C}_9\text{H}_{12}\text{OS}^+$ | | | | | | | |
|  | (≤ 8.56) IP from 81MOH/JIA. | (≤ 180) | (≤ 752) | -18 | -74 | *EST | 6378-07-0 |
| $\text{C}_9\text{H}_{12}\text{O}_2^+$ | | | | | | | |
|  | (≤ 7.95) | (≤ 122) | (≤ 511) | -61 | -256 | *EST | 494-99-5 |
|  | (9.0) IP is onset of photoelectron band (80FRO/WES). | (143) | (599) | -64 | -269 | *EST | 38476-46-9 |

Table 1. Positive Ion Table - Continued

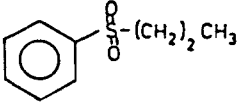
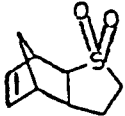
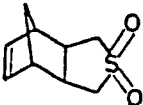
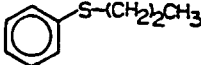
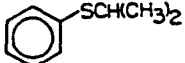
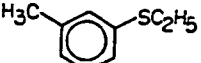
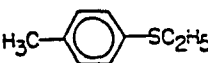
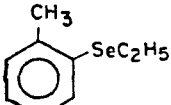
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₂O₂S⁺ | | | | | | | |
|  | (9.21±0.03) | (136) | (570) | -76.3±0.5 | -319±2 | *EST | 13596-75-3 |
|  | (9.1) IP is onset of photoelectron band (84AIT/GOS). | (176) | (735) | -34 | -143 | *EST | |
|  | (9.4) IP is onset of photoelectron band (84AIT/GOS). | (183) | (764) | -34 | -143 | *EST | |
| C₉H₁₂S⁺ | | | | | | | |
|  | (7.81±0.03) | (194) | (811) | 14 | 57 | *EST | 874-79-3 |
|  | (7.9) IP is onset of photoelectron band. | (195) | (814) | 12 | 52 | *EST | 3019-20-3 |
|  | (≤7.92) | (≤193) | (≤808) | 11 | 44 | *EST | 34786-24-8 |
|  | (8.0) IP is onset of photoelectron band. | (195) | (816) | 11 | 44 | *EST | 622-63-9 |
| C₉H₁₂Se⁺ | | | | | | | |
|  | (7.3) IP is onset of photoelectron band (81BAK/ARM). | (193) | (806) | 24 | 102 | *EST | 37773-42-5 |

Table 1. Positive Ion Table - Continued

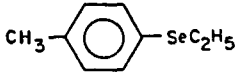
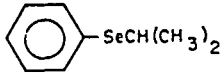
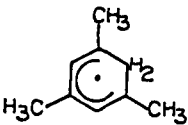
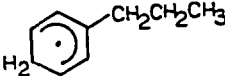
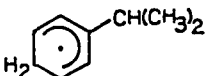
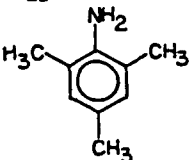
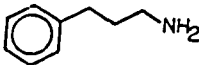
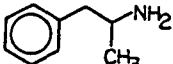
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₂Se⁺ | | | | | | | |
| Neutral  | (7.1) | (187) | (784) | 24 | 99 | *EST | 37773-43-6 |
| IP is onset of photoelectron band (81BAK/ARM). | | | | | | | |
|  | (≤ 8.3) | (≤ 217) | (≤ 908) | 26 | 107 | *EST | 22233-89-2 |
| IP from 81BAK/ARM. | | | | | | | |
| C₉H₁₃⁺ | | | | | | | |
|  | | 161 | 674 | | | | |
| From proton affinity of 1,3,5-(CH ₃) ₃ C ₆ H ₃ (RN 108-67-8). PA = 200.7 kcal/mol, 840. kJ/mol. | | | | | | | |
|  | | 175 | 733 | | | | |
| From proton affinity of n-C ₃ H ₇ C ₆ H ₅ (RN 103-65-1). PA = 192.4 kcal/mol, 805. kJ/mol. | | | | | | | |
|  | | 175 | 730 | | | | |
| From proton affinity of i-C ₃ H ₇ C ₆ H ₅ (RN 98-82-8). PA = 192.1 kcal/mol, 804. kJ/mol. | | | | | | | |
| C₉H₁₃N⁺ | | | | | | | |
|  | (7.15) | (164) | (686) | -1 | -4 | *EST | 88-05-1 |
| See also: 83CET/LAP | | | | | | | |
|  | ($\leq 8.89 \pm 0.12$) | (≤ 216) | (≤ 902) | 11 | 44 | *EST | 2038-57-5 |
|  | (8.5) | (203) | (849) | 7 | 29 | *EST | 300-62-9 |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

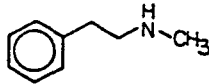
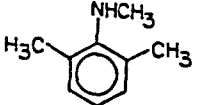
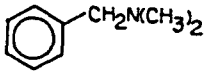
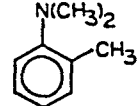
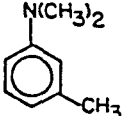
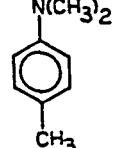
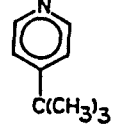
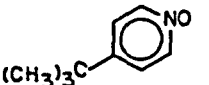
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{13}\text{N}^+$ | | | | | | | |
|  | (8.4) IP is onset of photoelectron band. | (205) | (857) | 11 | 47 | *EST | 589-08-2 |
|  | (7.34) | (182) | (763) | 13 | 55 | *EST | 767-71-5 |
|  | 7.69 See also: 81LOG/TAK, 79AUE/BOW. | (197) | (826) | 20 | 84 | *EST | 103-83-3 |
|  | 7.40±0.02 | (195) | (813) | 24 | 99 | *EST | 609-72-3 |
|  | 7.02 IP from charge transfer equilibrium constant determinations (85LIA/JAC). Reference standard: IP(C ₆ H ₅ N(CH ₃) ₂) = 7.12 eV. | (178) | (744) | 16 | 67 | *EST | 121-72-2 |
|  | 6.93 IP from charge transfer equilibrium constant determinations (85LIA/JAC, 84MAU/NEL); Reference standard: IP(C ₆ H ₅ N(CH ₃) ₂) = 7.12 eV. | (177) | (739) | 17 | 70 | *EST | 99-97-8 |
|  | (≤9.30±0.05) | (≤222) | (≤929) | 8 | 32 | *EST | 3978-81-2 |
| $\text{C}_9\text{H}_{13}\text{NO}^+$ | | | | | | | |
|  | (7.8) IP is onset of photoelectron band. | (169) | (705) | -11 | -48 | *EST | 23569-17-7 |

Table 1. Positive Ion Table - Continued

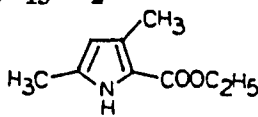
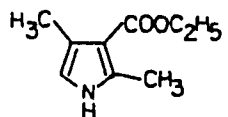
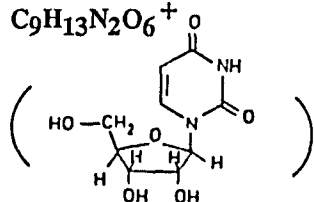
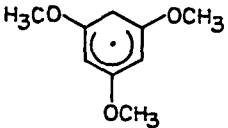
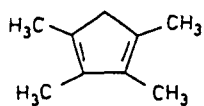
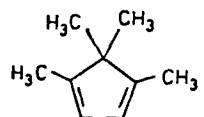


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|---|------------|---|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{13}\text{NO}_2^+$  | (≤ 7.91) IP from 81CAU/GIA. | (≤ 87) | (≤ 363) | -96 | -400 | *EST | |
|  | (≤ 7.95) IP from 81CAU/GIA. | (≤ 88) | (≤ 367) | -96 | -400 | *EST | |
| $\text{C}_9\text{H}_{13}\text{N}_2\text{O}_6^+$  | | | | -66 | -275 | From proton affinity of uridine (RN 58-96-8) PA = (208) kcal/mol, (870) kJ/mol. | |
| $\text{C}_9\text{H}_{13}\text{O}_3^+$  | | 55 | 228 | From proton affinity of 1,3,5- $\text{C}_6\text{H}_3(\text{OCH}_3)_3$ (RN 621-23-8) PA = 220.6 kcal/mol, 923. kJ/mol. | | | |
| $\text{C}_9\text{H}_{14}^+$  | 7.8 ± 0.1 | (183) | (768) | 4 | 15 | *EST | 4249-10-9 |
|  | 7.84 ± 0.05 | (183) | (765) | 2 | 9 | *EST | 4249-11-0 |
|  | (8.0) IP is onset of photoelectron band. | (195) | (817) | 11 | 45 | 81MAI/SCH | 17530-61-9 |
|  | (8.7) IP is onset of photoelectron band. | (200) | (835) | -1 ± 0.7 | -4 ± 3 | 83JOC/DEK | 7124-86-9 |

Table 1. Positive Ion Table - Continued

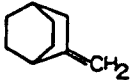





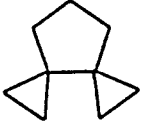
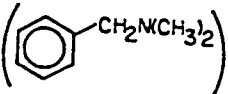
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|---|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{14}^+$ | | | | | | | |
|  | (≤ 8.87) | (≤ 202) | (≤ 847) | -2 ± 0.7 | -9 ± 3 | 77PED/RYL | 2972-20-5 |
|  | (8.3) IP is onset of photoelectron band (82SPA/GLE). | (210) | (879) | 19 | 78 | *EST | 81969-71-3 |
|  | (8.2) IP is onset of photoelectron band (82SPA/GLE). | (208) | (869) | 19 | 78 | *EST | 81969-72-4 |
|  | ($\leq 9.65 \pm 0.03$) | (≤ 236) | (≤ 987) | 13 | 56 | 73ENG/AND | 16526-28-6 |
|  | (8.8) IP is onset of photoelectron band. | (211) | (884) | 8 | 35 | 73ENG/AND | 16526-27-5 |
|  | (8.73) | (208) | (872) | 7 | 30 | *EST | 873-12-1 |
|  | (8.3) IP is onset of photoelectron band (82SPA/GLE). | (237) | (993) | 46 | 192 | *EST | 24973-90-8 |
| $\text{C}_9\text{H}_{14}\text{N}^+$ | | | | | | | |
|  | H^+ | (158) | (660) | From proton affinity of $\text{C}_6\text{H}_5\text{CH}_2\text{N}(\text{CH}_3)_2$ (RN 103-83-3). PA = 228.1 kcal/mol, 954. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

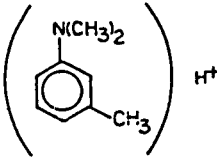
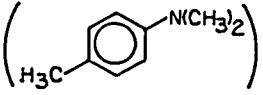
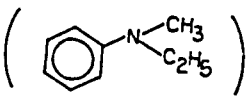
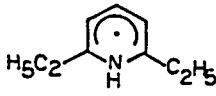
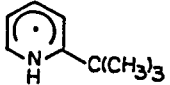
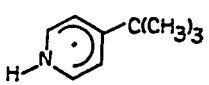
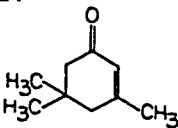
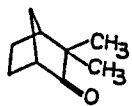
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-------------------------------|--------------------------|--------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₄N⁺ | | | | | | | |
|  | | 157 | 658 | From proton affinity of 3-CH ₃ C ₆ H ₄ N(CH ₃) ₂ (RN 121-72-2). PA = 224.5 kcal/mol, 939. kJ/mol. | | | |
|  | | 157 | 656 | From proton affinity of 4-CH ₃ C ₆ H ₄ N(CH ₃) ₂ (RN 99-97-8). PA = 225.6 kcal/mol, 944. kJ/mol. | | | |
|  | | (156) | (651) | From proton affinity of C ₆ H ₅ N(CH ₃)(C ₂ H ₅) (RN 613-97-8). PA = 227.1 kcal/mol, 950. kJ/mol. | | | |
|  | | (139) | (582) | From proton affinity of 2,6-diethylpyridine (RN 935-28-4). PA = 231.1 kcal/mol, 967. kJ/mol. | | | |
|  | | (145) | (607) | From proton affinity of 2-tert-butylpyridine (RN 5944-41-2). PA = (227.4) kcal/mol, (951.) kJ/mol. | | | |
|  | | (147) | (616) | From proton affinity of 4-tert-butylpyridine (RN 3978-81-2). PA = 225.9 kcal/mol, 945. kJ/mol. | | | |
| C₉H₁₄O⁺ | | | | | | | |
|  | (≤9.07) IP from 82PFI/GER. | (≤160) | (≤669) | -49 | -206 | *EST | 78-59-1 |
|  | (8.6) | (144) | (603) | -54 | -227 | *EST | 13211-15-9 |
| IP is onset of photoelectron band (80FRO/WES). | | | | | | | |

Table 1. Positive Ion Table - Continued

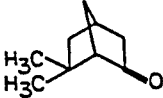
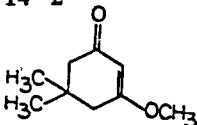
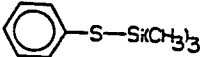
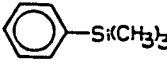
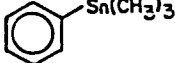


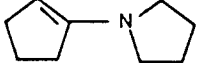
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{14}\text{O}^+$  | (8.75) IP is onset of photoelectron band (80FRO/WES). | (148) | (617) | -54 | -227 | *EST | 38476-45-8 |
| $\text{C}_9\text{H}_{14}\text{O}_2^+$  | (≤ 9.35) IP from 82PFI/GER. | (≤ 138) | (≤ 576) | -78 | -326 | *EST | 4683-45-8 |
| $\text{C}_9\text{H}_{14}\text{SSi}^+$  | (8.67 \pm 0.05) | (166) | (696) | -34 | -141 | *EST | 4551-15-9 |
| $\text{C}_9\text{H}_{14}\text{Si}^+$  | 8.22 IP is onset of photoelectron band (82TRA/RED). | (168) | (704) | -21 | -89 | *EST | 768-32-1 |
| $\text{C}_9\text{H}_{14}\text{Sn}^+$  | 8.83 \pm 0.05 | 231 | 965 | 27 \pm 1 | 113 \pm 5 | 77PED/RYL | 934-56-5 |
| $\text{C}_9\text{H}_{15}\text{N}^+$ ($\text{CH}_2=\text{CHCH}_2$) ₃ N  | (7.5) IP is onset of photoelectron band. See also: 79AUE/BOW. | (226) | (948) | 54 | 224 | *EST | 102-70-5 |
|  | (≤ 7.46) IP from 81MUL/PRE2. | (≤ 197) | (≤ 826) | 25 | 106 | *EST | 7326-44-5 |
|  | 7.1 IP from 79AUE/BOW. | (164) | (686) | 0.2 | 1 | *EST | 7148-07-4 |

Table 1. Positive Ion Table - Continued

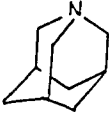
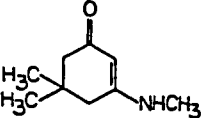
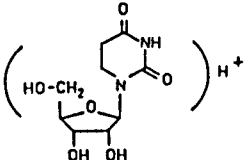
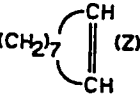


| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|-------------------------------|--------------------------|--------|------------------------------|-----------|---|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₅N⁺  | (7.57±0.02) | (165) | (692) | -9 | -38 | *EST | 281-27-6 |
| C₉H₁₅NO⁺  | (≤8.11) IP from 82PFI/GER. | (≤144) | (≤602) | -43 | -180 | *EST | 701-58-6 |
| C₉H₁₅N₂O₆⁺  | | | | -76 | -317 | From proton affinity of 5,6-dihydrouridine (RN 5627-05-4). PA = (208) kcal/mol, (870) kJ/mol. | |
| C₉H₁₆⁺ (E)-n-C ₅ H ₁₁ CH=CHCH=CH ₂ | (8.44) IP from 81MAS/MOU. | (193) | (809) | -1 | -5 | *EST | 56700-77-7 |
| 1-C ₉ H ₁₆ | (9.93±0.02) | (244) | (1020) | 15±0.7 | 62±3 | 79ROG/DAG | 3452-09-3 |
| 2-C ₉ H ₁₆ | 9.30±0.02 | 225 | 941 | 11±1 | 44±3 | 79ROG/DAG | 19447-29-1 |
| 3-C ₉ H ₁₆ | 9.20±0.01 | 222 | 930 | 10±0.7 | 42±3 | 79ROG/DAG | 20184-89-8 |
| 4-C ₉ H ₁₆ | (9.17±0.03) | (221) | (927) | 10±0.7 | 42±3 | 79ROG/DAG | 20184-91-2 |
|  (Z) | (8.81±0.15) | (190) | (795) | -13 | -55 | 78GRE/LIE | 933-21-1 |
|  | ≤9.36 | ≤210 | ≤878 | -6.0±0.3 | -25.2±1.4 | 84WIB/LUP | 39124-79-3 |
|  | (≤9.4) | (≤211) | (≤884) | -5.5±0.2 | -23±1 | 77PED/RYL | 286-60-2 |

Table 1. Positive Ion Table - Continued

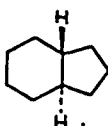
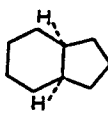

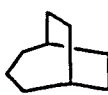
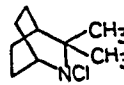
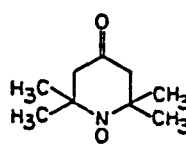
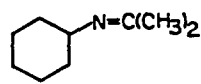
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{16}^+$  | (9.46±0.06) IP from 80MIK/ZAI. | (187) | (782) | -31±0.5 | -131±2 | 77PED/RYL | 3296-50-2 |
|  | (9.46±0.06) IP from 80MIK/ZAI. | (188) | (786) | -30±0.5 | -127±2 | 77PED/RYL | 4551-51-3 |
|  | (9.35) | (185) | (774) | -31±1 | -128±3 | 77PAR/STE | 280-65-9 |
|  | (9.0) IP is onset of photoelectron band. | (183) | (765) | -25 | -103 | 81MAI/SCH | 283-19-2 |
| $\text{C}_9\text{H}_{16}\text{ClN}^+$  | (≤8.34) IP from 82NEL/GAN. | (≤179) | (≤748) | -14 | -57 | *EST | 82666-06-6 |
| $\text{C}_9\text{H}_{16}\text{N}^+$ ($\text{CH}_2 = \text{CHCH}_2$) ₃ NH | | (189) | (792) | | | | |
| | From proton affinity of ($\text{CH}_2 = \text{CHCH}_2$) ₃ N (RN 102-70-5). PA = 230.0 kcal/mol, 962. kJ/mol. | | | | | | |
| $\text{C}_9\text{H}_{16}\text{NO}_2^+$  | 7.40±0.05 | 120 | 499 | -51±2 | -215±7 | 77PED/RYL | 2896-70-0 |
| $\text{C}_9\text{H}_{17}\text{N}^+$  | (8.23) | (183) | (763) | -7 | -31 | *EST | 6407-36-9 |

Table 1. Positive Ion Table - Continued

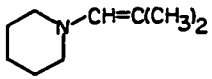
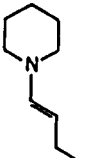
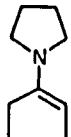
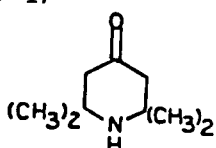
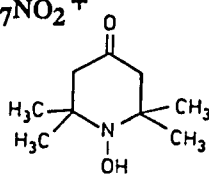
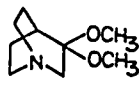
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₇N⁺ | | | | | | | |
|  | ($\leq 7.93 \pm 0.03$) See also: 81MUL/PRE2. | (≤ 172) | (≤ 717) | -11 | -48 | *EST | 673-33-6 |
|  | ≤ 7.46 IP from 81MUL/PRE2. | (≤ 160) | (≤ 670) | -12 | -50 | *EST | 7182-10-7 |
|  | (≤ 7.29) IP from 81MUL/PRE2. | (≤ 164) | (≤ 688) | -4 | -15 | *EST | 13750-57-7 |
| C₉H₁₇NO⁺ | | | | | | | |
|  | (7.74) | (113) | (474) | -65 \pm 1 | -273 \pm 4 | 77PED/RYL | 826-36-8 |
| C₉H₁₇NO₂⁺ | | | | | | | |
|  | (8.51 \pm 0.05) | (125) | (523) | -71 \pm 1 | -298 \pm 5 | 77PED/RYL | 3637-11-4 |
|  | (≤ 7.9) IP from 79AUE/BOW. | (≤ 104) | (≤ 436) | -78 | -326 | *EST | |
| C₉H₁₈⁺ | | | | | | | |
| 1-C ₉ H ₁₈ | (9.42 \pm 0.01) | (192) | (805) | -25 \pm 0.2 | -104 \pm 1 | 74ROG/KAN | 124-11-8 |
| n-C ₄ H ₉ C(CH ₃)=C(CH ₃) ₂ | (8.14 \pm 0.01) | (157) | (655) | -31 | -131 | *EST | 3074-64-4 |
| (E)-C ₃ H ₇ C(CH ₃)=C(CH ₃)C ₂ H ₅ | (8.08 \pm 0.01) | (155) | (649) | -31 | -130 | *EST | 3074-67-7 |
| (Z)-2-C ₉ H ₁₈ | (8.90 \pm 0.01) | (179) | (748) | -26 | -111 | *EST | 6434-77-1 |
| (E)-2-C ₉ H ₁₈ | (8.90 \pm 0.01) | (178) | (745) | -27 | -114 | *EST | 6434-78-2 |

Table 1. Positive Ion Table - Continued

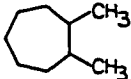
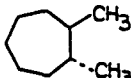
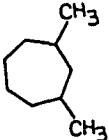
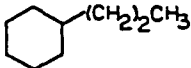
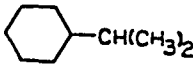
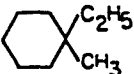
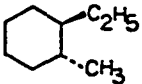
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{18}^+$ | | | | | | | | |
| (Z)-3- C_9H_{18} | | (8.84±0.01) | (178) | (743) | -26 | -110 | *EST | 20237-46-1 |
| (E)-3- C_9H_{18} | | 8.84±0.01 | (177) | (739) | -27 | -114 | *EST | 20063-92-7 |
| (Z)-4- C_9H_{18} | | (8.80±0.01) | (177) | (739) | -26 | -110 | *EST | 10405-84-2 |
| (E)-4- C_9H_{18} | | (8.81±0.01) | (176) | (736) | -27 | -114 | *EST | 10405-85-3 |
|  | | (10.21) IP from 81HER/SIC. | (196) | (818) | -40 | -167 | *EST | 13151-51-4 |
|  | | (10.31) IP from 81HER/SIC. | (196) | (820) | -42 | -175 | *EST | 13151-50-3 |
|  | | (10.41) IP from 81HER/SIC. | (197) | (824) | -43 | -180 | *EST | 13151-53-6 |
|  | | (9.46) From charge transfer equilibrium constants (82SIE/MAU). Reference standards, IP's of fluorobenzenes. | (172) | (720) | -46.0±0.2 | -192.7±0.7 | 77PED/RYL | 1678-92-8 |
|  | | (9.33) From charge transfer equilibrium constants (82SIE/MAU). Reference standards, IP's of fluorobenzenes. IP from photoionization threshold:9.55 eV. | (168) | (703) | -47 | -197 | *EST | 696-29-7 |
|  | | (9.34) From charge transfer equilibrium constants (82SIE/MAU). Reference standards, IP's of fluorobenzenes. | (169) | (706) | -47±0.5 | -195±2 | 77PED/RYL | 4926-90-3 |
|  | | (9.32) From charge transfer equilibrium constants (82SIE/MAU). Reference standards, IP's of fluorobenzenes. | (168) | (704) | -47±0.5 | -195±2 | 77PED/RYL | 4923-78-8 |

Table 1. Positive Ion Table - Continued

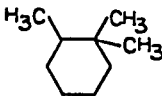
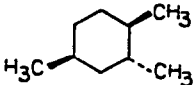
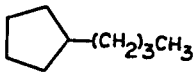
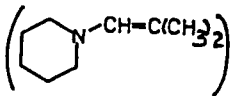
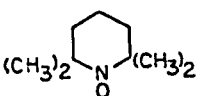
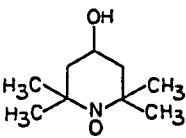
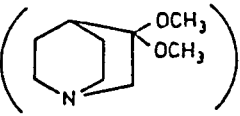
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₉H₁₈⁺ | | | | | | | |
|  | 9.39 | 160 | 671 | -56.2 | -235.1 | 69STU/WES | 3073-66-3 |
| | IP from charge transfer equilibrium constants (82SIE/MAU). Reference standards, IP's of fluorobenzenes. | | | | | | |
|  | (9.38) | (166) | (695) | -50 | -210 | *EST | 2234-75-5 |
| | IP from 82SIE/MAU. | | | | | | |
|  | (9.95±0.03) | (189) | (792) | -40 | -168 | 71ASTM | 2040-95-1 |
| C₉H₁₈N⁺ | | | | | | | |
|  | | (128) | (534) | | | | |
| | From proton affinity of 1-(N-piperidino)isobutene (RN 673-33-6). PA = (230.7) kcal/mol, (965.) kJ/mol. | | | | | | |
| C₉H₁₈NO⁺ | | | | | | | |
|  | (6.73) | (181) | (757) | 26±2 | 108±10 | *EST | 2564-83-2 |
| C₉H₁₈NO₂⁺ | | | | | | | |
|  | (7.4±0.1) | (101) | (423) | -70±2 | -291±9 | 77PED/RYL | 2226-96-2 |
|  | | (56) | (233) | | | | |
| | From proton affinity of 3,3-dimethoxy-1-azabicyclo[2.2.2]octane. PA = (232) kcal/mol, (971) kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued


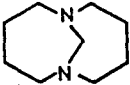
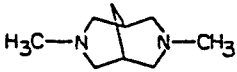
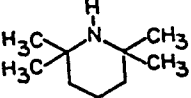

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{18}\text{N}_2^+$ | | | | | | | |
|  | 7.0 | (166) | (694) | 5 | 19 | *EST | |
| | IP is onset of photoelectron band (85HON/YAN). | | | | | | |
|  | 7.0 | (157) | (656) | -5 | -19 | *EST | |
| | IP is onset of photoelectron band (85HON/YAN). | | | | | | |
|  | (6.8) | (164) | (685) | 7 | 29 | *EST | 14789-33-4 |
| | IP is onset of photoelectron band (81LIV/ROB). | | | | | | |
| $\text{C}_9\text{H}_{18}\text{O}^+$ | | | | | | | |
| $n\text{-C}_7\text{H}_{15}\text{COCH}_3$ | (9.16) | (130) | (542) | -81 ± 0.5 | -340 ± 2 | 78SEL/STR2 | 821-55-6 |
| | IP from 81HOL/FIN. | | | | | | |
| $(n\text{-C}_4\text{H}_9)_2\text{CO}$ | (9.07) | (127) | (530) | -82.4 ± 0.3 | -344.9 ± 1.2 | 77PED/RYL | 502-56-7 |
| | IP from 81HOL/FIN. | | | | | | |
| $(\text{iso-C}_4\text{H}_9)_2\text{CO}$ | 9.04 ± 0.03 | 123 | 515 | -85.5 ± 0.3 | -357.6 ± 1.1 | 77PED/RYL | 108-83-8 |
| $(t\text{-C}_4\text{H}_9)_2\text{CO}$ | 8.67 ± 0.02 | 117 | 491 | -82.6 ± 0.3 | -345.8 ± 1.1 | 77PED/RYL | 815-24-7 |
| $\text{C}_9\text{H}_{19}^+$ | | | | | | | |
| $(n\text{-C}_4\text{H}_9)(n\text{-C}_3\text{H}_7)(\text{CH}_3)\text{C}$ | | 133 | 556 | | | 84LOS/HOL | 92056-65-0 |
| | From appearance potential measurement (84LOS/HOL). | | | | | | |
| $\text{C}_9\text{H}_{19}\text{N}^+$ | | | | | | | |
|  | 7.59 | 137 | 572 | -38 ± 0.7 | -160 ± 3 | 81SUR/HAC | 768-66-1 |
| | IP from 82ROZ/HOU, 79AUE/BOW. | | | | | | |
| $\text{C}_9\text{H}_{19}\text{N}_2^+$ | | | | | | | |
|  | | 166 | 696 | | | | |
| | From proton affinity of 1,5-diazabicyclo[3.3.3]undecane. PA = 232.4 kcal/mol, 972. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

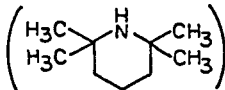
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{19}\text{O}^+$ (tert-C ₄ H ₉) ₂ COH | | 77 | 320 | | | | |
| | | From proton affinity of (tert-C ₄ H ₉) ₂ CO (RN 815-24-7). PA = 206.5 kcal/mol, 864. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{20}^+$ n-C ₉ H ₂₀ | (9.72) | (170) | (709) | -54.5±0.1 | -228.4±0.6 | 74SCO | 111-84-2 |
| | | (182) | (761) | -42.3±0.1 | -177.1±0.6 | | |
| | IP from charge transfer equilibrium constants (81MAU/SIE, 82LIA). Reference standards, fluorobenzenes. | | | | | | |
| $\text{C}_9\text{H}_{20}\text{N}^+$ | | | | | | | |
|  H ⁺ | | (96) | (401) | | | | |
| | From proton affinity of 2,2,6,6-tetramethylpiperidine (RN 768-66-1). PA = (231.7) kcal/mol, (969.) kJ/mol. | | | | | | |
| $\text{C}_9\text{H}_{21}\text{BO}_3^+$ (n-C ₃ H ₇ O) ₃ B | (10.02) | (-26) | (-109) | -257±1 | -1076±5 | 77PED/RYL | 688-71-1 |
| $\text{C}_9\text{H}_{21}\text{N}^+$ (n-C ₃ H ₇) ₃ N | (7.4) | (132) | (552) | -38±0.2 | -161±1 | *EST | 102-69-2 |
| | IP is onset of photoelectron band. | | | | | | |
| tert-C ₅ H ₁₁ (tert-C ₄ H ₉)NH | (7.81±0.1) | (134) | (563) | -46±1 | -191±4 | *EST | 58471-09-3 |
| | $\Delta_f H(\text{Ion})$ predicted from hydrogen affinities of secondary amines: 131 kcal/mol, 548 kJ/mol. Corresponding IP = 7.7 eV. | | | | | | |
| $\text{C}_9\text{H}_{22}\text{N}^+$ (n-C ₃ H ₇) ₃ NH | | (93) | (390) | | | | |
| | From proton affinity of (n-C ₃ H ₇) ₃ N (RN 102-69-2). PA = 234.0 kcal/mol, 979. kJ/mol. | | | | | | |
| (tert-C ₄ H ₉)C(CH ₃) ₂ NH(CH ₃) ₂ | | (97) | (404) | | | | |
| | From proton affinity of (tert-C ₄ H ₉)C(CH ₃) ₂ N(CH ₃) ₂ (RN 3733-36-6). PA = 235.1 kcal/mol, 984. kJ/mol. | | | | | | |
| (tert-C ₅ H ₁₁)(tert-C ₄ H ₉)NH ₂ | | (88) | (366) | | | | |
| | From proton affinity of (tert-C ₅ H ₁₁)(tert-C ₄ H ₉)NH (RN 58471-09-3). PA = 232.5 kcal/mol, 973. kJ/mol. | | | | | | |
| $\text{C}_9\text{H}_{22}\text{OP}^+$ (i-C ₃ H ₇) ₃ POH | | 17 | 73 | | | | |
| | From proton affinity of (i-C ₃ H ₇) ₃ PO (RN 17513-58-5) (85BOL/HOU). PA = 227.5 kcal/mol, 952. kJ/mol. | | | | | | |

Table 1. Positive Ion Table - Continued

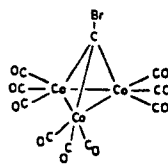
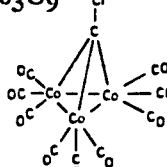
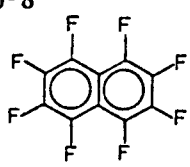
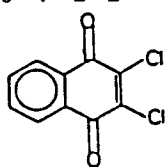
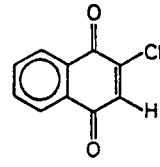
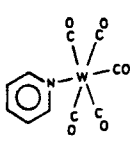
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_9\text{H}_{25}\text{N}_3\text{OP}^+$ $\text{HOP}(\text{CH}_2\text{N}(\text{CH}_3)_2)_3$ | | 51 | 214 | | | | |
| | | From proton affinity of $\text{OP}(\text{CH}_2\text{N}(\text{CH}_3)_2)_3$ (RN 2327-88-0) (85BOL/HOU). PA = 235. kcal/mol, 983. kJ/mol. | | | | | |
| $\text{C}_9\text{H}_{27}\text{NSi}_3^+$ $((\text{CH}_3)_3\text{Si})_3\text{N}$ | (≤ 8.60) | (≤ 38) | (≤ 160) | -160 ± 3 | -670 ± 12 | 77PED/RYL | 1586-73-8 |
| $\text{C}_{10}\text{BrC}_3\text{O}_9^+$ | 7.8 | -81 | -337 | -261 ± 2 | -1090 ± 9 | 82PIL/SKI | 19439-14-6 |
|  | | IP is onset of photoelectron band (81CHE/HAL, 82COS/LLO). See also: 82GRA/TON. | | | | | |
| $\text{C}_{10}\text{ClC}_3\text{O}_9^+$ | 7.8 | -76 | -316 | -255 ± 2 | -1069 ± 10 | 82PIL/SKI | 13682-02-5 |
|  | | IP is onset of photoelectron band (81CHE/HAL, 82GRA/TON, 82COS/LLO). | | | | | |
| $\text{C}_{10}\text{F}_8^+$ | 8.85 | (-88) | (-368) | -292 | -1222 | *EST | 313-72-4 |
|  | | IP from 84HOH/DIS, 82LEV/LIA. | | | | | |
| $\text{C}_{10}\text{H}_4\text{Cl}_2\text{O}_2^+$ | (9.5) | (180) | (754) | -39 ± 2 | -162 ± 10 | *EST | 117-80-6 |
|  | | IP is onset of photoelectron band (80RED/FRE). | | | | | |
| $\text{C}_{10}\text{H}_5\text{ClO}_2^+$ | (9.6) | (189) | (789) | -33 | -137 | *EST | 1010-60-2 |
|  | | IP is onset of photoelectron band (80RED/FRE). | | | | | |
| $\text{C}_{10}\text{H}_5\text{NO}_5\text{W}^+$ | 7.53 \pm 0.05 | 29 | 121 | -145 | -606 | 84ALT/CON2 | 14586-49-3 |
|  | | | | | | | |

Table 1. Positive Ion Table - Continued

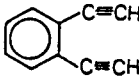
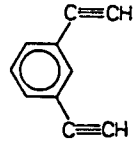

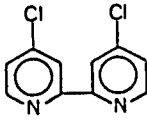
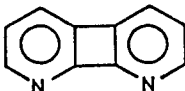
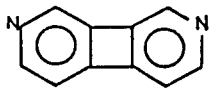
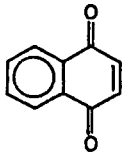
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₆⁺ | | | | | | | |
|  | (8.69±0.02) | (329) | (1378) | 129 | 540 | *EST | 21792-52-9 |
|  | (8.82±0.02) | (332) | (1390) | 129 | 539 | *EST | 1785-61-1 |
|  | (8.58±±0.02) | (327) | (1368) | 129 | 540 | *EST | 935-14-8 |
| C₁₀H₆Cl₂N₂⁺ | | | | | | | |
|  | (8.8) IP is onset of photoelectron band (83DOB/HIL). | (257) | (1074) | 54 | 225 | *EST | 1762-41-0 |
| C₁₀H₆N₂⁺ | | | | | | | |
|  | 7.7 IP is onset of photoelectron band (85YAM/HIG). | (309) | (1294) | 132 | 551 | *EST | |
|  | 8.3 IP is onset of photoelectron band (85YAM/HIG). | (323) | (1352) | 132 | 551 | *EST | |
| C₁₀H₆O₂⁺ | | | | | | | |
|  | 9.56±0.01 See also: 80RED/FRE. | 194 | 811 | -27±1 | -111±4 | 77PED/RYL | 130-15-4 |

Table 1. Positive Ion Table - Continued

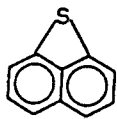
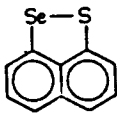
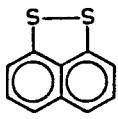
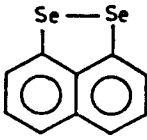
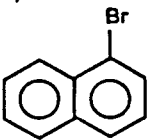
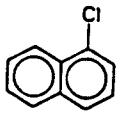
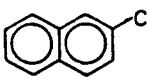
| ION Neutral | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------|----------------|--------------------------|------------|------------------------------|-----------|----------------------|------------------------|
| | eV | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_6\text{S}^+$  | (8.0) | (262) | (1096) | 77 | 324 | *EST | 3968-63-6 | |
| IP is onset of photoelectron band (81BOC/BRA). | | | | | | | | |
| $\text{C}_{10}\text{H}_6\text{SSe}^+$  | (≤ 7.14) | (≤ 236) | (≤ 986) | 71 | 297 | *EST | 64869-35-8 | |
| IP from 81BOC/BRA. | | | | | | | | |
| $\text{C}_{10}\text{H}_6\text{S}_2^+$  | 7.14 | (222) | (931) | 58 | 242 | *EST | 209-22-3 | |
| IP from 81BOC/BRA, 82LEV/LIA. | | | | | | | | |
| $\text{C}_{10}\text{H}_6\text{Se}_2^+$  | (7.06) | (247) | (1033) | 84 | 352 | *EST | 36579-71-2 | |
| IP from 81BOC/BRA. | | | | | | | | |
| $\text{C}_{10}\text{H}_7\text{Br}^+$  | (8.09) | (228) | (954) | 42 | 174 | *EST | 90-11-9 | |
| IP from 83KLA/KOV. | | | | | | | | |
| $\text{C}_{10}\text{H}_7\text{Cl}^+$  | (8.13) | (216) | (904) | 29 \pm 2 | 120 \pm 10 | 77PED/RYL | 90-13-1 | |
| IP from 83KLA/KOV. | | | | | | | | |
|  | (8.11) | (220) | (920) | 33 \pm 2 | 137 \pm 10 | 77PED/RYL | 91-58-7 | |
| IP from 83KLA/KOV. | | | | | | | | |

Table 1. Positive Ion Table - Continued

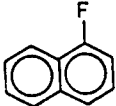
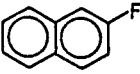
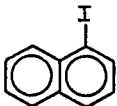
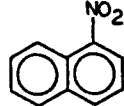
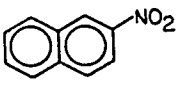
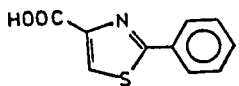
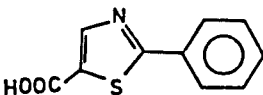
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₇F⁺ | | | | | | | | |
| |  | (8.15) IP from 83KLA/KOV. | (200) | (835) | 12 | 49 | *EST | 321-38-0 |
| |  | (8.23) IP from 83KLA/KOV. | (201) | (843) | 12 | 49 | *EST | 323-09-1 |
| C₁₀H₇I⁺ | | | | | | | | |
| |  | (8.03) IP from 83KLA/KOV. | (241) | (1009) | 56±2 | 234±9 | 77PED/RYL | 76279-71-5 |
| C₁₀H₇NO₂⁺ | | | | | | | | |
| |  | 8.60±0.01 See also: 83KLA/KOV. | 234 | 980 | 36±1 | 150±5 | 77PED/RYL | 86-57-7 |
| |  | 8.65±0.02 IP from 83KLA/KOV, 82LEV/LIA. | (232) | (970) | 32 | 135 | *EST | 581-89-5 |
| C₁₀H₇NO₂S⁺ | | | | | | | | |
| |  | (8.6) IP from 84DEM/SIM. | (168) | (705) | -30 | -125 | *EST | 7113-10-2 |
| |  | (8.7) IP from 84DEM/SIM. | (171) | (714) | -30 | -125 | *EST | 10058-38-5 |

Table 1. Positive Ion Table - Continued

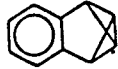
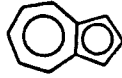

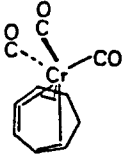
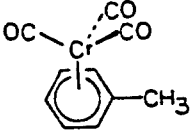
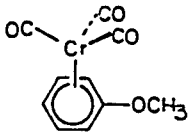
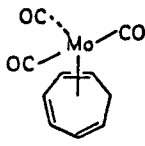
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₈⁺ | | | | | | | |
|  | (8.0) | (238) | (997) | 54 | 225 | *EST | 34305-47-0 |
| | IP is onset of photoelectron band (81GLE/GUB). | | | | | | |
|  | 7.41±0.02 | 240 | 1004 | 69±0.7 | 289±3 | 77PED/RYL | 275-51-4 |
|  | 8.14±0.01 | 223.6 | 935.8 | 35.9±0.3 | 150.4±1 | 82COL/JIM | 91-20-3 |
| C₁₀H₈CrO₃⁺ | | | | | | | |
|  | 6.9±0.2 | (108) | (452) | -51±2 | -214±9 | 77PED/RYL | 12125-72-3 |
|  | (6.6±0.2) | (61) | (257) | -91±1 | -380±5 | 84ALT/CON | 12083-24-8 |
| C₁₀H₈CrO₄⁺ | | | | | | | |
|  | (6.75±0.1) | (39) | (162) | -117 | -489 | 84ALT/CON | 12116-44-8 |
| C₁₀H₈MoO₃⁺ | | | | | | | |
|  | (7.0) | (111) | (466) | -50±1 | -209±7 | 82PIL/SKI | 12125-77-8 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

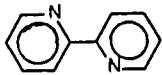
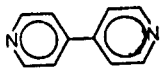
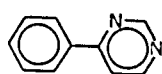
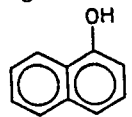
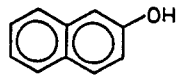
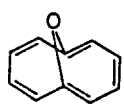
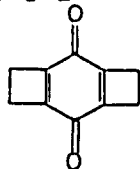
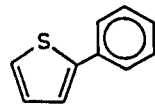
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₈N₂⁺ | | | | | | | |
|  | 8.35±0.02 See also: 83DOB/HIL. | (262) | (1095) | 69.1±1.2 | 289.0±5.2 | 85FAO/AKA | 366-18-7 |
|  | (9.10±0.02) | (283) | (1182) | 73 | 304 | *EST | 553-26-4 |
|  | (8.65) IP is onset of photoelectron band (84BAR/CAU). | (270) | (1131) | 71 | 296 | *EST | 3438-48-0 |
| C₁₀H₈O⁺ | | | | | | | |
|  | 7.76±0.03 | 172 | 719 | -7.1±0.2 | -29.9±1 | 77PED/RYL | 90-15-3 |
|  | 7.85±0.05 IP from 85OIK/ABE, 82LEV/LIA. | 174 | 727 | -7.2±0.3 | -30.3±1.2 | 77PED/RYL | 135-19-3 |
|  | (7.9) IP is onset of photoelectron band (84AND/CER). | (230) | (962) | 48±2 | 200±10 | 77PED/RYL | 4759-11-9 |
| C₁₀H₈O₂⁺ | | | | | | | |
|  | (9.3) IP is onset of photoelectron band (85GLE/JAH). | (241) | (1009) | 27 | 112 | *EST | 87258-06-8 |
| C₁₀H₈S⁺ | | | | | | | |
|  | (8.06) | (235) | (981) | 49 | 203 | *EST | 825-55-8 |

Table 1. Positive Ion Table - Continued

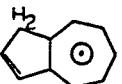
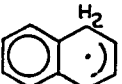
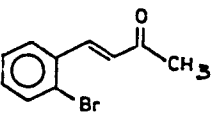
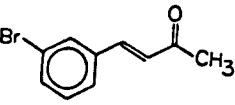
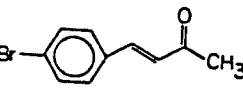
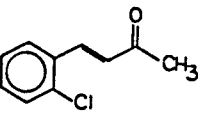
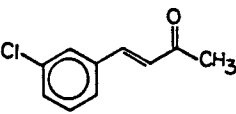
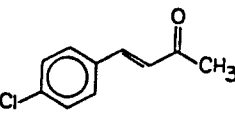
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---------------------------------------|---|--|---|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_9^+$ |  | | 215 | 898 | | | | |
| | | | From proton affinity of azulene (RN 275-51-4). PA = 220. kcal/mol, 921. kJ/mol. | | | | | |
| |  | | 207 | 865 | | | | |
| | | | From proton affinity of naphthalene (RN 91-20-3). PA = 194.7 kcal/mol, 815. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_9\text{BrO}^+$ | | | | | | | | |
| |  | (8.7±0.05) | (202) | (843) | 1 | 4 | 79SCH/GRU | |
| | | IP from 79SCH/GRU, 80GRU/SCH, 81SCH/GRO. | | | | | | |
| |  | (8.9) | (206) | (862) | 1 | 4 | *EST | 65300-30-3 |
| | | IP from 81SCH/GRO. | | | | | | |
| |  | (8.9) | (206) | (863) | 1 | 4 | *EST | 3815-31-4 |
| | | IP from 81SCH/GRO. | | | | | | |
| $\text{C}_{10}\text{H}_9\text{ClO}^+$ | | | | | | | | |
| |  | (8.8) | (190) | (795) | -13 | -54 | *EST | |
| | | IP from 80GRU/SCH, 81SCH/GRO. | | | | | | |
| |  | (8.9) | (192) | (805) | -13 | -54 | *EST | 30626-02-9 |
| | | IP from 81SCH/GRO. | | | | | | |
| |  | (8.7) | (188) | (785) | -13 | -54 | *EST | 30626-03-0 |
| | | IP from 81SCH/GRO. | | | | | | |

Table 1. Positive Ion Table - Continued

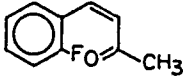
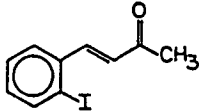
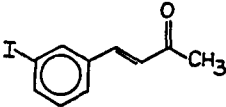
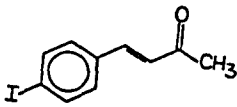
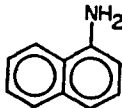
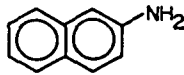
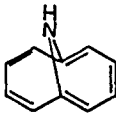
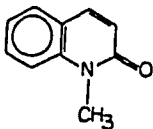
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₉FO⁺ | | | | | | | |
|  | (8.9) IP from 79SCH/GRU, 80GRU/SCH. | (154) | (646) | -51 | -213 | 79SCH/GRU | 2143-80-8 |
| C₁₀H₉IO⁺ | | | | | | | |
|  | (8.6±0.05) IP from 81SCH/GRO. See also: 80GRU/SCH. | (214) | (895) | 16 | 65 | *EST | |
|  | (8.7±0.05) IP from 81SCH/GRO. | (214) | (893) | 13 | 54 | *EST | |
|  | (8.4±0.05) IP from 81SCH/GRO. | (207) | (864) | 13 | 54 | *EST | |
| C₁₀H₉N⁺ | | | | | | | |
|  | (7.1) IP is onset of photoelectron band (83KLA/KOV). | (201) | (843) | 38±2 | 158±7 | 77PED/RYL | 134-32-7 |
|  | 7.10±0.02 See also: 83KLA/KOV. | 196 | 821 | 32±3 | 136±12 | 77PED/RYL | 91-59-8 |
|  | 7.75 IP from 84AND/CER. | (266) | (1115) | 88±2 | 367±7 | 77PED/RYL | 4753-55-3 |
| C₁₀H₉NO⁺ | | | | | | | |
|  | (8.0) IP is onset of photoelectron band (81PFI/GUI). | (174) | (727) | -11 | -45 | *EST | 606-43-9 |

Table 1. Positive Ion Table - Continued

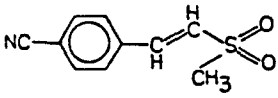
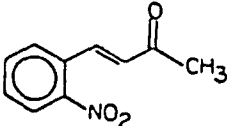
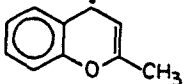
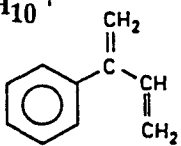
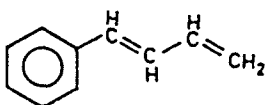
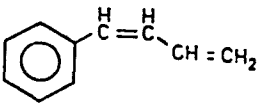
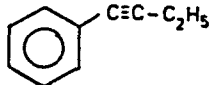
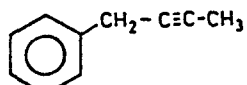
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₉NO₂S⁺  | (9.0) | (203) | (849) | -4 | -19 | *EST | 69957-44-4 |
| IP is onset of photoelectron band (84CAU/FUR). | | | | | | | |
| C₁₀H₉NO₃⁺  | (9.0) | (198) | (827) | -10 | -41 | 79SCH/GRU | 20766-40-9 |
| IP from 80GRU/SCH. | | | | | | | |
| C₁₀H₉O⁺  | | (147) | (617) | | | | 45883-76-9 |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations (79SCH/GRU). | | | | | | | |
| C₁₀H₁₀⁺  | 8.15±0.04 | (237) | (990) | 49 | 204 | *EST | 2288-18-8 |
| IP from 75DER/JOC, 83DAS/GRO. | | | | | | | |
|  | 8.06±0.07 | (235) | (982) | 49 | 204 | *EST | 16939-57-4 |
| IP from 74KOP/SCH, 83DAS/GRO. | | | | | | | |
|  | (8.39) | (243) | (1017) | 50 | 208 | *EST | 31915-94-3 |
|  | 8.35±0.02 | (259) | (1082) | 66 | 276 | *EST | 622-76-4 |
| IP from 82LEV/LIA, 81ELB/LIE. See also: 74KOP/SCH. | | | | | | | |
|  | (8.6) | (260) | (1089) | 62 | 259 | *EST | 33598-22-0 |
| IP from 74KOP/SCH. | | | | | | | |

Table 1. Positive Ion Table - Continued

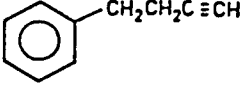


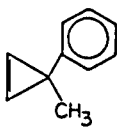
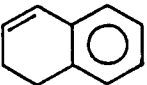
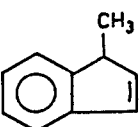
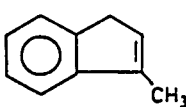
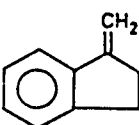
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
|  | (8.5) IP from 74KOP/SCH. | (263) | (1099) | 67 | 279 | *EST | 16520-62-0 |
|  | 8.20±0.02 IP from 75DER/JOC, 82LEV/LIA. See also: 83DAS/GRO. | (249) | (1041) | 60 | 250 | *EST | 3365-26-2 |
|  | (8.42) IP from 83DAS/GRO. | (258) | (1078) | 64 | 266 | *EST | 20211-64-7 |
|  | (8.3) IP from 84BAI/DOM. | (277) | (1160) | 86 | 359 | *EST | 65051-83-4 |
|  | 8.07±0.04 IP from 83DAS/GRO, 74KOP/SCH. | (214) | (897) | 28 | 119 | 77PED/RYL | 447-53-0 |
|  | (8.27) IP from 83DAS/GRO. | (226) | (945) | 35 | 147 | *EST | 767-59-9 |
|  | 8.05 IP from 83DAS/GRO, 74KOP/SCH. | (219) | (916) | 33 | 139 | *EST | 767-60-2 |
|  | (8.00±0.02) See also: 83DAS/GRO. | (220) | (921) | 36 | 149 | *EST | 1194-56-5 |

Table 1. Positive Ion Table - Continued

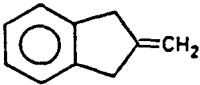
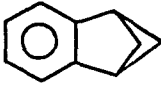
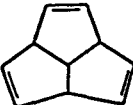

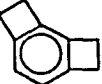

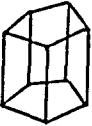
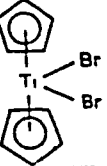
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{10}^+$ | | | | | | | | |
|  | | (8.34) See also: 83DAS/GRO. | (230) | (964) | 38 | 159 | *EST | 68846-65-1 |
|  | | (8.26) IP from 83DAS/GRO. | (254) | (1065) | 64 | 268 | *EST | |
|  | | (8.6) IP is onset of photoelectron band. | (252) | (1054) | 53.5±1 | 224±4 | 86LIE/PAQ | 6053-74-3 |
|  | | (8.17) | (265) | (1107) | 76 | 319 | *EST | 1610-51-1 |
|  | | (8.18) | (266) | (1114) | 78 | 325 | *EST | 58436-35-4 |
|  | | 8.09±0.05 | 266 | 1115 | 80±0.7 | 334±3 | 81MAN/SUN | 1005-51-2 |
|  | | (8.3) IP is onset of photoelectron band (82HON/EAT). | (310) | (1298) | 119±5 | 497±20 | 73ENG/AND | 4572-17-2 |
| $\text{C}_{10}\text{H}_{10}\text{Br}_2\text{Ti}^+$ | | | | | | | | |
|  | | ≤8.8 IP from 82BOH. See also: 82LEV/LIA. | (≤158) | (≤663) | -44 | -186 | *EST | 1293-73-8 |

Table 1. Positive Ion Table - Continued

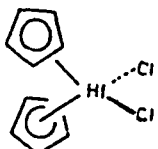
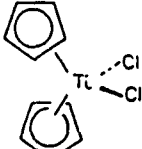
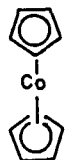
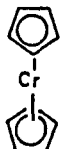
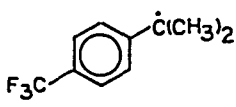
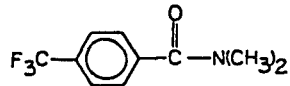
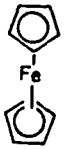
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{10}\text{Cl}_2\text{Hf}^+$  | (8.5) IP is onset of photoelectron band. | (93) | (391) | -103±0.7 | -429±3 | 82PIL/SKI | 12116-66-4 |
| $\text{C}_{10}\text{H}_{10}\text{Cl}_2\text{Ti}^+$  | (8.2) IP is onset of photoelectron band. See also: 82BOH. | (126) | (525) | -64±2 | -266±9 | 82PIL/SKI | 1271-19-8 |
| $\text{C}_{10}\text{H}_{10}\text{Co}^+$  | (5.2) | (193) | (809) | 73±1 | 307±5 | 77PED/RYL | 1277-43-6 |
| $\text{C}_{10}\text{H}_{10}\text{Cr}^+$  | 5.50 | 184 | 772 | 58±1 | 241±5 | 77PED/RYL | 1271-24-5 |
| $\text{C}_{10}\text{H}_{10}\text{F}_3^+$  | | 35 | 146 | | | | |
| | | From proton affinity of 4-CF ₃ C ₆ H ₄ C(CH ₃)=CH ₂ (RN 55186-75-9). PA = 199.6 kcal/mol, 835. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{10}\text{F}_3\text{NO}^+$  | (≤9.38) IP from 85GAL/GER. | (≤31) | (≤128) | -186 | -777 | *EST | 25771-21-5 |
| $\text{C}_{10}\text{H}_{10}\text{Fe}^+$  | 6.747 IP from 82BAR/HEI. See also: 86VON. | 213 | 893 | 58±0.7 | 242±3 | 77PED/RYL | 102-54-5 |

Table 1. Positive Ion Table - Continued

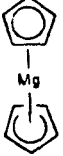
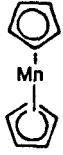
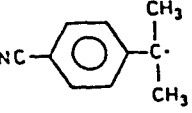
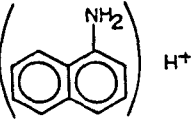
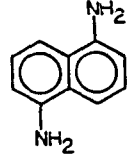
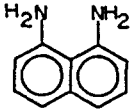
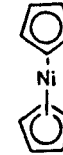
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{10}\text{Mg}^+$  | (8.11) | (218) | (913) | 31±2 | 131±8 | 77PED/RYL | 1284-72-6 |
| $\text{C}_{10}\text{H}_{10}\text{Mn}^+$  | 6.55 | 217 | 909 | 66 | 277 | 82PIL/SKI | 1271-27-8 |
| $\text{C}_{10}\text{H}_{10}\text{N}^+$  | | 219 | 915 | From appearance energy determinations (86ORL/MIS). | | | |
|  | | 187 | 781 | From proton affinity of 1-naphthalenamine (RN 134-32-7). PA = 216.9 kcal/mol, 907.5 kJ/mol. | | | |
| $\text{C}_{10}\text{H}_{10}\text{N}_2^+$  | (6.74±0.02) | (194) | (815) | 39 | 165 | *EST | 2243-62-1 |
|  | (6.65±0.02) | (199) | (835) | 46 | 193 | *EST | 479-27-6 |
| $\text{C}_{10}\text{H}_{10}\text{Ni}^+$  | 6.2 | 228 | 955 | 85±1 | 357±5 | 77PED/RYL | 1271-28-9 |

Table 1. Positive Ion Table - Continued

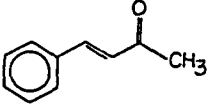
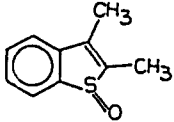
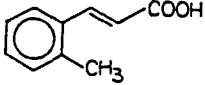
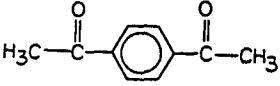
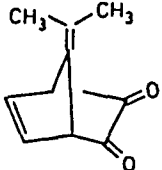
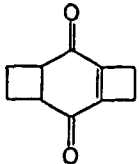
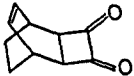
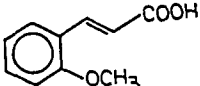
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{10}\text{O}^+$ | | | | | | | |
|  | (8.8±0.05) IP from 79SCH/GRU, 81SCH/GRO, 80GRU/SCH. | (197) | (824) | -6 | -25 | 79SCH/GRU | 122-57-6 |
| $\text{C}_{10}\text{H}_{10}\text{OS}^+$ | | | | | | | |
|  | (≤8.40) IP from 82BEN/DUR. | (≤134) | (≤561) | -60 | -249 | *EST | 70445-88-4 |
| $\text{C}_{10}\text{H}_{10}\text{O}_2^+$ | | | | | | | |
|  | (8.65±0.05) IP from 84SCH. | (134) | (563) | -65 | -272 | 84SCH | 939-57-1 |
|  | (≤9.61) IP from 85GAL/GER. | (≤160) | (≤670) | -61 | -257 | *EST | 1009-61-6 |
|  | (8.0) IP is onset of photoelectron band. | (173) | (726) | -11 | -46 | *EST | 60526-38-7 |
|  | (≤9.25) IP from 85GLE/JAH. | (≤240) | (≤1002) | 26 | 110 | *EST | 94499-49-7 |
|  | (≤9.02) IP from 85ALB/HEL. | (≤157) | (≤656) | -51 | -214 | *EST | 72590-52-4 |
| $\text{C}_{10}\text{H}_{10}\text{O}_3^+$ | | | | | | | |
|  | (8.50±0.05) IP from 84SCH. | (102) | (427) | -94 | -393 | 84SCH | 1011-54-7 |

Table 1. Positive Ion Table - Continued

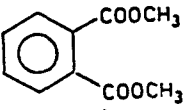
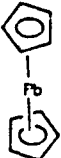
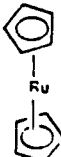

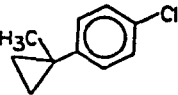
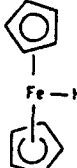
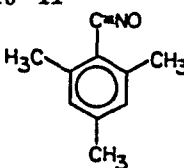
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|---|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{10}\text{O}_4^+$  | (9.64±0.07) | (66) | (276) | -156.2±4 | -654±17 | *EST | 131-11-3 |
| $\text{C}_{10}\text{H}_{10}\text{Pb}^+$  | (≤7.55) IP from 82BAX/COW, 82LEV/LIA. | (≤308) | (≤1288) | 134 | 559 | 85DEW/HOL | 1294-74-2 |
| $\text{C}_{10}\text{H}_{10}\text{Ru}^+$  | (7.1) IP is onset of photoelectron band. | (102) | (425) | -62 | -260 | *EST | 1287-13-4 |
| $\text{C}_{10}\text{H}_{10}\text{V}^+$  | (6.4) IP is onset of photoelectron band. | (196) | (822) | 49±2 | 204±10 | 77PED/RYL | 1277-47-0 |
| $\text{C}_{10}\text{H}_{11}\text{Cl}^+$  | (≤8.67) | (≤221) | (≤925) | 21±1 | 88±4 | *EST | 63340-05-6 |
| $\text{C}_{10}\text{H}_{11}\text{Fe}^+$  | | (214) | (893) | From proton affinity of Iron, bis(η^5 -cyclopentadienyl) (RN 102-54-5). PA = (210) kcal/mol, (879) kJ/mol. | | | |
| $\text{C}_{10}\text{H}_{11}\text{NO}^+$  | ≤8.37 IP is onset of photoelectron band. | (≤238) | (≤998) | 45 | 190 | *EST | 2904-57-6 |

Table 1. Positive Ion Table - Continued

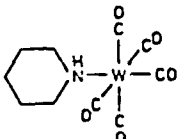
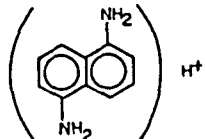
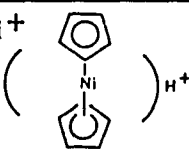
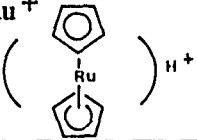
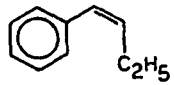
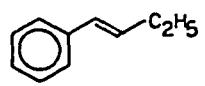
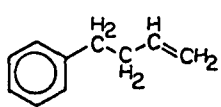
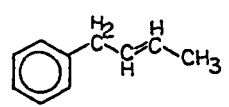
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{11}\text{NO}_5\text{W}^+$  | (7.0) | (-33) | (-140) | -195 | -815 | 84ALT/CON2 | 31082-68-5 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_{10}\text{H}_{11}\text{N}_2^+$  | | 188 | 787 | | | | |
| From proton affinity of 1,8-diaminonaphthalene (RN 479-27-6). PA = 223.8 kcal/mol, 936. kJ/mol. | | | | | | | |
| $\text{C}_{10}\text{H}_{11}\text{Ni}^+$  | | 228 | 954 | | | | |
| From proton affinity of Nickel, bis(η^5 -cyclopentadienyl) (RN 1271-28-9). PA = 223. kcal/mol, 933. kJ/mol. | | | | | | | |
| $\text{C}_{10}\text{H}_{11}\text{Ru}^+$  | | (86) | (358) | | | | |
| From proton affinity of Ruthenium, bis(η^5 -cyclopentadienyl) (RN 1287-13-4). PA = (218) kcal/mol, (912) kJ/mol. | | | | | | | |
| $\text{C}_{10}\text{H}_{12}^+$ | | | | | | | |
|  | (8.15) | (213) | (892) | 25 | 106 | *EST | 1560-09-4 |
| IP from onset of photoelectron band (81KOB/ARA). | | | | | | | |
|  | (8.0) | (208) | (873) | 24 | 101 | *EST | 1005-64-7 |
| IP from onset of photoelectron band (81KOB/ARA). | | | | | | | |
|  | (8.6) | (225) | (943) | 27 | 113 | *EST | 768-56-9 |
| IP from 78FU/DUN. | | | | | | | |
|  | (8.48) | (220) | (918) | 24 | 100 | *EST | 935-00-2 |
| IP from 78FU/DUN. | | | | | | | |

Table 1. Positive Ion Table - Continued

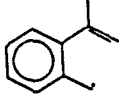
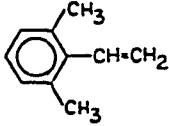
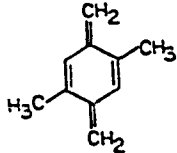
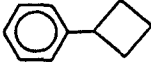
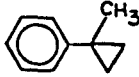
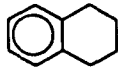
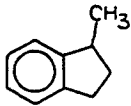
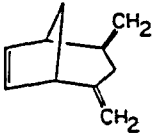
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{12}^+$  | (7.78±0.04) See also: 78FU/DUN. | (199) | (832) | 19 | 81 | 85DAS/GRO | 1587-04-8 |
|  | (8.10±0.02) | (212) | (886) | 25 | 104 | *EST | 2039-90-9 |
|  | (7.4) IP from onset of photoelectron band. See also: 82DEW. | (205) | (856) | 34 | 142 | *EST | 63238-49-3 |
|  | (8.4) IP from onset of photoelectron band. | (226) | (946) | 33 | 136 | *EST | 4392-30-7 |
|  | (8.4) IP from onset of photoelectron band. | (223) | (930) | 29±1 | 120±4 | *EST | 2214-14-4 |
|  | 8.47 See also: 80MAU. | 201 | 841 | 6±0.5 | 24±2 | 77PED/RYL | 119-64-2 |
|  | (8.47) | (205) | (856) | 9 | 39 | 85DAS/GRO | 767-58-8 |
|  | ≤8.98 | ≤268 | ≤1123 | 61 | 257 | 80MAR/HEL | 72569-84-7 |

Table 1. Positive Ion Table - Continued

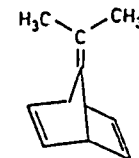
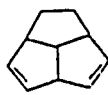
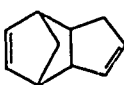
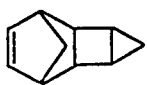

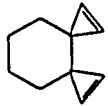
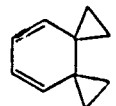
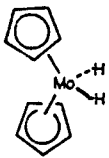
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|-----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{12}^+$ | | | | | | | |
|  | (7.97) | (249) | (1042) | 65 | 273 | *EST | 36456-22-1 |
|  | ≤ 9.00 | (≤ 238) | (≤ 996) | 30.5 ± 1 | 128 ± 4 | 86LIE/PAQ | 31678-74-7 |
|  | (8.79 ± 0.05) | (248) | (1038) | 45 ± 2 | 190 ± 9 | 80ROT/KLA | 77-73-6 |
|  | $\leq 8.83 \pm 0.03$ | ≤ 275 | ≤ 1152 | 72 | 300 | 80ROT/KLA | 6574-77-2 |
|  | 7.33 ± 0.05 | (260) | (1087) | 91 | 380 | *EST | 36262-33-6 |
|  | (8.3) IP is onset of photoelectron band (82SPA/KOR). | (298) | (1246) | 106 | 445 | *EST | 54440-40-3 |
|  | (≤ 7.74) | (≤ 269) | (≤ 1125) | 90 | 378 | *EST | 30353-70-9 |
| $\text{C}_{10}\text{H}_{12}\text{Mo}^+$ | | | | | | | |
|  | ($\leq 6.4 \pm 0.1$) | (≤ 220) | (≤ 920) | 72 ± 1 | 303 ± 6 | 86SIM/BEA | 1291-40-3 |

Table 1. Positive Ion Table - Continued

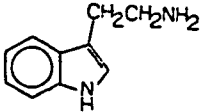
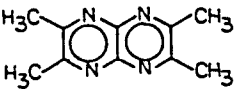
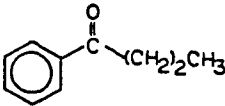
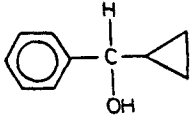
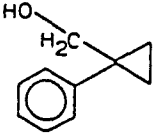
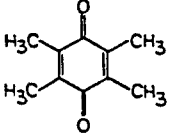
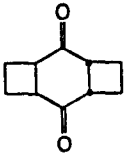
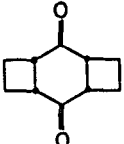
| ION | Ionization potential | | | $\Delta_f H$ (Neutral) | | Neutral reference | CAS registry number | |
|---------------------|---|---|----------------|------------------------|-----------------|-------------------|---------------------|------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | | | kJ/mol |
| $C_{10}H_{12}N_2^+$ |  | (7.7) | (211) | (882) | 33 | 139 | *EST | 61-54-1 |
| | | IP is onset of photoelectron band. | | | | | | |
| $C_{10}H_{12}N_4^+$ |  | (≤ 8.6) | (≤ 249) | (≤ 1040) | 50 | 210 | *EST | 6479-03-4 |
| | | IP from 84GLE/SPA2. | | | | | | |
| $C_{10}H_{12}O^+$ |  | 9.06 \pm 0.02 | 178 | 746 | -30.6 \pm 0.6 | -128.2 \pm 2.4 | 77PED/RYL | 495-40-9 |
| | | IP is average of values from 79MCL/TRA and 81DAL/NIB. | | | | | | |
| |  | (8.31) | (186) | (780) | -5 | -22 | 81DAL/NIB | 1007-03-0 |
| | | IP from 81DAL/NIB. | | | | | | |
| |  | (8.35) | (189) | (790) | -4 | -16 | 81DAL/NIB | 31729-66-5 |
| | | IP from 81DAL/NIB. | | | | | | |
| $C_{10}H_{12}O_2^+$ |  | (9.1) | (150) | (626) | -60 | -252 | *EST | 527-17-3 |
| | | IP from 80BOC/KAI, 82LEV/LIA. | | | | | | |
| |  | (≤ 9.0) | (≤ 180) | (≤ 752) | -28 | -116 | *EST | 87305-43-9 |
| | | IP is onset of photoelectron band (85GLE/JAH). | | | | | | |
| |  | (8.8) | (175) | (733) | -28 | -116 | *EST | 87305-42-8 |
| | | IP from 85GLE/JAH. | | | | | | |

Table 1. Positive Ion Table - Continued

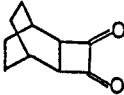
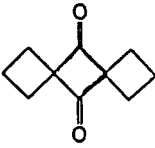
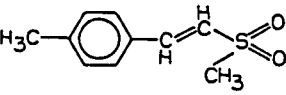
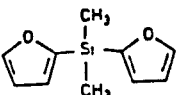
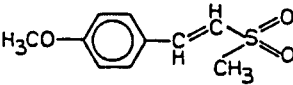
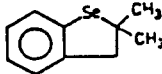
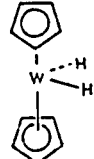
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₂O₂⁺ | | | | | | | |
|  | (≤ 9.06) IP from 85ALB/HEL. | (≤ 130) | (≤ 542) | -79 | -332 | *EST | |
|  | (9.0 ± 0.02) IP from 84OLI/FLE. | (216) | (904) | 9 | 36 | *EST | 4893-00-9 |
| C₁₀H₁₂O₂S⁺ | | | | | | | |
|  | (8.3) IP is onset of photoelectron band (84CAU/FUR). | (146) | (613) | -45 | -188 | *EST | 77355-29-4 |
| C₁₀H₁₂O₂Si⁺ | | | | | | | |
|  | (≤ 8.60) IP from 83ZYK/ERC. | (≤ 151) | (≤ 633) | -47 | -197 | *EST | 1578-44-5 |
| C₁₀H₁₂O₃S⁺ | | | | | | | |
|  | (≤ 8.52) IP from 84CAU/FUR. | (≤ 123) | (≤ 516) | -73 | -306 | *EST | 70784-98-4 |
| C₁₀H₁₂Se⁺ | | | | | | | |
|  | (7.3) IP is onset of photoelectron band (81BAK/ARM). | (144) | (603) | -24 | -101 | *EST | 60096-27-7 |
| C₁₀H₁₂W⁺ | | | | | | | |
|  | (6.35 ± 0.2) | (221) | (924) | 74 ± 1 | 311 ± 5 | 82PIL/SKI | 1271-33-6 |

Table 1. Positive Ion Table - Continued

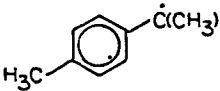
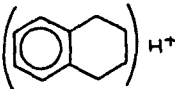
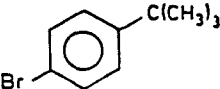
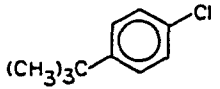
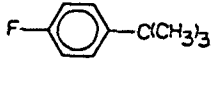
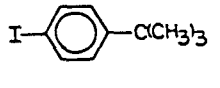

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{13}^+$ | | | | | | | |
|  | | (174) | (727) | | | | |
| | | From proton affinity of 4- $\text{CH}_3\text{C}_6\text{H}_4\text{C}(\text{CH}_3)_2\text{CH}_3$ (RN 1195-32-0). PA = 211.0 kcal/mol, 883. kJ/mol and from appearance energy determination (86ORL/MIS). | | | | | |
|  | | 177 | 740 | | | | |
| | | From proton affinity of 1,2,3,4-tetrahydronaphthalene (RN 119-64-2). PA = 194.7 kcal/mol, 815. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{13}\text{Br}^+$ | | | | | | | |
|  | 8.50±0.02 | (198) | (828) | 2 | 8 | 86ORL/MIS | 3972-65-4 |
| | IP from 86ORL/MIS, 85BAI/MIS. | | | | | | |
| $\text{C}_{10}\text{H}_{13}\text{Cl}^+$ | | | | | | | |
|  | (8.56±0.02) | (184) | (772) | -13 | -54 | 86ORL/MIS | 3972-56-3 |
| | IP from 86ORL/MIS. See also: 85BAI/MIS. | | | | | | |
| $\text{C}_{10}\text{H}_{13}\text{F}^+$ | | | | | | | |
|  | (8.59) | (146) | (609) | -52 | -219 | 86ORL/MIS | 701-30-4 |
| | IP from 85ORL/MIS. | | | | | | |
| $\text{C}_{10}\text{H}_{13}\text{I}^+$ | | | | | | | |
|  | (8.35±0.02) | (206) | (862) | 14 | 57 | 86ORL/MIS | 35779-04-5 |
| | IP from 86ORL/MIS. See also: 85BAI/MIS. | | | | | | |
| $\text{C}_{10}\text{H}_{13}\text{N}^+$ | | | | | | | |
|  | (6.8) | (185) | (773) | 28 | 117 | *EST | 4096-21-3 |
| | IP is onset of photoelectron band (82ROZ/HOU2). | | | | | | |

Table 1. Positive Ion Table - Continued

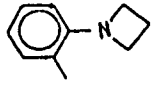
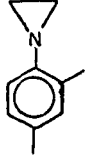
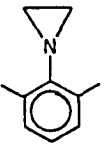
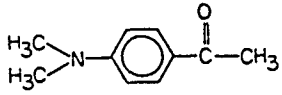
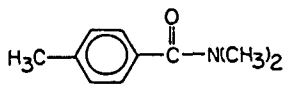
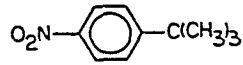
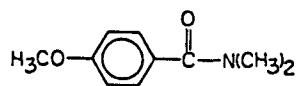
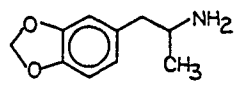
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₃N⁺ | | | | | | | | |
| |  | (7.1) | (222) | (927) | 58 | 242 | *EST | 19198-94-8 |
| | | IP is onset of photoelectron band (82ROZ/HOU2). | | | | | | |
| |  | (≤ 7.80) | (≤ 243) | (≤ 1017) | 63 | 265 | *EST | 78376-89-3 |
| | | IP from 82ROZ/HOU2. | | | | | | |
| |  | (7.6) | (232) | (969) | 56 | 236 | *EST | 78376-90-6 |
| | | IP is onset of photoelectron band (82ROZ/HOU2). See also: 82CRI/LIC. | | | | | | |
| C₁₀H₁₃NO⁺ | | | | | | | | |
| |  | ≤ 7.55 | (≤ 157) | (≤ 655) | -17 | -73 | *EST | 2124-31-4 |
| |  | (≤ 8.90) | (≤ 175) | (≤ 731) | -31 | -128 | *EST | 14062-78-3 |
| | | IP from 85GAL/GER. | | | | | | |
| C₁₀H₁₃NO₂⁺ | | | | | | | | |
| |  | (9.2) | (203) | (850) | -9 | -38 | 8SORL/MIS | 3282-56-2 |
| | | IP is onset of photoelectron band (85BAI/MIS2). See also: 86ORL/MIS. | | | | | | |
| |  | ≤ 8.40 | (≤ 135) | (≤ 564) | -59 | -246 | *EST | 7291-00-1 |
| | | IP from 85GAL/GER. | | | | | | |
| |  | ($\leq 8.01 \pm 0.06$) | (≤ 136) | (≤ 570) | -49 | -203 | *EST | 51497-09-7 |

Table 1. Positive Ion Table - Continued

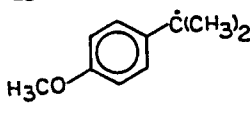
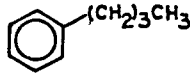
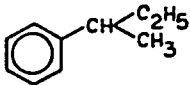
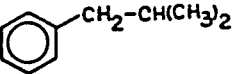
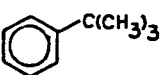
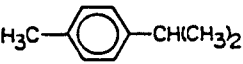
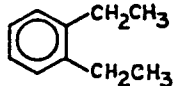
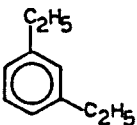
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|---|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{13}\text{O}^+$  | | | 139 | 580 | | | |
| | | | From proton affinity of 4- $\text{CH}_3\text{OC}_6\text{H}_4\text{C}(\text{CH}_3)=\text{CH}_2$ (RN 1712-69-2). PA = 217.4 kcal/mol, 910. kJ/mol. Value from appearance energy determination (86ORL/MIS) = 143 kcal/mol, 598 kJ/mol. | | | | |
| $\text{C}_{10}\text{H}_{14}^+$ | | | | | | | |
|  | 8.69±0.01 | 198 | 827 | -3.1±0.1 | -13.2±0.6 | 77PED/RYL | 104-51-8 |
| | | IP at 298 K from charge transfer equilibrium constant determinations (78LIA/AUS) is 8.71±0.01 eV. | | | | | |
|  | 8.68±0.01 | 196 | 820 | -4.1±0.2 | -17.3±1 | 77PED/RYL | 135-98-8 |
|  | 8.68±0.01 | 195 | 816 | -5.1±0.3 | -21.5±1 | 77PED/RYL | 538-93-2 |
|  | 8.64±0.02 | 194 | 812 | -5.1±0.3 | -21.5±1.2 | 77PED/RYL | 98-06-6 |
| | | IP is average of values from 80VAN (cited in 83BRA/BAE) and from 84HOW/GON. IP at 298 K from charge transfer equilibrium constant determinations (78LIA/AUS) is 8.72±0.01 eV. See also: 85DOM/LAK, 85BAI/MIS. | | | | | |
|  | (8.29) | (184) | (772) | -7 | -28 | *EST | 99-87-6 |
| | IP from 84HOW/GON. | | | | | | |
|  | ≤8.51 | ≤192 | ≤804 | -4±0.2 | -17±1 | 77PED/RYL | 135-01-3 |
|  | (8.49±0.01) | (191) | (798) | -5 | -21 | 77PED/RYL | 141-93-5 |
| | IP is 298 K value from charge transfer equilibrium constant determinations (78LIA/AUS). | | | | | | |

Table 1. Positive Ion Table - Continued

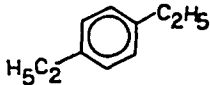
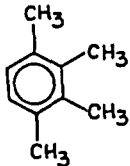
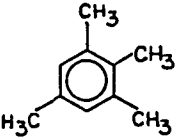
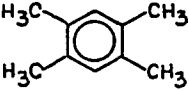
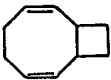

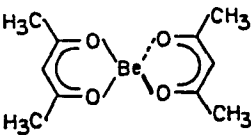
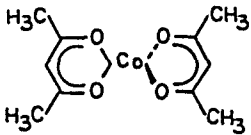
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{14}^+$ | | | | | | | |
|  | 8.40 See also: 80GLE/HOP. | 189 | 790 | -5±0.5 | -20±2 | 77PED/RYL | 105-05-5 |
|  | 8.16±0.02 IP from 82LEV/LIA, 84HOW/GON. | 180 | 754 | -8 | -33 | 75GOO | 488-23-3 |
|  | (8.07) IP from 84HOW/GON. | (176) | (738) | -10 | -41 | 75GOO | 527-53-7 |
|  | 8.04±0.01 See also: 82CAB/COW, 84HOW/GON. | 174 | 731 | -11 | -45 | 75GOO | 95-93-2 |
|  | (8.7) IP from 81BIS/GLE. | (248) | (1036) | 47 | 197 | *EST | 77614-69-8 |
|  | (≤8.48) | (≤259) | (≤1085) | 64 | 267 | *EST | 53143-76-3 |
| $\text{C}_{10}\text{H}_{14}\text{BeO}_4^+$ | | | | | | | |
|  | (8.1) IP is onset of photoelectron band. | (-90) | (-376) | -277±1 | -1158±4 | 80TEL/RAB | 10210-64-7 |
| $\text{C}_{10}\text{H}_{14}\text{CoO}_4^+$ | | | | | | | |
|  | 7.6 IP is onset of photoelectron band (82LEV/LIA, 83KIT/MOR). | (-17) | (-70) | -192±0.5 | -803±2 | 83KAK/GIE | 14024-48-7 |

Table 1. Positive Ion Table - Continued

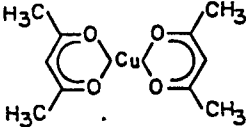
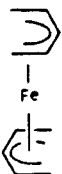
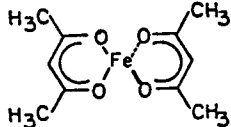
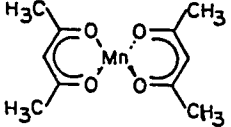
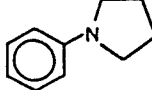
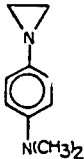
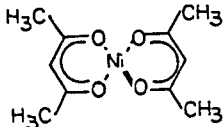
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|--------|------------------------------|--------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{14}\text{CuO}_4^+$ |  | (7.2) | (5) | (20) | -161±0.5 | -675±2 | 83KAK/GIE | 13395-16-9 |
| | | IP is onset of photoelectron band (83KIT/MOR). | | | | | | |
| $\text{C}_{10}\text{H}_{14}\text{Fe}^+$ |  | (6.6) | (192) | (803) | 40 | 166 | *EST | 74910-62-6 |
| | | IP is onset of photoelectron band (84GLE/BOH). | | | | | | |
| $\text{C}_{10}\text{H}_{14}\text{FeO}_4^+$ |  | (7.50±0.04) | (-25) | (-105) | -198±0.5 | -829±2 | 83KAK/GIE | 14024-17-0 |
| $\text{C}_{10}\text{H}_{14}\text{MnO}_4^+$ |  | (8.34±0.05) | (-37) | (-154) | -229±1 | -959±4 | 83KAK/GIE | 14024-58-9 |
| $\text{C}_{10}\text{H}_{14}\text{N}^+$ |  | | (171) | (716) | | | | |
| | | From proton affinity of N-phenylpyrrolidine (RN 4096-21-3). PA = 224.7 kcal/mol, 940. kJ/mol. | | | | | | |
| $\text{C}_{10}\text{H}_{14}\text{N}_2^+$ |  | (≤7.1) | (≤230) | (≤965) | 67 | 280 | *EST | 82027-08-5 |
| | | IP from 82CRI/LIC. | | | | | | |
| $\text{C}_{10}\text{H}_{14}\text{NiO}_4^+$ |  | (7.1) | (-31) | (-130) | -195±0.5 | -815±2 | 83KAK/GIE | 3264-82-2 |
| | | IP is onset of photoelectron band. See also: 83KIT/MOR. | | | | | | |

Table 1. Positive Ion Table - Continued

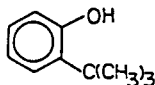
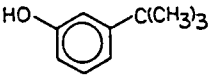
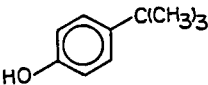
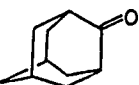
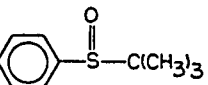
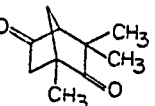
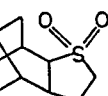
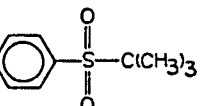
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_{10}H_{14}O^+$ | | | | | | | |
|  | (7.9) | (138) | (576) | -44 | -186 | *EST | 88-18-6 |
| IP is onset of photoelectron band (85BAI/MIS2). See also: 83CET/LAP. | | | | | | | |
|  | (≤ 8.40) | (≤ 145) | (≤ 608) | -48 | -202 | *EST | 585-34-2 |
| IP from 83CET/LAP. | | | | | | | |
|  | (7.8) | (132) | (551) | -48 | -202 | *EST | 98-54-4 |
| IP is onset of photoelectron band (85BAI/MIS). See also: 83CET/LAP. | | | | | | | |
|  | 8.62 | (144) | (601) | -55 \pm 1 | -231 \pm 5 | 78ARO/STE | 700-58-3 |
| IP is onset of photoelectron band. | | | | | | | |
| $C_{10}H_{14}OS^+$ | | | | | | | |
|  | (≤ 8.50) | (≤ 171) | (≤ 717) | -25 | -103 | *EST | 4170-71-2 |
| IP from 81MOH/JIA. | | | | | | | |
| $C_{10}H_{14}O_2^+$ | | | | | | | |
|  | (9.11) | (119) | (497) | -91 | -382 | *EST | 31211-08-2 |
| IP is onset of photoelectron band (80FRO/WES). | | | | | | | |
| $C_{10}H_{14}O_2S^+$ | | | | | | | |
|  | (9.15) | (165) | (690) | -46 | -192 | *EST | |
| IP is onset of photoelectron band (84AIT/GOS). | | | | | | | |
|  | (≤ 9.7) | (145) | (607) | -78 | -328 | *EST | 4170-72-3 |
| IP from 81MOH/JIA. | | | | | | | |

Table 1. Positive Ion Table - Continued

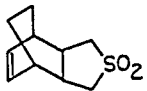
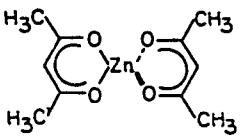
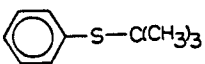
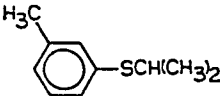
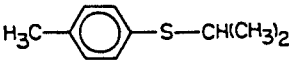
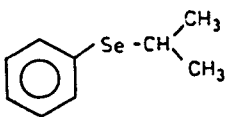
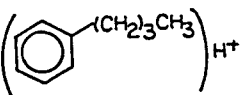
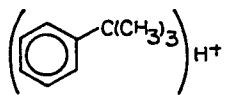
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|--|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{14}\text{O}_2\text{S}^+$  | (9.2) IP is onset of photoelectron band (84AIT/GOS). | (166) | (696) | -46 | -192 | *EST | |
| $\text{C}_{10}\text{H}_{14}\text{O}_4\text{Zn}^+$  | 7.8 IP is onset of photoelectron band (83KIT/MOR, 82LEV/LIA). | (-26) | (-107) | -206±2 | -860±10 | 83KAK/GIE | 14024-63-6 |
| $\text{C}_{10}\text{H}_{14}\text{S}^+$  | 8.39±0.05 | (197) | (825) | 4 | 15 | *EST | 3019-19-0 |
|  | (≤8.38) | (≤198) | (≤828) | 5 | 19 | *EST | 14905-80-7 |
|  | (8.5) IP is onset of photoelectron band. | (201) | (839) | 5 | 19 | *EST | 14905-81-8 |
| $\text{C}_{10}\text{H}_{14}\text{Se}^+$  | (7.2) IP is onset of photoelectron band (81BAK/ARM). | (184) | (772) | 18 | 77 | *EST | 78805-16-0 |
| $\text{C}_{10}\text{H}_{15}^+$  | | 170 | 713 | From proton affinity of n-C ₄ H ₉ C ₆ H ₅ (RN 104-51-8). PA = 192.1 kcal/mol, 804. kJ/mol. | | | |
|  | | 167 | 700 | From proton affinity of tert-C ₄ H ₉ C ₆ H ₅ (RN 98-06-6). PA = 193.0 kcal/mol, 807. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued




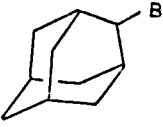
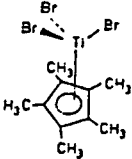
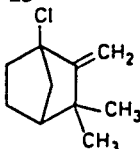
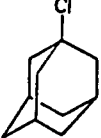
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{15}^+$ |  | (6.21) | (159) | (665) | 16 | 66 | 86KRU/BEA | 19740-18-2 |
| | | $\Delta_f H(\text{Ion})$ from chloride and hydride transfer equilibrium constant determinations (85SHA/SHA, 86KRU/BEA); $\Delta_f H$ (Adamantyl chloride) estimated as -43 kcal/mol, -180 kJ/mol. | | | | | | |
| |  | (6.73) | (168) | (704) | 13 | 54 | 86KRU/BEA | |
| | | IP from 86KRU/BEA. | | | | | | |
| $\text{C}_{10}\text{H}_{15}\text{Br}^+$ |  | 9.30±0.06 | (183) | (766) | -31 | -131 | *EST | 768-90-1 |
| | | IP from 84ABE/DEL, 82LEV/LIA. | | | | | | |
| |  | (9.31±0.05) | (185) | (772) | -30 | -126 | *EST | 7314-85-4 |
| $\text{C}_{10}\text{H}_{15}\text{Br}_3\text{Ti}^+$ |  | (8.0) | (30) | (126) | -154 | -646 | *EST | 33151-84-7 |
| | | IP is onset of photoelectron band (84TER/LOU). | | | | | | |
| $\text{C}_{10}\text{H}_{15}\text{Cl}^+$ |  | (≤9.11) | (≤193) | (≤809) | -17 | -70 | *EST | 4017-64-5 |
| | | IP from 81NES/BAI. | | | | | | |
| |  | (9.30) | (171) | (717) | -43 | -180 | *EST | 935-56-8 |

Table 1. Positive Ion Table - Continued

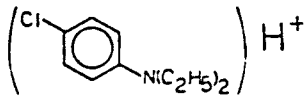
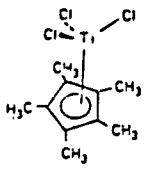
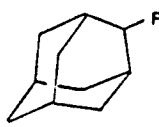

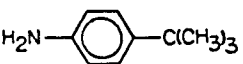
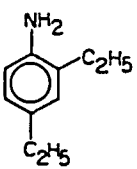
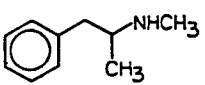
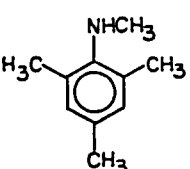
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{15}\text{ClN}^+$  | | 142 | 594 | | | | |
| | | From proton affinity of 4-ClC ₆ H ₄ N(C ₂ H ₅) ₂ (RN 2873-89-4). PA = 225.6 kcal/mol, 944. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{15}\text{Cl}_3\text{Ti}^+$  | (8.1) | (7) | (30) | -179 | -751 | *EST | 12129-06-5 |
| | | IP is onset of photoelectron band (84TER/LOU). | | | | | |
| $\text{C}_{10}\text{H}_{15}\text{F}^+$  | (9.46) | (141) | (592) | -77 | -321 | *EST | 16668-83-0 |
| $\text{C}_{10}\text{H}_{15}\text{I}^+$  | (8.6) | (182) | (760) | -17 | -70 | *EST | 768-93-4 |
| | | IP is onset of photoelectron band (84ABE/DEL). | | | | | |
| $\text{C}_{10}\text{H}_{15}\text{N}^+$  | (7.35±0.02) | (165) | (691) | -4.5 | -19 | 85ORL/MIS | 769-92-6 |
| | | IP from 85ORL/MIS. See also: 85BAI/MIS2. | | | | | |
|  | (≤7.77) | (≤176) | (≤736) | -3 | -14 | *EST | 579-66-8 |
| | | IP from 82ROZ/HOU2. | | | | | |
|  | (≤8.60±0.20) | (≤199) | (≤832) | 0.5 | 2 | *EST | 7632-10-2 |
|  | (7.22) | (171) | (717) | 5 | 20 | *EST | 13021-14-2 |

Table 1. Positive Ion Table - Continued

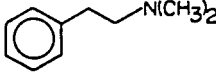
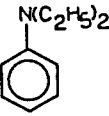
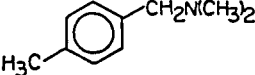
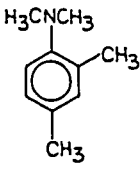
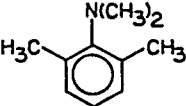
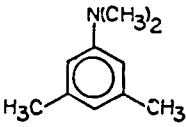
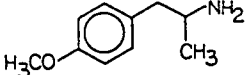
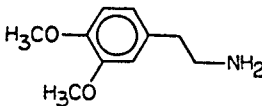
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₅N⁺ | | | | | | | |
|  | (7.70±0.05) See also: 81LOG/TAK. | (193) | (807) | 15 | 64 | *EST | 1126-71-2 |
|  | 6.98±0.02 IP from charge transfer equilibrium constant determinations; reference standard: IP (C ₆ H ₅ N(CH ₃) ₂ = 7.12eV) (84MAU/NEL, 8SLIA/JAC). | (171) | (714) | 9.5 | 40 | 69BEN/CRU | 91-66-7 |
|  | (7.61) See also: 81LOG/TAK. | (187) | (784) | 12 | 49 | *EST | 4052-88-4 |
|  | (≤7.79) IP from 82ROZ/HOU2. | (≤196) | (≤820) | 16 | 68 | *EST | 769-53-9 |
|  | (7.30±0.02) See also: 82ROZ/HOU2. | (190) | (797) | 22 | 93 | *EST | 769-06-2 |
|  | (6.95) IP from charge transfer equilibrium constant determinations; reference standard: IP (C ₆ H ₅ N(CH ₃) ₂ = 7.12 eV) (8SLIA/JAC). | (168) | (706) | 8 | 35 | *EST | 4913-13-7 |
| C₁₀H₁₅NO⁺ | | | | | | | |
|  | (≤8.16±0.06) | (≤184) | (≤768) | -5 | -19 | *EST | 23239-32-9 |
| C₁₀H₁₅NO₂⁺ | | | | | | | |
|  | 7.4 IP is onset of photoelectron band (81DOM/EAT, 82LEV/LIA). | (113) | (473) | -58 | -241 | *EST | 120-20-7 |

Table 1. Positive Ion Table - Continued

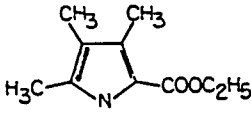
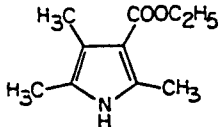
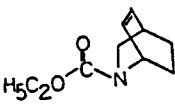
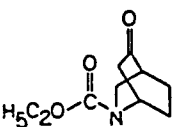
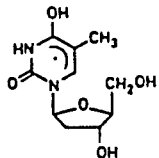
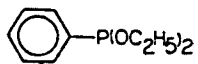
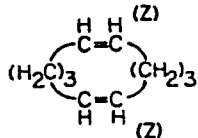
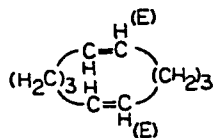
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|---|--------------------------|----------------|--|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_{10}H_{15}NO_2^+$ | | | | | | | | |
|  | | (≤ 7.71) IP from 81CAU/GIA. | (≤ 74) | (≤ 311) | -103 | -433 | *EST | 2199-46-4 |
|  | | (7.5) IP is onset of photoelectron band (81CAU/GIA). | (69) | (290) | -103 | -433 | *EST | 55770-78-0 |
|  | | (8.0) IP is onset of photoelectron band (81CAR/GAN). | (114) | (479) | -70 ⁻ | -293 | *EST | 3693-69-4 |
| $C_{10}H_{15}NO_3^+$ | | | | | | | | |
|  | | (8.5) IP is onset of photoelectron band (81CAR/GAN). | (53) | (220) | -143 | -600 | *EST | 37778-51-1 |
| $C_{10}H_{15}N_2O_5^+$ | | | | | | | | |
|  | | | (-72) | (-301) | From proton affinity of thymidine (RN 50-89-5). PA = (208) kcal/mol, (870) kJ/mol. | | | |
| $C_{10}H_{15}O_2P^+$ | | | | | | | | |
|  | | (8.2) IP is onset of photoelectron band (81ARS/ZVE, 81ZVE/VIL2). | (87) | (362) | -103 | -429 | *EST | 1638-86-4 |
| $C_{10}H_{16}^+$ | | | | | | | | |
|  | | (≤ 8.68) | (≤ 194) | (≤ 813) | -6 | -24 | 76JEN | 1124-79-4 |
|  | | (≤ 8.05) | (≤ 184) | (≤ 769) | -2 | -8 | 76JEN | 15840-81-0 |

Table 1. Positive Ion Table - Continued

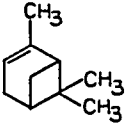
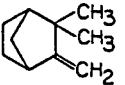

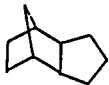
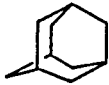

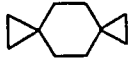

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{16}^+$  | (8.07) | (193) | (807) | 7 ± 0.5 | 28 ± 2 | 77PED/RYL | 80-56-8 |
|  | (≤ 8.86) IP from 81NES/BAI. | (≤ 198) | (≤ 827) | -7 | -28 | 77KOZ/BYC | 79-92-5 |
|  | (8.5) IP is onset of photoelectron band (82SPA/GLE). | (219) | (915) | 23 | 95 | *EST | 81969-73-5 |
|  | 9.35 ± 0.05 | 201.2 | 841.9 | -14.4 ± 1 | -60.2 ± 3 | 71BOY/SAN | 6004-38-2 |
|  | 9.24 ± 0.06 | 181 | 759 | -31.8 ± 0.3 | -132.7 ± 1.3 | 75CLA/KNO | 281-23-2 |
|  | (8.7) IP is onset of photoelectron band. | (209) | (875) | 9 | 36 | *EST | 53764-10-6 |
|  | (≤ 9.17) | (≤ 211) | (≤ 882) | -0.7 | -3 | *EST | 24518-94-3 |
|  | (8.5) IP is onset of photoelectron band (82SPA/GLE, 82LEV/LIA). | (231) | (967) | 35 | 147 | *EST | 24029-74-1 |

Table 1. Positive Ion Table - Continued

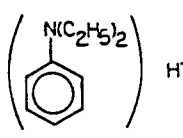
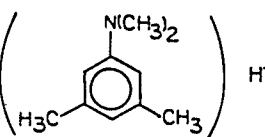
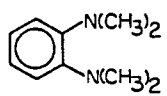
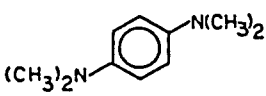
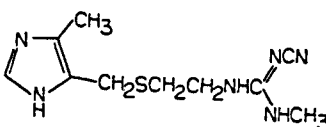
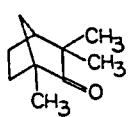
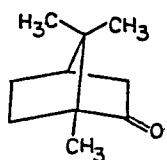
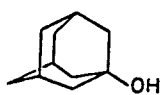
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{16}\text{N}^+$ | | | | | | | |
|  | | 148 | 617 | | | | |
| | | From proton affinity of $\text{C}_6\text{H}_5\text{N}(\text{C}_2\text{H}_5)_2$ (RN 91-66-7). PA = 227.6 kcal/mol, 952. kJ/mol. | | | | | |
|  | | 147 | 615 | | | | |
| | | From proton affinity of 3,5-(CH_3) $_2\text{C}_6\text{H}_3\text{N}(\text{CH}_3)_2$ (RN 4913-13-7). PA = 227.0 kcal/mol, 950. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{16}\text{N}_2^+$ | | | | | | | |
|  | (7.1) | (200) | (836) | 36 | 151 | *EST | 704-01-8 |
| | | IP is onset of photoelectron band (81NEL/GRE). | | | | | |
|  | 6.20±0.05 | 164 | 686 | 21 | 88 | 83MET/ARA | 100-22-1 |
| $\text{C}_{10}\text{H}_{16}\text{N}_6\text{S}^+$ | | | | | | | |
|  | (7.7) | (249) | (1042) | 72 | 300 | *EST | 51481-61-9 |
| | | IP is onset of photoelectron band (80KLA/BUT). | | | | | |
| $\text{C}_{10}\text{H}_{16}\text{O}^+$ | | | | | | | |
|  | (8.5) | (125) | (523) | -71 | -297 | *EST | 1195-79-5 |
| | | IP is onset of photoelectron band (80FRO/WES). | | | | | |
|  | (8.76±0.03) | (138) | (578) | -64±0.7 | -267±3 | 77STE | 76-22-2 |
|  | (9.09±0.05) | (136) | (566) | -74±0.7 | -311±3 | 78ARO/STE | 768-95-6 |

Table 1. Positive Ion Table - Continued

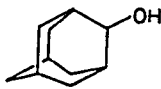
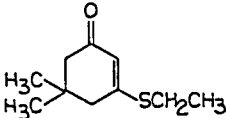
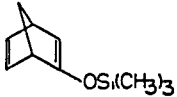
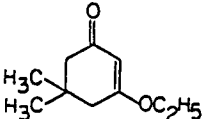
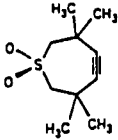
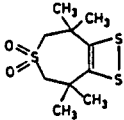
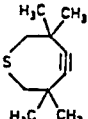
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{16}\text{O}^+$ | | | | | | | |
|  | (9.09±0.07) | (139) | (578) | -71±1 | -299±5 | 78ARO/STE | 700-57-2 |
| $\text{C}_{10}\text{H}_{16}\text{OS}^+$ | | | | | | | |
|  | (8.2) IP is onset of photoelectron band (82PFI/GER). | (147) | (613) | -43 | -178 | *EST | 52735-49-6 |
| $\text{C}_{10}\text{H}_{16}\text{OSi}^+$ | | | | | | | |
|  | (≤8.06) IP from 83HOU/RON. | (≤136) | (≤571) | -49 | -207 | *EST | 68364-22-7 |
| $\text{C}_{10}\text{H}_{16}\text{O}_2^+$ | | | | | | | |
|  | (≤8.87) IP from 82PFI/GER. | (≤117) | (≤489) | -88 | -367 | *EST | 6267-39-6 |
| $\text{C}_{10}\text{H}_{16}\text{O}_2\text{S}^+$ | | | | | | | |
|  | ≤9.75 IP from 83JIA/MOH. | (≤185) | (≤773) | -40 | -168 | *EST | |
| $\text{C}_{10}\text{H}_{16}\text{O}_2\text{S}_3^+$ | | | | | | | |
|  | ≤8.55 IP from 83JIA/MOH. | (≤152) | (≤637) | -45 | -188 | *EST | |
| $\text{C}_{10}\text{H}_{16}\text{S}^+$ | | | | | | | |
|  | ≤8.40 IP from 83JIA/MOH. | (≤237) | (≤992) | 43 | 182 | *EST | |

Table 1. Positive Ion Table - Continued

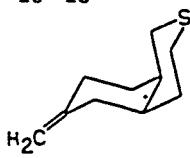

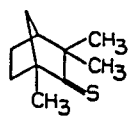
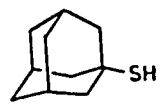
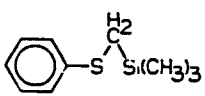
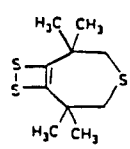
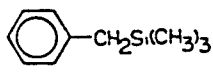
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₆S⁺ | | | | | | | |
|  | (8.22) IP from 80SAR/WOR. | (187) | (782) | -3 | -11 | *EST | 77471-74-0 |
|  | (8.26) IP from 80SAR/WOR. | (185) | (773) | -6 | -24 | *EST | 77471-73-9 |
|  | (8.13) IP from 80FRO/WES. See also: 82LEV/LIA. | (167) | (697) | -21 | -87 | *EST | 875-06-9 |
|  | (8.6) IP is onset of photoelectron band. | (158) | (663) | -40 | -167 | *EST | 34301-54-7 |
| C₁₀H₁₆SSi⁺ | | | | | | | |
|  | ($\leq 7.81 \pm 0.05$) | (≤ 121) | (≤ 506) | -59 | -248 | *EST | 17873-08-4 |
| C₁₀H₁₆S₃⁺ | | | | | | | |
|  | 7.8 IP is onset of photoelectron band (83JIA/MOH). | (219) | (915) | 39 | 162 | *EST | |
| C₁₀H₁₆Si⁺ | | | | | | | |
|  | 8.35 | (164) | (685) | -29 | -121 | *EST | 770-09-2 |

Table 1. Positive Ion Table - Continued

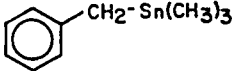
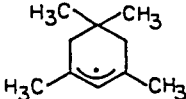
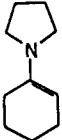
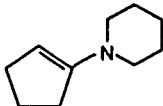
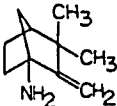
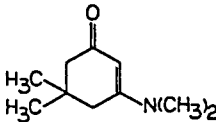
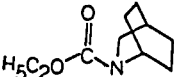
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₆Sn⁺ | | | | | | | |
|  | 8.08±0.05 | 206 | 863 | 20±1 | 83±6 | 77PED/RYL | 4314-94-7 |
| C₁₀H₁₇⁺ | | | | | | | |
|  | | (148) | (618) | | | | |
| | | From proton affinity of 1,5,5-trimethyl-3-methylenecyclohexene (RN 16609-28-2). PA = (216.1) kcal/mol, (904.) kJ/mol. | | | | | |
| C₁₀H₁₇N⁺ | | | | | | | |
|  | 7.10 See also: 81MUL/PRE2. | (165) | (689) | 1 | 4 | *EST | 1125-99-1 |
|  | (7.0) IP is onset of photoelectron band. | (138) | (580) | -23±1 | -95±4 | *EST | 1614-92-2 |
|  | (≤8.67) IP from 81NES/BAI. | (≤198) | (≤828) | -2 | -9 | *EST | 13487-72-4 |
| C₁₀H₁₇NO⁺ | | | | | | | |
|  | (≤7.88) IP from 82PFI/GER. | (≤143) | (≤596) | -39 | -164 | *EST | 31039-88-0 |
| C₁₀H₁₇NO₂⁺ | | | | | | | |
|  | (7.9) IP is onset of photoelectron band (81CAR/GAN). | (63) | (265) | -119 | -497 | *EST | 39926-11-9 |

Table 1. Positive Ion Table - Continued

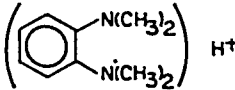
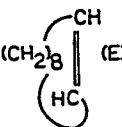
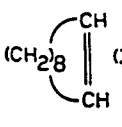
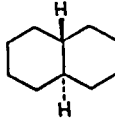
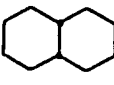
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|---|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{17}\text{N}_2^+$ | | | | | | | |
|  | | 167 | 697 | From proton affinity of 1,2-(N(CH ₃) ₂) ₂ C ₆ H ₄ (RN 704-01-8). PA = 235.2 kcal/mol, 984. kJ/mol. | | | |
| $\text{C}_{10}\text{H}_{18}^+$ | | | | | | | |
| 1-C ₁₀ H ₁₈ | (9.91±0.02) | (239) | (998) | 10±.7 | 42±3 | 79ROG/DAG | 764-93-2 |
| 2-C ₁₀ H ₁₈ | (9.30±0.02) | (220) | (921) | 6±0.7 | 24±3 | 79ROG/DAG | 2384-70-5 |
| 3-C ₁₀ H ₁₈ | 9.19±0.01 | 217 | 909 | 5±0.7 | 22±3 | 79ROG/DAG | 2384-85-2 |
| 4-C ₁₀ H ₁₈ | (9.17±0.02) | (216) | (905) | 5±0.7 | 20±3 | 79ROG/DAG | 2384-86-3 |
| 5-C ₁₀ H ₁₈ | 9.13±0.03 | (216) | (905) | 6 | 24 | *EST | 1942-46-7 |
| (tert-C ₄ H ₉)C=C(tert-C ₄ H ₉) | (9.05±0.01) | (206) | (861) | -3 | -13 | *EST | 17530-24-4 |
| | See also: 85ORL/BOG. | | | | | | |
|  | (8.80) | (199) | (832) | -4 | -17 | 78GRE/LIE | 2198-20-1 |
|  | (8.80) | (196) | (820) | -7 | -29 | 78GRE/LIE | 935-31-9 |
|  | 9.24 | 170 | 709 | -43.5±0.5 | -182.1±2.3 | 77PED/RYL | 493-02-7 |
| | From charge transfer equilibrium constant determinations (82SIE/MAU, 85AUS/LIA). Photoionization onset, 9.32 eV (80MIK/ZAI); onset of photoelectron band, 9.26 eV (77BIE/BUR). | | | | | | |
|  | 9.26 | 173 | 724 | -40.4±0.5 | -169.1±2.3 | 77PED/RYL | 493-01-6 |
| | From charge transfer equilibrium constant determination (85AUS/LIA). Photoionization onset, 9.32 eV (80MIK/ZAI); onset of photoelectron band, 9.26 eV. | | | | | | |

Table 1. Positive Ion Table - Continued

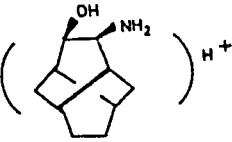
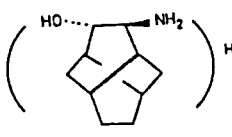
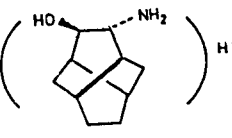
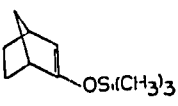
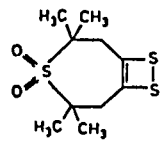
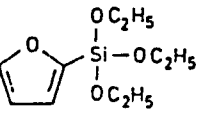
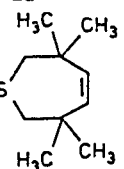
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₈NO⁺ | | | | | | | |
|  | | (95) | (396) | | | | |
| | | From proton affinity of cis-3-amino-2-twistanol. PA = 224.0 kcal/mol, 937. kJ/mol. | | | | | |
|  | | (95) | (398) | | | | |
| | | From proton affinity of trans-3-amino-2-twistanol (isomer 1). PA = 221.5 kcal/mol, 927. kJ/mol. | | | | | |
|  | | (97) | (405) | | | | |
| | | From proton affinity of trans-3-amino-2-twistanol (isomer 2). PA = 220.0 kcal/mol, 920. kJ/mol. | | | | | |
| C₁₀H₁₈OSi⁺ | | | | | | | |
|  | (≤8.09) | (≤101) | (≤424) | -85 | -357 | *EST | 57722-40-4 |
| | IP from 83HOU/RON. | | | | | | |
| C₁₀H₁₈O₂S₃⁺ | | | | | | | |
|  | (≤9.55) | (≤132) | (≤552) | -88 | -369 | *EST | |
| | IP from 83JIA/MOH. | | | | | | |
| C₁₀H₁₈O₄Si⁺ | | | | | | | |
|  | (8.0) | (-26) | (-110) | -211 | -882 | *EST | 55811-52-4 |
| | IP is onset of photoelectron band (83ZYK/ERC). | | | | | | |
| C₁₀H₁₈S⁺ | | | | | | | |
|  | ≤8.35 | (≤188) | (≤787) | -5 | -19 | *EST | |
| | IP from 83JIA/MOH. | | | | | | |

Table 1. Positive Ion Table - Continued

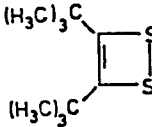
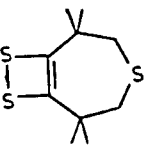
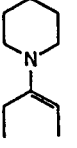

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₀H₁₈S₂⁺  | 7.65 IP is onset of photoelectron band (83JIA/MOH). | (204) | (853) | 27 | 115 | *EST | |
| C₁₀H₁₈S₃⁺  | 7.8 IP is onset of photoelectron band (83JIA/MOH). | 172 | 719 | -8 | -34 | *EST | |
| C₁₀H₁₉N⁺  | (≤ 7.61) IP from 81MUL/PRE2. | (≤ 161) | (≤ 673) | -14 | -61 | *EST | 21086-43-1 |
|  | (6.94 \pm 0.09) IP is onset of photoelectron band. | (166) | (695) | 6 | 25 | *EST | 31023-92-4 |
| C₁₀H₂₀⁺ | | | | | | | |
| 1-C ₁₀ H ₂₀ | 9.42 \pm 0.01 See also: 81HOL/FIN. | 188 | 786 | -29.5 \pm 0.5 | -123.3 \pm 2 | 77PED/RYL | 872-05-9 |
| (Z)-C ₁₀ H ₂₀ | 8.90 \pm 0.01 | (174) | (727) | -32 | -132 | *EST | 20348-51-0 |
| (E)-2-C ₁₀ H ₂₀ | 8.90 \pm 0.01 | (173) | (724) | -32 | -135 | *EST | 20063-97-2 |
| (Z)-3-C ₁₀ H ₂₀ | 8.83 \pm 0.01 | (172) | (721) | -31 | -131 | *EST | 19398-86-8 |
| (E)-3-C ₁₀ H ₂₀ | 8.83 \pm 0.01 | (171) | (717) | -32 | -135 | *EST | 19150-21-1 |
| (Z)-4-C ₁₀ H ₂₀ | 8.78 \pm 0.01 | (171) | (716) | -31 | -131 | *EST | 19398-88-0 |
| (E)-4-C ₁₀ H ₂₀ | 8.78 \pm 0.01 | (170) | (712) | -32 | -135 | *EST | 19398-89-1 |
| (Z)-5-C ₁₀ H ₂₀ | 8.77 \pm 0.01 | (171) | (715) | -31 | -131 | *EST | 7433-78-5 |
| (E)-5-C ₁₀ H ₂₀ | 8.76 \pm 0.01 | (170) | (710) | -32 | -135 | *EST | 7433-56-9 |
| (tert-C ₄ H ₉) ₂ C=CH ₂ | (8.79 \pm 0.01) | (164) | (688) | -38 | -161 | *EST | 5857-68-1 |
| n-C ₅ H ₁₁ C(CH ₃)=C(CH ₃) ₂ | (8.13 \pm 0.01) | (151) | (633) | -36 | -152 | *EST | 19781-18-1 |

Table 1. Positive Ion Table - Continued

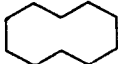
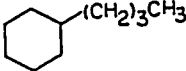
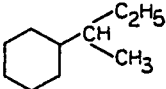
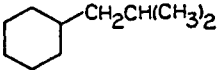
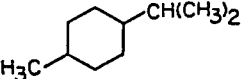
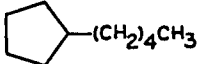
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{20}^+$ $\text{C}_4\text{H}_9\text{C}(\text{C}_2\text{H}_5) = \text{C}(\text{CH}_3)_2$ | (8.10±0.01) | (151) | (630) | -36 | -151 | *EST | 19780-61-1 |
| (tert-C ₄ H ₉)CH ₂ C(CH ₃) = C(CH ₃) ₂ | (8.10±0.01) | (146) | (610) | -41 | -172 | *EST | 33175-59-6 |
| (Z)-(t-C ₄ H ₉)CH = CH(t-C ₄ H ₉) | 8.69±0.01 | (171) | (717) | -29±0.7 | -121±3 | *EST | 692-47-7 |
| (E)-(t-C ₄ H ₉)CH = CH(t-C ₄ H ₉) | 8.74±0.01 | 162 | 677 | -40±0.7 | -166±3 | 79FUC/PEA | 692-48-8 |
|  | (9.5) | (182) | (762) | -36.9±0.4 | -154.3±1.5 | 77PED/RYL | 293-96-9 |
| | IP is onset of photoelectron band (77BIE/BUR). | | | | | | |
|  | 9.41 | 166 | 695 | -50.9±0.2 | -213.0±1 | 77PED/RYL | 1678-93-9 |
| | From charge transfer equilibrium constant determinations (82SIE/MAU); reference standard, fluorobenzenes. Threshold determinations give IP at 0 K of 9.51 eV. | | | | | | |
|  | 9.23 | (164) | (686) | -49 | -205 | *EST | 7058-01-7 |
| | From charge transfer equilibrium constant determinations (82SIE/MAU); reference standard, fluorobenzenes. Threshold determinations give IP at 0 K of 9.51 eV. | | | | | | |
|  | (9.54±0.03) | (171) | (716) | -49 | -204 | *EST | 1678-98-4 |
|  | 9.32 | 160 | 668 | -55±0.7 | -231±3 | 77PED/RYL | 99-82-1 |
| | From charge transfer equilibrium constant determinations (82SIE/MAU); reference standard fluorobenzenes. | | | | | | |
|  | (9.91±0.05) | (184) | (767) | -45 | -189 | 71ASTM | 3741-00-2 |

Table 1. Positive Ion Table - Continued



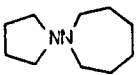
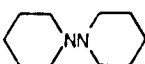
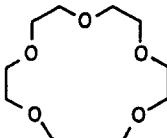
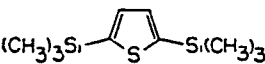
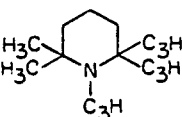
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number | |
|---|----------------------------|--|--------|------------------------------|----------------|----------------------|------------------------|------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | | |
| $\text{C}_{10}\text{H}_{20}\text{N}^+$ | | (142) | (592) | | | | | |
|  | | | | | | | | |
| | | From proton affinity of 1-azabicyclo[3.3.3]undecane (RN 31023-92-4). PA = 230.1 kcal/mol, 963. kJ/mol. | | | | | | |
| $\text{C}_{10}\text{H}_{20}\text{NO}^+$ | | 67 | 280 | | | | | |
|  | | | | | | | | |
| | | From proton affinity of 4-aminodecahydro-3-naphthalenol. PA = 222.1 kcal/mol, 929. kJ/mol. | | | | | | |
| $\text{C}_{10}\text{H}_{20}\text{N}_2^+$ | | (7.60) | (195) | (814) | 19 | 81 | *EST | 60678-75-3 |
|  | | | | | | | | |
| | | Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL. | | | | | | |
|  | | (7.89) | (189) | (791) | 7 | 30 | *EST | 6130-94-5 |
| | | See also: 84NEL. | | | | | | |
| $\text{C}_{10}\text{H}_{20}\text{O}_5^+$ | | (8.9) | (14) | (60) | -191 ± 0.5 | -799 ± 2 | 82BYS/MAN | 33100-27-5 |
|  | | | | | | | | |
| | | IP is onset of photoelectron band. See also: 83BAK/ARM. | | | | | | |
| $\text{C}_{10}\text{H}_{20}\text{SSi}_2^+$ | | (7.8) | (126) | (526) | -54 | -227 | *EST | 17906-71-7 |
|  | | | | | | | | |
| | | IP is onset of photoelectron band (83VES/HAR). | | | | | | |
| $\text{C}_{10}\text{H}_{21}\text{N}^+$ | | (7.23) | (127) | (530) | -40 | -167 | *EST | 79-55-0 |
|  | | | | | | | | |
| | | IP from 82ROZ/HOU. | | | | | | |

Table 1. Positive Ion Table - Continued

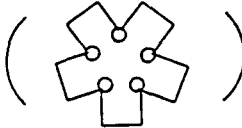
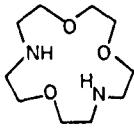
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{21}\text{O}_5^+$  H^+ | | -49 | -205 | | | | |
| | | From proton affinity of 1,4,7,10,13-pentaoxacyclopentadecane (15-Crown-5) (RN 33100-27-5). PA = 223.6 kcal/mol, 936. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{22}^+$ n- $\text{C}_{10}\text{H}_{22}$ | 9.65 | 163 | 682 | -59.6±0.2 | -249.5±0.9 | 77PED/RYL | 124-18-5 |
| | | 176 | 738 | -46.1±0.2 | -192.7±0.9 | | |
| | | From charge transfer equilibrium constants (81MAU/SIE). Reference standards, fluorobenzenes. | | | | | |
| $\text{C}_{10}\text{H}_{22}\text{N}_2\text{O}_3^+$  | (≤8.4) | (≤88) | (≤369) | -105 | -441 | *EST | 31249-95-3 |
| | | IP from 83BAK/ARM. | | | | | |
| $\text{C}_{10}\text{H}_{23}\text{N}^+$ n- $\text{C}_{10}\text{H}_{21}\text{NH}_2$ | (8.63±0.05) | (148) | (619) | -51 | -214 | *EST | 2016-57-1 |
| | | See also: 79AUE/BOW. | | | | | |
| $\text{C}_{10}\text{H}_{23}\text{O}^+$ (n- C_5H_{11}) $_2\text{OH}$ | | 72 | 304 | | | | |
| | | From proton affinity of (n- C_5H_{11}) $_2\text{O}$ (RN 693-65-2) (86SAN/BAL, 85HOU/ROL). PA = 203.5 kcal/mol, 851. kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{24}\text{N}^+$ n- $\text{C}_{10}\text{H}_{21}\text{NH}_3$ | | (94) | (393) | | | | |
| | | From proton affinity of n- $\text{C}_{10}\text{H}_{21}\text{NH}_2$. PA = (220.7) kcal/mol; (923.) kJ/mol. | | | | | |
| $\text{C}_{10}\text{H}_{24}\text{N}_2^+$ (n- C_3H_7) $_2\text{NN}(\text{C}_2\text{H}_5)_2$ | (≤7.87) | (≤191) | (≤797) | 9 | 38 | *EST | 52598-09-1 |
| | | Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL. | | | | | |
| (n- C_4H_9) $_2\text{NN}(\text{CH}_3)_2$ | (≤7.96) | (≤191) | (≤801) | 8 | 33 | *EST | 60678-67-3 |
| | | Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL. | | | | | |

Table 1. Positive Ion Table - Continued

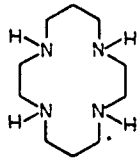
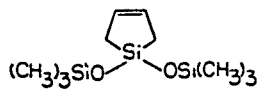
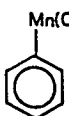
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|------------|--|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{10}\text{H}_{24}\text{N}_4^+$  | (7.7) IP is onset of photoelectron band (83BAK/ARM). | (182) | (761) | 4.3±0.8 | 18.0±3.3 | 83CLA/COR | 295-37-4 |
| $\text{C}_{10}\text{H}_{24}\text{O}_2\text{Si}_3^+$  | (≤9.36) IP from 81KHV/ZYK. | (≤-70) | (≤-293) | -286 | -1196 | *EST | 76795-95-4 |
| $\text{C}_{10}\text{H}_{25}\text{N}_2^+$ (CH ₃) ₂ NH(CH ₂) ₆ N(CH ₃) ₂ | | | 106 444 | | | | |
| | | | | From proton affinity of (CH ₃) ₂ N(CH ₂) ₆ N(CH ₃) ₂ (RN 111-18-2). PA = 245. kcal/mol, 1023. kJ/mol. | | | |
| $\text{C}_{10}\text{H}_{30}\text{N}_5\text{Ta}^+$ Ta(N(CH ₃) ₂) ₅ | (6.5) IP is onset of photoelectron band. | (93) | (390) | -57±4 | -237±15 | 82TN270 | |
| $\text{C}_{10}\text{H}_{30}\text{O}_3\text{Si}_4^+$ [(CH ₃) ₃ SiOSi(CH ₃) ₂] ₂ O | (≤10.24) IP from 82ERM/KIR. | (≤-226) | (≤-947) | -462±5 | -1935±23 | 77PED/RYL | 141-62-8 |
| $\text{C}_{10}\text{H}_{30}\text{Si}_4^+$ n-Si ₄ (CH ₃) ₁₀ | 7.29±0.01 | 32 | 135 | -136±6 | -568±24 | 77PED/RYL | 865-76-9 |
| $\text{C}_{10}\text{MnO}_{10}\text{Re}^+$ MnRe(CO) ₁₀ | 8.22±0.01 | (-184) | (-769) | -373 | -1562 | *EST | 14693-30-2 |
| $\text{C}_{10}\text{Mn}_2\text{O}_{10}^+$ Mn ₂ (CO) ₁₀ | (7.7) IP is onset of photoelectron band. See also: 81MIC/SVE. | (-201) | (-842) | -379±1 | -1585±5 | 82CON/ZAF | 10170-69-1 |
| $\text{C}_{10}\text{O}_{10}\text{Re}_2^+$ Re ₂ (CO) ₁₀ | (7.8) IP is onset of photoelectron band. See also: 81MIC/SVE. | (-193) | (-806) | -373±3 | -1559±11 | 83ALT/CON | 14285-68-8 |
| $\text{C}_{11}\text{H}_5\text{MnO}_5^+$  | (8.22±0.05) | (49) | (203) | -141±0.2 | -590±1 | 82CON/ZAF | 13985-77-8 |

Table 1. Positive Ion Table - Continued

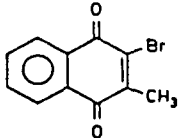
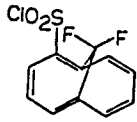
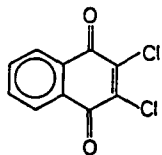
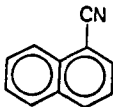
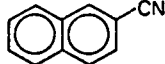
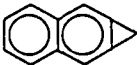
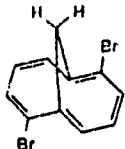
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₇BrO₂⁺ | | | | | | | |
|  | (9.25) IP is onset of photoelectron band (80RED/FRE). | (182) | (760) | -31 | -132 | *EST | 3129-39-3 |
| C₁₁H₇ClF₂O₂S⁺ | | | | | | | |
|  | (≤8.90) IP from 84AND/CER. | (≤35) | (≤146) | -170 | -713 | *EST | |
| C₁₁H₇ClO₂⁺ | | | | | | | |
|  | (9.4) IP is onset of photoelectron band (80RED/FRE). | (180) | (754) | -37 | -153 | *EST | 17015-99-5 |
| C₁₁H₇N⁺ | | | | | | | |
|  | (8.59) IP from 83KLA/KOV. | (244) | (1021) | 46 | 192 | *EST | 86-53-3 |
|  | (8.56) IP is onset of photoelectron band (83KLA/KOV). | (243) | (1016) | 45 | 190 | *EST | 613-46-7 |
| C₁₁H₈⁺ | | | | | | | |
|  | (8.03) IP from 80SCH/SCH. | (289) | (1210) | 104 | 435 | 73BIL/CHO | 286-85-1 |
| C₁₁H₈Br₂⁺ | | | | | | | |
|  | (7.85) IP is onset of photoelectron band (84AND/CER). | (252) | (1053) | 71 | 296 | *EST | 15825-93-1 |

Table 1. Positive Ion Table - Continued

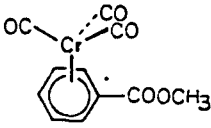
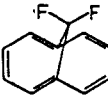
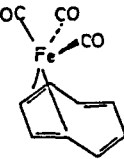
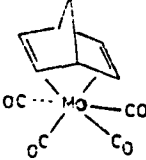
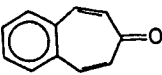
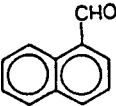
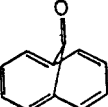
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{11}\text{H}_8\text{CrO}_5^+$  | (7.02±0.1) | (4) | (18) | -157 | -659 | 84ALT/CON | 12125-87-0 |
| $\text{C}_{11}\text{H}_8\text{F}_2^+$  | (8.19±0.03) IP from 84AND/CER. | (149) | (625) | -39 | -165 | *EST | 61997-36-2 |
| $\text{C}_{11}\text{H}_8\text{FeO}_3^+$  | (7.3) IP is onset of photoelectron band (82LEV/LIA, 80BOH/GLE). | (132) | (554) | -36±3 | -150±13 | 82PIL/SKI | 12093-05-9 |
| $\text{C}_{11}\text{H}_8\text{MoO}_4^+$  | (7.0) IP is onset of photoelectron band. | (81) | (339) | -80±3 | -336±11 | 82PIL/SKI | 12146-37-1 |
| $\text{C}_{11}\text{H}_8\text{O}^+$  | (8.3) IP is onset of photoelectron band. | (210) | (879) | 19 | 78 | *EST | 4443-91-8 |
|  | (8.33) IP from 83KLA/KOV. | (199) | (834) | 7 | 30 | *EST | 66-77-3 |
|  | (8.0) IP is onset of photoelectron band (84AND/CER). | (218) | (914) | 34 | 142 | *EST | 36628-80-5 |

Table 1. Positive Ion Table - Continued

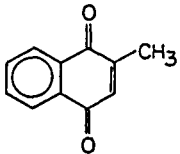
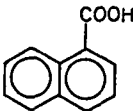
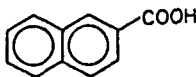
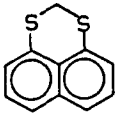
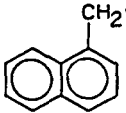
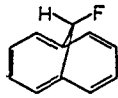
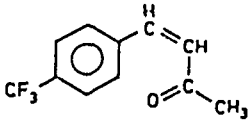
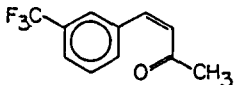
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{11}\text{H}_8\text{O}_2^+$ |  | (9.3) | (184) | (770) | -30 | -127 | *EST | 58-27-5 |
| | | IP is onset of photoelectron band (80RED/FRE). | | | | | | |
| |  | (8.29) | (138) | (577) | -53.3±0.2 | -223.1±0.9 | 77PED/RYL | 86-55-5 |
| | | IP from 83KLA/KOV. | | | | | | |
| |  | (8.26) | (135) | (564) | -55.6±0.4 | -232.5±1.6 | 77PED/RYL | 93-09-4 |
| | | IP from 83KLA/KOV. | | | | | | |
| $\text{C}_{11}\text{H}_8\text{S}_2^+$ |  | (7.3) | (237) | (991) | 69 | 287 | *EST | 204-14-8 |
| | | IP is onset of photoelectron band (81BOC/BRA). | | | | | | |
| $\text{C}_{11}\text{H}_9^+$ |  | (7.35±0.1) | (229) | (962) | 60 | 253 | 82MCM/GOL | 7419-60-5 |
| | | Appearance potential results (see 85HON/SEG and references cited therein) lead to value for $\Delta_f H(\text{Ion})$ of -252 kcal/mol, -1054 kJ/mol. | | | | | | |
| $\text{C}_{11}\text{H}_9\text{F}^+$ |  | (8.10±0.03) | (203) | (848) | 16 | 66 | *EST | 72791-63-0 |
| | | IP is onset of photoelectron band (84AND/CER). | | | | | | |
| $\text{C}_{11}\text{H}_9\text{F}_3\text{O}^+$ |  | (9.0±0.05) | (39) | (164) | -168 | -704 | 79SCH/GRU | 76293-37-3 |
| | | IP from 79SCH/GRU, 80GRU/SCH, 81SCH/GRO. | | | | | | |
| |  | (9.1±0.05) | (42) | (175) | -168 | -703 | *EST | |
| | | IP from 81SCH/GRO. | | | | | | |

Table 1. Positive Ion Table - Continued

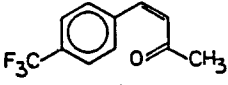
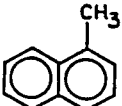
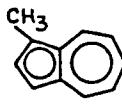
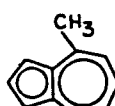
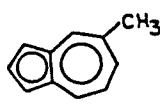
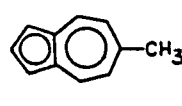
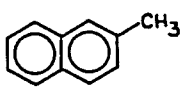
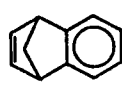
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{11}\text{H}_9\text{F}_3\text{O}^+$ | | | | | | | |
|  | (9.1±0.05) IP from 81SCH/GRO. | (42) | (175) | -168 | -703 | *EST | |
| $\text{C}_{11}\text{H}_{10}^+$ | | | | | | | |
|  | 7.85 IP from 82LEV/LIA, 83KLA/KOV. | 208 | 870 | 27±0.5 | 113±2 | 74SAB/CHA | 90-12-0 |
|  | (≤7.26±0.03) | (≤227) | (≤953) | 60 | 253 | *EST | 769-31-3 |
|  | (≤7.33±0.03) | (≤229) | (≤960) | 60 | 253 | *EST | 17647-77-7 |
|  | (≤7.30±0.03) | (≤227) | (≤950) | 59 | 246 | *EST | 1654-55-3 |
|  | (≤7.34±0.03) | (≤228) | (≤954) | 59 | 246 | *EST | 1654-52-0 |
|  | (7.8) IP is onset of photoelectron band (82LEV/LIA, 83KLA/KOV). | (206) | (864) | 27±0.5 | 111±2 | 74SAB/CHA | 91-57-6 |
|  | (8.1) IP is onset of photoelectron band. | (253) | (1057) | 66 | 276 | *EST | 4453-90-1 |

Table 1. Positive Ion Table - Continued

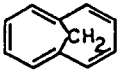
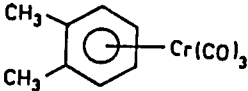
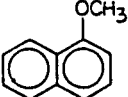
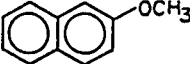
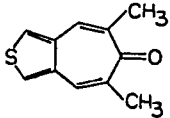
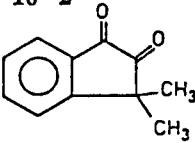
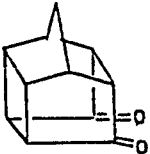
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{11}\text{H}_{10}^+$ |  | (7.7) | (253) | (1058) | 75±1 | 315±6 | 77PED/RYL | 2443-46-1 |
| | | IP is onset of photoelectron band (84AND/CER). | | | | | | |
| $\text{C}_{11}\text{H}_{10}\text{CrO}_3^+$ |  | (6.70±0.1) | (54) | (224) | -101 | -422 | *EST | 12129-29-2 |
| | | See also: 82GUI/PFI. | | | | | | |
| $\text{C}_{11}\text{H}_{10}\text{O}^+$ |  | 7.70 | (177) | (742) | -0.2 | -1 | *EST | 2216-69-5 |
| | | IP from 83KLA/KOV. | | | | | | |
| |  | (7.44) | (171) | (717) | -0.2 | -1 | *EST | 93-04-9 |
| | | IP is onset of photoelectron band (83KLA/KOV). | | | | | | |
| $\text{C}_{11}\text{H}_{10}\text{OS}^+$ |  | (≤8.40) | (≤201) | (≤840) | 7 | 30 | *EST | |
| | | IP from 84GLE/BIS. | | | | | | |
| $\text{C}_{11}\text{H}_{10}\text{O}_2^+$ |  | (8.5) | (145) | (607) | -51 | -213 | *EST | 20651-88-1 |
| | | IP is onset of photoelectron band. | | | | | | |
| |  | ≤9.1 | ≤195 | ≤814 | -15 | -64 | 64COO/CRU | |
| | | IP from 84MAR/KAY. | | | | | | |

Table 1. Positive Ion Table - Continued

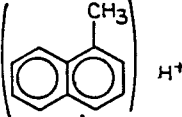
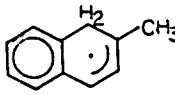
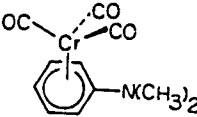
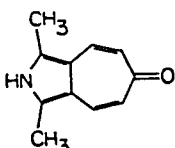
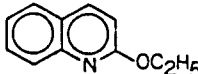
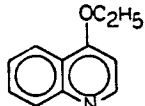
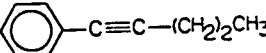
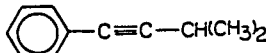
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_{11}H_{11}^+$ | | | | | | | |
|  | | 192 | 803 | | | | |
| | | From proton affinity of 1-methylnaphthalene (RN 90-12-0). PA = 200.7 kcal/mol, 840. kJ/mol. | | | | | |
|  | | 192 | 804 | | | | |
| | | From proton affinity of 2-methylnaphthalene (RN 91-57-6). PA = 200.0 kcal/mol, 837. kJ/mol. | | | | | |
| $C_{11}H_{11}CrNO_3^+$ | | | | | | | |
|  | (6.9) | (63) | (262) | -96±3 | -404±13 | 84ALT/CON | 12109-10-3 |
| $C_{11}H_{11}NO^+$ | | | | | | | |
|  | (7.55) | (182) | (763) | 8 | 35 | *EST | |
| | | IP from 84GLE/BIS. | | | | | |
|  | (8.0) | (181) | (756) | -4 | -16 | *EST | 46185-83-5 |
| | | IP is onset of photoelectron band (81PFI/GUI). | | | | | |
|  | (8.1) | (192) | (804) | 5 | 23 | *EST | 13720-91-7 |
| | | IP is onset of photoelectron band (81PFI/GUI). | | | | | |
| $C_{11}H_{12}^+$ | | | | | | | |
|  | (≤8.29±0.02) | (≤252) | (≤1055) | 61 | 255 | *EST | 4250-81-1 |
|  | (8.35±0.08) | (252) | (1053) | 59 | 247 | *EST | 1612-03-9 |
| | | IP is onset of photoelectron band (81ELB/LIE). | | | | | |

Table 1. Positive Ion Table - Continued

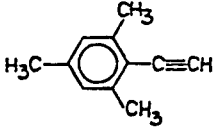
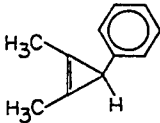
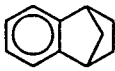
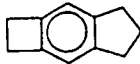
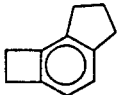
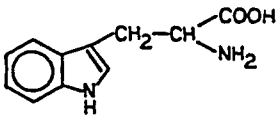
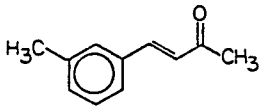
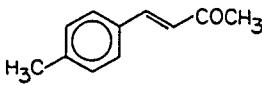
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_{11}H_{12}^+$ | | | | | | | |
|  | (8.16±0.08) IP from 81ELB/LIE. | (263) | (1102) | 75 | 315 | *EST | 769-26-6 |
|  | (≤8.17) IP from 84BAI/DOM. | (≤253) | (≤1058) | 65 | 270 | *EST | 23063-31-2 |
|  | (≤8.42±0.05) IP from 82HAS/NEU, 82LEV/LIA. | (≤224) | (≤938) | 30 | 126 | *EST | 4486-29-7 |
|  | (8.05) | (228) | (956) | 43 | 179 | *EST | 60582-10-7 |
|  | (8.19) | (232) | (969) | 43 | 179 | *EST | 60582-11-8 |
| $C_{11}H_{12}N_2O_2^+$ | | | | | | | |
|  | (≤7.5) See also: 83CAN/HAM. | (≤115) | (≤481) | -58 | -243 | *EST | 54-12-6 |
| $C_{11}H_{12}O^+$ | | | | | | | |
|  | (8.6±0.05) IP from 81SCH/GRO. | (182) | (762) | -16 | -68 | 81SCH/GRO | 15753-84-1 |
|  | (8.5±0.05) IP is onset of photoelectron band (81SCH/GRO). | (180) | (752) | -16 | -68 | *EST | 4023-84-1 |

Table 1. Positive Ion Table - Continued

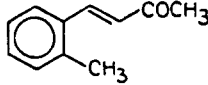
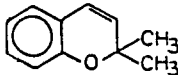

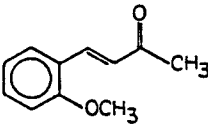
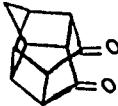
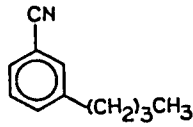
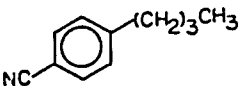
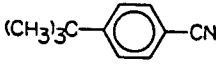
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₂O⁺ | | | | | | | |
|  | 8.5±0.05 IP from 79SCH/GRU, 80GRU/SCH, 81SCH/GRO. | 183 | 766 | -13 | -54 | 79SCH/GRU | 16927-82-5 |
|  | (7.8±0.05) IP from 79SCH/GRU. | (170) | (711) | -10 | -41 | 79SCH/GRU | 2513-25-9 |
|  | ≤8.8 IP from 84MAR/KAY. | (≤208) | (≤871) | 5 | 22 | *EST | |
| C₁₁H₁₂O₂⁺ | | | | | | | |
|  | (8.2) IP from 79SCH/GRU, 80GRU/SCH. | (146) | (612) | -43 | -179 | 79SCH/GRU | 10542-87-7 |
|  | (8.1) IP is onset of photoelectron band. | (205) | (860) | 19 | 78 | *EST | 60526-44-5 |
| C₁₁H₁₃N⁺ | | | | | | | |
|  | (9.77±0.1) | (254) | (1064) | 29±0.2 | 121±1 | *EST | 20651-74-5 |
|  | (10.08±0.1) | (261) | (1094) | 29±0.2 | 121±1 | *EST | 20651-73-4 |
|  | (8.8) IP is onset of photoelectron band (85BAI/MIS2). See also: 86ORL/MIS. | (229) | (959) | 26 | 110 | 85ORL/MIS | 4210-32-6 |

Table 1. Positive Ion Table - Continued

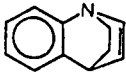
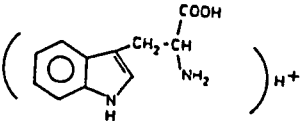
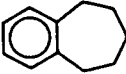
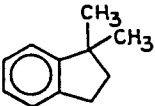
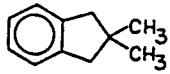
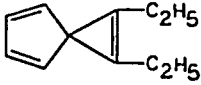

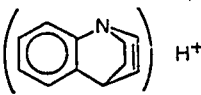
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|--|--------|----------------------|--|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₃N⁺  | (7.85±0.02) | (222) | (930) | 41 | 173 | *EST | 4363-25-1 |
| C₁₁H₁₃N₂O₂⁺  | | 82 | 344 | From proton affinity of L-tryptophan (RN 54-12-6). PA = 225.4 kcal/mol, 943. kJ/mol. | | | |
| C₁₁H₁₄⁺  | ≤8.40±0.02 | (≤198) | (≤827) | 4±0.7 | 17±3 | *EST | 1075-16-7 |
|  | (8.47) | (195) | (815) | -0.5±0.2 | -2±1 | 78OSB/SCO | 4912-92-9 |
|  | (8.47) | (195) | (815) | 0.5±0.2 | -2±1 | *EST | 20836-11-7 |
|  | (7.87) | (260) | (1089) | 79 | 330 | *EST | 49542-94-1 |
|  | (8.25) | (286) | (1196) | 96 | 400 | *EST | 58738-49-1 IP is onset of photoelectron band (82SPA/KOR). |
| C₁₁H₁₄N⁺  | | 175 | 732 | From proton affinity of 1,4-dihydro-1,4-ethanoquinoline (RN 4363-25-1). PA = 232.0 kcal/mol, 971. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

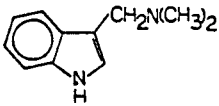
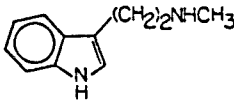
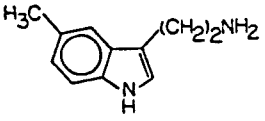
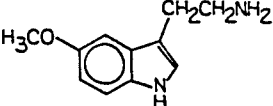
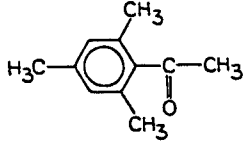
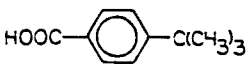
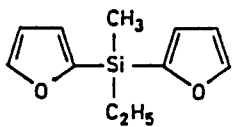
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{11}\text{H}_{14}\text{N}_2^+$ | | | | | | | |
|  | $(\leq 7.69 \pm 0.16)$ | (≤ 215) | (≤ 900) | 38 | 158 | *EST | 87-52-5 |
|  | (7.7) IP is onset of photoelectron band. | (206) | (864) | 29 | 121 | *EST | 61-49-4 |
|  | (7.6) IP is onset of photoelectron band. | (201) | (839) | 25 | 106 | *EST | 1821-47-2 |
| $\text{C}_{11}\text{H}_{14}\text{N}_2\text{O}^+$ | | | | | | | |
|  | $(\leq 7.68 \pm 0.12)$ | (≤ 174) | (≤ 729) | -3 | -12 | *EST | 608-07-1 |
| $\text{C}_{11}\text{H}_{14}\text{O}^+$ | | | | | | | |
|  | (8.2) IP is onset of photoelectron band (78CEN/FRA). | (140) | (586) | -49.0 ± 0.9 | -204.9 ± 3.6 | 77PED/RYL | 1667-01-2 |
| $\text{C}_{11}\text{H}_{14}\text{O}_2^+$ | | | | | | | |
|  | (8.6) IP is onset of photoelectron band (85BAI/MIS2). See also: 86ORL/MIS. | (103) | (431) | -95 | -399 | 85ORL/MIS | 98-73-7 |
| $\text{C}_{11}\text{H}_{14}\text{O}_2\text{Si}^+$ | | | | | | | |
|  | (8.1) IP is onset of photoelectron band (83ZYK/ERC). | (138) | (577) | -49 | -205 | *EST | |

Table 1. Positive Ion Table - Continued

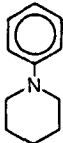
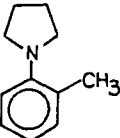
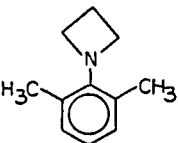
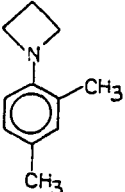
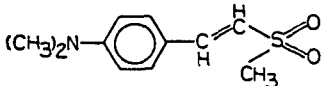
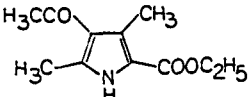
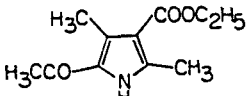
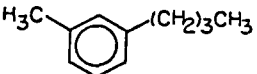
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₅N⁺  | (7.1) | (172) | (718) | 8 | 33 | *EST | 4096-20-2 |
| | IP is onset of photoelectron band (82ROZ/HOU2). | | | | | | |
|  | (6.8) | (184) | (771) | 27 | 115 | *EST | 41378-30-7 |
| | IP is onset of photoelectron band (82ROZ/HOU2). | | | | | | |
|  | (7.0) | (230) | (961) | 68 | 286 | *EST | 19199-06-5 |
| | IP is onset of photoelectron band (82ROZ/HOU2). | | | | | | |
|  | (≤ 7.48) | (≤ 223) | (≤ 933) | 50 | 211 | *EST | 81506-10-7 |
| | IP from 82ROZ/HOU2. | | | | | | |
| C₁₁H₁₅NO₂S⁺  | (7.0) | (129) | (538) | -33 | -137 | *EST | |
| | IP is onset of photoelectron band (84CAU/FUR). | | | | | | |
| C₁₁H₁₅NO₃⁺  | ≤ 8.26 | (≤ 54) | (≤ 227) | -136 | -570 | *EST | |
| | IP from 81CAU/GIA. | | | | | | |
|  | (≤ 8.23) | (≤ 54) | (≤ 224) | -136 | -570 | *EST | 6314-22-3 |
| | IP from 81CAU/GIA. | | | | | | |
| C₁₁H₁₆⁺  | (8.42 \pm 0.1) | (184) | (768) | -11 | -44 | *EST | 1595-04-6 |

Table 1. Positive Ion Table - Continued

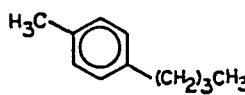
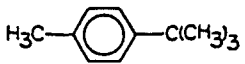
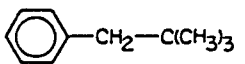
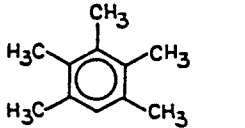
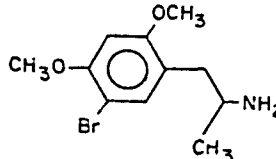
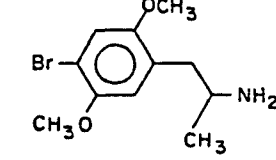
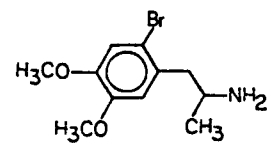
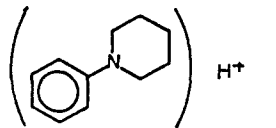
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|--------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₆⁺ | | | | | | | |
|  | (8.35±0.1) | (182) | (761) | -11 | -45 | *EST | 1595-05-7 |
|  | 8.28 IP from 86ORL/MIS. See also: 85BAI/MIS. | (178) | (744) | -13 | -55 | 85ORL/MIS | 98-51-1 |
|  | ≤8.7 | (≤187) | (≤784) | -13 | -55 | *EST | 1007-26-7 |
|  | 7.92±0.02 Sec: 84HOW/GON. | (165) | (690) | -18 | -74 | *EST | 700-12-9 |
| C₁₁H₁₆BrNO₂⁺ | | | | | | | |
|  | (7.4) IP is onset of photoelectron band (81DOM/EAT). | (97) | (406) | -74 | -308 | *EST | 60917-67-1 |
|  | (7.3) IP is onset of photoelectron band (81DOM/EAT, 82LEV/LIA). | (97) | (404) | -72 | -300 | *EST | 64638-07-9 |
|  | (7.4) IP is onset of photoelectron band (81DOM/EAT). | (102) | (425) | -69 | -289 | *EST | 32156-25-5 |
| C₁₁H₁₆N⁺ | | | | | | | |
|  | | 154 | 642 | From proton affinity of 1-phenylpiperidine (RN 4096-20-2). PA = 225.8 kcal/mol, 945. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

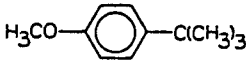
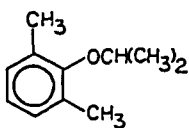
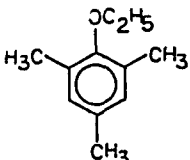
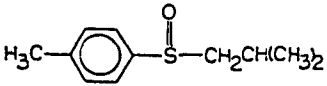
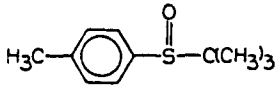
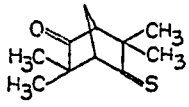
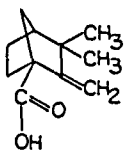
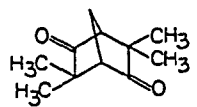
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₆O⁺ | | | | | | | |
|  | (7.77) IP from 86ORL/MIS. | (138) | (576) | -41.5 | -173.8 | 86ORL/MIS | 5396-38-3 |
|  | 8.49 | (162) | (676) | -34 | -143 | *EST | 54350-31-1 |
|  | (≤8.28) | (≤165) | (≤692) | -26 | -107 | *EST | 61248-63-3 |
| C₁₁H₁₆OS⁺ | | | | | | | |
|  | (≤8.50) IP from 81MOH/JIA. | (≤163) | (≤681) | -33 | -139 | *EST | 77919-66-5 |
|  | (≤8.33) IP from 81MOH/JIA. | (≤162) | (≤678) | -30 | -126 | *EST | 49833-45-6 |
|  | (8.25) IP is onset of photoelectron band (80FRO/WES). | (144) | (601) | -47 | -195 | *EST | 75503-13-8 |
| C₁₁H₁₆O₂⁺ | | | | | | | |
|  | (≤9.05) IP from 81NES/BAI. | (≤112) | (≤468) | -97 | -405 | *EST | 10309-20-3 |
|  | (8.86) IP is onset of photoelectron band (80FRO/WES). | (108) | (450) | -97 | -405 | *EST | 57239-03-9 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₆S⁺ | | | | | | | | |
| | | ($\leq 7.83 \pm 0.05$) | (≤ 179) | (≤ 747) | -2 | -8 | *EST | 7252-86-0 |
| | | (≤ 8.35) | (≤ 188) | (≤ 788) | -4 | -18 | *EST | 34786-26-0 |
| | | (≤ 8.31) | (≤ 187) | (≤ 784) | -4 | -18 | *EST | 7439-10-3 |
| C₁₁H₁₆S₂⁺ | | | | | | | | |
| | | (8.2) IP is onset of photoelectron band (80FRO/WES). | (193) | (806) | 4 | 15 | *EST | 75503-14-9 |
| C₁₁H₁₇N⁺ | | | | | | | | |
| | | (7.24) | (174) | (729) | 7 | 30 | *EST | 13021-15-3 |
| | | (6.90) | (160) | (670) | 1 | 4 | *EST | 91-67-8 |
| | | (6.83) | (160) | (671) | 3 | 12 | *EST | 613-48-9 |
| | | IP from charge transfer equilibrium constant determinations; reference standard: IP (C ₆ H ₅ N(CH ₃) ₂ = 7.12 eV). (85LIA/JAC). | | | | | | |
| | | IP from charge transfer equilibrium constant determinations; reference standard: IP (C ₆ H ₅ N(CH ₃) ₂ = 7.12 eV). (84MAU/NEL, 85LIA/JAC). | | | | | | |
| C₁₁H₁₇NO₂⁺ | | | | | | | | |
| | | (≤ 8.30) | (≤ 130) | (≤ 544) | -61 | -257 | *EST | 15402-81-0 |
| | | IP from 81DOM/EAT. | | | | | | |

Table 1. Positive Ion Table - Continued

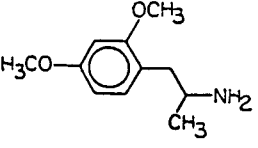
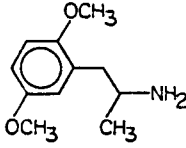
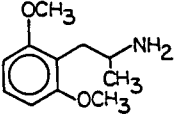
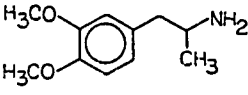
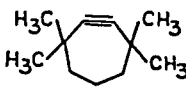
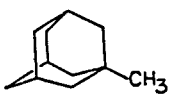
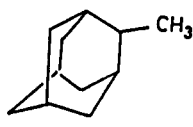

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₇NO₂⁺ | | | | | | | |
|  | (7.4) | (99) | (416) | -71 | -298 | *EST | 23690-13-3 |
| IP is onset of photoelectron band (81DOM/EAT, 82LEV/LIA). | | | | | | | |
|  | (7.1) | (94) | (395) | -69 | -290 | *EST | 13641-74-2 |
| IP is onset of photoelectron band (81DOM/EAT, 82LEV/LIA). | | | | | | | |
|  | (8.18) | (117) | (491) | -71 | -298 | *EST | 23690-14-4 |
| IP from 81DOM/EAT. | | | | | | | |
|  | ($\leq 8.03 \pm 0.06$) | (≤ 118) | (≤ 492) | -68 | -283 | *EST | 120-26-3 |
| See also: 81DOM/GAP. | | | | | | | |
| C₁₁H₁₈⁺ | | | | | | | |
|  | (8.4) | (225) | (940) | 31 | 130 | *EST | 33470-40-5 |
| IP is onset of photoelectron band. | | | | | | | |
|  | (9.17 \pm 0.02) | (170.9) | (715.0) | -40.6 \pm 0.3 | -169.8 \pm 1.4 | 79CLA/KNO | 768-91-2 |
|  | 9.24 | 176 | 737 | -36.9 | -154 | 79CLA/KNO | |
|  | (8.5) | (232) | (972) | 36 | 152 | *EST | 52879-54-6 |
| IP is onset of photoelectron band (82SPA/GLE). | | | | | | | |

Table 1. Positive Ion Table - Continued

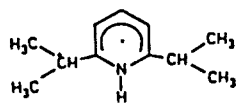
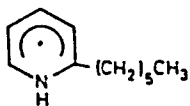
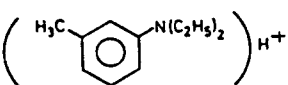
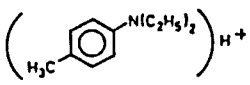
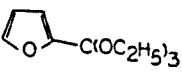
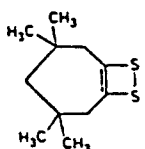
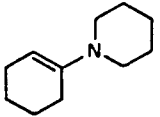
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₈N⁺ | | | | | | | |
|  | | 125 | 523 | | | | |
| | | From proton affinity of 2,6-diisopropylpyridine (RN 6832-21-9). PA = 232.9 kcal/mol, 974. kJ/mol. | | | | | |
|  | | 137 | 572 | | | | |
| | | From proton affinity of 2-n-hexylpyridine (RN 1129-69-7). PA = 228.9 kcal/mol, 958. kJ/mol. | | | | | |
|  | | 138 | 578 | | | | |
| | | From proton affinity of 3-CH ₃ C ₆ H ₄ N(C ₂ H ₅) ₂ (RN 91-67-8). PA = 228.9 kcal/mol, 956. kJ/mol. | | | | | |
|  | | 140 | 587 | | | | |
| | | From proton affinity of 4-CH ₃ C ₆ H ₄ N(C ₂ H ₅) ₂ (RN 613-48-9). PA = 228.6 kcal/mol, 956. kJ/mol. | | | | | |
| C₁₁H₁₈O₄⁺ | | | | | | | |
|  | (8.7) | (46) | (190) | -155 | -649 | *EST | 75905-10-1 |
| | | IP is onset of photoelectron band (83ZYK/ERC). | | | | | |
| C₁₁H₁₈S₂⁺ | | | | | | | |
|  | 7.65 | (203) | (848) | 26 | 110 | *EST | |
| | | IP is onset of photoelectron band (83JLA/MOH). | | | | | |
| C₁₁H₁₉N⁺ | | | | | | | |
|  | (≤7.44±0.03) | (≤162) | (≤676) | -10 | -42 | *EST | 2981-10-4 |

Table 1. Positive Ion Table - Continued

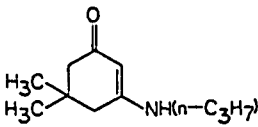
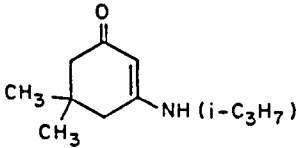
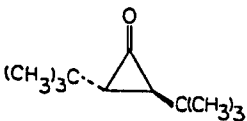
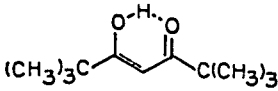
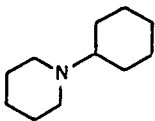
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|---|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₁H₁₉NO⁺ | | | | | | | | |
| |  | (≤ 8.03) IP from 82PFI/GER. | (≤ 131) | (≤ 548) | -54 | -227 | *EST | 56570-54-8 |
| |  | (≤ 7.94) IP from 82PFI/GER. | (≤ 125) | (≤ 525) | -58 | -241 | *EST | 80555-73-3 |
| C₁₁H₂₀⁺ | | | | | | | | |
| | (<i>tert</i> -C ₄ H ₉) ₂ C=C=CH ₂ | (≤ 8.55) | (≤ 206) | (≤ 860) | 8 | 35 | *EST | 22585-31-5 |
| | (CH ₃) ₃ CCH=C=CHC(CH ₃) ₃ IP is onset of photoelectron band (8SELS/VER). | (8.6) | (193) | (807) | -5 | -23 | *EST | 42066-39-7 |
| | 1-C ₁₁ H ₂₀ | (9.90 \pm 0.02) | (233) | (976) | 5 | 21 | *EST | 2243-98-3 |
| | 2-C ₁₁ H ₂₀ | (9.28 \pm 0.02) | (214) | (897) | 0.5 | 2 | *EST | 60212-29-5 |
| | 3-C ₁₁ H ₂₀ | (9.17 \pm 0.02) | (212) | (888) | 0.8 | 3 | *EST | 60212-30-8 |
| | 4-C ₁₁ H ₂₀ | (9.13 \pm 0.02) | (211) | (884) | 0.8 | 3 | *EST | 60212-31-9 |
| | 5-C ₁₁ H ₂₀ | (9.11 \pm 0.02) | (211) | (882) | 0.8 | 3 | *EST | 2294-72-6 |
| C₁₁H₂₀O⁺ | | | | | | | | |
| |  | (≤ 8.45) | (≤ 147) | (≤ 613) | -48 | -202 | *EST | 14743-58-9 |
| C₁₁H₂₀O₂⁺ | | | | | | | | |
| |  | (7.9) Heat of formation of neutral molecule refers to enol form. IP is onset of photoelectron band. | (56) | (234) | -126 \pm 1 | -528 \pm 4 | 81FER/RIB | 1118-71-4 |
| C₁₁H₂₁N⁺ | | | | | | | | |
| |  | ($\leq 7.93\pm 0.03$) | (≤ 146) | (≤ 609) | -37 | -156 | *EST | 3319-01-5 |

Table 1. Positive Ion Table - Continued

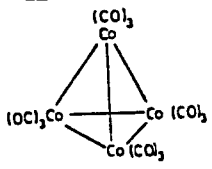
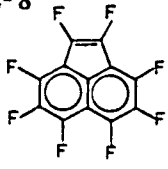
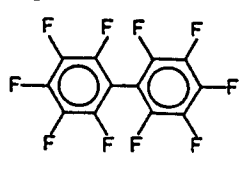
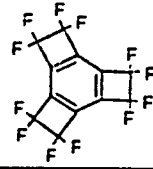
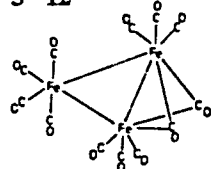
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|----------------------------|--------------------------|---------------|------------------------------|------------------|---|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{11}\text{H}_{22}^+$ | $\text{C}_2\text{H}_5\text{CH}_2\text{C}(\text{C}_2\text{H}_5)=\text{C}(\text{C}_2\text{H}_5)_2$ | $(8.04 \pm 0.02)^a$ | (145) | (606) | -41 | -170 | *EST | 50787-14-9 |
| $\text{C}_{11}\text{H}_{24}^+$ | $n\text{-C}_{11}\text{H}_{24}$ | (9.56) | (156) | (651) | -65 ± 0.6 | -271 ± 3 | 77PED/RYL | 1120-21-4 |
| | | | | | | | IP from charge transfer equilibrium constant determinations (81MAU/SIE, 82LIA). | |
| | $n\text{-C}_8\text{H}_{17}\text{CH}(\text{CH}_3)_2$ | (9.68) | (157) | (656) | -66 | -278 | *EST | 6975-98-0 |
| | | | | | | | IP from charge transfer equilibrium constant determinations (81MAU/SIE, 82LIA). | |
| $\text{C}_{12}\text{Co}_4\text{O}_{12}^+$ |  | 7.45 | (-246.2) | (-1030.1) | -418.0 ± 3.2 | -1748.9 ± 13 | 82PIL/SKI | 17786-31-1 |
| | | | | | | | IP is onset of photoelectron band. See also: 81GRE/MIN. | |
| $\text{C}_{12}\text{F}_8^+$ |  | $(\leq 9.1 \pm 0.1)$ | (≤ -56) | (≤ -234) | -266 | -1112 | *EST | 1554-93-4 |
| $\text{C}_{12}\text{F}_{10}^+$ |  | (9.40 ± 0.02) | (-115) | (-480) | -332 ± 3 | -1387 ± 12 | 79PRI/SAP2 | 434-90-2 |
| $\text{C}_{12}\text{F}_{12}^+$ |  | 11.14 | (-211) | (-884) | -468 | -1959 | *EST | 32937-02-3 |
| | | | | | | | IP is onset of photoelectron band (84HEI/WIR). | |
| $\text{C}_{12}\text{F}_{27}\text{N}^+$ | $(n\text{-C}_4\text{F}_9)_3\text{N}$ | (11.3) | (-1067) | (-4466) | -1328 ± 2 | -5556 ± 10 | 79ERA/KOL | 311-89-7 |
| | | | | | | | IP is onset of photoelectron band (82ELB/DIE, 83MOL/PIK3). | |
| $\text{C}_{12}\text{Fe}_3\text{O}_{12}^+$ |  | (7.44) | (-247) | (-1035) | -419 ± 6 | -1753 ± 27 | 82PIL/SKI | 17685-52-8 |
| | | | | | | | IP is onset of photoelectron band (82DEK/WON). | |

Table 1. Positive Ion Table - Continued

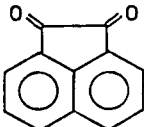
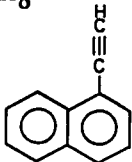
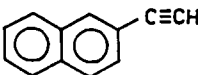

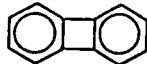
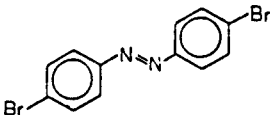
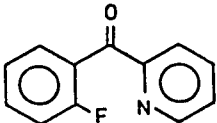
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_6\text{O}_2^+$ |  | (8.6) | (305) | (1275) | 106 | 445 | *EST | 82-86-0 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_{12}\text{H}_8^+$ |  | (8.03) | (279) | (1166) | 93 | 391 | *EST | 15727-65-8 |
| | | IP is onset of photoelectron band (81GLE/SCH). | | | | | | |
| |  | (8.11) | (280) | (1173) | 93 | 391 | *EST | 2949-26-0 |
| | | IP is onset of photoelectron band (81GLE/SCH). | | | | | | |
| |  | (8.22±0.04) | (252) | (1053) | 62±0.2 | 260±1 | 81KUD/KUD | 208-96-8 |
| |  | 7.56±0.02 | 279 | 1166 | 104±3 | 437±13 | 77PED/RYL | 259-79-0 |
| | | IP derived from charge transfer equilibrium constant determinations is in agreement (80MAU). See also: 85DEW/TIE. | | | | | | |
| $\text{C}_{12}\text{H}_8\text{Br}_2\text{N}_2^+$ |  | (9.24) | (324) | (1355) | 110.8 | 463.6 | *EST | 1601-98-5 |
| | | IP from 77NUY/MES. | | | | | | |
| $\text{C}_{12}\text{H}_8\text{FNO}^+$ |  | 9.11 | (222) | (927) | 11 | 48 | *EST | 6238-65-9 |
| | | IP from 80GRU/SCH. See also: 82LEV/LIA. | | | | | | |

Table 1. Positive Ion Table - Continued

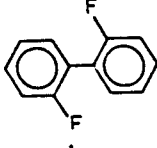
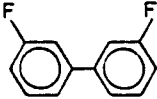
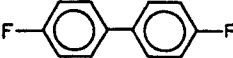
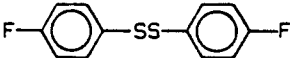
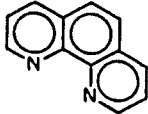
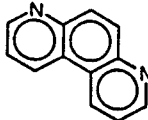
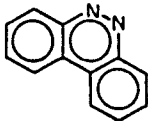
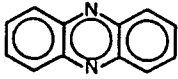
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_8\text{F}_2^+$ | | | | | | | |
|  | (8.35±0.02) | (147) | (616) | -45±1 | -190±5 | 64SMI/GOV | 388-82-9 |
|  | (8.35±0.02) | (146) | (611) | -47 | -195 | *EST | 396-64-5 |
|  | (8.00±0.02) | (138) | (577) | -47±1 | -195±5 | 64SMI/GOV | 398-23-2 |
| $\text{C}_{12}\text{H}_8\text{F}_2\text{S}_2^+$ | | | | | | | |
|  | (≤8.4) IP from 82GIO/BOC. | (≤109) | (≤456) | -85 | -354 | *EST | 405-31-2 |
| $\text{C}_{12}\text{H}_8\text{N}_2^+$ | | | | | | | |
|  | (8.3) IP is onset of photoelectron band. | (270) | (1130) | 79 | 329 | *EST | 66-71-7 |
|  | 8.35±0.02 | (269) | (1127) | 77 | 321 | *EST | 230-07-9 |
|  | (7.9) IP is onset of photoelectron band. | (277) | (1159) | 95 | 397 | 77SCH/PET | 230-17-1 |
|  | 8.33±0.02 | 274 | 1148 | 82±0.7 | 344±3 | 80ARS | 92-82-0 |

Table 1. Positive Ion Table - Continued

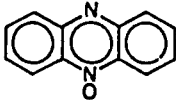
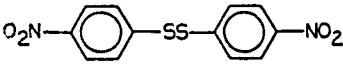
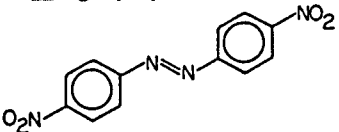
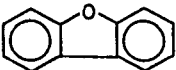
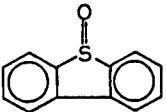
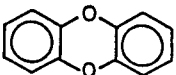
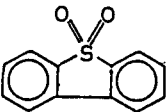
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_8\text{N}_2\text{O}^+$  | 8.00±0.02 | (247) | (1036) | 63 | 264 | *EST | 304-81-4 |
| $\text{C}_{12}\text{H}_8\text{N}_2\text{O}_4\text{S}_2^+$  | (≤8.98) IP from 82GIO/BOC. | (≤258) | (≤1080) | 51 | 214 | *EST | 100-32-3 |
| $\text{C}_{12}\text{H}_8\text{N}_4\text{O}_4^+$  | (9.97) IP from 77NUY/MES. | (314) | (1312) | 83.6 | 349.8 | *EST | |
| $\text{C}_{12}\text{H}_8\text{O}^+$  | 7.9±0.05 | 202 | 845 | 20±1 | 83±5 | 77PED/RYL | 132-64-9 |
| $\text{C}_{12}\text{H}_8\text{OS}^+$  | (8.1) IP is onset of photoelectron band. | (206) | (863) | 19 | 81 | *EST | 1013-23-6 |
| $\text{C}_{12}\text{H}_8\text{O}_2^+$  | (7.5) IP is onset of photoelectron band. | (158) | (661) | -15 | -63 | 82SHA | 262-12-4 |
| $\text{C}_{12}\text{H}_8\text{O}_2\text{S}^+$  | (8.9) IP is onset of photoelectron band. | (171) | (714) | -35 | -145 | *EST | 1016-05-3 |

Table 1. Positive Ion Table - Continued

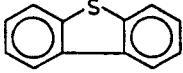
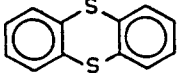
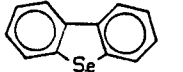
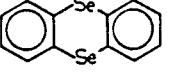
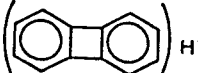
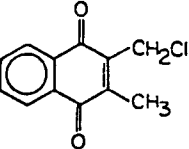
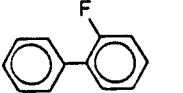
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|---------|---|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_8\text{S}^+$  | 7.90±0.03 | 231 | 967 | 49±0.7 | 205±1 | 79SAB | 132-65-0 |
| $\text{C}_{12}\text{H}_8\text{S}_2^+$  | (7.7) IP is onset of photoelectron band (81TRA/RED). | (201) | (840) | 23±1 | 97±6 | 77PED/RYL | 92-85-3 |
| $\text{C}_{12}\text{H}_8\text{Se}^+$  | (≤7.86) IP from 82TRA/ROD. | (≤243) | (≤1018) | 62 | 260 | *EST | 244-95-1 |
| $\text{C}_{12}\text{H}_8\text{Se}_2^+$  | (≤7.89) IP from 82TRA/ROD. | (≤231) | (≤968) | 49 | 207 | *EST | |
| $\text{C}_{12}\text{H}_9^+$  | | 267 | 1116 | From proton affinity of biphenylene. (RN 259-79-0). PA = 203.4 kcal/mol, 851. kJ/mol. | | | |
| $\text{C}_{12}\text{H}_9\text{ClO}_2^+$  | (9.25) IP is onset of photoelectron band (80RED/FRE). | (242) | (1010) | 28 | 118 | *EST | 31599-79-8 |
| $\text{C}_{12}\text{H}_9\text{F}^+$  | (8.20±0.02) | (185) | (774) | -4 | -17 | *EST | 321-60-8 |

Table 1. Positive Ion Table - Continued

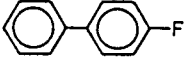
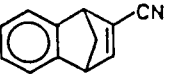
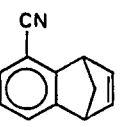
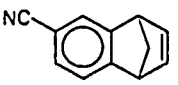
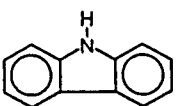
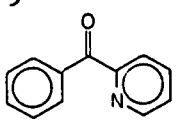
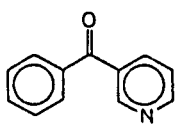
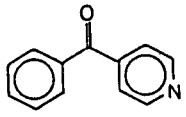
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₂H₉F⁺ | | | | | | | | |
| |  | (8.00±0.02) | (180) | (755) | -4 | -17 | *EST | 324-74-3 |
| C₁₂H₉N⁺ | | | | | | | | |
| |  | (8.5) IP is onset of photoelectron band. | (294) | (1228) | 98 | 408 | *EST | 71906-57-5 |
| |  | (8.7) IP is onset of photoelectron band. | (301) | (1259) | 100 | 420 | *EST | 61346-79-0 |
| |  | (8.7) IP is onset of photoelectron band. | (300) | (1255) | 99 | 416 | *EST | 16513-60-3 |
| |  | 7.57±0.03 | 229 | 959 | 55±0.2 | 229±1 | 81KUD/KUD2 | 86-74-8 |
| C₁₂H₉NO⁺ | | | | | | | | |
| |  | 9.06 IP from 82LEV/LIA, 80GRU/SCH. | (247) | (1032) | 38 | 158 | *EST | 91-02-1 |
| |  | (9.6±0.1) | (261) | (1090) | 39 | 164 | *EST | 5424-19-1 |
| |  | (9.6±0.1) | (261) | (1090) | 39 | 164 | *EST | 14548-46-0 |

Table 1. Positive Ion Table - Continued

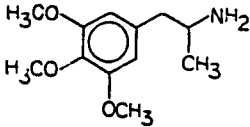
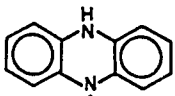
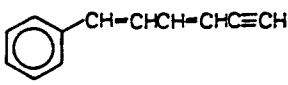
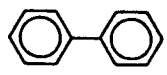
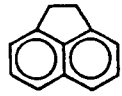
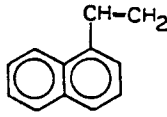
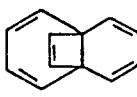
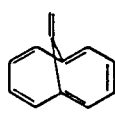
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|-----------------|------------------------------|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_9\text{NO}_3^+$  | ($\leq 8.16 \pm 0.06$) | (≤ 90) | (≤ 376) | -98 | -411 | *EST | 22199-17-3 |
| $\text{C}_{12}\text{H}_9\text{N}_2^+$  | | 224 | 938 | | | | |
| | | From proton affinity of phenazine (RN 92-82-0). PA = 223.7 kcal/mol, 936. kJ/mol. | | | | | |
| $\text{C}_{12}\text{H}_{10}^+$  | (7.9) IP from 74KOP/SCH. | (244) | (1019) | 61.5 | 257.3 | 62MOM/BRA | 940-50-1 |
|  | 7.95 ± 0.02 See also: 74KOP/SCH. | 226.9 | 949.4 | 43.6 ± 0.3 | 182.3 ± 1.4 | 77PED/RYL | 92-52-4 |
|  | (7.68) | (214) | (896) | 37 ± 0.2 | 155 ± 1 | 81KUD/KUD | 83-32-9 |
| | IP from charge transfer equilibrium constant determinations (80MAU, re-evaluated). | | | | | | |
|  | (7.7) | (229) | (958) | 51 | 215 | *EST | 826-74-4 |
| | IP is onset of photoelectron band (81GLE/SCH). | | | | | | |
|  | (≤ 8.1) | (≤ 286) | (≤ 1197) | 99 | 415 | *EST | 19539-78-7 |
|  | (7.5) | (269) | (1127) | 96 | 403 | *EST | 10474-24-5 |
| | IP is onset of photoelectron band (84AND/CER). | | | | | | |

Table 1. Positive Ion Table - Continued

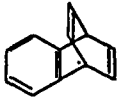
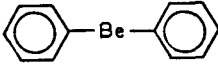
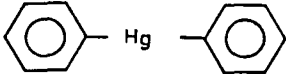
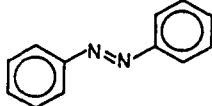
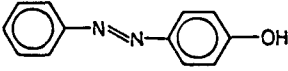
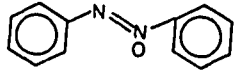
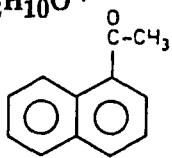
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{10}^+$  | (8.0) IP is onset of photoelectron band (82HAS/NEU). | (259) | (1083) | 74 | 311 | *EST | 7322-47-6 |
| $\text{C}_{12}\text{H}_{10}\text{Be}^+$  | (9.20±0.10) | (285) | (1193) | 73±5 | 305±21 | 80TEL/RAB | 22300-89-6 |
| $\text{C}_{12}\text{H}_{10}\text{Hg}^+$  | 8.30±0.03 See also: 81FUR/PIA. | 285 | 1192 | 93.5±0.8 | 391.4±3.2 | 77PED/RYL | 587-85-9 |
| $\text{C}_{12}\text{H}_{10}\text{N}_2^+$  | (8.2) IP is onset of photoelectron band. See also: 81NAT/FRA. | (286) | (1195) | 97±0.7 | 404±3 | 77SCH/PET | 17082-12-1 |
| $\text{C}_{12}\text{H}_{10}\text{N}_2\text{O}^+$  | 7.6 IP is onset of photoelectron band (81MIL/MIL, 82LEV/LIA). | (229) | (958) | 54 | 225 | *EST | 20714-70-9 |
|  | (8.1) IP is onset of photoelectron band (81MIL/CIL). | (269) | (1124) | 81.7±0.6 | 342±2.4 | 86KIR/ACR | 495-48-7 |
| $\text{C}_{12}\text{H}_{10}\text{O}^+$  | (8.23) | (185) | (773) | -5±2 | -21±8 | *EST | 941-98-0 |

Table 1. Positive Ion Table - Continued

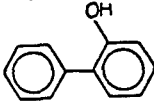
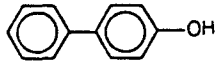
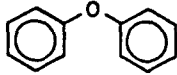
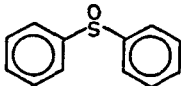
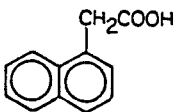
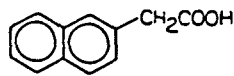
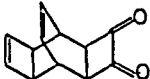
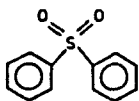
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{10}\text{O}^+$  | (7.80±0.02) | (181) | (756) | 0.7 | 3 | *EST | 90-43-7 |
|  | (7.78±0.03) | (180) | (754) | 0.7 | 3 | *EST | 92-69-3 |
|  | 8.09±0.03 | 183 | 766 | -3.6±0.4 | -14.9±1.8 | 77PED/RYL | 101-84-8 |
| <hr/> | | | | | | | |
| $\text{C}_{12}\text{H}_{10}\text{OS}^+$  | (8.3) IP is onset of photoelectron band. | (217) | (908) | 26±0.7 | 107±3 | 77PED/RYL | 945-51-7 |
| <hr/> | | | | | | | |
| $\text{C}_{12}\text{H}_{10}\text{O}_2^+$  | (7.71) IP is onset of photoelectron band (83KLA/KOV). | (111) | (464) | -67 | -280 | *EST | 86-87-3 |
|  | (8.05) IP from 83KLA/KOV. | (118) | (495) | -67 | -282 | *EST | 581-96-4 |
|  | (8.5) IP is onset of photoelectron band (85ALB/HEL). | (227) | (948) | 31 | 128 | *EST | |
| <hr/> | | | | | | | |
| $\text{C}_{12}\text{H}_{10}\text{O}_2\text{S}^+$  | 9.16±0.03 See: 81TRA/RED. | 183 | 765 | -28±0.7 | -119±3 | 77PED/RYL | 127-63-9 |

Table 1. Positive Ion Table - Continued

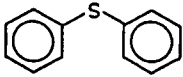
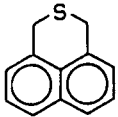
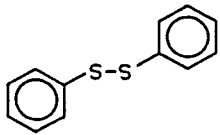
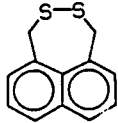
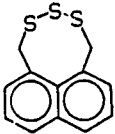
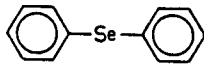
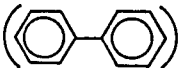
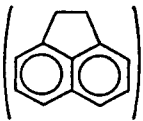
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|--|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₂H₁₀S⁺ | | | | | | | |
|  | 7.86±0.04 See also: 81TRA/RED. | 236 | 989 | 55±0.7 | 231±3 | 77PED/RYL | 139-66-2 |
|  | (7.91) IP is onset of photoelectron band (81GUT/BES). | (232) | (972) | 50 | 209 | *EST | |
| C₁₂H₁₀S₂⁺ | | | | | | | |
|  | ≤8.3 IP from 82GIO/BOC. | ≤250 | ≤1045 | 58±1 | 244±4 | 77PED/RYL | 882-33-7 |
|  | (7.4) IP is onset of photoelectron band (81GUT/BES). | (223) | (931) | 52 | 217 | *EST | 75574-98-0 |
| C₁₂H₁₀S₃⁺ | | | | | | | |
|  | (7.2) IP is onset of photoelectron band (81GUT/BES). | (221) | (925) | 55 | 230 | *EST | 75574-99-1 |
| C₁₂H₁₀Se⁺ | | | | | | | |
|  | (≤7.79) IP from 82TRA/ROD. | (≤248) | (≤1038) | 68.4±1.2 | 286.4±5.2 | 77PED/RYL | 1132-39-4 |
| C₁₂H₁₁⁺ | | | | | | | |
|  | H ⁺ | 213 | 892 | From proton affinity of biphenyl (RN 92-52-4). PA = 196.1 kcal/mol, 820. kJ/mol. | | | |
|  | H ⁺ | 199 | 834 | From proton affinity of acenaphthene (RN 83-32-9). PA = 203.5 kcal/mol, 851. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

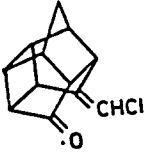
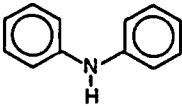
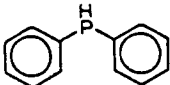
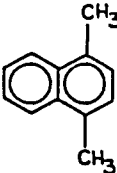
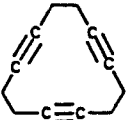
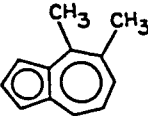
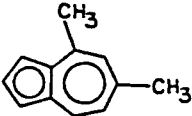
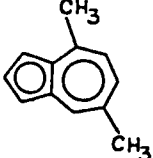
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{11}\text{ClO}^+$  | 9.0 IP from 84MAR/KAY. | (204) | (853) | -4 | -15 | *EST | |
| $\text{C}_{12}\text{H}_{11}\text{N}^+$  | 7.16±0.04 | 217 | 910 | 52±0.7 | 219±3 | 78STE | 122-39-4 |
| $\text{C}_{12}\text{H}_{11}\text{P}^+$  | (7.80±0.01) | (234) | (979) | 54 | 226 | *EST | 829-85-6 |
| $\text{C}_{12}\text{H}_{12}^+$  | 7.78±0.03 | 199 | 834 | 20 | 83 | 69STU/WES | 571-58-4 |
|  | (9.0) IP is onset of photoelectron band. | (344) | (1441) | 137 | 573 | 80BAR/STR | 60323-50-4 |
|  | (≤7.18±0.03) | (≤219) | (≤915) | 53 | 222 | *EST | 56594-77-5 |
|  | (≤7.29±0.03) | (≤221) | (≤924) | 53 | 221 | *EST | 56594-78-6 |
|  | (≤7.20±0.03) | (≤219) | (≤916) | 53 | 221 | *EST | 46030-99-3 |

Table 1. Positive Ion Table - Continued

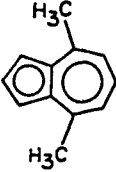
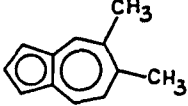
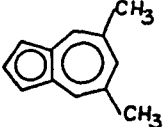
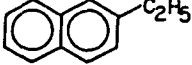
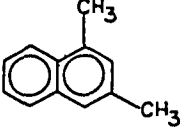
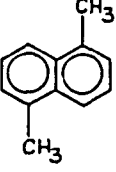
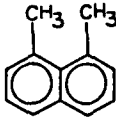
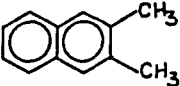
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|---------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{12}^+$ | | | | | | | |
|  | ($\leq 7.27 \pm 0.03$) | (≤ 221) | (≤ 923) | 53 | 222 | *EST | 7206-52-2 |
|  | ($\leq 7.17 \pm 0.03$) | (≤ 218) | (≤ 915) | 53 | 223 | *EST | 10556-12-4 |
|  | ($\leq 7.08 \pm 0.03$) | (≤ 216) | (≤ 906) | 53 | 223 | *EST | 56594-76-4 |
|  | 7.95 IP from 83KLA/KOV. | (203) | (853) | 20 | 86 | *EST | 939-27-5 |
|  | ($\leq 7.86 \pm 0.03$) | (≤ 201) | (≤ 840) | 20 | 82 | 69STU/WES | 575-41-7 |
|  | ($\leq 7.85 \pm 0.03$) | (≤ 201) | (≤ 839) | 20 | 82 | 69STU/WES | 571-61-9 |
|  | (7.5) IP is onset of photoelectron band (81GUT/BES). | (199) | (832) | 26.0 \pm 0.2 | 108.7 \pm 1 | 77PED/RYL | 569-41-5 |
|  | ($\leq 7.89 \pm 0.03$) | (≤ 202) | (≤ 845) | 20 | 84 | 69STU/WES | 581-40-8 |

Table 1. Positive Ion Table - Continued

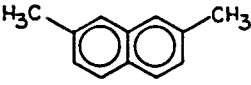
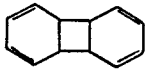
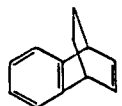
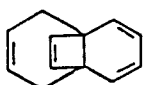
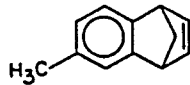
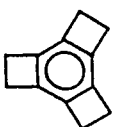
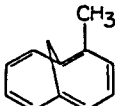
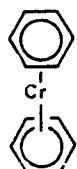
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|-----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{12}^+$  | ($\leq 7.89 \pm 0.03$) | (≤ 202) | (≤ 844) | 20 | 83 | 69STU/WES | 582-16-1 |
|  | (7.7) IP is onset of photoelectron band (81GLE/GUB). | (269) | (1125) | 91 | 382 | *EST | 21657-71-6 |
|  | (8.2) IP is onset of photoelectron band (82HAS/NEU). | (235) | (984) | 46 | 193 | *EST | 7322-46-5 |
|  | (≤ 8.0) | (≤ 257) | (≤ 1076) | 73 | 304 | *EST | 38310-32-6 |
|  | ($\leq 8.12 \pm 0.05$) | (≤ 247) | (≤ 1033) | 60 | 250 | *EST | 4897-73-8 |
|  | (8.15 ± 0.05) IP from 81HEI/KOV. | (312) | (1306) | 124 | 520 | *EST | 60323-52-6 |
|  | (7.5) IP from 84AND/CER. | (239) | (1002) | 66 | 278 | *EST | 58790-01-5 |
| $\text{C}_{12}\text{H}_{12}\text{Cr}^+$  | 5.40 IP from 82CAB/COW. | 177 | 741 | 53 ± 2 | 220 ± 8 | 77PED/RYL | 1271-54-1 |

Table 1. Positive Ion Table - Continued

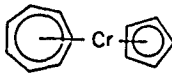
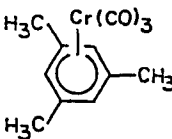
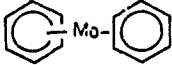
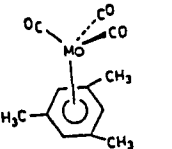
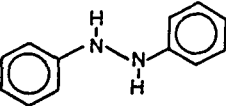
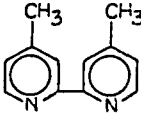
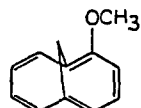
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₂H₁₂Cr⁺  | (5.4) IP is onset of photoelectron band (85DAV/GAR). | (260) | (1087) | 135 | 566 | *EST | 12093-81-1 |
| C₁₂H₁₂CrO₃⁺  | (6.8) IP is onset of photoelectron band. | (45) | (190) | -111±2 | -466±10 | 77PED/RYL | 12129-67-8 |
| C₁₂H₁₂Mo⁺  | (≤5.52±0.05) | (≤223) | (≤935) | 96±5 | 402±20 | 77PED/RYL | 12129-68-9 |
| C₁₂H₁₂MoO₃⁺  | (7.0) IP is onset of photoelectron band. | (60) | (251) | -101±3 | -424±13 | 82PIL/SKI | 12089-15-5 |
| C₁₂H₁₂N₂⁺  | | | | 78 | 326 | 69BEN/CRU | 122-66-7 |
|  | (8.2) IP is onset of photoelectron band (83DOB/HIL). | (241) | (1008) | 52 | 217 | *EST | 1134-35-6 |
| C₁₂H₁₂O⁺  | (7.0) IP is onset of photoelectron band (84AND/CER). | (186) | (778) | 25 | 103 | *EST | |

Table 1. Positive Ion Table - Continued

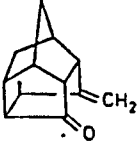
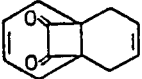
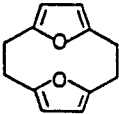
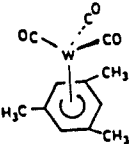
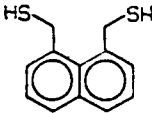
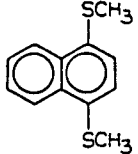
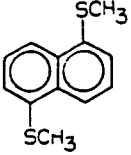
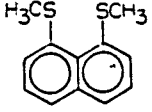
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|---------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{12}\text{O}^+$ | ≤ 8.95 IP from 84MAR/KAY. | (≤ 206) | (≤ 864) | 0 | 0 | *EST | |
|  | | | | | | | |
| $\text{C}_{12}\text{H}_{12}\text{O}_2^+$ | (8.3) IP is onset of photoelectron band. | (185) | (775) | -6 | -26 | *EST | 21377-44-6 |
|  | | | | | | | |
|  | (7.60) IP is onset of photoelectron band. | (148) | (620) | -27 | -113 | *EST | 73650-68-7 |
| $\text{C}_{12}\text{H}_{12}\text{O}_3\text{W}^+$ | (7.0) IP is onset of photoelectron band. | (74) | (309) | -87 ± 4 | -366 ± 15 | 84ALT/CON2 | 12129-69-0 |
|  | | | | | | | |
| $\text{C}_{12}\text{H}_{12}\text{S}_2^+$ | (7.7) IP is onset of photoelectron band (81GUT/BES). | (225) | (940) | 47 | 197 | *EST | 60948-99-4 |
|  | | | | | | | |
|  | 7.4 IP is onset of photoelectron band. | (214) | (895) | 43 | 181 | *EST | 10075-73-7 |
|  | (7.3) IP is onset of photoelectron band. | (212) | (885) | 43 | 181 | *EST | 10075-74-8 |
|  | (7.2) IP is onset of photoelectron band | (215) | (901) | 49 | 206 | *EST | 7343-31-9 |

Table 1. Positive Ion Table - Continued

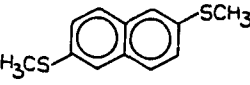
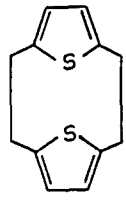
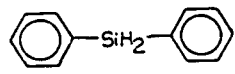
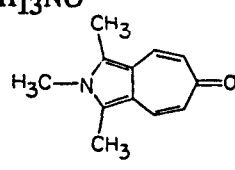
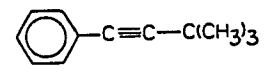
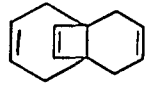
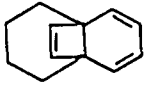
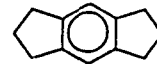
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_{12}H_{12}S_2^+$ | | | | | | | |
|  | 7.1 IP is onset of photoelectron band. | (207) | (866) | 43 | 181 | *EST | 10075-77-1 |
|  | (7.95) | (258) | (1079) | 75 | 312 | *EST | 73650-69-8 |
| $C_{12}H_{12}Si^+$ | | | | | | | |
|  | (8.8) IP is onset of photoelectron band. | (248) | (1037) | 45 | 188 | *EST | 775-12-2 |
| $C_{12}H_{13}NO^+$ | | | | | | | |
|  | (7.3) IP is onset of photoelectron band (84GLE/BIS). | (176) | (734) | 7 | 30 | *EST | |
| $C_{12}H_{14}^+$ | | | | | | | |
|  | (8.32±0.08) IP from 81ELB/LIE. See also: 85ORL/BOG. | (244) | (1020) | 52 | 217 | *EST | 4250-82-2 |
|  | (≤8.7) | (≤247) | (≤1032) | 46 | 193 | *EST | 20295-17-4 |
|  | (≤8.0) | (≤229) | (≤958) | 44 | 186 | *EST | 24139-33-1 |
|  | (7.94) | (193) | (809) | 10 | 43 | *EST | 495-52-3 |

Table 1. Positive Ion Table - Continued

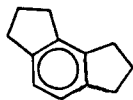
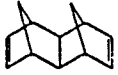
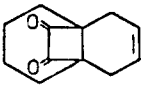
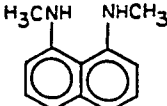
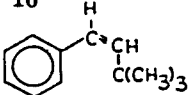
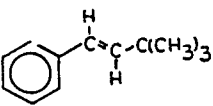
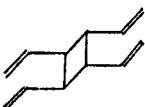

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{14}^+$  | (8.09) | (200) | (836) | 13 | 55 | *EST | 1076-17-1 |
|  | (8.2) IP from 81PAD/PAT. | (255) | (1068) | 66 | 277 | *EST | |
| $\text{C}_{12}\text{H}_{14}\text{O}_2^+$  | (8.1) IP is onset of photoelectron band. | (152) | (638) | -34 | -144 | *EST | 21377-45-7 |
| $\text{C}_{12}\text{H}_{15}\text{N}_2^+$  | | 181 | 757 | | | | |
| | | From proton affinity of N,N'-dimethyl-1,8-naphthalenediamine (RN 20734-56-9). PA = 230.0 kcal/mol, 962. kJ/mol. | | | | | |
| $\text{C}_{12}\text{H}_{16}^+$  | 8.29±0.04 See also: 81KOB/ARA. | (207) | (866) | 16±1 | 66±4 | 77PED/RYL | 3740-05-4 |
|  | 7.80±0.04 See also: 81KOB/ARA. | (188) | (786) | 8±2 | 33±9 | 77PED/RYL | 3846-66-0 |
|  | (≤9.16) IP from 83GLE/HAI2 | (≤300) | (≤1255) | 88.7 | 371.1 | 83GLE/HAI2 | 82865-42-7 |
|  | (≤9.02) IP from 83GLE/HAI2 | (≤294) | (≤1231) | 86.2 | 360.7 | 83GLE/HAI2 | 87753-95-5 |

Table 1. Positive Ion Table - Continued

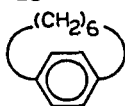
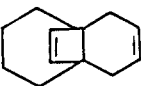
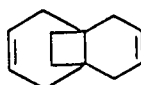
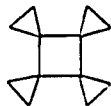
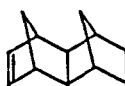
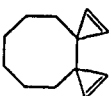
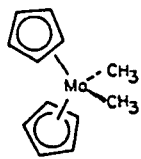
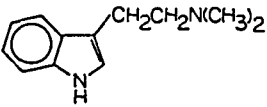
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|-----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{16}^+$  | (7.5) IP is onset of photoelectron band. | (196) | (821) | 23 | 97 | *EST | 53011-74-8 |
|  | (≤ 8.9) | (≤ 223) | (≤ 934) | 18 | 75 | *EST | 24139-32-0 |
|  | (8.7) IP is onset of photoelectron band. | (278) | (1161) | 77 | 322 | *EST | 5103-78-6 |
|  | (≤ 8.22) | (≤ 326) | (≤ 1365) | 137 | 572 | *EST | 24375-17-5 |
|  | (8.6) IP from 81PAD/PAT. | (231) | (968) | 33 | 138 | *EST | 262-30-6 |
|  | (8.2) IP from 82SPA/KOR. | (331) | (1385) | 142 | 594 | *EST | 64371-17-1 |
| $\text{C}_{12}\text{H}_{16}\text{Mo}^+$  | ($\leq 6.1 \pm 0.1$) | (≤ 226) | (≤ 943) | 85 ± 1 | 354 ± 6 | 82PIL/SKI | 39333-52-3 |
| $\text{C}_{12}\text{H}_{16}\text{N}_2^+$  | (7.3) IP is onset of photoelectron band. | (201) | (842) | 33 | 138 | *EST | 61-50-7 |

Table 1. Positive Ion Table - Continued

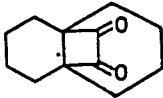
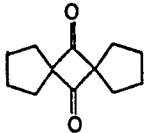
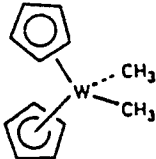
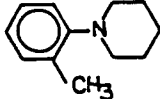
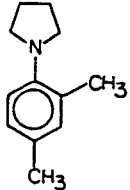
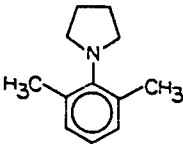
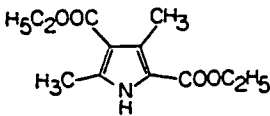
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₂H₁₆O₂⁺ | | | | | | | |
|  | 8.5 IP is onset of photoelectron band. | (133) | (558) | -63 | -262 | *EST | 21377-46-8 |
|  | (8.9±0.01) IP from 84OLI/FLE. | (161) | (675) | -44 | -184 | *EST | 5011-61-0 |
| C₁₂H₁₆W⁺ | | | | | | | |
|  | (5.8) IP is onset of photoelectron band. | (220) | (919) | 86±1 | 359±6 | 82PIL/SKI | 39333-53-4 |
| C₁₂H₁₇N⁺ | | | | | | | |
|  | (7.1) IP is onset of photoelectron band (82ROZ/HOU2). | (169) | (706) | 5 | 21 | *EST | 7250-70-6 |
|  | (≤7.60) IP from 82ROZ/HOU2. | (≤219) | (≤918) | 44 | 185 | *EST | 81506-12-9 |
|  | (7.0) IP is onset of photoelectron band (82ROZ/HOU2). | (199) | (834) | 38 | 159 | *EST | 64175-53-7 |
| C₁₂H₁₇NO₄⁺ | | | | | | | |
|  | (≤8.15) IP from 81CAU/GIA. | (≤4) | (≤15) | -184 | -771 | *EST | 2436-79-5 |

Table 1. Positive Ion Table - Continued

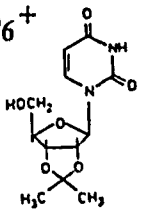
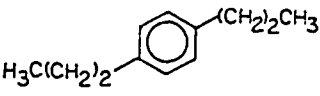
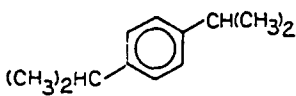
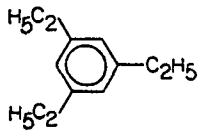
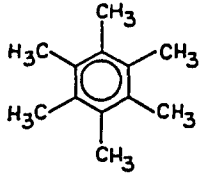
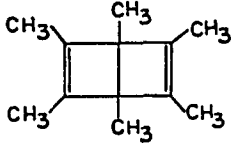
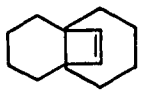
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{17}\text{N}_2\text{O}_6^+$ |  | | (-60) | (-251) | | | | |
| | | | From proton affinity of 2',3'-O'-isopropylideneuridine (RN 362-43-6). PA = (208) kcal/mol, (870) kJ/mol. | | | | | |
| $\text{C}_{12}\text{H}_{18}^+$ | $n\text{-C}_4\text{H}_9\text{C}\equiv\text{CC}\equiv\text{C}(n\text{-C}_4\text{H}_9)$ | (8.67) | (258) | (1077) | 58 | 241 | 77PED/RYL | 1120-29-2 |
| | (tert-C ₄ H ₉ C≡C) ₂ | (8.61±0.02) | (249) | (1040) | 50±1 | 209±5 | 77KUP/SHI | 6130-98-9 |
| |  | (≤8.31) | (≤176) | (≤736) | -16 | -66 | *EST | 4815-57-0 |
| | | IP from 80GLE/HOP. | | | | | | |
| |  | (8.35) | (175) | (732) | -18 | -75 | *EST | 100-18-5 |
| |  | (8.32) | (173) | (724) | -19 | -79 | *EST | 102-25-0 |
| | | IP from 84HOW/GON. | | | | | | |
| |  | 7.85 | 160 | 670 | -21±0.7 | -87±3 | 77PED/RYL | 87-85-4 |
| | | See also: 84HOW/GON. | | | | | | |
| |  | (≤7.83) | (≤219) | (≤917) | 39 | 162 | 78GRE/LIE | 7641-77-2 |
| |  | (≤9.05) | (≤198) | (≤830) | -10 | -43 | *EST | 38992-78-8 |

Table 1. Positive Ion Table - Continued



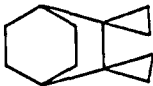
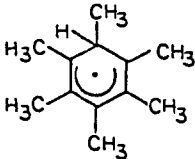
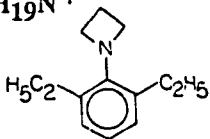

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|-----------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₂H₁₈⁺ | | | | | | | |
|  | ($\leq 9.50 \pm 0.03$) | (≤ 219) | (≤ 916) | -0.2 \pm 1 | -1 \pm 5 | 73ENG/AND2 | 53862-33-2 |
|  | ($\leq 9.57 \pm 0.03$) | (≤ 232) | (≤ 968) | 11 \pm 2 | 45 \pm 8 | 73ENG/AND2 | 15914-95-1 |
|  | (8.3) IP is onset of photoelectron band. | (232) | (972) | 41 | 171 | *EST | 40827-30-3 |
| C₁₂H₁₈Hg⁺ ((CH ₃) ₃ CC=C) ₂ Hg | | | | | | | |
| | (9.03) | (285) | (1194) | 77 | 323 | *EST | 73838-84-3 |
| | IP is onset of photoelectron band (81FUR/PIA). | | | | | | |
| C₁₂H₁₈O₄⁺ (CH ₃) ₃ C[C(=O)] ₄ C(CH ₃) ₃ | | | | | | | |
| | (8.5) | (25) | (105) | -171 | -715 | *EST | 19909-70-7 |
| | IP from 85GLE/DOB. | | | | | | |
| C₁₂H₁₉⁺ | | | | | | | |
|  | | 138 | 576 | | | | |
| | From proton affinity of hexamethylbenzene (RN 87-85-4). PA = 207.3 kcal/mol, 867. kJ/mol. | | | | | | |
| C₁₂H₁₉N⁺ | | | | | | | |
|  | (≤ 7.82) | (≤ 241) | (≤ 1007) | 60 | 253 | *EST | 81506-11-8 |
| | IP from 82ROZ/HOU2. | | | | | | |
|  | (6.90) | (158) | (661) | -1.2 | -4.9 | 85ORL/MIS | 2909-79-7 |
| | IP from 86ORL/MIS. See also: 85BAI/MIS2. | | | | | | |

Table 1. Positive Ion Table - Continued

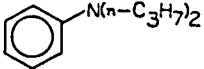
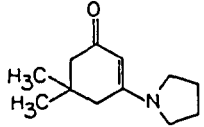
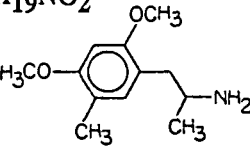
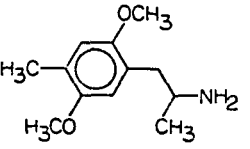
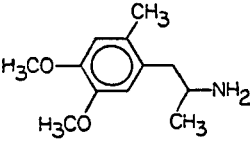
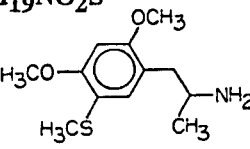
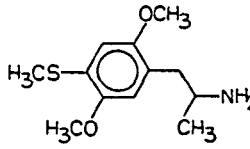
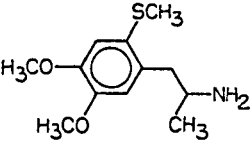
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{19}\text{N}^+$ |  | 6.93 | (161) | (672) | 1 | 3 | *EST | 2217-07-4 |
| | | IP from charge transfer equilibrium constant determinations; reference standard: IP $\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2 = 7.12\text{eV}$. (85LIA/JAC). | | | | | | |
| $\text{C}_{12}\text{H}_{19}\text{NO}^+$ |  | (≤ 7.54) | (≤ 138) | (≤ 579) | -35 | -148 | *EST | 3357-16-2 |
| | | IP from 82PFI/GER. | | | | | | |
| $\text{C}_{12}\text{H}_{19}\text{NO}_2^+$ |  | (7.2) | (87) | (366) | -79 | -329 | *EST | 79440-50-9 |
| | | IP is onset of photoelectron band (81DOM/EAT). | | | | | | |
| |  | (6.8) | (80) | (335) | -77 | -321 | *EST | 26011-50-7 |
| | | IP is onset of photoelectron band (81DOM/EAT, 82LEV/LIA). | | | | | | |
| |  | (6.9) | (85) | (356) | -74 | -310 | *EST | 56966-33-7 |
| | | IP is onset of photoelectron band (81DOM/EAT). | | | | | | |
| $\text{C}_{12}\text{H}_{19}\text{NO}_2\text{S}^+$ |  | (6.9) | (92) | (385) | -67 | -281 | *EST | 79440-52-1 |
| | | IP is onset of photoelectron band (81DOM/EAT). | | | | | | |
| |  | (6.8) | (91) | (383) | -65 | -273 | *EST | 61638-07-1 |
| | | IP is onset of photoelectron band (81DOM/EAT). | | | | | | |
| |  | (6.9) | (94) | (393) | -65 | -273 | *EST | |
| | | IP is onset of photoelectron band (81DOM/EAT). | | | | | | |

Table 1. Positive Ion Table - Continued

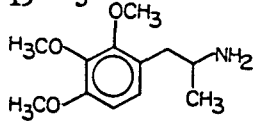
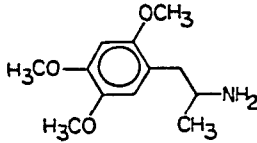
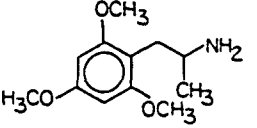
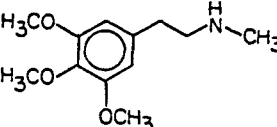
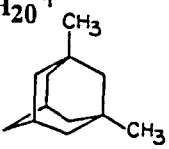
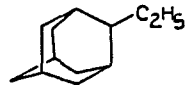
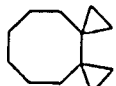
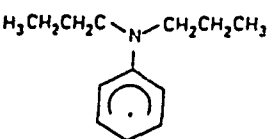
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------------------------|--------------|--|--------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{19}\text{NO}_3^+$ |  | $(\leq 8.09 \pm 0.06)$ | (≤ 88) | (≤ 370) | -98 | -411 | *EST | 22199-12-8 |
| |  | (7.0) | (54) | (224) | -108 | -451 | *EST | 22199-15-1 |
| | | IP is onset of photoelectron band (81DOM/EAT, 82LEV/LIA). | | | | | | |
| |  | $(\leq 7.76 \pm 0.06)$ | (≤ 67) | (≤ 279) | -112 | -470 | *EST | 22199-16-2 |
| |  | $(\leq 8.44 \pm 0.40)$ | (≤ 101) | (≤ 421) | -94 | -393 | *EST | 4838-96-4 |
| $\text{C}_{12}\text{H}_{20}^+$ |  | (9.15) | (159) | (664) | -52 ± 0.7 | -219 ± 3 | 77STE/WAT | 702-79-4 |
| |  | (9.2) | (250) | (1049) | 38 ± 0.5 | 161 ± 2 | 81GOD/SCH2 | 14451-87-7 |
| |  | (8.6) | (233) | (976) | 35 | 146 | *EST | 64601-40-7 |
| | | IP is onset of photoelectron band (82SPA/GLE). | | | | | | |
| $\text{C}_{12}\text{H}_{20}\text{N}^+$ |  | | 138 | 578 | From proton affinity of $\text{C}_6\text{H}_5\text{N}(\text{n-C}_3\text{H}_7)_2$ (RN 2217-07-4). PA = 228.6 kcal/mol, 956. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

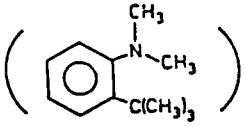
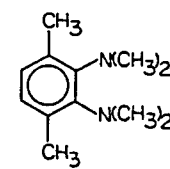
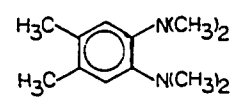
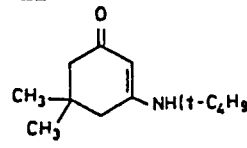
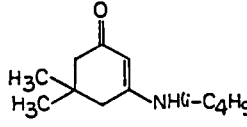
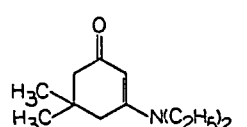
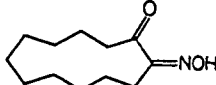
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------|----|---|--------|------------------------------|--------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{20}\text{N}^+$ | | | 152 | 634 | | | | |
|  | | | From proton affinity of 2-(t-C ₄ H ₉)C ₆ H ₄ N(CH ₃) ₂ (RN 22025-87-2). 229.3 kcal/mol, 959. kJ/mol. | | | | | |
| $\text{C}_{12}\text{H}_{20}\text{N}_2^+$ | | | | | | | | |
|  | (7.3) | | (196) | (822) | 28 | 118 | *EST | 66102-30-5 |
| | | | IP is onset of photoelectron band (81NEL/GRE). | | | | | |
|  | (6.4) | | (168) | (704) | 21 | 86 | *EST | 54929-05-4 |
| | | | IP is onset of photoelectron band (81NEL/GRE). | | | | | |
| $\text{C}_{12}\text{H}_{21}\text{N}^+$ | | | | | | | | |
| (CH ₂ =C(CH ₃)CH ₂) ₃ N | (7.8) | | (208) | (869) | 28 | 116 | *EST | |
| | | | IP from 79AUE/BOW. | | | | | |
| $\text{C}_{12}\text{H}_{21}\text{NO}^+$ | | | | | | | | |
|  | (≤7.69) | | (≤111) | (≤464) | -66 | -278 | *EST | 27336-61-4 |
| | | | IP from 82PFI/GER. | | | | | |
|  | (≤7.98) | | (≤122) | (≤513) | -61 | -257 | *EST | 82663-49-8 |
| | | | IP from 82PFI/GER. | | | | | |
|  | (7.3) | | (116) | (486) | -52 | -218 | *EST | 65115-73-3 |
| | | | IP is onset of photoelectron band (82PFI/GER). | | | | | |
| $\text{C}_{12}\text{H}_{21}\text{NO}_2^+$ | | | | | | | | |
|  | (8.99±0.03) | | (132) | (552) | -75 | -315 | *EST | 4422-06-4 |
| | | | IP from 79GOL/KUL. | | | | | |

Table 1. Positive Ion Table - Continued

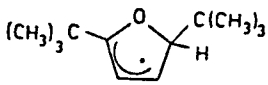
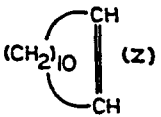
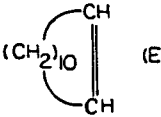
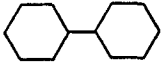
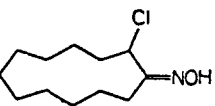
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|---------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₂H₂₁O⁺ | | | | | | | |
|  | | 87 | 366 | | | | |
| | | From the proton affinity of 2,5-Di- <i>t</i> -butylfuran (RN 4789-40-6) (85HOU/ROL). PA = 213.4 kcal/mol, 893. kJ/mol. | | | | | |
| C₁₂H₂₂⁺ | | | | | | | |
| 1-C ₁₂ H ₂₂ | (9.90±0.02) | (228) | (956) | 0.1 | 0.4 | *EST | 765-03-7 |
| 2-C ₁₂ H ₂₂ | (9.29±0.02) | (210) | (878) | -4 | -18 | *EST | 629-49-2 |
| 3-C ₁₂ H ₂₂ | (9.17±0.02) | (207) | (868) | -4 | -17 | *EST | 6790-27-8 |
| 4-C ₁₂ H ₂₂ | (9.14±0.03) | (207) | (865) | -4 | -17 | *EST | 22058-01-1 |
| 5-C ₁₂ H ₂₂ | (9.09±0.03) | (206) | (860) | -4 | -17 | *EST | 19780-12-2 |
| CH ₂ =C(<i>t</i> -C ₄ H ₉)C(<i>t</i> -C ₄ H ₉)=CH ₂ | (8.5) | (179) | (750) | -17 | -70 | *EST | 3378-20-9 |
| | | IP is onset of photoelectron band (84HON/ZHO). | | | | | |
| trans, trans-((<i>tert</i> -C ₄ H ₉)CH=CH) ₂ | (8.23±0.04) | (168) | (704) | -22 | -90 | *EST | 22430-49-5 |
|  | (8.78±0.15) | (173) | (727) | -29 | -120 | 76JEN | 1129-89-1 |
|  | (8.74±0.15) | (173) | (725) | -28 | -118 | 76JEN | 1486-75-5 |
|  | (9.41) | (164.8) | (689.5) | -52.2±.7 | -218.4±3.1 | 78MON/ROS | 92-51-3 |
| C₁₂H₂₂ClNO⁺ | | | | | | | |
|  | (9.18±0.03) | (160) | (668) | -52 | -218 | *EST | 4806-74-0 |
| | | IP from 79GOL/KUL. | | | | | |

Table 1. Positive Ion Table - Continued

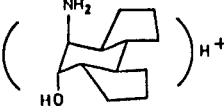
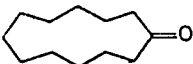
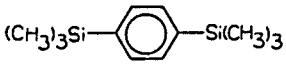
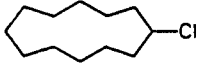
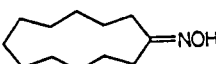
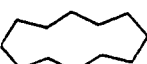
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{22}\text{N}^+$ | $(\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH}_2)_3\text{NH}$ | | (163) | (684) | | | | |
| | | | From proton affinity of $(\text{CH}_2 = \text{C}(\text{CH}_3)\text{CH}_2)_3\text{N}$. PA = (230.7) kcal/mol, (965.) kJ/mol. | | | | | |
| $\text{C}_{12}\text{H}_{22}\text{NO}^+$ |  | | 83 | 347 | | | | |
| | | | From proton affinity of 3-amino-tricyclo[7.3.0.0 ^{4,8}] dodecan-2-ol. PA = 220.0 kcal/mol, 920. kJ/mol. | | | | | |
| $\text{C}_{12}\text{H}_{22}\text{O}^+$ |  | (8.96±0.03) | (123) | (514) | -84 | -350 | *EST | 830-13-7 |
| | | IP from 79GOL/KUL. | | | | | | |
| $\text{C}_{12}\text{H}_{22}\text{Si}_2^+$ |  | (8.45) | (132) | (554) | -62 | -261 | *EST | 13183-70-5 |
| | | IP is onset of photoelectron band (82TRA/RED). | | | | | | |
| $\text{C}_{12}\text{H}_{23}\text{Cl}^+$ |  | (9.04±0.03) | (≤143) | (≤598) | -65 | -274 | *EST | 34039-83-3 |
| | | IP from 79GOL/KUL. | | | | | | |
| $\text{C}_{12}\text{H}_{23}\text{NO}^+$ |  | (8.84±0.03) | (154) | (643) | -50 | -210 | *EST | 946-89-4 |
| | | IP from 79GOL/KUL. | | | | | | |
| $\text{C}_{12}\text{H}_{24}^+$ | $(Z)\text{-(CH}_3)_3\text{CCH}_2\text{C(CH}_3)=\text{CHC(CH}_3)_3$ | (8.35±0.01) | (142) | (594) | -50 | -211 | *EST | 27656-50-4 |
| |  | (9.72±0.03) | (169) | (707) | -55±0.5 | -230±2 | 77PED/RYL | 294-62-2 |
| | | IP from 79GOL/KUL. | | | | | | |

Table 1. Positive Ion Table - Continued


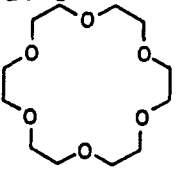
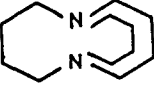
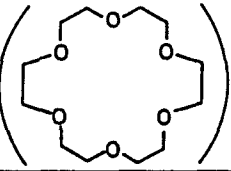
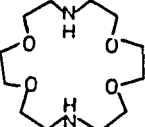
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------------------------|---------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{24}\text{O}^+$ |  | (9.26±0.03) | (119) | (499) | -94 | -394 | *EST | 1724-39-6 |
| | | IP from 79GOL/KUL. | | | | | | |
| $\text{C}_{12}\text{H}_{24}\text{O}_6^+$ |  | 8.9 | (-22) | (-91) | -227 | -950 | *EST | 17455-13-9 |
| | | IP is onset of photoelectron band (83BAK/ARM, 82LEV/LIA). | | | | | | |
| $\text{C}_{12}\text{H}_{25}\text{N}_2^+$ |  | | 117 | 490 | | | | |
| | | From proton affinity of 1,6-diazabicyclo[4.4.4]tetradecane (RN 71058-67-8). PA = 226.0 kcal/mol, 946. kJ/mol. | | | | | | |
| $\text{C}_{12}\text{H}_{25}\text{O}_6^+$ |  | | -91 | -382 | | | | |
| | | From proton affinity of 1,4,7,10,13,16-hexaoxacyclooctadecane (18-Crown-6) (RN 17455-13-9). PA = 230. kcal/mol, 962. kJ/mol. | | | | | | |
| $\text{C}_{12}\text{H}_{26}\text{N}_2\text{O}_4^+$ |  | (≤8.4) | (≤52) | (≤218) | -141 | -592 | *EST | 23978-55-4 |
| | | IP from 83BAK/ARM. | | | | | | |
| $\text{C}_{12}\text{H}_{27}\text{BO}_3^+$ | $(n\text{-C}_4\text{H}_9\text{O})_3\text{B}$ | (≤10.72±0.74) | (≤-27) | (≤-113) | -274±1 | -1147±4 | 77PED/RYL | 688-74-4 |
| $\text{C}_{12}\text{H}_{27}\text{N}^+$ | $(n\text{-C}_4\text{H}_9)_3\text{N}$ | (7.4) | (118) | (494) | -53±0.2 | -222±1 | *EST | 102-82-9 |
| | | IP is onset of photoelectron band (82ELB/DIE), giving $\Delta_f H(\text{Ion})$ in good agreement with value predicted from hydrogen affinity of tertiary amines. | | | | | | |
| $\text{C}_{12}\text{H}_{27}\text{P}^+$ | $(n\text{-C}_4\text{H}_9)_3\text{P}$ | (7.5) | (160) | (668) | -13±8 | -56±35 | 77PED/RYL | 998-40-3 |
| | | IP is onset of photoelectron band. | | | | | | |
| $\text{C}_{12}\text{H}_{28}\text{N}^+$ | $(n\text{-C}_4\text{H}_9)_3\text{NH}$ | | 77 | 323 | | | | |
| | | From proton affinity of $(n\text{-C}_4\text{H}_9)_3\text{N}$ (RN 102-82-9). PA = 235.4 kcal/mol, 985 kJ/mol (85BOL/HOU). | | | | | | |

Table 1. Positive Ion Table - Continued

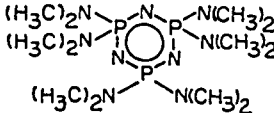
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------------------------------|--------------------------|----------------|------------------------------|---------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{H}_{28}\text{N}_2^+$ (n-C ₃ H ₇) ₂ NN(n-C ₃ H ₇) ₂ | (≤ 7.74) | (≤ 197) | (≤ 825) | 19 | 78 | *EST | 60678-69-5 |
| Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL. | | | | | | | |
| (n-C ₄ H ₉) ₂ NN(C ₂ H ₅) ₂ | (≤ 7.77) | (≤ 198) | (≤ 828) | 19 | 78 | *EST | 60678-68-4 |
| Reported values of IP's of hydrazines determined by threshold measurements are usually significantly higher than the adiabatic value because of the large geometry change associated with ionization. See also: 84NEL. | | | | | | | |
| $\text{C}_{12}\text{H}_{28}\text{Sn}^+$ (C ₃ H ₇) ₄ Sn | (≤ 8.82) | (≤ 168) | (≤ 705) | -35 \pm 1 | -146 \pm 6 | 77PED/RYL | 2176-98-9 |
| (iso-C ₃ H ₇) ₄ Sn | (≤ 8.46) | (≤ 166) | (≤ 693) | -29 \pm 2 | -123 \pm 7 | 77PED/RYL | 2949-42-0 |
| $\text{C}_{12}\text{H}_{30}\text{Ge}_2^+$ ((C ₂ H ₅) ₃ Ge) ₂ | 7.48 \pm 0.01 | 90 | 375 | -83 \pm 2 | -347 \pm 8 | 80TEL/RAB | 993-62-4 |
| $\text{C}_{12}\text{H}_{30}\text{N}_3\text{P}^+$ P(N(C ₂ H ₅) ₂) ₃ | (≤ 7.19) IP from 82WOR/HAR. | (≤ 112) | (≤ 468) | -54 \pm 2 | -226 \pm 10 | 77PED/RYL | 2283-11-6 |
| $\text{C}_{12}\text{H}_{30}\text{Sn}_2^+$ [(C ₂ H ₅) ₃ Sn] ₂ | (6.60 \pm 0.02) | (115) | (482) | -37 \pm 2 | -155 \pm 10 | 77PED/RYL | 993-63-5 |
| $\text{C}_{12}\text{H}_{31}\text{N}_3\text{OP}^+$ HOP(N(C ₂ H ₅) ₂) ₃ | | 9 | 37 | | | | |
| From proton affinity of OP(N(C ₂ H ₅) ₂) ₃ (RN 2622-07-3) (85BOL/HOU). PA = 230.0 kcal/mol, 962. kJ/mol. | | | | | | | |
| $\text{C}_{12}\text{H}_{36}\text{Mo}_2\text{N}_6^+$ Mo ₂ ((CH ₃) ₂ N) ₆ | (6.74) | (125) | (522) | -31 \pm 3 | -128 \pm 13 | 79ADE/CAV | 51956-20-8 |
| $\text{C}_{12}\text{H}_{36}\text{N}_6\text{W}^+$ W(N(CH ₃) ₂) ₆ | (6.3) | (209) | (876) | 64 \pm 3 | 268 \pm 14 | 79ADE/CAV | 54935-70-5 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_{12}\text{H}_{36}\text{N}_9\text{P}_3^+$  | (7.85 \pm 0.05) | (76) | (318) | -105 \pm 3 | -439 \pm 13 | 80TEL/RAB | 974-68-5 |
| $\text{C}_{12}\text{H}_{36}\text{Si}_5^+$ Si(Si(CH ₃) ₃) ₄ | (7.41 \pm 0.01) | (37) | (156) | -134 \pm 10 | -559 \pm 40 | 77PED/RYL | 4098-98-0 |

Table 1. Positive Ion Table - Continued

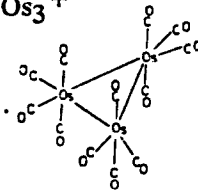
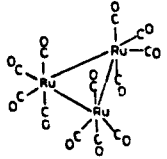
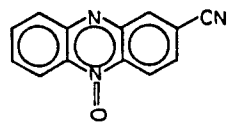
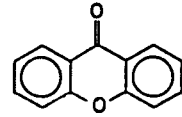
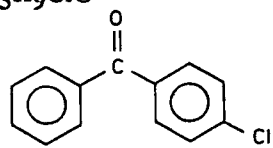
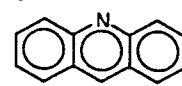
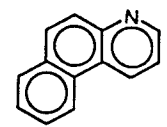
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{12}\text{O}_{12}\text{Os}_3^+$  | (7.6±0.3) | (-218) | (-911) | -393±7 | -1644±28 | 80CON | 15696-40-9 |
| IP is onset of photoelectron band. See also: 81GRE/MIN, 82SHE/HAL. | | | | | | | |
| $\text{C}_{12}\text{O}_{12}\text{Ru}_3^+$  | (7.3) | (-267) | (-1116) | -435±6 | -1820±26 | 77PED/RYL | 15243-33-1 |
| IP is onset of photoelectron band. See also: 81GRE/MIN. | | | | | | | |
| $\text{C}_{13}\text{H}_7\text{N}_3\text{O}^+$  | (8.44) | (290) | (1211) | 95 | 397 | *EST | 59019-84-0 |
| $\text{C}_{13}\text{H}_8\text{O}_2^+$  | (8.42±0.03) | (174) | (727) | -20±2 | -85±7 | 82JOH/KIM | 90-47-1 |
| $\text{C}_{13}\text{H}_9\text{ClO}^+$  | 9.64±0.04 | (229) | (959) | 7±2 | 29±8 | *EST | 134-85-0 |
| $\text{C}_{13}\text{H}_9\text{N}^+$  | 7.8 | 249 | 1044 | 69±0.2 | 291±1 | 81KUD/KUD2 | 260-94-6 |
|  | (8.14±0.02) | (244) | (1019) | 56±2 | 234±7 | 81STE/BAR | 85-02-9 |

Table 1. Positive Ion Table - Continued

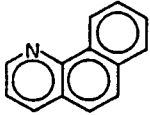
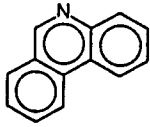
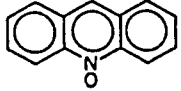
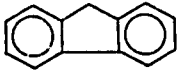
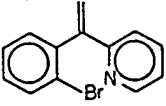
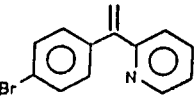
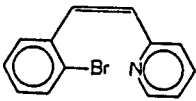
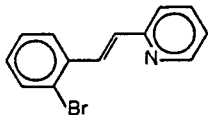
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₉N⁺ | | | | | | | |
|  | (8.04±0.02) | (240) | (1007) | 55±1 | 231±5 | 81STE/BAR | 230-27-3 |
|  | (8.31±0.02) | (250) | (1046) | 58±1 | 244±6 | 81STE/BAR | 229-87-8 |
| C₁₃H₉NO⁺ | | | | | | | |
|  | (7.45±0.02) | (222) | (930) | 50 | 211 | *EST | 10399-73-2 |
| C₁₃H₁₀⁺ | | | | | | | |
|  | 7.89±0.03 | 227 | 948 | 45±0.2 | 187±1 | 81KUD/KUD | 86-73-7 |
| | Value of IP from charge transfer equilibrium constant determinations (80MAU, re-evaluated) is in agreement. | | | | | | |
| C₁₃H₁₀BrN⁺ | | | | | | | |
|  | (8.6) | (274) | (1146) | 76 | 316 | *EST | 74309-56-1 |
| | IP from 80GRU/SCH, 82LEV/LIA. | | | | | | |
|  | (8.62) | (274) | (1148) | 76 | 316 | *EST | |
|  | (8.05) | (262) | (1098) | 77 | 321 | *EST | 76293-40-8 |
| | IP from 80GRU/SCH. | | | | | | |
|  | (8.05) | (258) | (1081) | 73 | 304 | *EST | 77275-12-8 |
| | IP from 80SCH/RAM. | | | | | | |

Table 1. Positive Ion Table - Continued

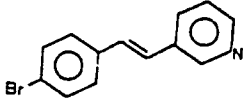
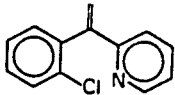
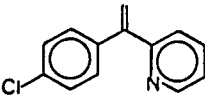
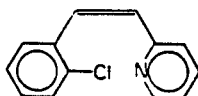
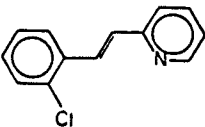
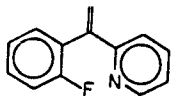
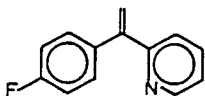
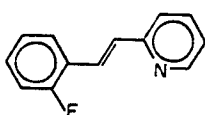
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₁₀BrN⁺ | | | | | | | |
|  | ($\leq 8.15 \pm 0.05$) | (≤ 258) | (≤ 1078) | 70 | 292 | *EST | 5847-71-2 |
| C₁₃H₁₀ClN⁺ | | | | | | | |
|  | (8.6) IP from 80GRU/SCH, 82LEV/LIA. | (262) | (1096) | 64 | 266 | *EST | 74309-55-0 |
|  | (8.58) | (261) | (1094) | 64 | 266 | *EST | |
|  | (8.07) IP from 80GRU/SCH. | (251) | (1051) | 65 | 272 | *EST | 5350-12-9 |
|  | 8.06 \pm 0.01 IP from 80SCH/RAM, 82LEV/LIA. | (247) | (1032) | 61 | 254 | *EST | 6772-77-6 |
| C₁₃H₁₀FN⁺ | | | | | | | |
|  | (8.66) IP from 82LEV/LIA, 80GRU/SCH. | (223) | (931) | 23 | 96 | *EST | 74309-53-8 |
|  | (8.68) | (223) | (933) | 23 | 96 | *EST | |
|  | (≤ 8.1) IP from 80SCH/RAM. | (≤ 207) | (≤ 866) | 20 | 84 | *EST | 77275-10-6 |

Table 1. Positive Ion Table - Continued

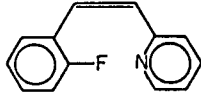
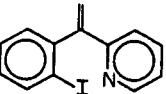
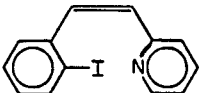
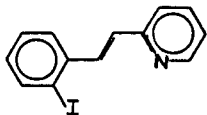
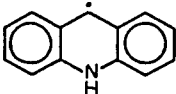
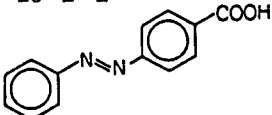
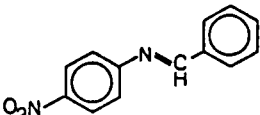
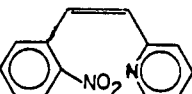
| ION | Neutral | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---------|----------------------|-------|--------------------------|--------|------------------------------|--------|-------------------|---------------------|
| | | eV | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₁₀FN⁺ | | | | | | | | | |
|  | | (8.18) | (213) | (890) | 24 | 101 | *EST | 76293-38-4 | |
| IP from 80GRU/SCH. | | | | | | | | | |
| C₁₃H₁₀IN⁺ | | | | | | | | | |
|  | | (8.3) | (281) | (1175) | 89 | 374 | *EST | 74309-57-2 | |
| IP from 80GRU/SCH, 82LEV/LIA. | | | | | | | | | |
|  | | (7.95) | (274) | (1147) | 91 | 380 | *EST | | |
| IP from 80GRU/SCH. | | | | | | | | | |
|  | | (7.95) | (270) | (1129) | 86 | 362 | *EST | 6772-85-6 | |
| IP from 80SCH/RAM. | | | | | | | | | |
| C₁₃H₁₀N⁺ | | | | | | | | | |
|  | | | 203 | 851 | | | | | |
| From proton affinity of acridine. (RN 260-94-6). PA = 231.9 kcal/mol, 970. kJ/mol. | | | | | | | | | |
| C₁₃H₁₀N₂O₂⁺ | | | | | | | | | |
|  | | (8.3) | (198) | (828) | 6 | 27 | *EST | 37790-20-8 | |
| IP is onset of photoelectron band. | | | | | | | | | |
|  | | (8.4) | (240) | (1005) | 47 | 195 | *EST | 69173-79-1 | |
| IP is onset of photoelectron band. | | | | | | | | | |
|  | | (8.30) | (259) | (1086) | 68 | 285 | *EST | 50385-24-5 | |
| IP from 80GRU/SCH. | | | | | | | | | |

Table 1. Positive Ion Table - Continued

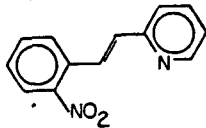
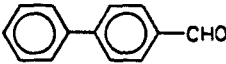
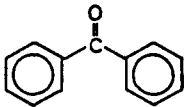
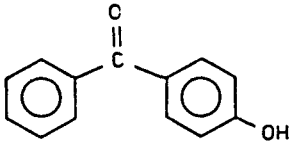
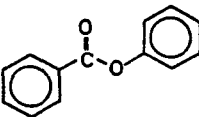
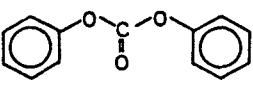
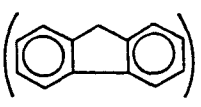
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|--------|--|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{13}\text{H}_{10}\text{N}_2\text{O}_2^+$ |  | (8.3) IP from 80SCH/RAM. | (255) | (1069) | 64 | 268 | *EST | 77340-84-2 |
| $\text{C}_{13}\text{H}_{10}\text{O}^+$ |  | (8.47±0.03) | (210) | (879) | 15±0.7 | 62±3 | *EST | 3218-36-8 |
| |  | 9.05±0.05 IP from 78CEN/FRA, 82LEV/LIA. | 221 | 923 | 12±0.7 | 50±3 | 78SAB/LAF3 | 119-61-9 |
| $\text{C}_{13}\text{H}_{10}\text{O}_2^+$ |  | (8.3) IP is onset of photoelectron band. | (160) | (670) | -31.2±2 | -131±8 | *EST | 1137-42-4 |
| |  | 8.99±0.02 | 173 | 724 | -34±0.7 | -143±3 | 77PED/RYL | 93-99-2 |
| $\text{C}_{13}\text{H}_{10}\text{O}_3^+$ |  | (9.01±0.05) | (134) | (558) | -74±2 | -311±9 | 77PED/RYL | 102-09-0 |
| $\text{C}_{13}\text{H}_{11}^+$ |  | H^+ | 210 | 880 | From proton affinity of fluorene (RN 86-73-7). PA = 200.0 kcal/mol, 837. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

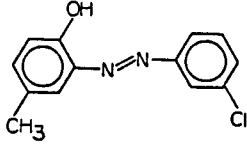
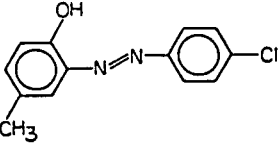
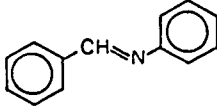
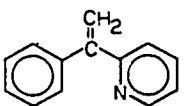
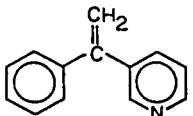
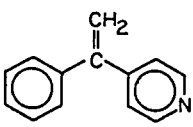
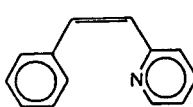
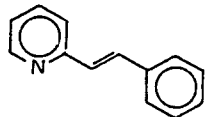
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₁₁ClN₂O⁺ | | | | | | | |
|  | (7.7) | (217) | (910) | 40 | 167 | *EST | 19116-23-5 |
| IP is onset of photoelectron band (81MIL/MIL). | | | | | | | |
|  | (7.7) | (217) | (910) | 40 | 167 | *EST | 2491-56-7 |
| IP is onset of photoelectron band (81MIL/MIL). | | | | | | | |
| C₁₃H₁₁N⁺ | | | | | | | |
|  | 7.9 | (232) | (972) | 50 | 210 | *EST | 538-51-2 |
| IP is onset of photoelectron band. | | | | | | | |
|  | (8.65) | (270) | (1130) | 71 | 295 | *EST | 15260-65-8 |
| See also: 80GRU/SCH. | | | | | | | |
|  | (8.73) | (269) | (1125) | 68 | 283 | *EST | 74309-58-3 |
|  | (8.90) | (277) | (1159) | 72 | 300 | *EST | 54813-56-8 |
|  | (8.15) | (264) | (1105) | 76 | 319 | *EST | 1519-59-1 |
| IP from 80GRU/SCH. | | | | | | | |
|  | (≤7.99±0.05) | (≤252) | (≤1054) | 68 | 283 | *EST | 538-49-8 |
| See also: 80SCH/RAM. | | | | | | | |

Table 1. Positive Ion Table - Continued

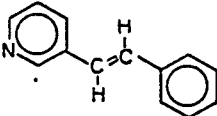
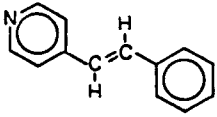
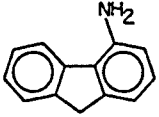
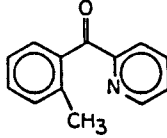
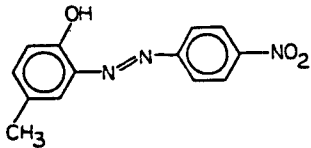
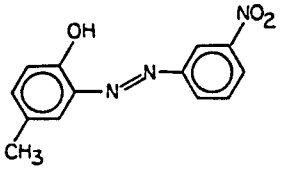
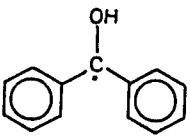
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|-----------------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{13}\text{H}_{11}\text{N}^+$ | | | | | | | |
|  | ($\leq 8.10 \pm 0.05$) See also: 80SCH/RAM. | (≤ 256) | (≤ 1072) | 69 | 290 | *EST | 5097-91-6 |
|  | ($\leq 8.34 \pm 0.05$) See also: 80SCH/RAM. | (≤ 261) | (≤ 1093) | 69 | 288 | *EST | 5097-93-8 |
|  | (7.25) IP is onset of photoelectron band (84GLE/SCH). | (216) | (904) | 49 | 205 | *EST | 7083-63-8 |
| $\text{C}_{13}\text{H}_{11}\text{NO}^+$ | | | | | | | |
|  | (8.72) IP from 82LEV/LIA, 80GRU/SCH. | (231) | (966) | 30 | 125 | *EST | |
| $\text{C}_{13}\text{H}_{11}\text{N}_3\text{O}_3^+$ | | | | | | | |
|  | (7.7) IP is onset of photoelectron band (81MIL/MIL). | (221) | (924) | 43 | 181 | *EST | 1435-68-3 |
|  | (≤ 8.19) IP from 81MIL/MIL. | (≤ 232) | (≤ 971) | 43 | 181 | *EST | 19020-84-9 |
| $\text{C}_{13}\text{H}_{11}\text{O}^+$ | | | | | | | |
|  | | 167 | 698 | From proton affinity of $(\text{C}_6\text{H}_5)_2\text{CO}$ (RN 119-61-9). PA = 210.9 kcal/mol, 882. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

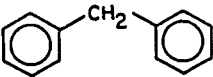
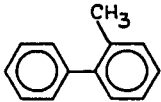
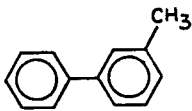
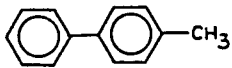
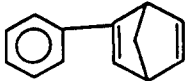
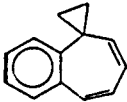
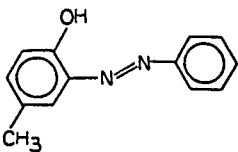
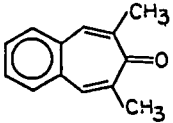
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₁₂⁺ | | | | | | | |
|  | 8.55±0.03 | 230 | 965 | 33±0.7 | 140±3 | 77PED/RYL | 101-81-5 |
|  | (8.10±0.02) | (228) | (954) | 41±2 | 172±7 | 77PED/RYL | 643-58-3 |
|  | (7.95±0.02) | (219) | (917) | 36±2 | 150±8 | *EST | 643-93-6 |
|  | (7.80±0.02) | (216) | (904) | 36±2 | 151±8 | *EST | 644-08-6 |
|  | (≤9.06) IP from 83HOU/RON. | (≤289) | (≤1210) | 79 | 330 | *EST | 74437-39-1 |
|  | (8.0±0.1) | (266) | (1115) | 82 | 343 | *EST | 29150-13-8 |
| C₁₃H₁₂N₂O⁺ | | | | | | | |
|  | (7.4) IP is onset of photoelectron band (81MIL/MIL). | (217) | (910) | 47 | 196 | *EST | 952-47-6 |
| C₁₃H₁₂O⁺ | | | | | | | |
|  | (8.1) IP is onset of photoelectron band. | (186) | (780) | -0.5±3 | -2±11 | 77PED/RYL | 2484-16-4 |

Table 1. Positive Ion Table - Continued

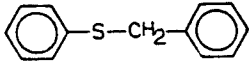
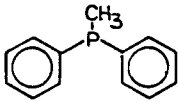
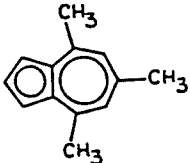
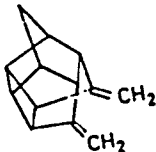
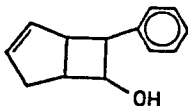
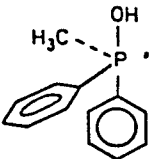
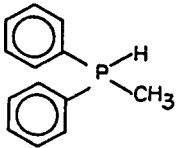
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₁₂S⁺ | | | | | | | |
|  | (7.87±0.02) | (227) | (950) | 46 | 191 | *EST | 831-91-4 |
| C₁₃H₁₃P⁺ | | | | | | | |
|  | (≤8.28±0.05) IP from 82IKU/KEB. | (≤235) | (≤984) | 44 | 185 | *EST | 1486-28-8 |
| C₁₃H₁₄⁺ | | | | | | | |
|  | (7.10) | (209) | (873) | 45 | 188 | *EST | 941-81-1 |
|  | ≤8.95 IP from 84MAR/KAY. | (≤222) | (≤929) | 16 | 65 | *EST | |
| C₁₃H₁₄O⁺ | | | | | | | |
|  | (7.82) IP from 81DAL/NIB. | (197) | (824) | 17 | 70 | 81DAL/NIB | 64353-61-3 |
| C₁₃H₁₄OP⁺ | | | | | | | |
|  | | 118 | 494 | | | | |
| | | From proton affinity of (C ₆ H ₅) ₂ CH ₃ PO (RN 2129-89-7) (86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | | | |
| C₁₃H₁₄P⁺ | | | | | | | |
|  | | 180 | 752 | | | | |
| | | From proton affinity of (C ₆ H ₅) ₂ (CH ₃)P (RN 1486-28-8). PA = 230.3 kcal/mol, 963.5 kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

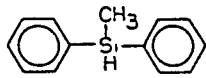
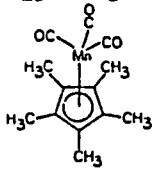
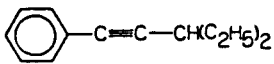
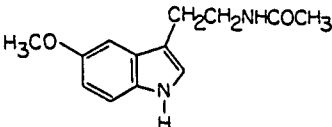
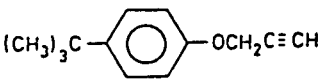
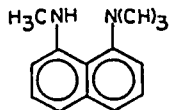
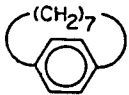
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{13}\text{H}_{14}\text{Si}^+$ | | | | | | | |
|  | (8.75±0.15) | (229) | (959) | 27 | 115 | *EST | 776-76-1 |
| $\text{C}_{13}\text{H}_{15}\text{MnO}_3^+$ | | | | | | | |
|  | (7.0) | (14) | (57) | -148 | -618 | *EST | 34807-89-1 |
| | IP is onset of photoelectron band (81CAL/HUB). | | | | | | |
| $\text{C}_{13}\text{H}_{16}^+$ | | | | | | | |
|  | (8.24±0.08) | (239) | (1000) | 49 | 205 | *EST | |
| | IP from 81ELB/LIE. | | | | | | |
| $\text{C}_{13}\text{H}_{16}\text{N}_2\text{O}_2^+$ | | | | | | | |
|  | (7.03) | (109) | (454) | -54 | -224 | *EST | |
| | IP is onset of photoelectron band (83CAN/HAM). | | | | | | |
| $\text{C}_{13}\text{H}_{16}\text{O}^+$ | | | | | | | |
|  | (7.85) | (196) | (820) | 15 | 63 | 8SORL/MIS | |
| | IP from 8SORL/MIS. | | | | | | |
| $\text{C}_{13}\text{H}_{17}\text{N}_2^+$ | | | | | | | |
|  | | 78 | 327 | | | | |
| | From proton affinity of N,N,N'-trimethyl-1,8-naphthalenediamine (RN 20723-57-0). PA = 235.6 kcal/mol, 986. kJ/mol. | | | | | | |
| $\text{C}_{13}\text{H}_{18}^+$ | | | | | | | |
|  | (8.0) | (195) | (817) | 11 | 45 | *EST | 3761-63-5 |
| | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

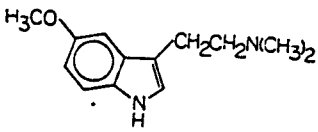
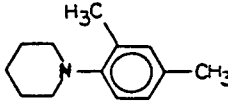
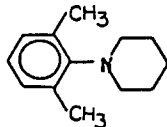
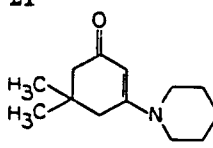
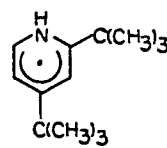
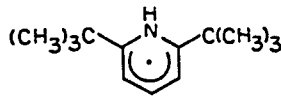
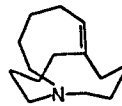
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₁₈N₂O⁺  | ($\leq 7.61 \pm 0.14$) | (≤ 172) | (≤ 721) | -3 | -13 | *EST | 1019-45-0 |
| C₁₃H₁₉N⁺  | (≤ 7.70) IP from 82ROZ/HOU2. | (≤ 202) | (≤ 844) | 24 | 101 | *EST | 81506-14-1 |
|  | (7.35) IP is onset of photoelectron band (82ROZ/HOU2). | (185) | (774) | 16 | 65 | *EST | 81506-15-2 |
| C₁₃H₂₁NO⁺  | (≤ 7.67) IP from 82PFI/GER. | (≤ 131) | (≤ 546) | -46 | -194 | *EST | 13358-76-4 |
| C₁₃H₂₂N⁺  | 115 483 From proton affinity of 2,4-di-tert-butylpyridine (RN 29939-31-9). PA = (231.4) kcal/mol, (968) kJ/mol. | | | | | | |
|  | 113 473 From proton affinity of 2,6-di-tert-butylpyridine (RN 585-48-4). PA = 233.4 kcal/mol, 976. kJ/mol. | | | | | | |
| C₁₃H₂₃N⁺  | (6.8) IP is onset of photoelectron band (82ALD/ARR). | (184) | (770) | 27 | 114 | *EST | 84509-55-7 |

Table 1. Positive Ion Table - Continued

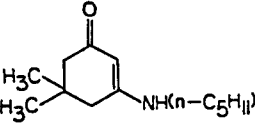
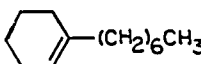
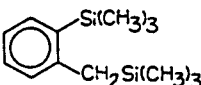
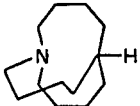
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₃H₂₃NO⁺  | (≤ 8.07) IP from 82PFI/GER. | (≤ 122) | (≤ 512) | -64 | -267 | *EST | 82663-50-1 |
| C₁₃H₂₄⁺ | | | | | | | |
| 1-C ₁₃ H ₂₄ | (9.90 \pm 0.02) | (223) | (934) | -5 | -21 | *EST | 26186-02-7 |
| 2-C ₁₃ H ₂₄ | (9.28 \pm 0.02) | (205) | (856) | -9 | -39 | *EST | 28467-75-6 |
| 3-C ₁₃ H ₂₄ | (9.14 \pm 0.03) | (202) | (844) | -9 | -38 | *EST | 60186-78-9 |
| 4-C ₁₃ H ₂₄ | (9.07 \pm 0.03) | (200) | (837) | -9 | -38 | *EST | 60186-79-0 |
| 5-C ₁₃ H ₂₄ | (9.09 \pm 0.03) | (201) | (839) | -9 | -38 | *EST | 60186-80-3 |
| 6-C ₁₃ H ₂₄ | (9.05 \pm 0.03) | (200) | (835) | -9 | -38 | *EST | 42371-66-4 |
|  | (8.37 \pm 0.02) | (232) | (969) | 39 \pm 1 | 161 \pm 5 | *EST | 15232-86-7 |
| C₁₃H₂₄Si₂⁺  | (8.26) IP is onset of photoelectron band (82TRA/RED). | (152) | (637) | -38 | -160 | *EST | 1899-74-7 |
| C₁₃H₂₅N⁺  | (7.3) IP from 82ALD/ARR. | (157) | (657) | -11 | -47 | 81ALD/ARR | |
| C₁₃H₂₆⁺ | | | | | | | |
| 1-C ₁₃ H ₂₆ | (9.38) IP from 81HOL/FIN. | (172) | (719) | -44.5 | -186.2 | *EST | 2437-56-1 |
| $((\text{CH}_3)_3\text{C})_2\text{C}=\text{CHCH}(\text{CH}_3)_2$ | (8.31 \pm 0.01) | (136) | (569) | -56 | -233 | *EST | 50787-12-7 |

Table 1. Positive Ion Table - Continued

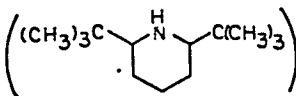
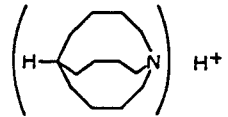
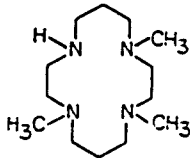
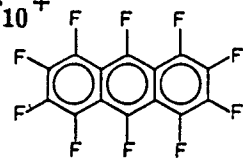
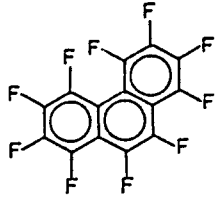
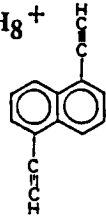
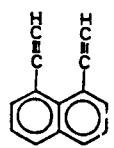
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{13}\text{H}_{26}\text{N}^+$ | | | | | | | |
|  | H ⁺ | | 56 233 | | | | |
| | | From proton affinity of 2,6-di-tert-butylpiperidine (RN 29939-31-9). PA = 234.3 kcal/mol, 980. kJ/mol. | | | | | |
|  | H ⁺ | | 140 586 | | | | |
| | | From proton affinity of out-6H-1-azabicyclo[4.4.4]tetradecane. PA = 214.3 kcal/mol, 896. kJ/mol. | | | | | |
| $\text{C}_{13}\text{H}_{30}\text{N}_4^+$ | | | | | | | |
|  | | (≤ 8.0) IP from 83BAK/ARM. | (≤ 185) | (≤ 775) | 1 | 3 | *EST |
| $\text{C}_{14}\text{F}_{10}^+$ | | | | | | | |
|  | | (8.28 ± 0.05) | (-165) | (-691) | -356 | -1490 | *EST 1580-19-4 |
|  | | (8.75 ± 0.05) | (-160) | (-669) | -362 | -1513 | *EST 1580-20-7 |
| $\text{C}_{14}\text{H}_8^+$ | | | | | | | |
|  | | (7.91) IP from 81GLE/SCH. | (324) | (1357) | 142 | 594 | *EST 67665-34-3 |
|  | | (7.88) IP from 81GLE/SCH, 84GLE/SCH. | (330) | (1381) | 148 | 621 | *EST 18067-44-2 |

Table 1. Positive Ion Table - Continued

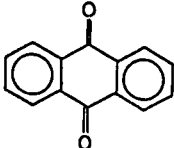
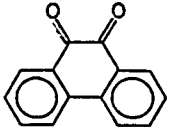
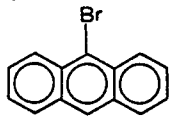
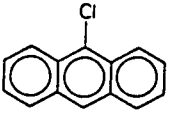
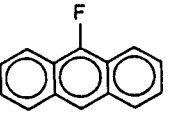
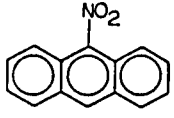
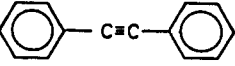
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_8\text{O}_2^+$  | 9.25±0.03 | 190 | 797 | -22.8±1.6 | -95.2±6.6 | 77PED/RYL | 84-65-1 |
|  | (8.64±0.03) | (166±1) | (695±5) | -33±1 | -139±5 | 77PED/RYL | 84-11-7 |
| $\text{C}_{14}\text{H}_9\text{Br}^+$  | (7.58) IP from 83KLA/KOV, 82LEV/LIA. | (236) | (986) | 61 | 255 | *EST | 1564-64-3 |
| $\text{C}_{14}\text{H}_9\text{Cl}^+$  | (7.45±0.03) IP from 82LEV/LIA, 83KLA/KOV. | (221) | (924) | 49 | 205 | *EST | 716-53-0 |
| $\text{C}_{14}\text{H}_9\text{F}^+$  | (7.46) IP from 83KLA/KOV, 82LEV/LIA. | (179) | (751) | 7 | 31 | *EST | 529-85-1 |
| $\text{C}_{14}\text{H}_9\text{NO}_2^+$  | 7.87±0.01 IP from 82LEV/LIA, 83KLA/KOV. | (233) | (974) | 51 | 215 | *EST | 602-60-8 |
| $\text{C}_{14}\text{H}_{10}^+$  | 7.90±0.02 See also: 81ELB/LIE. | 278 | 1165 | 96±1 | 403±4 | 82CHI/LIE | 501-65-5 |

Table 1. Positive Ion Table - Continued

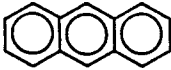
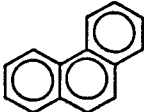
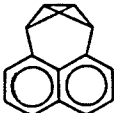
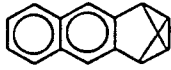
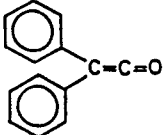
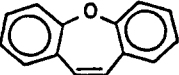
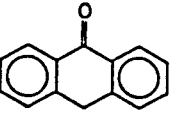
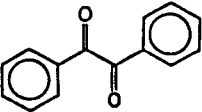
| ION Neutral | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|-------|--------------------------|---------|------------------------------|------------|----------------------|------------------------|
| | eV | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₄H₁₀⁺ | | | | | | | | |
|  | 7.45±0.03 | 227 | 949 | 55±0.2 | 230±1 | 79KUD/KUD4 | 120-12-7 | |
| | See also: 83KLA/KOV, 84STA/MAQ. Value of IP from charge transfer equilibrium constant determination (80MAU, re-evaluated) is in agreement. | | | | | | | |
|  | 7.86±0.02 | 230 | 965 | 49±0.2 | 207±1 | 79KUD/KUD4 | 85-01-8 | |
| | Value of IP from charge transfer equilibrium constant determination (80MAU, re-evaluated), 7.89 eV. | | | | | | | |
|  | (7.55) | (293) | (1226) | 119 | 498 | *EST | 40480-63-5 | |
| | IP from 81GLE/GUB. | | | | | | | |
|  | (7.71) | (247) | (1036) | 70 | 292 | *EST | 77669-79-5 | |
| | IP from 81GLE/GUB. | | | | | | | |
| C₁₄H₁₀O⁺ | | | | | | | | |
|  | (7.85) | (206) | (862) | 25 | 105 | 80DEM/WUL | 525-06-4 | |
|  | (7.45) | (207) | (866) | 35±4 | 147±18 | *EST | 257-05-6 | |
|  | (8.83±0.03) | (211) | (883) | 7 | 31 | 78KIM/WIN | 90-44-8 | |
| C₁₄H₁₀O₂⁺ | | | | | | | | |
|  | (8.5) | (183) | (764) | -13±0.7 | -56±3 | 77PED/RYL | 134-81-6 | |
| | IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

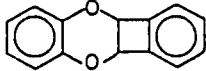
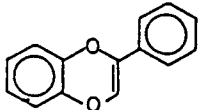
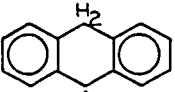
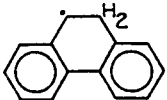
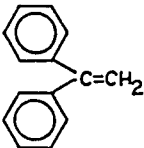
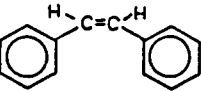
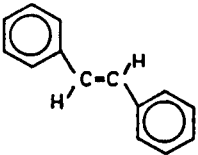
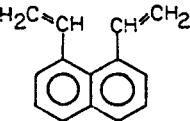
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|--|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{10}\text{O}_2^+$ | | | | | | | |
|  | (7.28±0.02) IP from 81BOU/DAG. | (153) | (639) | -15 | -63 | 81BOU/DAG | |
|  | (7.60±0.02) IP from 81BOU/DAG. | (171) | (716) | -4 | -17 | 81BOU/DAG | |
| $\text{C}_{14}\text{H}_{11}^+$ | | | | | | | |
|  | | 214 | 894 | From proton affinity of anthracene (RN 120-12-7). PA = 207.0 kcal/mol, 866. kJ/mol. | | | |
|  | | 216 | 906 | From proton affinity of phenanthrene (RN 85-01-8). PA = 198.7 kcal/mol, 831. kJ/mol. | | | |
| $\text{C}_{14}\text{H}_{12}^+$ | | | | | | | |
|  | (8.00±0.02) | (243) | (1018) | 59±1 | 246±4 | 77PED/RYL | 530-48-3 |
|  | (7.80±0.02) | (240) | (1005) | 60.3±0.4 | 252.4±1.6 | 77PED/RYL | 645-49-8 |
|  | 7.70±0.03 | 234 | 978 | 56±0.7 | 235±3 | 77PED/RYL | 103-30-0 |
|  | (7.72) IP from GLE/SCH. | (252) | (1054) | 74 | 309 | *EST | 17935-66-9 |

Table 1. Positive Ion Table - Continued

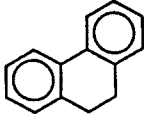
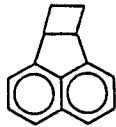
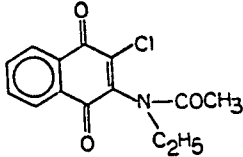
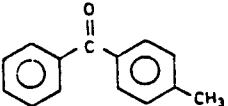
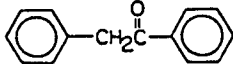
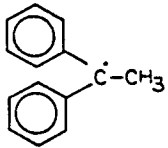
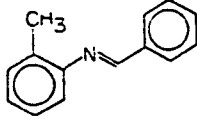
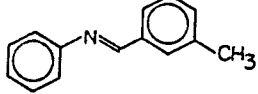
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $C_{14}H_{12}^+$ | | | | | | | |
|  | (7.55±0.02) | (216) | (903) | 42±2 | 175±8 | 77SHA/GOL | 776-35-2 |
|  | (7.60) IP from 82GLE/GUB. | (265) | (1107) | 89 | 374 | *EST | |
| $C_{14}H_{12}ClNO_3^+$ | | | | | | | |
|  | (8.7) IP is onset of photoelectron band (80RED/FRE). | (120) | (504) | -80 | -335 | *EST | 4497-72-7 |
| $C_{14}H_{12}O^+$ | | | | | | | |
|  | (9.13±0.05) | (217) | (907) | 6.1±1 | 26±4 | *EST | |
|  | (8.50) IP is onset of photoelectron band (78CEN/FRA). | (201) | (842) | 5±1 | 22±5 | 77PED/RYL | 451-40-1 |
| $C_{14}H_{13}^+$ | | | | | | | |
|  | | 213 | 889 | | | | |
| | | From proton affinity of $(C_6H_5)_2C=CH_2$ (RN 530-48-3). PA = 211.9 kcal/mol, 887. kJ/mol. | | | | | |
| $C_{14}H_{13}N^+$ | | | | | | | |
|  | (7.7) IP is onset of photoelectron band. | (236) | (988) | 59 | 245 | *EST | 5877-55-4 |
|  | (≤8.07) | (≤245) | (≤1024) | 59 | 245 | *EST | 6906-25-8 |

Table 1. Positive Ion Table - Continued

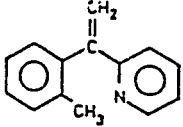
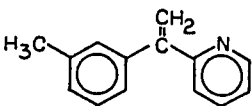
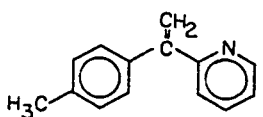
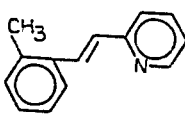
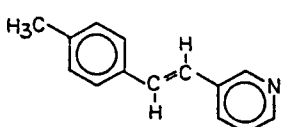
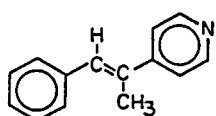
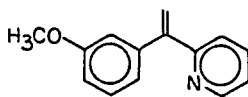
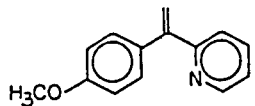
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------------------------------|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₄H₁₃N⁺  | 8.55 IP from 80GRU/SCH. | (261) | (1092) | 64 | 268 | *EST | 74309-54-9 |
|  | (8.48) | (258) | (1080) | 63 | 262 | *EST | |
|  | (8.45) | (257) | (1077) | 63 | 262 | *EST | |
|  | 8.01 IP from 80GRU/SCH, 80SCH/RAM. | (249) | (1040) | 64 | 267 | *EST | 77275-11-7 |
|  | ($\leq 7.90 \pm 0.05$) | (≤ 242) | (≤ 1012) | 60 | 250 | *EST | 6892-33-7 |
|  | ($\leq 8.39 \pm 0.05$) | (≤ 254) | (≤ 1063) | 60 | 253 | *EST | 18150-12-4 |
| C₁₄H₁₃NO⁺  | (8.27) | (225) | (942) | 34 | 144 | *EST | |
|  | (8.15) | (222) | (930) | 34 | 144 | *EST | |

Table 1. Positive Ion Table - Continued

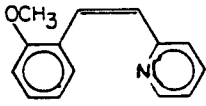
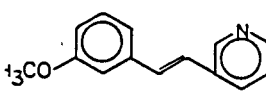
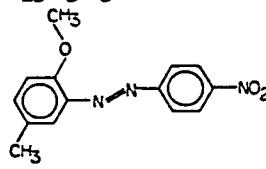
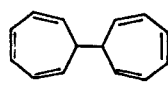
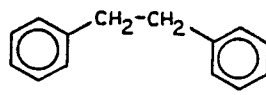
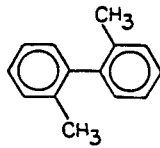
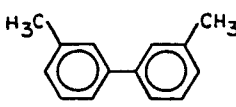
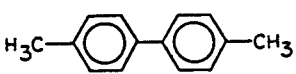
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|-----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{13}\text{NO}^+$ | | | | | | | |
|  | (7.87) IP from 80GRU/SCH. | (217) | (908) | 36 | 149 | *EST | 62205-27-0 |
|  | ($\leq 7.72 \pm 0.05$) | (≤ 211) | (≤ 883) | 33 | 138 | *EST | 5847-73-4 |
| $\text{C}_{14}\text{H}_{13}\text{N}_3\text{O}_3^+$ | | | | | | | |
|  | (7.8) IP is onset of photoelectron band (81MIL/MIL). | (229) | (959) | 49 | 206 | *EST | |
| $\text{C}_{14}\text{H}_{14}^+$ | | | | | | | |
|  | (8.2) IP is onset of photoelectron band. | (288) | (1203) | 99 | 412 | *EST | 39473-62-6 |
|  | 8.7 ± 0.1 | 235 | 982 | 34.2 ± 0.4 | 143.0 ± 1.8 | 77PED/RYL | 103-29-7 |
|  | (8.05 ± 0.02) | (214) | (895) | 28 | 118 | *EST | 605-39-0 |
|  | (7.85 ± 0.02) | (208) | (871) | 27 | 114 | *EST | 612-75-9 |
|  | (8.50) | (223) | (934) | 27 | 114 | *EST | 613-33-2 |

Table 1. Positive Ion Table - Continued

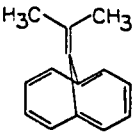
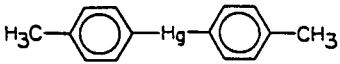
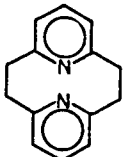
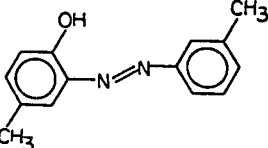
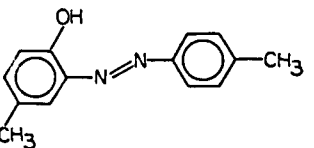
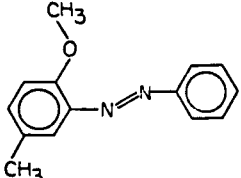
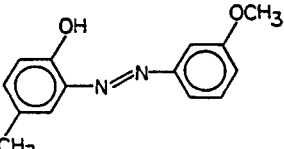
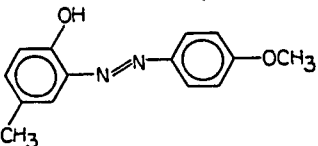
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{14}^+$  | (7.7) | (261) | (1094) | 84 | 351 | *EST | 88635-77-2 |
| IP is onset of photoelectron band (84AND/CER). | | | | | | | |
| $\text{C}_{14}\text{H}_{14}\text{Hg}^+$  | (7.94) | (261) | (1091) | 78 | 325 | *EST | 537-64-4 |
| IP is onset of photoelectron band (81FUR/PIA). | | | | | | | |
| $\text{C}_{14}\text{H}_{14}\text{N}_2^+$  | (8.35) | (265) | (1109) | 72 | 303 | *EST | 6574-83-0 |
| $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}^+$  | (≤ 7.88) | (≤ 221) | (≤ 923) | 39 | 163 | *EST | 19020-81-6 |
| IP from 81MIL/MIL. | | | | | | | |
|  | (≤ 7.88) | (≤ 221) | (≤ 923) | 39 | 163 | *EST | 17739-97-8 |
| IP from 81MIL/MIL. | | | | | | | |
|  | (7.3) | (221) | (925) | 53 | 221 | *EST | 77046-80-1 |
| IP is onset of photoelectron band (81MIL/MIL). | | | | | | | |
| $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_2^+$  | (≤ 7.85) | (≤ 192) | (≤ 802) | 11 | 45 | *EST | 23375-56-6 |
| IP from 81MIL/MIL. | | | | | | | |
|  | (≤ 7.76) | (≤ 190) | (≤ 794) | 11 | 45 | *EST | 15096-05-6 |
| IP from 81MIL/MIL. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------|---|--------------------------|----------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_2^+$ | | (7.72) IP from 77NUY/MES. | (198) | (830) | 20 | 85 | *EST | 501-58-6 |
| $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_3^+$ | | (≤ 8.06) IP from 81MIL/CIL. | (≤ 202) | (≤ 844) | 16 | 66 | *EST | 1562-94-3 |
| $\text{C}_{14}\text{H}_{14}\text{O}^+$ | | (≤ 7.78) | (≤ 161) | (≤ 674) | -18 | -77 | *EST | 5040-51-7 |
| $\text{C}_{14}\text{H}_{14}\text{OS}^+$ | | (8.1) IP is onset of photoelectron band (84GLE/BIS). | (227) | (950) | 40 | 169 | *EST | |
| | | (≤ 8.45) IP from 81MOH/JIA. | (≤ 204) | (≤ 855) | 10 | 40 | *EST | |
| $\text{C}_{14}\text{H}_{14}\text{OSi}^+$ | | (8.0 ± 0.1) | (158) | (661) | -27 | -111 | *EST | 18414-62-5 |
| | | (-7.0) IP from 82TRA/RED. | (-57) | (-240) | -104 | -435 | *EST | |

Table 1. Positive Ion Table - Continued

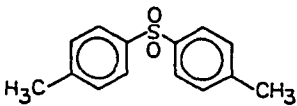
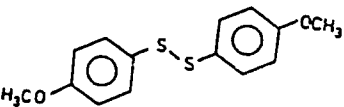
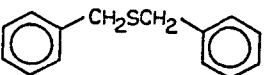
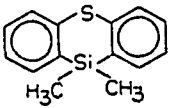
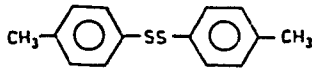
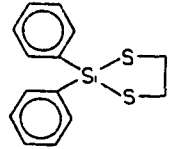
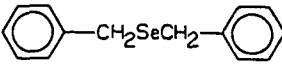
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{14}\text{O}_2\text{S}^+$  | (8.66±0.04) | (151) | (634) | -48±0.7 | -202±3 | 77PED/RYL | 599-66-6 |
| $\text{C}_{14}\text{H}_{14}\text{O}_2\text{S}_2^+$  | 7.6 IP from 82GIO/BOC. | (161) | (674) | -14 | -59 | *EST | 5335-87-5 |
| $\text{C}_{14}\text{H}_{14}\text{S}^+$  | (8.05±0.02) | (232) | (969) | 46±1 | 192±4 | 77PED/RYL | 538-74-9 |
| $\text{C}_{14}\text{H}_{14}\text{SSi}^+$  | (7.45) IP is onset of photoelectron band (82TRA/RED). See also: 81TRA/RED. | (206) | (864) | 35 | 145 | *EST | 61431-08-1 |
| $\text{C}_{14}\text{H}_{14}\text{S}_2^+$  | 7.5 IP is onset of photoelectron band (82GIO/BOC). | (215) | (901) | 42 | 177 | *EST | 103-19-5 |
| $\text{C}_{14}\text{H}_{14}\text{S}_2\text{Si}^+$  | (8.4) IP is onset of photoelectron band (83AND/CAU). | (220) | (919) | 26 | 109 | *EST | 57864-56-9 |
| $\text{C}_{14}\text{H}_{14}\text{Se}^+$  | (≤7.96) IP from 81BAK/ARM. | (≤243) | (≤1015) | 59 | 247 | *EST | 1842-38-2 |

Table 1. Positive Ion Table - Continued

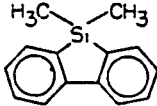
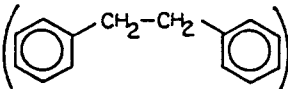
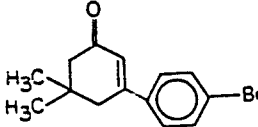
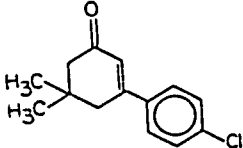
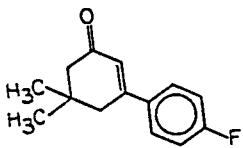
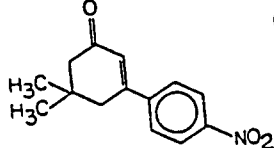
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------------|--|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₄H₁₄Si⁺ | (7.4) | (178) | (743) | 7 | 29 | *EST | 13688-68-1 |
|  | IP is onset of photoelectron band. | | | | | | |
| C₁₄H₁₅⁺ | | 205 | 859 | | | | |
|  | H ⁺ | From proton affinity of C ₆ H ₅ CH ₂ CH ₂ C ₆ H ₅ (RN 103-29-7). PA = 194.6 kcal/mol, 814. kJ/mol. | | | | | |
| C₁₄H₁₅BrO⁺ | (≤8.57) | (≤184) | (≤770) | -14 | -57 | *EST | 72036-54-5 |
|  | IP from 82PFI/GER. | | | | | | |
| C₁₄H₁₅ClO⁺ | (8.67) | (173) | (724) | -27 | -113 | *EST | 59344-32-0 |
|  | IP from 82PFI/GER. | | | | | | |
| C₁₄H₁₅FO⁺ | (8.90) | (124) | (518) | -82 | -341 | *EST | 72036-55-6 |
|  | IP from 82PFI/GER. | | | | | | |
| C₁₄H₁₅NO₃⁺ | (≤9.28) | (≤193) | (≤806) | -21 | -89 | *EST | 29339-45-5 |
|  | IP from 82PFI/GER. | | | | | | |

Table 1. Positive Ion Table - Continued

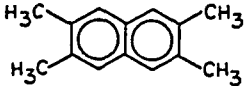

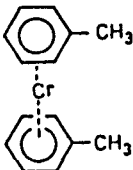
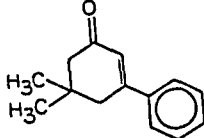
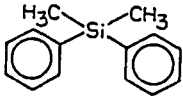
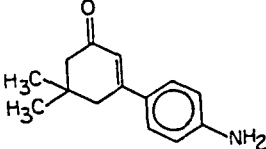
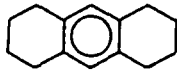
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{16}^+$ |  | ($\leq 7.60 \pm 0.03$) | (≤ 180) | (≤ 754) | 5 ± 0.7 | 21 ± 3 | *EST | 1134-40-3 |
| |  | (7.95 ± 0.05) IP from 81HEI/KOV. | (241) | (1007) | 57 | 240 | *EST | 54922-12-2 |
| $\text{C}_{14}\text{H}_{16}\text{Cr}^+$ |  | ($\leq 5.24 \pm 0.1$) See also: 82CAB/COW. | (≤ 86) | (≤ 360) | -35 | -146 | *EST | 12087-58-0 |
| $\text{C}_{14}\text{H}_{16}\text{O}^+$ |  | (≤ 8.90) IP from 82PFI/GER. | (≤ 187) | (≤ 781) | -19 | -78 | *EST | 36047-17-3 |
| $\text{C}_{14}\text{H}_{16}\text{Si}^+$ |  | (8.5) IP from 81TRA/RED. | (209) | (875) | 13 | 55 | *EST | 778-24-5 |
| $\text{C}_{14}\text{H}_{17}\text{NO}^+$ |  | (≤ 7.85) IP from 82PFI/GER. | (≤ 163) | (≤ 683) | -18 | -74 | *EST | 72036-57-8 |
| $\text{C}_{14}\text{H}_{18}^+$ |  | (7.86) IP from charge transfer equilibrium constant determinations (80MAU). | (172) | (721) | -9 ± 0.7 | -37 ± 3 | 77PED/RYL | 1079-71-6 |

Table 1. Positive Ion Table - Continued

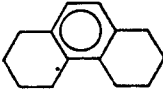
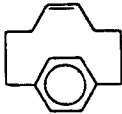
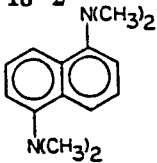
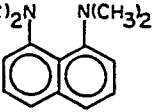
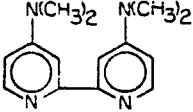
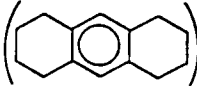
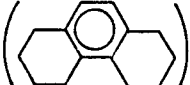
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|--|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{18}^+$ | | | | | | | |
|  | 7.89 | 174 | 727 | -8±2 | -34±8 | 77SHA/GOL | 5325-97-3 |
| | IP from charge transfer equilibrium constant determinations (80MAU). | | | | | | |
|  | (≤8.37) | (≤246) | (≤1029) | 53 | 221 | *EST | |
| | IP from 80GLE/HOP. | | | | | | |
| $\text{C}_{14}\text{H}_{18}\text{N}_2^+$ | | | | | | | |
|  | (6.70±0.02) | (200) | (839) | 46 | 193 | *EST | 10075-69-1 |
|  | (6.45±0.02) | (212) | (884) | 63 | 262 | *EST | 20734-58-1 |
| $\text{C}_{14}\text{H}_{18}\text{N}_4^+$ | | | | | | | |
|  | (7.3) | (244) | (1021) | 76 | 317 | *EST | 85698-56-2 |
| | IP is onset of photoelectron band (83DOB/HIL). | | | | | | |
| $\text{C}_{14}\text{H}_{19}^+$ | | | | | | | |
|  | H^+ | 154 | 645 | From proton affinity of 1,2,3,4,5,6,7,8-octahydroanthracene (RN 1079-71-6). PA = 202.6 kcal/mol, 848. kJ/mol. | | | |
|  | H^+ | 153 | 640 | From proton affinity of 1,2,3,4,5,6,7,8-octahydrophenanthrene (RN 5325-97-3). PA = 204.7 kcal/mol, 856. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

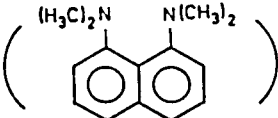
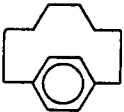
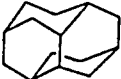
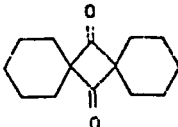
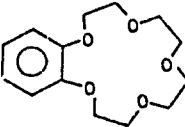
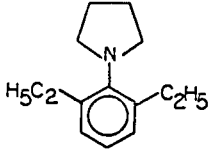
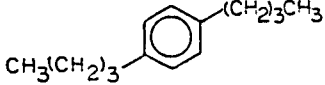
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------|---|---------------|------------------------------|-----------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{19}\text{N}_2^+$ | | | 186 | 780 | | | | |
|  H^+ | | | From proton affinity of N,N,N',N'-tetramethyl-1,8-naphthalenediamine (RN 20734-58-1). PA = 241.8 kcal/mol, 1012. kJ/mol. | | | | | |
| $\text{C}_{14}\text{H}_{20}^+$ | | | | | | | | |
|  | (≤ 8.17) | (≤ 190) | (≤ 795) | 2 | 7 | *EST | 4685-74-9 | |
| | IP from 80GLE/HOP. | | | | | | | |
|  | (8.93) | (171.1) | (717.1) | -34.5 | -144.5 | 79CLA/KNO | 2292-79-7 | |
| $\text{C}_{14}\text{H}_{20}\text{O}_2^+$ | | | | | | | | |
|  | 9.0 \pm 0.05 | (141) | (589) | -67 | -279 | *EST | 950-21-0 | |
| | IP from 84OLI/FLE. | | | | | | | |
| $\text{C}_{14}\text{H}_{20}\text{O}_5^+$ | | | | | | | | |
|  | (≤ 8.0) | (≤ 19) | (≤ 80) | -165 | -692 | *EST | 14098-44-3 | |
| | IP from 83BAK/ARM. | | | | | | | |
| $\text{C}_{14}\text{H}_{21}\text{N}^+$ | | | | | | | | |
|  | (≤ 7.60) | (≤ 208) | (≤ 869) | 33 | 136 | *EST | 81506-13-0 | |
| | IP from 82ROZ/HOU2. | | | | | | | |
| $\text{C}_{14}\text{H}_{22}^+$ | | | | | | | | |
|  | (≤ 8.40) | (≤ 163) | (≤ 681) | -31 \pm 1.4 | -129 \pm 6 | 84NES/VER | 1571-86-4 | |
| | IP from 80GLE/HOP. See also: 85BAI/MIS. | | | | | | | |

Table 1. Positive Ion Table - Continued

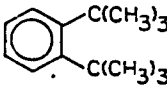
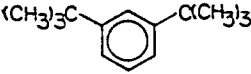
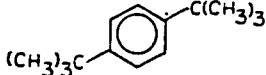
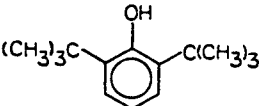
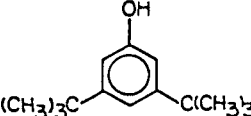
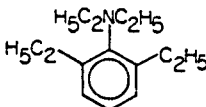
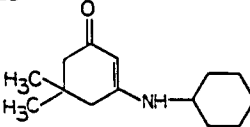
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{22}^+$ | | | | | | | |
|  | ($\leq 8.60 \pm 0.07$) | (≤ 185) | (≤ 774) | -13 | -56 | *EST | 1012-76-6 |
|  | (8.71 ± 0.07) | (171) | (713) | -30 ± 1.4 | -127 ± 6 | 84NES/VER | 1014-60-4 |
|  | 8.24 ± 0.01 | (161) | (673) | -29 | -122 | 85ORL/MIS | 1012-72-2 |
| | IP from 82LEV/LIA, 84HOW/GON, 86ORL/MIS. See also: 85BAI/MIS. | | | | | | |
| $\text{C}_{14}\text{H}_{22}\text{O}^+$ | | | | | | | |
|  | (7.70 ± 0.02) | (112) | (468) | -66 | -275 | *EST | 128-39-2 |
| | See also: 83CET/LAP. | | | | | | |
|  | (7.90 ± 0.02) | (109) | (455) | -73 | -307 | *EST | 1138-52-9 |
| $\text{C}_{14}\text{H}_{23}\text{N}^+$ | | | | | | | |
|  | (≤ 7.77) | (≤ 207) | (≤ 867) | 28 | 117 | *EST | 81506-16-3 |
| | IP from 82ROZ/HOU2. | | | | | | |
| $\text{C}_{14}\text{H}_{23}\text{NO}^+$ | | | | | | | |
|  | (≤ 7.75) | (≤ 116) | (≤ 486) | -63 | -262 | *EST | 1500-76-1 |
| | IP from 82PFI/GER. | | | | | | |

Table 1. Positive Ion Table - Continued

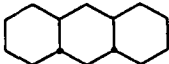
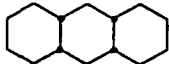
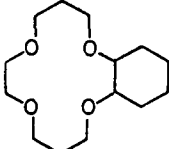
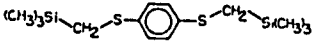
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{24}^+$ | | | | | | | |
|  | (8.8) | (145) | (606) | -58.1±0.9 | -243.2±3.8 | 77PED/RYL | |
| | IP is onset of photoelectron band (84HEI/HON). | | | | | | |
|  | (9.0) | (158) | (660) | -50 | -208 | 71ALL/WUE | |
| | IP is onset of photoelectron band (84HEI/HON). | | | | | | |
| $\text{C}_{14}\text{H}_{24}\text{O}_4^+$ | | | | | | | |
|  | (≤9.2) | (≤39) | (≤162) | -174 | -726 | *EST | |
| | IP from 83BAK/ARM. | | | | | | |
| $\text{C}_{14}\text{H}_{26}^+$ | | | | | | | |
| 1- $\text{C}_{14}\text{H}_{26}$ | (9.89±0.02) | (218) | (913) | -10 | -41 | *EST | 765-10-6 |
| 2- $\text{C}_{14}\text{H}_{26}$ | (9.26±0.03) | (199) | (833) | -14 | -60 | *EST | 638-60-8 |
| 3- $\text{C}_{14}\text{H}_{26}$ | (9.17±0.02) | (197) | (826) | -14 | -59 | *EST | 60212-32-0 |
| 4- $\text{C}_{14}\text{H}_{26}$ | (9.11±0.03) | (196) | (820) | -14 | -59 | *EST | 60212-33-1 |
| 5- $\text{C}_{14}\text{H}_{26}$ | (9.10±0.03) | (196) | (819) | -14 | -59 | *EST | 60212-34-2 |
| 6- $\text{C}_{14}\text{H}_{26}$ | (9.09±0.02) | (196) | (818) | -14 | -59 | *EST | 3730-08-3 |
| 7- $\text{C}_{14}\text{H}_{26}$ | (9.03±0.04) | (194) | (812) | -14 | -59 | *EST | 35216-11-6 |
| $\text{C}_{14}\text{H}_{26}\text{S}_2\text{Si}_2^+$ | | | | | | | |
|  | (7.0) | (23) | (96) | -138 | -579 | *EST | 69209-20-7 |
| | IP is onset of photoelectron band (82TRA/RED). | | | | | | |
| $\text{C}_{14}\text{H}_{28}^+$ | | | | | | | |
| $((\text{CH}_3)_3\text{C})_2\text{C}=\text{CHC}(\text{CH}_3)_3$ | (8.17±0.01) | (131) | (550) | -57 | -238 | 81HOL/FIN | 28923-90-2 |

Table 1. Positive Ion Table - Continued

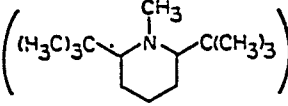
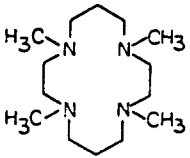
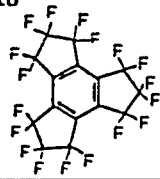
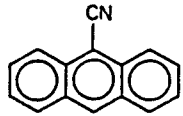
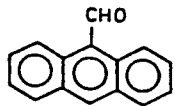
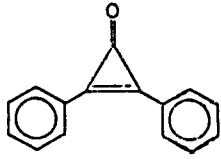
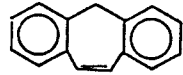
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|---|------------|------------------------------|-------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{14}\text{H}_{28}\text{N}^+$ | | | | | | | |
|  | | 60 | 250 | | | | |
| | H^+ | From proton affinity of 1-methyl-2,6-di-tert-butylpiperidine. PA = 239.2 kcal/mol, 1001. kJ/mol. | | | | | |
| $\text{C}_{14}\text{H}_{32}\text{N}_4^+$ | | | | | | | |
|  | ≤ 8.0 IP from 83BAK/ARM. | ≤ 184 | ≤ 770 | -0.5 | -2 | *EST | |
| $\text{C}_{15}\text{F}_{18}^+$ | | | | | | | |
|  | 11.3 IP is onset of photoelectron band (84HEI/WIR). | (-530) | (-2216) | -790 | -3306 | *EST | 33021-47-5 |
| $\text{C}_{15}\text{H}_9\text{N}^+$ | | | | | | | |
|  | (7.80 ± 0.03) See also: 83KLA/KOV. | (267) | (1114) | 87 ± 0.2 | 362 ± 1 | *EST | 1210-12-4 |
| $\text{C}_{15}\text{H}_{10}\text{O}^+$ | | | | | | | |
|  | 7.69 ± 0.03 | (204) | (852) | 26.3 ± 2 | 110 ± 8 | *EST | 642-31-9 |
|  | (8.1) IP is onset of photoelectron band. | (262) | (1099) | 76 ± 2 | 318 ± 8 | 85STE/GAM | 886-38-4 |
| $\text{C}_{15}\text{H}_{12}^+$ | | | | | | | |
|  | (7.6) IP is onset of photoelectron band. | (232) | (970) | 57 ± 0.7 | 237 ± 3 | *EST | 256-81-5 |

Table 1. Positive Ion Table - Continued

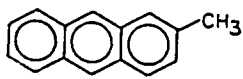
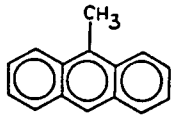
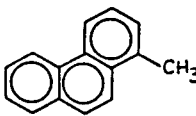
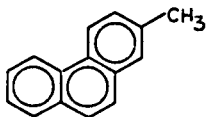
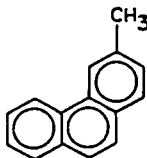
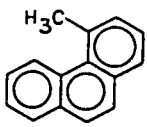
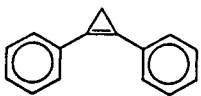
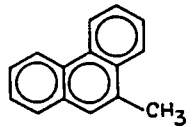
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{15}\text{H}_{12}^+$  | (7.37) | (215) | (898) | 45 | 187 | *EST | 613-12-7 |
| | IP from charge transfer equilibrium constant determinations (80MAU). | | | | | | |
|  | 7.24±0.03 | (215) | (899) | 48 | 201 | *EST | 779-02-2 |
| | See also: 80MAU, 83KLA/KOV. | | | | | | |
|  | 7.7±0.03 | (217) | (907) | 39±2 | 164±7 | *EST | 832-69-9 |
|  | (7.7) | (217) | (907) | 39±2 | 164±7 | *EST | 2531-84-2 |
|  | (7.68±0.01) | (216) | (905) | 39 | 164 | *EST | 832-71-3 |
|  | (7.70±0.02) | (222) | (929) | 44±4 | 186±15 | *EST | 832-64-4 |
|  | (7.45) | (282) | (1180) | 110 | 461 | *EST | 24168-52-3 |
|  | 7.46±0.03 | (214) | (897) | 42±0.2 | 177±1 | *EST | 883-20-5 |

Table 1. Positive Ion Table - Continued

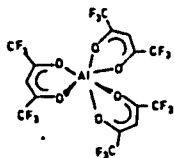
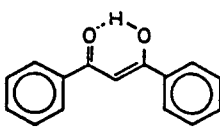
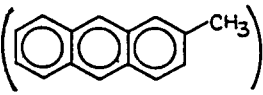
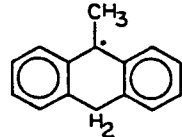
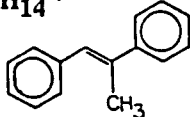
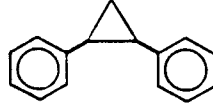
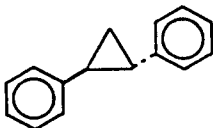

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|---|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{15}\text{H}_{12}\text{AlF}_9\text{O}_6^+$  | (8.7) IP is onset of photoelectron band. | (-580) | (-2428) | -781±3 | -3267±13 | 80TEL/RAB | 14354-59-7 |
| $\text{C}_{15}\text{H}_{12}\text{O}_2^+$  | (8.3) IP is onset of photoelectron band. | (131) | (550) | -60±0.7 | -251±3 | 81FER/RIB | 120-46-7 |
| $\text{C}_{15}\text{H}_{13}^+$  | H^+ | 200 | 837 | From proton affinity of 2-methylanthracene (RN 613-12-7). PA = 210.3 kcal/mol, 880. kJ/mol. | | | |
|  | | 200 | 836 | From proton affinity of 9-methylanthracene (RN 779-02-2). PA = 213.9 kcal/mol, 895. kJ/mol. | | | |
| $\text{C}_{15}\text{H}_{14}^+$  | (≤8.10±0.05) | (≤236) | (≤986) | 49 | 204 | *EST | 833-81-8 |
|  | (8.20) | (249) | (1043) | 60±0.5 | 252±2 | 77PED/RYL | 1138-48-3 |
|  | (8.05) | (243) | (1016) | 57±0.7 | 239±3 | 77PED/RYL | 1138-47-2 |
|  | (8.0) IP is onset of photoelectron band. | (351) | (1469) | 167 | 697 | *EST | 73045-27-9 |

Table 1. Positive Ion Table - Continued


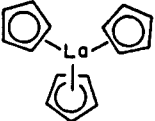
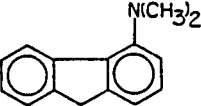
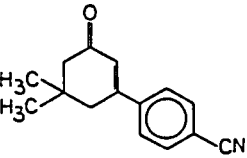
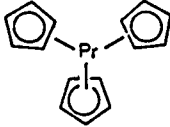
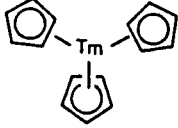
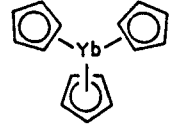
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₅H₁₄⁺  | (7.6) IP is onset of photoelectron band. | (386) | (1616) | 211 | 883 | *EST | 73050-58-5 |
| C₁₅H₁₅La⁺  | (7.9±0.3) | (218) | (912) | 36±2 | 150±7 | 77PED/RYL | 1272-23-7 |
| C₁₅H₁₅N⁺  | (7.1) IP is onset of photoelectron band (84GLE/SCH). | (216) | (905) | 53 | 220 | *EST | 92013-89-3 |
| C₁₅H₁₅NO⁺  | (≤9.20) IP from 82PFI/GER. | (≤226) | (≤946) | 14 | 58 | *EST | 72036-56-7 |
| C₁₅H₁₅Pr⁺  | (7.68±0.1) | (200) | (838) | 23±2 | 97±9 | 77PED/RYL | 11077-59-1 |
| C₁₅H₁₅Tm⁺  | (7.43±0.1) | (186) | (779) | 15±1 | 62±6 | 77PED/RYL | 1272-26-0 |
| C₁₅H₁₅Yb⁺  | (7.5±0.3) | (206) | (862) | 33±1 | 138±6 | 77PED/RYL | 1295-20-1 |

Table 1. Positive Ion Table - Continued

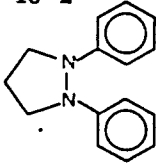
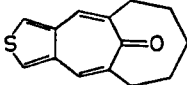
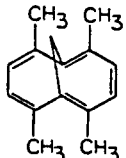
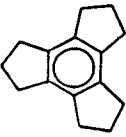
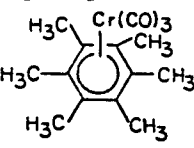
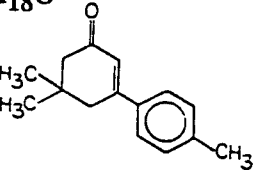
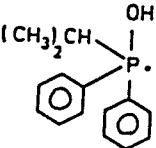
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|--|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{15}\text{H}_{16}\text{N}_2^+$  | (7.50±0.05) | (253) | (1059) | 80 | 335 | *EST | 63378-86-9 |
| $\text{C}_{15}\text{H}_{16}\text{OS}^+$  | (8.15) IP is onset of photoelectron band (84GLE/BIS). | (213) | (890) | 25 | 104 | *EST | |
| $\text{C}_{15}\text{H}_{18}^+$   | (7.1) IP is onset of photoelectron band (84AND/CER). | (219) | (918) | 56 | 233 | *EST | 88635-76-1 |
| $\text{C}_{15}\text{H}_{18}\text{CrO}_3^+$  | (6.35±0.1) | (10) | (42) | -136±3 | -571±13 | 77PED/RYL | 12088-11-8 |
| $\text{C}_{15}\text{H}_{18}\text{O}^+$  | (≤8.59) IP from 82PFI/GER. | (≤172) | (≤718) | -26 | -111 | *EST | 72036-52-3 |
| $\text{C}_{15}\text{H}_{18}\text{OP}^+$  | | 89 | 373 | From proton affinity of $i\text{-C}_3\text{H}_7(\text{C}_6\text{H}_5)_2\text{PO}$ (RN 2959-75-3)(86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | |

Table 1. Positive Ion Table - Continued

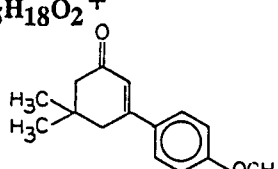
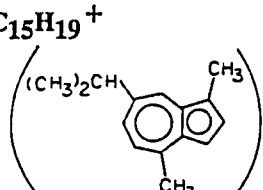
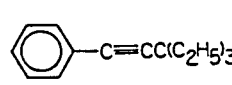
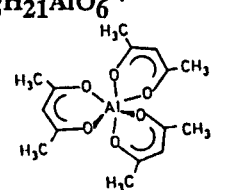
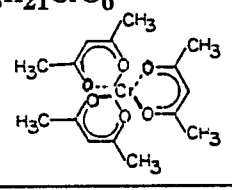
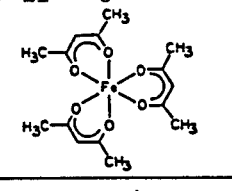
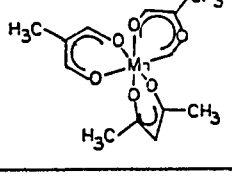
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--|----------------|------------------------------|---------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{15}\text{H}_{18}\text{O}_2^+$ |  | (≤ 8.26) IP from 82PFI/GER. | (≤ 136) | (≤ 568) | -55 | -229 | *EST | 29339-44-4 |
| $\text{C}_{15}\text{H}_{19}^+$ |  | | 165 | 694 | | | | |
| | | | From proton affinity of 1,4-dimethyl-7-isopropylazulene (RN 489-84-9). PA = 233. kcal/mol, 975. kJ/mol. | | | | ck | |
| $\text{C}_{15}\text{H}_{20}^+$ |  | (8.29 ± 0.08) IP from 81ELB/LIE. | (230) | (962) | 39 | 162 | *EST | 80025-09-8 |
| $\text{C}_{15}\text{H}_{21}\text{AlO}_6^+$ |  | (7.78 ± 0.05) IP from 81WES/REI. | (-220) | (-919) | -399 ± 1 | -1669 ± 4 | 80TEL/RAB | 13963-57-0 |
| $\text{C}_{15}\text{H}_{21}\text{CrO}_6^+$ |  | 6.95 ± 0.2 IP is onset of photoelectron band(81WES/REI). | (-182) | (-760) | -342 ± 2 | -1431 ± 7 | 82PIL/SKI | 21679-31-2 |
| $\text{C}_{15}\text{H}_{21}\text{FeO}_6^+$ |  | (7.55) IP is onset of photoelectron band. See also: 81WES/REI. | (-123) | (-515) | -297 ± 1 | -1244 ± 6 | 77PED/RYL | 14024-18-1 |
| $\text{C}_{15}\text{H}_{21}\text{MnO}_6^+$ |  | (7.58 ± 0.05) IP from 81WES/REI. | (-135) | (-564) | -310 ± 1 | -1295 ± 6 | 77PED/RYL | 14284-89-0 |

Table 1. Positive Ion Table - Continued

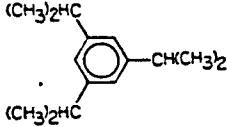
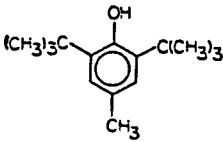
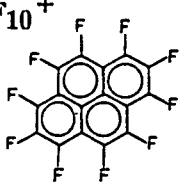
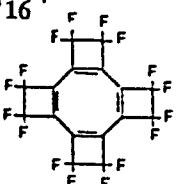
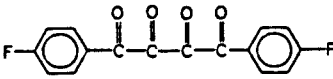
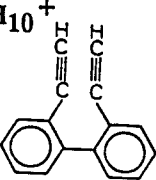

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------------------------|----------------|------------------------------|-------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{15}\text{H}_{24}^+$ |  | (8.24) IP from 84HOW/GON. | (158) | (661) | -32 | -134 | *EST | 717-74-8 |
| $\text{C}_{15}\text{H}_{24}\text{O}^+$ |  | (≤ 7.80) IP from 83CET/LAP. | (≤ 107) | (≤ 449) | -73 | -304 | *EST | 128-37-0 |
| $\text{C}_{16}\text{F}_{10}^+$ |  | (8.36 ± 0.05) | (-167) | (-697) | -359 | -1504 | *EST | 1493-68-1 |
| $\text{C}_{16}\text{F}_{16}^+$ |  | 10.1 IP is onset of photoelectron band (84HEI/WIR). | (-347) | (-1451) | -580 | -2425 | *EST | 42858-85-5 |
| $\text{C}_{16}\text{H}_8\text{F}_2\text{O}_4^+$ |  | (8.7) IP is onset of photoelectron band (85GLE/DOB). | (39) | (165) | -161 | -674 | *EST | 97245-28-8 |
| $\text{C}_{16}\text{H}_{10}^+$ |  | (8.2) IP is onset of photoelectron band (81GLE/SCH). | (340) | (1421) | 151 | 630 | *EST | 18442-29-0 |
| |  | 7.41 See also: 81CLA/ROB. IP value at 298 K from charge transfer equilibrium constant determinations (80MAU) = 7.50 eV. | 222 | 931 | 52 ± 0.2 | 216 ± 1 | 79KUD/KUD2 | 129-00-0 |

Table 1. Positive Ion Table - Continued

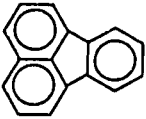
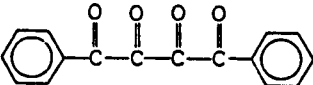
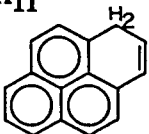
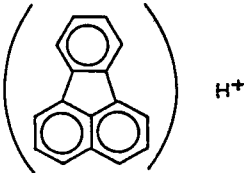
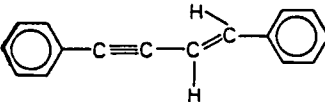
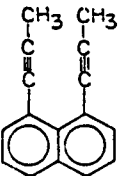
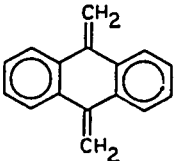
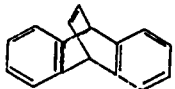
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|---|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{16}\text{H}_{10}^+$  | (7.95±0.04) | (253) | (1056) | 69.2±0.3 | 289.4±1.1 | 81KUD/KUD | 206-44-0 |
| $\text{C}_{16}\text{H}_{10}\text{O}_4^+$  | (8.5) IP is onset of photoelectron band (85GLE/DOB). | (130) | (544) | -66 | -276 | *EST | 19909-44-5 |
| $\text{C}_{16}\text{H}_{11}^+$  | | 211 | 884 | | | | |
| | | From proton affinity of pyrene (RN 129-00-0). PA = 206.1 kcal/mol, 862. kJ/mol. | | | | | |
|  | | 235 | 985 | | | | |
| | | From proton affinity of fluoranthene (RN 206-44-0). PA = 199.3 kcal/mol, 834. kJ/mol. | | | | | |
| $\text{C}_{16}\text{H}_{12}^+$  | (7.5) IP is onset of photoelectron band (80AND/BIC). | (276) | (1154) | 103 | 430 | *EST | 13343-79-8 |
|  | (7.48) IP from 84GLE/SCH. | (311) | (1304) | 139 | -582 | *EST | 22360-77-6 |
|  | (7.6) IP is onset of photoelectron band. | (233) | (976) | 58±3 | 243±12 | *EST | 3302-51-0 |
|  | (7.7) IP is onset of photoelectron band (82HAS/NEU). | (253) | (1060) | 76 | 317 | *EST | |

Table 1. Positive Ion Table - Continued

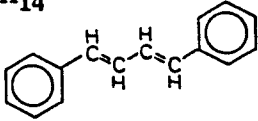
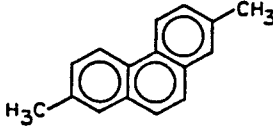
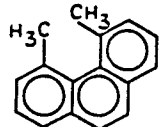
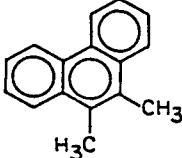
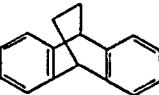
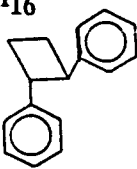
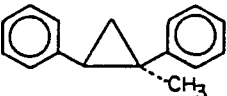
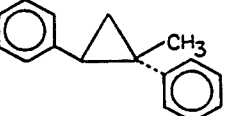
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{16}\text{H}_{14}^+$  | 7.55 | 237 | 993 | 63 | 265 | 69STU/WES | 538-81-8 |
|  | 7.99±0.04 | 218 | 914 | 34±0.5 | 143±2 | 77PED/RYL | 1576-69-8 |
|  | 7.56±0.1 | 220 | 924 | 46±1 | 194±6 | 77PED/RYL | 3674-69-9 |
|  | (8.01±0.05) | (225) | (940) | 40±2 | 167±9 | 77PED/RYL | 604-83-1 |
|  | (8.1) IP is onset of photoelectron band (82HAS/NEU). | (224) | (936) | 37 | 155 | *EST | |
| $\text{C}_{16}\text{H}_{16}^+$  | (8.2±0.1) IP from 84GRO/CHE. | (247) | (1035) | 58 | 244 | 84GRO/CHE | 20071-09-4 |
|  | (7.9) IP from 81KLY/SHU. | (235) | (984) | 53 | 222 | *EST | 14161-72-9 |
|  | (7.9) IP from 81KLY/SHU. | (232) | (971) | 50 | 209 | *EST | 14161-73-0 |

Table 1. Positive Ion Table - Continued

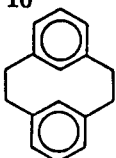
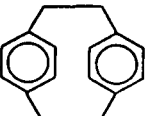
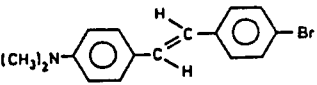
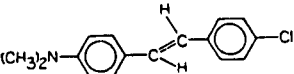
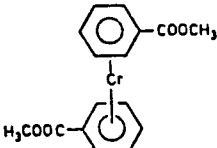
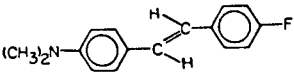
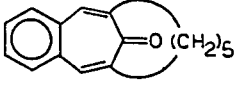
| ION | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------------------------|--------|------------------------------|--------|-------------------|---------------------|
| | Neutral | eV | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{16}\text{H}_{16}^+$ |  | 7.9 | (223) | (933) | 41±2 | 171±7 | 77PED/RYL | 2319-97-3 |
| | | IP is onset of photoelectron band. | | | | | | |
| |  | 7.8±0.1 | 239 | 998 | 59±0.7 | 246±3 | 80NIS/SAK | 1633-22-3 |
| | | IP from 82LEV/LIA, 82GLE/ECK. See also: 81ZHO/KOV. | | | | | | |
| $\text{C}_{16}\text{H}_{16}\text{BrN}^+$ |  | (≤7.04) | (≤200) | (≤835) | 37 | 156 | *EST | 2844-19-1 |
| | | IP from 85CAU/FUR. | | | | | | |
| $\text{C}_{16}\text{H}_{16}\text{ClN}^+$ |  | (≤7.05) | (≤190) | (≤796) | 28 | 116 | *EST | 69957-42-2 |
| | | IP from 85CAU/FUR. | | | | | | |
| $\text{C}_{16}\text{H}_{16}\text{CrO}_4^+$ |  | (≤5.77) | (≤35) | (≤147) | -98 | -410 | *EST | 1272-35-1 |
| | | IP from 82CAB/COW. | | | | | | |
| $\text{C}_{16}\text{H}_{16}\text{FN}^+$ |  | (6.39) | (160) | (671) | 13 | 54 | *EST | 38695-34-0 |
| | | IP is onset of photoelectron band (85CAU/FUR). | | | | | | |
| $\text{C}_{16}\text{H}_{16}\text{O}^+$ |  | (8.0) | (217) | (909) | 33±3 | 137±12 | 77PED/RYL | 25401-39-2 |
| | | IP is onset of photoelectron band. | | | | | | |

Table 1. Positive Ion Table - Continued

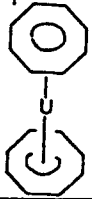
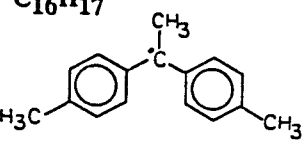
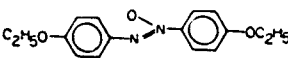
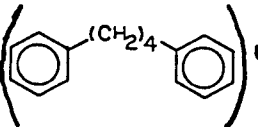
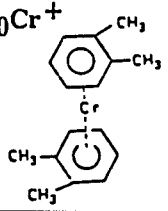
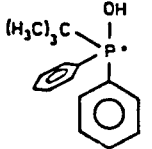
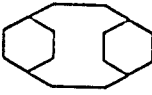
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--|-----------------|------------------------------|---------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{16}\text{H}_{16}\text{U}^+$  | $\leq 6.17 \pm 0.03$ | (≤ 247) | (≤ 1035) | 105 ± 3 | 439 ± 13 | 77TEL/RAB | 11079-26-8 |
| $\text{C}_{16}\text{H}_{17}^+$  | | 193 | 809 | | | | |
| | | From proton affinity of (4- $\text{CH}_3\text{C}_6\text{H}_4$) $_2\text{C}=\text{CH}_2$ (RN 2919-20-2). PA = 215.4 kcal/mol, 901. kJ/mol. | | | | | |
| $\text{C}_{16}\text{H}_{18}\text{N}_2\text{O}_3^+$  | (7.2) | (166) | (695) | 0 | 0 | *EST | 4792-83-0 |
| | IP is onset of photoelectron band (81MIL/CIL). | | | | | | |
| $\text{C}_{16}\text{H}_{19}^+$  | | 194 | 810 | | | | |
| | From proton affinity of $\text{C}_6\text{H}_5(\text{CH}_2)_4\text{C}_6\text{H}_5$ (RN 1083-56-3). PA = 195.9 kcal/mol, 820. kJ/mol. | | | | | | |
| $\text{C}_{16}\text{H}_{20}\text{Cr}^+$  | (≤ 5.21) | (≤ 105) | (≤ 439) | -15 | -64 | *EST | 12092-21-6 |
| | IP from 82CAB/COW. | | | | | | |
| $\text{C}_{16}\text{H}_{20}\text{OP}^+$  | | 83 | 347 | | | | |
| | From proton affinity of t- $\text{C}_4\text{H}_9(\text{C}_6\text{H}_5)_2\text{PO}$ (RN 56598-35-7) (86TRA/MUN). PA = 216 kcal/mol, 904 kJ/mol. | | | | | | |
| $\text{C}_{16}\text{H}_{28}^+$  | (9.1) | (173) | (726) | -36 ± 3 | -152 ± 13 | 77PED/RYL | 283-68-1 |
| | IP is onset of photoelectron band (84GLE/SPA). | | | | | | |

Table 1. Positive Ion Table - Continued

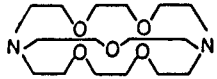
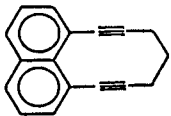
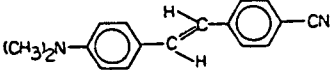
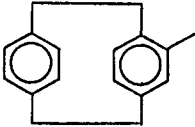
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|-----------------|------------------------------|---------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{16}\text{H}_{32}\text{N}_2\text{O}_5^+$  | (≤ 7.7) IP from 83BAK/ARM. | (≤ 7) | (≤ 31) | -170 | -712 | *EST | 31364-42-8 |
| $\text{C}_{16}\text{H}_{34}\text{N}_2^+$ (E)-((CH ₃) ₃ CCH ₂ C(CH ₃) ₂) ₂ N ₂ | (≤ 8.00) | (≤ 137) | (≤ 575) | -47 \pm 2 | -197 \pm 9 | 80ENG | 55204-43-8 |
| $\text{C}_{16}\text{H}_{36}\text{Sn}^+$ (C ₄ H ₉) ₄ Sn | (8.0) IP is onset of photoelectron band. | (132) | (553) | -52 \pm 1 | -219 \pm 4 | 77PED/RYL | 1461-25-2 |
| (iso-C ₄ H ₉) ₄ Sn | (≤ 8.68) | (≤ 165) | (≤ 689) | -35 | -148 | *EST | 3531-43-9 |
| $\text{C}_{16}\text{H}_{44}\text{Si}_4\text{Ti}^+$ [(CH ₃) ₃ SiCH ₂] ₄ Ti | (8.0) IP is onset of photoelectron band. | (-3) | (-14) | -188 \pm 8 | -786 \pm 33 | 86SIM/BEA | 33948-28-6 |
| $\text{C}_{16}\text{H}_{44}\text{Si}_4\text{Zr}^+$ ((CH ₃) ₃ SiCH ₂) ₄ Zr | (8.2) IP is onset of photoelectron band. | (-9) | (-36) | -198 \pm 8 | -827 \pm 33 | 86SIM/BEA | 32665-18-2 |
| $\text{C}_{17}\text{H}_{12}^+$  | (7.53) IP from 84GLE/SCH. | (349) | (1458) | (175) | (732) | *EST | 32137-40-9 |
| $\text{C}_{17}\text{H}_{16}\text{N}_2^+$  | (≤ 7.31) IP from 85CAU/FUR. | (≤ 261) | (≤ 1094) | 93 | 389 | *EST | |
| $\text{C}_{17}\text{H}_{18}^+$  | (7.6) IP is onset of photoelectron band (81ZHO/KOV). | (226) | (947) | 51 | 214 | *EST | 24262-07-5 |

Table 1. Positive Ion Table - Continued

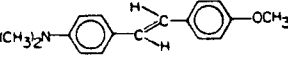
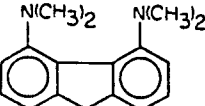
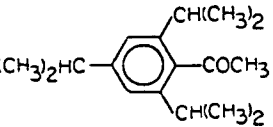
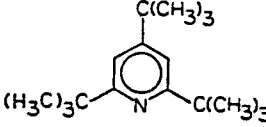
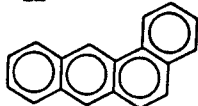
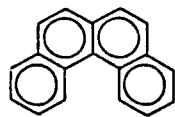
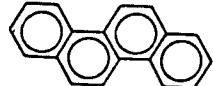
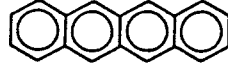
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₇H₁₉NO⁺  | (6.16) | (166) | (696) | 24 | 102 | *EST | 2844-24-8 |
| IP is onset of photoelectron band (85CAU/FUR). | | | | | | | |
| C₁₇H₂₀N₂⁺  | (6.7) | (226) | (945) | 71 | 299 | *EST | 86943-85-3 |
| IP is onset of photoelectron band (84GLE/SCH). | | | | | | | |
| C₁₇H₂₆O⁺  | (8.0) | (104) | (437) | -80 | -335 | *EST | 2234-14-2 |
| IP is onset of photoelectron band (78CEN/FRA). | | | | | | | |
| C₁₇H₂₉N⁺  | 8.20 | (133) | (558) | -56 | -233 | *EST | 20336-15-6 |
| IP is onset of photoelectron band. | | | | | | | |
| C₁₈H₁₂⁺  | 7.43±0.03 | 239 | 1001 | 68±0.2 | 284±1 | 79KUD/KUD2 | 56-55-3 |
| See also: 81AKI/HAR. | | | | | | | |
|  | 7.60 | 245 | 1024 | 70±0.2 | 291±1 | 79KUD/KUD2 | 195-19-7 |
|  | 7.59±0.02 | 243 | 1016 | 67.8±0.2 | 283.7±0.8 | 79KUD/KUD2 | 218-01-9 |
| Value of IP from charge transfer equilibrium constant determinations (80MAU) is in agreement. See also: 81SHA/AKI, 80SHU/BOY. | | | | | | | |
|  | 6.97±0.02 | 229 | 956 | 68±0.2 | 284±1 | 79KUD/KUD2 | 92-24-0 |
| See also: 84STA/MAQ, 80SHU/BOY. | | | | | | | |

Table 1. Positive Ion Table - Continued

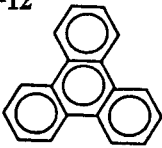
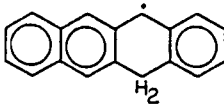
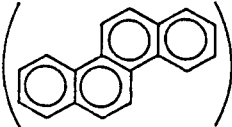
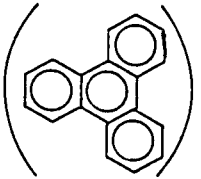
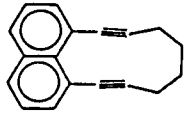
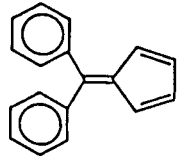
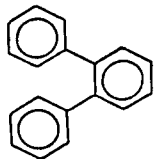
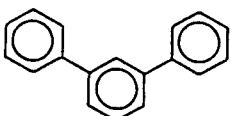
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|---------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{18}\text{H}_{12}^+$  | 7.84±0.01 | 245 | 1026 | 64±0.2 | 270±1 | 79KUD/KUD2 | 217-59-4 |
| | Value of IP from charge transfer equilibrium constant determinations (80MAU) is in agreement. See also: 80SHU/BOY. | | | | | | |
| $\text{C}_{18}\text{H}_{13}^+$  | | 216 | 903 | | | | |
| | From proton affinity of naphthalene (RN 92-24-0). PA = 217.8 kcal/mol, 911. kJ/mol. | | | | | | |
|  H ⁺ | | 227 | 950 | | | | |
| | From proton affinity of chrysene (RN 218-01-9). PA = 201.6 kcal/mol, 843. kJ/mol. | | | | | | |
|  H ⁺ | | 232 | 970 | | | | |
| | From proton affinity of triphenylene (RN 217-59-4). PA = 198.5 kcal/mol, 830.5 kJ/mol. | | | | | | |
| $\text{C}_{18}\text{H}_{14}^+$  | (7.50) | (343) | (1434) | 170 | 710 | *EST | 32137-39-6 |
| | IP from 84GLE/SCH. | | | | | | |
|  | (≤7.96) | (≤280) | (≤1170) | 96±4 | -402±15 | 77PED/RYL | 2175-90-8 |
|  | 8.0 | (252) | (1054) | 68 | 283 | *EST | 84-15-1 |
| | IP is onset of photoelectron band (83KOB, 82LEV/LIA). | | | | | | |
|  | 8.01±0.01 | (252) | (1056) | 68 | 283 | *EST | 92-06-8 |
| | IP from 82LEV/LIA. See also: 83KOB. | | | | | | |

Table 1. Positive Ion Table - Continued


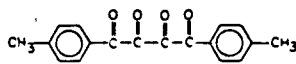
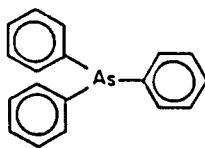
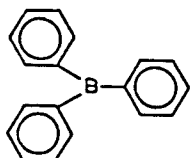
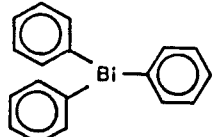
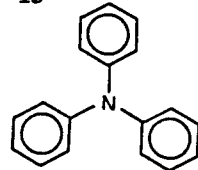
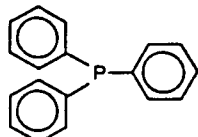
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₈H₁₄⁺  | 7.78±0.01 | (247) | (1034) | 68 | 283 | *EST | 92-94-4 |
| C₁₈H₁₄O₄⁺  | (≤8.8) IP from 85GLE/DOB. | (≤121) | (≤507) | -82 | -342 | *EST | 19909-64-9 |
| C₁₈H₁₅As⁺  | 7.32±0.05 | 266 | 1114 | 98±3 | 408±11 | 79STE | 603-32-7 |
| C₁₈H₁₅B⁺  | (8.60±0.03) | (229) | (960) | 31±2 | 130±8 | 77PED/RYL | 960-71-4 |
| C₁₈H₁₅Bi⁺  | 7.45±0.05 | 317 | 1328 | 146±2 | 609±10 | 79STE | 603-33-8 |
| C₁₈H₁₅N⁺  | 6.80±0.04 | 176 | 734 | 19±0.2 | 78±1 | 78STE | 603-34-9 |
| C₁₈H₁₅P⁺  | 7.39±0.03 IP from 82IKU/KEB, 77ROS/DRA, 82LEV/LIA. | 249 | 1041 | 78±5 | 328±21 | 79STE | 603-35-0 |

Table 1. Positive Ion Table - Continued

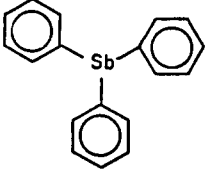
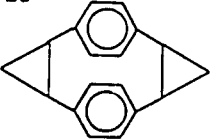
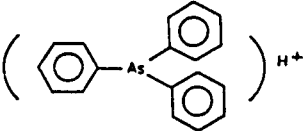
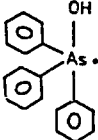
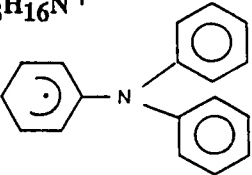
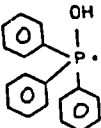
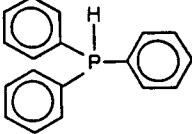
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{18}\text{H}_{15}\text{Sb}^+$ |  | 7.26±0.05 | 271 | 1135 | 104±4 | 435±19 | 79STE | 603-36-1 |
| $\text{C}_{18}\text{H}_{16}^+$ |  | (7.4) | (275) | (1152) | 105 | 438 | *EST | |
| | | | IP is onset of photoelectron band (82GLE/ECK). | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{As}^+$ |  | | 247 | 1034 | | | | |
| | | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{As}$ (RN 603-32-7)(86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{AsO}^+$ |  | | 198 | 827 | | | | |
| | | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{AsO}$ (RN 1153-05-5) (86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{N}^+$ |  | | 168 | 704 | | | | |
| H_2 | | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{N}$ (RN 603-34-9)(86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{OP}^+$ |  | | 154 | 644 | | | | |
| | | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{PO}$ (RN 791-28-6)(86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{P}^+$ |  | | 214 | 896 | | | | |
| | | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{P}$ (RN 603-35-0). PA = (230) kcal/mol, (962) kJ/mol. | | | | | |

Table 1. Positive Ion Table - Continued

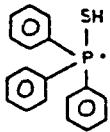
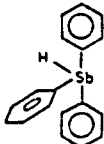
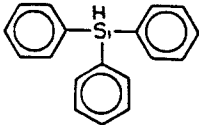
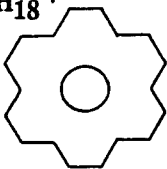
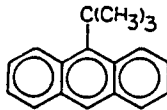
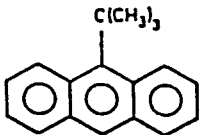
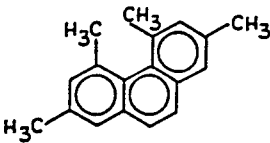
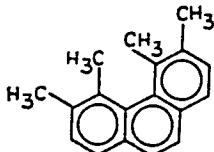
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{18}\text{H}_{16}\text{PS}^+$ | | | | | | | |
|  | | 206 | 860 | | | | |
| | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{PS}$ (RN 3878-45-3)(86TRA/MUN). PA = 216. kcal/mol, 904. kJ/mol. | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{Sb}^+$ | | | | | | | |
|  | | 267 | 1119 | | | | |
| | | From proton affinity of $(\text{C}_6\text{H}_5)_3\text{Sb}$ (RN 603-36-1)(86TRA/MUN). PA = 202. kcal/mol, 846. kJ/mol. | | | | | |
| $\text{C}_{18}\text{H}_{16}\text{Si}^+$ | | | | | | | |
|  | (8.4) | (257) | (1075) | 63 | 265 | *EST | 789-25-3 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{C}_{18}\text{H}_{18}^+$ | | | | | | | |
|  | (6.60) | (276) | (1156) | 124±5 | 519±20 | 74OTH/BUN | 2040-73-5 |
| | | IP from 82BAU/BUN. | | | | | |
|  | (7.13) | (201) | (843) | 37 | 156 | *EST | 62337-65-9 |
| | | IP from 78KLA/KOV, 83KLA/KOV. | | | | | |
|  | 7.13 | (201) | (843) | 37 | 155 | *EST | 13719-97-6 |
| | | IP from 83KLA/KOV. | | | | | |
|  | (7.8±0.1) | (210) | (879) | 30±1 | 126±6 | 77PED/RYL | 7396-38-5 |
|  | (7.5±0.1) | (211) | (881) | 38±1 | 157±6 | 77PED/RYL | 7343-06-8 |

Table 1. Positive Ion Table - Continued

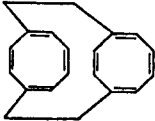
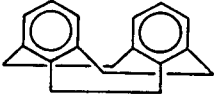
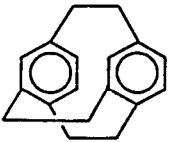
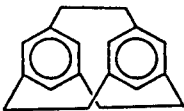
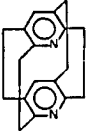
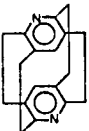
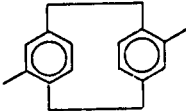
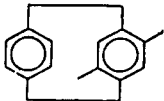
| ION | Neutral | Ionization potential | | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------|--------------------------|--------|------------------------------|--------|-------------------|---------------------|
| | | eV | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₈H₁₈⁺ | | | | | | | | | |
| |  | (7.3) | (221) | (926) | 53 | 222 | | *EST | |
| | | IP is onset of photoelectron band (84ZHO/HEI). | | | | | | | |
| |  | 7.8 | (222) | (931) | 43 | 178 | | *EST | |
| | | IP is onset of photoelectron band. | | | | | | | |
| |  | 7.8 | (236) | (987) | 56 | 234 | | *EST | 58002-98-5 |
| |  | 7.4 | (212) | (886) | 41 | 172 | | *EST | 27165-88-4 |
| | | IP is onset of photoelectron band. | | | | | | | |
| C₁₈H₁₈N₂⁺ | | | | | | | | | |
| |  | (7.4) | (275) | (1150) | 104 | 436 | | *EST | |
| | | IP is onset of photoelectron band (81ZHO/HEI). | | | | | | | |
| |  | (7.6) | (279) | (1169) | 104 | 436 | | *EST | |
| | | IP is onset of photoelectron band (81ZHO/HEI). | | | | | | | |
| C₁₈H₂₀⁺ | | | | | | | | | |
| |  | (≤7.85±0.05) | (≤225) | (≤939) | 43 | 182 | | *EST | |
| | | IP from 81ZHO/KOV. | | | | | | | |
| |  | (7.4) | (214) | (897) | 44 | 183 | | *EST | |
| | | IP is onset of photoelectron band (81ZHO/KOV). | | | | | | | |

Table 1. Positive Ion Table - Continued

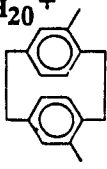
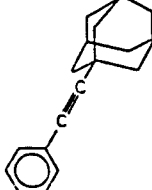
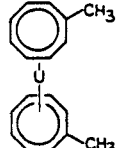
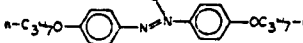
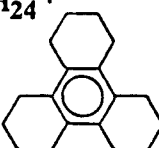
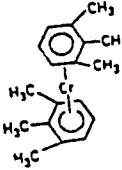
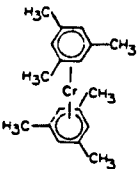
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--|--------------------------|--------------|------------------------------|-------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{18}\text{H}_{20}^+$ |  | $(\leq 7.85 \pm 0.05)$ IP from 81ZHO/KOV. | (≤ 225) | (≤ 939) | 43 | 182 | *EST | |
| |  | (8.12 ± 0.08) IP from 81ELB/LIE. | (240) | (1002) | 52 | 219 | *EST | |
| $\text{C}_{18}\text{H}_{20}\text{U}^+$ |  | (≤ 6.08) IP from 83GRE/PAY. | (≤ 229) | (≤ 959) | 89 | 373 | *EST | 41367-67-3 |
| $\text{C}_{18}\text{H}_{22}\text{N}_2\text{O}_3^+$ |  | (≤ 7.64) IP from 81MIL/CIL. | (≤ 167) | (≤ 697) | -10 | -40 | *EST | 23315-55-1 |
| $\text{C}_{18}\text{H}_{24}^+$ |  | (7.70 ± 0.05) IP from 81HEI/KOV. | (175) | (732) | -3 | -11 | *EST | 1610-39-5 |
| $\text{C}_{18}\text{H}_{24}\text{Cr}^+$ |  | (≤ 5.04) IP from 82CAB/COW. | (≤ 134) | (≤ 562) | 18 | 76 | *EST | 57820-96-9 |
| |  | 4.97 IP from 82CAB/COW. | 130 | 543 | 15 ± 3 | 64 ± 12 | 82PIL/SKI | 1274-07-3 |

Table 1. Positive Ion Table - Continued

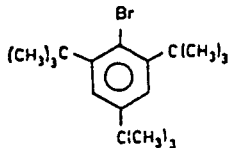
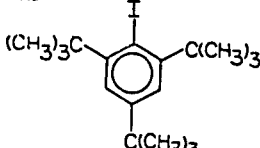
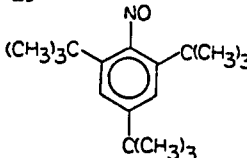
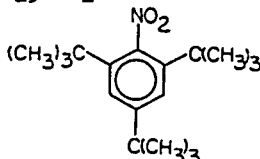
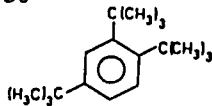
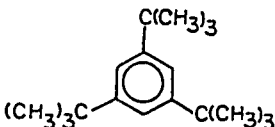
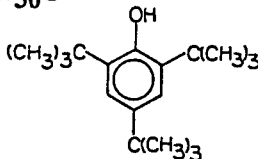
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|----------------|------------------------------|--------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₈H₂₉Br⁺  | 8.0 IP is onset of photoelectron band (83CET/LAP). | (151) | (631) | -33 | -140 | *EST | 3975-77-7 |
| C₁₈H₂₉I⁺  | 7.5 IP is onset of photoelectron band (83CET/LAP). | (156) | (651) | -17 | -73 | *EST | 31039-82-4 |
| C₁₈H₂₉NO⁺  | (≤ 8.69) IP from 83CET/LAP. | (≤ 184) | (≤ 768) | -17 | -70 | *EST | 24973-59-9 |
| C₁₈H₂₉NO₂⁺  | (≤ 8.78) IP from 83CET/LAP. | (≤ 154) | (≤ 646) | -48 | -201 | *EST | 3463-37-4 |
| C₁₈H₃₀⁺  | ($\leq 8.60 \pm 0.07$) | (≤ 164) | (≤ 687) | -34 ± 1.4 | -143 ± 6 | 67ARN/SAN | 1459-11-6 |
|  | (8.19) IP from 84HOW/GON. See also: 83CET/LAP. | (131) | (548) | -58 ± 1 | -242 ± 4 | 77PED/RYL | 1460-02-2 |
| C₁₈H₃₀O⁺  | (7.5) IP is onset of photoelectron band (83CET/LAP). | (80) | (335) | -93 | -389 | *EST | 732-26-3 |

Table 1. Positive Ion Table - Continued

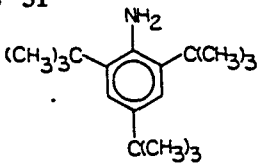
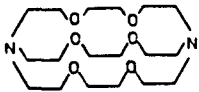
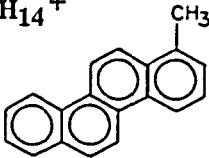
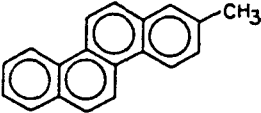
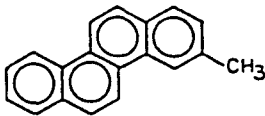
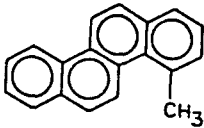
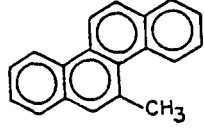
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{18}\text{H}_{31}\text{N}^+$  | (6.9) | (110) | (460) | -49 | -206 | *EST | |
| IP is onset of photoelectron band (83CET/LAP). | | | | | | | |
| $\text{C}_{18}\text{H}_{36}\text{N}_2\text{O}_6^+$  | (≤ 7.8) | (≤ -28) | (≤ -118) | -208 | -871 | *EST | 23978-09-8 |
| IP from 83BAK/ARM. | | | | | | | |
| $\text{C}_{18}\text{H}_{42}\text{N}_3\text{P}^+$ $\text{P}(\text{N}(\text{n}-\text{C}_3\text{H}_7)_2)_3$ | (≤ 7.05) | (≤ 78) | (≤ 325) | -85 | -355 | *EST | 5848-64-6 |
| IP from 82WOR/HAR. | | | | | | | |
| $\text{C}_{19}\text{H}_{14}^+$  | (7.46 ± 0.03) | (226) | (944) | 54 | 224 | *EST | 3351-28-8 |
| IP from 81SHA/AKI. | | | | | | | |
|  | (7.49 ± 0.03) | (226) | (947) | 54 | 224 | *EST | 3351-32-4 |
| IP from 81SHA/AKI. | | | | | | | |
|  | (7.46 ± 0.03) | (226) | (944) | 54 | 224 | *EST | 3351-31-3 |
| IP from 81SHA/AKI. | | | | | | | |
|  | (7.44 ± 0.03) | (230) | (963) | 59 | 245 | *EST | 3351-30-2 |
| IP from 81SHA/AKI. | | | | | | | |
|  | (7.40 ± 0.03) | (229) | (959) | 59 | 245 | *EST | 3697-24-3 |
| IP from 81SHA/AKI. | | | | | | | |

Table 1. Positive Ion Table - Continued

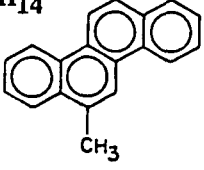
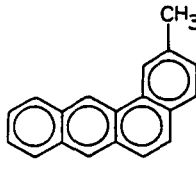
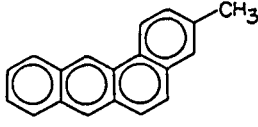
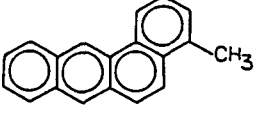
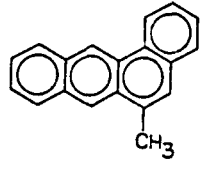
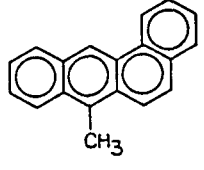
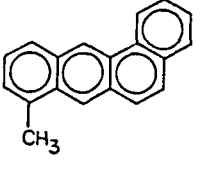
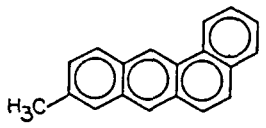
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{19}\text{H}_{14}^+$  | (7.44±0.03) IP from 81SHA/AKI. | (228) | (953) | 57 | 236 | *EST | 1705-85-7 |
|  | (7.30) IP from 81AKI/HAR. | (224) | (938) | 56 | 234 | *EST | 2498-76-2 |
|  | (7.29) IP from 81AKI/HAR. | (224) | (937) | 56 | 234 | *EST | 2498-75-1 |
|  | (7.30) IP from 81AKI/HAR. | (224) | (938) | 56 | 234 | *EST | 316-49-4 |
|  | (7.33) IP from 81AKI/HAR. | (225) | (941) | 56 | 234 | *EST | 316-14-3 |
|  | (7.24) IP from 81AKI/HAR. | (223) | (933) | 56 | 234 | *EST | 2541-69-7 |
|  | (7.33) IP from 81AKI/HAR. | (225) | (941) | 56 | 234 | *EST | 2381-31-9 |
|  | (7.31) IP from 81AKI/HAR. | (224) | (939) | 56 | 234 | *EST | 2381-16-0 |

Table 1. Positive Ion Table - Continued

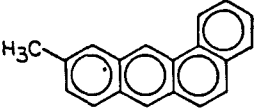
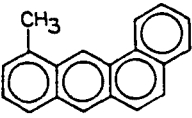
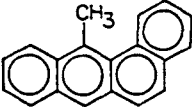
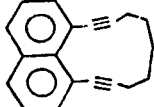
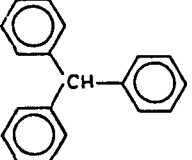
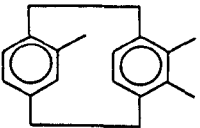
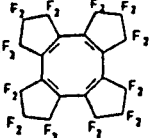
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₁₉H₁₄⁺ | | | | | | | |
|  | (7.30) IP from 81AKI/HAR. | (224) | (938) | 56 | 234 | *EST | 2381-15-9 |
|  | (7.30) IP from 81AKI/HAR. | (224) | (938) | 56 | 234 | *EST | 6111-78-0 |
|  | (7.27) IP from 81AKI/HAR. | (243) | (1015) | 75 | 314 | *EST | 2422-79-9 |
| C₁₉H₁₆⁺ | | | | | | | |
|  | (7.48) IP from 84GLE/SCH. | (337) | (1411) | 165 | 690 | *EST | 87842-94-2 |
|  | 8.34±0.03 | 257 | 1076 | 65±1 | 271±4 | 77PED/RYL | 519-73-3 |
| C₁₉H₂₂⁺ | | | | | | | |
|  | (7.3) IP is onset of photoelectron band (81ZHO/KOV). | (207) | (867) | 39 | 163 | *EST | |
| C₂₀F₂₄⁺ | | | | | | | |
|  | 10.75 IP is onset of photoelectron band (84HEI/WIR). | (-761) | (-3183) | -1009 | -4220 | *EST | 32936-99-5 |

Table 1. Positive Ion Table - Continued

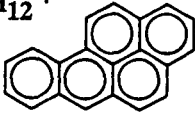
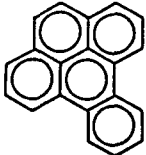
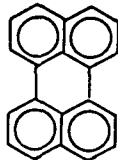
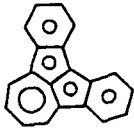
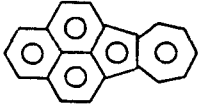
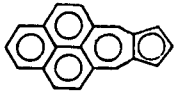
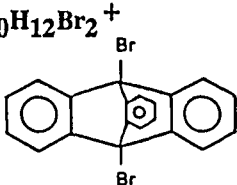
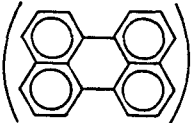
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{20}\text{H}_{12}^+$  | 7.12±0.01 | 233 | 976 | 69 | 289 | 77STE/GOL | 50-32-8 |
|  | 7.41 IP from 79CLA/SCH. | (233) | (976) | 62 | 261 | 77STE/GOL | 192-97-2 |
|  | 6.90±0.01 | 233 | 974 | 74±1 | 308±4 | 77PED/RYL | 198-55-0 |
|  | 6.84 IP from 81SAT/SEK. | (270) | (1129) | 112 | 469 | *EST | 4670-86-4 |
|  | (6.58) | (269) | (1126) | 117 | 491 | *EST | 54100-60-6 |
|  | (6.76) | (273) | (1143) | 117 | 491 | *EST | 6580-41-2 |
| $\text{C}_{20}\text{H}_{12}\text{Br}_2^+$  | (8.1) IP from 83MAR/MAY. | (265) | (1107) | 78 | 326 | *EST | |
| $\text{C}_{20}\text{H}_{13}^+$  | H^+ From proton affinity of perylene (RN 198-55-0). PA = 211.4 kcal/mol, 884. kJ/mol. | 228 | 954 | | | | |

Table 1. Positive Ion Table - Continued

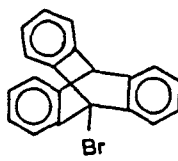
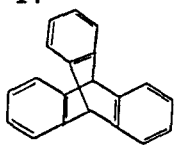
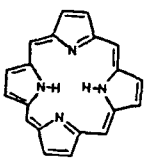
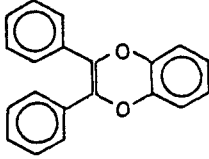
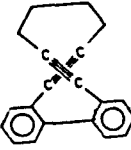
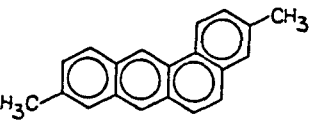
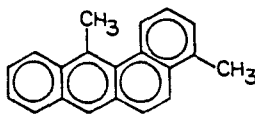
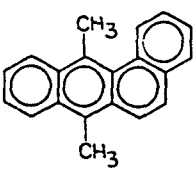
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{20}\text{H}_{13}\text{Br}^+$  | (7.9) IP is onset of photoelectron band (83MAR/MAY). | (260) | (1086) | 77 | 324 | *EST | |
| $\text{C}_{20}\text{H}_{14}^+$  | (7.8) IP is onset of photoelectron band (83MAR/MAY, 82HAS/NEU). | (257) | (1074) | 77 ± 3 | 322 ± 13 | 77PED/RYL | |
| $\text{C}_{20}\text{H}_{14}\text{N}_4^+$  | 6.6 IP from 80DUP/ROB. | -86.4 | -361.5 | -238.6 ± 0.4 | -998.3 ± 1.7 | 70LON/FIN | 101-60-0 |
| $\text{C}_{20}\text{H}_{14}\text{O}_2^+$  | (7.08 ± 0.02) IP from 81BOU/DAG. | (172) | (721) | 9 | 38 | 81BOU/DAG | 75694-46-1 |
| $\text{C}_{20}\text{H}_{16}^+$  | (7.9) IP is onset of photoelectron band (81GLE/SCH). | (354) | (1481) | 172 | 719 | *EST | |
|  | (7.20) IP from 81AKI/HAR. | (210) | (878) | 43.9 ± 0.9 | 183.7 ± 3.9 | 77PED/RYL | 316-51-8 |
|  | (7.18) IP from 81AKI/HAR. | (232) | (971) | 66 | 278 | *EST | 35187-19-0 |
|  | (7.10) IP from 81AKI/HAR. | (230) | (963) | 66.4 ± 1.0 | 277.7 ± 4.4 | 77PED/RYL | 57-97-6 |

Table 1. Positive Ion Table - Continued

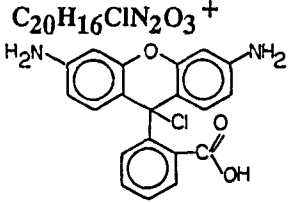
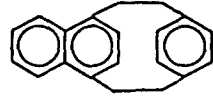
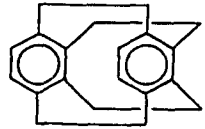
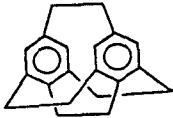
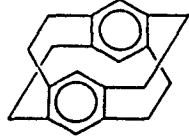
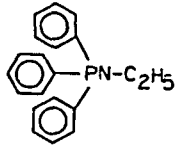
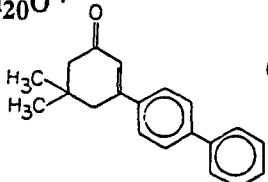
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{20}\text{H}_{16}\text{ClN}_2\text{O}_3^+$  | (7.80±0.05) IP from 81TIM/KOR. | (157) | (656) | -23 | -97 | *EST | |
| $\text{C}_{20}\text{H}_{18}^+$  | (7.4) IP is onset of photoelectron band. | (244) | (1020) | 73 | 306 | *EST | 4432-72-8 |
| $\text{C}_{20}\text{H}_{20}^+$  | (7.6) IP is onset of photoelectron band. | (233) | (975) | 58 | 242 | *EST | |
|  | (7.4) IP is onset of photoelectron band. | (226) | (944) | 55 | 230 | *EST | |
|  | (7.35) IP is onset of photoelectron band (81ZHO/HEI). | (237) | (993) | 68 | 284 | *EST | |
| $\text{C}_{20}\text{H}_{20}\text{NP}^+$  | (≤7.43) | (≤216) | (≤903) | 44±2 | 186±9 | 77PED/RYL | 47182-04-7 |
| $\text{C}_{20}\text{H}_{20}\text{O}^+$  | (≤8.88) IP from 82PFI/GER. | (≤186) | (≤778) | -19 | -79 | *EST | 72036-53-4 |

Table 1. Positive Ion Table - Continued

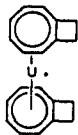
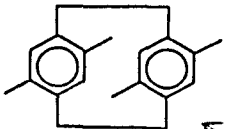
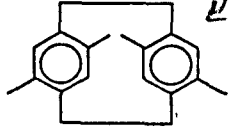
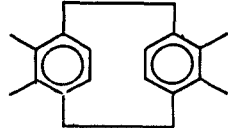
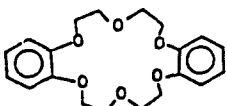
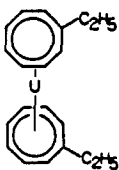
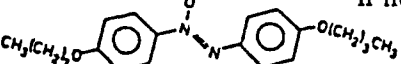
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|-----------------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{20}\text{H}_{20}\text{U}^+$  | (≤ 6.02) IP from 83GRE/PAY. | (≤ 300) | (≤ 1256) | 161 | 675 | *EST | 70377-87-6 |
| $\text{C}_{20}\text{H}_{24}^+$  | (7.2) IP is onset of photoelectron band (81ZHO/KOV). | (200) | (838) | 34 | 144 | *EST | |
|  | ($\leq 7.55 \pm 0.05$) IP from 81ZHO/KOV. | (≤ 209) | (≤ 872) | 34 | 144 | *EST | |
|  | $\leq 7.60 \pm 0.05$ IP from 81ZHO/KOV. | (≤ 210) | (≤ 877) | 34 | 144 | *EST | |
| $\text{C}_{20}\text{H}_{24}\text{O}_6^+$  | (7.5) IP is onset of photoelectron band (83BAK/ARM). | (1.8) | (7.6) | -171 | -716 | *EST | |
| $\text{C}_{20}\text{H}_{24}\text{U}^+$  | (5.9) IP is onset of photoelectron band (83GRE/PAY). | (215) | (900) | 79 | 331 | *EST | 37274-10-5 |
| $\text{C}_{20}\text{H}_{26}\text{N}_2\text{O}_3^+$  | (≤ 7.61) IP from 81MIL/CIL. | (≤ 156) | (≤ 654) | -19 | -80 | *EST | 17051-01-3 |

Table 1. Positive Ion Table - Continued

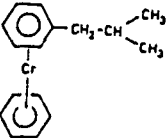
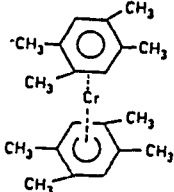
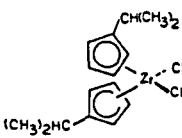
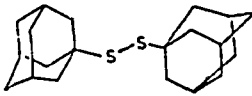
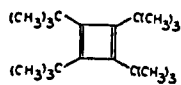
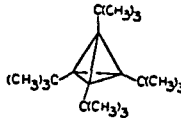
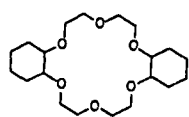
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|----------------------------|--------------------------|----------------|------------------------------|---------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{20}\text{H}_{28}\text{Cr}^+$ |  | (≤ 5.23) | (≤ 128) | (≤ 538) | 8 | 33 | *EST | 51951-64-5 |
| | IP from 82CAB/COW. | | | | | | | |
| |  | (≤ 4.85) | (≤ 108) | (≤ 454) | -3 | -14 | *EST | 57820-98-1 |
| IP from 82CAB/COW. | | | | | | | | |
| $\text{C}_{20}\text{H}_{30}\text{Cl}_2\text{Zr}^+$ |  | 7.1 | (60) | (252) | -103 ± 1 | -433 ± 4 | 82PIL/SKI | 54039-38-2 |
| IP is onset of photoelectron band (81CIL/CON). | | | | | | | | |
| $\text{C}_{20}\text{H}_{30}\text{S}_2^+$ |  | (7.5) | (125) | (524) | -48 | -200 | *EST | 34895-45-9 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_{20}\text{H}_{36}^+$ |  | (5.9) | (226) | (946) | 90 ± 5.5 | 377 ± 23 | *EST | 66809-05-0 |
| | IP is onset of photoelectron band. | | | | | | | |
| |  | (7.1) | (257) | (1076) | 93 ± 5.5 | -391 ± 23 | *EST | 66809-06-1 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_{20}\text{H}_{36}\text{O}_6^+$ |  | (8.6) | (-59) | (-248) | -258 | -1078 | *EST | 16069-36-6 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{C}_{20}\text{H}_{44}\text{Hf}^+$ | $((\text{CH}_3)_3\text{CCH}_2)_4\text{Hf}$ | (8.1) | (132) | (554) | -54 ± 9 | -228 ± 33 | 86SIM/BEA | 50654-35-8 |
| IP is onset of photoelectron band. | | | | | | | | |

Table 1. Positive Ion Table - Continued

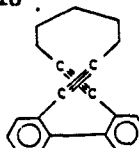
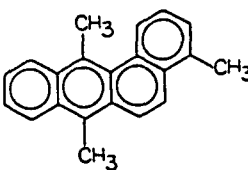
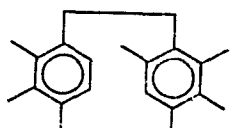
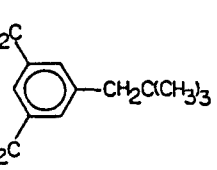
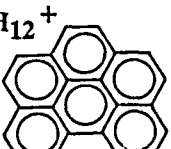
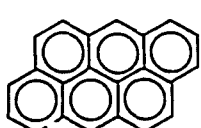
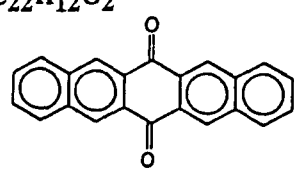
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{20}\text{H}_{44}\text{Ti}^+$ $((\text{CH}_3)_3\text{CCH}_2)_4\text{Ti}$ | (7.7) IP is onset of photoelectron band. | (140) | (586) | -38±8 | -157±33 | 86SIM/BEA | 36945-13-8 |
| $\text{C}_{21}\text{H}_{18}^+$  | (7.7) IP is onset of photoelectron band (81GLE/SCH). | (345) | (1442) | 167 | 699 | *EST | |
|  | (7.06±0.03) IP from 81AKI/HAR. | (217) | (909) | 54 | 228 | *EST | 35187-24-7 |
| $\text{C}_{21}\text{H}_{26}^+$  | (≤7.55±0.05) IP from 81ZHO/KOV. | (≤199) | (≤831) | 25 | 103 | *EST | |
| $\text{C}_{21}\text{H}_{36}^+$  | (8.17) IP from 84HOW/GON. | (111) | (466) | -77 | -322 | *EST | 21411-39-2 |
| $\text{C}_{22}\text{H}_{12}^+$  | 7.15 See also: 80MAU. | 237 | 992 | 72 | 302 | 77STE/GOL | 191-24-2 |
|  | (6.92±0.04) | (233) | (978) | 74 | 310 | 77STE/GOL | 191-26-4 |
| $\text{C}_{22}\text{H}_{12}\text{O}_2^+$  | (8.07±0.05) | (196) | (823) | 10±2 | 44±9 | 77PED/RYL | 3029-32-1 |

Table 1. Positive Ion Table - Continued

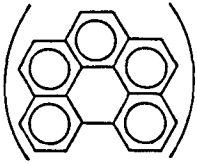
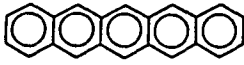
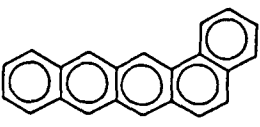
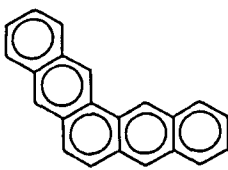
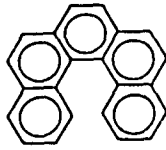
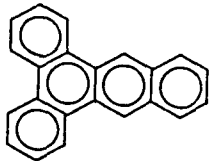
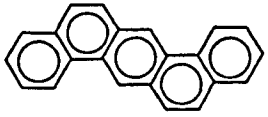
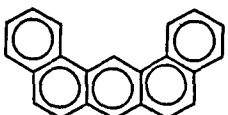
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|-----------------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{22}\text{H}_{13}^+$ | | | | | | | |
|  | H^+ | | 229 960 | | | | |
| | | From proton affinity of 1,12-benzoperylene (RN 191-24-2). PA = 208.5 kcal/mol, 872. kJ/mol. | | | | | |
| $\text{C}_{22}\text{H}_{14}^+$ | | | | | | | |
|  | | 6.61±0.02 | (237) (992) | | 85 354 | *EST | 135-48-8 |
| | | See also: 84STA/MAQ. | | | | | |
|  | | 7.00 | (244) (1020) | | 82 345 | *EST | 226-88-0 |
| | | See also: 75CLA/SCH. | | | | | |
|  | | 7.27±0.02 | (250) (1046) | | 82 345 | *EST | 222-93-5 |
| | | See also: 79CLA/SCH. | | | | | |
|  | | 7.47±0.04 | (269) (1127) | | 97 406 | *EST | 188-52-3 |
| | | See also: 75CLA/SCH. | | | | | |
|  | | 7.39±0.02 | 251 1049 | | 80 336 | 77STE/GOL | 215-58-7 |
| | | See also: 75CLA/SCH, 79CLA/SCH. | | | | | |
|  | | 7.38±0.04 | 250 1048 | | 80 336 | 77STE/GOL | 53-70-3 |
| | | See also: 75CLA/SCH. | | | | | |
|  | | (7.40±0.02) | (251) (1050) | | 80 336 | 77STE/GOL | 224-41-9 |
| | | See also: 75CLA/SCH. | | | | | |

Table 1. Positive Ion Table - Continued

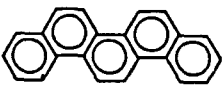
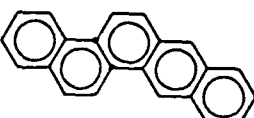
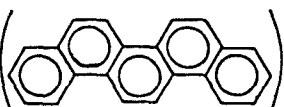
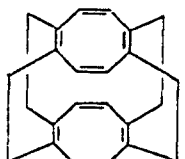

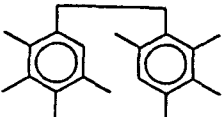
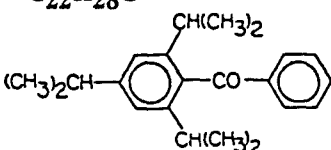
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₂₂H₁₄⁺ | | | | | | | |
|  | 7.48 | (250) | (1048) | 78 | 326 | *EST | 213-46-7 |
| | IP from charge transfer equilibrium constant determinations (80MAU), in agreement with value from 79CLA/SCH. | | | | | | |
|  | (7.14±0.04) | (245) | (1024) | 80 | 336 | *EST | 214-17-5 |
| C₂₂H₁₅⁺ | | | | | | | |
|  | | 240 | 1005 | | | | |
| H ⁺ | From proton affinity of picene (RN 213-46-7). PA = 203.4 kcal/mol, 851. kJ/mol. | | | | | | |
| C₂₂H₂₂⁺ | | | | | | | |
|  | (7.35) | (256) | (1072) | 87 | 363 | *EST | |
| | IP is onset of photoelectron band (84ZHO/HEI). | | | | | | |
| C₂₂H₂₂O₄⁺ | | | | | | | |
|  | (≤8.6) | (≤76) | (≤318) | -122 | -512 | *EST | 19909-65-0 |
| | IP from 85GLE/DOB. | | | | | | |
| C₂₂H₂₈⁺ | | | | | | | |
|  | (7.0) | (176) | (737) | 15 | 62 | *EST | |
| | IP is onset of photoelectron band (81ZHO/KOV). | | | | | | |
| C₂₂H₂₈O⁺ | | | | | | | |
|  | (7.9) | (137) | (574) | -45±2 | -189±7 | 82INA/MUR | 33574-11-7 |
| | IP is onset of photoelectron band (78CEN/FRA). | | | | | | |

Table 1. Positive Ion Table - Continued

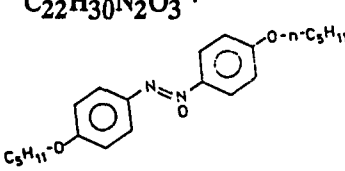
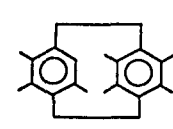
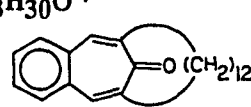
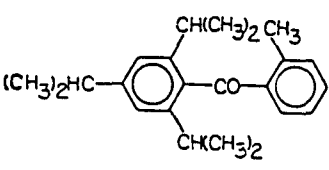
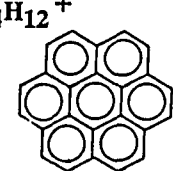
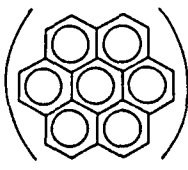
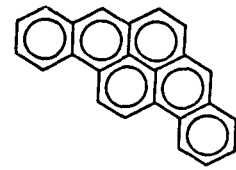
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|---|--------------------------|----------------|------------------------------|---------------|----------------------|---|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}_3^+$ |  | (≤ 7.63) IP from 81MIL/CIL. | (≤ 147) | (≤ 616) | -29 | -120 | *EST | 19482-05-4 |
| $\text{C}_{23}\text{H}_{30}^+$ |  | ($\leq 7.40 \pm 0.05$) IP from 81ZHO/KOV. | (≤ 182) | (≤ 763) | 12 | 49 | *EST | |
| $\text{C}_{23}\text{H}_{30}\text{O}^+$ |  | ($\leq 8.15 \pm 0.03$) | (≤ 150) | (≤ 626) | -38 ± 4 | -160 ± 15 | 77PED/RYL | 25401-43-8 |
| |  | (7.6) IP is onset of photoelectron band (78CEN/FRA). | (125) | (523) | -50 | -210 | *EST | 78823-28-6 |
| $\text{C}_{24}\text{H}_{12}^+$ |  | 7.29 IP at 298 K from charge transfer equilibria, 7.26 eV (80MAU, re-evaluated). See also: 81CLA/ROB. | 245 | 1026 | 77 | 323 | 77STE/GOL | 191-07-1 |
| $\text{C}_{24}\text{H}_{13}^+$ |  | | 238 | 995 | | | | From proton affinity of coronene (RN 191-07-1). PA = 205.0 kcal/mol, 858. kJ/mol. |
| $\text{C}_{24}\text{H}_{14}^+$ |  | (6.95) IP from 79CLA/SCH. | (243) | (1018) | 83 | 348 | 77STE/GOL | 189-55-9 |

Table 1. Positive Ion Table - Continued

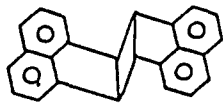
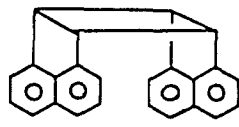
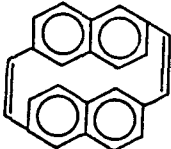
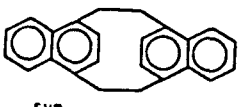
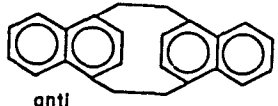
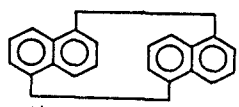
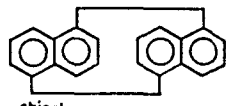
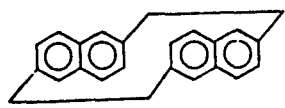
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{24}\text{H}_{16}^+$ | | | | | | | |
|  | (7.48) IP from 82GLE/GUB. | (321) | (1342) | 148 | 620 | *EST | 14620-98-5 |
|  | 7.58 IP from 82GLE/GUB. | (323) | (1351) | 148 | 620 | *EST | 15065-28-8 |
|  | (7.1) IP is onset of photoelectron band. | (305) | (1274) | 141 | 589 | *EST | 43012-17-5 |
| $\text{C}_{24}\text{H}_{20}^+$ | | | | | | | |
|  syn | (7.0) IP is onset of photoelectron band. | (250) | (1046) | 89 | 371 | *EST | 14724-91-5 |
|  anti | (7.3) IP is onset of photoelectron band. | (257) | (1075) | 89 | 371 | *EST | 17341-02-5 |
|  achiral | (6.8) IP is onset of photoelectron band. | (245) | (1027) | 89 | 371 | *EST | 54835-57-3 |
|  chiral | (7.1) IP is onset of photoelectron band. | (253) | (1058) | 89 | 373 | *EST | 54835-57-3 |
|  | (7.3) IP is onset of photoelectron band. | (240) | (1005) | 72 | 301 | *EST | 73608-51-2 |

Table 1. Positive Ion Table - Continued

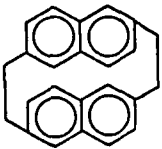
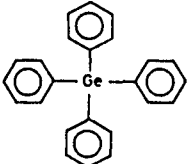
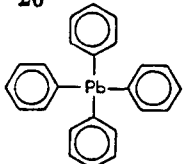
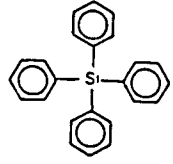
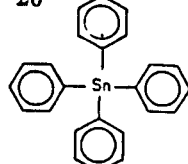
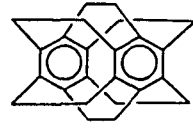
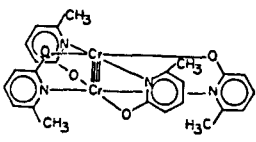
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{24}\text{H}_{20}^+$  | (7.0) IP is onset of photoelectron band. | (317) | (1325) | 155 | 650 | *EST | 7130-24-7 |
| $\text{C}_{24}\text{H}_{20}\text{Ge}^+$  | (8.1) IP is onset of photoelectron band (84NOV/POT). | (293) | (1226) | (106±6) | (445±24) | 77PED/RYL | 1048-05-1 |
| $\text{C}_{24}\text{H}_{20}\text{Pb}^+$  | (8.0) IP is onset of photoelectron band (84NOV/POT). | (346) | (1446) | 161±4 | 674±15 | 78STE | 595-89-1 |
| $\text{C}_{24}\text{H}_{20}\text{Si}^+$  | (8.50±0.03) See also: 84NOV/POT. | (278) | (1162) | 82±1 | 342±6 | 82PIL/SKI | 1048-08-4 |
| $\text{C}_{24}\text{H}_{20}\text{Sn}^+$  | (8.34±0.03) See also: 84NOV/POT. | (329) | (1378) | 137±2 | 573±8 | 77KAN/MOR | 595-90-4 |
| $\text{C}_{24}\text{H}_{24}^+$  | (7.3) IP is onset of photoelectron band. | (253) | (1057) | 84 | 353 | *EST | 60144-50-5 |
| $\text{C}_{24}\text{H}_{24}\text{Cr}_2\text{N}_4\text{O}_4^+$  | (6.5) IP is onset of photoelectron band. | (-41) | (-171) | -191±2 | -798±9 | 81CAV/GAR | 67634-82-6 |

Table 1. Positive Ion Table - Continued

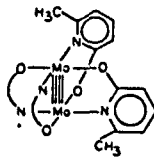
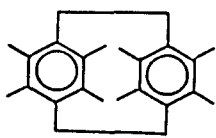
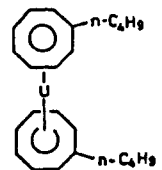
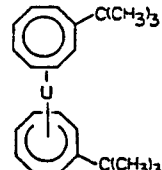
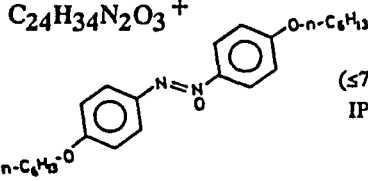
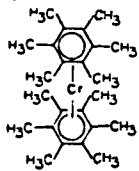
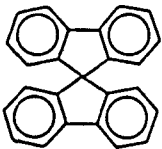
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{24}\text{H}_{24}\text{Mo}_2\text{N}_4\text{O}_4^+$  | (5.5) | (-16) | (-66) | -143±2 | -597±9 | 81CAV/GAR | 67634-80-4 |
| IP is onset of photoelectron band. | | | | | | | |
| $\text{C}_{24}\text{H}_{32}^+$  | (6.9) | (168) | (702) | 9 | 36 | *EST | |
| IP is onset of photoelectron band (81ZHO/KOV). | | | | | | | |
| $\text{C}_{24}\text{H}_{32}\text{U}^+$  | (≤6.05) | (≤199) | (≤831) | 59 | 247 | *EST | 37274-12-7 |
| IP from 83GRE/PAY. | | | | | | | |
|  | (≤6.03) | (≤193) | (≤809) | 54 | 227 | *EST | 63230-70-6 |
| IP from 83GRE/PAY. | | | | | | | |
| $\text{C}_{24}\text{H}_{34}\text{N}_2\text{O}_3^+$  | (≤7.55) | (≤136) | (≤568) | -38 | -160 | *EST | 2587-42-0 |
| IP from 81MIL/CIL. | | | | | | | |
| $\text{C}_{24}\text{H}_{36}\text{Cr}^+$  | (≤4.68) | (≤87) | (≤364) | -21±3 | -88±12 | 82PIL/SKI | 12243-39-9 |
| $\text{C}_{25}\text{H}_{16}^+$  | (7.5) | (286) | (1199) | 114 | 475 | *EST | 159-66-0 |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

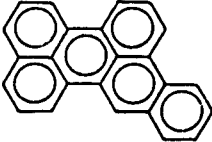
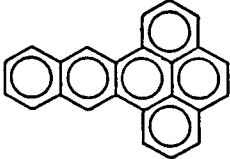
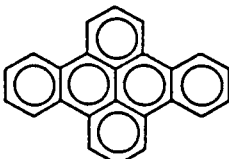
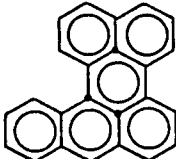
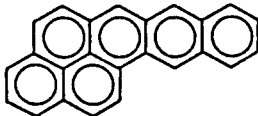
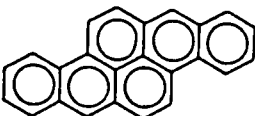
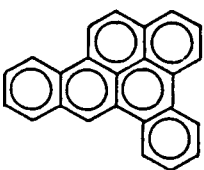
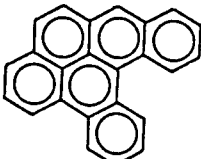
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{24}\text{H}_{14}^+$ | | | | | | | |
|  | (6.89) | (240) | (1004) | 81 | 339 | *EST | 197-70-6 |
|  | (7.35) IP from 79CLA/SCH. | (253) | (1057) | 83 | 348 | *EST | 193-09-9 |
|  | 7.39 IP from 79CLA/SCH. | (249) | (1042) | 79 | 329 | *EST | 192-51-8 |
|  | (6.71) | (237) | (991) | 82 | 344 | *EST | 191-85-5 |
|  | (6.82) IP from 79CLA/SCH. | (243) | (1015) | 85 | 357 | *EST | |
|  | (6.82) IP from 79CLA/SCH. | (240) | (1006) | 83 | 348 | 77STE/GOL | 189-64-0 |
|  | (7.11) IP from 79CLA/SCH. | (245) | (1025) | 81 | 339 | *EST | |
|  | (7.07) IP from 79CLA/SCH. | (245) | (1026) | 82 | 344 | *EST | |

Table 1. Positive Ion Table - Continued

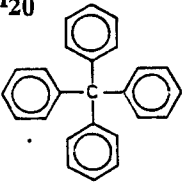
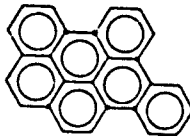
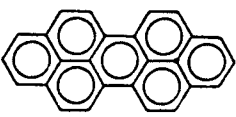
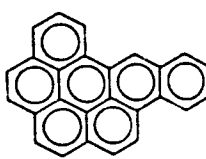
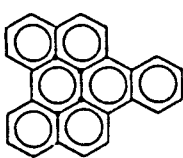
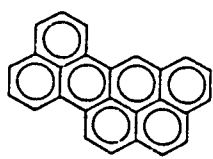
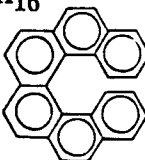
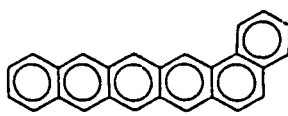
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{25}\text{H}_{20}^+$  | (8.0) | (280) | (1170) | 95±1 | 398±4 | 77PED/RYL | 630-76-2 |
| | IP is onset of photoelectron band (84NOV/POT). | | | | | | |
| $\text{C}_{26}\text{H}_{14}^+$  | (7.12) | (248) | (1038) | 84 | 351 | *EST | 190-95-4 |
|  | 6.72±0.02 | (241) | (1008) | 86 | 360 | *EST | 188-96-5 |
|  | (6.99) | (247) | (1034) | 86 | 360 | *EST | 5869-30-7 |
|  | (6.96) | (244) | (1022) | 84 | 351 | *EST | 190-84-1 |
|  | (6.82±0.04) | (243) | (1018) | 86 | 360 | *EST | 188-89-6 |
| $\text{C}_{26}\text{H}_{16}^+$  | (7.37) | (290) | (1213) | 120 | 502 | *EST | 187-83-7 |
|  | (6.61±0.02) | (251) | (1050) | 99 | 413 | *EST | 239-98-5 |

Table 1. Positive Ion Table - Continued

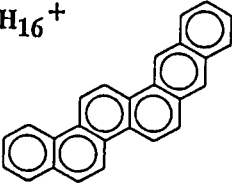
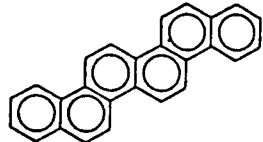
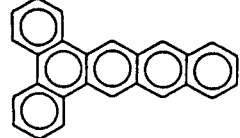
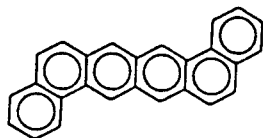
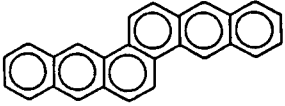
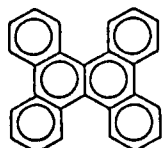

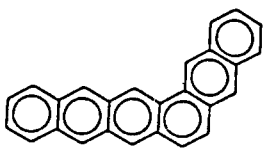
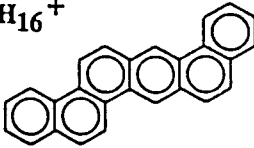
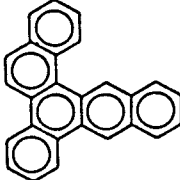
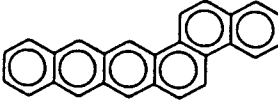
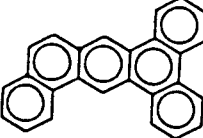
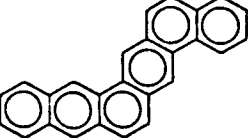
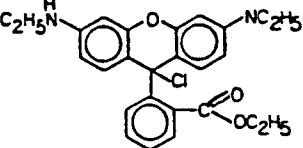
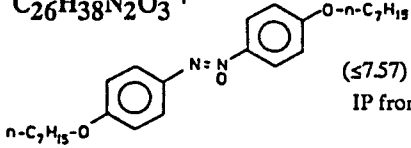
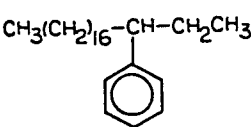
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------------|---|--|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{26}\text{H}_{16}^+$ |  | 7.17±0.02 | (267) | (1117) | 102 | 425 | *EST | 217-42-5 |
| |  | (7.36) | (262) | (1095) | 92 | 385 | *EST | 217-37-8 |
| |  | 6.97±0.02 See also: 75CLA/SCH, 79CLA/SCH. | (255) | (1067) | 94 | 394 | *EST | 216-00-2 |
| |  | (6.99±0.02) See also: 75CLA/SCH. | (258) | (1078) | 96 | 403 | *EST | 227-04-3 |
| |  | (6.97±0.04) | (257) | (1076) | 96 | 403 | *EST | 217-54-9 |
| |  | 7.20±0.02 See also: 75CLA/SCH, 79CLA/SCH. | (256) | (1072) | 90 | 377 | *EST | 191-68-4 |
| |  | (6.36±0.02) See also: 75CLA/SCH. | (247) | (1035) | 101 | 422 | *EST | 258-31-1 |
| |  | (6.92±0.02) | (258) | (1080) | 99 | 413 | *EST | 222-78-6 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|-----------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{26}\text{H}_{16}^+$  | (7.19±0.02) | (253) | (1057) | 87 | 363 | *EST | 220-77-9 |
|  | 7.15±0.02 | (258) | (1080) | 93 | 390 | *EST | 196-64-5 |
|  | (6.83±0.02) | (254) | (1062) | 96 | 403 | *EST | 220-82-6 |
|  | 7.40±0.02 See also: 75CLA/SCH. | (263) | (1099) | 92 | 385 | *EST | 215-26-9 |
|  | (7.20±0.02) | (262) | (1098) | 96 | 403 | *EST | 222-54-8 |
| $\text{C}_{26}\text{H}_{25}\text{ClN}_2\text{O}_3^+$  | (6.94±0.05) IP from 81TIM/KOR. | (115) | (483) | -45 | -187 | *EST | 989-38-8 |
| $\text{C}_{26}\text{H}_{38}\text{N}_2\text{O}_3^+$  | (≤7.57) IP from 81MIL/CIL. | (≤127) | (≤530) | -48 | -200 | *EST | 2635-26-9 |
| $\text{C}_{26}\text{H}_{46}^+$  | (8.95±0.10) | (123) | (516) | -83±1.2 | -348±5 | 77PED/RYL | 72557-70-1 |

Note in the case of the above is C.G. = CASM
FOR IUPAC + STRUCT ≠ REG

Table 1. Positive Ion Table - Continued

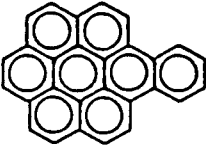
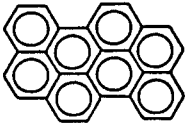
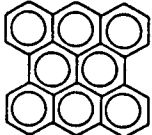
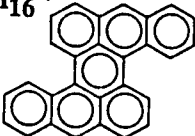
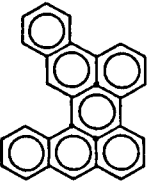
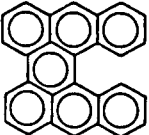
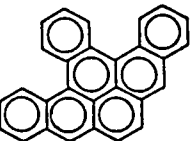
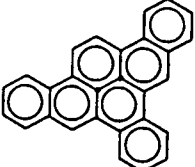
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{28}\text{H}_{14}^+$  | (7.08) | (248) | (1037) | 84 | 353 | *EST | 190-70-5 |
|  | (6.92±0.04) | (249) | (1040) | 89 | 372 | *EST | 190-71-6 |
|  | (6.30) | (234) | (980) | 89 | 372 | *EST | 190-39-6 |
| $\text{C}_{28}\text{H}_{16}^+$  | (6.51) | (252) | (1055) | 102 | 426 | *EST | 191-87-7 |
|  | (6.64) | (247) | (1033) | 94 | 392 | *EST | 191-81-1 |
|  | (6.51) | (247) | (1035) | 97 | 406 | *EST | 190-36-3 |
|  | (6.96) IP from 79CLA/SCH. | (262) | (1098) | 102 | 426 | *EST | 191-20-8 |
|  | (6.99) IP from 79CLA/SCH. | (256) | (1072) | 95 | 397 | *EST | 192-47-2 |

Table 1. Positive Ion Table - Continued

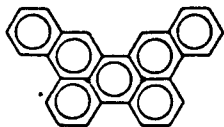
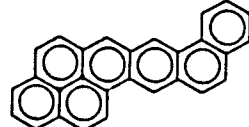
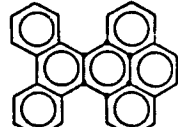
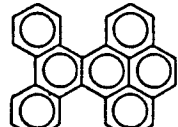
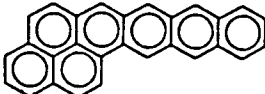
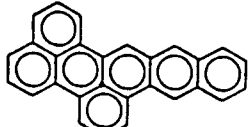
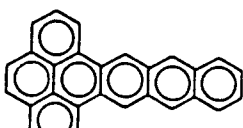
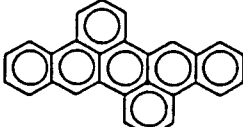
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{28}\text{H}_{16}^+$  | (6.85) | (253) | (1058) | 95 | 397 | *EST | 197-69-3 |
|  | (6.83) IP from 79CLA/SCH. | (257) | (1075) | 99 | 416 | *EST | |
|  | (7.00) IP from 79CLA/SCH. | (252) | (1055) | 91 | 380 | *EST | |
|  | (7.00±0.04) | (250) | (1044) | 88 | 369 | *EST | 385-14-8 |
|  | (6.57) IP from 79CLA/SCH. | (253) | (1059) | 102 | 425 | *EST | 196-45-2 |
|  | (6.82) | (254) | (1064) | 97 | 406 | *EST | 14147-38-7 |
|  | (6.95) IP from 79CLA/SCH. | (260) | (1086) | 99 | 416 | *EST | 193-11-3 |
|  | (6.86) | (253) | (1059) | 95 | 397 | *EST | 197-74-0 |

Table 1. Positive Ion Table - Continued

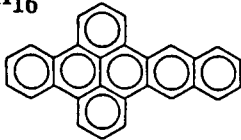
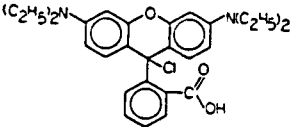
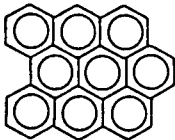
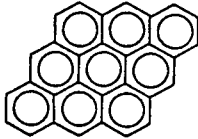
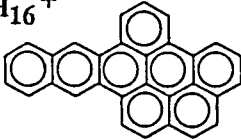
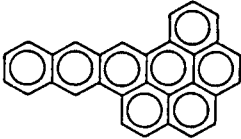
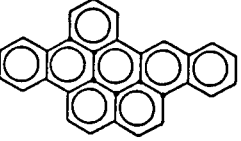
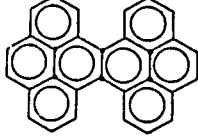
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---|-------------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{28}\text{H}_{16}^+$ |  | (7.33±0.04) See also: 79CLA/SCH. | (264) | (1104) | 95 | 397 | *EST | 192-59-6 |
| $\text{C}_{28}\text{H}_{32}\text{ClN}_2\text{O}_3^+$ |  | (6.70±0.05) IP from 81TIM/KOR. | (109) | (455) | -46 | -191 | *EST | |
| $\text{C}_{30}\text{H}_{14}^+$ |  | (6.50) | (244) | (1021) | 94 | 394 | *EST | 190-31-8 |
| |  | (6.42±0.02) | (244) | (1022) | 96 | 403 | *EST | 190-55-6 |
| $\text{C}_{30}\text{H}_{16}^+$ |  | (7.04) | (269) | (1125) | 107 | 446 | *EST | 14258-76-5 |
| |  | (6.78) | (265) | (1110) | 109 | 455 | *EST | 5869-31-8 |
| |  | (6.97) | (254) | (1063) | 93 | 391 | *EST | 190-87-4 |
| |  | (6.90±0.04) | (259) | (1086) | 100 | 420 | *EST | 385-13-7 |

Table 1. Positive Ion Table - Continued

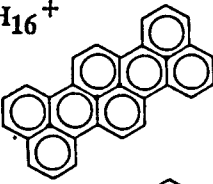
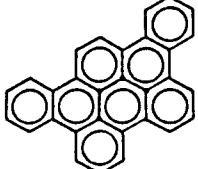
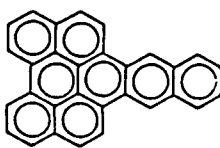
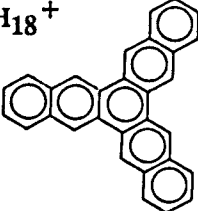
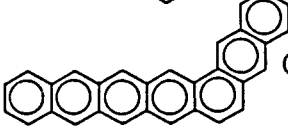
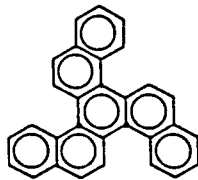
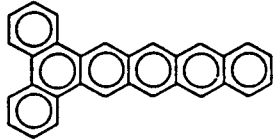
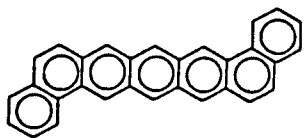
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{30}\text{H}_{16}^+$ | | | | | | | |
|  | (6.42±0.02) | (246) | (1029) | 98 | 409 | *EST | 188-72-7 |
|  | (7.13) | (260) | (1088) | 96 | 400 | *EST | 190-81-8 |
|  | (6.77) | (256) | (1072) | 100 | 419 | *EST | 190-85-2 |
| $\text{C}_{30}\text{H}_{18}^+$ | | | | | | | |
|  | (7.35±0.02) | (280) | (1171) | 110 | 462 | *EST | 196-62-3 |
|  | (6.59±0.02) | (280) | (1172) | 128 | 536 | *EST | 222-81-1 |
|  | (7.19±0.02) | (273) | (1144) | 108 | 450 | *EST | 27798-46-5 |
|  | 6.62±0.02 See also: 75CLA/SCH. | (263) | (1101) | 110 | 462 | *EST | 216-08-0 |
|  | (6.64±0.02) | (264) | (1103) | 110 | 462 | *EST | 227-09-8 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------|-------------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{30}\text{H}_{18}^+$ | | (7.17±0.02) | (276) | (1154) | 110 | 462 | *EST | 213-44-5 |
| | | (6.89±0.02) | (274) | (1145) | 115 | 480 | *EST | 222-75-3 |
| | | (7.04±0.02) | (275) | (1150) | 113 | 472 | *EST | 222-58-2 |
| | | (7.25) | (314) | (1315) | 147 | 616 | *EST | 16914-68-4 |
| | | 7.43±0.02 See also: 75CLA/SCH. | (275) | (1151) | 104 | 434 | *EST | 215-11-2 |
| | | (6.99±0.02) See also: 75CLA/SCH. | (269) | (1127) | 108 | 453 | *EST | 215-96-3 |
| $\text{C}_{30}\text{H}_{36}\text{ClN}_2\text{O}_3^+$ | | (6.58±0.05) IP from 81TIM/KOR. | (99) | (416) | -52 | -219 | *EST | |
| $\text{C}_{32}\text{H}_{14}^+$ | | 6.71 See also: 81CLA/ROB. | 254 | 1062 | 99 | 415 | 77STE/GOL | 190-26-1 |

Table 1. Positive Ion Table - Continued

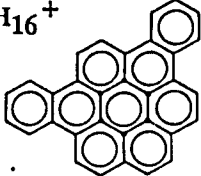
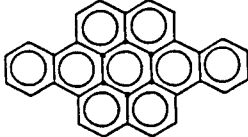
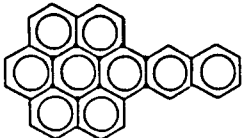
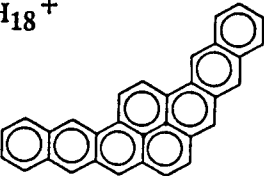
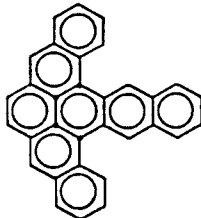
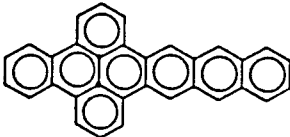
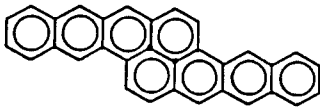
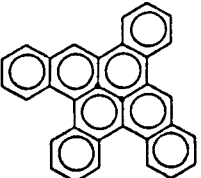
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------------|---|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{32}\text{H}_{16}^+$ |  | (7.04) | (263) | (1101) | 101 | 422 | *EST | 190-66-9 |
| |  | (6.92) | (260) | (1089) | 101 | 422 | *EST | 190-72-7 |
| |  | (6.88) | (264) | (1104) | 105 | 440 | *EST | 190-74-9 |
| $\text{C}_{32}\text{H}_{18}^+$ |  | (6.65) IP from 79CLA/SCH. | (269) | (1125) | 115 | 483 | *EST | 189-43-5 |
| |  | (6.94) IP from 79CLA/SCH. | (274) | (1145) | 114 | 475 | *EST | |
| |  | (6.91) IP from 79CLA/SCH. | (270) | (1132) | 111 | 465 | *EST | 192-60-9 |
| |  | (6.42) IP from 79CLA/SCH. | (264) | (1103) | 115 | 483 | *EST | |
| |  | (7.02) IP from 79CLA/SCH. | (276) | (1153) | 114 | 476 | *EST | |

Table 1. Positive Ion Table - Continued

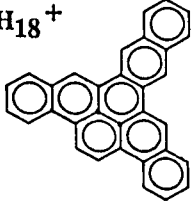
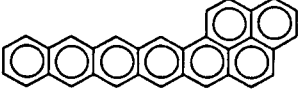
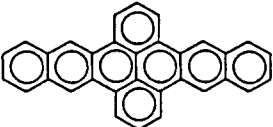
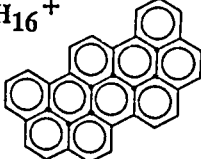
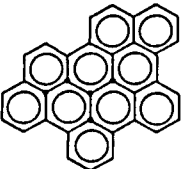
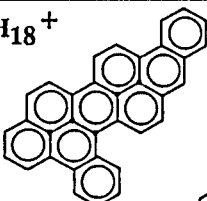
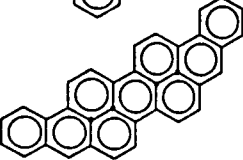
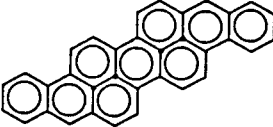
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{32}\text{H}_{18}^+$  | (6.99) | (277) | (1153) | 116 | 485 | *EST | |
| | IP from 79CLA/SCH. | | | | | | |
| | | | | | | | |
|  | (6.36) | (264) | (1107) | 118 | 493 | *EST | 196-46-3 |
| | IP from 79CLA/SCH. | | | | | | |
|  | (7.30±0.04) | (279) | (1169) | 111 | 465 | *EST | 192-54-1 |
| | | | | | | | |
| $\text{C}_{34}\text{H}_{16}^+$  | (6.74±0.02) | (261) | (1093) | 106 | 443 | *EST | 188-11-4 |
| | | | | | | | |
|  | 6.82±0.02 | (261) | (1092) | 104 | 434 | *EST | 187-94-0 |
| | | | | | | | |
| $\text{C}_{34}\text{H}_{18}^+$  | (6.59±0.02) | (265) | (1109) | 113 | 473 | *EST | |
| | | | | | | | |
| | | | | | | | |
|  | (6.48±0.02) | (263) | (1102) | 114 | 477 | *EST | |
| | | | | | | | |
|  | (6.42±0.02) | (262) | (1097) | 114 | 477 | *EST | 190-93-2 |
| | | | | | | | |

Table 1. Positive Ion Table - Continued

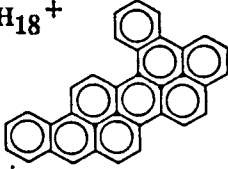
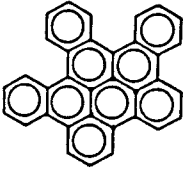
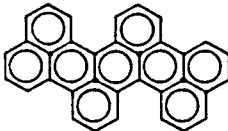
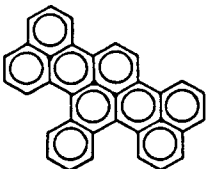
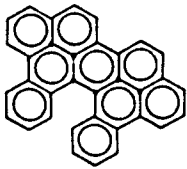
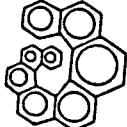
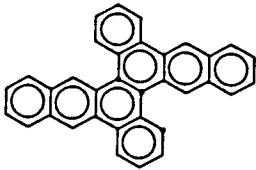
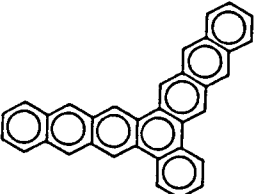
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{34}\text{H}_{18}^+$  | (6.59±0.02) | (265) | (1109) | 113 | 473 | *EST | 191-46-8 |
|  | (6.84) | (267) | (1119) | 110 | 459 | *EST | 313-63-3 |
|  | (6.27±0.02) | (256) | (1073) | 112 | 468 | *EST | 191-79-7 |
|  | (6.22±0.02) | (260) | (1088) | 117 | 488 | *EST | 188-13-6 |
|  | 6.58 | (276) | (1155) | 124 | 520 | *EST | 191-53-7 |
| $\text{C}_{34}\text{H}_{20}^+$  | (7.15) | (327) | (1370) | 162 | 680 | *EST | 20495-12-9 |
|  | (6.83±0.02) | (280) | (1172) | 123 | 513 | *EST | 385-15-9 |
|  | (6.90±0.02) | (286) | (1196) | 127 | 530 | *EST | 214-87-9 |

Table 1. Positive Ion Table - Continued

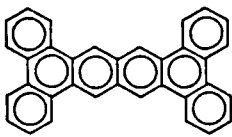
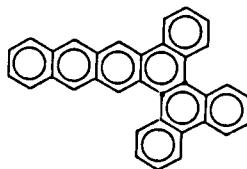
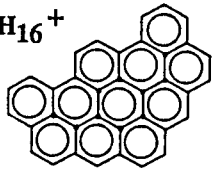
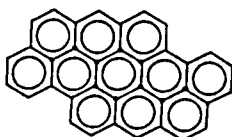
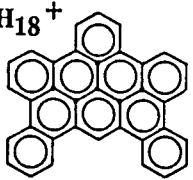
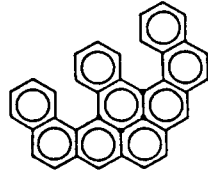
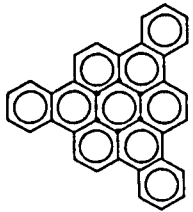
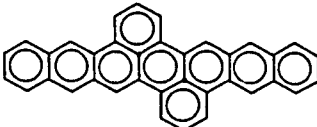
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------|--------------------------|---------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₃₄H₂₀⁺ | | | | | | | |
|  | (7.00±0.02) | (281) | (1177) | 120 | 502 | *EST | 215-95-2 |
|  | (6.73±0.02) | (278) | (1162) | 122 | 513 | *EST | 385-16-0 |
| C₃₆H₁₆⁺ | | | | | | | |
|  | (6.76±0.02) | (267) | (1117) | 111 | 464 | *EST | 53086-28-5 |
|  | (6.70±0.04) | (265) | (1111) | 111 | 464 | *EST | 190-47-6 |
| C₃₆H₁₈⁺ | | | | | | | |
|  | (≤7.10) | (≤284) | (≤1187) | 120 | 502 | *EST | 188-00-1 |
|  | (6.88) IP from 79CLA/SCH. | (306) | (1282) | 148 | 618 | *EST | |
|  | (6.88) | (271) | (1135) | 113 | 471 | *EST | 313-62-2 |
| C₃₆H₂₀⁺ | | | | | | | |
|  | (6.68) | (284) | (1190) | 130 | 545 | *EST | 197-73-9 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|------------------------------------|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{36}\text{H}_{20}^+$ | (6.82) | (287) | (1202) | 130 | 544 | *EST | 36474-85-8 |
| | (6.95) IP from 79CLA/SCH. | (286) | (1195) | 125 | 525 | *EST | |
| | (6.74) IP from 79CLA/SCH. | (283) | (1183) | 127 | 533 | *EST | |
| $\text{C}_{38}\text{H}_{16}^+$ | (6.81±0.02) | (271) | (1134) | 114 | 477 | *EST | 41163-25-1 |
| $\text{C}_{38}\text{H}_{18}^+$ | (6.38±0.02) | (270) | (1132) | 123 | 516 | *EST | 190-90-9 |
| | (6.50±0.02) | (277) | (1158) | 127 | 531 | *EST | 190-89-6 |
| $\text{C}_{38}\text{H}_{20}^+$ | (6.58) IP from 79CLA/SCH. | (282) | (1181) | 130 | 546 | *EST | |
| | (6.06±0.02) | (266) | (1112) | 126 | 528 | *EST | 187-96-2 |

Table 1. Positive Ion Table - Continued

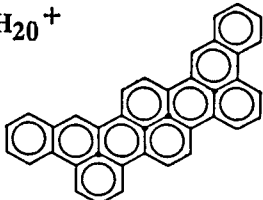
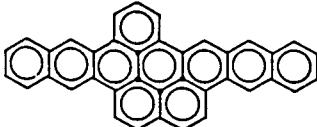
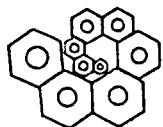
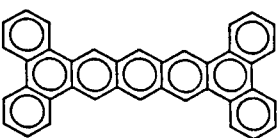
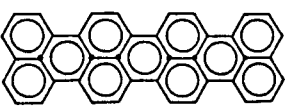
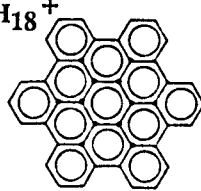
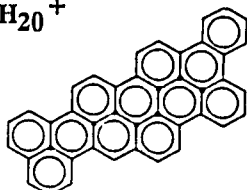
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-----------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{38}\text{H}_{20}^+$ | | | | | | | |
|  | (6.40±0.02) | (273) | (1144) | 126 | 527 | *EST | 34814-77-2 |
|  | (6.72) | (285) | (1193) | 130 | 545 | *EST | 14529-73-8 |
| $\text{C}_{38}\text{H}_{22}^+$ | | | | | | | |
|  | (7.07) | (365) | (1527) | 202 | 844 | *EST | 20495-14-1 |
|  | 6.65±0.02 See also: 75CLA/SCH. | (290) | (1212) | 136 | 570 | *EST | 216-07-9 |
| $\text{C}_{40}\text{H}_{20}^+$ | | | | | | | |
|  | (6.11±0.02) | (270) | (1128) | 129 | 539 | *EST | 188-73-8 |
| $\text{C}_{42}\text{H}_{18}^+$ | | | | | | | |
|  | 6.87±0.02 | (280) | (1170) | 121 | 508 | *EST | 190-24-9 |
| $\text{C}_{42}\text{H}_{20}^+$ | | | | | | | |
|  | (6.72±0.02) | (287) | (1199) | 132 | 551 | *EST | 34814-80-7 |

Table 1. Positive Ion Table - Continued

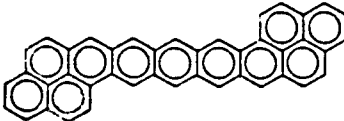
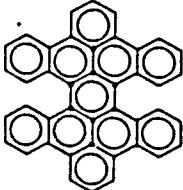
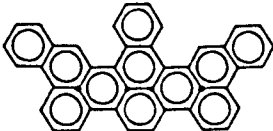
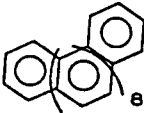
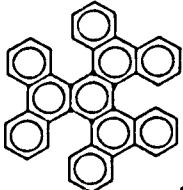
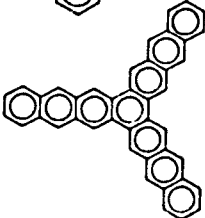
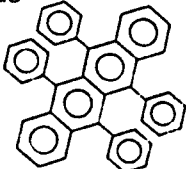
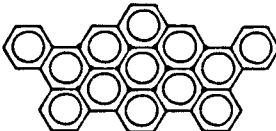
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| C₄₂H₂₂⁺ | | | | | | | |
|  | (6.22) IP from 79CLA/SCH. | (294) | (1231) | 151 | 631 | *EST | 190-09-0 |
|  | (6.71±0.02) | (283) | (1186) | 129 | 538 | *EST | 190-22-7 |
|  | (6.18±0.02) | (282) | (1181) | 140 | 585 | *EST | 34814-82-9 |
| C₄₂H₂₄⁺ | | | | | | | |
|  | (6.99) | (390) | (1633) | 229 | 959 | *EST | 57520-29-3 |
|  | (7.52±0.02) | (356) | (1491) | 183 | 765 | *EST | 190-23-8 |
|  | (6.85±0.02) | (317) | (1326) | 159 | 665 | *EST | 214-77-7 |
| C₄₂H₂₈⁺ | | | | | | | |
|  | 6.41 IP from 81SAT/SEK. | 334 | 1399 | 187±5 | 781±22 | 77PED/RYL | 517-51-1 |
| C₄₄H₂₀⁺ | | | | | | | |
|  | (6.79±0.02) | (287) | (1199) | 130 | 544 | *EST | 70346-75-7 |

Table 1. Positive Ion Table - Continued

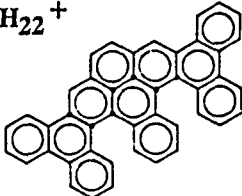
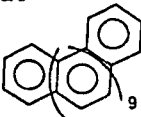
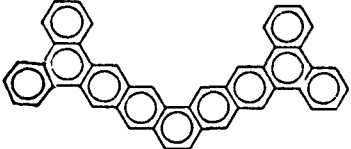
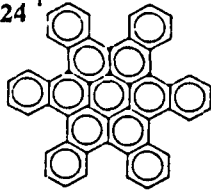
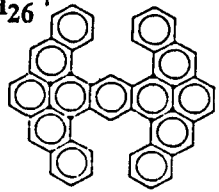
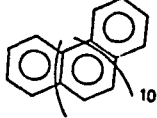
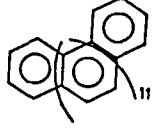
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------------|---|------------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{44}\text{H}_{22}^+$ |  | (6.80) IP from 79CLA/SCH. | (333) | (1394) | 176 | 738 | *EST | |
| $\text{C}_{46}\text{H}_{26}^+$ |  | (6.95) | (417) | (1744) | 256 | 1073 | *EST | 57468-45-8 |
| |  | (6.88±0.02) | (325) | (1360) | 166 | 696 | *EST | 62662-49-1 |
| $\text{C}_{48}\text{H}_{24}^+$ |  | (6.75) | (297) | (1242) | 141 | 590 | *EST | 1065-80-1 |
| $\text{C}_{50}\text{H}_{26}^+$ |  | (6.70) IP from 79CLA/SCH. | (329) | (1379) | 175 | 732 | *EST | 72382-92-4 |
| $\text{C}_{50}\text{H}_{28}^+$ |  | (6.93) | (444) | (1856) | 284 | 1187 | *EST | 57468-46-9 |
| $\text{C}_{54}\text{H}_{30}^+$ |  | (6.91) | (470) | (1968) | 311 | 1302 | *EST | 24386-06-9 |

Table 1. Positive Ion Table - Continued

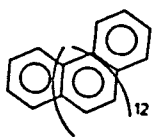
| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|------------------------------|------------------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{C}_{58}\text{H}_{32}^+$  | (6.88) | (497) | (2080) | 338 | 1416 | *EST | 57483-71-3 |
| Ca^+ Ca | 6.11321±0.00002 | <u>183.6</u> <u>183.5</u> | <u>768.0</u> <u>767.5</u> | 42.6 42.5 | 178.2 177.7 | 82TN270 | 7440-70-2 |
| CaCl^+ CaCl | 5.61±0.13 | 106 106 | 443 444 | -24.7±1.2 -24.5±1.2 | -103.4±5.0 -102.7±5.0 | 87GAR/PAR | 15606-71-0 |
| IP and $\Delta_f H(\text{Ion})$ derived from onset of endothermic reaction (84MEY/SCH). | | | | | | | |
| CaCl_2^+ CaCl ₂ | (≤10.0) | (≤118) (≤118) | (≤494) (≤493) | -113 -113 | -471 -472 | 82TN270 | 10043-52-4 |
| See also: 82EMO/KIE, 79LEE/POT2. | | | | | | | |
| CaH^+ CaH | (5.86±0.09) | (190) (190) | (794) (795) | 55 55 | 229 230 | 82TN270 | 14452-75-6 |
| Value for $\Delta_f H(\text{Ion})$ derived from onset energy of endothermic reaction is in good agreement (86ELK/ARI). | | | | | | | |
| CaHO^+ CaOH | 5.7 | (89) | (371) | -42.0 | -175.7 | 87GAR/PAR | 12177-67-2 |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (83MUR). See also: 81MUR. 0 K values. | | | | | | | |
| CaI^+ CaI | (6.1±0.3) | (139) (137) | (584) (572) | -1±21 -4 | -5±84 -17 | 79HUB/HER | 15923-87-2 |
| CaI_2^+ CaI ₂ | (8.7) | (139) (140) | (581) (584) | -62±4 -61±4 | -258±17 -255±17 | 85JANAF | 10102-68-8 |
| IP is onset of photoelectron band (79LEE/POT2). See also: 82EMO/KIE. | | | | | | | |
| CaO^+ CaO | (6.9) | (166) | (693) | (6±4) | (27±17) | 83PED/MAR | 1305-78-8 |
| IP from 83MUR. | | | | | | | |
| CaO_4W^+ CaWO ₄ | (9.8) | (0) | (0) | -226 | -946 | 76DEL/HAL | |
| Cd^+ Cd | 8.993 | <u>234.2</u> <u>234.2</u> | <u>979.7</u> <u>979.8</u> | 26.8 26.8 | 112.0 112.1 | 82TN270 | 7440-43-9 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-------------------|------------------------|---|--------------------------|------------------|------------------------------|------------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Ce^+ | Ce | 5.5387±0.0004 | 229 229 | 957 957 | 101 101.1 | 423 423.4 | 82TN270 | 7440-45-1 |
| CeI_3^+ | CeI_3 | 8.7 IP is onset of photoelectron band (83RUS/GOO). | 101 | 422 | -100 | -417 | 82TN270 | |
| CeO^+ | CeO | (4.90±0.1) | (81) (81) | (339) (341) | -32±3 -32 | -134±12 -132 | 83PED/MAR | 12014-74-3 |
| CeS^+ | CeS | (6.0±0.6) | (170) (170) | (710) (713) | 31 32 | 131 134 | 82TN270 | 12014-82-3 |
| Ce_2^+ | Ce_2 | (5.6±0.4) | (273) (274) | (1142) (1147) | 144 145 | 602 607 | 82TN270 | 12595-88-9 |
| Cf^+ | Cf | 6.3 See: 81CHE/GAB. | 192 | 804 | 47 | 196 | 85KLE/WAR | 7440-71-3 |
| Cl^+ | Cl | 12.967 | 328 328 | 1372 1371 | 29.0 28.6 | 121.3 119.6 | 85JANAF | 22537-15-1 |
| | | See also: 81KIM/KAT. | | | | | | |
| ClCs^+ | CsCl | (7.84±0.05) | (122) (122) | (510) (512) | -59 -58.4±1.8 | -247 -244.4±7.5 | 84PAR/WEX | 7647-17-8 |
| | | A value of 8.32±0.1 eV has also been reported for the ionization potential. | | | | | | |
| ClCsNa^+ | NaCsCl | 3.9±0.1 | (21) | (88) | -69 | -288 | *EST | 95860-64-3 |
| | | IP from 85KAP/RAD. 0 K values. | | | | | | |
| ClCs_2^+ | Cs_2Cl | 3.4±0.2 | -1 | -4 | -79±6 | -332±25 | 85KAP/RAD | 87331-16-6 |
| | | IP from 85KAP/RAD. 0 K values. | | | | | | |
| ClCu^+ | CuCl | (10.7±0.3) 0 K values. | (265) | (1110) | 19 | 78 | 79HUB/HER | 7758-89-6 |
| ClF^+ | ClF | 12.65±0.01 | 280 280 | 1170 1170 | -12.0±0.1 -12.0±0.1 | -50.3±0.4 -50.2±0.4 | 85JANAF | 7790-89-8 |
| | | See also: 84DYK/JOS. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ kcal/mol kJ/mol | $\Delta_f H(\text{Neutral})$ kcal/mol kJ/mol | Neutral reference | CAS registry number |
|--|---|---|---|----------------------|------------------------|
| ClFO_2^+ ClO_2F | (12.41±0.10) IP from 80BAL/NIK. | (278) (1164) | -8 -33 | 73BAR | 13637-83-7 |
| ClFO_2S^+ SO_2FCl | (12.4) IP is onset of photoelectron band. | (151) (632) | -135 -564 | 81WOO | 13637-84-8 |
| ClFO_3^+ ClO_3F | (12.945±0.005) | (293) (1225) (295) (1234) | -6 -24 -4 -15 | 82TN270 | 7616-94-6 |
| ClF_2^+ ClF_2 | (12.77±0.05) $\Delta_f H(\text{Ion})$ derived from appearance potential (13.78±0.07 eV) in ClF_3 is 261 kcal/mol, 1091 kJ/mol. | (269) (1127) | -25 -105 | 62ARM/KRI | 24801-48-7 |
| ClF_3^+ ClF_3 | (12.65±0.05) | (253) (1057) (254) (1061) | -39 -163 -38 -159 | 82BAU/COX | 7790-91-2 |
| ClF_5S^+ SF_5Cl | (12.335±0.005) | (34) (142) (37) (155) | -250 -1048 -247 -1035 | 82TN270 | 13780-57-9 |
| ClH^+ HCl | 12.747 IP for formation of $\text{HCl}^+(^2\Pi_{3/2})$ from 79HUB/HER, 82NAT/PEN, 77ROS/DRA, 82LEV/LIA. IP for formation of $\text{HCl}^+(^2\Pi_{1/2}) = 12.828$ eV. See also: 82VON/ASB, 84WAN/DIL, 81KIM/KAT. | 271.9 1137.6 271.4 1137.7 | -22.1±0.04 -92.3±0.2 -22.0±0.04 -92.1±0.2 | 85JANAF | 7647-01-0 |
| ClD^+ DCl | 12.754 IP for formation of $\text{DCl}(^2\Pi_{3/2})$ from 79HUB/HER, 83PEN/NAT. | 271.8 1137.3 271.9 1137.6 | -22.3±0.05 -93.3±0.2 -22.3±0.05 -93.1±0.2 | 85JANAF | 7698-05-7 |
| ClHO^+ HOCl | (11.12±0.01) | (238) (995) (239) (998) | -19 -78 -18 -75 | 82BAU/COX | 7790-92-3 |
| ClH_2^+ H_2Cl | 207 867 From proton affinity of HCl (RN 7647-01-0) (85MCM/KEB) re-evaluated relative to CO standard (84LIA/LIE). PA = 128.6 kcal/mol, 538 kJ/mol. | | | | |
| ClH_2N^+ NH_2Cl | (9.85±0.02) | (240) (1003) | 13 53 | *EST | 10599-90-3 |
| ClH_3Si^+ SiH_3Cl | 11.4 IP is onset of photoelectron band. | (215) (899) | -48 -201 | 81BEL/PER | 13465-78-6 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------------|----------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CII ⁺ ICI | 10.088±0.01 | 236.8 | 990.8 | 4.2±0.02 | 17.5±0.1 | 85JANAF | 7790-99-0 |
| | | 237.2 | 992.4 | 4.6±0.2 | 19.1±0.1 | | |
| See also: 84DYK/JOS, 71POT/PRI. | | | | | | | |
| CIIIn ⁺ CIIn | (9.51) | (201) | (843) | -18 | -75 | 82TN270 | 13465-10-6 |
| | | (204) | (852) | -15 | -65 | | |
| CIK ⁺ KCI | (8.0±0.4) | (133) | (557) | -51.3±0.1 | -214.7±0.4 | 85JANAF | 7447-40-7 |
| | | (134) | (559) | -50.9±0.1 | -212.9±0.4 | | |
| CIKNa ⁺ NaKCI | 4.0±0.1 | (26) | (107) | -67 | -279 | *EST | 95860-66-5 |
| | | IP from 85KAP/RAD. 0 K values. | | | | | |
| CIK ₂ ⁺ K ₂ CI | 3.5±0.2 | 10 | 44 | -70±4 | -294±17 | 85KAP/RAD | 95386-61-1 |
| | | IP from 85KAP/RAD. 0 K values. | | | | | |
| CILi ⁺ LiCI | 9.57 | 174 | 728 | -47±3 | -196±13 | 85JANAF | 7447-41-8 |
| | | 174 | 728 | -47±3 | -196±13 | | |
| CINO ⁺ NOCI | 10.87±0.01 | 263 | 1101 | 12 | 52 | 82BAU/COX | 2696-92-6 |
| | | 264 | 1103 | 13 | 54 | | |
| See also: 83BIN. | | | | | | | |
| CINO ₂ ⁺ CINO ₂ | (11.84) | (276) | (1155) | 3 | 13 | 82BAU/COX | 13444-90-1 |
| | | (277) | (1160) | 4 | 18 | | |
| CIN ₃ ⁺ CIN ₃ | (10.20±0.01) | (313.9) | (1313.4) | 78.7 | 329.3 | 83DEW/RZE | 13973-88-1 |
| | | | | | | | |
| CINa ⁺ NaCI | 8.92±0.06 | 162 | 679 | -43±0.5 | -181±2 | 85JANAF | 7647-14-5 |
| | | 163 | 681 | -43±0.5 | -180±2 | | |
| CINa ₂ ⁺ Na ₂ CI | 4.1±0.1 | 36 | 152 | -58±4 | -244±17 | 85KAP/RAD | 84008-89-9 |
| | | IP from 85KAP/RAD. 0 K values. | | | | | |
| CINi ⁺ NiCI | (11.4±0.5) | (306) | (1282) | 43±1 | 182±4 | 85JANAF | 13931-83-4 |
| | | (306) | (1282) | 43±1 | 182±4 | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| ClO ⁺ ClO | 10.95 | 277 | 1158 | 24.4 | 101.9 | 82BAU/COX | 14989-30-1 |
| | | 277 | 1158 | 24.3 | 101.8 | | |
| IP is onset of photoelectron band. | | | | | | | |
| ClOP ⁺ POCl | (11.5) IP from 83BIN. | (205) | (859) | -60 | -251 | 83BIN/LAK | 21295-50-1 |
| | | | | | | | |
| ClOSb ⁺ SbOCl | (10.7) IP from 83BIN. | (247) | (1032) | 0 | 0 | 83BIN | 7791-08-4 |
| | | | | | | | |
| ClO ₂ ⁺ OClO | 10.36±0.02 | 262 | 1097 | 23±2 | 97±8 | 82BAU/COX | 10049-04-4 |
| | | 263 | 1099 | 24 | 99 | | |
| See also: 80BAL/NIK. | | | | | | | |
| ClRb ⁺ RbCl | (8.50±0.03) | (141) | (591) | -55 | -229 | 82TN270 | 7791-11-9 |
| | | (142) | (593) | -54 | -227 | | |
| ClSr ⁺ SrCl | 5.10±0.06 | (88) | (368) | -30±2 | -124±8 | 85JANAF | 14989-33-4 |
| | | (88) | (370) | -29±2 | -122±8 | | |
| See also: 84MEY/SCH. | | | | | | | |
| ClTi ⁺ TiCl | 9.70±0.03 | 207 | 868 | -16 | -68 | 82TN270 | 7791-12-0 |
| | | 207 | 868 | -16 | -68 | | |
| See also: 83BAN/BRI. | | | | | | | |
| Cl ₂ ⁺ Cl ₂ | 11.480±0.005 | 265 | 1108 | 0 | 0 | *DEF | 7782-50-5 |
| | | 265 | 1108 | 0 | 0 | | |
| Cited ionization potential is for formation of Cl ₂ ⁺ (² Π _{3/2}) (77ROS/DRA, 82LEV/LIA, 84VAN/DEL2, 84DYK/JOS)). Formation of Cl ₂ ⁺ (² Π _{1/2}) requires 11.56 eV. See also: 81KIM/KAT. | | | | | | | |
| Cl ₂ Co ⁺ CoCl ₂ | (10.4) | (217) | (909) | -22±2 | -94±8 | 85JANAF | 7646-79-9 |
| | | (217) | (908) | -23±2 | -95±8 | | |
| IP is onset of photoelectron band. | | | | | | | |
| Cl ₂ Cr ⁺ CrCl ₂ | (9.4) | (186) | (779) | -31 | -128 | 82TN270 | 10049-05-5 |
| | | | | | | | |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

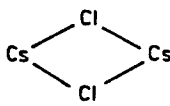
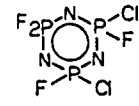
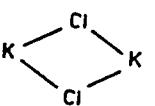
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------------|----------------------------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{Cl}_2\text{CrO}_2^+$ Cl_2CrO_2 | 11.6 | (139) (140) | (581) (585) | -129 -128 | -538 -534 | 82TN270 | 14977-61-8 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_2Cs_2^+  | (≤ 9.15) | (≤ 53) (≤ 54) | (≤ 223) (≤ 227) | -158 -157 | -660 -656 | 81LIN/BES | 12258-95-6 |
| $\text{Cl}_2\text{F}_4\text{N}_3\text{P}_3^+$  | (10.97 \pm 0.3) | (-150) | (-629) | -403 | -1687 | *EST | 29871-62-3 |
| IP from 81CLA/SOW. | | | | | | | |
| Cl_2Fe^+ FeCl_2 | (10.0) | (197) (197) | (824) (823) | -34 -34 | -141 -142 | 85JANAF | 7758-94-3 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_2Ge^+ GeCl_2 | (10.20 \pm 0.05) | (194) (195) | (813) (814) | -41 \pm 1 -41 | -171 \pm 5 -170 | 79TPIS | 10060-11-4 |
| IP from 82JON/VAN. | | | | | | | |
| Cl_2HN^+ NHCl_2 | (9.98 \pm 0.05) | (269) | (1124) | 38 | 161 | *EST | 3400-09-7 |
| $\text{Cl}_2\text{H}_2\text{Si}^+$ SiH_2Cl_2 | 11.4 | (183) | (765) | -80 | -335 | 81BEL/PER | 4109-96-0 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_2Hg^+ HgCl_2 | 11.380 \pm 0.003 | 227 | 952 | -35 \pm 1 | -146 \pm 6 | 71JANAF | 7487-94-7 |
| Cited ionization potential (from 83LIN/BRO) refers to formation of $\text{HgCl}_2^+ (^2\Pi_{3/2g})$. Ionization potential for formation of $\text{HgCl}_2^+ (^2\Pi_{1/2g})$ is 11.505 \pm 0.003 eV. See also: 81LEE/POT. | | | | | | | |
| Cl_2K_2^+  | (≤ 9.60) | (≤ 72) | (≤ 303) | -149 | -623 | 82TN270 | 12258-97-8 |

Table 1. Positive Ion Table - Continued

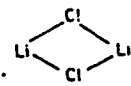
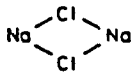
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|------------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Cl_2Li_2^+ | | | | | | | |
|  | 10.20 | 93 94 | 390 393 | -142 -141 | -594 -591 | 82TN270 | 12345-57-2 |
| Cl_2Mg^+ MgCl_2 | (8.5) | (102) (102) | (428) (427) | -94 -94 | -392 -393 | 85JANAF | 7786-30-3 |
| | | IP is onset of photoelectron band. See also: 77LEE/POT2. | | | | | |
| Cl_2Mn^+ MnCl_2 | (10.8) | (186) | (778) | -63 | -264 | 82TN270 | 7773-01-5 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{Cl}_2\text{MoO}_2^+$ MoO_2Cl_2 | (11.93±0.02) | (124) | (517) | -152 | -634 | 82TN270 | 13637-68-8 |
| Cl_2Na_2^+ | | | | | | | |
|  | (≤10.30) | (≤102) (≤103) | (≤428) (≤432) | -135±2 -134±2 | -566±8 -562±8 | 85JANAF | 12258-98-9 |
| Cl_2Ni^+ NiCl_2 | (10.8) | (231) (231) | (968) (968) | -18±0.1 -18±0.1 | -74±0.3 -74±0.3 | 82JANAF | 7718-54-9 |
| | | IP is onset of photoelectron band. | | | | | |
| Cl_2O^+ Cl_2O | 10.94 | 271 272 | 1136 1138 | 19 20 | 80 82 | 82BAU/COX | 7791-21-1 |
| Cl_2OS^+ SOCl_2 | 10.96 | (202) (203) | (844) (847) | -51 -50 | -213 -210 | 82TN270 | 7719-09-7 |
| $\text{Cl}_2\text{O}_2\text{S}^+$ SO_2Cl_2 | 12.05 | 193 195 | 808 814 | -85 -83 | -355 -349 | 85JANAF | 7791-25-5 |
| Cl_2Pb^+ PbCl_2 | (10.0) | (189) (189) | (791) (793) | -42±0.3 -41±0.3 | -174±1 -172±1 | 85JANAF | 7758-95-4 |
| | | IP is onset of photoelectron band. (See: 84NOV/POT2). | | | | | |

Table 1. Positive Ion Table - Continued

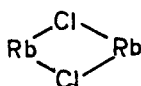
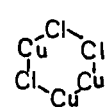
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|----------------------------------|----------------------------------|--------------------------------|------------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Cl_2Rb_2^+  | (≤ 9.30) | (≤ 64) (≤ 65) | (≤ 269) (≤ 271) | -150 -150 | -628 -626 | 82TN270 | 12265-61-1 |
| Cl_2S^+ SCl_2 | 9.45 \pm 0.03 IP from 81KAU/VAH. | 214 214 | 894 895 | -4 -4 | -18 -16 | 85JANAF | 10545-99-0 |
| Cl_2S_2^+ S_2Cl_2 | (9.66 \pm 0.03) IP from 81KAU/VAH. | (218) (219) | (914) (915) | -4 -4 | -18 -17 | 82TN270 | 10025-67-9 |
| Cl_2Se^+ SeCl_2 | 9.25 IP is onset of photoelectron band. | (206) | (860) | -8 | -32 | 82TN270 | 14457-70-6 |
| Cl_2Se_2^+ Se_2Cl_2 | (9.4) IP is onset of photoelectron band. | (221) | (924) | 4 | 17 | 82TN270 | 10025-68-0 |
| Cl_2Si^+ SiCl_2 | (10.93 \pm 0.10) | (212) (212) | (889) (889) | -40 -40 | -166 -166 | 82TN270 | 13569-32-9 |
| Cl_2Sn^+ SnCl_2 | (10.0) IP is onset of photoelectron band (84NOV/POT2, 82LEV/LIA). | (182) (183) | (762) (764) | -49 -49 \pm 2 | -203 -201 \pm 10 | 82TPIS | 7772-99-8 |
| Cl_2Sr^+ SrCl_2 | 9.70 \pm 0.1 See also: 82EMO/KIE, 79LEE/POT2. 0 K values. | (115) | (481) | -109 | -455 | 82EMO/KIE | 10476-85-4 |
| Cl_2Zn^+ ZnCl_2 | 11.85 IP is onset of photoelectron band. | 210 | 877 | -64 | -266 | 82TN270 | 7646-85-7 |
| Cl_3Cu_3^+  | (≤ 9.52) | (≤ 158) (≤ 158) | (≤ 660) (≤ 660) | -62 \pm 0.5 -62 \pm 0.5 | -259 \pm 2 -259 \pm 2 | 85JANAF | 11093-65-5 |

Table 1. Positive Ion Table - Continued

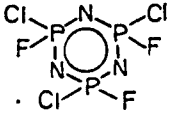

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|------------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{Cl}_3\text{F}_3\text{N}_3\text{P}_3^+$  | (10.76±0.03) IP from 81CLA/SOW. | (-5) | (-22) | -253 | -1060 | *EST | 25251-05-2 |
| Cl_3Ga^+ GaCl ₃ | 11.52 | 159 | 664 | -107 | -448 | 82TN270 | 13450-90-3 |
| Cl_3HSi^+ SiHCl ₃ | (11.7) IP is onset of photoelectron band. | (155) (156) | (647) (652) | -115 -114 | -482 -477 | 81BEL/PER | 10025-78-2 |
| Cl_3In^+ InCl ₃ | (11.4) | (173) (173) | (722) (724) | -90 -90±2 | -378 -376±7 | 82TPIS | 10025-82-8 |
| Cl_3La^+ LaCl ₃ | (10.6) IP is onset of photoelectron band (83RUS/GOO). | (67) | (282) | -177 | -741 | 82TN270 | 10099-58-8 |
| Cl_3Li_3^+  | (10.17) | (5) (6) | (19) (26) | -230 -228 | -962 -955 | 82TN270 | 59217-69-5 |
| Cl_3Lu^+ LuCl ₃ | (11.5±0.5) | (110) | (461) | -155 | -649 | 82TN270 | 10099-66-8 |
| Cl_3N^+ NCl ₃ | (10.12±0.1) | (297) | (1244) | 64 | 268 | *EST | 10025-85-1 |
| Cl_3NbO^+ NbOCl ₃ | ≤12.14 | ≤100.1 ≤100.8 | ≤419.0 ≤421.7 | -179.8 -179.1 | -752.3 -749.6 | 82TN270 | 13597-20-1 |
| Cl_3OP^+ POCl ₃ | 11.36±0.02 | 129 130 | 538 543 | -133 -132 | -558 -553 | 82TN270 | 10025-87-3 |
| Cl_3OV^+ VOCl ₃ | (11.6) IP is onset of photoelectron band. | (101) (102) | (423) (426) | -166 -166 | -696 -693 | 82TN270 | 7727-18-6 |

Table 1. Positive Ion Table - Continued

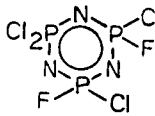
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Cl_3P^+ PCl_3 | 9.91 | 160 | 667 | -69 | -289 | 85JANAF | 7719-12-2 |
| | | 160 | 671 | -68 | -286 | | |
| See also: 83OZG, 81CHA/FIN. | | | | | | | |
| Cl_3PS^+ PSCl_3 | 9.71±0.03 | 137 | 574 | -87 | -363 | 71JANAF | 3982-91-0 |
| Cl_3Sb^+ SbCl_3 | (10.1±0.1) | (158) | (660) | -75 | -314 | 82TN270 | 10025-91-9 |
| IP is onset of photoelectron band. See also: 83OZG. | | | | | | | |
| Cl_3Si^+ SiCl_3 | | (108) | (454) | | | | 19165-34-5 |
| From appearance potentials, 11.91 eV in SiHCl_3 and 11.90 eV in CH_3SiCl_3 . | | | | | | | |
| $\text{Cl}_4\text{F}_2\text{N}_3\text{P}_3^+$  | (10.48±0.03) | (-53) | (-222) | -295 | -1233 | *EST | 25251-04-1 |
| IP from 81CLA/SOW. | | | | | | | |
| Cl_4Ge^+ GeCl_4 | 11.68±0.05 | (151) | (631) | -119 | -496 | 82TN270 | 10038-98-9 |
| | | (151) | (633) | -118 | -494 | | |
| Cl_4Hf^+ HfCl_4 | (11.7) | (59) | (246) | -211 | -883 | 81SPE | 13499-05-3 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_4Mo^+ MoCl_4 | (10.5±0.1) | (152) | (636) | -90 | -377 | 82TN270 | 13320-71-3 |
| IP from 83MAK/VER. | | | | | | | |
| Cl_4Si^+ SiCl_4 | 11.79±0.01 | 126 | 528 | -146 | -610 | 81BEL/PER | 10026-04-7 |
| | | 124 | 520 | -148 | -618 | | |
| Cl_4Sn^+ SnCl_4 | (11.88±0.05) | (161) | (674) | -113 | -472 | 82TN270 | 7646-78-8 |
| | | (162) | (677) | -112 | -469 | | |
| Cl_4Th^+ ThCl_4 | (12.7±0.3) | (62) | (259) | -231 | -966 | 82TN270 | 10026-08-1 |
| | | (62) | (260) | -230.8 | -965.6 | | |
| Cl_4Ti^+ TiCl_4 | 11.65±0.15 | (86) | (361) | -182 | -763 | 85JANAF | 7550-45-0 |
| | | (86) | (362) | -182 | -762 | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|----------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Cl_4U^+ UCl_4 | 9.18 | 18 | 76 | -193 | -810 | 82TN270 | 10026-10-5 |
| Cl_4V^+ VCl_4 | (9.2) | (86) (87) | (362) (364) | -126 -125 | -526 -524 | 82TN270 | 7632-51-1 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_4W^+ WCl_4 | (8.0) | (104) (104) | (436) (436) | -80±8 -80±8 | -336±33 -336±33 | 85JANAF | 13470-13-8 |
| Cl_4Zr^+ ZrCl_4 | (11.2) | (50) (51) | (211) (212) | -208 -208 | -870 -869 | 82TN270 | 10026-11-6 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_5Mo^+ MoCl_5 | (8.7) | (94) (94) | (392) (393) | -107 -106 | -448 -446 | 85JANAF | 10241-05-1 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_5Nb^+ NbCl_5 | (10.97) | (85) (85) | (355) (357) | -168 -168 | -703 -701 | 85JANAF | 10026-12-7 |
| Cl_5P^+ PCl_5 | 10.7 | (157) (158) | (657) (662) | -90 -88 | -375 -370 | 82TN270 | 10026-13-8 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_5Re^+ ReCl_5 | (9.2) | (136) | (570) | -76 | -318 | 82TN270 | 13596-35-5 |
| IP is onset of photoelectron band. | | | | | | | |
| Cl_5Sb^+ SbCl_5 | (10.8) | (155) (155) | (648) (650) | -94 -94 | -394 -392 | 82TN270 | 7647-18-9 |
| IP is onset of photoelectron band (81ELB/DIE). | | | | | | | |
| Cl_5Ta^+ TaCl_5 | 11.08 | 73 73 | 304 306 | -183 -182 | -765 -763 | 85JANAF | 7721-01-9 |
| Cl_5W^+ WCl_5 | (8.5) | (97) (98) | (407) (409) | -99±8 -98±8 | -413±33 -411±33 | 85JANAF | 13470-14-9 |
| IP is onset of photoelectron band. | | | | | | | |

Table 1. Positive Ion Table - Continued

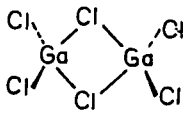
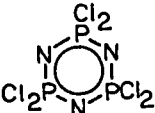
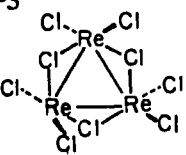
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-------------------------------------|---|----------------------------|--------------------------|--------|------------------------------|--------|----------------------|--|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Cl_6Ga_2^+ |  | (11.4) | (30) | (125) | -233 | -975 | 82TN270 | 15654-66-7 |
| | | | | | | | | IP is onset of photoelectron band. |
| $\text{Cl}_6\text{N}_3\text{P}_3^+$ |  | 9.8 | (51) | (213) | -175 | -732 | 69BEN/CRU | 940-71-6 |
| | | | | | | | | IP is onset of photoelectron band. See also: 81CLA/SOW. |
| Cl_6Si_2^+ | Si_2Cl_6 | (10.4) | (-4) | (-16) | -244 | -1019 | 81BEL/PER | 13465-77-5 |
| | | | | | | | | See also: 81KHV/ZYK. |
| Cl_6W^+ | WCl_6 | (9.5) | (83) | (347) | -136 | -570 | 81WOO | 13283-01-7 |
| | | | | | -136 | -570 | | |
| Cl_9Re_3^+ |  | (8.7) | (64) | (266) | -137 | -573 | 82TN270 | 14973-59-2 |
| | | | | | | | | IP is onset of photoelectron band. |
| Cm^+ | Cm | 6.09±0.02 | 233 | 974 | 92 | 386 | 85KLE/WAR | 7440-51-9 |
| | | | | | | | | See also: 81CHE/GAB. |
| Co^+ | Co | 7.864±0.001 | 283 | 1184 | 102 | 425 | 82TN270 | 7440-48-4 |
| | | | 282 | 1182 | 101 | 423 | | |
| | | | | | | | | See also: 82DYK/GRA. |
| CoH^+ | CoH | (7.3±0.1) | (287) | (1203) | (119) | (496) | 81ARM/BEA | 14994-20-8 |
| | | | | | | | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM4). See also: 81ARM/HAL. IP from 81ARM/BEA. 0 K values. |
| CoHO^+ | CoOH | | 220 | 920 | | | | 12314-24-8 |
| | | | | | | | | $\Delta_f H(\text{Ion})$ from photodissociation onset, proton affinity of CoO(84CAS/FRE). 0 K values. |
| CoO^+ | CoO | 8.9±0.2 | (277) | (1159) | 72±3 | 301±13 | 79HUB/HER | 1307-96-6 |
| | | | | | | | | $\Delta_f H(\text{Ion})$ from 81ARM/HAL, 82ARM/HAL. See also: 81KAP/STA. 0 K values. |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number | |
|---|----------------------------|--|------------------|------------------------------|--------------------|----------------------|------------------------|--|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | | |
| Cr^+ Cr | 6.766 | 250.8 250.3 | 1049.4 1047.3 | 94.8 94.3 | 396.6 394.5 | 82TN270 | 7440-47-3 | |
| CrF^+ CrF | (8.4±0.3) | (199) | (831) | 5 | 21 | 81WOO | 13943-42-5 | |
| CrF_2^+ CrF ₂ | (10.1±0.3) | (181) | (758) | -52 | -216 | 81WOO | 10049-10-2 | |
| CrF_3^+ CrF ₃ | (12.2±0.3) | (124) | (517) | -158 | -660 | 81WOO | 7788-97-8 | |
| CrH^+ CrH | | (274) | (1145) | | | | 13966-79-5 | |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM). See also: 81ARM/HAL. 0 K values. | | | | | | |
| CrO^+ CrO | 7.85±0.02 | (233) (230) | (975) (961) | 52±7 49±3 | 218±29 203±12 | 83PED/MAR | 12018-00-7 | |
| | | IP from 83DYK/GRA. $\Delta_f H(\text{Ion})$ at 0 K from 81ARM/HAL, 82ARM/HAL. See also: 81BAL/GIG, 81KAP/STA. | | | | | | |
| CrO_2^+ CrO ₂ | (10.3±0.5) 0 K values. | (223) | (935) | -14 | -59 | 82TN270 | 12018-01-8 | |
| CrO_2P^+ CrPO ₂ | (8.0±0.5) | (236) | (989) | 52±3 | 218±13 | 81BAL/GIG | | |
| | | IP from 81BAL/GIG. 0 K values. | | | | | | |
| CrO_3^+ CrO ₃ | (11.6±0.5) | (175) | (733) | -92 | -386 | 82TN270 | 1333-82-0 | |
| Cs^+ Cs | 3.894 | 108.0 108.3 | 451.8 453.3 | 18.2 18.5 | 76.1 77.6 | 82TN270 | 7440-46-2 | |
| | | See also: 84ASA/YAS, 85SCH/WEL. | | | | | | |
| CsF^+ CsF | (8.80±0.10) | (117) (117) | (488) (490) | -86 -85.8±1.8 | -361 -359.0±7.5 | 84PAR/WEX | 13400-13-0 | |
| CsHO^+ CsOH | (7.3±0.15) | (106) | (445) | -62 | -259 | 81LIN/BES | 21351-79-1 | |
| CsH_2O^+ CsOH ₂ | | 36 | 149 | | | | | |
| | | $\Delta_f H(\text{Ion})$ from equilibrium constant determination (69SEA/DZI). | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|------------------------------|--------------------------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CsI⁺ CsI | 7.10±0.05 | 127 128 | 531 535 | -37 -35.9±3.4 | -154 -150±14 | 84PAR/WEX | 7789-17-5 |
| See also: 82EMO/HOR, 82LEL/BAL, 84VIS/HIL. | | | | | | | |
| CsK⁺ KCs | (3.9±0.1) | (119) | (498) | (29) | (122) | 79HUB/HER | |
| IP from 85KAP/SCH. 0 K values. | | | | | | | |
| CsLi⁺ LiCs | (4.1±0.1) | (134) | (562) | (40) | (166) | 79HUB/HER | 12018-59-6 |
| IP from 85KAP/SCH. 0 K values. | | | | | | | |
| CsNa⁺ NaCs | (4.05±0.04) | (128) | (535) | (35) | (144) | 79HUB/HER | 12018-60-9 |
| IP from 85KAP/SCH. 0 K values. | | | | | | | |
| CsO⁺ CsO | 6.22 | 153.6 154.1 | 642.8 644.9 | 10.2 10.7±5 | 42.7 44.8±21 | 82TPIS | 24774-39-8 |
| IP from 84BUT/KUD. | | | | | | | |
| CsRb⁺ RbCs | 3.7±0.1 | 96 | 401 | 11 | 44 | 86IGE/WED | 12331-83-8 |
| IP from 85KAP/SCH. | | | | | | | |
| Cs₂⁺ Cs ₂ | 3.7±0.1 | (111) (112) | (464) (469) | 26±0.1 27±0.1 | 107±0.3 112±0.3 | 85JANAF | 12184-83-7 |
| IP from 85KAP/RAD, 85KAP/SCH. See also: 83HEL/MOL. | | | | | | | |
| Cs₂MoO₄⁺ Cs ₂ MoO ₄ | (7.0) | (-114) | (-479) | -276 | -1154 | 81LIN/BES | |
| Cs₂O⁺ Cs ₂ O | 4.41±0.03 | (80) | (333) | -22 | -92 | 81LIN/BES | 20281-00-9 |
| IP from 77ROS/DRA, 84BUT/KUD. | | | | | | | |
| Cu⁺ Cu | 7.72634±0.00002 | <u>259.0</u> <u>258.8</u> | <u>1083.8</u> <u>1082.7</u> | 80.9 80.6 | 338.3 337.2 | 82TN270 | 7440-50-8 |
| CuF⁺ CuF | 10.15±0.02 | (235) (231) | (984) (967) | 1 -3 | 5 -12 | 81WOO | 13478-41-6 |
| IP from 80DYK/FAY. See also: 77EHL/WAN. | | | | | | | |
| CuF₂⁺ CuF ₂ | (12.7) | (229) (230) | (958) (960) | -64 -63 | -267 -265 | 81WOO | 7789-19-7 |
| IP is onset of photoelectron band (80DYK/FAY). See also: 77EHL/WAN. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|------------------|----------------------------|--------------------------|---------------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| CuH^+ | CuH | (9.5) | (289) | (1208) | 70 | 291 | 79HUB/HER | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM4). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | | | |
| CuO^+ | CuO | | (286±8) | (1197±33) | 73±10 | 306±41 | 85JANAF | 1317-38-0 |
| $\Delta_f H(\text{Ion})$ from (81KAP/STA). Ion/molecule bracketing results. | | | | | | | | |
| CuSn^+ | CuSn | (7.2±1.0) | (277) | (1161) | 111 | 466 | 79HUB/HER | 12054-11-4 |
| 0 K values. | | | | | | | | |
| Cu_2^+ | Cu ₂ | 7.894±0.015 | 298 | 1247 | 116 | 485 | 85JANAF | 12190-70-4 |
| IP from 83POW/HAN. | | | | | | | | |
| | | | 298 | 1247 | 116 | 485 | | |
| D^+ | D | 13.602 | <u>366.6</u> | <u>1534.0</u> | 52.98 | 221.67 | 82TN270 | 16873-17-9 |
| | | | | | | | | |
| | | | <u>366.2</u> | <u>1532.1</u> | 52.52 | 219.76 | | |
| DH^+ | HD | 15.44477±0.00007 | <u>356.2</u> | <u>1490.5</u> | .077 | .32 | 85JANAF | 13983-20-5 |
| | | | | | | | | |
| | | | <u>356.2</u> | <u>1490.5</u> | .079 | .33 | | |
| DLi^+ | LiD | 7.7±0.1 | (211) | (884) | 33.7 | 141.1 | 82TN270 | 13587-16-1 |
| | | | | | | | | |
| | | | (211) | (884) | 33.7 | 141.1 | | |
| D_2^+ | D ₂ | 15.46660±0.0001 | <u>356.7</u> | <u>1492.2</u> | 0 | 0 | *DEF | 7782-39-0 |
| | | | | | | | | |
| | | | <u>356.7</u> | <u>1492.2</u> | 0 | 0 | | |
| IP from 79HUB/HER. | | | | | | | | |
| D_2O^+ | D ₂ O | 12.635±0.007 | <u>231.8</u> | <u>970.0</u> | -59.56 | -249.20 | 85JANAF | 7789-20-0 |
| | | | | | | | | |
| | | | <u>232.5</u> | <u>972.8</u> | -58.85 | -246.25 | | |
| D_3O^+ | D ₃ O | 4.3±0.1 | | | | | | 24847-51-6 |
| IP from 84GEL/POR. | | | | | | | | |
| Dy^+ | Dy | 5.9390±0.0006 | 206 | 863 | 69 | 290 | 82TN270 | 7429-91-6 |
| | | | | | | | | |
| | | | 207.0 | 866.1 | 70.0 | 293.1 | | |
| DyF^+ | DyF | (6.0±0.3) | (101) | (422) | -38 | -157 | 79HUB/HER | |
| 0 K values. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--------------------------------------|--------------------|---|--------------------------|----------------|------------------------------|---------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| DyO⁺ | DyO | (6.08±0.1) | (122) (121) | (512) (508) | -18±10 -19 | -75±42 -79 | 83PED/MAR | 12175-28-9 |
| Er⁺ | Er | 6.1077±0.0006 | 217 216.9 | 906 907.6 | 76 76.1 | 317 318.3 | 82TN270 | 7440-52-0 |
| ErF⁺ | ErF | (6.3±0.3) 0 K values. | (105) | (441) | -40 | -167 | 79HUB/HER | |
| ErF₂⁺ | ErF ₂ | (7.0±0.3) | (-3) | (-11) | -164 | -686 | 82TN270 | |
| ErI₃⁺ | ErI ₃ | 9.0 IP is onset of photoelectron band (83RUS/GOO). | 125 | 524 | -82 | -344 | 82TN270 | 13813-42-8 |
| ErO⁺ | ErO | (6.30±0.1) See also: 80MUR/HIL. | (135) (132) | (566) (554) | -10±5 -13 | -42±21 -54 | 83PED/MAR | 12280-61-4 |
| Es⁺ | Es | 6.52±0.10 | 182 | 762 | 31.8±3 | 133±13 | 85KLE/WAR | 7429-92-7 |
| Eu⁺ | Eu | 5.67045±0.0003 | 172 173.1 | 722 724.2 | 42 42.3 | 175 177.1 | 82TN270 | 7440-53-1 |
| EuO⁺ | EuO | (6.48±0.1) See also: 81BAL/GIG, 85BAL/GIG. | (139) | (582) | -10 | -43 | 83PED/MAR | 12020-60-9 |
| EuO₂V⁺ | EuVO ₂ | (8) IP from 83BAL/GIG. 0 K values. | (108) | (450) | -77 | -322 | 83BAL/GIG | 88762-30-5 |
| EuO₃Ti⁺ | EuTiO ₃ | (6.5±0.5) IP from 85BAL/GIG. | (-62) | (-260) | -212±7 | -887±28 | 85BAL/GIG | 12020-61-0 |
| EuO₃V⁺ | EuVO ₃ | 8.1±0.5 IP from 83BAL/GIG. 0 K values. | (4) | (17) | (-183) | (-764) | 83BAL/GIG | 39432-21-8 |
| EuS⁺ | EuS | (6.8±0.3) | (184) (180) | (769) (751) | 27 23 | 113 95 | 82TN270 | 12020-65-4 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|---------------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F ⁺ F | 17.422 | <u>420.7</u> | <u>1760.2</u> | 19.0±0.1 | 79.4±0.3 | 85JANAF | 14762-94-8 |
| | | <u>420.2</u> | <u>1758.2</u> | 18.5±0.1 | 77.4±0.3 | | |
| FGa ⁺ GaF | (9.6±0.5) | (167) | (699) | -54 | -227 | 79HUB/HER | 13966-78-4 |
| | | (167) | (700) | -54 | -226 | | |
| A value of 10.7±0.6 eV has also been reported for this ionization potential. | | | | | | | |
| FGe ⁺ GeF | 7.46 | 166 | 694 | -6 | -26 | 81WOO | 14929-46-5 |
| | | 166 | 694 | -6 | -26 | | |
| IP from 79HUB/HER. | | | | | | | |
| FHF ⁺ HF | 16.044±0.003 | 304.9 | 1275.5 | -65.1±0.2 | -272.5±0.8 | 85JANAF | 7664-39-3 |
| | | 304.9 | 1275.5 | -65.1±0.2 | -272.5±0.8 | | |
| See also: 81KIM/KAT, 81BIE/ASB. | | | | | | | |
| FHO ⁺ HOF | 12.71±0.01 | 270 | 1128 | -23±1 | -98±4 | 82BAU/COX | 14034-79-8 |
| | | | | | | | |
| FH ₂ ⁺ H ₂ F | | 184 | 767 | | | | |
| | | | | | | | |
| From proton affinity of HF (RN 7664-39-3). A value of 205 kcal/mol, 859 kJ/mol is derived from the appearance potential of this ion from (HF) ₂ . PA = 117 kcal/mol, 489.5 kJ/mol. | | | | | | | |
| FH ₃ Si ⁺ SiH ₃ F | 11.7 | (180) | (752) | -90±5 | -377±21 | 78JANAF | 13537-33-2 |
| | | | | | | | |
| IP is onset of photoelectron band. | | | | | | | |
| FHo ⁺ HoF | (6.1±0.3) | (103) | (431) | -38 | -158 | 79HUB/HER | 16087-66-4 |
| | | | | | | | |
| 0 K values. | | | | | | | |
| FI ⁺ IF | 10.62 | 222.2 | 929.9 | -22.7±0.9 | -94.8±3.8 | 85JANAF | 13873-84-2 |
| | | 222.7 | 931.8 | -22.2±0.9 | -92.9±3.8 | | |
| IP from 84DYK/JOS. | | | | | | | |
| FIn ⁺ InF | (9.6±0.5) | (177) | (740) | -44 | -186 | 79HUB/HER | 13966-95-5 |
| | | | | | | | |
| 0 K values. | | | | | | | |
| FMg ⁺ MgF | (7.68) | (120.5) | (504) | -56.6±2.0 | -236.8±8.4 | 85JANAF | 14953-28-7 |
| | | (121) | (505) | -56.5±2.0 | -236.4±8.4 | | |
| FMn ⁺ MnF | (8.3±0.2) | (173) | (723) | -19 | -78 | 81WOO | 13569-25-0 |
| | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|-------------------|----------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| FMo⁺ | MoF | (8.0±0.3) | (249) | (1044) | 65 | 272 | 81WOO | 60388-18-3 |
| FN⁺ | NF | 12.26±0.01 | 338 | 1415 | 55.5±0.5 | 232.2±2.1 | 84BER/GRE | 13967-06-1 |
| IP from 82DYK/JON. $\Delta_f H(\text{Ion})$ from 84BER/GRE. See also: 79DUD/BAL. 0 K values. | | | | | | | | |
| FNO⁺ | NOF | 12.63±0.03 | 275 | 1152 | -16 | -67 | 82TN270 | 7789-25-5 |
| | | | 276 | 1154 | -16 | -65 | | |
| FNO₂⁺ | NO ₂ F | (13.09) | (276) | (1154) | -26±5 | -109±21 | 85JANAF | 10022-50-1 |
| | | | (277) | (1160) | -25±5 | -103±21 | | |
| FNS⁺ | NSF | 11.51±0.04 | 260 | 1090 | -5±0.9 | -21±4 | 73LAR/JOH | 18820-63-8 |
| FNa₂⁺ | Na ₂ F | 4.0±0.1 | 5 | 20 | -87±3 | -366±13 | 85KAP/RAD | 87331-13-3 |
| IP from 85KAP/RAD. 0 K values. | | | | | | | | |
| FNd⁺ | FNd | (5.0±0.3) | (81) | (338) | -34 | -144 | 81WOO | |
| A 298 K heat of formation of -161 kJ/mol, -38.5 kcal/mol (79HUB/HER) or -159 kJ/mol, -38.0 kcal/mol (82TN270) has also been recommended for NdF. | | | | | | | | |
| FO⁺ | OF | 12.77 | 320 | 1341 | 26±2 | 109±8 | 82BAU/COX | 12061-70-0 |
| | | | 320 | 1341 | 26 | 109 | | |
| FO₂⁺ | O ₂ F | (12.6±0.2) | (294) | (1229) | 3±0.5 | 13±2 | 85JANAF | 15499-23-7 |
| | | | (294) | (1230) | 3±0.5 | 14±2 | | |
| FP⁺ | PF | (≤9.74±0.01) | (≤212) | (≤887) | -12.5±5 | -52±21 | 85JANAF | 16027-92-2 |
| | | | (≤212) | (≤888) | -12±5 | -51±21 | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determinations (84BER/GRE). IP from 82DYK/JON2. See also: 75TOR/WES, 82LEV/LIA. | | | | | | | | |
| FPb⁺ | PbF | (7.5±0.3) | (154) | (644) | -19 | -80 | 81WOO | 14986-72-2 |
| | | | (154) | (646) | -19 | -78 | | |
| FS⁺ | SF | 10.09 | (236) | (987) | 3±1 | 13±6 | 85JANAF | 16068-96-5 |
| | | | (233) | (973) | 2.9±1.5 | 12±6 | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination (80GOM/HAA); corresponding IP = 9.9 eV. IP from 85LOS/WIL. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------|---------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| FS_2^+ SSF | | 194 | 811 | | | | |
| | From appearance potential determinations (85LOS/WIL). | | | | | | |
| FSi^+ SiF | 7.28 | 163 | 682 | -5±6 | -20±25 | 83WAL | 11128-24-8 |
| | | 166 | 680 | -5±6 | -22±25 | | |
| | IP from 79HUB/HER. | | | | | | |
| FSm^+ SmF | (5.7±0.3) | (68) | (286) | -63 | -264 | 79HUB/HER | 17209-59-5 |
| | | (73) | (307) | -58 | -243 | | |
| FSn^+ SnF | (7.04) | (142) | (593) | -21 | -86 | 81WOO | 13966-74-0 |
| FSr^+ SrF | (5.0±0.3) | (45) | (188) | -70.4±2.0 | -294.6±8.4 | 85JANAF | 13569-27-2 |
| | | (45) | (189.5) | -70.0±2.0 | -292.9±8.4 | | |
| FTl^+ TlF | 10.52 | 199 | 833 | -43 | -182 | 82TN270 | 7789-27-7 |
| | | 199 | 833 | -43 | -182 | | |
| FW^+ WF | (8.5±1) | (282) | (1180) | 86 | 360 | 81WOO | 51621-16-0 |
| FXe^+ XeF | (10.3) | (252) | (1057) | 15.3 | 64.0 | 79HUB/HER | 16757-14-5 |
| | $\Delta_f H(\text{Ion})$ from appearance potential determinations. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | |
| FY^+ YF | (6.3±0.3) | (121) | (507) | -24 | -101 | 79HUB/HER | 13981-88-9 |
| | 0 K values. | | | | | | |
| F_2^+ F ₂ | 15.697±0.003 | 362.0 | 1514.5 | 0 | 0 | *DEF | 7782-41-4 |
| | | 362.0 | 1514.5 | | | | |
| | IP from 84VAN/DEL2. See also: 84DYK/JOS, 81KIM/KAT, 81BIE/ASB. | | | | | | |
| F_2Fe^+ FeF ₂ | (11.3±0.3) | (177) | (740) | -84 | -350 | 81WOO | 7789-28-8 |
| | | (177) | (741) | -83 | -349 | | |
| F_2Ge^+ GeF ₂ | (11.65) | (132) | (551) | -137 | -573 | 81WOO | 13940-63-1 |
| | IP from 82JON/VAN3. | | | | | | |
| F_2HN^+ HNF ₂ | (11.53±0.08) | (250) | (1047) | -16±1 | -65±6 | 69PAN/ZER | 10405-27-3 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|------------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{F}_2\text{HO}_2\text{S}^+$ F_2SOOH | | 38 | 159 | | | | |
| | | From proton affinity of F_2SO_2 (RN 2699-79-8) (85MCM/KEB, 85MCM/KEB2) re-evaluated relative to CO standard (84LIA/LIE). PA = 146.2 kcal/mol, 612. kJ/mol. | | | | | |
| $\text{F}_2\text{H}_2\text{Si}^+$ SiH_2F_2 | 12.2 | (92) (94) | (386) (395) | -189±5 -187±5 | -791±21 -782±21 | 85JANAF | 13824-36-7 |
| | | IP is onset of photoelectron band. | | | | | |
| F_2IP^+ PF_2I^+ | (9.8) | (84) | (350) | -142.3±1 | -595.4±4.2 | 84BER/GRE | 13819-11-9 |
| | | IP is onset of photoelectron band. 0 K values. | | | | | |
| F_2Kr^+ KrF_2 | 13.1±0.05 | 317 | 1325 | 14±0.7 | 60±3 | 67GUN | 13773-81-4 |
| F_2Mg^+ MgF_2 | (13.4±0.4) | (136) (137) | (569) (571) | -173 -172 | -724 -722 | 82TN270 | 7783-40-6 |
| F_2Mn^+ MnF_2 | (11.4) | (137) | (575) | -125 | -525 | 81WOO | 7782-64-1 |
| F_2Mo^+ MoF_2 | (9.00±0.15) | (167) | (700) | -40 | -168 | 81WOO | 20205-60-1 |
| F_2MoO_2^+ MoO_2F_2 | (13.0±0.3) | (56) | (236) | -243 | -1018 | 81WOO | 13824-57-2 |
| F_2N^+ NF_2 | 11.628±0.01 | 275 276.5 | 1153 1156.8 | 8 8.3 | 31 34.9 | 84BER/GRE | 3744-07-8 |
| | | IP from 84BER/GRE. See also: 79DUD/BAL. | | | | | |
| F_2NS^+ NSF_2 | | 253 | 1060 | | | | |
| | | From appearance potential (15.47 eV) in NSF_3 . | | | | | |
| F_2N_2^+ (E)- N_2F_2 | (12.8) | (315) (316) | (1316) (1321) | 19±1 21±1 | 81±5 86±5 | 85JANAF | 13776-62-0 |
| F_2Nd^+ F_2Nd | (5.6±0.3) | (-29) | (-120) | -158 | -660 | 81WOO | |
| F_2O^+ OF_2 | 13.11±0.01 | 308 309 | 1290 1292 | 5.9±0.4 6.4±0.4 | 24.5±1.6 26.8±1.6 | 85JANAF | 7783-41-7 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------------|--------------------------|----------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F_2OS^+ SOF ₂ | 12.25 | (164) (166) | (688) (693) | -118±8 -117±8 | -494±32 -489±32 | 87HER | 7783-42-8 |
| See also: 81COS/LLO. | | | | | | | |
| $\text{F}_2\text{O}_2\text{S}^+$ S ₂ O ₂ F ₂ | 13.04±0.01 | 119 121 | 499 508 | -181±2 -179±2 | -759±8 -750±8 | 87HER | 2699-79-8 |
| $\text{F}_2\text{O}_2\text{W}^+$ WO ₂ F ₂ | (12.5±0.3) IP from 81MAL/MEL. | (70) | (291) | -219 | -915 | 81WOO | 14118-73-1 |
| F_2P^+ PF ₂ | 8.847±0.010 | 90 90 | 378.5 375 | -115±0.5 -114.4±0.5 | -482±2.1 -478.6±2.1 | 84BER/GRE | 13873-52-4 |
| IP from 84BER/GRE. See also: 75TOR/WES, 82LEV/LIA. | | | | | | | |
| F_2Pb^+ PbF ₂ | (11.5) | (162) (163) | (677) (681) | -103 -102 | -432 -429 | 81WOO | 7783-46-2 |
| IP is onset of photoelectron band(83NOV/POT2). | | | | | | | |
| F_2Pt^+ PtF ₂ | (11.85±0.25) | (247) | (1032) | -26±6 | -111±25 | 83KOR/BON | 18820-56-9 |
| IP from 83KOR/BON. | | | | | | | |
| F_2S^+ SF ₂ | (10.08) | (161) (162) | (676) (678) | -71±4 -70±4 | -297±17 -295±17 | 87HER | 13814-25-0 |
| See also: 80GOM/HAA, 85LOS/WIL. | | | | | | | |
| F_2S_2^+ FSSF | 10.62±0.02 | 176 177 | 739 742 | -68±2 -67±2 | -286±10 -283±10 | 87HER | 13709-35-8 |
| IP from 85LOS/WIL. | | | | | | | |
| SSF ₂ | 10.41±0.02 | 169 | 707 | -71±2 | -297±10 | 87HER | 101947-30-2 |
| IP from 85LOS/WIL. See also: 84COO/KRO, 82LEV/LIA. | | | | | | | |
| F_2Si^+ SiF ₂ | 10.78±0.05 | (108) (108) | (450) (451) | -141±2 -141±2 | -590±8 -589±8 | 83WAL | 13966-66-0 |
| F_2Sn^+ SnF ₂ | (11.1) | (140) | (586) | -116 | -485 | 81WOO | 7783-47-3 |
| IP is onset of photoelectron band (83NOV/POT2). | | | | | | | |
| F_2Ti^+ TiF ₂ | (12.2±0.5) | (125) | (524) | -156 | -653 | 81WOO | 13814-20-5 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|----------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F_2W^+ WF ₂ | (9.0±0.3) | (182) | (763) | -25 | -105 | 81WOO | 33963-15-4 |
| F_2Xe^+ XeF ₂ | 12.35±0.01 | 259 | 1085 | -26±0.2 | -107±1 | 72JOH/MAL | 13709-36-9 |
| F_2Zr^+ F ₂ Zr | (12.0±0.5) | (143) (144) | (600) (602) | -133±5 -133±5 | -558±21 -556±21 | 85JANAF | 13842-94-9 |
| F_3Fe^+ FeF ₃ | (12.5±0.3) | (101) (101) | (421) (424) | -188 -187 | -785 -782 | 81WOO | 7783-50-8 |
| F_3HN^+ F ₂ NFH | | 199 | 832 | | | | |
| | | From proton affinity of NF ₃ (RN 7783-54-2) (85MCM/KEB) re-evaluated relative to CO standard (84LIA/LIE). PA = 136.9 kcal/mol, 573. kJ/mol. | | | | | |
| F_3HOP^+ P(OH)F ₃ | | -91 | -383 | | | | |
| | | From proton affinity of POF ₃ (RN 13478-20-1). PA = 167.8 kcal/mol, 702. kJ/mol. | | | | | |
| F_3HP^+ HPF ₃ | | -20 | -86 | | | | |
| | | From proton affinity of PF ₃ (RN 7783-55-3). PA = 166.5 kcal/mol, 697. kJ/mol. | | | | | |
| F_3HSi^+ SiHF ₃ | (14.0) | (36) (37) | (150) (157) | -287±5 -285±5 | -1201±21 -1194±21 | 85JANAF | 13465-71-9 |
| | | IP is onset of photoelectron band. | | | | | |
| $\text{F}_3\text{H}_2\text{OSi}^+$ SiF ₃ OH ₂ | | (-264) | (-1103) | | | | |
| | | From proton affinity of SiF ₃ OH (84REE/MUJ). PA = (162) kcal/mol, (676) kJ/mol. | | | | | |
| F_3Mn^+ MnF ₃ | (12.57±0.2) | (104) | (434) | -186 | -779 | 81WOO | 7783-53-1 |
| F_3Mo^+ MoF ₃ | (9.88±0.10) | (87) (87) | (361) (364) | -141 -141 | -592 -589 | 81WOO | 20193-58-2 |
| F_3MoS^+ MoSF ₃ | (13.0±0.3) | (134) | (559) | -166±6 | -695±27 | 80MAL/ALI | 67374-76-9 |
| | | IP from 80MAL/ALI, 80MAL/ALI2. | | | | | |
| F_3N^+ NF ₃ | 13.00±0.02 | 268 270 | 1122 1128 | -31±0.3 -30±0.3 | -132±1 -126±1 | 85JANAF | 7783-54-2 |
| | | See also: 84BER/GRE, 84BER/GRE2, 79DUD/BAL. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|------------------|------------------------------|----------------------|----------------------|--|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F_3NO^+ NOF_3 | 13.26±0.01 | (267) (269) | (1116) (1124) | -39±5 -37±5 | -163±21 -155±21 | 85JANAF | 13847-65-9 |
| F_3NS^+ NSF_3 | (12.0) IP is onset of photoelectron band. | (192) | (802) | -85±0.5 | -356±2 | 70O'H/HUB | 15930-75-3 |
| F_3OP^+ POF_3 | 12.76±0.01 See also: 83NES/MIL. | (-5) (-3) | (-23) (-14) | -300±2 -298±2 | -1254±8 -1245±8 | 85JANAF | 13478-20-1 |
| F_3OV^+ VOF_3 | (13.88±0.05) | (25) | (105) | -295.0±7.0 | -1234±29 | 75FLE/SVE | 13709-31-4 |
| F_3P^+ PF_3 | 11.44 IP from 84BER/GRE. See also: 83NES/MIL, 75TOR/WES, 82LEV/LIA. | (35) (36) | (146) (151) | -229±1 -228±1 | -958±4 -953±4 | 85JANAF | 7783-55-3 |
| F_3PS^+ F_3PS | ≤11.05±0.035 | (≤14) (≤16) | (≤57) (≤65) | -241±15 -239±15 | -1009±63 -1001±63 | 85JANAF | 2404-52-6 |
| F_3Sb^+ SbF_3 | (12.1) IP is onset of photoelectron band(83NOV/POT). | (80) | (334) | -199 | -833 | 81WOO | 7783-56-4 |
| F_3Si^+ SiF_3 | (9.3) From appearance potential (13.33 eV) in CH_3SiF_3 . IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | (-24) (-23) | (-99) (-96) | -239±5 -238±5 | -1000±21 -996±21 | 83WAL | |
| F_3Ti^+ TiF_3 | (10.5±0.5) | (-36) (-35) | (-151) (-147) | -278 -277 | -1164 -1160 | 81WOO | 7783-57-5 |
| F_3W^+ WF_3 | (9.0±0.2) | (81) | (337) | -127 | -531 | 81WOO | 51621-17-1 |
| F_3Xe^+ XeF_3 | | 234 | 981 | | | | From appearance potential (13.10 eV) in XeF_4 . |
| F_4Ge^+ GeF_4 | (15.5) IP is onset of photoelectron band (75LLO/ROB). | (73) | (306) | -284 | -1190 | 81WOO | 7783-58-6 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|---|----------------|------------------------------|----------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F_4HSi^+ SiF_3FH | | -140 | -588 | | | | |
| | | From proton affinity of SiF_4 (RN 7783-61-1)(84REE/MUJ). PA = 120.2 kcal/mol, 503. kJ/mol. | | | | | |
| F_4Mo^+ MoF_4 | (9.9) | (0) (1) | (1) (4) | -228 -227 | -954 -951 | 81WOO | 23412-45-5 |
| F_4MoO^+ MoOF_4 | 13.8 | 2 3 | 6 13 | -317 -315 | -1325 -1318 | 86BUR/FAW | 14459-59-7 |
| | | IP is onset of photoelectron band (81VOV/DUD). | | | | | |
| F_4MoS^+ MoSF_4 | (12.6±0.3) | (58) | (245) | -232±7 | -971±29 | 80MAL/ALI | 70487-60-4 |
| | | IP from 80MAL/ALI, 80MAL/ALI2. | | | | | |
| F_4N_2^+ N_2F_4 | 11.94±0.03 | 267 270 | 1119 1131 | -8 -5 | -33 -21 | 84BER/GRE | 10036-47-2 |
| | | IP from 84BER/GRE. | | | | | |
| F_4ORe^+ ReOF_4 | 10.5 | -22 | -91 | -264 | -1104 | 86BUR/FAW | 17026-29-8 |
| | | IP is onset of photoelectron band (81VOV/DUD). | | | | | |
| F_4OS^+ SOF_4 | (12.3) | (61) | (254) | -223±11 | -933±44 | 87HER | 13709-54-1 |
| | | IP is onset of photoelectron band (81COS/LLO). | | | | | |
| F_4OW^+ WOF_4 | 13.6 | -28 -27 | -119 -111 | -342 -340 | -1431 -1423 | 86BUR/FAW | 13520-79-1 |
| | | IP is onset of photoelectron band (81VOV/DUD). | | | | | |
| F_4P_2^+ P_2F_4 | ≤9.28 | ≤-56 | ≤-235 | -270 | -1130 | 84BER/GRE | 13824-74-3 |
| | | IP from 84BER/GRE, 82LEV/LIA. 0 K values. | | | | | |
| F_4Pb^+ PbF_4 | (10.4±0.3) | (42) (41) | (175) (170) | -198 -197 | -828 -823 | 81WOO | 7783-59-7 |
| F_4Pt^+ PtF_4 | (12.83±0.28) | (171) | (714) | -125±6 | -524±25 | 83KOR/BON | 13455-15-7 |
| | | IP from 83KOR/BON. | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|------------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F_4S^+ SF_4 | 12.03±0.05 | 95 98 | 397 408 | -182±5 -181±5 | -763±21 -757±21 | 85JANAF | 7783-60-0 |
| | See also: 81COS/LLO. | | | | | | |
| F_4SW^+ WSF_4 | (≤12.0±0.2) IP from 81MAL/ALI. | (≤5) | (≤21) | ≤-272±9 | ≤-1137±38 | 81MAL/ALI | 41831-80-5 |
| F_4S_2^+ F_3SSF | (10.15±0.10) IP from 80GOM/HAA. | (76) | (319) | -158±6 | -660±24 | 87HER | 27245-05-2 |
| F_4Si^+ SiF_4 | (15.7) | (-24) (-23) | (-100) (-94) | -386.0±0.3 -384.6±0.3 | -1615±1 -1609±1 | 85JANAF | 7783-61-1 |
| | IP is onset of photoelectron band. (75LLO/ROB, 82BIE/ASB, 82LEV/LIA) | | | | | | |
| F_4U^+ UF_4 | (9.51) | (-163) (-162) | (-681) (-676) | -382 -381 | -1599 -1594 | 82TN270 | 10049-14-6 |
| F_4W^+ WF_4 | (9.89±0.10) | (-18) | (-75) | -246 | -1029 | 81WOO | 13766-47-7 |
| F_4Xe^+ XeF_4 | 12.65±0.1 | 242 | 1015 | -49±0.2 | -206±1 | 72JOH/MAL | 13709-61-0 |
| F_5I^+ IF_5 | 12.943±0.005 | 106 108 | 445 453 | -201±0.5 -199±0.5 | -840±2 -832±2 | 85JANAF | 7783-66-6 |
| F_5Mo^+ MoF_5 | 10.5±0.3 | -54 -53 | -228 -223 | -297±1 -295±1 | -1241±4 -1236±4 | 85JANAF | 13819-84-6 |
| | IP from 80MAL/ALI2. | | | | | | |
| F_5ORe^+ ReOF_5 | (13.2±0.1) | (21) | (88) | -283 | -1186 | 81WOO | 23377-53-9 |
| F_5P^+ PF_5 | (15.1) | (-33) (-30) | (-139) (-127) | -381 -379 | -1596 -1584 | 82TN270 | 7647-19-0 |
| | IP is onset of photoelectron band. | | | | | | |
| F_5S^+ SF_5 | 10.5±0.1 | 23 25 | 97 106 | -219 -216 | -915.9 -906 | 81BAB/STR | 10546-01-7 |
| | IP from charge exchange bracketing experiments (81BAB/STR); $\Delta_f H(\text{Ion})$ from equilibrium: $\text{CF}_3^+ + \text{SF}_6 = \text{SF}_5^+ + \text{CF}_4$ (81BAB/STR). | | | | | | |

Table 1. Positive Ion Table - Continued

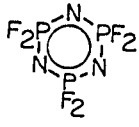
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--|------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| F_5U^+ UF_5 | (11.4) | (-200) (-199) | (-837) (-832) | -463 -462 | -1937 -1932 | 82TN270 | 13775-07-0 |
| F_5W^+ WF_5 | (10.03±0.10) | (-103) | (-429) | -334 | -1397 | 81WOO | 19357-83-6 |
| F_6Mo^+ MoF_6 | (14.5±0.1) | (-38) (-36) | (-159) (-152) | -372.4±0.2 -370.7±0.2 | -1558±1 -1551±1 | 85JANAF | 7783-77-9 |
| $\text{F}_6\text{N}_3\text{P}_3^+$  | 11.58 IP form 82LEV/LIA and 81CLA/SOW. | (-245) | (-1024) | -512 | -2141 | *EST | 15599-91-4 |
| F_6Re^+ ReF_6 | (11.0) IP from 80VOV/DUD. | (-69) | (-288) | -322 | -1349 | 84BAR/YEH | 10049-17-9 |
| F_6S^+ SF_6 | 15.33±0.03 See also: 82BIE/ASB. | 62 65 | 259 273 | -291.7±0.2 -288.3±0.2 | -1220.5±.8 -1206.5±.8 | 85JANAF | 2551-62-4 |
| F_6U^+ UF_6 | 14.00±0.10 | -190 -189 | -796 -791 | -513 -512 | -2147 -2141 | 82TN270 | 7783-81-5 |
| F_6Xe^+ XeF_6 | 12.19±0.02 | 214 | 897 | -67±0.5 | -279±2 | 72JOH/MAL | 13693-09-9 |
| F_7Re^+ ReF_7 | (14.1±0.1) | (-16) | (-69) | -342±3 | -1429±13 | 84BAR/YEH | 17029-21-9 |
| Fe^+ Fe | 7.870 See also: 82DYK/GRA. | 281 280 | 1175 1173 | 99 99 | 416 414 | 82TN270 | 7439-89-6 |
| FeH^+ FeH | | (283) | (1184) | | | | 15600-68-7 |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM3). See also: 81ARM/HAL, 84HAL/KLE. | | | | | |
| FeHO^+ FeOH | 7.9±0.2 | 211 (214) | 884 (895) | 32 | 133 | 80MUR | 12315-09-2 |
| | | IP from 80MUR. $\Delta_f H(\text{Ion})$ at 298 K from proton affinity of FeO (84CAS/FRE). | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------|----------------------------|--------------------------|--------|------------------------------|----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| FeI^+ | FeI | (7.8±0.5) | (247) | (1033) | (67) | (280) | 84GRA/ROS2 | |
| $\Delta_f H(\text{Ion})$ from 84GRA/ROS2. Cited IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See also: 85GRA/ROS. 0 K values. | | | | | | | | |
| FeI_2^+ | FeI_2 | 9.3 | (233) | (976) | (19) | (79) | 84GRA/ROS2 | 7783-86-0 |
| IP from 84GRA/ROS. See also: 84GRA/ROS2, 85GRA/ROS. 0 K values. | | | | | | | | |
| FeO^+ | FeO | 8.9±0.1 | 265.2 | 1109.7 | 60.0±5 | 251.0±21 | 85JANAF | 1345-25-1 |
| 265.3 1109.8 60.0±5 251.1±21 IP from 82ARM/HAL. See also: 84JAC/JAC, 81ARM/HAL, 81KAP/STA, 80MUR. | | | | | | | | |
| FeV^+ | VFe | (5.4) | (302) | (1264) | 177 | 743 | 85HET/FRE | |
| $\Delta_f H(\text{Ion})$ and IP from 85HET/FRE. 0 K values. | | | | | | | | |
| Fe_2^+ | Fe_2 | 6.30±0.01 | 325 | 1361 | 180 | 753 | 82SHI/GIN | 12596-01-9 |
| IP from 84ROH/COX. 0 K values. | | | | | | | | |
| Fm^+ | Fm | 6.64±0.11 | | | | | | 7440-72-4 |
| Ga^+ | Ga | 5.999 | 203 | 851 | 65.0 | 272.0 | 85JANAF | 7440-55-3 |
| 203 850 64.8 271.0 See also: 85HIR/STR. | | | | | | | | |
| GaI^+ | GaI | (9.0±0.3) | (219) | (915) | 11.1 | 46.4 | 79HUB/HER | 15605-68-2 |
| (219) (917) 11.6 48.5 | | | | | | | | |
| GaI_3^+ | GaI_3 | 9.40 | 183 | 765 | -34 | -142 | 82TN270 | 13450-91-4 |
| GaO^+ | GaO | (9.4±0.5) | (257) | (1074) | 40±10 | 167±42 | 83PED/MAR | 12024-08-7 |
| (257) (1074) 40 167 | | | | | | | | |
| Gd^+ | Gd | 6.1502±0.0006 | 237 | 991 | 95 | 398 | 82TN270 | 7440-54-2 |
| 237.2 992.3 95.3 398.9 | | | | | | | | |
| GdO^+ | GdO | (5.75±0.1) | (116) | (486) | -16±3 | -69±13 | 83PED/MAR | 12024-77-0 |
| (116) (484) -17 -71 See also: 80MUR/HIL. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--------------------------------|----------------------------|--------------------------|------------------|------------------------------|----------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| GdS⁺ | GdS | (6.9±0.6) 0 K values. | (197) | (825) | 38 | 159 | 82TN270 | 12134-74-6 |
| Ge⁺ | Ge | 7.899 | 272.2 271.5 | 1138.7 1135.9 | 90.0 89.3 | 376.6 373.8 | 82TN270 | 7440-56-4 |
| GeH₄⁺ | GeH ₄ | 11.33 | 283 285 | 1184 1195 | 22 24 | 91 102 | 64GUN/GRE | 7782-65-2 |
| GeH₅⁺ | GeH ₅ | | 221 | 926 | | | | |
| From proton affinity of GeH ₄ (RN 7782-65-2) (80SEN/ABE). PA = 166.2 kcal/mol, 695. kJ/mol. | | | | | | | | |
| GeI₂⁺ | GeI ₂ | (8.9) | (216) | (906) | 11.2 | 46.9 | 82TN270 | 13573-08-5 |
| IP is onset of photoelectron band (83JON/VAN). | | | | | | | | |
| GeI₄⁺ | GeI ₄ | (9.42) | (204) (205) | (852) (857) | -14 -12.3 | -57 -51.5 | 82TN270 | 13450-95-8 |
| GeO⁺ | GeO | 11.25±0.01 | 250 250 | 1044 1044 | -9.9±0.7 -9.9 | -41±3 -41 | 84RAU/SCH | 20619-16-3 |
| GeS⁺ | GeS | 9.98±0.02 | 252 | 1055 | 22 | 92 | 82TN270 | 12025-32-0 |
| GeSe⁺ | GeSe | (9.3) | (230) (237) | (964) (993) | 23 23 | 96 96 | 77PED/RYL | 12065-10-0 |
| IP is onset of photoelectron band. | | | | | | | | |
| GeSi⁺ | GeSi | 8.2±0.3 0 K values. | 315 | 1319 | 126 | 528 | 79HUB/HER | 12025-36-4 |
| Ge₂⁺ | Ge ₂ | (7.8) | (293) (293) | (1226) (1224) | 113 113 | 473 473 | 86KIN/NAG | 12596-05-3 |
| Ge₂H₆⁺ | Ge ₂ H ₆ | (12.5±0.3) | (327) | (1368) | 38.8 | 162 | 64GUN/GRE | 13818-89-8 |
| Ge₃H₈⁺ | Ge ₃ H ₈ | (9.6±0.3) | (276) | (1153) | 54.2 | 227 | 64GUN/GRE | 14691-44-2 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--------------------|----------------------------|------------------------------|--------------------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| H ⁺ | H | 13.598 | <u>365.7</u> <u>365.2</u> | <u>1530.0</u> <u>1528.0</u> | 52.10 51.63 | 217.999 216.035 | 85JANAF | 12385-13-6 |
| HHe ⁺ | HHe | | 323 | 1352 | | | | 13766-24-0 |
| $\Delta_f H(\text{Ion})$ from 79HUB/HER. Corresponding proton affinity of He = 42.5 kcal/mol, 178. kJ/mol. | | | | | | | | |
| HI ⁺ | HI | 10.386±0.001 | 245.8 246.3 | 1028.5 1030.6 | 6.3±0.05 6.8±0.05 | 26.4±0.2 28.5±0.2 | 85JANAF | 10034-85-2 |
| See also: 81KIM/KAT. | | | | | | | | |
| HK ⁺ | KH | (8.0±1.0) | (214) (215) | (895) (896) | 29±3 30±3 | 123±15 126±15 | 85JANAF | 7693-26-7 |
| IP from 82FAR/SRI. | | | | | | | | |
| HKO ⁺ | KOH | (7.50±0.15) | (117) (119) | (491) (497) | -56 -54 | -233 -227 | 81LIN/BES | 1310-58-3 |
| See also: 82FAR/SRI. | | | | | | | | |
| HKr ⁺ | KrH | | 264 | 1105 | | | | |
| From proton affinity of Kr (RN 7439-90-9). PA = 101.6 kcal/mol, 425. kJ/mol. | | | | | | | | |
| HLi ⁺ | LiH | 7.7 | (211) (211) | (882) (882) | 33.3±0.01 33.3±0.01 | 139.2±0.04 139.4±0.04 | 79HUB/HER | 7580-67-8 |
| IP from 79HUB/HER. | | | | | | | | |
| HLi ₂ O ⁺ | Li ₂ OH | | 37 | 155 | | | | |
| From reaction enthalpies of Li ₂ OH ⁺ ions (84BUT/KUD). Corresponding proton affinity of Li ₂ O (RN 12057-24-8) is 289 kcal/mol, 1208 kJ/mol. | | | | | | | | |
| HMgO ⁺ | MgOH | 7.5±0.3 | (143) | (599) | -30 | -125 | 81MUR | 12141-11-6 |
| $\Delta_f H(\text{Ion})$ from 81MUR. 0 K values. | | | | | | | | |
| HMn ⁺ | MnH | (7.8) | (242) | (1011) | (61) | (256) | 79HUB/HER | 14452-76-7 |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM2). See also: 81ARM/HAL. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | | | |
| HMn ₂ ⁺ | Mn ₂ H | | (284) | (1186) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ARM). 0 K value. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|----------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| HM_o⁺ MoH | | (331) | (1385) | | | | |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction. 0 K value. | | | | | |
| HN⁺ NH | 13.49±0.01 | 401.1 | 1678.2 | 90.0±4.0 | 376.6±16.7 | 85JANAF | 13774-92-0 |
| | | 401.1 | 1678.1 | 90.0±4.0 | 376.5±16.7 | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (85GIB/GRE). | | | | | |
| HNO⁺ NOH | | 274.3 | 1147.7 | | | | |
| | | 274.8 | 1149.8 | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (82KUT/GOO). | | | | | |
| HNO | (10.1) | (256.3) | (1072.3) | 24 | 100 | 82BAU/COX | 14332-28-6 |
| | | (256.8) | (1074.4) | | | | |
| | | $\Delta_f H(\text{Ion})$ from appearance potential determination (82KUT/GOO). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | |
| HNOS⁺ HN=S=O | (11.3) | (302) | (1265) | 41.7 | 175 | 82OLE/TUR | 13817-04-4 |
| | | IP is onset of photoelectron band. | | | | | |
| HNO₂⁺ HNO ₂ | ≤11.3 | 234 | 977 | -19 | -80 | 82BAU/COX | 7782-77-6 |
| | | ≤243 | ≤1018 | -17 | -72 | | |
| | | $\Delta_f H(\text{Ion})$ at 298 K from proton affinity of NO ₂ (RN 10102-44-0) (84POL/MUN). PA = 140 kcal/mol, 586 kJ/mol. | | | | | |
| HNO₃⁺ HNO ₃ | 11.95±0.01 | 244 | 1018 | -32 | -135 | 82BAU/COX | 7697-37-2 |
| | | 246 | 1028 | -30 | -125 | | |
| HN₂⁺ HN ₂ | | 247.5 | 1035.5 | | | | |
| | | From proton affinity of N ₂ ; threshold determination (82LEV/LIA) gives the same value. PA = 118.2 kcal/mol, 494.5 kJ/mol. | | | | | |
| HN₂O⁺ HNNO | | 246 | 1031 | | | | |
| | | From proton affinity of N ₂ O (RN 10024-97-2). See also: 85MCM/KEB, 85MCM/KEB2. PA = 138.8 kcal/mol, 581 kJ/mol. | | | | | |
| HN₃⁺ HN ₃ | 10.72±0.025 | 317.5 | 1328.3 | 70.3 | 294.1 | 82TN270 | 7782-79-8 |
| | | 318.9 | 1334.8 | 71.8 | 300.5 | | |
| HNb⁺ NbH | | (330) | (1380) | | | | |
| | | $\Delta_f H(\text{Ion})$ from onset of endothermic reaction. 0 K value. | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--------------------|----------------------------|--------------------------|------------------|--------------------------------|----------------------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| HNe ⁺ | NeH | | 318 | 1329 | | | | |
| $\Delta_f H(\text{Ion})$ from 68CHU/RUS. Corresponding proton affinity of Ne = 48.1 kcal/mol, 201. kJ/mol. | | | | | | | | |
| HNI ⁺ | NiH | (≤ 9.0) | (291) | (1216) | > (83) | > (347) | 79HUB/HER | 14332-32-2 |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM4). See also: 80ARM/BEA, 81ARM/HAL. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | | | |
| HO ⁺ | OH | 13.00 | 309.1 309.0 | 1293.3 1292.7 | 9.3 \pm 0.3 9.2 \pm 0.3 | 39.0 \pm 1.2 38.4 \pm 1.2 | 85JANAF | 3352-57-6 |
| $\Delta_f H(\text{Ion})$ from appearance potential measurements; IP derived from $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$ is in good agreement with the experimentally determined value of 13.01 eV. See also: 84VAN/DEL. | | | | | | | | |
| HOSr ⁺ | SrOH | 5.1 \pm 0.2 | 74 | 309 | -44 | -183 | 83MUR | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (83MUR); IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | | | |
| HO ₂ ⁺ | HO ₂ | 11.35 \pm 0.01 | 264.2 264.9 | 1105.5 1108.5 | 2.5 3.2 | 10.5 13.4 | 82TN270 | 3170-83-0 |
| IP from 81DYK/JON. Value of $\Delta_f H(\text{Ion})$ from appearance potential measurements corrected to 298 K: 264.8 kcal/mol, 1107.9 kJ/mol. | | | | | | | | |
| HO ₂ S ⁺ | OSOH | | 143 | 597 | | | | |
| From proton affinity of SO ₂ (RN 7446-09-5) (85MCM/KEB, 85MCM/KEB2) re-evaluated relative to CO standard (84LIA/LIE). PA = 152.1 kcal/mol, 636. kJ/mol. | | | | | | | | |
| HO ₃ S ⁺ | O ₂ SOH | | (133) | (557) | | | | |
| From proton affinity of SO ₃ (RN 7446-11-9). PA = (138) kcal/mol, (577) kJ/mol. | | | | | | | | |
| HP ⁺ | PH | 10.18 \pm 0.1 | 291 291 | 1218 1219 | 56 \pm 2 57 \pm 2 | 236 \pm 8 237 \pm 8 | 86BER/CUR | 13967-14-1 |
| $\Delta_f H(\text{Ion})$ from 79HUB/HER, 86BER/CUR. IP from 86BER/CUR. | | | | | | | | |
| HPd ⁺ | PdH | | (281) | (1176) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM). 0 K value. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------|----------------------------|--------------------------|------------------|------------------------------|------------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| HS⁺ | SH | 10.37±0.01 | 272.4 271.8 | 1139.8 1137.0 | 33.3±1.2 32.6±1.2 | 139.3±5.0 136.5±5.0 | 85JANAF | 13940-21-1 |
| IP from 79DUN/DYK, $\Delta_f H(\text{Ion})$ from 83PRE/TZE, in good agreement with earlier results. See also: 81SMI/ADA. | | | | | | | | |
| HSc⁺ | ScH | | 239 | 999 | | | | 33486-02-1 |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). See also: 87SUN/ARI. 0 K values. | | | | | | | | |
| HSe⁺ | SeH | (9.79) | (258) | (1080) | (32) | (135) | 79HUB/HER | 13940-22-2 |
| 0 K values. | | | | | | | | |
| HSi⁺ | SiH | 7.89±0.07 | 272.0 271.5 | 1138.0 1136.2 | 90.0±2.0 89.6±2.0 | 376.7±8.4 374.9±8.4 | 85JANAF | 13774-94-2 |
| $\Delta_f H(\text{Ion})$ from 84ELK/ARM. IP from 87BOO/ARM. | | | | | | | | |
| HTe⁺ | TeH | (9.09) | (244) | (1020) | 34 | 143 | 79HUB/HER | 13940-36-8 |
| HTi⁺ | TiH | (6.0) | (265) | (1109) | (127) | (532) | 79HUB/HER | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (86ELK/ARM). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. 0 K values. | | | | | | | | |
| HU⁺ | UH | | 256 | 1070 | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (77ARM/HOD). | | | | | | | | |
| HV⁺ | VH | | (282) | (1179) | | | | |
| $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM). 0 K value. | | | | | | | | |
| HXe⁺ | XeH | | 247 | 1034 | | | | |
| From proton affinity of Xe (RN 7440-63-3). PA = 118.6 kcal/mol, 496. kJ/mol. | | | | | | | | |
| HY⁺ | YH | | (238) | (995) | | | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction. 0 K value. | | | | | | | | |
| HZn⁺ | ZnH | (9.4) | (241) | (1008) | (25) | (106) | 79HUB/HER | |
| From proton affinity of Zn (RN 7440-66-6). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--------------------------|----------------------------|--------------------------|---------------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| HZr^+ | ZrH | | | | 123.4 | 516.3 | 85JANAF | |
| | | | (301) | (1260) | 123.6 | 517.3 | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction. 0 K value. | | | | | | | | |
| H_2^+ | H_2 | 15.42589±0.00005 | <u>355.7</u> | <u>1488.3</u> | 0 | 0 | *DEF | 1333-74-0 |
| | | | <u>355.7</u> | <u>1488.3</u> | 0 | 0 | | |
| See also: 81KIM/KAT. | | | | | | | | |
| H_2I^+ | H_2I | | 225 | 941 | | | | |
| From proton affinity of HI (RN 10034-85-2) results of 85MCM/KEB re-evaluated relative to CO standard (84LIA/LIE). PA = 147.1 kcal/mol, 615. kJ/mol. | | | | | | | | |
| $\text{H}_2\text{I}_2\text{Si}^+$ | SiH_2I_2 | (9.4) | (208) | (896) | -9±5 | -38±20 | 85JANAF | 13760-02-6 |
| | | | (206) | (861) | -7±5 | -28±20 | | |
| IP is onset of photoelectron band. | | | | | | | | |
| H_2KO^+ | KOH_2 | | (47) | (198) | | | | |
| $\Delta_f H(\text{Ion})$ from equilibrium constant determination (69SEA/DZI). | | | | | | | | |
| H_2N^+ | NH_2 | 11.14±0.01 | 302.0 | 1263.8 | 45.1±0.3 | 188.7±1.3 | 85GIB/GRE | 15194-15-7 |
| | | | 302.7 | 1266.4 | 45.8±0.3 | 191.6±1.3 | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination is in agreement. IP from 85GIB/GRE. | | | | | | | | |
| H_2NO^+ | H_2NO | | 224.6 | 939.7 | | | | |
| $\Delta_f H(\text{Ion})$ from appearance potential determination (82KUT/GOO). | | | | | | | | |
| H_2N_2^+ | (Z)-HN=NH | (9.52±0.05) | (275) | (1150) | 55 | 232 | 82CAS/GOD | 28647-38-3 |
| | (E)-HN=NH | (9.59±0.01) | (272) | (1137) | 51 | 212 | 82CAS/GOD | 3618-05-1 |
| H_2N_3^+ | H_2NNN | | (257) | (1075) | | | | |
| From proton affinity of HN_3 (RN 7782-79-8) (84BEA/EYE). PA = 179 kcal/mol, 749 kJ/mol. | | | | | | | | |
| H_2NaO^+ | NaOH_2 | | 71 | 296 | | | | |
| From proton affinity of NaOH (RN 1310-73-2) (69SEA/DZI). PA = 248 kcal/mol, 1036 kJ/mol. | | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|------------------------------|------------------------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| H_2O^+ H_2O | 12.612±0.010 | <u>233.0</u> <u>233.7</u> | <u>975.0</u> <u>977.9</u> | -57.80 -57.10 | -241.83 -238.92 | 85JANAF | 7732-18-5 |
| See also: 81KIM/KAT. | | | | | | | |
| H_2O_2^+ H_2O_2 | 10.54 | 210 212 | 881 887 | -32.6 -31.1 | -136.3 -130.0 | 82BAU/COX | 7722-84-1 |
| See also: 77ASH/BUR, 81KIM/KAT. | | | | | | | |
| H_2P^+ PH_2 | 9.824±0.002 | 261 261 | 1093 1090 | 33.3 34.0±0.6 | 139.5 142.2±2.5 | 86BER/CUR | |
| IP from 86BER/CUR. See also: 82DYK/JON2. 0 K values. | | | | | | | |
| H_2S^+ H_2S | 10.453±0.008 | 236 237 | 988 991 | -4.9 -4.2 | -20.5±0.8 -17.6±0.8 | 85JANAF | 7783-06-4 |
| IP is average of several spectroscopic and photoionization-onset determinations (77ROS/DRA, 82LEV/LIA, 84BLA/WAL, 83PRE/TZE). See also: 81SMI/ADA, 81WAL/BLA, 81KIM/KAT. | | | | | | | |
| H_2S_2^+ H_2S_2 | (9.3) | (218) | (913) | 4 | 16 | 82TN270 | 13465-07-1 |
| IP is onset of photoelectron band. | | | | | | | |
| H_2Sc^+ HScH | | (238) | (996) | | | | 13598-30-6 |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (84TOL/BEA). See also: 87SUN/ARI. 0 K value. | | | | | | | |
| H_2Se^+ H_2Se | 9.882±0.001 | 235 236 | 983 987 | 7 8 | 30 34 | 82TN270 | 7783-07-5 |
| H_2Si^+ SiH_2 | 8.92±0.07 | 276.1 | 1155.2 | 69±2 | 289±8 | 87BOO/ARM | 13825-90-6 |
| IP and $\Delta_f H(\text{Ion})$ from 87BOO/ARM, in agreement with unpublished data of R.R. Corderman and J.L. Beauchamp. See also: 83DYK/JON2, 84CHA/HIL. | | | | | | | |
| H_2Te^+ H_2Te | 9.138±0.005 | 235 | 982 | 24 | 100 | 82TN270 | 7783-09-7 |
| H_3^+ H_3 | | 264.5 265 | 1106.6 1107 | | | | 12184-91-7 |
| From proton affinity of H_2 (RN 1333-74-0) (84ADA/SMI). PA = 101.2 kcal/mol, 423.4 kJ/mol. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--|--------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| H_3Si^+ SiH ₃ I | (9.5) | (219) | (915) | -0.5±4 | -2±17 | 85JANAF | 13598-42-0 |
| | | (221) | (925) | 2±4 | 8±17 | | |
| | | IP is onset of photoelectron band. | | | | | |
| H_3N^+ NH ₃ | 10.16±0.01 | 223.2 | 934.0 | -11.0 | -45.9±0.4 | 85JANAF | 7664-41-7 |
| | | 224.9 | 941.0 | -9.3 | -38.9±0.4 | | |
| | | See also: 81KIM/KAT, 73RAB/KAR. | | | | | |
| H_3NO^+ NH ₂ OH | 10.00 | (221) | (923) | -10 | -42 | 69BEN/CRU | 7803-49-8 |
| | | IP from 83KOP/MOL. See also: 81KIM/KAT, 82KUT/GOO. | | | | | |
| H_3O^+ H ₃ O | | 141 | 591 | | | | |
| | | 143 | 597 | | | | |
| | | $\Delta_f H(\text{Ion})$ at 298 K from proton affinity of H ₂ O; $\Delta_f H(\text{Ion})$ at 0 K from appearance potential from (H ₂ O) ₂ (77NG/TRE). PA = 166.5 kcal/mol, 697. kJ/mol. | | | | | |
| H_3O_2^+ H ₂ OOH | | 171 | 716 | | | | |
| | | From proton affinity of H ₂ O ₂ (RN 7722-84-1). PA = 162. kcal/mol, 678. kJ/mol. | | | | | |
| $\text{H}_3\text{O}_4\text{S}^+$ (HO) ₃ SO | | (21) | (88) | | | | |
| | | From proton affinity of H ₂ SO ₄ (RN 7664-93-9). PA = (169) kcal/mol, (707) kJ/mol. | | | | | |
| H_3P^+ PH ₃ | 9.869±0.002 | 229 | 957 | 1.3±0.4 | 5.4±1.7 | 61GUN/GRE | 7803-51-2 |
| | | 231 | 966 | 3.1 | 13.3 | | |
| | | IP from 83MAR/REI, 86BER/CUR. See also: 82COW/KEM. | | | | | |
| H_3S^+ H ₃ S | | 190 | 797 | | | | |
| | | From proton affinity of H ₂ S (RN 7783-06-4). See also: 83PRE/TZE2, 84BLA/WAL, 83ERM/AKO. PA = 170.2 kcal/mol, 712. kJ/mol. | | | | | |
| H_3Sb^+ SbH ₃ | 9.54±0.03 | 255 | 1066 | 35 | 145 | 82TN270 | 7803-52-3 |
| | | 257 | 1074 | 37 | 153 | | |
| H_3Se^+ H ₃ Se | | 202 | 843 | | | | |
| | | From proton affinity of H ₂ Se (RN 7783-07-5). PA = 171.3 kcal/mol, 717. kJ/mol. | | | | | |
| H_3Si^+ SiH ₃ | 8.14±0.01 | 237.1 | 992 | 48.5±1.5 | 202.9±6.3 | 87BOO/ARM | 13765-44-1 |
| | | IP from 83DYK/JON2. See also: 84CHA/HIL, 85DIN/CAS, 87BOO/ARM. | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--|---------|------------------------------|-----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| H_3Te^+ TeH_3 | | 214 | 894 | | | | |
| | | $\Delta_f H(\text{Ion})$ from proton affinity of H_2Te (RN7783-09-7)(86KAR/JAS). PA = 176 kcal/mol, 736 kJ/mol. | | | | | |
| H_4N^+ NH_4 | (4.73±0.06) | (151) | (630) | | | | |
| | | $\Delta_f H(\text{Ion})$ from proton affinity of NH_3 (RN 7664-41-7). IP from neutralized ion-beam spectroscopy data (82GEL/CLE). PA = (204.0) kcal/mol, (854.) kJ/mol. | | | | | |
| H_4N_2^+ N_2H_4 | 8.1±0.15 | (210) | (876) | 22.8±0.2 | 95.3±0.8 | 85JANAF | 302-01-2 |
| | | (213) | (891) | 26.1±0.2 | 109.4±0.8 | | |
| | | From charge transfer equilibrium constant determinations(84MAU/NEL). See also: 81KIM/KAT. | | | | | |
| H_4N_4^+ (E)- $\text{H}_2\text{NN}=\text{NNH}_2$ | (≤8.99) | (≤260) | (≤1089) | 53 | 222 | 82TN270 | 54410-57-0 |
| H_4P^+ PH_4 | | 178 | 746 | | | | |
| | | From proton affinity of PH_3 (RN 7803-51-2). PA = 188.6 kcal/mol, 789. kJ/mol. | | | | | |
| H_4P_2^+ P_2H_4 | 8.8±0.1 | (219) | (918) | 16 | 69 | *EST | 13445-50-6 |
| | | IP is onset of photoelectron band. | | | | | |
| H_4Si^+ SiH_4 | 11.65 | 277 | 1159 | 8 | 35 | 81BEL/PER | 7803-62-5 |
| | | 280 | 1170 | 11 | 46 | | |
| H_4Sn^+ SnH_4 | (10.75) | (287) | (1200) | 39 | 163 | 82TN270 | 2406-52-2 |
| | | (290) | (1212) | 42 | 175 | | |
| | | The SnH_4^+ ion has not been observed. | | | | | |
| H_5N_2^+ NH_3NH_2 | | 184 | 770 | | | | |
| | | From proton affinity of hydrazine (RN 302-01-2). PA = 204.7 kcal/mol, 856. kJ/mol. | | | | | |
| H_5Si^+ SiH_5 | | (219) | (917) | | | | |
| | | From proton affinity of SiH_4 (RN 7803-62-5). PA = (155) kcal/mol, (648) kJ/mol. | | | | | |
| H_6Si_2^+ Si_2H_6 | (9.7) | (243) | (1016) | 19 | 80 | 81BEL/PER | 1590-87-0 |
| | | (247) | (1032) | 23 | 96 | | |
| | | IP is onset of photoelectron band. See also: 84CHA/HIL. | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|----------------------------|--------------------------|---------------|------------------------------|--------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| HgSi_3^+ | Si_3Hg | (9.2) | (241) | (1009) | 29 | 121 | 81BEL/PER | 7783-26-8 |
| IP is onset of photoelectron band. | | | | | | | | |
| $\text{H}_{10}\text{Si}_4^+$ | Si_4H_{10} | (8.9) | (244) | (1021) | 39 | 162 | 81WAL | 7783-29-1 |
| IP is onset of photoelectron band. | | | | | | | | |
| He^+ | He | 24.587 | <u>567.0</u> | <u>2372</u> | 0 | 0 | *DEF | 7440-59-7 |
| <u>567.0</u> <u>2372</u> 0 0 | | | | | | | | |
| HeNe^+ | HeNe | 20.87 | 481.3 | 2013.9 | -0.028 | -0.12 | 79HUB/HER | 12162-16-2 |
| $\Delta_f H(\text{Ion})$ from 78DAB/HER. 0 K values. | | | | | | | | |
| He_2^+ | He_2 | 22.223 | 512.4 | 2144.1 | -0.02 | -0.09 | 79HUB/HER | 12184-98-4 |
| IP from 79HUB/HER. 0 K values. | | | | | | | | |
| Hf^+ | Hf | 6.78 | 304 | 1273 | 148 | 619 | 82TN270 | 7440-58-6 |
| 304 1273 148 619 | | | | | | | | |
| IP from 76MEG/MOO. | | | | | | | | |
| HfO^+ | HfO | (7.55±0.1) | (190) | (795) | 16±3 | 67±13 | 83PED/MAR | 12029-22-0 |
| (192) (804) 18 76 | | | | | | | | |
| Hg^+ | Hg | 10.437 | <u>255.3</u> | <u>1068.3</u> | 14.7 | 61.3 | 82TN270 | 7439-97-6 |
| <u>256.1</u> <u>1071.5</u> 15.4 64.5 | | | | | | | | |
| See also: 84LIN/LIA. | | | | | | | | |
| HgI_2^+ | HgI_2 | 9.5088±0.0022 | 215.2 | 900.3 | -4.1 | -17.2 | 82TN270 | 7774-29-0 |
| 216.7 906.6 -2.6 -10.9 | | | | | | | | |
| Cited ionization potential (83LIN/TZE) refers to formation of $\text{HgI}_2^+(^2\Pi_{3/2})$. Ionization potential for formation of $\text{HgI}_2^+(^2\Pi_{1/2})$ is 10.1953±0.0025 eV. See also: 81LEE/POT. | | | | | | | | |
| Hg_2^+ | Hg_2 | 9.103±0.010 | 237 | 992 | 27 | 114 | 82HIL | 12596-25-7 |
| 239 1000 29 122 | | | | | | | | |
| IP from 84LIN/LIA. | | | | | | | | |
| Ho^+ | Ho | 6.0216±0.0006 | 211 | 882 | 72 | 301 | 82TN270 | 7440-60-0 |
| 211.2 883.6 72.3 302.6 | | | | | | | | |

Table 1. Positive Ion Table - Continued

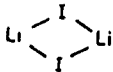
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| HoO⁺ HoO | (6.17±0.1) | (128) (126) | (534) (528) | -15±6 -16 | -61±25 -67 | 83PED/MAR | 12281-10-6 |
| See also: 80MUR/HIL. | | | | | | | |
| I⁺ I | 10.451 | 266.5 266.6 | 1115.2 1115.6 | 25.5 25.6 | 106.8 107.2 | 82BAU/COX | 14362-44-8 |
| See also: 81HOA/CAB, 85GRA/ROS. | | | | | | | |
| IK⁺ KI | (7.21±0.3) | (136) (137) | (570) (573) | -30.0±0.5 -29.2±0.5 | -125.5±2.1 -122.1±2.1 | 85JANAF | 7681-11-0 |
| See also: 82EMO/HOR. | | | | | | | |
| ILi⁺ LiI | (7.5) | (151) (152) | (633) (635) | -21.7±2.0 -21.3±2.0 | -91.0±8.4 -89.1±8.4 | 85JANAF | 10377-51-2 |
| IP is onset of photoelectron band. | | | | | | | |
| INa⁺ NaI | 7.64±0.02 | 157 158 | 659 662 | -18.6 -17.9 | -77.8 -74.9 | 82TN270 | 7681-82-5 |
| See also: 82EMO/HOR, 83HIL/GIN, 84HIL, 82LEL/BAL. | | | | | | | |
| IRb⁺ RbI | (7.12±0.1) | (132) (133) | (554) (558) | -32 -30.9 | -133 -129.3 | 79HUB/HER | 7790-29-6 |
| See also: 82EMO/HOR. | | | | | | | |
| ITI⁺ TII | 8.47±0.02 | 197 197 | 824 823 | 2 1 | 7 6 | 82TN270 | 7790-30-9 |
| See: 83BAN/BRI. | | | | | | | |
| I₂⁺ I ₂ | 9.3995±0.0012 | 231.7 232.4 | 969.3 972.4 | 14.9 15.7 | 62.4 65.5 | 82BAU/COX | 7553-56-2 |
| See also: 81HOA/CAB, 85GRA/ROS, 81KIM/KAT. | | | | | | | |
| I₂Li₂⁺  | (≤9.23±0.06) | (≤126) (≤127) | (≤529) (≤532) | -87±4 -85±4 | -362±17 -356±17 | 85JANAF | 37279-36-0 |
| I₂Mg⁺ MgI ₂ | (9.57±0.03) | (180) | (751) | -41 | -172 | 82TN270 | 10377-58-9 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|--|--------------------------|----------------|------------------------------|------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{I}_2\text{O}_2\text{W}^+$ WO_2I_2 | (10.4±0.4) | (137.5) | (575.4) | -102.3 | -428.0 | 76DEL/HAL | 14447-89-3 |
| I_2Pb^+ PbI_2 | 8.86±0.03 Onset of photoelectron band: 8.6 eV. See: 84NOV/POT2. 0 K values. | (205) | (856) | 0.2±1 | 1±4 | 85JANAF | 10101-63-0 |
| I_2Sn^+ SnI_2 | 8.83±0.1 IP from 83HIL/GIN, 84NOV/POT2. | 204 | 854 | 0.5 | 2 | 82TPIS | 10294-70-9 |
| I_2Sr^+ SrI_2 | (8.3) IP is onset of photoelectron band (79LEE/POT2). See also: 82EMO/KIE. | (126) (126) | (526) (529) | -65.7±1.5 -65.0±1.5 | -275±6 -272±6 | 85JANAF | 10476-86-5 |
| I_3La^+ LaI_3 | 8.8 IP is onset of photoelectron band (83RUS/GOO). | 119 | 498 | -84 | -351 | 82TN270 | |
| I_3Nd^+ NdI_3 | 8.7 IP is onset of photoelectron band (83RUS/GOO). | 124 | 519 | -76 | -320 | 82TN270 | 13813-24-6 |
| I_4Ti^+ TiI_4 | (9.1) IP is onset of photoelectron band. | (143) (145) | (600) (606) | -66±2 -65±2 | -278±8 -272±8 | 85JANAF | 7720-83-4 |
| I_4Zr^+ ZrI_4 | (9.3) IP is onset of photoelectron band. | (128) (130) | (534) (544) | -95±2 -85±2 | -363±8 -357±8 | 85JANAF | 13986-26-0 |
| In^+ In | 5.786 See also: 82GOM/CHA, 85KAP/LEL. | 191.7 191.6 | 801.9 802.6 | 58.2 58.2 | 243.7 243.3 | 82TN270 | 7440-74-6 |
| InS^+ InS | (7.0±0.5) 0 K values. | (218) | (911) | 57 | 236 | 79HUB/HER | 12030-14-7 |
| InSe^+ InSe | (7.1±0.5) 0 K values. | (218) | (913) | 55 | 228 | 79HUB/HER | 1312-42-1 |
| InTe^+ InTe | (7.6±0.5) 0 K values. | (230) | (962) | 55 | 229 | 79HUB/HER | 12030-19-2 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|--|--------------------------|----------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| In_2^+ In ₂ | (5.8±0.3) 0 K values. | (227) | (949) | 93 | 389 | 79HUB/HER | 61178-97-0 |
| In_2O^+ In ₂ O | 8.0±0.2 IP from 85KAP/LEL, 77ROS/DRA. See also: 82GOM/CHA. | (174) (175) | (728) (734) | -10 -9±5 | -43 -38±20 | 82TPIS | 12030-22-7 |
| Ir^+ Ir | 9.02 IP from 79RAU/ACK. | 367 366.8 | 1535 1534.6 | 159 158.8 | 665 664.3 | 82TN270 | 7439-88-5 |
| IrO^+ IrO | (10.1) 0 K values. | (367) | (1535) | (134) | (561) | 79HUB/HER | 12030-48-7 |
| IrO_3^+ IrO ₃ | (11.9) | (276) | (1156) | 2 | 8 | 82TN270 | 12030-50-1 |
| K^+ K | 4.341 | 121.4 121.6 | 507.8 508.7 | 21.3±0.1 21.5±0.1 | 89.0±0.4 89.9±0.4 | 85JANAF | 7440-09-7 |
| KLi^+ LiK | 4.57±0.04 IP from 85KAP/SCH. | 123 | 514 | 17 | 73 | 86IGE/WED | 12030-83-0 |
| KNa^+ NaK | 4.41636±0.00017 IP from 81LEU/HOF, 85KAP/RAD. 0 K values. | 134.2 | 561.5 | 32.4 | 135.4 | 86ZGE/WED | 12056-29-0 |
| KO^+ KO | 7.09±0.1 IP from 82LEV/LIA, 84BUT/KUD. | 178 179 | 745 747 | 15±5 15±5 | 61±21 -63±21 | 83PED/MAR | 12401-70-6 |
| KRb^+ KRb | (3.9±0.1) IP from 85KAP/SCH. 0 K values. | (120) | (500) | (30) | (124) | 79HUB/HER | 12333-39-0 |
| K_2^+ K ₂ | 4.0637±0.0002 IP from 85KAP/RAD, 85BRO/CHE, in good agreement with values from 81LEU/HOF, 78HER/SCH, and 84DAO/PET. | 124.0 124.8 | 518.9 522.3 | 30.3 31.1 | 126.9 130.3 | 79HUB/HER | 25681-80-5 |
| K_2O^+ K ₂ O | 4.96±0.2 IP from 84BUT/KUD. See also: 82FAR/SRI, 84DAO/PET. | (80) | (336) | -34±4 | -142±15 | 79BYK/ELI | 12136-45-7 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|------------------------|----------------------------|--------------------------|--------------|------------------------------|---------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| K_2O_2^+ | K_2O_2 | (5) | (44) | (184) | -71±4 | -298±15 | 79BYK/ELI | 17014-71-0 |
| Kr^+ | Kr | 13.9997±0.00001 | 322.8 | 1350.8 | 0 | 0 | *DEF | 7439-90-9 |
| | | | 322.8 | 1350.8 | 0 | 0 | | |
| See also: 81KIM/KAT. | | | | | | | | |
| KrXe^+ | KrXe | 11.760±0.014 | 270.8 | 1132.9 | -0.42 | -1.77 | 79HUB/HER | 12521-42-5 |
| See: 82DEH/PRA, 85PRA/DEH2. 0 K values. | | | | | | | | |
| Kr_2^+ | Kr_2 | 12.866±0.003 | 296.3 | 1240.9 | -0.36 | -1.51 | 79HUB/HER | 12596-40-6 |
| IP from 82PRA/DEH. 0 K values. | | | | | | | | |
| La^+ | La | 5.577 | 232 | 969 | 103 | 431 | 82TN270 | 7439-91-0 |
| | | | 231.7 | 969.4 | 103.1 | 431.3 | | |
| LaO^+ | LaO | 4.90±0.1 | 84 | 352 | -29±2 | -121±10 | 83PED/MAR | 12031-20-8 |
| | | | 85 | 354 | -28 | -119 | | |
| LaPt^+ | LaPt | (5.4±0.8) | (243) | (1018) | 119±5 | 497±21 | 81NAP/GIN | 12142-67-5 |
| IP from 81NAP/GIN. | | | | | | | | |
| Li^+ | Li | 5.392 | <u>162.4</u> | <u>679.6</u> | 38.1 | 159.4 | 82TN270 | 7439-93-2 |
| | | | <u>162.0</u> | <u>678.0</u> | 37.7 | 157.8 | | |
| See also: 81NAK/ASA. | | | | | | | | |
| LiNa^+ | LiNa | 5.05±0.04 | 137 | 572 | 20 | 85 | 86IGE/WED | 12333-49-2 |
| IP from 85KAP/SCH. | | | | | | | | |
| LiO^+ | LiO | (8.45±0.20) | (214) | (894) | 19±0.5 | 79±2 | 83PED/MAR | 12142-77-7 |
| | | | (214) | (894) | 19±0.5 | 79±2 | | |
| See also: 81NAK/ASA, 79WU/KUD, 84BUT/KUD. | | | | | | | | |
| LiOH_2^+ | LiOH_2 | | 69 | 289 | | | | |
| From proton affinity of LiOH (RN 1310-58-3). PA = 241 kcal/mol, 1007 kJ/mol. | | | | | | | | |
| LiRb^+ | LiRb | 4.3±0.1 | 116 | 485 | 17 | 70 | 86IGE/WED | 12031-70-8 |
| IP from 85KAP/SCH. | | | | | | | | |

Table 1. Positive Ion Table - Continued

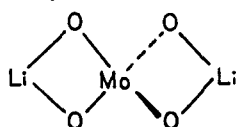
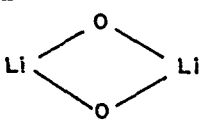
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|----------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Li_2^+ Li ₂ | 5.1127±0.0003 | 169.5 169.4 | 709.2 708.8 | 51.6±0.7 51.5±0.7 | 215.9±3.0 215.5±3.0 | 8SJANAF | 14452-59-6 |
| IP from 83MCG/SCH. See also: 82EIS/DEM. | | | | | | | |
| $\text{Li}_2\text{MoO}_4^+$  | (9.7±0.5) | (-22) | (-94) | -246 | -1030 | 81LIN/BES | |
| Li_2O^+ Li ₂ O | 6.19±0.20 | (103) (103) | (430) (431) | -40 -40 | -167 -166 | 81LIN/BES | 12057-24-8 |
| See also: 82IKE/TAM, 81NAK/ASA, 79WU/KUD. | | | | | | | |
| Li_2O_2^+  | (7.88±0.2) | (84) | (350) | -98±12 | -410±50 | 79WU/KUD | 12031-80-0 |
| IP from 79WU/KUD. 0 K values. | | | | | | | |
| $\text{Li}_2\text{O}_3\text{Si}^+$ Li ₂ SiO ₃ | 8.3±0.2 | -99 | -415 | -291 | -1216 | 81NAK/ASA | |
| IP from 81NAK/ASA. | | | | | | | |
| $\text{Li}_2\text{O}_4\text{W}^+$ Li ₂ WO ₄ | (9.2±0.5) | (-29) | (-122) | -241 | -1010 | 81LIN/BES | |
| Li_3O^+ Li ₃ O | (4.54±0.2) | (50) | (210) | -54±10 | -228±42 | 79WU/KUD | 69235-02-5 |
| IP from 79WU/KUD. 0 K values. | | | | | | | |
| Lu^+ Lu | 5.4259±0.00001 | 227 227.4 | 951 951.3 | 102 102.2 | 428 427.8 | 82TN270 | 7439-94-3 |
| LuO^+ LuO | (6.79±0.1) | (153) (153) | (640) (642) | -3.5 -3 | -14.6 -13 | 82TN270 | 12032-02-9 |
| A value of 2±17 kJ/mol, 0.5±4 kcal/mol has also been recommended for the 298 K heat of formation of LuO (83PED/MAR). See also: 80MUR/HIL. | | | | | | | |
| Md^+ Md | 6.74±0.12 | | | | | | 7440-11-1 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------|--|--------------------------|--------------|------------------------------|----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Mg⁺ | | | | | | | | |
| Mg | | 7.646 | <u>211.6</u> | <u>885.4</u> | 35.3 | 147.7 | 82TN270 | 7439-95-4 |
| | | | <u>211.3</u> | <u>884.2</u> | 35.0 | 146.5 | | |
| | | See also: 81SAS/HAR. | | | | | | |
| MgO⁺ | | | | | | | | |
| MgO | | 9.7 | (236±8) | (997±33) | 13.4 | 56.1 | 79HUB/HER | 1309-48-4 |
| | | | (238±8) | (997±33) | 13.5 | 56.5 | | |
| | | $\Delta_f H(\text{Ion})$ from 81KAP/STA. See also: 81MUR. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | |
| Mg₂H⁺ | | | | | | | | |
| MgHMg | | | 78 | 327 | | | | |
| | | From proton affinity of Mg ₂ (RN 29904-79-8). PA = (219) kcal/mol, (916) kJ/mol. | | | | | | |
| Mn⁺ | | | | | | | | |
| Mn | | 7.435 | 239 | 998 | 67 | 281 | 82TN270 | 7439-96-5 |
| | | | 238 | 996 | 67 | 279 | | |
| MnO⁺ | | | | | | | | |
| MnO | | 8.65±0.2 | (240) | (1005) | 41 | 170 | 67CHE/BAR | 1344-43-0 |
| | | IP from 82ARM/HAL. See also: 81ARM/HAL, 81KAP/STA. 0 K values. | | | | | | |
| Mn₂⁺ | | | | | | | | |
| Mn ₂ | | 6.9±0.4 | (280) | (1172) | (121±7) | (506±29) | 83ERV/LOH | 12596-53-1 |
| | | IP from 83ERV/LOH. 0 K values. | | | | | | |
| Mo⁺ | | | | | | | | |
| Mo | | 7.099 | 321.0 | 1343.1 | 157.3 | 658.1 | 82TN270 | 7439-98-7 |
| | | | 320.6 | 1341.5 | 156.9 | 656.6 | | |
| MoNa₂O₄⁺ | | | | | | | | |
| Na ₂ MoO ₄ | | (7.2) | (-87) | (-364) | -253 | -1059 | 82TN270 | |
| MoO⁺ | | | | | | | | |
| MoO | | (8.0±0.6) | (267) | (1119) | 83±5 | 347±21 | 83PED/MAR | 12058-07-0 |
| | | | (267) | (1119) | 83±5 | 347±21 | | |
| MoO₂⁺ | | | | | | | | |
| MoO ₂ | | (9.2) | (213) | (890) | -2±3 | -8±13 | 85JANAF | |
| | | | (213) | (892) | -2±3 | -6±13 | | |
| MoO₃⁺ | | | | | | | | |
| MoO ₃ | | (11.8±0.5) | (186) | (777) | -87 | -362 | 81WOO | 1313-27-5 |
| | | | (187) | (781) | -86 | -358 | | |

Table 1. Positive Ion Table - Continued

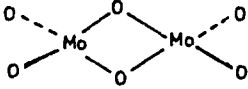
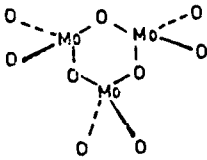
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|--------------------------------|--------------------------------|------------------------------|--------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Mo_2O_6^+  | 12.1±0.6 IP from 77ROS/DRA, 85KAP/LEL. | (201) | (841) | -78 | -326 | 82TN270 | 12412-19-0 |
| Mo_3O_9^+  | (12.0±1.0) See also: 85KAP/LEL. | (-174) | (-729) | -451 | -1887 | 82TN270 | 12163-83-6 |
| N^+ N | 14.534 | 448.2 447.7 | 1875.0 1873.1 | 113.0 112.5 | 472.7 470.8 | 82TN270 | 17778-88-0 |
| NO^+ NO | 9.26436±0.00006 | <u>235.33</u> <u>235.33</u> | <u>984.61</u> <u>984.65</u> | 21.82 21.69 | 91.28 90.78 | 82BAU/COX | 10102-43-9 |
| See: 83SEA/CHU, 84MUL/SAN, 83EBA/ANE for confirming high precision measurements. See also: 81KIM/STB, 82FAN/GIA, 81KIM/KAT. | | | | | | | |
| NO_2^+ NO_2 | 9.75±0.01 | 233 233 | 974 977 | 7.9 8.6 | 33.2 36.0 | 82BAU/COX | 10102-44-0 |
| Ionization involves a bent-linear transition with a broad Franck-Condon envelope and weak onset. Selected IP consistent with occurrence of reaction: $(\text{NO}_2^+ + \text{C}_6\text{H}_5\text{CF}_3 \rightarrow \text{C}_6\text{H}_5\text{CF}_3^+ + \text{NO}_2)$ (78AUS/LIA). See also: 81KIM/KAT, 82KAT/SHI. | | | | | | | |
| NP^+ NP | 11.85 | 298 299 | 1248 1249 | 25±1 25±1 | 105±5 106±5 | 85JANAF | 17739-47-8 |
| NS^+ NS | 8.87±0.01 | 268 268 | 1119 1119 | 63±25 63±25 | 264±105 263±105 | 85JANAF | 51801-08-2 |
| IP from 79HUB/HER. | | | | | | | |
| NTi^+ TiN | (6) 0 K values. | (250) | (1045) | 112 | 466 | 79HUB/HER | 25583-20-4 |
| NZr^+ ZrN | (7.9±0.4) | (352.7) (352.9) | (1475.6) (1476.5) | 170.5 170.7 | 713.4 714.3 | 85JANAF | 25658-42-8 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|------------------------------|------------------------------|-------------------------------|-------------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| N_2^+ N_2 | 15.5808 | 359.3 <i>359.3</i> | 1503.3 <i>1503.3</i> | 0 <i>0</i> | 0 <i>0</i> | *DEF | 7727-37-9 |
| IP from 79HUB/HER. See also: 84STE/MAR, 81ARM/TAR, 81KIM/KAT. | | | | | | | |
| N_2O^+ N_2O | 12.886 | 316.8 <i>317.6</i> | 1325.4 <i>1328.8</i> | 19.6 <i>20.4</i> | 82.1 <i>85.5</i> | 82BAU/COX | 10024-97-2 |
| See also: 81KIM/KAT. | | | | | | | |
| N_2O_4^+ N_2O_4 | 10.8±0.2 | (251) <i>(254)</i> | (1051) <i>(1061)</i> | 2 <i>5</i> | 9 <i>19</i> | 82BAU/COX | 10544-72-6 |
| See also: 82CHO/FRO. | | | | | | | |
| N_2O_5^+ N_2O_5 | (11.9) | (277) <i>(280)</i> | (1159) <i>(1173)</i> | 3 <i>6</i> | 11 <i>25</i> | 82BAU/COX | 10102-03-1 |
| IP is onset of photoelectron band. | | | | | | | |
| Na^+ Na | 5.139 | <u>144.1</u> <u>144.2</u> | <u>603.1</u> <u>603.4</u> | 25.6±0.2 <i>25.7±0.2</i> | 107.3±0.7 <i>107.6±0.7</i> | 85JANAF | 7440-23-5 |
| See also: 84PET/DAO. | | | | | | | |
| NaO^+ NaO | (7.41) | (190.9) <i>(191.2)</i> | (798.7) <i>(800.0)</i> | 20.0±10.0 <i>20.3±10.0</i> | 83.7±41.8 <i>85.0±41.8</i> | 85JANAF | 12401-86-4 |
| IP from 84BUT/KUD. | | | | | | | |
| NaRb^+ NaRb | 4.32±0.04 | 115 | 481 | 15 | 64 | 86IGE/WED | 12333-61-8 |
| IP from 85KAP/SCH. | | | | | | | |
| Na_2^+ Na_2 | 4.88898±0.00016 | <u>146.7</u> <u>147.3</u> | <u>613.8</u> <u>616.3</u> | 34.0±0.3 <i>34.6±0.3</i> | 142.1±1.2 <i>144.6±1.2</i> | 85JANAF | 25681-79-2 |
| IP from 81LEU/HOF, in agreement with values from 82MAR/CHE, 85KAP/RAD, 84PET/DAO, 78HER/SCH. $\Delta_f H(\text{Ion})$ in agreement with that derived from data of 83WAG/ISE. | | | | | | | |
| Na_2Cl^+ Na_2Cl | 4.15±0.22 | (59) | (245) | -37 | -155 | 83PET/DAO | |
| IP from 83PET/DAO. 0 K values. | | | | | | | |
| Na_2O^+ Na_2O | (5.06±0.4) | (110) <i>(111)</i> | (461) <i>(465)</i> | -6 <i>-5</i> | -27 <i>-23</i> | 83PET/DAO | 1313-59-3 |
| IP from 83PET/DAO. See also: 84BUT/KUD. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|---|----------------|------------------------------|---------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Nb ⁺ Nb | 6.88 | 334 | 1397 | 175±2 | 733±8 | 85JANAF | 7440-03-1 |
| | | 333 | 1394 | 174.5±2 | 730±8 | | |
| NbO ⁺ NbO | (6.1) | (186±53) | (778.5±222) | 48±5 | 200±21 | 85JANAF | 12034-57-0 |
| | | (187±53) | (780±222) | 48±5 | 199±21 | | |
| $\Delta_f H(\text{Ion})$ from 81KAP/STA. IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| Nd ⁺ Nd | 5.5250±0.0006 | 205.80 | 861.08 | 78 | 328 | 82TN270 | 7440-00-8 |
| | | 206.04 | 862.08 | 78.5 | 328.6 | | |
| NdO ⁺ NdO | (4.97±0.1) | (84) | (354) | -30±3 | -126±12 | 83PED/MAR | 12035-20-0 |
| | | (84) | (354) | -30±3 | -126±12 | | |
| Ne ⁺ Ne | 21.56471±0.00001 | <u>497.29</u> | <u>2080.66</u> | 0 | 0 | *DEF | 7440-01-9 |
| | | <u>497.29</u> | <u>2080.66</u> | 0 | 0 | | |
| NeKr ⁺ NeKr | (13.950±0.003) | (321.5) | (1345.3) | -0.15 | -0.62 | 79HUB/HER | |
| | | IP from 82PRA/DEH2. 0 K values. | | | | | |
| NeXe ⁺ NeXe | (12.094±0.004) | (278.7) | (1166.3) | -0.15 | -0.63 | 79HUB/HER | 58984-40-0 |
| | | IP from 82PRA/DEH2. See also: 85PRA/DEH2. 0 K values. | | | | | |
| Ne ₂ ⁺ Ne ₂ | 20.33±0.08 | 469 | 1961 | -0.047 | -0.195 | 79HUB/HER | 12185-05-6 |
| | | IP from 84TRE/POL. See also: 79HUB/HER. 0 K values. | | | | | |
| Ni ⁺ Ni | 7.635 | 278.9 | 1166.8 | 102.8 | 430.1 | 82TN270 | 7440-02-0 |
| | | 278.4 | 1164.8 | 102.3 | 428.1 | | |
| See also: 82DYK/GRA. | | | | | | | |
| NiO ⁺ NiO | 9.5±0.2 | (290) | (1214) | 71±4 | 297±17 | 83PED/MAR | 1313-99-1 |
| | | IP from 81ARM/HAL, 82ARM/HAL, 77ROS/DRA. See also: 81KAP/STA. | | | | | |
| No ⁺ No | 6.84±0.12 | | | | | | 10028-14-5 |
| | | | | | | | |
| Np ⁺ Np | 6.2657±0.0005 | 256 | 1070 | 111 | 465 | 85KLE/WAR | 7439-99-8 |
| | | | | | | | |
| NpO ⁺ NpO | (5.7±0.1) | (130) | (546) | -1±10 | -4±42 | 83PED/MAR | 12202-03-8 |
| | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|--------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| O ⁺ O | 13.618 | 373.6 | 1563.1 | 59.6 | 249.2±0.1 | 85JANAF | 17778-80-2 |
| | | 373.0 | 1560.7 | 59.0 | 246.8±0.1 | | |
| OP ⁺ PO | 8.39±0.01 | 186 | 777 | -8±3 | -33±13 | 83PED/MAR | 14452-66-5 |
| | | 186 | 778 | -8 | -32 | | |
| IP from 82DYK/MOR. See also: 81BAL/GIG. | | | | | | | |
| OPb ⁺ PbO | 9.08±0.10 | 224 | 939 | 15±3 | 63±13 | 79HUB/HER | 1317-36-8 |
| | | 227 | 949 | 17 | 73 | | |
| See also: 83SEM/RYSK, 84NIK/OVC. | | | | | | | |
| OPd ⁺ PdO | (9.1) 0 K values. | (293) | (1224) | 83 | 346 | 79HUB/HER | 1314-08-5 |
| | | | | | | | |
| OPr ⁺ PrO | (4.90±0.1) | (79) | (331) | -34±4 | -142±17 | 82TN270 | 12035-81-3 |
| | | (75) | (314) | -38 | -159 | | |
| OPt ⁺ PtO | (10.1±0.3) | (334) | (1397) | 101 | 423 | 79HUB/HER | 12035-82-4 |
| | | (339) | (1417) | 106 | 443 | | |
| ORb ⁺ RbO | 6.69 | 168.6 | 705.3 | 14.3 | 59.8 | 82TPIS | 12509-27-2 |
| | | 168.9 | 706.6 | 14.6±5 | 61.1±20 | | |
| IP from 84BUT/KUD. | | | | | | | |
| ORb ₂ ⁺ Rb ₂ O | 4.63 | 25.8 | 107.8 | -81.0±2.0 | -338.9±8.4 | 82TPIS | 18088-11-4 |
| | | 26.7 | 111.6 | -80.1 | -335.1 | | |
| IP from 84BUT/KUD. | | | | | | | |
| ORh ⁺ RhO | (9.3) | (309) | (1294) | 95±10 | 397±42 | 83PED/MAR | 12137-18-7 |
| | | | | | | | |
| ORu ⁺ RuO | (8.7) | (290) | (1211) | 89±10 | 372±42 | 83PED/MAR | 12143-05-4 |
| | | | | | | | |
| OS ⁺ SO | 10.32±0.02 | 239.2 | 1000.7 | 1.2±0.3 | 5.0±1.3 | 85JANAF | 13827-32-2 |
| | | 239.2 | 1000.7 | 1.2±0.3 | 5.0±1.3 | | |
| OS ₂ ⁺ S ₂ O | 10.54±0.04 | 231 | 965 | -12±0.2 | -52±1 | 86NIM/ELL | 20901-21-7 |
| | | 230 | 962 | -13±0.2 | -55±1 | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|----------------------------------|---------|------------------------------|------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| OSi ⁺ SiO | 11.43 | 239.6 | 1002.4 | -24.0±2 | -100.4±8.4 | 85JANAF | 10097-28-6 |
| | | 239.3 | 1001.2 | -24.3±2 | -101.6±8.4 | | |
| IP from 82LEV/LIA, 79HUB/HER. See also: 81NAK/ASA. | | | | | | | |
| OSm ⁺ SmO | 5.55±0.1 | 100 | 418 | -28±3 | -117±12 | 83PED/MAR | 12035-88-0 |
| | | 97 | 405 | -31 | -130 | | |
| OSn ⁺ SnO | 9.60±0.02 | 226 | 944 | 4.2 | 17.5 | 81LAU/BRI | 21651-19-4 |
| | | 226 | 945 | 4.6 | 19.2 | | |
| IP from 82DYK/MOR2. | | | | | | | |
| OSr ⁺ SrO | 7.0±0.15 | (158.2) | (662.8) | -3.2±4 | -13.4±16.7 | 85JANAF | 1314-11-0 |
| | | (158.7) | (663.9) | -2.7±4 | -11.5±16.7 | | |
| $\Delta_f H(\text{Ion})$ from onset of endothermic reaction (83MUR); IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. | | | | | | | |
| OTa ⁺ TaO | (7.92±0.1) | (228.6) | (956.7) | 46.0±15 | 192.5±62.8 | 85JANAF | 12035-90-4 |
| | | (228.9) | (957.9) | 46.3±15 | 193.7±62.8 | | |
| OTb ⁺ TbO | (5.62±0.1) | (113) | (471) | -17±3 | -71±12 | 83PED/MAR | 12035-91-5 |
| | | (111) | (463) | -19 | -79 | | |
| OTe ⁺ TeO | 8.72 | 218 | 910 | 16±5 | 69±21 | 83PED/MAR | 13451-17-7 |
| | | 218 | 912 | 17 | 71 | | |
| OTh ⁺ ThO | 6.1±0.1 | (133) | (557) | -7±2 | -31±10 | 83PED/MAR | 12035-93-7 |
| | | (134) | (559) | -7 | -29 | | |
| OTi ⁺ TiO | 6.56±0.03 | 164.3 | 687.3 | 13.0±2.0 | 54.4±8.4 | 85JANAF | 12137-20-1 |
| | | 164.2 | 686.8 | 12.9±2.0 | 53.9±8.4 | | |
| IP from 84DYK/GRA. See also: 80MUR/HIL, 82BAN/CHA, 85BAL/GIG, 81KAP/STA. | | | | | | | |
| OTm ⁺ TmO | (6.44±0.1) | (130) | (542) | -19 | -79 | 82TN270 | 12281-29-7 |
| | | See also: 80MUR/HIL. 0 K values. | | | | | |
| OU ⁺ UO | (5.65±0.2) | (136) | (570) | 6±2 | 25±10 | 83PED/MAR | 12035-97-1 |
| | | (136) | (568) | 6 | 23 | | |

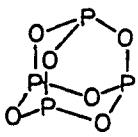
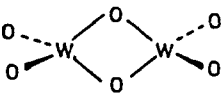
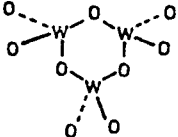
Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|----------------------------|--------------------------|----------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| OY⁺ VO | (7.5) | (203.4) (203.5) | (851.2) (851.4) | 30.5±5.0 30.5±5.0 | 127.6±20.9 127.8±20.9 | 85JANAF | 12035-98-2 |
| $\Delta_f H(\text{Ion})$ from onset energy of endothermic reaction (84ARI/ARM, 85ELK/ARM). IP is $\Delta_f H(\text{Ion}) - \Delta_f H(\text{Neutral})$. See also: 81KAP/STA. | | | | | | | |
| OW⁺ WO | (9.1±1) | (311.5) (311.6) | (1303.1) (1303.7) | 101.6±10.0 101.7±10.0 | 425.1±41.8 425.7±41.8 | 85JANAF | 12035-99-3 |
| OY⁺ YO | 5.85±0.15 | 124 124 | 518 519 | -11±2 -11 | -46±10 -45 | 83PED/MAR | 12036-00-9 |
| See also: 80MUR/HIL. | | | | | | | |
| OYb⁺ YbO | (6.55±0.1) | (147) | (615) | -4±2 | -17±8 | 83PED/MAR | 25578-79-4 |
| OZn⁺ ZnO | | (275±8) | (1151±33) | | | | |
| $\Delta_f H(\text{Ion})$ from 81KAP/STA. 0 K values. | | | | | | | |
| OZr⁺ ZrO | (6.1±0.3) | (154.7) (154.9) | (647.1) (648.2) | 14.0±12.0 14.2±12.0 | 58.6±50.2 59.6±50.2 | 85JANAF | 12036-01-0 |
| See also: 81KAP/STA. | | | | | | | |
| O₂⁺ O ₂ | 12.071±0.001 | 278.5 278.4 | 1165.3 1164.7 | 0 0 | 0 0 | *DEF | 7782-44-7 |
| See also: 81KIM/KAT. | | | | | | | |
| O₂P⁺ PO ₂ | (10.5±0.1) | (175) (176) | (733) (736) | -67 -66 | -280 -277 | 85JANAF | 12164-97-5 |
| O₂Pt⁺ PtO ₂ | (11.2±0.3) | (299) | (1253) | 41 | 172 | 82TN270 | 1314-15-4 |
| O₂Rh⁺ RhO ₂ | (10.0) | (275) | (1149) | 44 | 184 | 82TN270 | 12137-27-8 |
| O₂S⁺ SO ₂ | 12.32±0.02 | 213 214 | 892 894 | -70.9±0.1 -70.3±0.1 | -296.8±0.2 -294.3±0.2 | 85JANAF | 7446-09-5 |
| See also: 81SMI/STE, 84ORI/SRI, 81KIM/KAT. | | | | | | | |
| O₂Sn₂⁺ Sn ₂ O ₂ | (9.8±0.5) | (166) | (695) | -60 | -251 | 82TN270 | 12534-17-7 |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|------------------|------------------------------|--------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| O_2Th^+ ThO ₂ | (8.7±0.15) | (82) (82) | (341) (344) | -119 -118 | -498 -495 | 82TN270 | 1314-20-1 |
| O_2Ti^+ TiO ₂ | (9.54±0.1) | (149) (149) | (625) (623) | (-71) (-71) | (-299) (-297) | 85BAL/GIG | 13463-67-7 |
| See also: 82BAN/CHA, 85BAL/GIG. | | | | | | | |
| O_2U^+ UO ₂ | (5.4±0.1) | (13) (14) | (55) (57) | -111±1 -111 | -466±5 -464 | 80GRE | 1344-57-6 |
| O_2W^+ WO ₂ | (9.6±0.3) | (240) (240) | (1003) (1005) | 18±6 19±6 | 77±29 79±29 | 85JANAF | 12036-22-5 |
| See also: 81BAL/GIG. | | | | | | | |
| O_2Zr^+ ZrO ₂ | (9.5±0.3) | (151) (150) | (631) (629) | -68±11 -68±11 | -286±46 -284±46 | 85JANAF | 1314-23-4 |
| O_3^+ O ₃ | 12.43 | 321 321 | 1342 1344 | 34 35 | 143 145 | 82TN270 | 10028-15-6 |
| IP from 84KAT/SHI. | | | | | | | |
| O_3Ru^+ RuO ₃ | (11.2) | (240) | (1003) | -19 | -78 | 82TN270 | 12036-36-1 |
| O_3S^+ SO ₃ | 12.80±0.04 | 200 202 | 839 845 | -94.6±0.2 -93.2±0.2 | -395.8±0.7 -390.0±0.7 | 85JANAF | 7446-11-9 |
| See also: 81SMI/STE. | | | | | | | |
| O_3Sn_3^+ Sn ₃ O ₃ | (9.8±0.5) | (100) | (419) | -126 | -527 | 82TN270 | 12534-28-0 |
| O_3Ti_2^+ Ti ₂ O ₃ | (8.3±0.5) | (39) | (164) | (-152) | (-636) | 85BAL/GIG2 | 1344-54-3 |
| IP from 85BAL/GIG2. 0 K values. | | | | | | | |
| O_3U^+ UO ₃ | (10.5±0.5) | (51) | (213) | -191±5 | -800±20 | 80GRE | 1344-58-7 |
| O_3W^+ WO ₃ | (11.8±0.6) | (202) (203) | (846) (851) | -70 -69 | -293 -288 | 81WOO | 1314-35-8 |
| See also: 81BAL/GIG. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|----------------------------|--------------------------|--------------------|------------------------------|----------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| O_4Os^+ OsO_4 | 12.320 | 204 | 852 | -81 | -337 | 82TN270 | 20816-12-0 |
| O_4Ru^+ RuO_4 | 12.15±0.03 | 236 238 | 988 994 | -44 -43 | -184 -178 | 82TN270 | 20427-56-9 |
| O_4SnW^+ SnWO_4 | (10.8) 0 K values. | (41) | (172) | -208 | -870 | 82TN270 | 14553-36-7 |
| O_4Sn_4^+ Sn_4O_4 | (9.2±0.5) | (19) | (80) | -193 | -808 | 82TN270 | |
| O_4SrW^+ SrWO_4 | (9.4) | (-33) | (-139) | -250 | -1046 | 76DEL/HAL | |
| O_4Ti_2^+ Ti_2O_4 | (10.5±0.5) 0 K values. | (-19) | (-79) | (-261) | (-1092) | 85BAL/GIG3 | |
| $\text{O}_5\text{Sn}_2\text{W}^+$ Sn_2WO_5 | (8.4) 0 K values. | (-75) | (-315) | -269 | -1125 | 82TN270 | |
| O_6P_4^+  | (9.5) | (-293) (-288) | (-1227) (-1203) | -512±8 -507±6 | -2144±33 -2120±33 | 85JANAF | 10248-58-5 |
| O_6W_2^+  | (12.2±0.2) | (3) (4) | (11) (17) | -278 -277 | -1164 -1158 | 85JANAF | 12165-16-1 |
| O_7Re_2^+ Re_2O_7 | (12.7±0.2) | (30) | (124) | -263 | -1101 | 81WOO | 1314-68-7 |
| O_9W_3^+  | (12.0±0.2) | (-191) | (-800) | -468 | -1958 | 82TN270 | 12165-37-6 |

IP is onset of photoelectron band.

Table 1. Positive Ion Table - Continued

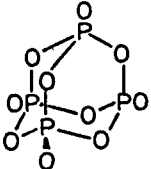
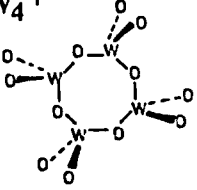
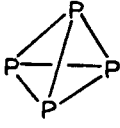
| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|-----------------------------|--------------------------|----------------------|------------------------------|------------------------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| $\text{O}_{10}\text{P}_4^+$  | (13.3±0.2) | (-371) (-363) | (-1551) (-1517) | -677±2 -669±2 | -2834±9 -2800±9 | 85JANAF | 16752-60-6 |
| $\text{O}_{12}\text{W}_4^+$  | (12.0±0.2) | (-372) | (-1557) | -649 | -2715 | 82TN270 | 12165-45-6 |
| Os^+ Os | 8.28 IP from 79RAU/ACK. | 380 | 1590 | 189 | 791 | 82TN270 | 7440-04-2 |
| P^+ P | 10.486 | 317 317 | 1328 1328 | 75.6±0.2 75.4±0.2 | 316.4±1.0 315.6±1.0 | 85JANAF | 7723-14-0 |
| PS^+ PS | (9.0) | (245) (245) | (1024) (1024) | 36±1 36±1 | 151±4 151±4 | 79HUB/HER | 12281-36-6 |
| PSe^+ PSe | (8.2) 0 K values. | (232) | (971) | 43 | 180 | 79HUB/HER | 12509-41-0 |
| P_2^+ P ₂ | 10.53 IP from 79HUB/HER. | (277.2) (277.6) | (1159.7) (1161.5) | 34.3±0.5 34.8±0.5 | 143.7±2.1 145.5±2.1 | 85JANAF | 12185-09-0 |
| P_3^+ P ₃ | (7.85±0.2) | (241) | (1006) | 59.4±4 | 249±16 | 74BEN/MAR | 55030-78-9 |
| P_4^+  | 9.08±0.05 | 223 225 | 935 942 | 14±0.5 16±0.5 | 59±2 66±2 | 85JANAF | 12185-10-3 |
| Pa^+ Pa | 5.89±0.12 | 270 | 1131 | 135 | 563 | 85KLE/WAR | 7440-13-3 |

Table 1. Positive Ion Table - Continued

| ION | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------------|---|--------------------------|--------|------------------------------|--------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Pb⁺ | | | | | | | |
| Pb | 7.416 | 217.6 | 910.5 | 46.6 | 195.0 | 82TN270 | 7439-92-1 |
| | | 217.8 | 911.1 | 46.8 | 195.6 | | |
| | See also: 83SEM/RYSK, 84NIK/OVC. | | | | | | |
| PbS⁺ | | | | | | | |
| PbS | (8.5±0.5) | (228) | (954) | 32 | 134 | 79HUB/HER | 1314-87-0 |
| | | (229) | (956) | 32 | 136 | | |
| PbSe⁺ | | | | | | | |
| PbSe | (8.4±0.5) 0 K values. | (224) | (935) | 30 | 125 | 79HUB/HER | 12069-00-0 |
| PbTe⁺ | | | | | | | |
| PbTe | (≤8.04) 0 K values. | (≤220) | (≤922) | 35 | 146 | 79HUB/HER | 1314-91-6 |
| Pb₂⁺ | | | | | | | |
| Pb ₂ | (6.1±0.3) | (214) | (897) | (74) | (308) | 79HUB/HER | 12596-92-8 |
| | | (215) | (901) | (75) | (312) | | |
| | IP from 82SAI/YAM. | | | | | | |
| Pd⁺ | | | | | | | |
| Pd | 8.34 | 283 | 1183 | 90 | 378 | 82TN270 | 7440-05-3 |
| | | 282 | 1182 | 90 | 377 | | |
| PdSi⁺ | | | | | | | |
| PdSi | (8.4±0.5) 0 K values. | (318) | (1329) | 124 | 519 | 79HUB/HER | 12137-77-8 |
| Pd₂⁺ | | | | | | | |
| Pd ₂ | (7.7±0.3) 0 K values. | (341) | (1426) | 163 | 683 | 79HUB/HER | 12596-93-9 |
| Pm⁺ | | | | | | | |
| Pm | 5.582 | | | | | | |
| Pr⁺ | | | | | | | |
| Pr | 5.464±0.006 | 211 | 883 | 85 | 356 | 82TN270 | 7440-10-0 |
| | | 211.2 | 883.9 | 85.2 | 356.7 | | |
| Pt⁺ | | | | | | | |
| Pt | 8.61 | 334 | 1396 | 135 | 565 | 82TN270 | 7440-06-4 |
| | | 333 | 1395 | 134.9 | 564.4 | | |
| | IP from 79RAU/ACK. See also: 81GUP/NAP. | | | | | | |
| Pu⁺ | | | | | | | |
| Pu | 6.03±0.10 IP from 81CHE/GAB. | 222 | 927 | 82.5 | 345 | 85KLE/WAR | 7440-07-5 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------------|-----------------|---|--------------------------|---------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Ra⁺ | Ra | 5.279 | 160 | 668 | 38 | 159 | 82TN270 | 7440-14-4 |
| | | IP from 70MOO. | | | | | | |
| Rb⁺ | Rb | 4.177 | 116 | 484 | 19.3±0.1 | 80.9±0.4 | 85JANAF | 7440-17-7 |
| | | | 116 | 485 | 19.6±0.1 | 82.2±0.4 | | |
| Rb₂⁺ | Rb ₂ | (3.9±0.1) | (117.0) | (489.6) | 27.1±0.6 | 113.3±2.5 | 85JANAF | |
| | | | (118.0) | (493.6) | 28.0±0.6 | 117.3±2.5 | | |
| | | IP from 85KAP/SCH. | | | | | | |
| Re⁺ | Re | 7.76 | 363 | 1519 | 184 | 770 | 82TN270 | 7440-15-5 |
| | | | 363 | 1518 | 184 | 769 | | |
| | | IP from 79RAU/ACK. | | | | | | |
| Rh⁺ | Rh | 7.46 | 305 | 1275 | 133 | 557 | 82TN270 | 7440-16-6 |
| | | | 305 | 1275 | 133 | 556 | | |
| | | See also: 81HAQ/GIN. | | | | | | |
| RhTi⁺ | TiRh | (8.2±1.0) | (342) | (1431) | 153 | 640 | 79HUB/HER | 12600-90-7 |
| | | 0 K values. | | | | | | |
| Rh₂⁺ | Rh ₂ | (7.1±1.0) | (363) | (1518) | 199 | 833 | 79HUB/HER | 12596-98-4 |
| | | 0 K values. | | | | | | |
| Rn⁺ | Rn | 10.748 | 247.9 | 1037.0 | 0 | 0 | *DEF | 10043-92-2 |
| | | | 247.9 | 1037.0 | 0 | 0 | | |
| | | IP from 70MOO. | | | | | | |
| Ru⁺ | Ru | (7.37) | (324) | (1354) | 154 | 643 | 82TN270 | 7440-18-8 |
| | | | (323) | (1352) | 153 | 641 | | |
| S⁺ | S | 10.360 | 305 | 1275 | 66.2±0.1 | 277.0±0.3 | 85JANAF | 7704-34-9 |
| | | | 304 | 1272 | 65.6±0.1 | 274.7±0.3 | | |
| | | See also: 86LIA/NG, 79DUN/DYK. | | | | | | |
| SSe⁺ | SeS | (9.2±0.2) | (243) | (1015) | 30 | 127 | 83GRA/WIE | 7446-34-6 |
| | | IP from 83GRA/WIE, 77LEV/LIA. 0 K values. | | | | | | |

Table 1. Positive Ion Table - Continued





| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|---|---|----------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| SSn^+ | SnS | (8.8) | (231) | (968) | 28 | 119 | 82TN270 | 1314-95-0 |
| IP is onset of photoelectron band. | | | | | | | | |
| STi^+ | TiS | (7.1±0.3) | (237) | (990) | 73 | 305 | 82TN270 | 12039-07-5 |
| SY^+ | YS | (6.0) | (180) | (754) | 42 | 175 | 82TN270 | 12210-79-6 |
| | | | (180) | (755) | 42 | 176 | | |
| S_2^+ | S_2 | 9.356±0.002 | 246.5 | 1031.3 | 30.7±0.1 | 128.6±0.3 | 85JANAF | 23550-45-0 |
| | | | 246.4 | 1031.0 | 30.7±0.1 | 128.3±0.3 | | |
| IP from 86LIA/NG. See also: 83ROS/GRA, 83GRA/WIE. | | | | | | | | |
| S_3^+ | S_3 | (9.68±0.03) | (257) | (1076) | 34±2 | 142±8 | 85JANAF | 12597-03-4 |
| See also: 83ROS/GRA. | | | | | | | | |
| S_4^+ |  | (10.1) | (270) | (1131) | 35±2 | 146±8 | 85JANAF | 19269-85-3 |
| $\Delta_f H(\text{Ion})$ from appearance potential of 11.94±0.05 in S_6 . IP from 83ROS/GRA. | | | | | | | | |
| S_5^+ |  | (8.60±0.05) | (224) | (939) | 26±2 | 109±8 | 85JANAF | 12597-10-3 |
| See also: 83ROS/GRA. | | | | | | | | |
| S_6^+ |  | (9.00±0.03) | (232) | (971) | 24±2 | 102±8 | 85JANAF | 13798-23-7 |
| See also: 83ROS/GRA. | | | | | | | | |
| S_7^+ |  | (8.67±0.03) | (227) | (951) | 27±2 | 114±8 | 85JANAF | 21459-04-1 |
| See also: 83ROS/GRA. | | | | | | | | |

Table 1. Positive Ion Table - Continued

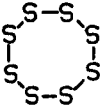
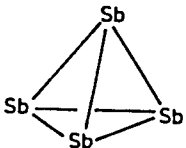
| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------------------|---|--|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| S₈⁺ |  | (9.04±0.03) | (232) | (972) | 24.0±0.2 | 100.4±0.6 | 85JANAF | 10544-50-0 |
| | | | (233) | (976) | 25.0±0.2 | 104.4±0.6 | | |
| | | See also: 83ROS/GRA. | | | | | | |
| Sb⁺ | Sb | 8.641 | 261.9 | 1096.0 | 62.7 | 262.3 | 82TN270 | 7440-36-0 |
| | | | 261.9 | 1095.7 | 62.6 | 262.0 | | |
| | | It has been suggested (83MAZ) that this value of the IP is too high. | | | | | | |
| Sb₂⁺ | Sb ₂ | (9.3±0.2) | (271) | (1133) | 56 | 236 | 82TN270 | 32679-33-7 |
| | | | (271) | (1134) | 57 | 237 | | |
| | | The cited ionization potential is from a spectroscopic determination. Threshold determinations have led to values of 8.4±0.3, 8.64±0.06, 8.7±0.3, 8.9±0.3, and 9.5±0.5 eV. | | | | | | |
| Sb₄⁺ |  | (7.40±0.10) | (220) | (919) | 49 | 205 | 82TN270 | 12597-17-0 |
| | | | (221) | (924) | 50 | 210 | | |
| | | IP from 84ELB/KUD. | | | | | | |
| Sc⁺ | Sc | 6.54 | 241 | 1009 | 90 | 378 | 82TN270 | 7440-20-2 |
| | | | 241 | 1007 | 90 | 376 | | |
| | | See: 85DYK/GRA. | | | | | | |
| Se⁺ | Se | 9.752 | 279.2 | 1168.0 | 54.27 | 227.07 | 82TN270 | 7782-49-2 |
| | | | 279.0 | 1167.3 | 54.11 | 226.40 | | |
| SeSn⁺ | SeSn | (8.6) | (229) | (959) | 31 | 129 | 79HUB/HER | 1315-06-6 |
| | | | (228) | (953) | 29 | 123 | | |
| | | IP is onset of photoelectron band. | | | | | | |
| SeTe⁺ | SeTe | (8.5±0.2) | (227) | (948) | 31 | 128 | 83GRA/WIE | 12067-42-4 |
| | | IP from 83GRA/WIE. | | | | | | |
| SeY⁺ | SeY | (6.1±1) | (90) | (376) | -51 | -213 | 79HUB/HER | 12067-44-6 |
| | | 0 K values. | | | | | | |

Table 1. Positive Ion Table - Continued

| ION Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|----------------------------------|----------------------------|--------------------------|----------|------------------------------|----------|----------------------|------------------------|
| | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Se_2^+ Se_2 | 8.70±0.05 | 236 | 985 | 35 | 146 | 82TN270 | 12185-17-0 |
| | | 236 | 987 | 35 | 148 | | |
| See also: 83POT/NOV, 83GRA/WIE. | | | | | | | |
| Si^+ Si | 8.15172±0.00003 | 295 | 1236 | 108±2 | 450±8 | 82TN270 | 7440-21-3 |
| | | 295 | 1233 | 106±2 | 446±8 | | |
| Si_2^+ Si_2 | (7.4) | (311.6) | (1303.9) | 141.0±3 | 589.9±13 | 85JANAF | 12597-35-2 |
| | | (311.0) | (1301.1) | 140.3±3 | 587.1±13 | | |
| Sm^+ Sm | 5.6437±0.0006 | 180 | 751 | 49.4 | 206.7 | 82TN270 | 7440-19-9 |
| | | 179 | 751 | 49.3 | 206.1 | | |
| Sn^+ Sn | 7.344 | 241.5 | 1010.7 | 72.2 | 302.1 | 82TN270 | 7440-31-5 |
| | | 241.5 | 1010.6 | 72.2 | 302.0 | | |
| Sr^+ Sr | 5.695 | 170 | 713 | 39±0.5 | 164±2 | 85JANAF | 7440-24-6 |
| | | 170 | 713 | 39±0.5 | 164±2 | | |
| Ta^+ Ta | 7.40 | 358 | 1496 | 187 | 782 | 85JANAF | 7440-25-7 |
| | | 357 | 1495 | 186.7 | 781.4 | | |
| IP from 79RAU/ACK. | | | | | | | |
| Tb^+ Tb | 5.8639±0.0006 | 228 | 955 | 93 | 389 | 82TN270 | 7440-27-9 |
| | | 228.6 | 956.4 | 93.4 | 390.6 | | |
| Tc^+ Tc | (7.28) | (330) | (1380) | 162 | 678 | 82TN270 | 7440-26-8 |
| Te^+ Te | 9.009 | 255 | 1066 | 47 | 197 | 82TN270 | 22541-49-7 |
| | | 255 | 1066 | 47 | 197 | | |
| TeY^+ YTe | (6.0±1.0) 0 K values. | (206) | (860) | 67 | 281 | 79HUB/HER | 12187-04-1 |
| Te_2^+ Te_2 | 8.29±0.03 | 223 | 933 | 32 | 133 | 79HUB/HER | 10028-16-7 |
| | | 224 | 936 | 32 | 136 | | |
| See also: 83GRA/WIE. | | | | | | | |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|--|---------------|---|--------------------------|------------------|------------------------------|------------------------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Th^+ | Th | 6.08 | 283 283 | 1184 1184 | 143 143 | 597 597 | 85KLE/WAR | 7440-29-1 |
| Ti^+ | Ti | 6.82 | 270 269 | 1128 1125 | 112 112 | 470 467 | 82TN270 | 7440-32-6 |
| Tl^+ | Tl | 6.108 | 184.4 184.5 | 771.6 772.1 | 43.5 43.7 | 182.2 182.8 | 82TN270 | 7440-28-0 |
| Tl_2^+ | Tl_2 | (6.5±0.5) IP from 80BAL/PIA. 0 K values. | (223) | (932) | 73 | 305 | 80BAL/PIA | 76939-73-6 |
| Tm^+ | Tm | 6.18 | 198 198 | 828 830 | 55 55.8 | 232 233.4 | 82TN270 | 7440-30-4 |
| U^+ | U | 6.1912 | 270 270 | 1128 1128 | 127 127 | 531 531 | 85KLE/WAR | 7440-61-1 |
| IP from 70EME/KHO, 76SOL/MAY. See also: 81CHE/GAB. | | | | | | | | |
| V^+ | V | 6.74 | 278 278 | 1165 1162 | 123±2 122±2 | 515±8 512±8 | 85JANAF | 7440-62-2 |
| See also: 85DYK/GRA. | | | | | | | | |
| W^+ | W | 7.60 | 379 378 | 1584 1582 | 203.4±1.5 203.7±1.5 | 851.0±6.3 849.8±6.3 | 85JANAF | 7440-33-7 |
| Xe^+ | Xe | 12.130 | 279.7 279.7 | 1170.4 1170.4 | 0 0 | 0 0 | *DEF | 7440-63-3 |
| See also: 81KIM/KAT. | | | | | | | | |
| Xe_2^+ | Xe_2 | 11.13±0.02 | 256 | 1072 | -0.53 | -2.22 | 79HUB/HER | 12185-19-2 |
| See also: 82POL/DEH. 0 K values. | | | | | | | | |
| Y^+ | Y | (6.22) | (244) (244) | (1021) (1021) | 101 100.5 | 421 420.4 | 82TN270 | 7440-65-5 |
| IP from 73GAR/REE. | | | | | | | | |
| Yb^+ | Yb | 6.254 | 180 180.7 | 755 756.2 | 36 36.5 | 152 152.8 | 82TN270 | 7440-64-4 |

Table 1. Positive Ion Table - Continued

| ION | Neutral | Ionization potential eV | $\Delta_f H(\text{Ion})$ | | $\Delta_f H(\text{Neutral})$ | | Neutral reference | CAS registry number |
|-----------------------|---------|----------------------------|--------------------------|--------|------------------------------|-----------|----------------------|------------------------|
| | | | kcal/mol | kJ/mol | kcal/mol | kJ/mol | | |
| Zn⁺ | | | | | | | | |
| Zn | | 9.394 | 247.8 | 1036.8 | 31.2 | 130.4±0.2 | 8SJANAF | 7440-66-6 |
| | | | 247.7 | 1036.3 | 31.0 | 129.9±0.2 | | |
| Zr⁺ | | | | | | | | |
| Zr | | 6.84 | 303 | 1270 | 146±2 | 610±8 | 8SJANAF | 7440-67-7 |
| | | | 302 | 1262 | 144±2 | 602±8 | | |

Table 2. Negative Ion Table

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|-------------------------|--|-------------|--|--|--------|--|-----------|
| e^- | | | | | | $\Delta_f H(AH) = 218$ | 82TN270 |
| e^- | | | | | | $BDE(A-H) = 1312$ | 82TN270 |
| e^- | 0 | | 1312 ^f | 1308 ^h | Def | | 82TN270 |
| Ag^- | | | | | | | |
| Ag^- | | | | | | $\Delta_f H(A) = 285$ | 82TN270 |
| Ag^- | 159±1 ^b | 1.302±0.007 | | | LPES | | 85HOT/LIN |
| Al^- | | | | | | $\Delta_f H(AH) = 259±1$ | 82TN270 |
| Al^- | | | | | | $BDE(A-H) = 285±3$ | 81KAN/MOO |
| Al^- | 284±5 ^a | 0.441±0.010 | 1554±4 ^c | | LPES | | 85HOT/LIN |
| $AlBeF_6^-$ | | | | | | | |
| $BeF_2 \cdot AlF_4^-$ | | | | | | | |
| $AlBeF_6^-$ | -2921±21 ^c | | 182±10 | | TDA's | | 80NIK/SOR |
| $AlBeF_7K^-$ | | | | | | | |
| $KBeF_3 \cdot AlF_4^-$ | | | | | | | |
| $AlBeF_7K^-$ | -3522±21 ^c | | 192±8 | | TDA's | | 80NIK/SOR |
| $AlBeF_7Na^-$ | | | | | | | |
| $NaBeF_3 \cdot AlF_4^-$ | | | | | | | |
| $AlBeF_7Na^-$ | -3497±21 ^c | | 192±8 | | TDA's | | 80NIK/SOR |
| AlF_2^- | | | | | | | |
| AlF_2^- | | | | | | $\Delta_f H(A) = -749±13$ | 81WOO |
| AlF_2^- | -971±13 | 2.25±0.13 | | | TDEq | EA: 111 kJ < EA(F), new EA(F) data used | 74SRI/UY |
| AlF_4^- | | | | | | | |
| $AlF_3 \cdot F^-$ | | | | | | | |
| AlF_4^- | -1945±10 ^c | | 488±8 | | TDA's | | 86NIK/IGO |
| AlF_4^- | | | | | | Summary of literature data plus new work. Recommended average value | |
| AlF_4^- | -1972±21 ^c | | 498±7 | | TDA's | F ⁻ A: 1100K; $\Delta_f H(AlF_4^-)$: 298K | 80SID/NIK |
| AlF_4^- | -1964±14 ^c | | 495±11 | | TDEq | F ⁻ A: 93±1 kJ > UF ₄ | 79NIK/SKO |
| AlF_4^- | -1954±12 ^c | | 500±8 | | TDEq | F ⁻ A: 17 kJ > ScF ₃ | 81NIK/SID |
| AlF_4^- | -2092±13 | | 628±42 ^k | | TDEq | 2AlF ₂ + AlF ₂ ⁻ = 2AlF + AlF ₄ ⁻ | 74SRI/UY |
| AlF_4^- | -1949±16 ^c | | 496±13 | | TDEq | KF ₂ ⁻ + KAlF ₄ = AlF ₄ ⁻ + 2KF | 80GUS/PYA |
| AlF_5K^- | | | | | | | |
| $KF \cdot AlF_4^-$ | | | | | | | |
| AlF_5K^- | -2397±33 ^c | | 120±8 | | TDA's | | 79GUS/GOR |
| AlF_7Mn^- | | | | | | | |
| $MnF_3 \cdot AlF_4^-$ | | | | | | | |
| AlF_7Mn^- | -2950±60 | | | | TDA's | | 84KOR/CHI |
| AlO^- | | | | | | | |
| AlO^- | | | | | | $\Delta_f H(A) = 67±8$ | 85JANAF |
| AlO^- | -282±21 ^b | 3.62±0.13 | | | TDEq | EA: near EA(Cl) | 72SRI/UY |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|----------------------|--|--|--------------|--|-----------------------------------|
| AlO_2^- AlO_2^- | | 4.05 ± 0.13 | | | TDEq | $\Delta_f H(A) = -130$ EA: 42 kJ > EA(Cl), new data for $\Delta_f H(AlO_2)$ and EA(Cl) used | 82KAS/CHE 72SRI/UY |
| $Al_2F_7^-$ $AlF_3 \cdot AlF_4^-$ | | | 204 ± 4 | 175 ± 8 | TDAs | | 80SID/NIK |
| | -3394 ± 15^c -3393 ± 33 | | | | TDAs | | 79GUS/GOR |
| $Al_2F_8K^-$ $KAlF_4 \cdot AlF_4^-$ | | | 147 ± 6 | 123 ± 10 | TDAs | | 80SID/NIK |
| | -4011 ± 42^c | | | | | | |
| $Al_2F_8Na^-$ $NaAlF_4 \cdot AlF_4^-$ | | | 166 ± 9 | 141 ± 13 | TDAs | | 80SID/NIK |
| | -4006 ± 42^c | | | | | | |
| $ArBr^-$ $Ar \cdot Br^-$ | | | 6 | | Mobl | | 84GAT |
| | -219^c | | | | | | |
| As^- As^- | | | | | | $\Delta_f H(A) = 303 \pm 2$ | 82TN270 |
| | 224 ± 5^b | 0.81 ± 0.03 | | | PD | | 85HOT/LIN |
| $AsBr^-$ $AsBr^-$ | | 1.3 | | | EIAP | From $AsBr_3$ | 76PAB/BEN |
| | 7 | | | | | | |
| $AsBr_2^-$ $AsBr_2^-$ | | 3.5 ± 0.1 3.5 | | | EIAP EIAP | From $AsBr_3$ From $AsBr_3$ | 78PAB/MAR 76PAB/BEN |
| | -303 | | | | | | |
| $AsCl^-$ $AsCl^-$ | | 1.3 | | | EIAP | From $AsCl_3$ | 76PAB/BEN |
| | -9 | | | | | | |
| $AsClF_3^-$ $AsF_3 \cdot Cl^-$ | | | 108 ± 8^g | 78 ± 8 | IMRE | | 85LAR/MCM |
| | -1121 ± 12^c | | | | | | |
| $AsCl_2^-$ $AsCl_2^-$ | | 2.2 2.2 ± 0.1 | | | EIAP EIAP | $\Delta_f H(A) = 67 \pm 21$ From $AsCl_3$ From $AsCl_3$ | 82TN270 76PAB/BEN 78PAB/MAR |
| | -273 | | | | | | |
| AsF^- AsF^- | | 1.3 | | | EIAP | From AsF_3 | 76PAB/BEN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--------------------------------|--|------------------------------------|--|--|--------------------|---|---|
| AsF_2^- AsF_2^- | -543 | 0.8 0.8±0.1 | | | EIAP EIAP | From AsF_3 From AsF_3 | 76PAB/BEN 78PAB/MAR |
| AsF_4^- $AsF_3 \cdot F^-$ | * -1236±13 ^c | | 202±8 ^g | 172±8 | IMRE | | 85LAR/MCM |
| AsH^- AsH^- | <167 | 1.0±0.1 <1.1 | | | PD IMRB IMRB | $As^- + AsH_3 \rightarrow$ From AsH_3 | 77RAC/FEL 74WYA/HOL 64EBI/KRA |
| AsH_2^- AsH_2^- | * 52±26 ^a * <41±20 ^a | 1.27±0.03 <1.1±0.5 ^d | 1515±26 ^g | 1483±25 | IMRB PD EIAP | $\Delta_f H(AH) = 67 \pm 1$ D-EA BDE(A-H) = 326±33 Between PH_3, H_2S From AsH_3 | 82TN270 74WYA/HOL 72SMY/BRA2 64EBI/KRA |
| As_2^- As_2^- | 180±19 | <0.8 0.1±0.2 ⁱ | | | PD EIAP | $\Delta_f H(A) = 190 \pm 3$ From As_4 | 73BEN/MAR 77FEL/RAC 73BEN/MAR |
| As_2H^- As_2H^- | <288 | | | | IMRB | $As^- + AsH_3 \rightarrow As_2H^- + H_2$ | 74WYA/HOL |
| As_3^- As_3^- | 160±18 | 0.8±0.4 ⁱ | | | EIAP | $\Delta_f H(A) = 241 \pm 16$ From As_4 | 73BEN/MAR 73BEN/MAR |
| Au^- Au^- | * 144±6 ^a | 2.309 | 1379±4 ^c | | LPD | $\Delta_f H(AH) = 295 \pm 2$ BDE(A-H) = 289±4 | 82TN270 82TN270 85HOT/LIN |
| AuF_6^- AuF_6^- | | >1.0 | | | NBIP | | 80COM/REI |
| B^- B^- | * 416±13 ^a | 0.277±0.010 | 1497±11 ^c | 1468±13 ^h | LPES | $\Delta_f H(AH) = 450 \pm 2$ BDE(A-H) = 211±10 | 82TN270 85JANAF 85HOT/LIN |
| $BBrCl_2^-$ BCl_2Br^- | -403±61 ^b | 0.7±0.2 | | | NBIP | $\Delta_f H(A) = -337 \pm 42$ | 85JANAF 80ROT/MAT |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|----------------|---|--|--|---|---|
| BBr ₂ Cl ⁻ BBr ₂ Cl ⁻ | -363±61 ^b | 0.9±0.2 | | | NBIP | $\Delta_f H(A) = -272 \pm 42$ | 85JANAF 80ROT/MAT |
| BBr ₃ ⁻ BBr ₃ ⁻ | -285±20 ^b | 0.8±0.2 | | | NBIP | $\Delta_f H(A) = -206$ | 82TN270 80ROT/MAT |
| BClF ₃ ⁻ BF ₃ · Cl ⁻ | -1473±12 ^c | | 109±8 ^g | 81±8 | IMRE | | 85LAR/MCM |
| BCl ₂ ⁻ BCl ₂ ⁻ | -144±109 ^b | 0.6 | | | Est2 | $\Delta_f H(AH) = -248 \pm 4$ Est: from IP, EA of isoelectronic NO ₂ , BF ₂ , AlF ₂ | 71JANAF 82JANAF |
| BCl ₃ ⁻ BCl ₃ ⁻ | -436±20 ^b | 0.3±0.2 | | | NBIP | $\Delta_f H(A) = -404 \pm 1$ | 82TN270 80ROT/MAT |
| BCl ₃ F ⁻ BCl ₃ · F ⁻ | < -890 | | > 238 ^k | | IMRB | F ⁻ A: > SF ₅ | 72STO/NEL |
| BCl ₄ ⁻ BCl ₃ · Cl ⁻ | -920±6 | | 289±8 ^k | | Latt | | 77KRI/TTT |
| BF ₂ ⁻ BF ₂ ⁻ | -796±26 ^b | 2.13±0.13 | 1468±29 ^e | 1436±31 ^h | IMRE | $\Delta_f H(AH) = -734 \pm 3$ $BDE(A-H) = 362 \pm 17$ EA: 122 kJ < EA(F), new EA(F) used | 85JANAF 85JANAF 74SRI/UY |
| BF ₂ O ⁻ BF ₂ O ⁻ | < -1002±25 | | < 1619±33 ^f | | IMRB | $\Delta_f H(AH) = -1092 \pm 8$ $BDE(A-H) = 473 \pm 23$ DO ⁻ + BF ₃ → | 82TN270 85JANAF 72STO/NEL |
| BF ₃ ⁻ BF ₃ ⁻ | | 0.0±0.2 2.6 | | | NBIP SI | $\Delta_f H(A) = -1137 \pm 2$ See also: 72STO/NEL | 85JANAF 80ROT/MAT 69PAG/GOO |
| BF ₄ ⁻ BF ₃ · F ⁻ | -1687±25 ^c -1716±44 ^c -1773±26 -1779±21 < -1812 < -1368 | | 301±21 ^g 330±40 385±25 ^k 393±21 ^k | 266±8 | IMRE TDEq Latt Latt TDEq IMRB | BF ₃ + BF ₂ ⁻ = BF ₄ ⁻ + BF F ⁻ A: > SF ₅ | 85LAR/MCM 84PYA/GUS 84MAL/ROS 77KRI/TTT 74SRI/UY 72STO/NEL |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------------------------|--|--|----------------------|---|---|
| BH₄⁻ BH ₃ · H ⁻ | < 197±28 -96±21 -85±8 | | 341 ^k 322±8 ^k | | IMRB Latt Latt | HO ⁻ + B ₂ H ₆ → | 68DUN 55ALT 77KRI/TTT |
| BKO₂⁻ KBO ₂ ⁻ | -785±31 ^b | 1.2±0.2 | | | EIAP | $\Delta_f H(A) = -672 \pm 10$ From K ₂ BO ₂ F | 85FAR/SRI 76SHE/ILJ |
| BNaO₂⁻ NaBO ₂ ⁻ | -782±63 ^b | 1.4±0.2 | | | EIAP | $\Delta_f H(A) = -644 \pm 42$ From Na ₂ BO ₂ F | 82TN270 76SHE/ILJ |
| BO⁻ BO ⁻ | -196±17 ^b | 2.84±0.09 > 2.48 | | | TDEq TDEq | EA(BO) < EA(Cl) by 75 kJ, new $\Delta_f H(BO_2)$ and EA(Cl) used | 71SRI/UY 70JEN |
| BO₂⁻ BO ₂ ⁻ | -736±18 -617±15 ^b | 4.51±0.21 ⁱ 3.28±0.13 | 1356±26 ^e | | TDEq TDEq | $\Delta_f H(AH) = -562 \pm 4$ $BDE(A-H) = 479 \pm 6$ | 82TN270 82TN270 83SID/RUD 71SRI/UY |
| | | 3.4±0.5 4.19±0.31 | | | EIAP TDEq | EA(BO ₂) < EA(Cl) by 32 kJ, new $\Delta_f H(BO_2)$ and EA(Cl) used From K ₂ BO ₂ F | 76SHE/ILJ 70JEN |
| BeF₃⁻ BeF ₂ · F ⁻ | -1477±15 ^c | | 407±10 | | TDEq | F ⁻ A: 83±7 kJ < AlF ₃ | 80NIK/SOR |
| BeH⁻ BeH ⁻ | 276 ^b | 0.7±0.1 | | | PD | $\Delta_f H(A) = 344$ | 79HUB/HER 77RAC/FEL |
| Be₂F₅⁻ Be ₂ F ₄ · F ⁻ | | | 464±8 | | TDEq | F ⁻ A: 26±8 kJ < AlF ₃ | 80NIK/SOR |
| Be₂F₆K⁻ KF · Be ₂ F ₅ ⁻ | -617 ^c | | 290±10 | | TDAs | | 80NIK/SOR |
| Be₂F₆Na⁻ NaF · Be ₂ F ₅ ⁻ | -564 ^c | | 273±10 | | TDAs | | 80NIK/SOR |
| Be₃F₇⁻ BeF ₂ · Be ₂ F ₅ ⁻ | -969 ^c | | 175±10 | | TDAs | | 80NIK/SOR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|--------------------------------------|--|----------------------------------|
| Be₃F₈K⁻ KBeF ₃ · Be ₂ F ₅ ⁻ | | | 188±8 | | TDA _s | | 80NIK/SOR |
| Bi⁻ Bi ⁻ | | | | | | $\Delta_f H(A) = 207 \pm 4$ | 82TN270 |
| • | 116±5 ^b | 0.946±0.010 | | | LPES | | 85HOT/LIN |
| Br⁻ Br ⁻ | | | | | | $\Delta_f H(AH) = -36$ $BDE(A-H) = 366$ | 85JANAF |
| • | -213±1 ^a | 3.365±0.003 | 1354 ^c 1349±9 ^g | 1331±1 ^h 1326±8 | PLA IMRE | | 82BAUJ/COX 85HOT/LIN 86TAF |
| BrClH⁻ HCl · Br ⁻ | | | | | | | |
| • | -387±9 ^c | | 82±8 | 54±11 | TDA _s | | 85CAL/KEB |
| BrCl₂⁻ BrCl ₂ ⁻ | | | | | | | |
| | < -464 | | > 251 ^k | | PDis | | 79LEE/SMI |
| BrCl₂P⁻ PBrCl ₂ ⁻ | | | | | | $\Delta_f H(A) = -188 \pm 42$ | |
| | -335±61 ^b | 1.5±0.2 | | | Est2 NBIP | | 76MAT/ROT |
| BrFH⁻ HF · Br ⁻ | | | | | | | |
| • | -557±10 ^c | | 71±8 | | Est | Extrapolated from other bihalide data | 84LAR/MCM3 |
| BrHI⁻ HBr · I ⁻ | | | | | | | |
| • | -292±9 ^c | | 67±8 | 43±11 | TDA _s | | 85CAL/KEB |
| BrHNO₃⁻ HBr · NO ₃ ⁻ | | | | | | | |
| | -438±10 ^c | | 94±8 | 78±7 | TDEq | | 77DAV/FEH |
| BrH₂O⁻ HOH · Br ⁻ | | | | | | | |
| | -517 ^c | | 62 53±8 | 37 29±8 | TDA _s TDA _s | | 82BUR/HAY 70ARS/YAM |
| BrI⁻ IBr ⁻ | | | | | | $\Delta_f H(A) = 41 \pm 1$ | 82TN270 |
| | -205±11 ^b | 2.5±0.1 | | | NBIP | | 72BAE |
| | | 2.5±0.1 | | | NBIP | | 73AUE/HUB |
| | | 2.7±0.2 | | | EnCT | | 71CHU/BER |
| | | 1.62±0.05 | | | NBIP | Vertical EA | 76HUB/KLE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|---|--|--|---|---|---|
| BrK⁻ KBr ⁻ | -239 ^b | 0.6 | | | Scat | $\Delta_f H(A) = -180 \pm 2$ | 85JANAF 79DEV/WIJ |
| BrKr⁻ Kr ⁻ · Br ⁻ | -221 ^c | | 8 | | Mobl | | 84GAT |
| BrNa⁻ NaBr ⁻ | -234 ^b | 0.9 | | | Scat | $\Delta_f H(A) = -143 \pm 8$ | 82TN270 79DEV/WIJ |
| BrO⁻ BrO ⁻ | < -20 ± 21 | > 1.5 ± 0.2 ⁱ | < 1590 ± 29 ^f | | Endo | $\Delta_f H(AH) = -79 \pm 8$ $BDE(A-H) = 423 \pm 13$ Br ⁻ + O ₂ → | 76BEN 82TN270 77VOG/DRE |
| BrO₂S⁻ SO ₂ · Br ⁻ | -590 ± 10 ^c | | 81 ± 8 | 53 ± 11 | TDA _s | | 85CAL/KEB |
| BrPb⁻ PbBr ⁻ | | 0.9 ± 0.2 | | | EIAP | From PbBr ₂ | 67HAS/BLO |
| BrXe⁻ Xe ⁻ · Br ⁻ | -227 ^c | | 14 | | Mobl | | 84GAT |
| Br₂⁻ Br ₂ ⁻ | | 2.5 ± 0.1 2.4 2.5 ± 0.1 2.6 ± 0.2 2.6 ± 0.2 2.9 ± 0.1 1.47 ± 0.05 | | | NBIP ECD EnCT NBIP EnCT EIAP NBIP | $\Delta_f H(A) = 31$ Vertical EA: 1.6 eV From CBr ₄ Vertical EA | 82BAU/COX 72BAE 81AYA/WEN 71CHU/BER 77DIS/LAC2 73HUG/LIF 71DEC/FRA 76HUB/KLE |
| Br₂ClP⁻ PBr ₂ Cl ⁻ | -309 ± 61 ^b | 1.6 ± 0.2 | | | Est2 NBIP | $\Delta_f H(A) = -152 \pm 42$ | 76MAT/ROT |
| Br₂Ge⁻ GeBr ₂ ⁻ | < -217 ^b | > 1.6 | | | EIAP | $\Delta_f H(A) = -63 \pm 8$ From GeBr ₄ | 82TN270 77PAB/MAR |
| Br₂H⁻ HBr · Br ⁻ | -336 ± 9 ^c | | 86 ± 8 | 58 ± 11 | TDA _s | | 85CAL/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|--------------|--|--|
| Br_2I^- IBr_2^- -251 | | | 95 ^k | | Latt | | 77FIN/GAT |
| Br_2Si^- $SiBr_2^-$ | | > 1.7 | | | EIAP | From $SiBr_4$ | 77PAB/MAR |
| Br_2Sn^- $SnBr_2^-$ -54 | | 1.3 | | | EIAP | From $SnBr_4$ | 77PAB/PER |
| Br_3Ge^- $GeBr_3^-$ | | > 0.9 | | | EIAP | From $GeBr_4$ | 77PAB/MAR |
| Br_3P^- PBr_3^- -293±23 ^b | 1.6±0.2 | | | | NBIP | $\Delta_f H(A) = -139 \pm 8$ | 82TN270 76MAT/ROT |
| Br_3Si^- $SiBr_3^-$ | > 1.5±0.2 | | | | EIAP | $\Delta_f H(AH) = -318 \pm 2$ From $SiBr_4$ | 82TN270 77PAB/MAR |
| Br_3Sn^- $SnBr_3^-$ -484 | 3.08±0.01 > 3.1 | | | | EIAP EIAP | $\Delta_f H(AH) = -318 \pm 8$ $BDE(A-H) = 349 \pm 28$ From $SnBr_4$ From $SnBr_4$ | 82TN270 77PAB/PER 78PAB/MAR 77PAB/PER |
| Br_3Ti^- $TiBr_3^-$ -452±15 | 0.76±0.01 0.8±0.3 | | | | EIAP EIAP | $\Delta_f H(A) = -377 \pm 42$ From $TiBr_4$ From $TiBr_4$ | 74BEN/PAB 78PAB/MAR 74BEN/PAB |
| C^- C^- * 595±1 ^a | 1.263 > 1.2±1.0 | | 1529±1 ^c | 1506±2 ^h | LPD EIAP | $\Delta_f H(AH) = 596$ $BDE(A-H) = 339 \pm 1$ From graphite | 82TN270 82TN270 85HOT/LIN 54HON |
| $CBrF_3^-$ CF_3Br^- -740 ^b | 0.9±0.2 | | | | NBIP | $\Delta_f H(A) = -652$ | 78KUD/KUD 78COM/REI2 |
| CBr_3^- Br_3C^- 40±29 ^b | 1.7±0.2 | | 1546±34 ^f | 1514±38 ^h | SI | $\Delta_f H(AH) = 24 \pm 5$ $BDE(A-H) = 402 \pm 7$ | 84BIC/MIN 82MCM/GOL 69PAG/GOO |
| CBr_4^- Br_4C^- -115 ^b | 2.1 | | | | SI | $\Delta_f H(A) = 84 \pm 3$ | 84BIC/MIN 69PAG/GOO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|---|---|---|--|---|--|
| CClF ₂ ⁻ CF ₂ Cl ⁻ | -431±35 ^a | 1.6±0.3 > 1.9 | 1583±33 ^c | 1550±37 ^h | NBAP EIAP | $\Delta_f H(AH) = -484 \pm 2$ $BDE(A-H) = 425 \pm 4$ From CF ₂ Cl ₂ From CF ₃ Cl | 77PED/RYL 82MCM/GOL 78DIS/LAC 79ILL/SCH |
| CClF ₂ O ⁻ CF ₂ =O ·· Cl ⁻ | * -920±11 ^c < -728±25 | | 52±8 ^g | 28±8 | IMRE IMRB | CICO ⁻ or ClF ⁻ + CF ₂ O → | 85LAR/MCM 76KAR/KLE |
| CClO ⁻ ClCO ⁻ | < -356 | | > 21 ^k | | EIAP | From Cl ₂ CO | 76KAR/KLE |
| CClO ₂ ⁻ CO ₂ ·· Cl ⁻ | * -654±2 ^c | | 33 32±8 | 9 9±8 | TDA _s TDEq | | 80KEE/LEE 86HIR/SHO |
| CClS ₂ ⁻ CS ₂ ·· Cl ⁻ | * -159±10 ^c | | 49±8 ^g | 24±8 | IMRE | | 85LAR/MCM |
| CCl ₂ ⁻ CCl ₂ ⁻ | -10±42 ^b | 1.8±0.3 2.5±0.6 | 1411±46 ^c | | NBAP EIAP | $\Delta_f H(AH) = 109 \pm 4$ $BDE(A-H) = 272 \pm 17$ From CCl ₄ , CFCl ₃ , CHCl ₃ From CCl ₄ , CHCl ₃ , CH ₂ Cl ₂ | 83WEI/BEN 85LIA/KAR 78DIS/LAC 80SCH/ILL |
| CCl ₂ F ⁻ CCl ₂ F ⁻ | | > 2.4±0.2 1.1±0.3 | < 1506±23 ^c | | EIAP NBAP | $\Delta_f H(AH) = -281$ $BDE(A-H) = 425 \pm 4$ From CF ₂ Cl ₂ From CFCl ₃ | 78KUDI/KUD 82MCM/GOL 79ILL/SCH 78DIS/LAC |
| CCl ₂ FO ⁻ CCl ₂ =O ·· F ⁻ | < -590±50 | | | | IMRB | FCO ⁻ + CCl ₂ O → | 76KAR/KLE |
| CCl ₂ F ₂ ⁻ CCl ₂ F ₂ ⁻ | -516±25 ^b | 0.4±0.2 | | | NBIP | $\Delta_f H(A) = -477 \pm 5$ | 77PED/RYL 78DIS/LAC |
| CCl ₃ ⁻ CCl ₃ ⁻ | * -141±28 ^a | 2.3±0.3 ^d 2.6±0.2 > 1.9 1.3±0.3 > 2.1±0.3 1.44±0.05 | 1494±26 ^g | 1461±25 | IMRB EIAP EIAP NBAP EIAP SI | $\Delta_f H(AH) = -105 \pm 2$ $BDE(A-H) = 401 \pm 4$ > acetone, ≤ C ₅ H ₆ From CCl ₄ From CFCl ₃ From CHCl ₃ , CCl ₄ From CFCl ₃ | 77PED/RYL 82MCM/GOL 72BOH/LEE 80SCH/ILL 79ILL/SCH 78DIS/LAC 61CUR 66GAI/KAY |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|---|--|--|----------------------|--|--|
| CCl_3F^- $CCl_3F^- \cdot$ | -374±37 ^b | 1.1±0.3 | | | NBIP | $\Delta_f H(A) = -268 \pm 8$ | 77PED/RYL 78DIS/LAC |
| CCl_3O^- $CCl_2 = O \cdot \cdot Cl^-$ | -498±9 ^c -460±8 | | 52±8 ^g | 27±8 | IMRE IMRB | | 85LAR/MCM 76KAR/KLE |
| CCl_4^- $CCl_4^- \cdot$ | -290±22 ^b | 2.0±0.2 2.0±0.2 2.1±0.1 | | | NBIP NBIP SI | $\Delta_f H(A) = -97 \pm 3$ | 77PED/RYL 83LAC/MAN 78DIS/LAC 66GAI/KAY |
| CCl_5^- $CCl_4 \cdot \cdot Cl^-$ | -384±7 ^c | | 59±3 | 25±4 | TDA's | | 74DOU/DAL |
| CF^- $CF^- \cdot$ | | > 3.3±0.3 | | | EIAP | $\Delta_f H(A) = 255$ From C_2F_4 | 85JANAF 70THY/MAC2 |
| CFN^- $FCN^- \cdot$ | -318 | 2.9 ⁱ > 4.0 | | | EIAP EIAP | $\Delta_f H(A) = 36 \pm 17$ From CF_3NC From PF_2CN | 85JANAF 86HEN/ILL 74HAR/RAN |
| CFO^- $FCO^- \cdot$ | -444±10 -435 ^a | 2.7 ⁱ 2.3±0.5 ^d 3.3 | 104 ^k 1475±19 | | EIAP EIAP EIAP | Est $\Delta_f H(AH) = -380$ $BDE(A-H) = 415 \pm 29$ From CF_2O From $HCOF$ From $(CF_3)_2CO$ | 81DYK/JON2 70THY/MAC 77KAR/KLE 70HAR/THY |
| $CFOS^-$ $COS \cdot \cdot F^-$ | -524±11 ^c | | 133±8 ^g | 103±8 | IMRE | | 85LAR/MCM |
| CFO_2^- $CO_2 \cdot \cdot F^-$ | -775±11 ^c -780±15 ^c | | 133±8 ^g 138±13 | 103±8 | IMRE IMRE | | 85LAR/MCM 78MCM/NOR |
| CFS_2^- $CS_2 \cdot \cdot F^-$ | -262±11 ^c | | 131±8 ^g | 101±8 | IMRE | | 85LAR/MCM |
| CF_2^- $CF_2^- \cdot$ | < -102 | < 1.3±0.8 > 0.2 | | | IMRB EIAP EIAP | $\Delta_f H(A) = -205 \pm 13$ $O^- + CH_2F_2 \rightarrow$ From $c-C_4F_8$ From C_2F_4 | 85LIA/KAR 76DAW/JEN 72HAR/THY2 70THY/MAC2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|------------------------------------|---|-------------------|---|---|-------------------|---|------------------------|
| CF_2^- CF_2^- | | 2.6 | | | SI | $\Delta_f H(A) = -205 \pm 13$ | 85LIA/KAR 69PAG/GOO |
| CF_2N^- CF_2N^- | | | | | EIAP | From CF_3NC | 86HEN/ILL |
| CF_3^- CF_3^- | | | | | | $\Delta_f H(AH) = -695 \pm 1$ $BDE(A-H) = 443 \pm 7$ | 78KUDI/KUD 86TSA |
| * | -648 ± 10^a | 1.84 ± 0.16^d | 1577 ± 9^g | 1545 ± 8 | IMRE | | 79BAR/SCO |
| * | | 2.82 ± 0.01 | | | PD | Vertical detachment energy | 75RIC/STE4 |
| | | 1.9 ± 0.3 | | | NBAP | From CF_4 | 78DIS/LAC |
| | | $> 2.0 \pm 0.2$ | | | EIAP | From C_3F_8 | 74HAR/FRA |
| | | $> 2.4 \pm 0.5$ | | | EIAP | From C_2F_6 | 74HAR/FRA |
| | | 2.2 ± 0.3 | | | EIAP | From CF_4 | 74HAR/FRA |
| -683 | | 2.2 | | | EIAP | From CF_4 | 74FRA/WAN |
| -651 | | 1.9 | | | EIAP | From CF_4 | 73WAN/MAR |
| | | 3.1 ± 0.2 | | | EIAP | From CF_4 | 72LIF/GRA |
| | | 2.1 ± 0.3 | | | EIAP | From C_2F_4 | 70THY/MAC2 |
| | | 1.8 ± 0.2 | | | EIAP | From CF_3OF | 70THY/MAC |
| | | 2.7 ± 0.2 | | | EIAP | From CF_4 | 70MAC/THY |
| | | 2.5 | | | EIAP | From $(CF_3)_2CO$ | 70HAR/THY |
| | | < 2.6 | | | EIAP | From C_2F_6 | 69MAC/THY |
| | | 2.0 | | | EIAP | From C_3F_8 | 69LIF/GRA |
| | | 3.3 | | | EIAP | From C_2F_6 | 63BIB/CAR |
| | | 2.0 ± 0.2 | | | SI | | 69PAG/GOO |
| | | | | 1539 ± 8 | IMRE ^o | | 79BAR/SCO |
| CF_3I^- CF_3I^- | | | | | | $\Delta_f H(A) = -590 \pm 21$ | 78KUDI/KUD |
| | | 1.6 ± 0.2 | | | NBIP | | 78COM/REI2 |
| | | 1.4 ± 0.2 | | | NBIP | | 76TAN/MAT |
| | | 2.2 ± 0.2 | | | NBIP | | 73MCN/LAC |
| CF_3N^- CF_3N^- | | | | | EIAP | From CF_3NC | 86HEN/ILL |
| CF_3NO^- CF_3NO^- | | | | | Est2 | $\Delta_f H(A) = -527$ | |
| | $< -720^b$ | $> 2.0 \pm 0.2$ | | | EIAP | From $(CF_3)_2NO$ | 77HAR |
| CF_3O^- $CF_2 = O \cdots F^-$ | | | | | | $\Delta_f H(AH) = -876 \pm 21$ $BDE(A-H) = 452 \pm 13$ | 79KLO/SEP 68CZA/CAS |
| * | -1062 ± 13^c | 4.35 ± 0.48^d | 178 ± 8^g | 142 ± 8 | IMRE | | 83LAR/MCM |
| | -1030 ± 17^c | | 146 ± 13 | | IMRE | | 78MCM/NOR |
| | -937 ± 25 | | | | IMRB | | 76KAR/KLE |
| | | $> 1.9 \pm 0.2$ | | | EIAP | From CF_3OOCF_3 | 72MAC/THY |
| | | 1.9 ± 0.1 | | | EIAP | From CF_3OF | 70THY/MAC |
| | | 1.3 | | | SI | | 69PAG/GOO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|------------------------|--|--|-------------------|--|-------------------------|
| $CF_3O_3S^-$ $CF_3SO_3^-$ | | 4.9±0.3 | | | IMRB EIAP | $\Delta_{acid}G: CF_3SO_3H < FSO_3H < H_2SO_4 < HPO_3 < HI, \Delta_{acid}G(CF_3SO_3H) < 1312 \text{ kJ}$ From $CF_3SO_3H_{anhydride}$ | 86VIG 86ADA/SMI |
| CF_3S^- CF_3S^- | | 1.8 | | | SI | | 69PAG/GOO |
| $CFeO^-$ $Fe(CO)^-$ • | | 1.260±0.022 | | | LPES EIAP | From $Fe(CO)_5$ | 79ENG/LIN2 76COM/STO |
| Cl^- Cl^- | | | | | | $\Delta_f H(A^-) = 552$ | 76REF/FRA2 |
| | <363 ^b | >2.0 | | | Endo | $I^- + CO \rightarrow$ | 77VOG/MIS |
| | 222±21 | 3.4±0.2 ⁱ | | | Endo | $I^- + CO \rightarrow$ | 76REF/FRA2 |
| ClO^- ICO^- | | | | | | | |
| | -412 | 3.1 | | | Endo | $I^- + COS \rightarrow$ | 76REF |
| ClO_2^- $CO_2 \cdot \cdot I^-$ • | | | | | | | |
| | -605±1 ^c | | 23 | 2 | TDA's | | 80KEE/LEE |
| CN^- CN^- • | | | | | | $\Delta_f H(AH) = 135$ $BDE(A-H) = 518 \pm 8$ | 82TN270 82MCM/GOL |
| | 74±9 ^a | 3.74±0.17 ^d | 1469±8 ^g | 1438±8 | IMRE | | 79BAR/SCO |
| | • | 3.82±0.02 | 1461±10 ^c | | PI | | 69BER/CHU |
| | 74 | | | | Endo | $I^- + (CN)_2 \rightarrow$ | 77REF/FRA |
| | -105±19 | 3.2±0.1 ⁱ | | | EIAP | From CH_3CN | 71DEC/BAF |
| | | 3.2 | | | SI | | 74CHA/PAG |
| | | 2.8 | | | SI | | 72PAG |
| | | 2.80±0.02 | | | SI | New DH(H-CN) used | 63NAP/PAG |
| | | | | 1447±8 | IMRE ^o | | 79BAR/SCO |
| CNO^- CNO^- | | | | | | | |
| | 142 | | | | EIAP | From $MeNO_2$ | 72DID/FRA |
| | 190±13 | | | | EIAP | From $MeNO_2$ | 69TSU/YOK2 |
| CNO^- NCO^- • | | | | | | $\Delta_f H(AH) = -105 \pm 12$ $BDE(A-H) = 477 \pm 26$ | 86SPI/PER 70OKA |
| | -192±21 ^a | 3.59±0.36 ^d | 1443±9 ^g | 1415±8 | IMRE | | 80WIG/BEA |
| | | >2.6±0.4 | | | EIAP | From PF_2NCO | 72THY |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|--|--|--|--|--|--|
| CNO ₄ ⁻ CO ₂ · · NO ₃ ⁻ • -740±2 ^c | | | 39 | 8±1 | TDA's | | 80KBE/LEE |
| CNS ⁻ SCN ⁻ | | 2.0 2.15±0.02 | 1375±25 ^g | 1343±21 | IMRB SI SI | $\Delta_f H(AH) = 128 \pm 2$ $BDE(A-H) = 464 \pm 6$ Acid: HNCS From (SCN) ₂ | 82TN270 63NAP/PAG 84BIE/GRA 72PAG 63NAP/PAG |
| CNSe ⁻ SeCN ⁻ | | 2.6 | | | SI | | 69PAG/GOO |
| CN ₂ ⁻ CN ₂ ⁻ · <649 | | | | | IMRB | O ⁻ + CH ₂ N ₂ or (CN) ₂ → | 79DAW/NOE |
| CN ₃ O ₆ ⁻ (NO ₂) ₃ C ⁻ | | 3.1 | | | EIAP | $\Delta_f H(AH) = -2 \pm 2$ From C(NO ₂) ₄ | 77PED/RYL 67JAE/HEN |
| CO ⁻ CO ⁻ · -243 | | 1.4 | | | EnCT | $\Delta_f H(A) = -110$ | 82TN270 76REF/FRA2 |
| COS ⁻ COS ⁻ · -187±20 ^b | | 0.5±0.2 >0.4 | | | NBIP ECD | $\Delta_f H(A) = -142 \pm 1$ | 77PED/RYL 75COM/REI 83CHE/WEN |
| CO ₂ ⁻ CO ₂ ⁻ · | | | | | | $\Delta_f H(A) = -394$ | 82TN270 75COM/REI |
| CO ₃ ⁻ CO · · O ₂ ⁻ >-210 ^c | | | <57 | | IMRB | CO · · O ₂ ⁻ + O ₂ → O ₄ ⁻ + CO | 70ADA/BOH |
| CO ₃ ⁻ CO ₂ · · O ⁻ · -503±6 ^c -502 | | 3.3±0.2 ⁱ >3.079 3.3±0.1 -469±12 ^c 2.7±0.1 >2.80 1.8±0.2 -469 ^c 3.1±0.2 | 218±4 183±10 174±10 183 | | PDis NBAP LPES PD PDis PD IMRE PDis PD PDis CIDT | From ethylene carbonate O ₃ ⁻ + CO ₂ = CO ₃ ⁻ + O ₂ | 80HIL/VES 83COM/REI 79NOV/ENG 77VES/MAU 77VES/MAU 77HON/WOO 77DOT/DAV 76MOS/COS 72BUR 79SMI/LEE 78TIE/WU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|------------------------------------|--|---|---|---|---|---|--|
| CO_3^- $CO_2 \cdots O^-$ | -455 ^c -520±23 ^c | | 169 234±21 ^g | 201±21 | PDIs IMRE | $O_3^- + CO_2 = CO_3^- + O_2$ | 78SMI/LEE 70ADA/BOH |
| CO_4^- $CO_2 \cdots O_2^-$ | -542±20 ^c -515±10 -510±9 ^c | | 106±19 | 51±5 | PDIs IMRE kine | $O_4^- + CO_2 = CO_4^- + O_2$ | 77VES/MAU 70ADA/BOH 66PAC/PHE2 |
| CO_5S^- $CO_2 \cdots SO_3^-$ | -981±17 ^c | | 27±1 | 1±1 | TDA's | | 80KEE/LEE |
| CS^- CS^- | 248 ^b | 0.205±0.021 > 1.6±0.3 | | | LPES EIAP | $\Delta_f H(A) = 268$ From COS | 79HUB/HER 82BUR/FEI 72THY |
| CS_2^- CS_2^- | 68±11 ^b | 0.51±0.10 0.53±0.11 0.895±0.200 0.6±0.1 1.0±0.2 0.5±0.2 0.9±0.3 | | | TDEq IMRE LPES ECD NBIP EnCT IMRB | $\Delta_f H(A) = 117±1$ Between NH_2^-, C^- | 77PED/RYL 87KEB/CHO 85GRI/CAL 86OAK/ELL 83CHE/WEN 75COM/REI 73HUG/LIF 61KRA/MUL |
| CH^- CH^- | 477±27 ^a 633 | 1.238±0.008 0.74±0.05 2.6±0.3 | 1616±18 ^c | 1588±20 ^h | LPES PD EIAP EIAP | $\Delta_f H(AH) = 390±8$ $BDE(A-H) = 423±18$ From CH_4, C_2H_2, C_2H_4 From CH_4 | 82TN270 79HUB/HER 75KAS/HER2 70FEL 70LOC/MOM 63TRE/NEU |
| $CHBrN^-$ $HCN \cdots Br^-$ | -145±9 ^c | | 67±8 | | Est | Extrapolated from other halide data | 84LAR/MCM3 |
| $CHClF_3^-$ $CHF_3 \cdots Cl^-$ | -992±12 ^c | | 70±10 ^g | 41±8 | IMRE | | 84LAR/MCM2 |
| $CHClN^-$ $HCN \cdots Cl^-$ | -180±10 ^c | | 88±8 ^g | 58±8 | IMRE | | 84LAR/MCM2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|----------------------|--|--|--------------|--|-------------------------------------|
| CHCl ₂ ⁻ CHCl ₂ ⁻ | -59±17 ^a | 1.7±0.2 ^d | 1567±16 ^g | 1535±13 | IMRB | $\Delta_f H(AH) = -96 \pm 1$ $BDE(A-H) = 422 \pm 5$ Comparable to DMSO | 77PED/RYL 83WEI/BEN 72BOH/LEE |
| CHCl ₂ F ₂ ⁻ CHF ₂ Cl · · Cl ⁻ | * -783±12 ^c | | 72±8 ^g | 43±8 | IMRE | | 84LAR/MCM2 |
| CHCl ₃ ⁻ CHCl ₃ ⁻ | -274±7 ^b | 1.76±0.05 | | | SI | $\Delta_f H(A) = -105 \pm 2$ | 77PED/RYL 66GAI/KAY |
| CHCl ₃ F ⁻ CHFC ₂ · · Cl ⁻ | * -582 ^c | | 74±8 ^g | 45±8 | IMRE | | 84LAR/MCM2 |
| CHCl ₃ Si ⁻ HCSiCl ₃ ⁻ | 142±42 | | | | EIAP | From MeSiCl ₃ | 68JAE/HEN |
| CHCl ₄ ⁻ CHCl ₃ · · Cl ⁻ | * -396±12 ^c | | 64±8 76±8 ^g | 45±8 47±8 | TDAs IMRE | | 71YAM/KEB 84LAR/MCM2 |
| | -412±6 ^c | | 80±3 | 49±4 43 | TDEq TDEq | | 74DOU/DAL 82FRE/IKU |
| CHF ⁻ HCF ⁻ | <116 | | | | IMRB | $\Delta_f H(A) = 109 \pm 12$ O ⁻ + CH ₃ F → | 85LIA/KAR 76DAW/JEN |
| CHFN ⁻ HCN · · F ⁻ | * -279±11 ^c | | 165±8 ^g | 138±8 | IMRE | | 83LAR/MCM |
| CHF ₂ ⁻ HCF ₂ ⁻ | -364±28 ^a | 1.3±0.3 ^d | 1618±28 ^g | 1586±25 | IMRB | $\Delta_f H(AH) = -453 \pm 1$ $BDE(A-H) = 432 \pm 4$ | 78KUD/KUD 83PIC/ROD 77SUL |
| CHF ₂ O ⁻ HCF=O · · F ⁻ | <-703 ^c | | >76 | | IMRB | FCO ⁻ + HCFO → | 77KAR/KLE |
| CHF ₄ ⁻ CF ₃ H · · F ⁻ | * -1057±12 ^c | | 113±8 ^g | 82±8 | IMRE | | 83LAR/MCM |
| CHN ⁻ HCN ⁻ | <38 ^b | >1.0 | | | EIAP | $\Delta_f H(A) = 135$ From CH ₃ CN | 82TN270 71TSU/YOK |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|---|-------------|--|--|--|--|--|
| $CHNO_2^-$ $CHNO_2^{\cdot -}$ <59 | | | | | IMRB | $O^- + CH_3NO_2 \rightarrow$ | 59HEN/MUC |
| CHN_2^- $CHN_2^{\cdot -}$ 262±38 ^a | | | 1561±22 ^g | 1527±17 | IMRB | $\Delta_f H(AH) = 230 \pm 17$ Near MeCN | 78VOG/WIL 83DEP/SCH |
| CHO^- HCO^- • 8±4 ^a | 0.313±0.005 | | 1646±3 ^c | 1613±5 ^h 1648±19 | LPES IMRB | $\Delta_f H(AH) = -109 \pm 1$ $BDE(A-H) = 364 \pm 3$ | 77PED/RYL 83MOO/SEI 86MUR/MIL 75KAR/KLE |
| CDO^- DCO^- | 0.301±0.005 | | | | LPES | | 86MUR/MIL |
| $CHOS^-$ $HCOS^-$ | | | | 1435±13 | IMRB | $\Delta_f H(AH) = -182 \pm 8$ | 85KAS/DEP 85KAS/DEP |
| CHO_2^- HCO_2^- • -464±13 ^a | 3.23±0.21 ^d | | 1444±12 ^g 1446±12 ^g | 1415±8 1416±8 | IMRE IMRE | $\Delta_f H(AH) = -379$ $BDE(A-H) = 444 \pm 8$ | 77PED/RYL 78CUM/KEB 81FUJ/MCI |
| CHO_3^- $CO_2 \cdot \cdot OH^-$ -897 ^c | | | 367 | | Endo | | 84HIE/PAU |
| CHO_3^- $HC(=O)OO^-$ -260±46 ^a | <1.1 | | 1551±4 ^g | <1523 | IMRB | Est2 $\Delta_f H(AH) = -280 \pm 42$ | 86BOW/DEP |
| CH_2^- $CH_2^{\cdot -}$ • 327±1 ^b | 0.652±0.006 0.670 0.210±0.015 0.208±0.031 <0.60±0.03 >0.9±0.4 <328±38 <290 | | 1712±2 ^c | 1679±3 ^h | LPES LPES LPES LPES PD EIAP EIAP EIAP | $\Delta_f H(AH) = 146 \pm 1$ $BDE(A-H) = 462 \pm 2$ Singlet-triplet splitting of $CH_2 = 37.7$ kJ Hot band problem Hot band problem, singlet-triplet splitting = 81.6 kJ From $CH_2 = CH_2$ From ketene From CH_4 | 81HEN/KNO 82TN270 85LEO/MUR 85LEO/MUR 81ENG/COR 76ZIT/ELL 77FEL/RAC 71THY/MAC 70COL/LOC 63TRE/NEU |
| CD_2^- $CD_2^{\cdot -}$ • | 0.645±0.006 1.043±0.010 | | | | LPES LPES | $\Delta_f H(A) = 390 \pm 1$ | 82TN270 85LEO/MUR 81ENG/COR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------------------------|---|---|--------------------|--|--------------------------------------|
| CH ₂ Br ⁻ CH ₂ Br ⁻ | | | | | | $\Delta_f H(AH) = -38 \pm 1$ $BDE(A-H) = 427 \pm 8$ | 84BIC/MIN 82MCM/GOL |
| * | 75 ± 18 ^a | 1.0 ± 0.3 ^d 1.9 | 1643 ± 16 ^g | 1614 ± 13 | IMRB SI | | 85ING/NIB2 69PAG/GOO |
| CH ₂ Cl ⁻ CH ₂ Cl ⁻ | | | | | | $\Delta_f H(AH) = -82 \pm 1$ $BDE(A-H) = 422 \pm 8$ | 79KUD/KUD 82MCM/GOL |
| * | 45 ± 16 ^a | 0.80 ± 0.24 ^d | 1657 ± 15 ^g | 1628 ± 13 1641 ± 17 | IMRE IMRB | | 85ING/NIB2 85HEN/HIE |
| CH ₂ ClO ₂ ⁻ HCO ₂ H · Cl ⁻ | | | | | | | |
| * | -721 ± 10 ^c | | 115 ± 8 107 ± 8 ^g 156 ± 8 | 84 ± 8 77 ± 8 106 ± 8 | TDA IMRE TDA | | 82FRE/IKU 84LAR/MCM2 71YAM/KEB |
| CH ₂ Cl ₃ ⁻ CH ₂ Cl ₂ · Cl ⁻ | | | | | | | |
| * | -389 ± 10 ^c | | 66 ± 8 ^g 65 ± 1 | 38 ± 8 37 ± 3 | IMRE TDEq | | 84LAR/MCM2 74DOU/DAL |
| CH ₂ FO ₂ ⁻ HCO ₂ H · F ⁻ | | | | | | | |
| * | -817 ± 11 ^c | | 190 ± 8 ^g | 159 ± 8 | IMRE | | 83LAR/MCM |
| CH ₂ I ⁻ CH ₂ I ⁻ | | | | | | $\Delta_f H(AH) = 15 \pm 1$ | 77PED/RYL |
| | 102 ± 25 ^a | | 1617 ± 24 ^g | 1587 ± 20 | IMRB | | 85ING/NIB2 |
| CH ₂ IO ₂ ⁻ HCO ₂ H · I ⁻ | | | | | | | |
| * | -646 ± 5 ^c | | 79 ± 4 | 53 ± 9 | TDA | | 84CAL/KEB |
| CH ₂ N ⁻ CH ₂ =N ⁻ | | | | | | $\Delta_f H(AH) = 135$ | 78DEF/HEH |
| | 230 ^a | 0.51 ± 0.07 | 1625 ± 22 ^g | 1594 ± 21 | IMRB | EA: between O ₂ and cyclooctatetraene | 85KAS/DEP |
| CH ₂ NO ⁻ CH ₂ =NO ⁻ | | | | | | Est2 $\Delta_f H(AH) = 29 \pm 13$ | |
| | < 56 | | < 1557 ^f | | EIAP | From MeNO ₂ | 72DID/FRA |
| CH ₂ NO ⁻ HN=CHO ⁻ | | | | | | $\Delta_f H(AH) = -186$ | 69BEN/CRU |
| * | -210 ^a | | 1506 ± 11 ^g | 1476 ± 8 | IMRE | | 86TAF |
| CH ₂ NO ⁻ HOH · CN ⁻ | | | | | | | |
| * | -225 ^c | | 58 ± 8 | 33 ± 8 | TDA | | 71PAY/YAM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|---|--|---|---|---|
| $CH_2NO_2^-$ $CH_2=NO_2^-$ • -114 ± 13^a | | | 1491 ± 12^g 1495 ± 12^g | 1463 ± 8 1467 ± 8 1467 ± 8 1473 ± 8 | IMRE IMRE IMRE EIAP IMRE ^o | $\Delta_f H(AH) = -75 \pm 1$ EA: $< NO_2$ From CH_3NO_2 | 77PED/RYL 79BAR/SCO 78CUM/KEB 78MAC/BOH 69TSU/YOK 79BAR/SCO |
| CH_2NS^- $CH_2=NS^-$ | | < 2.36 0.5 | | 1436 ± 15 | IMRB | | 85KAS/DEP |
| CH_2NS^- $H_2S \cdots CN^-$ • -29 ± 24^c | | | 83 ± 15^g | 52 ± 10 | IMRE | | 87LAR/MCM |
| $CH_2O_4^-$ $HOH \cdots CO_3^-$ • -793^c $> -906^c$ -792 | | 1.9 ± 0.2 | 48 ± 4^g < 161 | 28 ± 2 | IMRE PDis PD PDis | | 74FEH/FER 78SMI/LBE 72BUR 76COS/LIN |
| CH_2S^- $CH_2=S^-$ 56 ± 15^b | | 0.465 ± 0.023 | | | LPES | $\Delta_f H(A) = 100 \pm 13$ | 76BEN 87MOR/ELL |
| CH_3^- CH_3^- • 139 ± 8^a | 7.8 ± 0.030 < 0.5 $< 0.6^d$ 1.1 1.0 1.1 | | 1744 ± 7^c $> 1691^g$ | 1710 ± 7^h > 1657 | LPES PD IMRB SI SI SI | $\Delta_f H(A) = 147 \pm 5$ | 82MCM/GOL 78ELL/ENG 77FEL/RAC 72BOH/LEE 72PAG 69PAG/GOO 68GAI/PAG |
| $CH_3BF_3O^-$ $BF_3 \cdots MeO^-$ $< -1477 \pm 13^c$ | | | $> 92 \pm 8$ | | IMRB | $MeOH \cdots MeO^- + BF_3 \rightarrow$ | 73BLA/ISO |
| CH_3BrCl^- $MeBr \cdots Cl^-$ -311 ± 4^c | | | 46 ± 2 51 ± 13 | 30 ± 5 | TDA IMRB | Anchored: 84LAR/MCM | 74DOU/ROB 73RIV/BRE |
| $CH_3Br_2^-$ $MeBr \cdots Br^-$ -290 ± 4^c | | | 38 ± 2 | 21 ± 3 | TDA | | 74DOU/ROB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|-----------------------------|---|---|
| CH₃ClF⁻ MeF · Cl ⁻ • -522 ^c | | | 48±8 | | IMRE | | 84LAR/MCM2 |
| CH₃ClF₃Si⁻ MeSiF ₃ · Cl ⁻ • -1532 ^c | | | 67±8 ^g | 40±8 | IMRE | | 85LAR/MCM |
| CH₃ClI⁻ MeI · Cl ⁻ -253±3 ^c | | | 41±1 | 29±5 | TDA _s | | 74DOU/ROB |
| CH₃ClNO₂⁻ MeNO ₂ · Cl ⁻ -371±14 ^c | | | 68±13 | | IMRB | Anchored: 84LAR/MCM | 73RIV/BRE |
| CH₃Cl₂⁻ MeCl · Cl ⁻ • -360±10 ^c | | | 51±8 ^g 36±1 | 26±8 17±2 | IMRE TDA _s | | 84LAR/MCM2 74DOU/DAL |
| CH₃Cl₂Si⁻ MeSiCl ₂ ⁻ -105±21 | | | 1828±25 ^f | | EIAP | $\Delta_f H(AH) = -402 \pm 4$ From MeSiCl ₃ , probably ca. 300 kJ more stable | 77PEDI/RYL 68JAE/HEN |
| CH₃F₄Si⁻ MeSiF ₃ · F ⁻ • -1697 ^c | | | 211±8 ^g 257±21 | 180±8 | IMRE IMRB | | 85LAR/MCM 77MUR/BEA3 |
| CH₃I⁻ CH ₃ I ⁻ -13±20 ^b | 0.3±0.2 | | | | NBIP | $\Delta_f H(A) = 15 \pm 1$ Vertical EA | 77PEDI/RYL 74MOU/ATE |
| CH₃I₂⁻ MeI · I ⁻ -210±10 ^c | | | 38±8 | 17±1 | TDA _s | | 74DOU/ROB |
| CH₃NO₂⁻ CH ₃ NO ₂ ⁻ • -121±11 ^b | 0.48±0.10 0.49±0.11 0.45±0.05 0.4±0.2 | | | | TDEq IMRE ECD NBIP | $\Delta_f H(A) = -75 \pm 1$ | 77PEDI/RYL 87KEB/CHO 85GRI/CAL 83CHE/WEN 78COM/REI2 |
| CH₃N₂O⁻ HN=C(NH ₂)O ⁻ • -259±15 ^a | | | 1517±13 ^g | 1487±10 | IMRE | $\Delta_f H(AH) = -246 \pm 2$ | 77PEDI/RYL 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|-------------------|---|------------------------|
| CH_3O^- MeO ⁻ | | | | | | $\Delta_f H(AH) = -202$ $BDE(A-H) = 437 \pm 4$ | 77PED/RYL 82MCM/GOL |
| • | -139 ± 10^a | 1.62 ± 0.14^d | 1592 ± 9^g | 1565 ± 8 | IMRE | | 79BAR/SCO |
| • | | 1.570 ± 0.022 | 1597 ± 6^c | | LPES | | 78ENG/ELL |
| | | 1.59 ± 0.04 | | | PD | | 78JAN/ZIM |
| | | $< 1.59 \pm 0.04$ | | | PD | | 75REE/BRA |
| | | | 1595 ± 2 | 1569 ± 3 | TDEq | | 86MEO/SIE |
| | | 2.6 | | | EIAP | From MeOMe | 64TSU/HAM |
| | | 0.4 | | | SI | | 69PAG/GOO |
| | | | | 1559 ± 8 | IMRE ^o | | 79BAR/SCO |
| CD_3O^- CD ₃ O ⁻ | | | | | | | |
| | | 1.552 ± 0.022 | | | LPES | | 78ENG/ELL |
| $CH_3O_3^-$ HOH · · HCO ₂ ⁻ | | | | | | | |
| | -773^c | | 67 ± 4 | 38 ± 7 | TDA _s | | 86MEO/SIE2 |
| CH_3S^- MeS ⁻ | | | | | | $\Delta_f H(AH) = -23$ $BDE(A-H) = 364 \pm 9$ | 77PED/RYL 83SHU/BEN |
| • | -60 ± 13^a | 1.90 ± 0.22^d | 1493 ± 12^g | 1467 ± 8 | IMRE | | 79BAR/SCO |
| • | | 1.882 ± 0.022 | | | LPES | | 78ENG/ELL |
| | | 1.861 ± 0.004 | | | LPD | | 80JAN/REE |
| | | 1.861 ± 0.004 | | | LPD | | 80JAN/BRA |
| | | 1.4 | | | SI | | 69PAG/GOO |
| | | | | 1476 ± 8 | IMRE ^o | | 79BAR/SCO |
| CD_3S^- CD ₃ S ⁻ | | | | | | | |
| | | 1.858 ± 0.006 | | | LPD | | 80JAN/BRA |
| CH_4ClO^- MeOH · · Cl ⁻ | | | | | | | |
| • | -488 ± 10^c | | 59 ± 8 | 41 ± 8 | TDA _s | | 71YAM/KEB |
| | | | 70 ± 8^g | 41 ± 8 | IMRE | | 84LAR/MCM2 |
| | | | 73 ± 8 | 43 ± 11 | TDA _s | | 86YAM/FUR |
| | | | | 41 | TDEq | | 82FRE/KU |
| | | | 59 | 41 | TDA _s | | 73YAM/PAY |
| CH_4FO^- MeOH · · F ⁻ | | | | | | | |
| • | -574 ± 11^c | | 124 ± 8^g | 95 ± 8 | IMRE | | 83LAR/MCM |
| CH_4FS^- MeSH · · F ⁻ | | | | | | | |
| • | -415 ± 11^c | | 143 ± 8^g | 114 ± 8 | IMRE | | 83LAR/MCM |
| CH_4IO^- MeOH · · I ⁻ | | | | | | | |
| • | -437 ± 5^c | | 47 ± 4 | 25 ± 9 | TDA _s | | 84CAL/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|---|---|---|---|--|--|
| CH_4N^- MeNH ⁻ | * | 134 ± 5^a 0.45 ± 0.16^d < 1.6 | 1687 ± 5^E | 1656 ± 3 | IMRE EIAP | $\Delta_f H(AH) = -23$ $BDE(A-H) = 418 \pm 10$ From MeNH ₂ | 77PED/RYL 82MCM/GOL 76MAC/HEM 68COL/HUB |
| $CH_4O_3^-$ MeOH \cdots O ₂ ⁻ | * | -324 ± 5^c | 80 ± 4 | 52 ± 4 | TDA | | 73YAM/PAY |
| $CH_5O_2^-$ HOH \cdots MeO ⁻ | | -481^c | 100 ± 1 | 71 ± 1 | TDA | | 86MEO/SIE |
| C_2^- C ₂ ⁻ | * | 505 ± 2^b 3.391 ± 0.017 3.54 ± 0.05 $> 596 \pm 18^a$ $> 2.9 \pm 0.5$ 3.3 ± 0.2 $< 826 \pm 19$ > 2.9 4.0 | 1470 ± 7^c $> 1561 \pm 13^E$ | $> 1531 \pm 8$ | LPD PD IMRB EIAP EIAP EIAP EIAP | $\Delta_f H(AH) = 565 \pm 4$ $BDE(A-H) = 485 \pm 5$ From C ₂ H ₄ From C ₂ H ₂ , C ₂ H ₄ From ketene From C ₂ H ₄ From graphite | 82MCM/GOL 79HUB/HER 80JON/MEA 70FEL 75SCH/BOH 71THY/MAC 70LOC/MOM 70COL/LOC 63TRE/NEU 54HON |
| $C_2ClF_4O^-$ CF ₃ CF=O \cdots Cl ⁻ | * | -1339 ± 30^c | 70 ± 8^E | 42 ± 8 | IMRE | | 85LAR/MCM |
| $C_2Cl_2F_3O^-$ CF ₃ CCl=O \cdots Cl ⁻ | * | -1143 ± 30^c | 74 ± 8^E | 47 ± 8 | IMRE | | 85LAR/MCM |
| $C_2Cl_5^-$ C ₂ Cl ₅ ⁻ | | 1.5 | | | SI | Correct value probably 1 eV larger | 66GAI/KAY |
| $C_2Cl_6^-$ C ₂ Cl ₆ ⁻ | | 1.48 \pm 0.10 | | | SI | $\Delta_f H(A) = -150 \pm 5$ | 83KOL/PAP 66GAI/KAY |
| C_2F^- FC \equiv C ⁻ | | $> 3.4 \pm 0.8$ | $< 1536 \pm 98^c$ | $< 1504 \pm 100^h$ | EIAP | $\Delta_f H(AH) = 109$ $BDE(A-H) = 552 \pm 21$ From CH ₂ =CF ₂ | 80STA/VOG 71THY/MAC |
| $C_2F_2^-$ F ₂ C=C ⁻ | | < 15 -646 ± 58 1.7 \pm 0.2 | | | IMRB EIAP | O ⁻ + CH ₂ =CF ₂ \rightarrow From CF ₃ CHO | 76DAW/JEN 75HAR/THY |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|--|--|--|---|--|---|
| $C_2F_2O^-$ $F_2C=C=O^-$ | -156±58 | 2.4±0.6 | | | EIAP | From CF_3CHO | 75HAR/THY |
| $C_2F_3^-$ $C_2F_3^-$ | -391±19 | 2.1±0.2 ⁱ >1.6 | 1630±36 ^c | | EIAP | $\Delta_f H(AH) = -490 \pm 8$ $BDE(A-H) = 516 \pm 17$ From C_3F_8 | 77PED/RYL 83SPY/SAU |
| | -637±58 | | | | EIAP | From $CF_3CF=CFCF_3$ | 79SAU/CHR |
| | | 3.1±0.3 | | | EIAP | From CF_3CF_2CHO | 75HAR/THY |
| | -420±42 | 2.0±0.4 | | | EIAP | From CF_3CHO | 75HAR/THY |
| | | 2.0±0.4 | | | EIAP | From C_2F_4 | 72LIF/GRA |
| | | | | | EIAP | From C_2F_4 | 70THY/MAC2 |
| $C_2F_3O^-$ CF_3CO^- | >-707±29 ^a | <0.6 | >1623±17 ^c | | Est EIAP | $\Delta_f H(AH) = -800 \pm 13$ $BDE(A-H) = 368 \pm 17$ From $(CF_3)_2CO$ | 70HAR/THY |
| $C_2F_3O_2^-$ $CF_3CO_2^-$ | -1210±18 ^a | 4.20±0.27 ^d | 1351±17 ^g 1351±17 ^g | 1323±8 1324±8 | IMRE IMRE | $\Delta_f H(AH) = -1031 \pm 1$ $BDE(A-H) = 444 \pm 8$ | 77PED/RYL 78CUM/KEB 86TAF |
| $C_2F_3O_2^-$ $FCOCOF \cdot \cdot F^-$ | -1170±31 ^c | | 191±8 ^g | 155±8 | IMRE | | 85LAR/MCM |
| $C_2F_4N^-$ $CF_3CN \cdot \cdot F^-$ | -871±8 ^c | | 122±8 ^g | 92±8 | IMRE | | 85LAR/MCM |
| $C_2F_5^-$ $C_2F_5^-$ | -1067±23 ^a | 1.8±0.2 ^d 2.2±0.3 2.1±0.2 2.4 2.3 >3.3 >2.2±0.3 | 1567±17 ^g | 1535±13 | IMRB EIAP EIAP EIAP EIAP EIAP SI IMRB ^o | $\Delta_f H(AH) = -1105 \pm 6$ $BDE(A-H) = 430 \pm 2$ Between $tBuO^-$, F^- From $n-C_4F_{10}$ From C_3F_8 From C_2F_6 From C_3F_8 From C_3F_8 | 82MCM/GOL 82MCM/GOL 76SUL/BEA 73HAR/THY2 72HAR/THY 69MAC/THY 69LIF/GRA 63BIB/CAR 69PAG/GOO 76SUL/BEA |
| $C_2F_5O^-$ $CF_3CF=O \cdot \cdot F^-$ | -1481±31 ^c | | 191±8 ^g | 156±8 | IMRE | | 85LAR/MCM |
| $C_2FeO_2^-$ $Fe(CO)_2^-$ | -256±25 | 1.220±0.022 | | | LPES NBAP | From $Fe(CO)_5$ | 79ENG/LIN2 76COM/STO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------------|--|--|----------------------|--|-------------------------------------|
| C_2N^- C_2N^- | | $> 2.3 \pm 0.2$ | | | EIAP | From CH_3CN | 71TSU/YOK |
| C_2O^- CCO^- | | 1.848 ± 0.027 | | | LPES IMRB EIAP | $O^- + cis-CHF=CHF \rightarrow$ From ketene | 83OAK/JON 79DAW/NOE 70COL/LOC |
| $C_2O_3^-$ $C_2O_3^-$ | | | | | EIAP | From maleic anhydride | 73COO/COM |
| $C_2O_5^-$ $CO_2 \cdot CO_3^-$ | | | 30 | 3 ± 1 | TDA | | 80KEE/LEE |
| C_2H^- $HC \equiv C^-$ | | | | | | $\Delta_f H(AH) = 228 \pm 1$ $BDE(A-H) = 552 \pm 8$ | 77PED/RYL 85WOD/LEE |
| | 274 ± 10^a | 2.99 ± 0.19^d | 1576 ± 10^g | 1542 ± 8 | IMRE | | 79BAR/SCO |
| | | 2.940 ± 0.100 | 1585 ± 8 | 1546 ± 8 | LPD | Adiabatic EA: 3.18 ± 0.25 eV | 79JAN/BRA |
| | | | 1589 ± 2 | | TDEq | | 87MEO |
| | | 3.73 ± 0.05 | 1611 ± 4^g | 1577 ± 3 | PD | | 70FEL |
| | | | 1572 ± 38 | | TDEq | | 86MEO/SIE |
| | | | | | IMRE | | 74BOH/MAC |
| | | | | | Endo | | 73HUG/LIF |
| | | | | | | | 72BOH/LEE |
| | | $> 2.3 \pm 0.7$ | | | EIAP | From C_2H_4 | 71THY/MAC |
| | | 2.1 ± 0.3 | | | EIAP | From C_2H_2, C_2H_4 | 70LOC/MOM |
| | $< 515 \pm 19$ | | | | EIAP | From ketene | 70COL/LOC |
| | | > 2.8 | | | EIAP | From C_2H_4 | 63TRE/NEU |
| | | 2.6 | | | SI | | 69PAG/GOO |
| | | | | 1536 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_2HCIF_5^-$ $CF_3CF_2H \cdot Cl^-$ | | | 79 ± 8^g | 49 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_2HCIF_5O^-$ $CF_3OCF_2H \cdot Cl^-$ | | | | 51 ± 8 | IMRE | | 84LAR/MCM |
| $C_2HCl_2FN^-$ $CHCl_2F \cdot CN^-$ | | | 74 ± 15^g | 44 ± 10 | IMRE | | 87LAR/MCM |
| $C_2HCl_2O_2^-$ $CHCl_2CO_2^-$ | | | | | Est2 | $\Delta_f H(AH) = -427 \pm 17$ $BDE(A-H) = 444 \pm 8$ | |
| | -499 ± 28^a | 3.96 ± 0.20^d | 1374 ± 11^g | 1347 ± 8 | IMRE | | 78CUM/KEB |
| | -587 ± 28^a | | 1369 ± 11^g | 1342 ± 8 | IMRE | | 81FUJ/MCI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|------------------------------|--|---|
| $C_2HCl_3N^-$ $CHCl_3 \cdot \cdot CN^-$ • -106 ± 26^c | | | 76 ± 15^g | 45 ± 10 | IMRE | | 87LAR/MCM |
| $C_2HCl_3NO^-$ $HN=C(CCl_3)O^-$ • -329 ± 23^a | | | 1436 ± 11^g | 1406 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -235 \pm 13$ | 86TAF |
| C_2HF^- $FCH=C^-$ -139 | | | | | EIAP | $O^- + FCH=CH_2 \rightarrow$ | 76DAW/JEN |
| C_2HFN^- $CHFCN^-$ • | | | 1544 ± 11^g | 1513 ± 8 | IMRE | | 86TAF |
| $C_2HF_2O_2^-$ $CHF_2CO_2^-$ • -971 ± 29^a | 3.85 ± 0.21^d | | 1384 ± 12^g 1385 ± 12^g | 1354 ± 8 1355 ± 8 | IMRE IMRE | <i>Est2</i> $\Delta_f H(AH) = -824 \pm 17$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB 81FUJ/MCI |
| $C_2HF_3N^-$ $CHF_3 \cdot \cdot CN^-$ • -692 ± 25^c | | | 71 ± 15^g | 40 ± 10 | IMRE | | 87LAR/MCM |
| $C_2HF_3NO^-$ $HN=C(CF_3)O^-$ • -928 ± 23^a | | | 1438 ± 11^g | 1409 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -837 \pm 13$ | 86TAF |
| $C_2HF_4^-$ $F_2C=CFH \cdot \cdot F^-$ • -849 ± 19^c • -841 ± 36^c | | | 110 ± 8^g 102 ± 25 | 78 ± 8 | IMRE IMRB | | 83LAR/MCM 76SUL/BEA |
| $C_2HF_6^-$ $C_2F_5H \cdot \cdot F^-$ • -1480 ± 17^c | | | 127 ± 8^g | 94 ± 8 | IMRE | | 83LAR/MCM |
| $C_2HF_6O^-$ $CF_3OCF_2H \cdot \cdot F^-$ • | | | | 113 ± 8 | IMRE | | 84LAR/MCM |
| C_2HN^- $HCCN^-$ • < 422 • 309 ± 19 | 0.8 ± 0.4 > 1.1 | | 1569 ± 18^g | 1539 ± 13 | IMRB EIAP IMRB EIAP | Between H_2O_2 and mCl-toluene From CH_3CN $O^- + CH_3CN \rightarrow$ From CH_3CN | 87GRA/MEL 86HEN/ILL2 76DAW/JEN 71TSU/YOK |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|--|---|---|---|---|
| C_2HNO^- HCCNO $^-$ 502 | | | | | EIAP | From $CH_2 = CHNO_2$ | 72SHI/YAM |
| $C_2HN_2^-$ HCN \cdots CN $^-$ * 119 \pm 24 ^c 126 \pm 18 ^c | | | 91 \pm 15 ^g 84 \pm 8 | 57 \pm 10 | IMRE Est | | 87LAR/MCM 84LAR/MCM3 |
| C_2HO^- HC=CO $^-$ * * -51 \pm 13 ^a <-54 \pm 19 | 2.350 \pm 0.022 | | 1527 \pm 11 ^g | 1497 \pm 8 | D-EA LPES IMRE EIAP | $\Delta_f H(AH) = -48 \pm 3$ $BDE(A-H) = 441 \pm 9$ Acid: ketene From ketene | 77PED/RYL 83OAK/JON 83OAK/JON 70COL/LOC |
| $C_2H_2^-$ H $_2$ C=C $^-$ * 255 \pm 146 255 \pm 146 <0.4 | 0.470 \pm 0.020 | | | | LPES IMRB IMRB IMRB | $O^- + C_2H_4 \rightarrow C_2H_2^- + N_2O \rightarrow CH_2CN^-$ | 83BUR/STE 78DAW/NIB 76DAW/JEN 75LIN/ALB |
| $C_2D_2^-$ D $_2$ C=C $^-$ * 0.490 \pm 0.020 | | | | | LPES | | 83BUR/STE |
| $C_2H_2BrO_2^-$ BrCH $_2$ CO $_2^-$ * -528 \pm 19 ^a | 3.71 \pm 0.22 ^d | | 1397 \pm 13 ^g | 1370 \pm 8 | Est IMRE | $\Delta_f H(AH) = -395 \pm 6$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB |
| $C_2H_2ClF_4O^-$ (CF $_2$ H) $_2$ O \cdots Cl $^-$ * 71 \pm 8 | | | | | IMRE | | 84LAR/MCM |
| $C_2H_2ClO_2^-$ ClCH $_2$ CO $_2^-$ * -558 \pm 21 ^a | 3.61 \pm 0.21 ^d | | 1407 \pm 12 ^g 1407 \pm 12 ^g | 1376 \pm 8 1376 \pm 8 | IMRE IMRE | $\Delta_f H(AH) = -435 \pm 8$ $BDE(A-H) = 444 \pm 8$ | 77PED/RYL 78CUM/KEB 81FUJ/MCI |
| $C_2H_2Cl_2N^-$ CH $_2$ Cl $_2 \cdots$ CN $^-$ * -90 \pm 24 ^c | | | 68 \pm 15 ^g | 38 \pm 10 | IMRE | | 87LAR/MCM |
| $C_2H_2FO^-$ CH $_2 = CFO^-$ * -484 \pm 20 ^a * 2.22 \pm 0.09 | 2.4 \pm 0.3 ^d | | 148 \pm 8 ^k 148 \pm 8 ^g | 1460 \pm 15 115 \pm 8 1459 \pm 13 | IMRB PD IMRE IMRB ^o | $\Delta_f H(AH) = -444 \pm 3$ $BDE(A-H) = 406 \pm 8$ Between MeCOCH $_2$ F, cyclopentadiene | 77PED/RYL 80FAR/MCM 77ZIM/REE 83LAR/MCM 80FAR/MCM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|---------------------------|---|-------------------------|
| $C_2H_2FO_2^-$ $FCH_2CO_2^-$ | | | | | Est2 | $\Delta_f H(AH) = -586 \pm 8$ $BDE(A-H) = 444 \pm 8$ | |
| • | -700 ± 21^a | 3.52 ± 0.21^d | 1416 ± 12^g 1418 ± 12^g | 1385 ± 8 1386 ± 8 | IMRE IMRE | | 78CUM/KEB 81FUJ/MCI |
| $C_2H_2F_2O_2P^-$ $C_2H_2F_2O_2P^-$ <-1125 | | | | | IMRB | $CH_2 = CHO^- + PF_3O \rightarrow$ | 78SUL/BEA |
| $C_2H_2F_3^-$ $CF_2 = CH_2 \cdot F^-$ -697±17 | | | 112 ± 21^k | | IMRB | | 76SUL/BEA |
| $C_2H_2F_3O^-$ $CF_3CH_2O^-$ | | | | | | $\Delta_f H(AH) = -888 \pm 5$ $BDE(A-H) = 436 \pm 4$ | 77PED/RYL |
| • | -904 ± 20^a | 2.42 ± 0.20^d | 1514 ± 15^g | 1482 ± 8 1493 ± 8 | IMRE IMRE ^o | | 79BAR/SCO 79BAR/SCO |
| $C_2H_2F_3O_2S^-$ $CF_3SO_2CH_2^-$ | | | 1452 ± 11^g | 1422 ± 8 | IMRE | | 86TAF |
| $C_2H_2F_3O_3^-$ $HOH \cdot CF_3CO_2^-$ -1509 ^c | | | 57 ± 4 | 27 ± 7 | TDA _s | | 86MEO/SIE2 |
| $C_2H_2F_5O^-$ $(CHF_2)_2O \cdot F^-$ | | | 151 ± 8^g | 117 ± 8 | IMRE | | 83LAR/MCM |
| $C_2H_2N^-$ CH_2CN^- | | | | | | $\Delta_f H(AH) = 75 \pm 1$ $BDE(A-H) = 389 \pm 10$ | 83AN/MAN 82MCM/GOL |
| • | 105 ± 12^a | 1.46 ± 0.22^d | 1560 ± 11^g | 1528 ± 8 | IMRE | | 79BAR/SCO |
| • | | 1.543 ± 0.014 | 1562 ± 11^g | 1530 ± 8 | LPES IMRE | | 87MOR/ELL3 78CUM/KEB |
| | | 1.507 ± 0.018 | 1556 ± 12^c | 1523 ± 15^h | LPD | | 77ZIM/BRA |
| | | 1.560 ± 0.006 | | | LPD | | 86MAR/WET |
| | | | 1534 ± 19 | | EIAP | From CH_3CN | 86HEN/ILL2 |
| | 20 ± 19 | $> 1.6 \pm 0.2$ | | | EIAP | From CH_3CN , EtCN | 71TSU/YOK |
| | | | | 1525 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_2H_2N^-$ CH_2NC^- | | | | | | $\Delta_f H(AH) = 173 \pm 1$ | 77BAG/COL |
| | | 1.059 ± 0.024 | | | LPES | | 87MOR/ELL2 |
| $C_2D_2N^-$ CD_2CN^- | | | | | | | |
| | | 1.538 ± 0.012 | | | LPES | | 87MOR/ELL3 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|--|---|--|
| $C_2D_2N^-$ CD_2NC^- | | 1.070±0.024 | | | LPES | | 87MOR/BLL2 |
| $C_2H_2NO_2^-$ $H_2C=C=NO_2^-$ | | | 1515±19 <1563±3 ^g | <1531 | EIAP IMRB | From $CH_2=CHNO_2$ | 72SHI/YAM 80BAR |
| $C_2H_2O_2^-$ $O=CH-CH=O^-$ -272±25 | | | | | NBAP | $\Delta_f H(A) = -212 \pm 1$ From ethylene carbonate | 77PEDI/RYL 83COM/REI |
| $C_2H_3^-$ $C_2H_3^-$ * 221±9 ^a | 0.8±0.2 ^d >0.4 | 1699 | | >1661 | Bran IMRB IMRB | $\Delta_f H(AH) = 52$ $BDE(A-H) = 460 \pm 8$ | 77PEDI/RYL 82MCM/GOL 84DEP/BIE 75LIN/ALB 86FRO/FRE |
| $C_2H_3BrN^-$ $MeCN \cdots Br^-$ * -192±10 ^c | | | 54±8 | 33±8 | TDA _s | | 72YAM/KEB |
| $C_2H_3ClF_3^-$ $CF_2HCH_2F \cdots Cl^-$ * -1055±18 ^c | | | 79±8 ^g | 59±8 | IMRE | | 84LAR/MCM2 |
| $C_2H_3ClF_3O^-$ $CF_3CH_2OH \cdots Cl^-$ * -1216±15 ^c | | | 100±8 ^g | 69±8 | IMRE | | 84LAR/MCM2 |
| $C_2H_3ClN^-$ $MeCN \cdots Cl^-$ * -208±10 ^c | | | 56±8 57±8 44±8 ^g | 38±8 37±11 19±8 | TDA _s TDA _s IMRE | | 72YAM/KEB 86YAM/FUR 84LAR/MCM2 |
| $C_2H_3Cl_2O_2^-$ $ClCO_2Me \cdots Cl^-$ * | | | 59±8 ^g | 33±8 | IMRE | | 85LAR/MCM |
| $C_2H_3FN^-$ $MeCN \cdots F^-$ * -240±11 ^c | | | 67±8 | 50±8 | TDA _s | | 72YAM/KEB |
| $C_2H_3F_2^-$ $CHF=CH_2 \cdots F^-$ -453±27 ^c | | | 65±17 | | IMRB | | 76SUL/BEA |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|--|---|---|---|---|--|
| $C_2H_3F_2O^-$ $F_2CHCH_2O^-$ • -618 ± 16^a | | 2.23 ± 0.17^d | 1533 ± 12^g | 1503 ± 8 1505 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -620 \pm 4$ $BDE(A-H) = 436 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_2H_3F_4^-$ $CF_2HCH_2F \cdots F^-$ • -1108 ± 19^c | | | 111 ± 8^g | 79 ± 8 | IMRE | | 83LAR/MCM |
| $C_2H_3F_4O^-$ $CF_3CH_2OH \cdots F^-$ • -1300 ± 16^c | | | 164 ± 8^g | 130 ± 8 | IMRE | | 83LAR/MCM |
| $C_2H_3IN^-$ $MeCN \cdots I^-$ • -163 ± 10^c | | | 50 ± 8 | 27 ± 8 | TDA | | 72YAM/KEB |
| $C_2H_3NO_2^-$ $CH_2 = CHNO_2^-$ > 1.6 | | | | | IMRB | | 80BAR |
| $C_2H_3NO_2^-$ $MeCN \cdots O_2^-$ • -36 ± 6^c | | | 69 ± 4 | 47 ± 4 | TDA | | 73YAM/PAY |
| $C_2H_3O^-$ $CH_2 = CHO^-$ • -165 ± 13^a • • | | 1.817 ± 0.023 1.81 ± 0.06 | 1531 ± 12^g 1533 ± 12^g | 1502 ± 8 1505 ± 8 1505 ± 8 | D-EA IMRE LPES IMRE PD IMRE ^o | $\Delta_f H(AH) = -166$ $BDE(A-H) = 394 \pm 15$ | 77PEDI/RYL 79BAR/SCO 82ELL/ENG 78CUM/KEB 77ZIM/REE 79BAR/SCO |
| $C_2H_3O^-$ CH_3CO^- • -60 ± 11^a | | 0.4 ± 0.1^d | 1636 ± 11^g | 1604 ± 8 | IMRB | $\Delta_f H(AH) = -166$ $BDE(A-H) = 360 \pm 3$ | 77PEDI/RYL 82MCM/GOL 85DEP/BIE |
| $C_2D_3O^-$ $CD_2 = CDO^-$ • | | 1.817 ± 0.029 | | | LPES | | 82ELL/ENG |
| $C_2H_3O_2^-$ $HCO_2CH_2^-$ -249 ± 20^a | | | 1637 ± 19^g | 1607 ± 17 | IMRB | $\Delta_f H(AH) = -356 \pm 1$ | 77PEDI/RYL 85DEP/GRA |
| $C_2H_3O_2^-$ $MeCO_2^-$ • -504 ± 13^a | | 3.07 ± 0.21^d | 1459 ± 12^g 1457 ± 12^g 1459 ± 12^g | 1429 ± 8 1427 ± 8 1430 ± 8 | IMRE IMRE IMRE ECD EIAP | $\Delta_f H(AH) = -432$ $BDE(A-H) = 443 \pm 8$ | 78CHAI/ZWO 82MCM/GOL 78CUM/KEB 86TAF 81FUJ/MCI 68WEN/CHE 64TSU/HAM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|-----------------------------------|---|-------------------------------------|
| $C_2H_3O_4^-$ $HCO_2H \cdots HCO_2^-$ -997 ± 17^c | | | 154 ± 4 | 105 ± 7 | TDA's | | 86MEO/SIE2 |
| $C_2H_3Si^-$ $H_3SiC \equiv C^-$ < 322 | | | | | IMRB | $HC \equiv C^- + SiH_4 \rightarrow$ | 76PAY/TAN |
| $C_2H_4B_3^-$ $1,5-C_2B_3H_4^-$ | | | < 1795 | | EIAP | From closo-1,5- $C_2B_3H_5$ | 73ONA/HOW |
| $C_2H_4ClF_2^-$ $MeCHF_2 \cdots Cl^-$ * -787 ± 18^c | | | 62 ± 8^g | 34 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_2H_4ClF_3N^-$ $CF_3CH_2NH_2 \cdots Cl^-$ * -1004 ± 14^c | | | 75 ± 8^g | 45 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_2H_4ClO^-$ $MeCHO \cdots Cl^-$ * -453 ± 10^c | | | 60 ± 8^g | 33 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_2H_4ClO_2^-$ $MeCO_2H \cdots Cl^-$ * -750 ± 10^c | | | 90 ± 8 100 ± 8^g | 66 ± 8 70 ± 8 | TDA's IMRE | | 71YAM/KEB 84LAR/MCM2 |
| $C_2H_4Cl_2Si^-$ $HCSiMeCl_2^-$ 343±21 | | | | | EIAP | From Me_2SiCl_2 | 68JAE/HEN |
| $C_2H_4F^-$ $CH_2 = CH_2 \cdots F^-$ -221 ± 15^c | | | 25 ± 13 | | IMRB | Structure: 85ROY/MCM | 76SUL/BEA |
| $C_2H_4FO^-$ $FCH_2CH_2O^-$ -399 ± 25^a 2.1 ± 0.2^d | | | 1548 ± 16^g | 1521 ± 15 1527 ± 14 1520 ± 17 | IMRB IMRB IMRB ^o | <i>Est</i> $\Delta_f H(AH) = -417 \pm 8$ $BDE(A-H) = 436 \pm 4$ Between HF, acetone | 80CLA/MCM 77DAW/JEN 80CLA/MCM |
| $C_2H_4FO_2^-$ $MeCO_2H \cdots F^-$ * -865 ± 11^c | | | 185 ± 8^g | 153 ± 8 | IMRE | | 83LAR/MCM |
| $C_2H_4F_4N^-$ $CF_3CH_2NH_2 \cdots F^-$ * -1067 ± 19^c | | | 118 ± 8^g | 85 ± 8 | IMRE | | 83LAR/MCM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|-----------------------------------|---|---|
| $C_2H_4IO_2^-$ MeCO ₂ H · · I ⁻ • -691±5 ^c | | | 71±4 | 44±9 | TDA | | 84CAL/KEB |
| $C_2H_4N^-$ CH ₂ =NCH ₂ ⁻ 151±32 ^a | 0.8±0.3 | | 1610±23 ^g | 1582±21 | IMRB | $\Delta_f H(AH) = 71 \pm 8$ EA: between cyclooctatetraene, SO ₂ | 69BEN/CRU 85KAS/DEP |
| $C_2H_4NO^-$ HN=C(Me)O ⁻ • -339±12 ^a | | | 1429±11 ^g | 1400±8 | IMRE | $\Delta_f H(AH) = -238 \pm 1$ | 77PED/RYL 86TAF |
| $C_2H_4NO^-$ MeCH=NO ⁻ • -20±21 ^a | | | 1530±12 ^g | 1500±8 1503±8 | IMRE IMRE ^o | $\Delta_f H(AH) = -20 \pm 8$ | 69BEN/CRU 79BAR/SCO 79BAR/SCO |
| $C_2H_4NO^-$ MeN=CHO ⁻ • -809±15 ^a | | | 1508±11 ^g | 1479±8 | IMRE | Est2 $\Delta_f H(AH) = -787 \pm 4$ | 86TAF |
| $C_2H_4NO^-$ MeOH · · CN ⁻ • -196±24 ^c | | | 69±15 ^g | 38±10 | IMRE | | 87LAR/MCM |
| $C_2H_4NO_2^-$ H ₂ NCH ₂ CO ₂ ⁻ • -488±15 ^a | 3.35±0.19 ^d | | 1433±10 ^g | 1404±8 | IMRE | $\Delta_f H(AH) = -391 \pm 5$ BDE(A-H) = 444±8 | 77NGA/SAB 83LOC/MCI |
| $C_2H_4NO_2^-$ HN=C(OMe)O ⁻ • -433±15 ^a | | | 1514±11 ^g | 1485±8 | IMRE | Est $\Delta_f H(AH) = -417 \pm 4$ | 86TAF |
| $C_2H_4NO_2^-$ MeCH=NO ₂ ⁻ • -143±13 ^a | | | 1490±12 ^g 1496±12 ^g | 1462±8 1469±8 1472±8 | IMRE IMRE IMRE ^o | $\Delta_f H(AH) = -102$ | 77PED/RYL 79BAR/SCO 78CUM/KEB 79BAR/SCO |
| $C_2H_5^-$ MeCH ₂ ⁻ • 147±9 ^a | 1.0 0.9 | | 1761±8 | 1725±10 ^h | Bran SI SI | $\Delta_f H(AH) = -84$ BDE(A-H) = 421±2 | 74SCO 86BRO/LIG 84DEP/BIE 72PAG 69PAG/GOO |
| $C_2H_5B_4^-$ 1,2-C ₂ B ₄ H ₅ ⁻ | | | <1409±29 | | EIAP | From closo-1,2-C ₂ B ₄ H ₆ | 73ONA/HOW |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---|--|---|--|--|---|---|--|
| $C_2H_5B_4^-$ $1,6-C_2B_4H_5^-$ | | | <1891 | | ELAP | From closo-1,6- $C_2B_4H_6$ | 73ONA/HOW |
| $C_2H_5Br_2^-$ EtBr · Br ⁻ | * -324 ^c | | 49 | 25 | TDA _s | | 74DOU |
| $C_2H_5ClFO^-$ FCH ₂ CH ₂ OH · Cl ⁻ | * -730±18 ^c | | 86±8 ^g | 54±8 | IMRE | | 84LAR/MCM2 |
| $C_2H_5Cl_2^-$ EtCl · Cl ⁻ | -400±20 ^c | | 61±19 | | IMRB | Anchored: 84LAR/MCM | 73RIV/BRE |
| $C_2H_5Cl_2O^-$ ClCH ₂ CH ₂ OH · Cl ⁻ | * -579±14 ^c | | 90±8 ^g | 59±8 | IMRE | | 84LAR/MCM2 |
| $C_2H_5F_2O^-$ FCH ₂ CH ₂ OH · F ⁻ | * -811±19 ^c | | 146±8 ^g | 113±8 | IMRE | | 83LAR/MCM |
| $C_2H_5N^-$ EtN ⁻ | * 1.9±0.2 | | | | PD | | 74RIC/STE2 |
| $C_2H_5N_2O^-$ MeN(NO)CH ₂ ⁻ | * 113±19 ^a | | 1594±11 ^g | 1564±8 1567±8 | IMRE IMRE ^o | Est2 $\Delta_f H(AH) = 49±8$ | 85SING/NIB3 85SING/NIB3 |
| $C_2H_5O^-$ EtO ⁻ | * -186±10 ^a | 1.75±0.14 ^d 1.726±0.033 1.7±0.1 >1.7 0.6 | 1579±10 ^g 1582±8 ^e | 1551±8 | IMRE LPES EIAP EIAP SI IMRE ^o | $\Delta_f H(AH) = -235$ $BDE(A-H) = 436±4$ - From EtONO From EtOH | 77PED/RYL 82MCM/GOL 79BAR/SCO 82ELL/ENG 68WIL/HAM 63TRE/NEU 69PAG/GOO 79BAR/SCO |
| $C_2H_5O^-$ MeOCH ₂ ⁻ | * -11±9 ^a | | 1703±8 | 1666±12 ^h | Bran | $\Delta_f H(AH) = -184$ $BDE(A-H) = 389±4$ | 77PED/RYL 82MCM/GOL 84DEP/BIE |
| $C_2D_5O^-$ CD ₃ CD ₂ O ⁻ | * 1.702±0.033 | | | | LPES | | 82ELL/ENG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|-------------------|---|------------|
| $C_2H_5OS^-$ MeSOCH ₂ ⁻ | | | | | | $\Delta_f H(AH) = -151 \pm 1$ | 77PEDI/RYL |
| • | -119 ± 10^a | | 1563 ± 10^g | 1533 ± 8 | IMRE | | 79BAR/SCO |
| | | | 1566 ± 10^g | 1536 ± 8 | IMRE | | 78CUM/KEB |
| | | | | 1530 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_2H_5O_2S^-$ MeSO ₂ CH ₂ ⁻ | | | | | | $\Delta_f H(AH) = -373 \pm 3$ | 77PEDI/RYL |
| • | -373 ± 15^a | | 1531 ± 12^g | 1499 ± 8 | IMRE | | 79BAR/SCO |
| | -370 ± 15^a | | 1533 ± 12^g | 1502 ± 8 | IMRE | | 78CUM/KEB |
| | | | | 1502 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_2H_5O_3^-$ HOH · MeCO ₂ ⁻ | | | | | | | 86MEO/SIE2 |
| • | -813^c | | 67 ± 4 | 39 ± 7 | TDAs | | |
| $C_2H_5O_3^-$ MeOH · HCO ₂ ⁻ | | | | | | | 86MEO/SIE2 |
| • | -740 ± 17^c | | 74 ± 4 | 44 ± 7 | TDAs | | |
| $C_2H_5S^-$ EtS ⁻ | | | | | | $\Delta_f H(AH) = -46$ $BDE(A-H) = 364 \pm 9$ | 77PEDI/RYL |
| • | -90 ± 13^a | 1.97 ± 0.22^d | 1486 ± 12^g | 1460 ± 8 | IMRE | | 79BAR/SCO |
| • | | 1.953 ± 0.004 | 1488 ± 9^c | | LPD | | 80JAN/REE |
| | | 1.6 | | 1469 ± 8 | SI | | 69PAG/GOO |
| | | | | | IMRE ^o | | 79BAR/SCO |
| $C_2H_5S^-$ MeSCH ₂ ⁻ | | | | | | $\Delta_f H(AH) = -38$ | 77PEDI/RYL |
| • | 77 ± 11^a | | 1645 ± 11^g | 1615 ± 8 | IMRE | | 85SING/NIB |
| $C_2H_5Si^-$ MeSiCH ₂ ⁻ | | | | | | $\Delta_f H(AH) = 92 \pm 8$ | 86WAL |
| | 155 ± 32^a | | 1593 ± 23^g | 1565 ± 21 | IMRB | | 86DAM/DEP2 |
| $C_2H_6BF_2^-$ Me ₂ BF · F ⁻ | | | | | | | 77MUR/BEA2 |
| | -773^c | | 259 | | IMRB | F ⁻ A: Et ₃ B > Me ₂ BF > MeSiF ₃ > Me ₃ B > SF ₄ | |
| $C_2H_6BF_2O_2^-$ (MeO) ₂ BF · F ⁻ | | | | | | | 85LAR/MCM |
| • | | | 218 ± 21^g | 190 ± 21 | IMRE | | |
| $C_2H_6B_5^-$ 2,4-C ₂ B ₅ H ₆ ⁻ | | | | | | | 73ONA/HOW |
| | | | < 1891 | | EIAP | From closo-2,4-C ₂ B ₄ H ₇ | |
| $C_2H_6BrOS^-$ Me ₂ SO · Br ⁻ | | | | | | | 84MAG/CAL |
| • | -437 ± 6^c | | 72 ± 4 | 46 ± 9 | TDAs | | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|--------------------|---|--|
| $C_2H_6ClO^-$ EtOH \cdots Cl $^-$ • -535 ± 10^c | | | 72 ± 8^g | 44 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_2H_6ClOS^-$ Me $_2$ SO \cdots Cl $^-$ • -457 ± 6^c | | | 78 ± 4 | 52 ± 9 | TDA s | | 84MAG/CAL |
| $C_2H_6ClSi^-$ Me $_2$ SiCl $^-$ 67 ± 21 | | | | | EIAP | From Me $_2$ SiCl $_2$ | 68JAE/HEN |
| $C_2H_6FO^-$ EtOH \cdots F $^-$ • -615 ± 11^c | | | 132 ± 8^g | 101 ± 8 | IMRE | | 83LAR/MCM |
| $C_2H_6F_3Si^-$ Me $_2$ SiF $_2 \cdots$ F $^-$ | | | 232 ± 21 | | IMRB | F $^-$ A: SF $_4$ < Me $_2$ SiF $_2$ < Me $_3$ B | 77MUR/BEA3 |
| $C_2H_6IO^-$ EtOH \cdots I $^-$ • -474 ± 5^c | | | 51 ± 4 | 27 ± 9 | TDA s | | 84CAL/KEB |
| $C_2H_6IOS^-$ Me $_2$ SO \cdots I $^-$ • -405 ± 6^c | | | 66 ± 4 | 38 ± 9 | TDA s | | 84MAG/CAL |
| $C_2H_6N^-$ EtNH $^-$ • 93 ± 8^a | 0.66 ± 0.20^d | | 1671 ± 7^g | 1639 ± 3 | IMRE | $\Delta_f H(AH) = -48 \pm 1$ $BDE(A-H) = 423 \pm 13$ | 77PED/RYL 83MCM/GOL 76MAC/HEM |
| $C_2H_6N^-$ Me $_2$ N $^-$ • 109 ± 7^a | 0.39 ± 0.15^d 1.0 | | 1658 ± 6^g | 1628 ± 3 | IMRE SI | $\Delta_f H(AH) = -18$ $BDE(A-H) = 383 \pm 8$ | 77PED/RYL 82MCM/GOL 76MAC/HEM 69PAG/GOO |
| $C_2H_6O_4P^-$ (MeO) $_2$ PO $_2^-$ -1084 ± 149^a | | | 1463 ± 86^g | 1435 ± 84 | IMRB | Est2 $\Delta_f H(AH) = -1017 \pm 63$ | 80HOD/SUL |
| $C_2H_7O_2^-$ MeOH \cdots MeO $^-$ -461 ± 11^c -432 ± 18^c | | | 120 ± 1 91 ± 8 | 87 ± 2 64 ± 7 | TDA s TDA s | | 86MEO/SIE 84CAL/ROZ |
| The difference between 84CAL/ROZ and 86MEO/SIE2 has not been resolved. | | | | | | | |
| C_3^- C_3^- • | 1.981 ± 0.020 2.5 ± 1.0 | | | | LPES EIAP | From propene discharge From graphite | 86OAK/ELL 54HON |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|--|-------------|---|---|--------------------------------------|---|---|
| $C_3ClF_6O^-$ (CF_3) ₂ CO \cdots Cl $^-$ • -1720±30 ^c | | | 96±8 ^g | 68±8 | IMRE | | 85LAR/MCM |
| $C_3F_3^-$ $C_3F_3^-$ -425 -941 | | | | | EIAP EIAP | From $CF_2=CF-CF=CF_2$ From $CF_3CF=CF_2$ | 79SAU/CHR 72HAR/THY |
| $C_3F_3^-$ $CF_3C=C^-$ • -458±21 ^a | 3.92±0.31 ^d <5.6 | | 1486±9 ^g | 1454±8 | IMRE EIAP | $\Delta_f H(AH) = -414 \pm 13$ $BDE(A-H) = 552 \pm 21$ From $CF_3C=CCF_3$ | 86SMA 86TAF 79SAU/CHR |
| $C_3F_4O^-$ CF_3CFCO^- -926±58 | | | | | EIAP | From CF_3CF_2CHO | 75HAR/THY |
| $C_3F_5^-$ $C_3F_5^-$ -1052 -950±38 | 2.7±0.2 3.0 >2.7±0.2 2.6±0.4 2.7±0.1 | | | | EIAP EIAP EIAP EIAP EIAP | From $c-C_4F_8$ From $c-C_4F_8, 2-C_4F_8$ From $CF_3CF=CF_2$ From $CF_3CF=CF_2$ From $CF_3CF=CF_2$ | 72HAR/THY2 79SAU/CHR 72THY 72LIP/GRA 72HAR/THY |
| $C_3F_5O^-$ $CF_2=C(CF_3)O^-$ • -1318±42 ^a | 2.1±0.3 | | 1413±21 ^g | 1384±17 1356±10 | IMRB EIAP IMRB ^o | <i>Est</i> $\Delta_f H(AH) = -1201 \pm 21$ Between FCH_2CO_2H, HCl ; nearer to HCl From $(CF_3)_2CO$ | 80FAR/MCM 70HAR/THY 80FAR/MCM |
| $C_3F_6^-$ (CF_3) ₂ C $^-$ • -1181±17 ^a | 0.6 | | 1527±17 ^g | 1498±17 | IMRB EIAP | From $(CF_3)_2CO$ | 84MCD/CHO 70HAR/THY |
| $C_3F_6N^-$ $CF_3CF_2CN \cdots F^-$ • | | | 126±8 ^g | 97±8 | IMRE | | 83LAR/MCM |
| $C_3F_7^-$ (CF_3) ₂ CF $^-$ >2.7±0.2 >2.6±0.2 | | | | | EIAP EIAP | From $i-C_5F_{12}$ From $i-C_4F_{10}$ | 85SPY/HUN 83SPY/SAU |
| $C_3F_7^-$ $C_3F_7^-$ >3.4±0.3 >2.7±0.2 >2.6±0.1 >2.6±0.4 | | | | | EIAP EIAP EIAP EIAP | $\Delta_f H(A) = -1337 \pm 23$ From $neo-C_5F_{12}$ From $i-C_5F_{12}$ From $i-C_4F_{10}$ From C_3F_8 | 83EVA/WEE 85SPY/HUN 85SPY/HUN 85SPY/HUN 83SPY/SAU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|--|--|--|--|--|--|
| $C_3F_7^-$ $C_3F_7^-$ | | >2.8±0.1 >2.6±0.1 >2.5±0.4 >2.2±0.2 >2.3±0.2 >2.4 | | | EIAP EIAP EIAP EIAP EIAP EIAP | $\Delta_f H(A) = -1337 \pm 23$ From n-C ₆ F ₁₄ From n-C ₅ F ₁₂ From n-C ₄ F ₁₀ From n-C ₄ F ₁₀ From C ₃ F ₈ From C ₃ F ₈ | 83EVA/WEE 83SPY/SAU 83SPY/SAU 83SPY/SAU 73HAR/THY2 72HAR/THY 69LIF/GRA |
| $C_3F_7O^-$ (CF ₃) ₂ CO · · F ⁻ | | | | | | | |
| * | -1854±31 ^c | | 208±8 ^g | 174±8 | IMRE | | 85LAR/MCM |
| $C_3F_7O^-$ CF ₃ CF ₂ CFO · · F ⁻ | | | | | | | |
| * | -1919±31 ^c | | 197±8 ^g | 162±8 | IMRE | | 85LAR/MCM |
| $C_3FeO_3^-$ Fe(CO) ₃ ⁻ | | | | | | | |
| * | | 1.800±0.200 | | | LPES NBAP | From Fe(CO) ₅ | 79ENG/LIN2 76COM/STO |
| | | <-950 | | | | | |
| C_3N^- N≡CC=C ⁻ | | | | | | | |
| | 289 ^a -365±19 -512±21 318±29 | 4.11±0.32 ^d 2.4 2.4 | 1468±10 | 1438±10 | TDEq EIAP EIAP EIAP EIAP | $\Delta_f H(AH) = 351$ $BDE(A-H) = 552 \pm 21$ From CH ₂ =CHCN From TCNE From EtCN From HC≡C-C≡N | 85HAR 87MEO 86HEN/ILL2 72BRI/OLS 71TSU/YOK 61DIB/REE |
| C_3O^- C ₃ O ⁻ | | | | | | | |
| | | 1.340±0.150 | | | LPES | Large geometry change on detachment | 86OAK/ELL |
| $C_3O_2^-$ C ₃ O ₂ ⁻ | | | | | | | |
| | | 0.850±0.150 | | | LPES | | 86OAK/ELL |
| C_3H^- HC ₃ ⁻ | | | | | | | |
| | | 1.858±0.027 | | | LPES | From propene discharge | 86OAK/ELL |
| $C_3HClF_5O^-$ CF ₃ COCF ₂ H · · Cl ⁻ | | | | | | | |
| * | -1428 ^c | | | 68±8 | IMRE | | 84LAR/MCM |
| $C_3HCrO_3^-$ (CO) ₃ CrH ⁻ | | | | | | | |
| | <-287 | | | | IMRB | | 85LAN/SQU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|----------------------------------|---|--------------------------------------|
| $C_3HF_3^-$ $CF_3CH=C^-$ -614 | | | | | EIAP | $O^- + CF_3CH=CH_2 \rightarrow$ | 76DAW/JEN |
| $C_3HF_4O^-$ $CF_2=C(CHF_2)O^-$ • -1071±44 ^a | | | 1430±27 ^g | 1401±23 1400±25 | Est IMRB IMRB ^o | $\Delta_f H(AH) = -971 \pm 17$ Between HCO_2H , FCH_2CO_2H | 80FAR/MCM 80FAR/MCM |
| $C_3HF_5NO^-$ $CF_3OCF_2H \cdots CN^-$ • | | | 78±15 ^g | 47±10 | IMRE | | 87LAR/MCM |
| $C_3HF_6^-$ $(CF_3)_2CH^-$ • -1414±29 ^a 2.5±0.6 ^d | | | 1522±21 ^g | 1490±17 | IMRB | $\Delta_f H(AH) = -1406 \pm 8$ $BDE(A-H) = 452 \pm 33$ | 86KOLI/KOZ 84MCD/CHO 84MCD/CHO |
| $C_3HF_6O^-$ $(CF_3)_2CHO^-$ • -1623±19 ^a 3.19±0.16 ^d | | | 1443±11 ^g | 1415±8 1424 | Est IMRE IMRB | $\Delta_f H(AH) = -1536 \pm 8$ $BDE(A-H) = 438 \pm 4$ | 86TAF 81KOP/PIK |
| C_3HN^- $C=CHCN^-$ <402 | | | | | IMRB | $O^- + CH_2=CHCN \rightarrow$ | 76DAW/JEN |
| C_3HN^- $HC=C-CN^-$ 134±19 | | | | | EIAP | $\Delta_f H(A) = 351$ From $CH_2=CHCN$ | 85HAR 86HEN/ILL2 |
| $C_3HN_2^-$ $HC(CN)_2^-$ • 141±13 ^a | | | 1405±11 ^g 1406±11 ^g | 1373±8 1373±8 | IMRE IMRE | $\Delta_f H(AH) = 266 \pm 2$ | 77PED/RYL 81FUJ/MCI 78CUM/KEB |
| $C_3H_2^-$ $H_2C=C=C^-$ <191 | | 1.794±0.025 | | | LPES IMRB | From propene discharge $O^- + allene \rightarrow$ | 86OAK/BLL 76DAW/JEN |
| $C_3H_2Cl^-$ $ClCH_2C=C^-$ • 179±22 ^a | | | 1540±10 ^g | 1507±8 | Est IMRE | $\Delta_f H(AH) = 169 \pm 13$ $BDE(A-H) = 552 \pm 21$ | 86TAF |
| $C_3H_2ClF_4O^-$ $(CF_2H)_2CO \cdots Cl^-$ • -1198 ^c | | | | 76±8 | IMRE | | 84LAR/MCM |
| $C_3H_2F_3O^-$ $CH_2=C(CF_3)O^-$ • -880±22 ^a • 2.6±0.1 | | | 1461±10 ^g | 1431±8 | Est D-EA IMRE PD | $\Delta_f H(AH) = -811 \pm 13$ $BDE(A-H) = 398 \pm 22$ | 86TAF 77ZIM/RBE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|-----------------------------------|---|--|
| $C_3H_2F_3O^-$ $CH_2=C(CF_3)O^-$ | | | 1466 ± 15^g | 1436 ± 8 | Est D-EA IMRE | $\Delta_f H(AH) = -811 \pm 13$ $BDE(A-H) = 398 \pm 22$ | 78CUM/KEB |
| $C_3H_2F_3O_2^-$ $CF_3CH_2CO_2^-$ * -1215 ± 19^a | 3.68 ± 0.20^d | | 1401 ± 11^g | 1371 ± 8 | Est IMRE | $\Delta_f H(AH) = -1085 \pm 8$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_3H_2F_4NO^-$ $(CF_2H)_2O \cdots CN^-$ * | | | 92 ± 15^g | 63 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_2F_5O^-$ $CF_3CF_2CH_2O^-$ -1354 ± 33^a | 2.7 ± 0.4^d | | 1487 ± 30^g | 1459 ± 25 | IMRB | $\Delta_f H(AH) = -1310 \pm 3$ $BDE(A-H) = 435 \pm 8$ Between $(CF_3)_2CHOH$, CF_3CH_2OH | 77PED/RYL 77DAW/JEN |
| $C_3H_2F_7O^-$ $(CF_3)_2CHOH \cdots F^-$ * -1889 ± 19^c | | | 105 ± 8^g | 185 ± 8 | IMRE | | 83LAR/MCM |
| $C_3H_2N^-$ $CH_2=CCN^-$ * 207 ± 14^a | | | 1553 ± 12^g 1524 ± 19 | 1528 ± 8 1523 ± 8 | IMRE EIAP IMRE ^o | $\Delta_f H(AH) = 184 \pm 2$ From $CH_2=CHCN$ | 82CHU/NGU 80BAR 86HEN/ILL2 80BAR |
| $C_3H_2NO^-$ $CH_2=C(CN)O^-$ * -67^a | 2.87 ± 0.20^d | | 1441 ± 11^g | 1413 ± 8 1432 ± 21 | Est2 IMRE IMRB | $\Delta_f H(AH) = 22$ $BDE(A-H) = 406 \pm 8$ | 86TAF 68BRA/BLA |
| $C_3H_2NO_2^-$ $NCCH_2CO_2^-$ * -445 ± 32^a | 3.87 ± 0.20^d | | 1382 ± 11^g | 1354 ± 8 | Est2 IMRE | $\Delta_f H(AH) = -297 \pm 21$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_3H_2N_2^-$ pyrazolide ⁻ * | | | 1480 ± 11^g | 1449 ± 8 | IMRE | | 86TAF/ANV |
| $C_3H_3^-$ $CH_2=C=CH^-$ * 253 ± 12^a | 0.893 ± 0.026 2.3 | | 1592 ± 11^c | 1556 ± 13^h | LPES SI | $\Delta_f H(AH) = 191 \pm 1$ $BDE(A-H) = 367 \pm 8$ | 77PED/RYL 82MCM/GOL 83OAK/ELL 69PAG/GOO |
| $C_3H_3^-$ $MeC \equiv C^-$ * 251 ± 12^a * | 2.80 ± 0.32^d $> 2.602 \pm 0.043$ | | 1595 ± 10^g | 1562 ± 8 1556 ± 8 | IMRE LPES IMRE ^o | $\Delta_f H(AH) = 187 \pm 2$ $BDE(A-H) = 552 \pm 21$ | 77PED/RYL 79BAR/SCO 83OAK/ELL 79BAR/SCO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------------|--|--|---------------------------|---|------------------------|
| $C_3H_2D^-$ $CH_2=C=CD^-$ • 258 ± 25^a | | 0.880 ± 0.150 | 1601 ± 23^c | | LPES | $\Delta_f H(AH) = 187 \pm 2$ $BDE(A-H) = 374 \pm 8$ | 77PED/RYL 83OAK/ELL |
| $C_3HD_2^-$ $CD_2=C=CH^-$ • | | 0.907 ± 0.023 | | | LPES | | 83OAK/ELL |
| $C_3H_3F_2O^-$ $CF_2=C(Me)O^-$ > 1.0 ± 0.3 | | | | | EIAP | From CF_3COMe | 72THY |
| $C_3H_3F_2O^-$ $CHF=C(CH_2F)O^-$ • -625 ± 36^a | | | 1466 ± 15^E | 1436 ± 13 1433 ± 10 | IMRB IMRB ^o | Est $\Delta_f H(AH) = -561 \pm 21$ Between $PhCH_2CN$, CF_3COCH_3 | 80FAR/MCM 80FAR/MCM |
| $C_3H_3F_3N^-$ $CF_2HCH_2F \cdot CN^-$ • | | | 77 ± 15^E | 46 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_3F_3NO^-$ $CF_3CH_2OH \cdot CN^-$ • -916 ± 29^c | | | 103 ± 15^E | 69 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_3N^-$ $MeCCN^-$ < 401 | | | | | IMRB | $O^- + EtCN \rightarrow$ | 76DAW/JEN |
| $C_3H_3N_2^-$ $CH_2=CHCNN^-$ < -435 ± 21 | | | | | IMRB | $N_2O + CH_2=CHCH_2^- \rightarrow$ | 77BIE/DEP |
| $C_3H_3N_2^-$ $MeCN \cdot CN^-$ • 81 ± 24^c | | | 69 ± 15^E | 38 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_3N_2^-$ imidazolidine ⁻ • | | | 1465 ± 11^E | 1434 ± 8 | IMRE | | 86TAF/ANV |
| $C_3H_3N_3^-$ sym-triazine ⁻ 183^b | 0.5 | | | | ETS | $\Delta_f H(A) = 226 \pm 1$ | 82BYS 75NEN/SCH |
| $C_3H_4BrO_2^-$ $MeCHBrCO_2^-$ • -556 ± 19^a | | | 1407 ± 11^E | 1377 ± 8 | IMRE | Est2 $\Delta_f H(AH) = -432 \pm 8$ $BDE(A-H) = 444 \pm 8$ | 85CAL/MCM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|--|--|--|
| $C_3H_4ClO_2^-$ $ClCH_2CH_2CO_2^-$ • -585 ± 20^a | 3.41 ± 0.25^d | | 1426 ± 16^g | 1397 ± 8 | Est IMRE | $\Delta_f H(AH) = -481 \pm 4$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB |
| $C_3H_4ClO_2^-$ $MeCHClCO_2^-$ • -594 ± 22^a | 3.61 ± 0.19^d | | 1407 ± 10^g | 1380 ± 8 | Est IMRE | $\Delta_f H(AH) = -472 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB |
| $C_3H_4F^-$ $CH_2 = CFCH_2^-$ • | | | 1586 ± 14^g 1579 ± 10^g 63 ± 8^k | 1559 ± 13 1551 ± 8 1558 ± 17 1546 ± 13 | IMRB IMRB IMRB ^o IMRB ^o | | 84BAR/BUR 78MCM/NOR 84BAR/BUR 78MCM/NOR |
| $C_3H_4FO^-$ $CH_2 = C(CH_2F)O^-$ • -381 ± 41^a | 1.8 ± 0.3^d | | 1532 ± 21^g | 1503 ± 17 | Est IMRB | $\Delta_f H(AH) = -383 \pm 21$ $BDE(A-H) = 389 \pm 8$ | 80CLA/MCM |
| $C_3H_4FO^-$ $CHF = C(Me)O^-$ • -416 ± 39^a | | | 1497 ± 18^g | 1465 ± 15 1469 ± 10 | Est IMRB IMRB ^o | $\Delta_f H(AH) = -383 \pm 21$ Between pyrrole, MeNO ₂ | 80FAR/MCM 80FAR/MCM |
| $C_3H_4F_3O^-$ $CF_3CH(Me)O^-$ • -928 ± 19^a | | | 1507 ± 11^g | 1480 ± 8 1491 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -905 \pm 8$ $BDE(A-H) = 438 \pm 4$ | 85CAL/MCM 85CAL/MCM |
| $C_3H_4N^-$ $MeCHCN^-$ • 90 ± 11^a | 1.24 ± 0.16^d | | 1569 ± 11^g | 1537 ± 8 1532 ± 8 | IMRE IMRE ^o | $\Delta_f H(AH) = 51$ $BDE(A-H) = 377 \pm 4$ | 82CHU/NGU 82MCM/GOL 79BAR/SCO 79BAR/SCO |
| $C_3H_4NO^-$ $CH_2 = C(NO)CH_2^-$ | | | | 1586 ± 21 | IMRB | | 86KAS/FIL |
| $C_3H_4NO^-$ $CH_2 = CH-CH = NO^-$ | | | | 1504 ± 13 | IMRB | | 86KAS/FIL |
| $C_3H_4NO^-$ $MeOCHCN^-$ • -10 ± 23^a | | | 1556 ± 15^g | 1524 ± 8 1522 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -35 \pm 8$ | 79BAR/SCO 79BAR/SCO |
| $C_3H_4O^-$ $CH_2 = C(CH_2)O^-$ < 132 | | | | | IMRB | $O^- + Me_2CO \rightarrow$ | 79DAW/NOE2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|---------------------------|--|---|
| $C_3H_5^-$ $CH_2=CHCH_2^-$ | 125 ± 10^a | 0.41 ± 0.17^d | 1635 ± 10^g | 1607 ± 8 | IMRE | $\Delta_f H(AH) = 20$ $BDE(A-H) = 362 \pm 6$ | 77PED/RYL 79ROS/GOL 79BAR/SCO |
| • | | 0.362 ± 0.020 | | | LPES | | 84OAK/ELL |
| • | | 0.551 ± 0.052 | 1633 ± 4^g | 1605 ± 2 | LPD | | 77ZIM/BRA |
| • | | | | | IMRE | | 78MAC/LIE |
| $C_3H_5^-$ $CH_2=CMe^-$ | $> 184 \pm 3^a$ | | $> 1694 \pm 3^g$ | > 1661 | IMRB | $\Delta_f H(AH) = 20$ | 77PED/RYL 86FRO/FRE |
| $C_3H_5^-$ $MeCH=CH^-$ | $> 184 \pm 4^a$ | | $> 1694 \pm 4^g$ | > 1661 | IMRB | $\Delta_f H(AH) = 20$ | 77PED/RYL 86FRO/FRE |
| $C_3H_5^-$ cyclopropanide ⁻ | 247 ± 9^a $> 213 \pm 3^a$ | 0.3 ± 0.1^d | 1724 ± 8 $> 1690 \pm 3^g$ | 1687 ± 11^h > 1654 > 1654 | Bran IMRB IMRB | $\Delta_f H(AH) = 53 \pm 1$ $BDE(A-H) = 445 \pm 1$ | 77PED/RYL 82MCM/GOL 84DEP/BIE 72BOH/LBE 86FRO/FRE |
| $C_3H_4D^-$ $CH_2=CDCH_2^-$ | | 0.373 ± 0.020 | | | LPES | | 83OAK/ELL |
| $C_3D_5^-$ $CD_2=CDCD_2^-$ | | 0.380 ± 0.026 | | | LPES | | 83OAK/ELL |
| $C_3H_5ClNO^-$ $ClCH_2CH_2OH \cdot \cdot CN^-$ | -275 ± 28^c | | 88 ± 15^g | 56 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_5FNO^-$ $FCH_2CH_2OH \cdot \cdot CN^-$ | -428 ± 32^c | | 85 ± 15^g | 54 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_5F_2O^-$ $(FCH_2)_2CHO^-$ | -628 ± 25^a | | 1521 ± 21^g | 1492 ± 17 1498 ± 17 | IMRB IMRB ^o | Est $\Delta_f H(AH) = -620 \pm 4$ $BDE(A-H) = 436 \pm 4$ Between MeCHO, PhCOMe | 80CLA/MCM 80CLA/MCM |
| $C_3H_5F_2O^-$ $c-CH_2(O)CHCH_2F \cdot \cdot F^-$ | -610 ± 15^c | | 107 ± 8^g | 77 ± 8 | IMRE | | 83LAR/MCM |
| $C_3H_5N_2O_2^-$ $H_2NCON=C(Me)O^-$ | | | 1458 ± 12^g | 1427 ± 8 | IMRE | Est2 $\Delta_f H(AH) = -441 \pm 8$ Acid: acetylurea | 78CUM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|-------------------|--|-------------------------|
| $C_3H_5O^-$ $CH_2=C(Me)O^-$ | | | | | | $\Delta_f H(AH) = -217$ $BDE(A-H) = 411 \pm 11$ | 76CHAI/ZWO 70SOL/GOL |
| * | -203 ± 11^a | 1.86 ± 0.23^d | 1544 ± 11^g | 1514 ± 8 | IMRE | | 79BAR/SCO |
| * | | 1.757 ± 0.033 | | | LPES | | 82ELL/ENG |
| | | | 1546 ± 11^g | 1516 ± 8 | IMRE | | 78CUM/KEB |
| | | 1.76 ± 0.06 | | | PD | | 77ZIM/REE |
| | | | | 1513 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_3H_5O^-$ $MeCH=CHO^-$ | | | | | | $\Delta_f H(AH) = -187 \pm 2$ $BDE(A-H) = 372 \pm 12$ | 77PED/RYL |
| * | -189 ± 12^a | | 1528 ± 10^g | 1501 ± 8 | IMRE | | 79BAR/SCO |
| * | | 1.611 ± 0.023 | | | LPES | | 82ELL/ENG |
| | | | 1531 ± 10^g | 1504 ± 8 | IMRE | | 78CUM/KEB |
| | | 1.69 ± 0.06 | | | PD | | 77ZIM/REE |
| | | | | 1503 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_3H_5O_2^-$ $CH_2=C(OMe)O^-$ | | | | | | $\Delta_f H(AH) = -410 \pm 1$ $BDE(A-H) = 418 \pm 15$ | 77PED/RYL |
| * | -384 ± 10^a | | 1556 ± 10^g | 1528 ± 8 | IMRE | | 79BAR/SCO |
| * | | 1.80 ± 0.06 | | | PD | | 77ZIM/REE |
| | | | | 1524 ± 8 | IMRE ^o | | 79BAR/SCO |
| $C_3H_5O_2^-$ $EtCO_2^-$ | | | | | | $\Delta_f H(AH) = -448 \pm 2$ $BDE(A-H) = 445 \pm 8$ | 77PED/RYL 82MCM/GOL |
| * | -525 ± 14^a | 3.15 ± 0.21^d | 1454 ± 12^g | 1424 ± 8 | IMRE | | 78CUM/KEB |
| $C_3H_5O_3^-$ $MeOCH_2CO_2^-$ | | | | | Est2 | $\Delta_f H(AH) = -556 \pm 17$ $BDE(A-H) = 444 \pm 8$ | |
| * | -657 ± 28^a | 3.38 ± 0.20^d | 1429 ± 11^g | 1402 ± 8 | IMRE | | 86TAF |
| $C_3H_6ClF_2O^-$ $(FCH_2)_2CHOH \cdot Cl^-$ | | | | | | | |
| * | -946 ± 14^c | | 99 ± 8^g | 67 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_3H_6ClO^-$ $Me_2CO \cdot Cl^-$ | | | | | | | |
| * | -504 ± 10^c | | 59 ± 8^g | 34 ± 8 | IMRE | | 84LAR/MCM2 |
| | | | 57 ± 8 | 33 ± 8 | TDA _s | | 82FRE/IKU |
| $C_3H_6F_3O^-$ $(FCH_2)_2CHOH \cdot F^-$ | | | | | | | |
| * | -1026 ± 15^c | | 158 ± 8^g | 125 ± 8 | IMRE | | 83LAR/MCM |
| $C_3H_6IO_2^-$ $EtCO_2H \cdot I^-$ | | | | | | | |
| * | -706 ± 7^c | | 69 ± 4 | 44 ± 9 | TDA _s | | 84CAL/KEB |
| $C_3H_6NO^-$ $EtOH \cdot CN^-$ | | | | | | | |
| * | -233 ± 24^c | | 73 ± 15^g | 42 ± 10 | IMRE | | 87LAR/MCM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|-----------------------------------|---|--|
| $C_3H_6NO^-$ HCON(Me)CH ₂ ⁻ • -52±21 ^a | | | 1670±19 ^g | 1640±17 | IMRB | $\Delta_f H(AH) = -192 \pm 2$ | 77PED/RYL 85DEP/GRA |
| $C_3H_6NO^-$ Me ₂ C=NO ⁻ • -61±25 ^a | | | 1532±12 ^g | 1502±8 1505±8 | IMRE IMRE ^o | Est2 $\Delta_f H(AH) = -63 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_3H_6NO^-$ Me ₂ CO··CN ⁻ • -204±24 ^c | | | 62±15 ^g | 33±10 | IMRE | | 87LAR/MCM |
| $C_3H_6NO_2^-$ H ₂ NCH(Me)CO ₂ ⁻ • -519±14 ^a 3.42±0.19 ^d | | | 1425±10 ^g | 1396±8 | IMRE | $\Delta_f H(AH) = -414 \pm 4$ BDE(A-H) = 444±8 | 77NGA/SAB 83LOC/MCI |
| $C_3H_6NO_2^-$ HN=C(OEt)O ⁻ • -462±20 ^a | | | 1514±12 ^g | 1485±9 | IMRE | $\Delta_f H(AH) = -446 \pm 8$ | 75BER/BOU 86TAF |
| $C_3H_6NO_2^-$ Me ₂ C=NO ₂ ⁻ • -179±13 ^a | | | 1490±12 ^g 1491±12 ^g | 1464±8 1466±8 1474±8 | IMRE IMRE IMRE ^o | $\Delta_f H(AH) = -139 \pm 1$ | 77PED/RYL 79BAR/SCO 78CUM/KEB 79BAR/SCO |
| $C_3H_6NO_2^-$ MeNHCH ₂ CO ₂ ⁻ • -469±10 ^a 3.39±0.19 ^d | | | 1429±10 ^g | 1400±8 | IMRE | $\Delta_f H(AH) = -368 \pm 1$ BDE(A-H) = 444±8 | 77SAB/LAF 83LOC/MCI |
| $C_3H_6NS^-$ HCSN(Me)CH ₂ ⁻ • | | | 1587±11 ^g | 1558±8 1561±8 | IMRE IMRE ^o | | 85ING/NIB3 85ING/NIB3 |
| $C_3H_7^-$ Me ₂ CH ⁻ • 118±9 ^a | 0.7 | | 1753±8 | 1719±10 ^h | Bran SI | $\Delta_f H(AH) = -105$ BDE(A-H) = 398±4 | 74SCO 82MCM/GOL 84DEP/BIE 69PAG/GOO |
| $C_3H_7Br_2^-$ iPrBr··Br ⁻ | | | 51 | 26 | TDA's | | 74DOU |
| $C_3H_7Br_2^-$ nPrBr··Br ⁻ | | | 49 | 24 | TDA's | | 74DOU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|---|---|--|
| $C_3H_7ClSi^-$ HCSiMe ₂ Cl ⁻ -93±21 | | | | | EIAP | From Me ₃ SiCl | 68JAB/HEN |
| $C_3H_7N_2O^-$ HN=C(NMe ₂)O ⁻ * -259±25 ^a | | | 1514±13 ^g | 1484±10 | IMRE | Est2 $\Delta_f H(AH) = -243 \pm 13$ | 86TAF |
| $C_3H_7O^-$ iPrO ⁻ * -232±10 ^a * 1.839±0.029 1.7±0.1 >1.7 0.7 | 1.86±0.14 ^d | | 1571±10 ^g | 1543±8 | IMRE LPES EIAP EIAP SI IMRE ^o | $\Delta_f H(AH) = -273$ $BDE(A-H) = 438 \pm 4$ From iPrONO From iPrOH | 77PED/RYL 82MCM/GOL 79BAR/SCO 82ELL/ENG 68WIL/HAM 63TRE/NEU 69PAG/GOO 79BAR/SCO |
| $C_3H_7O^-$ nPrO ⁻ * -212±10 ^a * 1.789±0.033 1.9±0.1 >1.8 | 1.78±0.14 ^d | | 1573±9 ^g | 1546±8 | IMRE LPES EIAP EIAP IMRE ^o | $\Delta_f H(AH) = -255 \pm 1$ $BDE(A-H) = 433 \pm 4$ From nPrONO From nPrOH | 77PED/RYL 82MCM/GOL 79BAR/SCO 82ELL/ENG 68WIL/HAM 63TRE/NEU 79BAR/SCO |
| $C_3H_7O_2^-$ MeOCH ₂ CH ₂ O ⁻ * -332±16 ^a 1.90±0.17 ^d | | | 1564±12 ^g | 1535±8 1530±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = -366 \pm 4$ $BDE(A-H) = 436 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_3H_7S^-$ iPrS ⁻ * -128±13 ^a * 2.020±0.020 | 2.05±0.22 ^d | | 1479±12 ^g | 1452±8 1461±8 | IMRE LPD IMRE ^o | $\Delta_f H(AH) = -76 \pm 1$ $BDE(A-H) = 364 \pm 9$ | 77PED/RYL 79BAR/SCO 80JAN/RÉE 79BAR/SCO |
| $C_3H_7S^-$ nPrS ⁻ * -116±13 ^a * 2.000±0.020 | 2.02±0.22 ^d | | 1482±12 ^g | 1456±8 1465±8 | IMRE LPD IMRE ^o | $\Delta_f H(AH) = -68$ $BDE(A-H) = 364 \pm 9$ | 77PED/RYL 79BAR/SCO 80JAN/RÉE 79BAR/SCO |
| $C_3H_7Si^-$ CH ₂ =Si(Me)CH ₂ ⁻ * 104±41 ^a | | | 1613±25 ^g | 1586±21 | IMRB | $\Delta_f H(AH) = 21 \pm 17$ | 86WAL 86DAM/DEP |
| $C_3H_8B^-$ Me ₂ BCH ₂ ⁻ -120±39 ^a 1.8±0.5 ^d | | | 1532±29 ^g | 1502±25 1492±20 | IMRB IMRB ^o | $\Delta_f H(AH) = -123 \pm 10$ $BDE(A-H) = 397 \pm 21$ Between AsH ₃ , PH ₃ | 77PED/RYL 71BEL/PLA 76MUR/BEA 76MUR/BEA |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-----------------|--|--|--------|---|--------------------------------------|
| $C_3H_8ClO^-$ iPrOH \cdot Cl $^-$ | | | 74 ± 8^B | 45 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_3H_8ClO^-$ nPrOH \cdot Cl $^-$ | | | 74 ± 8^B | 45 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_3H_8FO^-$ iPrOH \cdot F $^-$ | | | 135 ± 8^B | 103 ± 8 | IMRE | | 83LAR/MCM |
| $C_3H_8FO^-$ nPrOH \cdot F $^-$ | | | 135 ± 8^B | 103 ± 8 | IMRE | | 83LAR/MCM |
| $C_3H_8IO^-$ iPrOH \cdot I $^-$ | | | 51 ± 4 | 27 ± 9 | TDA's | | 84CAL/KEB |
| $C_3H_8N^-$ Et(Me)N $^-$ | | | | | Est | $\Delta_f H(AH) = -46 \pm 4$ $BDE(A-H) = 383 \pm 8$ | 82MCM/GOL 85ING/NIB2 |
| $C_3H_8N^-$ Me $_2$ NCH $_2^-$ | | | $> 1700^B$ | > 1665 | IMRE | | 77PED/RYL 82MCM/GOL 78MAC/BOH2 |
| $C_3H_8N^-$ iPrNH $^-$ | | | | | | $\Delta_f H(AH) = -84 \pm 1$ $BDE(A-H) = 423 \pm 13$ | 77PED/RYL |
| | 49 ± 17^a | 0.8 ± 0.3^d | 1662 ± 16^B | 1631 ± 13 | IMRB | | 71BRA/BLA |
| $C_3H_8N^-$ nPrNH $^-$ | | | | | | $\Delta_f H(AH) = -70$ $BDE(A-H) = 423 \pm 13$ | 77PED/RYL |
| | 67 ± 17^a | 0.7 ± 0.3^d | 1667 ± 16^B | 1636 ± 13 | IMRB | | 71BRA/BLA |
| $C_3H_8NO^-$ iPrOH \cdot CN $^-$ | | | 76 ± 15^B | 45 ± 10 | IMRE | | 87LAR/MCM |
| $C_3H_8P^-$ Me $_2$ PCH $_2^-$ | | | | | | $\Delta_f H(AH) = -101 \pm 5$ | 77PED/RYL |
| | 5 ± 16^a | | 1636 ± 11^B | 1606 ± 8 | IMRE | | 85ING/NIB2 |
| $C_3H_9BF^-$ Me $_3$ B \cdot F $^-$ | | | 197 ± 8^B | 166 ± 8 | IMRE | | 85LAR/MCM |
| | -569 ± 21^c -616^c | | 245 | | IMRB | F $^-$ A: MeSiF $_3$ > Me $_3$ B > SF $_4$ | 77MUR/BEA2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-----------------|---|---|--------|---|--------------------|
| $C_3H_9BFO_3^-$ (MeO) ₃ B · F ⁻ | | | 176 ± 17^B | 142 ± 13 | IMRB | | 85LAR/MCM |
| $C_3H_9F_2Si^-$ Me ₃ SiF · F ⁻ | | | 160 ± 8^B | 132 ± 8 | IMRE | | 85LAR/MCM |
| | | | $< 226 \pm 42$ | | IMRB | | 77MUR/BEA3 |
| $C_3H_9O_2^-$ EtOH · MeO ⁻ | | | 85 ± 10^B | 57 ± 8 | IMRE | | 84CAL/ROZ |
| $C_3H_9Si^-$ Me ₃ Si ⁻ | -98 ± 23^a | 1.0 ± 0.3^d | 1595 ± 15^B | 1565 ± 13 | IMRB | $\Delta_f H(AH) = -163 \pm 8$ | 81WAL |
| | | | | | | $BDE(A-H) = 378 \pm 17$ | 81WAL 87THO/BAR |
| $C_3H_{10}NSi^-$ Me ₃ SiNH ⁻ | | | 1585 ± 15^B | 1552 ± 13 | IMRB | | 87THO/BAR |
| $C_4CoO_4^-$ Co(CO) ₄ ⁻ | | | | $< 1294 \pm 8$ | IMRB | | 87STE/BEA |
| $C_4F_4O_3^-$ tetrafluorosuccinic anhydride ⁻ | | 0.5 ± 0.2 | | | NBIP | | 74COO/COM |
| $C_4F_5^-$ C ₄ F ₅ ⁻ | -685 | 2.0 | | | EIAP | From $c-C_4F_6$ | 79SAU/CHR |
| | | | | | | | |
| $C_4F_6O^-$ CF ₃ CF ₂ CFCO ⁻ | | | -1331 ± 58 | | EIAP | From CF ₃ CF ₂ ĊF ₂ CHO | 75SHAR/THY |
| $C_4F_7^-$ C ₄ F ₇ ⁻ | -1457 ± 73^b | 3.0 ± 0.5 | | | EIAP | $\Delta_f H(A) = -1167 \pm 29$ | 83SPY/SAU |
| | | 0.9 ± 0.2 | | | | From $n-C_6F_{14}$ | 83SPY/SAU |
| | | 2.7 | | | | From CF ₃ CF = CFCF ₃ | 72LIF/GRA |
| | | | | | | SI | 69PAG/GOO |
| $C_4F_8^-$ C ₄ F ₈ ⁻ | | $> 0.7 \pm 0.4$ | | | EIAP | From $n-C_5F_{12}$ | 83SPY/SAU |
| $C_4F_8^-$ CF ₃ CF = CFCF ₃ ⁻ | $< -1670^b$ | $> 0.7 \pm 0.3$ | | | EnCT | $\Delta_f H(A) = -1602$ | 70BEN/O'N |
| | | | | | | | 73LIF/TIE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|---|---|---|----------------------|---|--------------------------------------|
| $C_4F_8^-$ $c-C_4F_8^-$ $< -1581 \pm 39^b > 0.4 \pm 0.3$ | | | | | EnCT | $\Delta_f H(A) = -1543 \pm 10$ | 77PED/RYL 73LIF/TIE |
| $C_4F_8N^-$ $CF_3CF_2CF_2CN \cdots F^-$ • | | | 129 ± 8^g | 99 ± 8 | IMRE | | 85LAR/MCM |
| $C_4F_9^-$ $(CF_3)_2CF_2CF_2^-$ | | 3.5 ± 0.5 3.5 ± 0.5 | | | EIAP EIAP | From $i-C_4F_{10}$ From $i-C_4F_{10}$ | 85SPY/HUN 83SPY/SAU |
| $C_4F_9^-$ $(CF_3)_3C^-$ | | 3.4 ± 0.1 3.4 ± 0.2 | | | Est EIAP EIAP | $\Delta_f H(A) = -1820$ From $(CF_3)_3CF$ From $(CF_3)_3CF$ | 85SPY/HUN 83SPY/SAU |
| $C_4F_9^-$ $C_4F_9^-$ | | $> 4.0 \pm 0.2$ $> 2.9 \pm 0.1$ 3.2 ± 0.3 | | | EIAP EIAP EIAP | From $n-C_4F_{10}$ From $n-C_5F_{12}$ From $n-C_4F_{10}$ | 83SPY/SAU 83SPY/SAU 73HAR/THY2 |
| $C_4F_9^-$ $CF_3CF_2CF(CF_3)^-$ $> 3.2 \pm 0.1$ | | | | | EIAP | From $i-C_5F_{12}$ | 85SPY/HUN |
| $C_4F_9O^-$ $(CF_3)_3CO^-$ • -2439 ± 33^a • -2442^a • -2451^a | 3.77 ± 0.21^d | | 1388 ± 12^g | 1356 ± 8 1352 ± 21 1345 ± 21 | IMRE IMRB IMRB | Est2 $\Delta_f H(AH) = -2297 \pm 21$ $BDE(A-H) = 439 \pm 8$ | 86TAF 81KOP/PIK 80CLA/MCM |
| $C_4FeO_4^-$ $Fe(CO)_4^-$ • -646 ± 52^b • 2.1 ± 0.3 | 2.398 ± 0.300 | | | | LPES EIAP | $\Delta_f H(A) = -414 \pm 23$ | 81SMI/LAI 79ENG/LIN2 76COM/STO |
| $Fe(CO)_5 + e^- \rightarrow Fe(CO)_4^- + CO$ "near thermoneutral". BDE from 81SMI/LAI | | | | | | | |
| C_4O^- C_4O^- | | 2.050 ± 0.150 | | | LPES | | 86OAK/ELL |
| $C_4HF_5NO^-$ $CF_3COCF_2H \cdots CN^-$ • -1234 ± 44^c | | | 108 ± 15^g | 75 ± 10 | IMRE | | 87LAR/MCM |
| $C_4HF_{10}O^-$ $(CF_3)_3COH \cdots F^-$ • -2617 ± 31^c | | | 71 ± 8^g | 151 ± 8 | IMRE | | 83LAR/MCM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|------------------------------|---|---|
| $C_4HFeO_4^-$ $Fe(CO)_4H^-$ | | | | 1313±23 | IMRB | | 87STE/BEA |
| $C_4H_2F_6NO^-$ $(CF_3)_2CHOH \cdots CN^-$ * -1566±32 ^c | | | 105±15 ^g | 70±10 | IMRE | | 87LAR/MCM |
| $C_4H_2F_7O^-$ $CF_3CF_2CF_2CH_2O^-$ -1626±51 ^a 2.9±0.4 ^d | | | 1465±30 ^g | 1437±25 | IMRB | Est2 $\Delta_f H(AH) = -1561 \pm 21$ $BDE(A-H) = 435 \pm 8$ Between $(CF_3)_2CHOH$, CF_3CH_2OH | 77DAW/JEN |
| $C_4H_2NO_2^-$ maleimide ⁻ | | | 1360±19 | | EIAP | Est2 $\Delta_f H(AH) = -287 \pm 8$ From maleimide | 73COO/COM |
| $C_4H_2N_2^-$ fumaronitrile ⁻ * 219 ^b | 1.24±0.10 1.25±0.09 0.8±0.1 | | | | TDEq TDEq SI | $\Delta_f H(A) = 339$ | 82CHU/NGU 87KEB/CHO 86CHO/KEB 67FAR/PAG |
| $C_4H_2O_3^-$ maleic anhydride ⁻ * -536±14 ^b | 1.44±0.10 1.41±0.11 1.38±0.05 1.4±0.2 | | | | TDEq IMRE IMRE NBIP | $\Delta_f H(A) = -397 \pm 4$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI 74COM/REI |
| $C_4H_3F_3NO^-$ $CH_3COCF_3 \cdots CN^-$ * -822±36 ^c | | | 85±15 ^g | 54±10 | IMRE | | 87LAR/MCM |
| $C_4H_3F_6O^-$ $(CF_3)_2C(Me)O^-$ * -1648±14 ^a | | | 1457±10 ^g | 1425±8 1431±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = -1576 \pm 4$ $BDE(A-H) = 440 \pm 4$ | 85CAL/MCM 85CAL/MCM |
| $C_4H_3N_2^-$ pyrimidinide ⁻ * 272±9 ^a | | | 1606±8 | 1569±8 | TDEq | $\Delta_f H(AH) = 197 \pm 1$ | 77PED/RYL 87MEO |
| $C_4H_3N_2O_3^-$ barbiturate ⁻ * -680±12 ^a | | | 1402±12 ^g | 1369±8 | IMRE | $\Delta_f H(AH) = -554 \pm 8$ Acid: barbituric acid | 72DOM 78CUM/KEB |
| $C_4H_3O^-$ $CH_2 = CHC=CO^-$ <35 | | | | | IMRB | $CH_2 = CHCH_2^- + CF_2 = O \rightarrow$ | 79DAW/NOE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|---------------|--|--|---|---|--|
| $C_4H_3O^-$ MeCOC \equiv C $^-$ • 44 ± 23^a | | | 1507 ± 10^B | 1474 ± 9 | Est2 IMRE | $\Delta_f H(AH) = 67 \pm 13$ BDE(A-H) = 552 ± 21 | 86TAF |
| $C_4H_3O_2^-$ MeOCOC \equiv C $^-$ • -171 ± 22^a | | | 1501 ± 10^B | 1469 ± 8 | Est2 IMRE | $\Delta_f H(AH) = -142 \pm 13$ BDE(A-H) = 552 ± 21 | 86TAF |
| $C_4H_4F_3^-$ CH $_2$ = C(CF $_3$)CH $_2^-$ • -615 ± 23^a | | | 1565 ± 10^B | 1537 ± 8 1532 ± 8 | Est IMRE IMRE O | $\Delta_f H(AH) = -649 \pm 13$ | 84BAR/BUR 84BAR/BUR |
| $C_4H_4F_3O_2S^-$ CF $_3$ SO $_2$ CH = CHCH $_2^-$ • -1023 ± 23^a | | | 1436 ± 11^B | 1407 ± 8 | Est2 IMRE | $\Delta_f H(AH) = -929 \pm 13$ | 86TAF |
| $C_4H_4F_7O^-$ (CF $_3$) $_2$ C(Me)OH $\cdot \cdot$ F $^-$ • -1933 ± 15^c | | | 109 ± 8^B | 189 ± 8 | IMRE | | 83LAR/MCM |
| $C_4H_4N^-$ CH $_2$ = C(CN)CH $_2^-$ • 151^a | | | 1551 ± 10^B | 1523 ± 8 1523 ± 8 | | $\Delta_f H(AH) = 130$ | 80WIL/BAE 84BAR/BUR 84BAR/BUR |
| $C_4H_4N^-$ CH $_2$ = CHCH = C = N $^-$ | | | | $< 1527 \pm 8$ | IMRB | Acid: CH $_2$ = CHCH $_2$ CN | 80DAW/NIB |
| $C_4H_4N^-$ c-(CH $_2$) $_2$ CCN $^-$ • 225 ± 13^a | | | 1571 ± 12^B | 1539 ± 8 1533 ± 8 | IMRE IMRE O | $\Delta_f H(AH) = 184 \pm 1$ Acid: cyanocyclopropane | 82FUC/HAL 79BAR/SCO 79BAR/SCO |
| $C_4H_4N^-$ pyrrolide $^-$ • 79 ± 13^a • 83 ± 13^a | | 2.4 ± 0.1 | 1501 ± 12^B 1505 ± 12^B | 1468 ± 8 1472 ± 8 1477 ± 8 | D-EA IMRE PD IMRE IMRE O | $\Delta_f H(AH) = 108$ BDE(A-H) = 419 ± 25 | 77PED/RYL 79BAR/SCO 75RIC/STE3 78CUM/KEB 79BAR/SCO |
| $C_4H_4NO_2^-$ succinimide $^-$ • -445 ± 18^a | | | 1445 ± 10^B 1379 ± 19 | 1414 ± 8 | IMRE EIAP | $\Delta_f H(AH) = -360 \pm 8$ | 69BEN/CRU 78CUM/KEB 73COO/COM |
| $C_4H_4NS^-$ 2-(thiofuryl)-NH $^-$ • | | | 1472 ± 11^B | 1441 ± 8 | IMRE | | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|---------------------------|-------------------------------|-------------------------------------|
| $C_4H_4N_2^-$ pyrazine ⁻ 157 ^b | | 0.4 | | | ETS | $\Delta_f H(A) = 196 \pm 1$ | 77PED/RYL 75NEN/SCH |
| $C_4H_4N_2^-$ pyridazine ⁻ 254 ^b | | 0.3 | | | ETS | $\Delta_f H(A) = 278 \pm 1$ | 77PED/RYL 75NEN/SCH |
| $C_4H_4N_2^-$ pyrimidine ⁻ | | 0.0 | | | ETS | $\Delta_f H(A) = 197 \pm 1$ | 77PED/RYL 75NEN/SCH |
| $C_4H_5ClN^-$ pyrrole ⁻ · Cl ⁻ * -198 ± 10 ^c | | | 79 ± 8 ^g | 49 ± 8 59 | IMRE TDEq | | 84LAR/MCM2 82FRE/IKU |
| $C_4H_5FN^-$ pyrrole ⁻ · F ⁻ * -283 ± 11 ^c | | | 143 ± 8 ^g | 111 ± 8 | IMRE | | 83LAR/MCM |
| $C_4H_5N^-$ EtCCN ⁻ < 381 | | | | | IMRB | O ⁻ + nPrCN → | 76DAW/JEN |
| $C_4H_5N_2^-$ 3-Me-pyrazolide ⁻ * | | | 1485 ± 11 ^g | 1452 ± 8 | IMRE | | 86TAF |
| $C_4H_5N_2^-$ 4-Me-pyrazolide ⁻ * | | | 1484 ± 11 ^g | 1454 ± 8 | IMRE | | 86TAF |
| $C_4H_5O^-$ CH ₂ = C(CH = CH ₂)O ⁻ * -148 ± 19 ^a | | | 1520 ± 11 ^g | 1492 ± 8 1500 ± 10 | IMRE IMRE ^o | $\Delta_f H(AH) = -138 \pm 8$ | 79VAJ/HAR 86BAR/KIP 86BAR/KIP |
| $C_4H_5O^-$ CH ₂ = C(CHO)CH ₂ ⁻ * | | | 1578 ± 16 ^g | 1549 ± 13 | IMRB | | 84BAR/BUR |
| $C_4H_5O^-$ CH ₂ = CHCH = CHO ⁻ * -149 ± 11 ^a | | | 1484 ± 10 ^g | 1456 ± 8 1466 ± 10 | IMRE IMRE ^o | $\Delta_f H(AH) = -104 \pm 2$ | 77PED/RYL 86BAR/KIP 86BAR/KIP |
| $C_4H_5O^-$ cyclobutanone enolate ⁻ * 1.84 ± 0.07 | | | | | Est2 PD | $\Delta_f H(AH) = -88 \pm 4$ | 78ZIM/JAC |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|----------------------------------|--|--|
| $C_4H_6BrO_2^-$ EtCHBrCO ₂ ⁻ • -600±23 ^a | | | 1407±11 ^g | 1378±8 | Est2 IMRE | $\Delta_f H(AH) = -477 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 85CAL/MCM |
| $C_4H_6ClO_2^-$ Cl(CH ₂) ₃ CO ₂ ⁻ • -586±20 ^a | 3.22±0.25 ^d | | 1445±16 ^g | 1416±8 | Est IMRE | $\Delta_f H(AH) = -501 \pm 4$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB |
| $C_4H_6ClO_2^-$ EtCHClCO ₂ ⁻ • -610±22 ^a | 3.56±0.19 ^d | | 1412±10 ^g | 1384±8 | Est IMRE | $\Delta_f H(AH) = -492 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB |
| $C_4H_6ClO_2^-$ MeCHClCH ₂ CO ₂ ⁻ • -616±20 ^a | 3.37±0.25 ^d | | 1431±16 ^g | 1401±8 | Est IMRE | $\Delta_f H(AH) = -516 \pm 4$ $BDE(A-H) = 444 \pm 8$ | 78CUM/KEB |
| $C_4H_6F_2NO^-$ (CH ₂ F) ₂ CHOH··CN ⁻ • -654±28 ^c | | | 109±15 ^g | 66±10 | IMRE | | 87LAR/MCM |
| $C_4H_6F_3O^-$ CF ₃ C(Me) ₂ O ⁻ • -928±14 ^a | 2.54±0.14 ^d | | 1507±10 ^g | 1479±8 1490±8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -905 \pm 4$ $BDE(A-H) = 440 \pm 4$ | 85CAL/MCM 85CAL/MCM |
| $C_4H_6N^-$ Me ₂ CCN ⁻ • 64±13 ^a | 1.08±0.21 ^d | | 1570±12 ^g | 1539±8 1534±8 | IMRE IMRE ^o | $\Delta_f H(AH) = 25 \pm 1$ $BDE(A-H) = 362 \pm 8$ | 77PED/RYL 82MCM/GOL 79BAR/SCO 79BAR/SCO |
| $C_4H_6NO_2^-$ MeCON=C(Me)O ⁻ • -509±19 ^a | | | 1451±15 ^g | 1422±8 | IMRE | $\Delta_f H(AH) = -430 \pm 4$ | 69BEN/CRU 78CUM/KEB |
| $C_4H_6NO_3^-$ HN=C(CO ₂ Et)O ⁻ • | | | 1472±11 ^g | 1442±8 | IMRE | | 86TAF |
| $C_4H_6O_2^-$ 2,3-butanedione ⁻ • -394±11 ^b | 0.69±0.10 0.70±0.11 1.1 | | | | TDEq IMRE ES | $\Delta_f H(A) = -327 \pm 1$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 66COM/CHR |
| $C_4H_7^-$ CH ₂ =C(Me)CH ₂ ⁻ • 86±11 ^a | 0.36±0.12 ^d | | 1633±10 ^g | 1602±9 | IMRE | $\Delta_f H(AH) = -17 \pm 1$ $BDE(A-H) = 356 \pm 1$ | 77PED/RYL 77LIA/AUS 84BAR/BUR |
| $C_4H_7O^-$ CH ₂ =C(Et)O ⁻ • -222±14 ^a | 1.75±0.06 | | 1549±14 ^e | 1520±18 ^h | PD | $\Delta_f H(AH) = -241$ $BDE(A-H) = 406 \pm 8$ | 77PED/RYL 77ZIM/REE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|---------------------------|--|---|
| $C_4H_7O^-$ $CH_2 = C(OMe)CH_2^-$ -60 ± 30^a | | | 1614 ± 26^g | 1586 ± 23 | Est IMRB | $\Delta_f H(AH) = -144 \pm 4$ | 84BAR/BUR |
| $C_4H_7O^-$ $EtCH = CHO^-$ * -206 ± 19^a | 1.67 ± 0.05 | | 1532 ± 18^c | 1504 ± 22^h | PD | $\Delta_f H(AH) = -208 \pm 2$ $BDE(A-H) = 381 \pm 13$ | 77PED/RYL 77ZIM/RBE |
| $C_4H_7O^-$ $MeCH = C(Me)O^-$ -231 ± 13^a * | 1.64 ± 0.19^d 1.67 ± 0.05 | | 1540 ± 12^g | 1512 ± 8 | IMRE PD | $\Delta_f H(AH) = -241$ $BDE(A-H) = 386 \pm 6$ | 77PED/RYL 82MCM/GOL 78CUM/KEB 77ZIM/RBE |
| $C_4H_7O_2^-$ $iPrCO_2^-$ * -562 ± 15^a | 3.17 ± 0.24^d | | 1449 ± 11^g | 1420 ± 8 | IMRE | $\Delta_f H(AH) = -482 \pm 4$ $BDE(A-H) = 444 \pm 13$ | 86TAF |
| $C_4H_7O_2^-$ $nPrCO_2^-$ * -553 ± 16^a | 3.17 ± 0.21^d | | 1450 ± 12^g | 1420 ± 8 | IMRE | $\Delta_f H(AH) = -473 \pm 4$ $BDE(A-H) = 444 \pm 8$ | 82BUT/FRA 78CUM/KEB |
| $C_4H_7O_4^-$ $MeCO_2H \cdot \cdot MeCO_2^-$ -1059 ± 17^c | | | 123 ± 4 | 85 ± 7 | TDA's | | 86MEO/SIE2 |
| $C_4H_8ClO^-$ $EtCOMe \cdot \cdot Cl^-$ * -530 ± 10^c | | | 62 ± 8^g | 36 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_4H_8IO_2^-$ $iPrCO_2H \cdot \cdot I^-$ * -740 ± 9^c | | | 70 ± 4 | 44 ± 9 | TDA's | | 84CAL/KEB |
| $C_4H_8NO^-$ $CH_2 = C(NMe_2)O^-$ * -196^a | | | 1569 ± 21^g | 1540 ± 8 1535 ± 8 | IMRE IMRE ^o | $\Delta_f H(AH) = -234$ | 78BEA/LEE 79BAR/SCO 79BAR/SCO |
| $C_4H_8NO^-$ $Me_2C(NO)CH_2^-$ 40 ± 34^a | | | 1613 ± 28^g | 1586 ± 25 | IMRB | $\Delta_f H(AH) = -43 \pm 6$ | 74CHO/MEN 80NOE/NIB |
| $C_4H_9^-$ Me_3C^- * 67 ± 9^a | 0.7 0.6 | | 1732 ± 8 | 1701 ± 10^h | Bran SI SI | $\Delta_f H(AH) = -135$ $BDE(A-H) = 390 \pm 8$ | 74SCO 82MCM/GOL 84DEP/BIE 72PAG 69PAG/GOO |
| $C_4H_9Br_2^-$ $iBuBr \cdot \cdot Br^-$ -374^c | | | 54 | 27 | TDA's | | 74DOU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|--|---|--|---|--|--|
| $C_4H_9Br_2^-$ tBuBr \cdots Br $^-$ | -397 ^c | | 52 | 28 | TDA _s | | 74DOU |
| $C_4H_9ClF^-$ tBuF \cdots Cl $^-$ | -614 \pm 18 ^c | | 56 \pm 8 ^B | 30 \pm 8 | IMRE | | 84LAR/MCM2 |
| $C_4H_9Cl_2^-$ tBuCl \cdots Cl $^-$ | -469 \pm 11 ^c | | 60 \pm 8 ^B | 33 \pm 8 | IMRE | | 84LAR/MCM2 |
| $C_4H_9F_2^-$ tBuF \cdots F $^-$ | -673 \pm 15 ^c | | 93 \pm 8 ^B | 64 \pm 8 | IMRE | | 83LAR/MCM |
| $C_4H_9O^-$ iBuO $^-$ | -246 \pm 11 ^a | 1.87 \pm 0.16 ^d | 1568 \pm 9 ^B | 1540 \pm 8 1535 \pm 8 | IMRE IMRE ^o | $\Delta_f H(AH) = -284 \pm 2$ $BDE(A-H) = 436 \pm 6$ | 77PEDI/RYL 79BAR/SCO 79BAR/SCO |
| $C_4H_9O^-$ nBuO $^-$ | -234 \pm 10 ^a | 1.78 \pm 0.15 ^d 1.9 \pm 0.1 0.9 | 1571 \pm 10 ^B 1569 \pm 12 | 1543 \pm 8 1541 \pm 13 ^h 1537 \pm 8 | IMRE CIDC EIAP SI IMRE ^o | $\Delta_f H(AH) = -275$ $BDE(A-H) = 431 \pm 5$ From nBuONO | 77PEDI/RYL 82MCM/GOL 79BAR/SCO 83BOA/HOU 68WIL/HAM 69PAG/GOO 79BAR/SCO |
| $C_4H_9O^-$ sBuO $^-$ | -259 \pm 10 ^a | 1.95 \pm 0.14 ^d | 1566 \pm 10 ^B 1565 \pm 11 | 1538 \pm 8 1538 \pm 13 ^h 1533 \pm 8 | IMRE CIDC IMRE ^o | $\Delta_f H(AH) = -295$ $BDE(A-H) = 441 \pm 4$ | 77PEDI/RYL 82MCM/GOL 86TAF 83BOA/HOU 79BAR/SCO |
| $C_4H_9O^-$ tBuO $^-$ | -275 \pm 12 ^a | 1.91 \pm 0.14 ^d 1.912 \pm 0.054 1.87 \pm 0.01 <1.87 \pm 0.04 | 1567 \pm 9 ^B | 1540 \pm 8 1534 \pm 8 | IMRE LPES PD PD IMRE ^o | $\Delta_f H(AH) = -313 \pm 3$ $BDE(A-H) = 440 \pm 4$ | 77PEDI/RYL 82MCM/GOL 79BAR/SCO 82BLL/ENG 78JAN/ZIM 75REE/BRA 79BAR/SCO |
| $C_4H_9O_3^-$ EtOH \cdots MeCO $_2^-$ | | | 87 \pm 4 | 50 \pm 7 | TDA _s | | 86MEO/SIE2 |
| $C_4H_9O_3^-$ HOH \cdots iPrCO $_2^-$ | | | 66 \pm 4 | 37 \pm 7 | TDA _s | | 86MEO/SIE2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|---------------------------------------|---|---|----------------------------------|--|--|
| $C_4H_9S^-$ iBuS ⁻ | * -149±13 ^a | 2.06±0.23 ^d | 1477±12 ^g | 1451±8 | IMRE | $\Delta_f H(AH) = -97 \pm 1$ $BDE(A-H) = 364 \pm 10$ | 77PED/RYL 86TAF |
| $C_4H_9S^-$ nBuS ⁻ | * -138±13 ^a * | 2.04±0.23 ^d 2.030±0.020 | 1480±12 ^g | 1454±8 | IMRE LPD | $\Delta_f H(AH) = -88 \pm 1$ $BDE(A-H) = 364 \pm 10$ | 77PED/RYL 86TAF 80JAN/REE |
| $C_4H_9S^-$ tBuS ⁻ | * -165±13 ^a * | 2.09±0.22 ^d 2.070±0.020 | 1475±12 ^g | 1449±8 1458±8 | IMRE LPD IMRE ^o | $\Delta_f H(AH) = -110 \pm 1$ $BDE(A-H) = 364 \pm 9$ | 77PED/RYL 79BAR/SCO 80JAN/REE 79BAR/SCO |
| $C_4H_{10}BF_2^-$ Et ₂ BF · F ⁻ | * -740±44 ^c -765 ^c | | 243±21 ^g 268 | 215±21 | IMRE IMRB | F ⁻ : iPr ₃ B > Et ₂ BF > Et ₃ B | 85LAR/MCM 77MUR/BEA2 |
| $C_4H_{10}ClO^-$ nBuOH · Cl ⁻ | * -576±10 ^c | | 74±8 ^g | 45±8 | IMRE | | 84LAR/MCM2 |
| $C_4H_{10}ClO^-$ tBuOH · Cl ⁻ | * -599±12 ^c | | 59±8 76±8 ^g | 46±8 46±8 | TDA IMRE | | 71YAM/KEB 84LAR/MCM2 |
| $C_4H_{10}FO^-$ nBuOH · F ⁻ | * -658±11 ^c | | 135±8 ^g | 103±8 | IMRE | | 83LAR/MCM |
| $C_4H_{10}FO^-$ tBuOH · F ⁻ | * -701±13 ^c | | 139±8 ^g | 107±8 | IMRE | | 83LAR/MCM |
| $C_4H_{10}IO^-$ tBuOH · I ⁻ | * | | 51±4 | 27±9 | TDA | | 84CAL/KEB |
| $C_4H_{10}NO^-$ Et ₂ NO ⁻ | * -15±23 ^a | 0.54±0.20 ^d | 1551±11 ^g | 1523±8 1520±8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -36 \pm 13$ $BDE(A-H) = 291 \pm 8$ | 78CAC/LIS 83BAR/BAS 83BAR/BAS |
| $C_4H_{11}O_2^-$ EtOH · EtO ⁻ | -507±21 ^c | | 115±4 86±10 ^g | 82±7 59±7 | TDEq IMRE | | 86MEO/SIE2 84CAL/ROZ |

The difference between 84CAL/ROZ and 86MEO/SIE2 has not been resolved.

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|---|--|--|----------------------|---|--|
| $C_4H_{11}O_2^-$ MeOH ··· nPrO ⁻ | | | 83 ± 10^B | 55 ± 8 | IMRE | | 84CAL/ROZ |
| $C_4H_{11}Si^-$ Me ₃ SiCH ₂ ⁻ | | 0.7 ± 0.3^d | 1661 ± 23^B | 1635 ± 21 | IMRB | $\Delta_f H(AH) = -233 \pm 3$ $BDE(A-H) = 415 \pm 8$ | 83STE 83STE2 84DEP/DAM |
| $C_4H_{12}FSi^-$ Me ₄ Si ··· F ⁻ | | | 125 ± 8^B | 99 ± 8 | IMRE | | 85LAR/MCM |
| $C_5ClFeO_5^-$ Fe(CO) ₅ ··· Cl ⁻ | | | 58 ± 13^B | 33 ± 13 | IMRB | | 85LAN/SAL |
| $C_5CrO_5^-$ Cr(CO) ₅ ⁻ | | > 2.3 | | | IMRB | | 85SAL/LAN |
| $C_5FFeO_5^-$ Fe(CO) ₅ ··· F ⁻ | | | 171 ± 8^B | 144 ± 8 | IMRE | | 85LAN/SAL |
| $C_5F_6O_3^-$ hexafluoroglutaric anhydride ⁻ | | 1.5 ± 0.2 | | | NBIP | | 74COO/COM |
| $C_5F_9^-$ C ₅ F ₉ ⁻ | | 4.6 ± 0.5 $> 3.1 \pm 0.3$ 3.1 | | | ELAP ELAP ELAP | $\Delta_f H(A) = -1573 \pm 29$ From n-C ₆ F ₁₄ From c-C ₄ F ₆ (CF ₃) ₂ From c-C ₄ F ₆ (CF ₃) ₂ | 83SPY/SAU 83SPY/SAU 72THY 70LIF/PEE |
| $C_5F_9O_2^-$ FCOCF ₂ CF ₂ CF ₂ CFO ··· F ⁻ | | | | 192 ± 19 | IMRE | | 84LAR/MCM |
| $C_5F_{10}^-$ C ₅ F ₁₀ ⁻ | | $< -2508 \pm 64^b$ $> 5.2 \pm 0.5$ | | | ELAP | $\Delta_f H(A) = -2007 \pm 21$ From n-C ₅ F ₁₂ | 83SPY/SAU 83SPY/SAU |
| $C_5F_{11}^-$ (CF ₃) ₃ CCF ₂ ⁻ | | 4.7 ± 0.3 | | | ELAP | From neo-C ₅ F ₁₂ | 85SPY/HUN |
| $C_5F_{11}^-$ C ₂ F ₅ (CF ₃) ₂ C ⁻ | | $> 4.2 \pm 0.3$ | | | ELAP | From i-C ₅ F ₁₂ | 85SPY/HUN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-----------------|--|--|-----------------------------------|--|--|
| $C_5F_{11}^-$ $C_5F_{11}^-$ | | $>4.5 \pm 0.2$ | | | EIAP | From n- C_5F_{12} | 83SPY/SAU |
| $C_5MnO_5^-$ $Mn(CO)_5^-$ | | | | 1309 ± 17 | IMRB | $\Delta_f H(A) = -740 \pm 10$ | 82CON/ZAF 87STE/BEA |
| C_5N^- $N \equiv CC \equiv CC \equiv C^-$ 682 | 2.3 | | 1643 ± 21^c | | Est2 EIAP | $\Delta_f H(AH) = 577 \pm 21$ $BDE(A-H) = 552 \pm 21$ From $HC \equiv C-(C \equiv C)_2-C \equiv N$ | 61DIB/REE |
| $C_5N_3^-$ $(NC)_2C = CCN^-$ | | 3.8 ± 0.5 | | | EIAP | From tetracyanoethylene | 72BRI/OLS |
| $C_5HFeO_5^-$ $Fe(CO)_5 \cdot H^-$ -858 ^c | | | 235 ± 13 | | IMRB | | 85LAN/SAL |
| $C_5HFeO_6^-$ $Fe(CO)_5 \cdot OH^-$ -1142 ^c <-1075 | | | 237 ± 17^g | 196 ± 17 | IMRB IMRB | | 85LAN/SAL 84LAN/LEE |
| $C_5H_3F_2^-$ difluorocyclopentadienide ⁻ <9 | | | | | IMRB | $CH_2 = CHCH_2^- + C_2F_4 \rightarrow$ | 79DAW/NOE |
| $C_5H_4^-$ cyclopentadienylide ⁻ 274 ± 27^a < 243 ± 19 | | | 1587 ± 16^g | 1556 ± 13 1546 ± 13 | IMRB EIAP IMRB ^o | $\Delta_f H(AH) = 217 \pm 10$ $D-EA$ $BDE(A-H) = 466 \pm 46$ | 82MCM/GOL 80MCD/CHO 72DID/HAR 80MCD/CHO |
| $C_5H_4F_3O_2^-$ $CF_3COCH = C(Me)O^-$ | | | 1374 ± 17^g 1374 ± 17^g | 1347 ± 8 1348 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -1003 \pm 4$ | 84ERA/KOL 81FUJ/MCI 78CUM/KEB |
| $C_5H_4F_6NO^-$ $(CF_3)_2C(Me)OH \cdot \cdot CN^-$ * -1609 ± 28^c | | | 108 ± 15^g | 74 ± 10 | IMRE | | 87LAR/MCM |
| $C_5H_4N^-$ pyridinide ⁻ * 250 ± 3^a | | 2.41 ± 0.03 | 1640 ± 2 | 1602 ± 2 < 1574 ± 8 | TDEq IMRB SI | $\Delta_f H(AH) = 140 \pm 1$ O^- deprotonates | 79KUD/KUD 87MEO 78BRU/FER 76FAI/JOY |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|--|--|---|
| $C_5H_4O^-$ $C_5H_4O^-$ 84±21 | | | | | Est2 NBAP | $\Delta_f H(A) = -35 \pm 35$ From benzoquinone. Possibly cyclopentadienone ⁻ ? | 75COO/NAF |
| $C_5H_5^-$ cyclopentadienide ⁻ • 82±16 ^a • 1.67±0.21 ^d • 1.786±0.020 • 1.839±0.030 79±8 <2.2±0.3 | | | 1481±12 ^g 1485±12 ^g | 1455±8 1459±8 1464±8 | IMRE LPES IMRE LPD EIAP IMRE ^o | $\Delta_f H(AH) = 131 \pm 4$ $BDE(A-H) = 329 \pm 8$ From cyclopentadiene | 77PED/RYL 82MCM/GOL 79BAR/SCO 77ENG/LIN2 78CUM/KEB 73RIC/STE 72DID/HAR 79BAR/SCO |
| $C_5H_5N_2^-$ pyrrole · · CN ⁻ • 101±24 ^c | | | 82±15 ^g | 51±10 | IMRE | | 87LAR/MCM |
| $C_5H_5N_2O_2^-$ EtOCOCN · · CN ⁻ • | | | 73±15 ^g | 42±10 | IMRE | | 87LAR/MCM |
| $C_5H_6Cl^-$ cyclopentadiene · · Cl ⁻ | | | | <10 | TDEq | | 82FRE/IKU |
| $C_5H_6NO^-$ $Me_2NCOC=C^-$ • 8±22 ^a | | | 1517±10 ^g | 1484±8 | IMRE | Est2 $\Delta_f H(AH) = 21 \pm 13$ $BDE(A-H) = 552 \pm 21$ | 86TAF |
| $C_5H_7^-$ $CH_2=C(CH=CH_2)CH_2^-$ 159±24 ^a | | | 1614±23 ^g | 1586±21 | IMRB | $\Delta_f H(AH) = 75 \pm 1$ Acid: isoprene | 77PED/RYL 79BAR/MCI |
| $C_5H_7^-$ $nPrC=C^-$ • 203±19 ^a • 2.85±0.37 ^d | | | 1589±15 ^g | 1556±8 1551±8 | IMRE IMRE ^o | $\Delta_f H(AH) = 144 \pm 4$ $BDE(A-H) = 552 \pm 21$ | 79ROG/DAG 79BAR/SCO 79BAR/SCO |
| $C_5H_7^-$ pentadienide ⁻ • 118±16 ^a • 0.91±0.03 | | | 1542±15 ^e | 1522±22 ^h | PD | $\Delta_f H(AH) = 106$ $BDE(A-H) = 318 \pm 13$ Acid: 1,4-pentadiene | 77PED/RYL 82MCM/GOL 78ZIM/GYG |
| $C_5H_7N_2^-$ 3,5-diMe-pyrazolide ⁻ • | | | 1481±11 ^g | 1450±8 | IMRE | | 86TAF |
| $C_5H_7O^-$ cyclopentanone enolate ⁻ • 1.62±0.06 | | | | | PD | $\Delta_f H(AH) = -194 \pm 2$ | 77PED/RYL 78ZIM/JAC |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|--------------|---|--------------------------------------|
| $C_5H_7O_2^-$ MeCOCH=C(Me)O ⁻ • -472±11 ^a | | | 1438±10 ^g 1438±10 ^g | 1408±8 1409±8 | IMRE IMRE | $\Delta_f H(AH) = -384 \pm 2$ | 79HAC/PIL 78CUM/KEB 86TAF |
| $C_5H_7O_3^-$ MeCO ₂ CH=C(Me)O ⁻ • -638±25 ^a | | | 1466±12 ^g | 1436±8 | IMRE | Est2 $\Delta_f H(AH) = -573 \pm 13$ BDE(A-H) = 377±8 | 78CUM/KEB |
| $C_5H_8Cl^-$ CH ₂ =CHCH ₂ CH=CH ₂ ·Cl ⁻ | | | | 15 | TDEq | | 82FRE/IKU |
| $C_5H_8ClO_2^-$ MeCOCH ₂ COMe·Cl ⁻ | | | | 56 | TDEq | | 82FRE/IKU |
| $C_5H_9O^-$ Me ₂ C(CHO)CH ₂ ⁻ -153±25 ^a | | | 1621±21 ^g | 1594±17 | IMRB | Est $\Delta_f H(AH) = -244 \pm 4$ | 80NOE/NIB |
| $C_5H_9O^-$ Me ₂ C=C(Me)O ⁻ • -257±13 ^a 1.46±0.26 ^d | | | 1535±12 ^g | 1508±8 | IMRE | $\Delta_f H(AH) = -262 \pm 1$ BDE(A-H) = 364±13 | 77PEDI/RYL 78CUM/KEB |
| $C_5H_9O^-$ MeCH=C(Et)O ⁻ • -246±13 ^a 1.65±0.30 ^d • 1.68±0.05 | | | 1542±12 ^g | 1512±8 | IMRE PD | $\Delta_f H(AH) = -259 \pm 1$ BDE(A-H) = 390±17 | 77PEDI/RYL 78CUM/KEB 77ZIM/RBE |
| $C_5H_9O_2^-$ iBuCO ₂ ⁻ • -596±17 ^a 3.17±0.20 ^d | | | 1449±11 ^g | 1420±8 | IMRE | $\Delta_f H(AH) = -515 \pm 6$ BDE(A-H) = 444±8 | 77PEDI/RYL 86TAF |
| $C_5H_9O_2^-$ nBuCO ₂ ⁻ • -572±11 ^a 3.2±0.2 ^d | | | 1449±10 | 1419±12 ^h | CIDC | $\Delta_f H(AH) = -490 \pm 2$ BDE(A-H) = 444±8 | 77PEDI/RYL 81MCL/CAM |
| $C_5H_9O_2^-$ tBuCO ₂ ⁻ • -600±15 ^a 3.25±0.20 ^d | | | 1442±11 ^g | 1412±8 | IMRE | Est $\Delta_f H(AH) = -512 \pm 4$ BDE(A-H) = 444±8 | 86TAF |
| $C_5H_{10}ClO^-$ Et ₂ CO·Cl ⁻ • -545±10 ^c | | | 59±8 ^g | 34±8 | IMRE | | 84LAR/MCM2 |
| $C_5H_{10}ClO^-$ tBuCHO·Cl ⁻ • -534±14 ^c | | | 63±8 ^g | 35±8 | IMRE | | 84LAR/MCM2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|-------------------|---|-------------------------------------|
| $C_5H_{10}FO^-$ tBuCHO $\cdot \cdot F^-$ • -596 ± 15^c | | | 103 ± 8^g | 70 ± 8 | IMRE | | 83LAR/MCM |
| $C_5H_{10}IO_2^-$ tBuCO ₂ H $\cdot \cdot I^-$ • -765 ± 9^c | | | 64 ± 4 | 37 ± 9 | TDA's | | 84CAL/KEB |
| $C_5H_{10}NO^-$ HN=C(tBu)O $^-$ • -354 ± 23^a | | | 1499 ± 11^g | 1469 ± 8 | IMRE | Est2 $\Delta_f H(AH) = -322 \pm 13$ | 86TAF |
| $C_5H_{10}NO^-$ tBuCH=NO $^-$ • -147 ± 23^a | | | 1518 ± 14^g | 1489 ± 10 1497 ± 8 | IMRE IMRE o | Est $\Delta_f H(AH) = -135 \pm 8$ | 79BAR/SCO 79BAR/SCO |
| $C_5H_{10}NO^-$ tBuOH $\cdot \cdot CN^-$ • -314 ± 26^c | | | 76 ± 15^g | 45 ± 10 | IMRE | | 87LAR/MCM |
| $C_5H_{10}NO_2^-$ tBuCH=NO ₂ $^-$ • -233 ± 16^a | | | 1486 ± 12^g | 1458 ± 8 1467 ± 8 | IMRE IMRE o | Est $\Delta_f H(AH) = -189 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_5H_{11}Br_2^-$ tBuCH ₂ Br $\cdot \cdot Br^-$ -418^c | | | 60 | 29 | TDA's | | 74DOU |
| $C_5H_{11}O^-$ Et ₂ CHO $^-$ • -286 ± 12^a 2.0 ± 0.2^d | | | 1559 ± 11 1556 ± 10 | 1532 ± 13^h | CIDC CIDC o | $\Delta_f H(AH) = -316 \pm 1$ $BDE(A-H) = 438 \pm 4$ | 77PED/RYL 83BOA/HOU 83BOA/HOU |
| $C_5H_{11}O^-$ iPrCH(Me)O $^-$ • -285 ± 13^a 2.0 ± 0.2^d | | | 1561 ± 11 1556 ± 10 | 1533 ± 13^h | CIDC CIDC o | $\Delta_f H(AH) = -316 \pm 1$ $BDE(A-H) = 438 \pm 4$ | 77PED/RYL 83BOA/HOU 83BOA/HOU |
| $C_5H_{11}O^-$ iPrCH ₂ CH ₂ O $^-$ • -274 ± 15^a 1.9 ± 0.2^d | | | 1563 ± 11 1559 ± 10 | 1535 ± 13^h 1531 ± 12^h | CIDC CIDC o | Est $\Delta_f H(AH) = -306 \pm 4$ $BDE(A-H) = 436 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_5H_{11}O^-$ nC ₅ H ₁₁ O $^-$ • -262 ± 13^a 1.9 ± 0.2^d | | | 1564 ± 11 1560 ± 10 | 1537 ± 13^h | CIDC CIDC o | $\Delta_f H(AH) = -297 \pm 2$ $BDE(A-H) = 436 \pm 4$ | 77PED/RYL 83BOA/HOU 83BOA/HOU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|---|--|--|------------------------------------|--|--|
| $C_5H_{11}O^-$ tBuCH ₂ O ⁻ | | | | | Est | $\Delta_f H(AH) = -318 \pm 2$ $BDE(A-H) = 428 \pm 6$ | 82MCM/GOL |
| * | -290 ± 14^a | 1.88 ± 0.19^d | 1559 ± 12^g | 1531 ± 8 | IMRE | | 79BAR/SCO |
| * | | 1.93 ± 0.05 | | | PD | | 78JAN/ZIM |
| | | $< 1.93 \pm 0.06$ | | 1528 ± 8 | PD IMRE ^o | | 75REE/BRA 79BAR/SCO |
| $C_5H_{11}O^-$ tPnO ⁻ | | | | | | $\Delta_f H(AH) = -331 \pm 1$ $BDE(A-H) = 440 \pm 4$ | 77PED/RYL |
| * | -300 ± 13^a | 2.0 ± 0.2^d | 1561 ± 11 1556 ± 10 | 1533 ± 13^h | CIDC CIDC ^o | | 83BOA/HOU 83BOA/HOU |
| $C_5H_{11}S^-$ nC ₅ H ₁₁ S ⁻ | | | | | | $\Delta_f H(AH) = -110 \pm 1$ $BDE(A-H) = 364 \pm 10$ | 77PED/RYL |
| * | -165 ± 13^a | 2.090 ± 0.020 | 1475^c | | LPD | | 80JAN/REE |
| $C_5H_{11}S^-$ tBuCH ₂ S ⁻ | | | | | | $\Delta_f H(AH) = -129 \pm 1$ $BDE(A-H) = 364 \pm 10$ | 77PED/RYL |
| * | -188 ± 13^a | 2.13 ± 0.23^d | 1472 ± 12^g | 1445 ± 8 | IMRE | | 86TAF |
| $C_5H_{12}FSi^-$ c-(CH ₂) ₃ Si(Me) ₂ · · F ⁻ | | | | | | | |
| | -544 ± 23^c | | 158 ± 9^g | 130 ± 9 | IMRE | | 81SUL/DEP |
| $C_5H_{13}O_2^-$ EtOH · · nPrO ⁻ | | | | | | | |
| | -531 ± 21^c | | 85 ± 10^g | 57 ± 8 | IMRE | | 84CAL/ROZ |
| $C_5H_{13}O_2^-$ MeOH · · tBuO ⁻ | | | | | | | |
| | -556 ± 23^c | | 107 ± 4 79 ± 10^g | 72 ± 7 51 ± 7 | TDEq IMRE | | 86MEO/SIE2 84CAL/ROZ |
| $C_5H_{15}Si^-$ nPnSiH ₃ · · H ⁻ | | | | | | | |
| | | | 45 ± 23 | | IMRB | | 86HAJ/SQU |
| $C_6Br_4O_2^-$ bromanil ⁻ | | | | | | | |
| | -218 ± 40^b | 2.4 ± 0.2 | | | Est NBIP | $\Delta_f H(A) = 18 \pm 21$ | 78COO/FRE |
| $C_6Cl_4O_2^-$ chloranil ⁻ | | | | | | | |
| * | -454 ± 21^b | 2.78 ± 0.10 2.68 ± 0.11 2.67 ± 0.05 2.8 ± 0.2 2.5 ± 0.3 | | | TDEq IMRE IMRE NBIP SI | $\Delta_f H(A) = -186 \pm 12$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI 78COO/FRE 66FAR/PAG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|--|---|---|--|--|--|
| $C_6Cl_5^-$ $C_6Cl_5^-$ | | 2.8 | 1510±13 ^c | | SI | $\Delta_f H(AH) = -40 \pm 9$ $BDE(A-H) = 464 \pm 13$ | 85PLA/SIM2 69PAG/GOO |
| $C_6F_4O_2^-$ fluoranil ⁻ | | | | | Est TDEq IMRE NBIP SI | $\Delta_f H(A) = -816 \pm 41$ | 87KEB/CHO 85FUK/MCI 78COO/FRE 69PAG/GOO |
| | ^a -1076±51 ^b -1043±46 | 2.70±0.10 2.36±0.05 | | | | | |
| | | 2.9±0.2 2.3 | | | | | |
| $C_6F_5^-$ $C_6F_5^-$ | | | | | | $\Delta_f H(AH) = -806 \pm 7$ $BDE(A-H) = 487 \pm 8$ | 77PED/RYL 82MCM/GOL |
| | ^b -797±34 -464 | 2.7±0.2 | 1539±28 ^c | 1506±29 ^h | NBAP Endo SI | From perfluorobenzene $I^- + C_6F_6 \rightarrow$ | 82COM/REI 73LIF/TIE 69PAG/GOO |
| | | 2.7 | | | | | |
| $C_6F_5O^-$ pentafluorophenoxide ⁻ | | | | | | $\Delta_f H(AH) = -957 \pm 2$ | 77PED/RYL |
| | | 3.06±0.09 | | | ECD | | 84HER/WEN |
| | < -857±8 | | < 1630±10 ^f | | IMRB | $HO^- + C_6F_6 \rightarrow$, acidity probably ca. 1340 kJ | 75BRI/RIV |
| $C_6F_6^-$ $C_6F_6^-$ | | | | | | $\Delta_f H(A) = -946 \pm 8$ | 79PRI/SAP |
| | ^a -996±18 ^b | 0.52±0.10 0.52±0.10 | | | TDEq TDEq EnCT SI | | 87KEB/CHO 86CHO/GRI 73LIF/TIE 69PAG/GOO |
| | | 1.8±0.3 1.20±0.07 | | | | | |
| $C_6F_{10}^-$ perfluorocyclohexene ⁻ | | | | | | $\Delta_f H(A) = -2369 \pm 8$ | 79PRI/SAP |
| | < -2504±37 ^b | 1.4±0.3 | | | EnCT | | 73LIF/TIE |
| $C_6F_{11}^-$ $C_6F_{11}^-$ | | | | | | | |
| | | > 4.2±0.2 3.5 | | | EIAP EIAP | From $c-C_4F_6(CF_3)_2$ From $c-C_6F_{12}$ | 72THY 70LIF/PBE |
| $C_6F_{13}^-$ $C_6F_{13}^-$ | | | | | | | |
| | | > 4.6±0.2 | | | EIAP | From $n-C_6F_{14}$ | 83SPY/SAU |
| $C_6N_4^-$ tetracyanoethylene ⁻ | | | | | | $\Delta_f H(A) = 705 \pm 6$ | 77PED/RYL |
| | ^a 400±25 ^b | 3.17±0.20 3.17±0.20 | | | TDEq TDEq LPD PD LPD SI | | 87KEB/CHO 86CHO/KEB 76LYO/PAL 73LYO/PAL 75LYO/PAL 67FAR/PAG |
| | | 2.300±0.300 2.03±0.05 1.700±0.300 2.88±0.06 | | | | | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|---|--|--|----------------------|--|--|
| $C_6HCl_3O_2^-$ triCl-benzoquinone ⁻ • -423 ± 17^b | | 2.52 ± 0.05 | | | Est2 IMRE | $\Delta_f H(A) = -180 \pm 13$ | 85FUK/MCI |
| $C_6HF_4O^-$ 2,3,5,6-tetrafluorophenoxide ⁻ | | 2.75 ± 0.09 | | | Est2 ECD | $\Delta_f H(AH) = -764 \pm 13$ $BDE(A-H) = 385 \pm 17$ | 84HER/WEN |
| $C_6H_2Cl_2O_2^-$ 2,5-diCl-benzoquinone ⁻ • -409 ± 22^b | | 2.43 ± 0.10 2.29 ± 0.05 | | | Est2 TDEq IMRE | $\Delta_f H(A) = -174 \pm 13$ | 87KEB/CHO 85FUK/MCI |
| $C_6H_2Cl_2O_2^-$ 2,6-diCl-benzoquinone ⁻ • -414 ± 21^b | | 2.48 ± 0.10 2.39 ± 0.11 2.40 ± 0.05 | | | TDEq IMRE IMRE | $\Delta_f H(A) = -174 \pm 12$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_2Cl_3O^-$ 3,4,5-triCl-phenoxide ⁻ • -310 ± 21^a | | 3.00 ± 0.26^d | 1384 ± 12^g | 1355 ± 8 | Est IMRE | $\Delta_f H(AH) = -164 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 81FUJ/MCI |
| $C_6H_2FO_2^-$ fluorobenzoquinonide ⁻ | | 2.4 ± 0.1 | | | Est2 SI | $\Delta_f H(AH) = -387 \pm 13$ | 66FAR/PAG |
| $C_6H_2N_3O_7^-$ 2,4,6-triNO ₂ -phenoxide ⁻ • -365 ± 25^a | | | $< 1324 \pm 4^g$ | < 1293 | Est2 IMRB | $\Delta_f H(AH) = -159 \pm 21$ $BDE(A-H) = 381 \pm 17$ Γ^- deprotonates | 74DZI/CAR |
| $C_6H_3ClNO_3^-$ 2-Cl-4-NO ₂ -phenoxide ⁻ • -328 ± 28^a | | | 1353 ± 11^g | 1323 ± 8 | Est2 IMRE | $\Delta_f H(AH) = -151 \pm 17$ $BDE(A-H) = 381 \pm 17$ | 86TAF |
| $C_6H_3Cl_2NO_2^-$ 2,3-diCl-nitrobenzene ⁻ • -116 ± 13^b | | 1.23 ± 0.05 | | | Est2 IMRE | $\Delta_f H(A) = 3 \pm 8$ | 85FUK/MCI |
| $C_6H_3Cl_2NO_2^-$ 3,4-diCl-nitrobenzene ⁻ • -125 ± 13^b | | 1.38 ± 0.05 | | | Est IMRE | $\Delta_f H(A) = 8 \pm 8$ | 85FUK/MCI |
| $C_6H_3Cl_2O^-$ 3,5-diCl-phenoxide ⁻ • -284 ± 19^a | | 2.85 ± 0.24^d | 1399 ± 11^g | 1370 ± 8 | Est IMRE | $\Delta_f H(AH) = -153 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 81FUJ/MCI |
| $C_6H_3FO_2^-$ fluorobenzoquinone ⁻ | | 1.5 ± 0.2 | | | Est2 SI | $\Delta_f H(A) = -387 \pm 13$ | 66FAR/PAG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|--------------|--|-------------------------------------|
| $C_6H_3F_2^-$ m-difluorophenide ⁻ < -264±3 ^a | | | < 1576±1 ^g | < 1543 | IMRB | $\Delta_f H(AH) = -309 \pm 1$ < iPrOH | 77PED/RYL 75BRI/RIV |
| $C_6H_3F_2^-$ o-difluorophenide ⁻ -242±18 ^a | 2.0±0.3 ^d | | 1582±16 ^g | 1547±13 | IMRB | $\Delta_f H(AH) = -294 \pm 1$ $BDE(A-H) = 460 \pm 13$ Between EtO ⁻ , iPrO ⁻ | 77PED/RYL 75BRI/RIV |
| $C_6H_3F_2^-$ p-difluorophenide ⁻ -247±16 ^a | 1.9±0.3 ^d | | 1590±15 ^g | 1555±13 | IMRB | $\Delta_f H(AH) = -307 \pm 1$ $BDE(A-H) = 460 \pm 13$ < MeOH, ≤ EtOH | 77PED/RYL 75BRI/RIV |
| $C_6H_3FeO_6^-$ Fe(CO) ₅ · OMe ⁻ -1095±37 ^c | | | 188±25 ^g | 149±25 | IMRB | | 85LAN/SAL |
| $C_6H_3N_3O_6^-$ 1,3,5-trinitrobenzene ⁻ -191 ^b | 2.6 | | | | SI | $\Delta_f H(A) = 62 \pm 2$ | 77PED/RYL 69PAG/GOO |
| $C_6H_3O_2^-$ benzoquinonide ⁻ 2.00±0.04 | | | | < 1607 | IMRB SI | $\Delta_f H(AH) = -123 \pm 3$ | 77PED/RYL 87JOH/SPE 66FAR/PAG |
| $C_6H_4^-$ o-benzyne ⁻ • 440±22 ^b < 433 | 0.560±0.010 | | | | LPES IMRB | $\Delta_f H(A) = 494 \pm 21$ O ⁻ + C ₆ H ₆ →, D label indicates ortho loss | 80POL/HEH 86LEO/MIL 78BRU/FER |
| $C_6D_4^-$ o-benzyne-d ₄ ⁻ • | 0.551±0.010 | | | | LPES | | 86LEO/MIL |
| $C_6H_4BrNO_2^-$ mBr-nitrobenzene ⁻ • -38±14 ^b | 1.32±0.10 | | | | Est TDEq | $\Delta_f H(A) = 90 \pm 4$ | 87KEB/CHO |
| $C_6H_4BrNO_2^-$ oBr-nitrobenzene ⁻ • -21±18 ^b | 1.17±0.10 | | | | Est2 TDEq | $\Delta_f H(A) = 92 \pm 8$ | 87KEB/CHO |
| $C_6H_4BrNO_2^-$ pBr-nitrobenzene ⁻ • -35±14 ^b | 1.29±0.10 | | | | Est TDEq | $\Delta_f H(A) = 90 \pm 4$ | 87KEB/CHO |
| $C_6H_4Cl^-$ chlorophenide ⁻ 144±24 ^a | 1.6±0.4 ^d | | 1620±23 ^g | 1586±21 | IMRB | $\Delta_f H(AH) = 54 \pm 1$ $BDE(A-H) = 460 \pm 13$ | 85PLA/SIM 79BAR/MCI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|-----------------------------|---|-------------------------------------|
| $C_6H_4ClF_2^-$ m- $C_6H_4F_2 \cdots Cl^-$ • -598 ± 11^c | | | 61 ± 8^g | 33 ± 8 32 ± 4 | IMRE TDEq | | 84LAR/MCM2 82FRE/IKU |
| $C_6H_4ClF_2^-$ o- $C_6H_4F_2 \cdots Cl^-$ • -581 ± 11^c | | | 60 ± 8^g | 33 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_6H_4ClF_2^-$ p- $C_6H_4F_2 \cdots Cl^-$ • -592 ± 11^c | | | 58 ± 8^g | 31 ± 8 | IMRE | | 84LAR/MCM2 |
| $C_6H_4ClNO_2^-$ mCl-nitrobenzene $^-$ • -85 ± 18^b | 1.28 ± 0.10 1.22 ± 0.11 1.20 ± 0.05 | | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = 38 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4ClNO_2^-$ oCl-nitrobenzene $^-$ • -68 ± 14^b | 1.14 ± 0.10 1.08 ± 0.11 1.05 ± 0.05 | | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = 42 \pm 4$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4ClNO_2^-$ pCl-nitrobenzene $^-$ • -84 ± 18^b | 1.26 ± 0.10 1.19 ± 0.11 1.17 ± 0.05 | | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = 38 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4ClO^-$ mCl-phenoxide $^-$ • -245 ± 29^a | 2.52 ± 0.30^d | | 1431 ± 21^g 1433 ± 21^g | 1402 ± 8 1404 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -146 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 77PED/RYL 81FUJ/MCI 77MCM/KEB |
| $C_6H_4ClO^-$ oCl-phenoxide $^-$ • -266 ± 30^a • $< 2.58 \pm 0.08$ | 2.87 ± 0.31^d | | 1437 ± 13^g | 1410 ± 8 | IMRE PD | Est2 $\Delta_f H(AH) = -173 \pm 17$ $BDE(A-H) = 402 \pm 17$ | 77MCM/KEB 75RIC/STB2 |
| $C_6H_4ClO^-$ pCl-phenoxide $^-$ • -248 ± 18^a | 2.47 ± 0.23^d | | 1436 ± 10^g 1438 ± 10^g | 1407 ± 8 1409 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -153 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 77PED/RYL 81FUJ/MCI 77MCM/KEB |
| $C_6H_4Cl_2^-$ o-dichlorobenzene $^-$ 24 b | 9.4 | | | | ECD | $\Delta_f H(A) = 33 \pm 2$ | 85PLA/SIM 69STB/WEN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|---|---|---|-----------------------------|--|-------------------------------------|
| $C_6H_4F^-$ fluorophenide ⁻ | -26 ± 26^a | 1.6 ± 0.4^d | 1620 ± 25^g | 1586 ± 22 | IMRB | $\Delta_f H(AH) = -116 \pm 1$ $BDE(A-H) = 460 \pm 13$ | 77PED/RYL 75BRI/RIV |
| $C_6H_4FNO_2^-$ mF-nitrobenzene ⁻ | -245 ± 18^b | 1.23 ± 0.10 1.18 ± 0.11 1.15 ± 0.05 | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = -126 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4FNO_2^-$ oF-nitrobenzene ⁻ | -221 ± 18^b | 1.07 ± 0.10 1.02 ± 0.11 1.04 ± 0.05 | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = -118 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4FNO_2^-$ pF-nitrobenzene ⁻ | -239 ± 18^b | 1.12 ± 0.10 1.05 ± 0.11 1.04 ± 0.05 | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = -131 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4FO^-$ mF-phenoxide ⁻ | -389 ± 18^a | 2.45 ± 0.19^d | 1438 ± 10^g 1441 ± 10^g 1422 ± 17^c | 1409 ± 8 1413 ± 8 1393 ± 18^h | IMRE IMRE ECD | $\Delta_f H(AH) = -297 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI 77MCM/KEB 84HER/WEN |
| $C_6H_4FO^-$ oF-phenoxide ⁻ | | | 1445 ± 12^g 1447 ± 12^g | 1418 ± 8 1420 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -285$ | 81FUJ/MCI 77MCM/KEB |
| $C_6H_4FO^-$ pF-phenoxide ⁻ | -370 ± 18^a | 2.31 ± 0.23^d | 1451 ± 10^g 1455 ± 10^g | 1422 ± 8 1426 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -291 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 81FUJ/MCI 77MCM/KEB |
| $C_6H_4F_2N^-$ 2,4-diF-anilide ⁻ | -497 ± 25^a | | 1510 ± 12^g | 1480 ± 8 1480 ± 8 | IMRE IMRE ^o | $\Delta_f H(AH) = -478 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_6H_4NO_2^-$ pNO-phenoxide ⁻ | -246 ± 19^a | | 1376 ± 11^g | 1345 ± 8 | IMRE | $\Delta_f H(AH) = -91 \pm 8$ | 86TAF |
| $C_6H_4NO_3^-$ mNO ₂ -phenoxide ⁻ | -244 ± 19^a | 2.85 ± 0.20^d | 1399 ± 11^g 1400 ± 11^g | 1370 ± 8 1371 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -113 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI 77MCM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|--|--|--|---|---|--|
| $C_6H_4NO_3^-$ oNO ₂ -phenoxide ⁻ * | | | | 1379±8 | Est2 IMRE | $\Delta_f H(AH) = -105 \pm 13$ | 77MCM/KEB |
| $C_6H_4NO_3^-$ pNO ₂ -phenoxide ⁻ * | -276±19 ^a | | 1372±11 ^g | 1343±8 | Est IMRE | $\Delta_f H(AH) = -117 \pm 8$ | 81FUJ/MCI |
| $C_6H_4N_2O_4^-$ mNO ₂ -nitrobenzene ⁻ * | -101±11 ^b | 1.65±0.10 1.57±0.11 1.57±0.05 | | | TDEq IMRE IMRE | $\Delta_f H(A) = 59 \pm 1$ | 76FER/PIA 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4N_2O_4^-$ oNO ₂ -nitrobenzene ⁻ * | -76±18 ^b | 1.65±0.10 | | | Est2 TDEq | $\Delta_f H(A) = 84 \pm 8$ | 87KEB/CHO |
| $C_6H_4N_2O_4^-$ pNO ₂ -nitrobenzene ⁻ * | -136±13 ^b | 2.00±0.10 1.89±0.11 1.89±0.05 | | | TDEq IMRE IMRE | $\Delta_f H(A) = 57 \pm 3$ | 76FER/PIA 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_6H_4N_3^-$ benzotriazolide ⁻ * | | | 1413±11 ^g | 1382±8 | IMRE | | 86TAF |
| $C_6H_4O_2^-$ o-benzoquinone ⁻ * | | 1.620±0.048 | | | Est2 LPD | $\Delta_f H(A) = -121 \pm 21$ | 85MAR/COM |
| $C_6H_4O_2^-$ p-benzoquinone ⁻ * | -307±13 ^b | 1.91±0.10 1.81±0.11 1.990±0.048 1.83±0.05 1.9±0.3 >0.0 1.37±0.08 | | | TDEq IMRE LPD IMRE NBIP ES SI | $\Delta_f H(A) = -123 \pm 3$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85MAR/COM 85FUK/MCI 75COO/NAF 70COL/CHR 66FAR/PAG |
| $C_6H_5^-$ phenide ⁻ * | 229±3 ^a | 1.03±0.11 ^d | 1677±2 | 1636±3 1632±27 | TDEq IMRB EIAP IMRB SI SI | $\Delta_f H(AH) = 83$ $BDE(A-H) = 464 \pm 8$ | 77PED/RYL 82MCM/GOL 86MEO/SIE 79BAR/MCI 86HEN/ILL2 71BOH/YOU 76FAI/JOY 72PAG |
| | 341±29 | 1.1±0.3 ^d 2.36±0.04 2.2 | 1665±25 ^g | 1628±23 | | From benzonitrile | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|-----------------------------------|---|---------------------------------------|
| $C_6H_5BrCl^-$ $PhBr \cdot \cdot Cl^-$ | | | | 28 | TDEq | | 82FRE/IKU |
| $C_6H_5ClF^-$ $PhF \cdot \cdot Cl^-$ | | | | 25 | TDEq | | 82FRE/IKU |
| $C_6H_5ClFO^-$ $pF-C_6H_4OH \cdot \cdot Cl^-$ • -629 ± 18^c | | | 110 ± 8 | 81 ± 8 | TDEq | | 77CUM/FRE |
| $C_6H_5ClI^-$ $PhI \cdot \cdot Cl^-$ | | | | 30 | TDEq | | 82FRE/IKU |
| $C_6H_5ClN^-$ $mCl-anilide^-$ • 26 ± 18^a | | | 1502 ± 10^g | 1471 ± 8 1480 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = 55 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_6H_5ClN^-$ $pCl-anilide^-$ • 33 ± 18^a | | | 1508 ± 10^g | 1477 ± 8 1482 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = 55 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_6H_5ClNO_2^-$ $PhNO_2 \cdot \cdot Cl^-$ | | | | 30 | TDEq | | 82FRE/IKU |
| $C_6H_5Cl_2^-$ $PhCl \cdot \cdot Cl^-$ • -230 ± 10^c | | | 57 ± 8^g 57 ± 4^g | 29 ± 8 29 ± 4 27 | IMRE IMRE TDEq | | 84LAR/MCM2 84LAR/MCM4 82FRE/IKU |
| $C_6H_5Cl_2O^-$ $pCl-C_6H_4OH \cdot \cdot Cl^-$ • -498 ± 18^c | | | 118 ± 8 | 87 ± 8 | TDEq | | 77CUM/FRE |
| $C_6H_5FN^-$ $mF-anilide^-$ • -132 ± 19^a | | | 1511 ± 11^g | 1481 ± 8 1489 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -113 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_6H_5FN^-$ $oF-anilide^-$ • -143 ± 29^a | 1.91 ± 0.30^d | | 1517 ± 12^g | 1487 ± 8 1495 ± 8 | Est2 IMRE IMRE ^o | $\Delta_f H(AH) = -130 \pm 17$ $BDE(A-H) = 389 \pm 17$ | 79BAR/SCO 79BAR/SCO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|--|---|---|--|---|---|
| $C_6H_5FN^-$ pF-anilide ⁻ | -115 ± 19^a | 1.62 ± 0.24^d | 1524 ± 11^g | 1494 ± 8 1499 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -109 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_6H_5F_5NS^-$ mSF ₅ -anilide ⁻ | * | 2.76 ± 0.24^d | 1414 ± 11^g | 1383 ± 8 | IMRE | $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_6H_5F_5NS^-$ pSF ₅ -anilide ⁻ | * | 2.92 ± 0.24^d | 1399 ± 11^g | 1368 ± 8 | IMRE | $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_6H_5N^-$ PhN ⁻ | 263 ± 26^a * | 1.461 ± 0.013 | 1556 ± 16^g | 1527 ± 13 1532 ± 13 | IMRB LPD IMRB ^o | Acidity near MeCN | 81MCD/CHO 84DRZ/BRA 81MCD/CHO |
| $C_6H_5NO_2^-$ nitrobenzene ⁻ | -30 ± 11^b | 1.01 ± 0.10 0.96 ± 0.11 0.97 ± 0.05 $> 0.7 \pm 0.2$ > 0.4 < 1.1 | | | TDEq IMRE IMRE EnCT ES IMRB | $\Delta_f H(A) = 67 \pm 1$ EA: $< SO_2$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI 73LIF/TIE 66COM/CHR 59HEN/MUC |
| $C_6H_5N_2O^-$ PhN=NO ⁻ | $< 308 \pm 25$ | | | | IMRB | Ph ⁻ + N ₂ O ⁻ ; thermochemical limit | 77BIE/DEP |
| $C_6H_5N_2O_2^-$ mNO ₂ -anilide ⁻ | 6 ± 13^a | 2.14 ± 0.24^d | 1474 ± 11^g | 1443 ± 8 | IMRE | $\Delta_f H(AH) = 62 \pm 2$ $BDE(A-H) = 368 \pm 13$ | 83NIS/SAK 86TAF |
| $C_6H_5N_2O_2^-$ pNO ₂ -anilide ⁻ | -38 ± 13^a | | 1437 ± 11^g | 1407 ± 8 | IMRE | $\Delta_f H(AH) = 55 \pm 2$ | 83NIS/SAK 86TAF |
| $C_6H_5O^-$ phenoxide ⁻ | -165 ± 10^a * | 2.21 ± 0.19^d $< 2.36 \pm 0.06$ | 1461 ± 10^g 1466 ± 10^g | 1432 ± 8 1437 ± 8 1441 ± 8 | IMRE PD IMRE IMRE ^o | $\Delta_f H(AH) = -96 \pm 1$ $BDE(A-H) = 362 \pm 8$ 86SHI/VOR: tautomer acidities $\Delta_{acid}H(\text{ortho}) = 1439 \pm 13$ kJ, (para) = 1423 ± 8 kJ | 77PED/RYL 82MCM/GOL 81FUJ/MCI 75RIC/STE2 78CUM/KEB 79BAR/SCO |
| $C_6H_5O_2^-$ mOH-phenoxide ⁻ | -354 ± 13^a | 2.32 ± 0.20^d | 1451 ± 11^g 1444 ± 11^g | 1422 ± 8 1415 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -274 \pm 2$ $BDE(A-H) = 362 \pm 8$ | 79KUD/KUD 81FUJ/MCI 77MCM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|------------------------|--|--|----------------------------------|---|--|
| $C_6H_5O_2^-$ oOH-phenoxide ⁻ • | | | 1421±11 ^g 1422±11 ^g | 1392±8 1393±8 | IMRE IMRE | $\Delta_f H(AH) = -272 \pm 4$ | 79KUD/KUD 81FUJ/MCI 77MCM/KEB |
| $C_6H_5O_2^-$ pOH-phenoxide ⁻ • | -328±13 ^a | | 1466±11 ^g | 1436±8 | IMRE | $\Delta_f H(AH) = -264 \pm 2$ | 79KUD/KUD 81FUJ/MCI |
| $C_6H_5S^-$ thiophenoxide ⁻ • | < 2.47±0.06 | | > 1409±15 ^c | | PD | $\Delta_f H(AH) = 113 \pm 1$ $BDE(A-H) = 335 \pm 9$ | 77PED/RYL 82MCM/GOL 75RIC/STB2 |
| $C_6H_6Cl^-$ $C_6H_6 \cdot \cdot Cl^-$ • | | | 41 ^g | 20±8 16 | IMRE TDEq | | 84LAR/MCM2 82FRE/IKU |
| $C_6H_6ClO^-$ PhOH··Cl ⁻ -432 ^c -426±10 ^c | | | 109 103±8 81±8 | 72 83±8 62±8 | TDA TDEq TDA | | 82FRE/IKU 77CUM/FRE 71YAM/KEB |
| $C_6H_6FO^-$ PhOH··F ⁻ • | -518±11 ^c | | 173±8 ^g | 140±8 | IMRE | | 83LAR/MCM |
| $C_6H_6N^-$ anilide ⁻ • | 90±12 ^a 1.704±0.030 | 1.53±0.20 ^d | 1533±11 ^g | 1502±8 1505±8 | IMRE LPD IMRE ^o | $\Delta_f H(AH) = 87 \pm 1$ $BDE(A-H) = 368 \pm 8$ | 77PED/RYL 82MCM/GOL 79BAR/SCO 84DRZ/BRA2 79BAR/SCO |
| $C_6H_6NO^-$ mNH ₂ -phenoxide ⁻ • | -153±11 ^a 2.15±0.19 ^d | | 1467±10 ^g 1469±10 ^g | 1438±8 1441±8 | IMRE IMRE | $\Delta_f H(AH) = -90 \pm 2$ $BDE(A-H) = 362 \pm 8$ | 86NUN/BAR 81FUJ/MCI 77MCM/KEB |
| $C_6H_6NO^-$ oNH ₂ -phenoxide ⁻ • | | | | 1428±8 | IMRE | Est2 $\Delta_f H(AH) = -105 \pm 17$ $BDE(A-H) = 391 \pm 17$ | 77MCM/KEB |
| $C_6H_6NO^-$ pNH ₂ -phenoxide ⁻ • | -137±11 ^a | | 1475±10 ^g 1483±10 ^g | 1446±8 1454±8 | IMRE IMRE | $\Delta_f H(AH) = -82 \pm 2$ $BDE(A-H) = 368 \pm 13$ | 86NUN/BAR 81FUJ/MCI 77MCM/KEB |
| $C_6H_6N_2O_2^-$ mNH ₂ -nitrobenzene ⁻ • | -33±11 ^b 0.95±0.10 | | | | TDEq | $\Delta_f H(A) = 59 \pm 1$ | 77PED/RYL 87KEB/CHO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------------|---|---|---------------------------|---|--|
| $C_6H_7^-$ 1,3-cyclohexadienide ⁻ | 138 ± 21^a 134 ± 75 | 0.6 ± 0.4^d | 1562 ± 21^g | 1531 ± 17 | IMRB IMRB | $\Delta_f H(AH) = 106$ $BDE(A-H) = 305 \pm 21$ Between SiH ₄ , tBuOH | 77PED/RYL 82MCM/GOL 86LEE/SQU 78DEP/BIE |
| $C_6H_7^-$ 1-methylcyclopentadienide ⁻ | * | 1.670 ± 0.039 | | | LPD | Est $\Delta_f H(AH) = 96 \pm 4$ | 73RIC/STE |
| $C_6H_7FN^-$ PhNH ₂ · F ⁻ | * -292 ± 11^c | | 131 ± 8^g | 98 ± 8 | IMRE | | 83LAR/MCM |
| $C_6H_7O^-$ cyclohexenone-4-enolate ⁻ | * -150 ± 22^a | | 1496 ± 10^g | 1464 ± 8 1473 ± 8 | IMRE IMRE ^o | Est2 $\Delta_f H(AH) = -116 \pm 13$ | 86BAR/KIP 86BAR/KIP |
| $C_6H_7O_2^-$ HOH · PhO ⁻ | -471 ± 14^c | | 64 ± 4 | 34 ± 7 | TDA's | | 86MEO/SIE2 |
| $C_6H_8B^-$ MeB(CH=CH) ₂ CH ⁻ | 17 ± 31^a | | 1402 ± 18^g | 1370 ± 17 | IMRB | Est $\Delta_f H(AH) = 146 \pm 13$ Acid: 3-methyl-3-bora-1,4-cyclohexadiene | 77SUL |
| $C_6H_9^-$ cyclohexenide ⁻ | 82 ± 25^a | | 1617 ± 25^g | 1586 ± 21 | IMRB | $\Delta_f H(AH) = -5$ | 77PED/RYL 86LEE/SQU |
| $C_6H_9^-$ tBuC≡C ⁻ | * 157 ± 15^a | 2.93 ± 0.34^d | 1582 ± 12^g | 1549 ± 8 1544 ± 8 | IMRE IMRE ^o | $\Delta_f H(AH) = 106 \pm 3$ $BDE(A-H) = 552 \pm 21$ | 77KUP/SHI 79BAR/SCO 79BAR/SCO |
| $C_6H_9O^-$ cyclohexanone enolate ⁻ | * 1.55 ± 0.05 | | | | PD | $\Delta_f H(AH) = -226 \pm 2$ | 77PED/RYL 78ZIM/JAC |
| $C_6H_{11}^-$ cyclohexanide ⁻ | $> 37 \pm 4^a$ | | 1690 ± 4^g | > 1665 | IMRB | $\Delta_f H(AH) = -123$ $BDE(A-H) = 400 \pm 4$ | 77PED/RYL 82MCM/GOL 72BOH/LEE |
| $C_6H_{11}O^-$ CH ₂ =C(tBu)O ⁻ | * -280 ± 15^a | 1.84 ± 0.07 | 1540 ± 15^c | 1512 ± 18^h | PD | $\Delta_f H(AH) = -290 \pm 1$ $BDE(A-H) = 406 \pm 8$ | 77PED/RYL 77ZIM/REE |
| $C_6H_{11}O^-$ tBuCH=CHO ⁻ | * -282 ± 21^a | 1.82 ± 0.06 | 1517 ± 18^c | 1490 ± 23^h | PD | Est $\Delta_f H(AH) = -269 \pm 2$ $BDE(A-H) = 381 \pm 13$ | 77ZIM/REE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|------------------------|---|---|---------------------------|---|-------------------------------------|
| $C_6H_{11}O_2^-$ nPrCO ₂ ⁻ | \bullet -597±11 ^a | 3.2±0.2 ^d | 1447±10 | 1418±12 ^h | CIDC | $\Delta_f H(AH) = -514 \pm 2$ $BDE(A-H) = 444 \pm 8$ | 77PED/RYL 81MCL/CAM |
| $C_6H_{11}O_2^-$ tBuCH ₂ CO ₂ ⁻ | \bullet -623±15 ^a | | 1444±11 ^g | 1415±8 | IMRE | $\Delta_f H(AH) = -538 \pm 4$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_6H_{11}S_2^-$ 5,5-dimethyl-1,3-dithianide ⁻ | \bullet -24±28 ^a | | 1566±11 ^g | 1535±8 1530±8 | IMRE IMRE ^o | $\Delta_f H(AH) = -59 \pm 17$ | 81BAR/HAY 81BAR/HAY |
| $C_6H_{13}O^-$ Et ₂ C(Me)O ⁻ | \bullet -330±15 ^a | 2.0±0.2 ^d | 1556±11 1553±10 | 1528±13 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -356 \pm 4$ $BDE(A-H) = 440 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_6H_{13}O^-$ iPrCH(Et)O ⁻ | \bullet -318±15 ^a | 2.0±0.2 ^d | 1554±11 1551±10 | 1527±13 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -342 \pm 4$ $BDE(A-H) = 438 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_6H_{13}O^-$ iPrCH ₂ CH ₂ CH ₂ O ⁻ | \bullet -296±14 ^a | 1.9±0.1 ^d | 1561±10 1557±10 | 1533±11 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -327 \pm 4$ $BDE(A-H) = 436 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_6H_{13}O^-$ nC ₆ H ₁₃ O ⁻ | \bullet -284±12 ^a | 1.9±0.2 ^d | 1561±11 1557±10 | 1533±13 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -315 \pm 1$ $BDE(A-H) = 436 \pm 4$ | 77PED/RYL 83BOA/HOU 83BOA/HOU |
| $C_6H_{13}O^-$ nPrC(Me) ₂ O ⁻ | \bullet -326±15 ^a | 2.0±0.2 ^d | 1557±11 1554±10 | 1529±13 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -352 \pm 4$ $BDE(A-H) = 440 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_6H_{13}O^-$ tBuCH(Me)O ⁻ | \bullet -328±16 ^a | 2.05±0.17 ^d | 1553±12 ^g | 1525±8 1523±8 | IMRE IMRE ^o | $\Delta_f H(AH) = -351 \pm 4$ $BDE(A-H) = 438 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_6H_{13}O^-$ tBuCH ₂ CH ₂ O ⁻ | \bullet -304±15 ^a | 2.0±0.2 ^d | 1559±11 1555±10 | 1531±13 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -332 \pm 4$ $BDE(A-H) = 436 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_6H_{13}O_2^-$ nPrOH \cdots CH ₂ =C(Me)O ⁻ | -518±23 ^c | | 61±10 ^g | 33±8 | IMRE | | 84CAL/ROZ |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|----------------------|---|--|
| $C_6H_{14}BF_2^-$ $iPr_2BF \cdots F^-$ < -817 °C | | | 278 | | IMRB | F ⁻ A: SiF ₄ > iPr ₂ BF > iPr ₃ B | 77MUR/BEA2 |
| $C_6H_{15}BCl^-$ $Et_3B \cdots Cl^-$ * -476±15 °C | | | 100±8 ^g | 72±8 | IMRE | | 85LAR/MCM |
| $C_6H_{15}BF^-$ $Et_3B \cdots F^-$ * -611±16 °C | | | 213±8 ^g 259 | 182±8 | IMRE IMRB | F ⁻ A: iPr ₃ B > Et ₃ B > MeSiF ₃ | 85LAR/MCM 77MUR/BEA2 |
| $C_6H_{15}BFO_3^-$ $(EtO)_3B \cdots F^-$ * -1434±17 °C | | | 184±13 ^g | 153±8 | IMRB | | 85LAR/MCM |
| $C_6H_{15}OSi^-$ Et_3SiO^- -580±19 °C ^a | | | 1508±11 ^g | 1479±8 | IMRE | Est2 $\Delta_f H(AH) = -559±8$ | 87THO/BAR |
| $C_6H_{15}O_2^-$ $EtOH \cdots tBuO^-$ -592±23 °C | | | 82±10 ^g | 54±8 | IMRE | | 84CAL/ROZ |
| $C_6H_{15}O_2^-$ $MeOH \cdots tBuCH_2O^-$ -569±25 °C | | | 78±10 ^g | 50±8 | IMRE | | 84CAL/ROZ |
| $C_6H_{15}O_2^-$ $nPrOH \cdots nPrO^-$ -554±21 °C | | | 88±10 ^g | 60±8 | IMRE | | 84CAL/ROZ |
| $C_6H_{17}Si^-$ $Et_3SiH \cdots H^-$ -98 °C | | | 43±23 | | IMRB | | 86HAJ/SQU |
| $C_6H_{18}NSi_2^-$ $(Me_3Si)_2N^-$ -497±15 °C ^a | 2.32 ^d | | 1509±10 ^g | 1477±8 | IMRE | $\Delta_f H(AH) = -477±6$ $BDE(A-H) = > 421$ | 77PED/RYL 78ROB/WIN 87THO/BAR |
| $C_7F_5N^-$ $C_6F_5CN^-$ * -852±22 °C ^b | 1.10±0.10 1.10±0.10 | | | | Est TDEq TDEq | $\Delta_f H(A) = -746±13$ | 87KEB/CHO 86CHO/GRI |
| $C_7F_8^-$ perfluorotoluene ⁻ * -1278±17 °C ^b | 0.94±0.10 0.91±0.10 > 1.7±0.3 | | | | TDEq IMRE EnCT | $\Delta_f H(A) = -1187±8$ | 77PED/RYL 87KEB/CHO 86CHO/GRI 73LIF/TIE |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-----------------------------|--|--|----------------------|---|-------------------------------------|
| $C_7F_{13}^-$ perfluoromethylcyclohexanide ⁻ | | -3063 ^b | 3.9 | | Est | $\Delta_f H(AH) = -2705 \pm 21$ | |
| | | | | | EIAP | From $c-C_6F_{11}(CF_3)$ | 70LIF/PBE |
| $C_7F_{14}^-$ perfluoromethylcyclohexane ⁻ | | -3002 \pm 11 ^b | 1.06 \pm 0.10 <1.6 | | TDEq IMRB | $\Delta_f H(A) = -2900 \pm 1$ | 77PED/RYL 85GRI/CHO 85GRI/CAL |
| $C_7H_3Cl_2N^-$ 2,6-diCl-benzonitrile ⁻ | | 87 \pm 22 ^b | 0.72 \pm 0.10 0.70 \pm 0.09 | | Est2 TDEq TDEq | $\Delta_f H(A) = 156 \pm 13$ | 87KEB/CHO 86CHO/KEB |
| $C_7H_3Cl_3O_2^-$ Me-triCl-benzoquinone ⁻ | | -449 \pm 21 ^b | 2.46 \pm 0.05 | | Est2 IMRE | $\Delta_f H(A) = -212 \pm 17$ | 85FUK/MCI |
| $C_7H_3F_5O^-$ pentafluoroanisole ⁻ | | -990 \pm 17 ^b | 0.54 \pm 0.09 | | Est2 ECD | $\Delta_f H(A) = -937 \pm 8$ | 84HER/WEN |
| $C_7H_3N_3O_4^-$ 3-NO ₂ -5-CN-nitrobenzene ⁻ | | -20 \pm 14 ^b | 2.16 \pm 0.10 | | Est TDEq | $\Delta_f H(A) = 188 \pm 4$ | 87KEB/CHO |
| $C_7H_4ClO_2^-$ mCl-benzoate ⁻ | | -473 \pm 15 ^a | 3.69 \pm 0.24 ^d | 1400 \pm 11 ^g | | $\Delta_f H(AH) = -342 \pm 4$ $BDE(A-H) = 444 \pm 13$ | 77PED/RYL 77MCM/KEB |
| $C_7H_4ClO_2^-$ oCl-benzoate ⁻ | | -454 \pm 14 ^a | 3.67 \pm 0.24 ^d | 1401 \pm 11 ^g | | $\Delta_f H(AH) = -325 \pm 3$ $BDE(A-H) = 444 \pm 13$ | 77PED/RYL 77MCM/KEB |
| $C_7H_4ClO_2^-$ pCl-benzoate ⁻ | | -472 \pm 14 ^a | 3.69 \pm 0.24 ^d | 1399 \pm 11 ^g | | $\Delta_f H(AH) = -341 \pm 3$ $BDE(A-H) = 444 \pm 13$ | 77PED/RYL 77MCM/KEB |
| $C_7H_4FO_2^-$ mF-benzoate ⁻ | | -617 \pm 15 ^a | 3.65 \pm 0.24 ^d | 1403 \pm 11 ^g | | $\Delta_f H(AH) = -490 \pm 4$ $BDE(A-H) = 444 \pm 13$ | 77MCM/KEB |
| $C_7H_4FO_2^-$ oF-benzoate ⁻ | | -623 \pm 25 ^a | 3.76 \pm 0.30 ^d | 1410 \pm 12 ^g | | $\Delta_f H(AH) = -502 \pm 13$ $BDE(A-H) = 460 \pm 17$ | 77MCM/KEB |
| $C_7H_4FO_2^-$ pF-benzoate ⁻ | | -620 \pm 14 ^a | 3.63 \pm 0.24 ^d | 1405 \pm 11 ^g | | $\Delta_f H(AH) = -495 \pm 3$ $BDE(A-H) = 444 \pm 13$ | 77PED/RYL 77MCM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------------------------------|---|---|-----------------------------|--|-------------------------------------|
| $C_7H_4F_3^-$ CF ₃ -phenide ⁻ | -511±24 ^a | 1.6±0.4 ^d | 1618±23 ^g | 1586±21 | IMRB | $\Delta_f H(AH) = -599 \pm 1$ $BDE(A-H) = 460 \pm 13$ | 77PEDI/RYL 79BAR/MCI |
| $C_7H_4F_3NO_2^-$ mCF ₃ -nitrobenzene ⁻ | * -740±18 ^b | 1.41±0.10 1.34±0.11 1.33±0.05 | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = -604 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_7H_4F_3NO_2^-$ oCF ₃ -nitrobenzene ⁻ | * -732±14 ^b | 1.33±0.10 | | | Est TDEq | $\Delta_f H(A) = -604 \pm 4$ | 87KEB/CHO |
| $C_7H_4F_3NO_2^-$ pCF ₃ -nitrobenzene ⁻ | * -746±14 ^b | 1.47±0.10 | | | Est TDEq | $\Delta_f H(A) = -604 \pm 4$ | 87KEB/CHO |
| $C_7H_4F_3O^-$ mCF ₃ -phenoxide ⁻ | * -875±18 ^a | 2.64±0.19 ^d | 1420±10 ^g | 1391±8 | IMRE | $\Delta_f H(AH) = -765 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI |
| $C_7H_4F_3O^-$ pCF ₃ -phenoxide ⁻ | * -885±19 ^a | 2.74±0.24 ^d | 1410±11 ^g | 1381±8 | IMRE | $\Delta_f H(AH) = -765 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 81FUJ/MCI |
| $C_7H_4F_3OS^-$ mSCF ₃ -phenoxide ⁻ | * 2.72±0.20 ^d | | 1411±11 ^g | 1382±8 | IMRE | $BDE(A-H) = 362 \pm 8$ | 86TAF |
| $C_7H_4F_3OS^-$ pSCF ₃ -phenoxide ⁻ | * 2.81±0.23 ^d | | 1403±10 ^g | 1374±8 | IMRE | $BDE(A-H) = 362 \pm 13$ | 86TAF |
| $C_7H_4F_3O_3S^-$ mSO ₂ CF ₃ -phenoxide ⁻ | * 3.06±0.20 ^d | | 1379±11 ^g | 1350±8 | IMRE | $BDE(A-H) = 362 \pm 8$ | 86TAF |
| $C_7H_4F_3O_3S^-$ pSO ₂ CF ₃ -phenoxide ⁻ | * 3.36±0.24 ^d | | 1350±11 ^g | 1321±8 | IMRE | $BDE(A-H) = 362 \pm 13$ | 86TAF |
| $C_7H_4F_4O^-$ 2,3,5,6-tetrafluoroanisole ⁻ | -866±13 ^b | 0.22±0.09 | | | Est ECD | $\Delta_f H(A) = -845 \pm 4$ | 84HER/WEN |
| $C_7H_4NO^-$ mCN-phenoxide ⁻ | * -82±18 ^a | 2.79±0.19 ^d | 1405±10 ^g 1405±10 ^g | 1376±8 1377±8 | IMRE IMRE | $\Delta_f H(AH) = 43 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI 77MCM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|--------|--|-----------|
| $C_7H_4NO^-$ oCN-phenoxide ⁻ | | | | | Est2 | $\Delta_f H(AH) = 25 \pm 13$ $BDE(A-H) = 378 \pm 17$ | |
| • | -105 ± 25^a | 3.00 ± 0.30^d | 1400 ± 12^g | 1369 ± 8 | IMRE | | 81FUJ/MCI |
| | | | 1400 ± 12^g | 1369 ± 8 | IMRE | | 77MCM/KEB |
| $C_7H_4NO^-$ pCN-phenoxide ⁻ | | | | | Est | $\Delta_f H(AH) = 43 \pm 8$ | |
| • | -97 ± 19^a | | 1390 ± 11^g | 1361 ± 8 | IMRE | | 81FUJ/MCI |
| | | | 1392 ± 11^g | 1363 ± 8 | IMRE | | 77MCM/KEB |
| $C_7H_4NO_4^-$ mNO ₂ -benzoate ⁻ | | | | | Est | $\Delta_f H(AH) = -310 \pm 8$ $BDE(A-H) = 444 \pm 13$ | |
| • | -458 ± 19^a | 3.88 ± 0.24^d | 1382 ± 11^g | 1350 ± 8 | IMRE | | 77MCM/KEB |
| | | | 1379 ± 11^g | 1347 ± 8 | IMRE | | 86TAF |
| $C_7H_4N_2O_2^-$ mCN-nitrobenzene ⁻ | | | | | Est | $\Delta_f H(A) = 204 \pm 8$ | |
| • | 53 ± 18^b | 1.56 ± 0.10 | | | TDEq | | 87KEB/CHO |
| | | 1.48 ± 0.11 | | | IMRE | | 85GRI/CAL |
| | | 1.49 ± 0.05 | | | IMRE | | 85FUK/MCI |
| $C_7H_4N_2O_2^-$ oCN-nitrobenzene ⁻ | | | | | Est | $\Delta_f H(A) = 204 \pm 4$ | |
| • | 48 ± 14^b | 1.61 ± 0.10 | | | TDEq | | 87KEB/CHO |
| $C_7H_4N_2O_2^-$ pCN-nitrobenzene ⁻ | | | | | Est | $\Delta_f H(A) = 202 \pm 4$ | |
| • | 36 ± 14^b | 1.72 ± 0.10 | | | TDEq | | 87KEB/CHO |
| | | 1.65 ± 0.11 | | | IMRE | | 85GRI/CAL |
| $C_7H_4N_3O_6^-$ 2,4,6-triNO ₂ -C ₆ H ₂ CH ₂ ⁻ | | | | | | $\Delta_f H(AH) = 39 \pm 2$ | 77PEL |
| | | | 1379 ± 25^g | 1351 ± 21 | IMRB | | 74DZI/CAR |
| $C_7H_5ClNO^-$ pCN-C ₆ H ₄ OH · Cl ⁻ | | | | | | | |
| | | | 141 ± 8 | 109 ± 8 | TDEq | | 77CUM/FRE |
| $C_7H_5ClO_2^-$ 2-Cl-5-Me-benzoquinone ⁻ | | | | | Est2 | $\Delta_f H(A) = -180 \pm 17$ | |
| • | -375 ± 21^b | 2.02 ± 0.05 | | | IMRE | | 85FUK/MCI |
| $C_7H_5FO^-$ mF-benzaldehyde ⁻ | | | | | Est | $\Delta_f H(A) = -230 \pm 8$ | |
| | -295 ± 13^b | 0.67 ± 0.05 | | | ECD | | 75WEN/KAO |
| $C_7H_5FO^-$ oF-benzaldehyde ⁻ | | | | | Est | $\Delta_f H(A) = -230 \pm 21$ | |
| | -292 ± 25^b | 0.64 ± 0.04 | | | ECD | | 75WEN/KAO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|----------------------------------|---|---|----------------------------------|--|--------------------------------------|
| $C_7H_5FO^-$ pF-benzaldehyde ⁻ • -273 ± 10^b | | 0.49 ± 0.02 | | | Est ECD | $\Delta_f H(A) = -226 \pm 8$ | 75WEN/KAO |
| $C_7H_5F_3N^-$ mCF ₃ -anilide ⁻ • -621 ± 18^a | | | 1493 ± 10^g | 1463 ± 8 1472 ± 8 | Est IMRE IMRE ⁰ | $\Delta_f H(AH) = -585 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_5F_3N^-$ pCF ₃ -anilide ⁻ • -636 ± 19^a | | | 1479 ± 10^g | 1448 ± 8 1457 ± 8 | Est IMRE IMRE ⁰ | $\Delta_f H(AH) = -585 \pm 10$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_5F_3NO_2S^-$ mSO ₂ CF ₃ -anilide ⁻ • 2.37 ± 0.24^d | | | 1451 ± 11^g | 1421 ± 8 | IMRE | $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_7H_5F_3NO_2S^-$ pSO ₂ CF ₃ -anilide ⁻ • 2.73 ± 0.24^d | | | 1417 ± 11^g | 1386 ± 8 | IMRE | $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_7H_5F_3NS^-$ mSCF ₃ -anilide ⁻ • 2.01 ± 0.24^d | | | 1487 ± 11^g | 1456 ± 8 | IMRE | $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_7H_5F_3NS^-$ pSCF ₃ -anilide ⁻ • 2.19 ± 0.24^d | | | 1469 ± 11^g | 1438 ± 8 | IMRE | $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_7H_5N^-$ benzotrile ⁻ 194 ± 4^b | | 0.26 ± 0.02 0.3 ± 0.1 | | | ECD ECD | $\Delta_f H(A) = 219 \pm 2$ | 82CHU/INGU 75WEN/KAO 83ZLA/LEE |
| $C_7H_5NO_3^-$ mCHO-nitrobenzene ⁻ • -188 ± 14^b | | 1.41 ± 0.10 | | | Est TDEq | $\Delta_f H(A) = -52 \pm 4$ | 87KEB/CHO |
| $C_7H_5NO_3^-$ oCHO-nitrobenzene ⁻ • -198 ± 14^b | | 1.51 ± 0.10 | | | Est TDEq | $\Delta_f H(A) = -52 \pm 4$ | 87KEB/CHO |
| $C_7H_5NO_3^-$ pCHO-nitrobenzene ⁻ • -213 ± 14^b | | 1.67 ± 0.10 | | | Est TDEq | $\Delta_f H(A) = -52 \pm 4$ | 87KEB/CHO |
| $C_7H_5N_2^-$ indazolide ⁻ • | | | 1456 ± 11^g | 1424 ± 8 | IMRE | | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------------|---|---|--------------|--|--------------------------------------|
| $C_7H_5N_2^-$ mCN-anilide ⁻ • 170 ± 14^a | | 2.11 ± 0.23^d | 1477 ± 10^g | 1446 ± 8 | Est IMRE | $\Delta_f H(AH) = 223 \pm 4$ $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_7H_5N_2^-$ pCN-anilide ⁻ • 146 ± 14^a | | | 1460 ± 10^g | 1429 ± 8 | Est IMRE | $\Delta_f H(AH) = 216 \pm 4$ | 86TAF |
| $C_7H_5O_2^-$ benzoate ⁻ • -407 ± 14^a | | | 1418 ± 12^g 1423 ± 12^g | 1388 ± 8 1393 ± 8 | IMRE IMRE | $\Delta_f H(AH) = -294 \pm 2$ $BDE(A-H) = 444 \pm 13$ | 77PEDI/RYL 78CUM/KEB 81FUJ/MCI |
| $C_7H_5O_2^-$ mCHO-phenoxide ⁻ • -319 ± 18^a | | 2.58 ± 0.19^d | 1425 ± 10^g | 1396 ± 8 | Est IMRE | $\Delta_f H(AH) = -213 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI |
| $C_7H_5O_2^-$ pCHO-phenoxide ⁻ • -350 ± 19^a | | | 1393 ± 11^g | 1364 ± 8 | Est IMRE | $\Delta_f H(AH) = -213 \pm 8$ | 81FUJ/MCI |
| $C_7H_5O_3^-$ mOH-benzoate ⁻ • -587 ± 19^a | | 3.54 ± 0.24^d | 1414 ± 11^g | 1382 ± 8 | Est IMRE | $\Delta_f H(AH) = -470 \pm 8$ $BDE(A-H) = 444 \pm 13$ | 77MCM/KEB |
| $C_7H_5O_3^-$ oOH-benzoate ⁻ • -660 ± 13^a | | | 1365 ± 12^g | 1332 ± 8 | Est IMRE | $\Delta_f H(AH) = -495$ | 77PEDI/RYL 77MCM/KEB |
| $C_7H_5O_3^-$ pOH-benzoate ⁻ • -598 ± 19^a | | 3.66 ± 0.24^d | 1402 ± 11^g | 1371 ± 8 | Est IMRE | $\Delta_f H(AH) = -470 \pm 8$ $BDE(A-H) = 444 \pm 13$ | 77MCM/KEB |
| $C_7H_6Cl^-$ mCl-C ₆ H ₄ CH ₂ ⁻ • 53 ± 19^a | | 1.07 ± 0.20^d | 1565 ± 11^g | 1535 ± 8 | Est IMRE | $\Delta_f H(AH) = 18 \pm 8$ $BDE(A-H) = 356 \pm 8$ | 80PRY 83CAL/BAR |
| $C_7H_6Cl^-$ pCl-C ₆ H ₄ CH ₂ ⁻ • 53 ± 19^a | | 1.11 ± 0.20^d | 1565 ± 11^g | 1535 ± 8 | Est IMRE | $\Delta_f H(AH) = 18 \pm 8$ $BDE(A-H) = 360 \pm 8$ | 80PRY 83CAL/BAR |
| $C_7H_6F^-$ mF-C ₆ H ₄ CH ₂ ⁻ • -109 ± 19^a | | 1.03 ± 0.20^d | 1571 ± 11^g | 1541 ± 8 | Est IMRE | $\Delta_f H(AH) = -150 \pm 8$ $BDE(A-H) = 358 \pm 8$ | 83CAL/BAR |
| $C_7H_6F^-$ pF-C ₆ H ₄ CH ₂ ⁻ • -90 ± 12^a | | 0.87 ± 0.24^d | 1588 ± 11^g | 1558 ± 8 | Est IMRE | $\Delta_f H(AH) = -148 \pm 1$ $BDE(A-H) = 360 \pm 13$ | 77PEDI/RYL 83CAL/BAR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|---------------------------|--|------------------------|
| $C_7H_6FNO_2^-$ 2-Me-4-F-nitrobenzene ⁻ • -246 ± 21 ^b 0.95 ± 0.05 | | | | | Est2 IMRE | $\Delta_f H(A) = -155 \pm 17$ | 85FUK/MCI |
| $C_7H_6FO^-$ mF-C ₆ H ₃ OMe ⁻ • -208 ± 23 ^a | | | 1589 ± 15 ^g | 1556 ± 13 | IMRB | Est $\Delta_f H(AH) = -267 \pm 8$ | 83ING/NIB |
| $C_7H_6FO^-$ oF-C ₆ H ₃ OMe ⁻ • -175 ± 32 ^a | | | 1618 ± 23 ^g | 1586 ± 21 | IMRB | Est $\Delta_f H(AH) = -264 \pm 8$ | 83ING/NIB |
| $C_7H_6FO^-$ pF-C ₆ H ₃ OMe ⁻ • -178 ± 32 ^a | | | 1618 ± 23 ^g | 1586 ± 21 | IMRB | Est $\Delta_f H(AH) = -267 \pm 8$ | 83ING/NIB |
| $C_7H_6NO^-$ HN=C(Ph)O ⁻ • -149 ± 23 ^a | | | 1482 ± 11 ^g | 1452 ± 8 | IMRE | $\Delta_f H(AH) = -101 \pm 13$ | 82TOR/SAB2 86TAF |
| $C_7H_6NO^-$ PhCH=NO ⁻ • 54 ± 28 ^a | | | 1477 ± 20 ^g | 1447 ± 8 1453 ± 8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = 108 \pm 8$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_6NO^-$ mNO-C ₆ H ₄ CH ₂ ⁻ • 64 ± 15 ^a 1.38 ± 0.24 ^d | | | 1539 ± 11 ^g | 1511 ± 8 | IMRE | Est $\Delta_f H(AH) = 55 \pm 4$ $BDE(A-H) = 360 \pm 13$ | 86TAF |
| $C_7H_6NO^-$ pCHO-anilide ⁻ • -101 ± 15 ^a | | | 1463 ± 11 ^g | 1432 ± 8 | IMRE | Est $\Delta_f H(AH) = -34 \pm 4$ | 86TAF |
| $C_7H_6NO^-$ pNO-C ₆ H ₄ CH ₂ ⁻ • -3 ± 15 ^a | | | 1472 ± 11 ^g | 1444 ± 8 | IMRE | Est $\Delta_f H(AH) = 55 \pm 4$ | 86TAF |
| $C_7H_6NO_2^-$ mNH ₂ -benzoate ⁻ • -393 ± 15 ^a 3.42 ± 0.24 ^d | | | 1426 ± 11 ^g | 1395 ± 8 | IMRE | $\Delta_f H(AH) = -289 \pm 4$ $BDE(A-H) = 444 \pm 13$ | 77NAB/SAB 77MCM/KEB |
| $C_7H_6NO_2^-$ mNO ₂ -C ₆ H ₄ CH ₂ ⁻ • 19 ± 15 ^a | | | 1518 ± 11 ^g | 1488 ± 8 | IMRE | $\Delta_f H(AH) = 31 \pm 4$ $BDE(A-H) = 360 \pm 13$ | 77PED/RYL 83CAL/BAR |
| $C_7H_6NO_2^-$ oNH ₂ -benzoate ⁻ • -422 ± 14 ^a | | | 1406 ± 12 ^g | 1377 ± 8 | IMRE | $\Delta_f H(AH) = -298 \pm 2$ | 77NAB/SAB 77MCM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|---|--|--|
| $C_7H_6NO_2^-$ oNO ₂ -C ₆ H ₄ CH ₂ ⁻ • 13±23 ^a | | | 1490±11 ^g | 1459±8 | Est2 IMRE | $\Delta_f H(AH) = 53 \pm 13$ | 86TAF |
| $C_7H_6NO_2^-$ pNH ₂ -benzoate ⁻ • -397±15 ^a | 3.40±0.24 ^d | | 1427±11 ^g | 1397±8 | IMRE | $\Delta_f H(AH) = -294 \pm 4$ $BDE(A-H) = 444 \pm 13$ | 77NAB/SAB 77MCM/KEB |
| $C_7H_6NO_2^-$ pNO ₂ -C ₆ H ₄ CH ₂ ⁻ • -25±14 ^a • -23±14 ^a | | | 1475±10 ^g 1477±10 ^g | 1445±8 1447±8 | IMRE IMRE | $\Delta_f H(AH) = 30 \pm 4$ | 77PED/RYL 86TAF 78CUM/KEB |
| $C_7H_6NO_3^-$ 2-Me-4-NO ₂ -phenoxide ⁻ • -297±21 ^a | | | 1375±12 ^g | 1343±8 | Est IMRE | $\Delta_f H(AH) = -142 \pm 8$ | 81FUJ/MCI |
| $C_7H_6N_2O_4^-$ 2-Me-3-NO ₂ -nitrobenzene ⁻ • 69±17 ^b | 1.39±0.05 | | | | Est2 IMRE | $\Delta_f H(A) = 203 \pm 13$ | 85FUK/MCI |
| $C_7H_6O^-$ benzaldehyde ⁻ -78±3 ^b | 0.429±0.009 0.39±0.05 0.42±0.01 | | | | ECD ECD ECD | $\Delta_f H(A) = -37 \pm 2$ | 77PED/RYL 75WEN/KAO 83ZLA/LBE 67WEN/CHE |
| $C_7H_6O_2^-$ methylbenzoquinone ⁻ • -334±18 ^b | 1.85±0.10 1.75±0.11 1.76±0.05 | | | | Est TDEq IMRE IMRE | $\Delta_f H(A) = -155 \pm 8$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_7H_7^-$ PhCH ₂ ⁻ • 113±10 ^a • | 0.90±0.15 ^d 0.863±0.013 0.885±0.065 2.35±0.07 1.1 0.8 0.8 | | 1593±10 ^g 1609±31 ^g | 1564±8 1579±29 1558±8 | IMRE LPD LPD IMRB SI SI SI SI IMRE ^o | $\Delta_f H(AH) = 50$ $BDE(A-H) = 368 \pm 4$ | 77PED/RYL 82MCM/GOL 79BAR/SCO 84DRZ/BRA2 75RIC/STE 71BOH/YOU 76FAI/JOY 72PAG 69PAG/GOO 68GAI/PAG 79BAR/SCO |
| $C_7H_7^-$ cycloheptatrienide ⁻ • 223±13 ^a | 0.49±0.21 ^d | | 1570±12 ^g | 1545±8 1539±8 | IMRE IMRE ^o | $\Delta_f H(AH) = 183 \pm 1$ $BDE(A-H) = 305 \pm 8$ | 77PED/RYL 82MCM/GOL 79BAR/SCO 79BAR/SCO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|--|-------------|---|---|------------------------------------|---|---|
| $C_7H_7^-$ norbornadienide ⁻ | | | | | | $\Delta_f H(AH) = 238 \pm 4$ | 80ROGI/CHO |
| • 380 ± 14^a | | | 1672 ± 10^g | 1637 ± 6 1628 ± 21 | IMRB IMRB | Between EtNH ₂ , nPrNH ₂ | 86LEE/SQU 81WRI/BEA |
| $C_7H_7ClNO_2^-$ pNO ₂ -C ₆ H ₄ CH ₃ · Cl ⁻ | | | | 31 | TDEq | | 82FRE/IKU |
| $C_7H_7F_2^-$ PhCH ₂ F · F ⁻ | | | | | | | |
| • -230^c | | | 102 ± 8^g | 69 ± 8 | IMRE | | 83LAR/MCM |
| $C_7H_7NO_2^-$ mMe-nitrobenzene ⁻ | | | | | | $\Delta_f H(A) = 31 \pm 4$ | |
| • -65 ± 14^b | 0.99 ± 0.10 0.93 ± 0.11 0.92 ± 0.05 0.8 ± 0.1 | | | | Est TDEq IMRE IMRE ECD | | 87KEB/CHO 85GRI/CAL 85FUK/MCI 83ZLA/LEE |
| $C_7H_7NO_2^-$ oMe-nitrobenzene ⁻ | | | | | | $\Delta_f H(A) = 53 \pm 8$ | |
| • -36 ± 18^b | 0.92 ± 0.10 0.87 ± 0.11 0.89 ± 0.05 | | | | TDEq IMRE IMRE | | 77PEDI/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_7H_7NO_2^-$ pMe-nitrobenzene ⁻ | | | | | | $\Delta_f H(A) = 31 \pm 4$ | |
| • -61 ± 13^b | 0.95 ± 0.10 0.89 ± 0.11 0.91 ± 0.05 | | | | TDEq IMRE IMRE | | 77PEDI/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_7H_7NO_3^-$ 3-Me-4-NO ₂ -phenoxide ⁻ | | | | | | $\Delta_f H(AH) = -142 \pm 17$ | |
| • -292 ± 28^a | | | 1380 ± 11^g | 1350 ± 8 | IMRE | | 81FUJ/MCI |
| $C_7H_7NO_3^-$ mOMe-nitrobenzene ⁻ | | | | | | $\Delta_f H(A) = -90 \pm 4$ | |
| • -191 ± 14^b | 1.04 ± 0.10 0.98 ± 0.11 | | | | Est TDEq IMRE | | 87KEB/CHO 85GRI/CAL |
| $C_7H_7NO_3^-$ pOMe-nitrobenzene ⁻ | | | | | | $\Delta_f H(A) = -90 \pm 4$ | |
| • -178 ± 14^b | 0.91 ± 0.10 0.85 ± 0.11 | | | | Est TDEq IMRE | | 87KEB/CHO 85GRI/CAL |
| $C_7H_7O^-$ PhCH ₂ O ⁻ | | | | | | $\Delta_f H(AH) = -100 \pm 1$ $BDE(A-H) = 436 \pm 4$ | 77PEDI/RYL |
| • -82 ± 13^a | 2.07 ± 0.17^d 2.142 ± 0.013 | | 1548 ± 12^g | 1520 ± 8 1519 ± 8 | IMRE LPD IMRE ^o | | 79BAR/SCO 85MOY/DOD 79BAR/SCO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|--------|---|-------------------------|
| $C_7H_7O^-$ mMe-phenoxide ⁻ | • -200 ± 11^a | 2.19 ± 0.19^d | 1463 ± 10^g | 1434 ± 8 | IMRE | $\Delta_f H(AH) = -132 \pm 1$ $BDE(A-H) = 362 \pm 8$ | 79KUD/KUD |
| | | | 1467 ± 10^g | 1438 ± 8 | IMRE | | 81FUJ/MCI 77MCM/KEB |
| $C_7H_7O^-$ oMe-phenoxide ⁻ | • -192 ± 13^a • $< 2.36 \pm 0.06$ | 2.19 ± 0.26^d | 1462 ± 12^g | 1431 ± 8 | IMRE | $\Delta_f H(AH) = -124 \pm 1$ $BDE(A-H) = 362 \pm 13$ | 79KUD/KUD |
| | | | | | PD | | 81FUJ/MCI 75RIC/STB2 |
| | | | 1465 ± 12^g | 1434 ± 8 | IMRE | | 77MCM/KEB |
| $C_7H_7O^-$ pMe-phenoxide ⁻ | • -190 ± 11^a | 2.16 ± 0.23^d | 1466 ± 10^g | 1437 ± 8 | IMRE | $\Delta_f H(AH) = -125 \pm 2$ $BDE(A-H) = 362 \pm 13$ | 79KUD/KUD |
| | | | 1466 ± 10^g | 1437 ± 8 | IMRE | | 81FUJ/MCI 79BAR/SCO |
| | | | 1471 ± 10^g | 1442 ± 8 | IMRE | | 77MCM/KEB |
| $C_7H_7O_2^-$ mOMe-phenoxide ⁻ | • -324 ± 18^a | 2.26 ± 0.19^d | 1456 ± 10^g | 1427 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -250 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI |
| | | | 1459 ± 10^g | 1431 ± 8 | IMRE | | 77MCM/KEB |
| $C_7H_7O_2^-$ oOMe-phenoxide ⁻ | • | | | 1433 ± 8 | IMRE | <i>Est2</i> $\Delta_f H(AH) = -264 \pm 17$ | 77MCM/KEB |
| | | | | | | | |
| $C_7H_7O_2^-$ pOMe-phenoxide ⁻ | • -306 ± 18^a | 2.15 ± 0.23^d | 1466 ± 10^g | 1437 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -242 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 81FUJ/MCI |
| | | | 1469 ± 10^g | 1440 ± 8 | IMRE | | 77MCM/KEB |
| $C_7H_7O_2S^-$ PhSO ₂ CH ₂ ⁻ | • -266 ± 13^a | | 1518 ± 10^g | 1487 ± 8 | IMRE | $\Delta_f H(AH) = -254 \pm 3$ | 77PEDI/RYL 78CUM/KEB |
| | | | | | | | |
| $C_7H_7O_2S^-$ mSOMe-phenoxide ⁻ | • -297 ± 18^a | 2.55 ± 0.19^d | 1428 ± 10^g | 1399 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -194 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI |
| | | | | | | | |
| $C_7H_7O_2S^-$ pSOMe-phenoxide ⁻ | • -312 ± 19^a | | 1412 ± 11^g | 1383 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -194 \pm 8$ | 81FUJ/MCI |
| | | | | | | | |
| $C_7H_7O_3S^-$ mSO ₂ Me-phenoxide ⁻ | • -567 ± 18^a | 2.77 ± 0.19^d | 1406 ± 10^g | 1377 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -443 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI |
| | | | | | | | |
| $C_7H_7O_3S^-$ pSO ₂ Me-phenoxide ⁻ | • -587 ± 19^a | | 1385 ± 11^g | 1356 ± 8 | IMRE | <i>Est</i> $\Delta_f H(AH) = -443 \pm 8$ | 81FUJ/MCI |
| | | | | | | | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|---------------------------|--|-------------------------------------|
| $C_7H_7S^-$ MeSC ₆ H ₄ ⁻ 185±25 ^a | | | 1617±23 ^g | 1586±21 1583±33 | IMRB IMRB ^o | $\Delta_f H(AH) = 98 \pm 1$ $BDE(A-H) = 460 \pm 13$ | 77PED/RYL 85ING/NIB 85ING/NIB |
| $C_7H_7S^-$ PhSCH ₂ ⁻ • 164±12 ^a | | | 1597±11 ^g | 1566±8 1560±8 | IMRE IMRE ^o | $\Delta_f H(AH) = 98 \pm 1$ | 77PED/RYL 85ING/NIB 85ING/NIB |
| $C_7H_8Cl^-$ PhMe \cdots Cl ⁻ | | | | 17 | TDEq | | 82FRE/IKU |
| $C_7H_8ClO^-$ PhOMe \cdots Cl ⁻ | | | | 31 | TDEq | | 82FRE/IKU |
| $C_7H_8ClO^-$ pMe-C ₆ H ₄ OH \cdots Cl ⁻ • -453±11 ^c | | | 101±8 | 69±8 | TDEq | | 77CUM/FRE |
| $C_7H_8N^-$ PhNMe ⁻ • 81±15 ^a | 1.57±0.20 ^d | | 1526±11 ^g | 1496±8 | IMRE | $\Delta_f H(AH) = 85 \pm 4$ $BDE(A-H) = 366 \pm 8$ | 78COL/BEN 82MCM/GOL 86TAF |
| $C_7H_8N^-$ mMe-anilide ⁻ • 59±18 ^a | 1.51±0.23 ^d | | 1535±10 ^g | 1505±8 1507±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = 54 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_8N^-$ pMe-anilide ⁻ • 65±15 ^a | 1.49±0.24 ^d | | 1537±11 ^g | 1507±8 1510±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = 59 \pm 4$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_8NO^-$ pOMe-anilide ⁻ • -53±18 ^a | 1.50±0.23 ^d | | 1536±10 ^g | 1505±8 1509±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = -59 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_8NO_2S^-$ mSO ₂ Me-anilide ⁻ • -291±15 ^a | 2.13±0.24 ^d | | 1475±11 ^g | 1445±8 | IMRE | Est $\Delta_f H(AH) = -236 \pm 4$ $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_7H_8NO_2S^-$ pSO ₂ Me-anilide ⁻ • -312±15 ^a | 2.34±0.24 ^d | | 1455±11 ^g | 1424±8 | IMRE | Est $\Delta_f H(AH) = -236 \pm 4$ $BDE(A-H) = 368 \pm 13$ | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------------|---|---|----------------------------------|--|-------------------------|
| $C_7H_8NS^-$ mSMe-anilide ⁻ • | 89 ± 18^a | 1.71 ± 0.23^d | 1515 ± 10^g | 1484 ± 8 1492 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = 104 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_9^-$ heptatrienide ⁻ • | | 1.27 ± 0.03 | | | Est PD | $\Delta_f H(AH) = 133 \pm 4$ | 78ZIM/GYG |
| $C_7H_9^-$ norbornenide ⁻ 242 ± 19^a | | | 1682 ± 15^g | 1648 ± 13 | IMRB | $\Delta_f H(AH) = 90 \pm 4$ between NH_3 , $EtNH_2$ | 80ROGI/CHO 86LEE/SQU |
| $C_7H_9O^-$ 2-norbornanone enolate ⁻ • | | 1.61 ± 0.05 | | | PD | $\Delta_f H(AH) = -168 \pm 3$ | 78STE 78ZIM/JAC |
| $C_7H_{11}O^-$ 2,5-diMe-cyclopentanone enolate ⁻ • | | 1.49 ± 0.04 | | | Est PD | $\Delta_f H(AH) = -272 \pm 4$ | 78ZIM/JAC |
| $C_7H_{11}O^-$ cycloheptanone enolate ⁻ • | | 1.48 ± 0.04 | | | PD | $\Delta_f H(AH) = -248 \pm 2$ | 77PED/RYL 78ZIM/JAC |
| $C_7H_{11}O_4^-$ $HC(CO_2Et)_2^-$ • | -912 ± 12^a | | 1457 ± 10^g | 1432 ± 8 | Est IMRE | $\Delta_f H(AH) = -839 \pm 2$ | 78CUM/KEB |
| $C_7H_{13}O^-$ $EtCH=C(nPr)O^-$ • | -296 ± 18^a | 1.72 ± 0.06 | 1535 ± 14^c | | Est PD | $\Delta_f H(AH) = -301 \pm 4$ $BDE(A-H) = 389 \pm 8$ | 77ZIM/REE |
| $C_7H_{13}O^-$ $Me_2C=C(iPr)O^-$ • | -307 ± 19^a | 1.47 ± 0.05 | 1535 ± 18^c | 1505 ± 23^h | PD | $\Delta_f H(AH) = -311 \pm 1$ $BDE(A-H) = 364 \pm 13$ | 77PED/RYL 77ZIM/REE |
| $C_7H_{13}O^-$ $c-C_6H_{11}-CH_2O^-$ • | -271 ± 14^a | 1.90 ± 0.19^d | 1564 ± 10^g | 1536 ± 8 | Est IMRE | $\Delta_f H(AH) = -305 \pm 4$ $BDE(A-H) = 435 \pm 8$ | 86TAF |
| $C_7H_{13}S^-$ $c-C_6H_{11}-CH_2S^-$ • | -171 ± 16^a | 2.09 ± 0.23^d | 1475 ± 12^g | 1449 ± 8 | Est IMRE | $\Delta_f H(AH) = -116 \pm 4$ $BDE(A-H) = 364 \pm 10$ | 86TAF |
| $C_7H_{15}O^-$ $(iPr)_2CHO^-$ • | -348 ± 15^a | 2.1 ± 0.2^d | 1551 ± 11 1549 ± 10 | 1523 ± 13^h | Est CIDC CIDC ^o | $\Delta_f H(AH) = -369 \pm 4$ $BDE(A-H) = 438 \pm 4$ | 83BOA/HOU 83BOA/HOU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|------------------------------------|---|---|----------------------------------|---|-------------------------------------|
| $C_7H_{15}O^-$ Et_3CO^- | -344 ± 15^a | 2.1 ± 0.2^d | 1552 ± 11 1549 ± 10 | 1524 ± 13^h | Est CIDC CIDC ^o | $\Delta_f H(AH) = -366 \pm 4$ $BDE(A-H) = 440 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_7H_{15}O^-$ $nBuC(Me)_2O^-$ | -348 ± 15^a | 2.0 ± 0.2^d | 1555 ± 11 1552 ± 10 | 1527 ± 13^h | Est CIDC CIDC ^o | $\Delta_f H(AH) = -373 \pm 4$ $BDE(A-H) = 440 \pm 4$ | 83BOA/HOU 83BOA/HOU |
| $C_7H_{15}O^-$ $nC_7H_{15}O^-$ | -307 ± 13^a | 2.0 ± 0.2^d | 1559 ± 11 1555 ± 10 | 1531 ± 13^h | CIDC CIDC ^o | $\Delta_f H(AH) = -336 \pm 2$ $BDE(A-H) = 436 \pm 4$ | 77PED/RYL 83BOA/HOU 83BOA/HOU |
| $C_7H_{15}O^-$ $tBuCH(Et)O^-$ | -353 ± 16^a | 2.10 ± 0.17^d | 1548 ± 12^g | 1520 ± 8 1519 ± 8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -371 \pm 4$ $BDE(A-H) = 438 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_7H_{15}OS_2^-$ $MeOH \cdots 5,5\text{-diMe-1,3-dithianide}^-$ | -287 ± 38^c | | 62 ± 10^g | 34 ± 7 | IMRE | | 84CAL/ROZ |
| $C_7H_{17}O_2^-$ $EtOH \cdots tBuCH_2O^-$ | -605 ± 24^c | | 80 ± 10^g | 53 ± 8 | IMRE | | 84CAL/ROZ |
| $C_7H_{17}O_2^-$ $nPrOH \cdots tBuO^-$ | -615 ± 23^c | | 85 ± 10^g | 57 ± 8 | IMRE | | 84CAL/ROZ |
| $C_8F_4N_2^-$ $pCN\text{-perfluorobenzonitrile}^-$ | -599 ± 26^b | 1.89 ± 0.10 1.89 ± 0.10 | | | Est TDEq IMRE | $\Delta_f H(A) = -417 \pm 17$ | 87KEB/CHO 86CHO/GRI |
| $C_8HN_2O_2^-$ $2,3\text{-diCN-benzoquinonide}^-$ | | 1.82 ± 0.09 | | | SI | | 66FAR/PAG |
| $C_8H_3F_5O^-$ $C_6F_5COCH_3^-$ | -1143 ± 26^b | 0.94 ± 0.10 0.94 ± 0.10 | | | Est TDEq IMRE | $\Delta_f H(A) = -1052 \pm 17$ | 87KEB/CHO 86CHO/GRI |
| $C_8H_3F_6NO_2^-$ $3,5\text{-diCF}_3\text{-nitrobenzene}^-$ | -1449 ± 14^b | 1.79 ± 0.10 | | | Est TDEq | $\Delta_f H(A) = -1276 \pm 4$ | 87KEB/CHO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|----------------------------|--|-------------------------------------|
| $C_8H_3F_6O^-$ 3,5-diCF ₃ -phenoxide ⁻ • -1636±28 ^a 3.05±0.24 ^d | | | 1380±11 ^g | 1351±8 | Est IMRE | $\Delta_f H(AH) = -1485 \pm 17$ $BDE(A-H) = 362 \pm 13$ | 86TAF |
| $C_8H_4F_3N^-$ mCF ₃ -benzotrile ⁻ • -499±18 ^b 0.67±0.10 0.67±0.09 | | | | | Est TDEq TDEq | $\Delta_f H(A) = -434 \pm 8$ | 87KEB/CHO 86CHO/KEB |
| $C_8H_4F_3N^-$ oCF ₃ -benzotrile ⁻ • -519±14 ^b 0.70±0.10 | | | | | Est TDEq | $\Delta_f H(A) = -452 \pm 4$ | 87KEB/CHO |
| $C_8H_4F_3N^-$ pCF ₃ -benzotrile ⁻ • -525±14 ^b 0.76±0.10 | | | | | Est TDEq | $\Delta_f H(A) = -452 \pm 4$ | 87KEB/CHO |
| $C_8H_4F_3O_2^-$ mCF ₃ -benzoate ⁻ • -1114±19 ^a 3.77±0.24 ^d | | | 1392±11 ^g | 1361±8 | Est IMRE | $\Delta_f H(AH) = -976 \pm 8$ $BDE(A-H) = 444 \pm 13$ | 86TAF |
| $C_8H_4F_3O_2^-$ pCF ₃ -benzoate ⁻ • -1115±19 ^a 3.78±0.24 ^d | | | 1391±11 ^g | 1361±8 | Est IMRE | $\Delta_f H(AH) = -976 \pm 8$ $BDE(A-H) = 444 \pm 13$ | 86TAF |
| $C_8H_4F_6N^-$ 3,5-diCF ₃ -anilide ⁻ • -1377±19 ^a 2.33±0.24 ^d | | | 1456±11 ^g | 1425±8 | Est IMRE | $\Delta_f H(AH) = -1302 \pm 8$ $BDE(A-H) = 368 \pm 13$ | 86TAF |
| $C_8H_4NO_2^-$ mCN-benzoate ⁻ • -309±23 ^a 3.90±0.24 ^d | | | 1379±11 ^g | 1348±8 | Est IMRE | $\Delta_f H(AH) = -158 \pm 13$ $BDE(A-H) = 444 \pm 13$ | 77MCM/KEB |
| $C_8H_4NO_2^-$ pCN-benzoate ⁻ • -314±23 ^a 3.95±0.24 ^d | | | 1374±11 ^g | 1345±8 | Est IMRE | $\Delta_f H(AH) = -158 \pm 13$ $BDE(A-H) = 444 \pm 13$ | 77MCM/KEB |
| $C_8H_4N_2^-$ mCN-benzotrile ⁻ • 275±12 ^b 0.91±0.10 0.91±0.09 | | | | | Est TDEq TDEq | $\Delta_f H(A) = 363 \pm 2$ | 80SAT/SAK 87KEB/CHO 86CHO/KEB |
| $C_8H_4N_2^-$ o-CN-benzotrile ⁻ • 271±22 ^b 0.95±0.10 0.95±0.09 1.1±0.1 | | | | | Est2 TDEq TDEq SI | $\Delta_f H(A) = 363 \pm 13$ | 87KEB/CHO 86CHO/KEB 67FAR/PAG |
| $C_8H_4N_2^-$ p-CN-benzotrile ⁻ • 257±18 ^b 1.10±0.10 1.10±0.09 | | | | | Est2 TDEq TDEq | $\Delta_f H(A) = 363 \pm 8$ | 87KEB/CHO 86CHO/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|------------------------------------|---|---|---------------------------|--|------------------------|
| $C_8H_4O_3^-$ phthalic anhydride ⁻ | | | | | | | |
| • | -487 ± 12^b | 1.21 ± 0.10 | | | TDEq | | 87KEB/CHO |
| • | | 1.20 ± 0.05 | | | IMRE | | 85FUK/MCI |
| $C_8H_5^-$ PhC≡C ⁻ | | | | | | $\Delta_f H(AH) = 306 \pm 2$ $BDE(A-H) = 552 \pm 21$ | 85DAV/ALL |
| • | 327 ± 15^a | 3.25 ± 0.36^d | 1551 ± 13^g | 1518 ± 8 1518 ± 8 | IMRE IMRE ^o | | 79BAR/SCO 79BAR/SCO |
| $C_8H_5Cl^-$ mCl-C ₆ H ₄ CHCN ⁻ | | | | | | Est $\Delta_f H(AH) = 156 \pm 4$ | |
| • | 68 ± 18^a | | 1441 ± 13^g | 1412 ± 8 | IMRE | | 81FUJ/MCI |
| $C_8H_5Cl^-$ pCl-C ₆ H ₄ CHCN ⁻ | | | | | | Est $\Delta_f H(AH) = 156 \pm 4$ | |
| • | 70 ± 18^a | | 1444 ± 13^g | 1416 ± 8 | IMRE | | 81FUJ/MCI |
| $C_8H_5FN^-$ mF-C ₆ H ₄ CHCN ⁻ | | | | | | Est $\Delta_f H(AH) = -7 \pm 4$ | |
| • | -70 ± 19^a | | 1467 ± 15^g | 1439 ± 8 | IMRE | | 86TAF |
| $C_8H_5FN^-$ pF-C ₆ H ₄ CHCN ⁻ | | | | | | Est $\Delta_f H(AH) = -7 \pm 4$ | |
| • | -77 ± 15^a | | 1460 ± 11^g | 1433 ± 8 | IMRE | | 86TAF |
| $C_8H_5F_3NO^-$ PhN=C(CF ₃)O ⁻ | | | | | | Est2 $\Delta_f H(AH) = -706 \pm 13$ | |
| • | -841 ± 23^a | | 1395 ± 11^g | 1366 ± 8 | IMRE | | 86TAF |
| $C_8H_5NO^-$ mCHO-benzonitrile ⁻ | | | | | | Est $\Delta_f H(A) = 99 \pm 4$ | |
| • | 2 ± 14^b | 1.00 ± 0.10 1.01 ± 0.09 | | | TDEq TDEq | | 87KEB/CHO 86CHO/KEB |
| $C_8H_5NO^-$ pCHO-benzonitrile ⁻ | | | | | | Est $\Delta_f H(A) = 99 \pm 4$ | |
| • | -19 ± 14^b | 1.22 ± 0.10 1.22 ± 0.09 | | | TDEq TDEq | | 87KEB/CHO 86CHO/KEB |
| $C_8H_5N_2O_2^-$ mNO ₂ -C ₆ H ₄ CHCN ⁻ | | | | | | Est $\Delta_f H(AH) = 171 \pm 4$ | |
| • | 53 ± 19^a | | 1412 ± 15^g | 1384 ± 8 | IMRE | | 86TAF |
| $C_8H_5N_2O_2^-$ pNO ₂ -C ₆ H ₄ CHCN ⁻ | | | | | | Est $\Delta_f H(AH) = 171 \pm 4$ | |
| • | 19 ± 18^a | | 1378 ± 13^g | 1350 ± 8 | IMRE | | 81FUJ/MCI |
| $C_8H_5O_3^-$ pCHO-benzoate ⁻ | | | | | | Est $\Delta_f H(AH) = -414 \pm 8$ $BDE(A-H) = 444 \pm 13$ | |
| • | -550 ± 19^a | 3.74 ± 0.24^d | 1395 ± 11^g | 1363 ± 8 | IMRE | | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|----------------------------------|-------------------------------|-------------------------------------|
| $C_8H_6ClO^-$ mCl-C ₆ H ₄ C(=CH ₂)O ⁻ • -152±18 ^a | | | 1495±10 ^g | 1466±8 | Est IMRE | $\Delta_f H(AH) = -116 \pm 8$ | 79BAR/SCO |
| $C_8H_6Cl_2O_2^-$ 2,5-diCl-3,6-diMe-benzoquinone ⁻ • -437±21 ^b 2.14±0.05 | | | | | Est2 IMRE | $\Delta_f H(A) = -230 \pm 17$ | 85FUK/MCI |
| $C_8H_6F_3^-$ mCF ₃ -C ₆ H ₄ CH ₂ ⁻ • -608±18 ^a | | | 1545±10 ^g | 1515±8 | Est IMRE | $\Delta_f H(AH) = -622 \pm 8$ | 83CAL/BAR |
| $C_8H_6F_3^-$ pCF ₃ -C ₆ H ₄ CH ₂ ⁻ • -617±18 ^a | | | 1536±10 ^g | 1505±8 | Est IMRE | $\Delta_f H(AH) = -622 \pm 8$ | 83CAL/BAR |
| $C_8H_6F_3O_2^-$ pSO ₂ CF ₃ -C ₆ H ₄ CH ₂ ⁻ • | | | 1454±11 ^g | 1425±8 | IMRE | | 86TAP |
| $C_8H_6N^-$ PhCHCN ⁻ • 123±18 ^a | | | 1467±13 ^g 1471±13 ^g | 1440±8 1443±8 1451±8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = 186 \pm 4$ | 81FUJ/MCI 78CUM/KEB 79BAR/SCO |
| $C_8H_6N^-$ indolide ⁻ • 89±15 ^a | | | 1461±11 ^g | 1431±8 | IMRE | $\Delta_f H(AH) = 157 \pm 5$ | 77PED/RYL 86TAP |
| $C_8H_6N^-$ mCN-C ₆ H ₄ CH ₂ ⁻ • 198±18 ^a | | | 1545±10 ^g | 1515±8 | Est IMRE | $\Delta_f H(AH) = 183 \pm 8$ | 83CAL/BAR |
| $C_8H_6N^-$ pCN-C ₆ H ₄ CH ₂ ⁻ • 162±19 ^a | | | 1510±11 ^g | 1479±10 | Est IMRE | $\Delta_f H(AH) = 182 \pm 8$ | 83CAL/BAR |
| $C_8H_6O_2^-$ p-CHO-benzaldehyde ⁻ -211 ^b 0.6 | | | | | Est ECD | $\Delta_f H(A) = -157 \pm 8$ | 68KUH/LEV |
| $C_8H_7ClO^-$ mCl-acetophenone ⁻ -173±9 ^b 0.583±0.006 | | | | | Est ECD | $\Delta_f H(A) = -117 \pm 8$ | 69STE/WEN |
| $C_8H_7ClO^-$ pCl-acetophenone ⁻ -172±9 ^b 0.567±0.005 | | | | | Est ECD | $\Delta_f H(A) = -117 \pm 8$ | 69STE/WEN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|------------------------------------|--|--|---|---|---|
| $C_8H_7ClO_2^-$ 2-Cl-3,6-diMe-benzoquinone $^-$ • -398 ± 21^b | | 1.93 ± 0.05 | | | Est2 IMRE | $\Delta_f H(A) = -212 \pm 17$ | 85FUK/MCI |
| $C_8H_7FO^-$ mF-acetophenone $^-$ • -336 ± 11^b | | 0.58 ± 0.03 | | | Est BCD | $\Delta_f H(A) = -280 \pm 8$ | 75WEN/KAO |
| $C_8H_7FO^-$ oF-acetophenone $^-$ • -323 ± 22^b | | 0.442 ± 0.009 | | | Est ECD | $\Delta_f H(A) = -280 \pm 21$ | 75WEN/KAO |
| $C_8H_7FO^-$ pF-acetophenone $^-$ • -318 ± 10^b | | 0.40 ± 0.01 | | | Est ECD | $\Delta_f H(A) = -280 \pm 8$ | 75WEN/KAO |
| $C_8H_7N^-$ oMe-benzonitrile $^-$ • 110 ± 11^b 110 ± 9^b | | 0.70 ± 0.10 0.70 ± 0.09 | | | Est TDEq TDEq | $\Delta_f H(A) = 178 \pm 1$ | 87KEB/CHO 86CHO/KEB |
| $C_8H_7N^-$ pMe-benzonitrile $^-$ • 109 ± 9^b | | 0.76 ± 0.09 | | | Est TDEq | $\Delta_f H(A) = 182 \pm 1$ | 86CHO/KEB |
| $C_8H_7NO_3^-$ mCOMe-nitrobenzene $^-$ • -229 ± 14^b | | 1.31 ± 0.10 | | | Est TDEq | $\Delta_f H(A) = -103 \pm 4$ | 87KEB/CHO |
| $C_8H_7NO_3^-$ oCOMe-nitrobenzene $^-$ • -217 ± 18^b | | 1.38 ± 0.10 | | | Est2 TDEq | $\Delta_f H(A) = -84 \pm 8$ | 87KEB/CHO |
| $C_8H_7NO_3^-$ pCOMe-nitrobenzene $^-$ • -252 ± 14^b | | 1.55 ± 0.10 | | | Est TDEq | $\Delta_f H(A) = -103 \pm 4$ | 87KEB/CHO |
| $C_8H_7O^-$ CH ₂ =C(Ph)O $^-$ • -105 ± 13^a • -101 ± 13^a | | 2.06 ± 0.08 | 1512 ± 11^g 1516 ± 11^g | 1483 ± 8 1487 ± 8 1491 ± 8 | D-EA IMRE PD IMRE IMRE ^o | $\Delta_f H(AH) = -87 \pm 2$ $BDE(A-H) = 399 \pm 18$ | 77PED/RYL 79BAR/SCO 77ZIM/REE 78CUM/KEB 79BAR/SCO |
| $C_8H_7O^-$ PhCH=CHO $^-$ • 2.10 ± 0.08 | | | | | Est PD | $\Delta_f H(AH) = -53 \pm 4$ | 77ZIM/REE |
| $C_8H_7O^-$ mCHO-C ₆ H ₄ CH ₂ $^-$ • -47 ± 19^a | | | 1554 ± 11^g | 1524 ± 8 | IMRE | $\Delta_f H(AH) = -71 \pm 8$ | 83CAL/BAR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|-----------|---|-------------------------------------|
| $C_8H_7O^-$ pCHO-C ₆ H ₄ CH ₂ ⁻ • -100±21 ^a | | | 1505±12 ^g | 1475±10 | IMRE | Est $\Delta_f H(AH) = -75 \pm 8$ | 86TAF |
| $C_8H_7O_2^-$ PhCH ₂ CO ₂ ⁻ • -423±15 ^a | 3.40±0.20 ^d | | 1428±11 ^g | 1398±8 | IMRE | Est $\Delta_f H(AH) = -320 \pm 4$ BDE(A-H) = 444±8 | 86TAF |
| $C_8H_7O_2^-$ mCOMe-phenoxide ⁻ • -361±18 ^a | 2.50±0.19 ^d | | 1433±10 ^g | 1404±8 | IMRE | Est $\Delta_f H(AH) = -264 \pm 8$ BDE(A-H) = 362±8 | 81FUJ/MCI |
| $C_8H_7O_2^-$ mMc-benzoate ⁻ • -437±12 ^a | 3.46±0.24 ^d | | 1422±11 ^g | 1391±8 | IMRE | $\Delta_f H(AH) = -329 \pm 1$ BDE(A-H) = 444±13 | 76COLJ/JIM 77MCM/KEB |
| $C_8H_7O_2^-$ oMe-benzoate ⁻ • -436±13 ^a | | | 1415±12 ^g | 1384±8 | IMRE | $\Delta_f H(AH) = -320 \pm 1$ | 76COLJ/JIM 77MCM/KEB |
| $C_8H_7O_2^-$ pCOMe-phenoxide ⁻ • -390±19 ^a | | | 1404±11 ^g | 1375±8 | IMRE | Est $\Delta_f H(AH) = -264 \pm 8$ | 81FUJ/MCI |
| $C_8H_7O_2^-$ pMe-benzoate ⁻ • -440±12 ^a | 3.46±0.24 ^d | | 1422±11 ^g | 1392±8 | IMRE | $\Delta_f H(AH) = -332 \pm 1$ BDE(A-H) = 444±13 | 76COLJ/JIM 77MCM/KEB |
| $C_8H_7O_3^-$ mCO ₂ Me-phenoxide ⁻ • -559±18 ^a | 2.44±0.19 ^d | | 1439±10 ^g | 1410±8 | IMRE | Est $\Delta_f H(AH) = -468 \pm 8$ BDE(A-H) = 362±8 | 81FUJ/MCI |
| $C_8H_7O_3^-$ mOMe-benzoate ⁻ • -559±12 ^a | 3.51±0.24 ^d | | 1417±11 ^g | 1386±8 | IMRE | $\Delta_f H(AH) = -446 \pm 1$ BDE(A-H) = 444±13 | 78COLJ/JIM 77MCM/KEB |
| $C_8H_7O_3^-$ oOMe-benzoate ⁻ • -567±19 ^a | | | 1415±11 ^g | 1386±8 | IMRE | Est $\Delta_f H(AH) = -452 \pm 8$ | 77MCM/KEB |
| $C_8H_7O_3^-$ pCO ₂ Me-phenoxide ⁻ • -587±19 ^a | | | 1411±11 ^g | 1382±8 | IMRE | Est $\Delta_f H(AH) = -468 \pm 8$ | 81FUJ/MCI |
| $C_8H_8^-$ cyclooctatetraene ⁻ 242±5 ^b | 0.58±0.04 <0.8 | | | | ECD PD | $\Delta_f H(A) = 297 \pm 1$ | 77PED/RYL 69WEN/RIS 79GYG/PET |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|--------------|--|-------------------------------------|
| $C_8H_8ClO^-$ PhCOMe \cdot ·Cl $^-$ | | | | 40 | TDEq | | 82FRE/IKU |
| $C_8H_8NO^-$ PhN=C(Me)O $^-$ | | | | | | $\Delta_f H(AH) = -129 \pm 1$ | 77PED/RYL |
| * -205 ± 11^a | | | 1454 $\pm 10^g$ 1476 $\pm 10^g$ | 1425 ± 8 1447 ± 8 | IMRE IMRE | | 86TAF 78CUM/KEB |
| $C_8H_8NO^-$ mCOMe-anilide $^-$ | | | | | Est | $\Delta_f H(AH) = -100 \pm 4$ $BDE(A-H) = 368 \pm 13$ | |
| * -125 ± 15^a | 1.82 $\pm 0.24^d$ | | 1505 $\pm 11^g$ | 1474 ± 8 | IMRE | | 86TAF |
| $C_8H_8NO^-$ pCOMe-anilide $^-$ | | | | | Est2 | $\Delta_f H(AH) = -88 \pm 4$ | |
| * -148 ± 15^a | | | 1470 $\pm 11^g$ | 1439 ± 8 | IMRE | | 86TAF |
| $C_8H_8NO_2^-$ mCO $_2$ Me-anilide $^-$ | | | | | Est | $\Delta_f H(AH) = -300 \pm 4$ $BDE(A-H) = 368 \pm 13$ | |
| * -322 ± 15^a | 1.78 $\pm 0.24^d$ | | 1509 $\pm 11^g$ | 1478 ± 8 | IMRE | | 86TAF |
| $C_8H_8NO_2^-$ pCO $_2$ Me-anilide $^-$ | | | | | Est | $\Delta_f H(AH) = -300 \pm 4$ | |
| * -356 ± 15^a | | | 1475 $\pm 11^g$ | 1444 ± 8 | IMRE | | 86TAF |
| $C_8H_8O^-$ acetophenone $^-$ | | | | | | $\Delta_f H(A) = -87 \pm 2$ | 77PED/RYL 75WEN/KAO 67WEN/CHE |
| -119 ± 2^b | 0.334 ± 0.004 0.334 ± 0.004 | | | | ECD ECD | | |
| $C_8H_8O^-$ mMe-benzaldehyde $^-$ | | | | | Est | $\Delta_f H(A) = -71 \pm 8$ | |
| -110 ± 10^b | 0.41 ± 0.01 | | | | ECD | | 75WEN/KAO |
| $C_8H_8O^-$ pMe-benzaldehyde $^-$ | | | | | Est | $\Delta_f H(A) = -75 \pm 8$ | |
| -111 ± 10^b | 0.37 ± 0.02 | | | | ECD | | 75WEN/KAO |
| $C_8H_8O_2^-$ 2,5-diMe-benzoquinone $^-$ | | | | | Est2 | $\Delta_f H(A) = -187 \pm 8$ | |
| * -358 ± 18^b | 1.77 ± 0.10 1.72 ± 0.11 | | | | TDEq IMRE | | 87KEB/CHO 85GRI/CAL |
| $C_8H_8O_2^-$ 2,6-diMe-benzoquinone $^-$ | | | | | Est | $\Delta_f H(A) = -187 \pm 8$ | |
| * -359 ± 18^b | 1.78 ± 0.10 1.67 ± 0.05 | | | | TDEq IMRE | | 87KEB/CHO 85FUK/MCI |
| $C_8H_8O_2^-$ mOMe-benzaldehyde $^-$ | | | | | Est | $\Delta_f H(A) = -182 \pm 8$ | |
| -224 ± 13^b | 0.43 ± 0.04 | | | | ECD | | 75WEN/KAO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|---|---|---|------------------------------|--|--|
| $C_8H_8O_2^-$ methyl benzoate ⁻ -305 ^b | | 0.2 | | | ECD | $\Delta_f H(A) = -288 \pm 7$ | 77PED/RYL 68KUH/LEV |
| $C_8H_8O_2^-$ oOH-acetophenone ⁻ • -184 \pm 17 ^b | | 0.86 \pm 0.05 | | | Est2 IMRE | $\Delta_f H(A) = -100 \pm 13$ | 85FUK/MCI |
| $C_8H_8O_4^-$ 2,6-diMeO-benzoquinone ⁻ • | | 1.73 \pm 0.10 | | | TDEq | | 87KEB/CHO |
| $C_8H_9^-$ 2-methylenenorborn-5-en-3-ide ⁻ • | | | 1632 \pm 10 ^g | 1603 \pm 8 | IMRB | | 86LEE/SQU |
| $C_8H_9^-$ PhCHMe ⁻ • 88 \pm 10 ^a | | 0.80 \pm 0.19 ^d | 1589 \pm 10 ^g | 1562 \pm 8 1556 \pm 8 | IMRE IMRE ^o | $\Delta_f H(AH) = 29$ $BDE(A-H) = 354 \pm 8$ | 77PED/RYL 81ROB/STE 79BAR/SCO 79BAR/SCO |
| $C_8H_9^-$ bicyclo[3.2.1]octa-2,6-dien-4-ide ⁻ • | | | 1588 \pm 11 ^g | 1559 \pm 8 | IMRE | | 86LEE/SQU |
| $C_8H_9^-$ mMe-C ₆ H ₄ CH ₂ ⁻ • 82 \pm 12 ^a | | 0.89 \pm 0.22 ^d | 1595 \pm 12 ^g | 1564 \pm 10 | IMRE | $\Delta_f H(AH) = 17$ $BDE(A-H) = 368 \pm 9$ | 77PED/RYL 86HAY/KRU 83CAL/BAR |
| $C_8H_9^-$ pMe-C ₆ H ₄ CH ₂ ⁻ • 86 \pm 12 ^a | | 0.84 \pm 0.22 ^d | 1598 \pm 11 ^g | 1568 \pm 10 | IMRE | $\Delta_f H(AH) = 18 \pm 1$ $BDE(A-H) = 367 \pm 10$ | 77PED/RYL 86HAY/KRU 79BAR/SCO |
| $C_8H_9NO_2^-$ 1,2-diMe-3-nitrobenzene ⁻ • -70 \pm 22 ^b | | 0.86 \pm 0.10 0.81 \pm 0.11 0.86 \pm 0.05 | | | Est2 TDEq IMRE IMRE | $\Delta_f H(A) = 13 \pm 13$ | 87KEB/CHO 85GRI/CAL 85FUK/MCI |
| $C_8H_9NO_2^-$ 1,2-diMe-4-nitrobenzene ⁻ • -85 \pm 13 ^b | | 0.87 \pm 0.05 | | | Est IMRE | $\Delta_f H(A) = -1 \pm 8$ | 85FUK/MCI |
| $C_8H_9NO_2^-$ 1,3-diMe-2-nitrobenzene ⁻ • -33 \pm 17 ^b | | 0.76 \pm 0.05 | | | Est IMRE | $\Delta_f H(A) = 40 \pm 13$ | 85FUK/MCI |
| $C_8H_9NO_2^-$ 1,3-diMe-4-nitrobenzene ⁻ • -60 \pm 13 ^b | | 0.83 \pm 0.05 | | | Est IMRE | $\Delta_f H(A) = 20 \pm 8$ | 85FUK/MCI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-----------------|--|--|--|------------------------|-----------|
| $C_8H_9O^-$ mEt-phenoxide ⁻ * -215 ± 11^a | 2.20 ± 0.19^d | 1461 ± 10^g | 1433 ± 8 | IMRE | $\Delta_f H(AH) = -146 \pm 2$ $BDE(A-H) = 362 \pm 8$ | 77PED/RYL 81FUJ/MCI | |
| $C_8H_9O^-$ pEt-phenoxide ⁻ * -210 ± 10^a | 2.18 ± 0.19^d | 1464 ± 10^g | 1435 ± 8 | IMRE | $\Delta_f H(AH) = -144 \pm 1$ $BDE(A-H) = 362 \pm 8$ | 77PED/RYL 81FUJ/MCI | |
| $C_8H_9OS^-$ pSOMe-C ₆ H ₄ CH ₂ ⁻ * -31 ± 19^a | | 1531 ± 11^g | 1503 ± 8 | IMRE | Est $\Delta_f H(AH) = -32 \pm 8$ | 86TAF | |
| $C_8H_9O_2S^-$ PhSO ₂ CHMe ⁻ * -283 ± 13^a | | 1527 ± 8^g | 1495 ± 8 | IMRE | Est $\Delta_f H(AH) = -280 \pm 4$ | 78CUM/KEB | |
| $C_8H_9O_2S^-$ pSO ₂ Me-C ₆ H ₄ CH ₂ ⁻ * -302 ± 14^a | | 1501 ± 11^g | 1473 ± 8 | IMRE | $\Delta_f H(AH) = -273 \pm 3$ | 77PED/RYL 86TAF | |
| $C_8H_9O_3^-$ PhOH · · MeCO ₂ ⁻ | | 109 ± 4 | 79 ± 7 | TDAs | | 86MEO/SIE2 | |
| $C_8H_{10}Cl^-$ PhEt · · Cl ⁻ | | | 21 | TDEq | | 82FRE/IKU | |
| $C_8H_{10}Cl^-$ m-xylene · · Cl ⁻ | | | 16 | TDEq | | 82FRE/IKU | |
| $C_8H_{10}Cl^-$ p-xylene · · Cl ⁻ | | | 16 | TDEq | | 82FRE/IKU | |
| $C_8H_{10}N^-$ PhNEt ⁻ * 50 ± 17^a | 1.60 ± 0.20^d | 1523 ± 11^g | 1493 ± 8 | IMRE | $\Delta_f H(AH) = -56 \pm 6$ $BDE(A-H) = 366 \pm 8$ | 77PED/RYL 86TAF | |
| $C_8H_{10}NO^-$ mNMe ₂ -phenoxide ⁻ * -148 ± 18^a | 2.15 ± 0.19^d | 1466 ± 10^g | 1437 ± 8 | IMRE | Est $\Delta_f H(AH) = -84 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI | |
| $C_8H_{10}NO^-$ pNMe ₂ -phenoxide ⁻ * -144 ± 18^a | | 1470 ± 10^g | 1441 ± 8 | IMRE | Est $\Delta_f H(AH) = -84 \pm 8$ | 81FUJ/MCI | |
| $C_8H_{10}N_2O_2^-$ mNMe ₂ -nitrobenzene ⁻ * -21 ± 6^b | 0.92 ± 0.05 | | | IMRE | $\Delta_f H(A) = 67 \pm 2$ | 84FUR/MUR 85FUK/MCI | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------------|---|---|---------------------------|---|------------------------|
| $C_8H_{11}^-$ 2-methylenenorbornan-3-ide ⁻ | | | | | Est2 | $\Delta_f H(AH) = 39 \pm 13$ | |
| • | 138 ± 18^a | | 1629 ± 5^g | 1600 ± 3 | IMRE | | 86LEE/SQU |
| $C_8H_{11}^-$ bicyclo[3.2.1]oct-2-en-4-ide ⁻ | | | | | Est2 | $\Delta_f H(AH) = 13 \pm 17$ | |
| • | 117 ± 22^a | | 1635 ± 5^g | 1604 ± 3 | IMRE | | 86LEE/SQU |
| $C_8H_{11}^-$ cyclooctadienide ⁻ | | | | | Est2 | $\Delta_f H(AH) = 42 \pm 13$ | |
| • | 88 ± 29^a | | 1576 ± 16^g | 1548 ± 13 | IMRB | between EtOH, nPrOH | 86LEE/SQU |
| $C_8H_{11}O^-$ 4,4-diMe-cyclohexenone-6-enolate ⁻ | | | | | Est2 | $\Delta_f H(AH) = -180 \pm 13$ | |
| • | -181 ± 22^a | | 1529 ± 10^g | 1497 ± 8 1500 ± 8 | IMRE IMRE ⁰ | | 86BAR/KIP 86BAR/KIP |
| $C_8H_{11}O_2^-$ 5,5-diMe-1,3-cyclohexandion-2-ide ⁻ | | | | | Est | $\Delta_f H(AH) = -287 \pm 13$ | |
| • | -399 ± 22^a | | 1418 ± 10^g | 1385 ± 8 | IMRE | Acid: dimedone | 78CUM/KEB |
| $C_8H_{11}O_2^-$ EtOH · PhO ⁻ | | | | | | | |
| | | | 81 ± 4 | 47 ± 7 | TDA's | | 86MEO/SIE2 |
| $C_8H_{12}B^-$ Me ₂ C(CH=CH) ₂ BCH ₂ ⁻ | | | | | Est | $\Delta_f H(AH) = 86 \pm 13$ | |
| • | 100 ± 32^a | | 1544 ± 19^g | 1515 ± 17 | IMRB | | 77SUL |
| $C_8H_{13}^-$ cyclooctenide ⁻ | | | | | | $\Delta_f H(AH) = -27 \pm 1$ | 77PED/RYL |
| • | 60 ± 26^a | | 1617 ± 25^g | 1586 ± 21 | IMRB | Between EtOH, nPrOH | 86LEE/SQU |
| $C_8H_{13}O^-$ cyclooctanone enolate ⁻ | | | | | | $\Delta_f H(AH) = -272 \pm 5$ | 77PED/RYL |
| • | 1.63 ± 0.06 | | | | PD | | 78ZIM/JAC |
| $C_8H_{13}O_2^-$ cC ₆ H ₁₁ -CH ₂ CO ₂ ⁻ | | | | | Est | $\Delta_f H(AH) = -523 \pm 8$ BDE(A-H) = 444 ± 8 | |
| • | -609 ± 19^a | 3.23 ± 0.20^d | 1444 ± 11^g | 1415 ± 8 | IMRE | | 86TAF |
| $C_8H_{15}O_4^-$ iPrCO ₂ H · iPrCO ₂ ⁻ | | | | | | | |
| | | | 125 ± 4 | 83 ± 7 | TDA's | | 86MEO/SIE2 |
| $C_8H_{17}O^-$ nC ₈ H ₁₇ O ⁻ | | | | | | $\Delta_f H(AH) = -355 \pm 1$ BDE(A-H) = 436 ± 4 | 77PED/RYL |
| • | -330 ± 12^a | 2.0 ± 0.2^d | 1556 ± 11 1553 ± 10 | 1528 ± 13^h | CIDC CIDC ⁰ | | 83BOA/HOU 83BOA/HOU |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|----------------------|---|---|---|---------|------------------------|
| $C_8H_{17}O^-$ tBuCH(iPr)O ⁻ * -379±16 ^a | 2.15±0.17 ^d | 1543±12 ^g | 1515±8 1514±8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -392 \pm 4$ $BDE(A-H) = 438 \pm 4$ | | 79BAR/SCO 79BAR/SCO |
| $C_8H_{17}O_5^-$ MeO(CH ₂ CH ₂ O) ₂ Me ⁻ ·MeCO ₂ ⁻ | | | 63±4 40±7 | TDA _s | | | 86MEO/SIE2 |
| $C_8H_{19}O_2^-$ nPrOH ⁻ ·tBuCH ₂ O ⁻ -627±25 ^c | | | 83±10 ^g | 55±8 IMRE | | | 84CAL/ROZ |
| $C_8H_{19}O_2^-$ tBuOH ⁻ ·tBuO ⁻ -673±26 ^c | | | 85±10 ^g | 58±8 IMRE | | | 84CAL/ROZ |
| $C_9HN_5^-$ 2,3,5,6-tetracyanopyridine ⁻ 2.17±0.07 | | | | Est SI | $\Delta_f H(A) = 669 \pm 17$ | | 67FAR/PAG |
| $C_9H_3F_6N^-$ 3,5-diCF ₃ -benzonitrile ⁻ * -1235±14 ^b 1.14±0.10 | | | | Est TDEq | $\Delta_f H(A) = -1125 \pm 4$ | | 87KEB/CHO |
| $C_9H_3F_6O_2^-$ 3,5-diCF ₃ -benzoate ⁻ * -1810±19 ^a 4.13±0.24 ^d | | | 1357±11 ^g | 1328±8 IMRE | $\Delta_f H(AH) = -1637 \pm 8$ $BDE(A-H) = 444 \pm 13$ | | 86TAF |
| $C_9H_4N^-$ pCN-C ₆ H ₄ C≡C ⁻ * 383±18 ^a 4.08±0.32 ^d | | | 1471±10 ^g | 1438±8 IMRE | $\Delta_f H(AH) = 443 \pm 8$ $BDE(A-H) = 552 \pm 21$ | | 86TAF |
| $C_9H_5CrO_3^-$ (CO) ₃ CrC ₆ H ₅ ⁻ -326±30 ^a | | | 1554±21 | IMRB | $\Delta_f H(AH) = -350 \pm 9$ | | 77PED/RYL 85LAN/SQU |
| $C_9H_5F_3N^-$ mCF ₃ -C ₆ H ₄ CHCN ⁻ * -585±19 ^a | | | 1431±15 ^g | 1403±8 IMRE | $\Delta_f H(AH) = -485 \pm 4$ | | 86TAF |
| $C_9H_5F_3N^-$ pCF ₃ -C ₆ H ₄ CHCN ⁻ * -595±19 ^a | | | 1420±15 ^g | 1393±8 IMRE | $\Delta_f H(AH) = -485 \pm 4$ | | 86TAF |
| $C_9H_5F_6^-$ 3,5-diCF ₃ -C ₆ H ₃ CH ₂ ⁻ * -1340±19 ^a 1.76±0.24 ^d | | | 1510±11 ^g | 1482±8 IMRE | $\Delta_f H(AH) = -1320 \pm 8$ $BDE(A-H) = 368 \pm 13$ | | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|------------------------|---|---|---------------------|---|---------------------------------|
| $C_9H_5N_2^-$ Ph-C(CN) $_2^-$ • | | | 1348±11 ^g | 1317±8 | IMRE | | 86TAF |
| $C_9H_5N_2^-$ mCN-C ₆ H ₄ CHCN ⁻ • | 211±18 ^a | | 1419±13 ^g | 1390±8 | IMRE | Est $\Delta_f H(AH) = 322 \pm 4$ | 81FUJ/MCI |
| $C_9H_5N_2^-$ pCN-C ₆ H ₄ CHCN ⁻ • | 192±18 ^a | | 1400±13 ^g | 1372±8 | IMRE | Est $\Delta_f H(AH) = 322 \pm 4$ | 81FUJ/MCI |
| $C_9H_6BrO_2^-$ 4-Br-cubyl-CO $_2^-$ • | | 3.61±0.21 ^d | 1407±12 ^g | 1378±8 | IMRE | BDE(A-H) = 444±8 | 86TAF |
| $C_9H_6F_3O^-$ pCOF ₃ -C ₆ H ₄ CH $_2^-$ • | -776±15 ^a | | 1470±11 ^g | 1439±8 | IMRE | Est $\Delta_f H(AH) = -715 \pm 4$ | 86TAF |
| $C_9H_6N^-$ quinolinide ⁻ • | 289±9 ^a | | 1608±8 | 1572±8 | TDEq | $\Delta_f H(AH) = 211 \pm 1$ | 79VIS 87MEO |
| $C_9H_6NO^-$ pCOCN-C ₆ H ₄ CH $_2^-$ • | -69±23 ^a | | 1446±11 ^g | 1418±8 | IMRE | Est2 $\Delta_f H(AH) = 15 \pm 13$ | 86TAF |
| $C_9H_7^-$ indenide ⁻ • | 106±12 ^a | 1.98±0.24 ^d | 1473±11 ^g | 1442±8 | IMRE | $\Delta_f H(AH) = 163 \pm 1$ BDE(A-H) = 351±13 | 80KUD/KUD 82MCM/GOL 86TAF |
| $C_9H_7F_3O^-$ mCF ₃ -acetophenone ⁻ -869±9 ^b | 0.663±0.009 | | | | Est ECD | $\Delta_f H(A) = -805 \pm 8$ | 75WEN/KAO |
| $C_9H_7F_3O^-$ oCF ₃ -acetophenone ⁻ -867±9 ^b | 0.642±0.009 | | | | Est ECD | $\Delta_f H(A) = -805 \pm 8$ | 75WEN/KAO |
| $C_9H_7F_3O^-$ pCF ₃ -acetophenone ⁻ -867±9 ^b | 0.642±0.009 | | | | Est ECD | $\Delta_f H(A) = -805 \pm 8$ | 75WEN/KAO |
| $C_9H_7NO^-$ pCOMe-benzonitrile ⁻ • | -61±14 ^b | 1.13±0.10 1.12±0.09 | | | Est TDEq TDEq | $\Delta_f H(A) = 49 \pm 4$ | 87KEB/CHO 86CHO/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|--|--|--|--|---------|--|
| $C_9H_7O_2^-$ cubyl-CO ₂ ⁻ * 136±33 ^a | 3.40±0.21 ^d | 1428±12 ^g | 1398±8 | Est2 IMRE | $\Delta_f H(AH) = 238 \pm 21$ $BDE(A-H) = 444 \pm 8$ | | 86TAF |
| $C_9H_7O_3^-$ pCOMe-benzoate ⁻ * -595±15 ^a | 3.69±0.24 ^d | 1399±11 ^g | 1369±8 | Est IMRE | $\Delta_f H(AH) = -464 \pm 4$ $BDE(A-H) = 444 \pm 13$ | | 86TAF |
| $C_9H_8^-$ indene ⁻ 146±3 ^b | 0.17±0.03 | | | ECD | $\Delta_f H(A) = 163 \pm 1$ | | 80KUD/KUD 81WOJ/FOL |
| $C_9H_8N^-$ pMe-C ₆ H ₄ CHCN ⁻ * 94±23 ^a | | 1471±15 ^g | 1443±8 | Est IMRE | $\Delta_f H(AH) = 153 \pm 8$ | | 86TAF |
| $C_9H_8NO^-$ pOMe-C ₆ H ₄ CHCN ⁻ * -29±19 ^a | | 1471±15 ^g | 1443±8 | Est IMRE | $\Delta_f H(AH) = 30 \pm 4$ | | 86TAF |
| $C_9H_8O^-$ PhCH=CHCHO ⁻ -59±13 ^b | 0.82±0.04 | | | Est ECD | $\Delta_f H(A) = 21 \pm 8$ | | 67WEN/CHE |
| $C_9H_9^-$ 1-phenylcyclopropanide ⁻ 260±17 ^a | | 1639±16 ^g | 1607±13 | IMRB | $\Delta_f H(AH) = 151 \pm 1$ | | 82FUCI/HAL 84AND/DEP |
| $C_9H_9^-$ CH ₂ =C(Ph)CH ₂ ⁻ 196±31 ^a | | 1613±27 ^g | 1586±23 | IMRB | $\Delta_f H(AH) = 113 \pm 4$ | | 69BEN/CRU 84BAR/BUR |
| $C_9H_9ClO_2^-$ Cl-triMe-benzoquinone ⁻ * -423±21 ^b | 1.86±0.05 | | | Est2 IMRE | $\Delta_f H(A) = -243 \pm 17$ | | 85FUK/MCI |
| $C_9H_9N^-$ 3,5-diMe-benzonitrile ⁻ * 39±21 ^b | 1.14±0.09 | | | Est2 TDEq | $\Delta_f H(A) = 149 \pm 13$ | | 86CHO/KEB |
| $C_9H_9O^-$ MeCH=C(Ph)O ⁻ * -131±23 ^a | | 1508±21 ^g 1509±21 ^g 1483±8 | 1481±8 1482±8 | IMRE IMRE IMRE ^o | $\Delta_f H(AH) = -109 \pm 2$ $BDE(A-H) = 389 \pm 8$ | | 77PED/RYL 79BAR/SCO 78CUM/KEB 79BAR/SCO |
| $C_9H_9O^-$ MeOH·PhC≡C ⁻ 70±26 ^c | | 56±10 ^g | 32±8 | IMRE | | | 84CAL/ROZ |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|-----------------------------------|---|--|
| $C_9H_9O^-$ PhCH=C(Me)O ⁻ • -163±16 ^a | | | 1465±15 ^g 1469±15 ^g | 1441±8 1445±8 1451±8 | IMRE IMRE IMRE ^o | $\Delta_f H(AH) = -98 \pm 2$ | 77PED/RYL 79BAR/SCO 78CUM/KEB 79BAR/SCO |
| $C_9H_9O^-$ pCOMe-C ₆ H ₄ CH ₂ ⁻ • -136±15 ^a | | | 1513±11 ^g | 1485±8 | IMRE | Est $\Delta_f H(AH) = -119 \pm 4$ | 86TAF |
| $C_9H_9O_2^-$ mOMe-C ₆ H ₄ C(=CH ₂)O ⁻ • -264±15 ^a | | | 1509±11 ^g | 1481±8 1490±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = -244 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_9H_9O_2^-$ pCO ₂ Me-C ₆ H ₄ CH ₂ ⁻ • -336±15 ^a | | | 1515±11 ^g | 1487±8 | IMRE | Est $\Delta_f H(AH) = -320 \pm 4$ | 86TAF |
| $C_9H_9O_2S^-$ PhSO ₂ -cyclopropanide ⁻ • -179±14 ^a | | | 1512±10 ^g | 1485±8 | IMRE | Est $\Delta_f H(AH) = -161 \pm 4$ | 78CUM/KEB |
| $C_9H_9O_3^-$ mCO ₂ Et-phenoxide ⁻ • -593±14 ^a 2.44±0.19 ^d | | | 1439±10 ^g | 1410±8 | IMRE | Est $\Delta_f H(AH) = -502 \pm 4$ BDE(A-H) = 362±8 | 81FUJ/MCI |
| $C_9H_{10}ClO^-$ PhCH ₂ COMe·Cl ⁻ | | | | 45 | TDEq | | 82FRE/IKU |
| $C_9H_{10}ClO_2^-$ 4-Cl-bicyclo[2.2.2]octene-CO ₂ ⁻ • -535±25 ^a 3.63±0.21 ^d | | | 1405±12 ^g | 1376±8 | IMRE | Est2 $\Delta_f H(AH) = -410 \pm 13$ BDE(A-H) = 444±8 | 86TAF |
| $C_9H_{10}O^-$ propiophenone ⁻ -143±3 ^b 0.351±0.004 | | | | | ECD | $\Delta_f H(A) = -109 \pm 2$ | 77PED/RYL 75WEN/KAO |
| $C_9H_{10}O_2^-$ benzyl acetate ⁻ -328±18 ^b 0.1±0.1 | | | | | ECD | Est $\Delta_f H(A) = -313 \pm 8$ | 83ZLA/LEE |
| $C_9H_{10}O_2^-$ triMe-benzoquinone ⁻ • -374±13 ^b 1.60±0.05 | | | | | IMRE | Est2 $\Delta_f H(A) = -220 \pm 8$ | 85FUK/MCI |
| $C_9H_{10}O_4^-$ 2,3-diMeO-5-Me-benzoquinone ⁻ • 1.86±0.10 | | | | | TDEq | | 87KEB/CHO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|---|---|---|------------------------------|---|--|
| $C_9H_{11}^-$ PhCMe ₂ ⁻ | | | | | | $\Delta_f H(AH) = 4 \pm 1$ $BDE(A-H) = 350 \pm 7$ | 77PED/RYL 81ROB/STE 79BAR/SCO 79BAR/SCO |
| • | 59 ± 11^a | 0.79 ± 0.18^d | 1586 ± 10^g | 1560 ± 8 1554 ± 8 | IMRE IMRE ^o | | |
| $C_9H_{11}NO_2^-$ 2,4,6-triMe-nitrobenzene ⁻ | | | | | | $\Delta_f H(A) = -17 \pm 13$ | 87KEB/CHO 85GRJ/CAL 85FUK/MCI |
| • | -84 ± 22^b | 0.70 ± 0.10 0.67 ± 0.11 0.72 ± 0.05 | | | Est2 TDEq IMRE IMRE | | |
| $C_9H_{11}O^-$ miPr-phenoxide ⁻ | | | | | | $\Delta_f H(AH) = -195 \pm 13$ $BDE(A-H) = 362 \pm 8$ | 77PED/RYL 81FUJ/MCI |
| • | -264 ± 22^a | 2.21 ± 0.19^d | 1461 ± 10^g | 1432 ± 8 | IMRE | | |
| $C_9H_{11}O^-$ oiPr-phenoxide ⁻ | | | | | | $\Delta_f H(AH) = -182 \pm 13$ | 77PED/RYL 81FUJ/MCI |
| • | -258 ± 25^a | | 1454 ± 12^g | 1423 ± 8 | IMRE | | |
| $C_9H_{11}O^-$ piPr-phenoxide ⁻ | | | | | | $\Delta_f H(AH) = -209 \pm 13$ $BDE(A-H) = 362 \pm 13$ | 77PED/RYL 81FUJ/MCI |
| • | -278 ± 22^a | 2.20 ± 0.23^d | 1461 ± 10^g | 1433 ± 8 | IMRE | | |
| $C_9H_{11}O_2^-$ bicyclo[2.2.2]octane-CO ₂ ⁻ | | | | | | $\Delta_f H(AH) = -363 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| • | -460 ± 25^a | 3.35 ± 0.21^d | 1433 ± 12^g | 1403 ± 8 | IMRE | | |
| $C_9H_{11}O_2S^-$ PhSO ₂ CHEt ⁻ | | | | | | $\Delta_f H(AH) = -319 \pm 2$ | 78CUM/KEB |
| • | -326 ± 10^a | | 1523 ± 8^g | 1491 ± 8 | IMRE | | |
| $C_9H_{12}BrO_2^-$ 4-Br-bicyclo[2.2.2]octane-CO ₂ ⁻ | | | | | | $\Delta_f H(AH) = -480 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| • | -598 ± 25^a | 3.56 ± 0.21^d | 1412 ± 12^g | 1382 ± 8 | IMRE | | |
| $C_9H_{12}Cl^-$ 1,3,5-triMe-benzene · Cl ⁻ | | | | | | | 82FRE/IKU |
| | | | | 19 | TDEq | | |
| $C_9H_{12}Cl^-$ PhiPr · Cl ⁻ | | | | | | | 82FRE/IKU |
| | | | | 23 | TDEq | | |
| $C_9H_{12}Cl^-$ PhnPr · Cl ⁻ | | | | | | | 82FRE/IKU |
| | | | | 21 | TDEq | | |
| $C_9H_{12}ClO_2^-$ 3-Cl-bicyclo[2.2.2]octane-CO ₂ ⁻ | | | | | | $\Delta_f H(AH) = -523 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| • | -629 ± 25^a | 3.44 ± 0.21^d | 1423 ± 12^g | 1394 ± 8 | IMRE | | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|------------------------|--|--|----------------------------------|---|-------------------------------------|
| $C_9H_{12}ClO_2^-$ 4-Cl-bicyclo[2.2.2]octane-CO ₂ ⁻ • -645±25 ^a | | 3.53±0.21 ^d | 1415±12 ^g | 1385±8 | Est2 IMRE | $\Delta_f H(AH) = -530 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_9H_{12}FO_2^-$ 4-F-bicyclo[2.2.2]octane-CO ₂ ⁻ • -790±25 ^a | | 3.51±0.21 ^d | 1417±12 ^g | 1387±8 | Est2 IMRE | $\Delta_f H(AH) = -677 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_9H_{12}N^-$ mNMe ₂ -C ₆ H ₄ CH ₂ ⁻ 134±32 ^a | | | 1597±23 ^g | 1569±21 | Est IMRB | $\Delta_f H(AH) = 67 \pm 8$ | 83CAL/BAR |
| $C_9H_{12}N^-$ pNMe ₂ -C ₆ H ₄ CH ₂ ⁻ 155±32 ^a | | | 1614±23 ^g | 1586±21 | Est IMRB | $\Delta_f H(AH) = 71 \pm 8$ | 83CAL/BAR |
| $C_9H_{12}NO_4^-$ 4-NO ₂ -bicyclo[2.2.2]octane-CO ₂ ⁻ • -664±25 ^a | | 3.65±0.21 ^d | 1403±12 ^g | 1374±8 | Est2 IMRE | $\Delta_f H(AH) = -537 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_9H_{13}O_2^-$ bicyclo[2.2.2]octane-CO ₂ ⁻ • -82±23 ^a | | 3.27±0.20 ^d | 1440±11 ^g | 1411±8 | Est2 IMRE | $\Delta_f H(AH) = 8 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_9H_{15}O^-$ cyclononane enolate ⁻ • 1.69±0.06 | | | | | Est PD | $\Delta_f H(AH) = -279 \pm 8$ | 78ZIM/JAC |
| $C_9H_{19}O^-$ (tBu) ₂ CHO ⁻ • -412±16 ^a | | 2.25±0.17 ^d | 1533±12 ^g | 1505±8 1509±8 | Est IMRE IMRE ^o | $\Delta_f H(AH) = -415 \pm 4$ $BDE(A-H) = 438 \pm 4$ | 79BAR/SCO 79BAR/SCO |
| $C_9H_{19}O^-$ nC ₉ H ₁₉ O ⁻ • -353±13 ^a | | 2.0±0.2 ^d | 1553±11 1551±10 | 1525±13 ^h | CIDC CIDC ^o | $\Delta_f H(AH) = -376 \pm 2$ $BDE(A-H) = 436 \pm 4$ | 77PED/RYL 83BOA/HOU 83BOA/HOU |
| $C_9H_{21}BF^-$ iPr ₃ B · F ⁻ -773 ^c | | | 272 | | IMRB | F ⁻ A: iPr ₂ BF > iPr ₃ B > Et ₂ BF > Et ₃ B | 77MUR/BEA2 |
| $C_9H_{21}O_2^-$ tBuOH · tBuCH ₂ O ⁻ -687±27 ^c | | | 85±10 ^g | 57±8 | IMRE | | 84CAL/ROZ |
| $C_{10}N_6^-$ hexacyanobutadiene ⁻ 3.3±0.1 3.3±0.1 | | | | | Est SI SI | $\Delta_f H(A) = 586 \pm 42$ | 69PAG/GOO 67FAR/PAG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|-----------------------------------|-------------------------------|--|
| $C_{10}HN_4^-$ 2,3,5,6-tetracyanophenide ⁻ | | 2.41±0.04 | | | SI | | 67FAR/PAG |
| $C_{10}H_2F_{12}O_6U^-$ UO ₂ ·(hexafluoroAcAc) ₂ ⁻ | | 1.9±0.3 | | | NBIP | | 82YOK/QUI |
| $C_{10}H_2N_4^-$ 1,2,4,5-tetracyanobenzene ⁻ | | 2.2±0.2 | | | Est2 SI | $\Delta_f H(A) = 627 \pm 13$ | 67FAR/PAG |
| $C_{10}H_4Cl_2O_2^-$ 2,3-diCl-1,4-naphthoquinone ⁻ • -374±20 ^b | 2.19±0.10 2.08±0.11 | | | | Est2 TDEq IMRE | $\Delta_f H(A) = -162 \pm 10$ | 87KEB/CHO 85GRI/CAL |
| $C_{10}H_5O_2^-$ 1,4-naphthoquinonide ⁻ | | | 1641±3 ^B | <1607 | IMRB | $\Delta_f H(AH) = -111 \pm 4$ | 77PED/RYL 87JOH/SPE |
| $C_{10}H_6Cl_4O_4^-$ dimethyl tetrachloroterephthalate ⁻ 711 ^b | 0.8 | | | | Est2 ECD | $\Delta_f H(A) = 785 \pm 13$ | 68KUH/LEV |
| $C_{10}H_6N_2O_4^-$ 1,3-diNO ₂ -naphthalene ⁻ • -52±14 ^b | 1.78±0.10 | | | | Est TDEq | $\Delta_f H(A) = 120 \pm 4$ | 87KEB/CHO |
| $C_{10}H_6N_2O_4^-$ 1,5-diNO ₂ -naphthalene ⁻ • -51±14 ^b | 1.77±0.10 | | | | Est TDEq | $\Delta_f H(A) = 120 \pm 4$ | 87KEB/CHO |
| $C_{10}H_6O_2^-$ 1,4-naphthoquinone ⁻ • -286±14 ^b | 1.81±0.10 1.71±0.11 1.71±0.05 >0.8 >0.6 | | | | TDEq IMRE IMRE ECD ES | $\Delta_f H(A) = -111 \pm 4$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85FUK/MCI 83CHE/WEN 70COL/CHR |
| $C_{10}H_7^-$ naphthalenide ⁻ • 272±6 ^a | | | 1651±5 | 1611±5 | TDEq | $\Delta_f H(AH) = 150 \pm 1$ | 82COL/JIM 87MEO |
| $C_{10}H_7Cl^-$ 1-Cl-naphthalene ⁻ 93±10 ^b | 0.277±0.003 | | | | ECD | $\Delta_f H(A) = 120 \pm 10$ | 77PED/RYL 69STE/WEN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|-------------|---|---|--|-----------------------------------|--|
| $C_{10}H_7NO_2^-$ 1-NO ₂ -naphthalene ⁻ • 31±12 ^b | | 1.23±0.10 | | | TDEq | $\Delta_f H(AH) = 150 \pm 2$ | 77PED/RYL 87KEB/CHO |
| $C_{10}H_7NO_2^-$ 2-NO ₂ -naphthalene ⁻ • 36±18 ^b | | 1.18±0.10 | | | Est2 TDEq | $\Delta_f H(AH) = 150 \pm 8$ | 87KEB/CHO |
| $C_{10}H_7N_2^-$ pMe-C ₆ H ₄ -C(CN) ₂ ⁻ • | | | 1354±12 ^g | 1323±8 | IMRE | | 86TAF |
| $C_{10}H_7O^-$ 2-naphthoxide ⁻ • -122±11 ^a | | | 1438±10 ^g | 1408±8 | IMRE | $\Delta_f H(AH) = -30 \pm 1$ | 77PED/RYL 86TAF |
| $C_{10}H_8^-$ azulene ⁻ • 223±13 ^b | 0.69±0.10 0.75±0.11 0.68±0.04 0.52±0.01 >0.5 0.656±0.008 | | | | TDEq IMRE Kinc ECD ES ECD | $\Delta_f H(AH) = 289 \pm 3$ | 77PED/RYL 87KEB/CHO 85GRI/CAL 85GRI/CHO2 81WOJ/FOL 70CHA/CHR 66BEC/CHE |
| $C_{10}H_8^-$ naphthalene ⁻ 137±6 ^b | 0.14±0.05 0.13±0.04 0.148±0.006 | | | | ECD ECD ECD | $\Delta_f H(AH) = 150 \pm 1$ | 82COL/JIM 83ZLA/LBE 81WOJ/FOL 66BEC/CHE |
| $C_{10}H_9BrCl_2O_2^-$ 2,6-diCl-Br-tBu-benzoquinone ⁻ • | | 2.42±0.05 | | | IMRE | | 85FUK/MCI |
| $C_{10}H_9N_2^-$ 1,5-diaminonaphthalenide ⁻ • 127±22 ^a | | | 1493±10 ^g | 1463±8 1472±8 | IMRE IMRE ^o | Est $\Delta_f H(AH) = 164 \pm 13$ | 82ARN/VEN 82ARN/VEN |
| $C_{10}H_9N_2^-$ 1,8-diaminonaphthalenide ⁻ • | | | | 1441±8 1450±8 | IMRE IMRE ^o | Est2 $\Delta_f H(AH) = 192 \pm 8$ | 82ARN/VEN 82ARN/VEN |
| $C_{10}H_9O_2^-$ PhCOCH=C(Me)O ⁻ • -358±12 ^a | | | 1422±10 ^g | 1393±8 | IMRE | Est $\Delta_f H(AH) = -250 \pm 2$ | 78CUM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|------------------------|--|--|--------|--|-----------|
| $C_{10}H_{10}Cl_2O_2^-$ 2,3-diCl-tBu-benzoquinone $^-$ * | | 2.25±0.05 | | | IMRE | | 85FUK/MCI |
| $C_{10}H_{10}F_3O_2^-$ 4-CF ₃ -bicyclo[2.2.2]octene-CO ₂ $^-$ * | -1153±25 ^a | 3.63±0.21 ^d | 1405±12 ^g | 1376±8 | IMRE | Est2 $\Delta_f H(AH) = -1028 \pm 13$ BDE(A-H) = 444±8 | 86TAF |
| $C_{10}H_{10}NO_2^-$ 4-CN-bicyclo[2.2.2]octene-CO ₂ $^-$ * | -367±25 ^a | 3.75±0.21 ^d | 1394±12 ^g | 1365±8 | IMRE | Est2 $\Delta_f H(AH) = -231 \pm 13$ BDE(A-H) = 444±8 | 86TAF |
| $C_{10}H_{10}O_4^-$ dimethyl isophthalate $^-$ -734 ^b | | 0.6 | | | ECD | Est2 $\Delta_f H(A) = -681 \pm 8$ | 68KUH/LEV |
| $C_{10}H_{10}O_4^-$ dimethyl phthalate $^-$ -707 ^b | | 0.6 | | | ECD | Est $\Delta_f H(A) = -654 \pm 17$ | 68KUH/LEV |
| $C_{10}H_{10}O_4^-$ dimethyl terephthalate $^-$ -743 ^b | | 0.6 | | | ECD | Est2 $\Delta_f H(A) = -681 \pm 8$ | 68KUH/LEV |
| $C_{10}H_{11}ClO_2^-$ 2-Cl-5-tBu-benzoquinone $^-$ * | | 2.06±0.05 | | | IMRE | | 85FUK/MCI |
| $C_{10}H_{11}N_2^-$ pNMe ₂ -C ₆ H ₄ CHCN $^-$ * | 143±23 ^a | | 1478±15 ^g | 1450±8 | IMRE | Est $\Delta_f H(AH) = 195 \pm 8$ | 86TAF |
| $C_{10}H_{11}O_2^-$ triMe-benzoquinone-CH ₂ $^-$ 0.80±0.09 | | | | | SI | Est $\Delta_f H(AH) = -252 \pm 8$ | 67FAR/PAG |
| $C_{10}H_{12}F_3O_2^-$ 4-CF ₃ -bicyclo[2.2.2]octane-CO ₂ $^-$ * | -1264±25 ^a | 3.55±0.21 ^d | 1413±12 ^g | 1384±8 | IMRE | Est2 $\Delta_f H(AH) = -1148 \pm 13$ BDE(A-H) = 444±8 | 86TAF |
| $C_{10}H_{12}NO^-$ mCONMe ₂ -C ₆ H ₄ CH ₂ $^-$ * | -95±19 ^a | | 1564±11 ^g | 1536±8 | IMRE | Est2 $\Delta_f H(AH) = -130 \pm 8$ | 86TAF |
| $C_{10}H_{12}NO^-$ pCONMe ₂ -C ₆ H ₄ CH ₂ $^-$ * | -131±19 ^a | | 1529±11 ^g | 1501±8 | IMRE | Est2 $\Delta_f H(AH) = -130 \pm 8$ | 86TAF |
| $C_{10}H_{12}NO_2^-$ 2-CN-bicyclo[2.2.2]octane-CO ₂ $^-$ * | -478±25 ^a | 3.63±0.21 ^d | 1405±12 ^g | 1376±8 | IMRE | Est2 $\Delta_f H(AH) = -354 \pm 13$ BDE(A-H) = 444±8 | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------------|--|--|--------|--|-----------|
| $C_{10}H_{12}NO_2^-$ 3-CN-bicyclo[2.2.2]octane-CO ₂ ⁻ | | | | | Est2 | $\Delta_f H(AH) = -354 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| • | -470 ± 25^a | 3.55 ± 0.21^d | 1413 ± 12^g | 1384 ± 8 | IMRE | | |
| $C_{10}H_{12}NO_2^-$ 4-CN-bicyclo[2.2.2]octane-CO ₂ ⁻ | | | | | Est2 | $\Delta_f H(AH) = -350 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| • | -476 ± 25^a | 3.64 ± 0.21^d | 1405 ± 12^g | 1375 ± 8 | IMRE | | |
| $C_{10}H_{12}O^-$ 2,4,6-triMe-benzaldehyde ⁻ | | | | | Est2 | $\Delta_f H(A) = -138 \pm 13$ | 69WEN/RIS |
| • | -180 ± 17^b | 0.44 ± 0.04 | | | ECD | | |
| $C_{10}H_{12}O_2^-$ 2-iPr-5-Me-benzoquinone ⁻ | | | | | | | 87KEB/CHO |
| • | | 1.79 ± 0.10 | | | TDEq | | |
| $C_{10}H_{12}O_2^-$ tetraMe-benzoquinone ⁻ | | | | | Est | $\Delta_f H(A) = -252 \pm 8$ | 87KEB/CHO |
| • | -405 ± 18^b | 1.59 ± 0.10 | | | TDEq | | 85FUK/MCI |
| | | 1.52 ± 0.05 | | | IMRE | | |
| $C_{10}H_{13}O^-$ mtBu-phenoxide ⁻ | | | | | Est | $\Delta_f H(AH) = -202 \pm 8$ $BDE(A-H) = 362 \pm 8$ | 81FUJ/MCI |
| • | -274 ± 18^a | 2.23 ± 0.19^d | 1459 ± 10^g | 1430 ± 8 | IMRE | | |
| $C_{10}H_{13}O^-$ otBu-phenoxide ⁻ | | | | | Est2 | $\Delta_f H(AH) = -186 \pm 13$ | 81FUJ/MCI |
| • | -270 ± 25^a | | 1447 ± 12^g | 1415 ± 8 | IMRE | | |
| $C_{10}H_{13}O^-$ ptBu-phenoxide ⁻ | | | | | Est | $\Delta_f H(AH) = -202 \pm 8$ $BDE(A-H) = 362 \pm 13$ | 81FUJ/MCI |
| • | -274 ± 18^a | 2.24 ± 0.23^d | 1458 ± 10^g | 1429 ± 8 | IMRE | | |
| $C_{10}H_{13}O_2^-$ 4-Me-bicyclo[2.2.2]octene-CO ₂ ⁻ | | | | | Est2 | $\Delta_f H(AH) = -395 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| • | -497 ± 25^a | 3.39 ± 0.21^d | 1428 ± 12^g | 1399 ± 8 | IMRE | | |
| $C_{10}H_{13}O_2S^-$ PhSO ₂ CHiPr ⁻ | | | | | Est | $\Delta_f H(AH) = -349 \pm 2$ | 78CUM/KEB |
| • | -362 ± 12^a | | 1517 ± 10^g | 1487 ± 8 | IMRE | | |
| $C_{10}H_{14}^-$ 1,2,3,5-tetramethylbenzene ⁻ | | | | | | $\Delta_f H(A) = -42 \pm 8$ | 75GOO |
| • | -52 ± 10^b | 0.11 ± 0.01 | | | ECD | | 81WOJ/FOL |
| $C_{10}H_{14}^-$ 1,2,4,5-tetramethylbenzene ⁻ | | | | | | $\Delta_f H(A) = -46 \pm 8$ | 75GOO |
| • | -51 ± 10^b | 4.80 ± 0.02 | | | ECD | | 81WOJ/FOL |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|---|--|--|---------------------------|--|-------------------------------------|
| $C_{10}H_{15}^-$ pentaMe-cyclopentadienide ⁻ | -54 ± 29^a | | 1510 ± 12^g | 1485 ± 8 | Est2 IMRE | $\Delta_f H(AH) = -33 \pm 17$ | 86TAF |
| $C_{10}H_{15}O_2^-$ 4-Me-bicyclo[2.2.2]octane-CO ₂ ⁻ | -609 ± 25^a | 3.31 ± 0.21^d | 1436 ± 12^g | 1407 ± 8 | IMRE | $\Delta_f H(AH) = -515 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_{10}H_{15}O_3^-$ 3-OMe-bicyclo[2.2.2]octane-CO ₂ ⁻ | -720 ± 25^a | 3.30 ± 0.21^d | 1438 ± 12^g | 1408 ± 8 | IMRE | $\Delta_f H(AH) = -628 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_{10}H_{15}O_3^-$ 4-OMe-bicyclo[2.2.2]octane-CO ₂ ⁻ | -732 ± 25^a | 3.38 ± 0.21^d | 1430 ± 12^g | 1400 ± 8 | IMRE | $\Delta_f H(AH) = -631 \pm 13$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_{10}H_{17}O^-$ cyclodecanone enolate ⁻ | 1.83 ± 0.07 | | | | Est PD | $\Delta_f H(AH) = -304 \pm 4$ | 78ZIM/JAC |
| $C_{10}H_{23}O_2^-$ tBuCH ₂ OH ··· tBuCH ₂ O ⁻ | -698 ± 26^c | | 90 ± 10^g | 62 ± 8 | IMRE | | 84CAL/ROZ |
| $C_{11}H_7N^-$ 1-naphthonitrile ⁻ | 221 ± 22^b | 0.68 ± 0.10 0.68 ± 0.09 | | | Est2 TDEq TDEq | $\Delta_f H(A) = 286 \pm 13$ | 87KEB/CHO 86CHO/KEB |
| $C_{11}H_7N^-$ 2-naphthonitrile ⁻ | 223 ± 22^b | 0.65 ± 0.10 0.65 ± 0.09 | | | Est2 TDEq TDEq | $\Delta_f H(A) = 286 \pm 13$ | 87KEB/CHO 86CHO/KEB |
| $C_{11}H_8O^-$ 1-naphthaldehyde ⁻ | -37 ± 18^b | 0.70 ± 0.10 0.68 ± 0.02 0.74 ± 0.07 | | | Est TDEq ECD ECD | $\Delta_f H(A) = 31 \pm 8$ | 87KEB/CHO 75WEN/KAO 67WEN/CHE |
| $C_{11}H_8O^-$ 2-naphthaldehyde ⁻ | -32 ± 18^b | 0.65 ± 0.10 0.62 ± 0.02 0.62 ± 0.04 | | | Est TDEq ECD ECD | $\Delta_f H(A) = 31 \pm 8$ | 87KEB/CHO 75WEN/KAO 67WEN/CHE |
| $C_{11}H_8O_2^-$ 2-Me-1,4-naphthoquinone ⁻ | -295 ± 18^b | 1.74 ± 0.10 1.66 ± 0.05 | | | Est TDEq IMRE | $\Delta_f H(A) = -127 \pm 8$ | 87KEB/CHO 85FUK/MCI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|------------------------|--|--|----------------------|--|------------------------|
| $C_{11}H_9NO_2^-$ 2-Me-1-NO ₂ -naphthalene ⁻ • 18±14 ^b | | 1.03±0.10 | | | Est TDEq | $\Delta_f H(A) = 117 \pm 4$ | 87KEB/CHO |
| $C_{11}H_9NO_3^-$ 4-MeO-1-NO ₂ -naphthalene ⁻ • -127±14 ^b | | 1.10±0.10 | | | Est TDEq | $\Delta_f H(A) = -21 \pm 4$ | 87KEB/CHO |
| $C_{11}H_9O_4^-$ 4-CO ₂ Me-cubyl-CO ₂ ⁻ • | | 3.55±0.21 ^d | 1413±12 ^g | 1384±8 | IMRE | BDE(A-H) = 444±8 | 86TAF |
| $C_{11}H_{10}^-$ 1-Me-naphthalene ⁻ 97±13 ^b | | 0.2±0.1 | | | ECD | $\Delta_f H(A) = 113 \pm 2$ | 74SAB/CHA 81WOJ/FOL |
| $C_{11}H_{10}^-$ 2-Me-naphthalene ⁻ 97±9 ^b | | 0.14±0.07 | | | ECD | $\Delta_f H(A) = 111 \pm 2$ | 74SAB/CHA 81WOJ/FOL |
| $C_{11}H_{13}O^-$ nPrOH · PhC≡C ⁻ 8±26 ^c | | | 64±10 ^g | 37±8 | IMRE | | 84CAL/ROZ |
| $C_{11}H_{14}ClO_2^-$ 3-Cl-1-adamantyl-CO ₂ ⁻ • -692±23 ^a | | | 1416±11 ^g | 1387±8 | IMRE | Est2 BDE(A-H) = 444±8 $\Delta_f H(AH) = -577 \pm 13$ | 86TAF |
| $C_{11}H_{14}O^-$ 2,4,6-triMe-acetophenone ⁻ -252±8 ^b | | 0.49±0.04 | | | ECD | $\Delta_f H(A) = -205 \pm 4$ | 77PED/RYL 69WEN/RIS |
| $C_{11}H_{15}NO_2^-$ p-t-amyl-nitrobenzene ⁻ -306±29 ^b | | 2.2±0.2 | | | Est2 CIDC | $\Delta_f H(A) = -97 \pm 13$ | 84BUR/FUK |
| $C_{11}H_{15}O_2^-$ 1-adamantyl-CO ₂ ⁻ • -624±23 ^a | | | 1438±11 ^g | 1408±8 | IMRE | Est2 BDE(A-H) = 444±8 $\Delta_f H(AH) = -532 \pm 13$ | 86TAF |
| $C_{11}H_{16}^-$ pentamethylbenzene ⁻ -93±10 ^b | | 0.18±0.01 | | | Est2 ECD | $\Delta_f H(A) = -75 \pm 8$ | 81WOJ/FOL |
| $C_{11}H_{25}O_2^-$ tBuCH ₂ OH · tBuCH(Me)O ⁻ -736±28 ^c | | | 89±10 ^g | 62±8 | IMRE | | 84CAL/ROZ |
| $C_{12}F_{10}^-$ C ₆ F ₅ -C ₆ F ₅ ⁻ • -873±22 ^b | | 0.91±0.10 0.91±0.10 | | | Est2 TDEq IMRE | $\Delta_f H(A) = -785 \pm 13$ | 87KEB/CHO 86CHO/GRI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|--|--|--|--------------------------|------------------------------|---|
| $C_{12}N_6^-$ hexacyanobenzene $^-$ | | 2.5±0.1 | | | Est SI | $\Delta_f H(A) = 900 \pm 21$ | 67FAR/PAG |
| $C_{12}H_4N_4^-$ tetracyanoquinodimethane $^-$ | | 2.8±0.1 2.8±0.3 2.8±0.1 2.9±0.2 | | | NBIP NBIP SI SI | $\Delta_f H(A) = 770 \pm 10$ | 77PED/RYL 74KLO/COM 77COM/COO 79NAZ/POK 67FAR/PAG |
| $C_{12}H_8^-$ acenaphthylene $^-$ | | 221±3 ^b 0.40±0.03 | | | ECD | $\Delta_f H(A) = 260$ | 81KUD/KUD 81WOJ/FOL |
| $C_{12}H_8N^-$ carbazolide $^-$ | | 143±15 ^a | 1444±11 ^g | 1412±8 | IMRE | $\Delta_f H(AH) = 229 \pm 4$ | 81KUD/KUD2 86TAF |
| $C_{12}H_8O_2^-$ 2-Ph-benzoquinone $^-$ | | 2.04±0.10 | | | TDEq | | 87KEB/CHO |
| $C_{12}H_9NO_2^-$ mPh-nitrobenzene $^-$ | | 58±18 ^b 1.13±0.10 | | | Est2 TDEq | $\Delta_f H(A) = 167 \pm 8$ | 87KEB/CHO |
| $C_{12}H_9NO_2^-$ oPh-nitrobenzene $^-$ | | 85±18 ^b 1.07±0.10 | | | Est2 TDEq | $\Delta_f H(A) = 188 \pm 8$ | 87KEB/CHO |
| $C_{12}H_9NO_2^-$ pPh-nitrobenzene $^-$ | | 52±18 ^b 1.20±0.10 | | | Est2 TDEq | $\Delta_f H(A) = 167 \pm 8$ | 87KEB/CHO |
| $C_{12}H_{10}^-$ biphenyl $^-$ | | 170±5 ^b 0.13±0.04 | | | ECD | $\Delta_f H(A) = 182 \pm 1$ | 77PED/RYL 81WOJ/FOL |
| $C_{12}H_{10}N^-$ Ph ₂ N $^-$ | | 157±14 ^a | 1468±11 ^g | 1438±8 | IMRE | $\Delta_f H(AH) = 219 \pm 3$ | 78STE 86TAF |
| $C_{12}H_{10}O^-$ 1-acetonaphthone $^-$ | | -79±11 ^b 0.60±0.03 | | | Est ECD | $\Delta_f H(A) = -21 \pm 8$ | 75WEN/KAO |
| $C_{12}H_{12}^-$ 1,4-diMe-naphthalene $^-$ | | 59±16 ^b 0.25±0.08 | | | ECD | $\Delta_f H(A) = 83 \pm 8$ | 69STU/WES 81WOJ/FOL |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|------------------------|---|---|---------------------|--|--|
| $C_{12}H_{12}^-$ 1-Et-naphthalene ⁻ | 82±14 ^b | 0.15±0.06 | | | Est ECD | $\Delta_f H(A) = 96 \pm 8$ | 81WOJ/FOL |
| $C_{12}H_{12}^-$ 2,3-diMe-naphthalene ⁻ | 67±21 ^b | 0.2±0.1 | | | ECD | $\Delta_f H(A) = 84 \pm 8$ | 69STU/WES 81WOJ/FOL |
| $C_{12}H_{12}^-$ 2,6-diMe-naphthalene ⁻ | 68±11 ^b | 0.16±0.07 | | | Est ECD | $\Delta_f H(A) = 84 \pm 4$ | 81WOJ/FOL |
| $C_{12}H_{12}^-$ 2-Et-naphthalene ⁻ | 67±14 ^b | 0.20±0.06 | | | Est ECD | $\Delta_f H(A) = 86 \pm 8$ | 81WOJ/FOL |
| $C_{12}H_{14}O_4^-$ diethyl phthalate ⁻ | -740 ^b | 0.5 | | | ECD | $\Delta_f H(A) = -688 \pm 12$ | 77PEDI/RYL 68KUH/LEV |
| $C_{12}H_{15}O_4^-$ 4-CO ₂ Et-bicyclo[2.2.2]octene-CO ₂ ⁻ | -865±28 ^a | 3.50±0.20 ^d | 1418±11 ^g | 1389±8 | Est2 IMRE | $\Delta_f H(AH) = -753 \pm 17$ $BDE(A-H) = 444 \pm 8$ | 86TAF |
| $C_{12}H_{18}^-$ hexamethylbenzene ⁻ | -98±4 ^b | 0.12±0.02 | | | ECD | $\Delta_f H(A) = -87 \pm 3$ | 77PEDI/RYL 81WOJ/FOL |
| $C_{12}H_{21}O^-$ cyclododecanone enolate ⁻ | * | 1.90±0.07 | | | Est PD | $\Delta_f H(AH) = -350 \pm 8$ | 78ZIM/JAC |
| $C_{12}H_{27}O_2^-$ tBuCH(Me)OH · tBuCH(Me)O ⁻ | -768±31 ^c | | 90±10 ^g | 62±8 | IMRE | | 84CAL/ROZ |
| $C_{13}F_{10}O^-$ (C ₆ F ₅) ₂ CO ⁻ | -2023±31 ^b | 1.61±0.10 1.61±0.10 | | | Est TDEq IMRE | $\Delta_f H(A) = -1868 \pm 21$ | 87KEB/CHO 85GRJ/CAL |
| $C_{13}H_8F_2O^-$ p,p'-diF-benzophenone ⁻ | -403±13 ^b | 0.79±0.05 | | | Est IMRE | $\Delta_f H(A) = -327 \pm 8$ | 85FUK/MCI |
| $C_{13}H_9^-$ fluorenyl ⁻ | 129±12 ^a | 1.86±0.24 ^d | 1472±11 ^g 1478±11 ^g | 1439±8 1446±8 | IMRE IMRE | $\Delta_f H(AH) = 187 \pm 1$ $BDE(A-H) = 339 \pm 13$ | 81KUD/KUD 70TRO/BAZ 86TAF 78CUM/KEB |

Table 2. Negative Ion Table -- Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|--|--|--|---|---|--|
| $C_{13}H_9^-$ perinaphthalenide ⁻ | | | | | Est2 | $\Delta_f H(AH) = 226 \pm 13$ | |
| * | | 1.1 ± 0.1 | | | PD | | 79GYG/PET |
| $C_{13}H_9ClO^-$ pCl-benzophenone ⁻ | | | | | Est | $\Delta_f H(A) = 29 \pm 8$ | |
| * | -53 ± 13 ^b | 0.85 ± 0.05 | | | IMRE | | 85FUK/MCI |
| $C_{13}H_9FO^-$ pF-benzophenone ⁻ | | | | | Est | $\Delta_f H(A) = -134 \pm 8$ | |
| * | -196 ± 18 ^b | 0.64 ± 0.10 | | | TDEq | | 87KEB/CHO |
| | | 0.74 ± 0.05 | | | IMRE | | 85FUK/MCI |
| $C_{13}H_9O_2^-$ mCOPh-phenoxide ⁻ | | | | | Est | $\Delta_f H(AH) = -131 \pm 8$ $BDE(A-H) = 362 \pm 8$ | |
| * | -232 ± 18 ^a | 2.54 ± 0.19 ^d | 1428 ± 10 ^g | 1400 ± 8 | IMRE | | 81FUJ/MCI |
| $C_{13}H_9O_2^-$ pCOPh-phenoxide ⁻ | | | | | Est | $\Delta_f H(AH) = -131 \pm 8$ | |
| * | -268 ± 19 ^a | | 1393 ± 11 ^g | 1364 ± 8 | IMRE | | 81FUJ/MCI |
| $C_{13}H_{10}^-$ fluorene ⁻ | | | | | | $\Delta_f H(A) = 188 \pm 1$ | 81KUD/KUD 81WOJ/FOL |
| | 162 ± 3 ^b | 0.28 ± 0.03 | | | ECD | | |
| $C_{13}H_{10}Cl^-$ mCl-C ₆ H ₄ -CH(Ph) ⁻ | | | | | Est | $\Delta_f H(AH) = 110 \pm 4$ | |
| * | 85 ± 16 ^a | | 1505 ± 12 ^g | 1482 ± 8 | IMRE | | 86TAF |
| $C_{13}H_{10}F^-$ mF-C ₆ H ₄ CH(Ph) ⁻ | | | | | Est | $\Delta_f H(AH) = -53 \pm 8$ | |
| * | -76 ± 21 ^a | | 1507 ± 12 ^g | 1479 ± 8 | IMRE | | 86TAF |
| $C_{13}H_{10}O^-$ benzophenone ⁻ | | | | | | $\Delta_f H(A) = 50 \pm 3$ | 78SAB/LAF3 87KEB/CHO 85GRI/CAL 85FUK/MCI 83CHE/WEN |
| * | -10 ± 13 ^b | 0.62 ± 0.10 0.61 ± 0.11 0.69 ± 0.05 0.64 ± 0.05 | | | TDEq IMRE IMRE ECD | | |
| $C_{13}H_{11}^-$ Ph ₂ CH ⁻ | | | | | | $\Delta_f H(AH) = 140 \pm 3$ $BDE(A-H) = 351 \pm 4$ | 77PED/RYL 82MCM/GOL 79BAR/SCO 78CUM/KEB 68GAI/PAG 79BAR/SCO |
| * | 131 ± 13 ^a | 1.47 ± 0.14 ^d | 1521 ± 10 ^g 1512 ± 10 ^g | 1499 ± 8 1489 ± 8 | IMRE IMRE SI IMRE ^o | | |
| | | 0.8 ± 0.3 | | 1502 ± 8 | | | |
| $C_{13}H_{11}FO^-$ mF-C ₆ H ₄ CH ₂ OPh ⁻ | | | | | Est | $\Delta_f H(A) = -166 \pm 4$ | |
| | | 0.28 ± 0.09 | | | ECD | | 84HER/WEN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|----------------------|--|--|--------------|--------------------------------|-------------------------------------|
| $C_{13}H_{11}F_{12}PUO_{10}^-$ $UO_2 \cdot (\text{hexafluoroAcAc})_2 \cdot OP(OMe)_3^-$ | | 1.5±0.3 | | | NBIP | | 82YOK/QUI |
| $C_{13}H_{11}O_2S^-$ $pSO_2Ph-C_6H_4CH_2^-$ • -187±23 ^a | | | 1504±11 ^g | 1473±8 | Est2 IMRE | $\Delta_f H(AH) = -160 \pm 13$ | 86TAP |
| $C_{13}H_{12}^-$ $Ph_2CH_2^-$ 125±7 ^b | | 0.16±0.04 | | | BCD | | 81WOJ/FOL |
| $C_{13}H_{12}Cl^-$ $Ph_2CH_2 \cdot \cdot Cl^-$ | | | | 31 | TDEq | | 82FRE/IKU |
| $C_{13}H_{15}O^-$ $tBuOH \cdot \cdot PhC=C^-$ -57±28 ^c | | | 72±10 ^g | 44±8 | IMRE | | 84CAL/ROZ |
| $C_{13}H_{21}O_2^-$ $tBuCH(Me)OH \cdot \cdot PhCH_2O^-$ -523±27 ^c | | | 90±10 ^g | 63±8 | IMRE | | 84CAL/ROZ |
| $C_{14}H_7ClO_2^-$ 1-Cl-9,10-anthraquinone ⁻ • -290±22 ^b | | 1.71±0.10 | | | Est2 TDEq | $\Delta_f H(A) = -125 \pm 13$ | 87KEB/CHO |
| $C_{14}H_7O_2^-$ 9,10-anthraquinonide ⁻ | | | | | | $\Delta_f H(AH) = -95 \pm 7$ | 77PED/RYL |
| | | | 1607±17 | | IMRB | | 87JOH/SPE |
| $C_{14}H_8O_2^-$ 9,10-anthraquinone ⁻ -249±16 ^b | | 1.59±0.10 1.1±0.1 | | | TDEq SI | $\Delta_f H(A) = -95 \pm 7$ | 77PED/RYL 87KEB/CHO 69PAG/GOO |
| $C_{14}H_9Cl^-$ 1-Cl-anthracene ⁻ • 126±14 ^b | | 0.78±0.10 | | | Est TDEq | $\Delta_f H(A) = 201 \pm 4$ | 87KEB/CHO |
| $C_{14}H_9Cl^-$ 2-Cl-anthracene ⁻ • 128±14 ^b | | 0.75±0.10 | | | Est TDEq | $\Delta_f H(A) = 201 \pm 4$ | 87KEB/CHO |
| $C_{14}H_9Cl^-$ 9-Cl-anthracene ⁻ • 118±14 ^b | | 0.86±0.10 | | | Est TDEq | $\Delta_f H(A) = 201 \pm 4$ | 87KEB/CHO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|--|--|--|---------------------------|-------------------------------------|--|
| $C_{14}H_9NO_2^-$ 9-NO ₂ -anthracene ⁻ * 75±14 ^b | | 1.43±0.10 | | | Est TDEq | $\Delta_f H(A) = 213 \pm 4$ | 87KEB/CHO |
| $C_{14}H_{10}^-$ PhC≡CPh ⁻ 371±11 ^b | | 0.32±0.07 | | | ECD | $\Delta_f H(A) = 402 \pm 4$ | 82CH/LIE 81WOJ/FOL |
| $C_{14}H_{10}^-$ anthracene ⁻ * 172±11 ^b | | 0.60±0.10 0.48±0.04 0.57±0.02 0.556±0.008 | | | TDEq ECD ECD ECD | $\Delta_f H(A) = 230 \pm 1$ | 79KUD/KUD4 87KEB/CHO 81WOJ/FOL 68LYO/MOR 66BEC/CHE |
| $C_{14}H_{10}^-$ phenanthrene ⁻ 181±4 ^b | | 0.27±0.04 0.307±0.007 | | | ECD ECD | $\Delta_f H(A) = 207 \pm 1$ | 79KUD/KUD4 81WOJ/FOL 66BEC/CHE |
| $C_{14}H_{10}F_3^-$ mCF ₃ -C ₆ H ₄ -CH(Ph) ⁻ * -574±14 ^a | | | 1484±10 ^g | 1462±8 | IMRE | Est $\Delta_f H(AH) = -528 \pm 4$ | 86TAF |
| $C_{14}H_{10}F_{12}O_7U^-$ UO ₂ ·(hexafluoroAcAc) ₂ ·THF ⁻ 1.6±0.2 | | | | | NBIP | | 82YOK/QUI |
| $C_{14}H_{10}N^-$ mCN-C ₆ H ₄ -CH(Ph) ⁻ * 216±14 ^a | | | 1470±10 ^g | 1448±8 | IMRE | Est $\Delta_f H(AH) = 276 \pm 4$ | 86TAF |
| $C_{14}H_{11}^-$ 2-Me-fluorenyl ⁻ * 100±21 ^a | | | 1475±12 ^g | 1443±8 | IMRE | Est2 $\Delta_f H(AH) = 155 \pm 8$ | 86TAF |
| $C_{14}H_{11}^-$ 9-Me-fluorenyl ⁻ * 47±23 ^a | | | 1468±11 ^g | 1437±8 | IMRE | Est2 $\Delta_f H(AH) = 109 \pm 13$ | 86TAF |
| $C_{14}H_{11}O^-$ pCOPh-C ₆ H ₄ CH ₂ ⁻ * 3±15 ^a | | | 1507±11 ^g | 1479±8 | IMRE | Est $\Delta_f H(AH) = 26 \pm 4$ | 86TAF |
| $C_{14}H_{11}O_2S^-$ 9-SO ₂ Me-fluorenyl ⁻ * -287±23 ^a | | | 1384±11 ^g | 1351±8 | IMRE | Est2 $\Delta_f H(AH) = -141 \pm 13$ | 86TAF |
| $C_{14}H_{12}^-$ (E)-PhCH=CHPh ⁻ 197±9 ^b | | 0.39±0.06 | | | ECD | $\Delta_f H(A) = 235 \pm 3$ | 77PED/RYL 81WOJ/FOL |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|--------------------------|--|--|---------------------|-----------------------------------|--------------------------------------|
| $C_{14}H_{12}^-$ Ph ₂ C=CH ₂ ⁻ | 208±10 ^b | 0.39±0.06 | | | ECD | | 81WOJ/FOL |
| $C_{14}H_{15}O_2^-$ PhCH ₂ OH · PhCH ₂ O ⁻ | -276±18 ^c | | 93±4 | 66±7 | TDA _s | | 84CAL/ROZ |
| $C_{14}H_{20}O_2^-$ 2,6-di-tBu-benzoquinone ⁻ | -516±22 ^b | 1.88±0.10 | | | Est2 TDEq | $\Delta_f H(A) = -335 \pm 13$ | 87KEB/CHO |
| $C_{15}H_9N^-$ 9-CN-anthracene ⁻ | 244±14 ^b | 1.27±0.10 | | | Est TDEq | $\Delta_f H(A) = 366 \pm 4$ | 87KEB/CHO |
| $C_{15}H_{10}O^-$ 9-anthraldehyde ⁻ | -16±18 ^b | 1.31±0.10 1.0±0.1 | | | Est2 TDEq ECD | $\Delta_f H(A) = 110 \pm 8$ | 87KEB/CHO 67WEN/CHE |
| $C_{15}H_{10}O^-$ 9-phenanthraldehyde ⁻ | 0±9 ^b | 0.724±0.009 0.7±0.1 | | | Est2 ECD ECD | $\Delta_f H(A) = 70 \pm 8$ | 75WEN/KAO 67WEN/CHE |
| $C_{15}H_{13}^-$ 9-Et-fluorenone ⁻ | 27±23 ^a | | 1469±11 ^g | 1437±8 | IMRE | Est2 $\Delta_f H(AH) = 88 \pm 13$ | 86TAF |
| $C_{15}H_{13}O^-$ PhCH ₂ OH · PhC=C ⁻ | 145±26 ^c | | 82±10 ^g | 54±8 | IMRE | | 84CAL/ROZ |
| $C_{16}H_{10}^-$ fluoranthene ⁻ | 228 ^b | 0.6 | | | ECD | $\Delta_f H(A) = 289 \pm 1$ | 81KUD/KUD 69MIC |
| $C_{16}H_{10}^-$ pyrene ⁻ | 168±4 ^b | 0.50±0.03 0.591±0.008 | | | ECD ECD | $\Delta_f H(A) = 216 \pm 1$ | 79KUD/KUD2 68LYO/MOR 66BEC/CHE |
| $C_{16}H_{12}O^-$ 9-COMe-anthracene ⁻ | -34±14 ^b | 0.97±0.10 | | | Est TDEq | $\Delta_f H(A) = 60 \pm 4$ | 87KEB/CHO |
| $C_{16}H_{12}O_2^-$ 2-Et-9,10-anthraquinone ⁻ | -299±14 ^b | 1.56±0.10 | | | Est TDEq | $\Delta_f H(A) = -149 \pm 4$ | 87KEB/CHO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|--------|-------------------------------|-------------------------|
| $C_{16}H_{15}^-$ 9-iPr-fluorenone ⁻ | | | | | Est2 | $\Delta_f H(AH) = 59 \pm 13$ | |
| * -2 ± 23^a | | | 1470 ± 11^B | 1437 ± 8 | IMRE | | 86TAF |
| $C_{17}H_{12}Cl^-$ 2-Ph-5-pCl-C ₆ H ₄ -cyclopentadienide ⁻ | | | | | Est2 | $\Delta_f H(AH) = 293 \pm 13$ | |
| * 170 ± 23^a | | | 1407 ± 11^B | 1376 ± 8 | IMRE | | 86TAF |
| $C_{17}H_{13}^-$ 1,4-diphenylcyclopentadienide ⁻ | | | | | Est | $\Delta_f H(AH) = 322 \pm 8$ | |
| * 205 ± 18^a | | | 1413 ± 10^B | 1383 ± 8 | IMRE | | 86TAF |
| $C_{17}H_{17}^-$ 9-iBu-fluorenone ⁻ | | | | | Est2 | $\Delta_f H(AH) = 42 \pm 13$ | |
| * -21 ± 23^a | | | 1468 ± 11^B | 1435 ± 8 | IMRE | | 86TAF |
| $C_{17}H_{17}^-$ 9-tBu-fluorenone ⁻ | | | | | Est2 | $\Delta_f H(AH) = 25 \pm 17$ | |
| * -36 ± 28^a | | | 1469 ± 11^B | 1438 ± 8 | IMRE | | 86TAF |
| $C_{18}H_{12}^-$ benz[a]anthracene ⁻ | | | | | ECD | $\Delta_f H(A) = 285 \pm 1$ | 79KUD/KUD2 66BEC/CHE |
| 224 ± 2^b | 0.630 ± 0.008 | | | | | | |
| $C_{18}H_{12}^-$ benzo[c]phenanthrene ⁻ | | | | | ECD | $\Delta_f H(A) = 293 \pm 1$ | 79KUD/KUD2 66BEC/CHE |
| 240 ± 2^b | 0.545 ± 0.008 | | | | | | |
| $C_{18}H_{12}^-$ chrysene ⁻ | | | | | ECD | $\Delta_f H(A) = 284 \pm 1$ | 79KUD/KUD2 66BEC/CHE |
| 246 ± 2^b | 0.397 ± 0.008 | | | | | | |
| $C_{18}H_{12}^-$ naphthacene ⁻ | | | | | ECD | $\Delta_f H(A) = 284 \pm 1$ | 79KUD/KUD2 68LYO/MOR |
| 199 ± 5^b | 0.88 ± 0.04 | | | | | | |
| $C_{18}H_{12}^-$ triphenylene ⁻ | | | | | ECD | $\Delta_f H(A) = 270 \pm 1$ | 79KUD/KUD2 66BEC/CHE |
| 242 ± 2^b | 0.285 ± 0.008 | | | | | | |
| $C_{18}H_{15}^-$ 2-Ph-5-p-tolyl-cyclopentadienide ⁻ | | | | | Est | $\Delta_f H(AH) = 288 \pm 8$ | |
| * 171 ± 19^a | | | 1413 ± 11^B | 1381 ± 8 | IMRE | | 86TAF |
| $C_{18}H_{16}O_2^-$ 2-tBu-9,10-anthraquinone ⁻ | | | | | Est | $\Delta_f H(A) = -202 \pm 4$ | |
| * -353 ± 14^b | 1.56 ± 0.10 | | | | TDEq | | 87KEB/CHO |
| $C_{18}H_{19}^-$ 9-tBuCH ₂ -fluorenone ⁻ | | | | | Est | $\Delta_f H(AH) = 4 \pm 17$ | |
| * -74 ± 28^a | | | 1452 ± 11^B | 1419 ± 8 | IMRE | | 86TAF |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|-------------|--|--|--------------|---|----------------------------|
| $C_{19}H_{11}^-$ fluoradenide ⁻ • | | | 1391±10 ^g | 1359±8 | IMRE | | 86TAF |
| $C_{19}H_{13}^-$ 9-Ph-fluorenone ⁻ • | 224±23 ^a | | 1436±11 ^g | 1404±8 | IMRE | <i>Est2</i> $\Delta_f H(AH) = 318 \pm 13$ | 86TAF |
| $C_{19}H_{15}^-$ Ph ₃ C ⁻ 242±16 ^a | | | 1501±12 ^g 1510±13 ^g | 1467±8 1476±10 | IMRE IMRE | $\Delta_f H(AH) = 271 \pm 4$ | 77PEDRYL 86TAF 84BAR |
| | 2.56 0.8 | | | | IMRE SI | Solution equilibrium + solvation cycle From hexaphenylethane | 30BEN 68GAI/PAG |
| $C_{19}H_{16}Cl^-$ Ph ₃ CH · · Cl ⁻ | | | | 17 | TDEq | | 82FRE/IKU |
| $C_{20}H_{12}^-$ benz[a]pyrene ⁻ 0.680±0.008 | | | | | ECD | $\Delta_f H(A) = 289 \pm 4$ | 77STEIGOL 66BEC/CHE |
| $C_{20}H_{12}^-$ benz[c]pyrene ⁻ 210±5 ^b | 0.534±0.008 | | | | ECD | <i>Est</i> $\Delta_f H(A) = 261 \pm 4$ | 66BEC/CHE |
| $C_{21}H_{15}^-$ 1,3-diphenylindenide ⁻ • | 244±18 ^a | | 1403±10 ^g | 1376±8 | IMRE | <i>Est</i> $\Delta_f H(AH) = 371 \pm 8$ | 86TAF |
| $C_{22}H_{14}^-$ dibenz[a,h]anthracene ⁻ 279±5 ^b | 0.595±0.008 | | | | ECD | $\Delta_f H(A) = 336 \pm 4$ | 77STEIGOL 66BEC/CHE |
| $C_{22}H_{14}^-$ dibenz[a,j]anthracene ⁻ 279±5 ^b | 0.591±0.008 | | | | ECD | <i>Est</i> $\Delta_f H(A) = 336 \pm 4$ | 66BEC/CHE |
| $C_{22}H_{14}^-$ picene ⁻ 274±9 ^b | 0.542±0.008 | | | | ECD | <i>Est</i> $\Delta_f H(A) = 326 \pm 8$ | 66BEC/CHE |
| $C_{27}H_{19}^-$ 1,2,3-triPh-indenide ⁻ • | | | 1404±11 ^g | 1373±8 | IMRE | | 86TAF |
| CaH ⁻ CaH ⁻ • | 139±47 ^b | 0.93±0.05 | | | PD | $\Delta_f H(A) = 229 \pm 42$ | 82TN270 77RAC/FEL |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|--|--|--|------------------------------|---|---|
| CeF ₄ ⁻ CeF ₃ · · F ⁻ | -2005±29 | 3.60±0.30 ⁱ | 459±29 ^k | | TDEq | | 81SID/SOR |
| CeI ⁻ CeI ⁻ · | | > 0.3±0.3 | | | EIAP | From CeI ₃ | 76CHA |
| CeI ₂ ⁻ CeI ₂ ⁻ · | < -221 ^b | > 0.3±0.2 | | | EIAP | $\Delta_f H(A) = -192$ From CeI ₃ | 76CHA 76CHA |
| CeI ₃ ⁻ CeI ₃ ⁻ · | < -368 ^b | > 0.3 | | | IMRB | $\Delta_f H(A) = -339$ CeI ₂ ⁻ + CeI ₃ → | 76CHA 76CHA |
| CeI ₄ ⁻ CeI ₄ ⁻ · | -808 ^c | | 280±33 | 245±42 | TDEq | | 76CHA |
| Cl ⁻ Cl ⁻ · | * -227±1 ^a | 3.617±0.003 | 1395±1 ^e 1396±9 ^g | 1372±1 ^h 1374±8 | LOG IMRE | $\Delta_f H(A) = 122$ | 85JANAF 85HOT/LIN 81FUJ/MCI |
| ClCrO ⁻ CrOCl ⁻ | -231±48 | 1.2±0.1 | | | EIAP | $\Delta_f H(A) = -116±48$ From CrO ₂ Cl ₂ | 69FLE/WHI 69FLE/WHI |
| ClCrO ₂ ⁻ CrO ₂ Cl ⁻ | -531±48 | 2.4±0.4 | | | EIAP | $\Delta_f H(A) = -309±48$ From CrO ₂ Cl ₂ | 69FLE/WHI 69FLE/WHI |
| ClF ⁻ ClF ⁻ · | -195±29 ^b | 1.5±0.3 > 1.5±0.2 2.9±0.2 > 1.5±0.4 | | | NBIP EIAP EIAP EIAP | $\Delta_f H(A) = -50$ From CF ₂ Cl ₂ From CFCl ₃ From ClF ₃ From SF ₅ Cl | 85JANAF 78DIS/LAC 79ILL/SCH 79DUD/GOR 72THY |
| ClFH ⁻ HF · · Cl ⁻ | * -591±10 ^c | | 91±8 ^g | 63±8 | IMRE | | 84LAR/MCM2 |
| ClFO ⁻ ClOF ⁻ · | | > 2.0±0.2 | | | Est2 EIAP | $\Delta_f H(A) = 54±21$ | 80BAL/NIK2 |
| ClFO ₂ ⁻ ClO ₂ F ⁻ · | -255 ^b | > 2.3 | | | EIAP | $\Delta_f H(A) = -33$ From ClO ₃ F | 73BAR 83ALE/FED |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---------------------------------------|---|------------------------|---|---|--------------------------------------|--|---|
| ClF_2^- ClF_2^- | | > 3.2±0.2 > 0.9±0.2 | | | EIAP EIAP | From ClF_5 From ClF_3 | 80BAL/NIK 79DUD/GOR |
| ClF_2OS^- $F_2SO \cdots Cl^-$ | | | 72±8 ^E | 43±8 | IMRE | | 85LAR/MCM |
| ClF_3^- ClF_3^- | | > 2.4±0.1 | | | EIAP | $\Delta_f H(A) = -163 \pm 2$ From ClF_5 | 82BAU/COX 80BAL/NIK |
| ClF_3OP^- $PF_3O \cdots Cl^-$ | | -1497±12 ^C | 58±8 ^E | 32±8 | IMRE | | 85LAR/MCM |
| ClF_3P^- $PF_3 \cdots Cl^-$ | | -1211±12 ^C | 65±8 ^E | 38±8 | IMRE | | 85LAR/MCM |
| ClF_4Si^- $SiF_4 \cdots Cl^-$ | | -1940±12 ^C | 98±8 ^E | 70±8 | IMRE | | 85LAR/MCM |
| $ClHI^-$ $HCl \cdots I^-$ | | -343±9 ^C | 62±8 | 37±11 | TDAs | | 85CAL/KEB |
| ClH_2O^- $HOH \cdots Cl^-$ | | -532 ^C | 62±1 62±8 60±8 ^E 62 55±8 | 38±6 38±11 35±8 37 30±8 | TDAs TDAs IMRE TDAs TDAs | | 80KEE/CAS2 86YAM/FUR 84LAR/MCM2 82BUR/HAY 71YAM/KEB |
| $ClH_2O_2^-$ $HOOH \cdots Cl^-$ | | -456±6 ^C | 92±4 | 65±4 | TDEq | Relative to $HOH \cdots Cl^-$, 80KEE/LBE | 84BOH/FAH |
| $ClH_2O_4S^-$ $HCl \cdots HSO_4^-$ | | | 66±4 | 47±4 | TDEq | Relative to $HOH \cdots HSO_4^-$, 84BOH/FAH | 84BOH/FAH |
| ClH_3N^- $NH_3 \cdots Cl^-$ | | -317±18 ^C | 44±17 ^E | 19±8 | IMRE | | 84LAR/MCM2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|-------------------------------------|--|------------------------|--|--|--------|--|----------------------|
| Cl⁻ | | | | | | | |
| ICl ⁻ | | | | | | $\Delta_f H(A) = 18$ | 85JANAF |
| | -215±10 ^b | 2.4±0.1 | | | NBIP | | 73AUE/HUB |
| | -155 | 1.8 ⁱ | | | Endo | I ⁻ + NOCl → | 77REF/FRA |
| | | 1.48±0.05 | | | NBIP | Vertical EA | 76HUB/KLE |
| ClK⁻ | | | | | | | |
| KCl ⁻ | | | | | | $\Delta_f H(A) = -215$ | 85JANAF |
| | -276 ^b | 0.6 | | | Scat | | 79DEV/WIJ |
| | | > 1.3 | | | EIAP | From (KCl) ₂ | 64EBI |
| ClLi⁻ | | | | | | | |
| LiCl ⁻ | | | | | | $\Delta_f H(A) = -196±8$ | 85JANAF |
| | -255±10 ^b | 0.610±0.020 | | | LPES | | 76CAR/PET |
| | | > 1.3 | | | EIAP | From (LiCl) ₂ | 64EBI |
| ClNa⁻ | | | | | | | |
| NaCl ⁻ | | | | | | $\Delta_f H(A) = -181±8$ | 85JANAF |
| | -255 ^b | 0.8 | | | Scat | | 79DEV/WIJ |
| | | > 1.3 | | | EIAP | From (NaCl) ₂ | 64EBI |
| ClO⁻ | | | | | | | |
| ClO ⁻ | | | | | | $\Delta_f H(AH) = -79±8$ $BDE(A-H) = 399±9$ | 82BAU/COX 82TN270 |
| * | -108±18 ^a | 2.170 | 1502±9 ^c | 1474±10 ^h | LPD | | 79LEE/SMI |
| | | 2.4±0.2 | | | EIAP | From Cl ₂ O | 80BAL/NIK2 |
| | | 1.9±0.3 | | | IMRB | | 78DOT/ALB |
| | < -54±21 | > 1.6±0.2 ⁱ | | | Endo | Cl ⁻ + O ₂ → | 77VOG/DRE |
| ClOV⁻ | | | | | | | |
| VOCl ⁻ | | | | | | | |
| | -310±48 | 1.4±0.4 | | | EIAP | From VOCl ₃ | 75FLE/SVE |
| ClO₂⁻ | | | | | | | |
| OCIO ⁻ | | | | | | $\Delta_f H(A) = 97±8$ | 82BAU/COX |
| | -29±50 ^b | 1.3±0.4 | | | ECD | | 81WEC/CHR |
| | | 1.8±0.2 | | | EIAP | From FClO ₃ | 80BAL/NIK2 |
| ClO₂S⁻ | | | | | | | |
| SO ₂ ·Cl ⁻ | | | | | | | |
| * | -617±11 ^c | | 93±8 | 66±8 | TDAs | | 85CAL/KEB |
| | | | 93±8 | 63±7 | TDEq | Relative to HOH·Cl ⁻ in 80KBE/LEE | 84BOH/FAH |
| | | | 91±1 | 62±1 | TDAs | | 80KBE/LEE |
| | | | 87±8 ^g | 62±8 | IMRE | | 85LAR/MCM |
| | < -565 | | | | IMRB | | 79ROB/FRA |
| | -326 | | | | IMRB | | 78SUL/BEA2 |
| ClO₃⁻ | | | | | | | |
| ClO ₃ ⁻ | | | | | | $\Delta_f H(A) = 126±21$ | 82TN270 |
| | -183±21 ^b | > 3.2 | | | EIAP | From ClO ₃ F | 83ALE/FED |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|---|---|--|
| ClO_3S^- $SO_3 \cdots Cl^-$ | | | < 1312±8 | | EIAP | From $ClSO_3H$ (Appearance Potential = 0 eV) | 86ADA/SMI |
| $ClO_4S_2^-$ $SO_2 \cdots SO_2Cl^-$ | | | | | IMRE IMRB | | 80KBE/CAS 79ROB/FRA |
| $ClPb^-$ $PbCl^-$ | | 1.0±0.2 | | | EIAP | From $PbCl_2$ | 67HAS/BLO |
| $ClRb^-$ $RbCl^-$ | | > 1.5 | | | EIAP | $\Delta_f H(A) = -229$ From $(RbCl)_2$ | 82TN270 64EBI |
| $ClXe^-$ $Xe \cdots Cl^-$ | | | 13 13 < 13 | | Mobl Mobl Mobl | | 84GAT 80THA/EIS 79DEV/WIJ2 |
| Cl_2^- Cl_2^- | | | | | NBIP ECD EnCT IMRB NBIP EIAP NBIP EnCT NBIP | Vertical EA: 1.02 eV From CCl_4 Vertical EA | 77DIS/LAC2 81AYA/WEN 73HUG/LIF 72DUN/FEH 72BAE 71DEC/FRA 76HUB/KLE 71CHU/BER 70LAC/HER |
| Cl_2CrO^- $CrOCl_2^-$ | | | | | EIAP | $\Delta_f H(A) = -309±48$ From CrO_2Cl_2 | 69FLE/WHI 69FLE/WHI |
| Cl_2Ge^- $GeCl_2^-$ | | | | | EIAP | $\Delta_f H(A) = -172±4$ From $GeCl_4$ | 79TPIS 77PAB/MAR |
| Cl_2H^- $HCl \cdots Cl^-$ | | | | | TDA IMRE TDA EIAP | | 85CAL/KEB 84LAR/MCM2 74YAM/KEB 80SCH/ILL |
| | | | 100±8 97±8 ^g 99±1 | 72±11 67±8 70±1 | | From $CHCl_3$ | |
| | | | < -521±48 | | | | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|------------------------------|---|------------------------|---|---|----------------------|--|--------------------------------------|
| Cl_2I^- ICl_2^- | -605 | | 377 ^k | | Latt | | 77FIN/GAT |
| Cl_2O^- Cl_2O^- | <-109 ^b | >2.2 | | | ECD | $\Delta_f H(A) = 105 \pm 2$ | 82TN270 81WEC/CHR |
| Cl_2OP^- Cl_2PO^- | | 3.8±0.3 | | | NBAP | From $POCl_3$ | 76MAT/ROT |
| Cl_2OV^- $VOCl_2^-$ | -590±48 | 3.2±0.5 | | | EIAP | From $VOCl_3$ | 75FLE/SVE |
| $Cl_2O_2S^-$ $SO_2Cl_2^-$ | <-598 ^b | >2.4 | | | IMRB | $\Delta_f H(A) = -364 \pm 2$ EA: > Cl_2^- | 82TN270 79ROB/FRA |
| Cl_2P^- PCl_2^- | <-891±19 | 0.9±0.1 | | | EIAP EIAP | From PCl_3 From $POCl_3$ | 78PAB/MAR 74HAL/KLE |
| Cl_2Si^- $SiCl_2^-$ | -228±21 | 0.8±0.1 >2.5 | | | EIAP EIAP | $\Delta_f H(A) = -166$ From $SiCl_4$ From $SiCl_4$ | 82TN270 77PAB/MAR 68JAE/HEN |
| Cl_2Sn^- $SnCl_2^-$ | -95 | 1.0 | | | EIAP | $\Delta_f H(A) = -203 \pm 4$ From $SnCl_4$ | 82TPIS 77PAB/PER |
| Cl_2V^- VCl_2^- | -1189±28 ^b | 1.2±0.2 | | | EIAP | $\Delta_f H(A) = -1073 \pm 8$ From $VOCl_3$ | 82TN270 75FLE/SVE |
| Cl_3^- Cl_3^- | -300 -300±21 >-410 ^c | >4.3±0.2 ⁱ | 70 ^k | <182 | IMRE IMRB PDis | | 79ROB/FRA2 79ROB/FRA 79LEE/SMI |
| Cl_3Ge^- $GeCl_3^-$ | | >2.6 1.8±0.1 1.8 | | | NBAP EIAP EIAP | From $GeCl_4$ From $GeCl_4$ From $GeCl_4$ | 79MAT/ROT 78PAB/MAR 77PAB/MAR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---------------------------------------|--|--|--|--|------------------------------|---|--|
| Cl_3OP^- Cl_3PO^- | -694 ± 20^b | 1.4 ± 0.2 | | | NBIP | $\Delta_f H(A) = -558 \pm 1$ | 82TN270 76MAT/ROT |
| Cl_3OV^- $VOCl_3^-$ | $< -1043 \pm 49^b$ | 3.6 ± 0.5 | | | IMRB | $\Delta_f H(A) = -696$ EA: $> Cl^-$ | 82TN270 75FLE/SVE |
| $Cl_3O_2S^-$ $SO_2Cl_2 \cdot Cl^-$ | $< -644 \pm 21$ | | | | IMRB | $Cl^-A: > Cl_2, SO_2$ | 79ROB/FRA |
| Cl_3P^- PCl_3^- | -368 ± 12^b | 0.8 ± 0.1 > 3.6 | | | NBIP IMRB | $\Delta_f H(A) = -289 \pm 2$ From PCl_5 | 85JANAF 76MAT/ROT 74HAL/KLE |
| Cl_3Si^- $SiCl_3^-$ | $< -510 \pm 13^a$ -589 ± 21 | > 2.0 3.5 ± 0.4 | $< 1501^e$ | | EIAP EIAP | $\Delta_f H(AH) = -481 \pm 8$ $BDE(A-H) = 382 \pm 4$ From $SiCl_4$ From $SiCl_4$ | 81BEL/PER 81WAL 77PAB/MAR 68JAE/HEN |
| Cl_3Sn^- $SnCl_3^-$ | | 3.4 ± 0.2 3.7 ± 0.5 2.53 ± 0.01 2.5 | | | NBAP NBAP EIAP EIAP | From $SnCl_4$ From $SnCl_4$ From $SnCl_4$ From $SnCl_4$ | 83LAC/MAN 79MAT/ROT 78PAB/MAR 77PAB/PER |
| Cl_3Ti^- $TiCl_3^-$ | -601 ± 18^b -597 ± 13 | 0.6 ± 0.2 0.6 ± 0.1 0.6 ± 0.2 | | | NBAP EIAP EIAP | $\Delta_f H(A) = -542 \pm 2$ From $TiCl_4$ From $TiCl_4$ From $TiCl_4$ | 82TN270 79MAT/ROT 78PAB/MAR 74BEN/PAB |
| Cl_3V^- VCl_3^- | -569 ± 48 | 2.2 ± 0.5 | | | EIAP | From $VOCl_3$ | 75FLE/SVE |
| Cl_4I^- ICl_4^- | -631 | | | | Latt | | 77FIN/GAT |
| Cl_4Nb^- $NbCl_4^-$ | $< -696^b$ | > 1.4 | | | EIAP | $\Delta_f H(A) = -561 \pm 2$ From $NbCl_5$ | 82TN270 75BEN/MAR |
| Cl_4Sn^- $SnCl_4^-$ | -684 ± 21^b | 2.2 ± 0.2 2.5 ± 0.2 | | | NBIP NBIP | $\Delta_f H(A) = -472 \pm 2$ | 82TN270 83LAC/MAN 79MAT/ROT |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|-----------------------------------|--|-------------------|--|--|--------------|--|-----------------------------------|
| Cl_4Ta^- $TaCl_4^-$ | | > 1.4 | | | EIAP | $\Delta_f H(A) = -561 \pm 2$ From $TaCl_5$ | 82TN270 75BEN/MAR |
| Cl_4Ti^- $TiCl_4^-$ | | 2.9 ± 0.2 | | | NBIP | $\Delta_f H(A) = -763 \pm 2$ | 85JANAF 79MAT/ROT |
| Cl_5Si^- $SiCl_4 \cdot Cl^-$ | -986 ± 12^c | | 101 ± 8^g | 74 ± 8 | IMRE | | 85LAR/MCM |
| Co^- Co^- | 361 ± 2^b 372^a | 0.662 ± 0.003 | 1437 ± 5^c | 1395 ± 13 | LPES IMRB | $\Delta_f H(A) = 425 \pm 2$ | 82TN270 86LEO/LIN 85SAL/LAN |
| CoH^- CoH^- | 412 ± 14^b | 0.671 ± 0.010 | | | LPES | $\Delta_f H(A) = 477 \pm 13$ | 81ARM/BEA 87MIL/FEI |
| CoH_2^- CoH_2^- | * | 1.450 ± 0.014 | | | LPES | | 86MIL/FEI |
| CoD_2^- CoD_2^- | * | 1.465 ± 0.013 | | | LPES | | 86MIL/FEI |
| Co_2^- Co_2^- | 576 ± 9^b | 1.110 ± 0.008 | | | LPES | $\Delta_f H(A) = 683 \pm 8$ | 82TN270 86LEO/LIN |
| Cr^- Cr^- | 332 ± 3^b | 0.666 ± 0.012 | | 1389 ± 13 | LPES IMRB | $\Delta_f H(A) = 397 \pm 2$ | 82TN270 85HOT/LIN 85SAL/LAN |
| $CrCl^-$ $CrCl^-$ | -145 ± 48 | 1.1 ± 0.2 | | | EIAP | $\Delta_f H(A) = 48 \pm 48$ From CrO_2Cl_2 | 69FLE/WHI 69FLE/WHI |
| $CrCl_2^-$ $CrCl_2^-$ | -309 ± 48 | 1.7 ± 0.2 | | | EIAP | $\Delta_f H(A) = -128 \pm 2$ From CrO_2Cl_2 | 82TN270 69FLE/WHI |
| CrF^- CrF^- | -67 ± 48 | 1.0 ± 0.4 | | | EIAP | $\Delta_f H(A) = 21$ From CrO_2F_2 | 81WOO 69FLE/WHI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|---|--|---|---|--------------|---|-------------------------------|
| CrFO ⁻ CrOF ⁻ | -367±48 | 0.7±0.2 | | | EIAP | $\Delta_f H(A) = -309 \pm 48$ From CrO ₂ F ₂ | 69FLE/WHI 69FLE/WHI |
| CrFO ₂ ⁻ CrO ₂ F ⁻ | -724±48 | 2.5±0.2 | | | EIAP | $\Delta_f H(A) = -473 \pm 48$ From CrO ₂ F ₂ | 69FLE/WHI 69FLE/WHI |
| CrF ₂ ⁻ CrF ₂ ⁻ | -540±48 | 1.5±0.4 | | | EIAP | $\Delta_f H(A) = -216$ From CrO ₂ F ₂ | 81WOO 69FLE/WHI |
| CrF ₂ O ⁻ CrOF ₂ ⁻ | -820±48 | 2.1±0.1 | | | EIAP | $\Delta_f H(A) = -618 \pm 48$ From CrO ₂ F ₂ | 69FLE/WHI 69FLE/WHI |
| CrH ⁻ CrH ⁻ | * | 0.563±0.010 | | | LPES | | 87MIL/FEI |
| CrHO ₃ ⁻ HCrO ₃ ⁻ | -1132±40 | 2.37±0.42 ⁱ | | | TDEq | | 72MIL |
| CrH ₂ ⁻ CrH ₂ ⁻ | | >2.500 | | | LPES | | 86MIL/FEI |
| CrKO ₄ ⁻ KCrO ₄ ⁻ | -1000±16 | | | | TDEq | | 85RUD/SID |
| CrO ⁻ CrO ⁻ | 183±48 | 1.3±0.7 | | | EIAP | $\Delta_f H(A) = 218 \pm 29$ From CrO ₂ F ₂ | 83PED/MAR 69FLE/WHI |
| CrO ₂ ⁻ CrO ₂ ⁻ | <-594±42 -280±48 | 2.3±0.7 | | | IMRB EIAP | $\Delta_f H(A) = -59 \pm 21$ From CrO ₂ F ₂ | 82TN270 72MIL 69FLE/WHI |
| CrO ₃ ⁻ CrO ₃ ⁻ | -674±27 -838±82 ^a | 3.70±0.30 ⁱ 4.04±0.42 ⁱ | 1437±40 | | TDEq TDEq | $\Delta_f H(A) = -386 \pm 2$ | 82TN270 85RUD/SID 72MIL |
| CrO ₄ ⁻ CrO ₄ ⁻ | -785±30 | | | | TDEq | | 85RUD/SID |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{\text{acid}}(AH)$ $\Delta H_{\text{aff}}(X \cdot \cdot Y^-)$ | $\Delta G_{\text{acid}}(AH)$ $\Delta G_{\text{aff}}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------|--|--|--------|-------------------------|----------------------|
| Cr_2O_6^- Cr_2O_6^- | | 1.6±0.3 | | | EIAP | From $(\text{CrO}_3)_3$ | 75WAN/MAR |
| Cr_3O_9^- Cr_3O_9^- | | 1.8 | | | EIAP | From $(\text{CrO}_3)_5$ | 75WAN/MAR |
| Cs^- Cs^- * | 31±5 ^a | 0.472 | 1445±3 ^c | | LPD | $\Delta_f H(A) = 76$ | 82TN270 85HOT/LIN |
| CsI_2^- $\text{CsI} \cdot \cdot \text{I}^-$ | | | 151±5 | | TDA | | 79GUS/GOR |
| Cs_2I_3^- $\text{Cs}_2\text{I}_2 \cdot \cdot \text{I}^-$ | | | 115±13 ^k | | TDA | | 79GUS/GOR |
| Cu^- Cu^- * | 220±1 ^b | 1.228±0.010 | 1459±22 ^f | | LPES | $\Delta_f H(A) = 338$ | 82TN270 85HOT/LIN |
| Cu_7^- Cu_7^- | | 1.870±0.080 | | | LPES | | 86ZHE/KAR |
| Cu_8^- Cu_8^- | | <1.440 | | | LPES | | 86ZHE/KAR |
| Cu_9^- Cu_9^- | | 2.270±0.060 | | | LPES | | 86ZHE/KAR |
| Cu_{10}^- Cu_{10}^- | | 2.010±0.060 | | | LPES | | 86ZHE/KAR |
| Cu_{11}^- Cu_{11}^- | | 2.380±0.060 | | | LPES | | 86ZHE/KAR |
| Cu_{12}^- Cu_{12}^- | | 2.140±0.070 | | | LPES | | 86ZHE/KAR |
| Cu_{13}^- Cu_{13}^- | | 2.605±0.175 | | | LPES | | 86ZHE/KAR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|-------------------------------|---|---------------------------------|---|---|------------------------------|---|---|
| Cu_{14}^- Cu_{14}^- | | 2.075±0.025 | | | LPES | | 86ZHE/KAR |
| Cu_{15}^- Cu_{15}^- | | 2.575±0.135 | | | LPES | | 86ZHE/KAR |
| Cu_{16}^- Cu_{16}^- | | 2.325±0.115 | | | LPES | | 86ZHE/KAR |
| Cu_{17}^- Cu_{17}^- | | <2.720 | | | LPES | | 86ZHE/KAR |
| Cu_{18}^- Cu_{18}^- | | 2.570±0.130 | | | LPES | | 86ZHE/KAR |
| Cu_{19}^- Cu_{19}^- | | 2.705±0.265 | | | LPES | | 86ZHE/KAR |
| F^- F^- | | | | | | $\Delta_f H(A) = 79$ | |
| • | -249±2 ^a | 3.399±0.003 | 1554±1 ^c | 1530±2 ^h 1530±8 | PLA IMRE ^o | | 85JANAF 85HOT/LIN 79BAR/SCO |
| FHI^- $HF \cdots I^-$ | | | | | | | |
| • | -524±10 ^c | | 63±8 | | Est | Extrapolated from other bihalide data | 84LAR/MCM3 |
| FH_2O^- $HOH \cdots F^-$ | | | | | | | |
| | -588 ^c | | 97±8 | 76±8 | TDA _s | | 70ARS/YAM |
| FH_2S^- $HSH \cdots F^-$ | | | | | | | |
| • | -414±11 ^c | | 145±8 ^g | 121±8 | IMRE | | 83LAR/MCM |
| FK^- KF^- | | | | | | | |
| | -427±22 ^b | 1.0±0.2 <1.50 0.2 >1.3 | | | EIAP IMRE Scat EIAP | $\Delta_f H(A) = -327 \pm 2$ From K_2BO_2F | 85JANAF 76SHE/ILJ 80SID/SKO 79DEV/WIJ 64EBI |
| FLi^- LiF^- | | | | | | | |
| | >1.4 | | | | EIAP | $\Delta_f H(A) = -340$ From $(LiF)_2$ | 82TN270 64EBI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|-------------------------|--|--|------------------------------------|---|--|
| FN ⁻ FN ⁻ | 184±2 ^b | 0.5 | | | EIAP | $\Delta_f H(A) = 232 \pm 2$ | 84BER/GRE 82SID |
| FNP ⁻ NPF ⁻ | < -285 | | | | IMRB | NH ₂ ⁻ + PF ₃ → | 78SUL/BEA |
| FNa ⁻ NaF ⁻ | -399±20 ^b | 1.1±0.2 0.4 > 1.4 | | | EIAP Scat EIAP | $\Delta_f H(A) = -291$ From Na ₂ BO ₂ F From (NaF) ₂ | 82TN270 76SHE/ILJ 79DEV/WIJ 64EBI |
| FNa ₂ ⁻ Na ₂ F ⁻ | -300±42 | | | | EIAP | From Na ₂ BO ₂ F | 76SHE/ILJ |
| FO ⁻ FO ⁻ | -89±13 ^b | 2.05±0.08 > 1.4±0.5 | | | EIAP EIAP | $\Delta_f H(AH) = -96 \pm 4$ $BDE(A-H) = 412 \pm 13$ From F ₂ O From CF ₃ OF | 82BAU/COX 82BAU/COX 84ALE/VOL 70THY/MAC |
| FOV ⁻ VOF ⁻ | -473±48 | 1.2±0.4 | | | EIAP | From VOF ₃ | 75FLE/SVE |
| FO ₂ S ⁻ SO ₂ ·F ⁻ | * -729±12 ^c < -595 -511 -715 | 2.8 | 183±8 ^g | 154±8 | IMRE EIAP IMRB SI EIAP | From SO ₂ F ₂ From SO ₂ F ₂ | 83LAR/MCM 80WAN/FRA 78SUL/BEA2 69PAG/GOO 58REE/DIB |
| FO ₃ S ⁻ SO ₃ ·F ⁻ | * -971±45 ^c < -971±13 ^a | 4.6±0.6 ^d | 326±42 ^g < 1312±8 | 297±42 | IMRB EIAP | From FSO ₃ H (Appearance Potential = 0 eV) | 85LAR/MCM 86ADA/SMI |
| FS ⁻ FS ⁻ | -180±55 ^b | 2.0±0.5 | | | Est | $\Delta_f H(A) = 13 \pm 6$ From trends in EA of SF _x | 85JANAF 82JANAF |
| FXe ⁻ Xe·F ⁻ | -276 ^c | | 27±4 | | Mobl | | 79DEV/WIJ2 |
| F ₂ ⁻ F ₂ ⁻ | -297 ^b | 3.1 2.9±0.2 | | | ECD EIAP | Vertical EA: 1.24 eV From NF ₃ | 81AYA/WEN 74HAR/FRA |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|-------------------------------|---|---|---|---|--|---|--|
| F_2^- $F_2^{\cdot-}$ | | 2.9±0.2 3.2 > 2.8±0.3 3.1±0.1 > 3.0 | | | ELAP ELAP ELAP EnCT ELAP | From BF_3 From SO_2F_2 From CF_2O From SO_2F_2 | 71DEC/FRA 80WAN/FRA 72THY 71CHU/BER 58REE/DIB |
| F_2Ge^- $GeF_2^{\cdot-}$ | | | | | ELAP | $\Delta_f H(A) = -573$ From GeF_4 | 81WOO 72HAR/CRA |
| F_2H^- $FH \cdots F^-$ | | | | | IMRE ELAP | - From $CHF = CHF$ | 83LAR/MCM 85HEN/ILL |
| F_2HNOP^- $HNOPF_2^-$ | | | | | IMRB | $NH_2^- + OPF_3 \rightarrow HNOPF_2^- + HF$; $HNOPF_2^- + OPF_3 \rightarrow (F_3PO)_2N^- + HF$ | 78SUL/BEA |
| F_2HNP^- $HNPf_2^-$ | | | | | IMRB | $NH_2^- + PF_3 \rightarrow$ | 78SUL/BEA |
| F_2K^- $KF \cdots F^-$ | | | | | TDA's TDEq | $F^- A: 1100K$; $\Delta H_f(KF_2^-): 298K$ | 81NIK/SID 80SID/NIK 79GUS/GOR |
| F_2Mn^- $MnF_2^{\cdot-}$ | | | | | TDEq | $\Delta_f H(A) = -525$ | 81WOO 82SID/GUB |
| F_2N^- $NF_2^{\cdot-}$ | | | | | IMRE IMRE EIAP EIAP EIAP SI | $\Delta_f H(AH) = -65 \pm 6$ $BDE(A-H) = 314 \pm 10$ | 69PAN/ZER 84BER/GRE 86TAF 81KOP/PIK 74HAR/FRA 79DUD/BAL 78DUD/BAL 69PAG/GOO |
| F_2NOP^- $ONPF_2^-$ | | | | | IMRB | $HNO^- + PF_3 \rightarrow NOPF_2^- + HF$ | 78SUL/BEA |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---------------------------------|---|--------------------------------|---|---|--------------------|---|---|
| F_2OP^- F_2PO^- | <-962 -448±19 | 3.4±0.2 | | | IMRB EIAP | $HNO^- + OPF_3 \rightarrow$ From F_3PO | 78SUL/BEA 71RHY/DIL |
| F_2OPS^- F_2POS^- | <-1033 | | | | IMRB | $HS^- + OPF_3 \rightarrow$ | 78SUL/BEA |
| F_2OV^- VOF_2^- | -925±48 | 2.8±0.5 | | | EIAP | From VOF_3 | 75FLE/SVE |
| $F_2O_2P^-$ $F_2PO_2^-$ | <-1167 | | | | IMRB | HO^- or $EtO^- + OPF_3 \rightarrow$ | 78SUL/BEA |
| $F_2O_2S^-$ $SO_2F_2^-$ | <3.1 | | | | IMRB | $\Delta_f H(A) = -759 \pm 8$ EA: $< F_2$ | 87HER 78GAL/FAI |
| $F_2O_2U^-$ $UO_2F_2^-$ | 3.36±0.52 | | | | TDEq | | 84GOR/PYA |
| F_2P^- PF_2^- | -636±51 ^b | 1.6±0.5 >1.6±0.5 1.5±0.5 | | | EIAP SI Est2 | $\Delta_f H(A) = -482 \pm 2$ From PF_2NCS | 84BERGRE 72THY 69PAG/GOO 82JANAF |
| F_2PS^- F_2PS^- | 2.6±1.0 | | | | EIAP | From PF_3S | 71RHY/DIL |
| F_2V^- VF_2^- | -703±48 | 0.4±0.5 | | | EIAP | From VOF_3^- | 75FLE/SVE |
| F_3Cr^- $CrF_2 \cdots F^-$ | -1124±15 | | | | TDEq | | 83IGO |
| F_3Cu^- $CuF_2 \cdots F^-$ | >5.26 ⁱ | 351±17 | | | TDEq | Anchor: $F^-A(FeF_3)$ | 84CHI/KOR 86KUZ/KOR |
| F_3Fe^- $FeF_2 \cdots F^-$ | -1069±12 ^c | 3.62±0.13 ⁱ | 359±9 374±17 | | TDEq TDEq | Corrections to 81SOR/SID, better neutral pressure determination | 86SID/BOR 86KUZ/KOR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|--|--|--|---------------------------|--|---|
| F_3Fe^- $FeF_2 \cdot \cdot F^-$ | | 4.30 ± 0.20^i | 441 ± 14 | | TDEq | $F^-A: 64 \pm 5 \text{ kJ} < AlF_3$ | 81SOR/SID |
| F_3Ge^- GeF_3^- | | 1.1 ± 0.4 1.6 3.1 ± 0.1 | | | EIAP EIAP EIAP | From GeF_4 From GeF_4 From GeF_4 | 74WAN/MAR 74FRA/WAN 72HAR/CRA |
| F_3Mn^- $MnF_2 \cdot \cdot F^-$ | | 4.36^i | 430 | | TDEq | | 81SID/SOR |
| F_3Ni^- $NiF_2 \cdot \cdot F^-$ | | | 338 ± 15 | | TDEq | Reanalyzed literature data, $150 \text{ kJ} < AlF_3$ | 86NIK/IGO |
| F_3OS^- $F_2SO \cdot \cdot F^-$ | | | 156 ± 8^g | 126 ± 8 | IMRE | | 83LAR/MCM |
| F_3OV^- VOF_3^- | | 3.1 ± 0.4 | | | IMRB | $\Delta_f H(A) = -1234 \pm 29$ EA: $> VOF_2^-$, $< F^-$ | 75FLE/SVE 75FLE/SVE |
| F_3OW^- WOF_3^- | | > 0.3 | | | EIAP | From WOF_4 | 77HIL |
| $F_3O_2S^-$ $SO_2F_2 \cdot \cdot F^-$ | | | 150 ± 8^g | 115 ± 8 | IMRE IMRB | | 83LAR/MCM 78GAL/FAI |
| F_3Pb^- PbF_3^- | | 3.7 4.3 | | | EIAP EIAP | $\Delta_f H(A) = -510 \pm 54$ From PbF_4 From PbF_4 | 75BEN/WAN 75BEN/WAN 74FRA/WAN |
| F_3S^- SF_3^- | | 3.1 ± 0.2 2.9 ± 0.1 2.7 ± 0.7 2.7 | | | NBAP EIAP Est SI | $\Delta_f H(A) = -488 \pm 25$ From SF_6 From SF_4 Reanalysis: 71HAR/THY | 87HER 78COM/REI 71HAR/THY 82JANAF 69PAG/GOO |
| F_3Se^- SeF_3^- | | | | | EIAP | From SeF_6 | 69BRI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---------------------------------|--|-----------------------------|---|---|----------------------------------|---|--|
| F_3Si^- SiF_3^- | -1284 ± 35^a -1176 -1176 | 2.9 ± 0.1 2.0 2.0 | 1446 ± 14^c | 1414 ± 15^h | PD EIAP EIAP EIAP SI | $\Delta_f H(AH) = -1201 \pm 21$ $BDE(A-H) = 419 \pm 4$ From SiF_4 From SiF_4 From SiF_4 | 85JANAF 81WAL 75RIC/STEA 74FRA/WAN 73WAN/MAR 70MAC/THY 69PAG/GOO |
| F_3Sn^- SnF_3^- | -632 -887 | > 1.2 > 1.2 2.6 | | | EIAP EIAP EIAP | From SnF_4 From SnF_4 From SnF_4 | 77PAB/PER 75BEN/WAN 74FRA/WAN |
| F_3V^- VF_3^- | -1033±48 | 1.6 ± 0.4 | | | EIAP | From VOF_3 | 75FLE/SVE |
| F_4Cr^- $CrF_3 \cdots F^-$ | -1467±15 | | | | TDEq | $FeF_3^- + CrF_4^- = FeF_4^- + CrF_3^-, \Delta_{rxn}H = 9 \text{ kJ}$ | 83IGO |
| F_4Fe^- $FeF_3 \cdots F^-$ | -1423±18 ^c -1490 ^c -1412±14 ^c | 5.45 ± 0.20^i | 451 ± 10 456 ± 14 439 ± 14 | | TDEq TDEq TDEq | $F^-A: 92.5 \text{ kJ} > FeF_2$ $F^-A: 37 \text{ kJ} < AlF_3$ $F^-A: 62 \text{ kJ} < AlF_3$ | 86SID/BOR 84CHI/KOR 81SOR/SID |
| $F_4Ge_2^-$ $Ge_2F_4^-$ | <-121 | | | | IMRB | | 72HAR/CRA |
| F_4La^- $LaF_3 \cdots F^-$ | -2004±33 | | | | TDEq | | 79GUS/GOR |
| F_4Mn^- $MnF_3 \cdots F^-$ | -1463±60 ^c -1466±60 | 5.23 ± 0.03^i | 421 ± 13 | | TDEq TDEq | $F^-A: 72 \pm 3 \text{ kJ} < AlF_3$ $F^-A: 79 \text{ kJ} < AlF_3$ | 84CHI/KOR 84KOR/CHI |
| F_4OP^- $F_3PO \cdots F^-$ | -1660 ± 13^c -1594±46 ^c | | 200 ± 8^g 134 ± 42 | 168 ± 8 | IMRE IMRB | $F^-A: SF_4 > F_3PO > SF_5$ | 85LAR/MCM 71RHY/DIL |
| F_4OU^- $UOF_3 \cdots F^-$ | | 3.80 ± 0.43 | | | TDEq | | 84GOR/PYA |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--------------------------------|--|---|--|--|----------------------|--|---|
| F_4P^- $PF_3 \cdot F^-$ | -1336 ± 13^c | | 168 ± 8^g 209 ± 21 | 136 ± 8 | IMRE IMRB | $F^-A: < OPF_3, > F, SF_4, Me_3SiF, HCN, SO_2$ | 83LAR/MCM 78SUL/BEA |
| F_4PS^- $F_3PS \cdot F^-$ | -1374 ± 106^c | | 134 ± 42 | | IMRB | $F^-A: \text{between } SF_4, SF_5$ | 71RHY/DIL |
| F_4Pt^- PtF_4^- | | 5.20 ± 0.16 | | | TDEq | $\Delta_f H(A) = -524 \pm 25$ EA: 2.5 kJ > MnF_4 | 83KOR/BON 84KOR/CHI |
| F_4Rh^- RhF_4^- | | 5.00 ± 0.20^i | 401 ± 14 | | TDEq | $F^-A: 22 \text{ kJ} < MnF_3$ | 84CHI/KOR |
| F_4S^- SF_4^- | -990 ± 31^b | 2.4 ± 0.1 0.8 ± 0.2 1.3 ± 0.1 | | | IMRB NBIP ES | $\Delta_f H(A) = -763 \pm 21$ EA: between NO_2^- , HS^- | 85JANAF 81BAB/STR2 78COM/REI 74DON/HAR |
| F_4Sc^- $ScF_3 \cdot F^-$ | -2013 ± 13^c -2009 ± 13^c | | 495 ± 10 487 ± 10 470 ± 10 | | TDEq TDEq TDEq | $F^-A: 10 \text{ kJ} < AlF_3$ $F^-A: 17 \text{ kJ} < AlF_3$ Reanalyzed data, $18 \text{ kJ} < AlF_3$ | 81SKO/NIK 81NIK/SID 86NIK/IGO |
| F_4Se^- SeF_4^- | | 1.7 ± 0.1 | | | EIAP EIAP | From SeF_6 From SeF_6 | 73HAR/THY 69BRI |
| F_4Te^- TeF_4^- | | 2.2 ± 0.1 | | | EIAP EIAP | From TeF_6 | 73HAR/THY 69BRI |
| F_4Ti^- TiF_4^- | | > 0.0 | | | EIAP | $\Delta_f H(A) = -1552 \pm 2$ | 82TN270 74BEN/PAB |
| F_4U^- UF_4^- | -1725 ± 30 | 1.24 ± 0.36^i < 1.8 | 415 ± 42^k | | TDEq IMRB | $\Delta_f H(A) = -1599 \pm 2$ Critical review | 82TN270 84PYA/GUS 80SID/SKO |
| F_4W^- WF_4^- | -1280^b | 2.6 $> 2.3 \pm 0.1$ | | | EIAP EIAP | $\Delta_f H(A) = -1029$ From WF_6 From WF_6 | 81WOO 77DEW/NEU 73THY/HAR2 |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|---|--------------------|--|--|---|---|--|
| F_4W^- WF_4^- | | > 1.0 | | | EIAP | $\Delta_f H(A) = -1029$ From WF_6 | 81WOO 77HIL |
| F_5Cr^- $CrF_4 \cdot \cdot F^-$ -1753±15 | | | | | TDEq | $FeF_3^- + CrF_5^- = FeF_4^- + CrF_4^-$, $\Delta_{rxn}H = -48$ kJ | 83IGO |
| F_5Fe^- $FeF_2 \cdot \cdot FeF_3^-$ * -1740±43 ^c -1769±17 ^c | 3.80±0.40 ⁱ | | 202±37 201±4 | | TDEq TDA | $\Delta_f H(A^-)$ at 0K | 86SID/BOR 81SOR/SID |
| F_5Ge^- $GeF_4 \cdot \cdot F^-$ -1856±26 < -2038 ^c | | | 418±29 ^k > 405 | | Latt IMRB | | 84MAL/ROS 72HAR/CRA |
| F_5Hf^- HfF_5^- -2386±17 ^c | | | 429±17 405±9 | | TDEq TDEq | Reanalyzed literature data, 59 kJ < AlF_3 F^-A : 84 kJ < AlF_3 | 86NIK/IGO 80NIK/SOR |
| F_5Mn^- MnF_5^- -1565±84 | | | 348±84 | | TDEq | F^-A : 73 kJ < MnF_3 | 84KOR/CHI |
| F_5Mo^- MoF_5^- | | > 3.5 > 3.3±0.4 | | | NBAP NBAP | $\Delta_f H(A) = -1241$ From MoF_6 From MoF_6 | 85JANAF 78COM/REI 77MAT/ROT |
| F_5Pt^- PtF_5^- | | 6.50 | | | TDEq | | 79SID/NIK |
| F_5S^- SF_5^- * -1195±31 ^c < -1263±33 -1251±25 -1269±33 ^b | 3.01±0.29 ⁱ > 3.7±0.3 ⁱ 3.7±0.2 2.7±0.2 > 2.9±0.1 > 2.8±0.1 > 2.8±0.2 3.2±0.2 3.66±0.04 | | 183±8 ^g | 151±8 | IMRE IMRB NBAP Est NBAP NBAP EnCT NBAP EIAP SI | F^-A : $SF_4 > SF_5$ From SF_6 Literature average From SF_6 From SF_6 From SF_6 From SF_6 , new EA(F^-) | 83LAR/MCM 81BAB/STR2 74LEF/TAN 82JANAF 78COM/REI 75HUB/LOS 73LIF/TIE 73COM/COO 61CUR2 64KAY/PAG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|-------------------------|--|------------------------|--|--|--------|---|------------|
| F_5Se^- | | | | | | | |
| SeF_5^- | | | | | | $\Delta_f H(A) = -476 \pm 42$ | 69BRI |
| | -1385±40 | > 5.1 ⁱ | | | NBAP | From SeF_6 | 78COM/REI |
| | | 3.3±0.1 | | | EIAP | From SeF_6 | 73HAR/THY |
| | < -1197 | | | | EIAP | From SeF_6 | 69BRI |
| F_5Si^- | | | | | | | |
| $SiF_4 \cdot \cdot F^-$ | | | | | | | |
| | -2115±19 ^c | | 251±17 ^E | 226±17 | IMRE | | 85LAR/MCM |
| | | | 285±21 | | IMRB | F ⁻ A: < BF_3 , > iPr_2BF | 77MUR/BEA3 |
| | < -2318 | | | | IMRB | | 70MAC/THY |
| F_5Te^- | | | | | | | |
| TeF_5^- | | | | | | $\Delta_f H(A) = -586 \pm 42$ | 69BRI |
| | | 4.5 | | | NBAP | From TeF_6 | 78COM/REI |
| | | 4.2±0.1 | | | EIAP | From TeF_6 | 73HAR/THY |
| | < -1397 | | | | EIAP | From TeF_6 | 69BRI |
| F_5Th^- | | | | | | | |
| $ThF_4 \cdot \cdot F^-$ | | | | | | | |
| | -2432±13 | | 436±15 ^k | | TDEq | F ⁻ A: 88 kJ < AlF_3 , 15 kJ < ZrF_4 | 83SID/ZHU |
| F_5U^- | | | | | | | |
| UF_5^- | | | | | | | |
| | -2275±19 ^c | 2.99±0.20 ⁱ | 427±15 | | TDEq | Reanalyzed literature data, 61 kJ < AlF_3 | 86NIK/IGO |
| | -2322±15 | 3.47±0.26 ⁱ | 424 ^k | | TDEq | Critical review, other literature data corrected | 84PYA/GUS |
| | -2256±4 ^c | 3.30±0.16 ⁱ | 410±1 | | TDEq | F ⁻ A: 93 kJ < AlF_3 | 80SID/SKO |
| | -2297±33 | 3.78±0.40 ⁱ | 448±36 ^k | | TDEq | | 80PYA/GUS |
| | -2297±33 | | | | TDEq | | 79GUS/GOR |
| | -2297±33 | > 1.9±0.4 ⁱ | | | NBAP | From UF_6 | 77MAT/ROT |
| | -2265±14 | 4.0±0.4 ⁱ | | | NBAP | From UF_6 | 77COM |
| F_5W^- | | | | | | | |
| WF_5^- | | | | | | $\Delta_f H(A) = -1397$ | 81WOO |
| | > -1631 | < 3.5 | | | IMRB | EA: WF_5 < WF_6 | 79GEO/BEA |
| | | > 1.8±0.3 | | | NBAP | From WF_6 | 78COM/REI |
| | < -1338±25 | > 0.4±0.2 | | | EIAP | From WF_6 | 77HIL |
| | | 1.2±0.3 | | | NBAP | From WF_6 | 77DIS/LAC |
| | | 1.3±0.2 | | | EIAP | From WF_6 | 77DEW/NEU |
| | -1409±29 | 0.8±0.2 | 270 ^k | | EIAP | From WF_6 | 73THY/HAR2 |
| F_5Zr^- | | | | | | | |
| $ZrF_4 \cdot \cdot F^-$ | | | | | | | |
| | -2338±31 ^c | | 415±8 | | TDEq | F ⁻ A: 92±3 kJ < AlF_3 | 82SKO/SOR |
| | -2343±17 ^c | | 403±4 | | TDEq | F ⁻ A: 97 kJ < AlF_3 | 81SKO/NIK |
| $F_6Fe_2^-$ | | | | | | | |
| $Fe_2F_6^-$ | | | | | | | |
| | -2071±38 | 4.45±0.24 ⁱ | | | TDEq | | 86SID/BOR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|------------------------|--|--|---|--|--|
| $F_6Ge_2^-$ $Ge_2F_6^-$ < -117 | | | | | IMRB | | 72HAR/CRA |
| F_6Ir^- IrF_6^- | | > 5.1±0.5 | | | NBIP | $\Delta_f H(A) = -544 \pm 21$ | 82TN270 78COM/REI |
| F_6Mo^- MoF_6^- | | > 5.1±0.5 > 4.5±0.4 | | | NBIP NBIP | $\Delta_f H(A) = -1558 \pm 1$ | 85JANAF 78COM/REI 77MAT/ROT |
| F_6P^- $PF_5 \cdot \cdot F^-$ • -2200±46 ^c | | | 356±42 ^g 423±33 | 308±42 | IMRE latt | | 85LAR/MCM 84MAL/ROS |
| F_6Pt^- PtF_6^- • -1448±57 ^b | 8.00±0.30 > 5.1±0.5 | | | | TDEq NBIP | $\Delta_f H(A) = -676 \pm 28$ EA: 272 kJ > PtF ₄ | 86KOF/NIK 81NIK/SID2 78COM/REI |
| F_6Re^- ReF_6^- | | > 5.1±0.5 | | | NBIP | $\Delta_f H(A) = -1349$ | 84BAR/YEH 78COM/REI |
| F_6S^- SF_6^- • -1322±10 ^b | 1.05±0.10 0.542 0.8 | | 159±15 ^k | | TDEq LPD Kine | $\Delta_f H(A) = -1221 \pm 1$ | 85JANAF 85GRI/CHO 82DRZ/BRA 83LIF |
| | | | | | Review: literature consistent with ΔS^\ddagger for detachment = -59 J/mol-K | | |
| | 0.5±0.2 1.4±0.1 > 0.7 1.2±0.3 | | | | NBIP Kine ECD Est | Literature average | 78COM/REI 83HEN/BEN 83CHE/WEN 82JANAF |
| < -1310 | | | | | CIDT NBIP NBIP EnCT NBIP NBIP IMRB IMRB ECD ES SI | | 78REF/FRA 75HUB/LOS 74LEF/TAN 73LIF/TIE 73COM/COO2 73COM/COO 71FEH 70LIF/HUG 68CHE/GEO 66COM/CHR 64KAY/PAG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--------------------------------------|--|--|---|--|--|---|--|
| F_6Sb^- $SbF_5 \cdot F^-$ | | > 6.0 | | | NBAP | From Sb_2F_{10} | 80COM/REI |
| F_6Se^- SeF_6^- | | -1397 ± 40^b | 2.9 ± 0.2 | | NBIP | $\Delta_f H(A) = -1117 \pm 21$ | 82TN270 78COM/REI |
| F_6Te^- TeF_6^- | | -1636 ± 31^b | 3.3 ± 0.1 3.3 ± 0.2 3.3 ± 0.2 | | NBIP NBIP NBIP | $\Delta_f H(A) = -1318 \pm 21$ | 82TN270 78COM/REI 73COM/COO2 73COM/COO |
| F_6U^- UF_6^- | | -2680 ± 25 -2649 ± 30 -2628 ± 24^c -2724 ± 42 | 5.58 ± 0.31^i 5.20 ± 0.34^i 4.89 ± 0.25^i 6.33 ± 0.50 5.8 ± 0.3^i > 3.6 > 5.5 > 4.3 \pm 0.4 > 5.1 4.9 ± 0.5 2.9 | 424 ± 17 | TDEq TDEq TDEq TDEq Latt IMRB IMRB NBIP NBIP IMRB SI | $\Delta_f H(A) = -2147 \pm 2$ Critical review $F^-A: 55 \pm 8 \text{ kJ} > UF_4$ $F^-A: (1100K) 14 \pm 1 \text{ kJ} > UF_4, \Delta_f H(A^-): 298K$ Endo F^- transfer to BF_3 at 1.5 eV observed | 82TN270 84PYA/GUS 83SKO/SOR 80SID/SKO 80PYA/GUS 84MAL/ROS 80STR/NEW 80ANN/STO 77MAT/ROT 77COM 76BEA 69PAG/GOO |
| F_6W^- WF_6^- | | -2046 ± 28^b -2061 ± 25 | 3.4 ± 0.2 3.5 ± 0.1 > 5.1 \pm 0.5 > 4.9 \pm 0.4 3.7 ± 0.2 2.7 | | IMRB IMRB NBIP NBIP NBIP SI | $\Delta_f H(A) = -1722 \pm 8$ EA: > F^- , < Cl^- | 82TN270 85VIG/PAU 79GEO/BEA 78COM/REI 77MAT/ROT 77DIS/LAC 69PAG/GOO |
| $F_7Fe_2^-$ $FeF_3 \cdot FeF_4^-$ | | -2379 ± 37^c -2280 ± 18 | 4.5 ± 0.2^i | 189 ± 24 204 ± 4 | TDA TDA | $\Delta_f H(A^-)$ at 0K | 86SID/BOR 81SOR/SID |
| F_7MnPt^- $MnF_3 \cdot PtF_4^-$ | | -2054 ± 105 | | | TDEq | | 84KOR/CHI |
| $F_7Mn_2^-$ $MnF_3 \cdot MnF_4^-$ | | -2517 ± 84 | | | TDA | | 84KOR/CHI |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------------|---|---|--------------|--|------------------------|
| $F_7 Sc_2^-$ $ScF_3 \cdots ScF_4^-$ | | | 228 ± 1 | 199 ± 1 | TDA's | | 81NIK/SID |
| $F_7 U^-$ $UF_6 \cdots F^-$ | -2630 ± 30 -2588 ± 46^c | | 237 ± 30^k 192 ± 42 | | TDEq IMRB | Critical review | 84PYA/GUS 76BEA |
| $F_7 W^-$ $WF_6 \cdots F^-$ | -2266 ± 24 | < 6.5 | 289 ± 21^k | | IMRB | F ⁻ A: SiF ₄ < WF ₆ < BF ₃ | 79GEO/BEA |
| $F_8 Ge_2^-$ $Ge_2F_8^-$ | < -151 | | | | IMRB | | 72HAR/CRA |
| $F_8 KSc_2^-$ $KScF_4 \cdots ScF_4^-$ | | | 144 ± 3 | 120 ± 4 | TDA's | | 81NIK/SID |
| $F_8 U_2^-$ $U_2F_8^-$ | $> -3598 \pm 1002.30^i$ | | | | TDEq | | 84PYA/GOR |
| $F_9 U_2^-$ $U_2F_9^-$ | -4130 ± 30 -4138 ± 33 | 4.30 ± 0.52 | 500 ± 50^k | | TDEq TDEq | | 84PYA/GOR 80PYA/GUS |
| $F_9 Zr_2^-$ $ZrF_4 \cdots ZrF_5^-$ | -4228 ± 15^c | | 214 ± 4 | | TDA's | | 82SKO/SOR |
| $F_{10} U_2^-$ $U_2F_{10}^-$ | -4490 ± 30 | 4.50 ± 0.40^i | 520 ± 50^k | | TDEq | | 84PYA/GOR |
| $F_{11} U_2^-$ $U_2F_{11}^-$ | -4850 ± 40 | 6.10 ± 0.70^i | 540 ± 50^k | | TDEq | | 84PYA/GOR |
| $F_{12} U_2^-$ $U_2F_{12}^-$ | $> -5200 \pm 79$ | 7.90 ± 0.80^i | | | TDEq | | 84PYA/GOR |
| Fe^- Fe^- | | | | | | $\Delta_f H(AH) = 471 \pm 29$ $BDE(A-H) = 163 \pm 29$ | 79DEN/VAN 79DEN/VAN |
| * | 402 ± 1^b | 0.151 ± 0.003 | 1461 ± 30^c | | LPES | | 86LEO/LIN |
| * | | | 1420 ± 13^g | 1389 ± 13 | IMRB | | 85SAL/LAN |
| | | 0.163 ± 0.035 | | | LPES | | 85HOT/LIN |
| | $> 361 \pm 42^a$ | $< 0.3 \pm 0.2$ | 1420 ± 13^c | | EIAP | From Fe(CO) ₅ | 76COM/STO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|---|--|------------------|--|--|------------------|--|-----------------------------------|
| FeH ⁻ FeH ⁻ • | | 0.934±0.010 | | | LPES | | 83STE/FEI |
| FeH ₂ ⁻ FeH ₂ ⁻ • | | 1.049±0.014 | | | LPES | | 86MIL/FEI |
| FeD ₂ ⁻ FeD ₂ ⁻ • | | 1.038±0.013 | | | LPES | | 86MIL/FEI |
| FeO ⁻ FeO ⁻ • | 107±56 ^a | 1.492±0.020 | 1504±40 ^c | | LPES | $\Delta_f H(AH) = 133 \pm 17$ $BDE(A-H) = 336 \pm 38$ | 80MUR 85JANAF 77ENG/LIN |
| Fe ₂ ⁻ Fe ₂ ⁻ • | | 0.902±0.008 | | | LPES | | 86LEO/LIN |
| Ga ⁻ Ga ⁻ • | 236±36 ^a | 0.3±0.1 | 1546±23 ^c | | DH PD | $\Delta_f H(AH) = 220 \pm 13$ $BDE(A-H) = 262 \pm 8$ | 81KAN/MOO 85HOT/LIN |
| Ge ⁻ Ge ⁻ • | 258±2 ^b | 1.233±0.003 | | | LPES | $\Delta_f H(A) = 377 \pm 2$ | 82TN270 86MIL/MIL |
| GeH ₃ ⁻ GeH ₃ ⁻ • | 50±17 ^a | 1.739±0.043 | 1490±15 ^c | 1455±15 ^h | LPD | $\Delta_f H(AH) = 91 \pm 2$ $BDE(A-H) = 345 \pm 10$ | 82TN270 83NOB/WAL 74REE/BRA |
| H ⁻ H ⁻ • | 145 ^a | 0.8 0.78±0.02 | 1675 ^c | 1649 ^h | Calc PD | $BDE(A-H) = 436$ Given: 0.754209(3) eV | 85JANAF 85HOT/LIN 70FEL |
| HIS ⁻ ISH ⁻ -48 | | 1.1 | | | Endo | $\Delta_f H(A) = 105$ $I^- + H_2S \rightarrow$ | 76REF 76REF |
| HI ₂ ⁻ HI··I ⁻ • | -233±9 ^c | | 71±8 | 41±11 | TDA _s | | 85CAL/KEB |
| HK ₂ O ⁻ K ₂ OH ⁻ -345±12 | | | | | TDE _q | | 84BUR/KUD |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|----------------------------|--|--|------------------|---|--|
| HMg ⁻ MgH ⁻ * | | 1.05±0.06 | | | PD | | 77RAC/FEL |
| HMn ⁻ MnH ⁻ * 172 ^b | | 0.869±0.010 | | | LPES | $\Delta_f H(A) = 256$ | 79HUB/HER 83STE/FEI |
| HMoO ₄ ⁻ HMoO ₄ ⁻ -1055±61 ^a | | | 1362±40 | | TDEq | $\Delta_f H(AH) = -887 \pm 21$ $e^- + H_2MoO_4 = HMoO_4^- + H$ | 82TN270 79MIL |
| HN ⁻ NH ⁻ * 340±21 ^a | | 0.381±0.014 0.380±0.030 | 1682±20 ^c | 1653±21 ^h | LPES LPES | $\Delta_f H(AH) = 189 \pm 1$ $BDE(A-H) = 406 \pm 18$ See also 85NEU/LYK | 85GIB/GRE 85JANAF 76ENG/LIN 74CEL/BEN |
| HNO ⁻ HNO ⁻ * 68±6 ^b | | 0.338±0.015 | | > 1498 | LPES IMRB | $\Delta_f H(A) = 100 \pm 4$ | 82BAU/COX 83ELL/ELL 77SUL |
| DNO ⁻ DNO ⁻ * | | 0.330±0.015 | | | LPES | | 83ELL/ELL |
| HNO ₃ ⁻ HNO ₃ ⁻ -190±15 ^b | | 0.6±0.1 0.6±0.2 | | | NBIP EnCT | $\Delta_f H(A) = -135$ | 82TN270 76MAT/ROT2 82PAU/DAL |
| HN ₂ O ⁻ HN=NO ⁻ < 247 < 130±21 | | | | | IMRB IMRB | RONO + NH ₂ ⁻ → CH ₂ =N ⁻ + N ₂ O → | 81KIN/MAR 85KAS/DEP |
| HN ₂ O ₄ ⁻ HONO · · NO ₂ ⁻ * -405±22 ^c | | | 136±4 | | TDA _s | | 80LEE/KEE |
| HN ₂ O ₆ ⁻ HNO ₃ · · NO ₃ ⁻ * -545±10 ^c | | | 103±8 | 87±7 | TDEq | | 77DAV/FEH |
| HNi ⁻ NiH ⁻ * | | 0.481±0.007 | | | LPES | | 87MIL/FEI |
| HO ⁻ HO ⁻ * -137 ^a | | 1.828 1.825±0.002 | 1635 ^c | 1607±1 ^h | LPD LPD | $\Delta_f H(AH) = -242$ $BDE(A-H) = 499$ Given: 1.827670(21) eV | 82TN270 85JANAF 82SCH/MEA 74HOT/PAT |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|----------------------------------|---|------------------------|---|---|-------------------|--|-----------|
| HO ⁻ | | | | | | $\Delta_f H(AH) = -242$ | 82TN270 |
| HO ⁻ | | | | | | $BDE(A-H) = 499$ | 85JANAF |
| | | 1.829±0.010 | | | LPES | | 74CEL/BEN |
| | | 1.83±0.04 | | | PD | | 66BRA |
| | | 1.8±0.2 | | | EIAP | From MeOH, EtOH, nPrOH | 64TSU/HAM |
| | | 1.9±0.1 | | | SI | | 69PAG/GOO |
| DO ⁻ | | | | | | $\Delta_f H(AH) = -249$ | 82TN270 |
| DO ⁻ | | | | | | $BDE(A-H) = 504 \pm 1$ | 82TN270 |
| | -139±1 ^a | 1.826 | 1640±1 ^c | 1615±1 ^h | LPD | Given: 1.822549(37) eV | 82SCH/MEA |
| | | 1.823±0.002 | | | LPD | | 74HOT/PAT |
| HO ₂ ⁻ | | | | | | $\Delta_f H(AH) = -136 \pm 1$ | 82TN270 |
| HOO ⁻ | | | | | | $BDE(A-H) = 365 \pm 3$ | 82TN270 |
| * | -94±10 ^a | 1.08±0.12 ^d | 1573±9 ^g | 1542±8 | IMRE | | 81BIE/SCH |
| * | | 1.078±0.017 | | | LPES | | 85OAK/HAR |
| | | 1.9±0.1 | | | Ther | From a solution phase thermodynamic cycle | 80BEN/NAN |
| | | 1.19±0.01 | | 1536±12 | IMRE ^o | | 81BIE/SCH |
| DO ₂ ⁻ | | | | | | | |
| DOO ⁻ | | | | | | | |
| * | | 1.089±0.017 | | | LPES | | 85OAK/HAR |
| HO ₂ S ⁻ | | | | | | | |
| HSO ₂ ⁻ | | | | | | | |
| | -415 ^c | | 264±67 ^g | 238±67 | IMRB | | 85LAH/HAY |
| HO ₃ S ⁻ | | | | | | | |
| SO ₂ ·OH ⁻ | | | | | | | |
| | <-802 ^c | | > 368 | | IMRB | CO ₂ ·HO ⁻ + SO ₂ → | 84HIE/PAU |
| HO ₄ S ⁻ | | | | | | $\Delta_f H(AH) = -735 \pm 8$ | 85JANAF |
| HSO ₄ ⁻ | | | | | | | |
| | <-953±17 ^a | | < 1312±8 | < 1281±10 ^h | EIAP | From H ₂ SO ₄ (Appearance Potential = 0eV) | 86ADA/SMI |
| | 4.5 ^d | | < 1320 ^g | < 1289 | IMRB | I ⁻ + H ₂ SO ₄ → | 80VIG/PER |
| HO ₄ W ⁻ | | | | | | $\Delta_f H(AH) = -906 \pm 4$ | 85JANAF |
| HWO ₄ ⁻ | | | | | | | |
| | -1084±46 ^a | | 1352±41 | 1322±48 ^h | TDEq | H ₂ WO ₄ + e ⁻ = HWO ₄ ⁻ + H measured | 70JEN/MIL |
| HP ⁻ | | | | | | $\Delta_f H(AH) = 139 \pm 3$ | 86BER/CUR |
| PH ⁻ | | | | | | $BDE(A-H) = 315 \pm 11$ | 86BER/CUR |
| * | 137±12 ^a | 1.028±0.010 | 1528±9 ^c | | LPES | | 76ZIT/LIN |
| | | 1.00±0.06 | | | PD | | 77RAC/FEL |
| | 218±18 | > 0.5±0.2 | | | EIAP | From PH ₃ | 69HAL/PLA |
| | | < 1.1 | | | IMRB | | 64EBI/KRA |
| HS ⁻ | | | | | | $\Delta_f H(AH) = -21 \pm 1$ | 82TN270 |
| HS ⁻ | | | | | | $BDE(A-H) = 381 \pm 1$ | 82TN270 |
| * | -81±10 ^a | 2.32±0.10 ^d | 1469±9 ^g | 1443±8 | IMRE | | 79BAR/SCO |
| * | | 2.310±0.010 | 1470±2 ^c | 1443±3 ^h | LPD | | 80JAN/REE |
| | | | 1473±9 ^g | 1446±8 | IMRE | | 78CUM/KEB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--------------------------------|--|------------------|--|--|-------------------|-------------------------------|------------|
| HS ⁻ | | | | | | $\Delta_f H(AH) = -21 \pm 1$ | 82TN270 |
| HS ⁻ | | | | | | $BDE(A-H) = 381 \pm 1$ | 82TN270 |
| | | 2.302±0.001 | | | LPD | | 74EYL/ATK |
| | | 2.32±0.01 | | | PD | | 68STE |
| -69 | | 2.2 ⁱ | | | Endo | | 76REF |
| | | 2.30±0.04 | | | SI | | 69PAG/GOO |
| | | | | 1453±8 | IMRE ^o | | 79BAR/SCO |
| HSe ⁻ | | | | | | $\Delta_f H(AH) = 30 \pm 8$ | 82TN270 |
| HSe ⁻ | | | | | | $BDE(A-H) = 368 \pm 19$ | 72DON/LIT |
| • -34±28 ^a | 2.213 | | 1466±19 ^c | 1440±19 ^h | LPD | | 86STO/LAR |
| | 2.21±0.03 | | | | PD | | 72SMY/BRA3 |
| | | | 1434±38 ^g | 1407±38 | IMRB | Between H ₂ S, HCl | 72DIX/HOL |
| HSi ⁻ | | | | | | $\Delta_f H(AH) = 237 \pm 16$ | 81DON/WAL |
| SiH ⁻ | | | | | | $BDE(A-H) = 353 \pm 8$ | 81DON/WAL |
| • 249±3 ^b | 1.277±0.009 | | | | LPES | | 75KAS/HER |
| HTe ⁻ | | | | | | $\Delta_f H(AH) = 100 \pm 2$ | 82TN270 |
| TeH ⁻ | | | | | | | |
| • | 2.102±0.015 | | | | LPES | | 86FRE/SNO |
| HZn ⁻ | | | | | | | |
| ZnH ⁻ | | <0.9 | | | PD | | 77RAC/FEL |
| H ₂ IO ⁻ | | | | | | | |
| HOH··I ⁻ | | | | | | | |
| • -472 ^c | | | 42±4 | 23±9 | TDAs | | 84CAL/KEB |
| | | | 46 | 22±1 | TDAs | | 80KBE/CAS2 |
| | | | 43±8 | 23±8 | TDAs | | 70ARS/YAM |
| H ₂ Mn ⁻ | | | | | | | |
| MnH ₂ ⁻ | | | | | | | |
| • | 0.444±0.016 | | | | LPES | | 86MIL/FEI |
| D ₂ Mn ⁻ | | | | | | | |
| MnD ₂ ⁻ | | | | | | | |
| • | 0.465±0.014 | | | | LPES | | 86MIL/FEI |
| H ₂ N ⁻ | | | | | | $\Delta_f H(AH) = -46$ | 82TN270 |
| NH ₂ ⁻ | | | | | | $BDE(A-H) = 449 \pm 3$ | 82TN270 |
| • 113±4 ^a | 0.75±0.06 ^d | | 1689±3 ^g | 1657±3 | IMRE | | 76MAC/HEM |
| • | 0.776±0.037 | | | | LPES | | 74CEL/BEN |
| | 0.744±0.022 | | | | LPD | | 72SMY/BRA2 |
| | 0.740±0.030 | | | | LPD | | 71SMY/MCI |
| | 0.76±0.04 | | | | PD | | 71FEL |
| 49±19 | | | | | EIAP | From NH ₃ | 68COL/HUB |
| | 1.1 | | | | SI | | 69PAG/GOO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot Y^-)$ | Method | Comment | Reference |
|--|--|---|--|--|---|--|--|
| $H_2NO_3^-$ HOH · NO ₂ ⁻ | • -494 ^c | | 64 60±8 | 34±1 33±8 34±24 | TDA TDA Endo LPD | | 80LEE/KEE 71PAY/YAM 82PAU/DAL 79SMI/LBE2 |
| | • -431 ^c | 2.850 | | | | | |
| $H_2NO_4^-$ HOH · NO ₃ ⁻ | • -610 ^c | | 61±1 52±8 | 30±1 28±8 | TDA TDA | | 80LEE/KEE 71PAY/YAM |
| $H_2NO_4^-$ HOOH · NO ₂ ⁻ | -410±15 ^c | | 85±4 | 60±4 | TDEq | Relative to HOH · NO ₂ ⁻ , 80KEE/LEE | 84BOH/FAH |
| $H_2NO_5^-$ HOOH · NO ₃ ⁻ | • -524±6 ^c | | 80±4 | 54±4 | TDEq | Relative to HOH · NO ₃ ⁻ , 80KEE/LEE | 84BOH/FAH |
| H_2NS^- H ₂ NS ⁻ | | | 1493±16 ^g | 1467±13 Between CF ₃ CH ₂ OH and H ₂ S, comparable to MeSH 1480±13 1476±13 | IMRB IMRB IMRB ^o | | 81DEP/BIE 84BIE/GRA 81DEP/BIE |
| H_2Ni^- NiH ₂ ⁻ | • | 1.934±0.008 | | | LPES | | 86MIL/FEI |
| D_2Ni^- NiD ₂ ⁻ | • | 1.926±0.007 | | | LPES | | 86MIL/FEI |
| H_2O^- HO · H ⁻ | -34±17 | | 287 ^k | | IMRB | | 84DEK/NIB |
| $H_2O_3^-$ HOH · O ₂ ⁻ | • -361 ^c | | 77±8 | 52±8 49±8 | TDA IMRE | | 70ARS/KEB 71PAR |
| H_2P^- PH ₂ ⁻ | • 27±10 ^a • | 1.19±0.14 ^d 1.271±0.010 1.25±0.03 1.300±0.030 | 1552±8 ^g 1544±6 ^c | 1520±8 | IMRE LPES PD LPD EIAP EIAP | $\Delta_f H(AH) = 5 \pm 2$ $BDE(A-H) = 354 \pm 5$ | 61GUN/GRE 86BER/CUR 79BAR/SCO 76ZIT/LIN 72SMY/BRA 71SMY/MCI 69HAL/PLA 64EBI/KRA |
| | <9±21 ^a | <1.4±0.3 ^d | <1524±19 <1534±19 | | | | |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|---|--|--|--|--|--|
| H_2P^- PH_2^- | | 1.6 | | 1519±8 | SI IMRE ^o | $\Delta_f H(AH) = 5 \pm 2$ $BDE(A-H) = 354 \pm 5$ | 61GUN/GRE 86BER/CUR 69PAG/GOO 79BAR/SCO |
| H_2Si^- SiH_2^- | * 181±10 ^b | 1.123±0.022 | 1508±17 ^c | | LPES | $\Delta_f H(AH) = 203 \pm 6$ $BDE(A-H) = 304 \pm 15$ | 87BOO/ARM 87BOO/ARM 75KAS/HER |
| H_3O^- $HOH \cdot \cdot H^-$ | -169±17 -199±49 | | 72±21 ^k | | Endo IMRB | $HOH \cdot \cdot HO^- + H_2 \rightarrow$ $H^- + HCO_2H \rightarrow$ | 84PAU/HEN 83KLE/NIB |
| H_3OSi^- H_3SiO^- | <-107 | | | | IMRB | $HO^- + SiH_4 \rightarrow$ | 76PAY/TAN |
| $H_3O_2^-$ $HOH \cdot \cdot OH^-$ | * -479 ^c * | 2.9±0.2 | 100±8 112±4 94±8 149±29 | 78±8 84±7 71±8 | TDAs PD TDEq TDAs CIDT | | 71PAY/YAM 68GOL/STE 86MEO/SIE2 70ARS/KEB 70DEP/GIA |
| $D_3O_2^-$ $DOD \cdot \cdot OD^-$ | -491 ^c | | 112±3 | 84±5 | TDAs | | 86MEO/SIE |
| $H_3O_5S^-$ $HOH \cdot \cdot HSO_4^-$ | | | 50±4 | 25±4 | TDAs | | 84BOH/FAH |
| $H_3O_6S^-$ $HOOH \cdot \cdot HSO_4^-$ | -1156±22 ^c | | 67±4 | 45±4 | TDEq | Relative to $HOH \cdot \cdot HSO_4^-$, 84BOH/FAH | 84BOH/FAH |
| $H_3P_2^-$ $P_2H_3^-$ | <66 | | | | IMRB | $PH_2^- + PH_3 \rightarrow$ | 72SMY/BRA |
| H_3Si^- SiH_3^- | * 63±10 ^a * 486±10 | 1.45±0.17 ^d 1.406±0.014 <1.440±0.030 | 1558±8 ^e 1562±10 ^c | 1522±8 1519±8 | IMRE LPES LPD EIAP IMRE ^o | $\Delta_f H(AH) = 35 \pm 2$ $BDE(A-H) = 386 \pm 8$ | 81BEL/PER 87BOO/ARM 79BAR/SCO 86NIM/ELL2 74REE/BRA 64EBI/KRA 79BAR/SCO |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|-------------|---|---|--------|---|--|
| D_3Si^- SiD_3^- | | 1.386±0.022 | | | LPES | | 86NIM/ELL2 |
| H_4N^- $NH_3 \cdots H^-$ | | 1.110±0.019 | | | LPES | | 87SNO/COE3 |
| | | 1.110±0.019 | | | LPES | | 85COE/SNO |
| $H_5N_2^-$ $NH_3 \cdots NH_2^-$ 16 ^c | | | 50 | | PDis | | 87SNO/COE2 |
| H_5Si^- $SiH_4 \cdots H^-$ 86 ^c | | | 94±19 | | IMRE | | 86HAJ/SQU |
| $H_7N_2^-$ $(NH_3)_2 \cdots H^-$ -114 ^c | | 1.460±0.019 | | | LPES | | 87SNO/COE3 |
| $H_8N_3^-$ $(NH_3)_2 \cdots NH_2^-$ | | 1.780±0.019 | | | LPES | | 87SNO/COE2 |
| I^- I^- • -188±1 ^a | 3.059 | | 1315 ^c | 1294±1 ^h | LOG | $\Delta_f H(A) = 107$ | 82BAU/COX 83WEB/MCD |
| IK_2^- K_2I^- -244±11 | | | | | TDEq | | 84BUR/KUD |
| ILi^- LiI^- <-199 ^b | >1.1 | | | | EIAP | $\Delta_f H(A) = -91±8$ From $(LiI)_2$ | 85JANAF 64EBI |
| IN^- IN^- 96±21 | 1.3±0.2 | | | | Endo | $\Delta_f H(A) = 215$ $I^- + NO_2 \rightarrow$ | 76REF2 76REF2 |
| IO^- IO^- <-66 -42±35 -79±21 -48 <-147 | >2.5 ⁱ >2.1±0.3 ⁱ 2.6 ⁱ 2.3 ⁱ >3.3 ⁱ | | | | Endo | $I^- + CO \rightarrow$ $I^- + O_2 \rightarrow$ $I^- + O_2 \rightarrow$ $I^- + SO_2 \rightarrow$ $O^- + I_2 \rightarrow$ | 77VOG/MIS 77VOG/DRE 76REF/FRA2 76REF/FRA 59HEN/MUC |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|----------------------|--|--|--------|--|--------------------|
| IOS ⁻ | | | | | | | |
| ISO ⁻ | | | | | Endo | I ⁻ + SO ₂ → | 76REF/FRA |
| IO ₂ S ⁻ | | | | | | | |
| SO ₂ · I ⁻ | | | | | | | |
| * | -545±10 ^c | | 60±8 | 38±11 | TDAs | | 85CAL/KEB |
| | -539±2 ^c | | 54 | 37±1 | TDAs | | 80KEB/LEE |
| IS ⁻ | | | | | | | |
| IS ⁻ | | | | | Endo | $\Delta_f H(A) = 310$ I ⁻ + H ₂ S and CS ₂ → | 76REF/FRA 76REF |
| | 47±29 | 2.7±0.3 ⁱ | | | | | |
| I ₂ ⁻ | | | | | | | |
| I ₂ ⁻ | | | | | | $\Delta_f H(A) = 62$ | 82BAU/COX |
| * | -181 ^b | 2.5±0.1 | | | NBIP | | 73BAE/AUE |
| | | 1.72±0.05 | | | NBIP | Vertical EA | 76HUB/KLE |
| | | 2.3 | | | ECD | Vertical EA: 1.7 eV | 81AYA/WEN |
| | | 2.4±0.2 | | | EnCT | | 73HUG/LIF |
| | | 2.4±0.1 | | | NBIP | | 71MOU/ATE |
| | | 2.6±0.1 | | | EIAP | From CHI ₃ | 71DEC/FRA |
| | | 2.6±0.1 | | | EnCT | | 71CHU/BER |
| I ₂ K ⁻ | | | | | | | |
| KI · I ⁻ | | | | | | | |
| | -483±8 | | | | TDEq | | 84BUR/KUD |
| I ₂ Sn ⁻ | | | | | | | |
| SnI ₂ ⁻ | | 1.7 | | | EIAP | From SnI ₄ | 77PAB/PER |
| I ₃ ⁻ | | | | | | | |
| I ₃ ⁻ | | | | | | | |
| | -482 | | 356 ^k | | Latt | | 77FIN/GAT |
| | < -207 | | | | IMRB | | 28HOG/HAR |
| $I_2^- + I_2 \rightarrow$; First negative ion/molecule reaction reported. | | | | | | | |
| I ₃ K ₂ ⁻ | | | | | | | |
| K ₂ I ₃ ⁻ | | | | | | | |
| | -760±15 | | | | TDEq | | 84BUR/KUD |
| I ₃ Sn ⁻ | | | | | | | |
| SnI ₃ ⁻ | | 3.21±0.01 | | | EIAP | From SnI ₄ | 78PAB/MAR |
| | | 3.2 | | | EIAP | From SnI ₄ | 77PAB/PER |
| I ₃ Ti ⁻ | | | | | | | |
| TiI ₃ ⁻ | | | | | | $\Delta_f H(A) = -150 \pm 33$ | 85JANAF |
| | < -240±18 | > 0.9 | | | EIAP | From TiI ₄ | 74BEN/PAB |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--------------------------------|--|--|--|--|--|--|---|
| In ⁻ | | | | | | | |
| In ⁻ | • 215±36 ^a | 0.3±0.2 0.8±0.2 | 1508±28 ^c | | PD EIAP | $\Delta_f H(A) = 244$ From InBr | 82TN270 85HOT/LIN 80BRU/COT |
| Ir ⁻ | | | | | | | |
| Ir ⁻ | • 514±3 ^b | 1.565±0.008 | | | LPES | $\Delta_f H(A) = 665±2$ | 82TN270 85HOT/LIN |
| K ⁻ | | | | | | | |
| K ⁻ | • 41 ^b | 0.501 | 1448±15 ^c | 1428±17 ^h | LPD | $\Delta_f H(A) = 89$ | 82TN270 85HOT/LIN |
| KO ₄ S ⁻ | | | | | | | |
| KSO ₄ ⁻ | -992±11 | | | | TDEq | | 85RUD/SID2 |
| Li ⁻ | | | | | | | |
| Li ⁻ | • 101 ^a | 0.618±0.001 | 1492 ^c | 1470±1 ^h | LPD | $\Delta_f H(A) = 161$ | 82TN270 85HOT/LIN |
| Mo ⁻ | | | | | | | |
| Mo ⁻ | • 586±3 ^b | 0.746±0.010 | | 1402±13 | LPES IMRB | $\Delta_f H(A) = 658±2$ | 82TN270 85HOT/LIN 85SAL/LAN |
| MoO ₃ ⁻ | | | | | | | |
| MoO ₃ ⁻ | -655±40 | 2.58±0.41 ⁱ | | | TDEq | $\Delta_f H(A) = -362$ $H + HMoO_4^- = H_2O + MoO_3^-$ | 81WOO 79MIL |
| NO ⁻ | | | | | | | |
| NO ⁻ | • 89±18 ^a | 2.4±0.010 0.1±0.1 2.0±0.1 0.0±0.1 0.7±0.2 2.500±0.007 >0.1±0.1 1.5±0.1 2.60±0.02 >6.0±0.1 >9.0 0.0±0.2 >0.7±0.1 0.8±0.1 0.8 0.9 | 1519±9 ^c | 1492±10 ^h | LPES ECD CIDT NBIP Endo ETS NBIP EnCT Kine EnCT EnCT NBIP EIAP EIAP SI SI | $\Delta_f H(A) = 91$ From NO ₂ From EtONO, nBuONO | 82BAU/COX 72SIE/CEL 83CHE/WEN 78TIE/WU 77DUR/PAR 76REF2 74BUR 73NAL/COM 73HUG/LIF 72PAR/SUG 71CHA 71BER/CHU 70LAC/HER 69STO/COM 68WIL/HAM 69PAG/GOO 64FAR/PAG |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|--|---|--|---|---|---|--|--|
| NOS⁻ | | | | | | | |
| NSO ⁻ | | | | | | | |
| | <-142 | | | 1414±21 | IMRB | NH ₂ ⁻ + SO ₂ → | 84BIE/GRA |
| NO₂⁻ | | | | | | | |
| NO ₂ ⁻ | | | | | | $\Delta_f H(AH) = -80 \pm 8$ $BDE(A-H) = 331 \pm 9$ | 82BAU/COX 82BAU/COX |
| * | -189±10 ^b | 2.30±0.10 2.359±0.100 2.31±0.11 2.1±0.2 2.350±0.100 2.800±0.050 <2.6 2.4±0.1 2.1 >2.5±0.1 2.50±0.05 2.3±0.1 2.38±0.06 2.5±0.1 1.8±0.2 <3.9 2.0 2.3±0.1 2.1±0.2 | 1421±18 ^c | 1389±18 ^h | TDEq LPES IMRE ECD LPD LPD IMRB CIDT EnCT NBIP NBIP IMRB IMRB NBIP NBIP PD EnCT EnCT IMRB | | 87KEB/CHO 74HER/PAT 85GRI/CAL 83CHE/WEN 79SMI/LEE2 74RIC/STE 72FER/DUN 78TIE/WU 76REF2 73NAL/COM 73LEF/JAC 73HUG/LIF 72DUN/FEH 72BAE 77DUR/PAR 71MIL/JAC 71BER/CHU 70LIF/HUG 69VOG |
| | 9±29 | | | | IMRE | CO ₃ ⁻ + NO → [isomer?] | 70ADA/BOH 69WAR |
| | | | | | PD | | 64TSU/HAM 62CUR |
| | | | | | EnCT | From MeNO ₂ , EtNO ₂ | 74RIC/STE 64FAR/PAG |
| | | | | | LPD | | |
| | | | | | SI | | |
| NO₃⁻ | | | | | | | |
| NO ₃ ⁻ | | | | | | $\Delta_f H(AH) = -135$ $BDE(A-H) = 424 \pm 21$ | 82BAU/COX 77DAV/FEH |
| * | -307±1 ^a | 3.92±0.24 ⁱ | 1358±1 | 1330±1 | TDEq | | 77DAV/FEH |
| | | | | | | Relative to HBr, reevaluated with current HBr acidity | |
| | | | 1358±2 | | TDEq | | 72FER/DUN |
| | | | 1371±24 | | Endo | I ⁻ + HNO ₃ → | 76REF/FRA3 |
| | | | 1380±20 | | NBAP | From HNO ₃ | 76MAT/ROT2 |
| | | 3.70±0.20 | | | IMRE | NO ₃ ⁻ + NO = NO ₂ ⁻ + NO ₂ | 72MCF/DUN |
| | | | 1491 | | Endo | | 71BER/CHU |
| | <-10 | | | | IMRB | O ₄ ⁻ + NO → NO ₃ ⁻ + O ₂ ; isomer? | 70ADA/BOH |
| | -135±21 ^c | | 276±21 | | PDis | | 78SMI/LEE |
| | | | <193 | | PDis | isomer: O ₄ ⁻ + NO → | 79SMI/LEE2 |
| NO₄S⁻ | | | | | | | |
| SO ₂ · NO ₂ ⁻ | | | | | | | |
| * | -594±12 ^c | | 108±1 | 62±1 | TDA | | 80KEE/LEE |
| | | | 102±4 | 62±4 | TDEq | Relative to HOH · NO ₂ ⁻ , 80KEE/LEE | 84BOH/FAH |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|---|--|-------------------|---|---|-----------------------------|--|---|
| NO_5^- $SO_2 \cdots NO_3^-$ • -676 ± 11^c | | | 72 ± 8 76 ± 5 | 40 ± 8 37 ± 4 | TDEq TDAs | Relative to $HOH \cdots NO_3^-$, 80KEE/LBE | 84BOH/FAH 83WLO/LUC |
| NS^- NS^- • | | 1.194 ± 0.011 | | | LPES | | 82BUR/FEI |
| N_2O^- N_2O^- 61 ± 10^b 67 ± 12^c | 0.2 ± 0.1 0.2 ± 0.1^i 0.3 ± 0.2 0.760 ± 0.100 | | 47 ± 12^k 41 ± 10 | | CIDT CIDT ECD LPES | $\Delta_f H(A) = 82$ Vertical detachment: -2.23 ± 0.20 eV Vertical detachment | 82BAU/COX 78TIE/WU 76HOP/WAH 73NAL/COM 71WEN/CHE 86COE/SNO |
| $N_2O_2^-$ $N_2 \cdots O_2^-$ > -99^c | | | < 57 | | IMRB | $N_2 \cdots O_2^- + O_2 \rightarrow O_4^-$ | 70ADA/BOH |
| $N_2O_3^-$ $N_2O \cdots O_2^-$ > -17^c | | | < 57 | | IMRB | $N_2O \cdots O_2^- + O_2 \rightarrow O_4^- + N_2O$ | 70ADA/BOH |
| N_3^- N_3^- • • 203 ± 15^a 199 ± 29 | 2.762 ± 0.043 2.7 ± 0.1 > 2.540 3.1 ± 0.3^i | | 1439 ± 13^g 1418 ± 21^f | 1414 ± 12 | LPD IMRB LPES BIAP | $\Delta_f H(AH) = 294 \pm 2$ $D-EA$ $BDE(A-H) = 387 \pm 21$ Acidity near HCO_2H From McN_3 and HN_3 | 82TN270 85ILL/COM 81PEL/JAC 76ENG/LIN 58FRA/DIB |
| $N_3O_2^-$ $N_2O \cdots NO^-$ 151 ± 18^b | 0.258 ± 0.009 | | 19^k | | LPES | | 87COE/SNO |
| $N_4O_2^-$ $(N_2O)_2^-$ | 0.950 ± 0.100 | | | | LPES | | 86COE/SNO |
| $N_5O_3^-$ $(N_2O)_2 \cdots NO^-$ 213 ± 18^c | 0.513 ± 0.022 | | 19^k | | LPES | | 87COE/SNO |
| Na^- Na^- • 54 ± 1^b | 0.548 | | 1455 ± 1^e | 1434 ± 3^h | LPD | $\Delta_f H(AH) = 130$ $BDE(A-H) = 195 \pm 1$ | 82TN270 85JANAF 85HOT/LIN |
| Nb^- Nb^- • 647 ± 11^b | 0.893 ± 0.025 | | | | LPES | $\Delta_f H(A) = 733 \pm 8$ | 85JANAF 85HOT/LIN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|-------------------------------|---|-----------------------|---|---|--------|---|------------|
| Ni ⁻ | | | | | | | |
| Ni ⁻ | | | | | | $\Delta_f H(A) = 430 \pm 2$ | 82TN270 |
| * 318 ± 3 ^b | 1.156 ± 0.010 | | | | LPES | | 85HOT/LIN |
| O ⁻ | | | | | | | |
| O ⁻ | | | | | | $\Delta_f H(AH) = 39 \pm 1$ $BDE(A-H) = 428 \pm 1$ | 85JANAF |
| * 108 ± 2 ^a | 1.461 | 1599 ± 1 ^c | 1574 ± 2 ^h | | LPD | Given: 1.461122(3) eV | 85JANAF |
| | 1.462 | | | | LPD | | 85NEU/LYK |
| | | | | | | | 85HOT/LIN |
| OP ⁻ | | | | | | | |
| OP ⁻ | | | | | | $\Delta_f H(A) = -33 \pm 13$ | 83PED/MAR |
| * -139 ± 14 ^b | 1.092 ± 0.010 | | | | LPES | | 76ZIT/LIN |
| OS ⁻ | | | | | | | |
| SO ⁻ | | | | | | $\Delta_f H(A) = 5 \pm 1$ | 85JANAF |
| * -100 ± 6 ^b | 1.09 ± 0.05 | | | | PD | | 70FEL |
| -111 | 1.12 ± 0.01 ⁱ | | | | Endo | $\Gamma + SO_2 \rightarrow$ | 76REF/FRA |
| | 1.1 | | | | EIAP | From SO ₂ | 73HAR/FRA |
| | > 1.2 ± 0.1 | | | | EIAP | From SO ₂ | 72THY |
| | < 1.1 | | | | IMRB | | 61KRA/MUL |
| | < 1.1 | | | | IMRB | EA: < SO ₂ | 59HEN/MUC |
| -100 | 1.2 | | | | EIAP | From SO ₂ | 58REE/DIB |
| OS ₂ ⁻ | | | | | | | |
| S ₂ O ⁻ | | | | | | $\Delta_f H(A) = -53$ | 86NIM/ELL |
| * -234 ± 1 ^b | 1.877 ± 0.008 | | | | LPES | | 86NIM/ELL |
| OSe ⁻ | | | | | | | |
| SeO ⁻ | | | | | | $\Delta_f H(A) = 53$ | 82TN270 |
| * -87 ± 2 ^b | 1.456 ± 0.020 | | | | LPES | | 86COE/SNO2 |
| OTe ⁻ | | | | | | | |
| TeO ⁻ | | | | | | $\Delta_f H(A) = 69 \pm 21$ | 83PED/MAR |
| * -95 ± 23 ^b | 1.697 ± 0.022 | | | | LPES | | 86FRE/COE |
| O ₂ ⁻ | | | | | | | |
| O ₂ ⁻ | | | | | | $\Delta_f H(AH) = 10 \pm 8$ $BDE(A-H) = 206 \pm 8$ | 82TN270 |
| * -42 ± 1 ^b | 0.440 ± 0.008 | 1476 ± 9 ^c | 1449 ± 9 ^h | | LPES | | 82TN270 |
| | 0.430 ± 0.030 | | | | LPES | | 72CEL/BEN |
| | 0.45 ± 0.05 | | | | ECD | | 71CEL/BEN |
| | 0.4 ± 0.1 | | | | CIDT | From O ₂ ⁻ | 83CHE/WEN |
| | 0.4 ± 0.1 | | | | NBIP | | 78TIE/WU |
| | 0.45 ± 0.02 | | | | ETS | | 77DUR/PAR |
| | 0.5 ± 0.1 | | | | NBIP | | 74BUR |
| | > 0.5 ± 0.1 | | | | EnCT | | 72BAE |
| | 0.46 ± 0.05 | | | | NBIP | | 71TIE/HUG |
| | > 0.6 ± 0.1 | | | | EnCT | | 71NAL/COM |
| | > 0.5 | | | | EnCT | | 71CHA |
| | 1.12 ± 0.07 | | | | IMRB | | 71BER/CHU |
| | 0.5 ± 0.2 | | | | NBIP | | 70VOG/HAU |
| | > 1.3 ± 0.2 | | | | EnCT | | 70LAC/HER |
| | > 1.1 ± 0.1 | | | | EIAP | From NO ₂ | 70BAI/MAH |
| | | | | | | | 69STO/COM |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|----------------------------|--|---|--|--|--|---|--|
| O_2^- $O_2^{\cdot -}$ | | 0.43±0.02 0.15±0.05 | | | Kine PD | $\Delta_f H(AH) = 10 \pm 8$ $BDE(A-H) = 206 \pm 8$ | 82TN270 82TN270 66PAC/PHE 58BUR/SMI |
| O_2P^- PO_2^- | | 3.80±0.22 ⁱ 3.00 | | | TDEq IMRE | $\Delta_f H(A) = -280 \pm 2$ | 85JANAF 86RUD/VOV 79WOR/KOB |
| O_2S^- SO_2^- | | 1.107±0.008 1.097±0.036 1.00±0.05 1.1 1.1±0.2 1.0±0.1 1.1±0.1 | | | LPES LPES PD EnCT NBIP EnCT IMRB | $\Delta_f H(A) = -297 \pm 1$ Between NH_2^- , C^- | 85JANAF 86NIM/ELL 74CEL/BEN 70FEL 76REF/FRA 75ROT/TAN 73HUG/LIF 61KRA/MUL |
| O_2Se^- SeO_2^- | | 1.823±0.040 | | | LPES | | 87SNO/COE |
| O_2Te^- TeO_2^- | | > 2.200 | | | LPES | | 87SNO/COE |
| O_3^- $O_3^{\cdot -}$ | | 2.103±0.003 1.9±0.1 2.2±0.4 ⁱ 2.06±0.06 ⁱ 2.2 ⁱ > 1.8 2.1±0.2 > 2.0 | 174±19 174±5 < 207 174±10 | | LPES PD PDis CIDT PDis CIDT IMRB NBIP EnCT | $\Delta_f H(A) = 143 \pm 2$ Excited state: 81 kJ up $I^- + O_3 \rightarrow$ | 82TN270 79NOV/ENG 71WON/VOR 78SMI/LEE 78LIF/WU 78COS/MOS 77WU/TIE 77DOT/DAV 75ROT/TAN 71BER/CHU |
| O_3P^- PO_3^- | | 4.49±0.53 ⁱ > 4.6 ^d 3.5 | < 1323±15 ^g | < 1293±13 | TDEq TDEq IMRB IMRB | $\Delta_f H(AH) = -565 \pm 63$ $BDE(A-H) = 456 \pm 167$ | 85HEN/VIG 85HEN/VIG 86RUD/VOV 83SID/RUD 85HEN/VIG 79WOR/KOB |
| O_3Re^- ReO_3^- | | 3.01±0.43 ⁱ > 2.5 | | | TDEq IMRB | $\Delta_f H(A) = -284 \pm 21$ | 75GOU/MIL 75GOU/MIL 72CEN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|---|--|---|--|--|---|---|--|
| O_3S^- SO_3^- | -560 ± 15^b -601 ± 8 | 1.7 ± 0.2 | | | NBIP TDEq | $\Delta_f H(A) = -396 \pm 1$ | 82TN270 75ROT/TAN 85RUD/SID2 |
| O_3W^- WO_3^- | -698 ± 40 | 3.64 ± 0.41^i > 2.5 | | | TDEq IMRB | $\Delta_f H(A) = -293$ $HWO_4^- + H = WO_3^- + H_2O$ | 81WOO 70JEN/MIL 72CEN |
| O_4^- $O_2 \cdot \cdot O_2^-$ | -194^c -99^c | 2.0 ± 0.2^i 1.9 ± 0.2 | 151 ± 19 | | PDis PD TDAAs IMRE IMRB | | 78SMI/LEE 72BUR2 68CON/NES 71PAR 70ADA/BOH |
| | | | | | $O_4^- + H_2O \rightarrow O_2^- \cdot \cdot H_2O + O_2$, anchored on 70ARS/KEB | | |
| O_4Re^- ReO_4^- | -976 ± 30 -867 ± 82^a | 4.46 ± 0.52^i > 2.5 | 1328 ± 40 | | TDEq TDEq IMRB | $\Delta_f H(AH) = -665 \pm 42$ | 82TN270 83SID/RUD 75GOU/MIL 72CEN |
| O_4S^- SO_4^- | -744 ± 10 | | | | TDEq | | 85RUD/SID2 |
| $O_4S_2^-$ $SO_2 \cdot \cdot SO_2^-$ | -801 ± 4^c | | 100 ± 1 | 58 ± 2 | TDAAs | | 80KEE/LEE |
| $O_5S_2^-$ $SO_2 \cdot \cdot SO_3^-$ | -912 ± 17^c | | 56 | 32 ± 1 | TDAAs | | 80KEE/LEE |
| P^- P^- | 244 ± 1^b | 0.747 0.77 ± 0.05 | 1538 ± 10^c | 1514 ± 10^h | LPD EIAP | $\Delta_f H(AH) = 236 \pm 8$ $BDE(A-H) = 298 \pm 10$ From P_4 | 86BER/CUR 85JANAF 85HOT/LIN 74BEN/MAR |
| P_2^- P_2^- | 88 ± 5^b 156 ± 20 | 0.589 ± 0.025 < 0.7 0.2 ± 0.2 | | | LPES PD EIAP | $\Delta_f H(A) = 144 \pm 2$ From P_4 | 85JANAF 85SNO/COE 77FEL/RAC 74BEN/MAR |
| P_3^- P_3^- | 160 ± 19 | 0.9 ± 0.4 | | | EIAP | $\Delta_f H(A) = 249 \pm 17$ From P_4 | 74BEN/MAR 74BEN/MAR |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdots Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdots Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdots Y^-)$ | Method | Comment | Reference |
|------------------------------|---|--------------------------------------|---|---|--------------|--|-----------------------------------|
| Pb ⁻ | | | | | | | |
| Pb ⁻ | • 160±3 ^b | 0.364±0.008 | | | LPES | $\Delta_f H(A) = 195 \pm 2$ | 82TN270 85HOT/LIN |
| Pd ⁻ | | | | | | | |
| Pd ⁻ | • 325±3 ^b | 0.557±0.008 | | | LPES | $\Delta_f H(A) = 378 \pm 2$ | 82TN270 85HOT/LIN |
| Pt ⁻ | | | | | | | |
| Pt ⁻ | • 360±2 ^b | 2.128±0.002 | | | LPD | $\Delta_f H(A) = 565 \pm 2$ | 82TN270 85HOT/LIN |
| Rb ⁻ | | | | | | | |
| Rb ⁻ | • 34 ^b | 0.486 | | | LPD | $\Delta_f H(A) = 81$ | 82TN270 85HOT/LIN |
| Re ₂ ⁻ | | | | | | | |
| Re ₂ ⁻ | • | 1.571±0.008 | | | LPES | | 86LEO/MIL2 |
| Rh ⁻ | | | | | | | |
| Rh ⁻ | • 447±3 ^b | 1.137±0.008 | | | LPES | $\Delta_f H(A) = 557 \pm 2$ | 82TN270 85HOT/LIN |
| S ⁻ | | | | | | | |
| S ⁻ | • 77 ^b | 2.077 | 1467±5 ^c | 1444±6 ^h | LPD | $\Delta_f H(AH) = 139 \pm 5$ $BDE(A-H) = 356 \pm 5$ | 85JANAF 85JANAF 85HOT/LIN |
| S ₂ ⁻ | | | | | | | |
| S ₂ ⁻ | • -32±5 ^b <46±10 | 1.663±0.040 >0.8±0.1 ⁱ | | | LPES IMRB | $\Delta_f H(A) = 129 \pm 1$ | 85JANAF 74CEL/BEN 68DIL/FRA |
| | | >2.5±0.8 | | | EIAP | $S^- + COS \rightarrow S_2^- + CO$. Also $S_2^- + COS \rightarrow S_3^- + CO$, etc. to n=6 From CS ₂ | 72THY |
| S ₃ ⁻ | | | | | | | |
| S ₃ ⁻ | • -60±11 ^b | 2.093±0.025 2.0±0.1 | | | LPES PD | $\Delta_f H(A) = 142 \pm 8$ | 85JANAF 86NIM/ELL 77FEL/RAC |
| Sb ⁻ | | | | | | | |
| Sb ⁻ | • 159±7 ^b | 1.07±0.05 | | | PD | $\Delta_f H(A) = 262 \pm 2$ | 82TN270 85HOT/LIN |
| Sc ⁻ | | | | | | | |
| Sc ⁻ | • 360±6 ^b | 0.188±0.020 | | | LPES | $\Delta_f H(A) = 378 \pm 4$ | 82TN270 85HOT/LIN |
| Se ⁻ | | | | | | | |
| Se ⁻ | • 32 ^b | 2.021 | | | LPD | $\Delta_f H(A) = 227$ | 82TN270 85HOT/LIN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|--|--|------------------------------------|--|--|--------------|---|------------------------|
| Se ₂ ⁻ Se ₂ ⁻ | | | | | | $\Delta_f H(A) = 146 \pm 8$ | 82TN270 |
| * | -41 ± 15^b | 1.940 ± 0.070 | | | LPES | | 87SNO/COE |
| Se ₃ ⁻ Se ₃ ⁻ | | | | | | | |
| | | > 2.200 | | | LPES | | 87SNO/COE |
| Si ⁻ Si ⁻ | | | | | | $\Delta_f H(AH) = 377 \pm 8$ $BDE(A-H) = 297 \pm 10$ | 85JANAF |
| * | 322 ± 3^b | 1.385 ± 0.005 | 1475 ± 11^c | 1453 ± 12^h | LPES | | 82TN270 75KAS/HER |
| Sn ⁻ Sn ⁻ | | | | | | $\Delta_f H(A) = 302 \pm 2$ | 82TN270 |
| * | 195 ± 4^b | 1.113 ± 0.020 1.1 ± 0.1 | | | LPES PD | | 86MIL/MIL 85HOT/LIN |
| Ta ⁻ Ta ⁻ | | | | | | $\Delta_f H(A) = 782$ | 82TN270 |
| * | 751 ± 1^b | 0.322 ± 0.012 | | | LPES | | 85HOT/LIN |
| Te ⁻ Te ⁻ | | | | | | $\Delta_f H(AH) = 143$ | 79HUB/HER |
| * | 6^b | 1.971 | | | LPD | | 85HOT/LIN |
| Te ₂ ⁻ Te ₂ ⁻ | | | | | | $\Delta_f H(A) = 168 \pm 8$ | 82TN270 |
| * | -17 ± 15^b | 1.920 ± 0.070 | | | LPES | | 87SNO/COE |
| Te ₃ ⁻ Te ₃ ⁻ | | | | | | | |
| | | < 2.700 | | | LPES | | 87SNO/COE |
| Ti ⁻ Ti ⁻ | | | | | | $\Delta_f H(AH) = 532$ | 79HUB/HER |
| * | 462 ± 3^b | 7.9 ± 0.014 | 1460^f | | LPES | | 85HOT/LIN |
| Tl ⁻ Tl ⁻ | | | | | | $\Delta_f H(A) = 182 \pm 1$ | 82TN270 |
| * | 163 ± 20^b | 0.2 ± 0.2 1.1 ± 0.2 | | | PD EIAP | From TlBr | 85HOT/LIN 80BRU/COT |
| V ⁻ V ⁻ | | | | | | $\Delta_f H(A) = 515 \pm 8$ | 85JANAF |
| * | 464 ± 10^b | 0.525 ± 0.012 | | 1389 ± 13 | LPES IMRB | | 85HOT/LIN 85SAL/LAN |
| W ⁻ W ⁻ | | | | | | $\Delta_f H(A) = 851 \pm 6$ | 85JANAF |
| * | 772 ± 7^b | 0.815 ± 0.008 | | | LPES | | 85HOT/LIN |

Table 2. Negative Ion Table - Continued

| Ion | $\Delta_f H(A^-)$ $\Delta_f H(X \cdot \cdot Y^-)$ | EA(A) eV | $\Delta H_{acid}(AH)$ $\Delta H_{aff}(X \cdot \cdot Y^-)$ | $\Delta G_{acid}(AH)$ $\Delta G_{aff}(X \cdot \cdot Y^-)$ | Method | Comment | Reference |
|-----------------|--|-------------------|--|--|--------|-----------------------------|-----------|
| Y ⁻ | | | | | | | |
| Y ⁻ | | | | | | $\Delta_f H(A) = 421 \pm 2$ | 82TN270 |
| • | 392 ± 3^b | 0.307 ± 0.012 | | | LPES | | 85HOT/LIN |
| Zr ⁻ | | | | | | | |
| Zr ⁻ | | | | | | $\Delta_f H(A) = 610 \pm 8$ | 85JANAF |
| • | 569 ± 10^b | 0.426 ± 0.014 | | | LPES | | 85HOT/LIN |

References to Tables 1 and 2

- 28HOG/HAR**
T.R. Hogness and R.W. Harkness, "The Ionization Processes of Iodine Interpreted by the Mass Spectrograph," *Phys. Rev.* **32**, 784 (1928).
- 29ROT/MUE**
W.A. Roth and F. Mueller, *Ber. Deut. Chem. Ges.* **62**, 1188 (1929).
- 30BEN**
H.E. Bent, "The Electron Affinity of Triphenylmethyl," *J. Am. Chem. Soc.* **52**, 1498 (1930).
- 47KAP/MAK**
A.F. Kaputinskii, I.A. Makalkin and L.I. Krishtalik, *Russ. J. Phys. Chem.* **21**, 125 (1947).
- 51COL/GIL**
L.G. Cole and E.C. Gilbert, *J. Am. Chem. Soc.* **73**, 5423 (1951).
- 54HON**
R.E. Honig, "Mass Spectrometric Study of the Molecular Sublimation of Graphite," *J. Chem. Phys.* **22**, 126 (1954).
- 55ALT**
A.P. Altschuller, "Lattice Energies and Related Thermodynamic Properties of the Alkali Metal Borohydrides and of the Borohydride Ion," *J. Am. Chem. Soc.* **77**, 5455 (1955).
- 56BRO/GIN**
H.C. Brown and D. Gintis, *J. Am. Chem. Soc.* **78**, 5378 (1956).
- 58BEN/BUS**
S.W. Benson and J.H. Buss, *J. Chem. Phys.* **29**, 546 (1958).
- 58BUR/SMI**
D.S. Burch, S.J. Smith and L.M. Branscomb, "Photodetachment of O₂⁻," *Phys. Rev.* **112**, 171 (1958).
- 58FRA/DIB**
J.L. Franklin, V.H. Dibeler, R.M. Reese and M. Krauss, "Ionization and Dissociation of Hydrazoic Acid and Methyl Azide by Electron Impact," *J. Am. Chem. Soc.* **80**, 298 (1958).
- 59HEN/MUC**
A. Henglein and G.A. Muccini, "Negative Ion-Molecule Reactions," *J. Chem. Phys.* **31**, 1426 (1959).
- 58REE/DIB**
R.M. Reese, V.H. Dibeler, J.L. Franklin, "Electron Impact Studies of Sulfur Dioxide and Sulfuryl Fluoride," *J. Chem. Phys.* **29**, 880 (1958).
- 61CUR**
R.K. Curran, "Positive and Negative Ion Formation in CCl₃F," *J. Chem. Phys.* **34**, 1069 (1961).
- 61CUR2**
R.K. Curran, "Low Energy Process for F⁻ Formation in SF₆," *J. Chem. Phys.* **34**, 2007 (1961).
- 61DIB/REE**
V.H. Dibeler, R.H. Reese and J.L. Franklin, "Mass Spectrometric Study of Cyanogen and Cyanoacetylenes," *J. Am. Chem. Soc.* **83**, 1813 (1961).
- 61GOW/JON**
B.G. Gowenlock, P.P. Jones and J.R. Majer, *Trans. Farad. Soc.* **57**, 23 (1961).
- 61GUN/GRE**
S.R. Gunn and L.G. Green, *J. Phys. Chem.* **65**, 779 (1961).
- 61KRA/MUL**
K. Kraus, W. Muller-Duysing and H. Neuert, "Uber Stosse Langsamer Negativer Ionen mit Ladungsübertragung," *Z. Naturfor.* **16A**, 1385 (1961).
- 61ZIM/GEI**
H. Zimmerman and H. Geisenfelder, *Z. Elektrochem.* **65**, 368 (1961).
- 62ARM/KRI**
G.T. Armstrong and L.A. Krieger, "Progress of International Research on Thermodynamic and Transport Properties" (ed. J.F. Masi and D.H. Tsai, Academic Press, New York, 1962).
- 62CUR**
R.K. Curran, "Formation of NO₂⁻ by Charge Transfer at Very Low Energies," *Phys. Rev.* **125**, 910 (1962).
- 62MOM/BRA**
J. Momigny, L. Brakier and L. D'Or, *Bull. Classe Sci. Acad. Roy. Belg.* **48**, 1002 (1962).
- 63BIB/CAR**
M.M. Bibby and G. Carter, "Ionization and Dissociation in Some Fluorocarbon Gases," *Trans. Farad. Soc.* **59**, 2455 (1963).
- 63NAP/PAG**
R. Napper and F.M. Page, "Determination of Electron Affinities. Part 5. Cyanide and Thiocyanate Radicals," *Trans. Farad. Soc.* **59**, 1086 (1963).
- 63TRE/NEU**
L. V. Trepka and H. Neuert, "Über die Entstehung von Negativen Ionen aus einigen Kohlenwasserstoffen und Alkoholen durch Elektronenstoss," *Z. Naturfor.* **18A**, 1295 (1963).
- 64COO/CRU**
R.C. Cookson, E. Crundwell, R.R. Hill and J. Hudec, *J. Chem. Soc.* **3062** (1964).
- 64EBI**
H.Z. Ebinghaus, "Negative Ionen aus Alkalihalogeniden und Elektronenaffinitäten der Alkalimetalle und Alkalihalogenide," *Z. Naturfor.* **19A**, 727 (1964).
- 64EBI/KRA**
H. Ebinghaus, K. Kraus, H. Neuert and W. Muller-Duysing, "Negative Ionen durch Elektronenresonanzeinfang in PH₃, AsH₃, und SiH₄," *Z. Naturfor.* **19A**, 732 (1964).
- 64FAR/PAG**
A.L. Farragher, F.M. Page and R.C. Wheeler, "Electron Affinities of the Nitrogen Oxides," *Disc. Faraday Soc.* **37**, 203 (1964).
- 64GUN**
S.R. Gunn, *J. Phys. Chem.* **68**, 949 (1964).
- 64GUN/GRE**
S.R. Gunn and L.G. Green, *J. Phys. Chem.* **68**, 946 (1964).
- 64KAY/PAG**
J. Kay and F.M. Page, "Determination of Electron Affinities. Part 7. Sulphur Hexafluoride and Disulphur Decafluoride," *Trans. Farad. Soc.* **60**, 1042 (1964).
- 64SMI/GOV**
N.K. Smith, G. Govid, W.D. Good and J.P. McCollough, *J. Phys. Chem.* **68**, 940 (1964).
- 64TSU/HAM**
S. Tsuda and W.H. Hamill, "Ionization Efficiency Measurements by the Retarding Potential Difference Method," *Adv. Mass Spectrom.* **3**, 249 (1964).
- 65BAC/BET**
R.A. Back and J. Betts, *Can. J. Chem.* **43**, 2157 (1965).
- 66BEC/CHE**
R.S. Becker and E. Chen, "Extension of Electron Affinities and Ionization Potentials of Aromatic Hydrocarbons," *J. Chem. Phys.* **45**, 2403 (1966).
- 66BRA**
L.M. Branscomb, "Photodetachment Cross Section, Electron Affinity, and Structure of the Negative Hydroxyl Ion," *Phys. Rev.* **148**, 11 (1966).
- 66COM/CHR**
R.N. Compton, L.G. Christophorou, G.S. Hurst and P.W. Reinhardt, "Nondissociative Electron Capture in Complex Molecules and Negative Ion Lifetimes," *J. Chem. Phys.* **45**, 4634 (1966).

- 66FAR/PAG**
A.L. Farragher and F.M. Page, "Experimental Determination of Electron Affinities. Part 9. - Benzoquinone and Chloranil and Related Compounds," *Trans. Farad. Soc.* **62**, 3072 (1966).
- 66GAI/KAY**
A.F. Gaines, J. Kay and F.M. Page, "Determination of Electron Affinities. Part 8. - CCl_4 , CHCl_3 , and CH_2Cl_2 ," *J. Chem. Soc. Faraday Trans.* **62**, 874 (1966).
- 66PAC/PHE**
J.L. Pack and A.V. Phelps, "Electron Attachment and Detachment. I. Pure O_2 at Low Energy," *J. Chem. Phys.* **44**, 1870 (1966).
- 66PAC/PHE2**
J.L. Pack and A.V. Phelps, "Electron Attachment and Detachment. II. Mixtures of O_2 and CO_2 and of O_2 and H_2O ," *J. Chem. Phys.* **45**, 4316 (1966).
- 66WAD**
I. Wadso, *Acta Chem. Scand.* **20**, 544 (1966).
- 67ARN/SAN**
E.M. Arnett, J.C. Sanda, J.M. Bollinger and M. Barber, *J. Am. Chem. Soc.* **89**, 5389 (1967).
- 67BID/MCI**
D.R. Bidinosti and N.S. McIntyre, *Can. J. Chem.* **45**, 641 (1967).
- 67CHE/BAR**
C.J. Chectam and R.F. Barrow, *Adv. High Temp. Chem.* **1**, 7 (1967).
- 67COL/HUB**
J.E. Collin, M.J. Hubin-Franskin and L. D'Or, "Negative Ions Produced by Electron Impact in Ammonia, Methylamine, and Deuterated Methylamine- Nd_2 ," *Adv. Mass Spectrom.* **4**, 713 (1967).
- 67FAR/PAG**
A.L. Farragher and F.M. Page, "Experimental Determination of Electron Affinities. Part 11. - Electron Capture by Some Cyanocarbons and Related Compounds," *Trans. Farad. Soc.* **63**, 2369 (1967).
- 67GUN**
S.R. Gunn, *J. Phys. Chem.* **71**, 2934 (1967).
- 67HAS/BLO**
J.W. Hastie, H. Bloom and J.D. Morrison, "Electron Impact Studies of PbCl_2 , PbBr_2 , and PbBrCl ," *J. Chem. Phys.* **47**, 158 (1967).
- 67HIR**
J.A. Hirsch, *Top. Stereochem.* **1**, 199 (1967).
- 67JAE/HEN**
K. Jaeger and A. Henglein, "Negative Ionen durch Elektronenstoss aus Organischen Nitroverbindungen. Athylnitrit und Athylnitrat," *Z. Naturfor.* **22A**, 700 (1967).
- 67KOR/PEP**
B.L. Korsunskii, V.I. Pepekin, Yu.A. Lebedev and A. Ya. Apin, *Bull. Acad. Sci. USSR Div. Chem. Sci.* 509 (1967).
- 67LOU/LAI**
L.F. Loucks and K.J. Laidler, *Can. J. Chem.* **45**, 2785 (1967).
- 67MAN/KOE**
A. Mannschreck and O. Koelle, *Tetrahedron Lett.* 863 (1967).
- 67SHV/TAY**
Y.H. Shvo, E.C. Taylor and J. Bartulin, *Tetrahedron Lett.* 3259 (1967).
- 67WEN/CHE**
W.E. Wentworth and E. Chen, "Experimental Determination of the Electron Affinity of Several Aromatic Aldehydes and Ketones," *J. Phys. Chem.* **71**, 1929 (1967).
- 68BOU/CHA**
R. Bougon, J. Chatelet, J.P. Desmolin and P. Plurien, *Compt. Rend. Acad. Sci.* **266C**, 176 (1968).
- 68BRA/BLA**
J.I. Brauman and L.K. Blair, "Gas Phase Acidities of Carbon Acids," *J. Am. Chem. Soc.* **90**, 5636 (1968).
- 68CHE/GEO**
E.C.M. Chen, R.D. George and W.E. Wentworth, "Experimental Determination of Rate Constants for Thermal Electron Attachment to Gaseous SF_6 and C_7F_{14} ," *J. Chem. Phys.* **49**, 1973 (1968).
- 68CHU/RUS**
W.A. Chupka and M.E. Russell, *J. Chem. Phys.* **49**, 5426 (1968).
- 68COL/HUB**
J.E. Collin, M.J. Hubin-Franskin, L. D'Or, "Negative Ions Produced by Electron Impact in Ammonia, Methylamine, and Deuterated Methylamine- Nd_2 ," *Adv. Mass Spectrom.* **4**, 713 (1968).
- 68CON/NES**
D.C. Conway, L.E. Nesbit, "Stability of O_4^- ," *J. Chem. Phys.* **48**, 509 (1968).
- 68CZA/CAS**
J. Czarnowski, E. Castellano and H. Schumacher, "The Energy of the O-F Bond in CF_3OF ," *Chem. Comm.*, 1255 (1968).
- 68DIL/FRA**
J.G. Dillard and J.L. Franklin, "Ion-Molecule Reactions of Negative Ions. I. Negative Ions of Sulfur," *J. Chem. Phys.* **48**, 2349 (1968).
- 68DUN**
R.C. Dunbar, "Ion-Molecule Chemistry of Diborane by Ion Cyclotron Resonance," *J. Am. Chem. Soc.* **90**, 5676 (1968).
- 68GAI/PAG**
A.F. Gaines and F.M. Page, "The Stabilities of Negative Ions. I. The Methyl-, Diphenylmethyl, and Triphenylmethyl Negative Ions," *Int. J. Mass Spectrom. Ion Phys.* **1**, 315 (1968).
- 68GOL/STE**
S. Golub and B. Steiner, "Photodetachment of $[\text{OH}(\text{H}_2\text{O})]^-$," *J. Chem. Phys.* **49**, 5191 (1968).
- 68JAE/HEN**
K. Jaeger and A. Henglein, "Die Bildung Negativer Ionen aus SiCl_4 und Organischen Siliciumchloriden durch Elektronenstoss," *Z. Naturfor.* **23A**, 1122 (1968).
- 68KUH/LEV**
W.F. Kuhn, R.J. Levins and A.C. Lilly, Jr., "Electron Affinities and Ionization Potentials of Phthalate Compounds," *J. Chem. Phys.* **49**, 5550 (1968).
- 68LYO/MOR**
L.E. Lyons, G.C. Morris and L.J. Warren, "Electron Affinities and the Electron Capture Method for Aromatic Hydrocarbons," *J. Phys. Chem.* **72**, 3677 (1968).
- 68LAC/SKI**
J.R. Lacher and H.A. Skinner, *J. Chem. Soc. A* 1034 (1968).
- 68LOW**
J.P. Lowe, *Prog. Phys. Org. Chem.* **6**, 1 (1968).
- 68STE**
B. Steiner, "Photodetachment of Electrons From SH^- ," *J. Chem. Phys.* **49**, 5097 (1968).
- 68WEN/CHE**
W.E. Wentworth, E. Chen and J.C. Steelhammer, "Determination of Electron Affinities of Radicals and Bond Dissociation Energies by Electron Attachment Studies at Thermal Energies - Electron Affinity of the Acetate Radical," *J. Phys. Chem.* **72**, 2671 (1968).
- 68TUR/GOE**
R.B. Turner, P. Goebel, B.J. Mallon, W. von E. Doering, J.F. Coburn, Jr. and M. Pomerantz, *J. Am. Chem. Soc.* **90**, 4315 (1968).
- 68WAL/PAP**
C. Walling and C.G. Papaioannou, *J. Phys. Chem.* **72**, 2260 (1968).
- 68WIL/HAM**
J.M. Williams and W.H. Hamill, "Ionization Potentials of Molecules and Free Radicals and Appearance Potentials by Electron Impact in the Mass Spectrometer," *J. Chem. Phys.* **49**, 4467 (1968).

- 69BEN/CRU**
S.W. Benson, F.R. Cruickshank, D.M. Golden, G.R. Haugen, H.E. O'Neal, A.S. Rogers, R. Shaw and R. Walsh, "Additivity Rules for the Estimation of Thermochemical Properties," *Chem. Rev.* **69**, 279 (1969).
- 69BER/CHU**
J. Berkowitz, W.A. Chupka and T.A. Walter, "Photoionization of HCN: The Electron Affinity and Heat of Formation of CN," *J. Chem. Phys.* **50**, 1497 (1969).
- 69BOG/GRI**
G.M. Bogolyubov, N.N. Grishin and A.A. Petrov, "Organic Derivatives of Group V and Group VI Elements. VIII. Mass Spectra of Phosphines and Diphosphines," *Zh. Obs. Khim.* **39**, 1808 (1969).
- 69BRI**
C.E. Brion, "Negative Ion Formation in the Hexafluorides of Sulphur, Selenium, and Tellurium," *Int. J. Mass Spectrom. Ion Phys.* **3**, 197 (1969).
- 69FLE/WHI**
G.D. Flesch, R.M. White and H.J. Svec, "The Positive and Negative Ion Mass Spectra of Chromyl Chloride and Chromyl Fluoride," *Int. J. Mass Spectrom. Ion Phys.* **3**, 339 (1969).
- 69GOL/BEN**
D.M. Golden and S.W. Benson, *Chem. Rev.* **69**, 125 (1969).
- 69HAL/PLA**
M. Halmann and I. Platzner, "Negative Ions Produced by Electron Capture in Phosphine," *J. Phys. Chem.* **73**, 4376 (1969).
- 69LIF/GRA**
C. Lifshitz and R. Grajower, "Dissociative Electron Capture and Dissociative Ionization in Perfluoropropane," *Int. J. Mass Spectrom. Ion Phys.* **3**, 211 (1969).
- 69MAC/THY**
K.A.G. MacNeil and J.C.J. Thynne, "Ionization and Dissociation of Hexafluoroethane, and of 1,1,1-Trifluoroethane and Fluoroform, by Electron Impact," *Int. J. Mass Spectrom. Ion Phys.* **2**, 1 (1969).
- 69MAC/THY2**
K.A.G. MacNeil and J.C.J. Thynne, "The Deconvolution of Negation Ion Data," *Int. J. Mass Spectrom. Ion Phys.* **3**, 35 (1969).
- 69MIC**
J. Michl, "Electronic Spectrum of Fluoranthene," *J. Mol. Spectrosc.* **30**, 66 (1969).
- 69OKA/MEL**
H. Okabe and A. Mele, *J. Chem. Phys.* **51**, 2100 (1969).
- 69PAG/GOO**
F.M. Page and G.C. Goode, "Negative Ions and the Magnetron," Wiley, NY 1969.
- 69PAN/ZER**
A.V. Pankratov, Z.N. Zhercheninov, V.I. Chesnokov and N.N. Zhadanaova, *Russ. J. Phys. Chem.* **43**, 212 (1969).
- 69SEA/DZI**
S.K. Searles, I. Dzidic and P. Kebarle, "Proton Affinities of the Alkali Hydroxides," *J. Am. Chem. Soc.* **91**, 281 (1969).
- 69STE/WEN**
J.C. Steelhammer and W.E. Wentworth, "Correlation of Electron Beam and Thermal Electron Attachment Studies for Some Chloro, Bromo, Iodo Aromatic Compounds," *J. Chem. Phys.* **51**, 1802 (1969).
- 69STO/COM**
J.A.D. Stockdale, R.N. Compton, G.S. Hurst and P.W. Reinhardt, "Collisions of Monoenergetic Electrons with NO₂: Possible Lower Limits to the Electron Affinities of O₂ and NO," *J. Chem. Phys.* **50**, 2176 (1969).
- 69STU/WES**
D.R. Stull, E.F. Westrum, Jr. and G.C. Sinke, "The Chemical Thermodynamics of Organic Compounds," (John Wiley & Sons, New York, 1969).
- 69TSU/YOK**
S. Tsuda, A. Yokohata and M. Kawai, "Measurement of Negative Ions Formed by Electron Impact. II. The Ionization Efficiency Curves of Negative Nitro, Oxygen Atoms, and Nitromethylene," *Bull. Chem. Soc. Japan* **42**, 614 (1969).
- 69TSU/YOK2**
S. Tsuda, A. Yokohata and M. Kawai, "Measurement of Negative Ions Formed by Electron Impact. III. The Ionization Efficiency Curves of Negative Ions of M/E 26 and 42 From Nitroalkanes," *Bull. Chem. Soc. Japan* **42**, 1515 (1969).
- 69VOG**
D. Vogt, "Über die Energieanhangigkeit und den Mechanismus von Reaktionen bei Stossen Langsamer Negativer Ionen auf Moleküle," *Int. J. Mass Spectrom. Ion Phys.* **3**, 81 (1969).
- 69WAR**
P. Warneck, "Photodetachment of NO₂⁻," *Chem. Phys. Lett.* **3**, 532 (1969).
- 69WEN/RIS**
W.E. Wentworth and W. Ristau, "Thermal Electron Attachment Involving a Change in Molecular Geometry," *J. Phys. Chem.* **73**, 2126 (1969).
- 70ADA/BOH**
N. Adams, D.K. Bohme, D.B. Dukin, D. Fehsenfeld and E.E. Ferguson, "Flowing Afterglow Studies of Formation and Reactions of Cluster Ions of O₂⁺, O₂⁻, and O⁻," *J. Chem. Phys.* **52**, 3133 (1970).
- 70ARS/KEB**
M. Arshadi and P. Kebarle, "Hydration of OH⁻ and O₂⁻ in the Gas Phase. Comparative Solvation of OH⁻ by Water and the Hydrogen Halides. Effect of Acidity," *J. Phys. Chem.* **74**, 1483 (1970).
- 70ARS/YAM**
M. Arshadi, R. Yamdagni and P. Kebarle, "Hydration of Halide Negative Ions in the Gas Phase. II. Comparison of Hydration Energies for the Alkali Positive and Halide Negative Ions," *J. Phys. Chem.* **74**, 1475 (1970).
- 70BAI/MAH**
T.L. Bailey and P. Mahadevan, "Electron Transfer and Detachment in Collisions of Low Energy Negative Ions with O₂," *J. Chem. Phys.* **52**, 179 (1970).
- 70BEN/O'N**
S.W. Benson and H.E. O'Neal, "Kinetic Data on Gas Phase Unimolecular Reactions," *NSRDS-NBS* **21**, (1970).
- 70CHA/CHR**
E.L. Chaney, L.G. Christophorou, P.M. Collins and J.C. Carter, "Electron Attachment in the Field of the Ground and Excited States of the Azulene Molecule," *J. Chem. Phys.* **52**, 4413 (1970).
- 70COL/CHR**
P.M. Collins, L.G. Christophorou, E.L. Chaney and J.G. Carter, "Energy Dependence of the Electron Attachment Cross Section and the Transient Negative Ion Lifetime for p-Benzoquinone and 1,4-Naphthoquinone," *Chem. Phys. Lett.* **4**, 646 (1970).
- 70COL/LOC**
J.E. Collin and R. Loch, "Positive and Negative Ion Formation in Ketene by Electron Impact," *Int. J. Mass Spectrom. Ion Phys.* **3**, 465 (1970).
- 70DEP/GIA**
M. De Paz, A.G. Giardini, L. Friedman, "Tandem-Mass Spectrometer Study of Solvated Derivatives of OD⁻. Total Hydration Energy of the Proton," *J. Chem. Phys.* **52**, 687 (1970).
- 70EME/KHO**
A.M. Emel'yanov, Y.S. Khodeev and L.N. Gorokhov, "Determination of the Ionization Potentials of Atomic Uranium by an Electron Impact Method. I. Measurement of the First Ionization Potential of Uranium," *Teplofiz. Vys. Temp.* **8**, 296 (1970).
- 70FEL**
D. Feldman, "Photoablosung von Elektronen bei einigen Stabilen Negativen Ionen," *Z. Naturfor.* **25A**, 621 (1970).

70FIN/GAR

A. Finch and P.J. Gardner, "Progress in Boron Chemistry," Vol. 3 (ed. R. Brotherton and H. Steinberg, Pergamon, New York, 1970).

70FUR/GOL

S. Furayama, D.M. Golden and S.W. Benson, *Int. J. Chem. Kinet.* 2, 93 (1970).

70HAR/THY

P.W. Harland and J.C.J. Thynne, "Positive and Negative Ion Formation in Hexafluoroacetone by Electron Impact," *J. Phys. Chem.* 74, 52 (1970).

70HEH/DIT

W.J. Hehre, R. Ditchfield, L. Radom and J.A. Pople, *J. Am. Chem. Soc.* 92, 4796 (1970).

70JEN

D.E. Jensen, "Electron Attachment and Compound Formation in Flames. II. Mass Spectrometry of Boron-Containing Flames," *J. Chem. Phys.* 52, 3305 (1970).

70JEN/MIL

D.E. Jensen and W.J. Miller, "Electron Attachment and Compound Formation in Flames. III. Negative Ion and Compound Formation in Flames Containing Tungsten and Potassium," *J. Chem. Phys.* 53, 3287 (1970).

70KLO/PAS

E. Kloster-Jensen, C. Pascual and J. Vogt, *Helv. Chim. Acta* 53, 2109 (1970).

70LAC/HER

K. Lacmann and D.R. Herschbach, "Collisional Excitation and Ionization of K Atoms by Diatomic Molecules: Role of Ion-pair States," *Chem. Phys. Lett.* 6, 106 (1970).

70LIF/HUG

C. Lifshitz, B.M. Hughes and T.O. Tiernan, "Electron Affinities for Endothermic Negative Ion Charge Transfer Reactions. NO₂ and SF₆," *Chem. Phys. Lett.* 7, 469 (1970).

70LIF/PEE

C. Lifshitz, A.M. Peers, R. Grajower and M. Weiss, "Breakdown Curves for Polyatomic Negative Ions," *J. Chem. Phys.* 53, 4605 (1970).

70LOC/MOM

R. Loch and J. Momigny, "Mass Spectrometric Determination of the Electron Affinities of Radicals," *Chem. Phys. Lett.* 6, 273 (1970).

70LON/FIN

F.R. Longo, J.D. Finarelli, E. Schmalzbach and A.D. Adler, *J. Phys. Chem.* 74, 3297 (1970).

70MAC/THY

K.A.G. MacNeil and J.C.J. Thynne, "The Formation of Negative Ions by Electron Impact on Silicon Tetrafluoride and Carbon Tetrafluoride," *Int. J. Mass Spectrom. Ion Phys.* 3, 455 (1970).

70MOO

C.E. Moore, "Ionization Potentials and Ionization Limits Derived from the Analyses of Optical Spectra," *Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.)* 34, (1970).

70OH/HUB

P.A.G. O'Hare, W.N. Hubbard, O. Glemser and J. Wegener, *J. Chem. Thermodyn.* 2, 71 (1970).

70OKA

H. Okabe, *J. Chem. Phys.* 53, 3507 (1970).

70SOL/GOL

R.K. Solly, D.M. Golden and S.W. Benson, "Kinetics of the Gas Phase Reaction of Acetone with Iodine: Heat of formation of the Acetyl Radical," *Int. J. Chem. Kinet.* 2, 11 (1970).

70THY/MAC

J.C.J. Thynne and K.A.G. MacNeil, "Ionisation and Dissociation of Carbonyl Fluoride and Trifluoromethyl Hypofluorite by Electron Impact," *Int. J. Mass Spectrom. Ion Phys.* 5, 95 (1970).

70THY/MAC2

J.C.J. Thynne and K.A.G. MacNeil, "Ionisation of Tetrafluoroethylene by Electron Impact," *Int. J. Mass Spectrom. Ion Phys.* 5, 329 (1970).

70VOG/HAU

D. Vogt, B. Hauffe and H. Neuert, "Ladungsaustausch-Reaktionen Einiger Negativer Ionen mit O₂ und die Elektronenaffinität des O₂," *Z. Phys.* 232, 439 (1970).

71ALL/WUE

N.L. Allinger and M.T. Wuesthoff, *J. Org. Chem.* 36, 2051 (1971).

71ASTM

"Physical Constants of Hydrocarbons C₁ to C₁₀," ASTM Data Series for Testing and Materials, TM DS 4A (Philadelphia, 1971).

71BEL/PLA

T.N. Bell and A.E. Platt, "The Reactions of CF₃ and CD₃ Radicals with Boron Trimethyl," *Int. J. Chem. Kinet.* 3, 307 (1971).

71BER/CHU

J. Berkowitz, W.A. Chupka and D. Gutman, "Electron Affinities of O₂, O₃, NO, NO₂, NO₃ by Endothermic Charge Transfer," *J. Chem. Phys.* 55, 2733 (1971).

71BOH/LEE

D.K. Bohme, E. Lee-Ruff and L.B. Young, "A Standard Acidity Scale. The pK_a of Alcohols in the Gas Phase," *J. Am. Chem. Soc.* 93, 4608 (1971).

71BOH/YOU

D.K. Bohme and L.B. Young, "Electron Affinities from Thermal Proton Transfer Reactions: C₆H₅ and C₆H₅CH₂," *Can. J. Chem.* 49, 2918 (1971).

71BOY/SAN

R.H. Boyd, S.N. Sanwal, S. Shary-Tehrany and D. McNally, *J. Phys. Chem.* 75, 1264 (1971).

71BRA/BLA

J.I. Brauman and L.K. Blair, "Gas Phase Acidities of Amines," *J. Am. Chem. Soc.* 93, 3911 (1971).

71CEL/BEN

R.J. Celotta, R.A. Bennett, J.L. Hall, J. Levine and M.W. Siegel, "Electron Affinity of O₂ by Laser Photodetachment," *Bull. Am. Phys. Soc.* 16, 212 (1971).

71CHA

P.J. Chantry, "Doppler Broadening in Beam Experiments," *J. Chem. Phys.* 55, 2746 (1971).

71CHU/BER

W.A. Chupka, J. Berkowitz and D. Gutman, "Electron Affinities of Halogen Diatomic Molecules as Determined by Endoergic Charge Exchange," *J. Chem. Phys.* 55, 2724 (1971).

71DEC/BAP

J.J. DeCorpo, D.A. Bafus and J.L. Franklin, "Correlation of Excess Energies of Dissociative Electron Attachment Processes with the Translational Energies of Their Products," *J. Chem. Phys.* 54, 1592 (1971).

71DEC/FRA

J.J. DeCorpo and J.L. Franklin, "Electron Affinities of the Halogen Molecules by Dissociative Electron Attachment," *J. Chem. Phys.* 54, 1885 (1971).

71FEH

F.C. Fehsenfeld, "Ion Chemistry of SF₆," *J. Chem. Phys.* 54, 438 (1971).

71FEL

D. Feldman, "Photoablosung von Elektronen bei Si⁻ und NH₂⁻," *Z. Naturfor.* 26A, 1100 (1971).

71HAR/THY

P.W. Harland and J.C.J. Thynne, "Autodetachment Lifetimes, Attachment Cross Sections, and Negative Ions Formed by Sulfur Hexafluoride and Sulfur Tetrafluoride," *J. Phys. Chem.* 75, 3517 (1971).

71JANAF

D.R. Stull and H. Prophet, "JANAF Thermochemical Tables," NSRDS-NBS 37, U.S. Gov't. Print. Off., Washington, DC, 1971.

Check and molecules



- 71KIN/GOL**
K.D. King, D.M. Golden and S.W. Benson, "Thermochemistry of the Equilibrium $\text{CH}_3\text{COCH}_3 + \text{Br}_2 = \text{CH}_3\text{COCH}_2\text{Br} + \text{HBr}$," *J. Chem. Therm.* **3**, 129 (1971).
- 71MOU/ATE**
A.M.C. Moutinho, J.A. Aten and J. Los, "Temperature Dependence of the Total Cross Section for Chemi-Ionization in Alkali Halide-Halogen Collisions," *Physica* **53**, 471 (1971).
- 71NAL/COM**
S.J. Nalley and R.N. Compton, "Collisional Ionization of Cesium by Oxygen: the Electron Affinity of O_2 ," *Chem. Phys. Lett.* **9**, 529 (1971).
- 71NUT/LAU**
R.L. Nuttall, A.H. Laufer and M.V. Kilday, *J. Chem. Thermodyn.* **3**, 107 (1971).
- 71PAR**
D.A. Parkes, "Electron Attachment and Negative Ion-Molecule Reactions in Pure O_2 ," *Trans. Farad. Soc.* **97**, 711 (1971).
- 71PAY/YAM**
J.D. Payzant, R. Yamdagni and P. Kebarle, "Hydration of CN^- , NO_2^- , NO_3^- , and HO^- in the Gas Phase," *Can. J. Chem.* **49**, 3308 (1971).
- 71POT/PRI**
A.W. Potts and W.C. Price, "Photoelectron Spectra of the Halogens and Mixed Halides, ICl and IBr ," *Trans. Faraday Soc.* **67**, 1242 (1971).
- 71RAP/WES**
N.J. Rapport, E.F. Westrum, Jr. and J.T.J. Andrews, *J. Am. Chem. Soc.* **93**, 4363 (1971).
- 71RHY/DIL**
T.C. Rhyne and J.G. Dillard, "Reactions of Gaseous Inorganic Negative Ions: III. SF_6^- with POF_3 and PSF_3 ," *Int. J. Mass Spectrom. Ion Phys.* **7**, 371 (1971).
- 71SEL**
P. Sellers, *Acta Chem. Scand.* **25**, 2194 (1971).
- 71SMY/MCI**
K.C. Smyth, R.T. McIver, J.I. Brauman and R.W. Wallace, "Photodetachment of Negative Ions Using a Continuously Tunable Laser and an ICR Spectrometer," *J. Chem. Phys.* **54**, 2758 (1971).
- 71SRI/UY**
R.D. Srivastava, O.M. Uy and M. Farber, "Effusion Mass Spectrometric Study of Thermodynamic Properties of BO^- and BO_2^- ," *Trans. Farad. Soc.* **67**, 2941 (1971).
- 71THY/MAC**
J.C.J. Thynne and K.A.G. MacNiel, "Negative Ion Formation by Ethylene and 1,1-difluoroethylene," *J. Phys. Chem.* **75**, 2584 (1971).
- 71TIE/HUG**
T.O. Tiernan, B.M. Hughes and C. Lifschitz, "Electron Affinities from Endothermic Negative Ion Charge Transfer Reactions. II. O_2 ," *J. Chem. Phys.* **55**, 5692 (1971).
- 71TSU/YOK**
S. Tsuda, A. Yokohata and T. Umaba, "Measurement of Negative Ions formed by Electron Impact. VIII. Ionization Efficiency Curves of Negative Ions from Methyl and Ethyl Cyanides," *Bull. Chem. Soc. Jpn.* **44**, 1486 (1971).
- 71WEN/CHE**
W.E. Wentworth, E. Chen and R. Freeman, "Thermal Electron Attachment to N_2O ," *J. Chem. Phys.* **55**, 2075 (1971).
- 71WIL/ZWO**
R.C. Wilhoit and B.J. Zwolinski, "Handbook of Vapor Pressures and Heats of Vaporization of Hydrocarbons and Related Compounds" (Thermodynamics Research Center, College Station, Texas, 1971).
- 71WON/VOR**
S.F. Wong, T.V. Vorburger and S.V. Woo, "Photodetachment of O_3^- ," *Bull. Am. Phys. Soc.* **16**, 213 (1971).
- 71YAM/KEB**
R. Yamdagni and P. Kebarle, "Hydrogen Bonding Energies to Negative Ions from Gas Phase Measurements of Ionic Equilibria," *J. Am. Chem. Soc.* **93**, 7139 (1971).
- 72BAE**
A.P.M. Baeda, "The Adiabatic Electron Affinities of Cl_2 , Br_2 , I_2 , IBr , NO_2 , and O_2 ," *Physica* **59**, 541 (1972).
- 72BOH/LEE**
D.K. Bohme, E. Lee-Ruff and L.B. Young, "Acidity Order of Selected Bronsted Acids in the Gas Phase at 300K," *J. Am. Chem. Soc.* **94**, 5153 (1972).
- 72BRI/OLS**
C.E. Brion and L.A.R. Olsen, "Negative Ion Formation in Tetracyanoethylene," *Int. J. Mass Spectrom. Ion Phys.* **9**, 413 (1972).
- 72BUR**
J.A. Burt, "Photodetachment Cross Sections for CO_3^- and Its First Hydrate," *J. Chem. Phys.* **57**, 4649 (1972).
- 72BUR2**
J.A. Burt, "Measurement of the Photodetachment Cross Section for O_4^- at High Pressure," *J. Geophys. Res.* **77**, 6280 (1972).
- 72CEL/BEN**
R.J. Celotta, R.A. Bennett, J.L. Hall, M.W. Siegel and J. Levine, "Molecular Photodetachment Spectrometry. II. The Electron Affinity of O_2 and the Structure of O_2^- ," *Phys. Rev.* **6A**, 631 (1972).
- 72CEN**
R.E. Center, "Ion-Molecule Experiments Involving Negative Ions of Tungsten and Rhenium Oxides," *J. Chem. Phys.* **56**, 371 (1972).
- 72CON/COL**
G. Conde-Caprace and J.E. Collin, "Electron Impact Induced Fragmentation of 1,3,6-Dioxathiocane," *Org. Mass Spectrom.* **6**, 341 (1972).
- 72DID/FRA**
A. DiDomenico and J.L. Franklin, "Negative Ions in the Mass Spectrum of Nitromethane," *Int. J. Mass Spectrom. Ion Phys.* **9**, 171 (1972).
- 72DID/HAR**
A. DiDomenico, P.W. Harland and J.L. Franklin, "Negative Ion Formation and Negative Ion-Molecule Reactions in Cyclopentadiene," *J. Chem. Phys.* **56**, 5299 (1972).
- 72DIX/HOL**
D.A. Dixon, D. Holtz and J.L. Beauchamp, "Acidity, Basicity, and Gas-Phase Ion Chemistry of Hydrogen Selenide by ICR Spectroscopy," *Inorg. Chem.* **11**, 960 (1972).
- 72DOM**
E.S. Domalski, *J. Phys. Chem. Ref. Data* **1**, 221 (1972).
- 72DON/LIT**
R.J. Donovan, D.J. Little, J. Konstantatos, "Vacuum Ultraviolet Spectra of Transient Molecules and Radicals," *J. Chem. Soc. Farad. Trans. II* **68**, 1812 (1972).
- 72DUN/FEH**
D.B. Dunkin, F.C. Fehsenfeld and E.E. Ferguson, "Thermal Energy Rate Constants for the Reactions $\text{NO}_2^- + \text{Cl}_2 \rightarrow \text{Cl}_2^-$, $\text{Cl}_2^- + \text{NO}_2 \rightarrow \text{Cl}^-$, $\text{HS}^- + \text{NO}_2 \rightarrow \text{NO}_2^-$, $\text{HS}^- + \text{Cl}_2 \rightarrow \text{Cl}_2^-$, and $\text{S}^- + \text{NO}_2 \rightarrow \text{NO}_2^-$," *Chem. Phys. Lett.* **15**, 257 (1972).
- 72FER/DUN**
E.E. Ferguson, D.B. Dunkin and F.C. Fehsenfeld, "Reactions of NO_2^- and NO_3^- with HCl and HBr ," *J. Chem. Phys.* **57**, 1459 (1972).
- 72GAF**
"M-PYROL: N-Methylpyrrolidone" (GAF Corporation, New York, 1972).
- 72GEI/RAU**
G. Geiseler and H.J. Rauh, *Z. Phys. Chem. (Leipzig)* **249**, 376 (1972).
- 72GOR**
A.S. Gordon, *Int. J. Chem. Kinet.* **4**, 541 (1972).
- 72GRE**
A. Greenberg, *J. Chem. Ed.* **49**, 575 (1972).

- 72GRO**
M.L. Gross, "An Ion Cyclotron Resonance Study of the Structure of $C_3H_6^+$ and the Mechanism of Its Reaction with Ammonia," *J. Am. Chem. Soc.* **94**, 3744 (1972).
- 72HAR/CRA**
P.W. Harland, S. Craddock and J.C.J. Thynne, "Positive- and Negative-Ion Formation Due to the Electron Bombardment of Germanium Tetrafluoride," *Int. J. Mass Spectrom. Ion Phys.* **10**, 169 (1972).
- 72HAR/THY**
P.W. Harland and J.C.J. Thynne, "Dissociative Electron Capture in Perfluoropropylene and Perfluoropropane," *Int. J. Mass Spectrom. Ion Phys.* **9**, 253 (1972).
- 72HAR/THY2**
P.W. Harland and J.C.J. Thynne, "Ionisation of Perfluorocyclobutane by Electron Impact," *Int. J. Mass Spectrom. Ion Phys.* **10**, 11 (1972).
- 72HEH/RAD**
W.J. Hehre, L.A. Radom and J.A. Pople, *J. Am. Chem. Soc.* **94**, 1496 (1972).
- 72JOH/MAL**
G.K. Johnson, J.G. Malm and W.N. Hubbard, *J. Chem. Thermodyn.* **4**, 879 (1972).
- 72KOZ/TIM**
M.P. Kozina, L.P. Timofeeva, S.M. Pinenova, V.A. Aleshna, N.A. Belikova, A.A. Bobyleva and A.F. Plate, *Russ. J. Phys. Chem.* **46**, 1689 (1972).
- 72LAU/OKA**
A.H. Laufer and H. Okabe, *J. Phys. Chem.* **76**, 3504 (1972).
- 72LIF/GRA**
C. Lifshitz and R. Grajower, "Dissociative Electron Capture and Dissociative Ionization in Perfluorocyclobutane," *Int. J. Mass Spectrom. Ion Phys.* **10**, 25 (1972).
- 72MAC/THY**
K.A.G. MacNeil and J.C.J. Thynne, "Negative Ion Formation at Low Electron Energies by Hexafluorodimethyl Peroxide," *Int. J. Mass Spectrom. Ion Phys.* **9**, 135 (1972).
- 72MCF/DUN**
M. McFarland, D.B. Dunkin, F.C. Fehsenfeld, A.L. Schmeltekopf and E.E. Ferguson, "Collisional Detachment Studies of NO^+ ," *J. Chem. Phys.* **56**, 2358 (1972).
- 72MIL**
E.S. Miller, "Electron Attachment and Compound Formation in Flames. V. Negative Ion Formation in Flames Containing Chromium and Potassium," *J. Chem. Phys.* **57**, 2354 (1972).
- 72PAG**
Page, F.M., "Experimental Determination of the Electron Affinities of Inorganic Radicals," *Adv. Chem. Ser.* **36**, 68 (1972).
- 72PAR/SUG**
D.A. Parkes and T.M. Sugden, "Electron Attachment and Detachment in Nitric Oxide," *J. Chem. Soc. Faraday II* **68**, 600 (1972).
- 72PIL**
G. Pilcher MTP Review of Science, Series 1, Vol. 10 (ed. H.A. Skinner, Butterworths, London, 1972).
- 72POT/SOR**
V.K. Potapov and V.V. Sorokin, "Kinetic Energies of Products of Dissociative Photoionization of Molecules. 1. Aliphatic Ketones and Alcohols," *Khim. Vys. Energ.* **6**, 387 (1972).
- 72SHA**
R. Shaw, *Int. J. Chem. Kinet.* **5**, 261 (1972).
- 72SHI/YAM**
T. Shiga, H. Yamaoka, K. Arakawa and T. Suguira, "A Negative Ion-Molecule Reaction in Nitroethylene," *Bull. Chem. Soc. Jpn.* **45**, 2065 (1972).
- 72SIE/CEL**
M.W. Siegel, R.J. Celotta, F.L. Hall, J. Levine and R.A. Bennett, "Molecular Photodetachment Spectroscopy. I. The Electron Affinity of Nitric Oxide and the Molecular Constants of NO^- ," *Phys. Rev. A* **6**, 607 (1972).
- 72SMY/BRA**
K.C. Smyth and J.I. Brauman, "Photodetachment of Electrons from Phosphide Ion; the Electron Affinity of PH_2^- ," *J. Chem. Phys.* **56**, 1132 (1972).
- 72SMY/BRA2**
K.C. Smyth and J.I. Brauman, "Photodetachment of Electrons from Amide and Arsenide Ions: the Electron Affinities of NH_2^- and AsH_2^- ," *J. Chem. Phys.* **56**, 4620 (1972).
- 72SMY/BRA3**
K.C. Smyth and J.I. Brauman, "Photodetachment of an Electron from Selenide Ion; The Electron Affinity and Spin-Orbit Coupling Constant for SeH^- ," *J. Chem. Phys.* **56**, 5993 (1972).
- 72SRI/UY**
R.D. Srivastava, O.M. Uy and M. Farber, "Effusion Mass Spectrometric Study of the Thermodynamic Properties of AlO^- and AlO_2^- ," *J. Chem. Soc. Faraday Trans. II* **1**, 1388 (1972).
- 72STO/NEL**
J.A.D. Stockdale, D.R. Nelson, F.J. Davis and R.N. Compton, "Studies of Electron Impact Excitation, Negative Ion Formation, and Negative Ion-Molecule Reactions in Boron Trifluoride and Boron Trichloride," *J. Chem. Phys.* **56**, 3336 (1972).
- 72THY**
J.C.J. Thynne, *Dyn. Mass Spectrom.* **3**, 67 (1972).
- 72TIR/ALL**
M.T. Tribble and N.L. Allinger, *Tetrahedron* **28**, 2147 (1972).
- 72WAL**
L.C. Walker, *J. Chem. Thermodyn.* **4**, 219 (1972).
- 72YAM/KEB**
R. Yamdagni and P. Kebarle, "Solvation of Negative Ions by Protic and Aprotic Solvents. Gas Phase Solvation of Halide Ions by Acetonitrile and Water Molecules," *J. Am. Chem. Soc.* **94**, 2940 (1972).
- 73ALF/GOL**
A.B. Alfassi, D.M. Golden and S.W. Benson, *J. Chem. Thermodyn.* **5**, 511 (1973).
- 73ARS/SHA**
M.R. Arshadi and M. Shabrang, *J. Chem. Soc. Perkins II*, 1732 (1973).
- 73AUE/HUB**
D.J. Auerbach, M.M. Hubers, A.P.M. Baeda and J. Los, "Chemi-Ionization in Alkali-Heteronuclear Halogen Collisions: Role of Excited Molecular Ion States," *Chem. Phys.* **2**, 107 (1973).
- 73BAE/AUE**
A.P.M. Baeda, J. Auerbach, and D.J. Los, "Fragmentation of Negative Ions Formed in Collisions of Alkali Atoms and Halogen Molecules," *Physica* **64**, 134 (1973).
- 73BAR**
P. Barbieri, *Inform. Sci. Tech, Commis. Energ. At.* **55**, 180 (1973). CA: 79:58336m (1973).
- 73BAT/MIL**
L. Batt and R.T. Milne, *Int. J. Chem. Kinet.* **5**, 1067 (1973).
- 73BEN/MAR**
S.L. Bennett, J.L. Margrave, J.L. Franklin and J.E. Hudson, "High Temperature Negative Ions: Electron Impact of As_4 Vapor," *J. Chem. Phys.* **59**, 5814 (1973).
- 73BIL/CHO**
W.E. Billups, W.Y. Chong, K.H. Leavell, E.S. Lewis, J.L. Margrave, R.L. Sass, J.J. Shieh, P.G. Werness and J.L. Wood, *J. Am. Chem. Soc.* **95**, 7878 (1973).
- 73BLA/ISO**
L.K. Blair, P.C. Isolani and J.M. Riveros, "Formation, Reactivity, and Relative Stability of Clustered Alkoxide Ions by ICR Spectroscopy," *J. Am. Chem. Soc.* **95**, 1057 (1973).

73COM/COO

R.N. Compton and C.D. Cooper, "Molecular Electron Affinities from Collisional Ionization of Cesium. II. SF₆ and TeF₆," *J. Chem. Phys.* **59**, 4140 (1973).

73COM/COO2

R.N. Compton, R.D. Cooper, W.T. Divver and P.W. Reinhardt, "Molecular Electron Affinities from Collisional Ionization of Cesium: SF₆," *Bull. Am. Phys. Soc.* **18**, 810 (1973).

73COO/COM

C.D. Cooper and R.N. Compton, "Electron Attachment and Cesium Collisional Ionization Studies of Tetrafluorosuccinic and Hexafluoroglutaric Anhydrides: Molecular Electron Affinities," *J. Chem. Phys.* **59**, 3550 (1973).

73COW/JOH

S.A. Cowling and R.A.W. Johnstone, *J. Electron Spectrosc. Relat. Phenom.* **2**, 161 (1973).

73EGG/COC

K.W. Egger and A.T. Cocks, "The Chemistry of the Carbon-Halogen Bond," S. Patai, Ed., Wiley, NY, 1973, Ch. 10.

73ENG/AND

E.M. Engler, J.D. Andose and P. von R. Schleyer, *J. Am. Chem. Soc.* **95**, 8003 (1973).

73GAR/REE

W.R.S. Garton, E.M. Reeves, F.S. Tomkins and B. Ercoli, "Rydberg Series and Autoionization Resonances in the Y I Absorption Spectrum," *Proc. R. Soc. Lond.* **A333**, 17 (1973).

73GOL/KOR

I.V. Gol'denfel'd, I.Z. Korostyshevskii, B.G. Mischanchuk and V.A. Pokrovskii, "Determination of Ionization Potentials of Atoms and Molecules Using a Field Mass Spectrometer Equipped with an Energy Analyzer," *Dokl. Akad. Nauk SSSR*, **213**, 626 (1973).

73HAA/MCD

J.C. Haartz and D.H. McDaniel, "Fluoride Ion Affinity of Some Lewis Acids," *J. Am. Chem. Soc.* **95**, 8562 (1973).

73HAR/FRA

P.W. Harland, J.L. Franklin and D.E. Carter, "Use of Translational Energy Measurements in the Evaluation of the Energetics for Dissociative Attachment Processes," *J. Chem. Phys.* **58**, 1430 (1973).

73HAR/THY

P.W. Harland and J.C.J. Thynne, "Comparison of Negative Ion Formation by the Hexafluorides of Sulphur, Selenium, Tellurium, and Tungsten," *Inorg. Nucl. Chem. Lett.* **9**, 265 (1973).

73HAR/THY2

P.W. Harland and J.C.J. Thynne, "Negative Ion Formation by Perfluoro-n-butane as the Result of Low Energy Electron Impact," *Int. J. Mass Spectrom. Ion Phys.* **11**, 445 (1973).

73HUG/LIF

B.M. Hughes, C. Lifshitz and T.O. Tiernan, "Electron Affinities from Endothermic Negative-ion Charge-Transfer Reactions. III. NO, NO₂, S₂, CS₂, Cl₂, Br₂, I₂, and C₂H₄," *J. Chem. Phys.* **59**, 3162 (1973).

73LAR/JOH

J.W. Larson, G.K. Johnson, P.A.G. O'Hare and O. Glemser, *J. Chem. Thermodyn.* **5**, 689 (1973).

73LAT/RAD

W.A. Lathan, L. Radom, P.C. Hariharan, W.J. Hehre and J.A. Pople, *Top. Curr. Chem.* **40**, 1 (1973).

73LEF/JAC

C.B. Leffert, W.M. Jackson and E.W. Rothe, "Measurement of the Electron Affinity of NO₂," *J. Chem. Phys.* **58**, 5801 (1973).

73LIF/TIE

C. Lifshitz, T.O. Tiernan and B.M. Hughes, "Electron Affinities from Endothermic Negative-Ion Charge Transfer Reactions. IV. SF₆, Selected Fluorocarbons, and other Polyatomic Molecules," *J. Chem. Phys.* **59**, 3182 (1973).

73LYO/PAL

L.E. Lyons and L.D. Palmer, "Photodetachment of Electrons from Tetracyanoethylene Negative Ions," *Chem. Phys. Lett.* **21**, 442 (1973).

73MCI/SCO

R.T. McIver, Jr., J.A. Scott and J.M. Riveros, "Effect of Solvation on the Intrinsic Relative Acidity of Methanol and Ethanol," *J. Am. Chem. Soc.* **95**, 2706 (1973).

73MCN/LAC

P.E. McNamee, K. Lacmann and D.R. Herschbach, *Faraday Disc. Chem. Soc.* **55**, 318 (1973).

73NAL/COM

S.J. Nalley, R.N. Compton, H.C. Schweinler and V.E. Anderson, "Molecular Electron Affinities from Collisional Ionization of Cesium. I. NO, NO₂, and N₂O," *J. Chem. Phys.* **59**, 4125 (1973).

73ONA/HOW

T. Onak, J. Howard and C. Brown, "Negative Ion Mass Spectrometry of closo-Carboranes," *J. Chem. Soc. Dalton* **76**, (1973).

73PEP/GAF

V.I. Pepekin, R.G. Gafurov, Yu.A. Lebedev, L.T. Eremenko, E.M. Soyomonyan and A. Ya. Apin, *Bull. Acad. Sci. USSR, Div. Chem. Sci.* **22**, 304 (1973).

73RAB/KAR

J.W. Rabalais, L. Karlsson, L.O. Werme, T. Bergmark and K. Siegbahn, "Analysis of Vibrational Structure and Jahn-Teller Effects in the Electron Spectrum of Ammonia," *J. Chem. Phys.* **58**, 3370 (1973).

73RAD/LAT

L. Radom, W.A. Lathan, W.J. Hehre and J.A. Pople, *J. Am. Chem. Soc.* **95**, 693 (1973).

73RIC/SIE

J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of Electrons from Large Molecular Systems: Cyclopentadienide and Methylcyclopentadienide Ions. An Upper Limit to the Electron Affinities of C₅H₅⁻ and CH₃C₅H₄⁻," *J. Chem. Phys.* **59**, 5068 (1973).

73RIV/BRE

J.M. Riveros, A.C. Breda and L.K. Blair, "Formation and Relative Stability of Chloride Ion Clusters in the Gas Phase by ICR Spectroscopy," *J. Am. Chem. Soc.* **95**, 4066 (1973).

73SEN/FRA

D.K. SenSharma and J.L. Franklin, "Heat of Formation of Free Radicals by Mass Spectrometry," *J. Am. Chem. Soc.* **95**, 6562 (1973).

73THY/HAR

T.C.J. Thynne and P.W. Harland, *Int. J. Mass Spectrom. Ion Phys.* **11**, 399 (1973).

73THY/HAR2

J.C.J. Thynne and P.W. Harland, "Negative Ion Formation by Tungsten Fluoride," *Int. J. Mass Spectrom. Ion Phys.* **11**, 137 (1973).

73WAN/MAR

J.L.-F. Wang, J.L. Margrave and J.L. Franklin, "Interpretation of Dissociative Electron Attachment Processes for Carbon and Silicon Tetrafluorides," *J. Chem. Phys.* **58**, 5417 (1973).

73YAM/PAY

R. Yamdagni, J.D. Payzant and P. Kebarle, "Solvation of Cl⁻ and O₂⁻ with H₂O, CH₃OH, and CH₃CN in the Gas Phase," *Can. J. Chem.* **51**, 2507 (1973).

74BAT/CHR

L. Batt, K. Christie, R.T. Milne and A.J. Summers, *Int. J. Chem. Kinet.* **6**, 877 (1974).

- 74BEA**
J.L. Beauchamp, "Chemical Applications of New Developments in Ion Cyclotron Resonance Spectroscopy," *Adv. Mass Spectrom.* 6, 717 (1974).
- 74BEA/MUE**
P. Beak, D.S. Mueller and J. Lee, *J. Am. Chem. Soc.* 96, 3867 (1974).
- 74BEN/MAR**
S.L. Bennett, J.L. Margrave and J.L. Franklin, "High Temperature Negative Ions. Electron Impact Study of Tetratomic Phosphorous Vapor," *J. Chem. Phys.* 61, 1647 (1974).
- 74BEN/PAB**
S.L. Bennett, S.E. Pabst, J.L. Margrave and J.L. Franklin, "Negative Ion Electron Impact Studies of Titanium Tetrahalides," *Int. J. Mass Spectrom. Ion Phys.* 15, 451 (1974).
- 74BEI/BAK**
D. Betteridge, A.D. Baker, P. Bye, S.K. Hasannudin, N.R. Kemp and M. Thompson, "A Cheap Versatile Ultraviolet Photoelectron Spectrometer," *J. Electron Spectrosc. Rel. Phenom.* 4, 163 (1974).
- 74BLI/MCM**
R.J. Blint, T.B. McMahon and J.L. Beauchamp, "Gas Phase Ion Chemistry of Fluoromethanes by Ion Cyclotron Resonance Spectroscopy. New Techniques for the Determination of Carbonium Ion Stabilities," *J. Am. Chem. Soc.* 96, 1269 (1974).
- 74BOH/MAC**
D.K. Bohme, G.I. MacKay, H.I. Schiff and R.S. Hemsworth, "Equilibrium $\text{OH}^- + \text{C}_2\text{H}_2 = \text{C}_2\text{H}^- + \text{H}_2\text{O}$ and the Determination of $\Delta H_{f298}(\text{C}_2\text{H}^-)$," *J. Chem. Phys.* 61, 2175 (1974).
- 74BUR**
P.D. Burrow, "Temporary Negative Ion Formation in NO and O_2 ," *Chem. Phys. Lett.* 26, 265 (1974).
- 74BUR/HAI**
J. Burgess, I. Haigh and R.D. Peacock, *J. Chem. Soc. Dalton Trans.* 1062 (1974).
- 74CEL/BEN**
R.J. Celotta, R.A. Bennett and J.L. Hall, "Laser Photodetachment Determination of the Electron Affinities of OH , NH_2 , NH , SO_2 , and S_2 ," *J. Chem. Phys.* 60, 1740 (1974).
- 74CHA/PAG**
A.T. Chamberlin, F.M. Page and M.R. Painter, "A Study of the Negative Ions Produced during Surface Ionization on Hot Metal Filaments," *Adv. Mass Spectrom.* 6, 311 (1974).
- 74CHA/ROD**
J. Chao, A.S. Rodgers, R.C. Wilhoit and B.J. Zwolinski, *J. Phys. Chem. Ref. Data* 3, 141 (1974).
- 74CHO/MEN**
K.Y. Choo, G.D. Mendenhall, D.M. Golden and S.W. Benson, "The Pyrolysis of Nitrosoisobutane and the Bond Dissociation Energies of Nitroso Compounds," *Int. J. Chem. Kinet.* 6, 813 (1974).
- 74COM/REI**
R.N. Compton, P.W. Reinhardt and C.D. Cooper, "Mass Spectrometry Utilizing Collisional Ionization of Cesium: Maleic Anhydride and Succinic Anhydride," *J. Chem. Phys.* 60, 2953 (1974).
- 74COO/COM**
C.D. Cooper and R.N. Compton, "Electron Attachment and Collisional Ionization Studies of Tetrafluorosuccinic and Hexafluoroglutaric Anhydrides: Molecular Electron Affinities," *J. Chem. Phys.* 60, 2424 (1974).
- 74DON/HAR**
R.J. Donovan, P.W. Harland, J.H. Knox, J.A. Makowski and J.C.J. Thynne, "Electron Affinity of SF_4 ," *Int. J. Mass Spectrom. Ion Phys.* 13, 464 (1974).
- 74DOU/DAL**
R.C. Dougherty, J. Dalton and J.D. Roberts, " $\text{S}_{\text{N}}2$ Reactions in the Gas Phase: Structure of the Transition State," *Org. Mass Spectrom.* 8, 77 (1974).
- 74DOU**
R.C. Dougherty, " $\text{S}_{\text{N}}2$ Reactions in the Gas Phase. Alkyl Group Structural Effects," *Org. Mass Spectrom.* 8, 85 (1974).
- 74DOU/ROB**
R.C. Dougherty and J.D. Roberts, " $\text{S}_{\text{N}}2$ Reactions in the Gas Phase. Nucleophilicity Effects," *Org. Mass Spectrom.* 8, 81 (1984).
- 74DZI/CAR**
I. Dzidic, D.I. Carroll, R.N. Stillwell and E.C. Horning, "Gas Phase Reactions. Ionization by Proton Transfer to Superoxide Anions," *J. Am. Chem. Soc.* 96, 5258 (1974).
- 74EYL/ATK**
J.R. Eyley and G.H. Atkinson, "Dye Laser-induced Photodetachment of Electrons from SH^- Studied by ICR Spectroscopy," *Chem. Phys. Lett.* 28, 217 (1974).
- 74FEH/FER**
F.C. Fehsenfeld and E.E. Ferguson, "Laboratory Studies of Negative Ion Reactions with Atmospheric Trace Constituents," *J. Chem. Phys.* 61, 3181 (1974).
- 74FRA/WAN**
J.L. Franklin, J.L.-F. Wang, S.L. Bennett, P.W. Harland and J.L. Margrave, "Studies of the Energies of Negative Ions at High Temperatures," *Adv. Mass Spectrom.* 6, 319 (1974).
- 74HAL/KLE**
M. Halmann and Y. Klein, "Positive and Negative Ion Mass Spectra of Phosphorous Compounds," *Adv. Mass Spectrom.* 3, 267 (1974).
- 74HAR/FRA**
P.W. Harland and J.L. Franklin, "Partitioning of Excess Energy in Dissociative Resonance Capture Processes," *J. Chem. Phys.* 61, 1621 (1974).
- 74HAR/RAN**
P.W. Harland, D.W.H. Rankin and J.C.J. Thynne, "Ionisation by Electron Impact of Phosphorus Trifluoride and Difluorocyanophosphine," *Int. J. Mass Spectrom. Ion Phys.* 13, 395 (1974).
- 74HER/PAT**
E. Herbst, T.A. Patterson and W.C. Lineberger, "Laser Photodetachment of NO_2^- ," *J. Chem. Phys.* 61, 1300 (1974).
- 74HOT/PAT**
H. Hotop, T.A. Patterson and W.C. Lineberger, "High Resolution Photodetachment Study of OH^- and OD^- in the Threshold Region 7000-6450 Å," *J. Chem. Phys.* 60, 1806 (1974).
- 74JANAF**
M.W. Chase, J.L. Curnutt, A.T. Hu, H. Prophet, A.N. Syverud and L.C. Walker, "JANAF Thermochemical Tables, 1974 Supplement," *J. Phys. Chem. Ref. Data* 3, 311 (1974).
- 74KIM/SET**
K.C. Kim, D.W. Setser and C.M. Bogan, "HF Infrared Chemiluminescence and Energy Partitioning, and $\text{D}(\text{H}-\text{GeH}_3)$ from the Reaction of F Atoms with Germane," *J. Chem. Phys.* 60, 1837 (1974).
- 74KOP/SCH**
C. Koppel, H. Schwarz and F. Bohlmann, "Elektronenstossinduzierte Fragmentierung von Acetylenverbindungen," *Org. Mass Spectrom.* 9, 324 (1974).
- 74LAT/CUR**
W.A. Lathan, L.A. Curtiss, W.J. Hehre, J.B. Lisle and J.A. Pople, *Prog. Phys. Org. Chem.* 11, 175 (1974).
- 74LEF/TAN**
C.B. Leffert, S.Y. Tang, E.W. Rothe and T.C. Cheng, "Collisional Ionization of Cs with SF_6 ," *J. Chem. Phys.* 61, 4929 (1974).
- 74LIE/GRE**
J.F. Liebman and A. Greenberg, *Biophys. Chem.* 1, 222 (1974).
- 74MCA**
T. McAllister, *Int. J. Mass Spectrom. Ion Phys.* 15, 303 (1974).

- 74MOU/ATE**
A.M.C. Moutinho, J.A. Aten and J. Los, "Chemi-Ionization in Alkali-Methylhalogen Collisions," *Chem. Phys.* **5**, 84 (1974).
- 74OTH/BUN**
J.F.M. Oth, J.-C. Bunzli and Y. de Julien de Zelicourt, *Helv. Chim. Acta* **57**, 2276 (1974).
- 74PAL/KEN**
M.H. Palmer and S.M.F. Kennedy, *J. Chem. Soc. Perkin II* **1893** (1974).
- 74PEA/SCH**
P.K. Pearson, H.F. Schaefer, III, J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Three Isomers of the NO₂⁻ Ion," *J. Am. Chem. Soc.* **96**, 6778 (1974).
- 74PIH/TAS**
K. Pihlaja and E. Taskinen, "Physical Methods in Heterocyclic Chemistry," Vol. 6 (ed. A.R. Katritzky, Academic Press, New York, 1974).
- 74REE/BRA**
K.J. Reed and J.I. Brauman, "Photodetachment of Electrons from Group IVa Binary Hydride Anions: The Electron Affinities of the SiH₃ and GeH₃ Radicals," *J. Chem. Phys.* **61**, 4830 (1974).
- 74RIC/STE**
J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of NO₂⁻: Experimental Evidence for a New Isomer," *Chem. Phys. Lett.* **25**, 318 (1974).
- 74RIC/STE2**
J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of Electrons from Phenoxides and Thiophenoxide," *Chem. Phys. Lett.* **25**, 321 (1974).
- 74RID/BEA**
D.P. Ridge and J.L. Beauchamp, "Chemical Consequences of Strong Hydrogen Bonding in the Reactions of Organic Ions in the Gas Phase. Base Induced Elimination Reactions," *J. Am. Chem. Soc.* **96**, 637 (1974).
- 74ROG/KAN**
D.W. Rogers and S. Kanupong, *J. Phys. Chem.* **78**, 2569 (1974).
- 74SAB/CHA**
R. Sabbah, R. Chastel and M. Laffitte, *Thermochim. Acta* **10**, 353 (1974).
- 74SCO**
D.W. Scott, "Chemical Thermodynamic Properties of Hydrocarbons and Related Substances: Properties of Alkane Hydrocarbons C₁ through C₁₀," APIRP62 Report 39, (U.S. Dept. of Interior, Bureau of Mines, 1974).
- 74SRJ/UJY**
R.D. Srivastava, O.M. Uy and M. Farber, "Experimental Determination of Heats of Formation of Negative Ions and Electron Affinities of Several Boron and Aluminum Fluorides," *J. Chem. Soc. Faraday Trans. I* **70**, 1033 (1974).
- 74WAN/MAR**
J.L.-F. Wang, J.L. Margrave and J.L. Franklin, "Enthalpy of Formation of Germanium Trifluoride," *J. Chem. Phys.* **60**, 2158 (1974).
- 74WAN/MAR2**
J.L.-F. Wang, J.L. Margrave and J.L. Franklin, "Interpretation of Dissociative-Electron Attachment Processes for Silicon Tetrachloride," *J. Chem. Phys.* **61**, 1357 (1974).
- 74WYA/HOL**
R.H. Wyatt, D. Holtz, T.B. McMahon and J.L. Beauchamp, "Acidity, Basicity, and Ion-Molecule Reactions of Arsine in the Gas Phase by ICR Spectroscopy," *Inorg. Chem.* **13**, 1511 (1974).
- 74YAM/KEB**
R. Yamdagni and P. Kebarle, "The Hydrogen Bond Energies in ClHCl⁻ and Cl⁻(HCl)_n," *Can. J. Chem.* **52**, 2449 (1974).
- 75ALF/GOL**
Z. Alfassi, D.M. Golden and S.W. Benson, *J. Chem. Thermodyn.* **5**, 411 (1975).
- 75ALL/MEY**
N.L. Allinger and A.Y. Meyer, *Tetrahedron* **31**, 1897 (1975).
- 75ASU/BLA**
O.I. Asubiojo, L.K. Blair and J.I. Brauman, "Tetrahedral Intermediates in Gas Phase Ionic Displacement Reactions at Carbonyl Carbons," *J. Am. Chem. Soc.* **97**, 6685 (1975).
- 75BAR/PIL**
D.S. Barnes and G. Pilcher, *J. Chem. Thermodyn.* **7**, 377 (1975).
- 75BEN/MAR**
S.L. Bennett, J.L. Margrave and J.L. Franklin, "Negative Ion Electron Impact Studies of Inorganic Halides. Niobium and Tantalum Pentachlorides," *J. Inorg. Nucl. Chem.* **37**, 937 (1975).
- 75BEN/WAN**
S.L. Bennett, J.L. Wang, J.L. Margrave and J.L. Franklin, "High Temperature Negative Ions. The Enthalpies of Formation of Gaseous PbF₃ and SnF₃ from Low-Energy Electron Impact Studies," *High Temp. Sci.* **7**, 142 (1975).
- 75BER/BOU**
M.A. Bernardi, Y. Boukari and F. Busnot, *Thermochim. Acta* **16**, 2677 (1975).
- 75BRI/RIV**
S.M.J. Briscese and J.M. Riveros, "Gas Phase Nucleophilic Reactions of Aromatic Systems," *J. Am. Chem. Soc.* **97**, 230 (1975).
- 75CHE/ROD**
S.S. Chen, A.S. Rodgers, J. Chao, R.C. Wilhoit and B.J. Zwolinski, *J. Phys. Chem. Ref. Data* **4**, 441 (1975).
- 75CHI**
J.S. Chickos, *J. Chem. Ed.* **52**, 134 (1975).
- 75CHO/GOL**
K.Y. Choo, D.M. Golden and S.W. Benson, *Int. J. Chem. Kinet.* **7**, 713 (1975).
- 75CLA/KNO**
T. Clark, T. McO. Knox, H. Mackle, M.A. McKervey and J.J. Rooney, *J. Am. Chem. Soc.* **97**, 3835 (1975).
- 75CLA/SCH**
E. Clar and W. Schmidt, "Correlations Between Photoelectron and Ultraviolet Absorption Spectra of Polycyclic Hydrocarbons and the Number of Aromatic Sextets," *Tetrahedron* **31**, 2263 (1975).
- 75COM**
F. Compennolle, *Org. Mass. Spectrom.* **10**, 289 (1975).
- 75COM/DES**
F. Compennolle and F. DeSchryver, *J. Am. Chem. Soc.* **97**, 3909 (1975).
- 75COM/REI**
R.N. Compton, P.W. Reinhardt and C.D. Cooper, "Collisional Ionization of Na, K, and Cs by CO₂, COS, and CS₂: Molecular Electron Affinities," *J. Chem. Phys.* **63**, 3821 (1975).
- 75COO/NAF**
C.D. Cooper, W.T. Naff and R.N. Compton, "Negative Ion Properties of p-Benzoquinone: Electron Affinity and Compound States," *J. Chem. Phys.* **63**, 2752 (1975).
- 75DER/JOC**
J.-L. Derocque and M. Jochem, "Studies in Mass Spectrometry. III. Formation of a Common Intermediate in the Primary Fragmentation Process of 1-Phenylcyclobutene and of 2-Phenyl-1,3-butadiene," *Org. Mass Spectrom.* **10**, 935 (1975).
- 75DIE/FRA**
H. tom Dieck, K.-D. Franz and W. Majunke, "Darstellung und Eigenschaften von Diacetylbis(methylimin)," *Z. Naturforsch.* **30b**, 922 (1975).
- 75FER**
L.N. Ferguson, "Organic Molecular Structure," (Willard Grant Press, Boston, 1975).
- 75FLE/SVE**
G.D. Flesch and H.J. Svec, "Thermochemistry Of Vanadium Oxytrichloride and Vanadium Oxytrifluoride by Mass Spectrometry," *Inorg. Chem.* **14**, 1817 (1975).

- 75GAL/WIL**
L. Galyer, G. Wilkinson and D.R. Lloyd, "The Photoelectron Spectra of Hexamethyltungsten and Pentamethyltantalum," *J. Chem. Soc. Chem. Commun.* 497 (1975).
- 75GOO**
W.D. Good, *J. Chem. Thermodyn.* 7, 49 (1975).
- 75GOU/MIL**
B.K. Gould and W.J. Miller, "Electron Attachment and Compound Formation in Flames. VI. Negative Ion and Compound Formation in Flames Containing Rhenium and Potassium," *J. Chem. Phys.* 62, 644 (1975).
- 75HAR/THY**
P.W. Harland and J.C.J. Thynne, "Dissociative Electron Capture in Trifluoroacetaldehyde, Pentafluoropropionaldehyde, and Heptafluorobutyraldehyde," *Int. J. Mass Spectrom. Ion Phys.* 18, 73 (1975).
- 75HEH/POP**
W.J. Hehre and J.A. Pople, *J. Am. Chem. Soc.* 97, 6941 (1975).
- 75HOT/LIN**
H. Hotop and W.C. Lineberger, "Binding Energies of Atomic Negative Ions," *J. Phys. Chem. Ref. Data* 4, 539 (1975).
- 75HUB/LOS**
M.M. Hubers and J. Los, "Ion Pair Formation in Alkali-SF₆ Collisions: Dependence on Collisional and Vibrational Energy," *Chem. Phys.* 10, 235 (1975).
- 75JANAF**
M.W. Chase, J.L. Curnutt, H. Prophet, R.A. McDonald and A.N. Syverud, "JANAF Thermochemical Tables, 1975 Supplement," *J. Phys. Chem. Ref. Data* 4, 1 (1975).
- 75KAR/KLE**
Z. Karpas and F.S. Klein, "Negative Ion-Molecule Reactions in a Mixture of Ammonia-Formaldehyde - An ICR Mass Spectrometry Study," *Int. J. Mass Spectrom. Ion Phys.* 18, 65 (1975).
- 75KAS/HER**
A. Kasdan, E. Herbst and W.C. Lineberger, "Laser Photoelectron Spectrometry of the Negative Ions of Silicon and its Hydrides," *J. Chem. Phys.* 62, 541 (1975).
- 75KAS/HER2**
A. Kasdan, E. Herbst and W.C. Lineberger, "Laser Photoelectron Spectrometry of CH₂," *Chem. Phys. Lett.* 31, 78 (1975).
- 75KAS/HER2**
A. Kasdan, E. Herbst and W.C. Lineberger, "Laser Photoelectron Spectrometry of the Negative Ions of Silicon and Its Hydrides," *J. Chem. Phys.* 62, 541 (1975).
- 75KIN/GOD**
K.D. King and R.D. Goddard, "Very-Low-Pressure Pyrolysis of Alkyl Cyanides. II. n-Propyl Cyanide and n-Butyl Cyanide. The Heat of Formation of the Cyanomethyl Radical," *Int. J. Chem. Kinet.* 7, 837 (1975).
- 75KOL/MCK**
P. Kollman, J. McKelvey, A. Johansson and S. Rothenberg, *J. Am. Chem. Soc.* 97, 855 (1975).
- 75LEB/MIR**
V.P. Lebedev, E. Miroshnickenko, Y.N. Matyushin, V.P. Laronov, V.S. Romanov, V.E. Bukolov, G.M. Denisov, A.A. Balepin and Y.A. Lebedev, *Russ. J. Phys. Chem.* 49, 1133 (1975).
- 75LIN/ALB**
W. Lindinger, A.L. Albritton, F.C. Fehsenfeld and E.E. Ferguson, "Reactions of O⁻ with N₂, N₂O, SO₂, NH₃, CH₄, C₂H₄, and C₂H₂ with O₂ from 300K to Relative Kinetic Energies of 2 eV," *J. Chem. Phys.* 63, 3238 (1975).
- 75LLO/ROB**
D.R. Lloyd and P.J. Roberts, "Photoelectron Spectra of Halides. VII. Variable Temperature He(I) and He(II) Studies of CF₄, SiF₄, and GeF₄," *J. Electron Spectrosc. Rel. Phenom.* 7, 325 (1975).
- 75LYO/PAL**
L.E. Lyons and L.D. Palmer, "A Surface Ionization Source and Quadrupole Mass Filter for Photodetachment Studies," *Int. J. Mass Spectrom. Ion Phys.* 16, 431 (1975).
- 75MCE/SAN**
D.M. McEachern, O. Sandoval and J.C. Iniguez, *J. Chem. Thermodyn.* 7, 299 (1975).
- 75MOS/PRI**
G.M. Moselby and H.O. Pritchard, *J. Chem. Thermodyn.* 7, 977 (1975).
- 75NEN/SCH**
I. Nenner and G.J. Schultz, "Temporary Negative Ions and Electron Affinities of Benzene and N-Heterocyclic Molecules: Pyridine, Pyridazine, Pyrimidine, Pyrazine, and s-Triazine," *J. Chem. Phys.* 62, 1747 (1975).
- 75OKA**
H. Okabe, *J. Chem. Phys.* 62, 2782 (1975).
- 75OKA/WHI**
E.N. Okafu and E. Whittle, *Int. J. Chem. Kinet.* 7, 213 (1975).
- 75REE/BRA**
K.J. Reed and J.I. Brauman, "Electron Affinities of Alkoxyl Radicals and the Bond Dissociation Energies in Aliphatic Alcohols," *J. Am. Chem. Soc.* 97, 1625 (1975).
- 75RIC/STE**
J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of Electrons from Large Molecular Systems: Benzyl Anion. An Upper Limit to the Electron Affinity of C₆H₅CH₂," *J. Chem. Phys.* 63, 74 (1975).
- 75RIC/STE2**
J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of Electrons from Phenoxides and Thiophenoxide," *J. Am. Chem. Soc.* 97, 2967 (1975).
- 75RIC/STE3**
J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of Electrons from Large Molecular Systems. Pyrrolate Ion. Electron Affinity of C₄H₄N," *J. Am. Chem. Soc.* 97, 1160 (1975).
- 75RIC/STEA**
J.H. Richardson, L.M. Stephenson and J.I. Brauman, "Photodetachment of Electrons from Trifluoromethyl and Trifluorosilyl Ions; The Electron Affinities of CF₃ and SiF₃," *Chem. Phys. Lett.* 30, 17 (1975).
- 75ROT/TAN**
E.W. Rothe, S.Y. Tang and G.P. Reck, "Measurement of Electron Affinities of O₃, SO₂, and SO₃ by Collisional Ionization," *J. Chem. Phys.* 62, 3829 (1975).
- 75SCH/BOH**
H.I. Schiff and D.K. Bohme, "Flowing Afterglow Studies at York University," *Int. J. Mass Spectrom. Ion Phys.* 16, 167 (1975).
- 75SOL/FIE**
J.J. Solomon and F.H. Field, "Reversible Reactions of Gaseous Ions. IX. The Stability of C₄-C₇ Tertiary Alkyl Carbonium Ions," *J. Am. Chem. Soc.* 97, 2625 (1975).
- 75SUB/ZWO**
D.J. Subach and B.J. Zwolinski, *J. Chem. Eng. Data* 20, 232 (1975).
- 75TOR/WES**
D.F. Torgerson and J.B. Westmore, "Energetics of the Ionization and Fragmentation of Phosphorus Trifluoride by Electron Impact," *Can. J. Chem.* 53, 933 (1975).
- 75TRC**
Selected Values of Properties of Chemical Compounds. Thermodynamic Research Center, Texas A & M (Table compiled 1975).
- 75VAN/RIN**
A.J. Vanderwielen, M.A. Ring and H.E. O'Neal, *J. Am. Chem. Soc.* 97, 993 (1975).
- 75WAN/MAR**
J.L.-F. Wang, J.L. Margrave and J.L. Franklin, "Low Energy Electron Attachment to Gaseous Chromium Oxides," *J. Inorg. Nucl. Chem.* 37, 1107 (1975).

- 75WEN/KAO**
W.E. Wentworth, L.W. Kao and R.S. Becker, "Electron Affinities of Substituted Aromatic Compounds," *J. Phys. Chem.* **79**, 1161 (1975).
- 76ALL**
N.L. Allinger, *Advan. Phys. Org. Chem.* **13**, 1 (1976).
- 76ANT/CAR**
M.E. Anthony, A.S. Carson, P.G. Carson, P.G. Laye and M. Yurckli, *J. Chem. Thermodyn.* **8**, 1009 (1976).
- 76ASH**
S.J. Ashcroft, *J. Chem. Eng. Data* **21**, 397 (1976).
- 76ASH/BUR**
A.J. Ashe III, F. Burger, M.Y. El-Sheik, E. Heilbronner, J.P. Maier and J.-F. Muller, "202. Angular- and Energy-Dependence of Band Intensities in the Photoelectron Spectra of Phosphabenzene and Arsabenzene," *Helv. Chim. Acta* **59**, 1944 (1976).
- 76AUD/PET**
H.E. Audiek, M. Fetizon, Y. Henry and T. Prange, *Org. Mass Spectrom.* **11**, 1047 (1976).
- 76BAE/TSA**
T. Baer, B.P. Tsai, D. Smith and P.T. Murray, "Absolute Unimolecular Decay Rates of Energy Selected Metastable Halobenzene Ions," *J. Chem. Phys.* **64**, 2460 (1976).
- 76BEA**
J.L. Beauchamp, "Ion Cyclotron Resonance Studies of Endothermic Reactions of UF_6^- Generated by Surface Ionization," *J. Chem. Phys.* **64**, 929 (1976).
- 76BEN**
S.W. Benson, "Thermochemical Kinetics," 2nd Ed., Wiley, NY, 1976.
- 76BOB/BAR**
M.V. Bobetic and J.A. Barker, *J. Chem. Phys.* **64**, 2367 (1976).
- 76BOW/STA**
J.H. Bowie and B.J. Stapleton, "Electron Impact Studies. C. Doubly Charged Negative Ions," *J. Am. Chem. Soc.* **98**, 6480 (1976).
- 76CAR/PET**
J.L. Carlstein, J.R. Peterson and W.C. Lineberger, "Binding of an Electron by the Field of a Molecular Dipole - $LiCl^-$," *Chem. Phys. Lett.* **37**, 5 (1976).
- 76CHA**
P.J. Chantry, "Negative Ion Formation in Cerium Triiodide," *J. Chem. Phys.* **65**, 4412 (1976).
- 76CHA/ZWO**
J. Chao and B.J. Zwolinski, *J. Phys. Chem. Ref. Data* **5**, 319 (1976).
- 76COL/JIM**
M. Colomina, P. Jimenez, R. Perez-Ossario and C. Turrion, *J. Chem. Thermodyn.* **8**, 439 (1976).
- 76COM/STO**
R.N. Compton and J.A.D. Stockdale, "Formation of Gas Phase Negative Ions in $Fe(CO)_5$ and $Ni(CO)_4$," *Int. J. Mass Spectrom. Ion Phys.* **22**, 47 (1976).
- 76COS/LIN**
P.C. Cosby, J.H. Ling, J.R. Peterson and J.T. Moseley, "Photodissociation and Photodetachment of Molecular Negative Ions. III. Ions Formed in $CO_2 \cdot O_2 \cdot H_2O$ Mixtures," *J. Chem. Phys.* **65**, 5267 (1976).
- 76DAW/JEN**
J.H.J. Dawson and K.R. Jennings, "Production of Gas Phase Radical Anions by Reaction of O^- Ions with Organic Substrates," *J. Chem. Soc. Faraday Trans. II* **72**, 700 (1976).
- 76DEL/HAL**
I. Dellienl, F.M. Hall and L.G. Hepler, *Chem. Rev.* **76**, 283 (1976).
- 76ENG/LIN**
P.C. Engleking and W.C. Lineberger, "Laser Photoelectron Spectrometry of NH^- : Electron Affinity and Intercombination Energy Difference in NH^- ," *J. Chem. Phys.* **65**, 4323 (1976).
- 76FAI/JOY**
R.L. Failes, J.T. Joyce and E.C. Walton, "The Behaviour of Some Dimethyl and Trimethyl Substituted Pyridines in the Magnetron," *J. Phys.*, D **9**, 1543 (1976).
- 76FER/PIA**
D. Ferro, V. Piacente, R. Gigli and G. D'Ascenzo, *J. Chem. Thermodyn.* **8**, 1137 (1976).
- 76GOR/MUN**
A. Goren and B. Munson, "Thermochemistry of Alkyl Ions," *J. Phys. Chem.* **80**, 2848 (1976).
- 76GUY/CHU**
P.M. Guyon, W.A. Chupka and J. Berkowitz, "Photoionization Mass Spectrometric Study of Formaldehyde H_2CO , $HDCO$, and D_2CO ," *J. Chem. Phys.* **65** 1419 (1976).
- 76HAM/THO**
W.S. Hamilton, P. Thompson and S. Pustejovsky, *J. Chem. Eng. Data* **21**, 428 (1976).
- 76HOP/BOS**
H.P. Hopkins, Jr., D. Bostwick and C.J. Alexander, *J. Am. Chem. Soc.* **98**, 1355 (1976).
- 76HOP/WAH**
D.G. Hopper, A.C. Wahl, R.L.C. Wu and T.O. Tiernan, "Theoretical and Experimental Studies of the N_2O^- and N_2O Ground State Potential Energy Surface. Implications for the $O^- + N_2 \rightarrow N_2O + e^-$ and Other Processes," *J. Chem. Phys.* **65**, 5474 (1976).
- 76HUB/KLE**
M.M. Hubers, A.W. Kleyn and J. Los, "Ion Pair Formation in Alkali-Halogen Collisions at High Velocities," *Chem. Phys.* **17**, 303 (1976).
- 76JEN**
J.L. Jensen, *Prog. Phys. Org. Chem.* **12**, 189 (1976).
- 76KAR/KLE**
Z. Karpas and F.S. Klein, "ICR Study of the Gas Phase Ion Chemistry of the Carbonyl Halides: Cl_2CO , F_2CO and $ClFCO$," *Int. J. Mass Spectrom. Ion Phys.* **22**, 189 (1976).
- 76KEB**
P. Kebarle, "Ion Thermochemistry and Solvation from Gas Phase Ion Equilibria," *Ann. Rev. Phys. Chem.* **28**, 445 (1977).
- 76LIA/AUS**
S.G. Lias, P. Ausloos and Z. Horvath, "Charge Transfer Reactions in Alkane and Cycloalkane Systems. Estimated Ionization Potentials," *Int. J. Chem. Kinet.* **8**, 725 (1976).
- 76LIN**
H.J. Lindner, *Tetrahedron* **32**, 753 (1976).
- 76LOS/TRA**
F.P. Lossing and J.C. Traeger, *Int. J. Mass Spectrom. Ion Phys.* **19**, 9 (1976).
- 76LYO/PAL**
L.E. Lyons and L.D. Palmer, "The Electron Affinity of Tetracyanoethylene and Other Organic Electron Acceptors," *Aust. J. Chem.* **29**, 1919 (1976).
- 76MAC/HEM**
G.J. MacKay, R.S. Hemsworth and D.K. Bohme, "Absolute Gas-Phase Acidities of CH_3NH_2 , $C_2H_5NH_2$, $(CH_3)_2NH$, and $(CH_3)_3N$," *Can. J. Chem.* **54**, 1624 (1976).
- 76MAS**
H. Massey, "Negative Ions," 3rd. Ed., Cambridge Univ. Press, 1976.
- 76MAT/ROT**
B.P. Mathur, E.W. Rothe, S.Y. Tang and G.P. Reck, "Negative Ions from Phosphorus Halides Due to Cesium Charge Exchange," *J. Chem. Phys.* **64**, 565 (1976).
- 76MAT/ROT2**
B.P. Mathur, E.W. Rothe, S.Y. Tang, K. Mahajan and G.P. Reck, "Negative Gaseous Ions from Nitric Acid," *J. Chem. Phys.* **64**, 1247 (1976).

- 76MAU/SOL**
M. Mautner (Meot-Ner), J.J. Solomon and F.H. Field, "Stability of Some C₇ Tertiary Carbonium Ions," *J. Am. Chem. Soc.* **98**, 1025 (1976).
- 76MEG/MOO**
W.F. Meggers and C.E. Moore, "The First Spectrum of Hafnium (Hf I)," *Nat. Bur. Stand. (U.S) Monogr.* **153** (1976).
- 76MEY/HOT**
E.F. Meyer and C.A. Hotz, *J. Chem. Eng. Data* **21**, 274 (1976).
- 76MOS/COS**
J.T. Moseley, P.C. Cosby and J.R. Peterson, "Photodissociation Spectroscopy of CO₃," *J. Chem. Phys.* **65**, 2512 (1976).
- 76MUR/BEA**
M.K. Murphy and J.L. Beauchamp, "Acid-Base Properties and Gas-Phase Ion Chemistry of (CH₃)₃B," *J. Am. Chem. Soc.* **98**, 1433 (1976).
- 76NUG/WU**
W.A. Nugent, M.M.-H. Wu, T.P. Fehlner and J.K. Kochi, "Enhanced Reactivity of exo-Norbornyl Derivatives. Evidence for σ -Participation in the Absence of Steric Effects," *J. Chem. Soc. Chem. Commun.* **456** (1976).
- 76PAB/BEN**
R.E. Pabst, S.L. Bennett, J.L. Franklin and J.L. Margrave, "Negative Ion Electron Impact Studies of Arsenic Trihalides: AsF₃, AsCl₃, and AsBr₃," *J. Chem. Phys.* **64**, 1550 (1976).
- 76PAL/KEN**
M.H. Palmer and S.M.F. Kennedy, *J. Chem. Soc. Perkin II* **81** (1976).
- 76PAY/TAN**
J.D. Payzant, K. Tanaka, K.D. Betowski and D.K. Bohme, "Gas Phase S_N2 Reactions at Silicon and Carbon Centers. An Experimental Appraisal of Theory," *J. Am. Chem. Soc.* **98**, 894 (1976).
- 76POP**
D. Poppinger, *Aust. J. Chem.* **29**, 465 (1976).
- 76REF**
K.M.A. Refaey, "Endoergic Ion-Molecule-Collision Processes of Negative Ions. II. Collisions of I⁻ on H₂S, CS₂, and COS," *J. Chem. Phys.* **65**, 2002 (1976).
- 76REF2**
K.M.A. Refaey, "Endoergic Ion-Molecule-Collision Processes of Negative Ions. IV. Collisions of I⁻ on NO₂, N₂O and NO," *Int. J. Mass Spectrom. Ion Phys.* **21**, 21 (1976).
- 76REF/TRA**
K.M.A. Refaey and J.L. Franklin, "Endoergic Ion-Molecule-Collision Processes of Negative Ions. V. Collision of I⁻ on HNO₃. The Electron Affinity of NO₃," *J. Chem. Phys.* **64**, 4810 (1976).
- 76REF/FRA2**
K.M.A. Refaey and J.L. Franklin, "Endoergic Ion-Molecule-Collision Processes of Negative Ions. I. Collision of I⁻ on SO₂," *J. Chem. Phys.* **65**, 1994 (1976).
- 76REF/FRA3**
K.M.A. Refaey and J.L. Franklin, "Endoergic Ion-Molecule-Collision Processes of Negative Ions, III. Collisions of I⁻ on O₂, CO and CO₂," *Int. J. Mass Spectrom. Ion Phys.* **20**, 19 (1976).
- 76ROD/CHA**
H.J. Rodriguez, J.-C. Chang and T.F. Thomas, *J. Am. Chem. Soc.* **98**, 2027 (1976).
- 76ROS**
F.D. Rossini, *J. Chem. Thermodyn.* **87**, 651 (1976).
- 76SHE/ILJ**
V.E. Shevchenko, M.K. Iljin, O.K. Nikitin and L.N. Sidorov, "Mass Spectrometric Study of Mixed Dimers M₂BO₂F," *Int. J. Mass Spectrom. Ion Phys.* **21**, 279 (1976).
- 76SOL/FIE**
J.J. Solomon and F.H. Field, "Reversible Reactions of Gaseous Ions. X. The Intrinsic Stability of the Norbornyl Cation," *J. Am. Chem. Soc.* **98**, 1567 (1976).
- 76SOL/MAY**
R.W. Solarz, C.A. May, L.R. Carlson, E.F. Worden, S.A. Johnson, J.A. Paisner and L.J. Radziemski, Jr., *Phys. Ref.* **A14**, 1129 (1976).
- 76SUL/BEA**
S.A. Sullivan and J.L. Beauchamp, "Competition Between Proton Transfer and Elimination in the Reactions of Strong Bases with Fluoroethanes in the Gas Phase. Influence of Base Strength on Reactivity," *J. Am. Chem. Soc.* **98**, 1160 (1976).
- 76TAN/MAT**
S.Y. Tang, B.P. Mathur, E.W. Roth and G.P. Reck, "Negative Ion Formation in Halocarbons by Charge Exchange with Cesium," *J. Chem. Phys.* **64**, 1270 (1976).
- 76WHA/WEI**
D.L. Whalen, J.F. Weimaster, A.M. Ross and R. Radhe, *J. Am. Chem. Soc.* **98**, 7319 (1976).
- 76WIL/LEB**
A.D. Williamson, P.R. LeBreton and J.L. Beauchamp, *J. Am. Chem. Soc.* **98**, 2705 (1976).
- 76ZIT/ELL**
P.F. Zittel, G.B. Ellison, S.V. O'Neil, E. Herbst, W.C. Lineberger and W.P. Reinhardt, "Laser Photoelectron Spectrometry of CH₂⁻. Singlet-Triplet Splitting and Electron Affinity of CH₂¹⁻," *J. Am. Chem. Soc.* **98**, 3731 (1976).
- 76ZIT/LIN**
P.F. Zittel and W.C. Lineberger, "Laser Photoelectron Spectrometry of PO⁻, PH⁻, and PH₂⁻," *J. Chem. Phys.* **65**, 1236 (1976).
- 77ALL/KLO**
M. Allan, E. Kloster-Jensen and J.P. Maier, "Emission Spectra of Cl-C=C-H⁺, Br-C=C-H⁺ and I-C=C-H⁺ Radical Cations: A² Π \rightarrow X² Π Band Systems and the Decay of the A² Π States," *J. Chem. Soc., Faraday Trans. II* **73**, 1406 (1977).
- 77ARM/HOD**
P. Armentrout, R. Hodges and J.L. Beauchamp, "Metal Atoms as Superbases: The Gas Phase Proton Affinity of Uranium," *J. Am. Chem. Soc.* **99**, 3162 (1977).
- 77ARN/CHA**
E.M. Arnett, B. Chawla, L. Bell, M. Taagepera, W.J. Hehre and R.W. Taft, "Solvation and Hydrogen Bonding of Pyridinium Ions," *J. Am. Chem. Soc.* **99**, 5729 (1977).
- 77ASH/BUR**
F.S. Ashmore and A.R. Burgess, "Study of Some Medium Size Alcohols and Hydroperoxides by Photoelectron Spectroscopy," *J. Chem. Soc., Faraday Trans. II* **73**, 1247 (1977).
- 77BAG/COL**
M.H. Baghal-Vayjooee, J.L. Collister and H.O. Pritchard, *Can. J. Chem.* **55**, 2634 (1977).
- 77BEC/LIP**
K.H. Becker, H. Lippmann and U. Schurath, *Ber. Bunsenges. Phys. Chem.* **81**, 567 (1977).
- 77BIE/BUR**
G. Bieri, F. Burger, E. Heilbronner and J.P. Maier, "Valence Ionization Energies of Hydrocarbons," *Helv. Chim. Acta* **60**, 2213 (1977).
- 77BIE/DEP**
V.M. Bierbaum, C.H. DePuy and R.J. Shapiro, "Gas Phase Reactions of Anions with Nitrous Oxide and Carbon Dioxide," *J. Am. Chem. Soc.* **99**, 5800 (1977).
- 77CAR/LAY**
J.A.S. Carson, P.G. Laye and M. Yureali, *J. Chem. Thermodyn.* **9**, 827 (1977).
- 77CIM/PER**
R. Cimraglia, M. Persico and J. Tomasi, *J. Phys. Chem.* **81**, 1876 (1977).
- 77COM**
R.N. Compton, "On the Formation of Positive and Negative Ions in Gaseous UF₆," *J. Chem. Phys.* **66**, 4478 (1977).

77COM/COO

R.N. Compton and C.D. Cooper, "Negative Ion Properties of Tetracyanoquinodimethane: Electron Affinity and Compound States," *J. Chem. Phys.* **66**, 4325 (1977).

77COW/GOO

A.H. Cowley, D.W. Goodman, N.A. Kuebler, M. Sanchez and J.G. Verkade, "Molecular Photoelectron Spectroscopic Investigation of Some Caged Phosphorus Compounds and Related Acyclic Species," *Inorg. Chem.* **16**, 854 (1977).

77CUM/PRE

J.B. Cummings, M.A. French and P. Kebarle, "Effect of Charge Delocalization on Hydrogen Bonding to Negative Ions and Solvation of Negative Ions. Substituted Phenols and Phenoxide Ions," *J. Am. Chem. Soc.* **99**, 6999 (1977).

77DAV/FEH

J.A. Davidson, F.C. Fehsenfeld and C.J. Howard, "The Heats of Formation of NO_3^- and NO_3^- Association Complexes with HNO_3 and HBr ," *Int. J. Chem. Kinet.* **9**, 17 (1977).

77DAW/JEN

J.H.J. Dawson and K.R. Jennings, "Relative Gas Phase Acidities of Some Fluoroalcohols," *Int. J. Mass Spectrom. Ion Phys.* **25**, 47 (1977).

77DEW/NEU

R. De Wall and H. Neuert, "Die Bildung Negativer Ionen bei Elektronenstoss auf WF_6 ," *Z. Naturfor.* **32A**, 1968 (1977).

77DIS/LAC

H. Dispert and K. Lacmann, "Formation of WF_6^- and Its Dissociative Products by Collisional Ionization," *Chem Phys. Lett.* **45**, 311 (1977).

77DIS/LAC2

H. Dispert and K. Lacmann, "Chemionization in Alkali-Halogen Reactions: Evidence for Ion Formation by Alkali Dimers," *Chem. Phys. Lett.* **47**, 533 (1977).

77DON/WAL

A.M. Doncaster and R. Walsh, "Kinetic Determination of the BDE $D(\text{Me}_3\text{Ge}-\text{H})$ and Its Implication for Bond Strengths in Germanes," *J. Chem. Soc. Chem. Comm.* **446** (1977).

77DOT/DAV

I. Dotan, J.A. Davidson, G.E. Streit, D.L. Albritton and F.C. Fehsenfeld, "A Study of the Reaction $\text{O}_3^- + \text{CO}_2 = \text{CO}_3^- + \text{O}_2$ and Its Implication on the Thermochemistry of CO_3^- and O_3^- and Their Negative Ions," *J. Chem. Phys.* **67**, 2874 (1977).

77DUR/PAR

M. Durup, G. Parlant, J. Appell, J. Durup and J.-B. Ozenne, "Translational Spectroscopy of Neutralization-Reionization Double Collision Processes of Ar^+ Ions at keV Energies," *Chem. Phys.* **25**, 245 (1977).

77EFR/HUA

A. Efraty, M.H.A. Huang and C.A. Weston, "Mass Spectra of Organometallic Compounds. 4. Electron-Impact Study of Some Cyclopentadienylmetal Carbonyl Dimers," *Inorg. Chem.* **16**, 79 (1977).

77EHL/WAN

T.C. Ehlert and J.S. Wang, "Thermochemistry of the Copper Fluorides," *J. Phys. Chem.* **81**, 2069 (1977).

77ENG/LIN

P.C. Engelking and W.C. Lineberger, "Laser Photoelectron Spectrometry of FeO^- : Electron Affinity, Electronic State Separations, and Ground State Vibrations of Iron Oxide, and a New Ground State Assignment," *J. Chem. Phys.* **66**, 5054 (1977).

77ENG/LIN2

P.C. Engelking and W.C. Lineberger, "Laser Photoelectron Spectrometry of C_5H_5^- : A Determination of the Electron Affinity and Jahn-Teller Coupling in Cyclopentadienyl," *J. Chem. Phys.* **67**, 1412 (1977).

77FEL/RAC

D. Feldman, R. Rackwitz, H.J. Kaiser and E. Heincke, "Photodetachment bei einigen Neagtiven Molekulionen: P_2^- , As_2^- , CH_2^- , CH_3^- , S_3^- ," *Z. Naturfor.* **32A**, 600 (1977).

77FIN/GAT

A. Finch, P.N. Gates and S.J. Peake, "Thermochemistry of Polyhalides. III. Cesium and Rubidium Tetrachloroiodates," *J. Inorg. Nucl. Chem.* **39**, 2135 (1977).

77FRO/MCD

D.C. Frost, C.A. McDowell and N.P.C. Westwood, "The Photoelectron Spectrum of Formyl Chloride," *Chem. Phys. Lett.* **51**, 607 (1977).

77HAR

P.W. Harland, "Electron Attachment to the Stable Free Radical Bis-trifluoromethylnitroxide in the Gas Phase," *Int. J. Mass Spectrom. Ion Phys.* **25**, 61 (1977).

77HIL

D.L. Hildebrand, "Studies of Some Negative Ion Processes Involving the Tungsten Fluorides," *Int. J. Mass Spectrom. Ion Phys.* **25**, 121 (1977).

77HIR/KEB

K. Hiraoka and P. Kebarle, "Condensation Reactions Involving Carbonium Ions in the Gas Phase Synthesis of Protonated Acids in Gaseous Methane Containing Carbon Monoxide and Water Vapor," *J. Am. Chem. Soc.* **99**, 366 (1977).

77HON/WOO

S.P. Hong, S.B. Woo and E.M. Helmy, "Photodetachment of Thermally Relaxed CO_3^- ," *Phys. Rev. A* **15**, 1563 (1977).

77KAN/MOR

A.S. Kana'an and T.I. Morrison, *J. Chem. Thermodyn.* **9**, 423 (1977).

77KAR/JAD

L. Karlsson, R. Jadrny, L. Mattsson, F.T. Chau and K. Siegbahn, *Phys. Scr.* **16**, 225 (1977).

77KAR/KLE

Z. Karpas and F.S. Klein, "The Gas Phase Ion Chemistry of Carbonyl Compounds: Formyl Fluoride and a Binary Mixture of $\text{H}_2\text{CO}-\text{F}_2\text{CO}$ or $\text{H}_2\text{CO}-\text{Cl}_2\text{CO}$," *Int. J. Mass Spectrom. Ion Phys.* **24**, 137 (1977).

77KOZ/BYC

M.P. Kozina, L.V. Bychikhina and G.L. Gal'chenko, *Russ. J. Phys. Chem.* **51**, 1258 (1977).

77KRE/PRI

M.J. Krech, S.J.W. Price and H.P. Sapiiano, *Can. J. Chem.* **55**, 4222 (1977).

77KRI/ITI

N.V. Krivtsov, K.V. Titova, V.Ya. Rosolovskii, "Thermochemical Study of Complex Borates," *Russ. J. Inorg. Chem.* **22**, 374 (1977).

77KUP/SHI

A.I. Kupreev and G.S. Shimonae, *Russ. J. Phys. Chem.* **49**, 1133 (1977).

77LEB/RYA

N.D. Lebedeva, V.L. Ryadnenko, N.N. Kiseleva and L.F. Nazarova, *Vses. Konf. Kalorim. (Rasshir. Tezisy Dokl.)* **7th**, 1, 91 (1977) (CA 93:75617q (1980)).

77LIA/AUS

S.G. Lias and P. Ausloos, "Ion-Molecule Reactions Involving Halomethyl Ions; Heats of Formation of Halomethyl Ions," *Int. J. Mass Spectrom. Ion Phys.* **23**, 273 (1977).

77LIU/LOU

M.T.H. Liu, L.F. Loucks and D.G. Hooper, "Pyrolysis of Trifluoroacetaldehyde," *Int. J. Chem. Kinet.* **9**, 589 (1977).

77MAT/ROT

B.P. Mathur, R.W. Rothe and G.P. Reck, "Ionization Reactions of Metal Hexafluorides with Alkali Atoms and Dimers," *J. Chem. Phys.* **67**, 377 (1977).

77MCM/KEB

T.B. McMahon and P. Kebarle, "Intrinsic Acidities of Substituted Phenols and Benzoic Acids Determined by Gas Phase Proton Transfer Equilibria," *J. Am. Chem. Soc.* **99**, 2222 (1977).

- 77MOF**
J.B. Moffat, *J. Mol. Struct.* **42**, 251 (1977).
- 77MUR/BEA**
M.K. Murphy and J.L. Beauchamp, *J. Am. Chem. Soc.* **99**, 2085 (1977).
- 77MUR/BEA2**
M.K. Murphy and J.L. Beauchamp, "Fluorine and Alkyl Substituent Effects on Gas-Phase Lewis Acidities of Boranes by ICR Spectroscopy," *Inorg. Chem.* **16**, 2437 (1977).
- 77MUR/BEA3**
M.K. Murphy and J.L. Beauchamp, "Methyl and Fluorine Substituent Effects on the Gas-Phase Lewis Acidities of Silanes by ICR Spectroscopy," *J. Am. Chem. Soc.* **99**, 4992 (1977).
- 77NAB/SAB**
M. Nabavian, R. Sabbah, R. Chastel and M. Laffitte, *J. Chim. Phys.* **74**, 115 (1977).
- 77NGA/SAB**
S.N. Ngauv, R. Sabbah and M. Laffitte, *Thermochim. Acta* **20**, 371 (1977).
- 77NUY/MES**
O. Nuyken and K. Messmer, "Massenspektrometrische Untersuchungen an Azoverbindungen. II- Ermittlung thermodynamischer Grossen," *Org. Mass Spectrom.* **12**, 106 (1977).
- 77OTH/OLS**
J.F.M. Oth, J. Olsen and J.P. Snyder, *J. Am. Chem. Soc.* **99**, 8504 (1977).
- 77PAB/MAR**
R.E. Pabst, J.L. Margrave and J.L. Franklin, "Electron Impact Studies of the Tetrachlorides and Tetrabromides of Silicon and Germanium," *Int. J. Mass Spectrom. Ion Phys.* **25**, 361 (1977).
- 77PAB/PER**
R.E. Pabst, D.L. Perry, J.L. Margrave and J.L. Franklin, "Electron Impact Studies of Tin Tetrahalides SnCl₄, SnBr₄ and SnI₄," *Int. J. Mass Spectrom. Ion Phys.* **24**, 323 (1977).
- 77PAR/STE**
W. Parker, W.V. Steele and I. Watt, *J. Chem. Thermodyn.* **9**, 307 (1977).
- 77PED/RYL**
J.B. Pedley and J. Rylance, "Sussex-N.P.L. Computer Analysed Thermochemical Data: Organic and Organometallic Compounds," University of Sussex (1977).
- 77PEL**
P.A. Peila, *J. Chem. Thermodyn.* **9**, 301 (1977).
- 77PIC/ROD**
J.M. Pickard and A.S. Rodgers, *Int. J. Chem. Kinet.* **9**, 759 (1977).
- 77POL/HEH**
S.K. Pollack and W.J. Hehre, *J. Am. Chem. Soc.* **99**, 4845 (1977).
- 77POP/RAD**
D. Poppinger, L. Radom and J.A. Pople, *J. Am. Chem. Soc.* **99**, 7806 (1977).
- 77PRA/HUB**
M.-T. Praet, M.-J. Hubin-Franskin, J.P. Delwiche and R. Schoos, *Org. Mass Spectrom.* **12**, 297 (1977).
- 77RAC/FEL**
R. Rackwitz, D. Feldman, H.J. Kaiser and E. Heincke, "Photodetachment bei einigen Zweiatomigen Negativen Hydridionen: BeH⁻, MgH⁻, CaH⁻, ZnH⁻, PH⁻, AsH⁻," *Z. Naturfor.* **32A**, 594 (1977).
- 77REF/PRA**
K.M.A. Refaey and J.L. Franklin, "Endoergic Ion-Molecule Collision Processes of Negative Ions. VI. Collisions of I⁻ on (CN)₂ and NOCl," *Int. J. Mass Spectrom. Ion Phys.* **23**, 13 (1977).
- 77REI/PRA**
R.C. Reid, J.M. Prausnitz and T.K. Sherwood "The Properties of Gases and Liquids" 3rd Edition, McGraw Hill, New York, (1977).
- 77ROS/DRA**
H.M. Rosenstock, K. Draxl, B.W. Steiner and J.T. Herron, "Energetics of Gaseous Ions," *J. Phys. Chem. Ref. Data* **6**, Supp. 1 (1977).
- 77ROS/SOL**
P. Rosmus, B. Solouki and H. Bock, "Ground and Excited States of Thioketene Radical Cation, H₂CCS⁺," *Chem. Phys.* **22**, 453 (1977).
- 77SAB/LAF**
R. Sabbah and M. Laffitte, *J. Chem. Thermo.* **9**, 1107 (1977).
- 77SAL/YOU**
P.P.S. Saluga, T.M. Young, R.F. Rodewald, F.H. Fuchs, D. Kohli and R. Fuchs, *J. Am. Chem. Soc.* **99**, 2949 (1977).
- 77SCH/PET**
F.W. Schulte, H.-J. Petrick, H.K. Cammenga and H. Klinge, *Z. Phys. Chem. (Frankfurt)* **107**, 1 (1977).
- 77SCH/SCH**
H. Schmidt, A. Schweig, W. Thiel and M. Jones, Jr., *Chem. Ber.* **111**, 1958 (1977).
- 77SHA/GOL**
R. Shaw, D.M. Golden and S.W. Benson, *J. Phys. Chem.* **81**, 1716 (1977).
- 77SLA/LIN**
J. Slater and W.C. Linberger, "High Resolution Photodetachment Study of P⁻ and Te⁻," *Phys. Rev. A* **15**, 2277 (1977).
- 77STA/WIE**
R.H. Staley, R.D. Wieting and J.L. Beauchamp, "Carbenium Ion Stabilities in the Gas Phase and Solution. An Ion Cyclotron Resonance Study of Bromide Transfer Reactions Involving Alkali Ions, Alkyl Carbenium Ions, Acyl Cations and Cyclic Halonium Ions," *J. Am. Chem. Soc.* **99**, 5964 (1977).
- 77SIE**
W.V. Steele, *J. Chem. Thermodyn.* **9**, 311 (1977).
- 77SIE/GOL**
S.E. Stein, D.M. Golden and S.W. Benson, *J. Phys. Chem.* **81**, 314 (1977).
- 77SIE/WAT**
W.V. Steele and I. Watt, *J. Chem. Thermodyn.* **9**, 843 (1977).
- 77SUL**
S.A. Sullivan California Institute of Technology Phd. thesis, 1977.
- 77TEL/RAB**
V.I. Tel'noi and I.B. Rabinovich, *Russ. Chem. Rev.* **46**, 689 (1977).
- 77VES/MAU**
M.L. Vestal and G.H. Mauclair, "Photodissociation of Negative Ions Formed in CO₂ and CO₂/O₂ Mixtures," *J. Chem. Phys.* **67**, 3758 (1977).
- 77VOG/DRE**
D. Vogt, W. Dreves and J. Mischke, "Energy Dependence of Differential Cross Sections in Endoergic Ion-Molecule Collision Processes of Negative Ions," *Int. J. Mass Spectrom. Ion Phys.* **24**, 285 (1977).
- 77VOG/MIS**
D. Vogt and J. Mischke, "Endoergic Ion-Molecule Collision Processes of Negative Ions in Collisions of I⁻ on CO," *Phys. Lett.* **60A**, 19 (1977).
- 77WU/TIE**
R.L.C. Wu, T.O. Tiernan and C. Lifschitz, "A Long-Lived Excited State of O₃⁻: Evidence From Collision Induced Dissociation," *Chem Phys. Lett.* **51**, 211 (1977).
- 77ZIM/BRA**
A.H. Zimmerman and J.I. Brauman, "Electron Photodetachment from Negative Ions of C_{2v} Symmetry. Electron Affinities of Allyl and Cyanomethyl Radicals," *J. Am. Chem. Soc.* **99**, 3565 (1977).

- 77ZIM/REE**
A.H. Zimmerman, K.J. Reed and J.I. Brauman, "Photodetachment of Electrons from Enolate Anions. Gas Phase Electron Affinities of Enolate Radicals," *J. Am. Chem. Soc.* **99**, 7203 (1977).
- 78ADA/VOG**
A.W. Adamson, A. Vogler, H. Kunkely and R. Wachter, *J. Am. Chem. Soc.* **100**, 1298 (1978).
- 78ARO/STE**
M. Arova and W.V. Steele, *J. Chem. Thermodyn.* **10**, 403 (1978).
- 78AUS/LIA**
P. Ausloos and S.G. Lias, "Reactions of NO_2^+ and Solvated NO_2^+ Ions with Aromatic Compounds and Alkanes," *Int. J. Chem. Kinetics* **10**, 657 (1978).
- 78BEA/LEE**
P. Beak, J.K. Lee and J.M. Ziegler, *J. Org. Chem.* **43**, 1536 (1978).
- 78BEN**
S.W. Benson, *Chem. Rev.* **78**, 23 (1978).
- 78BIE/JON**
G. Bieri and B.-O. Jonsson, "HNC⁺ Radical Cation Studied by Charge-Exchange Mass Spectrometry," *Chem. Phys. Lett.* **56**, 446 (1978).
- 78BRU/FER**
A.P. Bruins, A.J. Ferrer-Correia, A.G. Harrison, K.R. Jennings and R.K. Mithcum, "Negative Ion Chemical Ionization Mass Spectrometry of Some Aromatic Compounds Using O⁻ as the Reagent Ion," *Adv. Mass Spectrom.* **7A**, 355 (1978).
- 78CAC/LIS**
T. Caceres, E.A. Lissi and E. Sanhueza, "Autooxidation of Diethylhydroxylamine," *Int. J. Chem. Kinet.* **10**, 1167 (1978).
- 78CEN/FRA**
G. Centineo, I. Fragala, G. Bruno and S. Spampinato, "Photoelectron Spectroscopy of Benzophenone, Acetophenone and Their ortho-Alkyl Derivatives," *J. Molec. Struct.* **44**, 203 (1978).
- 78CHA/ZWO**
J. Chao and B.J. Zwolinski, *J. Phys. Chem. Ref. Data* **7**, 363 (1978).
- 78COL/BEN**
A.J. Colussi and S.W. Benson, *Int. J. Chem. Kinet.* **10**, 1139 (1978).
- 78COL/JIM**
M. Colomina, P. Jimenez, M.V. Roux and C. Turrion, *J. Chem. Thermodyn.* **10**, 661 (1978).
- 78COM/REI**
R.N. Compton, P.W. Reinhardt and C.D. Cooper, "Collisional Ionization between Fast Alkali Atoms and Selected Hexafluoride Molecules," *J. Chem. Phys.* **68**, 2023 (1978).
- 78COM/REI2**
R.N. Compton, P.W. Reinhardt and C.D. Cooper, "Collisional Ionization Between Alkali Atoms and Some Methane Derivatives: Electron Affinities for CH_3NO_2 , CF_3I , and CF_3Br ," *J. Chem. Phys.* **68**, 4360 (1978).
- 78COO/FRE**
C.D. Cooper, W.F. Frey and R.N. Compton, "Negative Ion Properties of Fluoranyl, Chloranyl, and Bromanyl: Electron Affinities," *J. Chem. Phys.* **69**, 2367 (1978).
- 78COS/MOS**
P.C. Cosby, J.T. Moseley, J.R. Peterson and J.H. Ling, "Photodissociation Spectroscopy of O_3 ," *J. Chem. Phys.* **69**, 2771 (1978).
- 78CUM/KEB**
J.B. Cumming and P. Kebarle, "Summary of Gas Phase Measurements Involving Acids AH. Entropy Changes in Proton Transfer Reactions Involving Negative Ions. Bond Dissociation Energies D(A-H) and Electron Affinities EA(A)," *Can. J. Chem.* **56**, 1 (1978).
- 78DAB/HER**
I. Dabrowski and G. Herzberg, "The Spectrum of HeNe⁺," *J. Molec. Spectrosc.* **73**, 183 (1978).
- 78DAW/NIB**
J.H.J. Dawson and N.M.M. Nibbering, "Concerning $\text{CH}_2=\text{C}^-$ and Its Reaction with $^{14}\text{N}^{15}\text{NO}$," *J. Am. Chem. Soc.* **100**, 1928 (1978).
- 78DEF/HEH**
D.J. DeFrees and W.J. Hehre, *J. Phys. Chem.* **82**, 391 (1978).
- 78DEP/BIE**
C.H. DePuy, V.M. Bierbaum, R.J. Schmitt and R.H. Shapiro, "Gas Phase Oxidation and Reduction Reactions with C_6H_7^- , HNO^- , and HO_2^- ," *J. Am. Chem. Soc.* **100**, 2970 (1978).
- 78DIS/LAC**
H. Dispert and K. Lacmann, "Negative Ion Formation in Collisions between Potassium and Fluoro- and Chloromethanes: Electron Affinities and Bond Dissociation Energies," *Int. J. Mass Spectrom. Ion Phys.* **28**, 49 (1978).
- 78DOT/ALB**
I. Dotan, D.L. Albritton, F.C. Fehsenfeld, G.E. Streit and E.E. Ferguson, "Rate Constants for the Reactions of O⁻, O₂⁻, NO₂⁻, CO₃⁻, and CO₄⁻ with HCl and ClO⁻ with NO, NO₂, SO₂, and CO₂ at 300K," *J. Chem. Phys.* **68**, 5414 (1978).
- 78DUD/BAL**
A.V. Dudin, A.V. Balaev and L.N. Gorokhov, "Mass Spectrometric Study of Tetrafluorohydrazine and the Products Resulting from Its Breakdown Under Electron Impact," *Izv. Akad. Nauk SSSR, Ser. Khim.* **1306** (1978).
- 78ELL/ENG**
G.B. Ellison, P.C. Engelking and W.C. Lineberger, "An Experimental Determination of the Geometry and Electron Affinity of CH₃," *J. Am. Chem. Soc.* **100**, 2556 (1978).
- 78ENG/ELL**
P.C. Engleking, G.B. Ellison and W.C. Lineberger, "Laser Photodetachment Electron Spectrometry of Methoxide, Deuteromethoxide, and Thiomethoxide: Electron Affinities and Vibrational Structure of CH₃O, and CH₃S," *J. Chem. Phys.* **69**, 1826 (1978).
- 78FU/DUN**
E.W. Fu and R.C. Dunbar, "Photodissociation Spectroscopy and Structural Rearrangements in Ions of Cyclooctatetraene, Styrene and Related Molecules," *J. Am. Chem. Soc.* **100**, 2283 (1978).
- 78GAL/PAI**
S.E. Galembeck, J.F.G. Faigle and J.M. Riveros, *An. Acad. Brasil Cienc.* **50**, 1 (1978).
- 78GAN/PEE**
T.H. Gan, J.B. Peel and G.D. Willett, "Photoelectron Spectra of the Gauche and Trans Conformers of 1,2-Dibromoethane," *J. Molec. Struct.* **44**, 211 (1978).
- 78GRE/LIE**
A. Greenberg and J.F. Liebman, "Strained Organic Molecules" (Academic Press, New York, 1978).
- 78GUN/HEA**
H.A. Gundry and A.J. Head, *J. Chem. Thermodyn.* **10**, 195 (1978).
- 78HAR/HEA**
D. Harrop and A.J. Head, *J. Chem. Thermodyn.* **10**, 705 (1978).
- 78HAV/MON**
J.J. Havel, R.L. Montgomery, C.-C. Lau and M. Grissom, *J. Chem. Eng. Data* **23**, 132 (1978).
- 78HEA/HEI²**
G.A. Heath, G.T. Hefter and W.V. Steele, *J. Chem. Thermodyn.* **10**, 395 (1978).
- 78HER/SCH**
A. Herrmann, E. Schumacher and L. Woste, "Preparation and Photoionization Potentials of Molecules of Sodium, Potassium and Mixed Atoms," *J. Chem. Phys.* **68**, 2327 (1978).
- 78JANAF**
M.W. Chase, J.L. Curnutt, R.A. McDonald and A.N. Syverud, "JANAF Thermochemical Tables, 1978 Supplement," *J. Phys. Chem. Ref. Data* **7**, 793 (1978).

- 78JAN/ZIM**
B.K. Janousek, A.H. Zimmerman, K.J. Reed and J.I. Brauman, "Electron Detachment from Aliphatic Molecular Anions. Gas Phase Electron Affinities of Methoxyl, tert-Butoxyl, and Neopentoxyl Radicals," *J. Am. Chem. Soc.* **100**, 6142 (1978).
- 78JOH/RAD**
I.G. John and L. Radom, *J. Am. Chem. Soc.* **100**, 3981 (1978).
- 78JOR/BUR**
K.D. Jordan and P.D. Burrow, "Studies of the Temporary Anion States of Unsaturated Hydrocarbons by Electron Transmission Spectroscopy," *Acc. Chem. Res.* **11**, 341 (1978).
- 78KAO/RAD**
J. Kao and L. Radom, *J. Am. Chem. Soc.* **100**, 760 (1978).
- 78KIM/WIN**
K.Y. Kim, R.E. Winans, W.N. Hubbard and C.E. Johnson, *J. Phys. Chem.* **82**, 402 (1978).
- 78KUD/KUD**
S.A. Kudchadker, A.P. Kudchadker, R.C. Wilhoit and B.J. Zwolinski, *J. Phys. Chem. Ref. Data* **7**, 417 (1978).
- 78KUD/KUD2**
S.A. Kudchadker and A.P. Kudchadker, *J. Phys. Chem. Ref. Data* **7**, 1285 (1978).
- 78LEB/TSV**
B.V. Lebedev, L.Y. Tsvetkova and I.B. Rabinovich, *J. Chem. Thermodyn.* **10**, 809 (1978).
- 78LEU/WIR**
W. Leupin and J. Wirz, *Helv. Chim. Acta* **61**, 1663 (1978).
- 78LIA/AUS**
S.G. Lias and P. Ausloos, "Ionization Energies of Organic Compounds by Equilibrium Measurements," *J. Am. Chem. Soc.* **100**, 6027 (1978).
- 78LIF/WU**
C. Lifshitz, R.L.C. Wu, T.O. Tiernan and T.E. Terwilliger, "Negative Ion-Molecule Reactions of Ozone and Their Implications on the Thermochemistry of O₂⁻," *J. Chem. Phys.* **68**, 247 (1978).
- 78LUC/WLO**
Z. Luczynski, S. Wlodek and H. Wincel, "Stabilities of HCOO⁻ (HCOOH)_n and Cl⁻(HCOOH)_n Clusters," *Int. J. Mass Spectrom. Ion Phys.* **26**, 103 (1978).
- 78MAC/BOH**
G.I. MacKay and D.K. Bohme, "Proton-Transfer Reactions in Nitromethane at 297K," *Int. J. Mass Spectrom. Ion Phys.* **26**, 327 (1978).
- 78MAC/BOH2**
G.I. Mackay and D.K. Bohme, "Bridging the Gap Between the Gas Phase and Solution: Transition in the Relative Acidity of Water and Methanol at 296±2 K," *J. Am. Chem. Soc.* **100**, 327 (1978).
- 78MAC/LIE**
G.I. Mackay, M.H. Lien, A.C. Hopkinson and D.K. Bohme, "Experimental and Theoretical Studies of Proton Removal from Propene," *Can. J. Chem.* **56**, 131 (1978).
- 78MAR/SCH**
H.-D. Martin, H.-J. Schiwiek, J. Spanget-Larsen and R. Gleiter, "Synthese, spektroskopische Eigenschaften und transannulare Wechselwirkungen tricyclischer 1,2-Cyclobutandione," *Chem. Ber.* **111**, 2557 (1978).
- 78MCC/FRE**
D.A. McCrery and B.S. Freiser, "Gas Phase Photodissociation of C₇H₇⁺," *J. Am. Chem. Soc.* **100**, 2902 (1978).
- 78MCC/HAM**
D.G. McCormick and W.S. Hamilton, *J. Chem. Thermodyn.* **10**, 275 (1978).
- 78MCM/NOR**
T.B. McMahon and C.J. Northcott, "The Fluoroformate Ion FCO₂⁻: An ICR Study of the Gas Phase Lewis Acidity of Carbon Dioxide and Related Isoelectronic Species," *Can. J. Chem.* **56**, 1068 (1978).
- 78MOF**
J.B. Moffat, *J. Mol. Struct.* **44**, 637 (1978).
- 78MON/ROS**
R.L. Montgomery and F.D. Rossini, *J. Chem. Eng. Data* **23**, 125 (1978).
- 78OLS/HOW**
J.F. Olsen and J.M. Howell, *Theor. Chim. Acta* **47**, 39 (1978).
- 78OSB/SCO**
A.G. Osborn and D.W. Scott, *J. Chem. Thermodyn.* **10**, 619 (1978).
- 78PAB/MAR**
R.E. Pabst, J.L. Margrave and J.L. Franklin, "Energy Distribution in the Products of Ionic Decomposition," *Adv. Mass Spectrom.* **7B**, 1217 (1978).
- 78PAL/KEN**
M.H. Palmer and S.M.F. Kennedy, *J. Mol. Struct.* **43**, 203 (1978).
- 78PAP/KOL**
T.S. Papina and V.P. Kolesov, *Vestn. Mosk. Univ. Ser. 2 Khim.* **19**, 500 (1978).
- 78PAU/KIM**
J.K. Pau, J.K. Kim and M.C. Caserio, "Mechanisms of Ionic Reactions in the Gas Phase. Displacement Reactions at Carbonyl Carbon," *J. Am. Chem. Soc.* **100**, 3831 (1978).
- 78POP/RAD**
D. Poppinger and L. Radom, *J. Am. Chem. Soc.* **100**, 3674 (1978).
- 78RAI/MOO**
L.J. Rains, H.E. Moore and R.T. McIver, Jr., "Equilibrium Electron-Transfer Reactions in the Gas Phase Involving Long-Lived Negative Ion Radicals," *J. Chem. Phys.* **68**, 3309 (1978).
- 78REF/TRA**
K.M.A. Refaey and J.L. Franklin, "Collisional Decomposition of SF₆⁻," *Int. J. Mass Spectrom. Ion Phys.* **26**, 125 (1978).
- 78ROB/WIN**
B.P. Roberts and J.N. Winter, "Generation and Some Reactions of the Bis(trimethylsilyl)aminy radical," *Chem. Comm.* **545** (1978).
- 78ROG/VON**
D.W. Rogers, H. von Voithenberg and N.L. Allinger, *J. Org. Chem.* **43**, 360 (1978) and personal communication from D.W. Rogers to the authors.
- 78ROT/BIE**
W.R. Roth, M. Biermann, H. Dekker, R. Jochem, C. Mosselman and H. Hermann, *Chem. Ber.* **111**, 3872 (1978).
- 78SAB/LAF**
R. Sabbah and M. Laffitte, *J. Chem. Thermodyn.* **10**, 101 (1978).
- 78SAB/LAF2**
R. Sabbah and M. Laffitte, *Bull. Soc. Chim. Fr.* **1**, 50 (1978).
- 78SAB/LAF3**
R. Sabbah and M. Laffitte, *Thermochim. Acta* **23**, 196 (1978).
- 78SEL/STR**
P. Sellers, G. Stridh and S. Sunner, *J. Chem. Eng. Data* **23**, 3 (1978).
- 78SEL/STR2**
P. Sellers, G. Stridh and S. Sunner, *J. Chem. Eng. Data* **23**, 250 (1978).
- 78SHA**
R. Shaw, "The Chemistry of the Carbon-Carbon Triple Bond" Part 1 (ed. S. Patai, John Wiley & Sons, New York, 1978).
- 78SMI/LEE**
G.P. Smith, L.C. Lee, P.C. Cosby, J.R. Peterson and J.T. Moseley, "Photodissociation and Photodetachment of Molecular Negative Ions. V. Atmospheric Ions from 7000 to 8400 Å," *J. Chem. Phys.* **68**, 3818 (1978).
- 78STE**
W.V. Steele, *J. Chem. Thermodyn.* **10**, 441 (1978).

- 78STE2**
W.V. Steele, *J. Chem. Thermodyn.* **10**, 445 (1978).
- 78STE3**
W.V. Steele, *J. Chem. Thermodyn.* **10**, 585 (1978).
- 78SUL/BEA**
S.A. Sullivan and J.L. Beauchamp, "Nucleophilic Reactions of Anions with PF₃ and OPF₃ in the Gas Phase by ICR Spectroscopy," *Inorg. Chem.* **17**, 1589 (1978).
- 78SUL/BEA2**
S.A. Sullivan and J.L. Beauchamp, "Positive and Negative Ion Chemistry of Sulfuryl Halides," *Int. J. Mass Spectrom. Ion Phys.* **28**, 69 (1978).
- 78TAF/TAA**
R.W. Taft, M. Taagepera, J.L.M. Abboud, J.F. Wolf, D.J. DeFrees, W.J. Hehre, J.E. Bartmess and R.T. McIver, Jr., "Regarding the Separation of Polarizability and Inductive Effects in Gas- and Solution-Phase Proton Transfer Equilibria," *J. Am. Chem. Soc.* **100**, 7765 (1978).
- 78TIE/WU**
T.O. Tiernan and R.L.C. Wu, "Thermochemical Data for Molecular Negative Ions from Collisional Dissociation Thresholds," *Adv. Mass Spectrom.* **7A**, 136 (1978).
- 78TPIS**
L.V. Gurvich, I.V. Veits, V.A. Medvedev, G.A. Khachkuruzov, V.S. Yungman and G.A. Bergman, et. al., "Termodinamicheskie Svoistva Individual'nykh Veshchestv" (Thermodynamic Properties of Individual Substances); Glushko, V.P., gen. ed., Vol. 1, parts 1 and 2 (1978).
- 78TRA/MCL**
J.C. Traeger and R.G. McLoughlin, "A Photoionization Study of the Energetics of C₇H₇⁺ Ion Formed from C₇H₈ Precursors," *Int. J. Mass Spectrom. Ion Phys.* **27**, 319 (1978).
- 78TRC**
Selected Values of Properties of Chemical Compounds. Thermodynamic Research Center, Texas A & M (Table compiled 1978).
- 78VAN/OSK**
H. Van Dam and A. Oskam, "He(I) and He(II) Photoelectron Spectra of Some Substituted Ethylenes," *J. Electron Spectrosc. Rel. Phenom.* **13**, 273 (1978).
- 78VOG/WIL**
J. Vogt, A.D. Williamson and J.L. Beauchamp, *J. Am. Chem. Soc.* **100**, 3478 (1978).
- 78ZAB/BEN**
F. Zabel, S.W. Benson and D.M. Golden, *Int. J. Chem. Kinet.* **10**, 295 (1978).
- 78ZIM/GYG**
A.H. Zimmerman, R. Gygas and J.I. Brauman, "Electron Photodetachment Spectroscopy of Polyene Anions. Electron Affinities of Pentadienyl and Heptatrienyl Radicals," *J. Am. Chem. Soc.* **100**, 5595 (1978).
- 78ZIM/JAC**
A.H. Zimmerman, R.L. Jackson, B.K. Janousek and J.J. Brauman, "Electron Photodetachment from Cyclic Enolate Anions in the Gas Phase: Electron Affinities of Cyclic Enolate Radicals," *J. Am. Chem. Soc.* **100**, 4674 (1978).
- 79ADE/CAV**
F.A. Adedeji, K.J. Cavell, S. Cavell, J.A. Connor, H.A. Skinner and M.T. Zafarani-Moattar, *J. Chem. Soc. Faraday Trans. 1* **75**, 603 (1979).
- 79AND/KOL**
G.M. Anderson, III, P.A. Kollman, L.N. Domelsmith and K.N. Houk, *J. Am. Chem. Soc.* **101**, 2344 (1979).
- 79AUE/BOW**
D.H. Aue and M.T. Bowers, "Stabilities of Positive Ions from Equilibrium Gas-Phase Basicity Measurements," in "Gas Phase Ion Chemistry," M.T. Bowers Editor, Vol. 2, pp. 1-51 Academic Press, New York (1979).
- 79BAG/NIK**
N.V. Bagarat'yan and O.T. Nikitin, "Electron Ionization of Vapor over Boron Oxide," *Vest. Mosk. Univ. Khim.* **34**, 539 (1979).
- 79BAR/MCI**
J.E. Bartmess and R.T. McIver, Jr., "The Gas Phase Acidity Scale," in "Gas Phase Ion Chemistry," V. 2, M.T. Bowers, Ed., Academic Press, NY, 1979, Ch. 11.
- 79BAR/SCO**
J.E. Bartmess, J.A. Scott and R.T. McIver, Jr., "The Gas Phase Acidity Scale from Methanol to Phenol," *J. Am. Chem. Soc.* **101**, 6047 (1979).
- 79BRE/ENG**
E.J. Bredford and F. Engelke, *J. Chem. Phys.* **71**, 1994 (1979).
- 79BUC/VOG**
U. Buchler and J. Vogt, *Org. Mass Spectrom.* **14**, 503 (1979).
- 79BYK/ELI**
H.J. Byker, I. Eliezer, R.C. Howard and T.C. Ehlert, *High Temp. Sci.* **11**, 153 (1979).
- 79CAR/MOU**
P. Carlier and G. Mouvier, "Etude par Spectrometrie de Photoelectrons de la Structure Electronique des Ynals et des Ynones Conjugues," *J. Electron Spectrosc. Rel. Phenom.* **17**, 169 (1979).
- 79CLA/KNO**
T. Clark, T. McO. Knox, M.A. McKervey, H. Mackle, J.J. Rooney, *J. Am. Chem. Soc.* **101**, 2404 (1979).
- 79CLA/SCH**
E. Clar and W. Schmidt, "Correlations between Photoelectron and UV Absorption Spectra of Polycyclic Hydrocarbons. The Pyrene Series," *Tetrahedron* **35**, 1027 (1979).
- 79COR**
M. Corval, "Elimination sous Impact Electronique de HCN et H a partir de l'Indole. Comparaison avec l'Indolizine," *Org. Mass Spectrom.* **14**, 213 (1979).
- 79COR/ENG**
R.R. Corderman, P.C. Engelking and W.C. Lineberger, "Laser Photoelectron Spectrometry of Co⁻ and Ni⁻," *J. Chem. Phys.* **70**, 4474 (1979).
- 79DAW/NOE**
J.H.J. Dawson, A.J. Noest and N.M.M. Nibbering, "The Gas Phase Allyl Anion," *Int. J. Mass Spectrom. Ion Phys.* **29**, 205 (1979).
- 79DAW/NOE2**
J.H.J. Dawson, A.J. Noest and N.M.M. Nibbering, "1,1 and 1,3 Eliminations of Water from the Reaction Complex of O⁻ with 1,1,1-Trideuterioacetone," *Int. J. Mass Spectrom. Ion Phys.* **30**, 189 (1979).
- 79DEK/OON**
C.G. DeKruif and H.A.J. Oonk, *J. Chem. Thermodyn.* **11**, 2877 (1979).
- 79DEK/VOO**
C.G. DeKruif, J. Voogd and J.C.A. Offringa, *J. Chem. Thermodyn.* **11**, 651 (1979).
- 79DEL**
J.E. Del Bene, *J. Am. Chem. Soc.* **101**, 6184 (1979) and personal communication of the results of unpublished ab-initio calculations.
- 79DEN/VAN**
R.J. Dendramas, R.J. Van Zee, W. Weltner, *J. Astrophys.* **231**, 632 (1979).
- 79DEV/WIJ**
C. De Vrengd, R.W. Wijnaendts van Resandt, J. Los and B. Smith, "Differential Cross Sections for Collisions of Negative Halogen Ions and Alkali Atoms," *Chem. Phys.* **42**, 305 (1979).
- 79DEV/WIJ2**
C. De Vrengd, R.W. Wijnaendts van Resandt and J. Los, "The Well Depths of XeF⁻ and XeCl⁻ from Differential Scattering Measurements," *Chem. Phys. Lett.* **65**, 93 (1979).
- 79DIL/GRE**
J.D. Dill, A. Greenberg and J.F. Liebman, *J. Am. Chem. Soc.* **101**, 6814 (1979).

- 79DIL/MCL**
J.D. Dill and F.W. McLafferty, "Collisional Activation and Theoretical Studies of Gaseous CSH_3^+ Ions," *J. Am. Chem. Soc.* **101**, 6526 (1979).
- 79DRA/GLA**
J.E. Drake, B.M. Glavincevski and K. Gorzelska, "The Photoelectron Spectra of Methylgermane, Trifluoro- and Trichloro-methylgermane," *J. Electron Spectrosc. Rel. Phenom.* **17**, 73 (1979).
- 79DRA/GLA2**
J.E. Drake, B.M. Glavincevski and K. Gorzelska, "The Photoelectron Spectra of Trimethylgermane, Chloro- and Fluoro-Trimethylgermane," *J. Electron Spectrosc. Rel. Phenom.* **16**, 331 (1979).
- 79DUD/BAL**
A.V. Dudin, A.V. Baluev and L.N. Gorokhov, "A Mass Spectrometric Investigation of Nitrogen Trifluoride by the Electron Shock Method," *Bull. Acad. Sci. USSR, Div. Chem. Sci. (Engl. Trans.)* **28**, 1996 (1979).
- 79DUD/GOR**
A.V. Dudin, L.N. Gorokhov and A.V. Balaev, "A Study of the Electron Impact Ionization of Chlorine Trifluoride and Its Decomposition Products by Mass Spectrometry," *Izv. Akad. Nauk SSSR, Ser. Khim.*, 2408 (1979).
- 79DUN/DYK**
S. Dunlavey, J. Dyke, N. Fayad, N. Jonathan and A. Morris, "Vacuum Ultraviolet Photoelectron Spectroscopy of Transient Species. Part 10. The $\text{SH}(\text{X}_{2\Pi})$ Radical and the $\text{S}(\text{P})$ Atom," *Mol. Phys.* **38**, 729 (1979).
- 79ELL/EAD**
M.R. Ellenberger, R.A. Eades, M.W. Thomsen, W.E. Farneth and D.A. Dixon, *J. Am. Chem. Soc.* **101**, 7151 (1979).
- 79ENG/LIN**
P.C. Engelking and W.C. Lineberger, *Phys. Rev. A* **19**, 149 (1979).
- 79ENG/LIN2**
P.C. Engelking and W.C. Lineberger, "Laser Photoelectron Spectrometry of the Negative Ions of Iron and Iron Carbonyls. Electron Affinity Determination for the Series $\text{Fe}(\text{CO})_n$, $n=0,1,2,3,4$," *J. Am. Chem. Soc.* **101**, 5569 (1979).
- 79ERA/KOL**
P.A. Erastov and V.P. Kolesov, *Russ. J. Gen. Chem.* **49**, 1186 (1979).
- 79FUC/PEA**
R. Fuchs and L.A. Peacock, *Can. J. Chem.* **57**, 2302 (1979).
- 79FUC/PEA2**
R. Fuchs and L.A. Peacock, *J. Phys. Chem.* **83**, 1975 (1979).
- 79GEO/BEA**
P.M. George and J.L. Beauchamp, "The Electron and Fluoride Affinities of Tungsten Hexafluoride by ICR Spectroscopy," *Chem. Phys.* **36**, 345 (1979).
- 79GOL/KUL**
A.E. Golubitskii, N.S. Kulikov, A.M. Zyakun, V.A. Valovoi, A.M. Alekseev and V.N. Volkov, "Photoionization Mass Spectra of Alicyclic Compounds with Various Substituents, and Their Ionization Energies and Appearance Energies," *Izvest. Akad. Nauk SSSR, Seriya Khimicheskaya* **11**, 2602 (1979).
- 79GUS/GOR**
A.V. Gusarov, L.N. Gorokhov, A.T. Pyatenko and I.V. Sidorova, "Negative Ions in the Vapors of Inorganic Compounds," *Adv. Mass Spectrom.* **8A**, 262 (1979).
- 79GYG/PET**
R. Gygas, H.L. Peters and J.I. Brauman, "Photodetachment of Electrons from Anions of High Symmetry. Electron Photodetachment Spectra of the Cyclooctatetraenyl and Perinaphthenyl Anions," *J. Am. Chem. Soc.* **101**, 2567 (1979).
- 79HA**
T.-K. Ha, *J. Mol. Struct.* **51**, 87 (1979).
- 79HAC/PIL**
J.M. Hacking and G. Pilcher, *J. Chem. Thermodyn.* **11**, 1015 (1979).
- 79HOL/LOS**
J.L. Holmes and F.P. Lossing, *Org. Mass Spectrom.* **14**, 572 (1979).
- 79HOL/TER**
J.L. Holmes and J.K. Terlouw, "Structures of $[\text{C}_4\text{H}_4\text{O}]^+$ Ions Produced from 2- and 4-Pyrone," *J. Am. Chem. Soc.* **101**, 4973 (1979).
- 79HOU**
F.A. Houle Ph. D. Thesis, California Institute of Technology (1979).
- 79HUB/HER**
K.P. Huber and G. Herzberg, "Molecular Spectra and Molecular Structure. IV. Constants of Diatomic Molecules," Van Nostrand Reinhold Co. (1979).
- 79ILL/SCH**
T. Illenberger, H. Scheunemann and H. Baumgartel, "Negative Ion Formation in CF_2Cl_2 , CF_3Cl , and CFCl_3 Following Low Energy (0-10eV) Impact with Near Monoenergetic Electrons," *Chem. Phys.* **37**, 21 (1979).
- 79JAN/BRA**
B.K. Janousek, J.I. Brauman and J. Simons, "An Experimental and Theoretical Determination of the Electron Affinity of the Ethynyl Radical HC_2 ," *J. Chem. Phys.* **71**, 2057 (1979).
- 79JAN/BRA2**
B. Janousek and J.I. Brauman, "Electron Affinities," in "Gas Phase Ion Chemistry," V. 2, M.T. Bowers, Ed., Academic Press, NY, 1979, Chapter 10.
- 79JOC/LOH**
H.W. Jochims, W. Lohr and H. Baumgartel, "Photoreactions of Small Organic Molecules. VI. Photoionization Processes of Difluoroethylenes," *Nouv. J. Chim.* **3**, 109 (1979).
- 79JOS**
R.M. Joshi, *J. Macromol. Sci. Chem.* **A13**, 1015 (1979).
- 79KAO/RAD**
J. Kao and L. Radom, *J. Am. Chem. Soc.* **101**, 311 (1979).
- 79KLO/SEP**
G. Kloter and K. Sepplet, "Trifluoromethanol (CF_3OH) and Trifluoromethylamine (CF_3NH_2)," *J. Am. Chem. Soc.* **101**, 347 (1979).
- 79KUD/KUD**
S.A. Kudchadker, A.P. Kudchadker, R.C. Wilhoit and B.J. Zwolinski, *Thermochem. Acta* **30**, 319 (1979).
- 79KUD/KUD2**
A.P. Kudchadker, S.A. Kudchadker and R.C. Wilhoit, "Four-Ring Condensed Aromatic Compounds," API Monograph 709-79, American Petroleum Institute, Washington, D.C. (1979).
- 79KUD/KUD3**
A.P. Kudchadker and S.A. Kudchadker, "Pyridine and Phenylpyridines," API Monograph 710-79, American Petroleum Institute, Washington, D.C. (1979).
- 79KUD/KUD4**
S.A. Kudchadker, A.P. Kudchadker and B.J. Zwolinski, *J. Chem. Thermodyn.* **11**, 1051 (1979).
- 79LEE/POT**
E.P.F. Lee and A. W. Potts, "The HeII α Photoelectron Spectra of the Barium Halides," *Chem. Phys. Lett.* **63**, 61 (1979).
- 79LEE/POT2**
E.P.F. Lee and A.W. Potts, "An Investigation of the Valence Shell Electronic Structure of Alkaline Earth Halides by Using Ab Initio S. C. F. Calculations and Photoelectron Spectroscopy," *Proc. Roy. Soc. Lond. A* **365**, 395 (1979).
- 79LEE/SMI**
L.C. Lee, G.P. Smith, J.T. Moseley, P.C. Cosby and J.A. Guest, "Photodissociation and Photodetachment of Cl_2^- , ClO^- , Cl_3^- , and BrCl_2^- ," *J. Chem. Phys.* **70**, 3237 (1979).
- 79LEI**
J. Leitch, *Int. J. Chem. Kinet.* **11**, 1249 (1979).

- 79MAJ/SVO**
V. Majer, V. Svoboda, J. Koubeck and J. Pick, *Collect. Czech. Chem. Commun.* **44**, 3521 (1979).
- 79MAR/KUN**
H.-D. Martin, M. Kunze and J.D. Beckhaus, *Tetrahedron Lett.* **3069** (1979).
- 79MAT/ROT**
E.P. Mathur, E.W. Rothe and G.P. Reck, "Negative Ions from the Reactions of Alkalis with SnCl₄, GeCl₄, and TiCl₄," *Int. J. Mass Spectrom. Ion Phys.* **31**, 77 (1979).
- 79MCL/TRA**
R.G. McLoughlin and J.C. Traeger, "A Photoionization Study of Some Benzoyl Compounds - Thermochemistry of [C₇H₅O]⁺ Formation," *Org. Mass Spectrom.* **14**, 434 (1979).
- 79MIL**
W.J. Miller, "The Use of Flames as Media for the Study of Ion-Molecule Thermochemistry," in, "Characterization of High Temperature Vapors and Gases," *NBS Spec. Pub.* **561/1**, 443 (1979).
- 79NAZ/POK**
V.A. Nazarenko and V.D. Pokhodenko, "The Formation of Negative Ions in High Electric Fields," *Int. J. Mass Spectrom. Ion Phys.* **31**, 381 (1979).
- 79NEL/KES**
S.F. Nelsen, C.R. Kessel and H.N. Brace, "Twisted and Bent Hydrazine Radical Cations," *J. Am. Chem. Soc.* **101**, 1874 (1979).
- 79NIK/SKO**
M.I. Nikitin, E.V. Skokan, I.D. Sorokin and L.N. Sidirov, *Dokl. Akad. Nauk SSSR* **247**, 151 (1979).
- 79NOV/ENG**
S.E. Novich, P.C. Engelking, P.L. Jones, J.H. Futrell and W.C. Lineberger, "Laser Photoelectron, Photodetachment, and Photodestruction Spectra of O₃," *J. Chem. Phys.* **70**, 2652 (1979).
- 79OLS**
J.F. Olsen, *J. Mol. Struct.* **57**, 245 (1979).
- 79OSA**
E. Osawa, *J. Am. Chem. Soc.* **101**, 5523 (1979).
- 79PET/MAJ**
L. Petros, V. Majer, J. Koubeck, V. Svoboda and J. Pick, *Collect. Czech. Chem. Commun.* **44**, 3533 (1979).
- 79PRI/SAP**
S.J.W. Price and H.J. Sapiiano, *Can. J. Chem.* **57**, 685 (1979).
- 79PRI/SAP2**
S.J.W. Price and H.J. Sapiiano, *Can. J. Chem.* **57**, 1468 (1979).
- 79RAU/ACK**
E.G. Rauh and R.J. Ackermann, "The First Ionization Potentials of the Transition Metals," *J. Chem. Phys.* **70**, 1004 (1979).
- 79ROB/FRA**
R. Robbiani and J.L. Franklin, "Formation of the Trihalide Ion Cl₃⁻ in the Gas Phase," *J. Am. Chem. Soc.* **101**, 764 (1979).
- 79ROB/FRA2**
R. Robbiani and J.L. Franklin, "Negative Ion-Molecule Reaction in Sulfuryl Halides," *J. Am. Chem. Soc.* **101**, 3709 (1979).
- 79ROG/DAG**
D.W. Rogers, O.A. Dagdagan and N.L. Allinger, *J. Am. Chem. Soc.* **101**, 671 (1979).
- 79ROS/GOL**
M. Rossi and D.M. Golden, "Absolute Rate Constants for the Metathesis Reactions of Allyl and Benzyl Radicals with HI(D). Heat of Formation of Allyl and Benzyl Radicals," *J. Am. Chem. Soc.* **101**, 1230 (1979).
- 79SAB**
R. Sabbah, *Bull. Soc. Chim. Fr.* **I-434** (1979).
- 79SAL/PEA**
P.P.S. Saluja, L.A. Peacock and R. Fuchs, *J. Am. Chem. Soc.* **101**, 1958 (1979).
- 79SAN/EPS**
D.J. Sandman, A.J. Epstein, J.S. Chickos, J. Ketchum and H.A. Scheraga, *J. Chem. Phys.* **70**, 305 (1979).
- 79SAU/CHR**
I. Savers, J.G. Christophorou and J.G. Carter, "Electron Attachment to Perfluorocarbon Compounds. III. Fragmentation of Aliphatic Perfluorocarbons of Interest to Gaseous Dielectrics," *J. Chem. Phys.* **71**, 3016 (1979).
- 79SCH/BIE**
R.J. Schmitt, V.M. Bierbaum and C.H. DePuy, "Gas Phase Reactions of Carbanions with Triplet and Singlet Molecular Oxygen," *J. Am. Chem. Soc.* **101**, 6443 (1979).
- 79SCH/GRU**
B. Schaldach and H.F. Grutzmacher, "Kinetics Energy Release and Position of the Transition State during the Intramolecular Substitution of Ionized Benzalacetones," *Int. J. Mass Spectrom. Ion Phys.* **31**, 271 (1979).
- 79SCH/SCH**
R. Schulz and A. Schweig, "Elucidation of Thermal Reactions by Variable Temperature Photoelectron Spectral Detection of Reactive Intermediates. The UV Photoelectron Spectra of Transient Fulveneketene, Fulvenethioketene, a Ketoketene, and Thiobenzpropiolactone," *Tetrahedron Lett.* **59** (1979).
- 79SCH/THI**
A. Schweig and W. Thiel, *J. Am. Chem. Soc.* **101**, 4742 (1979).
- 79SCH/THO**
A. Schweig, N. Thon and H. Vermeer, "On the Use of Photoelectron Spectroscopy in Studying Rotational Isomerism in Tetramethyldiphosphine," *J. Am. Chem. Soc.* **101**, 80 (1979).
- 79SID/NIK**
L.N. Sidirov, M.I. Nikitin and V.M. Korobov, "Determination of the Electron Affinity of Platinum Fluorides and Manganese Tetrafluoride by an Effusion Method," *Dokl. Akad. Nauk SSSR, Ser. Khim.* **248**, 1387 (1979).
- 79SMI/LEE**
G.P. Smith, L.C. Lee and P.C. Cosby, "Photodissociation and Photodetachment of Molecular Negative Ions. VII. Ions Formed in CO₂/O₂/H₂O Mixtures, 3500-5300 Å," *J. Chem. Phys.* **71**, 4464 (1979).
- 79SMI/LEE2**
G.P. Smith, L.C. Lee and J.T. Moseley, "Photodissociation and Photodetachment of Molecular Negative Ions. VIII. Nitrogen Oxides and Hydrates, 3500-8250 Å," *J. Chem. Phys.* **71**, 4034 (1979).
- 79STE**
W.V. Steele, *J. Chem. Thermodyn.* **11**, 1185 (1979).
- 79SUN/SVE**
S. Sunner, C. Svenson and A.S. Zelepuga, *J. Chem. Thermodyn.* **11**, 491 (1979).
- 79TER/BUR**
J.K. Terlouw, P.C. Burgers and J.L. Holmes, "Thermochemistry and Generation of Vinylketene," *J. Am. Chem. Soc.* **101**, 225 (1979).
- 79TPIS**
L.V. Gurvich, I.V. Veits, V.A. Medvedev, G.A. Khachkuruzov, V.S. Yungman and G.A. Bergman, et. al., "Termodinamicheskie Svoistva Individual'nykh Veshchestv" (Thermodynamic Properties of Individual Substances); Glushko, V.P., gen. ed., Vol. 2, parts 1 and 2 (1979). *Izdatel'stvo "Nauka" Moscow.*
- 79VAJ/HAR**
J.H. Vajda and A.G. Harrison, *Int. J. Mass Spectrom. Ion Phys.* **30**, 293 (1979).
- 79VIS**
D.S. Viswanath, "Quinoline," *API Monograph 711-79*, American Petroleum Institute, Washington, D.C. (1979).
- 79VIS/WIL**
D.S. Viswanath and R.C. Wilhoit, "Isoquinoline," *API Monograph 712-79*, American Petroleum Institute, Washington, D.C. (1979).

- 79WIB/SQU**
K.B. Wiberg and R.R. Squires, *J. Chem. Thermodyn.* **11**, 773 (1979).
- 79WIB/SQU2**
K.B. Wiberg and R.R. Squires, *J. Am. Chem. Soc.* **101**, 5512 (1979).
- 79WIL/WAT**
S.R. Wilson and I.D. Watson, *J. Chem. Thermodyn.* **11**, 911 (1979).
- 79WOL/HOL**
P. Wolkoff and J.L. Holmes, "Isomeric Cyclic [C₆H₁₀]⁺ Ions. The Energy Barrier to Ring Opening," *Can. J. Chem.* **57**, 348 (1979).
- 79WOR/KOB**
J.C. Wormhoudt and C.E. Kobb, "MS Determination of Negative and Positive Ion Concentrations in Coal Fired MHD Plasmas," in *Proc. 10th Materials Res. Symp. on Characterization of High Temp. Vapors*, NBS Spec. Pub. No. 561/1, 457 (1979).
- 79WU/KUD**
C.H. Wu, H. Kudo and H.R. Ihle, "Thermochemical Properties of Gaseous Li₃O and Li₂O₂," *J. Chem. Phys.* **70**, 1815 (1979).
- 80ADE/BOU**
P. Adeney, W.J. Bouma, L. Radom and W.R. Rodwell, *J. Am. Chem. Soc.* **102**, 4069 (1980).
- 80ALI/STE**
A.S. Alikhanyan, A.V. Steblevskii, V.B. Lazarev, V.T. Kalinnikov, Y.K. Grinberg, E.G. Zhukov, L.M. Agamirova and V.I. Gorgoraki, "Thermodynamic Properties of Gaseous Arsenic Iodides," *Izv. Akad. Nauk SSSR, Neorganisch. Mat.* **16**, 73 (1980).
- 80ALL/YUH**
N.L. Allinger, Y. Yuh and J.T. Sprague, *J. Comput. Chem.* **1**, 30 (1980).
- 80AND/BAL**
G.D. Andrews, J.E. Baldwin and K.E. Gilbert, *J. Org. Chem.* **45**, 1523 (1980).
- 80AND/BIC**
M.V. Andreocci, P. Bicev, C. Cauletti and M.N. Piancastelli, "The Electronic Structure of the π -Systems of Acetylenic Oligomers and Related Substances: An UPS Study of Diphenylbutadiyne and Diphenylbutenyne," *Gazz. Chim. Ital.* **110**, 31 (1980).
- 80AND/BOS**
M.V. Andreocci, M. Bossa, V. di Castro, C. Furlani, G. Mattogno and H. W. Roesky, "Electronic Structure of S₃N₂ Ring Derivatives: A Photoelectron Spectroscopy Study," *Gazz. Chim. Ital.* **110**, 1 (1980).
- 80AND/DEV**
M.V. Andreocci, F.A. Devillanova, C. Furlani, G. Mattogno, G. Verani and R. Zanoni, "Structural Characterization of Some Substituted Azolidine Molecules: UPS Photoelectron Spectroscopic Studies," *J. Molec. Str.* **69**, 151 (1980).
- 80ANN/STO**
B.K. Annis and J.A.D. Stockdale, "Internal Excitation of UF₆⁻ Ions in Collisions with Argon Atoms," *Chem. Phys. Lett.* **74**, 365 (1980).
- 80ARM/BEA**
P.B. Armentrout and J.L. Beauchamp, "Endothermic Reactions of Ni⁺ with H₂, HD and D₂," *Chem. Phys.* **50**, 37 (1980).
- 80ARN/PIE**
E.M. Arnett and N.J. Pienta, *J. Am. Chem. Soc.* **102**, 3239 (1980).
- 80ARS**
M.R. Arshadi, *J. Chem. Thermodyn.* **12**, 903 (1980).
- 80BAC/MOU**
M. Bachiri, G. Mouvier, P. Carlier and J.E. Dubois, "Evaluation Quantitative des Effets de Substituants sur les Premiers Potentiels d'Ionisation de Composés Monofonctionnels Aliphatiques," *J. Chim. Phys.* **77**, 899 (1980).
- 80BAE**
T. Baer, "Gas Phase Heats of Formation of C₂H₅⁺ and C₃H₇⁺," *J. Am. Chem. Soc.* **102**, 2482 (1980).
- 80BAL/LEB**
A.A. Balepin, V.P. Lebedev, A.A. Kuznetsova, K. Venters, M. Trusule, D. Lola and Y.A. Lebedev, *Izv. Akad. Nauk. SSSR, Ser. Khim.* **848** (1980).
- 80BAL/NIK**
A.V. Baluev, I.M. Nikitin, L.I. Fedorova and V.Ya. Rossolovskii, "Mass Spectrometric Study of Chlorine Pentafluoride Ionization by Electron Impact," *Izv. Akad. Nauk SSSR, Ser. Khim.*, 497 (1980).
- 80BAL/NIK2**
A.V. Baluev, Z.K. Nikitina, L.I. Fedorova and V.Ya. Rossolovskii, "Mass Spectrometric Investigation of Electron-Impact Induced Ionization of Chlorine Dioxide and Chloryl Fluoride Molecules," *Izv. Akad. Nauk. SSSR, Ser. Khim.* **1963** (1980).
- 80BAL/PIA**
G. Balducci and V. Piacente, "Dissociation Energy of the Tl₂(g) Molecule," *J. Chem. Soc. Chem. Comm.* **1287** (1980).
- 80BAR**
J.E. Bartmess, "Solvent Effects on Ion-Molecule Reactions. Vinyl Anions vs. Conjugate Addition," *J. Am. Chem. Soc.* **102**, 2483 (1980).
- 80BAR/STR**
A. Barkovich, E.S. Strauss and K.P.C. Vollhardt, *Isr. J. Chem.* **20**, 225 (1980).
- 80BEN/NAN**
S.W. Benson and P.S. Nangia, "Electron Affinity of HO₂⁻ and HO_x Radicals," *J. Am. Chem. Soc.* **102**, 2843 (1980).
- 80BOC/AYG**
H. Bock, S. Aygen, P. Rosmus and B. Solouki, "Analyse und Optimierung von Gasphasen-Reaktionen. XVII. Selenoketen," *Chem. Ber.* **113**, 3187 (1980).
- 80BOC/KAI**
H. Bock, W. Kaim, P.L. Timms and P. Hawker, "Durosemiquinone and its BF Analogue - Detection of 1,4-Diborane as an Unexpected Elimination Product," *Chem. Ber.* **113**, 3196 (1980).
- 80BOC/STE**
H. Bock, U. Stein and A. Semkov, "1,2-Dithiolan - Bindungsmodell für α -Liponsäure," *Chem. Ber.* **113**, 3208 (1980).
- 80BOH/CAR**
M.C. Bohm, R.V.C. Carr, R. Gleiter and L.A. Paquette, "Electronic Control of Stereoselectivity. 4. Effects of Neighboring Fused Bicyclic Frameworks on the Stereochemical Outcome of Diels-Alder Cycloadditions to Cyclopentadiene Rings," *J. Am. Chem. Soc.* **102**, 7218 (1980).
- 80BOH/DAU**
M.C. Bohm, J. Daub, R. Gleiter, P. Hofmann, M.F. Lappert and K. Ofele, "Die He(I) Photoelektronenspektren von Tetracarbonylisen(O)-Komplexen mit Carbenen," *Chem. Ber.* **113**, 3629 (1980).
- 80BOH/GLE**
M.C. Bohm and R. Gleiter, "The Electronic Structure and the He(I) Photoelectron Spectrum of Tricarbonylcyclooctatetraene-iron," *Z. Naturforsch.* **35b**, 1028 (1980).
- 80BOU/RAD**
W.J. Bouma, L. Radom and W.R. Rodwell, *Theor. Chim. Acta* **56**, 149 (1980).
- 80BRU/COT**
A. Brunot, M. Cottin, M.H. Donnart and J.C. Muller, "Mesure de L'electroaffinite du Thallium et de l'Indium par Attachement Electronique Dissociatif sur les Bromures et les Iodures de Thallium et d'Indium," *Int. J. Mass Spectrom. Ion Phys.* **33**, 417 (1980).

- 80CAR/COP**
T.A. Carslon, J. Copley, N. Duric, N. Elander, Perman, M. Larsson and M. Lyyra, "The Oscillator Strengths and the Dissociation Energy of SiH^+ as Determined From Time Resolved Precision Spectroscopy," *Astron. Astrophys.* **83**, 238 (1980).
- 80CLA/MCM**
R.L. Clair and T.B. McMahon, "An Ion Cyclotron Resonance Study of Base-Induced Elimination Reactions of Fluorinated Alcohols and Unimolecular Loss of HF from Chemically Activated Fluoroalkoxide Ions," *Int. J. Mass Spectrom. Ion Phys.* **33**, 21 (1980).
- 80COM/REI**
R.W. Compton and P.W. Reinhardt, "Reactions of Fast Cesium Atoms with Polymers of Antimony Pentafluoride and Gold Pentafluoride," *J. Chem. Phys.* **72**, 4655 (1980).
- 80CON**
J.A. Connor, "Transition Metal Clusters" (ed. B.F.G. Johnson, John Wiley & Sons, New York, 1980).
- 80CRA/SNY**
D.C. Crans and J.P. Snyder, *Chem. Ber.* **113**, 1201 (1980).
- 80DAV/FIN**
R.H. Davies, A. Finch and P.J. Gardner, *J. Chem. Thermodyn.* **12**, 291 (1980).
- 80DAW/NIB**
J.H.J. Dawson and N.M.M. Nibbering, "The Gas Phase Anionic Chemistry of Saturated and Unsaturated Aliphatic Nitriles," *Int. J. Mass Spectrom. Ion Phys.* **33**, 3 (1980).
- 80DEF/MCI**
D.J. DeFrees, R.T. McIver, Jr. and W.J. Hehre, "Heats of Formation of Gaseous Free Radicals via Ion Cyclotron Double Resonance Spectroscopy," *J. Am. Chem. Soc.* **102**, 3334 (1980).
- 80DEL/HUB**
J. Delwiche, M.-J. Hubin-Franskin, G. Caprace, P. Natalis and D. Roy, "On the He(I) and Ne(I) Photoelectron Spectra of OCS," *J. Electron Spectrosc. Rel. Phenom.* **21**, 205 (1980).
- 80DEM/WUL**
R.L. Deming and C.A. Wulff *The Chemistry of Ketenes, Allenes and Related Compounds*, S. Patai, Editor, part 1, pp. 155-164 J. Wiley and Sons, New York (1980).
- 80DEP/BIE**
C.H. DePuy, V.M. Bierbaum, L.A. Flippin, J.J. Brabowski, G.K. King, R.J. Schmidt and S.A. Sullivan, "Gas Phase Reactions of Anions with Substituted Silanes," *J. Am. Chem. Soc.* **102**, 5012 (1980).
- 80DEW/DAV**
M.J.S. Dewar and D.E. David, "Ultraviolet Photoelectron Spectrum of the Phenoxy Radical," *J. Am. Chem. Soc.* **102**, 7387 (1980).
- 80DUP/ROB**
P. Dupuis, R. Roberge and C. Sandorfy, "The Very Low Ionization Potentials of Porphyrins and the Possible Role of Rydberg States in Photosynthesis," *Chem. Phys. Lett.* **75**, 434 (1980).
- 80DYK/FAY**
J.M. Dyke, N.K. Fayad, G.D. Josland and A. Morris, "Study by High-temperature Photoelectron Spectroscopy of the Electronic Structure of the Transition Metal Difluorides, CuF_2 and ZnF_2 ," *J. Chem. Soc. Faraday Trans. II* **76**, 1672 (1980).
- 80DYK/JON**
J.M. Dyke, N. Jonathan, A. Morris and M.J. Winter, "The First Ionization Potential of the Formyl Radical, HCO (X^2A), Studied Using Photoelectron Spectroscopy," *Molec. Phys.* **39**, 629 (1980).
- 80EME/KOM**
D.P. Emerick, L. Komorowski, J. Lipinski, F.C. Nahm and K. Niedenzu, "Experimental and Theoretical Studies on Monomeric Iminoboranes," *Z. Anorg. Allg. Chem.* **468**, 44 (1980).
- 80ENG**
P.S. Engel, *Chem. Rev.* **80**, 99 (1980).
- 80FAR/MCM**
R. Farid and T.B. McMahon, "The Gas Phase Acidities of Fluorinated Acetones. An ICR Investigation of the Role of Fluorine Substituents in the Stabilization of Planar Carbanions," *Can. J. Chem.* **58**, 2307 (1980).
- 80FRO/WES**
D.C. Frost, N.P.C. Westwood and N.H. Werstiuk, "Ultraviolet Photoelectron Spectra of 2-Norbornanone, 2,5-Norbornanedione, Their Alkyl Derivatives and Thio-Analogues. An Investigation of Transannular Interactions by Photoelectron Spectroscopy," *Can. J. Chem.* **58**, 1659 (1980).
- 80FUC/PEA**
R. Fuchs and L.A. Peacock, *Can. J. Chem.* **58**, 2796 (1980).
- 80GAJ**
J.J. Gajewski, "Hydrocarbon Thermal Rearrangements" (Academic Press, New York, 1980).
- 80GLE/HOP**
R. Gleiter, H. Hopf, M. Eckert-Maksic and K.-L. Noble, "Photoelektronenspektren von [8]Paracyclophan und [8]Paracyclophan-4-en. Eine Ermittlung des induktiven und hyperkonjugativen Effekts für [n]Paracyclophane," *Chem. Ber.* **113**, 3401 (1980).
- 80GOF/YAR**
M.M. Gofman, V.G. Yarzhemsky and V.I. Nefedov, "Relative Intensities in the He(I) and He(II) Photoelectron Spectra of Benzoyl Chloride," *J. Electron Spectrosc. Rel. Phenom.* **21**, 171 (1980).
- 80GOM/HAA**
W. Gombler, A. Haas and H. Willner, "Chalkogenfluoride in niedrigen Oxydationsstufen. V. Die ungewöhnlichen chemischen Gleichgewichte $\text{F}_3\text{S-SF} = 2 \text{SF}_2$ und $\text{CF}_3\text{SF}_2\text{-SCF}_3 = 2 \text{CF}_3\text{SF}$," *Z. Anorg. Allg. Chem.* **469**, 135 (1980).
- 80GOR**
M.S. Gordon, *J. Am. Chem. Soc.* **102**, 7419 (1980).
- 80GRE**
D.W. Green, *Int. J. Thermophys.* **1**, 61 (1980).
- 80GRE/SED**
J.C. Green and E.A. Seddon, "UV Photoelectron Studies of $\text{Cr}(\eta\text{-C}_3\text{H}_5)_3$, $\text{Cr}_2(\eta\text{-C}_3\text{H}_5)_4$ and $\text{Mo}_2(\eta\text{-C}_3\text{H}_5)_4$," *J. Organometall. Chem.* **198**, C61 (1980).
- 80GRU/SCH**
H.-F. Grutzmacher, B. Schaldach, R. Schubert and D.V. Ramana, "Ion Kinetic Energy Release as a Transition State Probe in Intramolecular Aromatic Substitutions," *Adv. Mass Spectrom.* **8A**, 795 (1980).
- 80GUS/PYA**
A.V. Gusarov, A.T. Pyatenko and L.N. Gorokhov, "Tetrafluoroaluminate (AlF_4^-) and Fluoroaluminate (Al_2F_7^-) Ions in Aluminum Fluoride Vapors," *Teplofiz. Vys. Temperatur.* **18**, 961. CA: **94**, 10584y (1980).
- 80HIL/VES**
J.F. Hiller and M.L. Vestal, "Tandem Quadrupole Study of Laser Photodissociation of CO_3^- ," *J. Chem. Phys.* **72**, 4713 (1980).
- 80HOD/SUL**
R.V. Hodges, S.A. Sullivan and J.L. Beauchamp, "Nucleophilic Reactions of Anions with Trimethyl Phosphate in the Gas Phase by ICR Spectroscopy," *J. Am. Chem. Soc.* **102**, 935 (1980).
- 80HOL/LOS**
J.L. Holmes and F.P. Lossing, *J. Am. Chem. Soc.* **102**, 1591 (1980).
- 80HOL/LOS2**
J.L. Holmes and F.P. Lossing, *J. Am. Chem. Soc.* **102**, 3732 (1980).
- 80HOL/TER**
J.L. Holmes, J.K. Terlouw and P.C. Burgers, *Org. Mass Spectrom.* **15**, 140 (1980).
- 80HOT/NEI**
A. Hotzel, R. Neidlein, R. Schulz and A. Schweig, "Direct Detection of Dicyanoketene in the Gas Phase," *Angew. Chem. Int. Ed.* **19**, 739 (1980).

- 80HOU/SCH**
R. Houriet, H. Schwarz and W. Zummack, "Proton and Hydrogen Affinity of Furan and the Site of Protonation in the Gas Phase," *Angew. Chem. Int. Ed.* **19**, 905 (1980).
- 80HUN/SET**
D.F. Hunt and S.K. Sethi, "Gas Phase Ion/Molecule Isotope Exchange Reactions: Methodology for Counting Hydrogen Atoms in Specific Organic Structural Environments by Chemical Ionization Mass Spectrometry," *J. Am. Chem. Soc.* **102**, 6953 (1980).
- 80ING/HAN**
M.G. Inghram, G.R. Hanson and R. Stockbauer, "The Fragmentation of C_2F_6 ," *Int. J. Mass Spectrom. Ion Phys.* **33**, 253 (1980).
- 80JAN/BRA**
B.K. Janousek and J.I. Brauman, "Electron Photodetachment of Thiomethoxyl and Deuteriothiomethoxyl Anions: Electron Affinities, Vibrational Frequencies, and Spin-Orbit Splitting in CH_3S^- and CD_3S^- ," *J. Chem. Phys.* **72**, 694 (1980).
- 80JAN/REE**
B.K. Janousek, K.J. Reed and J.I. Brauman, "Electron Photodetachment from Mercaptyl Anions (RS^-): Electron Affinities of Mercaptyl Radicals and the S-H Bond Strength in Mercaptans," *J. Am. Chem. Soc.* **102**, 3125 (1980).
- 80JON/MEA**
P.L. Jones, R.D. Mead, B.E. Kohler, S.D. Rosner and W.C. Lineberger, "Photodetachment Spectroscopy of C_2^- Autodetaching Resonances," *J. Chem. Phys.* **73**, 4419 (1980).
- 80JOR/CAR**
F.S. Jorgensen, L. Carlsen and F. Duus, "The Electronic Structure of Propyl 3-Mercaptoacetate Studied by Photoelectron Spectroscopy," *Acta Chem. Scand.* **B34**, 695 (1980).
- 80KAI/TES**
W. Kaim, H. Tesmann and H. Bock, " Me_3C^- , Me_3Si^- , Me_3Ge^- , Me_3Sn^- und Me_3Pb^- -substituierte Benzol- und Naphthalin-Derivate und ihre Radikalanionen," *Chem. Ber.* **113**, 3221 (1980).
- 80KEE/CAS**
R.G. Keese and A.W. Castleman, Jr., "Heats of Formation of SO_2Cl^- and $(SO_2)_2Cl^-$," *J. Am. Chem. Soc.* **102**, 1446 (1980).
- 80KEE/CAS2**
R.G. Keese and A.W. Castleman, Jr., "Gas Phase Studies of Hydration Complexes of Cl^- and I^- and Comparison to Electrostatic Calculations in the Gas Phase," *Chem. Phys. Lett.* **74**, 139 (1980).
- 80KEE/LEE**
R.G. Keese, N. Lee and A.W. Castleman, Jr., "Properties of Clusters in the Gas Phase: V. Complexes of Neutral Molecules onto Negative Ions," *J. Chem. Phys.* **73**, 2195 (1980).
- 80KLA/BUT**
L. Klasinc, V. Butkovic, I. Novak, M. Mihalic, R. Toso and V. Sunjic, "Application of Photoelectron Spectroscopy to Biologically Active Molecules and Their Constituent Parts. VII. N-Cyanoazomethines," *Gazz. Chim. Ital.* **110**, 287 (1980).
- 80KOL**
H. Kollmar, *J. Am. Chem. Soc.* **102**, 2617 (1980).
- 80KOP/COM**
I. Koppel and M.B. Comisarow, "Ab Initio SCF LCAO MO Calculations of Molecules. I. Calculation of Proton Affinities. General Comparison with Experiment," *Organic Reactivity (Tartu State University)*, Engl. Ed. **XVII**, 495 (1980).
- 80KRA**
M.O. Krause, "Photoionization of Atomic Silver between 17 and 41 eV," *J. Chem. Phys.* **72**, 6474 (1980).
- 80KUD/KUD**
A.P. Kudchadker and S.A. Kudchadker, "Indan and Indenes" *API Monograph 714-80* (American Petroleum Institute, Washington, DC, 1980).
- 80LEB/MAS**
N.D. Lebedeva, T.N. Masalitinova, V.L. Ryadnenko and O.N. Mon'yakova, *Zh. Prikl. Khim.* **53**, 2444 (1980).
- 80LEE/KEE**
N. Lee, R.G. Keese and A.W. Castleman, Jr., "The Properties of Clusters in the Gas Phase. IV. Complexes of H_2O and HNO_x Clustering on NO_x^- ," *J. Chem. Phys.* **72**, 1089 (1980).
- 80LEU/HEI**
W. Leupin, E. Heilbronner and J. Wirz, "The Photoelectron Spectrum of 2-Methylbenzotriazole," *J. Molec. Str.* **68**, 329 (1980).
- 80LLO/AGO**
J.R. Lloyd, W.C. Agosta and F.H. Field, "Gaseous Anion Chemistry. Hydrogen-Deuterium Exchange in Mono- and Dialcohol Alkoxide Ions: Ionization Reactions in Dialcohols," *J. Org. Chem.* **45**, 3483 (1980).
- 80MAI/THO**
J.P. Maier and P. Thommen, "Radiative and Nonradiative Decay Rates of State Selected $H-(C\equiv C)-_2H^+$, $D-(C\equiv C)-_2D^+$, $A^2\Pi_u$, Determined by a Photoelectron-Photoion Coincidence Technique," *J. Chem. Phys.* **73**, 5616 (1980).
- 80MAJ/SVA**
V. Majer, L. Svab and V. Svoboda, *J. Chem. Thermodyn.* **12**, 843 (1980).
- 80MAJ/WAG**
V. Majer, Z. Wagner, V. Svoboda and V. Cadek, *J. Chem. Thermodyn.* **12**, 387 (1980).
- 80MAL/ALI**
I.P. Malkerova, A.S. Alikhanyan and V.I. Gorgoraki, "Heats of Formation of Molybdenum Sulphide Fluorides $MoSF_3$ and $MoSF_4$," *Zh. Neorganich. Khim.* **25**, 3181 (1980); *Engl. Trans., Russ. J. Inorg. Chem.* **25**, 1742 (1980).
- 80MAL/ALI2**
I.P. Malkerova, A.S. Alikhanyan, V.S. Pervov, V.D. Butskii and V.I. Gorgoraki, "Composition of the Gas Phase over Molybdenum Thiofluoride $MoSF_3$ [Molybdenum Sulphide Trifluoride]," *Zh. Neorganich. Khim.* **25**, 2067 (1980); *Engl. Trans., Russ. J. Inorg. Chem.* **25**, 1145 (1980).
- 80MAR/HEL**
H.-D. Martin, G. Heller, B. Mayer and H.-D. Beckhaus, *Chem. Ber.* **113**, 258 (1980).
- 80MAU**
M. Meot-Ner (Mautner), "Ion Thermochemistry of Low Volatility Compounds in the Gas Phase. 3. Polycyclic Aromatics: Ionization Energies, Proton, and Hydrogen Affinities. Extrapolations to Graphite," *J. Phys. Chem.* **84**, 2716 (1980).
- 80MCD/CHO**
R.N. McDonald, A.K. Chowdhury and D.W. Setser, "Gas Phase Generation of Phenylnitrene Anion Radical: Proton Affinity and Heat of Formation of PhN^- and Its Clustering with ROH Molecules," *J. Am. Chem. Soc.* **103**, 6599 (1981).
- 80MCD/CHO2**
R.N. McDonald, A.K. Chowdhury and D.W. Setser, "Hypovalent Radicals. 4. Gas Phase Studies of the Ion-Molecule Reactions of Cyclopentadienylidene Anion Radical in a Flowing Afterglow," *J. Am. Chem. Soc.* **102**, 6491 (1980).
- 80MCL/MCG**
F.W. McLafferty and D.C. McGilvery, "Gaseous HCN^+ , HNC^+ and $HCNH^+$ Ions," *J. Am. Chem. Soc.* **102**, 6189 (1980).
- 80MEI/HSI**
G.G. Meisels, T. Hsieh, and J.P. Gilman, "Ion Fragmentation from Noninterconverting Electronic States," *J. Chem. Phys.* **73**, 4126 (1980).

- 80MIK/ZA1**
A.I. Mikaya and V.G. Zaikin, "Determination of the Difference in Enthalpies of Formation of the cis- and trans-Isomers of Bicyclo[4.3.0]Nonane and Bicyclo[4.4.0]Decane Using Appearance Potentials," *Izvest. Akad. Nauk SSSR, Ser. Khim.* 6, 1286 (1980).
- 80MOF**
J.B. Moffat, *J. Mol. Struct.* 62, 213 (1980).
- 80MUR**
E. Murad, "Thermochemical Properties of Gaseous FeO and FeOH," *J. Chem. Phys.* 73, 1381 (1980).
- 80MUR/HIL**
E. Murad and D.L. Hildenbrand, "Dissociation Energies of GdO, HoO, ErO, TmO, and LuO. Correlation of Results for the Lanthanide Monoxide Series," *J. Chem. Phys.* 73, 4005 (1980).
- 80NAN/BEN**
P.S. Nangia and S.W. Benson, *J. Am. Chem. Soc.* 102, 3105 (1980).
- 80NEL/KES**
S.F. Nelsen, C.R. Kessel, L.A. Grezzo and D.J. Steffek, "Thermodynamic Destabilization of N-Centered Radical Cations by a γ -Keto Group," *J. Am. Chem. Soc.* 102, 5482 (1980).
- 80NIK/SOR**
M.I. Nikitin, I.D. Sorokin, E.V. Skokan and L.N. Sidirov, "Negative Ions in the Saturated Vapors of the Potassium Fluoride - Hafnium Tetrafluoride and Potassium Fluoride - Beryllium Difluoride Systems," *Russ. J. Phys. Chem.* 54, Russ:1337 (1980).
- 80NIS/SAK**
K. Nishiyama, M. Sakiyama, S. Seki, H. Horita, T. Otsubo and S. Misumi, *Bull. Chem. Soc. Jpn.* 53, 869 (1980).
- 80NOE/NIB**
A.J. Noest and N.M.M. Nibbering, "Homoconjugation vs. Charge Dipole Stabilization Interaction Effects in the Stabilization of Carbanions in the Gas Phase," *J. Am. Chem. Soc.* 102, 6427 (1980).
- 80PAL/NIS**
M.H. Palmer and J.D. Nisbet, "The Molecular and Electronic Structure of Homoaromatic Compounds: cis,cis,cis-Cyclonona-1,4,7-triene and 1,4,7-Trioxonin; A Study by Photoelectron Spectroscopy and Ab Initio Molecular Orbital Methods," *J. Molec. Str.* 67, 65 (1980).
- 80POL/HEH**
S.K. Pollack and W.J. Hehre, *Tetrahedron Lett.* 21, 2483 (1980).
- 80PRY**
W.A. Pryor, "Frontiers of Free Radical Chemistry," Academic Press, NY, 1980.
- 80PYA/GUS**
A.T. Pyatenko, A.V. Gusarov and L.N. Gorokhov, "Thermochemical Properties of Negative Ions in Vapor over UF₄," *Teplofiz. Vys. Temperatur.* 18, 1154. CA: 92, 7246p (1980).
- 80RED/TRE**
V.V. Redchenko, Y.F. Freimanis and Y.Y. Dregeris, "Photoelectron Spectroscopy of 2,3-Disubstituted Naphthoquinones," *Zh. Obsh. Khim.* 50, 1847 (1980); English trans.: *J. Gen. Chem. USSR* 50, 1507 (1980).
- 80ROG/CHO**
D.W. Rogers, L.S. Choi, R.S. Giirellini, T.J. Holmes and N.L. Allinger, *J. Phys. Chem.* 84, 1810 (1980).
- 80ROS/STO**
H.M. Rosenstock, R. Stockbauer and A.C. Parr, "Photoelectron-photoion Coincidence Study of the Bromobenzene Ion," *J. Chem. Phys.* 73, 773 (1980).
- 80ROS/STO2**
H.M. Rosenstock, R. Stockbauer and A.C. Parr, "Photoelectron-photoion Coincidence Study of Benzonitrile," *J. Chim. Phys.* 77, 745 (1980).
- 80ROT/KLA**
W.R. Roth, F.-G. Klarnner and H.W. Lennart, *Chem. Ber.* 113, 1818 (1980).
- 80ROT/MAT**
E.W. Rothe, B.P. Mathur and G.P. Reck, "Measurement of Boron Trihalide Electron Affinities: Correlation with Boron-Nitrogen Adduct Strengths," *Inorg. Chem.* 19, 829 (1980).
- 80SAB**
R. Sabbah, *Thermochim. Acta* 41, 33 (1980).
- 80SAB2**
R. Sabbah, *Thermochim. Acta* 35, 73 (1980).
- 80SAB/SKO**
R. Sabbah and S. Skoulika, *Thermochim. Acta* 36, 179 (1980).
- 80SAR/WOR**
R. Sarneel, C.W. Worrell, P. Pasman, J.W. Verhoeven and G.F. Mes, "The Photoelectron Spectra of 4-Methylene Thiacyclohexane Derivatives Through-Bond Interaction," *Tetrahedron* 36, 3241 (1980).
- 80SAT/SAK**
T. Sato-Toshima, M. Sakiyama and S. Seki, *Bull. Chem. Soc. Jpn.* 53, 2462 (1980).
- 80SCH/ILL**
H.U. Scheunemann, E. Illenberger and H. Baumgartel, "Dissociative Electron Attachment to CCl₄, CHCl₃, CH₂Cl₂, and CH₃Cl," *Ber. Bunsenges Phys. Chem.* 84, 580 (1980).
- 80SCH/RAM**
R. Schubert, D.V. Ramana and H.-F. Grutzmacher, "Freisetzung kinetischer Energie und Hammond-Postulat bei der intramolekularen aromatischen Substitution in 2-Stilbazol-Ionen," *Chem. Ber.* 113, 3758 (1980).
- 80SCH/SCH**
R. Schulz, A. Schweig, C. Wentrup and H.-W. Winter, "2-Vinylidene-2H-indene," *Angew. Chem. Int. Ed.* 19, 821 (1980).
- 80SCH/SCH2**
R. Schulz and A. Schweig, "Direct Detection of Dicyanothioetene in the Gas Phase," *Angew. Chem. Int. Ed.* 19, 740 (1980).
- 80SCH/THI**
A. Schweig and W. Thiel, *J. Comput. Chem.* 1, 129 (1980).
- 80SCH/THO**
A. Schweig, N. Thon, S.F. Nelsen and L.A. Grezzo, "Conformational Study of 1,2-Dimethylhexahydropyridazine by Variable-Temperature Photoelectron Spectroscopy," *J. Am. Chem. Soc.* 102, 7438 (1980).
- 80SEN/ABE**
S.N. Senzer, R.N. Abernathy and F.W. Lampe, "GeH₅⁺ and the Proton Affinity of Monogermane," *J. Phys. Chem.* 84, 3066 (1980).
- 80SHU/BOY**
B. Shushan and R.K. Boyd, "Unimolecular and Collision Induced Fragmentations of Molecular Ions of Polycyclic Aromatic Hydrocarbons," *Org. Mass Spectrom.* 15, 445 (1980).
- 80SID/NIK**
L.N. Sidorov, M.I. Nikitin, E.V. Skokan and I.D. Sorokin, "Mass-Spectrometric Determination of Enthalpies of Dissociation of Gaseous Complex Fluorides into Neutral and Charged Particles. II. Heats of Formation of AlF₄⁻ and KF₂⁻," *Int. J. Mass Spectrom. Ion Phys.* 35, 203 (1980).
- 80SID/SKO**
L.N. Sidorov, E.V. Skokan, M.I. Nitikin and I.D. Sorokin, "Mass Spectrometric Determination of Enthalpies of Dissociation of Gaseous Complex Fluorides into Neutral and Charged Particles. III. Heat of Formation of UF₅⁻ and Electron Affinity of UF₅⁻," *Int. J. Mass Spectrom. Ion Phys.* 35, 215 (1980).
- 80STA/VOG**
J.P. Stadelmann and J. Vogt, *Int. J. Mass Spectrom. Ion Phys.* 35, 83 (1980).
- 80SIR/NEW**
G.E. Streit and T.W. Newton, "Negative Ion-Uranium Hexafluoride Charge Transfer Reactions," *J. Chem. Phys.* 73, 3178 (1980).

- 80SVO/UCH**
V. Svoboda, V. Uchytlova, V. Majer and J. Pick, Collect. Czech. Chem. Commun. 45, 3233 (1980).
- 80TED/VID**
J.M. Tedder and P.H. Vidaud, "Charge Exchange Mass Spectra of Thiophene, Pyrrole and Furan," J. Chem. Soc. Faraday Trans. II 76, 1516 (1980).
- 80TEL/RAB**
V.I. Tel'noi and I.B. Rabinovich, Russ. Chem. Rev. 49, 603 (1980).
- 80TEP/YAN**
A.B. Teplitsky, I.K. Yanson, O.T. Glukhova, A. Zielenkiewicz, W. Zielenkiewicz and K.L. Wierzchowski, Biophys. Chem. 11, 17 (1980).
- 80TER/HEE**
J.K. Terlouw, W. Heerma, J.L. Holmes and P.C. Burgers, "Structure and Formation of Gaseous $[C_4H_6O]^+$ Ions. 1-The Enolic Ions $[CH_2=C(OH)-CH=CH_2]^+$ and $[CH_2=CH-CH=CH(OH)]^+$ and Their Relationship with Their Keto Counterparts," Org. Mass Spectrom. 15, 582 (1980).
- 80THA/EIS**
M.G. Thackston, F.L. Eisele, W.M. Pope, H.W. Ellis, E.W. McDaniel and I.R. Gatland, "Mobility of Cl^- Ions in Xe Gas and the Cl^- -Xe Interaction Potential," J. Chem. Phys. 73, 3183 (1980).
- 80TRA**
J.C. Traeger, Int. J. Mass Spectrom. Ion Phys. 32, 309 (1980).
- 80TRE**
A.B. Trentwith, J. Chem. Soc. Trans. Farad. I 76, 166 (1980).
- 80TSV/ALE**
V.G. Tsvetkov, V.A. Aleksandrov, V.N. Glushakova, N.A. Skorodumora and G.M. Kol'yakova, Russ. J. Gen. Chem. 50, 198 (1980).
- 80VAN**
J. Van der Greef Ph. D. Thesis, Univ. Amsterdam (1980).
- 80VAN/TER**
H. Van Dam, A. Terpstra, D.J. Stufkens and A. Oskam, "UV Photoelectron Spectroscopic Studies of the Metal-Olefin Bond. 2. Bonding in (β -Diketonato)rhodium(I) and -iridium(I) Carbonyl and Olefin Complexes," Inorg. Chem. 19, 3448 (1980).
- 80VER/SAL**
A.N. Vereshchagin, A.M. Salikhova, V.V. Zverev, F.G. Saitkulova and Y.Y. Villem, "Physical Properties, Conformations and Intramolecular Interactions of α,α -Dichlorodimethyl Ether," Izv. Akad. Nauk SSSR, Ser. Khim. 997 (1980); Engl. trans., Bull. Acad. Sci. USSR, Div. Chem. Sci. 29, 706 (1980).
- 80VIG/PER**
A.A. Vigiano, R.A. Perry, D.L. Albritton, E.E. Ferguson and F.C. Fehsenfeld, "The Role of H_2SO_4 in Stratospheric Negative Ion Chemistry," J. Geophys. Res. 85, 4551 (1980).
- 80VIL/PER**
R. Vilcu and S. Perisanu, Rev. Roum. Chim. 25, 619 (1980).
- 80VON/BIE**
W. Von Niessen, G. Bieri and L. Asbrink, "30.4 nm He(II) Photoelectron Spectra of Organic Molecules. Part III. Oxo-compounds (C, H, O)," J. Electron Spectrosc. Rel. Phenom. 21, 175 (1980).
- 80VOV/DUD**
V.I. Vovna, A.S. Dudin, S.N. Lopatin and E.G. Rakov, "Photoelectronic Spectra of Hexafluorides of Transition Metals with Open Shells (ReF_6 and OsF_6)," Koord. Khim. 6, 1580 (1980).
- 80WAN/TRA**
J.-S. Wang and J.L. Franklin, "Reactions and Energy Distributions in Dissociative Electron Capture Processes in Sulfuryl Halides," Int. J. Mass Spectrom. Ion Phys. 36, 233 (1980).
- 80WIG/BEA**
C.A. Wight and J.L. Beauchamp, "Acidity, Basicity, and Ion/Molecule Reactions of Isocyanic Acid in the Gas Phase by ICR Spectroscopy," J. Phys. Chem. 84, 2503 (1980).
- 80WIL/BAE**
G.D. Willett and T. Baer, J. Am. Chem. Soc. 102, 6774 (1980).
- 80WLO/LUC**
S. Wlodek, Z. Luczynski and H. Wincel, "Stabilities of Gas-Phase $NO_3^-(HNO_3)_n$, $n \leq 6$, Clusters," Int. J. Mass Spectrom. Ion Phys. 35, 39 (1980).
- 80WOL/HOL**
P. Wolkoff, J.L. Holmes and F.P. Lossing, "Ubiquitous [Cyclopentenium] $^+$ Formation from C_6H_{10} Molecular Ions by Methyl Loss, and from Higher Homologues," Adv. Mass Spectrom. 8A, 743 (1980).
- 80ZVE/VIL**
V.V. Zverev and Y.Y. Villem, "Ionization Potentials of Phosphoryl Compounds," Zh. Strukt. Khim. 21, 30 (1980); English trans., J. Struct. Chem. (USSR), 21, 22 (1980).
- 81AJO/CAS**
Flexibility of the Dehydroalanine Derivatives: Molecular and Electronic Structure of (Z)-N-Acetyldehydrophenylalanine," Tetrahedron 37, 3507 (1981).
- 81AJO/CAS2**
D. Ajo, M. Casarin, G. Granozzi and I. Fragala, "UV Photoelectron Spectra of 5- and 6-Azaauracil," Chem. Phys. Lett. 80, 188 (1981).
- 81AKI/HAR**
I. Akiyama, R.G. Harvey and P.R. LeBreton, "Ultraviolet Photoelectron Studies of Methyl-Substituted Benz[a]anthracenes," J. Am. Chem. Soc. 103, 6330 (1981).
- 81ALD/ARR**
R.W. Alder, R.J. Arrowsmith, A. Casson, R.B. Sessions, E. Heilbronner, B. Kovac, H. Huber and M. Taagepera, J. Am. Chem. Soc. 103, 6137 (1981).
- 81ALL/GLA**
N.L. Allinger, J.A. Glaser, H.E. Davis and D.W. Rogers, J. Org. Chem. 46, 658 (1981).
- 81ALL/MIG**
S.H. Allam, M.D. Migahed and A. El Khodary, "Electron Impact Study of Nitrobenzene and Nitromethane," Int. J. Mass Spectrom. Ion Phys. 39, 117 (1981).
- 81AND/DEK**
E.L. Andersen, R.L. DeKock and T.P. Fehlner, "Electronic Structure of Diiron Ferraboranes," Inorg. Chem. 20, 3291 (1981).
- 81ARI/ARM**
N. Aristov and P. Armentrout, "Bond Energy-Bond Order Relations in Transition-Metal Bonds: Vanadium," J. Am. Chem. Soc. 106, 4065 (1984).
- 81ARM/BEA**
P.B. Armentrout and J.L. Beauchamp, "Ion Beam Studies of the Reactions of Atomic Cobalt Ions with Alkanes: Determination of Metal-Hydrogen and Metal-Carbon Bond Energies and an Examination of the Mechanism by which Transition Metals Cleave Carbon-Carbon Bonds," J. Am. Chem. Soc. 103, 784 (1981).
- 81ARM/BEA2**
P. Armentrout and J. Beauchamp, "Cobalt Carbene Ion: Reactions of Co^+ with C_2H_4 , cyclo- C_3H_6 , and cyclo- C_2H_4O ," J. Chem. Phys. 74, 2819 (1981).
- 81ARM/HAL**
P. B. Armentrout, L. F. Halle and J. L. Beauchamp, "Periodic Trends in Transition Metal-Hydrogen, Metal-Carbon, and Metal-Oxygen Bond Dissociation Energies. Correlation with Reactivity and Electronic Structure," J. Am. Chem. Soc. 103, 6501 (1981).

81ARM/TAR

P. B. Armentrout, S. M. Tarr, A. Dori and R. S. Freund, "Electron Impact Ionization Cross Section of Metastable $N_2(A^2\Sigma_u^+)$," *J. Chem. Phys.* **75**, 2788 (1981).

81ARS/ZVE

R. P. Arshinova, V. V. Zverev, Y. Y. Villem and N. V. Villem, "Ionization of Potentials, Electron Structures, and Steric Structures of Tervalent-Phosphorus Di- and Tri-Esters," *Zh. Obs. Khim.* **51**, 1757 (1980); English translation, *J. Gen. Chem. (USSR)* **51**, 1503 (1982).

81ASB/SVE

L. Asbrink, A. Svensson, W. Von Niessen and G. Bieri, "30.4 nm He(II) Photoelectron Spectra of Organic Molecules," *J. Electron Spectrosc. Rel. Phenom.* **24**, 293 (1981).

81AUE/PED

D. Aue, M. Pedley, and M.T. Bowers, unpublished result cited in: A. J. Illies, S. Liu and M. T. Bowers, "Formation and Structure of $C_2H_4N^+$. Effect of Pressure and Reaction Exothermicity on Collision-Induced Dissociation Spectra," *J. Am. Chem. Soc.* **103**, 5674 (1981).

81AUS

P. Ausloos, *J. Am. Chem. Soc.* **103**, 3931 (1981).

81AYA/WEN

J.A. Ayala, W.E. Wentworth and E.C.M. Chen, "Electron Attachment to Halogens," *J. Phys. Chem.* **85**, 768 (1981).

81BAB/STR

L.M. Babcock and G.E. Streit, "Ion-Molecule Reactions of SF_6 : Determination of I. P. (SF_6), A. P. (SF_5^+/SF_6), and D (SF_5-F)," *J. Chem. Phys.* **74**, 5700 (1981).

81BAB/STR2

L.M. Babcock and G.E. Streit, "Negative Ion-Molecule Reactions of SF_4 ," *J. Chem. Phys.* **75**, 3864 (1981).

81BAI/CHI

V.N. Baidin, Y.V. Chizhov, M.M. Timoshenko, Y.A. Ustyniyuk and I.I. Kritskaya, "He(I) Photoelectronic Spectra of Organic Compounds with a Mercury Atom Attached to an sp^2 Hybridized Carbon," *Izv. Akad. Nauk SSSR, Ser. Khim.* **12**, 2831 (1981).

81BAI/CHI2

V.N. Baidin, Y.V. Chizhov, M.M. Timoshenko, O.K. Sokolikova, Y.K. Grishin and Y.A. Ustyniyuk, "The Photoelectron Spectra of Alkylmercury Chlorides," *Zh. Strukt. Khim.* **22**, 164 (1981).

81BAK/ARM

A.D. Baker, G.H. Armen and Y. Guang-di, "Photoelectron Spectra of Alkyl Aryl Selenides. Electronic and Steric Factors in the Observation of Rotamers," *J. Org. Chem.* **46**, 4127 (1981).

81BAL/GIG

G. Balducci, G. Gigli and M. Guido, "Dissociation Energies of the Molecules $CrPO_2(g)$ and $CoO(g)$ by High-temperature Mass Spectrometry," *J. Chem. Soc. Faraday Trans. II* **77**, 1107 (1981).

81BAL/GIG2

G. Balducci, G. Gigli and M. Guido, "Thermodynamic Study of Gaseous Ternary Oxide Molecules. The Europium-Vanadium-Oxygen System," *J. Chem. Phys.* **79**, 5623 (1981).

81BAR/HAY

J.E. Bartmess, R.L. Hays, H.N. Khatri, R.N. Misra and S.W. Wilson, "Elimination, Fragmentation and Proton Transfer in 1,3-Dithianes and 1,3-Dithiolanes in the Gas Phase," *J. Am. Chem. Soc.* **103**, 4746 (1981).

81BEC/HOF

E. Beck, P. Hofmann and A. Sieber, "Zur Elektronenstruktur Vicinaler Triketone: Bicyclo[3.1.0]hexan-2,3,4-trione," *Tetrahedron Lett.* **22**, 4683 (1981).

81BEL/PER

T.N. Bell, K.A. Perkins and P.G. Perkins, *J. Chem. Soc. Faraday Trans. I* **77**, 1779 (1981).

81BER/BEA

D.W. Berman, J.L. Beauchamp and L.R. Thorne, "Ion Cyclotron Resonance and Photoionization Investigations of the Thermochemistry and Reactions of Ions Derived from CF_3I ," *Int. J. Mass Spectrom. Ion Phys.* **39**, 47 (1981).

81BER/BOM

D.W. Berman, D.S. Bomse and J.L. Beauchamp, "Photoionization Threshold Measurements for CF_2 Loss from Perfluoropropylene, Perfluorocyclopropane, and Trifluoromethylbenzene. The Heat of Formation of CF_2 and the Potential Energy Surface for C_3F_6 Neutrals and Ions," *Int. J. Mass Spectrom. Ion Phys.* **39**, 263 (1981).

81BER/GAR

M. Berry, C.D. Garner, I.H. Hillier and A.A. MacDowell, "Electronic Structure and Photoelectron Spectrum of $Tris(\eta\text{-}3\text{-}C_3H_5)_3Cr$," *Inorg. Chem.* **20**, 1962 (1981).

81BEV/SAN

J.W. Bevan, C. Sandorfy, F. Pang and J.E. Bogg, "The Photoelectron Spectrum of Methylisocyanide-borane," *Spectrochimica Acta* **37A**, 601 (1981).

81BIA/LIF

S.E. Biali, C. Lifshitz and Z. Rappoport, "Thermochemistry and Unimolecular Reactions of Ionized 1,2-Dimesityl-2-phenylethanone and 2,2-Dimesityl-1-phenylethanone and Their Enols and Enol Acetates in the Gas Phase," *J. Am. Chem. Soc.* **103**, 2896 (1981).

81BIE/ASB

G. Bieri, L. Asbrink and W. Von Niessen, "30.4 nm He(II) Photoelectron Spectra of Organic Molecules. Part IV. Fluoro-compounds (C, H, F)," *J. Electron Spectrosc. Rel. Phenom.* **23**, 281 (1981).

81BIE/SCH

V.M. Bierbaum, R.J. Schmidt, C.H. DePuy, R.H. Mead, P.A. Schulz and W.C. Lineberger, "Reactions of Carbanions with Triplet and Singlet Molecular Oxygen," *J. Am. Chem. Soc.* **103**, 6262 (1981).

81BIE/VON

G. Bieri, W. Von Niessen, L. Asbrink and A. Svensson, "The He(II) Photoelectron Spectra of the Fluorosubstituted Ethylenes and Their Analysis by the Green's Function Method," *Chem. Phys.* **60**, 61 (1981).

81BIS/COL

P. Biscarini, F.P. Colonna, M. Guerra and G. Distefano, "Mercury-Sulphur Bonding in Some Di(alkylthio)mercury(II) Compounds Studied by Means of Ultraviolet Photoelectron Spectroscopy," *Inorg. Chim. Acta* **50**, 243 (1981).

81BIS/GLE

P. Bischof, R. Gleiter, K. Gubernator, R. Haider, H. Musso, W. Schwarz, W. Trautmann and H. Hopf, "Photoelectronen-spektroskopische Untersuchungen an Divinylcyclobutanen," *Chem. Ber.* **114**, 994 (1981).

81BOC/BRA

H. Bock, G. Brahler, D. Dauplaise and J. Meinwald, "One-Electron Oxidation of 1,8-Chalcogen-Bridged Naphthalenes," *Chem. Ber.* **114**, 2622 (1981).

81BOC/DAM

H. Bock, R. Dammel and L. Horner, "Die Pyrolyse von Methylazid," *Chem. Ber.* **114**, 220 (1981).

81BOC/ECK

M.C. Bohm, M. Eckert-Maksic, R. Gleiter, J. Grobe and D. Le Van, "Die He(I)-Photoelektronenspektren von $(CH_3)_2PSCH_3$, $(CH_3)_2AsSCH_3$, $(CH_3)_2AsSeCH_3$ und $(CH_3)_2PSeCH_3$," *Chem. Ber.* **114**, 2300 (1981).

81BOC/HIR

H. Bock, T. Hirabayashi and S. Mohmand, "Thermische Erzeugung von Alkyl- und Halogenketenen," *Chem. Ber.* **114**, 2595 (1981).

81BOC/RIE

H. Bock, W. Ried and U. Stein, "Analyse und Optimierung von Gasphasen-Reaktionen, 19. Pyrolyse von Cyclobuten-1,2-dionen zu Acetylenen," *Chem. Ber.* **114**, 673 (1981).

81BOC/SCH

H. Bock, W. Schulz and M. Schmidt, "P. E. Spektren und Molekuleigenschaften. 89. Ionisationsmuster und Konformation von Cyclo-Polythianen (H₂CS)_n," *Z. Anorg. Allg. Chem.* **474**, 199 (1981).

81BOC/SCH2

H. Bock, W. Schulz and U. Stein, "Radikationen, 47. Notiz: Die Strukturänderung während der Enelektronen-Oxidation von Bis(dimethylamino)disulfid," *Chem. Ber.* **114**, 2632 (1981).

81BOH/ECK

M.C. Bohm, M. Eckert-Maksic, R. Gleiter, J. Grobe and D. Le Van, "Die He(I)-Photoelektronenspektren von (CH₃)₂PSCH₃, (CH₃)₂AsSCH₃, (CH₃)₂AsSeCH₃ und (CH₃)₂PSeCH₃," *Chem. Ber.* **114**, 2300 (1981).

81BOM/BER

D.S. Bomse, D.W. Berman and J.L. Beauchamp, *J. Am. Chem. Soc.* **103**, 3967 (1981).

81BOM/DAN

R. Bombach, J. Dannacher, J.-P. Stadelmann and J. Vogt, "The Fragmentation of Formaldehyde Molecular Cations: The Lifetime of CD₂O⁺ (A²B₁)," *Chem. Phys. Lett.* **77**, 399 (1981).

81BOU/DAG

G. Bouchoux and J. Dagaut, "Mechanisms of Formation of [M-HCO]⁺ and [M-C₆H₅CO]⁺ Ions from Isomers of 1,4-Benzodioxin Derivatives," *Org. Mass Spectrom.* **16**, 246 (1981).

81BOU/HOP

G. Bouchoux and Y. Hoppilliard, "Fragmentation Mechanisms of Isoxazole," *Org. Mass Spectrom.* **16**, 459 (1981).

81BRU/CIL

G. Bruno, E. Ciliberto, I. Fragala and G. Granozzi, "The Electronic Structure of Hydrotris(1-pyrazolyl)borate Ligand by He-I and He-II Photoelectron Spectroscopy," *Inorg. Chim. Acta* **48**, 61 (1981).

81BUC/FOR

G. Buckley, W.G.F. Ford and A.S. Rodgers, *Thermochim. Acta* **49**, 199 (1981).

81BUF/PAR

R.D. Buff, A.C. Parr and A.J. Jason, "The Photoionization of Allyl Chloride from Onset to 20 eV," *Int. J. Mass Spectrom. Ion Phys.* **40**, 31 (1981).

81BUR/JEN

B.E. Bursten, J.R. Jensen, D.J. Gordon, P.M. Treichel and R.F. Fenske, "Electronic Structure of Transition-Metal Nitrosyls. X(α)-SW and Configuration Interaction Calculations of the Valence Ionization Potentials of Co(CO)₃NO and Mn(CO)₄NO," *J. Am. Chem. Soc.* **103**, 5226 (1981).

81CAB/COW

D.E. Cabelli, A.H. Cowley and M.J.S. Dewar, "UPE Studies of Conjugation Involving Group 5A Elements. 2. Substituted tert-Butylacetylenes," *J. Am. Chem. Soc.* **103**, 3290 (1981).

81CAB/COW2

D.E. Cabelli, A.H. Cowley and M.J.S. Dewar, "UPE Studies of Conjugation Involving Group 5A Elements. 1. Phenylphosphines," *J. Am. Chem. Soc.* **103**, 3286 (1981).

81CAL/HUB

D.C. Calabro, J.L. Hubbard, C.H. Blevins II, A.C. Campbell and D.L. Lichtenberger, "The Effects of Methyl Group Substitution on Metal-Coordinated Cyclopentadienyl Rings. The Core and Valence Ionizations of Methylated Tricarbonyl(η⁵-cyclopentadienyl)metal Complexes," *J. Am. Chem. Soc.* **103**, 6839 (1981).

81CAL/LIC

D.C. Calabro and D.L. Lichtenberger, "Valence Ionizations of Olefins Coordinated to Metals. Olefin Dicarbonyl(η⁵-(methyl and pentamethyl)cyclopentadienyl)manganese Complexes," *J. Am. Chem. Soc.* **103**, 6846 (1981).

81CAR/GAN

F. Carnovale, R.-H. Gan, J.B. Peel and A.B. Holmes, "Photoelectron Spectroscopic Studies of Some 2-Azabicyclo[2.2.2]octan-5-one and Bicyclo[2.2.2]octanone Derivatives," *J. Chem. Soc. Perkin Trans. II* **7**, 991 (1981).

81CAU/GIA

C. Cauletti, C. Giancaspro, A. Monaci, M.N. Piancastelli, "Free Energy Relationships of Ionization Energies Measured by Ultraviolet Photoelectron Spectroscopy in Substituted Pyrroles," *J. Chem. Soc. Perkin Trans. II* **7**, 656 (1981).

81CAV/CON

K.J. Cavell, J.A. Connor, G. Pilcher, M.A.V. Ribeiro da Silva, D.M.C. Ribeiro da Silva, Y. Vipmani and M.T. Zafarani-Moattar, *J. Chem. Soc. Faraday Trans. I* **77**, 1585 (1981).

81CAV/GAR

K.J. Cavell, C.D. Garner, J.A. Martinho-Simoes, G. Pilcher, H. Al-Samman, H.A. Skinner, G. Al-Tekhin, I.B. Walton and M.T. Zafarani-Moattar, *J. Chem. Soc. Faraday Trans. I* **77**, 2927 (1981).

81CHA/FIN

S. Chattorpadhyay, G.L. Findley and S.P. McGlynn, "Photoelectron Spectroscopy of Phosphites and Phosphates," *J. Electron Spectrosc. Rel. Phenom.* **24**, 27 (1981).

81CHE/GAB

A.P. Chetverikov, V.Y. Gabeskiriya and V.V. Puchkov, "Ionization of Uranium, Plutonium, Americium, Rhenium, Curium, and Californium on a Rhenium Surface," *Sov. Phys. Tech. Phys.* **26**, 73 (1981).

81CHE/HAL

P.T. Chesky and M.B. Hall, "Electronic Structure of Metal Clusters. 1. Photoelectron Spectra and Molecular Orbital Calculations on (Alkylidyne)tricobalt Nonacarbonyl Clusters," *Inorg. Chem.* **20**, 4419 (1981).

81CHI/HYM

J.S. Chickos, A.S. Hyman, L.H. Ladon and J.F. Liebman, *J. Org. Chem.* **46**, 4294 (1981).

81CHO/KIR

D.P. Chong, C. Kirby, W.M. Lau, T. Minato and N.P.C. Westwood, "Difluoroborane, HBF₂. A Study by HeI Photoelectron Spectroscopy, and Ab Initio Methods Including Perturbation Corrections to Koopmans' Theorem," *Chem. Phys.* **59**, 75 (1981).

81CIL/CON

E. Ciliberto, G. Condorelli, P.J. Fagan, J.M. Manriquez, I. Fragala and T. J. Marks, "Photoelectron Spectroscopy of f-Element Organometallic Complexes. 4. Comparative Studies of Bis(pentamethylcyclopentadienyl)Dichloride and Dimethyl Complexes of Uranium(IV), Thorium(IV), and Zirconium(IV)," *J. Am. Chem. Soc.* **103**, 4755 (1981).

81CLA/ROB

E. Clar, J.M. Robertson, R. Schlogl and W. Schmidt, "Photoelectron Spectra of Polynuclear Aromatics. 6. Application to Structural Elucidation: 'Circumanthracene'," *J. Am. Chem. Soc.* **103**, 1320 (1981).

81CLA/SOW

P. Clare and D.B. Sowerby, "Electron Impact Ionisation Energies of Some Halo-cyclotriphosphazenes," *J. Inorg. Nucl. Chem.* **43**, 477 (1981).

81COL/FRO

D. Colbourne, D.C. Frost, C.A. McDowell and N.P.C. Westwood, "The He(I) Photoelectron Spectra of the Alkyl Hypochlorites, ROCl (R = Me, Et and t-Bu)," *J. Electron Spectrosc. Rel. Phenom.* **23**, 109 (1981).

- 81COS/LLO**
N.C.V. Costa, D.R. Lloyd, P.J. Roberts, D.W.J. Cruickshank, E. Avramides, A. Chablo, G.A.D. Collins, B. Dobson and I.H. Hillier, "Experimental and Theoretical Study of the Electronic Structures of Thionyl Fluoride, Sulphur Tetrafluoride and Sulphur Tetrafluoride Oxide," *J. Chem. Soc. Faraday Trans. II* **77**, 899 (1981).
- 81DAB/HER**
I. Dabrowski and G. Herzberg, "The Spectrum of HeAr⁺," *J. Molec. Spectrosc.* **89**, 491 (1981).
- 81DAL/NIB**
J.W. Dallinga, N.M.M. Nibbering and G.J. Louter, "Formation and Structure of [C₈H₈O]⁺ Ions, Generated from Gas Phase Ions of Phenylcyclopropylcarbinol and 1-Phenyl-1-(hydroxymethyl)cyclopropane," *Org. Mass Spectrom.* **16**, 4 (1981).
- 81DEH/POL**
P.M. Dehmer and E.D. Poliakoff, "Photoionization of the Ar₂ Dimer," *Chem. Phys. Lett.* **77**, 326 (1981).
- 81DEP/BIE**
C.H. DePuy and V.M. Bierbaum, "Gas Phase Sulfur Anions: Synthesis and Reactions of H₂NS⁻ and Related Ions," *Tetrahedron Lett.* **22**, 5129 (1981).
- 81DOM/EAT**
L.N. Domelsmith, T.A. Eaton, K.N. Houk, G.M. Anderson III, R.A. Glennon, A.T. Shulgin, N. Castagnoli, Jr. and P.A. Kollman, "Photoelectron Spectra of Psychotropic Drugs. 6. Relationships between Physical Properties and Pharmacological Actions of Amphetamine Analogues," *J. Med. Chem.* **24**, 1414 (1981).
- 81DON/WAL**
A.M. Doncaster and R. Walsh, "Kinetics of the Gas Phase Reaction between Iodine and Monosilane and the Bond Dissociation Energy D(H₃Si-H)," *Int. J. Chem. Kinet.* **13**, 503 (1981).
- 81DRA/GOR**
J. E. Drake and K. Gorzelska, "The Photoelectron Spectra of Methyl(bromo)germanes," *J. Electron Spectrosc. Rel. Phenom.* **21**, 365 (1981).
- 81DRE/BEC**
C. Dreier, J. Becher, E. G. Frandsen and L. Henriksen, "Pyridinethiones - V. Spectroscopic Investigation and Electronic Structure of 3-Formyl-2(1H)-pyridones, -thiones and -selones," *Tetrahedron* **37**, 2663 (1981).
- 81DUM/DUP**
J.-M. Dumas, P. Dupuis, G. Pfister-Guillouzo and C. Sandorfy, "Ionization Potentials and Ultraviolet Absorption Spectra of Fluorocarbon Anesthetics," *Can. J. Spectrosc.* **26**, 102 (1981).
- 81DYK/JON**
J. M. Dyke, N. B. H. Jonathan, A. Morris and M. J. Winters, "Vacuum Ultraviolet Photoelectron Spectroscopy of Transient Species. Part 13. Observation of the X³A' State of HO₂," *Molec. Phys.* **44**, 1059 (1981).
- 81DYK/JON2**
J. Dyke, N. Jonathan, A. Morris and M. Winter, "First Ionization Potential of the FCO (X²A') Radical Studied Using Photoelectron Spectroscopy," *J. Chem. Soc. Faraday Trans. II* **77**, 667 (1981).
- 81EAD/WEI**
R.A. Eades, D.A. Weil, M.R. Ellenberger, D.A. Dixon and C.H. Douglass, Jr., *J. Am. Chem. Soc.* **103**, 5372 (1981).
- 81EIC/HEY**
K. Eichler and H. Heydtmann, *Int. J. Chem. Kinet.* **13**, 1107 (1981).
- 81ELB/DIE**
S. Elbel and H. tom Dieck, "Die Stiborane Me₃SbX₂ (X = Me, Cl, Br, I) und SbCl₅ - im Vergleich mit Nebengruppenanaloga TaMe₅, TaCl₅ und NbCl₅," *Z. Anorg. Allg. Chem.* **483**, 33 (1981).
- 81ELB/LIE**
S. Elbel, K. Lienert, A. Krebs and H. tom Dieck, "Phenylethin - Mustersonde fur Substituenteneffekte," *Liebigs Ann. Chem.* **1785** (1981).
- 81ELL/DIX**
M.R. Ellenberger, D.A. Dixon and W.E. Farneth, *J. Am. Chem. Soc.* **103** 5377 (1981).
- 81ELS/ALL**
T. M. El-Sherbini, S. H. Allam, M. D. Migahed and A. M. Dawoud, "Mass Spectrometric Investigation of Aliphatic Aldehydes," *Z. Naturforsch.* **36a**, 1334 (1981).
- 81ENG/COR**
P.E. Engelking, R.R. Corderman, J.J. Wenddoski, G.B. Ellison, V.S. O'Neil and W.C. Lineberger, "Laser Photoelectron Spectroscopy of CH₂⁻, and the Singlet-Triplet Splitting in Methylene," *J. Chem. Phys.* **74**, 5460 (1981).
- 81ERA/KOL**
P.A. Erastov, V.P. Kolesov, L.N. Dityat'eva and Y.G. Golovanova, *J. Chem. Thermodyn.* **13**, 663 (1981).
- 81FAR/SRI**
M. Farber and R. D. Srivastava, "Electron Impact Ionization of Ba(OH)₂ (g)," *J. Chem. Phys.* **74**, 2160 (1981).
- 81FEI/COR**
C.S. Feigerle, R.R. Corderman, S.V. Bobashev and W.C. Lineberger, "Binding Energies and Structure of Transition Metal Negative Ions," *J. Chem. Phys.* **74**, 1580 (1981). [Superceded: 85HOT/LIN].
- 81FER/RIB**
M.L.C.C.H. Ferrao, M.A.V. Ribeiro de Silva, S. Suradi, G. Pilcher and H.A. Skinner, *J. Chem. Thermodyn.* **13**, 567 (1981).
- 81FOR/MAI**
P. Forster, J. P. Maier and F. Thommen, "Radiative and Non-Radiative Decay Rates of Alkyl Substituted Diacetylene Cations at Selected Energies within their 2A States Determined via Photoelectron-Photoion Coincidence Measurements," *Chem. Phys.* **59**, 85 (1981).
- 81FRO/KIR**
D.C. Frost, C. Kirby, C.A. McDowell and N.P.C. Westwood, "Preparation and HeI Photoelectron Spectra of the Dihaloboranes, HBX₂ (X = Cl and Br)," *J. Am. Chem. Soc.* **103**, 4428 (1981).
- 81FRO/MAC**
D.C. Frost, C.B. MacDonald, C.A. McDowell and N.P.C. Westwood, "Preparation and HeI Photoelectron Spectra of the Halogen Thiocyanates, XSCN (X = Cl and Br)," *J. Am. Chem. Soc.* **103**, 4423 (1981).
- 81FUJ/MCI**
M. Fujio, R.T. McIver, Jr. and R.W. Taft, "Effects on the Acidities of Phenols from Specific Substituent-Solvent Interactions. Inherent Substituent Parameters from Gas Phase Acidities," *J. Am. Chem. Soc.* **103**, 4017 (1981).
- 81FUR/PIA**
C. Furlani, M.N. Piancastelli, C. Cauletti, F. Faticanti and G. Ortaggi, "He(I) and He(II) Photoelectron Spectra of Some Organomercury Compounds with Carbon π-Systems," *J. Electron Spectrosc. Rel. Phenom.* **22**, 309 (1981).
- 81FUS/NOT**
H. Füssstetter and H. Noth, "Zur Darstellung von 1,1-Bis(diorganylboryl)-2,2-dimethylhydrazinen," *Liebigs Ann. Chem.* **633** (1981).
- 81GAL/KLA**
V. Galasso, L. Klasinc, A. Sabluic, N. Trinajstic, G. C. Pappalardo and W. Steglich, "Conformation and Photoelectron Spectra of 2-(2-Furyl)pyrrole and 2-(2-Thienyl)pyrrole," *J. Chem. Soc. Perkins II*, 127 (1981).
- 81GIN/PEL**
K.A. Gingerich, M. Pelino and R. Haque, *High Temp. Sci.* **14**, 137 (1981).
- 81GLE/BAR**
R. Gleiter and R. Bartetzko, "The Structures of S₄N, S₃N₂O₂, and S₄N₃," *Z. Naturforsch.* **36b**, 492 (1981).

- 81GLE/BOH**
R. Gleiter, M. C. Bohm and M. Baudler, "Photoelektronenspektroskopische Untersuchungen an Phosphor-Drei- und -Vierling-Systemen," *Chem. Ber.* **114**, 1004 (1981).
- 81GLE/GUB**
R. Gleiter, K. Gubernator, M. Eckert-Maksic, J. Spanget-Larsen, B. Bianco, G. Gandillon and U. Berger, "120. The Electronic Structure of Phenylene and Naphthylene Bicyclobutanes. Photoelectron Spectroscopy and Model Calculations," *Helv. Chim. Acta* **64**, 1312 (1981).
- 81GLE/GUB2**
R. Gleiter, K. Gubernator and W. Grimme, "Evidence for a Strong Through-Bond Interaction in anti-Tricyclo[6.4.0.0.2,7]dodecatetraene," *J. Org. Chem.* **46**, 1247 (1981).
- 81GLE/SCH**
R. Gleiter, W. Schafer and M. Eckert-Maksic, "Transannulare Wechselwirkungen zwischen Acetylenen - Photoelektronenspektroskopische Untersuchungen an 1,8-Diethylnaphthalin und cyclischen Derivaten von 2,2'-Diethylnaphthalin," *Chem. Ber.* **114**, 2309 (1981).
- 81GOD/SCH**
S.A. Godleski, P. von R. Schleyer, E. Osawa and W.T. Wipke, *Prog. Phys. Org. Chem.* **13**, 67 (1981).
- 81GRA/TON**
G. Granozzi, E. Tondello, M. Casarin and D. Ajo, "Electronic Structure of μ -Methylene-Bis-[Dicarbonyl (η^5 -Cyclopentadienyl)- Manganese] by UV Photoelectron Spectroscopy," *Inorg. Chim. Acta* **48**, 73 (1981).
- 81GRE/MIN**
J. C. Green, D. M. P. Mingos and E. A. Seddon, "Ultraviolet Photoelectron Studies on Bonding in Some Metal Carbonyl and Metal Hydrido Carbonyl Clusters," *Inorg. Chem.* **20**, 2595 (1981).
- 81GRI/LOS**
D. Griller and F. P. Lossing, "On the Thermochemistry of α -Aminoalkyl Radicals," *J. Am. Chem. Soc.* **103**, 1586 (1981).
- 81GRO/SCH**
G. Gross, R. Schulz, A. Schweig and C. Wentrup, "Isobenzofulvene," *Angew. Chem. Int. Ed.* **20**, 1021 (1981).
- 81GRU**
H.-F. Grutzmacher, "The Loss of ortho Halogeno Substituents from Substituted Thiobenzamide Ions," *Org. Mass Spectrom.* **16**, 448 (1981).
- 81GUP/NAP**
S. K. Gupta, B. M. Nappi and K. A. Gingerich, "Mass Spectrometric Study of the Stabilities of the Gaseous Molecules Pt₂ and PtY," *Inorg. Chem.* **20**, 966 (1981).
- 81GUS/VOL**
L. E. Guse'nikov, V. V. Volkova, V. G. Zaikin, N. A. Tarasenko, A. A. Tishenkov, N. S. Nametkin, M. G. Voronkov and S. V. Kirpichenko, "Mass Spectra of 3,3-Dimethyl-3-silathietane and 3,3-Diethyl-2,4-dimethyl-3-silathietane. First Observation of Silathione Ions and Calculation of Their Heats of Formation," *J. Organometall. Chem.* **215**, 9 (1981).
- 81GUS/VOL2**
L. E. Guse'nikov, V. V. Volkova, N. A. Tarasenko, A. A. Tishenkov, V. G. Zaikin, E. I. Eremina and N. S. Nametkin, "Deuterium Labelling and Thermochemical Studies of Dissociative Ionization of Methyl Substituted Monosilacyclobutanes. Ring Expansion in the Molecular Ion of 1,1,3-Trimethyl-1-silacyclobutane," *Org. Mass Spectrom.* **16**, 242 (1981).
- 81GUT/BES**
H. G. Guttenberger, H. J. Bestmann, F. L. Dickert, F. S. Jorgensen and J. P. Snyder, "Sulfur-Bridged peri-Naphthalenes: Synthesis, Conformational Analysis, and Photoelectron Spectroscopy of the Mono-, Di-, and Trisulfides of 1,8-Dimethylnaphthalene," *J. Am. Chem. Soc.* **103**, 159 (1981).
- 81HAL/ARM**
L. F. Halle, P. B. Armentrout and J. L. Beauchamp, "Formation of Chromium Carbene Ions by Reaction of Electronically Excited Chromium Ions with Methane in the Gas Phase," *J. Am. Chem. Soc.* **103**, 962 (1981).
- 81HAQ/GIN**
R. Haque and K. A. Gingerich, "Identification and Atomization Energies of Gaseous Molecules ScC₂, ScC₃, ScC₄, ScC₅, and ScC₆ by High Temperature Mass Spectrometry," *J. Chem. Phys.* **74**, 6407 (1981).
- 81HEI/KOV**
E. Heilbronner, B. Kovac, W. Nutakul, A. D. Taggart and R. P. Thummel, "Trisannulated Benzenes. Preparation, Properties, and Photoelectron Spectra," *J. Org. Chem.* **46**, 5279 (1981).
- 81HEN/ISA**
L. Henriksen, R. Isaksson, T. Liljefors and J. Sandstrom, "Ultraviolet Absorption and Photoelectron Spectra of Some Cyclic and Open-Chain Mono- and Dithiooxamides," *Acta Chem. Scand. B* **35**, 489 (1981).
- 81HEN/KNO**
S.P. Henegan, P.A. Knoot and S.W. Benson, *Int. J. Chem. Kinet.* **13**, 677 (1981).
- 81HER/SIC**
R. Herzsuh and A. Sicker, "Stereochemische Einflüsse auf die Ionisations- und Auftretensenergien cis/trans-isomerer Dimethylcycloalkane," *Z. Chem.* **21**, 409 (1981).
- 81HOA/CAB**
A. Hoareau, B. Cabaud and P. Melinon, "Time-of-Flight Mass Spectroscopy of Supersonic Beam of Metallic Vapours: Intensities and Appearance Potentials of M_x Aggregates," *Surface Sci.* **106**, 195 (1981).
- 81HOL/BUR**
J.L. Holmes, P.C. Burgers and J.K. Terlouw, "Water Elimination from the Keto and Enol Tautomers of Ionised Ethylacetate," *Can. J. Chem.* **59**, 1805 (1981).
- 81HOL/TIN**
J.L. Holmes, M. Fingas and F.P. Lossing, "Towards a General Scheme for Estimating the Heats of Formation of Organic Ions in the Gas Phase. Part I. Odd-Electron Cations," *Can. J. Chem.* **59**, 80 (1981).
- 81HOT**
K. Hottmann, "Über einige Zusammenhänge zwischen massenspektrometrischen Ionenhäufigkeiten und Molekuleigenschaften an element-organischen Verbindungen der Gruppe IVb," *J. f. Prakt. Chemie* **323**, 399 (1981).
- 81JAC/PEL**
R.L. Jackson, M.J. Pellerite and J.I. Brauman, "Photodetachment of the Azide Ion in the Gas Phase. Electron Affinity of the Azide Radical," *J. Am. Chem. Soc.* **103**, 1802 (1981).
- 81JEN/RAN**
W.B. Jennings, D. Randall, S.D. Worley and J.H. Hargis, "Conformation and Stereodynamics of 2-Dialkylamino-1,3-dimethyl-2,3-dihydro-1H-1,3,2-benzodiazaphospholes. An Experimental Nuclear Magnetic Resonance, Ultraviolet Photoelectron, and Theoretical MNDO Investigation," *J. Chem. Soc. Perkin Trans. II*, 1411 (1981).
- 81JON/MOO**
G. Jonkers, R. Mooyman and C.A. De Lange, "Ultraviolet Photoelectron Spectroscopy of Unstable Species: Nitrosyl Cyanide (ONCN)," *Chem. Phys.* **57**, 97 (1981).
- 81JOR**
F.S. Jorgensen, "Photoelectron Spectrum and Molecular Orbital (MNDO and PRDDO) Study of Dimethoxymethane," *J. Chem. Res.* **212** (1981).

81JOR/CAR

F.S. Jorgensen, L. Carlsen and F. Duus, "The Electronic Structure of β -Thioxoketones. A Photoelectron Spectroscopic Study of the Enol-Enethiol Tautomerism of Thioacetylacetone and Related Compounds," *J. Am. Chem. Soc.* **103**, 1350 (1981).

81JOR/NOR

F.S. Jorgensen, L. Norskov-Lauritsen, R.B. Jensen and G. Schroll, "Polyethers. Structural Analysis of the 1,4,5,8-Tetraoxadecalins and 2,2'-bis(1,3-Dioxolane) by Photoelectron Spectroscopy, Molecular Mechanics and Molecular Orbital Calculations," *Tetrahedron* **37**, 3671 (1981).

81KAI

W. Kaim, "Organometal-Stabilized 1,4-Dihydropyrazines: Extremely Electron-Rich Heterocycles," *Angew. Chem. Int. Ed.* **20**, 600 (1981).

81KAN/MOO

A. Kant and K.A. Moon, "Mass Spectrometric Determination of the Dissociation Energies of Gaseous AlH, GaH, InH, ScH, CoH and Estimation of the Maximum Dissociation Energies of TiH, CrH, MnH and FeH," *High Temp. Sci.* **14**, 23 (1981).

81KAP/STA

M.M. Kappes and R.H. Staley, "Oxidation of Transition-Metal Cations in the Gas Phase. Oxygen Bond Dissociation Energies and Formation of an Excited State Product," *J. Phys. Chem.* **85**, 942 (1981).

81KAU/VAH

R. Kaufel, G. Vahl, R. Minkwitz and H. Baumgartel, "Die Photoionenspektren von SCl_2 , S_2Cl_2 und S_2Br_2 ," *Z. Anorg. Allg. Chem.* **481**, 207 (1981).

81KHV/ZYK

V.I. Khvostenko, B.G. Zykov, V.P. Yuriev, V.F. Mironov, G.I. Kovel'zon, A.A. Panasenko, V.D. Sheludyakov and I.A. Gailyunas, "Study of $d(\pi)$ - $p(\pi)$ Interaction in Vinyl- and Alkylsilicon-Containing Compounds by Photoelectron Spectroscopy," *J. Organometall. Chem.* **218**, 155 (1981).

81KIM/KAT

K. Kimura, S. Katsumata, Y. Achiba, T. Yamazaki and S. Iwata, "Ionization Energies, Ab Initio Assignments and Valence Electronic Structure for 200 Molecules," *Handbook of HeI Photoelectron Spectra of Fundamental Organic Compounds*, Japan Scientific Soc. Press, Tokyo (1981) and Halsted Press, New York (1981).

81KIM/STE

Y.B. Kim, K. Stephan, E. Mark and T.D. Mark, "Single and Double Ionization of Nitric Oxide by Electron Impact from Threshold up to 180 eV," *J. Chem. Phys.* **74**, 6771 (1981).

81KIN/MAR

G.K. King, M.M. Maricq, V.M. Bierbaum and C.H. DePuy, "Gas Phase Reaction of Negative Ions with Alkyl Nitrites," *J. Am. Chem. Soc.* **103**, 7133 (1981).

81KLY/SHU

N.A. Klyuev, Y.V. Shurukhhin, I.I. Grandberg, I.K.81pad Yakushchenko and A.E. Bolubitskii, "Stereospecific Fragmentation of 1-Methyl-1,2-Diarylcyclopropanes," *Z. Org. Khim.* **17**, 556 (1981).

81KOB/ARA

T. Kobayashi, T. Arai, H. Sakuragi, K. Tokumaru and C. Utsunomiya, "A New Method for Conformational Analysis by Photoelectron Spectroscopy with Application to Alkyl-Substituted Styrenes," *Bull. Chem. Soc. Jpn.* **54**, 1658 (1981).

81KOE/MCK

T. Koenig and W. McKenna, "First Ionization Band of 1,1-Dimethylsilaethylene by Transient Photoelectron Spectroscopy," *J. Am. Chem. Soc.* **103**, 1212 (1981).

81KOM/DYK

A. Komornicki, C.E. Dykstra, M.A. Vincent and L. Radom, *J. Am. Chem. Soc.* **103**, 1652 (1981).

81KOP/PIK

I. Koppel, R. Pikver, A. Sugis, E. Suurmaa and E. Lippmaa, *Org. Reac.* **18**, 3 (1981).

81KUD/KUD

A.P. Kudchadker, S.A. Kudchadker, R.C. Wilhoit and S.K. Gupta, "Acenaphthylene, Acenaphthene, Fluorene, and Fluoranthene," *API Monograph*, **715-81**, American Petroleum Institute, Washington, D.C. (1981).

81KUD/KUD2

A.P. Kudchadker, S.A. Kudchadker, R.C. Wilhoit and S.K. Gupta, "Carbazole, 9-Methylcarbazole and Acridine," *API Monograph*, **716-81**, American Petroleum Institute, Washington, D.C. (1981).

81KUD/KUD3

A.P. Kudchadker, S.A. Kudchadker, R.C. Wilhoit and S.K. Gupta, "Thiophene, 2,3- and 2,5-Dihydrothiophene and Tetrahydrothiophene," *API Monograph* **717-81**, American Petroleum Institute, Washington, D.C. (1981).

81LAU/BRI

K.H. Lau, R.D. Brittain and D.L. Hildenbrand, *Chem. Phys. Lett.* **81**, 227 (1981).

81LEB/YEV

B.D. Lebedev, A.A. Yevstropov and Y.G. Kiparisova, *Int. J. Chem. Kinet.* **13**, 1185 (1981).

81LEE/POT

E.P.F. Lee and A.W. Potts, "Fine Structure in the He(I)/He(II) Photoelectron Spectra of the Metal Valence (d) Shells of the Group IIB Dihalides," *J. Electron Spectrosc. Rel. Phenom.* **22**, 247 (1981).

81LEU/HOF

S. Leutwyler, M. Hofmann, H.-P. Harri and E. Schumacher, "The Adiabatic Ionization Potentials of the Alkali Dimers Na_2 , NaK and K_2 ," *Chem. Phys. Lett.* **77**, 257 (1981).

81LIF/TZI

C. Lifshitz and E. Tzidony, "Kinetic Energy Release Distributions for $\text{C}_3\text{H}_6\text{O}^+$ Ion Dissociations: A Further Test of the Applicability of the Energy Randomization Hypothesis to Unimolecular Fragmentations," *Int. J. Mass Spectrom. Ion Phys.* **39**, 181 (1981).

81LIN/BES

T.B. Lindemer, T.M. Bessman and C.E. Johnson, *J. Nucl. Mater.* **100**, 178 (1981).

81LIV/ROB

P. Livant, K.A. Roberts, M.D. Eggers and S.D. Worley, "The Gas Phase Conformation of 3,7-Dimethyl-3,7-Diazabicyclo[3.3.1]nonane," *Tetrahedron* **37**, 1853 (1981).

81LOG/TAK

Y. Loguinov, V.V. Takhistov and L.P. Vatlina, "Photoionization Studies of Substituted Trimethylamines," *Org. Mass Spec.* **16**, 239 (1981).

81LOS/LAM

F.P. Lossing, Y.-T. Lam and A. Maccoll, "Gas Phase Heats of Formation of Alkyl Immonium Ions," *Can. J. Chem.* **59**, 2228 (1981).

81MAI

J.P. Maier, "Structure and Decay of Gaseous Organic Radical Cations Studied by Their Radiative Decay, Exemplified by the 1,3-Pentadiyne Cation," *Angew. Chem. Int. Ed. Engl.* **20**, 638 (1981).

81MAI/MIS

J.P. Maier, L. Misev and F. Thommen, "191. Laser-Induced Fluorescence and Photoelectron-Photon Coincidence Studies of 3,5-Octadiyne Cation," *Helv. Chim. Acta* **64**, 1985 (1981).

81MAI/SCH

W.F. Maier and P. von R. Schleyer, *J. Am. Chem. Soc.* **103**, 1891 (1981).

81MAI/THO

J.P. Maier and F. Thommen, "Fluorescence Quantum Yields and Lifetimes of Fluorobenzene Cations in Selected Levels of Their B and C States Determined by Photoelectron-Photoion Coincidence Spectroscopy," *Chem. Phys.* **57**, 319 (1981).

81MAI/THO2

J.P. Maier and F. Thommen, "Photoelectron-photon coincidence Measurements of the Fluorescence quantum Yields of cis-1,2-Difluoroethylene Cation in Selected Levels of the A^2A_1 State," *J. Chem. Soc. Faraday Trans. II* **77**, 845 (1981).

81MAL/ALI

I.P. Malkerova, A.S. Alikhanyan, V.D. Butskii, V.S. Pervov and V.I. Gorkoraki, "The Behavior of Tungsten Sulphide Tetrafluoride WSF_4 When Heated," *Zh. Neorganich. Khim.* **26**, 1955 (1981). *Engl. Trans.: Russ. J. Inorg. Chem.* **26**, 1055 (1981).

81MAL/MEL

I.P. Malkerova, E.I. Mel'nichenko, A.S. Alikhanyan, E.G. Rakov and V.I. Gorkoraki, "Thermal Decomposition of the Tungsten Oxide Fluorides $W_2O_4F.O.2(NOF)$ and W_2O_4F ," *Zh. Neorganich. Khim.* **26**, 17 (1981); *Engl. Trans.: Russ. J. Inorg. Chem.* **26**, 9 (1981).

81MAN/SUN

M. Mansson and S. Sunner, *J. Chem. Thermodyn.* **13**, 671 (1981).

81MAS/MOU

P. Masclet, G. Mouvier and J.F. Bocquet, "Effets Electroniques et Effets Steriques dus a la Substitution Alcoyle dans les Dienes Conjugues," *J. Chim. Phys.* **78**, 99 (1981).

81MAU/SIE

M. Meot-Ner (Mautner), L.W. Sieck and P. Ausloos, "Ionization of Normal Alkanes: Enthalpy, Entropy, Structural, and Isotope Effects," *J. Am. Chem. Soc.* **103**, 5342 (1981).

81MCD/CHO

R.N. McDonald, A.K. Chowdhury and D.W. Setser, "Gas Phase Generation of Phenylnitrene Anion Radical - Proton Affinity and Heat of Formation of PhN^- and Its Clustering with ROH Molecules," *J. Am. Chem. Soc.* **103**, 6599 (1981).

81MCL/CAM

S.A. McLuckey, D. Cameron and R.G. Cooks, "Proton Affinities from the Dissociation of Proton Bound Dimers," *J. Am. Chem. Soc.* **103**, 1313 (1981).

81MEE/WAH

J. Meeks, A. Wahlborg and S.P. McGlynn, "Photoelectron Spectroscopy of Carbonyls: Benzoic Acid and Its Derivatives," *J. Elec. Spectrosc. Rel. Phenom.* **22**, 43 (1981).

81MIC/SVE

G.D. Michels and H.J. Svec, "Characterization of $MnTc(CO)_{10}$ and $TcRe(CO)_{10}$," *Inorg. Chem.* **20**, 3445 (1981).

81MIL/CIL

S. Millefiori, E. Ciliberto, A. Millefiori and M.A. Zerbo, "Gas Phase U. V. Photoelectron Investigation of Azoxybenzene and 4,4'-di(n-alkoxy)azoxybenzenes," *Spectromchim. Acta* **37A**, 605 (1981).

81MIL/MIL

S. Millefiori and A. Millefiori, "Spectroscopic and Electrochemical Properties of Intramolecularly Hydrogen-Bonded Compounds. Ortho-Hydroxyazobenzenes," *Can. J. Chem.* **59**, 821 (1981).

81MOD/DIS

A. Modelli and G. Distefano, "L. C. B. O.: An Easy Method to Predict Valence Ionization Energies. Application to Substituted Benzenes," *Z. Naturforsch.* **36a**, 1344 (1981).

81MOD/DIS2

A. Modelli and G. Distefano, "He(I) Photoelectron Spectra of Chloro-, Vinyl- and Acetyl-Pyridines," *J. Elec. Spectrosc. Rel. Phenom.* **23**, 323 (1981).

81MOH/HIR

S. Mohmand, T. Hirabayashi and H. Bock, "Gasphasen-Reaktionen, 22. Thermische Erzeugung von C_4H_4O : Vinylketen und Ethylidenketen," *Chem. Ber.* **114**, 1609 (1981).

81MOH/JIA

M. Mohraz, W. Jian-qi, E. Heilbronner, A. Solladie-Cavallo and F. Matloubi-Moghadam, "11. Some Comments on the Conformations of Methyl Phenyl Sulfides, Sulfoxides and Sulfones," *Helv. Chim. Acta* **64**, 97 (1981).

81MOR/KOL

B.J. Morris-Sherwood, B.W.S. Kolthammer and M.B. Hall, "Photoelectron Spectra of and Molecular Orbital Calculations on (η^5 -Cyclopentadienyl)dinitrosylhalochromium and -Tungsten," *Inorg. Chem.* **20**, 2771 (1981).

81MUL/PRE

K. Muller and F. Previdoli, "247. Enamines. III. A Theoretical and Photoelectron Spectroscopic Study of the Molecular and Electronic Structures of Aziridine Enamines," *Helv. Chim. Acta* **64**, 2508 (1981).

81MUL/PRE2

K. Muller, F. Previdoli and H. Desilvestro, "246. Enamines. II. A Theoretical and Photoelectron Spectroscopic Study of the Molecular and Electronic Structure of Aliphatic Enamines," *Helv. Chim. Acta* **64**, 2497 (1981).

81MUR

E. Murad, "Thermochemical Properties of the Gaseous Alkaline Earth Monohydroxides," *J. Chem. Phys.* **75**, 4080 (1981).

81NAK/ASA

H. Nakagawa, M. Asano and K. Kubo, "Mass Spectrometric Study of the Vaporization of Lithium Metasilicate," *J. Nucl. Mat.* **102**, 292 (1981).

81NAP/GIN

B.M. Nappi and K.A. Gingerich, "Dissociation Energy and Standard Heat of Formation of Gaseous LaPt," *Inorg. Chem.* **20**, 522 (1981).

81NAT/FRA

P. Natalis and J.L. Franklin, "Ionization and Dissociation of Diphenyl and Condensed-Ring Aromatics by Electron Impact. III. Azobenzene," *Int. J. Mass Spectrom. Ion Phys.* **40**, 35 (1981).

81NEL/GRE

S.F. Nelsen, L.A. Grezzo and W.C. Hollinsed, "Effects of Structure on the Ease of Electron Removal from o-Phenylenediamines. 2. Photoelectron Spectra of o-Phenylenediamines," *J. Org. Chem.* **46**, 283 (1981).

81NES/BAI

A.N. Nesmeyanov, V.N. Baiden, Y.V. Chizhov, M.M. Timoshenko, Y.S. Nekrasov and I.I. Kritskaya, "He(I) Photoelectron Investigation of Functionally Substituted Camphene Derivatives," *Dokl. Akad. Nauk SSSR* **256**, 121 (1981).

81NIK/SID

M.I. Nikitin, L.N. Sidorov, E.V. Skokan and I.D. Sorokin, "Mass Spectrometric Determination of the Heats of Formation of ScF_4^- and KF_2^- ," *Russ. J. Phys. Chem.* **55**, 1107 (1981).

81NIK/SID2

M.I. Nikitin, L.N. Sidorov and M.V. Korobov, "The Electron Affinity of Platinum Hexafluoride," *Int. J. Mass Spectrom. Ion Proc.* **37**, 13 (1981).

81NOV/CVI

I. Novak, T. Cvitas and L. Klasinc, "Photoelectron Spectrum of Tetravinylstannane, $Sn(CH=CH_2)_4$," *J. Organometall. Chem.* **220**, 145 (1981).

81NOV/CVI2

I. Novak, T. Cvitas and L. Klasinc, "Photoelectron Spectrum of Tetraiodomethane," *Chem. Phys. Lett.* **79**, 154 (1981).

81NOV/CVI3

I. Novak, T. Cvitas, L. Klasinc and H. Gusten, "Photoelectron Spectra of Some Halogenomethanes," *J. Chem. Soc. Faraday Trans. II* **77**, 2049 (1981).

- 81ONO/OSU**
Y. Ono, E.A. Osuch and C.Y. Ng, "Molecular Beam Photoionization Study of OCS, (OCS)₂, (OCS)₃, and OCS-CS₂," *J. Chem. Phys.* **74**, 1645 (1981).
- 81PAD/PAT**
M.N. Paddon-Row, H.K. Patney, R.S. Brown and K.N. Houk, "Observation of a Very Large Orbital Interaction through Four Bonds. An Alternative Model of Orbital Interactions through Bonds," *J. Am. Chem. Soc.* **103**, 5575 (1981).
- 81PAL/SIM**
M.H. Palmer, I. Simpson and J.R. Wheeler, "Gas Phase Tautomerism in the Triazoles and Tetrazoles: A Study by Photoelectron Spectroscopy and ab Initio Molecular Orbital Calculations," *Z. Naturforsch.* **36a**, 1246 (1981).
- 81PAP/ERA**
T. Papina, P.A. Erastov and V.P. Kolesov, *J. Chem. Thermodyn.* **13**, 683 (1981).
- 81PAQ/HER**
L.A. Paquette, L.W. Hertel, R. Gleiter, M.C. Bohm, M.A. Beno and G.G. Christoph, "Electronic Control of Stereoselectivity. 8. The Stereochemical Course of Electrophilic Additions to Aryl-Substituted 9-Isopropylidenebenzonorbornenes," *J. Am. Chem. Soc.* **103**, 7106 (1981).
- 81PEL/JAC**
M.J. Pellerite, R.L. Jackson and J.I. Brauman, "Proton Affinity of the Gaseous Azide Ion. The N-H Bond Dissociation Energy in HN₃," *J. Phys. Chem.* **85**, 1624 (1981).
- 81PFI/GUI**
G. Pfister-Guillouzo, C. Guimon, J. Frank, J. Ellison and A.R. Katritzky, "Tautomeric Pyridines, 26. A Photoelectron Spectral Study of the Vapour Phase Tautomerism of 2- and 4-Quinolone," *Liebigs Ann. Chem.* **366** (1981).
- 81PLE/VIL**
V.V. Plemenkov, Y.Y. Villem, N.V. Villem, I.G. Bolesov, L.S. Surmina, N.I. Yakushkina and A.A. Formanovskii, "Photoelectron Spectra of Polyalkylcyclopropenes and Polyalkylcyclopropanes," *Zh. Obshchei. Khim.* **51**, 2076 (1981).
- 81POL/RAI**
S.K. Pollack, B.C. Raine and W.J. Hehre, "Determination of the Heats of Formation of the Isomeric Xylylenes by Ion Cyclotron Resonance Spectroscopy," *J. Am. Chem. Soc.* **103**, 6308 (1981).
- 81PRA/CHU**
S.T. Pratt and W.A. Chupka, "Photoionization Study of the Kinetics of Unimolecular Decomposition of Halobenzene Ions," *Chem. Phys.* **62**, 153 (1981).
- 81PRO/RAD**
A. Pross and L. Radom, *Prog. Phys. Org. Chem.* **13**, 1 (1981).
- 81RAB/HEL**
M.A. Rabbih, A.I. Helal and M.A. Fahmey, "Mass Spectrometric Studies of Methylacetophenone Isomers by Electron Impact," *Indian J. Pure Appl. Phys.* **19**, 335 (1981).
- 81RAN/WRI**
D.W.H. Rankin and J.G. Wright, "The Preparation and Properties of 1-Difluorophosphino-pyrrole," *J. Fluorine Chem.* **17**, 469 (1981).
- 81RID/RAY**
D.M. Rider, G.W. Ray, E.J. Darland and G.E. Leroi, "A Photoionization Mass Spectrometric Investigation of CH₃CN and CD₃CN," *J. Chem. Phys.* **74**, 1652 (1981).
- 81ROB/STE**
D.A. Robaugh and S.E. Stein, *Int. J. Chem. Kinet.* **13**, 445 (1981).
- 81ROT/SCH**
W.R. Roth and B.P. Scholz, *Chem. Ber.* **114**, 3741 (1981).
- 81RUS/KLA**
B. Ruscic, L. Klasinc, A. Wolf and J.V. Knop, "Photoelectron Spectra of and Ab Initio Calculations on Chlorobenzenes. 3. Hexachlorobenzene," *J. Phys. Chem.* **85**, 1495 (1981).
- 81RUS/KLA2**
B. Ruscic, L. Klasinc, A. Wolf and J.V. Knop, "Photoelectron Spectra of and Ab Initio Calculations on Chlorobenzenes. 1. Chlorobenzene and Dichlorobenzenes," *J. Phys. Chem.* **85**, 1486 (1981).
- 81RUS/KLA3**
B. Ruscic, L. Klasinc, A. Wolf and J.V. Knop, "Photoelectron Spectra of and Ab Initio Calculations on Chlorobenzenes. 2. Trichlorobenzenes, Tetrachlorobenzenes, and Pentachlorobenzene," *J. Phys. Chem.* **85**, 1490 (1981).
- 81SAB/MIN**
R. Sabbah and C. Mindakis, *Thermochim. Acta* **43**, 269 (1981).
- 81SAS/HAR**
T. Sasamoto, H. Hara and T. Sata, "Mass Spectrometric Study of the Vaporization of Magnesium Oxide from Magnesium Aluminate Spinel," *Bull. Chem. Soc. Jpn.* **54**, 3327 (1981).
- 81SAT/SEK**
N. Sato, K. Seki and H. Inokuchi, "Polarization Energies of Organic Solids Determined by Ultraviolet Photoelectron Spectroscopy," *J. Chem. Soc. Faraday Trans. II* **77**, 1621 (1981).
- 81SCH/GRO**
B. Schaldach, B. Grottemeyer, J. Grottemeyer and H.-F. Grutzmacher, "Kinetic and Thermodynamic Effects on Intramolecular Aromatic Substitution in meta and para Substituted Benzalacetones," *Org. Mass Spectrom.* **16**, 410 (1981).
- 81SCH/SCH**
R. Schulz and A. Schweig, "Cyclopentadienethione," *Angew. Chem. Int. Ed.* **20**, 570 (1981).
- 81SEN/KEB**
D.K. Sen Sharma and P. Kebarle, "Stability and Reactivity of the Benzyl and Tropylium Cations in the Gas Phase," *Can. J. Chem.* **59**, 1592 (1981).
- 81SHA/AKI**
M. Shahbaz, I. Akiyama and P. LeBreton, "Ultraviolet Photoelectron Studies of Methyl Substituted Chrysenes," *Biochemical and Biophysical Research Communications* **103**, 25 (1981).
- 81SID/SOR**
L.N. Sidorov, I.D. Sorokin, N.I. Nitikin and E.V. Skokan, "Effusion Method for Determining the Electron Affinity and Heat of Formation of Negative Ions," *Int. J. Mass Spectrom. Ion Phys.* **39**, 311 (1981).
- 81SKO/NIK**
E.V. Skokan, M.I. Nikitin, I.D. Sorokin, A.V. Gusarov and L.N. Sidorov, "Determination of the Heat of Formation of the Tetrafluoroscandate and Pentafluorozirconate Ions by the Effusion Method," *Russ. J. Phys. Chem.* **55**, 1062 (1981).
- 81SMI/ADA**
D. Smith, N.G. Adams and W. Lindinger, "Reactions of the H_nS Ions (n = 0 to 3) with Several Molecular Gases at Thermal Energies," *J. Chem. Phys.* **75**, 3365 (1981).
- 81SMI/LAI**
G.P. Smith and R.M. Laine, "Organometallic Bond Dissociation Energies. Laser Pyrolysis of Fe(CO)₅," *J. Phys. Chem.* **85**, 1620 (1981).
- 81SMI/STE**
O.I. Smith and J.S. Stevenson, "Determination of Cross Sections for Formation of Parent and Fragment Ions by Electron Impact from SO₂ and SO₃," *J. Chem. Phys.* **74**, 6777 (1981).
- 81SOR/SID**
I.D. Sorokin, L.N. Sidorov, M.I. Nikitin and E.V. Skokan, "Mass-Spectrometric Determination of the Enthalpies of Dissociation of Gaseous Complex Fluorides into Neutral and Charged Particles. V. Heats of Formation of FeF₃⁻ and FeF₄⁻," *Int. J. Mass Spectrom. Ion Phys.* **41**, 45 (1981).

- 81SPE**
P.J. Spencer, "Hafnium: Physicochemical Properties of its compounds and Alloys" ("Atomic Energy Review Special Issue 8," ed. K.L. Komarek, IAEA, Vienna, 1981).
- 81STE/BAR**
S.E. Stein and B.D. Barton, *Thermochim. Acta* **44**, 265 (1981).
- 81SUL/DEP**
S.A. Sullivan, C.H. DePuy and R. Damrauer, "Gas Phase Reactions of Cyclic Silanes," *J. Am. Chem. Soc.* **103**, 480 (1981).
- 81SUR/HAC**
S. Suradi, J.M. Hacking, G. Pilcher, I. Gumrukcu and M.F. Lappert, *J. Chem. Thermodyn.* **13**, 857 (1981).
- 81SZE/KOR**
L. Szepes, T. Koranyi, G. Naray-Szabo, A. Modelli and G. Distefano, "Ultraviolet Photoelectron Spectra of Group IV Hexamethyl Derivatives Containing a Metal-Metal Bond," *J. Organometall. Chem.* **217**, 35 (1981).
- 81TAA/SUM**
M. Taagepera, K.D. Summerhays, W.J. Hehre, R.D. Topsom, A. Pross, L. Radom and R.W. Taft, *J. Org. Chem.* **46**, 891 (1981).
- 81TER/HEE**
J.K. Terlouw, W. Heerma and G. Dijkstra, "On the Structure of the Odd Electron $[C_2H_6O]^+$ Ions in the Mass Spectrum of [1,3-Propanediol]⁺," *Org. Mass Spectrom.* **16**, 326 (1981).
- 81TIM/KOR**
M.M. Timoshenko, I.V. Korkoshko, V.T. Kleimenov, N.E. Petrachenko, Y.V. Chizhov, V.V. Ryl'kov and M.E. Akopyan, "Ionization Potentials of Rhodamine Dyes," *Dokl. Akad. SSSR* **260**, 138 (1981).
- 81TPIS**
L.V. Gurvich, I.V. Veits, V.A. Medvedev, G.A. Khachkuruzov, V.S. Yungman and G.A. Bergman, et al., "Termodinamicheskie Svoistva Individual'nykh Veshchestv" (Thermodynamic Properties of Individual Substances); Glushko, V.P., gen. ed., Vol. 3, parts 1 and 2 (1981), Izdatel'stvo "Nauka" Moscow.
- 81TRA**
J.C. Traeger, "Heat of Formation of sec-Butyl Cation in the Gas Phase," *Org. Mass Spectrom.* **16**, 193 (1981).
- 81TRA/MCL**
J.C. Traeger and R.G. McLoughlin, "Absolute Heats of Formation for Gas Phase Cations," *J. Am. Chem. Soc.* **103**, 3647 (1981).
- 81TRA/RED**
V.F. Traven', V.V. Redchenko, M.Y. Eismont and B.I. Stepanov, "Photoelectron Spectra and Electronic and Steric Structures of Silicon and Sulfur-Containing Analogs of 9,10-Dihydroanthracene," *Zh. Obs. Khim.* **51**, 1297 (1981); English Trans.: *J. Gen. Chem. (USSR)* **51**, 1099 (1981).
- 81TRA/RED2**
F. Traven', V.V. Redchenko and B.I. Stepanov, "Photoelectron Spectrum of Thianthrene," *Zh. Obs. Khim.* **51**, 1293 (1981); English Trans.: *J. Gen. Chem. (USSR)* **51**, 1094 (1981).
- 81TRO/NED**
B.A. Trofimov, N.A. Nedolya, N.B. Lebedev, V.L. Ryadnenko, T.N. Masalitinova, S.L. Dobyichin, R.K. Zacheslavskaya and G.N. Petrov, *Bull. Acad. Sci. USSR Div. Chem. Sci.* **537** (1981).
- 81TSA**
W. Tsang, "Shock Tubes in Chemistry," (A. Lifshitz, Editor; Dekker, 1981), p. 59.
- 81VAN/TER**
H. Van Dam, A. Terpstra, A. Oskam and J.H. Teuben, "UV Photoelectron Spectra of Some Bent Bis(η^5 -cyclopentadienyl)Niobium and Tantalum Complexes," *Z. Naturforsch.* **36b**, 420 (1981).
- 81VOV/DUD**
V.I. Vovna, A.S. Dudin, A.M. Kleshchevnikov, S.N. Lopatin and E.G. Rakov, "Photoelectron Spectra and Electronic Structure of Molybdenum, Tungsten, Rhenium, and Osmium Oxotetrafluorides," *Koord. Khim.* **7**, 575 (1981).
- 81WAL**
R. Walsh, "Bond Dissociation Energy Values in Silicon-Containing Compounds and Some of Their Implications," *Acc. Chem. Res.* **14**, 246 (1981).
- 81WAL/BLA**
E.A. Walters and N.C. Blais, "Molecular Beam Photoionization of $(H_2S)_n$, $n = 1 - 7$," *J. Chem. Phys.* **75**, 4208 (1981).
- 81WEC/CHR**
D. Wecker, A.A. Christodoulides and R.N. Schindler, "Studies by the Electron Cyclotron Resonance (ECR) Technique. XV. Interactions of Thermal-Energy Electrons with ClO_2 and Cl_2O ," *Int. J. Mass Spectrom. Ion Phys.* **38**, 391 (1981).
- 81WES/REI**
J.B. Westmore, M.L.J. Reimer and C. Reichert, "Ionization Energies of Metal Chelates. Acetylacetonates, Trifluoroacetylacetonates, and Hexafluoroacetylacetonates of Trivalent Metals of the First Transition Series," *Can. J. Chem.* **59**, 1797 (1981).
- 81WHI/FRI**
R.A. Whiteside, M.J. Frisch, J.S. Binkley, D.J. DeFrees, B. Schlegel, K. Raghavachari and J.A. Pople, "Carnegie-Mellon Quantum Chemistry Archive" (2nd Edn., Pittsburgh, 1981).
- 81WIB/SQU**
K.B. Wiberg and R.R. Squires, *J. Am. Chem. Soc.* **103**, 4473 (1981).
- 81WIB/WAS**
K.B. Wiberg and D.J. Wasserman, *J. Am. Chem. Soc.* **103**, 6563 (1981).
- 81WOJ/FOL**
L. Wojnarovits and G. Foldiak, "Electron Capture Detection of Aromatic Hydrocarbons," *J. Chromatogr.* **206**, 511 (1981).
- 81WON/DUT**
K.S. Wong, T.L. Dutta and T.P. Fehlner, "The Proton as a Probe of Cluster Bonding. The UV Photoelectron Spectra of Two Hydroido Transition Metal Clusters," *J. Organometall. Chem.* **215**, C48 (1981).
- 81WOO**
A.A. Woolf, *Adv. Inorg. Chem. Radiochem.* **24**, 1 (1981).
- 81WOR/GIB**
S.D. Worley, D.H. Gibson and W.-L. Hsu, "Electronic Structures of Some η^3 -Allyl Transition Metal Complexes," *Inorg. Chem.* **20**, 1327 (1981).
- 81WRI/BEA**
C.A. Wright and J.L. Beauchamp, "Infrared Spectra of Gas Phase Ions and Their Use in Elucidating Reaction Mechanisms. Identification of $C_7H_7^-$ Structural Isomers by Multiphoton Electron Detachment Using a Low-Powered Laser," *J. Am. Chem. Soc.* **103**, 6499 (1981).
- 81YU/ODO**
C. Yu, T.J. O'Donnell and P.R. LeBreton, "Ultraviolet Photoelectron Studies of Volatile Nucleoside Models. Vertical Ionization Potential Measurements of Methylated Uridine, Thymidine, Cytidine, and Adenosine," *J. Phys. Chem.* **85**, 3851 (1981).
- 81ZHO/HEI**
Y. Zhong-zhi, E. Heilbronner, H.C. Kang and V. Boekelheide, "The Photoelectron Spectra of 4,13-Diaza- and 4,16-Diaza[2,4](1,2,4,5)cyclophanes" *Helv. Chim. Acta* **64**, 2029 (1981).
- 81ZHO/KOV**
Y. Zhong-zhi, B. Kovak, E. Heilbronner, S. Eltamany and H. Hopf, "192. Ionization Energies of Methyl-substituted [2.2] Paracyclophanes," *Helv. Chim. Acta* **64**, 1991 (1981).

- 81ZVE/ERM**
V.V. Zverev and L.V. Ermolaeva, "Ionization Potentials and Intramolecular Charge Transfer. II. The Photoelectron Spectrum and Electronic Structure of Trichloromethylbenzene," *Zh. Strukt. Khim.* **22**, 22 (1981).
- 81ZVE/VIL**
V.V. Zverev, Y.Y. Villem, N.V. Villem, B.G. Liorber and Y.P. Kitaev, "Photoelectron Spectra of Some Unsaturated Three- and Four-Coordinate Phosphorus Compounds," *Zh. Obs. Khim.* **51**, 303 (1981); English trans.: *J. Gen. Chem. (USSR)*, **51**, 242 (1981).
- 81ZVE/VIL2**
V.V. Zverev, Y.Y. Villem and R.P. Arshinova, "Photoelectron Spectra of Di- and Triesters of Trivalent Phosphorus," *Dokl. Akad. Nauk, SSSR* **256**, 1412 (1981).
- 81ZYK/KHV**
B.G. Zykov, V.I. Khvostenko, M.G. Voronkov, V.P. Yur'ev, G.S. Lomakin and E.N. Suslova, "Photoelectron Spectra of Si-Substituted Chloromethylsilanes," *Dokl. Akad. Nauk SSSR* **258**, 135 (1981).
- 82ALD/ARR**
R.W. Alder, R.J. Arrowsmith, C.S.J. Boothby, E. Heilbronner and Y. Zhong-zhi, "1-Azabicyclo[4.4.4]tetradec-5-ene," *J. Chem. Soc. Chem. Commun.* **940** (1982).
- 82ALL/DOD**
N.L. Allinger, H. Dodziuk, D.W. Rogers and S.N. Naik, *Tetrahedron* **38**, 1593 (1982).
- 82ALL/MIG**
S.H. Allam, M.D. Migahed and A. El-Khodary, "Electron Impact Ionization and Dissociation of Deuterated and non-Deuterated Methanol, Methyl Cyanide, Nitromethane and Nitrobenzene," *Egypt. J. Phys.* **13**, 167 (1982).
- 82ARM/HAL**
P.B. Armentrout, L.F. Halle and J.L. Beauchamp, "Reaction of Cr^+ , Mn^+ , Fe^+ , Co^+ , and Ni^+ with O_2 and N_2O . Examination of the Translational Energy Dependence of the Cross Sections of Endothermic Reactions," *J. Chem. Phys.* **76**, 2449 (1982).
- 82ARN/VEN**
E.M. Arnett, K.G. Venkatasubramanian, R.T. McIver, E.K. Fukuda, F.G. Bordwell and F.D. Press, "Stabilization of the Monoanion of 1,8-Diaminonaphthalene by Intramolecular Hydrogen Bonding. A Novel Case of Anion Homoconjugation in Superbase Solution," *J. Am. Chem. Soc.* **104**, 325 (1982).
- 82AUD/BOU**
H.E. Audier, G. Bouchoux, Y. Hoppilliard and A. Milliet, "The Mechanism of Formation of $[\text{C}_4\text{H}_9\text{O}]^+$ Ions from Isobutyl Ethyl Ether," *Org. Mass Spectrom.* **17**, 382 (1982).
- 82BAI/CON**
M.A. Baig, J.P. Connerade and J. Hormes, "Autoionisation Resonances in the 4p(II) Spectrum of Methyl Bromide," *J. Phys. B: At. Mol. Phys.* **15**, L5 (1982).
- 82BAL/CAR**
M.A. Baldwin, D.M. Carter and J. Gilmore, "Loss of Hydroxyl Radical from Isomeric Ethylnitrobenzenes," *Org. Mass Spectrom.* **17**, 45 (1982).
- 82BAN/CHA**
S. Banon, C. Chatillon and M. Allibert, "High Temperature Mass Spectrometric Study of Ionization and Fragmentation of TiO and TiO_2 Gas Under Electron Impact," *High Temp. Sci.* **15**, 17 (1982).
- 82BAN/CHA2**
G.M. Bancroft, T. Chan, R.J. Puddephatt and J.S. Tse, "Role of the Au 5d Orbitals in Bonding: Photoelectron Spectra of $[\text{AuMe}(\text{PMc}_3)]$," *Inorg. Chem.* **21**, 2946 (1982).
- 82BAN/PEL**
G.M. Bancroft, E. Pellach, A.P. Sattelberger and K.W. McLaughlin, "The Photoelectron Spectrum of Quadruply Bonded $\text{W}_2(\text{O}_2\text{CCF}_3)_4$," *J. Chem. Soc. Chem. Commun.* **752** (1982).
- 82BAR**
J.E. Bartmess, "Gas Phase Ion Chemistry of 5-Methylene-1,3-cyclohexadiene (o-Isotoluene) and 3-Methylene-1,4-cyclohexadiene (p-Isotoluene)," *J. Am. Chem. Soc.* **104**, 335 (1982).
- 82BAR/CAU**
V. Barone, C. Cauletti, F. Lelu, M.N. Piancastelli and N. Russo, "Relative Ordering and Spacing of n and π Levels in Isomeric Bipyrimidines. A Theoretical and Gas-Phase UV Photoelectron Spectroscopic Study," *J. Am. Chem. Soc.* **104**, 4571 (1982).
- 82BAR/HEI**
R. Bar, T. Heinis, C. Nager and M. Jungen, "Photoionization of Ferrocene," *Chem. Phys. Lett.* **91**, 440 (1982).
- 82BAU/BUN**
H. Baumann, J.-C. Bunzli and J.F.M. Oth, "58. The Photoelectron Spectrum of [18]Annulene," *Helv. Chim. Acta* **65**, 582 (1982).
- 82BAU/COX**
D.L. Baulch, R.A. Cox, P.J. Crutzen, R.F. Hampson, Jr., J.A. Kerr, J. Troe and R.T. Watson, *J. Phys. Chem. Ref. Data* **11**, 327 (1982).
- 82BAX/COW**
S.G. Baxter, A.H. Cowley, J.G. Lasch, M. Lattman, W.P. Sharum and C.A. Stewart, "Electronic Structures of Bent-Sandwich Compounds of the Main-Group Elements: A Molecular Orbital and UV Photoelectron Spectroscopic Study of Bis(cyclopentadienyl)tin and Related Compounds," *J. Am. Chem. Soc.* **104**, 4064 (1982).
- 82BEN/DUR**
A. Bened, R. Durand, D. Pioch, P. Geneste, J.-P. Declercq, G. Germain, J. Rambaud, R. Roques, C. Guimon and G. Pfister-Guillouzo, "Isoxazolines by Cycloadditions of Mesitronitrile Oxide with Benzo[b]thiophene S-Oxide and S,S-Dioxide. Structural Studies, Theoretical Explanations, and Kinetics," *J. Org. Chem.* **47**, 2461 (1982).
- 82BER/HEN**
U. Berg, L. Henriksen, K.A. Lerstrup and J. Sandstrom, "The Torsional Barrier of the Dimethylamino Group in N,N-Dimethyltellurobenzamide. A Comparison with N,N-Dimethylbenzamide and its Thio and Seleno Analogues," *Acta Chem. Scand. B* **36**, 19 (1982).
- 82BIE/ASB**
G. Bieri, L. Asbrink and W. Von Niessen, "30.4-nm He(II) Photoelectron Spectra of Organic Molecules," *J. Electron Spectrosc. Rel. Phenom.* **27**, 129 (1982).
- 82BLÓ/COR**
E. Block, E.R. Corey, R.E. Penn, T.L. Renken, P.F. Sherwin, H. Bock, T. Hirabayashi, S. Mohmand and B. Solouki, "Synthesis and Thermal Decomposition of 1,3-Dithetane and Its S-Oxides," *J. Am. Chem. Soc.* **104**, 3119 (1982).
- 82BOC/MOH**
H. Bock, S. Mohmand, T. Hirabayashi and A. Semkow, "Thioacrolein," *J. Am. Chem. Soc.* **104**, 312 (1982).
- 82BOC/MOH2**
H. Bock, S. Mohmand, T. Hirabayashi and A. Semkow, "Thioacrolein: Das stabilste $\text{C}_3\text{H}_4\text{S}$ -Isomers und sein PE-spektroskopischer Nachweis in der Gasphase," *Chem. Ber.* **115**, 1339 (1982).
- 82BOC/WIT**
H. Bock, J. Wittmann, J. Mintzer and J. Russow, "Ni/Pd-Katalysierte Gasphasen-Bromierung von Trifluormethan," *Chem. Ber.* **115**, 2346 (1982).
- 82BOH**
M.C. Bohm, "The Photoelectron Spectra of Bis(cyclopentadienyl)titanium Derivatives - a Green's Function Approach," *Inorg. Chim. Acta* **62**, 171 (1982).

82BOH/ECK

M.C. Bohm, M. Eckert-Maksic, R.D. Ernst, D.R. Wilson and R. Gleiter, "Electronic Structure of Bis(pentadienyl)iron. Semiempirical Calculations and Photoelectron Spectra," *J. Am. Chem. Soc.* **104**, 2699 (1982).

82BOH/GLE

M.C. Bohm, R. Gleiter and W. Petz, "The He(I) Photoelectron Spectrum of $\text{Fe}(\text{CO})_4\text{CS}$," *Inorg. Chim. Acta* **59**, 255 (1982).

82BOH/GLE2

M.C. Bohm and R. Gleiter, "The Electronic Structure and the He(I) Photoelectron Spectrum of Bis(π -Pentadienyl)dinickel," *Chem. Phys.* **64**, 183 (1982).

82BOM/DAN

R. Bombach, J. Dannacher, J.-P. Stadelmann, J. Vogt, L.R. Thorne and J.L. Beauchamp, "Photoelectron-Photoion Coincidence Study of CF_3I . Implications for the CW IR Laser Multiphoton Dissociation of CF_3I^+ ," *Chem. Phys.* **66**, 403 (1982).

82BRE/SCH

M. Breitenstein, R. Schulz and A. Schweig, "Photoelectron Spectrum and Infrared Spectrum of Thermally Generated Transient Benzodithiete," *J. Org. Chem.* **47**, 1979 (1982).

82BRO/NYB

A.G. Brook, S.C. Nyburg, F. Abdesaken, B. Gutekunst, G. Gutekunst, R.K.M. R. Kallury, Y.C. Poon, Y.-M. Chang and W. Wong-Ng, "Stable Solid Silaethylenes," *J. Am. Chem. Soc.* **104**, 5667 (1982).

82BRU/CIL

G. Bruno, E. Ciliberto, R.D. Fischer, I. Fragala and A.W. Spieg, "Photoelectron Spectroscopy of f-Element Organometallic Complexes. 5. Comparative Study of Ring-Substituted Uranocenes," *Organometall.* **1**, 1060 (1982).

82BUR/FEI

S.M. Burnett, C.S. Feigerle, A.E. Stevens and C.W. Lineberger, "Photoelectron Spectroscopy of CS^- and NS^- ," *J. Phys. Chem.* **86**, 4486 (1982).

82BUR/HAY

N.A. Burdett and A.N. Hayhurst, "Hydration of Gas Phase Ions and the Measurement of Boundary Layer Cooling During Flame Sampling into a Mass Spectrometer," *J. Chem. Soc. Farad. I* **78**, 2997 (1982).

82BUR/HOL

P.C. Burgers, J.L. Holmes, F.P. Lossing, A.A. Mommers, F.R. Povel and J.K. Terlouw, "Isomeric and Tautomeric $[\text{C}_4\text{H}_4\text{O}]^+$ Ions. Their Thermochemistry and Collisionally Induced Fragmentation Characteristics," *Can. J. Chem.* **60**, 2246 (1982).

82BUR/HOL2

P.C. Burgers and J.L. Holmes, "Metastable Ion Studies. XIII. The Measurement of Appearance Energies of Metastable Peaks," *Org. Mass Spectrom.* **17**, 123 (1982).

82BUR/PAW

H. Burger, G. Pawelke, R. Dammel and H. Bock, "Effects of Fluorine Substitution on Methyl Amines," *J. Fluorine Chem.* **19**, 565 (1982).

82BUR/TER

P.C. Burgers, J.K. Terlouw and K. Levsen, "Gaseous $[\text{C}_7\text{H}_8]^+$ Ions: [Methylene Cyclohexadiene] $^+$, a Stable Species in the Gas Phase," *Org. Mass Spectrom.* **17**, 295 (1982).

82BUR/TER2

P.C. Burgers, J.K. Terlouw and J.L. Holmes, "The Vinyloxonium Cation, $\text{CH}_2=\text{CH}-\text{OH}_2^+$, a Stable $[\text{C}_2\text{H}_5\text{O}]^+$ Species in the Gas Phase," *Org. Mass Spectrom.* **17**, 369 (1982).

82BUS/WEI

B. Busse and K.G. Weil, "Mass Spectrometric Studies on the Vapour Phase Composition over Solid and Fused Cs_3Sb ," *Ber. Bunsenges. Phys. Chem.* **86**, 93 (1982).

82BUT/BAE

J.J. Butler and T. Baer, "Photoionization Study of the Heat of Formation of HCS^+ ," *J. Am. Chem. Soc.* **104**, 5016 (1982).

82BUT/FRA

J.J. Butler, M.L. Fraser-Monteiro, L. Fraser-Monteiro, T. Baer and J.R. Hass, "Thermochemistry and Dissociation Dynamics of State-Selected $\text{C}_4\text{H}_8\text{O}_2^+$ Ions. 2. Butanoic Acid," *J. Phys. Chem.* **86**, 747 (1982).

82BYS

K. Bystrom, *J. Chem. Thermodyn.* **14**, 865 (1982).

82BYS/MAN

K. Bystrom and M. Mansson, *J. Chem. Soc. Perkins Trans. II*, 565 (1982).

82CAB/COW

D.E. Cabelli, A.H. Cowley and J.J. Lagowski, "The Bonding in Some Bis(arene)Chromium Compounds as Indicated by U. V. Photoelectron Spectroscopy," *Inorg. Chim. Acta* **57**, 195 (1982).

82CAR/KIB

F. Carnovale, M.H. Kibel, G.L. Nyberg and J.B. Peel, "Photoelectron Spectroscopic Assignment of the p-States of Benzenethiol," *J. Electron Spectrosc. Rel. Phenom.* **25**, 171 (1982).

82CAS/CIL

M. Casarin, E. Ciliberto, I. Fragala and G. Granozzi, "Photoelectron Spectroscopy of f-Element Coordination Compounds. 2. He-II Spectra of β -Diketonate Complexes of Uranium(IV), Thorium(IV), Zirconium(IV) and Dioxouranium(VI)," *Inorg. Chim. Acta* **64**, L247 (1982).

82CAS/GOD

C.J. Casewitt and W.A. Goddard, III, *J. Am. Chem. Soc.* **104**, 3280 (1982).

82CHE/LAP

E.K. Chess, R.L. Lapp and M.L. Gross, "The Question of Tautomerism of Alkynitrile and Isonitrile Radical Cations," *Org. Mass Spectrom.* **17**, 475 (1982).

82CHI/LIE

J.S. Chickos and J.F. Liebman unpublished result.

82CHO/FRO

D.P. Chong, D.C. Frost, W.M. Lau and C.A. McDowell, "Shake-Up Satellites in HeI Photoelectron Spectra: N_2O_4 and CH_3NO ," *Chem. Phys. Lett.* **90**, 332 (1982).

82CHU/NGU

J.Y. Chu, T.T. Nguyen and K.D. King, *J. Phys. Chem.* **86**, 443 (1982).

82COL/JIM

M. Colomina, P. Jimenez and C. Turrion, *J. Chem. Thermodyn.* **14**, 779 (1982).

82COM/REI

R.N. Compton and P.W. Reinhardt, "Collisional Ionization between Fast Alkali Atoms and Hexafluorobenzene," *Chem. Phys. Lett.* **91**, 268 (1982).

82CON/ZAF

J.A. Connor, M.T. Zafarani-Moattar, J. Bickerton, N.I. El-Saied, S. Suradi, R. Carson, G. Al Takkhin and H.A. Skinner, *Organometallics* **1**, 1166 (1982).

82COS/LLO

N.C.V. Costa, D.R. Lloyd, P. Brint, W.K. Pelin and T.R. Spalding, "The Photoelectron Spectra of Enneacarbonyl- η^3 -methylidyne-tricobalt and Some Derivatives," *J. Chem. Soc. Dalton Trans.* 201 (1982).

82COW/KEM

A.H. Cowley, R.A. Kemp, M. Lattman and M.L. McKee, "Lewis Base Behavior of Methylated and Fluorinated Phosphines. A Photoelectron Spectroscopic Investigation," *Inorg. Chem.* **21**, 85 (1982).

82COW/LAT

A.H. Cowley, M. Lattman, P.M. Stricklen and J.G. Verkade, "UV Photoelectron Spectroscopic Investigation of Some Polycyclic Group 5A Compounds and Related Acyclic Species. 1. Free and Coordinated Aminophosphines and Related Compounds," *Inorg. Chem.* **21**, 543 (1982).

- 82CRE**
D. Cremer, *J. Comput. Chem.* **3**, 165 (1982).
- 82CRI/LIC**
K. Crimaldi, R.L. Lichter and A.D. Baker, "Nitrogen-15 Nuclear Magnetic Resonance and Photoelectron Spectroscopy of Substituted N-Phenylaziridines," *J. Org. Chem.* **47**, 3524 (1982).
- 82DEH/PRA**
P.M. Dehmer and S.T. Pratt, "Photoionization of ArKr, ArXe, and KrXe and Bond Dissociation Energies of the Rare Gas Dimer Ions," *J. Chem. Phys.* **77**, 4804 (1982).
- 82DEH/PRA2**
P.M. Dehmer and S.T. Pratt, "Photoionization of Argon Clusters," *J. Chem. Phys.* **76**, 843 (1982).
- 82DEK/WON**
R.L. DeKock, K.S. Wong and T.P. Fehlner, "Effects of Bridging Hydrogens on Metal-Metal Bonds. 2. UV Photoelectron and UV-Visible Spectra and Quantum Chemical Calculations for $\text{Fe}_3(\mu\text{-H})_3(\text{CO})_9(\mu^3\text{-CCH}_3)$ and $\text{Co}_3(\text{CO})_9(\mu^3\text{-CCH}_3)$," *Inorg. Chem.* **21**, 3203 (1982).
- 82DES/DUT**
P. Deshmukh, T.K. Dutta, J.L.-S. Hwang, C.E. Housecroft and T.P. Fehlner, "Photoelectron Spectroscopic Measurements of the Relative Charge on Carbyne Fragments Bound to Polynuclear Cobalt Carbonyl Clusters," *J. Am. Chem. Soc.* **104**, 1740 (1982).
- 82DEW**
M.J.S. Dewar, "Ionization Energies of p-Quinodimethane and 2,5-Dimethyl-p-quinodimethane," *J. Am. Chem. Soc.* **104**, 1447 (1982).
- 82DOL/MED**
W.R. Dolbier, Jr., K.S. Medinger, A. Greenberg and J.F. Liebman, *Tetrahedron* **38**, 2415 (1982).
- 82DRA/EUJ**
J.E. Drake, R. Eujen and K. Gorzelska, "ESCA and UV Photoelectron Spectra of Methyl(trifluoromethyl)germanes, $(\text{CF}_3)_4\text{-nGe}(\text{CH}_3)_n$ ($n = 1-3$), and Tetrakis(trifluoromethyl)germane, $(\text{CF}_3)_4\text{Ge}$," *Inorg. Chem.* **21**, 1784 (1982).
- 82DRA/GOR**
J.E. Drake, K. Gorzelska, G.S. White and R. Eujen, "Photoelectron Spectra of Trifluoromethyl(bromo)- and Trifluoromethyl(chloro)-Germanes $(\text{CF}_3)_4\text{-nGeX}_n$ ($X = \text{Br}, \text{Cl}; n = 1-3$)," *J. Electron Spectrosc. Rel. Phenom.* **26**, 1 (1982).
- 82DRA/GOR2**
J.E. Drake, K. Gorzelska, R. Helbing and R. Eujen, "UV and X-Ray Excited Photoelectron Spectra of Trifluoromethylgermanes $(\text{CF}_3)_4\text{-nGeH}_n$ ($n = 1-3$)," *J. Electron Spectrosc. Rel. Phenom.* **26**, 19 (1982).
- 82DRZ/BRA**
P.S. Drzaic and J.I. Brauman, "Electron Photodetachment of Sulfur Hexafluoride Anion. Comments on the Structure of SF_6^- ," *J. Am. Chem. Soc.* **104**, 13 (1982).
- 82DYK/GRA**
J.M. Dyke, B.W.J. Gravenor, R.A. Lewis and A. Morris, "Gas-Phase High Temperature Photoelectron Spectroscopy: An Investigation of the Transition Metals Iron, Cobalt, and Nickel," *J. Phys. B: At. Mol. Phys.* **15**, 4523 (1982).
- 82DYK/JON**
J.M. Dyke, N. Jonathan, A.E. Lewis and A. Morris, "Vacuum Ultraviolet Photoelectron Spectroscopy of Transient Species. Part 14. A Study of the Ground State of NF^+ via the Ionization Processes $\text{NF}^+(\text{X } 2\Pi) \rightarrow \text{NF}(\text{X } 3\Sigma, \text{a } 1\Delta)$," *J. Chem. Soc. Faraday Trans. II* **78**, 1445 (1982).
- 82DYK/JON2**
J. M. Dyke, N. Jonathan and A. Morris, "Recent Progress in the Study of Transient Species with Vacuum Ultraviolet Photoelectron Spectroscopy," *Int. Rev. Phys. Chem.* **2**, 3 (1982).
- 82DYK/JOS**
J.M. Dyke, G.D. Josland, R.A. Lewis and A. Morris, "Improved First Ionization Potential of the Dimethylsilaethylene Molecule Obtained with High-Temperature Photoelectron Spectroscopy," *J. Phys. Chem.* **86**, 2913 (1982).
- 82DYK/MOR**
J.M. Dyke, A. Morris and A. Ridha, "Study of the Ground State of PO^+ Using Photoelectron Spectroscopy," *J. Chem. Soc. Faraday Trans. II* **78**, 2077 (1982).
- 82DYK/MOR2**
J.M. Dyke, A. Morris, A.M.A. Ridha and J.G. Snijders, "Gas Phase High Temperature Photoelectron Spectroscopy: The Tin Monoxide Molecule," *Chem. Phys.* **67**, 245 (1982).
- 82EIS/DEM**
D. Eisel and W. Demtroder, "Accurate Ionization Potential of Li_2 from Resonant Two-Photon Ionization," *Chem. Phys. Lett.* **88**, 481 (1982).
- 82ELB/DIE**
S. Elbel, H.T. Dieck and R. Demuth, "Photoelectron Spectra of Group V Compounds. IX. The Relative Perfluoroalkyl Substituent Effect," *J. Fluorine Chem.* **19**, 349 (1982).
- 82ELL/ENG**
G.B. Ellison, P.C. Englekling and W.C. Lineberger, "Photoelectron Spectroscopy of Alkoxide and Enolate Negative Ions," *J. Phys. Chem.* **86**, 4873 (1982).
- 82EMO/HOR**
H.-H. Emons, W. Horlbeck and D. Kiessling, "Massenspektrometrische Untersuchung der Gasphase über Alkalimetalliodiden," *Z. Anorg. Allg. Chem.* **488**, 212 (1982).
- 82EMO/KIE**
H.-H. Emons, D. Kiessling and W. Horlbeck, "Dampfdruckmessungen und massenspektrometrische Untersuchungen der Gasphase über Erdalkalimetallhalogeniden," *Z. Anorg. Allg. Chem.* **488**, 219 (1982).
- 82ENG/SOL**
P.S. Engel, L.R. Soltero, S.A. Baughman, C.J. Nalepa, P.A. Cahill and R.B. Weisman, *J. Am. Chem. Soc.* **104**, 1698 (1982).
- 82ERA/KOL**
P.A. Erastov and V.P. Kolesov, *J. Chem. Thermodyn.* **14**, 103 (1982).
- 82ERM/KIR**
A.I. Ermakov, E.A. Kirichenko, N.I. Pimkin, Y.V. Chizhov and V.I. Kleimenov, "Photoelectronic Spectra and Electronic Structure of Organotri- and Organotetrasiloxanes of Linear and Cyclic Structure," *Zh. Struk. Khim.* **23**, 61 (1982).
- 82FAN/GIA**
R. Fantoni, A. Giardini-Guidoni and R. Tiribelli, "(c,2e) Spectroscopy of Valence States of the NO Molecule," *J. Electron Spectrosc. Rel. Phenom.* **26**, 99 (1982).
- 82FAR/SRI**
M. Farber, R.D. Srivastava and J.W. Moyer, "Mass Spectrometric Determination of the Thermodynamics of Potassium Hydroxide and Minor Potassium-Containing Species Required in Magnetohydrodynamic Power Systems," *J. Chem. Thermodyn.* **14**, 1103 (1982).
- 82FRA/FRA**
M.L. Fraser-Monteiro, L. Fraser-Monteiro, J.J. Butler, T. Baer and J.R. Hass, "Thermochemistry and Dissociation Dynamics of State-Selected $\text{C}_4\text{H}_8\text{O}_2^+$ Ions. 1. 1,4-Dioxane," *J. Phys. Chem.* **86**, 739 (1982).
- 82FRA/FRA2**
L. Fraser-Monteiro, M.L. Fraser-Monteiro, J.J. Butler and T. Baer, "Thermochemistry and Dissociation Dynamics of State-Selected $\text{C}_4\text{H}_8\text{O}_2^+$ Ions. 3. Ethyl Acetate," *J. Phys. Chem.* **86**, 752 (1982).
- 82FRE/IKU**
M.A. French, S. Ikuta and P. Kebarle, "Hydrogen Bonding of O-H and C-H Hydrogen Donors to Cl^- . Results from Mass Spectrometric Measurement of the Ion-Molecule Equilibria $\text{RH} + \text{Cl}^- = \text{RHCl}^-$," *Can. J. Chem.* **60**, 1907 (1982).

82FRO/LAU

D.C. Frost, W.M. Lau, C.A. McDowell and N.P.C. Westwood, "A Study by He I Photoelectron Spectroscopy of Monomeric Nitrosomethane, the Cis and Trans Dimers, and Formaldoxime," *J. Phys. Chem.* **86**, 3577 (1982).

82FUC

R. Fuchs, personal communication of unpublished heat of vaporization data.

82FUC/HAL

R. Fuchs, J.H. Hallman and M.O. Perlman, *Can. J. Chem.* **60**, 1832 (1982).

82FUC/PEA

R. Fuchs, L.A. Peacock and W.K. Stephenson, *Can. J. Chem.* **60**, 1953 (1982).

82FUR/SAK

J. Furukawa, M. Sakiyama, S. Seki, Y. Saito and K. Kusano, *Bull. Chem. Soc. Jpn.* **55**, 3329 (1982).

82GEL/CLE

G. Gellene, D. Cleary and R. Porter, "Stability of the Ammonium and Methylammonium Radicals from Neutralized Ion-Beam Spectroscopy," *J. Chem. Phys.* **77**, 3471 (1982).

82GIO/BOC

J. Giordon and H. Bock, "Radical Ions, 51. Oxidative Rearrangement of Diphenyl Disulfides to Thianthrenes," *Chem. Ber.* **115**, 2548 (1982).

82GLE/DOB

R. Gleiter, W. Dobler and M. Eckert-Maksic, "The Electronic Structure of o-Tropoquinone, p-Tropoquinone, and Cyclopentene-1,2,3-trione - PE Spectroscopic Investigations," *Nouv. J. Chim.* **6**, 123 (1982).

82GLE/ECK

R. Gleiter, M. Eckert-Maksic, W. Schafer and E.A. Truesdale, "Quest for a Strong Through Bond Interaction in [2.2]Paracyclophane," *Chem. Ber.* **115**, 2009 (1982).

82GLE/GUB

C.R. Gleiter and K. Gubernator, "Through-Bond- und Through-Space-Wechselwirkungen bei den Photodimeren von Acenaphthylen," *Chem. Ber.* **115**, 3811 (1982).

82GOM/CHA

M. Gomez, C. Chatillon and M. Allibert, "Thermodynamics of Gaseous and Condensed Indium Oxides by Mass Spectrometry with Controlled Oxygen Pressure," *J. Chem. Thermodyn.* **14**, 447 (1982).

82GRA/AJO

G. Granozzi, D. Ajo, C. Boschi and R. Roulet, "UV Photoelectron Spectra of Iron Tricarbonyl Complexes of 2,3,5,6-Tetrakis(methylene)-7-oxabicyclo[2.2.2]heptane," *J. Organometall. Chem.* **224**, 147 (1982).

82GRA/CAS

G. Granozzi, M. Casarin, D. Ajo and D. Osella, "Electron Structure of $[\{\text{Ni}(\eta^5\text{-C}_5\text{H}_5)(\mu\text{-CO})\}_2]$ by He(I) and He(II) Photoelectron Spectroscopy," *J. Chem. Soc. Dalton Trans.* **2047** (1982).

82GRA/TON

G. Granozzi, E. Tondello, D. Ajo, M. Casarin, S. Aime and D. Osella, "Gas Phase Helium I Photoelectron Spectra of Methyltricobalt Enneacarbonyl Clusters," *Inorg. Chem.* **21**, 1081 (1982).

82GRE/KEL

J.C. Green, M.R. Kelly, P.D. Grebenik, C.E. Briant, N.A. McEvoy and D.M.P. Mingos, "UV Photoelectron Spectral and Theoretical Studies on Tris(butadiene)-Molybdenum and -Tungsten," *J. Organometall. Chem.* **228**, 239 (1982).

82GRE/LIE

A. Greenberg and J.F. Liebman, *J. Org. Chem.* **47**, 2084 (1982).

82GRE/MCC

M.M. Green, R.J. McCluskey and J. Vogt, "A Comparison between the Stereoselective Thermal-Induced and Ionization-Induced Elimination of Acetic Acid from 2-Butyl Acetate," *J. Am. Chem. Soc.* **104**, 2262 (1982).

82GRE/PAY

J.C. Green, M. Payne, E.A. Seddon and R.A. Andersen, "He-I and He-II Photoelectron Studies of Bonding in Metal Silylamido-complexes, $\text{M}\{\text{N}(\text{SiMe}_3)_2\}_n$ ($n = 1, 2, \text{ or } 3$)," *J. Chem. Soc. Dalton Trans.* **887** (1982).

82GUI/KHA

C. Guimon, S. Khayar, G. Pfister-Guillouze, R.M. Claramunt and J. Elguero, "A Direct Photoelectron Spectroscopy Study of the 1-Azidopyridine Pyrolysis," *Spectrosc. Letters* **15**, 435 (1982).

82GUI/PFI

C. Guimon, G. Pfister-Guillouze and E. Rose, "Applications de la Spectroscopie Photoelectronique aux Proprietes Moleculaires. X. Influence de la Conformation des Arenes Chrome Tricarbonyle ortho Disubstitues sur Leur Structure Electronique et Leur Reactivite Nucleophile: Etude par Spectroscopie Photoelectronique UV(He(I), He(II)) et Calculs EHT," *J. Organometall. Chem.* **224**, 125 (1982).

82HA/NGU

T.-K. Ha and M.T. Nguyen, *Theochem.* **4**, 355 (1982).

82HAS/NEU

E. Haselbach, L. Neuhaus, R.P. Johnson, K.N. Houk and M.N. Paddon-Row, " π -Orbital Interactions in Mobius-Type Molecules as Studied by Photoelectron Spectroscopy," *Helv. Chim. Acta* **65**, 1743 (1982).

82HAY/IWA

T. Hayaishi, S. Iwata, M. Sasanuma, E. Ishiguro, Y. Morioka, Y. Iida and M. Nakamura, "Photoionisation Mass Spectrometric Study of Acetylene in the VUV Region," *J. Phys. B: At. Mol. Phys.* **15**, 79 (1982).

82HEP/TRE

J.W. Hepburn, D.J. Trevor, J.E. Pollard, D.A. Shirley and Y.T. Lee, "Multiphoton Ionization Photoelectron Spectroscopy of CCl_2F_2 and CCl_3F ," *J. Chem. Phys.* **76**, 4287 (1982).

82HIL

K. Hilpert, *J. Chem. Phys.* **77**, 1425 (1982).

82HIR/MOH

T. Hirabayashi, S. Mohmand and H. Bock, "Thermische Zersetzung ofenkettiger kialkyl-Sulfide, -disulfide und -diselenide," *Chem. Ber.* **115**, 483 (1982).

82HIT/HAO

A.P. Hitchcock, N. Hao, N.H. Werstiuk, M.J. McGlinchey and T. Ziegler, "Electronic Structure of $\text{Zr}(\text{BH}_4)_4$ and $\text{Hf}(\text{BH}_4)_4$ Studied by Photoelectron Spectroscopy and LCAO-HFS Calculations," *Inorg. Chem.* **21**, 793 (1982).

82HOL/BUR

J.L. Holmes, P.C. Burgers and Y.A. Mollah, "Alkane Elimination from Ionized Alkanols," *Org. Mass Spectrom.* **17**, 127 (1982).

82HOL/LOS

J.L. Holmes, F.P. Lossing, J.K. Terlouw and P.C. Burgers, "The Radical Cation $[\text{CH}_2\text{OH}_2]^+$ and Related Stable Gas Phase Ion-Dipole Complexes," *J. Am. Chem. Soc.* **104**, 2931 (1982).

82HOL/LOS2

J.L. Holmes and F.P. Lossing, "Towards a General Scheme for Estimating the Heats of Formation of Organic Ions in the Gas Phase. Part II. The Effect of Substitution at Charge-Bearing Sites," *Can. J. Chem.* **60**, 2365 (1982).

82HOL/LOS3

J.L. Holmes and F.P. Lossing, "Heats of Formation of the Ionic and Neutral Enols of Acetaldehyde and Acetone," *J. Am. Chem. Soc.* **104**, 2648 (1982).

82HOL/SMI

J.B. Holbrook, B.C. Smith, C.E. Housecroft and K. Wade, *Polyhedron* **1**, 701 (1982).

82HON/EAT

E. Honegger, P.E. Eaton, B.K. Ravi Shankar and E. Heilbronner, "195. The Electronic Structure of Pentaprismane ($\text{C}_{10}\text{H}_{10}$) as Revealed by its Photoelectron Spectrum," *Helv. Chim. Acta* **65**, 1982 (1982).

- 82HUB/LIC**
J.L. Hubbard and D.L. Lichtenberger, "Vibrational Fine Structure in the Valence Ionizations of Transition-Metal Hexacarbonyls: New Experimental Indication of Metal-to-Carbonyl π Bonding," *J. Am. Chem. Soc.* **104**, 2132 (1982).
- 82IKE/TAM**
Y. Ikeda, M. Tamaki, G. Matsumoto, K. Amioka and T. Mizuno, "Mass Spectrometric Studies of Lithium-Containing Oxides at High Temperatures," *Spectrochim. Acta* **37B**, 647 (1982).
- 82IKU/KEB**
S. Ikuta, P. Kebarle, G.M. Bancroft, T. Chan and R.J. Puddephatt, "Basicities of Methyl-, Methylphenyl-, and Phenylphosphines in the Gas Phase," *J. Am. Chem. Soc.* **104**, 5899 (1982).
- 82INA/MUR**
S. Inagaki, S. Murata, M. Sakiyama, Y. Ito, Y. Umihara, T. Hijiya and T. Matsura, *Bull. Chem. Soc. Jpn.* **55**, 2803 (1982).
- 82INA/MUR2**
S. Inagaki, S. Murata and M. Sakiyama, *Bull. Chem. Soc. Jpn.* **55**, 2808 (1982).
- 82JANAF**
M.J. Chase, Jr., J.L. Curnutt, J.R. Downy Jr., R.A. McDonald, A.N. Syverud and E.A. Valenzuela, "JANAF Thermochemical Tables 1982 Supplement," *J. Phys. Chem. Ref. Data* **11**, 695 (1982).
- 82JOH/KIM**
C.E. Johnson, K.Y. Kim and M. Mansson, Personal communication.
- 82JOH/POW**
K. Johnson, I. Powis and C.J. Danby, "A Photoelectron-Photoion Coincidence Study of Acetaldehyde and Ethylene Oxide Molecular Ions," *Chem. Phys.* **70**, 329 (1982).
- 82JON/DEL**
G. Jonkers, C.A. De Lange and J.G. Snijders, "Effects of Relativity in the He(I) Photoelectron Spectrum of Cl_4 ," *Chem. Phys.* **69**, 109 (1982).
- 82JON/GRA**
G. Jonkers, O. Grabandt, R. Mooyman and C.A. De Lange, "He(I) Photoelectron Spectroscopy of Transient Species: Fluorothiocyanate (FSCN)," *J. Electron Spectrosc. Rel. Phenom.* **26**, 147 (1982).
- 82JON/VAN**
G. Jonkers, S.M. Van Der Kerk and C.A. De Lange, "He(I) Photoelectron Spectroscopy of Transient Species: Germanium Dichloride and Germanium Dibromide," *Chem. Phys.* **70**, 69 (1982).
- 82JON/VAN2**
G. Jonkers, S.M. Van Der Kerk, R. Mooyman, C.A. De Lange and J.G. Snijders, "He(I) Photoelectron Spectroscopy of Tetraiodoethylene (C_2I_4)," *Chem. Phys.* **69**, 115 (1982).
- 82JON/VAN3**
G. Jonkers, S.M. Van Der Kerk, R. Mooyman and C.A. De Lange, "UV Photoelectron Spectroscopy of Transient Species: Germanium Difluoride (GeF_2)," *Chem. Phys. Lett.* **90**, 252 (1982).
- 82JOS**
R.M. Joshi, *J. Macromol. Sci. Chem.* **A18**, 861 (1982).
- 82KAS/CHE**
O.E. Kashireninov, A.D. Chervonnyi and V.A. Piven, *High Temp. Sci.* **15**, 79 (1982).
- 82KAT/SHI**
S. Katsumata, H. Shiromaru, K. Mitani, S. Iwata and K. Kimura, "Photoelectron Angular Distribution and Assignments of Photoelectron Spectra of Nitrogen Dioxide, Nitromethane and Nitrobenzene," *Chem. Phys.* **69**, 423 (1982).
- 82KLA/SAB**
L. Klasinc, A. Sabljic, G. Kluge, J. Rieger and M. Scholz, "Chemistry of Excited States. Part 13. Assignment of Lowest π -Ionizations in Photoelectron Spectra of Thiophen, Furan, and Pyrrole," *J. Chem. Soc. Perkin Trans. II* **539** (1982).
- 82KOB/KUB**
T. Kobayashi, T. Kubota, K. Ezumi and C. Utsunomiya, "Photoelectron Angular Distribution Study of Some Isoxazoles Combined with Perturbation Theoretic Approach," *Bull. Chem. Soc. Jpn.* **55**, 3915 (1982).
- 82KOZ/MAS**
M.P. Kozina, V.S. Mastryunov and E.M. Mil'vitshaya, *Russ. Chem. Rev.* **1**, 765 (1982).
- 82KRE/SCH**
J. Kreile, A. Schweig and W. Thiel, "Experimental and Theoretical Investigation of the Photoionization of Hydrogen Cyanide," *Chem. Phys. Lett.* **87**, 473 (1982).
- 82KUT/EDW**
R. Kutina, A. Edwards, G. Goodman and J. Berkowitz, "Photoionization Mass Spectrometry of CH_3SH , CD_3SH , and CH_3SD : Heats of formation of CH_3S^+ , $(\text{CH}_2\text{SH})^+$, CH_2S^+ , CH_2S , and HCS^+ ," *J. Chem. Phys.* **77**, 5508 (1982).
- 82KUT/GOO**
R.E. Kutina, G.L. Goodman and J. Berkowitz, "Photoionization Mass Spectrometry of NH_2OH : Heats of Formation of HNO^+ and NOH^+ ," *J. Chem. Phys.* **77**, 1664 (1982).
- 82LEL/BAL**
L. Lelik, V.K. Balthazarne, O.E. Kaposi and G. Sajo, "Homo- es heterokomplexek kezodesenek magashomersekleto tomegspektrometriás vizsgalata NaI/CsI rendszér egyensulyi gozterebeben, I.," *Magyar Kem. Fol.* **88**, 513 (1982).
- 82LEV/LIA**
R. Levin and S.G. Lias, "Ionization Potential and Appearance Potential Measurements, 1971-1981," *Nat. Stand. Ref. Data Ser., Nat. Bur. Stand. (U.S.)* **71**, (1982).
- 82LIA**
S. Lias, "Thermochemical Information from Ion-Molecule Rate Constants," *Ion-Cyclotron Resonance Spectrometry II* (H. Hartmann and K.-P. Wanczek, editors), Springer-Verlag, Berlin (1982), p. 409.
- 82LIF**
C. Lifshitz, "Time-Dependent Mass Spectra and Breakdown Graphs. 2. The Kinetic Shift in Pyridine," *J. Phys. Chem.* **86**, 606 (1982).
- 82LIF2**
C. Lifshitz, "A Surprisal Analysis of the Dissociation Dynamics of $\text{C}_3\text{H}_6\text{O}^+$ Cations," *Int. J. Mass Spectrom. Ion Phys.* **43**, 179 (1982).
- 82LOU/AND**
J.N. Louwen, R.R. Andrea, D.J. Stufkens and A. Oskam, "He(I) and He(II) Photoelectron Spectra of $\text{M}[\text{Co}(\text{CO})_4]_2$ and $\text{M}[\text{Mn}(\text{CO})_5]_2$ Complexes ($\text{M} = \text{Zn}, \text{Cd}, \text{and Hg}$)," *Z. Naturforsch.* **37b**, 711 (1982).
- 82LOU/HAR**
J.N. Louwen, J. Hart, D.J. Stufkens and A. Oskam, "The HeI and HeII Photoelectron Spectra of $[\text{Fe } \eta^3\text{-C}_3\text{H}_5(\text{CO})_3\text{X}]$ and $\text{Fe } \eta^3\text{-C}_4\text{H}_7(\text{CO})_3\text{X}$ ($\text{X} = \text{Cl}, \text{Br}, \text{I}$) and $[\text{CoC}_3\text{H}_5(\text{CO})_3]$," *Z. Naturforsch.* **37b**, 179 (1982).
- 82LOU/VAN**
J.N. Louwen, H. Van Dam, D.J. Stufkens, A. Oskam and H.H. Jaffe, "The Electronic Structures of Some Aromatic Sulfinylamines," *J. Electron Spectrosc. Rel. Phenom.* **26**, 235 (1982).
- 82MAC**
A. Maccoll, "Ion Enthalpies and Their Application in Mass Spectrometry," *Org. Mass Spectrom.* **17**, 1 (1982).
- 82MAI/MIS**
J.P. Maier, L. Misev and F. Thommen, "Dicyanoacetylene Cation. Laser-Induced Fluorescence and Photoelectron-Photon Coincidence Studies," *J. Phys. Chem.* **86**, 514 (1982).

82MAI/THO

J.P. Maier and F. Thommen, "Relaxation of Dihaloacetylene Cations in Their A and B States Studied by Photoelectron-Photoion Coincidence Spectroscopy," *Chem. Phys.* **70**, 325 (1982).

82MAI/THO2

J.P. Maier and F. Thommen, "Photoelectron-Photoion Coincidence Studies of Halobenzene Cations in Their Excited Electronic States," *J. Chem. Phys.* **77**, 4427 (1982).

82MAR/CHE

S. Martin, J. Chevalere, S. Valignat, J.P. Perrot, M. Broyer, B. Cabaud and A. Hoareau, "Autoionizing Rydberg States of the Na₂ Molecule," *Chem. Phys. Lett.* **87**, 235 (1982).

82MAR/THR

D.S. Marynick, L. Throckmorton and R. Backquet, *J. Am. Chem. Soc.* **104**, 1 (1982).

82MAS/BOH

R.S. Mason, D.K. Bohme and K.R. Jennings, "Gas Phase Basicities of Halogenotoluenes," *J. Chem. Soc. Faraday Trans. I* **78**, 1943 (1982).

82MCI/FUK

R.T. McIver, Jr. and E.K. Fukuda, "Equilibrium Electron Affinities," *Lec. Notes in Chem.* **31**, 165. Data are anchored to EA(SO₂) Ref. 74CEL/BEN.

82MCM/GOL

D.F. McMillen and D.M. Golden, "Hydrocarbon Bond Dissociation Energies," *Ann. Rev. Phys. Chem.* **33**, 493 (1982).

82MEI/HAN

H. Meier, N. Hanold and H. Kolshorn, *Angew. Chem. Int. Ed.* **21**, 66 (1982).

82MIK/TRU

A.I. Mikaya, E.A. Trusova, V.G. Zaikin, E.N. Karaulova and L.M. Petrova, "Ionization Energy and Appearance Energy in Organic Chemistry. Communication 5. Energy Differences of Stereoisomers of 2-Methyl-1-Thiadecalin," *Izv. Akad. Nauk SSSR, Ser. Khim.* **1479** (1982); English trans.: *Bull. Acad. Sci. USSR Div. Chem. Sci.*, 1319 (1982).

82MIN/SAB

C. Mindakis and R. Sabbah, *Thermochim. Acta* **55**, 147 (1982).

82MIS/POK

B.G. Mishchanchuk, V.A. Pokrovskii, V.P. Shabel'nikov and E.N. Korol, "Mass Spectrometric Study of Energy Characteristics of Methanol and Ethanol Ions During Ionization by a Strong Electric Field," *Teor. Eks. Khim.* **18**, 307 (1982).

82MOR/MER

J.-P. Morizur, J. Mercier and M. Sarraf, "2-Substituted-2,3-dihydro-4H-pyrans: Competition between 'Retro Diels-Alder' Fragmentation and Substituent Loss," *Org. Mass Spectrom.* **17**, 327 (1982).

82NAT/PEN

P. Natalis, P. Pannetier, L. Longton and J.E. Collin, "Ionisation Energy Values for the Vibronic Transitions from HCl X¹Σ⁺ (v' = 0) to HCl⁺ Ionic States X²Π (v' = 0-13) and A²Σ⁺ (v' = 0-12), Determined by Photoelectron Spectroscopy," *J. Electron Spectrosc. Rel. Phenom.* **27**, 267 (1982).

82NEL/GAN

S.F. Nelsen and P.M. Gannett, "The 3,3-Dimethyl-2-azabicyclo[2.2.2]octyl System as a Bredt's Rule Kinetically Stabilized Dialkylamino Group in Electron Transfer Studies," *J. Am. Chem. Soc.* **104**, 5292 (1982).

82NIS/SHI

N. Nishi, H. Shinohara and I. Hanazaki, *J. Chem. Phys.* **77**, 246 (1982).

82OLE/TUR

J.J. Oleksik and A.G. Turner, *Inorg. Chim. Acta* **55**, 165 (1984).

82PAD/PAT

M.N. Paddon-Row, H.K. Patney and R.S. Brown, "Orbital Interactions. XI. Application of Photoelectron Spectroscopy to the Study of Orbital Interactions in Some 2- and 9-Substituted Octahydrodimethanonaphthalenes," *Aust. J. Chem.* **35**, 293 (1982).

82PAM/ROG

M. Pamidimukkala, D. Rogers and G.B. Skinner, *J. Phys. Chem. Ref. Data* **11**, 83 (1982).

82PAP/KOL

T.S. Papina and V.P. Kolesov, *Russ. J. Phys. Chem.* **56**, 675 (1982).

82PAU/DAL

J.F. Paulson and J. Dale, "Reactions of OH·-H₂O with NO₂," *J. Chem. Phys.* **77**, 4006 (1982).

82PAU/HEH

C.F. Pau and W.J. Hehre, *J. Phys. Chem.* **86**, 321 (1982).

82PAU/HEH2

C.F. Pau and W.J. Hehre, *J. Phys. Chem.* **86**, 1282 (1982).

82PFI/GER

G. Pfister-Guillouzo, S. Geribaldi and J.-F. Gal, "Spectres Photoelectroniques de Cyclohexene-2-ones-1 Diversement Substituees en Position 3. Correlations avec la Reactivite," *Can. J. Chem.* **60**, 1163 (1982).

82PIE/HEH

W.J. Pietro and W.J. Hehre, *J. Am. Chem. Soc.* **104**, 4329 (1982).

82PIL/SKI

G. Pilcher and H.A. Skinner, "The Chemistry of the Metal-Carbon Bond" (ed. F.R. Hartley and S. Patai, John Wiley & Sons, New York, 1982).

82POL/DEH

E.D. Poliakoff, P.M. Dehmer, J.L. Dehmer and R. Stockbauer, "Photoelectron-photoion Coincidence Spectroscopy of Gas-Phase Clusters," *J. Chem. Phys.* **76**, 5214 (1982).

82PRA/DEH

S.T. Pratt and P.M. Dehmer, "Photoionization of the Kr₂ Dimer," *Chem. Phys. Lett.* **87**, 533 (1982).

82PRA/DEH2

S.T. Pratt and P.M. Dehmer, "Photoionization of the Neon-Rare Gas Dimers NeAr, NeKr, and NeXe," *J. Chem. Phys.* **76**, 3433 (1982).

82ROG

D.W. Rogers, *Am. Lab.*, p. 15, et. seq. (Jan. 1982).

82ROS/BUF

H.M. Rosenstock, R. Buff, M.A.A. Ferreira, S.G. Lias, A.C. Parr, R.L. Stockbauer and J.L. Holmes, "Fragmentation Mechanism and Energetics of Some Alkyl Halide Ions," *J. Am. Chem. Soc.* **104**, 2337 (1982).

82ROS/DAN

H.M. Rosenstock, J. Dannacher and J.F. Liebman, "The Role of Excited Electronic States in Ion Fragmentation: C₆H₆⁺," *Radiat. Phys. Chem.* **20**, 7 (1982).

82ROY/MCM

M. Roy and T.B. McMahon, *Org. Mass Spectrom.* **17**, 392 (1982).

82ROZ/HOU

M.D. Rozeboom and K.N. Houk, "Stereospecific Alkyl Group Effects on Amine Lone-Pair Ionization Potentials: Photoelectron Spectra of Alkylpiperidines," *J. Am. Chem. Soc.* **104**, 1189 (1982).

82ROZ/HOU2

M.D. Rozeboom, K.N. Houk, S. Searles and S.E. Seyedrezai, "Photoelectron Spectroscopy of N-Aryl Cyclic Amines. Variable Conformations and Relationships to Gas- and Solution-Phase Basicities," *J. Am. Chem. Soc.* **104**, 3448 (1982).

82SAB/GOM

R. Sabbah and L.A.T. Gomez, *Thermochim. Acta* **52**, 285 (1982).

- 82SAI/YAM**
Y. Saito, K. Yamauchi, K. Mihama and T. Noda, "Formation and Ionization Potentials of Lead Clusters," *Japanese J. Appl. Phys.* **21**, L396 (1982).
- 82SCH/MEA**
P.A. Schulz, R.D. Mead, P.L. Jones and W.C. Lineberger, "OH⁻ and OD⁻ Threshold Photodetachment," *J. Chem. Phys.* **77**, 1153 (1982).
- 82SCH/SCH**
R. Schulz and A. Schweig, "Photoelectron Spectra of Thermally Generated Unstable Organic Compounds - 2-Propene-1-imine," *J. Elec. Spectros. Rel. Phenom.* **28**, 33 (1982).
- 82SCO/ERD**
L.T. Scott, I. Erden, W.R. Brunsvold, T.H. Schultz, K.N. Houk and M.N. Paddon-Row, "Competitive [6 + 2], [4 + 2], and [2 + 2] Cycloadditions. Experimental Classification of Two-Electron Cycloaddends," *J. Am. Chem. Soc.* **104**, 3659 (1982).
- 82SEL/HEL**
E.T.M. Selim and A.I. Helal, "The Study of C₁-C₃ Monosubstituted Alkyl Benzenes by the Inverse Convolution of First Differential Ionization Efficiency Curves," *Org. Mass Spectrom.* **17**, 539 (1982).
- 82SHA**
W.M. Shaub, *Thermochim. Acta* **55**, 59 (1982).
- 82SHE/HAL**
D.E. Sherwood, Jr. and M.B. Hall, "Electronic Structure of Metal Clusters. 2. Photoelectron Spectrum and Molecular Orbital Calculations of Decacarbonyldihydridotriosmium," *Inorg. Chem.* **21**, 3458 (1982).
- 82SHI/GIN**
I. Shim and K.A. Gingerich, *J. Chem. Phys.* **77**, 2490 (1982).
- 82SID**
L.N. Sidirov, "Molecules With A High Electron Affinity," *Russ. Chem. Rev.* **51**, 356 (1982).
- 82SID/GUB**
L.N. Sidirov and G.D. Gubarevich, "Dissociation of the Gaseous Complex Fluorides. Alkali Metal Trifluoromanganates and Tetrafluoromanganates," *Koord. Khim.* **8**, 463. CA: 97, 12708c (1982).
- 82SIE/MAU**
L.W. Sieck and M. Mautner (Meot-Ner), "Ionization Energies and Entropies of Cycloalkanes. Kinetics of Free Energy Controlled Charge-Transfer Reactions," *J. Phys. Chem.* **86**, 3646 (1982).
- 82SKO/SOR**
E.V. Skokan, I.D. Sorokin, M.I. Nikitin, N.S. Chilingarov and L.N. Sidorov, "The Electron Affinity of UF₆," *Russ. J. Phys. Chem.* **57**, 1745 (1983).
- 82SOL/BOC**
B. Solouki, H. Bock, R. Appel, A. Westerhaus, G. Becker and G. Uhl, "Photoelektronen-Spektren von Methylidindiphosphanen R-CP," *Chem. Ber.* **115**, 3747 (1982).
- 82SPA/GLE**
J. Spanget-Larsen, R. Gleiter, K. Gubernator, R.J. Ternansky and L.A. Paquette, "Electronic and Molecular Structure of Simple Bicyclopropyls. Photoelectron Spectroscopy and Model Calculations," *J. Org. Chem.* **47**, 3082 (1982).
- 82SPA/KOR**
J. Spanget-Larsen, C. de Korswagen, M. Eckert-Maksic and R. Gleiter, "90. Electronic and Molecular Structure of Simple 3,3'-Bicyclopropenyls. Photoelectron Spectroscopy and Model Calculations," *Helv. Chim. Acta* **65**, 968 (1982).
- 82SPL/CAL**
J.S. Splitter and M. Calvin, "Conformational Dependence in the Mass Spectrum of Cyclohexanecarboxaldehyde," *J. Org. Chem.* **47**, 4545 (1982).
- 82SQU/DEP**
R.R. Squires and C.H. DePuy, "Flowing Afterglow Studies of the Reactions Between Negative Ions and Trimethylsilyl Enol Ethers. Regiospecific Generation of Gas Phase Enolate Ions," *Org. Mass Spectrom.* **17**, 187 (1982).
- 82SUR/ELS**
S. Suradi, N. El-Saiad, G. Pilcher and H.A. Skinner, *J. Chem. Thermodyn.* **14**, 45 (1982).
- 82TN270**
D.D. Wagman, W.H. Evans, V.B. Parker, R.H. Schumm, I. Halow, S.M. Bailey, K.L. Churney, R.L. Nuttall, "The NBS Tables of Chemical Thermodynamic Properties (NBS Tech Note 270)," *J. Phys. Chem. Ref. Data* **11**, Supl. 1 (1982).
- 82TOR/SAB**
L.A.F. Torres Gomez and R. Sabbah, *Thermochim. Acta* **57**, 67 (1982).
- 82TOR/SAB2**
L.A.F. Torres Gomez and R. Sabbah, *Thermochim. Acta* **58**, 311 (1982).
- 82TIPIS**
L.V. Gurvich, I.V. Veits, V.A. Medvedev, G.A. Khachkuruzov, V.S. Yungman and G.A. Bergman, et al., "Termodinamicheskoe Svoistva Individual'nykh Veshchestv" (Thermodynamic Properties of Individual Substances); Glushko, V.P., gen. ed., Vol. 4, parts 1 and 2(1982), Izdatel'stvo "Nauka" Moscow.
- 82TRA/MCL**
J.C. Trager, R.G. McLouglin and A.J.C. Nicholson, "Heat of Formation for Acetyl Cation in the Gas Phase," *J. Am. Chem. Soc.* **104**, 5318 (1982).
- 82TRA/RED**
V.F. Traven', V.V. Redchenko and B.I. Stepanov, "Parameterization of Quantum Chemical Calculations of Organic Compounds of Silicon by Means of Photoelectron Spectroscopy and the Electronic Spectroscopy of CTC," *Zh. Obs. Khim.* **52**, 2262 (1982); English trans.: *J. Gen. Chem. (USSR)* **52**, 2015 (1982).
- 82TRA/RED2**
V.F. Traven', V.V. Redchenko, V.Y. Bartenev and B.I. Stepanov, "Photoelectron Spectra of Permethyl-di-, -tri-, and -tetra-germanes," *Zh. Obs. Khim.* **52**, 358 (1982). English trans.: *J. Gen. Chem. (USSR)* **52**, 310 (1982).
- 82TRA/ROD**
V.F. Traven', O.G. Rodin, V.V. Redchenko, T.A. Chibisova and B.I. Stepanov, "Photoelectron Spectra of Selenanthrene and Dibenzoselenophene," *Zh. Obs. Khim.* **52**, 2650 (1982); English trans.: *J. Gen. Chem. (USSR)* **52**, 2345 (1982).
- 82VON/ASB**
W. Von Niessen, L. Asbrink and G. Bieri, "30.4 nm He(II) Photoelectron Spectra of Organic Molecules. Part VI. Halogeno-Compounds (C, H, X: X = Cl, Br, I)," *J. Electron Spectrosc. Rel. Phenom.* **26**, 173 (1982).
- 82WAR/PEA**
P.M. Warner and S. Peacock, *J. Comput. Chem.* **3**, 417 (1982).
- 82WIB/WEN**
K.B. Wiberg and J.J. Wendolowski, *J. Am. Chem. Soc.* **104**, 5679 (1982).
- 82WOR/HAR**
S.D. Worley, J.H. Hargis, L. Chang, G.A. Mattson and W.B. Jennings, "The UPS of Some Compounds Containing the Heteroatoms Phosphorus, Nitrogen and Oxygen," *J. Electron Spectrosc. Rel. Phenom.* **25**, 135 (1982).
- 82WOR/WEB**
S.D. Worley, T.R. Webb and T.Y. Ou, "He(I) Photoelectron Spectra of Three Novel Ruthenium Tricarbonyl π Complexes," *J. Electron Spectrosc. Rel. Phenom.* **28**, 129 (1982).
- 82YOK/QUI**
A. Yokozeki, E.L. Quitevis and D.R. Herschbach, "Electron Attachment to Volatile Uranyl Molecules," *J. Phys. Chem.* **86**, 617 (1982).

"Thermochemistry of peroxides" in *The Chemistry of Functional Groups, Peroxides* ed. S. Patai

82ZVE/ASH

V.V. Zverev, L.K. Ashrafullina, Y.Y. Villem, N.V. Villem and R.R. Shagidullin, "Photoelectron Spectrum and Electronic and Steric Structure of o-Aminothiophenol," *Izv. Akad. Nauk SSSR, Ser. Khim.* 2165 (1982).

82ZVE/VIL

V.V. Zverev, Y.Y. Villem, N.V. Villem, E.N. Klimovitskii and B.A. Arbuзов, "Photoelectron Spectra and Intramolecular Interactions of Dimethoxymethane and 4,7-Dihydro-1,3-dioxepin," *Zh. Obsh. Khim.* 52, 1888 (1982); *Engl. trans.. J. Gen. Chem. USSR* 52, 1775 (1982).

83ALE/FED

V.L. Alekseev, L.I. Fedorova and A.V. Baluev, "Mass Spectrometric Study of Thermochemical Characteristics of Perchloryl Fluoride and its Decomposition Product Chlorosyl Fluoride," *Izv. Akad. Nauk SSR, Ser. Khim.* 1084 (1983).

83ALI/CON

G. Al-Takhin, J.A. Connor and H.A. Skinner, *J. Organomet. Chem.* 259, 313 (1983).

83AN/MAN

X.-W. An and M. Mansson, *J. Chem. Thermodyn.* 15, 287 (1983).

83AND/CAU

E. Andoni, C. Cauletti and C. Furlani, "Electronic Structure and Bonding in 2,2-Diphenyl-1,3-dithiacyclopentanes of IV A Group Elements Studied by UV Photoelectron Spectroscopy," *Inorg. Chim. Acta* 76, L35 (1983).

83AUD/MIL

H.E. Audier, A. Milliet and J.-P. Denhez, "Mecanismes d'Isomerisation des Indanols-2 en Spectrometrie de Masse," *Bull. Soc. Chim. France II-202* (1983).

83BAK/ARM

A.D. Baker, G.H. Armen and S. Funaro, "Oral Levels of Crown Ethers and Related Macrocycles Studies by Ultraviolet Photoelectron Spectroscopy: Relationship to Complexation Studies," *J. Chem. Soc. Dalton Trans.*, 2519 (1983).

83BAL/GIG

G. Balducci, G. Gigli and M. Guido, "Thermodynamic Study of Gaseous Ternary Oxide Molecules. The Europium-Vanadium-Oxygen System," *J. Chem. Phys.* 79, 5623 (1983).

83BAL/NEU

T. Bally, L. Neuhaus, S. Nitsche, E. Haselbach, J. Janssen and W. Luttke, "Cross Conjugated Polyenes Derived from 2-Vinylbutadiene: Electronic States of Their Radical Cations and Triplet Energy," *Helv. Chim. Acta* 66, 1288 (1983).

83BAN/BRI

G.M. Bancroft and D.J. Bristow, "High Resolution HeI and HeII Photoelectron Spectra of the Thallium Halides: Valence Bands and Tl 5d Ligand Field Splittings," *Can. J. Chem.* 61, 2669 (1983).

83BAN/DAV

J.A. Bandy, C.E. Davies, J.C. Green, M.L.H. Green, K. Prout and D.P.S. Rodgers, "Synthesis, Crystal Structures, and Bonding of the Molybdenum Cubane Compounds $[Mo(\mu-C_5H_4Pr)(\mu^3-S)]_4^{n+}$, where $n = 0, 1, \text{ and } 2$," *J. Chem. Soc. Chem. Commun.* 1395 (1983).

83BAR/BAS

J.E. Bartmess, T. Basso and R.M. Georgiadis, "The Electron Affinity of a Nitroxide Radical," *J. Phys. Chem.* 87, 912 (1983).

83BIE/MOR

H.W. Biermann and T.H. Morton, "Reversible Tautomerization of Radical Cations. Photoionization of 2-Methoxyethanol and 3-Methoxy-1-propanol," *J. Am. Chem. Soc.* 105, 5025 (1983).

83BIN

M. Binnewies, "Bildung und Stabilität von gasförmigem AsOCl und SbOCl Massenspektrometrische Untersuchungen," *Z. Anorg. Allg. Chem.* 505, 32 (1983).

83BIN/LAK

M. Binnewies, M. Lakenbrink and H. Schnockel, *Z. Anorg. Chem.* 497, 7 (1983).

83BOA/HOU

G. Boand, R. Houriet and T. Baumann, "The Gas Phase Acidity of Aliphatic Alcohols," *J. Am. Chem. Soc.* 105, 2203 (1983).

83BOC/MOH

H. Bock, S. Mohmand, T. Hirabayashi, G. Maier and H.P. Reisenauer, "Photoelektronen-spektroskopischer Nachweis und Matrix-Isolierung von Thio-para-benzochinonen," *Chem. Ber.* 116, 273 (1983).

83BOC/ROT

H. Bock and B. Roth, "Radical Ions. 49. Redox Reactions of Some Thiophene Derivatives," *Phosphorus and Sulfur* 14, 211 (1983).

83BOH/GLE2

M.C. Bohm, R. Gleiter, T.A. Albright and V. Sriyungwat, "The Low-Lying Cationic Hole-States of η^3 -Allyl Tetracarbonyl Complexes of Mn and Re. Photoelectron Spectra and Green's Function Calculations," *Molec. Phys.* 50, 113 (1983).

83BOM/DAN

R. Bombach, J. Dannacher and J.-P. Stadelmann, "The Rate-Energy Functions for the Formation of Tropylium and Benzylium Ions from Toluene Molecular Cations," *Chem. Phys. Lett.* 95, 259 (1983).

83BOM/DAN2

R. Bombach, J. Dannacher and J.-P. Stadelmann, "Energy and Time Dependence of the Decay Processes of Toluene Molecular Cations," *J. Am. Chem. Soc.* 105, 4205 (1983).

83BOM/DAN3

R. Bombach, J. Dannacher, J.-P. Stadelmann and R. Neier, "Fundamental Aspects of Ionic Dissociations: The Fragmentation Pathways of Excited Bicyclobutane Cations," *Helv. Chim. Acta* 66, 701 (1983).

83BOM/DAN4

R. Bombach, J. Dannacher, E. Honegger, J.-P. Stadelmann and R. Neier, "Unimolecular Dissociations of Excited $C_3H_6^+$: A Photoelectron-Photoion Coincidence Study of Cyclopropanol and Allyl Alcohol," *Chem. Phys.* 82, 459 (1983).

83BRA/BAE

W.A. Brand and T. Baer, "Unimolecular Dissociation of Energy-Selected t-Butylbenzene Ions and Effect of Thermal Energy on Data Analysis," *Int. J. Mass Spectrom. Ion Phys.* 49, 103 (1983).

83BRA/BAE2

W.A. Brand, T. Baer and C.E. Klots, "Kinetic Energy Release Distributions in the Fragmentation of Energy-Selected Iodopropane Ions," *Chem. Phys.* 76, 111 (1983).

83BRO/BUS

R.S. Brown, J.M. Buschek, K.R. Kopecky and A.J. Miller, "Photoelectron Spectra of syn- and anti-Sesquinorbornene. Evidence for Vertical $\Sigma-\pi$ Delocalization in Bicyclo[2.2.1]heptene," *J. Org. Chem.* 48, 3692 (1983).

83BUR/CAS

T.J. Burkey, A.L. Castelhana, D. Griller and F.P. Lossing, "Heats of Formation and Ionization Potentials of Some α -Aminoalkyl Radicals," *J. Am. Chem. Soc.* 105, 4701 (1983).

83BUR/HOL

P.C. Burgers, J.L. Holmes, A.A. Mommers and J.E. Szulejko, "The Collisionally Induced Dissociation of Allyl and 2-Propenyl Cations," *Org. Mass Spectrom.* 18, 596 (1983).

83BUR/HOL2

P.C. Burgers, J.L. Holmes, J.E. Szulejko, A.A. Mommers and J.K. Terlouw, "The Gas Phase Ion Chemistry of the Acetyl Cation and Isomeric $[C_2H_3O]^+$ Ions. On the Structure of the $[C_2H_3O]^+$ Daughter Ions Generated from the Enol of Acetone Radical Cation," *Org. Mass Spectrom.* 18, 254 (1983).

- 83BUR/HOL3**
P.C. Burgers, J.L. Holmes, F.P. Lossing, F.R. Povel and J.K. Terlouw, "The Role of Charge-site Location in Fragmenting Ions. 1- $[\text{CH}_3\text{CHXCOOCH}_3]^+$ and $[\text{CH}_2\text{XCH}_2\text{COOCH}_3]^+$ Ions and the Structures of Daughter Ions Derived from the Loss of $\text{X}(\text{X} = \text{I, Br, Cl and CH}_3)$," *Org. Mass Spectrom.* **18**, 335 (1983).
- 83BUR/MOM**
P.C. Burgers, A.A. Mommers and J.L. Holmes, "Ionized Oxycarbenes: $[\text{COH}]^+$, $[\text{HCOH}]^+$, $[\text{C}(\text{OH})_2]^+$, $[\text{HCO}_2]^+$, and $[\text{COOH}]^+$, Their Generation, Identification, Heat of Formation, and Dissociation Characteristics," *J. Am. Chem. Soc.* **105**, 5976 (1983).
- 83BUR/STE**
S.M. Burnett, A.E. Stevens, C.S. Feigerle and W.C. Lineberger, "Observation of X^1A_1 Vinylidene by Photoelectron Spectroscopy of the C_2H_2^- Anion," *Chem. Phys. Lett.* **100**, 124 (1983).
- 83BUT/BAE**
J.J. Butler, T. Baer and S.A. Evans, Jr., "Energetics and Structures of Organosulfur Ions: $\text{CH}_3\text{SSCH}_3^+$, CH_3SS^+ , $\text{C}_2\text{H}_5\text{S}^+$, and CH_2SH^+ ," *J. Am. Chem. Soc.* **105**, 3451 (1983).
- 83BUT/BAE2**
J.J. Butler and T. Baer, "A Photoionization Study of Organosulfur Ring Compounds: Thiirane, Thietane and Tetrahydrothiophene," *Org. Mass Spectrom.* **18**, 248 (1983).
- 83BYS**
K. Bystrom, *J. Comput. Chem.* **4**, 308 (1983).
- 83CAL**
W.L. Calhoun, *J. Chem. Eng. Data* **28**, 147 (1983).
- 83CAL/BAR**
G. Caldwell and J.E. Bartmess, unpublished results.
- 83CHE/WEN**
E.C.M. Chen and W.E. Wentworth, "Determination of Molecular Electron Affinities Using the Electron Capture Detector in the Pulse Sampling Mode at Steady State," *J. Phys. Chem.* **87**, 45 (1983).
- 83CAM/CIU**
R. Cambi, G. Ciullo, A. Sgamellotti, F. Tarantelli, R. Fantoni, A. Giardini-Guidoni, I.E. McCarthy and V. di Martino, "An (ϵ , 2ϵ) Spectroscopic Investigation and a Green's Function Study of the Ionization of Chloro- and Bromoethylene," *Chem. Phys. Lett.* **101**, 477 (1983).
- 83CAN/HAM**
P.H. Cannington and N.S. Ham, "He(I) and He(II) Photoelectron Spectra of Glycine and Related Molecules," *J. Electron Spectrosc. Rel. Phenom.* **32**, 139 (1983).
- 83CAR/LIV**
F. Carnovale, M.K. Livett and J.B. Peel, "Identification of the Gas Phase Trimer $(\text{CH}_3)_2\text{S}(\text{HF})_2$ by Photoelectron Spectroscopy," *J. Am. Chem. Soc.* **105**, 6788 (1983).
- 83CAS/AJO**
M. Casarin, D. Ajo, G. Granozzi, E. Tondello and S. Aime, "Gas-Phase Ultraviolet Photoelectron Spectra of $\{[\text{Ni}(\eta\text{-C}_5\text{H}_5)]_2(\mu\text{-C}_2\text{R}_2)\}$ ($\text{R} = \text{H or CF}_3$) Complexes," *J. Chem. Soc. Dalton Trans.*, 869 (1983).
- 83CAS/KIM**
M.C. Caserio and J.K. Kim, "Thioacylium Ions. Gas Phase Ion-Molecule Reactions of Thioic and Dithioic Acid Derivatives," *J. Am. Chem. Soc.* **105**, 6896 (1983).
- 83CAT/ELG**
J. Catalan, J. Elguero, R. Flammang and A. Maquestiau, "The Relative Basicities of Imidazole and Benzimidazole," *Angew. Chem. Int. Ed.* **22**, 323 (1983).
- 83CET/LAP**
B. Cetinkaya, M.F. Lappert and R.J. Suffolk, "Photoelectron Spectra of Some Sterically Hindered Phenols and Related Compounds," *J. Chem. Res. (S)*, 316 (1983).
- 83CET/LAP2**
B. Cetinkaya, M.F. Lappert, J.G. Stamper and R.J. Suffolk, "The He(I) and He(II) Photoelectron Spectra of bis(2,4,6-Tri-*t*-butylphenyl)-diphosphene," *J. Electron Spectrosc. Rel. Phenom.* **32**, 133 (1983).
- 83CLA/COR**
R.M. Clay, S. Corr, G. Keenan and W.V. Steele, *J. Am. Chem. Soc.* **105**, 2071 (1983).
- 83COM/REI**
R.N. Compton, P.W. Reinhardt and H.C. Schweinler, "Formation of Gas-Phase Negative Ions in Vinylene Carbonate," *Int. J. Mass Spectrom. Ion Phys.* **49**, 113 (1983).
- 83DAM/BOC**
R. Dammel, H. Bock and J.-M. Denis, "The Photoelectron Spectrum of 1-Azetine," *Chem. Phys. Lett.* **102**, 239 (1983).
- 83DAN/ROS**
J. Dannacher, H.M. Rosenstock, R. Buff, A.C. Parr, R.L. Stockbauer, R. Bombach and J.-P. Stadelmann, "Benchmark Measurement of Iodobenzene Ion Fragmentation Rates," *Chem. Phys.* **75**, 23 (1983).
- 83DAS/GRO**
C. Dass and M.L. Gross, "Electrocyclic Ring Opening of 1-Phenylcyclobutene and 3-Phenylcyclobutene Radical Cations," *J. Am. Chem. Soc.* **105**, 5724 (1983).
- 83DEK/DES**
R.L. DeKock, P. Deshmukh, T.P. Fehlner, C.E. Housecroft, J.S. Plotkin and S.G. Shore, "UV Photoelectron Spectra and Electronic Structure of $(\eta^5\text{-C}_5\text{H}_5)(\text{CO})_2\text{FeB}_2\text{H}_5$. Comparison of the Fe-B Bonding with the Fe-C Bonding in $(\text{CO})_4\text{FeC}_2\text{H}_4$," *J. Am. Chem. Soc.* **105**, 815 (1983).
- 83DEP/SCH**
C.H. Depuy and R. Schmitt, Personal Communication.
- 83ELL/ELL**
H.B. Ellis Jr. and G.B. Ellison, "Photoelectron Spectroscopy of HNO and DNO ," *J. Chem. Phys.* **78**, 6541 (1983). *appears later ✓*
- 83DEW/HEA**
M.J.S. Dewar and E. Healy, *J. Comput. Chem.* **4**, 542 (1983).
- 83DEW/RZE**
M.J.S. Dewar and H.S. Rzepa, *J. Comput. Chem.* **4**, 158 (1983).
- 83DIE/BRU**
H. tom Dieck, B. Bruder and K.-D. Franz, "Synthese offenkettiger und cyclischer, silylierter 1,2-Ethendiamine," *Chem. Ber.* **116**, 136 (1983).
- 83DOB/HIL**
B. Dobson, I.H. Hillier, J.A. Connor, D. Moncrieff, M.L. Scanlan and C.D. Garner, "Electronic Structure and Low-energy Photoelectron Spectra of 4,4'-Disubstituted 2,2'-Bipyridines," *J. Chem. Soc. Faraday Trans. II* **79**, 295 (1983).
- 83DUD/GRE**
N. Dudeney, J.C. Green, P. Grebenik and O.N. Kirchner, "Synthesis and Structural Characterization of the Electron Rich Complexes $\text{Co}(\eta\text{-C}_5\text{Me}_5)(\text{R}_2\text{PCH}_2\text{CH}_2\text{PR}_2)$ ($\text{R} = \text{Me, Ph}$). Photoelectron Spectroscopic Studies of Some Pentamethylcyclopentadienylphosphinecobalt Complexes," *J. Organometall. Chem.* **252**, 221 (1983).
- 83DYK/GRA**
J.M. Dyke, B.W.J. Gravenor, R.A. Lewis and A. Morris, "A Study of the First Ionization Potential of the $\text{CrO}(\text{X}^2\text{Pi})$ Molecule with High-temperature Photoelectron Spectroscopy," *J. Chem. Soc., Faraday Trans. II* **79**, 1083 (1983).
- 83DYK/JON**
J.M. Dyke, N. Jonathan, A.E. Lewis, J.D. Mills and A. Morris, "Vacuum Ultraviolet Photoelectron Spectroscopy of Transient Species. Part 16. The $\text{NCO}(\text{X}^2\text{Pi})$ Radical," *Molec. Phys.* **50**, 77 (1983).
- 83DYK/JON2**
J.M. Dyke, N. Jonathan, A. Morris, A. Ridha and M.J. Winter, "Vacuum Ultraviolet Photoelectron Spectroscopy of Transient Species. XVII. The $\text{SiH}_3(\text{X}^2\text{A}_1)$ Radical," *Chem. Phys.* **81**, 481 (1983).

83DYK/KIR

J.M. Dyke, C. Kirby and A. Morris, "Study of the Ionization Process $\text{BF}^+ (\text{X}^2\Sigma^+) \leftarrow \text{BF}(\text{X}^1\Sigma^+)$ by High-temperature Photoelectron Spectroscopy," *J. Chem. Soc. Faraday Trans. II* **79**, 483 (1983).

83EBA/ANE

T. Ebata, Y. Anezaki, M. Fujii, N. Mikami and M. Ito, "High Rydberg States of NO Studied by Two-Color Multiphoton Spectroscopy," *J. Phys. Chem.* **87**, 4773 (1983).

83EDM

R.S. Edmundson, "Cyclic Organophosphorus Compounds. XVIII-The Mass Spectroscopy of Some 2-Amino-1,3,2-dioxaphosphorinans, Cyclic Phosphorodiamidates and Phosphorodiamidodithionates," *Org. Mass Spectrom.* **18**, 150 (1983).

83ELL/ELL

H.B. Ellis, Jr. and G.B. Ellison, "Photoelectron Spectroscopy of HNO^- and DNO^- ," *J. Chem. Phys.* **78**, 6541 (1983).

83ERM/AKO

A.I. Ermolenko, M.E. Akopyan and Y.L. Sergeev, "Decomposition of Dimethyl Sulfide Molecular Ions. Randomization of States during Photoionization Dissociation of Molecules," *Khim. Vys. Ener.* **17**, 25 (1983); English trans.: *High Energy* **17**, 19 (1983).

83ERV/LOH

K. Ervin, S.K. Loh, N. Aristov and P.B. Armentrout, "Metal Cluster Ions: The Bond Energy of Mn_2^+ ," *J. Phys. Chem.* **87**, 3593 (1983).

83EVA/WBE

B.S. Evans, S.I. Weeks and E. Whittle, "Bromination of Fluoroalkanes. Part 5 Kinetics of Forward and Reverse Reactions in the System $\text{Br}_2 + i\text{-C}_3\text{F}_7\text{H} = \text{HBr} + i\text{-C}_3\text{F}_7\text{Br}$," *J. Chem. Soc. Faraday Trans. I* **79**, 1471 (1983).

83FUC/HAL

R. Fuchs and J.H. Hallman, *Can. J. Chem.* **61**, 503 (1983).

83FUC/SMI

R. Fuchs and N. Smith, Personal communication of unpublished heat of formation and heat of vaporization data.

83GAU/HOU

T. Gaumann, R. Houriet, D. Stahl, J.-C. Tabet, N. Heinrich and H. Schwarz, "Further Examples of Skeletal Rearrangement of the Wagner-Meerwein Type in Chemical Ionization Mass Spectrometry: the Case of $[\text{C}_6\text{H}_9^+]$ Ions," *Org. Mass Spectrom.* **18**, 215 (1983).

83GIL/HSI

J.P. Gilman, T. Hsieh and G.G. Meisels, "Competition between Isomerization and Fragmentation of Gaseous Ions. II. Nitromethane and Methylnitrite Ions," *J. Chem. Phys.* **78**, 1174 (1983).

83GIL/HSI2

J.P. Gilman, T. Hsieh and G.G. Meisels, "The Unimolecular Decomposition Rates of Energy Selected Methylnitrite and Deuterated Methylnitrite Ions," *J. Chem. Phys.* **78**, 3767 (1983).

83GLE/BOH

R. Gleiter, M.C. Bohm, A. de Meijere and T. Preuss, "Electronic Structure and Reactivity of Homobarrelene Derivatives," *J. Org. Chem.* **48**, 796 (1983).

83GLE/BOH2

R. Gleiter, M.C. Bohm, M. Eckert-Maksic, W. Schafer, M. Baudler, Y. Aktalay, G. Fritz and K.-D. Hoppe, "The Electronic Structure of Phosphorus Cages with the Nortricyclane Skeleton. Model Calculations and Photoelectron Spectroscopic Investigations," *Chem. Ber.* **116**, 2972 (1983).

83GLE/GOO

R. Gleiter, W.D. Goodmann, W. Schafer, J. Grobe and J. Apel, "He(I)-PE Spektren der Verbindungen $(\text{CH}_3)_n\text{E}(\text{CF}_3)_{3-n}$ für $\text{E} = \text{P, As, Sb}$ und $n = 0-3$," *Chem. Ber.* **116**, 3745 (1983).

83GLE/HAI

R. Gleiter, R. Haider, P. Bischof and J.-J. Lindner, "Zur Konformation von Tetravinyl- und Trivinylmethan - Vergleich der PE-Spektren von Tetravinylmethan mit trans,trans,trans-1,2,3,4-Tetravinylcyclobutan," *Chem. Ber.* **116**, 3736 (1983).

83GLE/HAI2

R. Gleiter, R. Haider, K. Gubernator and P. Bischof, "cis,trans,cis- and trans,trans,trans-1,2,3,4-Tetravinylcyclobutan - Preparation and Some Spectroscopic Properties," *Chem. Ber.* **116**, 2983 (1983).

83GLE/SAA

R. Gleiter, R.W. Saalfrank, W. Paul, D.O. Cowan and M. Eckert-Maksic, "Das Photoelektronenspektrum von 1,1-Diethoxy-3,3-bis(trifluormethyl)allen. Der Effekt von Trifluormethylgruppen auf kumulierte Systeme," *Chem. Ber.* **116**, 2888 (1983).

83GLE/SPA

R. Gleiter, J. Spanget-Larsen, R. Bartetzko, H. Neunhoeffer and M. Clausen, "Photoelectron Spectra of 1,2,3-Triazine and Its Methyl Derivatives," *Chem. Phys. Lett.* **97**, 94 (1983).

83GON/PII

D. Gonbeau, G. Pfister-Guillouzo, J. Escudie, C. Couret and J. Satge, "Application de la Spectroscopie Moléculaire aux Propriétés Moléculaires. XVII. Etude Structurale de Diphosphènes par Spectroscopie Photoelectronique," *J. Organometal. Chem.* **247**, C17 (1983).

83GOR/BOU

M.S. Gordon, P. Boudjouk and F. Anwar, *J. Am. Chem. Soc.* **105**, 4972 (1983).

83GRA/WIE

M. Grade, J. Wienecke, W. Rosinger and W. Hirschwald, "Electron Impact Investigation of the Molecules $\text{SeS}(\text{g})$ and $\text{TeSe}(\text{g})$ under High-Temperature Equilibrium Conditions," *Ber. Bunsenges. Phys. Chem.* **87**, 355 (1983).

83GRE/PAY

J.C. Green, M.P. Payne and A. Streitwieser, Jr., "He I and He II Photoelectron Spectral Studies of Alkyluranocenes," *Organometal.* **2**, 1707 (1983).

83GRE/TOM

A. Greenberg, R.P.T. Tomkins, M. Dobrovolny and J.F. Liebman, *J. Am. Chem. Soc.* **105**, 6855 (1983).

83GUI/PII

C. Guimon, G. Pfister-Guillouzo and M. Begtrup, "Tautomerism of Pyrazoline-5-thione and Trazoline-5-thione Studied by Variable Temperature Photoelectron Spectroscopy," *Can. J. Chem.* **61**, 1197 (1983).

83GUI/PII2

G. Guimon, G. Pfister-Guillouzo, H. Lavayssiere, G. Dousse, J. Barrau and J. Satge, "Generation of Dimethylgermathione and Dimethylsilathione and Their Detection in the Gas Phase by Photoelectron Spectroscopy," *J. Organometal. Chem.* **249**, C17 (1983).

83HAR/OHN

Y. Harada, K. Ohno and H. Mutoh, "Penning Ionization Electron Spectroscopy of CO and $\text{Fe}(\text{CO})_5$. Study of Electronic Structure of $\text{Fe}(\text{CO})_5$ from Electron Distribution of Individual Molecular Orbitals," *J. Chem. Phys.* **79**, 3251 (1983).

83HEL/MOL

H. Helm and R. Moller, "Bound-Free Spectroscopy of Cs_2^+ ," *Phys. Rev. A* **27**, 2493 (1983).

83HEN/BEN

S.P. Heneghan and S.W. Benson, "Kinetics and Thermochemistry of Electron Attachment to SF_6 ," *Int. J. Chem. Kinet.* **15**, 109 (1983).

83HIL

K. Hilpert, *Ber. Buns. Gesell. Phys. Chem.* **87**, 161 (1983).

- 83HIL/GIN**
K. Hilpert and K.A. Gingerich, "Mass Spectrometric Study on the Evaporation of Phases of the System NaI-SnI₂," *Int. J. Mass Spectrom. Ion Phys.* **47**, 247 (1983).
- 83HOL**
J.L. Holmes, Unpublished result.
- 83HOL/BUR**
J.L. Holmes, P.C. Burgers, J.K. Terlouw, H. Schwarz, B. Ciommer and H. Halim, "Stable C₂H₅X⁺ (X = Cl, Br) Radical Cations of Structure [CH₃CHXH⁺]: Their Energetics and Dissociation Characteristics," *Org. Mass Spectrom.* **18**, 208 (1983).
- 83HOL/LOS**
J.L. Holmes and F.P. Lossing, "The Need for Adequate Thermochemical Data for the Interpretation of Fragmentation Mechanisms and Ion Structure Assignments," *Int. J. Mass Spectrom. Ion Phys.* **47**, 133 (1983).
- 83HOL/LOS2**
J.L. Holmes, F.P. Lossing, J.K. Terlouw and P.C. Burgers, "Novel Gas-Phase Ions. The Radical Cations [CH₂XH]⁺ (X = F, Cl, Br, I, OH, NH₂, SH) and [CH₂CH₂NH₃]⁺," *Can. J. Chem.* **61**, 2305 (1983).
- 83HOU/RON**
K.N. Houk, N.G. Rondan, M.N. Paddon-Row, C.W. Jefford, P.T. Huy, P.D. Burrow and K.D. Jordan, "Ionization Potentials, Electron Affinities, and Molecular Orbitals of 2-Substituted Norbornadienes. Theory of 1,2 and Homo-1,4 Carbene Cycloaddition Selectivities," *J. Am. Chem. Soc.* **105**, 5563 (1983).
- 83ING/NIB**
S. Ingemann and N.M.M. Nibbering, "Gas Phase Reactions of Anions with 2-, 3-, and 4-Fluoroanisole," *J. Org. Chem.* **48**, 183 (1983).
- 83JIA/MOH**
W. Jian-qi, M. Mohraz, E. Heilbronner, A. Krebs, K. Schutz, J. Voss and B. Kopke, "79. The He(I α) Photoelectron Spectra of Substituted 1,2-Dithietes," *Helv. Chim. Acta* **66**, 801 (1983).
- 83JOC/DEK**
R. Jochem, H. Dekker, C. Mosselman and G. Soumen, *J. Chem. Thermodyn.* **15**, 95 (1983).
- 83JON/VAN**
G. Jonkers, S.M. Van Der Kerk, R. Mooyman, C.A. De Lange and J.G. Snijders, "UV Photoelectron Spectroscopy of Transient Species: Germanium Diiodide (GeI₂)," *Chem. Phys. Lett.* **94**, 585 (1983).
- 83JOR/CAR**
F.S. Jorgensen and L. Carlsen, "1,2-Oxathiolane - A Photoelectron Spectroscopic Study," *Chem. Ber.* **116**, 2374 (1983).
- 83JOR/MCC**
F.S. Jorgensen and P.H. McCabe, "Cis-Disulfides. Photoelectron Spectrum of a 6,7-Dithiabicyclo[3.2.1]octane," *Tetrahedron Lett.* **24**, 319 (1983).
- 83JOR/PAD**
F.S. Jorgensen, M.N. Paddon-Row and H.K. Patney, "Photoelectron Spectra of Some Decahydrotrimethanoanthracenes: Observation of Large Π Orbital Interactions through Six Bonds and an Apparent Violation of the trans Rule," *J. Chem. Soc. Chem. Commun.* **573** (1983).
- 83KAG/UJS**
N.D. Kagramanov, K. Ujszaszy, J. Tamas, A.K. Maltsev and O.M. Nefedov, "Mass Spectrometric Detection of Allylic and Perfluoroallylic Free Radicals and the Determination of Their Ionization Potentials," *Izv. Akad. Nauk SSSR, Ser. Khim.* **7**, 1683 (1983); English trans.: *Bull. Acad. Sci. USSR Div. Chem. Sci.* **7**, 1531 (1984).
- 83KAK/GIE**
W. Kakolowicz and E. Giera, *J. Chem. Thermodyn.* **15**, 203 (1983).
- 83KIR/DOM**
D.B. Kirklin and E.S. Domalski, *J. Chem. Thermodyn.* **15**, 941 (1983).
- 83KIT/MOR**
S. Kitagawa, I. Morishima and K. Yoshikawa, "UV Photoelectron Spectra of Some Transition Metal(II) Acetylacetonates," *Polyhedron* **2**, 43 (1983).
- 83KLA/KOV**
L. Klasinc, B. Kovac and H. Gusten, "Photoelectron Spectra of Acenes. Electronic Structure and Substituent Effects," *Pure & Appl. Chem.* **55**, 289 (1983).
- 83KLA/MAI**
D. Klapstein, J.P. Maier and W. Zambach, "Emission Spectra of Rotationally Cooled Dihaloacetylene Cations in the Gas Phase: A₂ Π (n, g) \rightarrow x₂ Π (n, u) Band Systems," *Chem. Phys.* **77**, 463 (1983).
- 83KLE/NIB**
J.C. Kleingeld and N.M.M. Nibbering, "The Long-lived H₃O⁺ Ion in the Gas Phase: Its Formation, Structure, and Reactions," *Int. J. Mass Spectrom. Ion Phys.* **49**, 311 (1983).
- 83KOB**
T. Kobayashi, "Conformational Analysis of Terphenyls by Photoelectron Spectroscopy," *Bull. Chem. Soc. Jpn.* **56**, 3224 (1983).
- 83KOL/PAP**
V.P. Kolesov and T.S. Papina, *Russ. Chem. Rev.* **52**, 754 (1983).
- 83KOP/MOL**
I.A. Koppel, U.H. Molder and R.J. Pikver, "Photoelectron Spectra of Molecules. I. Alcohols," *Org. Reactivity* **20**, 45 (1983).
- 83KOR/BON**
M.V. Korobov, A.A. Bondarenko, L.N. Sidorov and V.V. Nikulin, "Enthalpies of Formation of the Gaseous Platinum Fluorides, PtF₂ and PtF₄," *High Temp. Sci.* **16**, 411 (1983).
- 83LAC/MAN**
K. Lacmann, M.J.P. Maneira, A.M.C. Moutinho and U. Weigman, "Total and Double Differential Cross Sections of Ion-Pair Formations in Collisions of K Atoms with SnCl₄ and CCl₄," *J. Chem. Phys.* **78**, 1767 (1983).
- 83LAR/MCM**
J.W. Larson and T.B. McMahon, "Strong Hydrogen Bonding in Gas-Phase Anions. An Ion Cyclotron Resonance Determination of Fluoride Binding Energetics to Bronsted Acids from Gas-Phase Fluoride Exchange Equilibria Measurements," *J. Am. Chem. Soc.* **105**, 2944 (1983).
- 83LIE**
J.F. Liebman, "The Cyclophanes" (ed. P.M. Keehn and S. Rosenfeld, Academic Press, New York, 1983).
- 83LIF**
C. Lifshitz, "Energy-Entropy Trade-offs in the Unimolecular Decompositions of SF₆⁻," *J. Phys. Chem.* **87**, 3474 (1983).
- 83LIF/BER**
C. Lifshitz, P. Berger and E. Tzidon, "Kinetic Energy Release Distributions (KERDs) for the Dissociation of Metastable Enol Ions," *Chem. Phys. Lett.* **95**, 109 (1983).
- 83LIF/EAT**
C. Lifshitz and P.E. Eaton, "Time-Dependent Mass Spectra and Breakdown Graphs. III. The Cubane Cation Complete or Partial Instability," *Int. J. Mass Spectrom. Ion Phys.* **49**, 337 (1983).
- 83LIN/BRO**
S.H. Linn, J.M. Brom, Jr., W.-B. Tzeng and C.Y. Ng, "Molecular Beam Photoionization Study of HgCl₂," *J. Chem. Phys.* **78**, 37 (1983).
- 83LIN/TZE**
S.H. Linn, W.-B. Tzeng, J.M. Brom, Jr. and C.Y. Ng, "Molecular Beam Photoionization Study of HgBr₂ and HgI₂," *J. Chem. Phys.* **78**, 50 (1983).
- 83LOC/MCI**
M.J. Locke and R.T. McIver, Jr., "Effect of Solvation on the Acid/Base Properties of Glycine," *J. Am. Chem. Soc.* **105**, 4226 (1983).

83MAK/VER

A.V. Makarov, E.N. Verkhoturov and O.T. Nikitin, "Mass Spectrometric Study of the Processes of Ionization of the Vapor above Molybdenum Dichloride," *Vest. Mosk. Univ. Khim.* **38**, 350 (1983); English trans.: *Moscow Univ. Chem. Bull.* **38**, 42 (1983).

83MAL/MIL

W.G. Mallard, J.H. Miller and K.C. Smyth, "The ns Rydberg Series of 1,3-trans-Butadiene Observed Using Multiphoton Ionization," *J. Chem. Phys.* **79**, 5900 (1983).

83MAR/MAY

H.-D. Martin, B. Mayer, R. Gleiter, W. Schafer and F. Vogtle, "Photoelektronenspektroskopische Untersuchung transanularer π -, σ - und n-Wechselwirkungen in bruckenkopfsubstituierten Triptycenen," *Chem. Ber.* **116**, 2546 (1983).

83MAR/REI

R. Maripuu, I. Reineck, H. Agren, W. Nian-Zu, J. Rong, H. Veenhuizen, S. Al-Shamma, L. Karlsson and K. Siegbahn, "The HeI Excited Electron Spectrum of Phosphine. An Experimental and Theoretical Study," *Molec. Phys.* **48**, 1255 (1983).

83MAU

M. Meot-Ner (Mautner), Personal communication.

83MAZ

M. Mazzoni, " ^{210}P Photoionization Cross Section of Sb I," *Phys. Lett.* **97A**, 381 (1983).

83MCA/HUD

D.J. McAdoo and C.E. Hudson, "The Decompositions of Metastable $[\text{C}_4\text{HgO}]^+$ Ions and the $[\text{C}_4\text{HgO}]^+$ Potential Surface," *Org. Mass Spectrom.* **18**, 466 (1983).

83MCG/SCH

M.W. McGeogh and R.E. Schlier, "Autoionizing Rydberg States of the Li_2 Molecule: Molecular Constants for Li_2^+ ," *Chem. Phys. Lett.* **99**, 347 (1983).

83MET/ARA

R.M. Metzger and E.S. Ararat, *J. Chem. Phys.* **78**, 2696 (1983).

83MOL/PIK

U.H. Molder, R.J. Pikver and I.A. Koppel, "Photoelectron Spectra of Molecules. 2. Ethers," *Org. Reactivity* **20**, 208 (1983).

83MOL/PIK2

U.H. Molder, R.J. Pikver and I.A. Koppel, "Photoelectron Spectra of Molecules. 3. Nitriles," *Org. Reactivity* **20**, 230 (1983).

83MOL/PIK3

U.H. Molder, R.J. Pikver and I.A. Koppel, "Photoelectron Spectra of Molecules. 4. Amines," *Org. Reactivity* **20**, 355 (1983).

83MOO/SEI

G.K. Moortgart, W. Seiler and P. Warnck, "Photodissociation of HCHO in Air: CO and H_2 Quantum Yields at 220 and 300K," *J. Chem. Phys.* **78**, 1185 (1983).

83MUR

E. Murad, "Abstraction Reactions of Ca^+ and Sr^+ Ions," *J. Chem. Phys.* **78**, 6611 (1983).

83NES/MIL

O. Neskovic, M. Miletic, M. Veljkovic, D. Golobocantin and K.F. Zmbov, "Ionization and Fragmentation of Phosphorous Oxyfluoride by Electron Impact," *Int. J. Mass Spectrom. Ion Proc.* **47**, 141 (1983).

83NIS/SAK

K. Nishiyama, M. Sakiyama and S. Seki, *Bull. Chem. Soc. Jpn.* **56**, 3171 (1983).

83NOB/WAL

P.N. Noble and R. Walsh, "Kinetics of the Gas Phase Reaction between Iodine and Monogermene and the Bond Dissociation Energy $\text{D}(\text{H}_3\text{Ge-H})$," *Int. J. Chem. Kinet.* **15**, 547 (1983).

83NOV/POT

I. Novak and A.W. Potts, "The Ultraviolet Photoelectron Spectra of Gas Phase and Condensed Bismuth Halides and Antimony Trifluoride," *J. Chem. Soc. Dalton Trans.*, 635 (1983).

83NOV/POT2

I. Novak and A.W. Potts, "The Ultraviolet Photoelectron Spectra and Electron Structure of Gas Phase and Condensed SnF_2 and PbF_2 ," *J. Chem. Soc. Dalton Trans.*, 2211 (1983).

83OAK/ELL

J.M. Oakes and B.G. Ellison, "Photoelectron Spectroscopy of the Allenyl Anion $\text{CH}_2 = \text{C} = \text{CH}^-$," *J. Am. Chem. Soc.* **105**, 2969 (1983).

83OAK/ION

J.M. Oakes, M.E. Jones, V.M. Bierbaum and G.B. Ellison, "Photoelectron Spectroscopy of CCO^- and HCCO^- ," *J. Phys. Chem.* **87**, 4810 (1983).

83OGD/SHA

I.K. Ogden, N. Shaw, C.J. Danby and I. Powis, "Competing Dissociation Channels of Nitromethane and Methyl Nitrite Ions and the Role of Electronic and Internal Modes of Excitation," *Int. J. Mass Spectrom. Ion Proc.* **54**, 41 (1983).

83OHN/IMA

K. Ohno, K. Imai, S. Matsumoto and Y. Harada, "Penning Ionization Electron Spectroscopy of $\text{C}_2\text{H}_5\text{X}$ ($\text{X} = \text{NH}_2, \text{OH}, \text{H}, \text{Cl}, \text{I}$). Relative Reactivity of Orbital Localizing on Functional Groups upon Electrophilic Attack by Metastable Helium Atoms," *J. Phys. Chem.* **87**, 4346 (1983).

83OZG

T. Ozgen, "Mass Spectrometric Determination of the Appearance Potentials of $\text{PCl}_3, \text{AsCl}_3, \text{SbCl}_3, \text{BiCl}_3$ and Their Fragments," *Int. J. Mass Spectrom. Ion Proc.* **48**, 427 (1983).

83PED/MAR

J.B. Pedley and E.M. Marshall, *J. Phys. Chem. Ref. Data* **12**, 957 (1983).

83PEN/NAT

P. Pernetreau, P. Natalis, L. Longton and J.E. Collin, "Ionization Energies for the Vibrionic Transitions from $\text{DCI } X^1\Sigma^+$ ($v' = 0$) to $\text{DCI}^+ X^2\Pi$ ($V' = 0-18$) and $A^2\Sigma^+$ ($v' = 0-17$) Determined by Photoelectron Spectroscopy," *J. Electron Spectrosc. Rel. Phenom.* **28**, 295 (1983).

83PET/DAO

K.I. Peterson, P.D. Dao and A.W. Castleman, Jr., "Photoionization Studies of Na_2Cl and Na_2O and Reactions of Metal Clusters," *J. Chem. Phys.* **79**, 777 (1983).

83PIA/KEL

M.N. Piancastelli, P.R. Keller, J.W. Taylor, F.A. Grimm and T.A. Carlson, "Angular Distribution Parameter as a Function of Photon Energy for Some Mono- and Diazabenzene and Its Use for Orbital Assignment," *J. Am. Chem. Soc.* **105**, 4235 (1983).

83PIC/ROD

J.M. Pickard and A.S. Rodgers, "Kinetics of the Gas Phase Reaction $\text{CH}_3\text{F} + \text{I}_2 = \text{CH}_3\text{FI} + \text{HI}$: The C-H Bond Dissociation Energy in Methyl and Methylene Fluorides," *Int. J. Chem. Kinet.* **15**, 569 (1983).

83PLA/SIM

V.A. Platonov and Y.N. Simulin, *Russ. J. Phys. Chem.* **57**, 840 (1983).

83PLE/MAR

P. Plessis, P. Marmet and R. Dutil, "Ionization and Appearance Potentials of CH_4 by Electron Impact," *J. Phys. B: At. Mol. Phys.* **16**, 1283 (1983).

83POT/NOV

A.W. Potts and I. Novak, "Ultraviolet Photoelectron Spectra of Selenium and Tellurium," *J. Electron Spectrosc. Rel. Phenom.* **28a**, 267 (1983).

83POT/NOV2

A.W. Potts, I. Novak and M.L. Lyus, "The Valence Shell Electronic Structure and UV Photoelectron Spectra of the Tetrahaloethylenes," *J. Electron Spectrosc. Rel. Phenom.* **31**, 57 (1983).

- 83POW**
I. Powis, "The Unimolecular Dissociation of Electronic State-Selected Methyl Iodide Cations," *Chem. Phys.* **74**, 421 (1983).
- 83POW/HAN**
D.E. Powers, S.G. Hansen, M.E. Geusic, D.L. Michalopoulos and R.E. Smalley, "Supersonic Copper Clusters," *J. Chem. Phys.* **78**, 2866 (1983).
- 83PRE/TZE**
H.F. Prest, W.-B. Tzeng, J.M. Brom, Jr. and C.Y. Ng, "Molecular Beam Photoionization Study of H₂S," *Int. J. Mass Spectrom. Ion Proc.* **50**, 315 (1983).
- 83PRE/TZE2**
H.F. Prest, W.-B. Tzeng, J.M. Brom, Jr. and C.Y. Ng, "Photoionization Study of (H₂S)₂ and (H₂S)₃," *J. Am. Chem. Soc.* **105**, 7531 (1983).
- 83RAB/SEL**
M.A. Rabbih and E.T.M. Selim, "A Mass Spectrometric Appearance Energies Study of Cyclohexanol," *Egypt. J. Phys.* **14**, 243 (1983).
- 83RAK/BOH**
A.B. Rakshit and D.K. Bohme, *Int. J. Mass Spectrom. Ion Proc.* **49**, 275 (1983).
- 83REI/NOH**
I. Reineck, C. Nohre, P. Lodin, R. Maripuu, B. Lindberg, L. Karlsson, K. Siegbahn, A.-B. Hornfeldt and S. Gronowitz, "Electronic and Vibrational Structure of 2-Substituted Selenophenes Studied by He(I) Photoelectron Spectroscopy," *Chem. Scripta* **22**, 209 (1983).
- 83ROS/GRA**
W. Rosinger, M. Grade and W. Hirschwald, "Detection of Ion States of S₂ to S₈ by Electron Impact," *Int. J. Mass Spectrom. Ion Proc.* **47**, 239 (1983).
- 83RUS/TRE**
D.H. Russell, B.S. Freiser, E.H. McBay and D.C. Canada, "The Structure of Decomposing [C₇H₇O]⁺ Ions: Benzyl versus Tropylium Ion Structures," *Org. Mass Spectrom.* **18**, 474 (1983).
- 83RUS/GOO**
B. Ruscic, G.L. Goodman and J. Berkowitz, "Photoelectron Spectra of the Lanthanide Trihalides and Their Interpretation," *J. Chem. Phys.* **78**, 5443 (1983).
- 83SCH/BAL**
W. Schmidt, H.-J. Ballschmidt, M. Klessinger, A. Heesing and W. Herdering, "MINDO/3-Rechnungen und PE-Untersuchungen zur Reaktionsweise von Azabicyclen," *Chem. Ber.* **116**, 1097 (1983).
- 83SCH/SCH**
R. Schulz, A. Schweig, K. Hartke and J. Koster, "Variable Temperature Photoelectron Spectral Study of 1,3-Dithiol-2-one and 4,5-Disubstituted 1,3-Dithiol-2-ones. Thermal Generation of 1,2-Dithiete, 3,4-Disubstituted 1,2-Dithietes, and Dialkyl Tetrathiooxalates" *J. Am. Chem. Soc.* **105**, 4519 (1983).
- 83SEA/CHU**
M. Seaver, W.A. Chupka, S.D. Colson and D. Gauyacq, "Double Resonance Multiphoton Ionization Studies of High Rydberg States in NO," *J. Phys. Chem.* **87**, 2226 (1983).
- 83SEM/RYSK**
V.I. Semikhin, A.N. Rykov and L.N. Sidorov, "Mass Spectrometric Study of the Evaporation of Lead Monoxide," *Zh. Fiz. Khim.* **57**, 1663 (1983); *Engl. trans.: Russ. J. Phys. Chem.* **57**, 1008 (1983).
- 83SHU/BEN**
L.G.S. Shum and S.W. Benson, "Thermochemistry and Kinetics of the Reaction of Methyl Mercaptan with Iodine," *Int. J. Chem. Kinet.* **15**, 433 (1983).
- 83SID/RUD**
L.N. Sidirov, E.B. Rudnyi, M.I. Nikitin and I.D. Sorokin, "Gas Phase Anion Exchange Reactions and the Determination of the Heats of Formation of Metaphosphate (PO₃-), metaborate (BO₂-), and perrhenate (ReO₄-)," *Dokl. Akad. Nauk SSSR, Ser. Khim.* **272**, 1172 (1983).
- 83SID/ZHU**
L.N. Sidirov, L.V. Zhuravlena, M.V. Varkov, E.V. Skokan, I.D. Sorokin, Yu. M. Koronev and P.A. Akishima, "Mass-Spectrometric Determination of Enthalpies of Dissociation of Gaseous Complex Fluorides into Neutral and Charged Particles. VII. MF-ThF₄ Systems," *Int. J. Mass Spectrom. Ion Proc.* **51**, 291 (1983).
- 83SKO/SAB**
S. Skoulika and R. Sabbah, *Thermochim. Acta* **61**, 203 (1983).
- 83SKO/SOR**
E.V. Skokan, I.D. Sorokin, M.I. Nikitin, N.S. Chilingarov and L.N. Sidirov, "The Electron Affinity of UF₆," *Russ. J. Phys. Chem.* **57**, 1745 (1983).
- 83SMI**
O.I. Smith, "Cross Sections for Formation of Parent and Fragment Ions by Electron Impact from C₂N₂," *Int. J. Mass Spectrom. Ion Proc.* **54**, 55 (1983).
- 83SPY/SAU**
S.M. Spyrou, I. Sauers and L.G. Christophorou, "Electron Attachment to the Perfluoroalkanes n-C_nF_{2n}+2 (n = 1-6) and i-C₄F₁₀," *J. Chem. Phys.* **78**, 7200 (1983).
- 83STE**
S.E. Stein, personal communication to the authors.
- 83STE/FEI**
A.E. Stevens, C.S. Feigerle and W.C. Lineberger, "Laser Photoelectron Spectroscopy of MnH⁻ and FeH⁻: Electronic Structures of the Metal Hydrides, Identification of a Low Spin Excited State of MnH," *J. Chem. Phys.* **78**, 5420 (1983).
- 83STE2**
W.V. Steele, *J. Chem. Thermodyn.* **15**, 595 (1983).
- 83TER/HOL**
J.K. Terlouw, J.L. Holmes and F.P. Lossing, "Ionized Ethylidene Ketene and Its Homologue Methylene Ketene," *Can. J. Chem.* **61**, 1722 (1983).
- 83TER/WEZ**
J.K. Terlouw, J. Wezenberg, P.C. Burgers and J.L. Holmes, "New Stable Isomers of [C₂H₄O]⁺ and [C₂H₄O₂]⁺, the Radical Cations [CH₂COH]⁺ and [CH₃COH]⁺," *J. Chem. Soc. Chem. Commun.* **1121** (1983).
- 83THO/GLI**
C. Thomson and C. Glidewell, *J. Comput. Chem.* **4**, 1 (1983).
- 83TOM/AMM**
G.M.R. Tombo, H.J. Ammann, K. Muller and C. Ganter, "5. Nucleophilic Addition to C,C-Double Bonds. VII. Study of Proximity Effects in Olefinic Alcohols and Amines by Photoelectron Spectroscopy," *Helv. Chim. Acta* **66**, 50 (1983).
- 83TUR/HAN**
F. Turecek and V. Hanus, "Stereolectronic Control of Ion Fragmentations: Loss of Hydrogen from Cyclic Ethers during Electron Impact Spectrometry," *Tetrahedron* **39**, 1499 (1983).
- 83VES/HAR**
T. Veszpremi, Y. Harada, K. Ohno and H. Mutoh, "Photoelectron and Penning Ionization Electron Spectroscopic Investigation of Trimethylsilyl- and t-Butyl-thiophenes," *J. Organometall. Chem.* **252**, 121 (1983).
- 83WAG/ISE**
G. Wagner and N.R. Isenor, "Formation and Dissociation of Na₂⁺ by Ruby Laser Radiation," *Can. J. Phys.* **61**, 40 (1983).
- 83WAL**
R. Walsh, "Thermochemistry of Silicon-containing Compounds," *J. Chem. Soc., Faraday Trans. I* **79**, 2233 (1983).
- 83WAN/LER**
F.C.-Y. Wang and G.E. Leroi, "Photoionization and Fragmentation of Halogenated Methanes," *Ann. Israel Phys. Soc.* **6**, 210 (1983).

- 83WEB/MCD**
C.R. Webster, I.S. McDermid and C.T. Rettner, "Laser Optogalvanic Photodetachment Spectroscopy: A new technique for studying photodetachment thresholds with application to I⁻," *J. Chem. Phys.* **78**, 646 (1983).
- 83WEI/BEN**
M. Weissman and S.W. Benson, *J. Phys. Chem.* **87**, 243 (1983).
- 83WLO/LUC**
S. Wlodek, Z. Luczynski and H. Wincel, "Gas Phase Complexes of NO₂⁻ and NO₃⁻ with SO₂," *Int. J. Mass Spectrom. Ion Proc.* **49**, 301 (1983).
- 83ZLA/LEE**
A. Zlatkis, C.K. Lee, W.E. Wentworth and E.C.M. Chen, "Constant Current Linearization for Determination of Electron Capture Mechanisms," *Anal. Chem.* **55**, 1596 (1983).
- 83ZVE/BAZ**
V.V. Zverev, Z.G. Bazhanova, N.V. Villem and Y.Y. Villem, "Photoelectron Spectra and Electronic Structure of Organophosphorus Compounds of Tricoordinated Phosphorus with a Phosphorus-Sulfur Bond," *Zh. Obs. Khim.* **53**, 1968 (1983); *Engl. trans.: J. Gen. Chem. (USSR)* **53**, 1775 (1983).
- 83ZYK/ERC**
B.G. Zykov, N.P. Erchak, V.I. Khvostenko, E. Lukevits, V.F. Matorykina and N.L. Asfandiarov, "Photoelectron Spectra of Furylsilanes and Their Carbon Analogs," *J. Organometall. Chem.* **253**, 301 (1983).
- 84ABE/DEL**
R.S. Abeywickrema, E.W. Della, P.E. Pigou, M.K. Livett and J.B. Peel, "Orbital Interactions in Some Polycycloalkyl Halides: A Photoelectron Spectroscopic Study," *J. Am. Chem. Soc.* **106**, 732 (1984).
- 84ADA/SMI**
N.G. Adams and D. Smith, "A Further Study of the Near-Thermoneutral Reaction O₂H⁺ + H₂ = H₃⁺ + O₂," *Chem. Phys. Lett.* **105**, 604 (1984).
- 84AIT/GOS**
R.A. Aitken, I. Gosney, H. Farries, M.H. Palmer, I. Simpson, J.I.G. Cadogan and E.J. Tinley, "Chemical Repercussions of Orbital Interactions through Bond and through Space. The Reactivity of the Double Bond in Unsaturated Cyclic Sulphones towards Aziridine Formation and Epoxidation," *Tetrahedron* **40**, 2487 (1984).
- 84AJO/CAS**
D. Ajo, M. Casarin, G. Granozzi, H.C.J. Ottenheijm and R. Plate, "An Investigation of the Electronic Structure of α,β -Unsaturated Acetyl amino Acid Ethyl Esters Using He(I) and He(II) Photoelectron Spectroscopy," *Recueil J. Royal Neth. Chem. Soc.* **103**, 365 (1984).
- 84ALA/RYE**
M. Alai and R.T.B. Rye, "Fragmentation of Selected C₆H₁₀O Isomers: Evidence for a Common [C₅H₇O]⁺ Ion Generated by Methyl Loss from Cyclohexene Oxide and 5,6-Dihydro-4-methyl-2H-pyran," *Org. Mass Spectrom.* **19**, 506 (1984).
- 84ALB/ALL**
B. Albrecht, M. Allan, E. Haselbach, L. Neuhaus and P.-A. Carrupt, "26. Molecular Ions of Transient Species: Vinyl-Alcohol Cation," *Helv. Chim. Acta* **67**, 216 (1984).
- 84ALB/ALL2**
B. Albrecht, M. Allan, E. Haselbach, L. Neuhaus and P.-A. Carrupt, "27. Molecular Ions of Transient Species: Vinylamine-Cation," *Helv. Chim. Acta* **67**, 220 (1984).
- 84ALE/VOL**
V.I. Alekseev, V.M. Volkov, L.I. Fedorova and A.V. Baluev, "Mass Spectrometric Study of the Ionization of an Oxygen Difluoride Molecule," *Izv. Akad. Nauk SSR, Ser. Khim* **1302** (1984).
- 84ALT/CON**
G. Al-Takhin, J.A. Connor, H.A. Skinner and M.T. Zaharani-Moettar, *J. Organomet. Chem.* **260**, 189 (1984).
- 84ALT/CON2**
G. Al-Takhin, J.A. Connor, G. Pilcher and H.A. Skinner, *J. Organomet. Chem.* **265**, 263 (1984).
- 84AND/CER**
R.R. Andrea, H. Cerfontain, H.J.A. Lamberchts, J.N. Louwen and A. Oskam, "He I and He II Photoelectron Spectra and CNDO/S and MNDO MO Calculations of Some Bridged [10] Annulenes," *J. Am. Chem. Soc.* **106**, 2531 (1984).
- 84AND/DEP**
A.H. Andrist, C.H. DePuy and R.R. Squires, "Structures of Isomeric Anions in the Gas Phase: Arylallyl and Arylcyclopropyl Anions," *J. Am. Chem. Soc.* **106**, 845 (1984).
- 84AND/DYK**
L. Andrews, J.M. Dyke, N. Jonathan, N. Keddar, A. Morris and A. Ridha, "A Photoelectron Spectroscopic Study of the Ground States of CH₂F⁺ and CD₂F⁺," *J. Phys. Chem.* **88**, 2364 (1984).
- 84AND/DYK2**
L. Andrews, J.M. Dyke, N. Jonathan, N. Keddar and A. Morris, "Photoelectron Spectroscopic Study of the Ground States of CH₂Cl⁺, CHCl₂⁺, and CHFCl⁺," *J. Am. Chem. Soc.* **106**, 299 (1984).
- 84AND/DYK3**
L. Andrews, J.M. Dyke, N. Jonathan, N. Keddar and A. Morris, "The First Bands in the Photoelectron Spectra of the CH₂Br, CD₂Br, CHBr₂, and CH₂I Free Radicals," *J. Phys. Chem.* **88**, 1950 (1984).
- 84ARI/YOS**
M. Arimura and Y. Yoshikawa, "Ionization Efficiency and Ionization Energy of Cyclic Compounds by Electron Impact," *Mass Spectrosc.* **32**, 375 (1984).
- 84ASA/YAS**
M. Asano, Y. Yasue and K. Kubo, "Mass Spectrometric Study of Ions Formed from Cesium Metaborate Vapor under Electron Impact," *J. Nucl. Sci. Tecn.* **21**, 614 (1984).
- 84ASF/ZYK**
N.L. Asfandiarov and B.C. Zykov, "Photoelectron Spectra of Methyl-Substituted 1,3-Dioxanes," *Izv. Akad. Nauk SSSR, Ser. Khim.* **2293** (1983). *English trans.: Bull. Acad. Sci. USSR Div. Chem. Sci.* **2069** (1983).
- 84BAI/DOM**
V.N. Baidin, I.N. Domnin, O.A. Prokhorenko, S. Elbel and A. de Meijere, "Photoelectron Spectra of Phenyl Substituted Cyclopropene Derivatives," *J. Electron Spectrosc. Rel. Phenom.* **34**, 103 (1984).
- 84BAN/YAT**
G.M. Bancroft, B.W. Yates, K.H. Tan and L.L. Coatsworth, "High Resolution Gas Phase Photoelectron Spectra Using Synchrotron Radiation. Selective Enhancement of the I 5p Cross Section in CF₃I Due to Photoexcitation of I 4d Electrons," *J. Chem. Soc. Chem. Commun.* **1613** (1984).
- 84BAR**
J.E. Bartmess, "The Gas Phase Thermochemistry of Ph₃C⁻, Ph₃C[•], and Ph₃C⁺," *32nd Ann. Conf. on Mass Spectrom. Allied Topics, San Antonio TX 27 May - 1 June, 1984. Abstracts p.* **472**.
- 84BAR/BUR**
J.E. Bartmess and R. Burnham, "Effect of Central Substituents on the Gas Phase Acidities of Propenes," *J. Org. Chem.* **49**, 1382 (1984).
- 84BAR/CAU**
V. Barone, C. Caulletti, L. Commisso, F. Leij, M.N. Piancastelli and N. Russo, "Quantum-mechanical and Ultraviolet Photoelectron Spectroscopic Studies of Azabiphenyls. The Case of 4-Phenylpyridine" *J. Chem. Research (S)* **338** (1984).
- 84BAR/YEH**
N. Bartlett, S. Yeh, K. Kourtakis and T. Mallouk, *J. Fluor. Chem.* **26**, 97 (1984).

- 84BEA/EYE**
D.B. Beach, C.J. Eyermann, S.P. Smit, S.F. Xiang and W.L. Jolly, "Applications of the Equivalent Cores Approximation. The Determination of Proton Affinities and Isocyanide-to-Nitrile Isomerization Energies from Core Binding Energies," *J. Am. Chem. Soc.* **106**, 536 (1984).
- 84BER/GRE**
J. Berkowitz, J.P. Greene, J. Foropoulos, Jr. and O.M. Neskovic, "Bonding and Ionization Energies of N-F and P-F Compounds," *J. Chem. Phys.* **81**, 6166 (1984).
- 84BER/GRE2**
J. Berkowitz and J.P. Greene, "The Barrier to Inversion in NF_3^+ ," *J. Chem. Phys.* **81**, 3383 (1984).
- 84BIC/MIN**
J. Bickerton, M.E. Minas da Piedade and G. Pilcher, *J. Chem. Thermodyn.* **16**, 661 (1984).
- 84BIC/PIL**
J. Bickerton, G. Pilcher and G. Al-Takhin, *J. Chem. Thermodyn.* **16**, 373 (1984).
- 84BIE/GRA**
V.M. Bierbaum, J.J. Grabowski and C.H. DePuy, "Gas-Phase Synthesis and Reactions of Nitrogen- and Sulfur-Containing Anions," *J. Phys. Chem.* **88**, 1389 (1984).
- 84BLA/WAL**
E.A. Walters and N.C. Blais, "Molecular Beam Photoionization and Fragmentation of D_2S , $(\text{H}_2\text{S})_2$, $(\text{D}_2\text{S})_2$, and $\text{H}_2\text{S}\cdot\text{H}_2\text{O}$," *J. Chem. Phys.* **80**, 3501 (1984).
- 84BOC/AYG**
H. Bock, S. Aygen, P. Rosmus, B. Solouki and E. Weissflog, "Gasphasen-Reaktionen, 40. Selenoformaldehyd: hochkorrelierte Wellenfunktion und photoelektronenspektroskopischer Nachweis," *Chem. Ber.* **117**, 187 (1984).
- 84BOC/ROS**
H. Bock, P. Rosmus, B. Solouki and G. Maier, "Gas Phase Reactions. XXXXVI. Silabenzene: Photoelectron Spectrum from 1-Sila-2,5-hexadiene Pyrolysis and Assignment of Valence Ionisation Energies," *J. Organometall. Chem.* **271**, 145 (1984).
- 84BOH/FAH**
H. Bohringer, D.W. Fahey, F.C. Fehsenfeld and E.E. Ferguson, "Bond Energies of the Molecules H_2O , SO_2 , H_2O_2 , and HCl to Various Atmospheric Negative Ions," *J. Chem. Phys.* **81**, 2805 (1984).
- 84BOL/HOU**
J.-C. Bollinger, R. Houriet and T. Yvernault, "Gas Phase Basicity of Hexamethylphosphotriamide (HMPT) Phosphinooxide Derivatives," *Phosphorus and Sulfur* **19**, 379 (1984).
- 84BOU/BRA**
C. Bousquet, N. Bras and Y. Majdi, *J. Phys. B* **17**, 1831 (1984).
- 84BOU/DAG**
G. Bouchoux and J. Dagaut, "The Loss of Br from $[(\text{CH}_3)_2\text{CBrCOCH}_3]^+$," *Org. Mass Spectrom.* **19**, 291 (1984).
- 84BOU/FLA**
G. Bouchoux, R. Flammang, Y. Hoppilliard, P. Jaudon, A. Maquestiau and P. Meyrant, "Mecanismes de Formation des Ions $[\text{C}_5\text{H}_9\text{O}]^+$ a Partir des Cetones $[\text{CH}_3\text{COC}_4\text{H}_9]^+$," *Int. J. Spectrosc.* **3**, 1 (1984).
- 84BOU/HOP**
G. Bouchoux, Y. Hoppilliard, P. Jaudon and R. Houriet, "The Heats of Formation of Some Protonated Olefinic Carbonyl Compounds: $[\text{C}_5\text{H}_9\text{O}]^+$ and $[\text{C}_4\text{H}_7\text{O}_2]^+$ Ions," *Org. Mass Spectrom.* **19**, 394 (1984).
- 84BOU/PFI**
J.P. Boutique, G. Pfister-Guillouzo, J. Riga, J.J. Verbist, J.G. Fripiat, J. Delhalle, R.C. Haddon and M.L. Kaplan, "Gas Phase UPS Study of Naphtho[1,8-cd:4,5-c'd']bis[1,2,6]thiadiazine," *J. Electron Spectrosc. Rel. Phenom.* **34**, 199 (1984).
- 84BOU/VER**
J.P. Boutique, J.J. Verbist, J.G. Fripiat, J. Delhalle, G. Pfister-Guillouzo and G.J. Ashwell, "3,5,11,13-Tetraazacycl[3.3.3]azine: Theoretical (ab-Initio) and Experimental (X-ray and Ultraviolet Photoelectron Spectroscopy) Studies of the Electronic Structure," *J. Am. Chem. Soc.* **106**, 4374 (1984).
- 84BOW/MAC**
R.D. Bowen and A. Maccoll, "Low Energy, Low Temperature Mass Spectra," *Org. Mass Spectrom.* **19**, 379 (1984).
- 84BRA/BAE**
W.A. Brand and T. Baer, "Dissociation Dynamics of Energy-Selected $\text{C}_5\text{H}_{10}^+$ Ions," *J. Am. Chem. Soc.* **106**, 3154 (1984).
- 84BUR/FUK**
D.J. Burinsky, E.K. Fukuda and J.E. Campana, "Electron Affinities from Dissociations of Mixed Negative Ion Dimers," *J. Am. Chem. Soc.* **106**, 2270 (1984).
- 84BUR/HOL**
P.C. Burgers, J.L. Holmes and J.K. Terlouw, "Gaseous $[\text{H}_2, \text{C}, \text{N}]^+$ and $[\text{H}_3, \text{C}, \text{N}]^+$ Ions. Generation of Formation, and Dissociation Characteristics of $[\text{H}_2\text{CN}]^+$, $[\text{HCNH}]^+$, $[\text{CNH}_2]^+$, $[\text{H}_2\text{CNH}]^+$, and $[\text{HCN}]^+$," *J. Am. Chem. Soc.* **106**, 2762 (1984).
- 84BUR/HOL2**
P.C. Burgers and J.L. Holmes, "The Generation of Triplet Methoxy Cations," *Org. Mass Spectrom.* **19**, 452 (1984).
- 84BUR/HOL3**
P.C. Burgers and J.L. Holmes, "Fragmentation Rate Constants and Appearance Energies for Reactions Having a Large Kinetic Shift and the Energy Partitioning in Their Metastable Decomposition," *Int. J. Mass Spectrom. Ion Proc.* **58**, 15 (1984).
- 84BUR/KUD**
G.G. Burdukovskaya, L.S. Kudin, M.F. Butman and K.S. Krasnov, "Ionic Forms in the Vapour over Potassium Iodide," *Zh. Neorgan. Khim.* **29**, 3020 (1984).
- 84BUT/HOL**
J.J. Butler, D.M.P. Holland, A.C. Parr and R. Stockbauer, "A Threshold Photoelectron-Photoion Coincidence Spectrometric Study of Dimethyl Ether (CH_3OCH_3)," *Int. J. Mass Spectrom. Ion Proc.* **58**, 1 (1984).
- 84BUT/KUD**
M.F. Butman, L.S. Kudin and K.S. Krasnov, "The Mass Spectrometric Determination of the Proton Affinity of the Molecules M_2O ($\text{M} = \text{Na}, \text{K}, \text{Rb}, \text{Cs}$)," *Zh. Neorgan. Khim.* **29**, 2150 (1984); English trans.: *Russ. J. Inorg. Chem.* **29**, 1228 (1984).
- 84CAL/KEB**
G. Caldwell and P. Kebarle, "Binding Energies and Structural Effects in Halide Anion-ROH and -RCOOH Complexes from Gas Phase Equilibria Measurements," *J. Am. Chem. Soc.* **106**, 967 (1984).
- 84CAL/ROZ**
G. Caldwell, M.D. Rozeboom, J.P. Kiplinger and J.E. Bartmess, "Anion-Alcohol Hydrogen Bond Strengths in the Gas Phase," *J. Am. Chem. Soc.* **106**, 4660 (1984).
- 84CAO/BAC**
J.-R. Cao and M.H. Back, *Int. J. Chem. Kinet.* **16**, 961 (1984).
- 84CAR/FAH**
T.A. Carlson, A. Fahlman, W.A. Svensson, M.O. Krause, T.A. Whitley, F.A. Grimm, M.N. Piancastelli and J.W. Taylor, "Angle-Resolved Photoelectron Cross Section of CF_4 ," *J. Chem. Phys.* **81**, 3828 (1984).
- 84CAS/FRE**
C.J. Cassidy and B.S. Freiser, "Determination of the $\text{Fe}^+ \cdot \text{OH}$ and $\text{Co}^+ \cdot \text{OH}$ Bond Energies by Deprotonation Reactions and by Photodissociation," *J. Am. Chem. Soc.* **106**, 6176 (1984).
- 84CAT/PAZ**
J. Catalan, J.L.G. de Paz, M. Yanez and J. Elguero, "Relationship between Substituent-Induced Energy and Charge Effects in Proton-Transfer Equilibria Involving Heteroaromatic Nitrogen Systems. The "Lone Pair Charge" Approach," *J. Am. Chem. Soc.* **106**, 6552 (1984).

- 84CAU/FUR**
C. Cauletti, C. Furlani, G. Nicotra, K.-D. Schleinitz and W. Wegener, "Substituent Effects on the Electronic Structure of Some Styryl Methyl Sulphones Studied by Ultraviolet Photoelectron Spectroscopy," *J. Chem. Soc. Perkin Trans. II* 533 (1984).
- 84CHA/HIL**
H. Chatham, D. Hils, R. Robertson and A. Gallagher, "Total and Partial Electron Collisional Ionization Cross Sections for CH₄, C₂H₆, SiH₄, and Si₂H₆," *J. Chem. Phys.* 81, 1770 (1984).
- 84CHE/HAL**
P.T. Chesky and M.B. Hall, "Electronic Structure of Triple-Decker Sandwiches. Photoelectron Spectra and Molecular Orbital Calculations of Bis(η^5 -cyclopentadienyl)(μ , η^6 -benzene)divanadium and Bis(η^5 -cyclopentadienyl)(μ , η^6 -mesitylene)divanadium," *J. Am. Chem. Soc.* 106, 5186 (1984).
- 84CHI/KOR**
N.S. Chilingarov, M.V. Korobov, L.N. Sidirov, V.N. Mitkin, V.A. Shipachev and S.V. Zemskov, "Electron Affinity of Rhodium Tetrafluoride," *J. Chem. Therm.* 16, 965 (1984).
- 84CIL/DOR**
E. Ciliberto, K.A. Doris, W.J. Pietro, G.M. Reisner, D.E. Ellis, I. Fraga, F.H. Herbstein, M.A. Ratner and T.J. Marks, " π - π Interactions and Bandwidths in 'Molecular Metals'. A Chemical, Structural, Photoelectron Spectroscopic, and Hartree-Fock-Slater Study of Monomeric and Cofacially Joined Dimeric Silicon Phthalocyanines," *J. Am. Chem. Soc.* 106, 7748 (1984).
- 84COL/JIM**
M. Colomina, P. Jimenez, M.V. Roux and C. Turrion, *J. Chem. Thermodyn.* 16, 1121 (1984).
- 84COO/KRO**
T.A. Cooper, H.W. Kroto, C. Kirby and N.P.C. Westwood, "A Photoelectron Spectroscopic Study of the (FBS)_n System (n = 1-3)," *J. Chem. Soc. Dalton Trans.* 1047 (1984).
- 84CZI/TAM**
G. Czira, J. Tamas and G. Kalas, "Effects of Stereoisomerism on the Electron Impact Fragmentation of Some Compounds with an Eburnane Skeleton," *Org. Mass Spectrom.* 19, 555 (1984).
- 84DAO/PET**
P.D. Dao, K.I. Peterson and A.W. Castleman, Jr., "The Photoionization of Oxidized Metal Clusters," *J. Chem. Phys.* 80, 563 (1984).
- 84DEH/PAR**
J.L. Dehmer, A.C. Parr, S.H. Southworth and D.M.P. Holland, "Angle-Resolved Photoelectron Study of the Valence Levels of BF₃ in the Range 17 < hv < 28 eV," *Phys. Rev. A* 30, 1783 (1984).
- 84DEK/NIB**
L.J. de Koning and N.M.M. Nibbering, "Formation of the Long-Lived H₂O⁻ Ion in the Gas Phase," *J. Am. Chem. Soc.* 106, 7971 (1984).
- 84DEL/ABE**
E.W. Della, R.S. Abeywickrema, M.K. Livett and J.B. Peel, "The Photoelectron Spectra of Some 1-Halogenobicycloalkanes," *J. Chem. Soc. Perkin Trans. II* 1653 (1984).
- 84DEL/PIG**
E.W. Della, P.E. Pigou, M.K. Livett and J.B. Peel, "The Photoelectron Spectrum and Molecular Geometry of Bicyclo[2.1.1]Hexane," *J. Elec. Spectrosc. Rel. Phenom.* 33, 163 (1984).
- 84DEM/SIM**
H. Demian, I. Simiti and N. Palibroda, "Contributions to the Study of Some Heterocycles. 58-Mass Spectra of 2-Phenyl-4-carboxy- and 5-Carboxy-thiazoles," *Org. Mass Spectrom.* 19, 196 (1984).
- 84DEN/AUD**
J.P. Denhez and H.E. Audier, "Isomerization of [C₆H₅-C₃H₅O]⁺ Ions: The Case of 1-Phenylpropenol," *Org. Mass Spectrom.* 19, 407 (1984).
- 84DEP/BIE**
C.H. DePuy, V.M. Bierbaum and R. Damrauer, "Relative Gas-Phase Acidities of the Alkanes," *J. Am. Chem. Soc.* 106, 4051 (1984).
- 84DEP/DAM**
C.H. DePuy and R. Damrauer, "Reactions of Organosilane Anionic Species with Nitrous Oxide, *Organomet.* 3, 362 (1984).
- 84DEW/HEA**
M.J.S. Dewar, E.F. Healy and J.J.P. Stewart, *J. Comput. Chem.* 5, 4 (1984).
- 84DIX/KOM**
D.A. Dixon, A. Komornicki and W.P. Kraemer, "Energetics of the Protonation of CO: Implications for the Observation of HOC⁺ in Dense Interstellar Clouds," *J. Chem. Phys.* 81, 3603 (1984).
- 84DOG/POU**
J.P. Dognon, C. Pouchan, A. Dargelos and J.P. Flament, "Ab Initio CI Study and Vibronic Analysis of the Photoelectron Spectra of Formaldoxime," *Chem. Phys. Lett.* 109, 492 (1984).
- 84DRZ/BRA**
P.S. Drzaic and J.I. Brauman, "A Determination of the Singlet-Triplet Splitting in Phenylnitrene via Photoelectron Detachment," *J. Am. Chem. Soc.* 106, 3443 (1984).
- 84DRZ/BRA2**
P.S. Drzaic and J.I. Brauman, "A Determination of the Singlet-Triplet Splitting in Phenylnitrene via Photoelectron Detachment," *J. Phys. Chem.* 88, 5285 (1984).
- 84DRZ/MAR**
P.S. Drzaic, J. Marks and J.I. Brauman, "Electron Photodetachment from Gas Phase Molecular Anions," in "Gas Phase Ion Chemistry," V. 3, M.T. Bowers, Ed., Academic Press, NY, 1984, Ch. 21.
- 84DUD/GRE**
N. Dudeney, J.C. Green, O.N. Kirchner and F.S.J. Smallwood, "A Study of the Electronic Structure of the Dimers [Co₂(η -C₅Me₅)₂(CO)₂], [Rh₂(η -C₅Me₅)₂], [Co₂(η -C₅Me₅)₂(NO)₂] and [Co₂(η -C₅Me₅)₂(CO)(NO)] by He-I and He-II Photoelectron Spectroscopy," *J. Chem. Soc. Dalton Trans.* 1883 (1984).
- 84DUD/KIR**
N. Dudeney, O.N. Kirchner, J.C. Green and P. Maitlis, "Electronic Structure and Reactivity of Pentamethylcyclopentadienyl Complexes of Cobalt, Rhodium, and Iridium: He-I and He-II Photoelectron Spectroscopic Investigation," *J. Chem. Soc. Dalton Trans.* 1877 (1984).
- 84DUN/HON**
R.C. Dunbar and J.P. Honovich, "Threshold Ion Photodissociation. Bromobenzene and Iodobenzene Ions," *Int. J. Mass Spectrom. Ion Proc.* 58, 25 (1984).
- 84DYK/ELL**
J.M. Dyke, A.R. Ellis, N. Keddar and A. Morris, "A Reinvestigation of the First Band in the Photoelectron Spectrum of the Ethyl Radical," *J. Phys. Chem.* 88, 2565 (1984).
- 84DYK/ELL2**
J.M. Dyke, A.R. Ellis, N. Jonathan, N. Keddar and A. Morris, "Observation of the CH₂OH Radical in the Gas Phase by Vacuum Ultraviolet Photoelectron Spectroscopy," *Chem. Phys. Lett.* 111, 207 (1984).
- 84DYK/GRA**
J.M. Dyke, B.W.J. Gravenor, G.D. Josland, R.A. Lewis and A. Morris, "A Gas Phase Investigation of Titanium Monoxide and Atomic Titanium Using High Temperature Photoelectron Spectroscopy," *Molec. Phys.* 53, 465 (1984).

- 84DYK/JOS**
J.M. Dyke, G. D. Josland, J. G. Snijders and P.M. Boerrigter, "Ionization Energies of the Diatomic Halogens and Interhalogens Studied with Relativistic Hartree-Fock-Slater Calculations," *Chem. Phys.* **91**, 419 (1984).
- 84DYK/KIR**
J.M. Dyke, C. Kirby, A. Morris, B.W.J. Gravenor, R. Klein and P. Rosmus, "A Study of Aluminium Monofluoride and Aluminium Trifluoride by High-Temperature Photoelectron Spectroscopy," *Chem. Phys.* **88**, 289 (1984).
- 84DYK/LBW**
J.M. Dyke, A.E. Lewis and A. Morris, "A Photoelectron Spectroscopic Study of the Ground State of CF^+ via the Ionization Process $CF^+(X^1\Sigma^+) \rightarrow CF(X^2\Pi)$," *J. Chem. Phys.* **80**, 1382 (1984).
- 84ELB/KUD**
S. Elbel, J. Kudnig, M. Grodzicki and H.J. Lempka, "Photoelectron Spectra of Group V Compounds. The Elements: Sb_4 ," *Chem. Phys. Lett.* **109**, 312 (1984).
- 84ELK/ARM**
J.L. Elkind and P.B. Armentrout, "Threshold Behavior for Chemical Reactions: Line-of-Centers Cross Section for $Si^+(^2P) + H_2 \rightarrow SiH^+ + H$," *J. Phys. Chem.* **88**, 5454 (1984).
- 84ERA/KOL**
P.A. Erastov, V.P. Kolesov and I.K. Igumenov, *Russ. J. Phys. Chem.* **58**, 1311 (1984).
- 84FAR/SRI**
M. Farber and R.D. Srivastava, "Electron and Thermal Dissociation of $BF_3(g)$," *J. Chem. Phys.* **81**, 241 (1984).
- 84FLA/MAQ**
R. Flammang, A. Maquestiau, J. Catalan, P. Periz and J. Elguero, "Basicity of Azoles. Experimental Gas Phase Basicities Determined by Mass Spectrometry Towards ab initio Calculated Protonation Energies," *Org. Mass Spectrom.* **19**, 627 (1984).
- 84FRA/FRA**
M.L. Fraser-Monteiro, L. Fraser-Monteiro, J. de Wit and T. Baer, "Dissociation Dynamics of Energy-Selected Phenol Ions," *J. Phys. Chem.* **88**, 3622 (1984).
- 84FUC/HAL**
R. Fuchs and J.H. Hallman Unpublished results, personal communication to the authors.
- 84FUK/YOS**
K. Fuke, H. Yoshiuchi, K. Kaya, Y. Achiba, K. Sato and K. Kimura, "Multiphoton Ionization Photoelectron Spectroscopy and Two-Color Multiphoton Ionization Threshold Spectroscopy on the Hydrogen Bonded Phenol and 7-Azaindole in a Supersonic Jet," *Chem. Phys. Lett.* **108**, 179 (1984).
- 84FUR/MUR**
J. Furukawa, S. Murata, M. Sakiyama and S. Seki, *Bull. Chem. Soc. Jpn.* **57**, 3058 (1984).
- 84GAN/LIV**
T.H. Gan, M.K. Livett and J.B. Pecl, "Comparative He I and He II Photoelectron Spectroscopic Studies of the Benzoyl Halides," *J. Chem. Soc. Faraday Trans. II* **80**, 1281 (1984).
- 84GAT**
I.R. Gatland, "Swarms of Ions and Electrons In Gases," W. Lindinger, Ed., Springer-Verlag NY 1984. pg. 44.
- 84GEF/LIF**
S. Gefen and C. Lifshitz, "Time-Dependent Mass Spectra and Breakdown Graphs. V. The Kinetic Shift in Iodobenzene," *Int. J. Mass Spectrom. Ion Proc.* **58**, 251 (1984).
- 84GEL/POR**
G.I. Gellene and R.F. Porter, "Experimental Evidence for Metastable States of D_3O and Its Monohydrate by Neutralized Ion Beam Spectroscopy," *J. Chem. Phys.* **81**, 5570 (1984).
- 84GLE/BAR**
R. Gleiter, R. Bartetzko and D. Cremer, "Electronic Structure of 1,5-Dithia-2,4,6,8-tetraazocine. Model Calculations and Spectroscopic Investigations," *J. Am. Chem. Soc.* **106**, 3437 (1984).
- 84GLE/BIS**
R. Gleiter, P. Bischof, M.C. Bohm, R. Guillard and H. Yamaguchi, "The Electronic Structure of Heterotropones. Photoelectron Spectra and Molecular Orbital Calculations," *Bull. Chem. Soc. Jpn.* **57**, 856 (1984).
- 84GLE/BOH**
R. Gleiter, M.C. Bohm and R.D. Ernst, "The He(I) Photoelectron Spectrum of Bis(pentadienyl)-iron: A Comparison with the Ferrocene Spectrum," *J. Electron Spectrosc. Rel. Phenom.* **33**, 269 (1984).
- 84GLE/DOB**
R. Gleiter, W. Dobler, M. Eckert-Maksic, A.J. Bloodworth, H.J. Eggelte and D. Cremer, "Photoelectron Spectra of Dioxabicyclo[n.2.1]alkanes," *J. Org. Chem.* **49**, 3716 (1984).
- 84GLE/FRI**
R. Gleiter, G. Friedrich, M. Yoshifuji, K. Shibayama and N. Inamoto, "Photoelectron Spectra of Diaryldiphosphenes," *Chem. Lett.* **313** (1984).
- 84GLE/HAI**
R. Gleiter, R. Haider, P. Bischof, N.S. Zefirov and A.M. Boganov, "Electronic Structure of the Tricyclo[2.1.0.0^{2,5}]pentane System. Photoelectron Spectroscopic Investigations of 1,5-Dimethyl-3-exo-methylenetricyclo[2.1.0.0^{2,5}]pentane and 1,5-Dimethyltricyclo[2.1.0.0^{2,5}]pentan-3-one," *J. Org. Chem.* **49**, 375 (1984).
- 84GLE/SCH**
R. Gleiter, W. Schafer and A. Flatow, "Interaction between Triple Bonds in 1,8-Diethynyl-naphthalenes," *J. Org. Chem.* **49**, 372 (1984).
- 84GLE/SCH2**
R. Gleiter, W. Schafer, H.A. Staab and T. Saupe, "Electronic Structure of 4,5-Bis(dimethylamino)fluorene. PE Spectroscopic Investigations," *J. Org. Chem.* **49**, 4463 (1984).
- 84GLE/SPA**
R. Gleiter, J. Spanget-Larsen, H. Hopf and C. Mlynek, "Photoelectron Spectra of Some Reduction Products of [2.2]Paracyclophane," *Chem. Ber.* **117**, 1987 (1984).
- 84GLE/SPA2**
R. Gleiter, J. Spanget-Larsen and W.L.F. Armarego, "Through-bond Effects in Pyrazino[2,3-b]pyrazine. Photoelectron Spectroscopy and Model Calculations," *J. Chem. Soc. Perkin Trans. II* **1517** (1984).
- 84GOR/PYA**
L.N. Gorokhov, A.T. Pyatenko, I.V. Sidorova and V.K. Smirnov, "Thermochemistry of Molecules and Negative Ions in a Uranium-Oxygen-Fluorine System," *Probl. Kalorim. Khim. Termodyn. Dokl. Vses. Konf.* **2**, 66 (1984). CA 104, 11261c.
- 84GRA/ROS**
M. Grade, W. Rosinger and P.A. Dowben, "Core and Valence Electron Binding Energies of FeI_2 and Stabilities of Gas Phase Species," *Ber. Bunsenges. Phys. Chem.* **88**, 65 (1984).
- 84GRA/ROS2**
M. Grade and W. Rosinger, "A Mass Spectrometric Investigation of Iron(II)-Iodide," *Ber. Bunsenges. Phys. Chem.* **88**, 767 (1984).
- 84GRE/POW**
J.C. Green, P. Powell and J.E. van Tilborg, "He I and He II Photoelectron Spectroscopic Studies of the Bonding in Cobalt, Rhodium, and Iridium Cyclopentadienyl Diene Complexes," *Organometall.* **3**, 211 (1984).
- 84GRE/STB**
A. Greenberg, S.E. Stein and R.L. Brown, *Sci. Totl. Env.* **40**, 219 (1984).
- 84GRO/CHE**
G.S. Groenwold, E.K. Chess and M.L. Gross, "Structure of the Intermediate Formed in the Reaction of the Styrene Radical Cation and Neutral Styrene," *J. Am. Chem. Soc.* **106**, 539 (1984).

- 84GRO/GRO**
G.S. Groenwold and M.L. Gross, "Cation Radical Diels-Alder Reaction of 1,3-Butadiene: A Two-Step Cycloaddition," *J. Am. Chem. Soc.* **106**, 6569 (1984).
- 84GRU/WHE**
S.G. Grubb, R.L. Whetten, A.C. Albrecht and E.R. Grant, "A Precise Determination of the First Ionization Potential of Benzene," *Chem. Phys. Lett.* **108**, 420 (1984).
- 84GUB/GER**
A.I. Gubareva, P.A. Gerasimov and V.V. Beregovykh, *Zh. Prikl. Khim.* **57**, 2297 (1984).
- 84HAL/KLE**
L.F. Halle, F.S. Klein and J.L. Beauchamp, "Properties and Reactions of Organometallic Fragments in the Gas Phase. Ion Beam Studies of FeH^+ ," *J. Am. Chem. Soc.* **106**, 2543 (1984).
- 84HAR/HOU**
A.G. Harrison, R. Houriet and T.T. Tidwell, "Gas Phase Basicities of Substituted Styrenes. Comparison of Gas Phase and Solution Reactivities," *J. Org. Chem.* **49**, 1302 (1984).
- 84HEI/BAR**
T. Heins, R. Bar, K. Borlin and M. Jungen, "Photoionization of 1,1-Difluoroethane: The Structure of the $\text{C}_2\text{H}_3\text{F}_2^+$ Ions," *Chem. Phys. Lett.* **105**, 327 (1984).
- 84HEI/HON**
E. Heilbronner, E. Honegger, W. Zambach, P. Schmitt and H. Gunther, "The Persistence of Ribbon Orbitals in Polycyclic Alkanes," *Helv. Chim. Acta* **67**, 1681 (1984).
- 84HEI/WIR**
E. Heilbronner, J. Wirz and R.L. Soulen, "6. The $\text{He}(\text{I}\alpha)$ Photoelectron Spectra of the Perfluoroderivatives of Trisannulated Benzenes and Tetrakisannulated Cyclooctatetraenes," *Helv. Chim. Acta* **67**, 47 (1984).
- 84HER/WEN**
N. Hernandez-Gill, W.E. Wentworth and E.C.M. Chen, "Electron Affinities of Fluorinated Phenoxy Radicals," *J. Phys. Chem.* **88**, 6181 (1984).
- 84HIE/PAU**
P.M. Hierl and J.F. Paulson, "Translational Energy Dependence of Cross Sections for Reactions of OH^+ (H_2O)_n with CO_2 and SO_2 ," *J. Chem. Phys.* **80**, 4890 (1984).
- 84HIL**
K. Hilpert, "Vaporization of Sodium Iodide and Thermochemistry of $(\text{NaI})_2(\text{g})$ and $(\text{NaI})_3(\text{g})$: An Experimental and Theoretical Study," *Ber. Bunsenges. Phys. Chem.* **88**, 132 (1984).
- 84HOH/DIS**
G. Hohlneicher, D. Distler, M. Muller and H.-J. Freund, "Identification of Shake-Up Satellites in Valence Photoelectron Spectra of Organic Compounds by Comparison with Electronic Absorption Spectra of Radical Cations. Case Study: Octafluoronaphthalene," *Chem. Phys. Lett.* **111**, 151 (1984).
- 84HOL/LOS**
J.L. Holmes and F.P. Lossing, *Int. J. Mass Spectrosc. Ion Proc.* **58**, 113 (1984).
- 84HOL/MOM**
J.L. Holmes, A.A. Mommers, J.E. Szulejko and J.K. Terlouw, "Two New Stable $[\text{C}_3\text{H}_6\text{O}]^+$ Isomers: the Radical Cations $[\text{C}_3\text{H}_6\text{OH}_2]^+$," *J. Chem. Soc. Chem. Commun.* **165** (1984).
- 84HOL/SZU**
J.L. Holmes and J.E. Szulejko, "The Generation and Identification of the Transient Vinylidene Cation," *Chem. Phys. Lett.* **107**, 301 (1984).
- 84HON/HEI**
E. Honegger, E. Heilbronner, A. Dratva and C.A. Grob, "197. 'Lone Pair' and 'C-I Bond' Ionization Energies of exo- and endo-2-Norbornyl Iodides," *Helv. Chim. Acta* **67**, 1691 (1984).
- 84HON/ZHO**
E. Honegger, Y. Zhong-zhi, E. Heilbronner, W.v.E. Doering and J.C. Schmidhauser, "Photoelectron-Spectroscopic Characterization of 3,4-Dimethyl-2,4-hexadienes," *Helv. Chim. Acta* **67**, 640 (1984).
- 84HOP/JAH**
H.P. Hopkins, Jr., D.V. Jahagirdar, P.S. Moulik, D.H. Aue, H.M. Webb, W.R. Davidson and M.D. Pedley, "Basicities of the 2-, 4-, 2,4-Di and 2,6-Disubstituted t-Butyl Pyridines in the Gas Phase and Aqueous Phase: Steric Effects in the Solvation of tert-Butyl-Substituted Pyridines and Pyridinium Cations," *J. Am. Chem. Soc.* **106**, 4341 (1984).
- 84HOW/GON**
J.O. Howell, J.M. Goncalves, C. Amatore, L. Klasinc, R.M. Wightman and J.K. Kochi, "Electron Transfer from Aromatic Hydrocarbons and Their π -Complexes with Metals. Comparison of the Standard Oxidation Potentials and Vertical Ionization Potentials," *J. Am. Chem. Soc.* **106**, 3968 (1984).
- 84JAC/JAC**
T.C. Jackson, D.B. Jacobson and B.S. Freiser, "Gas Phase Reactions of FeO^+ with Hydrocarbons," *J. Am. Chem. Soc.* **106**, 1252 (1984).
- 84JEN/MOR**
H.D.B. Jenkins and D.F.C. Morris, "Proton Affinity of Gaseous Ammonia," *J. Chem. Soc. Faraday Trans. II* **80**, 1167 (1984).
- 84JOR/THO**
F.S. Jorgensen and T. Thomsen, "Synthesis and Structural Analysis of the Bridged peri-Naphthalene: 1,3-Dihydro-2-phenalenone," *Acta Chem. Scand.* **B38**, 113 (1984).
- 84KAT/SHI**
S. Katsumata, H. Shiromaru and T. Kimura, "Photoelectron Angular Distribution and Assignment of Photoelectron Spectrum of Ozone," *Bull. Chem. Soc. Jpn.* **57**, 1784 (1984).
- 84KIN/KRO**
M.A. King and H.W. Kroto, "He I Photoelectron Study of Cyanogen Isothiocyanate, NCNCS , Produced by Thermal Isomerization of Sulfur Dicyanide, $\text{S}(\text{CN})_2$," *J. Am. Chem. Soc.* **106**, 7347 (1984).
- 84KIR/POP**
K. Kirste, R. Poppek and P. Rademacher, "Photoelektronenspektren und Konformationsverhalten von Azinen," *Chem. Ber.* **117**, 1061 (1984).
- 84KLA/KUH**
D. Klapstein, R. Kuhn, J.P. Maier, L. Misev and M. Ochsner, "138. Spectroscopic Characterization of Open-Shell Cations: Emission and Laser Excitation Spectra of Rotationally Cooled $\text{CH}_3(-\text{C}=\text{C}-)_2\text{X}^+$, $\text{X} = \text{Cl}, \text{Br}$," *Helv. Chim. Acta* **67**, 1222 (1984).
- 84KLA/MAI**
D. Klapstein, J.P. Maier, M. Ochsner and W. Zambach, "Emission and Laser Excitation Spectra of the $\text{A}^2\Pi = \text{X}^2\Pi$ Transition of Rotationally Cooled Bromochloroacetylene Cation in the Gas Phase," *J. Electron Spectrosc. Rel. Phenom.* **34**, 161 (1984).
- 84KOB**
T. Kobayashi, "Angle-Resolved Photoelectron Spectroscopy of Intramolecular Hydrogen Bond Systems: 2-Chloroethanol and 2-Bromoethanol," *Phys. Lett.* **103A**, 424 (1984).
- 84KOB/YOS**
T. Kobayashi, Z. Yoshida, H. Awaji, T. Kawase and S. Yoneda, "Intramolecular Orbital Interaction in 6,6'-Bi(1,4-dithiafulvenyl) Studied by Photoelectron Spectroscopy," *Bull. Chem. Soc. Jpn.* **57**, 2591 (1984).

- 84KOR/CHI**
M.V. Korobov, N.S. Chilingarov, N.A. Igolkina, M.I. Nikitin and L.N. Sidorov, "Molecules with a High Electron Affinity. Negative Ions in the Saturated Vapor of the Platinum-Manganese Trifluoride System," *Russ. J. Phys. Chem.* **58**, 1368 (1984).
- 84KUH/MAI**
R. Kuhn, J.P. Maier and F. Thommen, "Fluorescence Quantum Yields and Lifetimes of Fluorobenzene Cations in Selected Levels of Their B and C States Determined by Photoelectron-Photon Coincidence Spectroscopy," *Chem. Phys.* **57**, 319 (1981).
- 84LAN/LEE**
K.R. Lane, R.E. Lee, L. Sallans and R.R. Squires, "Formation and Reactivity of $\text{Fe}(\text{CO})_4\text{COOH}^-$ in the Gas Phase. Implication for the $\text{Fe}(\text{CO})_5$ Catalyzed Water Gas Shift Reaction," *J. Am. Chem. Soc.* **106**, 5767 (1984).
- 84LAR/MCM**
J.W. Larson and T.B. McMahon, "Fluoride and Chloride Affinities of Main Group Oxides, Fluorides, Oxofluorides, and Alkyls. Quantitative Scales of Lewis Acidities from Ion Cyclotron Resonance Halide-Exchange Equilibria," *J. Phys. Chem.* **88**, 1083 (1984).
- 84LAR/MCM2**
J.W. Larson and T.B. McMahon, "Hydrogen Bonding in Gas Phase Anions. An Experimental Investigation of the Interaction between Chloride Ion and Bronsted Acids from ICR Chloride Exchange Equilibria," *J. Am. Chem. Soc.* **106**, 517 (1984).
- 84LAR/MCM3**
J.W. Larson and T.B. McMahon, "Gas Phase Bihalide and Pseudohalide Ions. An ICR Determination of Hydrogen Bond Energies in XHY^- Species (X, Y = F, Cl, Br, CN)," *Inorg. Chem.* **23**, 2029 (1984).
- 84LAR/MCM4**
J.W. Larson and T.B. McMahon, "Gas Phase Negative Ion Chemistry of Alkylchloroformates," *Can. J. Chem.* **62**, 675 (1984).
- 84LEB/GUT**
N.D. Lebedeva, N.M. Gutner, Y.A. Katin, N.M. Kozlova, N.N. Kiseleva, E.F. Makhina and S.K. Dobyshin, *Zh. Prikl. Khim.* **57**, 2297 (1984).
- 84LEW/GOL**
K.E. Lewis, D.M. Golden and G.P. Smith, "Organometallic Bond Dissociation Energies: Laser Pyrolysis of $\text{Fe}(\text{CO})_5$, $\text{Cr}(\text{CO})_6$, $\text{Mo}(\text{CO})_6$, and $\text{W}(\text{CO})_6$," *J. Am. Chem. Soc.* **106**, 3905 (1984).
- 84LIA/BUC**
S.G. Lias and T.J. Buckley, "Structures and Reactions of C_3H_6^+ Ions Generated in Cyclopropane," *Int. J. Mass Spectrom. Ion Proc.* **56**, 123 (1984).
- 84LIA/LIE**
S.G. Lias, J.F. Liebman and R.D. Levin, "Evaluated Gas Phase Basicities and Proton Affinities of Molecules; Heats of Formation of Protonated Molecules," *J. Phys. Chem. Ref. Data* **13**, 695 (1984).
- 84LIC/BLE**
D.L. Lichtenberger and C.H. Blevins II, "Contribution of a σ -Orbital Electron to a Quadruple Metal-Metal Bond. A Direct Experimental Measure from Vibrational Fine Structure in the Σ Ionization of $\text{Mo}_2(\text{O}_2\text{CCH}_3)_4$," *J. Am. Chem. Soc.* **106**, 1636 (1984).
- 84LIC/CAL**
D.L. Lichtenberger, D.C. Calabro and G.E. Kellogg, "Electronic Structure and Bonding Characteristics of Cyclopentadienyl d^8 Metal-Ligand Complexes. Core and Valence Ionization Study of $\text{CpM}(\text{CO})_2$ Where M = Co and Rh and Cp = $\eta^5\text{-C}_5\text{H}_5$ and $\eta^5\text{-C}_5(\text{CH}_3)_5$," *Organometall.* **3**, 1623 (1984).
- 84LIF/MAL**
C. Lifshitz and Y. Malinovich, "Time Resolved Photoionization Mass Spectrometry in the Millisecond Range," *Int. J. Mass Spec. Ion Proc.* **60**, 99 (1984).
- 84LIN/LIA**
S.H. Linn, C.L. Liao, C.X. Liao, J.M. Brom, Jr. and C.Y. Ng, "Photoionization Study of Hg_2 ," *Chem. Phys. Lett.* **105**, 645 (1984).
- 84LOS/HOL**
F.P. Lossing and J.L. Holmes, "Stabilization Energy and Ion Size in Carbocations in the Gas Phase," *J. Am. Chem. Soc.* **106**, 6917 (1984).
- 84LOU/HEN**
J.N. Louwen, R. Hengelmolen, D.M. Grove, A. Oskam and R.L. DeKock, "Ultraviolet Photoelectron Spectra of Square-Planar Complexes of Nickel Triad Metals. 3. He I and He II Spectra of trans- $\{(\text{PEt}_3)_2\text{MXY}\}$ (M = Pd, Pt. X = Y = $\text{C}\equiv\text{CH}$, $\text{C}\equiv\text{CCH}_3$, $\text{C}\equiv\text{N}$. X = Cl, Y = $\text{C}\equiv\text{N}$) and Hartree-Fock-Slater Calculations on Model Compounds," *Organometall.* **3**, 908 (1984).
- 84LOU/STU**
J.N. Louwen, D.J. Stufkens and A. Oskam, "He(I) and He(II) Photoelectron Spectra of Some Organozinc and Organoaluminum Radicals Containing 1,2-Bis(t-butylimino)ethane," *J. Chem. Soc. Dalton Trans.* 2683 (1984).
- 84MAG/CAL**
T.F. Magnera, G. Caldwell, J. Sumner, S. Ikuta and P. Kebarle, "Solvation of the Halide Anions in Dimethyl Sulfoxide. Factors Involved in Enhanced Reactivity of Negative Ions in Dipolar Aprotic Solvents," *J. Am. Chem. Soc.* **106**, 6140 (1984).
- 84MAI/THO**
J.P. Maier and F. Thommen, "Relaxation Dynamics of Open-Shell Cations Studied by Photoelectron-Photon Coincidence Spectroscopy" *Gas Phase Ion Chemistry*, Vol. 3 (M. T. Bowers, editor), Academic Press, Inc., p. 357 (1984).
- 84MAL/ROS**
T.E. Mallouk, G.L. Rosenthal, G. Muller, R. Brusasco and N. Bartlett, "Fluoride Ion Affinities of GeF_4 and BF_4 from Thermodynamic and Structural Data for $(\text{SF}_2)_2\text{GeF}_6$, ClO_2GeF_5 , and ClO_2BF_4 ," *Inorg. Chem.* **23**, 3167 (1984).
- 84MAR/KAY**
A.P. Marchand, R. Kaya and A.D. Baker, "A Photoelectron Spectroscopic Study of Proximity Effects in 8,11-Disubstituted Pentacyclo(5.4.0.0^{2,6}.0^{3,10}.0^{5,9})undecanes," *Tetrahedron Lett.* **25**, 795 (1984).
- 84MAU/NEL**
M. Meot-Ner (Mautner), S.F. Nelsen, M.R. Willi and T.B. Frigo, "Special Effects of an Unusually Large Neutral to Radical Cation Geometry Change. Adiabatic Ionization Energies and Proton Affinities of Alkylhydrazines," *J. Am. Chem. Soc.* **106**, 7384 (1984).
- 84MCA/HUD**
D.J. McAdoo, C.E. Hudson, F.W. McLafferty and T.E. Parks, "The Reactions of Metastable $[\text{C}_5\text{H}_{10}\text{O}]^+$ Ions with the Oxygen on the Second Carbon," *Org. Mass Spectrom.* **19**, 353 (1984).
- 84MCD/CHO**
R.N. McDonald, A.K. Chowdhury and W.D. McGhee, "Gas-Phase Generation of 1,1,1,3,3,3-Hexafluoroisopropylidene Anion Radical: Proton Affinity and Heat of Formation of $(\text{CF}_3)_2\text{C}^-$ and $(\text{CF}_3)_2\text{CH}^-$," *J. Am. Chem. Soc.* **106**, 4112 (1984).
- 84MCD/CHO2**
R.N. McDonald, A.K. Chodhury and P.L. Schell, "Generations and Reactions of $(\text{OC})_3\text{Fe}^-$ in a Flowing Afterglow Apparatus," *J. Am. Chem. Soc.* **106**, 6095 (1984).
- 84MCM/HAL**
J.E. McMurry, G.J. Haley, J.R. Matz, J.C. Clardy and G. Van Duyn, "Tetracyclo[8.2.2.2^{2,5}.2^{6,9}]octadeca-1,5,9-triene," *J. Am. Chem. Soc.* **106**, 5018 (1984).
- 84MEA/STE**
R.D. Mead, A.E. Stevens and W.C. Lineberger, "Photodetachment in Negative Ion Beams," *Gas Phase Ion Chemistry*, Vol. 3 (M.T. Bowers, editor), Academic Press, NY, (1984), Ch. 22.

- 84MEY/SCH**
H.-J. Meyer, T. Schulze and U. Ross, "Molecular Beam Study of the Chemi-Ionization in the Reactive Scattering of Ca and Sr with Cl₂ and Br₂ at Collision Energies C(c.m.) < 4.5 eV," *Chem. Phys.* **90**, 185 (1984).
- 84MIL/BAE**
B.E. Miller and T. Baer, "Kinetic Energy Release Distribution in the Fragmentation of Energy-Selected Vinyl and Ethyl Bromide Ions," *Chem. Phys.* **85**, 39 (1984).
- 84MOM/BUR**
A.A. Mommers, P.C. Burgers, J.L. Holmes and J.K. Terlouw, "Isomeric [C₃H₄]⁺ Ions: Their Identification and Generation in Dissociative Ionizations," *Org. Mass Spectrom.* **19**, 7 (1984).
- 84MOR/POW**
B.J. Morris-Sherwood, C.B. Powell and M.B. Hall, "Photoelectron Spectra and Molecular Orbital Calculations on Bis(cyclopentadienyldicarbonylchromium, -molybdenum, and -tungsten): Nature of the Bonding of Linear Semibridging Carbonyls," *J. Am. Chem. Soc.* **106**, 5079 (1984).
- 84MOS/MOS**
J. Moskal, A. Moskal and K. Nagraba, "Conjugated Schiff's Bases. 16 - Substituent Effect on Electron Impact Fragmentation of Some 1-Oxa-4-azabutadienes," *Org. Mass Spectrom.* **19**, 87 (1984).
- 84MOY/BRA**
C.R. Moyland and J.I. Brauman, "Bond Dissociation Energies in Alcohols: Kinetic and Photochemical Evidence Regarding Ion Thermochemistry," *J. Phys. Chem.* **88**, 3175 (1984).
- 84MUL/SAN**
K. Muller-Dethlefs, M. Sander and E.W. Schlag, "Two-Colour Photoionization Resonance Spectroscopy of NO: Complete Separation of Rotational Levels of NO⁺ at the Ionization Threshold," *Chem. Phys. Lett.* **112**, 291 (1984).
- 84NAD/REI**
I. Nadler, H. Reisler, M. Noble and C. Wittig, *Chem. Phys. Lett.* **108**, 115 (1984).
- 84NEL**
S.F. Nelsen, "Ionization from Nitrogen and Oxygen Lone Pairs: A Comparison of Trialkylamine, Dialkyl Ether, Tetraalkylhydrazine, and Dialkyl Peroxide Photoelectron Spectroscopic Ionization Potentials," *J. Org. Chem.* **49**, 1891 (1984).
- 84NES/VER**
T.N. Nesterova, S.P. Verekin, S.Y. Karaseva, A.M. Rozhnov and V.F. Tsyetkov, *Russ. J. Phys. Chem.* **58**, 491 (1984).
- 84NIK/OVC**
E.N. Nikolaev, K.V. Ovchinnikov and G.A. Semenov, "Composition of Vapor over Lead Molybdate and Tungstate," *Zh. Obs. Khim.* **54**, 977 (1984). English trans.: *J. Gen. Chem. (USSR)* **54**, 869 (1984).
- 84NOV/POT**
I. Novak and A.W. Potts, "The UV Gas-Phase Photoelectron Spectra of Group IVB Tetraphenyl Derivatives," *J. Organometall. Chem.* **262**, 17 (1984).
- 84NOV/POT2**
I. Novak and A.W. Potts, "Ultraviolet Photoelectron Spectra of Gas Phase and Condensed Tin and Lead Dihalides (MX₂, M = Sn, Pb; X = Cl, Br, I)" *J. Electron Spectrosc. Rel. Phenom.* **33**, 1 (1984).
- 84OAK/ELL**
J.M. Oakes and G.B. Ellison, "Photoelectron Spectroscopy of the Allylic Anion," *J. Am. Chem. Soc.* **106**, 7734 (1984).
- 84OHN/MAT**
K. Ohno, S. Matumoto, K. Imai and Y. Haraa, "Penning Ionization Electron Spectroscopy of Nitriles," *J. Phys. Chem.* **88**, 206 (1984).
- 84OLI/FLE**
J.A. Olivares, G.D. Flesch and H.J. Svec, "Mass Spectrometry of Five Ketene Dimers," *Int. J. Mass Spectrom. Ion Proc.* **56**, 293 (1984).
- 84OLI/GUE**
P.R. Olivato, S.A. Guerrero, A. Modelli, G. Granozzi, D. Jones and G. Distefano, "Electronic Interaction in Heterosubstituted Acetones Studied by Means of Ultraviolet Photoelectron and Electron Transmission Spectroscopy," *J. Chem. Soc. Perkin Trans. II* **1505** (1984).
- 84ORI/SRI**
O.J. Orient and S.K. Srivastava, "Mass Spectrometric Determination of Partial and Total Electron Impact Ionization Cross Sections of SO₂ from Threshold up to 200 eV," *J. Chem. Phys.* **80**, 140 (1984).
- 84PAN/BAE**
M. Panczel and T. Baer, "A Photoelectron Photoion Coincidence (PEPICO) Study of Fragmentation Rates and Kinetic Energy Release Distributions in Nitrobenzene," *Int. J. Mass Spectrom. Ion Proc.* **58**, 43 (1984).
- 84PAQ/DAN**
L.A. Paquette, R.G. Daniels and R. Gleiter, "Synthesis, Reactivity, and Electronic Structure of (2-(Trimethylsilyl)-1,3-cyclohexadiene)iron Tricarbonyl Complexes," *Organometall.* **3**, 560 (1984).
- 84PAR/WEX**
E.K. Parks and S. Wexler, *J. Phys. Chem.* **88**, 4492 (1984).
- 84PAU/HEN**
J.F. Paulson and M.J. Henchman, "On the Formation of H₃O⁺ in an Ion-Molecule Reaction," in "Ionic Processes in the Gas Phase," M.A. Almoester Ferreira, Ed., Reidel, Dordrecht, 1984, p.331.
- 84PED/UND**
A.H. Pedersen and K. Undheim, "N-Quaternary Compounds. Part LIX. Facile Synthesis of 3-Vinyl-4(3H)-pyrimidinethiones," *J. Heterocyc. Chem.* **21**, 1149 (1984).
- 84PET/DAO**
K.I. Peterson, P.D. Dao, R.W. Farley and A.W. Castleman, Jr., "Photoionization of Sodium Clusters," *J. Chem. Phys.* **80**, 1780 (1984).
- 84PLA/SIM**
V.A. Platonov and Y.N. Simulin, *Russ. J. Phys. Chem.* **58**, 1630 (1984).
- 84POL/MUN**
C.W. Polley, Jr. and B. Munson, "The Proton Affinity of Nitrogen Dioxide," *Int. J. Mass Spectrom. Ion Proc.* **59**, 333 (1984).
- 84POL/TRE**
J.E. Pollard, D.J. Trevor, J.E. Reutt, Y.T. Lee and D.A. Shirley, "Torsional Potential and Intramolecular Dynamics in the C₂H₄⁺ Photoelectron Spectra," *J. Chem. Phys.* **81**, 5302 (1984).
- 84PYA/GOR**
A.T. Pyatenko and L.N. Gorokhov, "Electron Affinities of the U₂F_n Molecules (8 < n < 12)," *Chem. Phys. Lett.* **105**, 205 (1984). [Duplicates, *Russ. J. Phys. Chem.* **58**, 1624 (1984).]
- 84PYA/GUS**
A.T. Pyatenko, A.V. Gusarov and L.N. Gorokhov, "Thermochemistry of Negative Ions in the U-F System," *Russ. J. Phys. Chem.* **58**, 1 (1984).
- 84RAU/SCH**
H. Rau and E. Schnedler, *J. Chem. Thermodyn.* **16**, 673 (1984).
- 84REE/MUJ**
W.D. Reents, Jr. and A.M. Mujsce, "Ion/Molecule Reactions of Silicon Tetrafluoride," *Int. J. Mass Spectrom. Ion Proc.* **59**, 65 (1984).
- 84ROH/COX**
E.A. Rohlffing, D.M. Cox and A. Kaldor, "Production and Characterization of Supersonic Carbon Cluster Beams," *J. Chem. Phys.* **81**, 3322 (1984).

- 84ROH/COX2**
E.A. Rohlffing, D.M. Cox, A. Kaldor and K.H. Johnson, "Photoionization Spectra and Electronic Structure of Small Iron Clusters," *J. Chem. Phys.* **81**, 3846 (1984).
- 84ROL/HOU**
E. Rolli and R. Houriet, "Basicite en Phase Gaseuse et Site de Protonation des p-Nitroanilines $R_2NC_6H_4NO_2$ ($R = H, CH_3$)," *Int. J. Spectrosc.* **3**, 177 (1984).
- 84ROT**
W.R. Roth, Personal communication.
- 84RUS/CUR**
B.M. Ruscic, L.A. Curtiss and J. Berkowitz, "Photoelectron Spectrum and Structure of B_2O_2 ," *J. Chem. Phys.* **80**, 3962 (1984).
- 84SCH**
B. Schaldach, "Chemistry of Gaseous Ions. Part VIII. Bond Strength and Mass Spectrometric Fragmentation of 2-Methoxycinnamic Acid," *Int. J. Mass Spectrom. Ion Proc.* **56**, 237 (1984).
- 84SCH/HOU**
J.C. Schultz, F.A. Houle and J.L. Beauchamp, "Photoelectron Spectroscopy of 1-Propyl, 1-Butyl, Isobutyl, Neopentyl and 2-Butyl Radicals: Free Radical Precursors to High Energy Carbonium Ion Isomers," *J. Am. Chem. Soc.* **106**, 3917 (1984).
- 84SCH/HOU2**
J.C. Schultz, F.A. Houle and J.L. Beauchamp, "Photoelectron Spectroscopy of Isomeric C_4H_7 Radicals. Implications for the Thermochemistry and Structures of the Radicals and Their Corresponding Carbonium Ions," *J. Am. Chem. Soc.* **106**, 7336 (1984).
- 84SHA/URA**
M. Shahbaz, S. Urano, P.R. LeBreton, M.A. Rossman, R.S. Hosmane and N.J. Leonard, "Tri-s-triazine: Synthesis, Chemical Behavior and Spectroscopic and Theoretical Probes of Valence Orbital Structure," *J. Am. Chem. Soc.* **106**, 2805 (1984).
- 84SHI/GIN**
I. Shin and K.A. Gingerich, *J. Chem. Phys.* **81**, 5937 (1984).
- 84SMI/ADA**
D. Smith, N.G. Adams and E.E. Ferguson, "The Heat of Formation of $C_3H_2^+$," *Int. J. Mass Spectrom. Ion Proc.* **61**, 15 (1984).
- 84SMI/HAG**
M.A. Smith, J.W. Hager and S.C. Wallace, "Two Color Photoionization Spectroscopy of Jet Cooled Aniline: Vibrational Frequencies of the Aniline X^2B_1 Radical Cation," *J. Chem. Phys.* **80**, 3097 (1984).
- 84SMI/HAG2**
M.A. Smith, J.W. Hager and S.C. Wallace, "Two-Color Laser Photoionization Spectroscopy in a Collisionless Free-Jet Expansion: Spectroscopy and Excited-State Dynamics of Diazabicyclooctane," *J. Phys. Chem.* **88**, 2250 (1984).
- 84STA/MAQ**
D. Stahl and F. Maquin, "Charge-Stripping Mass Spectrometry of Molecular Ions from Polyacenes and Molecular Orbital Theory," *Chem. Phys. Lett.* **108**, 613 (1984).
- 84STE/MAR**
K. Stephen, T.D. Mark, J.H. Futrell and H. Helm, "Electron Impact Ionization of $(N_2)_2$: Appearance Energies of N_3^+ and N_4^+ ," *J. Chem. Phys.* **80**, 3185 (1984).
- 84STO/SPL**
J.A. Stone and D.E. Splinter, "A High-Pressure Mass Spectrometric Study of the Binding of $(CH_3)_3Sn^+$ to Lewis Bases in the Gas Phase," *Int. J. Mass Spectrom. Ion Proc.* **59**, 169 (1984).
- 84SZE/BAE**
L. Szepes and T. Baer, "Dissociation Dynamics of Energy Selected Hexamethyldisilane Ions and the Heats of Formation of $(CH_3)_3Si^+$ and $(CH_3)_3Si$," *J. Am. Chem. Soc.* **106**, 273 (1984).
- 84SZE/BAE2**
L. Szepes and T. Baer, " $(CH_3)_3SiX$ típusu vegyületek vizsgálatá fotoelektron-fotoion koincideneciával, I. Meghatározott belső energiájú hexametil-diszilán-ionok disszociációjának dinamikája," *Mag. Kem. Foly.* **90**, 104 (1984).
- 84TER/LOU**
A. Terpstra, J.N. Louwen, A. Oskam and J.H. Teuben, "The He(I) and He(II) Photoelectron Spectra of Some η^5 -Cyclopentadienyl-Titanium, -Zirconium and -Hafnium Trihalide Complexes," *J. Organometall. Chem.* **260**, 207 (1984).
- 84TOB/TAJ**
S. Tobita, S. Tajima and T. Tsuchiya, "The Substituent Effect on the Single and Double Hydrogen Atom Transfer Reactions in para-Substituted Benzoic Acid Isobutyl Esters," *Org. Mass Spectrom.* **19**, 326 (1984).
- 84TOL/BEA**
M.A. Tolbert and J.L. Beauchamp, "Activation of Carbon-Hydrogen and Carbon-Carbon Bonds by transition-Metal Ions in the Gas Phase. Exhibition of Unique Reactivity by Scandium Ions," *J. Am. Chem. Soc.* **106**, 8117 (1984).
- 84TRA**
J.C. Traeger, "A Study of the Allyl Cation Thermochemistry by Photoionization Mass Spectrometry," *Int. J. Mass Spectrom. Ion Proc.* **58**, 259 (1984).
- 84TRE/POL**
D.J. Trevor, J.E. Pollard, W.D. Brewer, S.H. Southworth, C.M. Truesdale, D.A. Shirley and Y.T. Lee, "Photoionization Mass Spectrometry of Ne Dimers," *J. Chem. Phys.* **80**, 6083 (1984).
- 84TUM/FOS**
W. Tumas, R.F. Foster and J.I. Brauman, "Unimolecular Decompositions of Gas-Phase Alkoxide Anions," *J. Am. Chem. Soc.* **106**, 4053 (1984).
- 84TUR**
F. Turecek, "2-Hydroxybutadiene: Preparation, Ionization Energy and Heat of Formation," *Tetrahedron Lett.* **25**, 5133 (1984).
- 84TUR2**
F. Turecek, "(E)- and (Z)-Prop-1-en-ol: Gas Phase Generation and Determination of Heats of Formation by Mass Spectrometry," *J. Chem. Soc. Chem. Commun.* 1374 (1984).
- 84TUR/HAN**
F. Turecek and V. Hanus, "Loss of Methyl from $[H_2C=C(OH)-CH_3]^+$ Ions Prepared by Electron Impact Ionization of Unstable 2-Hydroxypropene," *Org. Mass Spectrom.* **19**, 631 (1984).
- 84VAN/DEL**
H. Van Lonkhuizen and C.A. De Lange, "U. V. Photoelectron Spectroscopy of OH and OD Radicals," *Molec. Phys.* **51**, 551 (1984).
- 84VAN/DEL2**
H. Van Lonkhuizen and C.A. De Lange, "High-Resolution UV Photoelectron Spectroscopy of Diatomic Halogens," *Chem. Phys.* **89**, 313 (1984).
- 84VES/HAR**
T. Veszpremi, Y. Harada, K. Ohno and H. Mutoh, Jr., "Photoelectron and Penning Electron Spectroscopic Investigation of Phenylhalosilanes," *Organometall. Chem.* **266**, 9 (1984).
- 84VIS/HIL**
R. Viswanathan and K. Hilpert, "Mass Spectrometric Study of the Vaporization of Cesium Iodide and Thermochemistry of $(CsI)_2(g)$ and $(CsI)_3(g)$," *Ber. Bunsenges. Phys. Chem.* **88**, 125 (1984).
- 84WAN/CAP**
H. Wankenne, G. Caprace and J. Momigny, "Unimolecular Decay of Metastable Ions in Formaldehyde," *Int. J. Mass Spectrom. Ion Proc.* **57**, 149 (1984).

- 84WAN/DIL**
R.-G. Wang, M.A. Dillon and D. Spence, "Electron Spectroscopy of Hydrogen Chloride from 5 to 19 eV," *J. Chem. Phys.* **80**, 63 (1984).
- 84WIB/LUP**
K.B. Wiberg, E.C. Lupton, D.J. Wasserman, A. de Meijere and S.R. Kass, *J. Am. Chem. Soc.* **106**, 1740 (1984).
- 84WIB/WAS**
K.B. Wiberg, D.J. Wasserman and E. Martin, *J. Phys. Chem.* **88**, 3684 (1984).
- 84ZHO/HEI**
Y. Zhong-Zhi, E. Heilbronner, V. Boekelheide and J. Garbe, "The He(Ia) PE Spectra of Cyclophanes Containing a Cyclooctatetraene Ring," *Int. J. Mass Spectrom. Ion Proc.* **58**, 233 (1984).
- 84ZWI/HAR**
J.J. Zwinselman and A.G. Harrison, "An Energy-resolved Study of the Fragmentation of Ionized 1-Penten-3-ol" *Org. Mass Spectrom.* **19**, 573 (1984).
- 85ADA/SMI**
N.G. Adams and D. Smith, "A Study of the Nearly Thermoneutral Reactions of N^+ with N_2 , HD and D_2 ," *Chem. Phys. Lett.* **117**, 67 (1985).
- 85AJO/CAS**
D. Ajo, M. Casarin, R. Bertonecello, V. Busetti, H.C.J. Ottenheim and R. Plate, "Molecular and Electronic Structure of the Dehydroalanine Derivatives: The Cyclic Dipeptide of Dehydrophenylalanine," *Tetrahedron* **41**, 5543 (1985).
- 85ALA/ATT**
M. Alai, G.G. Attardo and R.T.B. Rye, "Formation Threshold Structures of Some $[C_5H_7O]^+$ Ions: Use of General Schemes for Estimation of Heats of Formation of Gas Phase Ions," *Can. J. Chem.* **63**, 833 (1985).
- 85ALB/GEH**
R.A. Albery and C.A. Gehrig, *J. Phys. Chem. Ref. Data* **14**, 803 (1985).
- 85ALB/HEL**
B. Albert, C. Heller, R. Iden, G. Martin, H.-D. Martin, B. Mayer and A. Oftring, "Design and Synthesis of α -Diketones. The Cyclobutane-1,2-dione Chromophore: Synthesis, Dienophilic Reactivity and Electronic Properties of Cyclobutenedione and Polycyclic Cyclobutenediones," *Israel J. Chem.* **25**, 74 (1985).
- 85AUD/MIL**
H. Audier, A. Milliet and G. Sozzi, "Isomerisation des Cation Radicaux Acides Pentanoique et Methyl-3-butanoique in Phase Gazeuse," *Bull. Soc. Chim. France* **5**, 833 (1985).
- 85AUS/LIA**
P. Ausloos and S.G. Lias, unpublished results.
- 85BAI/MIS**
V.N. Baidin, A.D. Misharev and V.V. Takhistov, "Effect of Alkyl Substituents on the Ionization Potentials of Halogenobenzenes," *Zh. Org. Khim.* **21**, 817 (1985).
- 85BAI/MIS2**
V.N. Baidin, A.D. Misharev and V.V. Takhistov, "Investigation of Substituted tert-Butylbenzenes by Photoelectron Spectroscopy," *Zh. Org. Khim.* **21**, 1237 (1985).
- 85BAJ/HUM**
M. Bajic, K. Humski, L. Klasinc and B. Ruscic, "Substitution Effects on Electronic Structure of Thiophene," *Z. Naturforsch.* **40b**, 1214 (1985).
- 85BAK/KIN**
J. Baker, E.E. Kingston, W.J. Bouma, A.G. Brenton and L. Radom, "Is the Methylenemethonium Radical Cation ($CH_2CH_4^+$) a Stable Species?" *J. Chem. Soc. Chem. Commun.* **1625** (1985).
- 85BAL/GIG**
G. Balducci, G. Gigli and M. Guido, "Mass Spectrometric Study of the Thermochemistry of Gaseous $EuTiO_3$ and TiO_2 ," *J. Chem. Phys.* **83**, 1909 (1985).
- 85BAL/GIG2**
G. Balducci, G. Gigli and M. Guido, "Identification and Stability Determinations for the Gaseous Titanium Oxide Molecules Ti_2O_3 and Ti_2O_4 ," *J. Chem. Phys.* **83**, 1913 (1985).
- 85BAL/HAS**
T. Bally, D. Hasselmann and K. Loosen, "41. The Molecular Ion of 5-Methylene-1,3-cyclohexadiene: Electronic Absorption Spectrum and Revised Enthalpy of Formation," *Helv. Chim. Acta* **68**, 345 (1985).
- 85BAN/MTE**
J.A. Bandy, V.S.B. Mtetwa, K. Prout, J.C. Green, C.E. Davies, M.L.H. Green, N.J. Hazel, A. Izquierdo and J.J. Martin-Polo, "Synthesis, Structure, and Bonding of Fulvene Complexes of Titanium, Molybdenum, and Tungsten," *J. Chem. Soc. Dalton Trans.* **2037** (1985).
- 85BIN/GRO**
M. Binnewies, J. Grosse and D. Le Van, "Reaktive $E = C(p-p)\pi$ -Systeme II: Massenspektrometrische Untersuchung von $F_3CP = CF_2$, $F_3CAS = CF_2$, $S = CF_2$ und $Se = CF_2$," *Phosphorus und Sulfur* **21**, 349 (1985).
- 85BIS/GLE**
P. Bischof, R. Gleiter, R. Haider and C.W. Rees, "The Photoelectron Spectrum of 7b-Methyl-7bH-cyclopent[cd]indene," *J. Chem. Soc. Perkin Trans. II* **1001** (1985).
- 85BOL/HOU**
J.C. Bollinger, R. Houriet, C.W. Kern, D. Perret, J. Weber and Y. Yvernault, "Experimental and Theoretical Studies of the Gas Phase Protonation of Aliphatic Phosphine Oxides and Phosphoramides," *J. Am. Chem. Soc.* **107**, 5352 (1985).
- 85BRO/CHE**
M. Broyer, J. Chevalere, G. Delacretaz, P. Fayet and L. Woste, "One- and Two-Photon Ionization of Alkaline Clusters," *Surface Sci.* **156**, 342 (1985).
- 85BRO/COX**
M. Brookhart, K. Cox, F.G.N. Cloke, J.C. Green, M.L.H. Green, P.M. Hare, J. Bashkin, A.E. Derome and P.D. Grebenik, "Hexakis(trimethylphosphine)molybdenum Chemistry: Dinitrogen, Ethylene, Butadiene, η -Cyclopentadienyl, and Related Derivatives," *J. Chem. Soc. Dalton Trans.* **423** (1985).
- 85BUR**
P.C. Burgers, "Isotope Effects Associated With the Unimolecular Fragmentation of D-Labeled Formic Acid Radical Cations," *Org. Mass Spectrom.* **20**, 426 (1985).
- 85BUR/HOL**
P.C. Burgers, J.L. Holmes, J.K. Terlouw and B. van Baar, "Three New Isomers of $[C_2H_6O]^+$: The Radical Cations $[CH_3O(H)CH_2]^+$, $[CH_3CHOH_2]^+$ and a Low-energy Isomer of Unassigned Structure," *Org. Mass Spectrom.* **20**, 202 (1985).
- 85BUT/LEB**
J.J. Butler, G.E. Leroi, A.C. Parr and R. Stockbauer, "The Rate Versus Energy Dependence of CH_3 Loss from t-Butylbenzene Cation," to be submitted.
- 85CAL/KEB**
G. Caldwell and P. Kebarle, "The Hydrogen Bond Energies of the Bihalide Ions XHX^- and YHX^- ," *Can. J. Chem.* **63**, 1399 (1985).
- 85CAL/MCM**
G. Caldwell, T.B. McMahon, P. Kebarle, J.E. Bartmess and J.P. Kiplinger, "Methyl Substituent Effects in the Gas Phase Acidities of Halosubstituted Oxygen Acids. A Realignment with Substituent Effects in Solution," *J. Am. Chem. Soc.* **107**, 80 (1985).
- 85CAN/HAM**
P.H. Cannington and N.S. Ham, "He(II) Photoelectron Spectra of Esters," *J. Electron Spectrosc. Rel. Phenom.* **36**, 203 (1985).

- 85CAU/FUR**
C. Cauletti, C. Furlani, A. Palma, M.N. Piancastelli, K.D. Schleinitz and D. Gloyna, "Electronic Structure and Free-Energy Relationships for Some 4'-Substituted 4-Dimethylamino trans-Stilbenes by U. V. Photoelectron Spectroscopy," *J. Prakt. Chem.* **327**, 829 (1985).
- 85CLE/MUN**
D. Clemens and B. Munson, "Selective Reagents in Chemical Ionization Mass Spectrometry: Trimethylsilyl Adduct Ions," *Anal. Chem.* **57**, 2022 (1985).
- 85COE/SNO**
J.V. Coe, J.T. Snodgrass, C.B. Friedhoff, K.M. McHugh and K.H. Bowen, "Negative Ion Photoelectron Spectroscopy of the negative ion $H^-(NH_3)$," *J. Chem. Phys.* **83**, 3169 (1985).
- 85DAS/GRO**
C. Dass and M.L. Gross, "The Question of Cyclic Versus Acyclic Ions: The Structure of $[C_6H_{10}]^+$ Gas Phase Ions," *Org. Mass Spectrom.* **20**, 34 (1985).
- 85DAS/NIS**
P.R. Das, T. Nishimura and G.G. Meisels, "Fragmentation of Energy-Selected Hexacarbonylchromium Ion," *J. Phys. Chem.* **89**, 2808 (1985).
- 85DAV/ALL**
H.E. Davis, N.L. Allinger and D.W. Rogers, *J. Org. Chem.* **50**, 3601 (1985).
- 85DAV/GAR**
C.E. Davies, I.M. Gardiner, J.C. Green, M.L.H. Green, N.J. Hazel, P.D. Grebenik, V.S.B. Mtetwa and K. Prout, "Mono- η -cycloheptatrienyltitanium Chemistry: Synthesis, Molecular and Electronic Structures, and Reactivity of the Complexes $[Ti(\eta-C_7H_7)L_2X]$ (L = Tertiary Phosphine, O- or N-Donor Ligand. X = Cl or Alkyl)," *J. Chem. Soc. Dalton Trans.* **669** (1985).
- 85DEL/PIG**
E.W. Della, P.E. Pigou, M.K. Livett and J.B. Peel, "The Photoelectron Spectrum of 1-Bromotricyclene," *Aust. J. Chem.* **38**, 69 (1985).
- 85DEP/BIE**
C.H. DePuy, V.M. Bierbaum, R. Damrauer and J.A. Soderquist, "Gas-Phase Reactions of the Acetyl Anion," *J. Am. Chem. Soc.* **107**, 3385 (1985).
- 85DEP/GRA**
C.H. DePuy, J.J. Grabowski, V.M. Bierbaum, S. Ingemann and N.M.M. Nibbering, "Gas-Phase Reactions of Anions with Methyl Formate and N,N-Dimethylformamide," *J. Am. Chem. Soc.* **107**, 1093 (1985).
- 85DEW/GRA**
M.J.S. Dewar, G.L. Grady, K.M. Merz, Jr. and J.J.P. Stewart, *Organometallics* **4**, 1964 (1985).
- 85DEW/HOL**
M.J.S. Dewar, M.K. Holloway, G.L. Grady and J.J.P. Stewart, *Organometallics* **4**, 1973 (1985).
- 85DEW/MER**
M.J.S. Dewar and K.M. Merz, Jr., *J. Am. Chem. Soc.* **107**, 6175 (1985).
- 85DEW/TIE**
M.J.S. Dewar and T.-P. Tien, "Photoelectron Spectrum of Benzyne," *J. Chem. Soc. Chem. Commun.* **1243** (1985).
- 85DIN/CAS**
A. Ding, R. Cassidy, L. Cordis and F. Lampe, "The Photoionization Spectra of Effusing and Supersonic Molecular Beams of Monosilane," *J. Chem. Phys.* **83**, 3426 (1985).
- 85DIS/GIU**
G. Distefano, A.G. Giumanini, A. Modelli and G. Poggi, "Reinvestigation of the Formaldehyde-Aniline Condensation. Part 4. Ultraviolet Photoelectron and Electron transmission Spectra of N-Methyleaniline and its Symmetric Dimethyl Ring-Substituted Homologues and Semiempirical Theoretical Evaluation," *J. Chem. Soc. Perkin Trans. II*, 1623 (1985).
- 85DOM/LAK**
I.N. Dornin, A.M. Lakshin, A.D. Misharev, V.M. Orlov and V.V. Takhistov, "Thermochemical Investigation of Some Hydrocarbon Ions Generated By Photoionization," *Zh. Org. Khim.* **21**, 1262 (1985).
- 85DUN**
R. Dunbar, "Collisional Quenching of Iodobenzene Ion One-Photon Photofragmentation," *Chem. Phys. Lett.* **155**, 349 (1985).
- 85DYK/ELL**
J. Dyke, A. Ellis, N. Jonathan and A. Morris, "Vacuum Ultraviolet Photoelectron Spectroscopy of Transient Species. Part 18. The Cyclopropyl, Isopropyl, an n-Propyl Radicals," *J. Chem. Soc. Faraday II*, **81**, 1573 (1985).
- 85DYK/GRA**
J.M. Dyke, B.W.J. Gravenor, M.P. Hastings, G.D. Josland and A. Morris, "Gas Phase High Temperature Photoelectron Spectroscopy: An Investigation of the Transition Metals Scandium and Vanadium," *J. Electron Spectrosc. Rel. Phenom.* **35**, 65 (1985).
- 85ELB/ELL**
S. Elbel, A. Ellis, E. Niecke, H. Egsgaard and L. Carlsen, "A Photoelectron Spectroscopic Study of Di-t-butylphosphazene," *J. Chem. Soc. Dalton Trans.* **879** (1985).
- 85ELK/ARM**
J. Elkind and P. Armentrout, "Effect of Kinetic and Electronic Energy on the Reaction of V^+ with H_2 , HD and D_2 ," *J. Phys. Chem.* **89**, 5626 (1985).
- 85ELS/VER**
C.J. Elsevier, P. Vermeer, A. Gedanken and W. Runge, "Excited States of the Allene Chromophore: Photoelectron, Circular Dichroism, and Absorption Spectroscopy of Alkyl- and Halogenoallenes," *J. Am. Chem. Soc.* **107**, 2537 (1985).
- 85FAO/AKA**
M. Faour and T.S. Akasheh, *J. Chem. Soc. Perkin Trans. II* **311** (1985).
- 85FAR/SRI**
M. Farber and R.D. Srivastava, "Electron Impact and Thermodynamic Studies of Potassium Metaborate," *J. Chem. Soc. Faraday Trans. I* **81**, 913 (1985).
- 85FUC**
R. Fuchs, Unpublished data, personal communication.
- 85FUK/MCI**
E.K. Fukuda and R.T. McIver, Jr., "Relative Electron Affinities of Substituted Benzophenones, Nitrobenzenes, and Quinones. [Anchored to $EA(SO_2)$ from 74CEL/BEN]," *J. Am. Chem. Soc.* **107**, 2291 (1985).
- 85GAD/GUB**
S.N. Gadzhiev, A.I. Gubareva and V.I. Khun, *Izv. Vyssh. Uchebn. Zaved. Khim. Khim. Technol.* **28**, 48 (1984) CA **103**:130117z (1985).
- 85GAL/GER**
J.-F. Gal, S. Geribaldi, G. Pfister-Guillouzo and D.G. Morris, "Basicity of the Carbonyl Group. Part 12. Correlations between Ionization Potentials and Lewis Basicities in Aromatic Carbonyl Compounds," *J. Chem. Soc. Perkin Trans. II* **103** (1985).
- 85GAL/TAM**
G.L. Gal'chenko, N. Tamm, E.P. Brykina, D.B. Bekker, A.B. Petrunin and A.F. Zhigach, *Russ. J. Phys. Chem.* **59**, 1610 (1985).
- 85GAR/GON**
J.-L. Garcia, D. Gonbeau, G. Pfister-Guillouzo, M. Roch and J. Weber, "Germathione - silathione spectres photoelectroniques et structures electroniques par la methode $MS X\alpha$," *Can. J. Chem.* **63**, 1518 (1985).
- 85GIB/GRE**
S. Gibson, J. Greene and J. Berkowitz, "Photoionization of the Amidogen Radical," *J. Chem. Phys.* **83**, 4319 (1985).

85GLE/DOB

R. Gleiter and W. Dobler, "Die Elektronenstruktur von 1,4-disubstituierten Butantetronen," *Chem. Ber.* **118**, 1917 (1985).

85GLE/JAH

R. Gleiter, G. Jahne, M. Oda and M. Iyoda, "Effect of Through Bond Interaction via Strained Σ Bonds in Cyclohexane-1,4-dione Derivatives," *J. Org. Chem.* **50**, 678 (1985).

85GLE/KRE

R. Gleiter, G. Krennrich, D. Cremer, K. Yamamoto and I. Murata, "Electronic Structure and Thermal Stability of Thiopins. Photoelectron Spectroscopic Investigations," *J. Am. Chem. Soc.* **107**, 6874 (1985).

85GLE/SCH

R. Gleiter, P. Schang, M. Bloch, E. Heilbronner, J.-C. Bunzli, D.C. Frost and L. Weiler, "The He(I α) PE Spectra and Electronic Absorption Spectra of Hexafluorocyclobutanone and of Tetrafluoro-1,2-cyclobutanedione," *Chem. Ber.* **118**, 2127 (1985).

85GLE/ZIM

R. Gleiter, H. Zimmermann, W.-D. Fessner and H. Prinzbach, "Dominance of Through-Bond Interaction in a syn-Tricyclo[6.4.0.0^{2,7}]dodecatetraene Moiety," *Chem. Ber.* **118**, 3856 (1985).

85GRA/BER

G. Granozzi, R. Bertonecello, E. Tondello and D. Ajo, "He(I)/He(II) Sn 5p Photoionization Cross Sections: Definitive Evidence from the Spectra of Sn₂(CH₃)₆," *J. Electron Spectrosc. Rel. Phenom.* **36**, 207 (1985).

85GRA/ROS

M. Grade and W. Rosinger, "Correlation of Electronic Structures and Stabilities of Gaseous FeI₂, Fe₂I₂ and Fe₂I₄ Molecules, Solid [FeI₂], and Iodine Adsorbed on [Fe]," *Surface Sci.* **156**, 920 (1985).

85GRE/PAZ

J.C. Green, M. Paz-Sandoval and P. Powell, "He I and He II Photoelectron Spectra of Open-chain Pentadienyl Complexes of Manganese and Rhenium," *J. Chem. Soc. Dalton Trans.* **2677** (1985).

85GRI/CAL

E. Grimsrud, G. Caldwell and P. Kebarle, "Electron Affinities from Electron Transfer Equilibria: A⁻ + B = A + B⁻," *J. Am. Chem. Soc.* **107**, 4627 (1985).

85GRI/CHO

E.P. Grimsrud, S. Chowdhury and P. Kebarle, "Electron Affinity of SF₆ and Perfluoromethylcyclohexane. The Unusual Kinetics of Electron Transfer Reactions A⁻ + B⁻ + A, where A = SF₆ or Perfluorinated Cycloalkanes," *J. Chem. Phys.* **83**, 1059 (1985).

85GRI/CHO2

E.P. Grimsrud, S. Chowdhury and P. Kebarle, "Thermal Energy Electron Detachment Rate Constants. The Electron Detachment from Azulene⁻ and the Electron Affinity of Azulene," *J. Chem. Phys.* **83**, 3983 (1985).

85GRU/SPI

H.-F. Grutzmacher and R. Spilker, "Loss of CO from 4,6-Dimethyl-2-pyrone and 2,6-Dimethyl-4-pyrone Radical Cations," *Org. Mass Spectrom.* **20**, 258 (1985).

85GUI/PFI

C. Guimon, G. Pfister-Guillouzo, B. Chaudret and R. Poilblanc, "Application of Photoelectron Spectroscopy to Molecular Properties. Part 19. Electronic Structure of Tris(α -di-imino) Complexes of Ruthenium(0)," *J. Chem. Soc. Dalton Trans.* **43** (1985).

85GUI/PFI2

G. Guimon, G. Pfister-Guillouzo, G. Rima, M. El Amine and J. Barrau, "Generation, Detection and Electronic Structure of Dimethyl Germanone by Photoelectron Spectroscopy and Quantum Calculations," *Spectrosc. Lett.* **18**, 7 (1985).

85GUI/PFI3

C. Guimon, G. Pfister-Guillouzo, J. Dubac, A. Laporterie, G. Manuel and H. Ioughmane, "Electronic Structure of Group 14 (η^4 -Metallole)tricarbonyliron Complexes," *Organometall.* **4**, 636 (1985).

85HAG/IVA

J. Hager, M. Ivanco, M.A. Smith and S.C. Wallace, "Solvation Effects in Jet-Cooled Van der Waals Clusters: Two-Color Threshold Photoionization Spectroscopy of Indole, Indole-Argon, Indole-Methane, Indole-Water and Indole-Methanol," *Chem. Phys. Lett.* **113**, 503 (1985).

85HAG/SMI

J. Hager, M. Smith and S. Wallace, "Autoionizing Rydberg Structure Observed in the Vibrationally Selective, Two-Color Threshold Photoionization Spectrum of Jet-Cooled Aniline," *J. Chem. Phys.* **83**, 4820 (1985).

85HAR

P.W. Harland, unpublished results, cited in B.S. Knight, C.G. Freeman, M.J. McEwan, N.G. Adams and D. Smith, *Int. J. Mass Spectrom. Ion Proc.* **67**, 317 (1985).

85HAR/MCI

P.W. Harland and B.J. McIntosh, "Enthalpies of Formation for the Isomeric Ions H_xCCN⁺ and H_xCNC⁺ (x = 0-3) by, "Monochromatic" Electron Impact on C₂N₂, CH₃CN and CH₃NC," *Int. J. Mass Spectrom. Ion Proc.* **67**, 29 (1985).

85HEI/BAR

T. Heinis, R. Bar, K. Borlin and M. Jungen, "Photoionization Mass Spectrometry of 1,1-Difluoroethane," *Chem. Phys.* **94**, 235 (1985).

85HEN/HIE

M. Henchman, P.M. Hierl and J.F. Paulson, "Nucleophilic Displacement vs. Proton Transfer: The System OH⁻(H₂O)_{0,1,2} + CH₃Cl in the Relative Energy Range 0.03-5 eV," *J. Am. Chem. Soc.* **107**, 2812 (1985).

85HEN/ILL

M. Heni and E. Illenberger, "The Stability of the Bifluoride Ion (HF₂⁻) in the Gas Phase," *J. Chem. Phys.* **83**, 6056 (1985).

85HEN/VIG

M. Henchman, A.A. Viggiano, J.F. Paulson, A. Freedman and J. Wormhoudt, "Thermodynamic and Kinetic Properties of the Metaphosphate Anion, PO₃⁻, in the Gas Phase," *J. Am. Chem. Soc.* **107**, 1453 (1985).

85HET/FRE

R. Hettich and B. Freiser, "Heteronuclear Transition-Metal Cluster Ions in the Gas Phase. Photodissociation and Reactivity of VFe⁺," *J. Am. Chem. Soc.* **107**, 6222 (1985).

85HIL/BEN

K. Hilpert, L. Bencivenni and B. Saha, "Thermochemistry of the Molecule (ZnI₂)₂(g) and the Vaporization of ZnI₂(s)," *J. Chem. Phys.* **83**, 5227 (1985).

85HIR/STR

C. Hirayama, R.D. Straw and H.M. Hobgood, "Equilibria Over GaAs in the Knudsen Cell Range," *J. of the Less-Common Metals* **109**, 331 (1985).

85HOD/BEA

R.V. Hodges, J.L. Beauchamp, A.J. Ashe, III and W.-T. Chan, "Proton Affinities of Pyridine, Phosphabenzene and Arsabenzene," *Organometall.* **4**, 457 (1985).

85HOL/MOM

J. Holmes, A. Mommers, C. DeKoster, W. Heerma and J. Terlouw, "Four Isomeric [C, H₃, O₂] Ions," *Chem. Phys. Lett.* **115**, 437 (1985).

85HON/HBI

E. Honegger, E. Heilbronner, T. Urbanek and H.-D. Martin, "Inverted Hyperconjugation in Symmetrical 1,4-Dihalocubanes," *Helv. Chim. Acta* **68**, 23 (1985).

85HON/HEI2

E. Honegger, E. Heilbronner, N. Hess and H.-D. Martin, "Photoelectron Spectra of Symmetrical 1,4-Dihalonorbornanes," *Chem. Ber.* **118**, 2927 (1985).

85HON/HUB

E. Honegger, H. Huber, E. Heilbronner, W.P. Dailey and K.B. Wiberg, "Photoelectron Spectrum of [1.1.1]Propellane: Evidence for a Nonbonding MO?," *J. Am. Chem. Soc.* **107**, 7172 (1985).

85HON/SEG

J. Honovich, J. Segall and R. Dunbar, "Fragmentation Thermochemistry of Gas-Phase Ions by Threshold Photodissociation and Charge-Exchange Ionization. Methylanthalene and Methylstyrene Ions," *J. Phys. Chem.* **89**, 3617 (1985).

85HON/YAN

E. Honegger, Z.-Z. Yang and E. Heilbronner, "Lone-Pair Ionization Energies of Diazabicycloalkanes," *J. Electron Spectrosc. Rel. Phenom.* **36**, 297 (1985).

85HOT/LIN

H. Hotop and W.C. Lineberger, "Binding Energies in Atomic Negative Ions. II," *J. Phys. Chem. Ref. Data* **14**, 731 (1985).

85HOU/ROL

R. Houriet, E. Rolli, G. Bouchoux and Y. Hoppilliard, "215. Gas-Phase Basicities of Furan Compounds. The Role of Alkyl Substitution on Proton Affinity and on the Site of Protonation," *Helv. Chim. Acta* **68**, 2037 (1985).

85HOU/SCO

K.N. Houk, L.T. Scott, N.G. Rondan, D.C. Spellmeyer, G. Reinhardt, J.L. Hyun, G.J. DeCicco, R. Weiss, M.H.M. Chen, L.S. Bass, J. Clardy, F.S. Jorgensen, T.A. Eaton, V. Sarkozi, C.M. Petit, L. Ng and K.D. Jordan, "Pericyclics: 'Exploded Cycloalkanes' with Unusual Orbital Interactions and Conformational Properties. MM2 and STO-3G Calculations, X-ray Crystal Structures, Photoelectron Spectra, and Electron Transmission Spectra," *J. Am. Chem. Soc.* **107**, 6556 (1985).

85ILL/COM

E. Illenberger, P. Comita, J.I. Brauman, H.-P. Fenzlaff, M. Heni, N. Heinrich, W. Koch and G. Frenking, "Experimental and Theoretical Investigation of the Azide Anion (N_3^-) in the Gas Phase," *Ber. Bunsen.* **89**, 1026 (1985).

85ING/NIB

S. Ingemann and N.M.M. Nibbering, "Gas Phase Chemistry of Alpha-Thio Carbanions," *Can. J. Chem.* **62**, 2273 (1985).

85ING/NIB2

S. Ingemann and N.M.M. Nibbering, "Gas-Phase Acidity of CH_3X [$X = P(CH_3)_2, SCH_3, F, Cl, Br, I$] Compounds," *J. Chem. Soc. Perkin Trans. II* **837** (1985).

85ING/NIB3

S. Ingemann and N.M.M. Nibbering, "Gas Phase Chemistry of Dipole Stabilized Carbanions Derived From N,N-Dimethylthioformamide and N,N-Dimethylnitrosamine," *Acta Chem. Scand. B* **39**, 697 (1985).

85JANAF

M.W. Chase, Jr., C.A. Davies, J.R. Downey, Jr., D.J. Frurip, R.A. McDonald and A.N. Syverud, "JANAF Thermochemical Tables (Third Edition) Suppl. 1," *J. Phys. Chem. Ref. Data* **14**, (1985).

85JEO/RAK

S. Jeon, A. Raksit and G. Gellene, "Formation of Hypervalent Ammoniated Radicals by Neutralized Ion Beam Techniques," *J. Am. Chem. Soc.* **107**, 4129 (1985).

85JOH/WIL

C.L. Johlman and C.L. Wilkins, "Gas-Phase Reactions of Nucleophiles with Methyl Formate," *J. Am. Chem. Soc.* **107**, 327 (1985).

85JOR/GAJ

F.S. Jorgensen, M. Gajhede and B. Frei, "227. Cyclic Acetals. Structural Analysis of 1,3-Dioxepine and Related Compounds," *Helv. Chem. Acta* **68**, 2148 (1985).

85KAI

W. Kaim, "Electronic Structure of 1,4-Dihydro-1,2,4,5-tetrazines and of Related 1,4-Dihydroaromatic Compounds," *J. Chem. Soc. Perkin Trans. II* **1633** (1985).

85KAP/LEL

O. Kaposi, L. Lelik, G.A. Semenov and E.N. Nikolajev, "Gazfasisu indium-molibdenat kepzodoshojenek tomegspektrometrias meghatarozasa," *Magy. Kem. Foly.* **91**, 31 (1985).

85KAP/RAD

M.M. Kappes, P. Radi, M. Schar and E. Schumacher, "Photoionization Measurements on Dialkali Monohalides Generated in Supersonic Nozzle Beams," *Chem. Phys. Lett.* **113**, 243 (1985).

85KAP/RAD2

M.M. Kappes, P. Radi, M. Schar and E. Schumacher, "Probes for Electronic and Geometrical Shell Structure Effects in Alkali-Metal Clusters. Photoionization Measurements on K_xLi , K_xMg and K_xZn ($x < 25$)," *Chem. Phys. Lett.* **119**, 11 (1985).

85KAP/SCH

M.M. Kappes and E. Schumacher, "Generation, Spectroscopic and Chemical Characterization of Metal Clusters (M_x , $x < 65$), A Progress Report," *Surface Sci.* **156**, 1 (1985).

85KAR

Z. Karpas, "The Proton Affinity of H_2Se , $SeCO$ and H_2CSe and Reactions of Positive Ions with H_2Se ," *Chem. Phys. Lett.* **120**, 53 (1985).

85KAR/STE

Z. Karpas, W.J. Stevens, T.J. Buckley and R. Metz, "The Proton Affinity and Gas Phase Ion Chemistry of Methyl Isocyanate, Methyl Isothiocyanate, and Methyl Thiocyanate," *J. Phys. Chem.* **89**, 5274 (1985).

85KAS/DEP

S.R. Kass and C.H. DePuy, "Gas Phase Ion Chemistry of Azides. The Generation of $CH=N^+$ and $CH_2=NCH_2^+$," *J. Org. Chem.* **50**, 2874 (1985).

85KIE/WEI

J.H. Kiefer, H.C. Wei, R.D. Kern and C.H. Wu, *Int. J. Chem. Kinet.* **17**, 225 (1985).

85KIS/MOR

W. Kischlat and H. Morgner, "Comparative Study of He(2^3S)-Penning Ionization and He(I) Photoionization of CF_4 , CCl_4 , and the Chlorofluoromethanes by Electron-Ion Coincidence," *J. Electron Spectrosc. Rel. Phenom.* **35**, 273 (1985).

85KLE/WAR

P.D. Kleinschmidt, J.W. Ward, G.M. Matlack and R.G. Haire, *High Temp. Sci.* **19**, 267 (1985).

85KNI/TRE

J.S. Knight, C.G. Freeman and M.J. McEwan, "Selected-Ion Flow Tube Studies of HC_3N ," *Int. J. Mass Spectrom. Ion Phys.* **67**, 317 (1985).

85KOL/MEI

H. Kolshornand and H. Meier, *Chem. Ber.* **118**, 176 (1985).

85KRO/MCN

H.W. Kroto and D. McNaughton, "Photoelectron Spectra of the Aminodifluoroboranes NH_2BF_2 , $NHMeBF_2$, and NMe_2BF_2 ," *J. Chem. Soc. Dalton Trans.* **1767** (1985).

85LAD/HAR

K.R. Laderoute and A.G. Harrison, "Cyclopropane Intermediates in the Rearrangement and Fragmentation of Olefinic Molecular Ions," *Org. Mass. Spectrom.* **20**, 624 (1985).

85LAH/HAY

J. Lahnstein, R.N. Hayes and J.H. Bowie, "Gas Phase Ion Chemistry of Ambident Nucleophiles. Reactions of Alkoxide and Thiomethoxide Negative Ions with Hydrogen Free Molecules," *Nouv. J. Chim.* **9**, 205 (1985).

85LAN/SAL

K.R. Lane, L. Sallans and R.R. Squires, "Anion Affinities of Transition Metal Carbonyls. A Thermochemical Correlation for Iron Tetracarbonyl Acyl Negative Ions," *J. Am. Chem. Soc.* **107**, 5369 (1985).

85LAN/SQU

K.R. Lane and R.R. Squires, "Formation of $\text{HCr}(\text{CO})_3^-$ from the Remarkable Reaction of Hydride Ion with Benzenechromium Tricarbonyl," *J. Am. Chem. Soc.* **107**, 6403 (1985).

85LAR/MCM

J.W. Larson and T.B. McMahon, "Fluoride and Chloride Affinities of the Main Group Oxides, Fluorides, Oxofluorides, and Alkyls. Quantitative Scales of Lewis Acidities from ICR Halide Exchange Equilibria," *J. Am. Chem. Soc.* **107**, 766 (1985).

85LAU/BRI

K.H. Lau, R.D. Brittain and D.L. Hildenbrand, *J. Phys. Chem.* **89**, 4369 (1985).

85LAU/WES

W.M. Lau, N.P.C. Westwood and M.H. Palmer, "A Photoelectron/Photoionisation and Ab Initio Study of the S_3N_3 Radical Produced by Vaporisation of $(\text{SN})_x$," *J. Chem. Soc. Chem. Commun.* **752** (1985).

85LEE/LIV

K.-J. Lee, P.D. Livant, M.L. McKee and S.D. Worley, "Photoelectron, Infrared, and Theoretical Study of 1-Aza-5-boratricyclo[3.3.3.0^{1,5}]undecane and Related Compounds," *J. Am. Chem. Soc.* **107**, 5901 (1985).

85LEO/MUR

D.G. Leopold, K.K. Murray, A.E.S. Miller and W.C. Lineberger, "Methylene: A Study of the X^3B_1 and the $1A_1$ States by Photoelectron Spectroscopy of CH_2^- and CD_2^- ," *J. Chem. Phys.* **83**, 4849 (1985).

85LIA/AUS

S.G. Lias and P. Ausloos, "Structures of C_6H_7^+ Ions Formed in Unimolecular and Bimolecular Reactions," *J. Chem. Phys.* **82**, 3613 (1985).

85LIA/JAC

S.G. Lias, J.-A.A. Jackson, H. Argentar and J.F. Liebman, "Substituted N,N-Dialkylanilines: Relative Ionization Energies and Proton Affinities through Determinations of Ion-Molecule Reaction Equilibrium Constants," *J. Org. Chem.* **50**, 333 (1985).

85LIA/KAR

S.G. Lias, Z. Karpas and J.F. Liebman, "Halomethylenes: Effects of Halogen Substitution on Absolute Heats of Formation," *J. Am. Chem. Soc.* **107**, 6089 (1985).

85LIN/BRO

S.H. Linn, J.M. Brom, Jr., W.-B. Tzeng and C.Y. Ng, "Photoionization Study of HgAr^+ ," *J. Chem. Phys.* **82**, 648 (1985).

85LOS/WIL

O. Losking and H. Willner, "Thermochemische Daten und Photoionisation-Massenspektren von SSF_2 , FSSF , SF_3SF und SF_3SSF ," *Z. Anorg. Allg. Chem.* **530**, 169 (1985).

85MAL/ARA

Y. Malinovich, R. Arakawa, G. Haase and C. Lifshitz, "Time-Dependent Mass Spectra and Breakdown Graphs. VI. Slow Unimolecular Dissociation of Bromobenzene Ions at Near Threshold Energies," *J. Phys. Chem.* **89**, 2253 (1985).

85MAR/COM

J. Marks, P.B. Comita and J.I. Brauman, "Threshold Resonances in Electron Photodetachment Spectra. Structural Evidence for Dipole-Supported States," *J. Am. Chem. Soc.* **107**, 3718 (1985).

85MAR/MAN

K.N. Marsh and M. Mansson, *J. Chem. Thermodyn.* **17**, 995 (1985).

85MAR/MAY

H.-D. Martin, B. Mayer, R.W. Hoffmann, A. Riemann and P. Rademacher, "PE- und ^{13}C -NMR-spektroskopische Untersuchungen zur Homokonjugation in 7-Alkylidennorbormadienen," *Chem. Ber.* **118**, 2514 (1985).

85MAR/MOD

F. Marcuzzi, G. Modena and C. Paradisi, "Gas Phase Basicity of Ring-Substituted Phenylacetylenes," *J. Org. Chem.* **50**, 4973 (1985).

85MCM/KEB

T.B. McMahon and P. Kebarle, "Bridging the Gap. A Continuous Scale of Gas Phase Basicities from Methane to Water from Pulsed Electron Beam High Pressure Mass Spectrometric Equilibria Measurements," *J. Am. Chem. Soc.* **107**, 2612 (1985).

85MCM/KEB2

T.B. McMahon and P. Kebarle, "The Formyl and Isoformyl Cations. A Pulsed Electron Beam High Pressure Mass Spectrometric Study of the Energetics of HCO^+ and HOC^+ ," *J. Chem. Phys.* **83**, 3919 (1985).

85MCM/KEB3

T.B. McMahon and P. Kebarle, "Proton Affinities of the Hydrogen, Methyl and Trifluoromethyl Halides from Pulsed Electron Beam High Pressure Mass Spectrometric Equilibria Measurements," *Can. J. Chem.* **63**, 3160 (1985).

85MCN/SUF

D. McNaughton and R.J. Suffolk, "The Production and Photoelectron Spectrum of Propa-1,2-dien-3-one, $\text{C}_3\text{H}_2\text{O}$," *J. Chem. Res.* **32** (1985).

85MEA/HEF

R.D. Mead, U. Hefter, P.A. Schulz and W.C. Lineberger, "Ultrahigh Resolution Spectroscopy of C_2^- : The $A^2\Pi_u$ State Characterized by Deperturbation Methods," *J. Chem. Phys.* **82**, 1723 (1985).

85MEE/SEK

J.T. Meek, E. Sekreta, W. Wilson, K.S. Viswanathan and J.P. Reilly, "The Laser Photoelectron Spectrum of Gas Phase Aniline," *J. Chem. Phys.* **82**, 1741 (1985).

85MEI/KON

H. Meier, P. Konig, T. Molz, R. Gleiter and W. Schafer, "Untersuchungen zur Konformation und elektronischen Struktur von Cyclooctadienen," *Chem. Ber.* **118**, 210 (1985).

85MIC/GIU

H.A. Michelsen, R.P. Giugliano and J.J. BelBruno, "Photochemistry and Photophysics of Small Heterocyclic Molecules. 1. Multiphoton Ionization and Dissociation of N-Isopropylidimethyloxaziridine," *J. Phys. Chem.* **89**, 3034 (1985).

85MOC/WOR

K. Mochida, S. Worley and J. Kochi, "UV Photoelectron Spectra of Peralkylated Catenates of Group 4B Elements (Silicon, Germanium, and Tin)," *Bull. Chem. Soc. Jpn.* **58**, 3389 (1985).

85MOY/DOD

C.R. Moylan, J.A. Dodd and J.I. Brauman, "Electron Photodetachment Spectroscopy of Solvated Anions. A Probe of Structure and Energetics," *Chem. Phys. Lett.* **118**, 38 (1985).

85NAC/PRO

E. Nachbaur and P. Prosegger, "HeI-Photoelektronen-Spektrum von Cl-NSO," *Mon. Chem.* **116**, 1385 (1985).

85NEU

A. Neubert, "Investigation of Gaseous LiBO by Knudsen Effusion Mass Spectrometry," *J. Chem. Phys.* **82**, 939 (1985).

85NEU/LYK

D.M. Neumark, K.R. Lykke, T. Andersen and W.C. Lineberger, "Laser Photodetachment Measurement of the Electron Affinity of Atomic Oxygen," *Phys. Rev. A* **32**, 1890 (1985).

85NOV/POT

I. Novak, A.W. Potts, F. Quinn, G.V. Marr, B. Dobson, I.H. Hillier and J.B. West, "Photoelectron Asymmetry Measurements for CHF_3 and CF_4 in the Photon Energy Range 19 to 80 eV," *J. Phys. B: At. Mol. Phys.* **18**, 1581 (1985).

85OAK/HAR

J.M. Oakes, L.B. Harding and G.B. Ellison, "The Photoelectron Spectroscopy of HO_2^- ," *J. Chem. Phys.* **83**, 5400 (1985).

- 85OHN/IMA**
K. Ohno, K. Imai and Y. Harada, "Variations in Reactivity of Lone-Pair Electrons due to Intramolecular Hydrogen Bonding as Observed by Penning Ionization Electron Spectroscopy," *J. Am. Chem. Soc.* **107**, 8078 (1985).
- 85OHN/ISH**
K. Ohno, T. Ishide, Y. Naitoh and Y. Izumi, "Study of Stereochemical Properties of Molecular Orbitals by Penning Ionization Electron Spectroscopy," *J. Am. Chem. Soc.* **107**, 8082 (1985).
- 85OIK/ABE**
A. Oikawa, H. Abe, N. Mikami and M. Ito, "Electronic Spectra and Ionization Potentials of Rotational Isomers of Several Disubstituted Benzenes," *Chem. Phys. Letters* **116**, 50 (1985).
- 85OKU/TTO**
J. Okubo, H. Ito, T. Hishi and T. Kobayashi, "Intramolecular Orbital Interactions in and Conformation of N,N'-Diphenylcarbodiimide Studied by Photoelectron Spectroscopy," *Tetrahedron Lett.* **26**, 643 (1985).
- 85ORL/BOG**
V.M. Orlov, A.M. Boganov, T.V. Siretskaya, V.V. Takhistov, "Thermochemical Characteristics of the Molecular and Fragmentation Ions of Substituted tert-Butylacetylenes," *Izv. Akad. Nauk SSSR Ser. Khim.* **12**, 2795 (1985).
- 85PAN/BAE**
M. Panczel and T. Baer, "A nitro-benzol bomlasanak vizsgalata fotoelektron-fotoion koincidenciaval, I. A bomlas dinamikaja," *Magy. Kem. Foly.* **91**, 136 (1985).
- 85PAN/BAE2**
M. Panczel and T. Baer, "A nitro-benzol bomlasanak vizsgalata fotoelektron-fotoion koincidenciaval, II. A bomlas termokimijaja," *Magy. Kem. Foly.* **91**, 153 (1985).
- 85PAP/KOL**
T.S. Papina and V.P. Kolesov, *Zh. Fiz. Khim.* **59**, 2169 (1985).
- 85PFI/GUI**
G. Pfister-Guillouze and C. Guimon, "Studies by P.E.S. of Highly Reactive Sulfur Species," *Phosph. and Sulfur* **23**, 197 (1985).
- 85PIM/NES**
A.A. Pimerzin, T.N. Nesterova and A.M. Rozhnov, *J. Chem. Thermodyn.* **17**, 641 (1985).
- 85PLA/SIM**
V.A. Platonov and Y.N. Simulin, *Russ. J. Phys. Chem.* **59**, 179 (1985).
- 85PLA/SIM2**
V.A. Platonov, Y. N. Simulin and M.M. Rozenberg, *Russ. J. Phys. Chem.* **59**, 814 (1985).
- 85PRA/DEH**
S. Pratt, P. Dehmer and J. Dehmer, "The Photoelectron Spectrum of ArXe Obtained Using Resonantly Enhanced Multiphoton Ionization," *J. Chem. Phys.* **82**, 5758 (1985).
- 85PRA/DEH2**
S. Pratt, P. Dehmer and J. Dehmer, "Electron Spectra of NeXe, ArXe, and KrXe Using Resonantly Enhanced Multiphoton Ionization," *J. Chem. Phys.* **83**, 5380 (1985).
- 85ROD/CHI**
A.A. Rodin, A.B. Chistyakov, Y.S. Sarkisov, Y.L. Sergeev, K.A. V'yunov and A.V. Golovin, "Electronic Structure and Geometric Structure of Three-Member Heterocycles. I. Photoelectron Spectra of Glycidyl and Thioglycidyl Ethers," *Zh. Fiz. Khim.* **59**, 764 (1985); English trans.: *Russ. J. Phys. Chem.* **59**, 444 (1985).
- 85ROS/LEO**
M.A. Rossman, N.J. Leonard, S. Urano and P.R. LeBreton, "Synthesis and Valence Orbital Structures of Azacycl[3.3.3]azines in a Systematic Series," *J. Am. Chem. Soc.* **107**, 3884 (1985).
- 85ROT/BOC**
B. Roth, H. Bock and H. Gotthardt, "Radikationen 66. Thioparabansaure-Derivate: Ionisation zum Radikalkation und Reduktion zum Radikalanion," *Phosphorus and Sulfur* **22**, 109 (1985).
- 85RUD/SID**
E.B. Rudny, L.N. Sidirov, L.A. Kuligina and G.A. Semenov, "Heterolytic Dissociation of Potassium Chromate in the Gas Phase and the Electron Affinity of Chromium Oxides," *Int. J. Mass Spectrom. Ion Phys.* **64**, 95 (1985).
- 85RUD/SID2**
E.B. Rudny, L.N. Sidirov and O.M. Voyk, "Heterolytic Dissociation of Potassium Sulfate in the Gas Phase and Heats of Formation for Trioxosulfate(1⁻), Tetraoxosulfate(1⁻), and Potassium Sulfate (KSO₄⁻) Ions," *Teplotfiz. Vys. Temp.* **23**, 291 (1985).
- 85SAL/LAN**
L. Sallans, K.R. Lane, R.R. Squires and B.S. Freiser, "Generation and Reactions of Atomic Metal Anions in the Gas Phase. Determination of the Heterolytic and Homolytic Bond Energies of VH, CrH, FeH, CoH, and MoH," *J. Am. Chem. Soc.* **107**, 4379 (1985).
- 85SCH/WEI**
T. Scheuring and K.G. Weil, "Intermetallic Species in the Vapour above Alkali Metal-Antimony Mixtures," *Surface Sci.* **156**, 457 (1985).
- 85SEE/MOL**
R. Seefeldt, W. Moller and M. Schmidt, "Zur Elektronenstossionisierung des Hexamethyldisiloxans (HMDS)," *Z. Phys. Chemie, Leipzig* **266**, 797 (1985).
- 85SHA/HOJ**
D.K.S. Sharma, S.M. De Hojer and P. Kebarle, "Stabilities of Halonium Ions from a Study of Gas Phase Equilibria $R^+ + XR' = (RXR')^+$," *J. Am. Chem. Soc.* **107**, 3557 (1985).
- 85SHA/SHA**
R.B. Sharma, D.K.S. Sharma, K. Hiraoka and P. Kebarle, "Kinetics and Equilibria of Chloride Transfer Reactions. Stabilities of Carbocations Based on Chloride and Hydride Transfer Equilibria Measurements," *J. Am. Chem. Soc.* **107**, 3747 (1985).
- 85SHU/BEN**
L.G.S. Shum and S.W. Benson, *Int. J. Chem. Kinet.* **17**, 749 (1985).
- 85SMI/ADA**
D. Smith and N.G. Adams, "The Proton Affinity of CS," *J. Phys. Chem.* **89**, 3964 (1985).
- 85SNO/COE**
J.T. Snodgrass, J.V. Coe, C.B. Freidhoff, K.M. McHugh and K.H. Bowen, "Negative Ion Photoelectron Spectroscopy of P₂," *Chem Phys. Lett.* **122**, 352 (1985).
- 85SPY/HUN**
S.M. Spyrou, S.R. Hunter and L.G. Christophorou, "A Study of the Isomeric Dependence of Low-Energy (< 10 eV) Electron Attachment: Perfluoroalkanes," *J. Chem. Phys.* **83**, 641 (1985).
- 85STE/FAH**
S.E. Stein and A. Fahr, *J. Phys. Chem.* **89**, 3714 (1985).
- 85STE/GAM**
W.V. Steele, B.E. Gammon, N.K. Smith, J.S. Chickos, A. Greenberg and J.F. Liebman, *J. Chem. Thermodyn.* **14**, 505 (1985).
- 85SVY/IOF**
V.A. Svyatkin, A.I. Ioffe and D.M. Nefedov, *Izv. Akad. Nauk, SSSR Ser. Khim.* **578** (1985).
- 85TAJ/TOB**
S. Tajima, S. Tobita and T. Tsuchiya, "A Kinetic Energy Release Study of CO Loss from C₆H₅CO⁺ Ions Generated from Benzoic Acid Alkyl Esters," *Mass Spectrosc.* **33**, 39 (1985).
- 85TRA**
J.C. Traeger, "Heat of Formation for the Propanoyl Cation by Photoionization Mass Spectrometry," *Org. Mass Spectrom.* **20**, 223 (1985).

- 85TRA2**
J.C. Traeger, "Heat of Formation for the Formyl Cation by Photoionization Mass Spectrometry," *Int. J. Mass Spectrom. and Ion Proc.* **66**, 271 (1985).
- 85TRE/RAD**
L. Treschanke and P. Rademacher, "Electronic Structure and Conformational Properties of the Amide Linkage," *J. Molec. Str. (Theochem)* **122**, 47 (1985).
- 85TSA**
W. Tsang, "The Stability of Alkyl Radicals," *J. Am. Chem. Soc.* **107**, 2872 (1985).
- 85TUR/PAN**
F. Turecek, J. Pancir, D. Stahl and T. Gaumann, "Stereo-electronic Effects on the Retro-Diels-Alder Fragmentation of Ionized Bicyclo[4.3.0]nona-3,7-dienes," *Org. Mass Spectrom.* **20**, 360 (1985).
- 85VAN/LEA**
V.J. Vandiver, C.S. Leasure and G.A. Eiceman, "Proton Affinity for Polycyclic Aromatic Hydrocarbons at Atmospheric Pressure in Ion Mobility Spectrometry," *Int. J. Mass Spectrom. Ion Proc.* **66**, 223 (1985).
- 85VEK/TAM**
K. Vekey, J. Tamas, E. Berenyi and P. Benko, "Mass Spectrometric Study on 2-Amino-as-triazino[6,5-c]quinoline and Some of Its Derivatives," *Org. Mass Spectrom.* **20**, 416 (1985).
- 85VIG/PAU**
A.A. Viggiano, J.F. Paulson, F. Dale, M. Henchman, N.G. Adams and D. Smith, "Ion Chemistry and Electron Affinity of WF_6 ," *J. Phys. Chem.* **89**, 2264 (1985).
- 85WAG/KEM**
W. Wagner-Redeker, P.R. Kemper, M.F. Jarrold and M.T. Bowers, "The Formation and Reactivity of HOC^+ : Interstellar Implications," *J. Chem. Phys.* **83**, 1121 (1985).
- 85WEI/PLA**
D.A. Weil, I. Platzner, L.L. Miller and D.A. Dixon, "Positive Ion-Molecule Reactions in OCS/Hydrocarbon Mixtures," *Org. Mass Spectrom.* **20**, 115 (1985).
- 85WIB/DAI**
K.B. Wiberg, W.P. Dailey, F.H. Walker, S.T. Waddell, L.S. Crocker and M. Newton, *J. Am. Chem. Soc.* **107**, 7247 (1985).
- 85WIB/WAS**
K.B. Wiberg, D.V. Wasserman, E.J. Martin and M.A. Murcko, *J. Am. Chem. Soc.* **107**, 6019 (1985).
- 85WOD/LEE**
A.M. Wodtke and Y.T. Lee, "Photodissociation of Acetylene at 193.3 nm," *J. Phys. Chem.* **89**, 4744 (1985).
- 85YAM/HIG**
H. Yamaguchi, M. Higashi, J.A.H. MacBride and R. Gleiter, "Photoelectron Spectra of Diazabiphenylenes," *J. Chem. Soc. Faraday Trans. II* **81**, 1831 (1985).
- 85ZAY/PER**
A.Yu. Zayats, A.A. Perov and A.P. Simonov, "Formation of Long-lived Rydberg Atoms Upon Excitation of NF_3 and CF_2Cl_2 Molecules by Electron Impact," *Sov. J. Chem. Phys.* **2**, 1906 (1985).
- 86ADA/SMI**
N.G. Adams, D. Smith, A.A. Viggiano, J.F. Paulson and M.J. Henchman, "Dissociative Attachment Reactions of Electron with Strong Acid Molecules," *J. Chem. Phys.* **84**, 6728 (1986).
- 86AND/TER**
R.R. Andrea, A. Terpstra, A. Oskam, P. Bruin and J.H. Teuben, "He(I) and He(II) Photoelectron Spectra of Some Mixed Sandwich Compounds of Titanium, Zirconium and Hafnium," *J. Organomet. Chem.* **307**, 307 (1986).
- 86ARI/ARM**
N. Aristov and P. Armentrout, "Reaction Mechanisms and Thermochemistry of $V^+ + C_2H_p$ ($p = 1-3$)," *J. Am. Chem. Soc.* **108**, 1806 (1986).
- 86ARM**
P. Armentrout, "Laser Application in Chemistry and Biochemistry," *Proc. SPIE* **620**, 38 (1986).
- 86BAA/WEI**
B. Van Baar, T. Weiske, J. Terlouw and H. Schwarz, "Hydroxyacetylene: Generation and Characterization of the Neutral Molecule, Radical Cation and Dication in the Gas Phase," *Angew. Chem. Int. Ed.* **25**, 282 (1986).
- 86BAL/JON**
R.E. Ballard, J. Jones, E. Sutherland, D. Read and A. Inchley, "The He(I) Photoelectron Spectrum of Benzyl Alcohol - Liquid Surface and Gas Phase," *Chem. Phys. Lett.* **126**, 311 (1986).
- 86BAR/KIP**
J.E. Bartmess and J.P. Kiplinger, "Kinetic vs. Thermodynamic Acidities of Enones in the Gas Phase," *J. Org. Chem.* **51**, 2173 (1986).
- 86BEC/HUN**
K. Beck, S. Hunig, R. Poppek, F. Prokschy and P. Rademacher, "Photoelektronenspektroskopische Untersuchungen an Hexahydro-1,3,5-triazinen," *Chem. Ber.* **119**, 554 (1986).
- 86BEC/HUN2**
K. Beck, S. Hunig, G. Kleefeld, H.-D. Martin, K. Peters, F. Prokschy and H.G. von Schnering, "Photoelectron and UV Spectroscopic Investigations of Homoconjugative Interactions between Parallel C=C and N=N Bonds," *Chem. Ber.* **119**, 543 (1986).
- 86BER/CUR**
J. Berkowitz, L. Curtiss, S. Gibson, J. Greene, G. Hillhouse and J. Pople, "Photoionization Mass Spectrometric Study and Ab Initio Calculation of Ionization and Bonding in P-H Compounds. Heats of Formation, Bond Energies, and the $^3B_1-^1A_1$ Separation in PH_2^+ ," *J. Chem. Phys.* **84**, 375 (1986).
- 86BOC/BAN**
H. Bock and M. Bankmann, " $H_3C-P=CH_2$: An Ylide with Two-Coordinate Phosphorus?" *Angew. Chem. Int. Ed.* **25**, 265 (1986).
- 86BOU/DJA**
G. Bouchoux, F. Djazi, Y. Hoppilliard, R. Houriet and E. Rolli, "Gas Phase Protonation of Unsaturated Ethers - Experimental and Theoretical Study of 2,3- and 2,5-Dihydrofuran and Related Compounds," *Org. Mass Spectrom.* **21**, 209 (1986).
- 86BOU/HAN**
G. Bouchoux, I. Hanna, R. Houriet and E. Rolli, "Gas Phase Basicity of Dihydropyran and Dihydro-1,4-dioxin," *Can. J. Chem.* **64**, 1345 (1986).
- 86BOW/DEP**
J.H. Bowie, C.H. DePuy, S.A. Sullivan and V.M. Bierbaum, "Gas Phase Reactions of the Hydroperoxide and Peroxyformate Anions," *Can. J. Chem.* **64**, 1046 (1986).
- 86BRO/LIG**
M. Brouard, P.D. Lightfoot and M.J. Pilling, "Observation of Equilibration in the System $H + C_2H_4 = C_2H_5$. The Determination of the Heat of Formation of C_2H_5 ," *J. Phys. Chem.* **90**, 445 (1986).
- 86BUD/KRA**
P.H.M. Budzelaar, E. Kraka, D. Cremer and P.v.R. Schleyer, *J. Am. Chem. Soc.* **108**, 561 (1986).
- 86BUR/FAW**
J. Burgess, J. Fawcett and R.D. Peacock, *J. Fluor. Chem.* **31**, 25 (1986).
- 86BUR/HOL**
P.C. Burgers, J.L. Holmes, C.E.C.A. Hop and J.K. Terlouw, "Gas Phase Ion Chemistry of Methyl Acetate, Methyl Propanoate and Their Enolic Tautomers," *Org. Mass Spectrom.* **21**, 549 (1986).
- 86CAS/KEE**
A.W. Castleman, Jr. and R.G. Keesee, "Ionic Clusters," *Chem. Rev.* **86**, 589 (1986).
- 86CAU/DIV**
C. Cauletti, M.L. Di Vona, P. Gargano, F. Grandinetti, C. Galli and C. Lillocci, "Ring-size Effects on the Ionization Potentials of N-Substituted Azacycloalkanes," *J. Chem. Soc. Perkin Trans. II* **667** (1986).

86CHI/ANN

J.S. Chickos, R. Annunziata, L.H. Ladon, A.S. Hyman and J.F. Liebman, *J. Org. Chem.* **51**, 4311 (1986).

86CHO/GRI

S. Chowdury, E.P. Grimsrud, T. Heinis and P. Kebarle, "Electron Affinities of Perfluorobenzene and Perfluorophenyl Compounds," *J. Am. Chem. Soc.* **108**, 3630 (1986).

86CHO/KEB

S. Chowdhury and P. Kebarle, "Electron Affinities of Di- and Tetracyanoethylene and Cyanobenzenes Based on Measurements of Gas-Phase Electron Transfer Equilibria," *J. Am. Chem. Soc.* **108**, 5453 (1986).

86COE/SNO

J.V. Coe, J.T. Snodgrass, C.B. Freidhoff, K.M. McHugh and K.H. Bowen, "Negative Ion Photoelectron Spectroscopy of N_2O^- and $(N_2O)_2^-$," *Chem. Phys. Lett.* **124**, 274 (1986).

86COE/SNO2

J.V. Coe, J.T. Snodgrass, C.B. Freidhoff, K.M. McHugh and K.H. Bowen, "Photoelectron Spectroscopy of the Negative Ion SeO^- ," *J. Chem. Phys.* **84**, 618 (1986).

86DAM/DEP

R. Damrauer, C.H. DePuy, I.M.T. Davidson and K.J. Hughes, "Gas Phase Ion Chemistry of Dimethylsilene," *Organomet.* **5**, 2050 (1986).

86DAM/DEP2

R. Damrauer, C.H. DePuy, I.M.T. Davidson and K.J. Hughes, "Gas Phase Ion Chemistry of Dimethylsilylene," *Organomet.* **5**, 2054 (1986).

86DAS/GIL

P.R. Das, J.P. Gilman and G.G. Meisels, "Unimolecular Decomposition of Energy-Selected Anisole Ions. Breakdown Graph and Metastable Decay Rates," *Int. J. Mass Spectrom. Ion Proc.* **68**, 155 (1986).

86DON/WAL

A.M. Doncaster and R. Walsh, *J. Chem. Soc. Faraday Trans. II* **82**, 707 (1986).

86ECK/GLE

M. Eckert-Maksic and R. Gleiter, "Photoelectron Spectra of N-substituted 1,4-Dihydro-4,4-dimethylpyridines," *Chem. Ber.* **119**, 2381 (1986).

86ELK/ARI

J. Elkind, N. Aristov, R. Georgiadis, L. Sunderlin and P. Armentrout to be published. Referred to in, "Structure, Reactivity and Thermochemistry of Ions" (P. Ausloos and S.G. Lias, editors), D. Reidel Publ. Co. (1986).

86ELK/ARM

J. Elkind and P. Armentrout, "Transition-Metal Hydride Bond Energies: First and Second Row," *Inorg. Chem.* **25**, 1078 (1986).

86ELK/ARM2

J. Elkind and P. Armentrout, "Effect of Kinetic and Electronic Energy on the Reactions of Mn^+ with H_2 , HD, and D_2 ," *J. Chem. Phys.* **84**, 4862 (1986).

86ELK/ARM3

J. Elkind and P. Armentrout, "Effect of Kinetic and Electronic Energy on the Reactions of Fe^+ with H_2 , HD, and D_2 : State-Specific Cross Sections for $Fe^+(^6D)$ and $Fe^+(^4F)$," *J. Phys. Chem.* **90**, 5736 (1986).

86ELK/ARM4

J. Elkind and P. Armentrout, "Effect of Kinetic and Electronic Energy on the Reactions of Co^+ , Ni^+ , and Cu^+ with H_2 , HD, and D_2 ," *J. Phys. Chem.* **90**, 6576 (1986).

86FRE/COE

C.B. Freidhoff, J.V. Coe, J.T. Snodgrass, K.M. McHugh and K.H. Bowen, "Negative Ion Photoelectron Spectroscopy of TeO^- ," *Chem. Phys. Lett.* **124**, 268 (1986).

86FRE/SNO

C.B. Freidhoff, J.T. Snodgrass, J.V. Coe, K.M. McHugh and K.H. Bowen, "Negative Ion Photoelectron Spectroscopy of TeH^- ," *J. Chem. Phys.* **84**, 1051 (1986).

86FRO/FRE

S.W. Froelicher, B.S. Freiser and R.R. Squires, "The $C_3H_5^-$ Isomers. Experimental and Theoretical Studies of the Tautomeric Propenyl Ions and the Cyclopropyl Anion in the Gas Phase," *J. Am. Chem. Soc.* **108**, 2853 (1986).

86FUJ/OHN

S. Fujisawa, K. Ohno, S. Masuda and Y. Harada, "Penning Ionization Electron Spectroscopy of Monohalobenzenes: C_6H_5F , C_6H_5Cl , C_6H_5Br , and C_6H_5I ," *J. Am. Chem. Soc.* **108**, 6505 (1986).

86GEO/ARM

R. Georgiadis and P. Armentrout, "Neutral and Ionic Metal Methyl Bond Energies: Zn," *J. Am. Chem. Soc.* **108**, 2119 (1986).

86GLE/BIS

R. Gleiter, P. Bischof and M. Christl, "Electronic Structure of Octavalene. Photoelectron Spectroscopic Investigations," *J. Org. Chem.* **51**, 2895 (1986).

86GLE/KRE

R. Gleiter, G. Krennrich and U.H. Brinker, "Electronic Structure of Spiropentane and Some Derivatives," *J. Org. Chem.* **51**, 2899 (1986).

86GLE/KRE2

R. Gleiter, G. Krennrich, P. Bischof, T. Tsuji and S. Nishida, "103. PE Spectra of Dewar Benzenes, Bridged by a Cyclohexadiene or a Butadiene Unit," *Helv. Chim. Acta* **69**, 962 (1986).

86GLE/USC

R. Gleiter, J. Uschmann and M. Baudler, "The Electronic Structure of Substituted 1,2,4,5-Tetraphosphaspiro[2.2]pentanes and 1,2,4,5-Tetraphospha-3-silaspiro[2.2]pentanes," *J. Chem. Soc. Dalton Trans.* 1659 (1986).

86GLE/USC2

R. Gleiter and J. Uschmann, "Electronic Structure of Heterospirenes - PE Spectroscopic Investigations," *J. Org. Chem.* **51**, 370 (1986).

86HAJ/SQU

D.J. Hajdasz and R.R. Squires, "Hypervalent Silicon Hydrides: SiH_5^- ," *J. Am. Chem. Soc.* **108**, 3139 (1986).

86HAW/GRI

J.A. Hawari, D. Griller and F.P. Lossing, "Thermochemistry of Perthiyl Radicals," *J. Am. Chem. Soc.* **108**, 3273 (1986).

86HAY/KRU

K. Hayashibara, G.H. Kruppa and J.L. Beauchamp, "Photoelectron Spectroscopy of the o-, m-, and p-Methylbenzyl Radicals. Implications for the Thermochemistry of the Radicals and Ions," *J. Am. Chem. Soc.* **108**, 5441 (1986).

86HEN/ILL

M. Heni, E. Illenberger and D. Lentz, "The Isomers CF_3CN and CF_3NC . Formation and Dissociation of the Anions Formed on Electron Attachment," *Int. J. Mass Spectrom. Ion Proc.* **71**, 199 (1986).

86HEN/ILL2

M. Heni and E. Illenberger, "Electron Attachment by Saturated Nitriles. Acrylonitrile (CH_2H_3CN), and Benzonitrile (C_6H_5CN)," *Int. J. Mass Spectrom. Ion Phys.* **73**, 127 (1986).

86HEI/FRE

R.L. Hettich and B.S. Freiser, "Gas-Phase Photodissociation of $FeCH_2^+$ and $CoCH_2^+$: Determination of the Carbide, Carbyne, and Carbene Bond Energies," *J. Am. Chem. Soc.* **108**, 2537 (1986).

86HIR/SHO

K. Hiraoka, T. Shoda, K. Morise, S. Yamabe, E. Kawai and K. Hirao, "Stability and Structure of Cluster Ions in the Gas Phase: Carbon Dioxide with Cl^- , H_3O^+ , HCO_2^+ and HCO^+ ," *J. Chem. Phys.* **84**, 2091 (1986).

86HOL/LOS

J.L. Holmes and F.P. Lossing, Unpublished result.

- 86HOU/SCH**
R. Houriet, T. Schwitzguebel, P.-A. Carrupt and P. Vogel, "Experimental and Theoretical Studies on the Homoconjugation in the Bicyclic Carbenium and Oxonium Ions in the Gas Phase," *Tetrahedron Lett.* **27**, 37 (1986).
- 86HOV/MCM**
J.K. Hovey and T.B. McMahon, "C-Xe Bond Strength in the Methylxenonium Cation Determined from Ion Cyclotron Resonance Methyl Cation Exchange Equilibria," *J. Am. Chem. Soc.* **108**, 528 (1986).
- 86HUS/WIL**
N.S. Hush, G.D. Willett, M.N. Paddon-Row, H.K. Patney and J.B. Peel, "Orbital Interactions. Part 13. The Observation of Through-bond Orbital Interactions between Benzene and Double Bonds in Some Dimethanoanthracenes," *J. Chem. Soc. Perkin Trans. II* **827** (1986).
- 86IGE/WED**
G. Igel-Mann, U. Wedig, P. Fuentealba and H. Stoll, *J. Chem. Phys.* **84**, 5007 (1986).
- 86KAM/BOS**
J. Kamphuis, H.J.T. Bos, C.W. Worrell and W. Runge, "The Molecular Structure of Allenes and Ketenes. Part 19. Photoelectron Spectra and Conformations of Donor-substituted Allenes," *J. Chem. Soc. Perkin Trans. II* **1509** (1986).
- 86KAM/YOU**
A. Kamar, A.B. Young and R.E. March, "Experimentally Determined Proton Affinities of 4-Methyl-3-Penten-2-one, 2-Propyl Ethanoate and 4-Hydroxy-4-methyl-2-pentanone in the Gas Phase," *Can. J. Chem.*, in press.
- 86KAR/JAS**
Z. Karpas and P. Jasien, "The Proton Affinity of Hydrogen Telluride," *Int. J. Mass Spectrom. Ion Proc.* **69**, 115 (1986).
- 86KAS/FIL**
S.R. Kass, J. Filley, J.M. Van Doren and C.H. DePuy, "Nitrous Oxide in Gas-Phase Ion-Molecule Chemistry: A Versatile Reagent for the Determination of Carbanion Structure," *J. Am. Chem. Soc.* **108**, 2849 (1986).
- 86KAT/ELI**
A. Katrib, B.D. El-Issa and A.W. Potts, "The He(I) and X-Ray Photoelectron Spectra of p-N,N-dimethylaminobenzal-malonitrile," *Can. J. Chem.* **64**, 528 (1986).
- 86KEE/CAS**
R.G. Keesee and A.W. Castleman, Jr., "Thermochemical Data on Gas-Phase Ion-Molecule Association and Clustering Reactions," *J. Phys. Chem. Ref. Data* **15**, 1011 (1986).
- 86KIN/NAG**
J.E. Kingcade, H.M. Nagarathna-Naik, I. Shim and K.A. Gingench, *J. Phys. Chem.* **90**, 2830 (1986).
- 86KIR/ACR**
J.E. Kirchner, W.E. Acree, Jr., G. Pilcher and L. Shofeng, *J. Chem. Thermodyn.* **18**, 793 (1986).
- 86KNI/FRE**
J.S. Knight, C.G. Freeman and M.J. McEwan, "Isomers of $C_2H_4N^+$ and the Proton Affinities of CH_3CN and CH_3NC ," *J. Am. Chem. Soc.* **108**, 1404 (1986).
- 86KOB/YOS**
T. Kobayashi, Z. Yoshida, Y. Asako, S. Miki and S. Kato, "Avoided Crossing Between the Ground and First Excited $2A'$ Cation States of Dicyanobicycloalkadiene Series Revealed by Photoelectron Spectroscopy," *Chem. Phys. Lett.* **125**, 586 (1986).
- 86KOL/KOZ**
V.P. Kolesov and M.P. Kozina, "Thermochemistry of Organic and Organohalogen Compounds" *Russian Chem. Revs.* **55**, 912 (1986).
- 86KOR/NIK**
M.V. Korobov, V.V. Nikulin, N.S. Chilingarov, L.N. Sidorov, *J. Chem. Thermodyn.* **18**, 235 (1986).
- 86KRU/BEA**
G.H. Kruppa and J.L. Beauchamp, "Energetics and Structure of the 1- and 2-Adamantyl Radicals and Their Corresponding Carbonium Ions by Photoelectron Spectroscopy," *J. Am. Chem. Soc.* **108**, 2162 (1986).
- 86KUZ/KOR**
S.V. Kuznetsov, M.V. Korobov, L.N. Savinova and L.N. Sidorov, "Enthalpy of the Addition of the F^- Ion to Copper and Iron Difluorides," *Russ. J. Phys. Chem.* **60**, 766 (1986).
- 86LAF/GON**
C. Lafon, D. Gonbeau, G. Pfister-Guillouzo, M. Lasne, J. Ripoll and J. Denis, "Etheneamine: Spectre Photoelectronique," *Nouv. J. Chem./New J. Chem.* **10**, 70 (1986).
- 86LAN/SAL**
K.R. Lane, L. Sallans and R.R. Squires, "Gas-Phase Nucleophilic Addition Reactions of Negative Ions with Transition Metal Carbonyls," *J. Am. Chem. Soc.* **108**, 4368 (1986).
- 86LAU/WES**
W.M. Lau, N.P.C. Westwood and M.H. Palmer, "Vaporization of (SN)_x: He I Photoelectron Spectrum and ab Initio Calculations for the S₃N₃ Radical," *J. Am. Chem. Soc.* **108**, 3229 (1986).
- 86LEB/URA**
P.R. LeBreton, S. Urano, M. Shahbaz, S.L. Emery and J.A. Morrison, "He I Photoelectron Spectra, Valence Electronic Structure, and Back Bonding in the Deltahedral Boron Chlorides, B₄Cl₄, B₈Cl₈, and B₉Cl₉," *J. Am. Chem. Soc.* **108**, 3937 (1986).
- 86LEE/SQU**
R.E. Lee and R.R. Squires, "Anionic Homoaromaticity: a Gas Phase Experimental Study," *J. Am. Chem. Soc.* **105**, 5078 (1986).
- 86LEO/LIN**
D.G. Leopold and W.C. Lineberger, "A Study of the Low-Lying Electronic States of Fe₂ and Co₂ by Negative Ion Photoelectron Spectroscopy," *J. Chem. Phys.* **85**, 51 (1986).
- 86LEO/MIL**
D.G. Leopold, A.G. Miller and W.C. Lineberger, "Determination of the Singlet-Triplet Splitting and Electron Affinity of, o-Benzynes by Negative Ion Photoelectron Spectroscopy," *J. Am. Chem. Soc.* **108**, 1379 (1986).
- 86LEO/MIL2**
D.G. Leopold, T.M. Miller and W.C. Lineberger, "Flowing Afterglow Negative Ion Photoelectron Spectroscopy of Dirhenium: Evidence for Multiple Bonding in Re₂ and Re₂⁻," *J. Am. Chem. Soc.* **108**, 178 (1986).
- 86LEU/MAG**
W. Leupin, D. Magde, G. Persy and J. Wirz, "1,4,7-Triazacycl[3.3.3]azine: Basicity, Photoelectron Spectrum, Photophysical Properties," *J. Am. Chem. Soc.* **108**, 17 (1986).
- 86LIA/NG**
C. Liao and C. Ng, "Molecular Beam Photoionization Study of S₂," *J. Chem. Phys.* **84**, 788 (1986).
- 86LIC/KEL**
D.L. Lichtenberger and G.E. Kellogg, "Electronic Structure Factors of Carbon-Hydrogen Bond Activation. The Photoelectron Spectroscopy of (Cyclohexenyl)manganese Tricarbonyl," *J. Am. Chem. Soc.* **108**, 2560 (1986).
- 86LIE/PAQ**
J.F. Liebman, L.A. Paquette, J.L. Peterson and D.W. Rogers, *J. Am. Chem. Soc.* **108**, 8267 (1986).
- 86MAI/OLE**
J. Main-Boob, S. Loesik, W. Gase, T. Baer, A. Mommers and J. Holmes, "The Thermochemistry and Dissociation Dynamics of Internal-Energy Selected Pyrazole and Imidazole Ions," *J. Am. Chem. Soc.* **108**, 677 (1986).
- 86MAL/LIF**
Y. Malinovich and C. Lifshitz, "Time-Dependent Mass Spectra and Breakdown Graphs. 7. Time-Resolved Photoionization Mass Spectrometry of Iodobenzene. The Heat of Formation of C₆H₅⁺," *J. Phys. Chem.* **90**, 2200 (1986).

86MAR/TOP

S. Marriott, R.D. Topsom, C.B. Lebrilla, I. Koppel, M. Mishima and R.W. Taft, "Proton Affinities of Substituted Cyanides," *J. Molec. Str. (Theochem)* **137**, 133 (1986).

86MAR/WET

J. Marks, D.M. Wetzel, P.B. Comita and J.I. Brauman, "A Dipole Supported State in Cyanomethyl Anion, the Conjugate Base of Acetonitrile. Rotational Band Assignments in the Electron Photodetachment Spectrum of $-\text{CH}_2\text{CN}^-$," *J. Chem. Phys.* **84**, 5284 (1986).

86MAT/AKA

H. Matsumoto, K. Akaiwa, Y. Nagai, K. Ohno, K. Imai, S. Masuda and Y. Harada, "Analysis of Stereochemical Properties of Molecular Orbitals of (Trimethylsilyl)acetylenes by Penning Ionization Electron Spectroscopy," *Organomet.* **5**, 1526 (1986).

86MAU

M. Meot-Ner (Mautner), Personal communication.

86MAU/KAR

M. Meot-Ner (Mautner), Z. Karpas and C. A. Deakne, "Ion Chemistry of Cyanides and Isocyanides. I. The Carbon Lone Pair as Proton Acceptor: Proton Affinities of Isocyanides," *J. Am. Chem. Soc.*, in press.

86MAU/LIE

M. Meot-Ner (Mautner), J.F. Liebman and J.E. Del Bene, "Proton Affinities of Azoles. Experimental and Theoretical Studies," *J. Org. Chem.*, in press.

86MCE/ALL

S.W. McElvaney and J. Allison, "Gas-Phase Chemistry of Transition-Metal-Containing Anions with Alcohols, Chloroalkanes, and Bifunctional Organic Molecules," *Organomet.* **5**, 416 (1986).

86MCM/HAL

J.E. McMurry, G.J. Haley, J.R. Matz, J.C. Clardy, G. Van Duyne, R. Gleiter, W. Schafer and D.H. White, "Synthesis and Chemistry of Tetracyclo[8.2.2.2.5.2.9]-1,5,9-octadecatriene," *J. Am. Chem. Soc.* **108**, 2932 (1986).

86MCM/KEB

T.B. McMahon and P. Kebarle, "Strong Hydrogen Bonding in Gas-Phase Ions: A High Pressure Mass Spectrometric Study of Formation and Energetics of Methyl Fluoride Proton Bound Dimer," *J. Am. Chem. Soc.* **108**, 6502 (1986).

86MEO/SIE

M. Meot-ner and L.W. Sieck, "Relative Acidities of Water and Methanol, and the Stabilities of the Dimer Adducts," *J. Phys. Chem.* **90**, 6687 (1986).

86MEO/SIE2

M. Meot-Ner and L.W. Sieck, "The Ionic Hydrogen Bond and Ion Solvation. 5. $\text{OH}\cdots\text{O}^-$ Bonds. Gas Phase Solvation and Clustering of Alkoxide and Carboxylate Anions," *J. Am. Chem. Soc.* **108**, 7525 (1986).

86MIL/FEI

A.E.S. Miller, C.S. Feigerle and W.C. Lineberger, "Laser Photoelectron Spectroscopy of MnH_2^- , FeH_2^- , CoH_2^- , and NiH_2^- : Determination of the Electron Affinities for the Metal Dihydrides," *J. Chem. Phys.* **84**, 4127 (1986).

86MIL/FEI2

A.E.S. Miller, C.S. Feigerle and W.C. Lineberger, "Laser Photoelectron Spectroscopy of CrH^- , CoH^- , and NiH^- : Periodic Trends in the Electronic Structure of the Transition-Metal Hydrides," *J. Chem. Phys.* **84**, (1986).

86MIL/MIL

T.M. Miller, A.E.S. Miller and W.C. Lineberger, "Electron Affinities of Ge and Sn," *Phys. Rev. A*, Spring 1986.

86MUR/MIL

K.K. Murray, T.M. Miller, D.G. Leopold and W.C. Lineberger, "Laser Photoelectron Spectroscopy of the Formyl Anion," *J. Chem. Phys.* **84**, 2520 (1986).

86NIK/IGO

M.I. Nikitin, N.A. Igolkina, E.V. Skokan, I.D. Sorokin and L.N. Sidirov, "Enthalpies of Formation of the AlF_4^- Ion," *Russ. J. Phys. Chem.* **60**, 22 (1986).

86NIM/ELL

M.R. Nimlos and G.B. Ellison, "Photoelectron Spectroscopy of SO_2^- , S_3^- , and S_2O^- ," *J. Phys. Chem.* **90**, 2574 (1986).

86NIM/ELL2

M.R. Nimlos and G.B. Ellison, "Photoelectron Spectroscopy of SiH_3^- and SiD_3^- ," *J. Am. Chem. Soc.* **108**, 6522 (1986).

86NIS/DAS

T. Nishimura, P.R. Das and G.G. Meisels, "On the Dissociation Dynamics of Energy-Selected Nitrobenzene Ion," *J. Chem. Phys.* **84**, 6190 (1986).

86NOT/PRI

H. Noth and H. Prigge, "Kernresonanz- und He(I)-Photoelektronenspektroskopische Untersuchung an Disopropyl- und Di-tert-butylboranen," *Chem. Ber.* **119**, 338 (1986).

86NUN/BAR

L. Nunez, L. Barral, S.G. Largo and S. Pilcher, *J. Chem. Thermo.* **18**, 575 (1986).

86OAK/ELL

J.M. Oakes and G.B. Ellison, "Photoelectron Spectroscopy of Radical Anions," *Tetrahedron* **42**, 6263 (1986).

86ORL/MIS

V. Orlov, A. Misharev and V. Takhistox, "Determination of the Enthalpy of Formation of para-Substituted α,α -Dimethyl Cations in the Gas Phase," *Izv. Akad. Nauk SSSR, Ser. Khim.* **9**, 2006 (1986).

86POS/RUT

R. Postma, P. Ruttink, J. Terlouw and J. Holmes, "The Ketene-Water Radical Cation Dipole Complex $[\text{CH}_2\text{CO H}_2\text{O}]^+$," *J. Chem. Soc. Chem. Commun.* 683 (1986).

86REU/WAN

J.E. Reutt, L.S. Wang, Y.T. Lee and D.A. Shirley, "Molecular Beam Photoelectron Spectroscopy of $\text{Ni}(\text{CO})_4$," *Chem. Phys. Lett.* **126**, 399 (1986).

86ROT/LEN

W.R. Roth, H.W. Lennartz, I. Vogel, M. Leiendecker, M. Oda, *Chem. Ber.* **19**, 837 (1986).

86RUD/VOV

E.B. Rudnyi, O.M. Vovk, L.N. Sidirov, I.D. Sorokin and A.S. Alikhanyan, "Enthalpy of Formation of PO_2^- , PO_3^- , and NaPO_2 ," *High Temp.* **24**, 56 (1986).

86RUM

D. Rumack, Personal communication.

86SAN/BAL

I. Santos, D.W. Balogh, C.W. Doecke, A.G. Marshall and L.A. Paquette, "Gas-Phase Basicities of Cubane, Dodecahedrane, Methyl- and 1,16-Dimethyldodecahedrane as Measured by Fourier Transform Ion Cyclotron Resonance Mass Spectrometry," *J. Am. Chem. Soc.* **108**, 8183 (1986).

86SHI/BEA

S.K. Shin and J.L. Beauchamp, "Proton Affinity and Heat of Formation of Silylene," *J. Phys. Chem.* **90**, 1507 (1986).

86SHI/VOR

C.S. Shiner, P.E. Vornor and S.R. Kass, "Gas Phase Acidities and Heats of Formation of 2,4- and 2,5- Cyclohexadien-1-one, the Keto Tautomers of Phenol," *J. Am. Chem. Soc.* **108**, 5699 (1986).

86SID/BOR

L.N. Sidirov, A.Ya. Borshchevsky, O.V. Boltalina, I.D. Sorokin and E.V. Skokan, "Electron Affinities of Gaseous Iron Fluorides and Dimers," *Int. J. Mass Spectrom. Ion Proc.* **73**, 1 (1986).

86SIM/BEA

J.A. Martinho-Simoes and J.L. Beauchamp, *Chem. Rev.* review article in preparation.

86SMA

B. Smart, in "Molecular Structures and Energetics" Vol. 3 (ed. J.F. Liebman and A. Greenberg, VCH Publishers, Deerfield Beach, FL. 1986).

- 86SPA/RAD**
G. Spanka and P. Rademacher, "Transannular Interactions in Difunctional Medium Rings. 1. n/π Interactions in Cyclic Amino Ketones and Aminoalkenes Studied by Photoelectron Spectroscopy," *J. Org. Chem.* **51**, 592 (1986).
- 86SPI/GRU**
R. Spilker and H.-F. Grutzmacher, "Isomerization and Fragmentation of Methylfuran Ions and Pyran Ions in the Gas Phase," *Org. Mass. Spectrom.* **21**, 459 (1986).
- 86SPI/PER**
T.A. Spiglanin, R.A. Perry and D.W. Chandler, *J. Phys. Chem.* **90**, 6184 (1986).
- 86STE/BEA**
A.E. Stevens and J.L. Beauchamp, "Gas-Phase Acidities of $(CO)_5MnH$, $(CO)_4FeH_2$, and $(CO)_4CoH$," *J. Am. Chem. Soc.*, in press.
- 86STO/LAR**
R.C. Stoneman and D.J. Larson, "Photodetachment Spectroscopy of SeH^- in a Magnetic Field," *J. Phys. B* **19**, L405 (1986).
- 86TAF**
R.W. Taft, Personal communication of unpublished data, 1986.
- 86TAF/ANV**
R.W. Taft, F. Anvia, M. Taagepera, J. Catalan and J. Elguero, "Electrostatic Proximity Effects in the Relative Basicities of Pyrazole, Imidazole, Pyridazine, and Pyrimidine," *J. Am. Chem. Soc.* **108**, 3237 (1986).
- 86TAF/GAL**
R.W. Taft, J.-F. Gal, S. Geribaldi and P.-C. Maria, "Unique Basicity Properties of Conjugated Amino Cyclohexenone Derivatives. The Effects of Molecular Structure on the Disparate Basicities toward H Acids," *J. Am. Chem. Soc.* **108**, 861 (1986).
- 86TRA**
J.C. Traeger, "Heat of Formation for the 1-Methylallyl Cation," by Photoionization Mass Spectrometry," *J. Phys. Chem.* **90**, 4114 (1986).
- 86TRA/MCA**
J.C. Traeger and D.J. McAdoo, "Decomposition Thresholds and Associated Translational Energy Releases for Eight $C_4H_8O^+$ Isomers," *Int. J. Mass Spectrom. Ion Proc.* **68**, 35 (1986).
- 86TRA/MUN**
V.T. Tran and B. Munson, "Proton Affinities by Reactant Ion Monitoring: Triphenyl Group Va Compounds," *Org. Mass Spectrom.* **21**, 41 (1986).
- 86TSA**
W. Tsang, "Single Pulse Shock Tube Study on the Stability of Perfluorobromomethane" to be published.
- 86TUR/HAV**
F. Turecek, Z. Havlas, F. Maquin, N. Hill and T. Gaumann, "(E)- and (Z)-1-Hydroxy-1,3-butadiene: New Kinetically Unstable C_4H_6O Isomers," *J. Org. Chem.* **51**, 4061 (1986).
- 86TUR/HAV2**
F. Turecek, Z. Havlas, F. Maquin and T. Gaumann, "72. The Mass Spectra of Organic Compounds 1-Buten-3-yn-2-ol. A New Kinetically Unstable C_4H_4O Isomer," *Helv. Chim. Acta* **69**, 683 (1986).
- 86TUR/HAV3**
F. Turecek and Z. Havlas, "Energy Barriers to the Diels-Alder Cycloadditions and Cycloreversions of Cation-radicals in the Gas Phase," *J. Chem. Soc. Perkin Trans. II* **1011** (1986).
- 86VIG**
A.A. Viggiano, Personal communication (1986).
- 86VON**
T. Vondrak, "Electronic Structure of Ferrocene Derivatives Studied by He(I) Photoelectron Spectroscopy and CNDO/2 Method," *J. Organomet. Chem.* **306**, 89 (1986).
- 86VOR/BRO**
M.G. Voronkov, E.I. Brodskaya, V.V. Belyaeva, D.D. Chuvashev, D.D. Toryashinova, A.F. Ermikov and V.P. Baryshok, "Through-Bond Interaction in Compounds Containing an Si-O-C-C-N Group," *J. Organomet. Chem.* **311**, 9 (1986).
- 86WAL**
R. Walsh, "Thermochemical Kinetics: A Success Story," *J. Phys. Chem.* **90**, 389 (1986).
- 86WER**
N.H. Werstiuk, "Thermolysis of N-alkylated Ethylenediamines: An Ultraviolet Photoelectron Spectroscopy Study," *Can. J. Chem.* **64**, 2175 (1986).
- 86YAM/FUR**
S. Yamabe, Y. Furumiya, K. Hiraoka and K. Morise, "Theoretical Van't Hoff Plots of Gas Phase Ion Equilibria of Cl^- Ion in Water, Methanol and Acetonitrile," *Chem Phys. Lett.* **131**, 261 (1986).
- 86ZHE/KAR**
L.-S. Zheng, C.M. Karner, P.J. Brucat, S.H. Yang, C.L. Pettiette, M.J. Craycraft and R.E. Smalley, "Photodetachment Studies of Metal Clusters: Electron Affinity Measurements for Cu_x^- ," *J. Chem. Phys.* **85**, 1681 (1986).
- 87BOO/ARM**
B.H. Boo and P.B. Armentrout, "Reaction of Silicon Ion ($2P$) with Silane (SiH_4 , SiD_4). Heats of Formation of SiH_n^+ , SiH_n^+ ($n = 1, 2, 3$), and $Si_2H_n^+$ ($n = 0, 1, 2, 3$). Remarkable Isotope Exchange Reaction Involving Four Hydrogen Shifts," *J. Am. Chem. Soc.* **109**, 3549 (1987).
- 87CHI**
J.S. Chickos, "Molecular Structure and Energetics," Vol. 2 (ed. J.F. Liebman and A. Greenberg, VCH Publishers, Inc. Deerfield Beach, FL, 1987).
- 87COE/SNO**
J.V. Coe, J.T. Snodgrass, C.B. Freidhoff, K.M. McHugh and K.H. Bowen, "Photoelectron Spectroscopy of the Negative Cluster Ions, $NO^-(N_2O)_n = 1, 2$," *J. Chem. Phys.* **87**, 4302 (1987).
- 87DEA/MAU**
C.A. Deakne, M. Meot-Ner (Mautner), T.J. Buckley and R. Metz, "Proton Affinities of Diacetylene, Cyanoacetylene, and Cyanogen," *J. Chem. Phys.* **86**, 2334 (1987).
- 87FER/ROH**
E.E. Ferguson, J. Roncin and L. Bonazzola, "Heats of Formation and Bond Energies of H_3CO^+ and H_3COOH^+ Ions," *Int. J. Mass Spectrom. Ion Proc.*, **79**, 215 (1987).
- 87GAR/PAR**
D. Garvin, V.B. Parker and H.J. White, Jr., "CODATA Thermodynamic Tables: Selections for Some Compounds of Calcium and Related Mixtures: A Prototype Set of Tables" Hemisphere Publ. Corp. (1987).
- 87GRA/MEL**
J.J. Grabowski and S.J. Melly, "Formation of Carbene Radical Anions: Gas Phase Reaction of the Atomic Oxygen Anion with Organic Neutrals," *Int. J. Mass Spectrom. Ion Proc.* **xx,xxx** (1987).
- 87HER**
J.T. Herron, "Thermochemical Data on Gas Phase Compounds of Sulfur, Fluorine, and Oxygen Related to Pyrolysis and Oxidation of Sulfur Hexafluoride," *J. Phys. Chem. Ref Data* **16**, 1 (1987).
- 87JOH/SPE**
C.L. Johlman, L. Spencer, D.T. Sawyer and C.L. Wilkins, "Hydroxide Initiated Gas Phase Chemistry of Anthraquinone and Related Quinones," *J. Org. Chem.* **52**, 3027 (1987).
- 87KEB/CHO**
P. Kebarle and S. Chowdhury, "Electron Affinities and Electron Transfer Reactions," *Chem. Rev.* **87**, 513 (1987).
- 87LAR/MCM**
J.W. Larson and T.B. McMahon, "Hydrogen Bonding in Gas Phase Anions. The Energetics of Interaction Between Cyanide Ion and Bronsted Acids," *J. Am. Chem. Soc.* **109**, 6230 (1987).

87LIA/AUS

S.G. Lias and P. Ausloos, "Structures and Heats of Formation of $C_4H_7^+$ Ions in the Gas Phase," *Int. J. Mass Spectrom. Ion Proc.*, in press.

87MAS/FER

R. Mason, M.T. Fernandez and K.R. Jennings, *J. Chem. Soc. Faraday Trans. II* 83, 89 (1987).

87MEO

M. Meot-ner, Personal communication of unpublished results.

87MIL/FEI

A.E.S. Miller, C.S. Feigerle and W.C. Lineberger, "Laser Photoelectron Spectroscopy of CrH^+ , CoH^+ , and NiH^+ : Periodic Trends in the Electronic Structure of the Transition-Metal Hydrides," *J. Chem. Phys.* 86, 1549 (1987).

87MOR/ELL

S. Moran and G.B. Ellison, "Electron Spectroscopy of CH_2S^- ," *Int. J. Mass Spectrom. Ion Proc.* xx, xxx (1987).

87MOR/ELL2

S. Moran, H.B. Ellis, Jr., D.J. DeFrees, A.D. McLean, S.E. Paulson and G.B. Ellison, "Carbanion Spectroscopy: CH_2NC^- ," *J. Am. Chem. Soc.* 109, xxx (1987).

87MOR/ELL3

S. Moran, H.B. Ellis, Jr., D.J. DeFrees, A.D. McLean and G.B. Ellison, "Carbanion Spectroscopy: CH_2CN^- ," *J. Am. Chem. Soc.* 109, xxx (1987).

87SNO/COE

J.T. Snodgrass, J.V. Coe, K.M. McHugh, C.B. Freidhoff and

K.H. Bowen, "Photoelectron Spectroscopy of the Selenium and Tellurium Containing Negative Ions: SeO_2^- , Se_2^- and Te_2^- ," *Chem. Phys.* 87, xxx (1987).

87SNO/COE2

J.T. Snodgrass, J.V. Coe, C.B. Freidhoff, K.M. McHugh and K.H. Bowen, "Photoelectron Spectroscopy of the Negative Cluster Ions, $NH_2^-(NH_3)_n = 1,2$," *J. Chem. Phys.* 87, xxx (1987).

87SNO/COE3

J.T. Snodgrass, J.V. Coe, C.B. Freidhoff, K.M. McHugh and K.H. Bowen, "The Negative Ion Photoelectron Spectroscopy of $H^-(NH_3)$, $D^-(ND_3)$, and $H^-(NH_3)_2$," *J. Chem. Phys.* 87, xxx (1987).

87STA/NOR

S.W. Staley and T.D. Norden, Personal communication.

87STE/BEA

A.E. Stevens and J.L. Beauchamp, "Gas Phase Acidities of $(CO)_5MnH$, $(CO)_4FeH_2$, and $(CO)_4CoH$," *J. Am. Chem. Soc.* 109, xxx (1987).

87SUN/ARI

L. Sunderlin, N. Aristov and P. Armentrout, "Reaction of Scandium Ions with Ethane. First and Second Hydride-Scandium Ion Bond Energies," *J. Am. Chem. Soc.* 109, 78 (1987).

87THO/BAR

D. Thomas and J.E. Bartmess, Unpublished work.

Journal of Physical and Chemical Reference Data

Cumulative Listing of Reprints and Supplements

Reprints from Volume 1

| | |
|---|--------|
| 1. Gaseous Diffusion Coefficients, <i>T.R. Marrero and E.A. Mason</i> , Vol. 1, No. 1, pp. 1-118 (1972) | \$7.00 |
| 2. Selected Values of Critical Supersaturation for Nucleation of Liquids from the Vapor, <i>G.M. Pound</i> , Vol. 1, No. 1, pp. 119-134 (1972) | \$3.00 |
| 3. Selected Values of Evaporation and Condensation Coefficients for Simple Substances, <i>G.M. Pound</i> , Vol. 1, No. 1, pp. 135-146 (1972) | \$3.00 |
| 4. Atlas of the Observed Absorption Spectrum of Carbon Monoxide between 1060 and 1900 Å, <i>S.G. Tilford and J.D. Simmons</i> , Vol. 1, No. 1, pp. 147-188 (1972) | \$4.50 |
| 5. Tables of Molecular Vibrational Frequencies, Part 5, <i>T. Shimanouchi</i> , Vol. 1, No. 1, pp. 189-216 (1972) (superseded by No.103) | \$4.00 |
| 6. Selected Values of Heats of Combustion and Heats of Formation of Organic Compounds Containing the Elements C, H, N, O, P, and S, <i>Eugene S. Domalski</i> , Vol. 1, No. 2, pp. 221-278 (1972) | \$5.00 |
| 7. Thermal Conductivity of the Elements, <i>C.Y. Ho, R.W. Powell, and P.E. Liley</i> , Vol. 1, No. 2, pp. 279-422 (1972) | \$7.50 |
| 8. The Spectrum of Molecular Oxygen, <i>Paul H. Krupenie</i> , Vol. 1, No. 2, pp. 423-534 (1972) | \$6.50 |
| 9. A Critical Review of the Gas-Phase Reaction Kinetics of the Hydroxyl Radical, <i>Wm. E. Wilson, Jr.</i> , Vol. 1, No. 2, pp. 535-574 (1972) | \$4.50 |
| 10. Molten Salts: Volume 3, Nitrates, Nitrites, and Mixtures, Electrical Conductance, Density, Viscosity, and Surface Tension Data, <i>G.J. Janz, Ursula Krebs, H.F. Siegenthaler, and R.P.T. Tomkins</i> , Vol. 1, No. 3, pp. 581-746 (1972) | \$8.50 |
| 11. High Temperature Properties and Decomposition of Inorganic Salts—Part 3 Nitrates and Nitrites, <i>Kurt H. Stern</i> , Vol. 1, No. 3, pp. 747-772 (1972) | \$4.00 |
| 12. High-Pressure Calibration: A Critical Review, <i>D.L. Decker, W.A. Bassett, L. Merrill, H.T. Hall, and J.D. Barnett</i> , Vol. 1, No. 3, pp. 773-836 (1972) | \$5.00 |
| 13. The Surface Tension of Pure Liquid Compounds, <i>Joseph J. Jasper</i> , Vol. 1, No. 4, pp. 841-1009 (1972) | \$8.50 |

| | |
|--|--------|
| 14. Microwave Spectra of Molecules of Astrophysical Interest, I. Formaldehyde, Formamide, and Thioformaldehyde, <i>Donald R. Johnson, Frank J. Lovas, and William H. Kirchhoff</i> , Vol. 1, No. 4, pp. 1011-1046 (1972) | \$4.50 |
| 15. Osmotic Coefficients and Mean Activity Coefficients of Uni-univalent Electrolytes in Water at 25° C, <i>Walter J. Hamer and Yung-Chi Wu</i> , Vol. 1, No. 4, pp. 1047-1099 (1972) | \$5.00 |
| 16. The Viscosity and Thermal Conductivity Coefficients of Gaseous and Liquid Fluorine, <i>H.J.M. Hanley and R. Prydz</i> , Vol. 1, No. 4, pp. 1101-1113 (1972) | \$3.00 |

Reprints from Volume 2

| | |
|---|--------|
| 17. Microwave Spectra of Molecules of Astrophysical Interest, II Methylenimine, <i>William H. Kirchhoff, Donald R. Johnson, and Frank J. Lovas</i> , Vol. 2, No. 1, pp. 1-10 (1973) | \$3.00 |
| 18. Analysis of Specific Heat Data in the Critical Region of Magnetic Solids, <i>F.J. Cook</i> , Vol. 2, No. 1, pp. 11-24 (1973) | \$3.00 |
| 19. Evaluated Chemical Kinetic Rate Constants for Various Gas Phase Reactions, <i>Keith Schofield</i> , Vol. 2, No. 1, pp. 25-84 (1973) | \$5.00 |
| 20. Atomic Transition Probabilities for Forbidden Lines of the Iron Group Elements. (A Critical Data Compilation for Selected Lines), <i>M.W. Smith and W.L. Wiese</i> , Vol. 2, No. 1, pp. 85-120 (1973) | \$4.50 |
| 21. Tables of Molecular Vibrational Frequencies, Part 6, <i>T. Shimanouchi</i> , Vol. 2, No. 1, pp. 121-162 (1973) (superseded by No. 103) | \$4.50 |
| 22. Compilation of Energy Band Gaps in Elemental and Binary Compound Semiconductors and Insulators, <i>W.H. Strehlow and E.L. Cook</i> , Vol. 2, No. 1, pp. 163-200 (1973) | \$4.50 |
| 23. Microwave Spectra of Molecules of Astrophysical Interest, III. Methanol, <i>R.M. Lees, F.J. Lovas, W.H. Kirchhoff, and D.R. Johnson</i> , Vol. 2, No. 2, pp. 205-214 (1973) | \$3.00 |
| 24. Microwave Spectra of Molecules of Astrophysical Interest, IV. Hydrogen Sulfide, <i>Paul Helminger, Frank C. De Lucia, and William H. Kirchhoff</i> , Vol. 2, No. 2, pp. 215-224 (1973) | \$3.00 |

Journal of Physical and Chemical Reference Data Reprint and Supplement Orders

To: American Chemical Society
Distribution Office
1155 Sixteenth Street, N.W.
Washington, DC 20036

Name: _____

Title: _____

Organization: _____

Address: _____

City: _____ State: _____

Country: _____ Zip: _____

I am a member of _____
(ACS, AIP, or Affiliated Society)

ORDERS FOR REPRINTS AND SUPPLEMENTS MUST BE PREPAID
*Foreign orders for Reprints, add \$2.50 for each reprint for postage and handling Foreign orders for Reprint Packages, add \$5.00 for each Reprint Package for postage and handling Make checks payable to the American Chemical Society.

BULK RATES Subtract 20% from the listed price for orders of 50 or more of any one item.

Please ship the following reprints and supplements:

Reprint No./Package _____, _____ copies \$ _____

Reprint No./Package _____, _____ copies \$ _____

Reprint No./Package _____, _____ copies \$ _____

Vol. 2, Suppl 1 Hardcover _____ copies \$ _____

Softcover _____

Vol. 3, Suppl. 1 Hardcover _____ copies \$ _____

Softcover _____

Vol 6, Suppl. 1 Hardcover _____ copies \$ _____

Softcover _____

Vol 10, Suppl. 1 Hardcover _____ copies \$ _____

Vol. 11, Suppl. 1 Hardcover _____ copies \$ _____

Vol. 11, Suppl 2 Hardcover _____ copies \$ _____

Vol. 13, Suppl. 1 Hardcover _____ copies \$ _____

Vol. 14, Suppl. 1 Hardcover _____ copies \$ _____

Vol 14, Suppl 2 Hardcover _____ copies \$ _____

Vol. 16, Suppl. 1 Hardcover _____ copies \$ _____

Other Suppl: _____ copies \$ _____

Total Enclosed \$ _____

(Continuation of Cumulative Listing of Reprints)

25. Tables of Molecular Vibrational Frequencies, Part 7, *T. Shimanouchi*, Vol. 2, No. 2, pp. 225-256 (1973) (superseded by No. 103) \$4.00
26. Energy Levels of Neutral Helium (⁴He I), *W.C. Martin*, Vol. 2, No. 2, pp. 257-266 (1973) \$3.00
27. Survey of Photochemical and Rate Data for Twenty-eight Reactions of Interest in Atmospheric Chemistry, *R.F. Hampson*, Editor, *W. Braun, R.L. Brown, D. Garvin, J.T. Herron, R.E. Huie, M.J. Kurylo, A.H. Laufer, J.D. McKinley, H. Okabe, M.D. Scheer, W. Tsang, and D.H. Stedman*, Vol. 2, No. 2, pp. 267-312 (1973) \$4.50
28. Compilation of the Static Dielectric Constant of Inorganic Solids, *K.F. Young and H.P.R. Frederikse*, Vol. 2, No. 2, pp. 313-410 (1973) \$6.50
29. Soft X-Ray Emission Spectra of Metallic Solids: Critical Review of Selected Systems, *A.J. McAlister, R.C. Dobbyn, J.R. Cuthill, and M.L. Williams*, Vol. 2, No. 2, pp. 411-426 (1973) \$3.00
30. Ideal Gas Thermodynamic Properties of Ethane and Propane, *J. Chao, R.C. Wilhoit, and B.J. Zwolinski*, Vol. 2, No. 2, pp. 427-438 (1973) \$3.00
31. An Analysis of Coexistence Curve Data for Several Binary Liquid Mixtures Near Their Critical Points, *A. Stein and G.F. Allen*, Vol. 2, No. 3, pp. 443-466 (1973) \$4.00
32. Rate Constants for the Reactions of Atomic Oxygen (O³P) with Organic Compounds in the Gas Phase, *John T. Herron and Robert E. Huie*, Vol. 2, No. 3, pp. 467-518 (1973) \$5.00
33. First Spectra of Neon, Argon, and Xenon 136 in the 1.2-4.0 μm Region, *Curtis J. Humphreys*, Vol. 2, No. 3, pp. 519-530 (1973) \$3.00
34. Elastic Properties of Metals and Alloys, I. Iron, Nickel, and Iron-Nickel Alloys, *H.M. Ledbetter and R.P. Reed*, Vol. 2, No. 3, pp. 531-618 (1973) \$6.00
35. The Viscosity and Thermal Conductivity Coefficients of Dilute Argon, Krypton, and Xenon, *H.J.M. Hanley*, Vol. 2, No. 3, pp. 619-642 (1973) \$4.00
36. Diffusion in Copper and Copper Alloys, Part I. Volume and Surface Self-Diffusion in Copper, *Daniel B. Butrymowicz, John R. Manning, and Michael E. Read*, Vol. 2, No. 3, pp. 643-656 (1973) \$3.00
37. The 1973 Least-Squares Adjustment of the Fundamental Constants, *E. Richard Cohen and B.N. Taylor*, Vol. 2, No. 4, pp. 663-734 (1973) \$5.50
38. The Viscosity and Thermal Conductivity Coefficients of Dilute Nitrogen and Oxygen, *H.J.M. Hanley and James F. Ely*, Vol. 2, No. 4, pp. 735-756 (1973) \$4.00
39. Thermodynamic Properties of Nitrogen Including Liquid and Vapor Phases from 63 K to 2000 K with Pressures to 10,000 Bar, *Richard T. Jacobsen and Richard B. Stewart*, Vol. 2, No. 4, pp. 757-922 (1973) \$8.50
40. Thermodynamic Properties of Helium 4 from 2 to 1500 K at Pressures to 10⁸ Pa, *Robert T. McCarty*, Vol. 2, No. 4, pp. 923-1042 (1973) \$7.00
44. Critical Analysis of Heat-Capacity Data and Evaluation of Thermodynamic Properties of Ruthenium, Rhodium, Palladium, Iridium, and Platinum from 0 to 300 K. A Survey of the Literature Data on Osmium, *George T. Furukawa, Martin L. Reilly, and John S. Gallagher*, Vol. 3, No. 1, pp. 163-209 (1974) \$4.50
45. Microwave Spectra of Molecules of Astrophysical Interest, V. Water Vapor, *Frank C. De Lucia, Paul Helming, and William H. Kirchhoff*, Vol. 3, No. 1, pp. 211-219 (1974) \$3.00
46. Microwave Spectra of Molecules of Astrophysical Interest, VI. Carbonyl Sulfide and Hydrogen Cyanide, *Arthur G. Maki*, Vol. 3, No. 1, pp. 221-244 (1974) \$4.00
47. Microwave Spectra of Molecules of Astrophysical Interest, VII. Carbon Monoxide, Carbon Monosulfide, and Silicon Monoxide, *Frank J. Lovas and Paul H. Krupenie*, Vol. 3, No. 1, pp. 245-257 (1974) \$3.00
48. Microwave Spectra of Molecules of Astrophysical Interest, VIII. Sulfur Monoxide, *Eberhard Tiemann*, Vol. 3, No. 1, pp. 259-268 (1974) \$3.00
49. Tables of Molecular Vibrational Frequencies, Part 8, *T. Shimanouchi*, Vol. 3, No. 1, pp. 269-308 (1974) (superseded by No. 103) \$4.50
50. JANAF Thermochemical Tables, 1974 Supplement, *M.W. Chase, J.L. Curnutt, A.T. Hu, H. Prophet, A.N. Syverud, and L.C. Walker*, Vol. 3, No. 2, pp. 311-480 (1974) \$8.50
51. High Temperature Properties and Decomposition of Inorganic Salts, Part 4. Oxy-Salts of the Halogens, *Kurt H. Stern*, Vol. 3, No. 2, pp. 481-526 (1974) \$4.50
52. Diffusion in Copper and Copper Alloys, Part II. Copper-Silver and Copper-Gold Systems, *Daniel B. Butrymowicz, John R. Manning, and Michael E. Read*, Vol. 3, No. 2, pp. 527-602 (1974) \$5.50
53. Microwave Spectral Tables I. Diatomic Molecules, *Frank J. Lovas and Eberhard Tiemann*, Vol. 3, No. 3, pp. 609-770 (1974) \$8.50
54. Ground Levels and Ionization Potentials for Lanthanide and Actinide Atoms and Ions, *W.C. Martin, Lucy Hagan, Joseph Reader, and Jack Sugar*, Vol. 3, No. 3, pp. 771-780 (1974) \$3.00
55. Behavior of the Elements at High Pressures, *John Francis Cannon*, Vol. 3, No. 3, pp. 781-824 (1974) \$4.50
56. Reference Wavelengths from Atomic Spectra in the Range 15 Å to 25000 Å, *Victor Kaufman and Bengt Edlén*, Vol. 3, No. 4, pp. 825-895 (1974) \$5.50
57. Elastic Properties of Metals and Alloys II. Copper, *H.M. Ledbetter and E.R. Naimon*, Vol. 3, No. 4, pp. 897-935 (1974) \$4.50
58. A Critical Review of H-Atom Transfer in the Liquid Phase: Chlorine Atom, Alkyl, Trichloromethyl, Alkoxy, and Alkylperoxy Radicals, *D.G. Hendry, T. Mill, L. Piskiewicz, J.A. Howard, and H.K. Eigenmann*, Vol. 3, No. 4, pp. 937-978 (1974) \$4.50
59. The Viscosity and Thermal Conductivity Coefficients for Dense Gaseous and Liquid Argon, Krypton, Xenon, Nitrogen, and Oxygen, *H.J.M. Hanley, R.D. McCarty, and W.M. Haynes*, Vol. 3, No. 4, pp. 979-1017 (1974) \$4.50

Reprints from Volume 3

41. Molten Salts: Volume 4, Part 1, Fluorides and Mixtures, Electrical Conductance, Density, Viscosity, and Surface Tension Data, *G.J. Janz, G.L. Gardner, Ursula Krebs, and R.P.T. Tomkins*, Vol. 3, No. 1, pp. 1-115 (1974) \$7.00
42. Ideal Gas Thermodynamic Properties of Eight Chloro- and Fluoromethanes, *A.S. Rodgers, J. Chao, R. C. Wilhoit, and B.J. Zwolinski*, Vol. 3, No. 1, pp. 117-140 (1974) \$4.00
43. Ideal Gas Thermodynamic Properties of Six Chloroethanes, *J. Chao, A.S. Rodgers, R.C. Wilhoit, and B.J. Zwolinski*, Vol. 3, No. 1, pp. 141-162 (1974) \$4.00

Reprints from Volume 4

60. JANAF Thermochemical Tables, 1975 Supplement, *M.W. Chase, J.L. Curnutt, H. Prophet, R.A. McDonald, and A.N. Syverud*, Vol. 4, No. 1, pp. 1-175 (1975) \$8.50
61. Diffusion in Copper and Copper Alloys, Part III. Diffusion in Systems Involving Elements of the Groups IA, IIA, IIIB, IVB, VB, VIB, and VIIB, *Daniel B. Butrymowicz, John R. Manning, and Michael E. Read*, Vol. 4, No. 1, pp. 177-249 (1975) \$6.00

(Continuation of Cumulative Listing of Reprints)

| | | | |
|--|---------|---|---------|
| 62. Ideal Gas Thermodynamic Properties of Ethylene and Propylene, <i>Jing Chao and Bruno J. Zwolinski</i> , Vol. 4, No. 1, pp. 251-261 (1975) | \$3.00 | 82. Tables of Critically Evaluated Oscillator Strengths for the Lithium Isoelectronic Sequence, <i>G.A. Martin and W.L. Wiese</i> , Vol. 5, No. 3, pp. 537-570 (1976) | \$4.50 |
| 63. Atomic Transition Probabilities for Scandium and Titanium (A Critical Data Compilation of Allowed Lines), <i>W.L. Wiese and J.R. Fuhr</i> , Vol. 4, No. 2, pp. 263-352 (1975) | \$6.00 | 83. Ideal Gas Thermodynamic Properties of Six Chlorofluoromethanes, <i>S.S. Chen, R. C. Wilhoit, and B.J. Zwolinski</i> , Vol. 5, No. 3, pp. 571-580 (1976) | \$3.00 |
| 64. Energy Levels of Iron, Fe I through Fe XXVI, <i>Joseph Reader and Jack Sugar</i> , Vol. 4, No. 2, pp. 353-440 (1975) | \$6.00 | 84. Survey of Superconductive Materials and Critical Evaluation of Selected Properties, <i>B.W. Roberts</i> , Vol. 5, No. 3, pp. 581-821 (1976) | \$12.50 |
| 65. Ideal Gas Thermodynamic Properties of Six Fluoroethanes, <i>S.S. Chen, A.S. Rodgers, J. Chao, R.C. Wilhoit, and B.J. Zwolinski</i> , Vol. 4, No. 2, pp. 441-456 (1975) | \$3.00 | 85. Nuclear Spins and Moments, <i>Gladys H. Fuller</i> , Vol. 5, No. 4, pp. 835-1092 (1976) | \$11.50 |
| 66. Ideal Gas Thermodynamic Properties of the Eight Bromo- and Iodomethanes, <i>S.A. Kudchadker and A.P. Kudchadker</i> , Vol. 4, No. 2, pp. 457-470 (1975) | \$3.00 | 86. Nuclear Moments and Moment Ratios as Determined by Mössbauer Spectroscopy, <i>J.G. Stevens and B.D. Dunlap</i> , Vol. 5, No. 4, pp. 1093-1121 (1976) | \$4.00 |
| 67. Atomic Form Factors, Incoherent Scattering Functions, and Photon Scattering Cross Sections, <i>J.H. Hubbell, Wm.J. Veigele, E.A. Briggs, R.T. Brown, D.T. Cromer, and R.J. Howerton</i> , Vol. 4, No. 3, pp. 471-538 (1975) | \$5.50 | 87. Rate Coefficients for Ion-Molecule Reactions, I. Ions Containing C and H, <i>L. Wayne Sieck and Sharon G. Lias</i> , Vol. 5, No. 4, pp. 1123-1146 (1976) | \$4.00 |
| 68. Binding Energies in Atomic Negative Ions, <i>H. Hotop and W.C. Lineberger</i> , Vol. 4, No. 3, pp. 539-576 (1975) | \$4.50 | 88. Microwave Spectra of Molecules of Astrophysical Interest, XI. Silicon Sulfide, <i>Eberhard Tiemann</i> , Vol. 5, No. 4, pp. 1147-1156 (1976) | \$3.00 |
| 69. A Survey of Electron Swarm Data, <i>J. Dutton</i> , Vol. 4, No. 3, pp. 577-856 (1975) | \$12.00 | 89. Property Index and Author Index to Volumes 1-5 (1972-1976), Vol. 5, No. 4, pp. 1161-1183 | \$4.00 |
| 70. Ideal Gas Thermodynamic Properties and Isomerization of <i>n</i> -Butane and Isobutane, <i>S.S. Chen, R.C. Wilhoit, and B.J. Zwolinski</i> , Vol. 4, No. 4, pp. 859-869 (1975) | \$3.00 | | |
| 71. Molten Salts: Volume 4, Part 2, Chlorides and Mixtures, Electrical Conductance, Density, Viscosity, and Surface Tension Data, <i>G.J. Janz, R.P.T. Tomkins, C.B. Allen, J.R. Downey, Jr., G.L. Gardner, U. Krebs, and S.K. Singer</i> , Vol. 4, No. 4, pp. 871-1178 (1975) | \$13.00 | | |
| 72. Property Index to Volumes 1-4 (1972-1975), Vol. 4, No. 4, pp. 1179-1192 (1975) | \$3.00 | | |

Reprints from Volume 5

| | |
|--|--------|
| 73. Scaled Equation of State Parameters for Gases in the Critical Region, <i>J.M.H. Levelt Sengers, W.L. Greer, and J.V. Sengers</i> , Vol. 5, No. 1, pp. 1-51 (1976) | \$5.00 |
| 74. Microwave Spectra of Molecules of Astrophysical Interest, IX. Acetaldehyde, <i>A. Bauder, F.J. Lovas, and D.R. Johnson</i> , Vol. 5, No. 1, pp. 53-77 (1976) | \$4.00 |
| 75. Microwave Spectra of Molecules of Astrophysical Interest, X. Isocyanic Acid, <i>G. Winnewisser, W.H. Hocking, and M.C.L. Gerry</i> , Vol. 5, No. 1, pp. 79-101 (1976) | \$4.00 |
| 76. Diffusion in Copper and Copper Alloys, Part IV. Diffusion in Systems Involving Elements of Group VIII, <i>Daniel B. Butrymowicz, John R. Manning, and Michael E. Read</i> , Vol. 5, No. 1, pp. 103-200 (1976) | \$6.50 |
| 77. A Critical Review of the Stark Widths and Shifts of Spectral Lines from Non-Hydrogenic Atoms, <i>N. Konjevic and J.R. Roberts</i> , Vol. 5, No. 2, pp. 209-257 (1976) | \$5.00 |
| 78. Experimental Stark Widths and Shifts for Non-Hydrogenic Spectral Lines of Ionized Atoms (A Critical Review and Tabulation of Selected Data), <i>N. Konjevic and W.L. Wiese</i> , Vol. 5, No. 2, pp. 259-308 (1976) | \$5.00 |
| 79. Atlas of the Absorption Spectrum of Nitric Oxide (NO) between 1420 and 1250 Å, <i>E. Miescher and F. Albert</i> , Vol. 5, No. 2, pp. 309-317 (1976) | \$3.00 |
| 80. Ideal Gas Thermodynamic Properties of Propanone and 2-Butanone, <i>Jing Chao and Bruno J. Zwolinski</i> , Vol. 5, No. 2, pp. 319-328 (1976) | \$3.00 |
| 81. Refractive Index of Alkali Halides and Its Wavelength and Temperature Derivatives, <i>H.H. Li</i> , Vol. 5, No. 2, pp. 329-528 (1976) | \$9.50 |

Reprints from Volume 6

| | |
|---|--------|
| 90. Diffusion in Copper and Copper Alloys, Part V. Diffusion in Systems Involving Elements of Group VA, <i>Daniel B. Butrymowicz, John R. Manning, and Michael E. Read</i> , Vol. 6, No. 1, pp. 1-50 (1977) | \$5.00 |
| 91. The Calculated Thermodynamic Properties of Superfluid Helium-4, <i>James S. Brooks and Russell J. Donnelly</i> , Vol. 6, No. 1, pp. 51-104 (1977) | \$5.00 |
| 92. Thermodynamic Properties of Normal and Deuterated Methanols, <i>S.S. Chen, R.C. Wilhoit, and B.J. Zwolinski</i> , Vol. 6, No. 1, pp. 105-112 (1977) | \$3.00 |
| 93. The Spectrum of Molecular Nitrogen, <i>Alf Lofthus and Paul H. Krupenie</i> , Vol. 6, No. 1, pp. 113-307 (1977) | \$9.50 |
| 94. Energy Levels of Chromium, Cr I through Cr XXIV, <i>Jack Sugar and Charles Corliss</i> , Vol. 6, No. 2, pp. 317-383 (1977) | \$5.50 |
| 95. The Activity and Osmotic Coefficients of Aqueous Calcium Chloride at 298.15 K, <i>Bert R. Staples and Ralph L. Nuttall</i> , Vol. 6, No. 2, pp. 385-407 (1977) | \$4.00 |
| 96. Molten Salts: Volume 4, Part 3, Bromides and Mixtures; Iodides and Mixtures-Electrical Conductance, Density, Viscosity, and Surface Tension Data, <i>G.J. Janz, R.P.T. Tomkins, C.B. Allen, J.R. Downey, Jr., and S.K. Singer</i> , Vol. 6, No. 2, pp. 409-596 (1977) | \$9.00 |
| 97. The Viscosity and Thermal Conductivity Coefficients for Dense Gaseous and Liquid Methane, <i>H.J.M. Hanley, W.M. Haynes, and R.D. McCarty</i> , Vol. 6, No. 2, pp. 597-609 (1977) | \$3.00 |
| 98. Phase Diagrams and Thermodynamic Properties of Ternary Copper-Silver Systems, <i>Y. Austin Chang, Daniel Goldberg, and Joachim P. Neumann</i> , Vol. 6, No. 3, pp. 621-673 (1977) | \$5.00 |
| 99. Crystal Data Space-Group Tables, <i>Alan D. Mighell, Helen M. Ondik, and Bettijoyce Breen Molino</i> , Vol. 6, No. 3, pp. 675-829 (1977) | \$8.00 |
| 100. Energy Levels of One-Electron Atoms, <i>Glen W. Erickson</i> , Vol. 6, No. 3, pp. 831-869 (1977) | \$4.50 |
| 101. Rate Constants for Reactions of ClO _x of Atmospheric Interest, <i>R.T. Watson</i> , Vol. 6, No. 3, pp. 871-917 (1977) | \$4.50 |
| 102. NMR Spectral Data: A Compilation of Aromatic Proton Chemical Shifts in Mono- and Di-Substituted Benzenes, <i>B.L. Shapiro and L.E. Mohrmann</i> , Vol. 6, No. 3, pp. 919-991 (1977) | \$5.50 |

(Continuation of Cumulative Listing of Reprints)

103. Tables of Molecular Vibrational Frequencies. Consolidated Volume II. *T. Shimanouchi*, Vol. 6, No. 3, pp. 993-1102 (1977) (supersedes Nos. 5, 21, 25, 49) \$6.50
104. Effects of Isotopic Composition, Temperature, Pressure, and Dissolved Gases on the Density of Liquid Water, *George S. Kell*, Vol. 6, No. 4, pp. 1109-1131 (1977) \$4.00
105. Viscosity of Water Substance—New International Formulation and Its Background, *A. Nagashima*, Vol. 6, No. 4, pp. 1133-1166 (1977) \$4.50
106. A Correlation of the Existing Viscosity and Thermal Conductivity Data of Gaseous and Liquid Ethane, *H.J.M. Hanley, K.E. Gubbins, and S. Murad*, Vol. 6, No. 4, pp. 1167-1180 (1977) \$3.00
107. Elastic Properties of Zinc: A Compilation and a Review, *H.M. Ledbetter*, Vol. 6, No. 4, pp. 1181-1203 (1977) \$4.00
108. Behavior of the AB-Type Compounds at High Pressures and High Temperatures, *Leo Merrill*, Vol. 6, No. 4, pp. 1205-1252 (1977) \$4.50
109. Energy Levels of Manganese, Mn I through Mn XXV, *Charles Corliss and Jack Sugar*, Vol. 6, No. 4, pp. 1253-1329 (1977) \$5.50
123. Thermal Conductivity of Ten Selected Binary Alloy Systems, *C.Y. Ho, M.W. Ackerman, K.Y. Wu, S.G. Oh, and T.N. Havill*, Vol. 7, No. 3, pp. 959-1177 (1978) \$10.00
124. Semi-Empirical Extrapolation and Estimation of Rate Constants for Abstraction of H from Methane by H, O, HO, and O₂, *Robert Shaw*, Vol. 7, No. 3, pp. 1179-1190 (1978) \$3.00
125. Energy Levels of Vanadium, V I through V XXIII, *Jack Sugar and Charles Corliss*, Vol. 7, No. 3, pp. 1191-1262 (1978) \$5.50
126. Recommended Atomic Electron Binding Energies, 1s to 6p_{3/2}, for the Heavy Elements, Z = 84 to 103, *F.T. Porter and M.S. Freedman*, Vol. 7, No. 4, pp. 1267-1284 (1978) \$4.00
127. Ideal Gas Thermodynamic Properties of CH_{4-(a+b+c+d)}F_aCl_bBr_cI_d Halomethanes, *Shanti A. Kudchadker and Arvind P. Kudchadker*, Vol. 7, No. 4, pp. 1285-1307 (1978) \$4.00
128. Critical Review of Vibrational Data and Force Field Constants for Polyethylene, *John Barnes and Bruno Fanconi*, Vol. 7, No. 4, pp. 1309-1321 (1978) \$3.00
129. Tables of Molecular Vibrational Frequencies, Part 9, *Takehiko Shimanouchi, Hiroatsu Matsuura, Yoshiki Ogawa, and Issei Harada*, Vol. 7, No. 4, pp. 1323-1443 (1978) \$7.00

130. Microwave Spectral Tables. II. Triatomic Molecules, *Frank J. Lovas*, Vol. 7, No. 4, pp. 1445-1750 (1978) \$13.00

Reprints from Volume 7

110. Tables of Atomic Spectral Lines for the 10 000 Å to 40 000 Å Region, *Michael Outred*, Vol. 7, No. 1, pp. 1-262 (1978) \$11.50
111. Evaluated Activity and Osmotic Coefficients for Aqueous Solutions: The Alkaline Earth Metal Halides, *R.N. Goldberg and R.L. Nuttall*, Vol. 7, No. 1, pp. 263-310 (1978) \$4.50
112. Microwave Spectra of Molecules of Astrophysical Interest XII. Hydroxyl Radical, *Robert A. Beaudet and Robert L. Poynter*, Vol. 7, No. 1, pp. 311-362 (1978) \$5.00
113. Ideal Gas Thermodynamic Properties of Methanoic and Ethanoic Acids, *Jing Chao and Bruno J. Zwolinski*, Vol. 7, No. 1, pp. 363-377 (1978) \$3.00
114. Critical Review of Hydrolysis of Organic Compounds in Water Under Environmental Conditions, *W. Mabey and T. Mill*, Vol. 7, No. 2, pp. 383-415 (1978) \$4.50
115. Ideal Gas Thermodynamic Properties of Phenol and Creosols, *S.A. Kudchadker, A.P. Kudchadker, R.C. Wilhoit, and B.J. Zwolinski*, Vol. 7, No. 2, pp. 417-423 (1978) \$3.00
116. Densities of Liquid CH_{4-a}X_a (X = Br, I) and CH_{4-(a+b+c+d)}F_aCl_bBr_cI_d Halomethanes, *A.P. Kudchadker, S.A. Kudchadker, P.R. Patnaik, and P.P. Mishra*, Vol. 7, No. 2, pp. 425-439 (1978) \$3.00
117. Microwave Spectra of Molecules of Astrophysical Interest XIII. Cyanoacetylene, *W.J. Lafferty and F.J. Lovas*, Vol. 7, No. 2, pp. 441-493 (1978) \$5.00
118. Atomic Transition Probabilities for Vanadium, Chromium, and Manganese (A Critical Data Compilation of Allowed Lines), *S.M. Younger, J.R. Fuhr, G.A. Martin, and W.L. Wiese*, Vol. 7, No. 2, pp. 495-629 (1978) \$7.50
119. Thermodynamic Properties of Ammonia, *Lester Haar and John S. Gallagher*, Vol. 7, No. 3, pp. 635-792 (1978) \$8.00
120. JANAF Thermochemical Tables, 1978 Supplement, *M.W. Chase, Jr., J.L. Curnutt, R.A. McDonald, and A.N. Syverud*, Vol. 7, No. 3, pp. 793-940 (1978) \$8.00
121. Viscosity of Liquid Water in the Range -8°C to 150°C, *Joseph Kestin, Mordechai Sokolov, and William A. Wakeham*, Vol. 7, No. 3, pp. 941-948 (1978) \$3.00
122. The Molar Volume (Density) of Solid Oxygen in Equilibrium with Vapor, *H.M. Roder*, Vol. 7, No. 3, pp. 949-957 (1978) \$3.00

Reprints from Volume 8

131. Energy Levels of Titanium, Ti I through Ti XXII, *Charles Corliss and Jack Sugar*, Vol. 8, No. 1, pp. 1-62 (1979) \$5.00
132. The Spectrum and Energy Levels of the Neutral Atom of Boron (B I), *G.A. Odintzova and A.R. Strganov*, Vol. 8, No. 1, pp. 63-67 (1979) \$3.00
133. Relativistic Atomic Form Factors and Photon Coherent Scattering Cross Sections, *J.H. Hubbell and I. Øverbø*, Vol. 8, No. 1, pp. 69-105 (1979) \$4.50
134. Microwave Spectra of Molecules of Astrophysical Interest. XIV. Vinyl Cyanide (Acrylonitrile), *M.C.L. Gerry, K. Yamada, and G. Winnewisser*, Vol. 8, No. 1, pp. 107-123 (1979) \$4.00
135. Molten Salts: Volume 4, Part 4, Mixed Halide Melts. Electrical Conductance, Density, Viscosity, and Surface Tension Data, *G.J. Janz, R.P.T. Tomkins, and C.B. Allen*, Vol. 8, No. 1, pp. 125-302 (1979) \$9.00
136. Atomic Radiative and Radiationless Yields for K and L Shells, *M.O. Krause*, Vol. 8, No. 2, pp. 307-327 (1979) \$4.00
137. Natural Widths of Atomic K and L Levels, K α X-ray Lines and Several KLL Auger Lines, *M.O. Krause and J.H. Oliver*, Vol. 8, No. 2, pp. 329-338 (1979) \$3.00
138. Electrical Resistivity of Alkali Elements, *T.C. Chi*, Vol. 8, No. 2, pp. 339-438 (1979) \$6.50
139. Electrical Resistivity of Alkaline Earth Elements, *T.C. Chi*, Vol. 8, No. 2, pp. 439-497 (1979) \$5.00
140. Vapor Pressures and Boiling Points of Selected Halomethanes, *A.P. Kudchadker, S.A. Kudchadker, R.P. Shukla, and P.R. Patnaik*, Vol. 8, No. 2, pp. 499-517 (1979) \$4.00
141. Ideal Gas Thermodynamic Properties of Selected Bromoethanes and Iodoethane, *S.A. Kudchadker and A.P. Kudchadker*, Vol. 8, No. 2, pp. 519-526 (1979) \$3.00
142. Thermodynamic Properties of Normal and Deuterated Naphthalenes, *S.S. Chen, S.A. Kudchadker, and R.C. Wilhoit*, Vol. 8, No. 2, pp. 527-535 (1979) \$3.00

(Continuation of Cumulative Listing of Reprints)

143. Microwave Spectra of Molecules of Astrophysical Interest. XV. Propyne, *A. Bauer, D. Boucher, J. Burne, J. Demaison, and A. Dubrulle*, Vol. 8, No. 2, pp. 537-558 (1979) \$4.00
144. A Correlation of the Viscosity and Thermal Conductivity Data of Gaseous and Liquid Propane, *P.M. Holland, H.J.M. Hanley, K.E. Gubbins, and J.M. Haile*, Vol. 8, No. 2, pp. 559-575 (1979) \$4.00
145. Microwave Spectra of Molecules of Astrophysical Interest. XVI. Methyl Formate, *A. Bauder*, Vol. 8, No. 3, pp. 583-618 (1979) \$4.50
146. Molecular Structures of Gas-Phase Polyatomic Molecules Determined by Spectroscopic Methods, *Marlin D. Harmony, Victor W. Laurie, Robert L. Kuczkowski, R.H. Schwendeman, D.A. Ramsay, Frank J. Lovas, Walter J. Lafferty, and Arthur G. Maki*, Vol. 8, No. 3, pp. 619-721 (1979) \$6.50
147. Critically Evaluated Rate Constants for Gaseous Reactions of Several Electronically Excited Species, *Keith Schofield*, Vol. 8, No. 3, pp. 723-798 (1979) \$5.50
148. A Review, Evaluation, and Correlation of the Phase Equilibria, Heat of Mixing, and Change in Volume on Mixing for Liquid Mixtures of Methane + Ethane, *M.J. Hiza, R.C. Miller, and A.J. Kidnay*, Vol. 8, No. 3, pp. 799-816 (1979) \$4.00
149. Energy Levels of Aluminum, Al_I through Al_{XIII}, *W.C. Martin and Romuald Zalubas*, Vol. 8, No. 3, pp. 817-864 (1979) \$4.50
150. Energy Levels of Calcium, Ca_I through Ca_{XX}, *Jack Sugar and Charles Corliss*, Vol. 8, No. 3, pp. 865-916 (1979) \$5.00
151. Evaluated Activity and Osmotic Coefficients for Aqueous Solutions: Iron Chloride and the Bi-univalent Compounds of Nickel and Cobalt, *R.N. Goldberg, R.L. Nutall, and B.R. Staples*, Vol. 8, No. 4, pp. 923-1003 (1979) \$6.00
152. Evaluated Activity and Osmotic Coefficients for Aqueous Solutions: Bi-univalent Compounds of Lead, Copper, Manganese, and Uranium, *R.N. Goldberg*, Vol. 8, No. 4, pp. 1005-1050 (1979) \$4.50
153. Microwave Spectra of Molecules of Astrophysical Interest. XVII. Dimethyl Ether, *F.J. Lovas, H. Lutz, and H. Dreizler*, Vol. 8, No. 4, pp. 1051-1107 (1979) \$5.00
154. Energy Levels of Potassium, K_I through K_{XIX}, *Charles Corliss and Jack Sugar*, Vol. 8, No. 4, pp. 1109-1145 (1979) \$4.50
155. Electrical Resistivity of Copper, Gold, Palladium, and Silver, *R.A. Matula*, Vol. 8, No. 4, pp. 1147-1298 (1979) \$8.00
161. A Compilation of Kinetic Parameters for the Thermal Degradation of *n*-Alkane Molecules, *D.L. Allara and Robert Shaw*, Vol. 9, No. 3, pp. 523-559 (1980) \$5.50
162. Refractive Index of Silicon and Germanium and Its Wavelength and Temperature Derivatives, *H.H. Li*, Vol. 9, No. 3, pp. 561-658 (1980) \$7.50
163. Microwave Spectra of Molecules of Astrophysical Interest XIX. Methyl Cyanide, *D. Boucher, J. Burne, A. Bauer, A. Dubrulle, and J. Demaison*, Vol. 9, No. 3, pp. 659-719 (1980). \$6.00
164. A Review, Evaluation, and Correlation of the Phase Equilibria, Heat of Mixing, and Change in Volume on Mixing for Liquid Mixtures of Methane + Propane, *R.C. Miller, A.J. Kidnay, and M.J. Hiza*, Vol. 9, No. 3, pp. 721-734 (1980) \$4.00
165. Saturation States of Heavy Water, *P.G. Hill and R.D. Chris MacMillan*, Vol. 9, No. 3, pp. 735-749 (1980) \$4.00
166. The Solubility of Some Sparingly Soluble Lead Salts: An Evaluation of the Solubility in Water and Aqueous Electrolyte Solution, *H. Lawrence Clever and Francis J. Johnston*, Vol. 9, No. 3, pp. 751-784 (1980) \$5.50
167. Molten Salts Data as Reference Standards for Density, Surface Tension, Viscosity, and Electrical Conductance: KNO₃ and NaCl, *George J. Janz*, Vol. 9, No. 4, pp. 791-829 (1980) \$5.50
168. Molten Salts: Volume 5, Part 1, Additional Single and Multi-Component Salt Systems. Electrical Conductance, Density, Viscosity, and Surface Tension Data, *G.J. Janz and R.P. Tomkins*, Vol. 9, No. 4, pp. 831-1021 (1980) \$10.50
169. Pair, Triplet, and Total Atomic Cross Sections (and Mass Attenuation Coefficients) for 1 MeV-100 GeV Photons in Elements *Z* = 1 to 100, *J.H. Hubbell, H.A. Gimm, and I. Øverbø*, Vol. 9, No. 4, pp. 1023-1147 (1980) \$8.00
170. Tables of Molecular Vibrational Frequencies, Part 10, *Takehiko Shimanouchi, Hiroatsu Matsuura, Yoshiki Ogawa, and Issei Harada*, Vol. 9, No. 4, pp. 1149-1254 (1980) \$7.50
171. An Improved Representative Equation for the Dynamic Viscosity of Water Substance, *J.T.R. Watson, R.S. Basu, and J.V. Sengers*, Vol. 9, No. 4, pp. 1255-1290 (1980) \$5.50
172. Static Dielectric Constant of Water and Steam, *M. Uematsu and E. U. Franck*, Vol. 9, No. 4, pp. 1291-1306 (1980) \$4.00
173. Compilation and Evaluation of Solubility Data in the Mercury (I) Chloride-Water System, *Y. Marcus*, Vol. 9, No. 4, pp. 1307-1329 (1980) \$5.00

Reprints from Volume 9

156. Energy Levels of Magnesium, Mg_I through Mg_{XII}, *W.C. Martin and Romuald Zalubas*, Vol. 9, No. 1, pp. 1-58 (1980) \$6.00
157. Microwave Spectra of Molecules of Astrophysical Interest. XVIII. Formic Acid, *Edmond Willemot, Didier Dangois, Nicole Monnanteuil, and Jean Bellet*, Vol. 9, No. 1, pp. 59-160 (1980) \$7.50
158. Refractive Index of Alkaline Earth Halides and Its Wavelength and Temperature Derivatives, *H.H. Li*, Vol. 9, No. 1, pp. 161-289 (1980). \$8.50
159. Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry, *D.L. Baulch, R.A. Cox, R.F. Hampson, Jr., J.A. Kerr, J. Troe, and R.L. Watson*, Vol. 9, No. 2, pp. 295-471 (1980) \$10.00
160. Energy Levels of Scandium, Sc_I through Sc_{XXII}, *Jack Sugar and Charles Corliss*, Vol. 9, No. 2, pp. 473-511 (1980) \$5.50

Reprints from Volume 10

174. Evaluated Activity and Osmotic Coefficients for Aqueous Solutions: Bi-Univalent Compounds of Zinc, Cadmium, and Ethylene Bis(Trimethylammonium) Chloride and Iodide, *R. N. Goldberg*, Vol. 10, No. 1, pp. 1-55 (1981) \$6.00
175. Tables of the Dynamic and Kinematic Viscosity of Aqueous KCl Solutions in the Temperature Range 25-150°C and the Pressure Range 0.1-35 MPa, *Joseph Kestin, H. Ezzat Khalifa, and Robert J. Correia*, Vol. 10, No. 1, pp. 57-70 (1981) \$4.00
176. Tables of the Dynamic and Kinematic Viscosity of Aqueous NaCl Solutions in the Temperature Range 20-150°C and the Pressure Range 0.1-35 MPa, *Joseph Kestin, H. Ezzat Khalifa, and Robert J. Correia*, Vol. 10, No. 1, pp. 71-87 (1981) \$5.00
177. Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. I. Selenium, *Umesh Gaur, Hua-Cheng Shu, Aspy Mehta, and Bernhard Wunderlich*, Vol. 10, No. 1, pp. 89-117 (1981) \$5.00

(Continuation of Cumulative Listing of Reprints)

178. Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. II. Polyethylene, *Umesh Gaur and Bernhard Wunderlich*, Vol. 10, No. 1, pp. 119–152 (1981) \$5.50
179. Energy Levels of Sodium, Na I through Na XI, *W. C. Martin and Romuald Zalubas*, Vol. 10, No. 1, pp. 153–195 (1981) \$5.50
180. Energy Levels of Nickel, Ni I through Ni XXVIII, *Charles Corliss and Jack Sugar*, Vol. 10, No. 1, pp. 197–289 (1981) \$7.00
181. Ion Product of Water Substance, 0–1000 °C, 1–10,000 bars New International Formulation and Its Background, *William L. Marshall and E. U. Franck*, Vol. 10, No. 2, pp. 295–304 (1981) \$4.00
182. Atomic Transition Probabilities for Iron, Cobalt, and Nickel (A Critical Data Compilation of Allowed Lines), *J. R. Fuhr, G. A. Martin, W. L. Wiese, and S. M. Younger*, Vol. 10, No. 2, pp. 305–565 (1981) \$12.50
183. Thermodynamic Tabulations for Selected Phases in the System CaO-Al₂O₃-SiO₂-H₂O at 101.325 kPa (1 atm) between 273.15 and 1800 K, *John L. Haas, Jr., Gilpin R. Robinson, Jr., and Bruce S. Hemingway*, Vol. 10, No. 3, pp. 575–669 (1981) \$7.00
184. Evaluated Activity and Osmotic Coefficients for Aqueous Solutions: Thirty-Six Uni-Bivalent Electrolytes, R. N. Goldberg, Vol. 10, No. 3, pp. 671–764 (1981) \$7.00
185. Activity and Osmotic Coefficients of Aqueous Alkali Metal Nitrites, *Bert R. Staples*, Vol. 10, No. 3, pp. 765–778 (1981) \$4.00
186. Activity and Osmotic Coefficients of Aqueous Sulfuric Acid at 298.15 K, *Bert R. Staples*, Vol. 10, No. 3, pp. 779–798 (1981) \$5.00
187. Rate Constants for the Decay and Reactions of the Lowest Electronically Excited Singlet State of Molecular Oxygen in Solution, *Francis Wilkinson and James G. Brummer*, Vol. 10, No. 4, pp. 809–999 (1981) \$10.00
188. Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. III. Polyoxides, *Umesh Gaur and Bernhard Wunderlich*, Vol. 10, No. 4, pp. 1001–1049 (1981) \$5.50
189. Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. IV. Polypropylene, *Umesh Gaur and Bernhard Wunderlich*, Vol. 10, No. 4, pp. 1051–1064 (1981) \$4.00
190. Tables of N₂O Absorption Lines for the Calibration of Tunable Infrared Lasers from 522 cm⁻¹ to 657 cm⁻¹ and from 1115 cm⁻¹ to 1340 cm⁻¹, *W. B. Olson, A. G. Maki, and W. J. Lafferty*, Vol. 10, No. 4, pp. 1065–1084 (1981) \$5.00
191. Microwave Spectra of Molecules of Astrophysical Interest. XX. Methane, *I. Ozier, M. C. L. Gerry, and A. G. Robiette*, Vol. 10, No. 4, pp. 1085–1095 (1981) \$4.00
192. Energy Levels of Cobalt, Co I through Co XXVII, *Jack Sugar and Charles Corliss*, Vol. 10, No. 4, pp. 1097–1174 (1981) \$6.50
193. A Critical Review of Henry's Law Constants for Chemicals of Environmental Interest, *Donald Mackay and Wan Ying Shiu*, Vol. 10, No. 4, pp. 1175–1199 (1981) \$5.00
194. Property, Materials, and Author Indexes to the Journal of Physical and Chemical Reference Data, Vol. 1–10, pp. 1205–1225 (1972–1981) \$5.00
197. Ideal Gas Thermodynamic Properties of CH₃, CD₃, CD₄, C₂D₂, C₂D₄, C₂D₆, C₂H₆, CH₃N₂CH₃, and CD₃N₂CD₃, *Krishna M. Pamidimukkala, David Rogers, and Gordon B. Skinner*, Vol. 11, No. 1, pp. 83–99 (1982) \$6.00
198. Peak Absorption Coefficients of Microwave Absorption Lines of Carbonyl Sulphide, *Z. Kisiel and D. J. Millen*, Vol. 11, No. 1, pp. 99–116 (1982) \$6.00
199. Vibrational Contributions to Molecular Dipole Polarizabilities, *David M. Bishop and Lap M. Cheung*, Vol. 11, No. 1, pp. 119–133 (1982) \$5.00
200. Energy Levels of Iron, Fe I through Fe XXVI, *Charles Corliss and Jack Sugar*, Vol. 11, No. 1, pp. 135–241 (1982) \$11.00
201. Microwave Spectra of Molecules of Astrophysical Interest. XXI. Ethanol (C₂H₅OH) and Propionitrile (C₂H₅CN), *Frank J. Lovas*, Vol. 11, No. 2, pp. 251–312 (1982) \$8.00
202. Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. V. Polystyrene, *Umesh Gaur and Bernhard Wunderlich*, Vol. 11, No. 2, pp. 313–325 (1982) \$5.00
203. Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry: Supplement 1, CODATA Task Group on Chemical Kinetics, *D. L. Baulch, R. A. Cox, P. J. Crutzen, R. F. Hampson, Jr., J. A. Kerr (Chairman), J. Troe, and R. T. Watson*, Vol. 11, No. 2, pp. 327–496 (1982) \$15.00
204. Molten Salts Data: Diffusion Coefficients in Single and Multi-Component Salt Systems, *G. J. Janz and N. P. Bansal*, Vol. 11, No. 3, pp. 505–693 (1982) \$16.00
205. JANAF Thermochemical Tables, 1982 Supplement, *M. W. Chase, Jr., J. L. Curnutt, J. R. Downey, Jr., R. A. McDonald, A. N. Syverud, and E. A. Valenzuela*, Vol. 11, No. 3, pp. 695–940 (1982) \$20.00
206. Critical Evaluation of Vapor-Liquid Equilibrium, Heat of Mixing, and Volume Change of Mixing Data. General Procedures, *Buford D. Smith, OI Muthu, Ashok Dewan, and Matthew Gierlach*, Vol. 11, No. 3, pp. 941–951 (1982) \$5.00
207. Rate Coefficients for Vibrational Energy Transfer Involving the Hydrogen Halides, *Stephen R. Leone*, Vol. 11, No. 3, pp. 953–996 (1982) \$7.00
208. Behavior of the AB₂-Type Compounds at High Pressures and High Temperatures, *Leo Merrill*, Vol. 11, No. 4, pp. 1005–1064 (1982) \$8.00
209. Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. VI. Acrylic Polymers, *Umesh Gaur, Suk-fai Lau, Brent B. Wunderlich, and Bernhard Wunderlich*, Vol. 11, No. 4, pp. 1065–1089 (1982) \$6.00
210. Molecular Form Factors and Photon Coherent Scattering Cross Sections of Water, *L. R. M. Morin*, Vol. 11, No. 4, pp. 1091–1098 (1982) \$5.00
211. Evaluation of Binary P_{Txy} Vapor-Liquid Equilibrium Data for C₆ Hydrocarbons. Benzene + Cyclohexane, *Buford D. Smith, OI Muthu, Ashok Dewan, and Matthew Gierlach*, Vol. 11, No. 4, pp. 1099–1126 (1982) \$6.00
212. Evaluation of Binary Excess Enthalpy Data for C₆ Hydrocarbons. Benzene + Cyclohexane, *Buford D. Smith, OI Muthu, Ashok Dewan, and Matthew Gierlach*, Vol. 11, No. 4, pp. 1127–1149 (1982) \$6.00
213. Evaluation of Binary Excess Volume Data for C₆ Hydrocarbons. Benzene + Cyclohexane, *Buford D. Smith, OI Muthu, Ashok Dewan, and Matthew Gierlach*, Vol. 11, No. 4, pp. 1151–1169 (1982) \$6.00

Reprints from Volume 11

195. A Fundamental Equation of State for Heavy Water, *P. G. Hill, R. D. Chris MacMillan, and V. Lee*, Vol. 11, No. 1, pp. 1–14 (1982) \$5.00
196. Volumetric Properties of Aqueous Sodium Chloride Solutions, *P. S. Z. Rogers and Kenneth S. Pitzer*, Vol. 11, No. 1, pp. 15–81 (1982) \$9.00

Reprints from Volume 12

214. Thermodynamic Properties of Steam in the Critical Region, *J. M. H. Levelt Sengers, B. Kamgar-Parsi, F. W. Balfour, and J. V. Sengers*, Vol. 12, No. 1, pp. 1–28 (1983) \$6.00

(Continuation of Cumulative Listing of Reprints)

- 215.Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. VII. Other Carbon Backbone Polymers, *Umesh Gaur, Brent B. Wunderlich, and Bernhard Wunderlich*, Vol. 12, No. 1, pp. 29-63 (1983) \$7.00
- 216.Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. VIII. Polyesters and Polyamides, *Umesh Gaur, Suk-fai Lau, Brent B. Wunderlich, and Bernhard Wunderlich*, Vol. 12, No. 1, pp. 65-89 (1983) \$6.00
- 217.Heat Capacity and Other Thermodynamic Properties of Linear Macromolecules. IX. Final Group of Aromatic and Inorganic Polymers, *Umesh Gaur, Suk-fai Lau, and Bernhard Wunderlich*, Vol. 12, No. 1, pp. 91-108 (1983) \$6.00
- 218.An Annotated Compilation and Appraisal of Electron Swarm Data in Electronegative Gases, *J. W. Gallagher, E. C. Beatty, J. Dutton, and L. C. Pitchford*, Vol. 12, No. 1, pp. 109-152 (1983) \$7.00
- 219.The Solubility of Oxygen and Ozone in Liquids, *Rubin Battino, Timothy R. Rettich, and Toshihiro Tominaga*, Vol. 12, No. 2, pp. 163-178 (1983) \$5.00
- 220.Recommended Values for the Thermal Expansivity of Silicon from 0 to 1000 K, *C. A. Swenson*, Vol. 12, No. 2, pp. 179-182 (1983) \$5.00
- 221.Electrical Resistivity of Ten Selected Binary Alloy Systems, *C. Y. Ho, M. W. Ackerman, K. Y. Wu, T. N. Havill, R. H. Bogaard, R. A. Matula, S. G. Oh, and H. M. James*, Vol. 12, No. 2, pp. 183-322 (1983) \$13.00
- 222.Energy Levels of Silicon, Si I through Si XIV, *W. C. Martin and Romuald Zalubas*, Vol. 12, No. 2, pp. 323-380 (1983) \$8.00
- 223.Evaluation of Binary PT_{xy} Vapor-Liquid Equilibrium Data for C_6 Hydrocarbons. Benzene + Hexane, *Buford D. Smith, Ol Muthu, and Ashok Dewan*, Vol. 12, No. 2, pp. 381-387 (1983) \$5.00
- 224.Evaluation of Binary Excess Enthalpy Data for C_6 Hydrocarbons. Benzene + Hexane, *Buford D. Smith, Ol Muthu, and Ashok Dewan*, Vol. 12, No. 2, pp. 389-393 (1983) \$5.00
- 225.Evaluation of Binary Excess Volume Data for C_6 Hydrocarbons. Benzene + Hexane, *Buford D. Smith, Ol Muthu, and Ashok Dewan*, Vol. 12, No. 2, pp. 395-401 (1983) \$5.00
- 226.Atlas of the High-Temperature Water Vapor Spectrum in the 3000 to 4000 cm^{-1} Region, *A. S. Pine, M. J. Coulombe, C. Camy-Peyret, and J-M. Flaud*, Vol. 12, No. 3, pp. 413-465 (1983) \$8.00
- 227.Small-Angle Rayleigh Scattering of Photons at High Energies: Tabulations of Relativistic HFS Modified Atomic Form Factors, *D. Schuapp, M. Schumacher, F. Smend, P. Rullhusen, and J. H. Hubbell*, Vol. 12, No. 3, pp. 467-512 (1983) \$7.00
- 228.Thermodynamic Properties of D_2O in the Critical Region, *B. Kamgar-Parsi, J. M. H. Levelt Sengers, and J. V. Sengers*, Vol. 12, No. 3, pp. 513-529 (1983) \$6.00
- 229.Chemical Kinetic Data Sheets for High-Temperature Chemical Reactions, *N. Cohen and K. R. Westberg*, Vol. 12, No. 3, pp. 531-590 (1983) \$8.00
- 230.Molten Salts: Volume 5, Part 2. Additional Single and Multi-Component Salt Systems. Electrical Conductance, Density, Viscosity and Surface Tension Data, *G. J. Janz and R. P. T. Tomkins*, Vol. 12, No. 3, pp. 591-815 (1983) \$19.00
- 231.International Tables of the Surface Tension of Water, *N. B. Vargaftik, B. N. Volkov, and L. D. Voljak*, Vol. 12, No. 3, pp. 817-820 (1983) \$5.00
- 232.Evaluated Theoretical Cross Section Data for Charge Exchange of Multiply Charged Ions with Atoms. I. Hydrogen Atom-Fully Stripped Ion Systems, *R. K. Janev, B. H. Bransden, and J. W. Gallagher*, Vol. 12, No. 4, pp. 829-872 (1983) \$7.00
- 233.Evaluated Theoretical Cross Section Data for Charge Exchange of Multiply Charged Ions with Atoms. II. Hydrogen Atom-Partially Stripped Ion Systems, *J. W. Gallagher, B. H. Bransden, and R. K. Janev*, Vol. 12, No. 4, pp. 873-890 (1983) \$6.00
- 234.Recommended Data on the Electron Impact Ionization of Light Atoms and Ions, *K. L. Bell, H. B. Gilbody, J. G. Hughes, A. E. Kingston, and F. J. Smith*, Vol. 12, No. 4, pp. 891-916 (1983) \$6.00
- 235.A Correlation of the Viscosity and Thermal Conductivity Data of Gaseous and Liquid Ethylene, *P. M. Holland, B. E. Eaton, and H. J. M. Hanley*, Vol. 12, No. 4, pp. 917-932 (1983) \$5.00
- 236.Transport Properties of Liquid and Gaseous D_2O over a Wide Range of Temperature and Pressure, *N. Matsunaga and A. Nagashima*, Vol. 12, No. 4, pp. 933-966 (1983) \$7.00
- 237.Thermochemical Data for Gaseous Monoxides, *J. B. Pedley and E. M. Marshall*, Vol. 12, No. 4, pp. 967-1031 (1983) \$9.00
- 238.Vapor Pressure of Coal Chemicals, *J. Chao, C. T. Lin, and T. H. Chung*, Vol. 12, No. 4, pp. 1033-1063 (1983) \$6.00

Reprints from Volume 13

- 239.Thermodynamic Properties of Aqueous Sodium Chloride Solutions, *Kenneth S. Pitzer, J. Christopher Peiper, and R. H. Busey*, Vol. 13, No. 1, pp. 1-102 (1984) \$11.00
- 240.Refractive Index of ZnS, ZnSe, and ZnTe and Its Wavelength and Temperature Derivatives, *H. H. Li*, Vol. 13, No. 1, pp. 103-150 (1984) \$7.00
- 241.High Temperature Vaporization Behavior of Oxides. I. Alkali Metal Binary Oxides, *R. H. Lamoreaux and D. L. Hildenbrand*, Vol. 13, No. 1, pp. 151-173 (1984) \$6.00
- 242.Thermophysical Properties of Fluid H_2O , *J. Kestin, J. V. Sengers, B. Kamgar-Parsi, and J. M. H. Levelt Sengers*, Vol. 13, No. 1, pp. 175-183 (1984) \$5.00
- 243.Representative Equations for the Viscosity of Water Substance, *J. V. Sengers and B. Kamgar-Parsi*, Vol. 13, No. 1, pp. 185-205 (1984) \$6.00
- 244.Atlas of the Schumann-Runge Absorption Bands of O_2 in the Wavelength Region 175-205 nm, *K. Yoshino, D. E. Freeman, and W. H. Parkinson*, Vol. 13, No. 1, pp. 207-227 (1984) \$6.00
- 245.Equilibrium and Transport Properties of the Noble Gases and Their Mixtures at Low Density, *J. Kestin, K. Knierim, E. A. Mason, B. Najafi, S. T. Ro, and M. Waldman*, Vol. 13, No. 1, pp. 229-303 (1984) \$9.00
- 246.Evaluation of Kinetic and Mechanistic Data For Modeling of Photochemical Smog, *Roger Atkinson and Alan C. Lloyd*, Vol. 13, No. 2, pp. 315-444 (1984) \$13.00
- 247.Rate Data for Inelastic Collision Processes in the Diatomic Halogen Molecules, *J. I. Steinfeld*, Vol. 13, No. 2, pp. 445-553 (1984) \$11.00
- 248.Water Solubilities of Polynuclear Aromatic and Heteroaromatic Compounds, *Robert S. Pearlman, Samuel H. Yalkowsky, and Sujit Banerjee*, Vol. 13, No. 2, pp. 555-562 (1984) \$5.00
- 249.The Solubility of Nitrogen and Air in Liquids, *Rubin Battino, Timothy R. Rettich, and Toshihiro Tominaga*, Vol. 13, No. 2, pp. 563-600 (1984) \$7.00
- 250.Thermophysical Properties of Fluid D_2O , *J. Kestin, J. V. Sengers, B. Kamgar-Parsi, and J. M. H. Levelt Sengers*, Vol. 13, No. 2, pp. 601-609 (1984) \$5.00
- 251.Experimental Stark Widths and Shifts for Spectral Lines of Neutral Atoms (A Critical Review of Selected Data for the Period 1976 to 1982), *N. Konjević, M. S. Dimitrijević, and W. L. Wiese*, Vol. 13, No. 3, pp. 619-647 (1984) \$6.00

(Continuation of Cumulative Listing of Reprints)

252. Experimental Stark Widths and Shifts for Spectral Lines of Positive Ions (A Critical Review and Tabulation of Selected Data for the Period 1976 to 1982), *N. Konjević, M. S. Dimitrijević, and W. L. Wiese*, Vol. 13, No. 3, pp. 649–686 (1984) \$7.00
253. A Review of Deuterium Triple-Point Temperatures, *L. A. Schwalbe and E. R. Grilly*, Vol. 13, No. 3, pp. 687–693 (1984) \$5.00
254. Evaluated Gas Phase Basicities and Proton Affinities of Molecules; Heats of Formation of Protonated Molecules, *Sharon G. Lias, Joel F. Liebman, and Rhoda D Levin*, Vol. 13, No. 3, pp. 695–808 (1984) \$12.00
255. Isotopic Abundances and Atomic Weights of the Elements, *Paul De Bièvre, Marc Gallet, Norman E. Holden, and I. Lynus Barnes*, Vol. 13, No. 3, pp. 809–891 (1984) \$10.00
256. Representative Equations for the Thermal Conductivity of Water Substance, *J. V. Sengers, J. T. R. Watson, R. S. Basu, B. Kamgar-Parsi, and R. C. Hendricks*, Vol. 13, No. 3, pp. 893–933 (1984) \$7.00
257. Ground-State Vibrational Energy Levels of Polyatomic Transient Molecules, *Marilyn E. Jacox*, Vol. 13, No. 4, pp. 945–1068 (1984) \$12.00
258. Electrical Resistivity of Selected Elements, *P. D. Desai, T. K. Chu, H. M. James, and C. Y. Ho*, Vol. 13, No. 4, pp. 1069–1096 (1984) \$6.00
259. Electrical Resistivity of Vanadium and Zirconium, *P. D. Desai, H. M. James, and C. Y. Ho*, Vol. 13, No. 4, pp. 1097–1130 (1984) \$7.00
260. Electrical Resistivity of Aluminum and Manganese, *P. D. Desai, H. M. James, and C. Y. Ho*, Vol. 13, No. 4, pp. 1131–1172 (1984) \$7.00
261. Standard Chemical Thermodynamic Properties of Alkane Isomer Groups, *Robert A. Alberty and Catherine A. Gehrig*, Vol. 13, No. 4, pp. 1173–1197 (1984) \$6.00
262. Evaluated Theoretical Cross-Section Data for Charge Exchange of Multiply Charged Ions with Atoms. III. Nonhydrogenic Target Atoms, *R. K. Janev and J. W. Gallagher*, Vol. 13, No. 4, pp. 1199–1249 (1984) \$8.00
263. Heat Capacity of Reference Materials: Cu and W, *G. K. White and S. J. Collocott*, Vol. 13, No. 4, pp. 1251–1257 (1984) \$5.00
264. Evaluated Kinetic and Photochemical Data for Atmospheric Chemistry: Supplement II. CODATA Task Group on Gas Phase Chemical Kinetics, *D. L. Baulch, R. A. Cox, R. F. Hampson, Jr., J. A. Kerr (Chairman), J. Troe, and R. T. Watson*, Vol. 13, No. 4, pp. 1259–1380 (1984) \$12.00
265. Thermodynamic Properties of Key Organic Oxygen Compounds in the Carbon Range C₁ to C₄. Part 1. Properties of Condensed Phases, *Randolph C Wilhoit, Jing Chao, and Kenneth R Hall*, Vol. 14, No. 1, pp. 1–175 (1985) \$15.00
266. Standard Chemical Thermodynamic Properties of Alkylbenzene Isomer Groups, *Robert A. Alberty*, Vol. 14, No. 1, pp. 177–192 (1985) \$5.00
267. Assessment of Critical Parameter Values for H₂O and D₂O, *J. M. H. Levelt Sengers, J. Straub, K. Watanabe, and P. G. Hill*, Vol. 14, No. 1, pp. 193–207 (1985) \$5.00
268. The Viscosity of Nitrogen, Oxygen, and Their Binary Mixtures in the Limit of Zero Density, *Wendy A Cole and William A Wakeham*, Vol. 14, No. 1, pp. 209–226 (1985) \$6.00
269. The Thermal Conductivity of Fluid Air, *K Stephan and A Laesecke*, Vol. 14, No. 1, pp. 227–234 (1985) \$5.00
270. The Electronic Spectrum and Energy Levels of the Deuterium Molecule, *Robert S Freund, James A. Schiavone, and H. M. Crosswhite*, Vol. 14, No. 1, pp. 235–383 (1985) \$14.00
271. Microwave Spectra of Molecules of Astrophysical Interest. XXII. Sulfur Dioxide (SO₂), *F. J. Lovas*, Vol. 14, No. 2, pp. 395–488 (1985) \$10.00
272. Evaluation of the Thermodynamic Functions for Aqueous Sodium Chloride from Equilibrium and Calorimetric Measurements below 154 °C, *E. Colin W. Clarke and David N. Glew*, Vol. 14, No. 2, pp. 489–610 (1985) \$12.00
273. The Mark–Houwink–Sakurada Equation for the Viscosity of Linear Polyethylene, *Herman L. Wagner*, Vol. 14, No. 2, pp. 611–617 (1985) \$5.00
274. The Solubility of Mercury and Some Sparingly Soluble Mercury Salts in Water and Aqueous Electrolyte Solutions, *H. Lawrence Clever, Susan A. Johnson, and M. Elizabeth Derrick*, Vol. 14, No. 3, pp. 631–680 (1985) \$8.00
275. A Review and Evaluation of the Phase Equilibria, Liquid-Phase Heats of Mixing and Excess Volumes, and Gas-Phase PVT Measurements for Nitrogen + Methane, *A. J. Kidnay, R. C. Miller, E. D. Sloan, and M. J. Hiza*, Vol. 14, No. 3, pp. 681–694 (1985) \$5.00
276. The Homogeneous Nucleation Limits of Liquids, *C. T. Avedisian*, Vol. 14, No. 3, pp. 695–729 (1985) \$7.00
277. Binding Energies in Atomic Negative Ions II, *H. Hotop and W. C. Lineberger*, Vol. 14, No. 3, pp. 731–750 (1985) \$6.00
278. Energy Levels of Phosphorus, P I through P xv, *W. C. Martin, Romuald Zalubas, and Arlene Musgrove*, Vol. 14, No. 3, pp. 751–802 (1985) \$8.00
279. Standard Chemical Thermodynamic Properties of Alkene Isomer Groups, *Robert A. Alberty and Catherine A. Gehrig*, Vol. 14, No. 3, pp. 803–820 (1985) \$6.00
280. Standard Chemical Thermodynamic Properties of Alkyl-naphthalene Isomer Groups, *Robert A. Alberty and Theodore M. Bloomstein*, Vol. 14, No. 3, pp. 821–837 (1985) \$6.00
281. Carbon Monoxide Thermophysical Properties from 68 to 1000 K at Pressures to 100 MPa, *Robert D. Goodwin*, Vol. 14, No. 4, pp. 849–932 (1985) \$10.00
282. Refractive Index of Water and Its Dependence on Wavelength, Temperature, and Density, *I. Thormählen, J. Straub, and U. Griggull*, Vol. 14, No. 4, pp. 933–945 (1985) \$5.00
283. Viscosity and Thermal Conductivity of Dry Air in the Gaseous Phase, *K. Kadoya, N. Matsunaga, and A. Nagashima*, Vol. 14, No. 4, pp. 947–970 (1985) \$6.00
284. Charge Transfer of Hydrogen Ions and Atoms in Metal Vapors, *T. J. Morgan, R. E. Olson, A. S. Schlachter, and J. W. Gallagher*, Vol. 14, No. 4, pp. 971–1040 (1985) \$9.00
285. Reactivity of HO₂/O₂⁻ Radicals in Aqueous Solution, *Benon H. J. Bielski, Diane E. Cabelli, Ravindra L. Arudi, and Alberta B. Ross*, Vol. 14, No. 4, pp. 1041–1100 (1985) \$8.00
286. The Mark–Houwink–Sakurada Equation for the Viscosity of Atactic Polystyrene, *Herman L. Wagner*, Vol. 14, No. 4, pp. 1101–1106 (1985) \$5.00
287. Standard Chemical Thermodynamic Properties of Alkylcyclopentane Isomer Groups, Alkylcyclohexane Isomer Groups, and Combined Isomer Groups, *Robert A. Alberty and Young S. Ha*, Vol. 14, No. 4, pp. 1107–1132 (1985) \$6.00

Reprints from Volume 15

288. Triplet–Triplet Absorption Spectra of Organic Molecules in Condensed Phases, *Ian Carmichael and Gordon L. Hug*, Vol. 15, No. 1, pp. 1–250 (1986) \$20.00
289. Recommended Rest Frequencies for Observed Interstellar Molecular Microwave Transitions—1985 Revision, *F. J. Lovas*, Vol. 15, No. 1, pp. 251–303 (1986) \$8.00

(Continuation of Cumulative Listing of Reprints)

- 290 New International Formulations for the Thermodynamic Properties of Light and Heavy Water, *J. Kestin and J. V. Sengers*, Vol. 15, No. 1, pp. 305-320 (1986) \$5.00
- 291 Forbidden Lines in ns^2np^k Ground Configurations and $nsnp$ Excited Configurations of Beryllium through Molybdenum Atoms and Ions, *Victor Kaufman and Jack Sugar*, Vol. 15, No. 1, pp. 321-426 (1986) \$11.00
- 292 Thermodynamic Properties of Twenty-One Monocyclic Hydrocarbons, *O. V. Dorofeeva, L. V. Gurvich, and V. S. Jonsh*, Vol. 15, No. 2, pp. 437-464 (1986) \$6.00
- 293 Evaluated Kinetic Data for High-Temperature Reactions Volume 5 Part 1 Homogeneous Gas Phase Reactions of the Hydroxyl Radical with Alkanes, *D. L. Baulch, M. Bowers, D. G. Malcolm, and R. T. Tuckerman*, Vol. 15, No. 2, pp. 465-592 (1986) \$12.00
- 294 Thermodynamic Properties of Ethylene from the Freezing Line to 450 K at Pressures to 260 MPa, *Majid Jahangiri, Richard T. Jacobsen, Richard B. Stewart, and Robert D. McCarty*, Vol. 15, No. 2, pp. 593-734 (1986) \$13.00
- 295 Thermodynamic Properties of Nitrogen from the Freezing Line to 2000 K at Pressures to 1000 MPa, *Richard T. Jacobsen, Richard B. Stewart, and Majid Jahangiri*, Vol. 15, No. 2, pp. 735-909 (1986) \$15.00
- 296 A Critical Review of Aqueous Solubilities, Vapor Pressures, Henry's Law Constants, and Octanol-Water Partition Coefficients of the Polychlorinated Biphenyls, *Wan Ying Shiu and Donald Mackay*, Vol. 15, No. 2, pp. 911-929 (1986) \$6.00
- 297 Computer Methods Applied to the Assessment of Thermochemical Data Part I The Establishment of a Computerized Thermochemical Data Base Illustrated by Data for $TiCl_4(g)$, $TiCl_4(l)$, $TiCl_3(cr)$, and $TiCl_2(cr)$, *S. P. Kirby, E. M. Marshall, and J. B. Pedley*, Vol. 15, No. 3, pp. 943-965 (1986) \$6.00
- 298 Thermodynamic Properties of Iron and Silicon, *P. D. Desai*, Vol. 15, No. 3, pp. 967-983 (1986) \$6.00
- 299 Cross Sections for Collisions of Electrons and Photons with Nitrogen Molecules, *Y. Itikawa, M. Hayashi, A. Ichimura, K. Onda, K. Sakimoto, K. Takayanagi, M. Nakamura, H. Nishimura, and T. Takayanagi*, Vol. 15, No. 3, pp. 985-1010 (1986) \$6.00
- 300 Thermochemical Data on Gas-Phase Ion-Molecule Association and Clustering Reactions, *R. G. Keesee and A. W. Castleman, Jr.*, Vol. 15, No. 3, pp. 1011-1071 (1986) \$8.00
- 301 Standard Reference Data for the Thermal Conductivity of Liquids, *C. A. Nieto de Castro, S. F. Y. Li, A. Nagashima, R. D. Trengove, and W. A. Wakeham*, Vol. 15, No. 3, pp. 1073-1086 (1986) \$5.00
- 302 Chemical Kinetic Data Base for Combustion Chemistry Part I Methane and Related Compounds, *W. Tsang and R. F. Hampson*, Vol. 15, No. 3, pp. 1087-1279 (1986) \$17.00
- 303 Improved International Formulations for the Viscosity and Thermal Conductivity of Water Substance, *J. V. Sengers and J. T. R. Watson*, Vol. 15, No. 4, pp. 1291-1314 (1986) \$6.00
- 304 The Viscosity and Thermal Conductivity of Normal Hydrogen in the Limit of Zero Density, *M. J. Assael, S. Mixafendi, and W. A. Wakeham*, Vol. 15, No. 4, pp. 1315-1322 (1986) \$5.00
- 305 The Viscosity and Thermal Conductivity Coefficients of Gaseous and Liquid Argon, *B. A. Younglove and H. J. M. Hanley*, Vol. 15, No. 4, pp. 1323-1337 (1986) \$5.00
- 306 Standard Chemical Thermodynamic Properties of Alkyne Isomer Groups, *Robert A. Alberty and Ellen Burmenko*, Vol. 15, No. 4, pp. 1339-1349 (1986) \$5.00
- 307 Recent Progress in Deuterium Triple-Point Measurements, *L. A. Schwabe*, Vol. 15, No. 4, pp. 1351-1356 (1986) \$5.00
- 308 Rate Constants for Reactions of Radiation-Produced Transients in Aqueous Solutions of Actinides, *S. Gordon, J. C. Sullivan, and Alberta B. Ross*, Vol. 15, No. 4, pp. 1357-1367 (1986) \$5.00
- 309 Thermodynamic Properties of Key Organic Oxygen Compounds in the Carbon Range C_1 to C_4 . Part 2 Ideal Gas Properties, *Jing Chao, Kenneth R. Hall, Kenneth N. Marsh, and Randolph C. Wilhoit*, Vol. 15, No. 4, pp. 1369-1436 (1986) \$9.00

Reprints from Volume 16

- 310 Thermochemical Data on Gas Phase Compounds of Sulfur, Fluorine, Oxygen, and Hydrogen Related to Pyrolysis and Oxidation of Sulfur Hexafluoride, *John T. Herron*, Vol. 16, No. 1, pp. 1-6 (1987) \$5.00
- 311 The Thermochemical Measurements on Rubidium Compounds: A Comparison of Measured Values with Those Predicted from the NBS Tables of Chemical and Thermodynamic Properties, *V. B. Parker, W. H. Evans, and R. L. Nuttall*, Vol. 16, No. 1, pp. 7-59 (1987) \$8.00
- 312 Standard Thermodynamic Functions of Gaseous Polyatomic Ions at 100-1000 K, *Aharon Loewenschuss and Yitzhak Marcus*, Vol. 16, No. 1, pp. 61-89 (1987) \$6.00
- 313 Thermodynamic Properties of Manganese and Molybdenum, *P. D. Desai*, Vol. 16, No. 1, pp. 91-108 (1987) \$6.00
- 314 Thermodynamic Properties of Selected Binary Aluminum Alloy Systems, *P. D. Desai*, Vol. 16, No. 1, pp. 109-124 (1987) \$5.00
- 315 ^{13}C Chemical Shielding in Solids, *T. M. Duncan*, Vol. 16, No. 1, pp. 125-151 (1987) \$6.00
- 316 The Mark-Houwink-Sakurada Relation for Poly(Methyl Methacrylate), *Herman L. Wagner*, Vol. 16, No. 2, pp. 165-173 (1987) \$5.00
- 317 The Viscosity of Carbon Dioxide, Methane, and Sulfur Hexafluoride in the Limit of Zero Density, *R. D. Trengove and W. A. Wakeham*, Vol. 16, No. 2, pp. 175-187 (1987) \$5.00
- 318 The Viscosity of Normal Deuterium in the Limit of Zero Density, *M. J. Assael, S. Mixafendi, and W. A. Wakeham*, Vol. 16, No. 2, pp. 189-192 (1987) \$5.00
- 319 Standard Chemical Thermodynamic Properties of Alkane-thiol Isomer Groups, *Robert A. Alberty, Ellen Burmenko, Tae H. Kang, and Michael B. Chung*, Vol. 16, No. 2, pp. 193-208 (1987) \$5.00
- 320 Evaluation of Binary Excess Volume Data for the Methanol + Hydrocarbon Systems, *R. Srivastava and B. D. Smith*, Vol. 16, No. 2, pp. 209-218 (1987) \$5.00
- 321 Evaluation of Binary Excess Enthalpy Data for the Methanol + Hydrocarbon Systems, *R. Srivastava and B. D. Smith*, Vol. 16, No. 2, pp. 219-237 (1987) \$6.00
- 322 Extinction Coefficients of Triplet-Triplet Absorption Spectra of Organic Molecules in Condensed Phases: A Least-Squares Analysis, *Ian Carmichael, W. P. Helman, and G. L. Hug*, Vol. 16, No. 2, pp. 239-260 (1987) \$6.00
- 323 Evaluated Chemical Kinetic Data for the Reactions of Atomic Oxygen $O(^3P)$ with Unsaturated Hydrocarbons, *R. J. Cvetanović*, Vol. 16, No. 2, pp. 261-326 (1987) \$9.00
- 324 Spectral Data for Molybdenum Ions, $Mo\text{ VI-Mo XII}$, *Toshizo Shirai, Yohta Nakai, Kunio Ozawa, Keishi Ishii, Jack Sugar, and Kazuo Mori*, Vol. 16, No. 2, pp. 327-377 (1987) \$8.00

(Continuation of Cumulative Listing of Reprints)

| | | | |
|---|---------|--|---------|
| 325. Standard Chemical Thermodynamic Properties of Alkanol Isomer Groups, <i>Robert A. Alberty, Michael B Chung, and Theresa M Flood</i> , Vol. 16, No. 3, pp. 391-417 (1987) | \$6 00 | 334. Rate Data for Inelastic Collision Processes in the Diatomic Halogen Molecules. 1986 Supplement, <i>J. I. Steinfeld</i> , Vol. 16, No 4, pp. 903-910 (1987) | \$5.00 |
| 326. High-Temperature Vaporization Behavior of Oxides II. Oxides of Be, Mg, Ca, Sr, Ba, B, Al, Ga, In, Tl, Si, Ge, Sn, Pb, Zn, Cd, and Hg, <i>R. H. Lamoreaux, D. L. Hildenbrand, and L. Brewer</i> , Vol. 16, No 3, pp. 419-443 (1987) | \$6.00 | 335. Critical Survey of Data on the Spectroscopy and Kinetics of Ozone in the Mesosphere and Thermosphere, <i>Jeffrey I Steinfeld, Steven M Adler-Golden, and Jean W. Gallagher</i> , Vol 16, No 4, pp. 911-951 (1987) | \$7 00 |
| 327. Equilibrium and Transport Properties of Eleven Polyatomic Gases at Low Density, <i>A. Boushehr, J Bzowski, J. Kestin, and E. A. Mason</i> , Vol. 16, No. 3, pp. 445-466 (1987) | \$6.00 | 336. Critical Compilation of Surface Structures Determined by Low-Energy Electron Diffraction Crystallography, <i>Philip R. Watson</i> , Vol. 16, No 4, pp. 953-992 (1987) | \$7 00 |
| 328. The Thermochemistry of Inorganic Solids IV. Enthalpies of Formation of Compounds of the Formula MX_aY_b , <i>Mohamed W. M. Hisham and Sidney W. Benson</i> , Vol. 16, No. 3, pp. 467-470 (1987) | \$5.00 | 337. Viscosity and Thermal Conductivity of Nitrogen for a Wide Range of Fluid States, <i>K. Stephan, R Krauss, and A Laesecke</i> , Vol 16, No. 4, pp. 993-1023 (1987) | \$6 00 |
| 329. Chemical Kinetic Data Base for Combustion Chemistry. Part 2. Methanol, <i>Wing Tsang</i> , Vol. 16, No. 3, pp. 471-508 (1987) | \$7.00 | Reprints from Volume 17 | |
| 330. Phase Diagrams and Thermodynamic Properties of the 70 Binary Alkali Halide Systems Having Common Ions, <i>James Sangster and Arthur D Pelton</i> , Vol. 16, No. 3, pp. 509-561 (1987) | \$8 00 | 338. Pressure and Density Series Equations of State for Steam as Derived from the Haar-Gallagher-Kell Formulation, <i>R. A. Dobbins, K Mohammed, and D. A. Sullivan</i> , Vol 17, No 1, pp 1-8 (1988) | \$5 00 |
| 331. Thermophysical Properties of Fluids. II. Methane, Ethane, Propane, Isobutane, and Normal Butane, <i>B. A. Younglove and J. F. Ely</i> , Vol. 16, No. 4, pp. 577-798 (1987) | \$18.00 | 339. Absolute Cross Sections for Molecular Photoabsorption, Partial Photoionization, and Ionic Photofragmentation Processes, <i>J W. Gallagher, C E Bron, J. A. R Samson, and P W Langhoff</i> , Vol 17, No. 1, pp. 9-153 (1988) | \$14.00 |
| 332. Methanol Thermodynamic Properties from 176 to 673 K at Pressures to 700 Bar, <i>Robert D. Goodwin</i> , Vol. 16, No. 4, pp. 799-892 (1987) | \$10.00 | 340. Energy Levels of Molybdenum, Mo_I through Mo_{XVII} , <i>Jack Sugar and Arlene Musgrove</i> , Vol. 17, No 1, pp 155-239 (1988) | \$10 00 |
| 333. International Equations for the Saturation Properties of Ordinary Water Substance, <i>A. Saul and W. Wagner</i> , Vol. 16, No. 4, pp. 893-901 (1987) | \$5.00 | 341. Standard Chemical Thermodynamic Properties of Polycyclic Aromatic Hydrocarbons and Their Isomer Groups I Benzene Series, <i>Robert A. Alberty and Andrea K. Reif</i> , Vol 17, No 1, pp. 241-253 (1988) | \$5.00 |

Special Reprints Packages

These special reprints packages offer selected articles in specific subject areas from the JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA, and they are offered at a better rate than when purchased individually. You will have available a complete library of literature for your specific requirements at a fraction of the cost of purchasing back issues of the journal.

Look over the reprints packages available—they are listed by subject area. In the Cumulative Listing of Reprints you will find the titles corresponding to the reprint numbers. You are sure to find building your information bank in this manner to be thorough and economical.

Package C1 (5 Parts) MOLECULAR VIBRATIONAL FREQUENCIES. Consisting of Reprint Nos. 103, 129, 170, 257, NSRD 39.
If purchased individually: \$ 33.00
Special package price: \$ 26.00

Package C2 (22 Parts) ATOMIC ENERGY LEVELS. Consisting of Reprint Nos. 26, 54, 64, 68, 94, 100, 109, 125, 126, 131, 132, 149, 150, 154, 156, 160, 179, 180, 192, 200, 222, 278.
If purchased individually: \$121.00
Special package price: \$ 96.00

Package C3 (6 Parts) ATOMIC SPECTRA. Consisting of Reprint Nos. 33, 56, 77, 78, 110, 132.
If purchased individually: \$ 33.00
Special package price: \$ 27.00

Package C4 (5 Parts) ATOMIC TRANSITION PROBABILITIES. Consisting of Reprint Nos. 20, 63, 82, 118, 182.
If purchased individually: \$ 35.00
Special package price: \$ 28.00

Package C5 (7 Parts) MOLECULAR SPECTRA. Consisting of Reprint Nos. 4, 8, 53, 79, 93, 130, 146.
If purchased individually: \$ 51.50
Special package price: \$ 41.00

Package C6 (9 Parts) THERMODYNAMIC PROPERTIES OF ELECTROLYTE SOLUTIONS. Consisting of Reprint Nos. 15, 95, 111, 151, 152, 174, 184, 185, 186.
If purchased individually: \$ 46.00
Special package price: \$ 37.00

Package C7 (12 Parts) IDEAL GAS THERMODYNAMIC PROPERTIES. Consisting of Reprint Nos. 30, 42, 43, 62, 65, 66, 70, 80, 83, 113, 115, 141.
If purchased individually: \$ 38.00
Special package price: \$ 31.00

Package C8 (7 Parts) RESISTIVITY. Consisting of Reprint Nos. 138, 139, 155, 221, 258, 259, 260.
If purchased individually: \$ 47.50
Special package price: \$ 39.00

Package C9 (7 Parts) MOLTEN SALTS. Consisting of Reprint Nos. 10, 41, 71, 96, 135, 167, 168.
If purchased individually: \$ 62.50
Special package price: \$ 44.00

Package C10 (4 Parts) REFRACTIVE INDEX. Consisting of Reprint Nos. 81, 158, 162, 240.
If purchased individually: \$ 32.50
Special package price: \$ 26.00

Supplements to JPCRD

When the topic demands it, and the quality of the data justifies it, the JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA issues a special Supplement. Each Supplement is a monograph—collected tables of highly significant physical or chemical property data in one complete volume. Listed below are the special Supplements to JPCRD that have been published. Each is a valuable resource for the physical chemist and chemical physicist.

ATOMIC AND IONIC SPECTRUM LINES BELOW 2000 ANGSTROMS: HYDROGEN THROUGH KRYPTON, by Raymond L. Kelly. (Supplement No. 1 to Volume 16) 1987, 1689 pages, 3 volumes. Hardcover. U.S. & Canada: \$75.00
Abroad: \$90.00

ATOMIC ENERGY LEVELS OF THE IRON-PERIOD ELEMENTS: POTASSIUM THROUGH NICKEL by J. Sugar and C. Corliss. (Supplement No. 2 to Volume 14) 1985, 664 pages. Hardcover. U.S. & Canada: \$50.00
Abroad: \$58.00

JANAF THERMOCHEMICAL TABLES, Third Edition by M. W. Chase, Jr., C. A. Davies, J. R. Downey, Jr., D. J. Frurip, R. A. McDonald, and A. N. Syverud. (Supplement No. 1 to Volume 14) 1985, 1896 pages, 2 volumes. Hardcover. U.S. & Canada: \$130.00
Abroad: \$156.00

HEAT CAPACITIES AND ENTROPIES OF ORGANIC COMPOUNDS IN THE CONDENSED PHASE by E.S. Domalski, W.H. Evans, and E.D. Hearing. (Supplement No. 1 to Volume 13) 1984, 288 pages. Hardcover. U.S. & Canada: \$40.00
Abroad: \$48.00

THE NBS TABLES OF CHEMICAL THERMODYNAMIC PROPERTIES. SELECTED VALUES FOR INORGANIC AND C₁ AND C₂ ORGANIC SUBSTANCES IN SI UNITS by

D.D. Wagman, W.H. Evans, V.B. Parker, R.H. Schumm, I. Halow, S.M. Bailey, K.L. Churney, and R.L. Nuttall. (Supplement No. 2 to Volume 11) 1982, 394 pages. Hardcover. U.S. & Canada: \$40.00
Abroad: \$48.00

THERMOPHYSICAL PROPERTIES OF FLUIDS. 1. ARGON, ETHYLENE, PARAHYDROGEN, NITROGEN, NITROGEN TRIFLUORIDE, AND OXYGEN by B.A. Younglove. (Supplement No. 1 to Volume 11) 1982, 368 pages. Hardcover. U.S. & Canada: \$40.00
Abroad: \$48.00

EVALUATED KINETIC DATA FOR HIGH TEMPERATURE REACTIONS: VOLUME 4, HOMOGENEOUS GAS PHASE REACTIONS OF HALOGEN- AND CYANIDE-CONTAINING SPECIES by D.L. Baulch, J. Duxbury, S.J. Grant, and D.C. Montague. (Supplement No. 1 to Volume 10) 1981, 721 pages. Hardcover. U.S. & Canada: \$80.00
Abroad: \$96.00

THERMAL CONDUCTIVITY OF THE ELEMENTS: A COMPREHENSIVE REVIEW by C.Y. Ho, R.W. Powell, and P.E. Liley. (Supplement No. 1 to Volume 3) 1974, 796 pages.* U.S. & Canada: \$60/\$55
Abroad: \$72/\$66

PHYSICAL AND THERMODYNAMIC PROPERTIES OF ALIPHATIC ALCOHOLS by R.C. Wilhoit and B.J. Zwolinski. (Supplement No. 1 to Volume 2) 1973, 420 pages.* U.S. & Canada: \$33/\$30
Abroad: \$40/\$36

*Prices are for hardcover/softcover.

JANAF THERMOCHEMICAL TABLES

Third Edition

A Major Supplement from JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA

Presenting Reliable Data Utilized by Chemists, Chemical Engineers, and Materials Scientists from Around the World for Over 25 Years

JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA is very pleased to publish the Third Edition of the JANAF THERMOCHEMICAL TABLES.

Since the first version appeared 25 years ago, the JANAF THERMOCHEMICAL TABLES have been among the most widely used data tables in science and engineering.

You'll find:

- Reliable tables of thermodynamic properties of substances of wide interest
- A highly professional approach with critical evaluations of the world's thermochemical and spectroscopic literature
- A concise and easy-to-use format

This Third Edition presents an extensive set of tables including thermodynamic properties of more than 1800 substances, expressed in SI units. The notation has been made consistent with current international recommendations.

There is no other reference source of thermodynamic data that satisfies the needs of such a broad base of users.

Order your 2-volume set of the JANAF THERMOCHEMICAL TABLES today! You'll get over 1890 pages of valuable information that is crucial to your research—in two hardback volumes.

SUBSCRIPTION INFORMATION

The JANAF THERMOCHEMICAL TABLES, THIRD EDITION is a two-volume supplement of *Journal of Physical and Chemical Reference Data*.

1896 pages, 2 volumes, hardcover
ISBN 0-88318-473-7
Supplement Number 1 to Volume 14, 1985

U.S. & Canada \$130.00
All Other Countries \$156.00
(Postage included.)

All orders for supplements must be prepaid.

Foreign payment must be made in U.S. currency by international money order, UNESCO coupons, U S bank draft, or order through your subscription agency. For rates in Japan, contact Maruzen Co., Ltd. Please allow four to six weeks for your copy to be mailed.

For more information, write American Chemical Society, Marketing Communications Department, 1155 Sixteenth Street, NW, Washington, DC 20036.

In a hurry? Call TOLL FREE 800-227-5558 and charge your order!



Published by the American Chemical Society and the American Institute of Physics for the National Bureau of Standards

Editors:

M.W. Chase, Jr.
Nat'l Bureau of Standards

C.A. Davies
Dow Chemical U.S.A.

J.R. Downey, Jr.
Dow Chemical U.S.A.

D.J. Frurip
Dow Chemical U.S.A.

R.A. McDonald
Dow Chemical U.S.A.

A.N. Syverud
Dow Chemical U.S.A.

The Best
Source of
Quantitative
Numerical
Data of
Physics and
Chemistry
Published
Today!

JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA

Editor: David R. Lide
National Bureau of Standards

Published quarterly by the American Chemical Society and the American Institute of Physics, the *JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA* provides you with compilations and reviews produced under the National Standard Reference Data System (NSRDS) of the National Bureau of Standards.

Chemists, physicists, materials scientists, engineers, and information specialists can all benefit from the reliable information available in this journal. You'll find data on:

- Atomic and Molecular Science
- Chemical Kinetics
- Spectroscopy
- Thermodynamics
- Transport Phenomena
- Crystallography
- Materials Science
- And much more!

The *JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA* contains recommended values, uncertainty limits, critical commentary on methods of measurement, and full references to the original papers.

Join the thousands of professionals who rely on the *JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA* as a working tool for their research.

Backed by the ACS Guarantee, the *JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA* is guaranteed to be a reliable, up-to-date reference source—you won't want to miss a single issue!

1988 SUBSCRIPTION INFORMATION

JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA is published quarterly, one volume per year. Volume 17 (1988), ISSN: 0074-2689

| | U.S. | Canada & Mexico | Europe, Mideast & N. Africa* | Asia & Oceania* |
|------------------------|--------------------------------|--------------------------------|--------------------------------|--------------------------------|
| Individual (U.S. only) | <input type="checkbox"/> \$60 | <input type="checkbox"/> \$ 70 | <input type="checkbox"/> \$ 80 | <input type="checkbox"/> \$ 80 |
| Library/Institution | <input type="checkbox"/> \$265 | <input type="checkbox"/> \$275 | <input type="checkbox"/> \$285 | <input type="checkbox"/> \$285 |

* Includes postage per \$70 (members) and \$275 (nonmembers) to all countries. Postage included for personal use only. * Air service included.

To order your *JOURNAL OF PHYSICAL AND CHEMICAL REFERENCE DATA* subscription, call TOLL FREE 800-227-5558 and charge your order! (U.S. only) or outside the U.S. call (202) 872-4363, or write American Chemical Society, 1155 Sixteenth Street, NW, Washington, DC 20036, U.S.A.